## Chemical Science



### **EDGE ARTICLE**

View Article Online
View Journal | View Issue



Cite this: Chem. Sci., 2016, 7, 4372

# Expeditious synthesis of the fused hexacycle of puberuline C *via* a radical-based cyclization/translocation/cyclization process†

Koichi Hagiwara, Toshiki Tabuchi, Daisuke Urabe and Masayuki Inoue\*

The fused 6/7/5/6/6/6-hexacyclic ring system of puberuline C was assembled in 18 steps from 2-(ethoxycarbonyl)cyclohexanone. After the azabicyclo[3.3.1]nonane derivative was sequentially coupled with propargyl magnesium bromide, 2-iodo cyclopentenone and allyl bromide, the pentacycle was constructed in a single step *via* a radical-based cyclization/translocation/cyclization process. The C11-bridgehead radical generated *via* C-Br homolysis participated in a 7-endo cyclization, and the 1,5-hydrogen translocation of the resultant radical was followed by transannular 6-exo cyclization to simultaneously realize the construction of the two rings and the introduction of the five contiguous stereocenters. The last 6-exo cyclization was induced by the Mukaiyama aldol reaction, and the C16-ketone was stereoselectively reduced by the action of Sml<sub>2</sub>/t-BuOH, leading for the first time to the synthesis of the entire hexacycle of puberuline C.

Received 13th February 2016 Accepted 11th March 2016

DOI: 10.1039/c6sc00671j

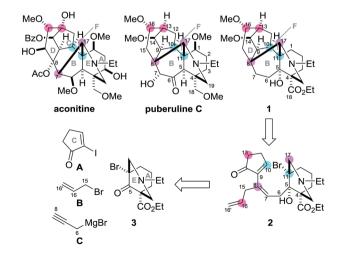
www.rsc.org/chemicalscience

Plants of the genera *Aconitum* and *Delphinium* have been used for centuries in traditional oriental medicines for their anti-inflammatory, analgesic, and anti-rheumatic activities. Efforts to identify the pharmacologically important compounds in these plant families have resulted in the determination of over 600 structurally complex C19–diterpenoid alkaloids, of which aconitine is a representative example (Scheme 1). In 2009, puberuline C was isolated from a traditional Chinese medical plant, *Aconitum barbatum* var. *puberulum*, and was found to belong to the C19–diterpenoid alkaloid family. Its architecturally complex hexacyclic system is composed of fused 6/7/5/6/6-membered (A/B/C/D/E/F) rings containing one nitrogen group and six oxygen-based polar functionalities. These structural features significantly increase the challenge of chemically synthesizing puberuline C.4

Puberuline C differs structurally from the majority of C19-diterpenoid alkaloids in its C17-bond connection. Specifically, the C8–17 bond of puberuline C constitutes the six-membered F-ring, while the C7–17 bond of aconitine forms a five-membered counterpart. To date, considerable synthetic effort has been focused on the aconitine-type alkaloids, <sup>5,6</sup> culminating in the historic total syntheses of talatisamine, chasmanine and 13-desoxydelphonine by Wiesner's group, <sup>7</sup> neofinaconitine by Gin's group, <sup>8</sup> and weisaconitine D and liljestrandinine by

Sarpong's group.9 In sharp contrast, the molecular framework of puberuline C has not been chemically assembled.10 In this manuscript, we describe the efficient synthesis of the unique hexacyclic ring system of puberuline C by utilizing a radical-based cyclization/translocation/cyclization process and the Mukaiyama aldol reaction as the two key transformations.

To devise a novel strategy for the total synthesis of puberuline C, we designed a simplified puberuline C (1) as a model target compound (Scheme 1). Although the oxidation states at the C1, 5, 6, 7 and 18 positions of 1 are different from those of puberuline C, 1 retains its entire hexacyclic framework and nine



Scheme 1 Structures of aconitine and puberuline C, and a retrosynthetic plan of the model target compound  ${\bf 1}$ .

Graduate School of Pharmaceutical Sciences, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-0033, Japan. E-mail: inoue@mol.f.u-tokyo.ac.jp; Fax: +81-3-5841-0568

† Electronic supplementary information (ESI) available. CCDC 1452192. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c6sc00671j

**Edge Article Chemical Science** 

stereocenters (C4, 8, 9, 10, 11, 13, 14, 16, 17). In our retrosynthesis, the three ring structures of 1 were disconnected at the C10-11, C13-16 and C8-17 bonds to identify 2 as a pivotal intermediate. Compound 2 was proposed as the precursor of the bridgehead radical at C11. We presumed that the bridgehead radical would enable stereoselective construction of the sterically hindered C10-11 bond because of its potent reactivity, stereochemically predestined nature, and high orthogonality to diverse polar functional groups.11 In the synthetic direction, the C11-bridgehead radical generated through C-Br homolysis of triene 2 would chemoselectively react with the C9-10 double bond of the C-ring enone, leading to the formation of the sevenmembered B-ring. After radical cyclization, the remaining C16-16' and C7-8 double bonds were to be utilized to connect the C13/16 and C8/17 atoms to transannulate the six-membered Dand F-rings of 1, respectively, thereby establishing the stereochemistry of these four positions. As the radical precursor 2 has only three stereocenters (C4, 5, 11), the route to 2 would be greatly simplified. Highly unsaturated 2 would be prepared through a series of carbon chain extensions by step-wise attachment of three achiral units (five-membered C-ring A, B and C) to the chiral 6/6-fused AE-ring 3.

The synthesis of 1 commenced by preparing the known material 3 12 (Scheme 2). 2-(Ethoxycarbonyl)cyclohexanone (4) was first brominated to 5, which was then treated with ethylamine and formaldehyde to induce a double Mannich reaction, giving rise to 3 equipped with the C11-bromo group. Propargyl

Scheme 2 Synthesis of the three substrates for the radical reaction. Reagents and conditions: (a) Br<sub>2</sub>, Et<sub>2</sub>O, 0 °C to RT; (b) aq. HCHO, aq. EtNH<sub>2</sub>, MeOH, 0 °C to RT; (c)  $\mathbf{C}$ , ZnBr<sub>2</sub> (5 mol%), CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, 49% (dr at C5 = 3.2:1); recrystallization 29% for 6 (3 steps); (d) A, PdCl<sub>2</sub> (PPh<sub>3</sub>)<sub>2</sub> (5 mol%), CuI (10 mol%), i-Pr<sub>2</sub>NEt, CH<sub>3</sub>CN, 0 °C, 65%; (e) n-Bu<sub>3</sub>SnH, PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (5 mol%), THF; (f) B, Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (4 mol%), Ph<sub>3</sub>As (16 mol%), THF, 60 °C, 59% (2 steps); (g) OsO<sub>4</sub> (10 mol%), NMO, THF,  $H_2O$ , 81% (dr = 1 : 1); (h)  $H_5IO_6$ , THF, 0 °C, 72%; (i) BnOH, In(OTf)<sub>3</sub>,  $CH_2Cl_2$ , 0 °C, 79%. Bn = benzyl, dba = dibenzylideneacetone, NMO = N-methylmorpholine-N-oxide, THF = tetrahydrofuran

magnesium bromide C, which was prepared from 3-bromopropyne, magnesium turnings, and catalytic ZnBr<sub>2</sub>, <sup>13</sup> then attacked the C5-ketone of 3 to afford the three carbon extended 6 as the major product (dr = 3.2 : 1). In this reaction, the tertiary amine of 3 would assist the stereoselective delivery of the organomagnesium reagent from the β-face of the molecule through chelation. Next, three Pd-catalyzed reactions from 6 completed the synthesis of the key intermediate 2. The fivemembered C-ring A was attached to the terminal alkyne of 6 through a Sonogashira coupling14 [PdCl2(PPh3)2 and CuI], providing tricyclic compound 7. Upon treatment of 7 with PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> and n-Bu<sub>3</sub>SnH,<sup>15</sup> the C8 position of the internal triple bond was regioselectively functionalized with the n-Bu<sub>3</sub>Sn group to form 8, presumably because the C8 position was less sterically shielded by the bulky azabicyclic AE-ring. Additionally, stereoselective hydrostannylation defined the syn-relationship of the AE- and C-rings, which later served as an important structural factor to facilitate 7-endo cyclization. The thus introduced C8-stannyl moiety of 8 was changed to the allyl group by π-allyl Stille coupling. Namely, Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> and Ph<sub>3</sub>As<sup>16,17</sup> effected the coupling between 8 and allyl bromide B to furnish triene 2. Therefore, only six steps were required for the conversion of monocycle 4 to tricycle 2, which possesses all the requisite carbons for the synthesis of the skeleton of 1. To prepare for the key radical cyclization, substrates 9 and 11a were derivatized from 2 via chemoselective manipulations of the C8allyl group. Dihydroxylation of the least sterically shielded olefin of triene 2 afforded triol 9 as a 1:1 C16-diastereomixture. Alternatively, dibenzyl acetal 11a was formed by oxidative glycolcleavage with H5IO6, followed by In(OTf)3-catalyzed acetalization with benzyl alcohol.18

Remarkably, the single radical reaction of 2 promoted cyclization not only of the requisite seven-membered B-ring, but also of the six-membered F-ring (entry 1, Table 1). Treatment of

Table 1 Radical-based cyclization/translocation/cyclization process<sup>a</sup>

Entry	Substrate	$R^2$	Product	Yield
1	2	16	12	29%
2	$9^b$	HO \$	13 <sup>c</sup>	55%
3	11a	BnO 16 OBn	14a	54%

<sup>a</sup> Conditions: substrate (1.0 equiv.), n-Bu<sub>3</sub>SnH (5 equiv.), V-40 (0.4 equiv.), toluene (0.02 M), reflux. n-Bu<sub>3</sub>SnH and V-40 (0.2 equiv.) in toluene were added by syringe pump over 3 h. Reactions were performed on a 0.10 mmol scale. b 9 was used as a 1:1 C16diastereomixture. <sup>c</sup> Product 13 was obtained as the hemiacetal forms  $(13\alpha : 13\beta = 1 : 1.5)$ . V-40 = 1,1'-azobis(cyclohexanecarbonitrile).

**Chemical Science** 

tricycle 2 with *n*-Bu<sub>3</sub>SnH and V-40 in refluxing toluene provided pentacycle **12**, with the formation of two C–C bonds (C10–11, C8–17) and five stereocenters (C8, 9, 10, 11, 17). The low yield from the convex face of the BC-ring to generate the C9-stereo-

pentacycle 12, with the formation of two C-C bonds (C10-11, C8-17) and five stereocenters (C8, 9, 10, 11, 17). The low yield (29%) of 12 from 2 was attributed to involvement of the C16olefin in undesired radical pathways. Consequently, compounds 9 and 11a with no C16 radical acceptor were next submitted to the same reaction conditions (entries 2 and 3): the yields of the corresponding pentacyclic products 13 and 14a from 9 and 11a, respectively, almost doubled (55% for 13 and 54% for 14a). 19,20 After the reaction in entry 2, 13 was obtained as C16-epimeric hemiacetals ( $13\alpha : 13\beta = 1 : 1.5$ ) due to addition of the C16-secondary hydroxy group to the C14-ketone. The structure of crystalline C16-β-alcohol 13β was unambiguously determined by X-ray crystallographic analysis (Fig. 1), which revealed its unusually complicated shape. On the other hand, the newly formed ring systems of 12 and 14a were established by judicious NMR analyses and chemical derivatizations.21

The present cascade pathway involves three essential radical reactions: 7-endo cyclization at C10 and C11, 1,5-radical translocation from C7 to C17, and 6-exo cyclization at C17 and C8. This intricate reaction course is illustrated in Scheme 3A using the transformation of 11a to 14a as an example. The bridgehead radical Int-1a is first generated from 11a by the action of the stannyl radical. To maximize the SOMO/LUMO interaction, the electron rich C11-radical of Int-1a selectively reacts with the electron deficient sp<sup>2</sup>-carbon atom at C10, thereby forming the C10-11 bond of Int-2Aa through a 7-endo cyclization. The C10stereochemistry is controlled at this stage, while the configuration of the fixed C11-bridgehead position is retained. The vinylogous ketone moiety of Int-2Aa bestows an electron deficient character to the C7-radical, which preferentially reacts with the electron rich N-α C-H bond at the proximal C17 position via facile 1,5-hydrogen abstraction.22 Next, a transannular 6-exo cyclization of the translocated nucleophilic C17-radical of Int-3a with the electrophilic C8-9 double bond produces Int-4a. This intramolecular radical addition occurs within the fused

HO OH HO CO<sub>2</sub>Et 13β' 13β CO<sub>2</sub>Et

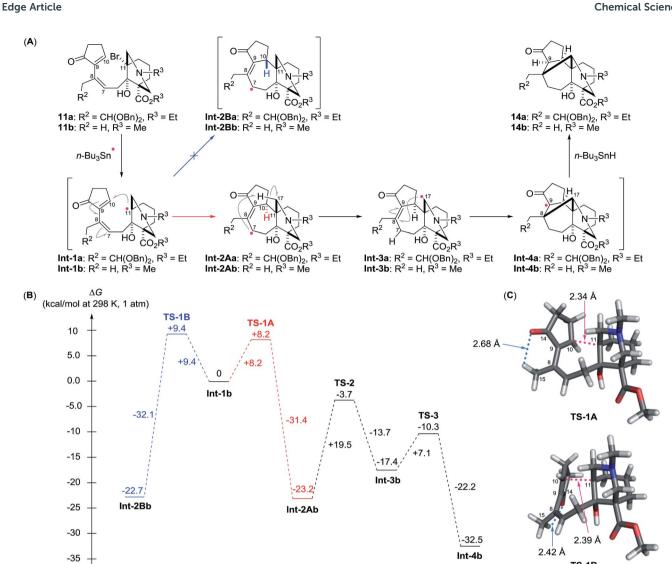
X-ray structure of 13β (CCDC1452192)

Fig. 1 X-ray structure of C16-β-alcohol 13β.

center of 14a. The multiple radical reactions proposed above were evaluated by DFT calculations of the energy diagram at the UM06-2X/6-31G(d) level of theory (Scheme 3B).23 To facilitate the calculations, we used the structurally abbreviated radical intermediates, in which R2 and R3 were changed from CH(OBn)<sub>2</sub> and Et to H and Me, respectively (see 11a and 11b). The calculated activation energy from Int-1b to Int-2Ab is smaller than that from the same Int-1b to the C10-epimeric Int-2Bb ( $\Delta G^{\ddagger} = +8.2 \text{ kcal mol}^{-1} \text{ for TS-1A}, +9.4 \text{ kcal mol}^{-1} \text{ for}$ TS-1B), supporting the observed C10-stereoselectivity of the 7endo cyclization. The higher energy of TS-1B than TS-1A would originate from the close C14O-C15H contact in TS-1B (2.42 Å, Scheme 3C) within the sum of the van der Waals radii (2.72 Å) in comparison with that in TS-1A (2.68 Å). After formation of the stable delocalized radical Int-2Ab through the endothermic reaction  $(-23.2 \text{ kcal mol}^{-1})$ , 1,5-hydrogen abstraction (Int-2Ab → Int-3b) requires a relatively large activation energy (+19.5 kcal mol<sup>-1</sup>) and gives a less stable intermediate (+5.8 kcal mol<sup>-1</sup>). However, TS-3 is lower in energy than TS-2 by -6.6kcal mol<sup>-1</sup>. Thus, the forward reaction from **Int-3b** to **Int-4b** is favored over the reverse reaction to Int-2Ab. Furthermore, 6exo cyclization gives the thermodynamically preferred radical **Int-4b**, rather than **Int-3b**  $(-15.1 \text{ kcal mol}^{-1})$ . The gain in energy of the overall process from the starting radical Int-1b to the end radical Int-4b is -32.5 kcal mol<sup>-1</sup>, corroborating the high efficiency of the present cyclization/translocation/cyclization process.

Having constructed the five fused ring system with the seven stereocenters, the remaining tasks for the synthesis of the target 1 were the construction of the six-membered D-ring and the introduction of the C13,14 and 16 stereogenic centers (Scheme 4). We anticipated that an intramolecular aldol reaction between the C14-ketone and C16-aldehyde of 15a would stereoselectively form the C13-16 bond. The aldol substrate 15a was readily prepared from hemiacetal 13\$\beta\$ through oxidative cleavage by H5IO6. However, treatment of 15a under acidic or basic conditions (e.g., aq. HCl/dioxane or DBU/benzene) resulted in either decomposition or recovery of 15a. These negative results were rationalized by DFT calculations of 15b and 16b, in which Et (R3) was replaced with Me. Although the reacting C13 and C16 atoms of 15b are in close proximity, C13-16 bond formation significantly increases the potential energy of 16b (+17.6 kcal mol<sup>-1</sup>). The bond angle ( $\theta = 101.3^{\circ}$ ) between C14–13 and C13-16 of 16b deviates significantly from the ideal value, indicating its unusually strained character. Accordingly, the retro-aldol reaction from 16a to 15a would readily occur, even when 16a was produced.

The above data and considerations led us to employ an irreversible Mukaiyama aldol reaction for cyclization of the strained D-ring (Scheme 5).<sup>24,25</sup> To realize the transformation, an alternative aldol substrate 17 was designed to have the silyl enol ether as the nucleophile and the dibenzyl acetal as the activatable electrophile. The requisite TBS enol ether structure



Scheme 3 (A) Plausible reaction pathway for the radical cascade reaction. (B) Calculated energy diagram for the cascade reaction. The changes in Gibbs free energy from Int-1b were calculated at the UM06-2X/6-31G(d) level and are shown in kcal mol<sup>-1</sup> (298 K, 1 atm, gas phase). (C) The pink dotted line represents the C10-11 bond being formed, and the cyan dotted line represents the C140-C15H interaction in the TS-1A and TS-1A represents the C140-C15H interaction in the TS-1A and TS-1B represents the C140-C15H interaction in the TS-1A represents the C140-C15H interaction in the TS-1A represents the C140-C15H represents th 1B structures.

reaction coordinate

of 17 was regioselectively constructed by applying TBSOTf and Et<sub>3</sub>N to C14-ketone **14a**. A number of Lewis acids (e.g., SnCl<sub>2</sub>, Sn(OTf)<sub>2</sub>, ZnBr<sub>2</sub>, BF<sub>3</sub>·OEt<sub>2</sub>, AlCl<sub>3</sub> and TiCl<sub>4</sub>) activated the acetal moiety, yet failed to give the requisite product because the C14oxygen atom instead of the C13 atom reacted with the C16cation. Eventually, it was found that SnCl<sub>4</sub> attained the requisite C13–16 bond formation. Treatment of 17 with SnCl<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -78 to 0 °C permitted effective construction of hexacycle 19, with installation of the C13 and 16-stereocenters. Therefore, capping the C16-hydroxy group with the benzyl group indeed inhibited the retro-aldol type reaction of adduct 19 under the reaction conditions. As illustrated by 18, the newly generated C16-stereochemistry of 19 would originate from nucleophilic attack of the silvl enol ether from the Si-face of the oxocarbenium ion, which would be fixed by the binding of SnCl<sub>4</sub> between the oxygen and nitrogen atoms.

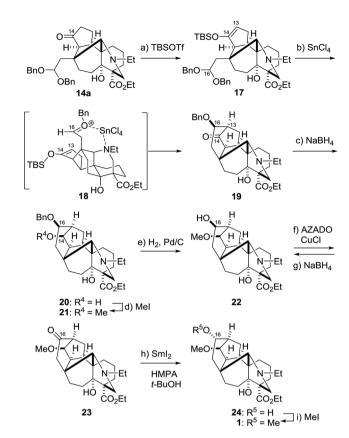
The synthesis of 1 was finalized by functional group manipulations at the C14 and C16 positions (Scheme 5). The C14-ketone of 19 was stereoselectively reduced from the convex face to provide the hydroxy group of 20. Alcohol 20 was in turn converted to methyl ether 21 using MeI and t-BuOK. Then, the C16-configuration was inverted by an oxidation/reduction sequence. After removal of the benzyl group from 21, the resultant hydroxy group of 22 was chemoselectively oxidized to the ketone within 23 in the presence of tertiary amine using the reagent combination AZADO/CuCl/2,2'-bipyridine/DMAP under air.26 However, the NaBH4 reduction of 23 only afforded its precursor 22: hydride selectively attacked from the convex face to generate the undesired β-oriented C16-alcohol. Conversely, SmI<sub>2</sub>, a one-electron reducing agent, produced the correct C16epimer.27,28 When 23 was treated with SmI2 in THF/HMPA in the presence of t-BuOH, the  $\alpha$ -oriented C16-alcohol 24 was obtained

**Chemical Science** 

Scheme 4 Attempted D-ring formation by the aldol reaction. The relative Gibbs free energy values were calculated at the UM06-2X/6-31G(d) level (298 K, 1 atm, gas phase).

as the major product (22: 24 = 1:8). It is noteworthy that the use of  $\rm H_2O$  or MeOH instead of  $\it t$ -BuOH decreased the stereoselectivity for 24 (22: 24 = 1.5:1),  $^{29}$  indicating that the slower protonation of the C16-carbanion intermediate had a beneficial effect for generating 24. Lastly, methylation of the C16-hydroxy group of 24 provided the targeted puberuline C skeleton 1 with its ten contiguous stereocenters.  $^{21}$ 

In summary, we have accomplished the expeditious synthesis of the entire 6/7/5/6/6-membered ring system 1 of the C19-diterpenoid alkaloid puberuline C [18 steps from 2-(ethoxycarbonyl)cyclohexanone (4)]. The six ABCDEF rings and the ten stereogenic centers were constructed by a series of powerful transformations. First, the double Mannich reaction cyclized the AE-ring 3, with implementation of the two tetrasubstituted carbons (C4, 11). Second, carbon elongation from 3 to 2 was realized by nucleophilic addition of C and palladiumcatalyzed couplings of A and B, introducing the C5-stereocenter and the C-ring structure. Third, the C11-bridgehead radical reaction of the highly unsaturated substrate 11a underwent a 7endo cyclization/1,5-radical translocation/6-exo cyclization process to form the B- and F-rings, establishing the stereochemistries of the two quaternary (C8, 11) and three tertiary carbons (C9, 10, 17) in a single step. Fourth, the Mukaiyama aldol reaction, followed by hydride and one-electron reductions, constructed the D-ring and the C13,14,16stereocenters. Importantly, detailed DFT-calculations fully validated the reaction course and the stereoselective outcome of the salient radical cascade reaction. Further synthetic studies of



Scheme 5 Synthesis of the hexacyclic ring system of puberuline C. Reagents and conditions: (a) TBSOTf, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 90%; (b) SnCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 to 0 °C, 46%; (c) NaBH<sub>4</sub>, MeOH, 0 °C, 96%; (d) MeI, t-BuOK, THF, 0 °C, 93%; (e) H<sub>2</sub>, Pd/C, aq. HCl, MeOH, 95%; (f) AZADO (15 mol%), CuCl (15 mol%), 2,2'-bipyridine (15 mol%), DMAP (30 mol%), CH<sub>3</sub>CN, air, 82%; (g) NaBH<sub>4</sub>, MeOH, 0 °C, 50%; (h) Sml<sub>2</sub>, t-BuOH, HMPA, THF, 22 : 24 = 1 : 8, 75% for 24; (i) MeI, t-BuOK, THF, 0 °C, 93%. AZADO = 2-azaadamantane *N*-oxyl, HMPA = hexamethylphosphoric triamide, TBS = tert-butyldimethylsilyl, Tf = trifluoromethanesulfonyl.

puberuline C based on the newly developed strategy are underway in our laboratory.

### Acknowledgements

This research was financially supported by the Funding Program a Grant-in-Aid for Scientific Research (A) to M.I. (JSPS) and (C) (JSPS) and on Innovative Areas (MEXT) to D.U. A fellowship from JSPS to K.H. is gratefully acknowledged. We thank Dr Kenji Yoza (Bruker AXS) for X-ray crystallographic analysis.

#### Notes and references

- 1 F.-P. Wang and Q.-H. Chen, The C<sub>19</sub>-Diterpene Alkaloids, in *The Alkaloids: Chemistry and Biology*, ed. G. A. Cordell, Elsevier Science, New York, 2010, vol. 69, p. 1.
- 2 K. Wiesner, M. Götz, D. L. Simmons, L. R. Fowler, F. W. Bachelor, R. F. C. Brown and G. Büchi, *Tetrahedron Lett.*, 1959, 1, 15.

**Edge Article** 

3 (a) L.-M. Sun, H.-L. Huang, W.-H. Li, Z.-D. Nan, G.-X. Zhao and C.-S. Yuan, *Helv. Chim. Acta*, 2009, **92**, 1126. Khairitdinova reported the isolation of the same compound from *Aconitum septentrionale K.*, and designated it as septontrionine. In this paper, we used puberuline C as the compound name; (b) E. D. Khairitdinova, E. M. Tsyrlina, L. V. Spirikhin, A. A. Balandina, S. K. Latypov and M. S. Yunusov, *Russ. J. Org. Chem.*, 2008, **44**, 536.

- 4 For selected reviews on total synthesis of terpenoids, see: (a) K. C. Nicolaou, D. Vourloumis, N. Winssinger and P. S. Baran, *Angew. Chem., Int. Ed.*, 2000, 39, 44; (b) R. M. Wilson and S. J. Danishefsky, *J. Org. Chem.*, 2007, 72, 4293; (c) T. J. Maimone and P. S. Baran, *Nat. Chem. Biol.*, 2007, 3, 396; (d) D. Urabe, T. Asaba and M. Inoue, *Chem. Rev.*, 2015, 115, 9207.
- 5 For representative synthetic studies on aconitine-type alkaloids, see: (a) J. L. van der Baan, J. W. F. K. Barnick, G. van Beek and F. Bickelhaupt, *Tetrahedron*, 1992, **48**, 2773; (b) G. A. Kraus and S. Kesavan, *Tetrahedron Lett.*, 2005, **46**, 1111; (c) D. F. Taber, J.-L. Liang, B. Chen and L. Cai, *J. Org. Chem.*, 2005, **70**, 8739; (d) R. M. Conrad and J. Du Bois, *Org. Lett.*, 2007, **9**, 5465; (e) H. Cheng, F.-H. Zeng, D. Ma, M.-L. Jiang, L. Xu and F.-P. Wang, *Org. Lett.*, 2014, **16**, 2299.
- 6 For reviews on synthetic studies of the related diterpenoid alkaloids, see: (a) K. J. Goodall, D. Barker and M. A. Brimble, *Synlett*, 2005, 1809; (b) E. C. Cherney and P. S. Baran, *Isr. J. Chem.*, 2011, 51, 391; (c) A. M. Hamlin, J. K. Kisunzu and R. Sarpong, *Org. Biomol. Chem.*, 2014, 12, 1846; (d) X.-Y. Liu and Y. Qin, *Asian J. Org. Chem.*, 2015, 4, 1010; (e) G. Zhu, R. Liu and B. Liu, *Synthesis*, 2015, 47, 2691.
- 7 Total synthesis of aconitine-type alkaloids from Wiesner's group. Talatisamine: (a) K. Wiesner, T. Y. R. Tsai, K. Huber, S. E. Bolton and R. Vlahov, *J. Am. Chem. Soc.*, 1974, **96**, 4990; (b) K. Wiesner, *Pure Appl. Chem.*, 1975, **41**, 93. Chasmanine: (c) S.-F. Lee, G. M. Sathe, W. W. Sy, P.-T. Ho and K. Wiesner, *Can. J. Chem.*, 1976, **54**, 1039; (d) T. Y. R. Tsai, C. S. J. Tsai, W. W. Sy, M. N. Shanbhag, W. C. Liu, S. F. Lee and K. Wiesner, *Heterocycles*, 1977, 7, 217; (e) K. Wiesner, T. Y. R. Tsai and K. P. Nambiar, *Can. J. Chem.*, 1978, **56**, 1451. 13–Desoxydelphonine: (f) K. Wiesner, *Pure Appl. Chem.*, 1979, **51**, 689.
- 8 Y. Shi, J. T. Wilmot, L. U. Nordstrøm, D. S. Tan and D. Y. Gin, *J. Am. Chem. Soc.*, 2013, 135, 14313.
- 9 C. J. Marth, G. M. Gallego, J. C. Lee, T. P. Lebold, S. Kulyk, K. G. M. Kou, J. Qin, R. Lilien and R. Sarpong, *Nature*, 2015, 528, 493.
- 10 For conversion of the naturally occurring aconitine-type alkaloids to the puberuline frameworks, see: (a) T. Amiya and T. Shima, Bull. Chem. Soc. Jpn., 1967, 40, 1957; (b)
  O. E. Edwards, Can. J. Chem., 1981, 59, 3039; (c) P. Tang, L. Wang, Q.-F. Chen, Q.-H. Chen, X.-X. Jian and F.-P. Wang, Tetrahedron, 2012, 68, 5031.
- 11 α-Alkoxy bridgehead radicals were utilized for the intra- and intermolecular reactions in our laboratory: (a) D. Urabe,
   H. Yamaguchi and M. Inoue, Org. Lett., 2011, 13, 4778; (b)

- D. Urabe, M. Nagatomo, K. Hagiwara, K. Masuda and M. Inoue, *Chem. Sci.*, 2013, 4, 1615; (*c*) K. Murai, S. Katoh, D. Urabe and M. Inoue, *Chem. Sci.*, 2013, 4, 2364; (*d*) D. Kamimura, D. Urabe, M. Nagatomo and M. Inoue, *Org. Lett.*, 2013, 15, 5122; (*e*) M. Nagatomo, M. Koshimizu, K. Masuda, T. Tabuchi, D. Urabe and M. Inoue, *J. Am. Chem. Soc.*, 2014, 136, 5916; (*f*) T. Asaba, Y. Katoh, D. Urabe and M. Inoue, *Angew. Chem., Int. Ed.*, 2015, 54, 14457; (*g*) M. Nagatomo, K. Hagiwara, K. Masuda, M. Koshimizu, T. Kawamata, Y. Matsui, D. Urabe and M. Inoue, *Chem.-Eur. J.*, 2016, 22, 222.
- 12 G. A. Kraus and J. Shi, J. Org. Chem., 1991, 56, 4147.
- 13 H. P. Acharya, K. Miyoshi and Y. Kobayashi, *Org. Lett.*, 2007, 9, 3535.
- 14 For a review, see: J. A. Marsden and M. M. Haley, Cross-Coupling Reactions to sp Carbon Atoms, in *Metal-Catalyzed Cross-Coupling Reactions*, ed. A. de Meijere and F. Diederich, Wiley-VCH, Weinheim, 2004, vol. 1, p. 317.
- 15 J.-F. Betzer, F. Delaloge, B. Muller, A. Pancrazi and J. Prunet, J. Org. Chem., 1997, 62, 7768.
- 16 For a review, see: V. Farina, V. Krishnamurthy and W. J. Scott, *Org. React.*, 1997, **50**, 1.
- 17 V. Farina and B. Krishnan, J. Am. Chem. Soc., 1991, 113, 9585.
- 18 B. M. Smith and A. E. Graham, *Tetrahedron Lett.*, 2006, 47, 9317.
- 19 For recent reviews on radical cascade reactions, see: (a) A. Baralle, A. Baroudi, M. Daniel, L. Fensterbank, J.-P. Goddard, E. Lacôte, M.-H. Larraufie, G. Maestri, M. Malacria and C. Ollivier, Radical Cascade Reactions, in *Encyclopedia of Radicals in Chemistry, Biology and Materials*, ed. C. Chatgilialoglu and A. Studer, John Wiley & Sons Ltd, Chichester, 2012, vol. 2, p. 729; (b) U. Wille, *Chem. Rev.*, 2013, 113, 813; (c) J.-R. Chen, X.-Y. Yu and W.-J. Xiao, *Synthesis*, 2015, 47, 604.
- 20 For recent reviews on cascade reactions in natural product synthesis, see: (a) K. C. Nicolaou, D. J. Edmonds and P. G. Bulger, *Angew. Chem., Int. Ed.*, 2006, **45**, 7134; (b) H. Pellissier, *Chem. Rev.*, 2013, **113**, 442.
- 21 For the structural determinations, see ESI.†
- 22 For selected reports on the radical translocation/cyclization process, see: (a) D. C. Lathbury, P. J. Parsons and I. Pinto, J. Chem. Soc., Chem. Commun., 1988, 81; (b) D. P. Curran, D. Kim, H. T. Liu and W. Shen, J. Am. Chem. Soc., 1988, 110, 5900; (c) V. Snieckus, J. C. Cuevas, C. P. Sloan, H. Liu and D. P. Curran, J. Am. Chem. Soc., 1990, 112, 896; (d) J. Robertson, M. A. Peplow and J. Pillai, Tetrahedron Lett., 1996, 37, 5825; (e) F. Beaufils, F. Dénès, B. Becattini, P. Renaud and K. Schenk, Adv. Synth. Catal., 2005, 347, 1587; (f) M. Fujitani, M. Tsuchiya, K. Okano, K. Takasu, M. Ihara and H. Tokuyama, Synlett, 2010, 822; (g) J. K. Vellucci and C. M. Beaudry, Org. Lett., 2015, 17, 4558. For reviews, see: ; (h) J. Robertson, J. Pillai and R. K. Lush, Chem. Soc. Rev., 2001, 30, 94; (i) F. Dénès, F. Beaufils and P. Renaud, Synlett, 2008, 2389.
- 23 (a) Y. Zhao and D. G. Truhlar, Theor. Chem. Acc., 2008, 120, 215; (b) Y. Zhao and D. G. Truhlar, Acc. Chem. Res., 2008, 41, 157.

24 T. Mukaiyama, T. Izawa and K. Saigo, Chem. Lett., 1974, 323.

**Chemical Science** 

- 25 For a review on Mukaiyama aldol reaction in natural product synthesis, see: S. B. J. Kan, K. K.-H. Ng and I. Paterson, *Angew. Chem., Int. Ed.*, 2013, **52**, 9097.
- 26 Y. Sasano, S. Nagasawa, M. Yamazaki, M. Shibuya, J. Park and Y. Iwabuchi, *Angew. Chem., Int. Ed.*, 2014, **53**, 3236.
- 27 P. Girard, J. L. Namy and H. B. Kagan, *J. Am. Chem. Soc.*, 1980, **102**, 2693.
- 28 For a book, see: *Organic Synthesis using Samarium Diiodide A Practical Guide*, ed. D. J. Procter, R. A. Flowers II and T. Skrydstrup, RSC publishing, Cambridge, 2010.
- 29 For the effects of proton source on stereoselective reduction of ketones, see: (*a*) S. K. Upadhyay and S. Hoz, *J. Org. Chem.*, 2011, **76**, 1355; (*b*) P. R. Chopade, E. Prasad and R. A. Flowers ll, *J. Am. Chem. Soc.*, 2004, **126**, 44.