



Cite this: *J. Anal. At. Spectrom.*, 2017, **32**, 1571

Quantitative determination of the sulfur-containing antioxidant ergothioneine by HPLC/ICP-QQQ-MS†

Nina Kroepfl, ^a Talke A. Marschall, ^b Kevin A. Francesconi, ^a Tanja Schwerdtle ^b and Doris Kuehnelt ^{a*}

Interest in the sulfur-containing antioxidant ergothioneine calls for reliable analytical methods for its quantification. In this work, a method based on reversed-phase high performance liquid chromatography (RP-HPLC) coupled with elemental mass spectrometry detection in mass shift mode (inductively coupled plasma triple quadrupole mass spectrometry, ICP-QQQ-MS) using oxygen as the reaction gas was developed for the element-selective determination of ergothioneine in complex biological matrices. Application of an instrumental setup using a 6-port-valve and the introduction of a methanol gradient allowed the time-efficient analysis of samples containing strongly retained sulfur species besides ergothioneine without compromising ICPMS detection. In aqueous solution, limits of detection and quantification (LOD and LOQ) of the optimized method for m/z 32 \rightarrow 48 (SO^+) were 0.23 $\mu\text{g S per L}$ and 0.80 $\mu\text{g S per L}$, respectively; measurements in a complex matrix (human hepatocyte carcinoma cells, HepG2) resulted in an LOD of 0.6 $\mu\text{g S per L}$ and an LOQ of 2.3 $\mu\text{g S per L}$. Recoveries of ergothioneine from cell pellets spiked with the analyte before cell lysis ($97 \pm 3\%$) matched those obtained for cell culture medium spiked before syringe filtration ($96 \pm 9\%$) demonstrating that sample preparation did not impair the quantitative determination of ergothioneine. When HepG2 cells were exposed to ergothioneine via the culture medium, they showed low absorption; approximately 3% of the added ergothioneine was found in cell lysates, while most of it ($\geq 85\%$) remained in the cell culture medium. The method is capable of separating ergothioneine from other biologically relevant sulfur-containing species and is expected to be of broad future use. Furthermore, the potential use for the simultaneous separation of selenium species, thereby extending the scope of possible applications, was demonstrated by applying it to water extracts of oyster mushrooms.

Received 23rd January 2017
Accepted 26th June 2017

DOI: 10.1039/c7ja00030h

rsc.li/jaas

Introduction

Ergothioneine, a natural sulfur-containing histidine derivative, was first isolated in 1909 from *Claviceps purpurea*, an ergot fungal infection of rye grain.¹ Synthesized only by non-yeast-like fungi, bacteria of the order Actinomycetales, and cyanobacteria,^{2–4} ergothioneine is taken up by plants and animals through their roots and diet.^{5–7} The interest in ergothioneine derives from its antioxidant activity and cytoprotective effects,

demonstrated by various *in vitro* studies;⁸ further investigations into the effects of ergothioneine call for analytical methods to accurately quantify this sulfur species.

Quantitative determination of ergothioneine in biological samples is usually performed by HPLC with detection by either molecular mass spectrometry^{4,9–12} or UV absorbance.^{13–19} Element-selective detection by ICPMS represents an interesting alternative to these techniques as it is robust, sensitive, and allows the recording of the complete speciation of an element in a sample. Although due to these advantages HPLC/ICPMS is a common technique for elemental speciation analysis it has not been widely applied to sulfur species, mainly because of isobaric interferences on the main sulfur isotope $^{32}\text{S}^+$. Approaches applied to reduce these interferences include the use of high resolution ICPMS, reaction/collision cell technology,²⁰ or ICP-triple quadrupole-MS (ICP-QQQ-MS),^{21,22} with the last approach being particularly promising.

For the measurement of S by ICP-QQQ-MS, the first quadrupole selects ions of the target mass (m/z 32), which then pass through to the oxygen-charged reaction cell. Here, the analyte

^aInstitute of Chemistry, Analytical Chemistry, NAWI Graz, University of Graz, Universitaetsplatz 1, 8010 Graz, Austria. E-mail: doris.kuehnelt@uni-graz.at; Tel: +43 316 380 5316

^bInstitute of Nutritional Science, University of Potsdam, Arthur-Scheunert-Allee 114-116, 14558 Nuthetal, Germany

† Electronic supplementary information (ESI) available: Examples of external calibration plots for the quantitative determination of ergothioneine, HPLC/ICP-QQQ-MS chromatograms of native and spiked 1 + 199 dilutions of an ergothioneine-treated cell culture medium and high resolution mass spectra of an aqueous ergothioneine standard solution, an ergothioneine-treated cell pellet and a water extract of oyster mushrooms. See DOI: 10.1039/c7ja00030h



ions yield SO^+ ions, with a mass shift to m/z 48, which are then selected for by the third quadrupole. Thus in comparison with single quadrupole/reaction cell systems, the triple quadrupole system provides great selectivity and lower background counts, which translates into better detection limits for sulfur.²³ ICP-QQQ-MS has been so far mainly applied to the determination of sulfur in the field of protein quantification,^{23–27} with samples including solutions of sulfur-containing compounds,^{23,25} DNA–protein cross-linking reaction products,²⁴ human plasma^{25,26} and snake venom.²⁷

We present here a HPLC/ICP-QQQ-MS method for the quantitative determination of the health-relevant sulfur compound ergothioneine and its application to cell culture media and cell pellets. Reversed-phase chromatography was used to separate ergothioneine from other biologically relevant sulfur-containing species, and we applied a methanol gradient in combination with a 6-port-valve for the time-efficient removal of strongly retained sulfur species without compromising the stability of the ICP. With a view to future investigations comparing the biological chemistry of sulfur and selenium, we also demonstrate the potential of the method to provide simultaneous measurements of selenium and sulfur species by a first analysis of aqueous extracts of oyster mushrooms.

Experimental

Chemicals, reagents and standards

Solutions were prepared from analytical grade chemicals and Milli-Q water (18.2 MΩ cm) throughout. Ammonium formate ($\geq 95\%$), ammonia solution (25%, p.a.) and hydrochloric acid (fuming, 37%, p.a.) were obtained from Carl Roth GmbH + Co. KG (Karlsruhe, Germany). Methanol (HPLC gradient grade) and acetonitrile (HPLC grade) were purchased from VWR International (Fontenay-sous-Bois, France) and Chem Lab NV (Zedelgem, Belgium), respectively. Glutathione reduced ($\geq 98\%$) and glutathione oxidized ($\geq 98\%$) were obtained from Carl Roth GmbH + Co. KG. L-Ergothioneine ($\geq 98\%$) and L-cysteine (99+%) were purchased from Tetrahydron (Paris, France) and Acros Organics (Geel, Belgium), respectively. L-Cystine (puriss.), D,L-methionine ($\geq 99.0\%$) and D,L-selenomethionine ($>99\%$) were obtained from Fluka (Buchs, St Gallen, Switzerland). D,L-Selenocystine was purchased from Sigma Aldrich (Karlsruhe, Germany). The selenosugars methyl-2-acetamido-2-deoxy-1-seleno- β -D-galactopyranoside (selenosugar 1), methyl-2-acetamido-2-deoxy-1-seleno- β -D-glucopyranoside (selenosugar 2) and methyl-2-amino-2-deoxy-1-seleno- β -D-galactopyranoside (selenosugar 3) were synthesized in-house as previously reported.^{28,29} Sulfur and selenium standards were prepared by dilution of aqueous stock standard solutions of the respective sulfur or selenium compounds with water. To prepare stock solutions of selenocystine and cystine, total dissolution was ensured by adding 5 (selenocystine) or 10 (cystine) drops of a HCl solution (37% HCl diluted 1 + 1 with water) to the compounds before adding water.

Basic cell culture medium (minimum essential medium (MEM) (Eagle)) and non-essential amino acids (NEAs) were from Biochrom (Berlin, Germany). Penicillin/streptomycin (PS) and trypsin were obtained from Sigma Aldrich (Steinheim, Germany)

and fetal calf serum (FCS) was purchased from PAA Laboratories (Cölbe, Germany). Cell culture medium used for cell experiments consisted of 440 mL basic cell culture medium, 50 mL FCS, 10 mL PS and 5 mL NEA. The mobile phase (20 mM aqueous ammonium formate, pH 9.0) for the HPLC separation of sulfur and selenium species was prepared by weighing in the appropriate amount of ammonium formate, dissolving it in water, and adjusting the pH to 9.0 by addition of ammonia solution.

Samples

Human hepatocyte carcinoma cells (HepG2, liver cells) were obtained from the European Collection of Cell Cultures (ECACC; number 85011430, Salisbury, UK). As described earlier³⁰ HepG2 cell culturing was performed in plastic culture dishes as monolayers in supplemented MEM at 37 °C, 5% CO_2 and 100% atmospheric humidity. Every two to three days, cells were split using 0.25% trypsin. For the preparation of cell pellets 25 000 cells per cm^2 were seeded. After three days of proliferation, cells were incubated with 1 mM ergothioneine for 1 h. Control cells remained untreated. Subsequently, cell culture medium was removed, and cells were washed with 37 °C warm phosphate-buffered saline (PBS) and detached with trypsin at 37 °C. Thereafter, cells were suspended in ice-cold PBS and the cell volume and cell number were measured using an automated cell counting system (Casy TTCs, Roche Innovatis AG, Bielefeld, Germany). Following centrifugation (5 min, 4 °C, $300 \times g$), the supernatant was removed and the cell pellet was carefully washed with ice-cold PBS. The final cell pellet was obtained by one more centrifugation (5 min, 4 °C, $600 \times g$) and removal of the supernatant. The cell pellets and cell culture medium either native (control medium and pellets) or incubated with 1 mM (32 mg S per L) ergothioneine for one hour (ergothioneine-treated medium and pellets) were prepared at the University of Potsdam, Germany, and sent to Graz, Austria, on dry ice, where they were immediately stored at -80 °C until analysis.

Muscle from fresh tuna (*Thunnus albacares* from the Indian Ocean, stored on ice) for the preparation of a fraction naturally containing 2-selenyl- $N_{\alpha},N_{\alpha},N_{\alpha}$ -trimethyl-L-histidine (selenoneine) was purchased from a local fish supplier. Oyster mushrooms were obtained from two local commercial suppliers (referred to as “mushroom sample 1” and “mushroom sample 2”) and stored in the refrigerator at 4 °C before sample preparation.

Sample preparation

Samples stored at -80 °C were thawed and subjected to further sample preparation procedures.

Cell culture medium. Thawed cell culture medium was filtered through 0.2 μm syringe filters with a nylon membrane (VWR International, Vienna, Austria). After syringe filtration, control cell culture medium was diluted 1 + 1 and ergothioneine-treated cell culture medium was diluted 1 + 199 with water.

Cell pellets. Cell pellets (control pellets or ergothioneine-treated pellets) were subjected to cell lysis by adding 1.0 mL of cold water, mixing, letting the sample sit on ice for 10



minutes, mixing again, and centrifuging for 10 minutes at 4 °C and $21\,380 \times g$ (D3024R High Speed Refrigerated Micro-Centrifuge, SCIOLOGEX, LLC., Rocky Hill, Connecticut, USA). After centrifugation, the supernatant was filtered through a syringe filter (0.2 µm nylon membrane).

Tuna muscle. A fraction containing selenoneine was collected from a water extract of fresh tuna muscle as previously described.³¹ Aliquots of 1.0 mL of the fraction were evaporated to dryness, stored at −80 °C and re-dissolved by adding 1.0 mL of water after thawing.

Oyster mushrooms. Mushrooms were manually cleaned of soil, chopped into small pieces using a kitchen knife and transferred to a plastic bag. The plastic bag and contents was placed briefly in liquid nitrogen, and the frozen mushrooms were then transferred to a commercially available kitchen grinder (Moulinex) for mincing. For the simultaneous extraction of water soluble sulfur and selenium species portions of homogenized mushrooms (1 g) were dispersed in 10 mL of water and sonicated for 4 × 5 min in an ultrasonic bath (Transsonic 700 H, Elma, Singen, Germany; ultrasonic power RMS 160 W, ultrasonic maximum peak power 640 W). Samples were manually shaken in between the extraction cycles. The extracts were centrifuged for 10 min at $3756 \times g$ and filtered through 0.2 µm syringe filters (nylon membrane).

Determination of sulfur and selenium species by HPLC/ICP-QQQ-MS

Instrumentation. Ergothioneine determination was performed by HPLC/ICP-QQQ-MS. Separation of ergothioneine from other sulfur-containing species was carried out using a combination of an Agilent 1100 and an Agilent 1200 HPLC system (Agilent Technologies, Waldbronn, Germany) consisting of a degasser (G 1379A), a quaternary pump (G 1311A), an autosampler (G 1367C) with a thermostat (G 1330B), and a thermostatted column compartment (G 1316A) equipped with a 6-port-valve. Sulfur and selenium species were separated on a reversed-phase column (Gemini C6-Phenyl 4.6 × 150 mm; Phenomenex, Torrance, USA) protected with a matching guard column (Security Guard cartridge AJ0-7915, Phenomenex). The HPLC system was connected directly to the MicroMist nebulizer of an 8800-ICP-QQQ-MS system (Agilent Technologies) via PEEK capillary tubing (i.d. 0.125 mm). The ICP-QQQ-MS was operated in oxygen mode, using O₂ at a flow rate of ca. 0.3 mL min^{−1} as a cell gas. Sulfur and selenium were monitored in mass shift mode following their reaction with oxygen forming the analyte ions SO⁺ (m/z 32 → 48 and m/z 34 → 50) and SeO⁺ (m/z 77 → 93, m/z 78 → 94, m/z 80 → 96 and m/z 82 → 98) with integration times of 0.1 s and 0.3 s for sulfur- and selenium-containing ions, respectively. Typical ICP-QQQ-MS settings are given in Table 1. ICP-QQQ-MS settings were optimized before each period of measurements using a 1 µg L^{−1} solution of Li, Y, Tl, Ce, Co, Fe, Cr, Se, As in 2% HNO₃. Quantification of ergothioneine was performed by external calibration using m/z 32 → 48 (for external calibration plots see ESI Fig. S-1†). A sulfate standard containing 334 mg S per L (1000 mg SO₄^{2−} per L ± 0.2%; Carl Roth GmbH + Co. KG) was used for quality control (recovery:

109 ± 2% with $n = 3$). The instrumental drift was assessed by regular injection of an ergothioneine standard solution over the whole measurement sequence.

Procedures. Cell pellets and aqueous extracts of oyster mushrooms were analyzed by isocratic elution on the reversed-phase column (Table 1). Owing to strongly retained sulfur-containing species present in the cell culture medium, isocratic elution resulted in analysis times of up to two hours per sample. To reduce the analysis time, the use of a 6-port-valve and the application of a methanol gradient were tested. The method was adapted from the method introduced by Kokarnig *et al.* for the determination of selenium metabolites in human urine.³² The gradient was optimized in terms of flow rates, holding time of the highest methanol content and re-equilibration time to reduce the stress put on the chromatographic column. Flow rates between 0.4 and 0.7 mL min^{−1} (steps of 0.1 mL min^{−1}) were tested, finding a flow rate of 0.5 mL min^{−1} to be a suitable compromise between fast analysis and reduced pressure fluctuations. Optimized gradient elution was performed as shown in Table 1, with the mobile phase, injection volume and column temperature being analogous to isocratic elution.

Analytical figures of merit

Limit of detection (LOD) and limit of quantification (LOQ).

The LOD and LOQ of the developed analytical method were determined according to DIN 32645 (calibration method) using aqueous ergothioneine standards in the concentration range between 1 and 10 µg S per L (4 independent dilutions of each standard). Integrated peak areas of ergothioneine obtained from the m/z 32 → 48 signals were used to calculate the LOD and the LOQ ($k = 3$). To estimate LOD and LOQ in a real matrix a HepG2 control cell pellet was spiked with 2.5 µg S per L ergothioneine and analyzed ten times. LOD and LOQ were calculated according to Eurachem³³ from the three- and ten-fold standard deviation, respectively, of the ten injections.

Spiking recoveries. Spiking experiments in real matrix were performed with both cell pellets and cell culture medium. Spiking of cell pellets was performed either before or after cell lysis. For spiking before cell lysis, frozen cell pellets were spiked with ergothioneine to a final concentration of 100 µg S per L after thawing and addition of water. When cell pellets were spiked after lysis, pellets with added water were mixed thoroughly after sitting on ice and 0.5 mL of the lysate was removed for spiking with ergothioneine (spiked concentrations of ergothioneine: control cell pellets, 2.5 and 10 µg S per L and ergothioneine-treated cell pellets, 20 µg S per L). Native and spiked lysates of cell pellets were then subjected to centrifugation and filtration as described above.

Cell culture medium was spiked with ergothioneine either before or after syringe filtration. Spiking concentrations ranged between 2.5 µg S per L and 300 µg S per L ergothioneine for control and ergothioneine-treated cell culture medium.

HPLC-electrospray-Orbitrap-MS identification of ergothioneine. For the identification of ergothioneine in a cell pellet lysate and a mushroom extract by high resolution MS a Dionex Ultimate 3000 (Thermo Scientific, Waltham, USA) HPLC system



Table 1 HPLC and ICP-QQQ-MS settings including the optimized methanol gradient for the elution of strongly retained sulfur species present in cell culture medium

Typical ICP-QQQ-MS settings			
RF power	1550 W	Spray chamber temperature	2 °C
RF matching	1.88 V	Extract 1	0.0 V
Sampling depth	8.0 mm	Extract 2	−150.0 V
Carrier gas flow	0.97 L min ^{−1}	Omega bias	−90 V
Makeup gas flow	0.13 L min ^{−1}	Omega lens	8.0 V
Q1 bias	−3.0 V	Q1 entrance	−6.1 V
Q1 prefilter bias	−22.0 V	Q1 exit	0.0 V
Q1 postfilter bias	−20.0 V	Cell focus	0.0 V
Cell gas (O ₂ in Ar)	30%	Cell entrance	−50 V
Octopole bias	−5.0 V	Cell exit	−60 V
Octopole RF	191 V	Deflect	4.0 V
Energy discriminator	−7.0 V	Plate bias	−60 V
HPLC settings for isocratic elution			
Mobile phase	20 mM ammonium formate, pH 9.0	Flow rate	0.7 mL min ^{−1}
Injection volume	10 µL	Column temperature	30 °C
HPLC settings for gradient elution			
Eluent A	20 mM ammonium formate, pH 9.0	Eluent B	Methanol
Gradient	Time [min]	Eluent B [%]	Flow rate [mL min ^{−1}]
	0–9	0	0.7
	9–11	0–80	0.5
	11–18	80	0.5
	18–20	80–0	0.5
	20–24	0	0.5
	24–25	0	0.5–0.7
	25–31	0	0.7
6-Port-valve	0–9 min	Position 1; effluent from column to ICP	
	9–29 min	Position 2; effluent from column to waste	
	29–31 min	Position 1; effluent from column to ICP	

consisting of a Rapid Separation (RS) pump, RS autosampler and RS column compartment was coupled to a Q Exactive Orbitrap Mass Spectrometer (Thermo Scientific, Waltham, USA) equipped with a heated electrospray ionization (HESI) source from the same manufacturer.

HPLC conditions were the same as for HPLC/ICP-QQQ-MS. MS was performed in the positive mode and source settings were: gas temperature 465 °C, flow rates 58 (sheath) and 17 (aux) instrument units, spray voltage 3500 V and capillary temperature 285 °C. The full scan range was m/z 200–600 and a resolution of 70 000 (full width half-maximum, FWHM), an automatic gain control (AGC) target of 3×10^6 , and a maximum injection time (IT) of 200 ms were used.

MS/MS for the precursor ion at m/z 230 was performed after recording of full scans. A collision energy of 20–30–35 (stepped) instrument units was used. MS/MS settings were: maximum injection time 50 ms, AGC target 1×10^5 , resolution 17 500 FWHM. An isolation window of 4 (offset 1) Thomson (mass/charge) was selected.

Results and discussion

Determination of sulfur and selenium species by HPLC/ICP-QQQ-MS

Element-selective detection by ICPMS is sensitive and robust towards matrix effects, which is advantageous with respect to quantification, and additionally offers the possibility of obtaining a complete picture of the speciation of an element or even several elements in one run. The triple quadrupole system enables the detection of sulfur despite spectral interferences on the main sulfur isotopes (m/z 32 and m/z 34) by using oxygen as the reaction gas and operating the instrument in mass shift mode. Oxygen readily reacts with sulfur, shifting the analyte ion from S^+ to SO^+ . Because selenium behaves similarly forming the SeO^+ ion, the mass shift mode using oxygen as a reaction gas is applicable to the simultaneous determination of selenium and sulfur species. With respect to the determination of ergothioneine, we expected ICP-QQQ-MS detection to be more selective and sensitive than UV detection and more robust towards matrix effects than molecular MS detection.



We performed preliminary tests investigating various separation mechanisms (reversed-phase, cation-exchange, hydrophilic interaction liquid chromatography) to separate several biologically relevant sulfur and selenium species. The various functional groups in the target molecule ergothioneine and its selenium analogue selenoneine can be deprotonated or protonated depending on the pH of the mobile phase, but at pH 9.0 both have a net charge of zero making them amenable to reversed-phase chromatography. Thus, the Gemini C6-Phenyl column, chosen because of its stability over the wide pH range (pH 1 to 12), in combination with a mobile phase of 20 mM ammonium formate at pH 9.0 was capable of separating ergothioneine from the other biologically relevant sulfur-containing species tested (Fig. 1). The same chromatographic

conditions separated selenoneine from all of the tested selenium species (Fig. 2).

Analysis of cell culture matrices

As *in vitro* studies into the fate and action of ergothioneine are amongst the potential applications for the developed HPLC/ICP-QQQ-MS method, we employed it for the analysis of human hepatocyte carcinoma cells (HepG2) and the corresponding cell culture medium.

Several sulfur species were detected in the cell culture medium (Fig. 3a). Peak identification for the two sulfur-containing constituents thiamine and phenol red of the cell culture medium was not attempted due to the lack of available standard

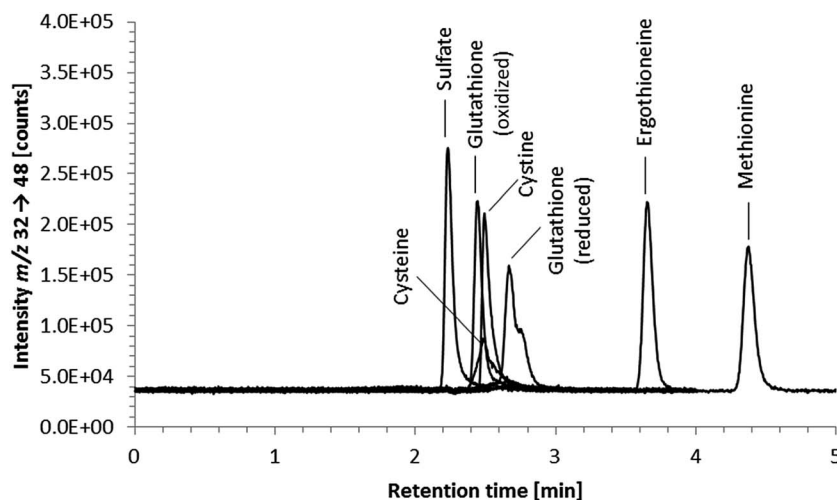


Fig. 1 HPLC/ICP-QQQ-MS chromatograms of 100 μg S per L standard solutions of various biologically relevant sulfur-containing species; chromatographic conditions as follows: Gemini C6-Phenyl 4.6 \times 150 mm; column temperature, 30 $^{\circ}\text{C}$; injection volume, 10 μL ; mobile phase, 20 mM ammonium formate pH 9.0; flow rate, 0.7 mL min^{-1} .

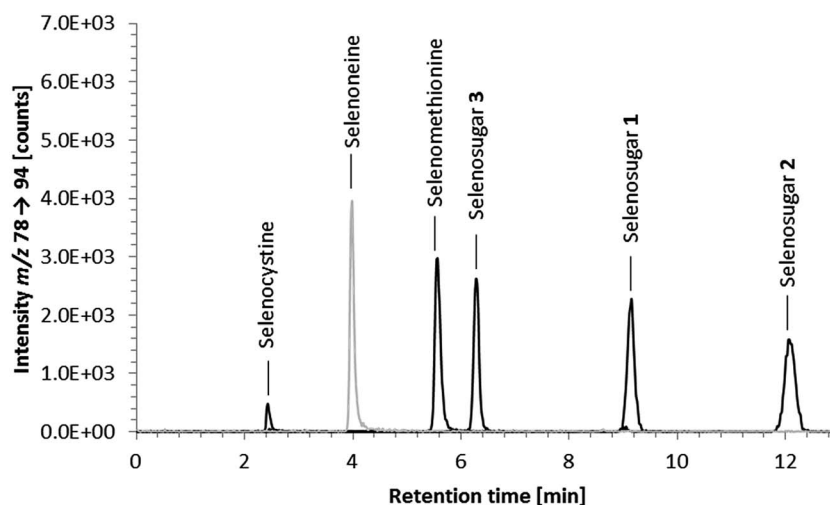


Fig. 2 HPLC/ICP-QQQ-MS chromatograms of 20 μg Se per L standard solutions of various biologically relevant selenium-containing species and a fraction from a water extract of tuna muscle containing selenoneine; 20 μg Se per L standard solutions of selenium-containing species (solid black line) and fraction containing selenoneine (solid light grey line); chromatographic conditions as follows: Gemini C6-Phenyl 4.6 \times 150 mm; column temperature, 30 $^{\circ}\text{C}$; injection volume, 10 μL ; mobile phase, 20 mM ammonium formate pH 9.0; flow rate, 0.7 mL min^{-1} .



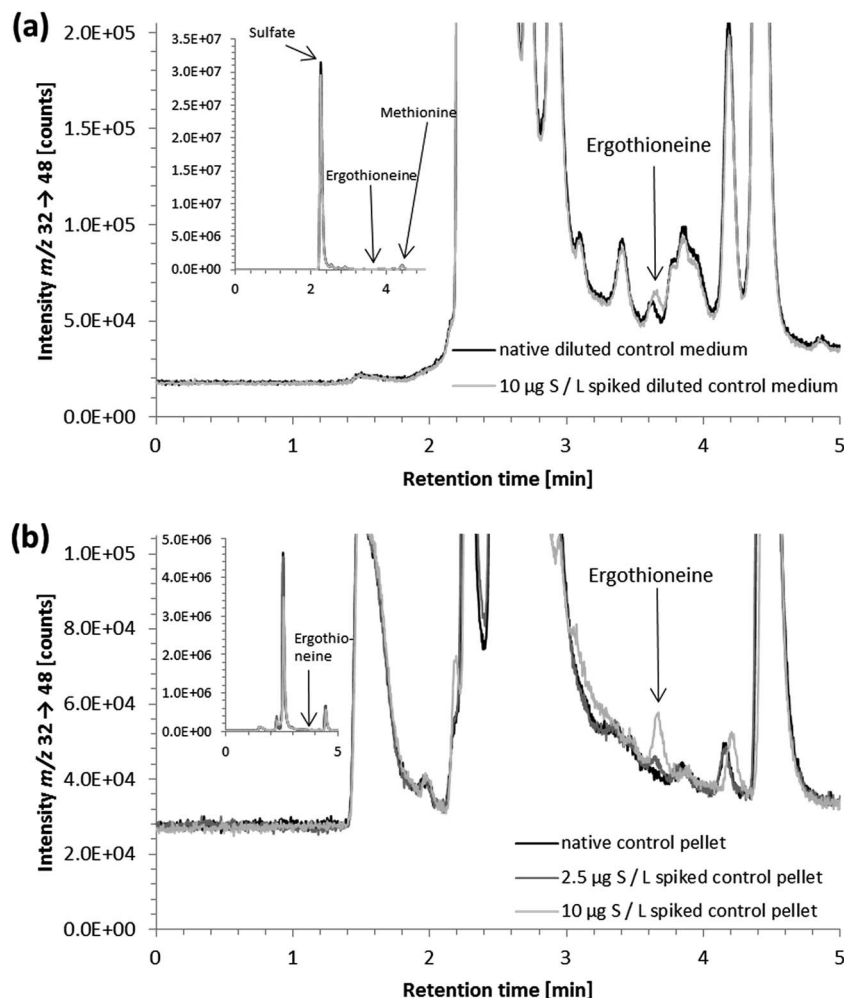


Fig. 3 HPLC/ICP-QQQ-MS chromatograms of (a) native and spiked 1 + 1 dilution of control cell culture medium (a retention time shift upon spiking indicated that another sulfur-containing species eluting close to ergothioneine was present in the cell culture medium at a very low concentration) and (b) native and spiked lysate of a control cell pellet (recoveries for control cell pellets spiked with 10 or 2.5 $\mu\text{g S per L}$ ergothioneine: $96 \pm 1\%$ and $88 \pm 2\%$ ($n = 3$ in both cases), respectively); insets showing the whole chromatograms; native sample (solid black line), 2.5 $\mu\text{g S per L}$ ergothioneine spiked sample (solid dark grey line) and 10 $\mu\text{g S per L}$ ergothioneine spiked sample (solid light grey line); chromatographic conditions as follows: Gemini C6-Phenyl 4.6 \times 150 mm; column temperature, 30 $^{\circ}\text{C}$; injection volume, 10 μL ; mobile phase, eluent A: 20 mM ammonium formate pH 9.0 and eluent B: methanol (gradient elution using the 6-port-valve in the case of (a), isocratic elution with eluent A in the case of (b)); flow rate, (a) 0.5–0.7 mL min^{-1} and (b) 0.7 mL min^{-1} .

compounds. Besides sulfate and methionine contained in the minimum essential medium (Eagle), we observed the presence of a sulfur compound with a similar retention time to the ergothioneine standard (Fig. 3a). This signal was also detected in FCS, when analyzed separately (data not shown). However, due to the results of spiking experiments showing slight differences in the retention times of unspiked and spiked cell culture medium (Fig. 3a), it is doubtful that the control cell culture medium contained ergothioneine. Due to the complex matrix of the cell culture medium and the vast number of sulfur-containing species present in biological samples, this peak is likely to derive from another sulfur-containing species in the medium. As its concentration was low (approximately 10 $\mu\text{g S per L}$ when quantified against ergothioneine) it was negligible for the quantification of ergothioneine in the *in vitro* experiment discussed later.

Furthermore, strongly retained sulfur species with retention times up to two hours were detected. They were shown to result from the penicillin/streptomycin mixture that was an additive to the cell culture medium; their presence in the samples necessitated exceedingly long chromatographic runtimes. Neither the addition of acetonitrile (for protein precipitation) nor cut-off filtration adapted from previous procedures^{34–36} resulted in the removal of these species (data not shown). To overcome problems resulting from the strongly retained sulfur species, we then tried a different approach based on a method reported for the reversed-phase separation of selenium species applying a methanol gradient.³² Thus, a 6-port-valve was introduced to the instrumental setup directing the column effluent either to the ICP-QQQ-MS or to the waste. The column effluent was directed to the ICP-QQQ-MS for the first 9 minutes of analysis, before the methanol gradient was started and the effluent from



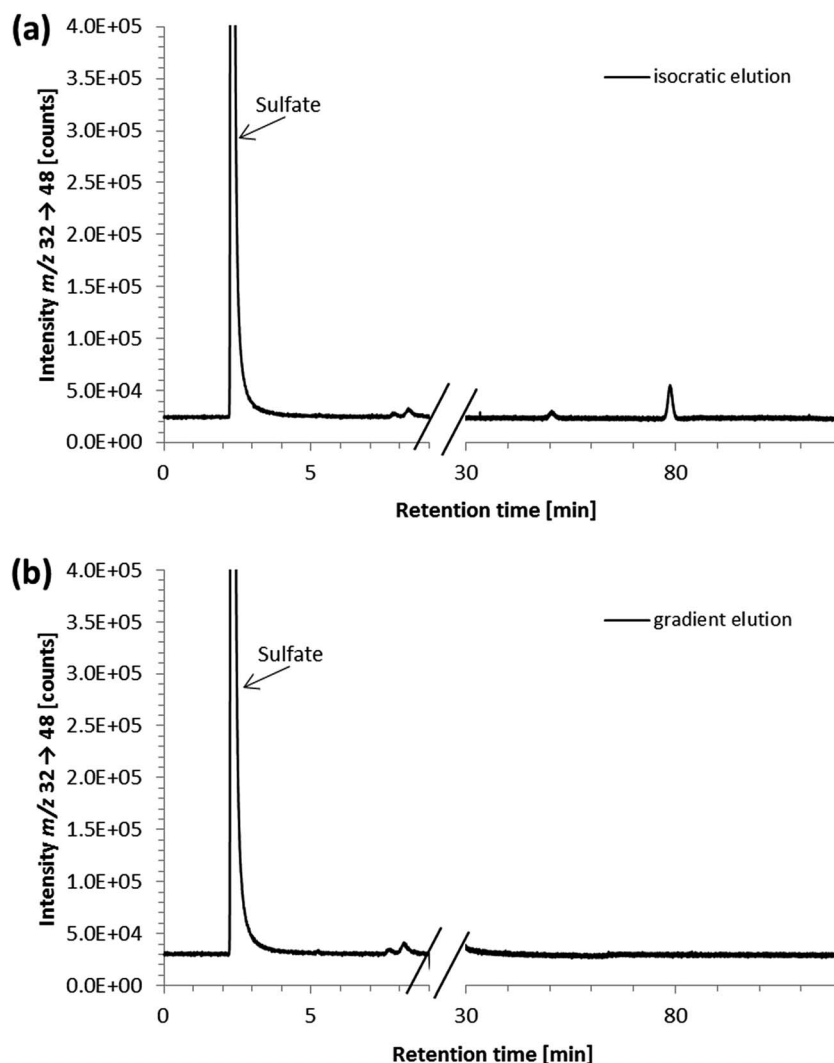


Fig. 4 HPLC/ICP-MS chromatograms of a 1 + 99 dilution of a penicillin/streptomycin (PS) cell culture additive subjected to HPLC separation with (a) isocratic elution or (b) gradient elution; the sulfate peak derived from streptomycin, present as streptomycin sulfate in the PS additive; to prove the removal of strongly retained sulfur species after application of the methanol gradient, signals of a penicillin/streptomycin cell culture additive were monitored for 1.5 hours after re-establishing starting conditions (100% 20 mM ammonium formate, pH 9.0 with a flow rate of 0.7 mL min^{-1}); chromatographic conditions as follows: Gemini C6-Phenyl $4.6 \times 150\text{ mm}$; column temperature, $30\text{ }^{\circ}\text{C}$; injection volume, $10\text{ }\mu\text{L}$; mobile phase, eluent A: 20 mM ammonium formate pH 9.0 and eluent B: methanol (isocratic elution with eluent A in the case of (a), gradient elution using the 6-port-valve in the case of (b)); flow rate, (a) 0.7 mL min^{-1} and (b) $0.5\text{--}0.7\text{ mL min}^{-1}$.

the column was directed to the waste to prevent the plasma from extinguishing due to the high amount of organic solvent. To avoid a high backpressure on the chromatographic column during gradient elution, the flow rate was reduced and the time for re-equilibration at the lower flow rate was optimized.

We then tested the applicability of the gradient elution method for determining ergothioneine in cell culture medium with the penicillin/streptomycin mixture containing the strongly retained sulfur species. No peaks were detected in the chromatograms obtained after gradient elution, ensuring the removal of the strongly retained sulfur-containing species with the methanol gradient from the column (Fig. 4). Both the peak area and the retention time of the ergothioneine peak were unchanged, compared to the isocratic elution method. This approach is also expected to be applicable to other matrices

containing strongly retained sulfur species. Thus, with the developed method using gradient elution and the 6-port-valve, it was possible to reduce the time for analysis of cell culture medium from 120 to 31 minutes per sample. This analysis time is still substantially longer than that of LC/MSMS methods designed for the determination of ergothioneine in food,^{9,11} but comparable to that of a method using UV detection.¹⁹

In lysates of the control cell pellets, HPLC/ICP-MS showed various sulfur-containing species between 1.3 and 5.0 minutes (Fig. 3b). Small peaks eluting after 5.0 minutes were also detected, nevertheless, chromatographic separation of sulfur species in lysates of cell pellets using isocratic elution was completed within 23 minutes obviating the need for gradient elution for the analysis of cell pellets. In contrast to the cell culture medium, no peaks were



detected near the retention time of ergothioneine in the control cell pellets. Identification of sulfur-containing species other than ergothioneine in cell pellets was not attempted.

Analytical figures of merit

Limit of detection (LOD) and limit of quantification (LOQ). Using the calibration method (DIN 32645), the limits of detection and quantification for ergothioneine obtained with our HPLC/ICP-QQQ-MS method were 0.23 $\mu\text{g S per L}$ and 0.80 $\mu\text{g S per L}$ for m/z 32 \rightarrow 48. The LOD was comparable to or slightly lower than those obtained for the determination of ergothioneine by LC/molecular MSMS^{9,11} or other sulfur containing species by LC-ICP-QQQ-MS^{23–26} and about 1 to 2 orders of magnitude lower compared to LC/UV methods^{14,17,19} for the determination of ergothioneine. However, LOD and LOQ values obtained with aqueous ergothioneine standards, as is commonly reported, do not incorporate the influence of matrix components, which were expected to be significant in case of the complex matrix of the cell samples. Spiking experiments with low concentrations of ergothioneine, down to 2.5 $\mu\text{g S per L}$, added to lysates of cell pellets were performed to check the practicality of the LOD and the LOQ determined with the calibration method (Fig. 3b).

Even with an elevated baseline in the region of ergothioneine elution, due to the sulfur-rich matrix of the cell pellet sample, the HPLC/ICP-QQQ-MS method was able to detect 2.5 $\mu\text{g S per L}$ ergothioneine spiked to the lysate of a control cell pellet with a recovery of $88 \pm 2\%$ ($n = 3$); the LOD and LOQ, estimated according to Eurachem,³³ were 0.6 $\mu\text{g S per L}$ and 2.3 $\mu\text{g S per L}$, respectively. Matrix influence was more severe in cell culture medium because of the presence of the sulfur compound with a similar retention time to the ergothioneine standard as discussed before.

Spiking recoveries. Possible loss of ergothioneine during the complex sample preparation procedure for the cells was investigated by determining the recovery of the analyte spiked to control cell pellets before cell lysis at a level of 100 $\mu\text{g S per L}$. Recoveries for pellets were found to be $97 \pm 3\%$ (3 spiked control cell pellets injected 3 times each), indicating no significant losses of the target analyte during sample preparation. Possible adsorption of ergothioneine to the filter membrane was also specifically tested at lower spiked levels (20 $\mu\text{g S per L}$ spiked after cell lysis) and shown to be negligible for the two ergothioneine-treated cell pellets investigated (recoveries: $104 \pm 1\%$ and $100 \pm 2\%$; $n = 3$ injections in both cases) (Fig. 5).

Sample preparation of cell culture medium (syringe filtration) was demonstrated not to compromise the quantitative determination of ergothioneine (recovery of 300 $\mu\text{g S per L}$ ergothioneine spiked to control cell culture medium: $96 \pm 9\%$, 3 dilutions of 3 spiked control cell culture media; for chromatograms of spiked ergothioneine-treated cell culture medium see ESI Fig. S-2†).

Applications

Uptake of ergothioneine in HepG2 cells. Since ergothioneine is of health relevance its investigation in *in vitro* experiments is of interest. The developed HPLC/ICP-QQQ-MS method was therefore applied in a first small *in vitro* experiment estimating ergothioneine uptake by HepG2 cells grown in cell culture medium containing 1 mM ergothioneine. Whereas ergothioneine was below the LOD in lysates of control cell pellets not exposed to ergothioneine, it was present in low amounts in the ergothioneine-treated cell pellets (Table 2). To rigorously confirm the identity of the ergothioneine peak in this complex matrix, high resolution molecular MS was performed on a cell pellet lysate and compared to that of an ergothioneine standard solution (ESI Fig. S-3 and S-4†). The full scan mass spectrum as

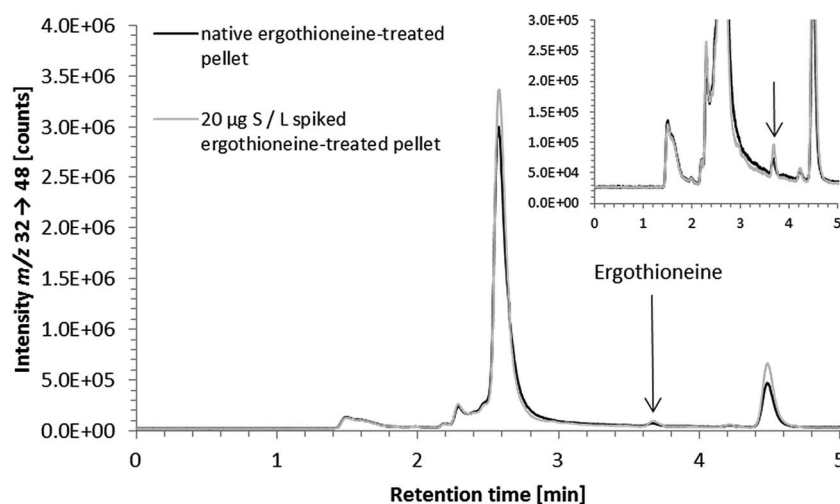


Fig. 5 HPLC/ICP-QQQ-MS chromatograms of a native and a spiked lysate of an ergothioneine-treated cell pellet; inset showing a zoom into the area of ergothioneine elution with the arrow indicating the ergothioneine peak; native ergothioneine-treated cell pellet (solid black line) and 20 $\mu\text{g S per L}$ ergothioneine spiked ergothioneine-treated cell pellet (solid light gray line); chromatographic conditions as follows: Gemini C6-Phenyl 4.6 \times 150 mm; column temperature, 30 $^{\circ}\text{C}$; injection volume, 10 μL ; mobile phase, 20 mM ammonium formate pH 9.0; flow rate, 0.7 mL min^{-1} .



Table 2 Ergothioneine concentrations in HepG2 cell pellets and cell culture medium treated with ergothioneine

Ergothioneine-treated cell pellets			
	Pellet 1	Pellet 2	Pellet 3
ng S in the form of ergothioneine in the pellet ^a	11.8 ± 0.2	15.4 ± 0.2	15.0 ± 0.3
Ergothioneine concentration ^a [mg S per L cell volume]	0.78 ± 0.06	1.13 ± 0.07	1.08 ± 0.11
Percentage of added ergothioneine ^a [%]	2.4 ± 0.2	3.5 ± 0.2	3.4 ± 0.3
Ergothioneine-treated cell culture medium			
	Medium 1	Medium 2	Medium 3
Ergothioneine concentration ^b [mg S per L]	27.3 ± 0.3	29.9 ± 0.5	30.5 ± 0.3
Percentage of added ergothioneine ^b [%]	85 ± 0.9	93 ± 1.5	95 ± 0.2

^a *n* = 3 injections. ^b *n* = 3 dilutions.

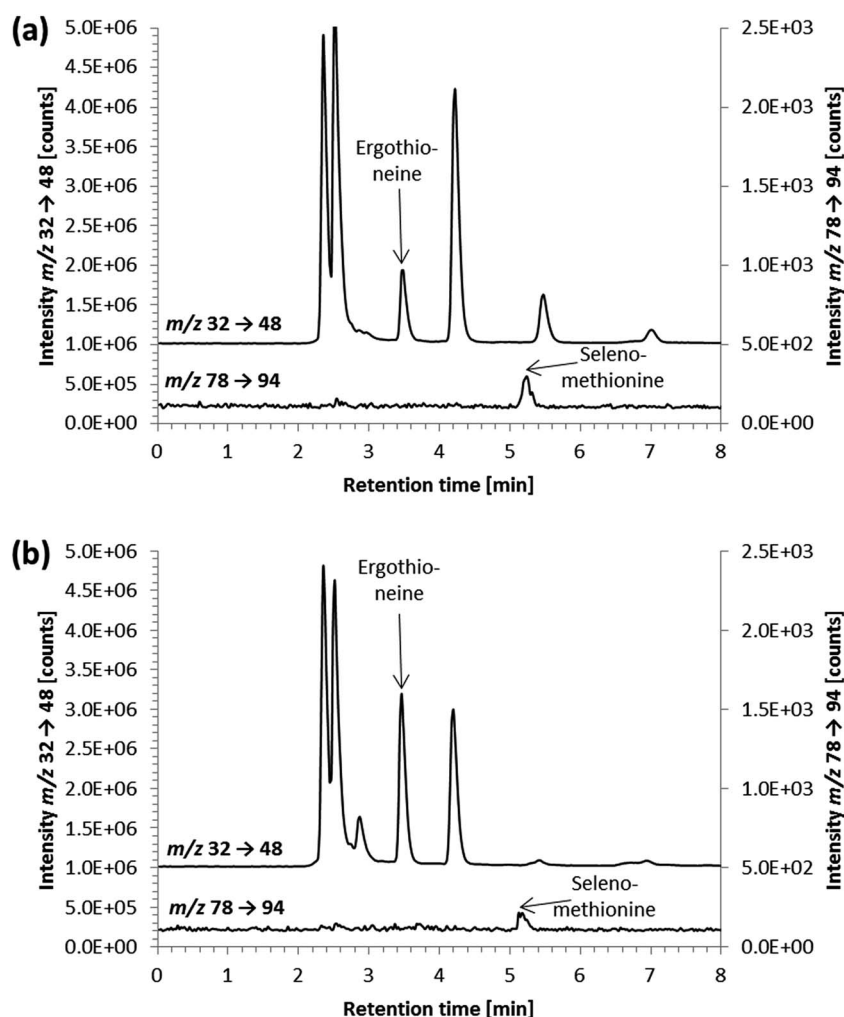


Fig. 6 HPLC/ICP-QQQ-MS chromatograms of water extracts of (a) oyster mushroom sample 1 and (b) oyster mushroom sample 2; *m/z* 32 → 48 *y*-axis offset of 1×10^6 counts and *m/z* 78 → 94 *y*-axis offset of 1×10^2 counts; chromatographic conditions as follows: Gemini C6-Phenyl 4.6×150 mm; column temperature, 30 °C; injection volume, 10 μ L; mobile phase, 20 mM ammonium formate 1% methanol pH 9.0; flow rate, 0.7 mL min⁻¹.

well as fragmentation of the pseudo-molecular ion (Fig. S-4†) clearly demonstrated that the peak in the cell pellet lysate corresponded to ergothioneine.

No cytotoxic effects were observed, when cell numbers of control cells and cells treated with 1 mM ergothioneine were compared, probably due to the low uptake by the cells. This low



absorption of ergothioneine was not unexpected because there was no detectable gene expression of the ergothioneine specific transporter OCTN1 in HepG2 cells.³⁷ In accord with the low uptake of ergothioneine by the cells, the concentrations of ergothioneine in the incubated cell culture media remained high – the measured concentration of 29.2 ± 2.5 mg S per L, represented around $91 \pm 8\%$ of the added ergothioneine (Table 2).

Simultaneous monitoring of sulfur and selenium species and determination of ergothioneine in oyster mushrooms (*Pleurotus* sp.) as a natural food source. To explore both the wider applicability of the method for the determination of ergothioneine in food and the simultaneous determination of selenium species, an attempt was made to analyze extracts of two samples of oyster mushrooms purchased from local food stores. In this first test with a real food matrix water was chosen as a compromise extractant for ergothioneine and selenium species. In both samples, a clear and well-resolved signal for ergothioneine was obtained (Fig. 6), the identity of which was again rigorously established by high resolution MS (ESI Fig. S-5†). Water extractable ergothioneine concentrations were 9.0 and 11 mg S per kg wet mass (*i.e.* 64 and 79 mg ergothioneine per kg wet mass) for duplicate analysis of mushroom sample 1, and 23 and 24 mg S per kg wet mass (165 and 172 mg ergothioneine per kg wet mass) for mushroom sample 2. These concentrations are in the range of those from an LC/MSMS study employing extraction with 4 mM perchloric acid and reporting 97 to 123 mg ergothioneine per kg wet mass in different samples of oyster mushrooms.⁹ No selenium signal other than a tiny one tentatively assigned as selenomethionine by co-elution with the respective standard could be observed in the water extracts of both oyster mushroom samples (Fig. 6). However, water extraction is not capable of making protein-bound selenium species reported to be contained in selenium enriched mushrooms^{38,39} available for analysis. Further optimization of the extraction procedure required for achieving optimum extraction efficiencies for both sulfur and selenium species was beyond the scope of this work and will be the topic of future studies. Nevertheless, this first application of our method clearly demonstrates the high potential of HPLC/ICP-QQQ-MS for the determination of ergothioneine in food sources and for the simultaneous speciation analysis of sulfur and selenium in biological samples.

Conclusions

A HPLC/ICP-QQQ-MS method was developed for the quantitative determination of the antioxidant ergothioneine in biological samples. It was successfully applied to quantify this health-relevant sulfur species in complex matrices such as cell pellets and culture medium and aqueous mushroom extracts. With the method, the fate of ergothioneine in *in vitro* experiments can be investigated to further elucidate its biological role. Furthermore, the method is expected to offer a broad field of future applications like the monitoring of ergothioneine in food or body fluids. We also show that the method has the potential to simultaneously determine sulfur and selenium species thereby

yielding a picture of the speciation of both elements in one run, and, hence, could be used in studies investigating similarities and differences between these two elements and their species in biological systems with likely significant applications in human health research in the future.

Author contributions

The manuscript was written with contributions from all authors. All authors have given approval to the final version of the manuscript.

Conflicts of interest

The authors declare that they have no competing interests.

Acknowledgements

We thank the Austrian Science Fund (FWF, project number I 2262-N28) and the “Deutsche Forschungsgemeinschaft” (DFG, project grant number SCHW 903/9-1) for funding this research. We also thank the “NutriAct – Competence Cluster Nutrition Research Berlin-Potsdam” funded by the Federal Ministry of Education and Research (project number 01EA1408B) and the “Fonds der Chemischen Industrie” for financial support. We thank Kenneth Jensen and the NAWI Graz Central Lab – Metabolomics for high resolution mass spectra.

References

- 1 M. C. Tanret, *C. R. Hebd. Seances Acad. Sci.*, 1909, **149**, 222–224.
- 2 D. S. Genghof and O. van Damme, *J. Bacteriol.*, 1964, **87**, 852–862.
- 3 D. S. Genghof, *J. Bacteriol.*, 1970, **103**, 475–478.
- 4 C. Pfeiffer, T. Bauer, B. Surek, E. Schömig and D. Gründemann, *Food Chem.*, 2011, **129**, 1766–1769.
- 5 G. W. Jones, S. Doyle and D. A. Fitzpatrick, *Gene*, 2014, **549**, 161–170.
- 6 D. Gründemann, *Prev. Med.*, 2012, **54**, S71.
- 7 C. E. Hand and J. F. Honek, *J. Nat. Prod.*, 2005, **68**, 293–308.
- 8 I. K. Cheah and B. Halliwell, *Biochim. Biophys. Acta, Mol. Basis Dis.*, 2012, **1822**, 784–793.
- 9 J. Ey, E. Schömig and D. Taubert, *J. Agric. Food Chem.*, 2007, **55**, 6466–6474.
- 10 P. Tepwong, A. Giri, F. Sasaki, R. Fukui and T. Ohshima, *Food Chem.*, 2012, **131**, 247–258.
- 11 Y. Sapozhnikova, W. C. Byrdwell, A. Lobato and B. Romig, *J. Agric. Food Chem.*, 2014, **62**, 3034–3042.
- 12 L.-Z. Wang, W.-L. Thuya, D. S.-L. Toh, M. G.-L. Lie, J.-Y. A. Lau, L.-R. Kong, S.-C. Wan, K.-N. Chua, E. J.-D. Lee and B.-C. Goh, *J. Mass Spectrom.*, 2013, **48**, 406–412.
- 13 N. Dubost, B. Ou and R. Beelman, *Food Chem.*, 2007, **105**, 727–735.
- 14 T. H. Nguyen, A. Giri and T. Ohshima, *Food Chem.*, 2012, **133**, 585–591.



- 15 T. H. Nguyen, R. Nagasaka and T. Ohshima, *Int. J. Food Sci. Technol.*, 2012, **47**, 1193–1205.
- 16 S.-Y. Chen, K.-J. Ho, Y.-J. Hsieh, L.-T. Wang and J.-L. Mau, *LWT–Food Sci. Technol.*, 2012, **47**, 274–278.
- 17 S. Sotgia, A. Zinellu, G. Pintus, G. A. Pinna, L. Deiana and C. Carru, *J. Sep. Sci.*, 2013, **36**, 1002–1006.
- 18 A. Z. Woldegiorgis, D. Abate, G. D. Haki and G. R. Ziegler, *Food Chem.*, 2014, **157**, 30–36.
- 19 Q. Liu, W. Zhang, H. Wang, Y. Li, W. Liu, Q. Wang, D. Liu, N. Chen and W. Jiang, *J. Chromatogr. Sci.*, 2016, **54**, 934–938.
- 20 D. Pröfrock and A. Prange, *Appl. Spectrosc.*, 2012, **66**, 843–868.
- 21 J. Giner Martínez-Sierra, O. Galilea San Blas, J. M. Marchante Gayón and J. I. García Alonso, *Spectrochim. Acta, Part B*, 2015, **108**, 35–52.
- 22 L. Balcaen, E. Bolea-Fernandez, M. Resano and F. Vanhaecke, *Anal. Chim. Acta*, 2015, **894**, 7–19.
- 23 S. D. Fernández, N. Sugishama, J. R. Encinar and A. Sanz-Medel, *Anal. Chem.*, 2012, **84**, 5851–5857.
- 24 J. Gong, M. J. Solivio, E. J. Merino, J. A. Caruso and J. A. Landero-Figueroa, *Anal. Bioanal. Chem.*, 2015, **407**, 2433–2437.
- 25 G. Hermann, L. H. Møller, B. Gammelgaard, J. Hohlweg, D. Mattanovich, S. Hann and G. Koellensperger, *J. Anal. At. Spectrom.*, 2016, **31**, 1830–1835.
- 26 L. H. Møller, A. Macherius, T. H. Hansen, H. M. Nielsen, C. Cornett, J. Østergaard, S. Stürup and B. Gammelgaard, *J. Anal. At. Spectrom.*, 2016, **31**, 1877–1884.
- 27 F. Calderón-Celis, S. Díez-Fernández, J. M. Costa-Fernández, J. R. Encinar, J. J. Calvete and A. Sanz-Medel, *Anal. Chem.*, 2016, **88**, 9699–9706.
- 28 P. Traar, F. Belaj and K. A. Francesconi, *Aust. J. Chem.*, 2004, **57**, 1051–1053.
- 29 D. Kuehnelt, N. Kienzl, P. Traar, N. H. Le, K. A. Francesconi and T. Ochi, *Anal. Bioanal. Chem.*, 2005, **383**, 235–246.
- 30 T. A. Marschall, J. Bornhorst, D. Kuehnelt and T. Schwerdtle, *Mol. Nutr. Food Res.*, 2016, **60**, 2622–2632.
- 31 N. Kroepfl, K. B. Jensen, K. A. Francesconi and D. Kuehnelt, *Anal. Bioanal. Chem.*, 2015, **407**, 7713–7719.
- 32 S. Kokarnig, N. Kroepfl, D. Kuehnelt and K. A. Francesconi, *Anal. Methods*, 2014, **6**, 1603–1607.
- 33 *Eurachem Guide. The Fitness for Purpose of Analytical Methods – A Laboratory Guide to Method Validation and Related Topics*, ed. B. Magnusson and U. Örnemark, 2nd edn, 2014.
- 34 R. Raml, G. Raber, A. Rumpler, T. Bauernhofer, W. Goessler and K. A. Francesconi, *Chem. Res. Toxicol.*, 2009, **22**, 1534–1540.
- 35 S. Kokarnig, D. Kuehnelt, M. Stiboller, U. Hartleb and K. A. Francesconi, *Anal. Bioanal. Chem.*, 2011, **400**, 2323–2327.
- 36 K. Flouda, J. M. Dersch, C. Gabel-Jensen, S. Stürup, S. Misra, M. Björnstedt and B. Gammelgaard, *Anal. Bioanal. Chem.*, 2016, **408**, 2293–2301.
- 37 G. Ahlin, C. Hilgendorf, J. Karlsson, C. A.-K. Szegartyo, M. Uhlen and P. Artursson, *Drug Metab. Dispos.*, 2009, **37**, 2275–2283.
- 38 V. Gergely, K. M. Kubachka, S. Mounicou, P. Fodor and J. A. Caruso, *J. Chromatogr. A*, 2006, **1101**, 94–102.
- 39 V. Díaz Huerta, A. Sanz-Medel and M. L. Fernández Sánchez, *Anal. Bioanal. Chem.*, 2006, **384**, 902–907.

