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A highly diastereoselective “super silyl” governed aldol reaction: synthesis of α,β -dioxyaldehydes and 1,2,3-triols†

Wafa Gati* and Hisashi Yamamoto*

A highly diastereoselective approach for the synthesis of protected α,β -dioxyaldehydes derived from (Z)-tris(trimethylsilyl)silyl “super silyl” enol ethers is described. A general and highly *syn*-stereoselective aldol reaction directed by the “super silyl” group catalyzed by triflimide (HNTf₂) is developed providing α,β -dioxyaldehydes and 1,2,3-triol fragments which can be a useful platform for the elaboration of natural and unnatural sugar derivatives.

Introduction

Polyols are among the most interesting motifs present in various natural and synthetic products. Over the past several decades, organic chemists have made great efforts to invent simpler and more efficient strategies to access stereodefined polyol motifs toward the synthesis of complex sugar frameworks. Because the majority of molecules containing this motif require a multistep protocol for access, chemists have been engaged with creating multiple stereocenters in a one-pot procedure. Although there are numerous routes to C–C bond formation, the aldol reaction remains the most promising and straightforward method for creating two new adjacent stereogenic centers toward the construction of the required polyol subunits.^{1,2}

Recently our group has actively investigated the Mukaiyama aldol reaction of tris(trimethylsilyl)silyl “super silyl” enol ethers for the highly diastereoselective synthesis of β -super silyloxy aldehydes and α -halo- β -super silyloxy aldehydes employing Lewis acid catalysis.³ This efficient methodology allows for a rapid and stereoselective construction of mono-, bis- and tris-hydroxyaldehydes through mono, double and triple cross aldol processes, respectively, affording polyketide-like scaffolds which are particularly useful for the oriented construction of complex natural polyketides. In our continuous studies on Mukaiyama aldol reactions of super silyl enol ethers, we questioned whether a similar strategy might provide access to α,β -

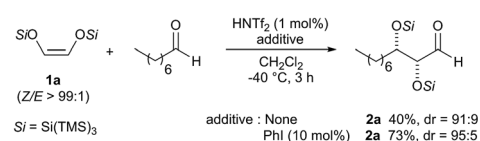
dioxygenated aldehydes which could be a useful building block for construction of complex sugar moieties.⁴

Results and discussion

Herein we describe the first highly diastereoselective aldol reaction with dioxy enol ethers to give protected α,β -dioxygenated aldehydes in moderate to good yields and with exclusively high *syn* selectivities.

The super silyl enol ether derived from silyloxy acetaldehyde⁵ was prepared according to the general procedure recently developed in our laboratory.^{2a} We began our studies by establishing optimal conditions for the Mukaiyama aldol reaction of bisuper silyloxy enol ether **1a** with 1-octanal using 1 mol% of HNTf₂ as catalyst in dichloromethane at –40 °C (Scheme 1).

We were pleased to find that the aldol adduct was obtained in high diastereoselectivity (dr = 91 : 9) but with moderate yield (40%). Thus, in an attempt to optimize the conditions we performed the reaction in the presence of 10 mol% of iodobenzene, which has previously been found to be very useful for increasing the reactivity and the rate of the aldol reaction.^{3d} Gratifyingly, we found that the reaction works more efficiently and the adduct **2a** was obtained with much better yield (73%) and a slightly improved diastereoselectivity (dr = 95 : 5). Although we are not sure about the exact role of iodobenzene, we believe that it acts as a co-catalyst that stabilizes the silylenium cation



Molecular Catalyst Research Center, Chubu University, 1200 Matsumoto-cho, Kasugai, Aichi 487-8501, Japan. E-mail: wafagati@isc.chubu.ac.jp; hyamamoto@isc.chubu.ac.jp

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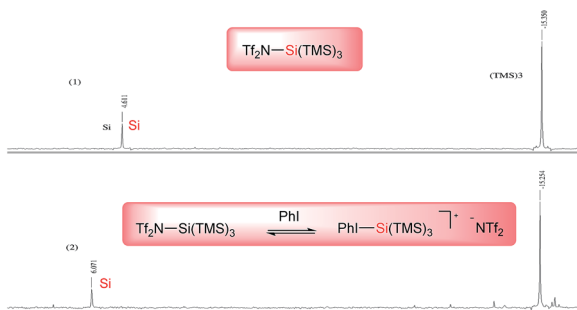
Scheme 1 Influence of the additive on the aldol reaction. Yields of isolated aldehydes are shown. The dr values are determined from crude ¹H NMR.



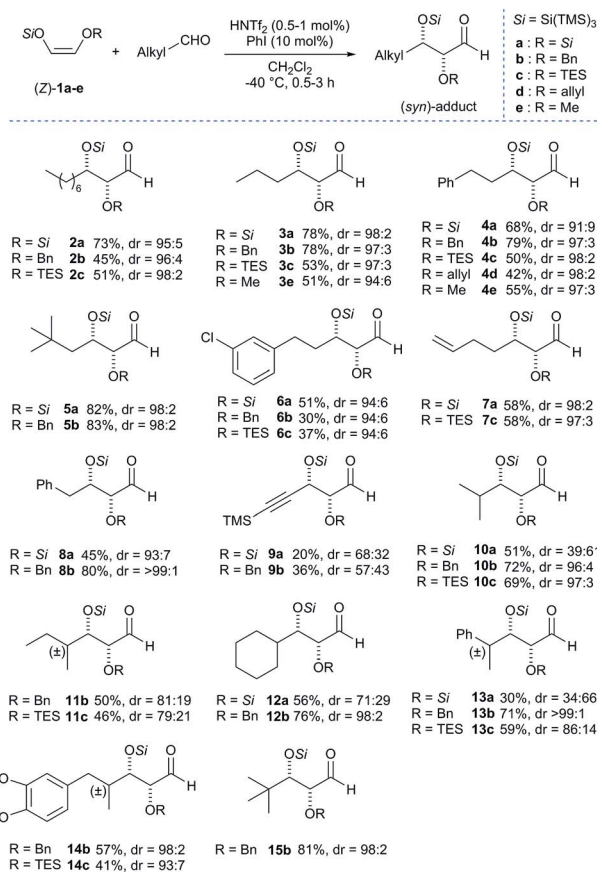
formed *in situ*. Because the additive seemed to be playing a critical role in affecting the rate of the reaction, we conducted a ^{29}Si NMR study with the hypothesis that $[\text{PhI-Si}(\text{TMS})_3]^+$ is the real active catalytic species. We first recorded a reference ^{29}Si NMR spectrum using a simple test substrate (allyltris(trimethylsilyl)silane) in the presence of triflimide (Scheme 2, (1)). We detected a first singlet corresponding to three trimethylsilyl groups that appears at -15.35 ppm and a second singlet corresponding to central silicon that appears at 4.61 ppm. Iodobenzene was then added to the NMR tube and a second ^{29}Si NMR spectrum was recorded after 45 min at room temperature.

Surprisingly, we found that the second singlet was shifted up to 6.07 ppm (Scheme 2, (2)). Surprised by the large effect that iodobenzene had on the outcome of the NMR experiment, we decided to perform other experiments varying the stoichiometry and the reaction time. Interestingly, we found that with higher amounts of iodobenzene, the silicon shift is more pronounced, and the singlets also shift more with longer reaction times (see ESI†). To the best of our knowledge, this is the first NMR proof of the role of organoiodide compounds in the Mukaiyama aldol reaction and the ^{29}Si NMR study was proof of our principle considering $[\text{PhI-Si}(\text{TMS})_3]^+$ as a more active catalytic species than $\text{Ti}_2\text{N-Si}(\text{TMS})_3$.

Satisfied with these results, we applied our general conditions to the reaction of various super silyl enol ethers with a broad array of aldehydes to afford protected α,β -dioxygenated aldehydes (Scheme 3). Most linear aliphatic aldehydes reacted very smoothly and selectively with super silyl enol ether **1a-e** providing the desired α,β -dioxyaldehydes (**2-8**) in moderate to high yields (up to 83% for compound **5b**) and with excellent and exclusive *syn*-selectivities (up to 98 : 2). Fortunately, the major diastereomer of compound **3a** was crystalline, and the *syn* stereochemistry was directly determined from X-ray analysis.⁶ An aldehyde bearing an unsaturation (alkynyl group) in alpha to the carbonyl group was also tested and found to react rather sluggishly with super silyl enol ethers **1a** and **1b** to afford the corresponding adducts **9a** and **9b** respectively with low yields and poor selectivity. We next investigated aliphatic aldehydes bearing an additional substitution in alpha to the carbonyl which were also tolerated but with moderate yields (up to 51%)



Scheme 2 ^{29}Si NMR study on the influence of iodobenzene on the aldol reaction. 1 equiv. of iodobenzene was used. Experiments conducted in NMR tube in CD_2Cl_2 under nitrogen atmosphere and at room temperature.



Scheme 3 Synthesis of protected α,β -dioxyaldehydes: substrate scope of aliphatic aldehydes. Unless otherwise noted, all reactions were carried out on a 0.2 mmol scale. Yields of isolated aldehydes are shown. The dr values are based on the integration of the ^1H NMR signals of crude material. The attribution of *syn* and *anti*-ratios was based on the coupling constants of characteristic protons.

and selectivities (up to 71 : 29 dr) (**10a**, **12-13a**) due to the presence of the extraordinarily bulky silyloxy group. Nevertheless when we tested these branched aldehydes with a less bulky silyl enol ether by substitution of one of the super silyloxy groups with a benzyloxy (**1b**) or a triethylsilyloxy group (**1c**), we found that the previously obtained yields and diastereoselectivities were incredibly improved (**10a** vs. **10b** and **10c**, **12a** vs. **12b**, **13a** vs. **13b** and **13c**). (*Z*)-1-Supersilyloxy-2-benzyloxy enol ether **1b** reacted as expected with remarkably high selectivities (up to >99 : 1) and better yields (up to 72%) obtained in almost all products (**2-15**). Notably, pivalaldehyde, which was unreactive with other super silyl enol ethers, was found to react smoothly with (*Z*)-1-supersilyloxy-2-benzyloxy enol ether **1b** to afford the corresponding aldol adduct **15b** with excellent yield and diastereoselectivity (81%, dr = 98 : 2). On the other hand, (*Z*)-1-supersilyloxy-2-triethylsilyloxy enol ether **1c** was found to react less effectively affording the corresponding aldol adducts with diminished yields, probably due to the competitive reaction of the triethylsilyloxy group with our catalyst, although we did not observe the formation of the corresponding regioisomer, and with no remarkable changes in the



diastereoselectivity ratios obtained with **1a** or **1b**. Subsequently, additional super silyl enol ethers bearing allyloxy (**1d**) or methoxy (**1e**) groups were also briefly investigated. We subjected (*Z*)-1-allyloxy-2-supersilyloxy enol ether **1d** and (*Z*)-1-methoxy-2-supersilyloxy enol ether **1e** to our optimized reaction conditions, affording the corresponding desired aldol adducts with comparable yields and diastereoselectivities.

After the exploration of the scope of aliphatic aldehydes, we next turned our attention to the scope of aromatic aldehydes which were found to be more challenging. When we first investigated the reactivity of benzaldehyde with **1a** using 1 mol% of triflimide catalyst without any additive, we found that the reaction did not proceed and only trace amounts (<5%) of the desired adduct were detected. However, when the reaction was performed with 10 mol% of iodobenzene, the results were remarkably improved and the reaction provided the desired aldol adduct **16a** in high yield (78%) but with moderate diastereoselectivity (Scheme 4). Despite the encouraging results regarding the increased yield using iodobenzene, all other attempts to improve the diastereoselectivity ratios for aromatic aldehydes failed, probably due to the presence of two very bulky super silyl groups. Even so, we were interested in examining the scope of aromatic substrates with our super silyl enol ethers. The reaction with **1a** was found to have poor selectivity (**16a**, **18–23a**) due to steric hindrance with the two silyloxy groups. Nevertheless, we were delighted to find that the diastereoselectivity could be improved up to 98% (for compound **20b**) starting from **1b** and up to 93% (for compound **18d**) starting from **1d**. Then we considered the use of heteroaromatic aldehydes and experiments have shown that an electron-withdrawing group on the heteroaromatic ring is necessary for the

reaction to proceed. Our scope was then extended and compounds **22a–b** and **23a** were obtained in acceptable yields and diastereoselectivities. It is worth noting that all the protected *syn*- α,β -dioxyaldehydes obtained are stable in almost all cases and can be kept for weeks in the freezer, since these compounds are known to be rather sensitive to both elimination and epimerisation.

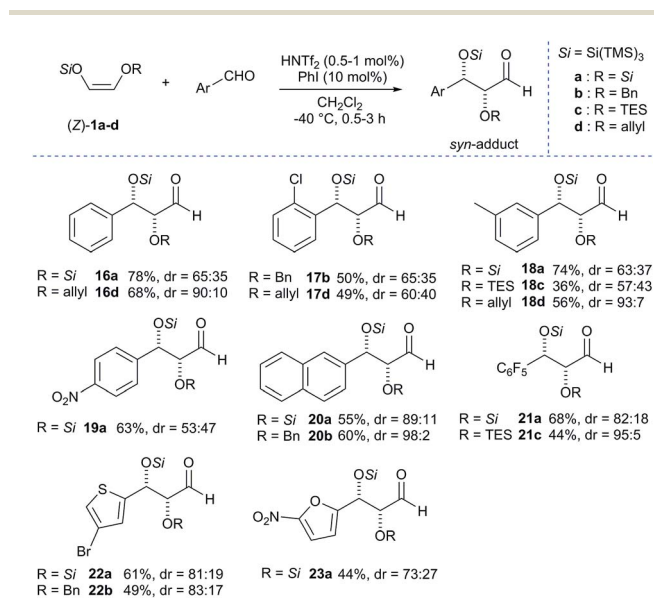
The scope of the reaction was further examined by reacting an optically pure aldehyde with different super silyl enol ethers. In this case, it is known that the stereochemical outcome of the reaction can be controlled by the chirality of the substrate (1,2-asymmetric induction).^{7,8}

Indeed, the use of (*R*)-2-phenylpropanal exhibited, as expected, a high Felkin control in conjunction with *syn* selectivity to afford **24a,b,d** with three adjacent stereocenters in excellent diastereoselectivity ratios (up to >99 : 1 *syn*-*syn* for compound **24b**) (Scheme 5).⁹

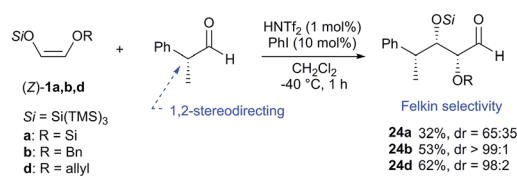
Next, we investigated the possibility of subsequent one-pot sequential transformation of the obtained protected α,β -dioxygenated aldehydes (Table 1). The addition of alkyl, vinyl, alkynyl, thiophen-2-yl or aryl Grignard reagents to the crude material proceeded smoothly to afford trishydroxy products **25–34** with good to excellent yields (50–84%) and exceptionally high *syn*-*syn* diastereoselectivities (>99 : 1 : 0) which was confirmed by single crystal X-ray analysis of triol **27**.¹⁰ In the same fashion as in Scheme 5, we considered the use of an aldehyde with a defined α -stereocenter for a 1,2-asymmetric induction investigation. After reaction of (*R*)-2-phenylpropanal with super silyl enol ether **1b** and addition of phenylmagnesium chloride we obtained the desired triol **33** in moderate yield (50%) and diastereoselectivity (dr = 83 : 17 : 0).

Then we decided to test the more reactive (*Z*)-supersilyloxy-2-allyloxy enol ether **1d** and introduce an additional vinyl group, as it is a rather valuable handle for further transformations. By the addition of a vinyl Grignard reagent we were delighted to easily isolate the synthetically useful vinylic triol **34** generating three new adjacent stereocenters in a one-pot sequential manner in 63% yield and excellent all-*syn* diastereoselectivity (>99 : 1 : 0). Moreover, the reaction was also successful using the lithiated nucleophile EtLi affording the desired alkyl triols **35** and **36** in high diastereoselectivity but with dramatically decreased yields (33% and 36% respectively).

Inspired by the important skeleton of the vinylic all-*syn* triol **34** and in an attempt to further probe the utility of our highly



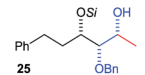
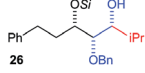
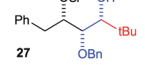
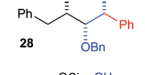
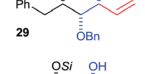
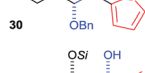
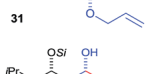
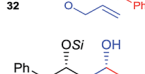

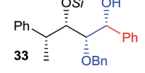

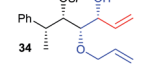
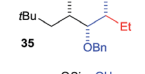
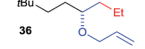
Scheme 4 Synthesis of protected α,β -dioxyaldehydes: substrate scope of aromatic aldehydes. Unless otherwise noted, all reactions were carried out on a 0.2 mmol scale. Yields of isolated aldehydes are shown. The dr values are based on the integration of the ^1H NMR signals of crude material. The attribution of *syn* and *anti*-ratios was based on the coupling constants of characteristic protons.



Scheme 5 1,2-Stereodirected aldol reaction. Unless otherwise noted, all reactions were carried out on a 0.2 mmol scale. Yields of isolated aldehydes are shown. The dr values are based on the integration of the ^1H NMR signals of crude material. The attribution of *syn* and *anti*-ratios was based on the coupling constants of characteristic protons.



Table 1 Diastereoselective one-pot sequential reactions^a

Entry	R	R'	Nucleophile ^b	Major product	%Yield ^c (dr) ^d
1	Bn	CH ₂ CH ₂ Ph	MeMgBr		84% (dr > 99 : 1 : 0)
2	Bn	CH ₂ CH ₂ Ph	<i>i</i> PrMgBr		81% (dr > 99 : 1 : 0)
3	Bn	CH ₂ Ph	<i>t</i> BuMgBr		78% (dr > 99 : 1 : 0)
4	Bn	CH ₂ Ph	PhMgCl		59% (dr > 99 : 1 : 0)
5	Bn	CH ₂ Ph		68% (dr > 99 : 1 : 0)	
6	Bn	CH ₂ CH(CH ₃) ₂		75% (dr > 99 : 1 : 0)	
7	Allyl	CH ₂ CH ₂ Ph		82% (dr > 99 : 1 : 0)	
8	Allyl	CH ₂ CH(CH ₃) ₂	Ph-C≡C-MgBr		52% (dr = 88 : 12 : 0)
9	Bn		PhMgCl		50% (dr = 83 : 17 : 0)
10	Allyl			63% (dr > 99 : 1 : 0)	
11 ^e	Bn	CH ₂ (CH ₃) ₃	EtLi		33% (dr > 99 : 1 : 0)
12 ^e	Allyl	CH ₂ (CH ₃) ₃	EtLi		36% (dr > 99 : 1 : 0)

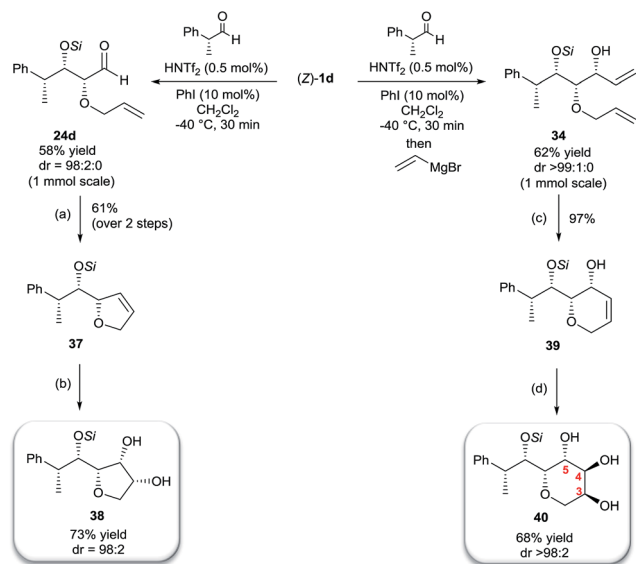
^a Unless otherwise noted, all reactions were carried out on a 0.2 mmol scale. ^b 1.5 equiv. of nucleophile was used. ^c Yields of isolated products are shown. ^d The dr are based on the integration of the ¹H NMR signals of crude material. ^e The reaction was slowly warmed to -20 °C after addition of 2.0 equiv. of nucleophile.

diastereoselective one-pot sequential aldol reaction, we targeted pentose and hexose-like scaffolds which are usually difficult to access without employing natural sugar as starting material.¹¹ We first applied our strategy to establish the desired α -allyloxy- β -supersilyloxyaldehyde **24d** which was obtained at a slightly decreased yield (58%) on a 1 mmol scale but with no loss of selectivity (98 : 2 *syn-syn*). Olefination through Wittig reaction and ring closing metathesis using Grubbs second generation catalyst yielded the five member ring compound **37** in 61% yield (over 2 steps). The last step of the asymmetric dihydroxylation was performed under optimal conditions using catalytic AD-mix- β in biphasic solution at 0 °C for four days¹² which afforded a single diastereomer of **38** in 73% yield containing five adjacent stereocenters in excellent all-*syn* selectivity (Scheme 6). The stereochemistry of compound **38** was determined based on ¹H,

NOE and NOESY experiments (see the ESI†) in comparison with the literature.

Finally, we considered the possibility of hexose-like scaffold construction, which can be a useful building block to access complex natural and unnatural sugar targets. First we employed our highly diastereoselective Lewis acid catalyzed one-pot sequential aldol scaled-up reaction (1 mmol scale) starting from (*R*)-2-phenylpropanal and silyl enol ether **1d** followed by nucleophilic addition of vinyl magnesium bromide to obtain the desired vinylic triol **34** with no loss of reactivity or diastereoselectivity (62%, 99% dr). Next, a very low loading of the Grubbs second generation catalyst (2 mol%) gave access to six membered ring **39** in excellent yield (97%). A quick optimization of the asymmetric dihydroxylation step (see ESI†) showed that *cis*-osmilation using osmium tetroxide in presence of





Scheme 6 Synthesis of pentose and hexose-like scaffolds. (a) $1) \text{CH}_3\text{PPh}_3$, $n\text{-BuLi}$, THF (2) Grubbs 2nd generation (2 mol%), CH_2Cl_2 , 40°C , 2 h (b) AD-mix- β , MeSO_2NH_2 , $t\text{-BuOH}/\text{H}_2\text{O}$ (1 : 1), 0°C , 4 d (c). Grubbs 2nd generation (2 mol%), CH_2Cl_2 , 40°C , 2 h (d). OsO_4 , NMO, $t\text{-BuOH}$, acetone/ H_2O (1.7 : 1), r.t., 12 h.

excess of *N*-methylmorpholine *N*-oxide provided the desired hexose-like structure.¹³ Thus, compound **40**, containing six adjacent stereocenters, was obtained in 68% yield and with an exclusive 4,5-*anti* stereochemistry. The determination of the stereochemistry of the latter compound was based on the optimization reactions where we obtained the same single isomer using both chiral AD-mix- α or β with comparable selectivity but with a slower reaction rate (50% conversion after 4 days), which can be explained by the preferred attack of osmium from the opposite side of the free hydroxy group present in **39**. In addition, a very high coupling constant value ($J_{4-5} > 10.6$ Hz) was detected which emphasizes an *anti*-like relationship between C4-H and C5-H.¹⁴

Conclusions

In summary, a very useful strategy to generate synthetically important protected *syn*- α,β -dioxaldehydes using Lewis acid catalysis has been described. To the best of our knowledge, this is the first synthesis of α -hydroxyaldehydes using Mukaiyama aldol reaction. Furthermore, a ²⁹Si NMR study was performed providing the first proof of the role of iodobenzene as additive in increasing the reactivity of the active silylenium cation formed *in situ*. Since the ability of using different protecting groups in the same molecule is an attractive tool to discriminate among chemically similar hydroxyl groups, super silyloxy, benzyloxy, triethylsilyloxy, allyloxy and methoxy have proved to be suitable for the construction of α,β -dioxaldehydes and 1,2,3-triols. Various nucleophiles were found to react smoothly in a sequential manner allowing for the highly stereoselective construction of all-*syn* 1,2,3-triols. We have finally demonstrated the utility of our methodology as a key step for the

elegant construction of pentose and hexose-like scaffolds. Further applications using super silyl governed aldol reactions targeting complex sugar construction are currently underway in our laboratory and will be reported in due course.

Acknowledgements

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Notes and references

- For reviews on the aldol reaction in the synthesis of carbohydrates and polyketides, see: (a) D. A. Evans, J. V. Nelson and T. Taber, in *Topics in Stereochemistry*, Wiley, New York, 1982, vol. 13, p. 1; (b) T. D. Machajewski and C.-H. Wong, *Angew. Chem., Int. Ed.*, 2000, **39**, 1352; (c) C. Palomo, M. Oiarbide and J. M. García, *Chem.-Eur. J.*, 2002, **8**, 36.
- For selected references on the Mukaiyama aldol reaction, see: (a) T. Mukaiyama, K. Narasaka and K. Banno, *Chem. Lett.*, 1973, 1011; (b) H. Gröger, E. M. Vogl and M. Shibasaki, *Chem.-Eur. J.*, 1998, **47**, 1137; (c) R. Mahrwald, *Chem. Rev.*, 1999, **99**, 1095; (d) K. Ishihara and H. Yamamoto, in *Modern Aldol Reactions*, ed. R. Mahrwald, Wiley-VCH, Weinheim, 2004, vol. 2, p. 25; (e) B. M. Trost and C. S. Brindle, *Chem. Soc. Rev.*, 2010, **39**, 1600; (f) J.-I. Matsuo and M. Murukami, *Angew. Chem., Int. Ed.*, 2013, **52**, 9109.
- (a) M. Boxer and H. Yamamoto, *J. Am. Chem. Soc.*, 2006, **128**, 48; (b) M. Boxer and H. Yamamoto, *J. Am. Chem. Soc.*, 2007, **129**, 2762; (c) M. Boxer and H. Yamamoto, *J. Am. Chem. Soc.*, 2008, **130**, 1580; (d) B. J. Albert and H. Yamamoto, *Angew. Chem., Int. Ed.*, 2010, **49**, 2747; (e) B. J. Albert, Y. Yamaoka and H. Yamamoto, *Angew. Chem., Int. Ed.*, 2011, **50**, 2610; (f) J. Saadi, M. Akakura and H. Yamamoto, *J. Am. Chem. Soc.*, 2011, **133**, 14248; (g) P. B. Brady and H. Yamamoto, *Angew. Chem., Int. Ed.*, 2012, **51**, 1942; (h) P. B. Brady, B. J. Albert, M. Akakura and H. Yamamoto, *Chem. Sci.*, 2013, **4**, 3223; (i) A. Izumiseki and H. Yamamoto, *J. Am. Chem. Soc.*, 2014, **136**, 1308; (j) P. B. Brady, S. Oda and H. Yamamoto, *Org. Lett.*, 2014, **16**, 3864.
- For selected examples on the synthesis of sugar derivatives using the aldol reaction, see: (a) N. S. Chowdari, D. B. Ramachary, A. Córdoba and F. Barbas, *Tetrahedron Lett.*, 2002, **43**, 9591; (b) A. B. Northrup and D. W. C. MacMillan, *Science*, 2004, **305**, 1752; (c) J. Casas, M. Engqvist, I. Ibrahim, B. Kaynak and A. Córdoba, *Angew. Chem., Int. Ed.*, 2005, **44**, 1343; (d) J. Carpenter, A. B. Northrup, D. Chung, J. J. M. Wiener, S. Kim and D. W. C. MacMillan, *Angew. Chem., Int. Ed.*, 2008, **47**, 3568; (e) M. Peifer, R. Berger, V. W. Shurtleff, J. C. Conrad and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2014, **136**, 5900.



- 5 Supersilyloxy- and allyloxyacetaldehydes were prepared from ethyl glycolate following a described procedure: S. R. Angle, I. Choi and F. S. Tham, *J. Org. Chem.*, 2008, **73**, 6268.
- 6 The *syn* selectivity was confirmed by single crystal X-ray analysis of compound **3a**.†
- 7 For references on stereodirected aldol reactions see: (a) R. Mahrwald, *Chem. Rev.*, 1999, **99**, 1095; (b) L. C. Dias, E. C. Polo, E. C. de Lucca and M. A. B. Ferreira, in *Modern Methods in Stereoselective Aldol Reactions*, ed. R. Mahrwald, Wiley-VCH Verlag GmbH & Co. KGaA, 2013, pp. 293–375.
- 8 IUPAC definition of asymmetric induction: IUPAC, *Compendium of Chemical Terminology*, (the “Gold Book”), ed. A. D. McNaught and A. Wilkinson, Blackwell Scientific Publications, Oxford, 2nd edn, 1997, <http://goldbook.iupac.org>, 2006, created by M. Nic, J. Jirat and B. Kosata, updates compiled by A. Jenkins, ISBN 0-9678550-9-8, DOI: 10.1351/goldbook.
- 9 The determination of the Felkin selectivity was achieved by comparison of coupling constant values with previous reports of similar work in our laboratory, see ref. **3a** and **3h** and references therein.
- 10 The *syn-syn* selectivity was confirmed by single crystal X-ray analysis of compound **27**.†
- 11 J. L. Clark, L. Hollecker, J. C. Mason, L. J. Stuyver, P. M. Tharnish, S. Lostia, T. R. McBrayer, R. F. Schinazi, K. A. Watanabe, M. J. Otto, P. A. Furman, W. J. Stec, S. E. Patterson and K. W. Pankiewicz, *J. Med. Chem.*, 2005, **48**, 5504.
- 12 For recent asymmetric dihydroxylation procedures using AD-mix- β in a five member ring see: (a) K. B. Lindsay and S. G. Pyne, *J. Org. Chem.*, 2002, **67**, 7774; (b) N. S. Karanjule, S. D. Markad and D. D. Dhavale, *J. Org. Chem.*, 2006, **71**, 6273.
- 13 For recent asymmetric dihydroxylation procedures in six member rings see: (a) K. Matsumoto, T. Ebata, K. Koseki, H. Kawakami and H. Matsushita, *Bull. Chem. Soc. Jpn.*, 1991, **64**, 2309; (b) F. Konno, T. Ishikawa, M. Kawahata and K. Yamaguchi, *J. Org. Chem.*, 2006, **71**, 9818; (c) G. Errasti, C. Koundé, O. Mirguet, T. Lecourt and L. Micouin, *Org. Lett.*, 2009, **11**, 2912; (d) M. T. Mwangi, M. D. Schultz and N. B. Bowden, *Org. Lett.*, 2009, **11**, 33; (e) T. V. Robinson, D. S. Pederson, D. K. Taylor and E. R. T. Tiekink, *J. Org. Chem.*, 2009, **74**, 5093; (f) C. Taffin, G. Kreutler, D. Bourgeois, E. Clot and C. Périgaud, *New J. Chem.*, 2010, **34**, 517; (g) W. Lim, J. Kim and Y. H. Rhee, *J. Am. Chem. Soc.*, 2014, **136**, 13618.
- 14 Please see the ESI† for more details.

