

Cite this: *Chem. Sci.*, 2016, **7**, 4453

Highly efficient synthesis and stereoselective migration reactions of chiral five-membered aza-spiroindolenines: scope and mechanistic understanding†

Qing-Feng Wu, Chao Zheng,* Chun-Xiang Zhuo and Shu-Li You*

Received 14th January 2016
Accepted 29th March 2016

DOI: 10.1039/c6sc00176a

www.rsc.org/chemicalscience

Introduction

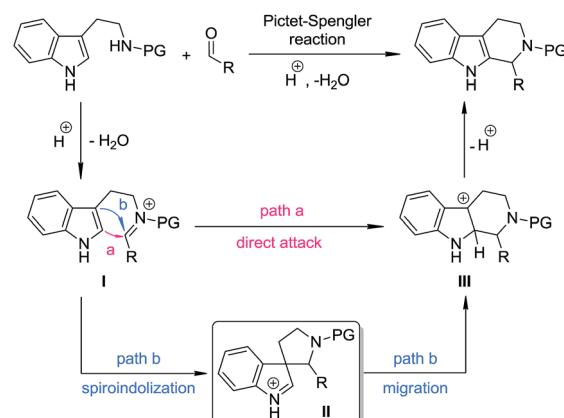
The Pictet–Spengler reaction¹ has long been recognized as one of the most powerful methods for the synthesis of tetrahydro- β -carbolines and other indole alkaloids with more complex structures.² Despite their widespread use, mechanistically, Pictet–Spengler-type reactions still need further investigation.^{1c} In general, there are two possible pathways (Scheme 1): the reaction can occur through either the direct attack of the C2 position of indole towards the iminium moiety (path a) or the stepwise spiroindolization/migration sequence (path b). Although experimental evidence supporting the latter mechanism is documented a number of times in the literature,³ to the best of our knowledge, the five-membered aza-spiroindolenine, the key intermediate (**II**) for the asymmetric Pictet–Spengler reaction, has not been isolated yet.⁴

As a part of our ongoing program of developing the catalytic asymmetric dearomatization (CADA) reaction,⁵ we reported in 2012 the enantioselective synthesis of carbocyclic spiroindolenines **2** with excellent enantioselectivity *via* Ir-catalyzed allylic dearomatization reactions of indole derivatives **1**.^{6–8} Highly stereoselective allyl migration of **2** was achieved when treated with a catalytic amount of TsOH (eqn (1), Scheme 2).⁹ A mechanism involving a “three-center-two-electron” transition state is proposed according to DFT calculations and it is suggested that the electronic properties of the migratory group exert a great influence on the profile of the migration.¹⁰

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China.
E-mail: zhengchao@sioc.ac.cn; slyou@sioc.ac.cn; Web: <http://shuliyou.sioc.ac.cn/>

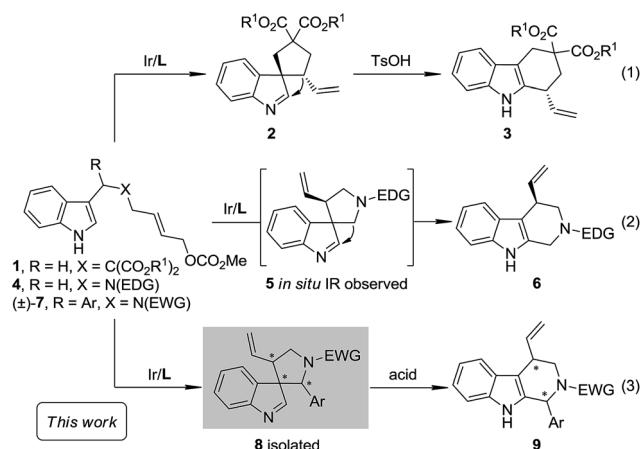
† Electronic supplementary information (ESI) available: Experimental procedures, compound characterization data and computational details. CCDC 1018536–1018541. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c6sc00176a

Intriguingly, the reaction of the *N*-tethered substrate analogue **4** with an electron-donating group (EDG) attached to the nitrogen atom under Ir-catalysis yields the unexpected tetrahydro- β -carbolines **6**, with the tether that was originally attached to the C3 position of indole moving to the C2 position. A dearomatized five-membered aza-spiroindolenine intermediate **5** was observed by *in situ* IR spectroscopy, but the attempts to isolate this intermediate were all unsuccessful (eqn (2), Scheme 2).^{11–13} Recently, by tuning the electronic properties of the *N*-linker, we achieved the synthesis of five-membered aza-spiroindolenines **8** with three contiguous stereogenic centers, from indole derivatives (\pm)-**7** by Ir-catalysis. Upon isolation, **8** were exposed to acidic conditions, and the tetrahydro- β -carbolines **9** were delivered *via* aryl iminium migration (eqn (3), Scheme 2). Interestingly, the reaction pattern of the migration step highly depends on the relative stereochemistry of **8**. Guided by further mechanistic investigations including DFT calculations, a one-



Scheme 1 The possible pathways for Pictet–Spengler reactions.





Scheme 2 Ir-Catalyzed intramolecular asymmetric allylic dearomatization and subsequent migration.

pot asymmetric synthesis of Pictet-Spengler-type products **9**, with two stereogenic centers, from (\pm) -**7** was realized under enhanced reaction conditions. Herein, we report the full account of this study.

Results and discussion

Asymmetric synthesis of five-membered aza-spiroindolines

Our study began by testing indol-3-yl phenyl allylic carbonates (\pm) -**7** in Ir-catalyzed intramolecular allylic dearomatization

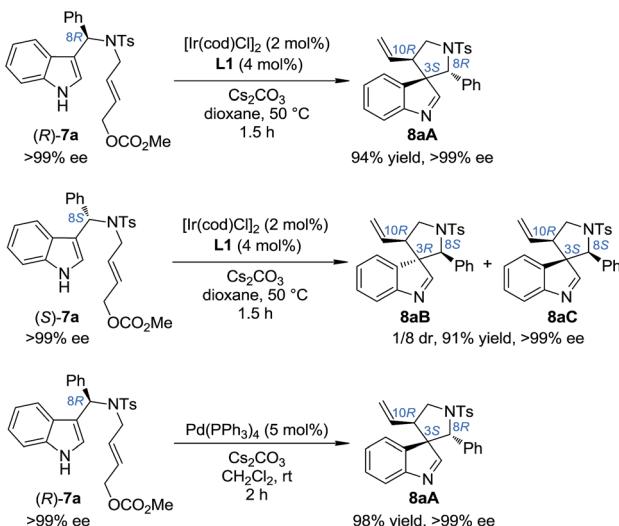
reactions (Table 1).¹⁴ We envisioned that the addition of an electron-withdrawing group (EWG) to the *N*-linker might stabilize the desired five-membered aza-spiroindolines **8** by reducing their activity toward the migration process. However, the existence of a benzylic substituent in **7** posed additional challenges for stereochemical control. Under the optimal conditions, three diastereoisomers of **8** were obtained in relatively good yields and excellent enantiomeric excess (ee). The structures and absolute configurations [(*3S,8R,10R*) for **8aA**; (*3R,8S,10R*) for **8aB** and (*3S,8S,10R*) for **8aC**] were determined unambiguously by X-ray crystallographic analysis.¹⁶ When the protecting group on the nitrogen atom was Ts, the reactions of substrates bearing either an EDG (5-OMe, **7b** and 6-Me, **7c**) or an EWG (5-Br, **7d**) on the indole core all led to their corresponding spiroindoline products in excellent yields and ee. Substrates bearing various aryl groups (Ar) on the tether (**7e–k**) were well tolerated but showed very different activities. *p*-Tol substituted **7e** gave the best diastereomeric ratio (dr) value of all of the N-Ts linked substrates tested. Notably, when an *o*-ClC₆H₄ substituted substrate **7j** was used, the two diastereoisomers **8jA** and **8jC** could not be separated by column chromatography, giving a good combined isolated yield with an excellent ee. On the other hand, the reaction proceeded smoothly for the substrates containing different *N*-protecting groups (**7l–q**), affording the five-membered aza-spiroindoline products in good yields and ee with a moderate to excellent dr. Among them, the *N*-Boc protected substrates bearing either an EDG (**7o** and **7p**) or an EWG (**7q**) at various positions all gave a good dr and excellent ee.

Table 1 Scope of the Ir-catalyzed allylic dearomatization reaction^a

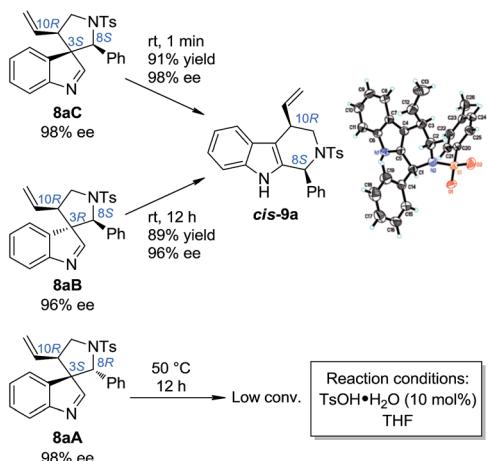
Substrate	Yield	dr	ee	Yield	dr	ee	Yield	dr	ee
8a(A–C) dr: 10/1/8 yield: 42%, 4%, 37% ee: 98%, 97%, 98%	42%	4%	37%	44%	10%, 32%	92%, 94%, 98%	42%	7%	34%
8b(A–C) dr: 10/2/7 yield: 44%, 10%, 32% ee: 92%, 94%, 98%	44%	10%	32%	42%	23%, 21%	95%, 96%, 98%	42%	7%	34%
8c(A–C) dr: 10/2/7 yield: 48%, 11%, 30% ee: 97%, 96%, 97%	48%	11%	30%	42%	23%, 21%	95%, 96%, 97%	42%	7%	34%
8d(A–C) dr: 10/5/5 yield: 48%, 11%, 30% ee: 97%, 96%, 97%	48%	11%	30%	42%	23%, 21%	95%, 96%, 97%	42%	7%	34%
8e(A–C) dr: 10/1/9 yield: 44%, 4%, 39% ee: 95%, 80%, 95%	44%	4%	39%	42%	23%, 21%	95%, 96%, 97%	42%	7%	34%
8f(A–C) dr: 10/3/6 yield: 38%, 13%, 30% ee: 94%, 89%, 96%	38%	13%	30%	38%	13%, 30%	94%, 89%, 96%	42%	7%	34%
8i(A–C) dr: 10/2/8 yield: 42%, 7%, 34% ee: 91%, 76%, 92%	42%	7%	34%	38%	13%, 30%	91%, 76%, 92%	42%	7%	34%
8j(A–C) dr: 10/3/7 yield = 79% (A+C), 15% ee: 93%, 98%, 95%	79%	15%	93%, 98%, 95%	79%	15%	93%, 98%, 95%	42%	7%	34%
8k(A–C) dr: 10/4/6 yield: 43%, 15%, 25% ee: 90%, 83%, 95%	43%	15%	25%	41%	14%, 40%	90%, 83%, 95%	41%	13%, 37%	90%, 83%, 95%
8l(A–C) dr: 10/3/7 yield: 44%, 8%, 35% ee: 93%, 90%, 95%	44%	8%	35%	41%	3%, 40%	93%, 90%, 95%	43%	5%, 36%	90%, 83%, 95%
8m(A–C) dr: 10/2/7 yield: 44%, 14%, 40% ee: 92%, 78%, 92%	44%	14%	40%	41%	3%, 40%	92%, 78%, 92%	43%	5%, 36%	90%, 83%, 95%
8n(A–C) dr: 10/1/9 yield: 41%, 3%, 40% ee: 93%, N. D., 94%	41%	3%	40%	41%	3%, 40%	93%, N. D., 94%	43%	5%, 36%	90%, N. D., 94%
8o(A–C) dr: 10/1/9 yield: 41%, 3%, 37% ee: 90%, N. D., 94%	41%	3%	37%	41%	3%, 37%	90%, N. D., 94%	43%	6%, 38%	89%, N. D., 94%
8p(A–C) dr: 10/1/9 yield: 43%, 5%, 36% ee: 90%, N. D., 92%	43%	5%	36%	43%	5%, 36%	90%, N. D., 92%	43%	6%, 38%	89%, N. D., 92%
8q(A–C) dr: 10/1/8 yield: 39%, 6%, 38% ee: 89%, N. D., 91%	39%	6%	38%	39%	6%, 38%	89%, N. D., 91%	39%	6%, 38%	89%, N. D., 91%

^a Reaction conditions: 0.004 mmol of [Ir(cod)Cl]₂, 0.008 mmol of **L1** and 0.2 mmol of **7**, 0.4 mmol of Cs₂CO₃ in dioxane (2.0 mL) at 50 °C. The catalyst was prepared via ¹PrNH₂ activation.¹⁵ The dr values are determined by ¹H NMR of the crude reaction mixture. The ee values are determined by HPLC analysis. The isolated yields are reported. N. D.: not determined.





Scheme 3 The reaction patterns of enantiopure substrates.



Scheme 4 Regio- and stereoselective migration of the five-membered aza-spiroindolenines.

On the origin of the three diastereomers of the five-membered aza-spiroindolenines

In order to gain deep insights into the origin of the three diastereoisomers of **8** from (\pm) -**7**, the enantiopure substrates **(R)**-**7a** and **(S)**-**7a** obtained by preparative HPLC were subjected to the standard dearomatization reaction conditions (Scheme 3). **(R)**-**7a** was converted to **8aA** in a 94% yield and >99% ee. On the other hand, **(S)**-**7a** was transformed to **8aB** and **8aC** in a 1/8 dr and >99% ee. The ratio of the three diastereoisomers of **8a**(**A–C**) and their stereochemistry are in agreement with the reaction results for (\pm) -**7a** (Table 1). In addition, **8aA** could also be obtained from **(R)**-**7a** in 98% yield with >99% ee in 2 h, with $\text{Pd}(\text{PPh}_3)_4$ as an achiral catalyst.

Stereoselective aryl iminium migration

With the optimized method for the construction of the highly enantioenriched aryl substituted five-membered aza-

spiroindolenines in hand, we next sought to identify conditions that are suitable for the subsequent migration process. It was quite interesting to find that the three diastereoisomers of **8a** exhibit extremely different reactivities when exposed to our previously established migration conditions (10 mol% $\text{TsOH}\cdot\text{H}_2\text{O}$ in THF at room temperature) (Scheme 4).⁹ **8aC** was easily converted to tetrahydro- β -carboline **cis-9a** in a satisfactory yield and preserved ee within 1 minute. X-Ray crystallographic analysis of **cis-9a** showed that the aryl iminium moiety migrated to the C2 position of indole and the stereochemistry (*8S,10R*) of the benzylic position can be maintained, which is consistent with our understanding of the migration mechanism.¹⁰ The minor diastereoisomer **8aB** also underwent similar regio- and stereoselective migration to afford **cis-9a** smoothly, albeit with a longer reaction time (12 h) at room temperature. However, the other major diastereoisomer **8aA** remained intact even at an elevated temperature (50 °C).¹⁷

To test the generality of the regio- and stereoselective N-Ts aryl iminium migration reactions, the most reactive diastereoisomer **8C** possessing different substituents was subjected to the above reaction conditions (Table 2). In all cases, the absolute configuration at the benzylic positions and the ee values of the compounds were well maintained, leading to the corresponding tetrahydro- β -carbolines **cis-9** smoothly, in a highly enantioenriched form.

DFT calculations

DFT calculations¹⁸ (PBE1PBE/6-311+G(d,p) level of theory) were carried out at this stage to investigate why the diastereoisomers of **8** exhibit dramatically different reactivities toward the migration process (Fig. 1). The complexes of the three diastereoisomers (**8aA**, **8aB** and **8aC**) with TsOH , named **A**· TsOH (0.0 kcal mol⁻¹), **B**· TsOH (-0.9 kcal mol⁻¹) and **C**· TsOH (1.5 kcal mol⁻¹), respectively, were set as the starting points for their corresponding migration processes. All of the aryl N-Ts iminium migrations were characterized to be stepwise.¹⁹ The **C**· TsOH complex was the most prone to the reaction. The C3–C8 bond was cleaved *via* the ring-opening transition state **TS1-C** (17.8 kcal mol⁻¹), leading to the key intermediate **INT1-C** (16.7 kcal mol⁻¹) with an *E*-iminium moiety formed. Natural bond orbital (NBO) analysis of this intermediate revealed the existence of interactions between the indole ring and the N-Ts iminium part. According to the second-order perturbation theory analysis, the stabilization energy *E*(2) from the occupied $\pi(\text{C}2-\text{C}3)$ orbital to the unoccupied $\pi^*(\text{N}-\text{C}8)$ orbital is estimated to be 5.69 kcal mol⁻¹. It is speculated that this kind of interaction might account for the highly stereoselective migration phenomenon. Subsequently, the iminium carbon C8 is attacked by the indole C2 position *via* a chair-like six-membered-ring transition state **TS2-C** (18.3 kcal mol⁻¹) to afford the protonated tetrahydro- β -carboline complex **INT2-C** (6.8 kcal mol⁻¹). After the following deprotonation step (not shown in Fig. 1), the final complex **cis-Pro** (-10.1 kcal mol⁻¹) is generated. For the second reactive species **B**· TsOH , a similar reaction profile was found, except a *Z*-iminium intermediate **INT1-B** (16.9 kcal mol⁻¹) is involved, which makes the ring-



Table 2 Scope of the regio- and stereoselective migration^a

9a 91% yield 98% ee (98%)	9b 90% yield 98% ee (98%)
9c 97% yield 97% ee (97%)	9d 90% yield 92% ee (92%)
9e 93% yield 96% ee (95%)	9f 91% yield 94% ee (96%)
9g 95% yield 93% ee (92%)	9h 90% yield 96% ee (95%)
9i 91% yield 96% ee (95%)	9j 90% yield 94% ee (95%)
9k 91% yield 96% ee (95%)	9l 90% yield 94% ee (95%)
9m 86% yield 93% ee (92%)	9n 90% yield 97% ee (97%)
9o 95% yield 92% ee (94%)	9p 94% yield 93% ee (92%)
9q 94% yield 93% ee (92%)	9r 94% yield 93% ee (92%)

^a Reaction conditions: 0.1 mmol of **8**, 0.01 mmol of $\text{TsOH} \cdot \text{H}_2\text{O}$ in THF (2 mL) at rt. The ee values are determined by HPLC analysis. The numbers in the parentheses are the ee values of the starting materials **8C**. The isolated yields are reported.

closure (**TS2-B**, 22.0 kcal mol⁻¹), the rate-limiting step, slightly higher in energy. If an *E*-iminium intermediate was forced to be generated from **B**· TsOH , the ring-closure step must proceed through an unfavorable boat-like six-membered-ring transition state which further raises the energetic barrier. The migration

process of complex **A**· TsOH shares something in common with that of **B**· TsOH because an unfavorable *Z*-iminium intermediate (**INT1-A**, 17.6 kcal mol⁻¹) is also generated. An additional detrimental factor for the migration process of **A**· TsOH was also identified in the key ring-closure transition state **TS2-A**

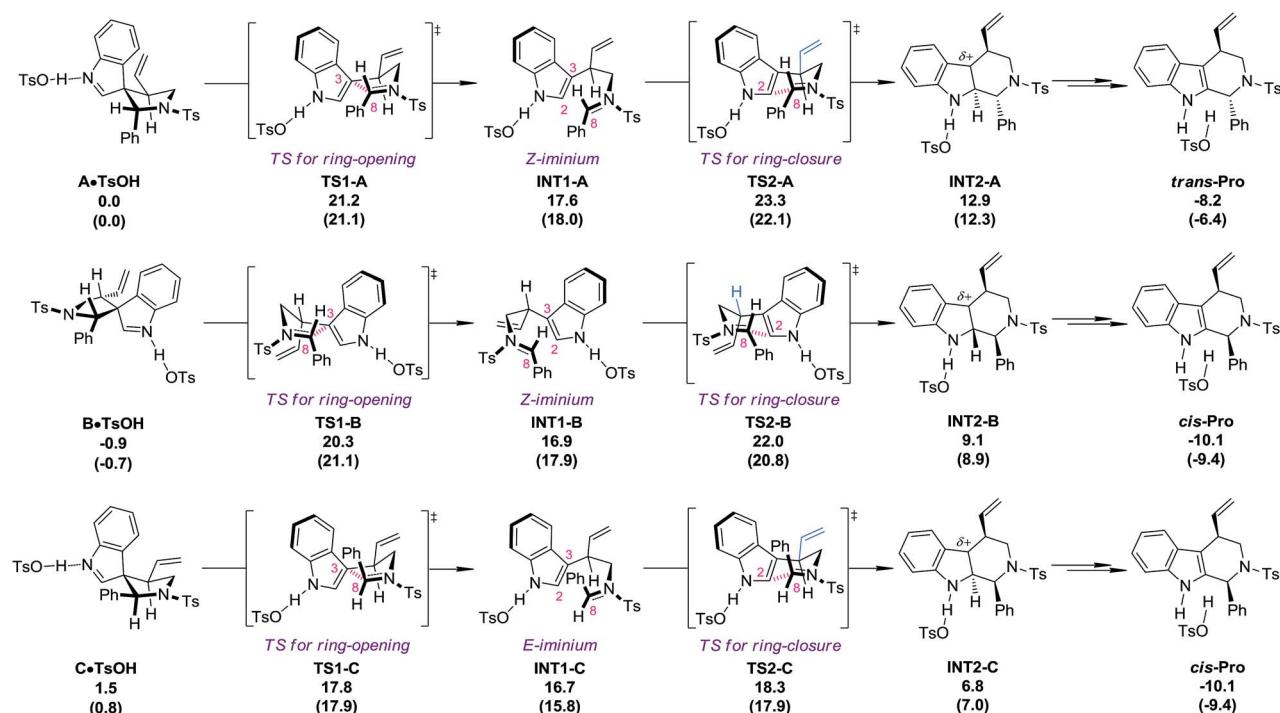
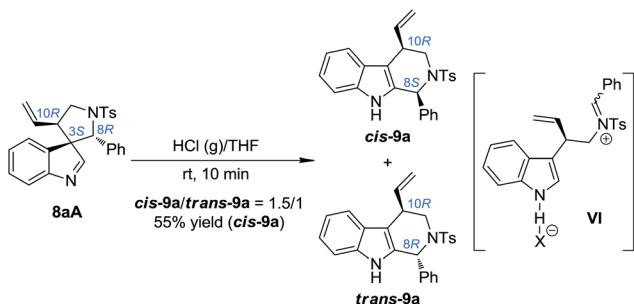
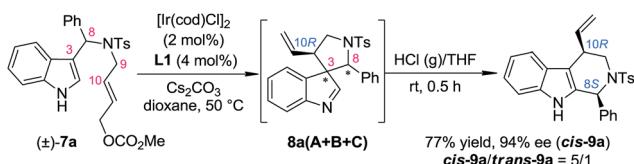


Fig. 1 The reaction profiles for the TsOH catalyzed migration processes of the diastereoisomers of **8**. Calculated at a PBE1PBE/6-311+G(d,p) level of theory. ΔG_{THF} and (ΔE_{THF}) values are in kcal mol⁻¹. The transition states of the deprotonation steps are not shown.





Scheme 5 Migration reaction for the diastereoisomer 8aA.



Scheme 6 Optimized reaction conditions for the one-pot synthesis of tetrahydro-β-carbolines.

(23.3 kcal mol⁻¹). The vinyl moiety on the tether points toward the indole ring, which will cause a stronger steric repulsion compared with that observed in **TS2-B** where a hydrogen atom instead of the vinyl group is found in that place (highlighted in blue). As a result, **TS2-A** ranks as the most high-energy one among the three ring-closure transition states. In general, the order of reactivity toward the TsOH-catalyzed migration of the three diastereoisomers is qualitatively reproduced by computations. Besides, the calculated energetic barriers for the allyl migration are always higher than those of the aryl N-Ts iminium migration, which is also consistent with experimental observations.

One-pot asymmetric allylic dearomatization/migration procedure

Based on the above mechanistic understanding, we decided to develop a one-pot highly stereoselective procedure to synthesize tetrahydro-β-carbolines **9** from (±)-7, which requires the full conversion of all of the three diastereoisomers of the intermediate **8**. However, the unreactive **8A** is the major challenge. Since the absolute configuration of the benzylic stereogenic center (C8) of **8A** is the opposite of that of **8B** and **8C**, even if **8A** could take part in the migration reaction, and at the same time all of the chiral information at the benzylic position was preserved, only a mixture of **cis-9** and **trans-9** with a 1 : 1 ratio could be obtained [the ratio of **8A** : (**8B** + **8C**) is about 1 : 1]. We speculated that using a stronger acid as the catalyst might be a possible solution to this issue. The reasons for this choice include: (1) the spiroindolenines can be better activated by stronger acids and thus the ring-opening step can be better facilitated; (2) stronger acids might lead to a more loosely bound ion-pair in **INT-1**, which in turn might weaken or even destroy the interaction between the indole ring and the aryl

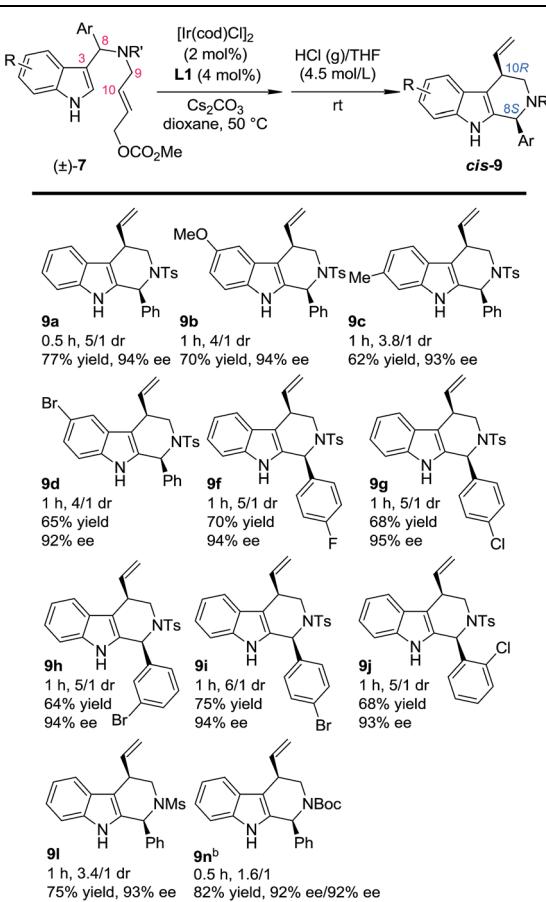
iminium moiety, thus reducing the possibility of the maintenance of the chiral information possessed by the migratory group. To our delight, we discovered that freshly prepared saturated HCl (g)/THF was an efficient acid system to promote the migration of **8aA**, providing two diastereoisomers of the tetrahydro-β-carbolines **cis-9a** and **trans-9a** in a 1.5/1 dr (Scheme 5).¹⁴ Intriguingly, the major diastereoisomer obtained (**cis-9a**) was the same as that from **8aB** or **8aC** (8S,10R). This means that the absolute configuration of the benzylic stereogenic center of **8aA** can indeed be inverted during the migration process in a more acidic medium, and a free iminium species (**IV**) might be formed during the migration. The diastereoselectivity of the migration step might be induced by the allylic stereogenic center adjacent to the C3 position of the indole core. Encouraged by these findings, we further identified enhanced reaction conditions for the one-pot synthesis of tetrahydro-β-carbolines **9** from (±)-7 via a dearomatization/migration sequence.¹⁴ Under the optimal conditions, the desired product **cis-9a** was obtained in a 77% yield and good dr (**cis-9a/****trans-9a** = 5/1) (Scheme 6).

With this newly established one-pot asymmetric allylic dearomatization/migration procedure, various substrates were tested to examine the generality of the reaction (Table 3). Reactions of substrates bearing either an EDG (5-OMe, **7b** and 6-Me, **7c**) or an EWG (5-Br, **7d**) on the indole core all led to their corresponding Pictet-Spengler-type products in moderate dr and yields with excellent ee (3.8/1-4/1 dr, 62–70% yield, 92–94% ee). Substrates bearing aryl groups with various substituted patterns (*p*-F, **9f**; *p*-Cl, **9g**; *m*-Br, **9h**; *p*-Br, **9i** and *o*-Cl, **9j**) on the tether underwent the dearomatization/migration reaction smoothly to give the tetrahydro-β-carbolines with satisfactory results (5/1–6/1 dr, 64–75% yields, 93–95% ee). The different *N*-protecting groups such as Ms and Boc (**9l** and **9n**) were also tolerated, affording the corresponding products in good yields and enantioselectivity (92–93% ee), albeit with a decreased dr.

Conclusions

In summary, we have developed highly efficient synthesis and stereoselective migration reactions of enantioenriched five-membered aza-spiroindolenines on the basis of a deep understanding of the reaction mechanism and the rational design of the substrates. Ir-catalyzed intramolecular allylic dearomatization of the racemic indol-3-yl aryl substituted allylic carbonates led to the highly enantioenriched five-membered aza-spiroindolenines which are isolated and fully characterized for the first time. A detailed investigation of the reaction patterns of the three diastereoisomers paved the way to a highly regio- and stereoselective migration whose mechanism was further understood by DFT calculations. In addition, a one-pot asymmetric allylic dearomatization/migration sequence was developed under enhanced reaction conditions, providing a novel strategy to synthesize the Pictet-Spengler-type products with an additional allylic stereogenic center adjacent to the C3 position of indole.



Table 3 Scope of the one-pot asymmetric allylic dearomatization/migration procedure^a

^a Reaction conditions: for step 1: the same as shown in Table 1. For step 2: the reaction residue of step 1 containing intermediate 8 in HCl (g)/THF (4.0 mL, 4.5 mol L⁻¹) at rt. The dr values are determined by ¹H NMR of the crude reaction mixture. The ee values are determined by HPLC analysis. The isolated yields are reported. ^b Combined yield of the two diastereoisomers.

Acknowledgements

We thank the National Basic Research Program of China (973 Program 2015CB856600) and the National Natural Science Foundation of China (21332009, 21302209 and 21421091) for generous financial support.

Notes and references

- For reviews: (a) E. D. Cox and J. M. Cook, *Chem. Rev.*, 1995, **95**, 1797; (b) J. Royer, M. Bonin and L. Micouin, *Chem. Rev.*, 2004, **104**, 2311; (c) J. Stöckigt, A. P. Antonchick, F. Wu and H. Waldmann, *Angew. Chem., Int. Ed.*, 2011, **50**, 8538.
- For reviews: (a) S. Cacchi and G. Fabrizi, *Chem. Rev.*, 2005, **105**, 2873; (b) M. Bandini, A. Melloni, S. Tommasi and A. Umani-Ronchi, *Synlett*, 2005, 1199; (c) M. Bandini and A. Eichholzer, *Angew. Chem., Int. Ed.*, 2009, **48**, 9608; (d)

L. Joucla and L. Djakovitch, *Adv. Synth. Catal.*, 2009, **351**, 673; (e) M. Zeng and S.-L. You, *Synlett*, 2010, 1289; (f) G. Bartoli, G. Bencivenni and R. Dalpozzo, *Chem. Soc. Rev.*, 2010, **39**, 4449; (g) S. Cacchi and G. Fabrizi, *Chem. Rev.*, 2011, **111**, PR215.

- (a) A. H. Jackson and A. E. Smith, *Tetrahedron*, 1965, **21**, 989; (b) A. H. Jackson and A. E. Smith, *Tetrahedron*, 1968, **24**, 403; (c) A. H. Jackson and P. Smith, *Tetrahedron*, 1968, **24**, 2227; (d) A. H. Jackson, B. Naidoo and P. Smith, *Tetrahedron*, 1968, **24**, 6119; (e) J. S. L. Ibaceta-Lizana, A. H. Jackson, N. Prasitpan and P. V. R. Shannon, *J. Chem. Soc., Perkin Trans. 2*, 1987, 1221; (f) Q. Nguyen, T. Nguyen and T. G. Driver, *J. Am. Chem. Soc.*, 2013, **135**, 620.
- The five-membered aza-spiroindolines can be obtained via condensation of tryptamine derivatives and aldehydes under reduction conditions: (a) J. R. Williams and L. R. Unger, *Chem. Commun.*, 1970, 1605; (b) F. Ungemach and J. M. Cook, *Heterocycles*, 1978, **9**, 1089; (c) P. D. Bailey, *J. Chem. Res., Synop.*, 1987, 202; (d) T. Kawate, M. Nakagawa, K. Ogata and T. Hino, *Heterocycles*, 1992, **33**, 801; (e) R. V. Edwankar, C. R. Edwankar, O. A. Namjoshi, J. R. Deschamps and J. M. Cook, *J. Nat. Prod.*, 2012, **75**, 181; (f) M. Jia, G. Cera, D. Perrotta, M. Monari and M. Bandini, *Chem.-Eur. J.*, 2014, **20**, 9875.
- For selected examples of CADA reactions of indoles: (a) J. F. Austin, S.-G. Kim, C. J. Sinz, W.-J. Xiao and D. W. C. MacMillan, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 5482; (b) S. B. Jones, B. Simmons and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2009, **131**, 13606; (c) L. M. Repka, J. Ni and S. E. Reisman, *J. Am. Chem. Soc.*, 2010, **132**, 14418; (d) T. Benkovic, I. A. Guzei and T. P. Yoon, *Angew. Chem., Int. Ed.*, 2010, **49**, 9153; (e) S. Zhu and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2012, **134**, 10815; (f) Z. Zhang and J. C. Antilla, *Angew. Chem., Int. Ed.*, 2012, **51**, 11778; (g) J. E. Spangler and H. M. L. Davies, *J. Am. Chem. Soc.*, 2013, **135**, 6802; (h) H. Xiong, H. Xu, S. Liao, Z. Xie and Y. Tang, *J. Am. Chem. Soc.*, 2013, **135**, 7851; (i) M.-C. Tong, X. Chen, J. Li, R. Huang, H. Tao and C.-J. Wang, *Angew. Chem., Int. Ed.*, 2014, **53**, 4680; (j) H. M. Nelson, S. H. Reisberg, H. P. Shunatona, J. S. Patel and F. D. Toste, *Angew. Chem., Int. Ed.*, 2014, **53**, 5600; (k) L. Liao, C. Shu, M. Zhang, Y. Liao, X. Hu, Y. Zhang, Z. Wu, W. Yuan and X. Zhang, *Angew. Chem., Int. Ed.*, 2014, **53**, 10471; (l) C. Romano, M. Jia, M. Monari, E. Manoni and M. Bandini, *Angew. Chem., Int. Ed.*, 2014, **53**, 13854; (m) Y.-C. Zhang, J.-J. Zhao, F. Jiang, S.-B. Sun and F. Shi, *Angew. Chem., Int. Ed.*, 2014, **53**, 13912; (n) C. Shen, R.-R. Liu, R.-J. Fan, Y.-L. Li, T.-F. Xu, J.-R. Gao and Y.-X. Jia, *J. Am. Chem. Soc.*, 2015, **137**, 4936; (o) X. Zhao, X. Liu, H. Mei, J. Guo, L. Lin and X. Feng, *Angew. Chem., Int. Ed.*, 2015, **54**, 4032.
- For reviews on Ir-catalyzed allylic substitution reaction: (a) R. Takeuchi and S. Kezuka, *Synthesis*, 2006, 3349; (b) G. Helmchen, A. Dahnz, P. Dübon, M. Schelwies and R. Weihofen, *Chem. Commun.*, 2007, 675; (c) G. Helmchen, in *Iridium Complexes in Organic Synthesis*, ed. L. A. Oro and C. Claver, Wiley-VCH, Weinheim, Germany, 2009, p. 211;



(d) J. F. Hartwig and L. M. Stanley, *Acc. Chem. Res.*, 2010, **43**, 1461; (e) J. F. Hartwig and M. J. Pouy, *Top. Organomet. Chem.*, 2011, **34**, 169; (f) W.-B. Liu, J.-B. Xia and S.-L. You, *Top. Organomet. Chem.*, 2012, **38**, 155; (g) P. Tosatti, A. Nelson and S. P. Marsden, *Org. Biomol. Chem.*, 2012, **10**, 3147.

7 For a review on transition-metal catalyzed asymmetric allylic dearomatization reactions: (a) C.-X. Zhuo, C. Zheng and S.-L. You, *Acc. Chem. Res.*, 2014, **47**, 2558. For selected examples from our group: (b) Q.-F. Wu, H. He, W.-B. Liu and S.-L. You, *J. Am. Chem. Soc.*, 2010, **132**, 11418; (c) Q.-L. Xu, L.-X. Dai and S.-L. You, *Chem. Sci.*, 2013, **4**, 97; (d) X. Zhang, L. Han and S.-L. You, *Chem. Sci.*, 2014, **5**, 1059; (e) X. Zhang, W.-B. Liu, H.-F. Tu and S.-L. You, *Chem. Sci.*, 2015, **6**, 4525.

8 For selected examples on transition-metal catalyzed asymmetric allylic dearomatization reactions from other groups: (a) M. Kimura, M. Futamata, R. Mukai and Y. Tamaru, *J. Am. Chem. Soc.*, 2005, **127**, 4592; (b) M. Bandini, A. Melloni, F. Piccinelli, R. Sinisi, S. Tommasi and A. Umani-Ronchi, *J. Am. Chem. Soc.*, 2006, **128**, 1424; (c) B. M. Trost and J. Quancard, *J. Am. Chem. Soc.*, 2006, **128**, 6314; (d) N. Kagawa, J. P. Malerich and V. H. Rawal, *Org. Lett.*, 2008, **10**, 2381; (e) M. Bandini and A. Eichholzer, *Angew. Chem., Int. Ed.*, 2009, **48**, 9533; (f) G. Cera, P. Crispino, M. Monari and M. Bandini, *Chem. Commun.*, 2011, **47**, 7803; (g) Y. Zhu and V. H. Rawal, *J. Am. Chem. Soc.*, 2012, **134**, 111; (h) M. Bandini, A. Bottoni, M. Chiarucci, G. Cera and G. P. Mischione, *J. Am. Chem. Soc.*, 2012, **134**, 20690; (i) G. Cera, M. Chiarucci, A. Mazzanti, M. Mancinelli and M. Bandini, *Org. Lett.*, 2012, **14**, 1350; (j) Y. Liu and H. Du, *Org. Lett.*, 2013, **15**, 740; (k) J. Chen and M. J. Cook, *Org. Lett.*, 2013, **15**, 1088; (l) T. D. Montgomery, Y. Zhu, N. Kagawa and V. H. Rawal, *Org. Lett.*, 2013, **15**, 1140; (m) A. Lin, J. Yang and M. Hashim, *Org. Lett.*, 2013, **15**, 1950; (n) T. M. Kaiser and J. Yang, *Eur. J. Org. Chem.*, 2013, 3983; (o) Z.-S. Liu, W.-K. Li, T.-R. Kang, L. He and Q.-Z. Liu, *Org. Lett.*, 2015, **17**, 150.

9 Q.-F. Wu, C. Zheng and S.-L. You, *Angew. Chem., Int. Ed.*, 2012, **51**, 1680.

10 C. Zheng, Q.-F. Wu and S.-L. You, *J. Org. Chem.*, 2013, **78**, 4357.

11 C.-X. Zhuo, Q.-F. Wu, Q. Zhao, Q.-L. Xu and S.-L. You, *J. Am. Chem. Soc.*, 2013, **135**, 8169.

12 The intermediates **5** can be captured by *in situ* reduction using NaBH₄ or LiBH₄ in some special cases to deliver their corresponding spiroindoline products: (a) Z.-P. Yang, C.-X. Zhuo and S.-L. You, *Adv. Synth. Catal.*, 2014, **356**, 1731; (b) C.-X. Zhuo, Y. Zhou, Q. Cheng, L. Huang and S.-L. You, *Angew. Chem., Int. Ed.*, 2015, **54**, 14146.

13 Enantioselective synthesis of pyrrole-based spirocyclic analogs by Ir-catalyzed asymmetric allylic dearomatization and controllable migration reactions has been recently realized: C.-X. Zhuo, Q. Chen, W.-B. Liu, Q. Zhao and S.-L. You, *Angew. Chem., Int. Ed.*, 2015, **54**, 8475.

14 See ESI† for the details of the evaluation of reaction conditions.

15 (a) C. A. Kiener, C. Shu, C. Incarvito and J. F. Hartwig, *J. Am. Chem. Soc.*, 2003, **125**, 14272; (b) C. Shu, A. Leitner and J. F. Hartwig, *Angew. Chem., Int. Ed.*, 2004, **43**, 4797.

16 CCDC 1018536–1018541 contain the ESI† crystallographic data for **8aA**, **8aB**, **8aC**, **cis-9a**, **trans-9a**, and **cis-9l**.

17 It was also found in a related case of multisubstituted five-membered carbo-spiroindolenines that the activity toward allyl migration is dependent on the relative configuration of diastereomers of carbo-spiroindolenines. See ref. 8(o).

18 See the ESI† for details of the DFT calculations.

19 The nature of the migration process in the Pictet–Spengler-type reactions is highly dependent on the electronic property of the migratory group. For the transformation from **2** to **3**, the allyl migration is concerted, while for the transformations from **5** to **6** and **8** to **9**, only stepwise *N*-aryl iminium migration involving an intermediate like **INT-1** exists. For a detailed study on this issue, see ref. 10.

