



Cite this: *Phys. Chem. Chem. Phys.*,
2017, **19**, 6006

Time-resolved terahertz spectroscopy reveals the influence of charged sensitizing quantum dots on the electron dynamics in ZnO

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Photoinitiated charge carrier dynamics in ZnO nanoparticles sensitized by CdSe quantum dots is studied using transient absorption spectroscopy and time-resolved terahertz spectroscopy. The evolution of the transient spectra shows that electron injection occurs in a two-step process, where the formation of a charge transfer state (occurring in several picoseconds) is followed by its dissociation within tens of picoseconds. The photoconductivity of electrons injected into the ZnO nanoparticles is lower than that of charges photogenerated directly in ZnO. We conclude that the motion of injected electrons in ZnO nanoparticles is strongly influenced by their interaction with positive charges left in the sensitizing quantum dots.

Received 2nd November 2016,
Accepted 26th January 2017

DOI: 10.1039/c6cp07509f

rsc.li/pccp

A Introduction

Semiconductor quantum dots (QDs) are promising building blocks in photovoltaic devices due to their size-tunable band gap, strong absorption, long-term photostability and possible carrier multiplication.¹⁻⁷ In a typical QD-based solar cell, incident light is absorbed in QDs producing electron-hole pairs. The QDs can be attached to wide-band-gap semiconductor nanoparticles. In such semiconductors, the bottom of the conduction band can have lower energy than the excited electron in the QDs. Consequently, energetically downhill electron injection from the QD into the semiconductor occurs. The electron injection competes with various recombination processes, therefore high speed of reaching stabilized separated charges is required to achieve a high charge collection efficiency of the solar cells.^{8,9}

The time-scale of the injection and its dependence on the electronic band alignment, sensitization geometry and linker

molecules have been widely investigated *via* time-resolved photoluminescence and visible transient absorption spectroscopy.^{10–13} These conventional approaches can only probe the depopulation of the excited states in the QDs. However, probing the electron dynamics in the semiconductor is also important, particularly in systems where electron dynamics includes a charge transfer state (CTS) due to an electrostatic interaction between the injected electron and the hole remaining in the donor.^{8,14,15}

In this work, we combine transient absorption (TA) spectroscopy and time-resolved terahertz (THz) spectroscopy to investigate photoinitiated processes in ZnO nanoparticles (NPs) sensitized by CdSe QDs (referred to as CdSe/ZnO in the following text). While TA probes mainly the oxidation state of the sensitizer, THz spectroscopy probes predominantly the population of free charges in the NPs and their transport properties on a nanometer distance scale.¹⁶ A combination of these results with Monte-Carlo simulations provides a detailed picture of the electron dynamics, including the interaction of the injected electrons with positively charged QDs. We show that the Coulombic interaction not only affects the injection rate, but it also considerably reduces electron mobility in the ZnO NPs. We also show that all the processes are controlled by the effective permittivity of the system, which implies that significant differences appear between various forms of ZnO nanostructures.

B Experimental details

ZnO films were deposited by the doctor-blade technique using colloidal ZnO NPs (Sigma Aldrich) with various sizes; subsequently,

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they were sintered in air at 450 °C for 30 minutes, resulting in about 1 micron thick films. Although the films became mesoporous after annealing, we can still distinguish the boundaries of individual nanoparticles (Fig. 1). The mean NP diameters identified from the SEM image (25, 48, and 61 nm) are close to the nominal sizes specified by the manufacturer (35, 50 and 70 nm, respectively).

Sensitization by mercaptopropionic-acid-capped CdSe QDs⁸ was accomplished by immersing the preheated nanoparticulate ZnO films into QD solution for 2 h. Based on the investigations in ref. 10, this leads to approximately monolayer adsorption of QDs on the NP surface. The steady-state absorption spectrum of neat QDs exhibits a pronounced $1S_{3/2}$ – $1S_e$ exciton band at 542 nm (Fig. 1), which corresponds to the mean QD diameter of ~ 2.9 nm.^{8,17} The blue shift of the exciton peak for CdSe/ZnO is

due to structural and environmental changes upon adsorption to ZnO.¹⁸ All pump-probe studies were performed under a N₂ atmosphere in order to eliminate possible effects of cumulative photodegradation.^{8,15,19–21}

The transient absorption spectra were recorded in a standard pump-probe setup⁸ based on a regenerative Ti:sapphire laser amplifier (Spitfire, Spectra Physics; 1 kHz repetition rate). The excitation beam was tuned to 470 nm; the photon fluence of 1.4×10^{15} photons per cm² per pulse ensures a sufficient signal to noise ratio. Although this intensity generates ~ 2.6 electron-hole pairs per QD,¹⁷ we verified that it is still low enough to prevent non-linear phenomena.²²

C Results and discussion

The probing wavelength (530 nm) was tuned to the band-edge transition of the QDs, therefore the instantaneous decay observed in the kinetics of bare CdSe QDs (Fig. 2a) corresponds to the state filling of the QD $1S$ electron levels^{8,12,23–26} (the hole contribution to the TA signal is negligible due to their higher effective mass). The nanosecond decay of the TA kinetics for the neat CdSe QDs represents a slow recombination of the photoexcited electron-hole pairs.

The decay of the TA dynamics of CdSe/ZnO is significantly faster due to electron injection into the ZnO NPs.²² The observed decay kinetics is rather complex, since the electron transfer from the QDs to the ZnO NPs can occur *via* multiple channels; the dynamics at longer time scales is further complicated by electron-hole recombination. We can identify three distinct characteristic decay times: a fast one (about 4 ps), a slow one (30–40 ps), and a long-lived one (~ 600 ps). This kind of dynamics has been observed in many similar QD-sensitized metal-oxide nanostructures.^{18,27–31}

The components faster than a few hundreds of picoseconds are due to electron injection from QDs directly attached to the metal-oxide;^{8,18,27} the bi-exponential nature can be explained by injection *via* a CTS.^{8,22,32} The long-lived component is typically associated with QDs without a direct connection to the ZnO NPs: the injection is then either very slow or even inhibited.^{10,23} A certain fraction of the sensitizing QDs can be attached to the metal-oxide indirectly *via* other QDs in which case energy transfer may take place.¹⁰ *Via* energy transfer the excitation will ultimately reach the QDs which are directly attached to a ZnO NP.

Transient THz conductivity spectra were recorded in a usual setup for time-resolved THz spectroscopy²¹ driven by the same laser amplifier as in the TA measurements. In order to reveal the influence of a QD sensitizer on the electron dynamics in ZnO NPs, we investigated both CdSe and ZnO upon photo-excitation of the QDs (542 nm), and directly excited neat ZnO NPs (266 nm). The wavelength of 542 nm is well below the band gap of ZnO; therefore, it is absorbed only by the QDs. Conversely, the excitation at 266 nm is well above the band gap of ZnO, leading to the generation of electron-hole pairs directly inside the NPs.

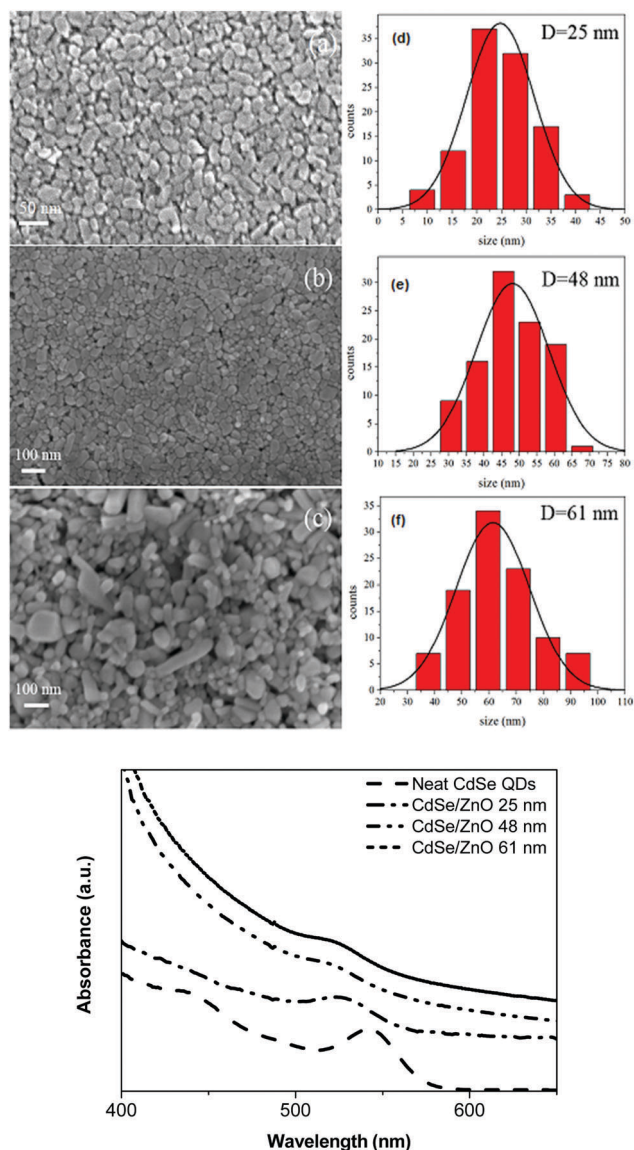


Fig. 1 (a–c) SEM images of ZnO NPs with small, middle, and big sizes. (d–f) Histograms of NP diameters identified in the images, and Gaussian fits of these distributions. (g) Steady-state absorption spectra of neat CdSe QDs and of CdSe/ZnO.



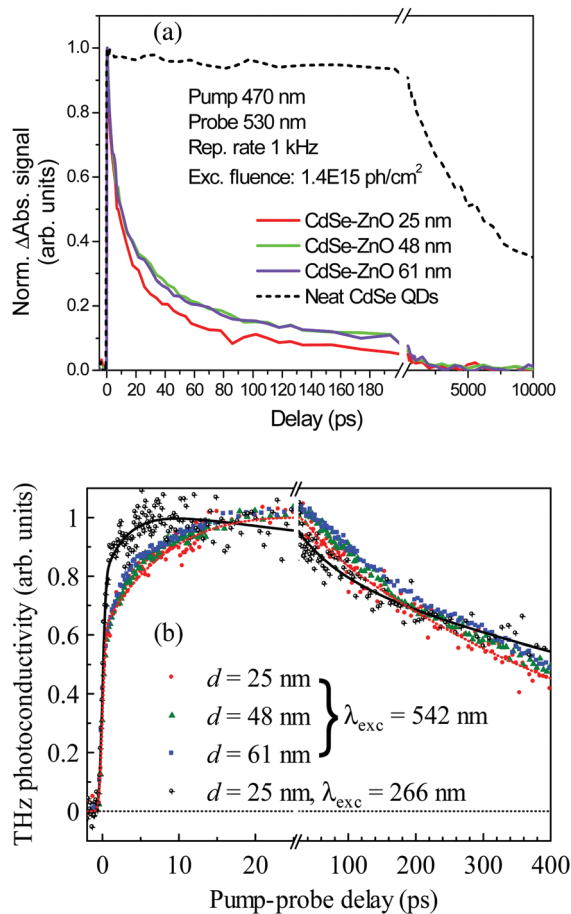


Fig. 2 (a) Transient absorption kinetics of CdSe QDs and CdSe/ZnO. (b) Time dependence of the transient THz photoconductivity of CdSe/ZnO (excitation wavelength 542 nm) and bare ZnO NPs (excitation wavelength 266 nm). The lines serve only to guide the eye. All data are normalized to unity.

Measurement of transient THz transmittance spectra $\Delta T/T$ ($=\Delta E/E$ where E is the spectrum of the electric field transmitted through the sample without photoexcitation, and ΔE is the photo-induced change in the transmitted electric field spectrum) allows retrieval of charge mobility μ and quantum yield of mobile charge carriers. For a thin photoconducting layer deposited on a substrate with refractive index n_s and completely absorbing the excitation power,

$$\frac{\Delta T}{T}(\omega) = -\frac{Z_0}{n_s + 1} \Delta \Sigma_{\text{eff}}(\omega) \quad (1)$$

where $\Delta \Sigma_{\text{eff}}$ is the effective transient sheet photoconductivity of the film and Z_0 is the vacuum impedance.^{32,33} Due to the inhomogeneity of the sample, effective medium approximation is employed to calculate the measured effective conductivity $\Delta \sigma_{\text{eff}}$ from the microscopic conductivity $\Delta \sigma$ of the NPs. In the limit of weak photoexcitation, $\Delta \sigma_{\text{eff}} = (V + B) \Delta \sigma$,³² where the constants B and V reflect the morphology of the sample ($V + B \leq 1$).^{33,34} Since the transient sheet conductivity reads $\Delta \Sigma_{\text{eff}} = \Delta \sigma_{\text{eff}}/\alpha$ (α is the optical absorption coefficient) and the microscopic transient conductivity $\Delta \sigma(\omega, t) = e_0 \phi(t) n_{\text{exc}} \mu(\omega, t)$ is proportional to the

product of the quantum yield of charge formation ϕ and the electron mobility spectrum $\mu(\omega)$, eqn (1) simplifies to

$$(B + V) \phi \mu(\omega) = \frac{\Delta \sigma_{\text{eff}}(\omega)}{e_0 n_{\text{exc}}} = \underbrace{\frac{n_s + 1}{Z_0 e_0 F}}_{\Delta T_{\text{norm}}} \cdot \frac{\Delta T}{T}(\omega) \quad (2)$$

where F is the excitation fluence (photon exposure in photons per m^2), $n_{\text{exc}} = F\alpha$ is the excitation density and e_0 is the elementary charge. This equation is the fundamental link between the measured transient transmittance (or more conveniently, normalized transient transmission ΔT_{norm}) and the yield-mobility product. The normalized transient transmission has the meaning of an effective conductivity per unit charge.

For carriers directly photogenerated in neat ZnO NPs, the real part of the measured THz conductivity increases with frequency, whereas the imaginary part is mostly negative (Fig. 3a), indicating a certain degree of charge localization. In order to better assess the properties of these bare nanoparticles, we carried out Monte-Carlo calculations of conductivity of charges confined in nanospheres.³⁵ In brief, Newton's equations of motion are solved to obtain thermal trajectories of an ensemble of charges. When a charge hits the boundary of a NP, it is either scattered into a neighboring NP with probability p_F , or it is scattered back into the original NP with probability $p_B = 1 - p_F$. The Kubo formula is subsequently employed to retrieve the mobility spectrum from the calculated velocity autocorrelation function. Unlike the parameters V and B (which describe the distribution of the THz electric field in the inhomogeneous system), the probabilities p_B and p_F are related solely to the charge transport among nanocrystals. With electron effective mass $m_{\text{eff}} = 0.24m_e$ and scattering time $\tau_s = 27$ fs (which stems from the Drude dc mobility $\mu = e_0 \tau_s / m_{\text{eff}}$),³⁶ the fits of the measured spectra (Fig. 3a) yield $p_B = 99\%$, 89% and 86% and $V + B = 0.07$, 0.18 and 0.11 for samples with mean NP diameters of 25, 48 and 61 nm, respectively (the initial quantum yield ϕ is 1). The sum $V + B$ is higher for the 48 nm NPs, which indicates that they are better percolated (the NPs are better electrically connected with their neighbors) than the others; this is also the reason why the transient conductivity of these particles is higher than that of larger NPs. The large values of p_B provide evidence that NP boundaries constitute a significant barrier for charge transport in all studied films. The deviations between the calculated and measured imaginary parts in Fig. 3a are attributed to the possible minor drift of the experimental setup, which can mix a small part of the dominating real component of the signal with the imaginary component.³⁷

The shape of the normalized transient conductivity spectra of carriers injected into the ZnO NPs (Fig. 3b) is very similar to that of the photogenerated carriers, but the amplitude is a factor of 4 to 5 lower. We attribute the difference in amplitude to the Coulombic interaction between the electron injected into the ZnO NP and the hole left in the CdSe QD, similarly to that observed for zinc-porphyrin-sensitized ZnO NPs.¹⁵ We can rule out other effects, possibly responsible for the decrease in the conductivity amplitude.

- Low injection efficiency is incompatible with TA kinetics. The TA kinetics decays to $<20\%$ of its peak value within 100 ps; the injection efficiency thus exceeds 80%.



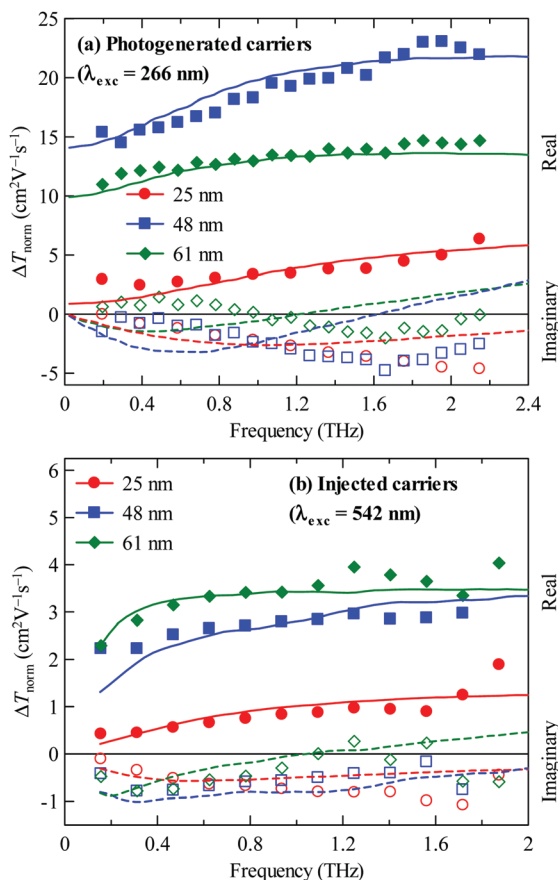


Fig. 3 Normalized transient transmission of (a) photogenerated carriers in pure ZnO NPs ($F = 1.2 \times 10^{13}$ photons per cm^2 , pump-probe delay 10 ps) and of (b) carriers injected from CdSe QDs into ZnO NPs ($F = 5.4 \times 10^{14}$ photons per cm^2 , pump-probe delay 5 ps). Symbols: measured data (closed = real part, open = imaginary part); fits by Monte-Carlo calculations (lines).

- Structural differences between bare and sensitized samples are improbable. This is supported by ref. 15, where sensitization by zinc-porphyrin dye did not change the THz photoconductivity of photogenerated electrons.

- Ultrafast recombination or trapping of electrons in ZnO NPs would appear as a rapid initial decay component of THz photoconductivity, while the measured kinetics varies only slowly (Fig. 2b).

The THz kinetics (which is a measure of the population change of mobile charges) exhibits similar features to the TA kinetics. The 4 ps increase is due to the injection of electrons from directly attached CdSe QDs. Less pronounced is the increase with 37 ps time constant (which is not present when electrons in ZnO are generated directly by above-band-gap photoexcitation). In principle, it could represent a slower injection from QDs not directly attached to the ZnO NPs; however, our previous research revealed that such a process is mediated by inter-QD energy transfer which appears on a nanosecond timescale.¹⁰ For this reason, we attribute the observed process to CTS-mediated electron injection.^{8,15,38,39} The CTS then undergoes continuous dissociation which increases the THz

photoconductivity up to 40 ps. At this time—when the amplitude is maximum—, most of the electrons are already injected into the ZnO NPs.⁸ Finally, the THz kinetic decays on the hundreds-picoseconds timescale due to trapping or recombination of the electrons in ZnO NPs.

The behavior of injected electrons is strongly affected by Coulombic interaction with positively charged QDs. Here we focus on two such effects. Firstly, we discuss the dynamics of the CTS and estimate the injection rate of electrons into the NPs. Secondly, we develop Monte-Carlo calculations to assess the influence on electron transport after CTS dissociation.

The dissociation rate K_{diss} of a CTS can be estimated using the Braun–Onsager model:²⁵

$$K_{\text{diss}} = \frac{\mu e_0}{\varepsilon_{\text{MO}} \varepsilon_0} \cdot \frac{3}{4\pi a^3} \cdot \exp(-E_b/k_B T) \quad (3)$$

where μ is the charge carrier mobility in ZnO, $\varepsilon_{\text{MO}} = (\varepsilon_{\text{cc}} + 2\varepsilon_{\text{aa}})/3 = 8.2$ is the relative permittivity of ZnO,⁴⁰ E_b is the Coulombic binding energy of the electron in the CTS, and a is the thermalization length (=distance at which the Coulombic energy is equal to the thermal energy $k_B T$). The positively charged QD generates a displacement field D . Outside of the QD—in the distance r from the QD center—Gauss theorem yields $D(r) = -e_0/(4\pi r^2)$. The energy required to move an electron from the surface of the NP (bound CTS) to a distance x thus reads

$$W(x) = \int_{R_{\text{QD}}+h}^{R_{\text{QD}}+h+x} \frac{e_0 D(r)}{\varepsilon_0 \varepsilon_{\text{eff}}} dr = \frac{e_0^2}{4\pi \varepsilon_0 \varepsilon_{\text{eff}}} \left[\frac{1}{R_{\text{QD}}+h} - \frac{1}{R_{\text{QD}}+h+x} \right] \quad (4)$$

where R_{QD} is the radius of the QD and $h = 0.65$ nm is the distance between the QD and the surface of the NP, which is given by the length of the linker molecule (mercaptopropionic acid). In this notation, dissociation occurs for $x \rightarrow \infty$, i.e. $E_b = W(\infty)$, and the thermalization length a is calculated from the equation $W(a) = k_B T$. In eqn (4), we neglected the exact spatial profile of the relative permittivity outside of the QD and we treated the surroundings as a homogenous medium with effective permittivity ε_{eff} . We estimate ε_{eff} from the Maxwell-Garnett model, where we assume that the ZnO NP with permittivity ε_{MO} form a percolated network containing $\sim 30\%$ of air voids (between closely packed spherical NPs); $\varepsilon_{\text{eff}} \sim 5.5$. Using these values, we obtain $E_b = 120$ meV and $a = 9$ nm. The determining charge motion in eqn (3) is that occurring on the distance of the thermalization length; in our case, $a = 9$ nm therefore charges have only a small chance to interact with the surface of the NP and the pertinent mobility μ is then that of bulk ZnO ($200 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$,³⁶). The dissociation time estimated from eqn (3) is thus $1/K_{\text{diss}} = 12$ ps, which corresponds well with the slow components in TA and THz kinetics (30–40 ps).

According to eqn (3), the dissociation rate depends very sensitively on the binding energy. Coulombic interaction attracts the hole towards the electron: the decreased distance between the electron and the hole would lead to a higher binding energy, and in turn, to a slower dissociation rate. A particular environment of each individual QD is another issue influencing the binding energy through the effective permittivity



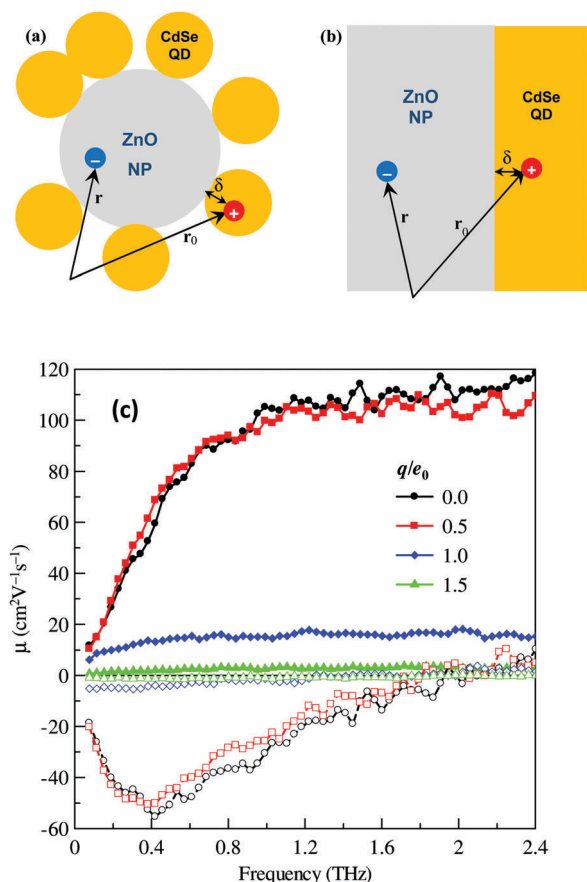


Fig. 4 Approximation of the Coulombic potential of the hole (+) acting on the electron (–) injected into the ZnO. (a) Realistic geometry and (b) geometry employed for simplified evaluation of the potential. (c) Mobility spectra calculated for isolated spherical ZnO NP with a diameter of 48 nm. A positive charge is placed in the distance $\delta = 1$ nm from the surface of the nanoparticle, and the effective permittivity of the surroundings was considered to be $\tilde{\epsilon} = 6$.

ϵ_{eff} . For example, the packing of NPs leads to much larger filling factors than for nanowires; in turn, the effective permittivity of the nanowire system is lower.¹¹ In this situation, the binding energy would increase [eqn (4)], leading to slower dissociation [eqn (3)]. This is in agreement with the longer CTS dissociation time (200 ps) observed in CdSe-QD-sensitized ZnO nanowires.⁸ The faster injection rates in NPs would be favorable for the overall charge injection efficiency in solar cells.⁴¹

The effective permittivity depends on the permittivity, filling factor, shape and percolation degree of the electron acceptor. Our results thus clearly show that the electron injection efficiency depends on the shape of the electron acceptor (nanoparticles or nanowires) as well as on the detailed morphology of the acceptor network. In other words, shape and morphology are fundamental parameters which need to be considered in the studies of electron injection.

In order to simulate the electron mobility in QD sensitized ZnO nanoparticles the Coulombic interaction in Monte-Carlo calculations was represented using an electric potential $\Phi(\mathbf{r})$ in which electrons move. We approximated it by a potential of a

positive charge q placed in the distance δ from a planar interface (Fig. 4a and b):

$$\Phi(\mathbf{r}) = \frac{q}{4\pi\epsilon_0\tilde{\epsilon}} \cdot \frac{1}{|\mathbf{r} - \mathbf{r}_0|}$$

where \mathbf{r}_0 is the position of the positive charge and $\tilde{\epsilon}$ is the reduced permittivity defined as⁴² $\tilde{\epsilon} = (\epsilon_{\text{MO}} + \epsilon_{\text{QD}})/2$. In order to illustrate the effects of this interaction, we considered mutually isolated NPs (*i.e.*, $p_{\text{F}} = 0$). When the Coulombic binding energy is smaller than the thermal energy of charges, the electron response is only weakly influenced by the Coulombic interaction. With increasing interaction strength, electrons are more attracted to the positive charges and their motion becomes more confined. This is accompanied by a decrease in the mobility amplitude (Fig. 4c), and for even stronger interaction, by a change in the shape of the mobility spectrum. This behavior is in agreement with experimental observations, showing a substantial difference in amplitudes of normalized conductivities of injected and directly photogenerated charges, and only insignificant differences in the spectra shapes.

D Conclusions

In summary, we have investigated photoinduced charge carrier dynamics in CdSe QD sensitized ZnO NPs. The Coulombic interaction between injected electrons in ZnO and the holes in the QDs significantly influences the electron dynamics after photoexcitation in various aspects: first, the electron injection from CdSe to ZnO is mediated by a CTS formed within a few picoseconds and dissociated within tens of picoseconds. We extended the Braun–Onsager model to reflect the environment of the constituents and applied it for the estimation of the dissociation rate in CdSe-QD-sensitized ZnO NPs. This model also explains the different dissociation rate in the nanowires system as a result of different effective permittivities of NPs and nanowires systems. More generally, dependence of the injection rate on the effective permittivity of the system implies a fundamental influence of the shape and morphology of the nanostructures forming the electron acceptor network on the charge injection efficiency. In addition, the Coulombic interaction between the electron and the hole reduces mobility of injected electrons by a factor of 4–5. Such a behavior was explained by Monte-Carlo calculations of THz conductivity of electrons moving in an electrostatic potential.

Acknowledgements

This work was supported by the Knut and Alice Wallenberg Foundation, Swedish Research Council, Czech Science Foundation (Grant 17-03662S) and Carl Tryggers Foundation in Stockholm (Project CTS 14:390). KZ acknowledges support from NPRP grant #NPRP7-227-1-034 from the Qatar National Research Fund (a member of Qatar Foundation).

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