

Cite this: *Chem. Sci.*, 2016, 7, 781

# Transition voltages respond to synthetic reorientation of embedded dipoles in self-assembled monolayers†

Andrii Kovalchuk,<sup>a</sup> Tarek Abu-Husein,<sup>b</sup> Davide Fracasso,<sup>a</sup> David A. Egger,<sup>cd</sup> Egbert Zojer,<sup>c</sup> Michael Zharnikov,<sup>e</sup> Andreas Terfort<sup>b</sup> and Ryan C. Chiechi<sup>\*a</sup>

We studied the influence of embedded dipole moments in self-assembled monolayers (SAMs) formed on template stripped Au surfaces with liquid eutectic Ga–In alloy as a top electrode. We designed three molecules based on a *p*-terphenyl structure in which the central aromatic ring is either phenyl or a dipole-inducing pyrimidyl in one of two different orientations. All three form well defined SAMs with similar thickness, packing density and tilt angle, with dipole moments embedded in the SAM, isolated from either interface. The magnitude of the current density is dominated by the tunneling distance and is not affected by the presence of dipole moments; however, transition voltages ( $V_T$ ) show a clear linear correlation with the shift in the work function of Au induced by the collective action of the embedded dipoles. This observation demonstrates that  $V_T$  can be manipulated synthetically, without altering either the interfaces or electrodes and that trends in  $V_T$  can be related to experimental observables on the SAMs before installing the top contact. Calculated projected density of states of the SAMs on Au surfaces that relate HOMO-derived states to  $V_T$  further show that energy level alignment within an assembled junction can be predicted and adjusted by embedding dipoles in a SAM without altering any other properties of the junction. We therefore suggest that trends in  $V_T$  can be used analogously to  $\beta$  in systems for which length-dependence is physically or experimentally inaccessible.

Received 20th August 2015

Accepted 17th October 2015

DOI: 10.1039/c5sc03097h

www.rsc.org/chemicalscience

## 1 Introduction

The field of molecular electronics aims to investigate and realize electronic devices with functionality defined by molecular properties. Two main approaches are currently used to contact molecules, which is a key step in the examination of charge transport: single-molecule and large-area (*i.e.*, ensembles) measurements. In both cases the molecules under investigation are placed in between two metal electrodes that are on the order of 2 nm apart (the exact distance is defined by the dimensions of the molecules under investigation). In these systems interfaces play an important role in defining the characteristics of

a junction and both approaches suffer from an uncertainty—is transport dominated by molecules or by interfaces?<sup>1,2</sup> Electron transport in large-area junctions is affected by defects in self-assembled monolayers (SAMs) that can dominate transport in certain cases,<sup>3</sup> while single-molecule junctions exhibit background currents in which tunneling charges flow directly from one electrode to the other, by-passing the molecule in between.<sup>4</sup> Thus, the magnitude of  $J$  or  $I$  (current-density or current) by itself varies considerably and therefore carries little useful information on the intrinsic electronic properties of the molecules in the junction.

One of the most reliable metrics that seeks to resolve these issues is  $\beta$ , which is an empirical parameter derived from a form of the Simmons equation  $J = J_0 e^{-\beta d}$ , where  $J$  is the current density,  $d$  is the tunneling distance defined by the length of the molecular backbone and  $J_0$  is the theoretical value of  $J$  at  $d = 0$ . Values of  $\beta$  are derived from measurements of series of molecules that differ only by length, while both top and bottom interfaces are kept constant, thus isolating the molecular component in charge transport.<sup>5,6</sup> This approach to data analysis is particularly robust when comparing saturated molecules (*i.e.*, where the backbone comprises mostly  $sp^3$ -hybridized C atoms), for which the consensus value of  $\beta$  is  $\sim 0.75 \text{ \AA}^{-1}$ .<sup>6</sup> Saturated molecules have frontier orbitals that are typically not accessible in the typical bias windows used in molecular electronics and they are not very

<sup>a</sup>Stratingh Institute for Chemistry & Zernike Institute for Advanced Materials, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands. E-mail: r.c.chiechi@rug.nl

<sup>b</sup>Institut für Anorganische und Analytische Chemie, Universität Frankfurt, Max-von-Laue-Straße 7, 60438 Frankfurt, Germany

<sup>c</sup>Institute of Solid State Physics, NAWI Graz, Graz University of Technology, Graz, Austria

<sup>d</sup>Department of Materials and Interfaces, Weizmann Institute of Science, Rehovot 76100, Israel

<sup>e</sup>Angewandte Physikalische Chemie, Universität Heidelberg, Im Neuenheimer Feld 253, 69120 Heidelberg, Germany

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c5sc03097h

polarizable. With the exception of end groups that introduce accessible gap states<sup>7</sup> these properties tend to make saturated molecules less sensitive to the details of the contacts, in general; *e.g.*, tail-groups,<sup>8–10</sup> anchoring groups,<sup>11,12</sup> and minor alterations to the backbone<sup>13</sup> have little impact on the tunneling transport in terms of the magnitudes of  $I$  or  $J$ . Unsaturation, by contrast, adds significant complexity and even subtle changes in conjugation patterns can have pronounced and non-distance dependent effects on transport.<sup>14–18</sup> Tuning the length of fully conjugated molecules is also synthetically challenging and not always possible, since a minimal step size is a  $\pi$  bond (*i.e.*, two carbons) or aromatic ring (usually phenylene) and, unlike alkanes, conjugated molecules become markedly less soluble with increasing length.<sup>19</sup> Thus, a parameter other than  $\beta$ , but that is comparably independent from non-molecular variables (*e.g.*, interfaces), could greatly assist in the description of tunneling transport phenomena in conjugated molecules and, importantly, in the deconvolution of molecular properties from those of the experimental platform.

Beebe *et al.*<sup>20</sup> introduced the transition voltage ( $V_T$ ) as an approximate measure of the tunneling barrier height, which was later related to level alignment—*i.e.*, the difference between the energy of the accessible frontier orbital of a molecule and the Fermi level of the electrode (*e.g.*,  $E_{\text{LUMO}} - E_F$  or  $E_F - E_{\text{HOMO}}$ ) in an assembled junction. The parameter  $V_T$  can be extracted from the minimum of a Fowler–Nordheim plot,  $\ln(I/V^2)$  versus  $1/V$ . The possibility of determining the level alignment of a junction by simply re-plotting conductance data has led to a number of experimental<sup>21–28</sup> and theoretical studies.<sup>29–33</sup>

While  $\beta$  provides information about the effective tunneling distance (and barrier height),  $V_T$  provides information about energy level alignment. Multiple experiments showed a correlation between  $V_T$  and apparent energetic separation between the Fermi energy level ( $E_F$ ) and the dominant frontier molecular orbital.<sup>34,35</sup> However, the precise physical meaning of  $V_T$  is still under debate; *e.g.*, current becomes “superquadratic” with bias and might not always correlate to energy spectral transition.<sup>30,36</sup> Soththwes *et al.*,<sup>37</sup> studied vacuum gaps in ultra-high vacuum STM junctions and found that transition voltage is inversely proportional to  $1/d$ ; *i.e.*, that work showed that  $V_T$  can even be measured in the absence of molecules.

Summarizing the above considerations, we assert that the interpretation of  $V_T$  is not straightforward and that  $V_T$  is highly dependent on interfaces and is a conflation of two effects—interfacial and molecular—underscoring the importance of separating one from the other. This paper describes the control over  $V_T$  by manipulating a single parameter—embedded dipoles—while keeping the interfaces and electrodes constant, allowing the unambiguous assignment of trends in  $V_T$  and energy level alignment to an intrinsic molecular property.

## 2 Results and discussion

### $J/V$ measurements

We investigated the influence of embedded dipoles on electron transport of SAMs placing them in EGaIn junctions of the form Au<sup>TS</sup>/SAM//Ga<sub>n</sub>O<sub>m</sub>/EGaIn (where “/” denotes an interface

defined by chemisorption and “//” by physisorption).<sup>38</sup> Here EGaIn stands for eutectic alloy of Ga and In (75.5% Ga and 24.5% In by weight, mp = 15.7 °C) which is covered by a superficial layer of ~0.7 nm of conductive Ga<sub>n</sub>O<sub>m</sub>. Multiple studies have shown that the oxide layer has a negligible effect on transport properties in EGaIn junctions and is orders of magnitude more conductive than the contacts.<sup>6,38–40</sup> We designed three structures (depicted and assigned in Fig. 1) for this study that possess identical length, surface chemistry, and nearly identical gas-phase frontier orbital energies; for TP1-down and -up they are identical (as is their empirical formula). All three compounds form well-defined SAMs on template-stripped Au (Au<sup>TS</sup>)<sup>41</sup> and were extensively characterized by a number of complementary surface-analytical techniques,<sup>42</sup> exhibiting comparable film thickness and packing densities (see Table 1). The discernible difference is the dipole moment associated with the central aromatic ring (either a pyrimidine or benzene).

An immediate consequence of the collective effect of SAMs of polar pyrimidyl groups is the modification of the electrostatic potential profile, which shifts the vacuum level and the energy separation between  $E_F$  and frontier molecular orbitals. Transition voltages offer insight into the effects of electrostatic fields induced by SAMs because they carry information about the level alignment between the frontier molecular orbitals and the Fermi energies of the electrodes. This information is inaccessible experimentally and is challenging to model theoretically, as the details of alignment between molecular and electrode levels are difficult to predict.<sup>43,44</sup> Our experimental approach is to vary an internal, molecular property—in this case dipole

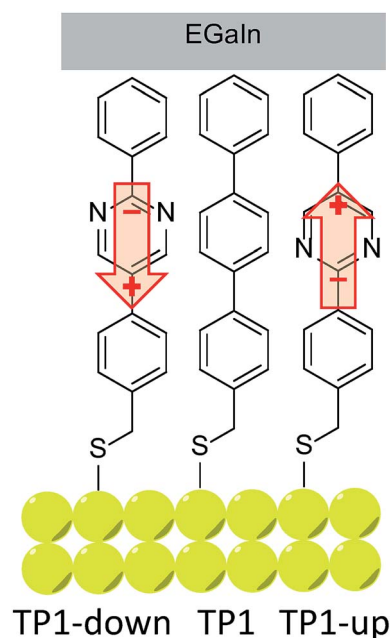


Fig. 1 Schematic of a junction with two pyrimidyl-containing compounds (TP1-down and TP1-up) and the reference compound (TP1). Arrows indicate directions of dipole moments associated with the embedded pyrimidine rings (from negative to positive charge).



**Table 1** X-ray photoemission spectroscopy derived effective thickness and packing density of TP1, TP1-up and TP1-down SAMs; X-ray absorption spectroscopy derived tilt angles; WF shifts with respect to pristine gold<sup>c</sup>

SAM	Effective thickness (nm)	Packing density (molecules per cm <sup>2</sup> )	Tilt angle	WF shift <sup>a</sup> (eV)
TP1	1.78 ± 0.04	4.6 × 10 <sup>14</sup>	18 ± 3°	0.98
TP1-up	1.74 ± 0.05	4.2 × 10 <sup>14</sup>	18 ± 3°	1.41(+0.43 <sup>b</sup> )
TP1-down	1.75 ± 0.05	4.3 × 10 <sup>14</sup>	17 ± 3°	0.43(−0.55 <sup>b</sup> )

<sup>a</sup> Measured with a Kelvin probe; we use opposite sign conventions for  $\Phi$ . <sup>b</sup> Difference from TP1. <sup>c</sup> Experimental values are from ref. 42.

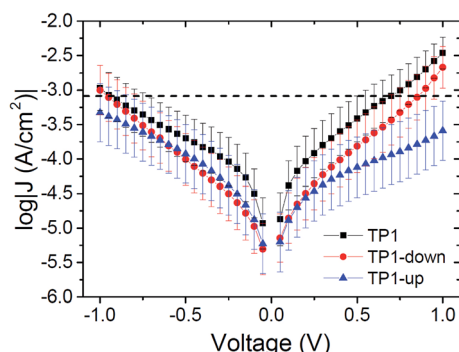
moment—and measure the effect in a SAM supported by a bottom electrode (*i.e.*, *ex situ*) before the top contact is installed. We chose shifts in the work function of the bottom electrode ( $\Phi$ ) because work function shift ( $\Delta\Phi$ ) is defined by the collective effect of embedded dipoles in the SAM.<sup>45</sup> This collective effect is preserved when the top contact is installed (*i.e.*, *in situ*), because the dipoles are embedded in the SAM and are isolated from both interfaces. After assembling the junction and performing electrical measurements, we extracted  $V_T$  and plotted it against  $\Delta\Phi$  to give us two experimental parameters, one intrinsic to the SAM/bottom-contact ( $\Delta\Phi$ ) and one to the bottom-contact/SAM/top-contact ( $V_T$ ). This approach is similar to that of the analysis of  $\beta$ , where tunneling distance  $d$  (which is an *ex situ* parameter and can be calculated and measured in multiple ways) is correlated to current density  $J$  (an *in situ* characteristic of an assembled junction). It is important to compare trends because the absolute magnitude of  $V_T$  is still affected by the details of the contacts.<sup>26,36</sup>

Fig. 2 summarizes measurements of tunneling current through SAMs of TP1, TP1-down, and TP1-up. These data were acquired by sweeping the potential in EGaIn junctions through a range of  $\pm 1$  V (see ESI† for a detailed description of data acquisition and analysis). As expected, the conductances of all SAMs are nearly identical. The magnitude of current is dominated by the tunneling distance, which is identical along the series, and is influenced only slightly, if at all, by the embedded

dipoles ( $\beta \approx 0.4 \text{ \AA}^{-1}$  for these backbones<sup>46</sup>). All of the curves are slightly asymmetric,<sup>47</sup> with TP1-up showing opposite asymmetry—it conducts slightly more at negative bias as opposed to TP1 and TP1-down, which are slightly less conductive at negative bias (but values of  $J(+)$  and  $J(-)$  are within error for most values of  $V$  for all three SAMs, see the ESI†). Though there is evidence that terminal pyrimidine rings can induce asymmetry in  $J/V$  traces<sup>48–50</sup> (which can theoretically be caused by internal dipole moments as well,<sup>51</sup>) we are hesitant to ascribe the observed asymmetry solely to the presence of molecular dipoles, since TP1 (which does not possess an embedded dipole) and TP1-down exhibit comparable degrees of asymmetry. However we can eliminate packing, tilt, and the molecule–electrode interfaces, as these parameters are effectively identical for the three SAMs. The difference in the symmetry of the  $J/V$  curve of TP1-up may be related to the effect of the direction of the dipole moments on the hybridization of the HOMO with states in the gold electrode (see below).

### Transition voltage measurements

We calculated  $V_T$  by re-plotting raw  $I/V$  data as  $\ln(I/V^2)$  versus  $1/V$  for both positive and negative biases for each  $J/V$  curve and extrapolating the minimum (see the ESI† for details on  $V_T$  acquisition). The peak values of Gaussian fits ( $\mu$ ) to the resulting distributions are taken as  $V_T$  and the error is derived from the widths ( $\sigma$ ). These data are summarized in Table 2 along with gas-phase HOMO energies and dipole moments calculated using structural information from the characterization of the SAMs (as described in ref. 46). The HOMO energies serve only to highlight the electronic similarities between the three compounds, not the SAMs. The values of  $V_T$  at negative bias (denoted  $V_T^-$ ) are systematically higher than the corresponding values of  $V_T^+$ , which is common for EGaIn junctions,<sup>26,46</sup> but they follow the same trend; increasing from TP1-down to TP1 to TP1-up. The value of  $V_T^+$  for TP1 is in good agreement with the



**Fig. 2** Plots of log current-density versus applied potential for SAMs of TP1 (black squares), TP1-down (red circles) and TP1-up (blue triangles). Values of  $\log|J|$  at  $V = 0$  are omitted for clarity. Error bars represent 95% confidence intervals. The three traces are hardly distinguishable at negative bias, while, at positive bias, TP1-up deviates from the rest showing opposite asymmetry ( $J(+1 \text{ V})$  is slightly higher than  $J(-1 \text{ V})$  for TP1 and TP1-down and opposite for TP1-up).

**Table 2** Values of  $V_T$  for all SAMs for positive ( $V_T^+$ ) and negative bias ( $V_T^-$ ) as well as gas-phase calculated HOMO energies. Errors are 95% CI

SAM	$V_T^+$ (V)	$V_T^-$ (V)	HOMO <sup>a</sup> (eV)	$\mu_{\text{net}}^a$ (D)
TP1	0.52 ± 0.05	−0.65 ± 0.05	−5.65	+0.01
TP1-up	0.80 ± 0.06	−0.85 ± 0.03	−6.08	−2.75
TP1-down	0.40 ± 0.02	−0.43 ± 0.04	−6.08	+2.34

<sup>a</sup> Gas-phase HSE06/6-311+g(2d,2p) DFT calculations.



previously reported value of  $0.55 \pm 0.10$  V.<sup>46</sup> The general trend is also in agreement; “down” dipole moments lower both  $V_T^+$  and  $V_T^-$  with respect to “up” dipole moments.

### Level alignment

As a result of collective effect of individual dipoles, SAMs of TP1-down and TP1-up shift the electrostatic energy within the junction, which alters the relative positions of the frontier orbitals and the Fermi levels of the electrodes leading to a change in  $V_T$ . The magnitude of the shift can be approximated by measuring the work function of the bare Au<sup>TS</sup> substrate and the substrate supporting a SAM using Kelvin probe technique or UPS.<sup>46</sup> Kim *et al.*<sup>24</sup> demonstrated correlation of  $V_T$  versus  $\Delta\Phi$  using conducting AFM tips to contact SAMs; however, they adjusted  $\Phi$  by varying materials of either bottom or top electrodes, not the characteristics of the molecules. Another study found a correlation between  $V_T$  and interfacial dipoles, but could not unambiguously assign it to a molecular property.<sup>46</sup> The effects of embedded dipolar groups have also been investigated in aliphatic SAMs (*i.e.*, comprising CH<sub>2</sub> backbones), including a study of the physical and electronic structure effects of embedded esters<sup>52</sup> as well as a study of the  $J/V$  properties of embedded amides,<sup>9</sup> however, no correlation to  $V_T$  has been established. Taking TP1 as a reference point, the shifts in TP1-down and up are  $\Delta\Phi = -0.55$  and  $+0.43$  eV (see Table 1), respectively; they are shifted by approximately the same amount, but opposite in sign, from TP1. Fig. 3 shows plots of  $V_T^+$  and  $V_T^-$  versus  $\Delta\Phi$ . The plots are approximately linear, fitting with  $R^2 = 0.77$  and  $0.99$  respectively, demonstrating that  $V_T$  correlates to the shift in vacuum level of Au<sup>TS</sup> induced by the embedded dipoles of the SAMs. A symmetric offset is apparent for  $V_T^-$ , which differs from TP1 by  $\sim \pm 0.2$  V, but less so for  $V_T^+$ ; however, the correlation of the latter to  $\Delta\Phi$  is also less robust. Thus, it appears that the simple picture in Fig. 1 is a reasonable, qualitative description of the synthetic manipulation of  $V_T$ .

### DFT calculations

Valuable insight can be gained from the level alignment of the molecular states relative to the Fermi energy of the Au substrate

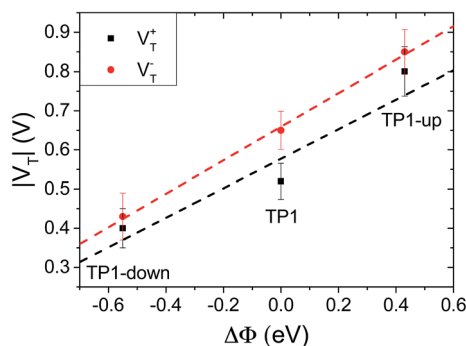


Fig. 3 Plot of  $V_T^+$  (black squares, fitting with the slope of 0.38 and  $R^2 = 0.77$ ) and  $V_T^-$  (red circles, slope of 0.43 and  $R^2 = 0.99$ ) versus the SAM-induced work-function shift. Values of  $\Delta\Phi$  are taken from Table 1.

in the absence of the EGaIn (top) electrode. Thus, we plot the DFT calculated projected densities of states (PDOS) associated with the three studied monolayers in Fig. 4. We used the hybrid functional HSE<sup>53,54</sup> for the periodic band-structure calculations (performed with the VASP code,<sup>55</sup> see the ESI† for details) for the metal-SAM systems as, due to the mixing of short-range Fock and semi-local exchange, orbital self-interactions errors that would distort the electronic structure of pyrimidyl-containing systems can be reduced.<sup>56,57</sup> However, the absolute values of the calculated level alignment, especially for upright-standing molecules,<sup>58</sup> cannot quantitatively reproduce the experiment even with the hybrid-functionals used here.<sup>43,59,60</sup> Nevertheless, for chemically similar systems such as the ones studied here, advanced hybrid DFT-calculations allow for predicting trends in the level alignment.

In Fig. 4, one clearly sees that in TP1-down the highest occupied states are shifted towards  $E_F$  compared to the reference TP1 system, while they are shifted away in the TP1-up case. These shifts can be understood from the peculiar distribution of the electrostatic energy within the SAM where, due to collective electrostatic effects<sup>61,62</sup> (*i.e.*, the superposition of the fields of the pyrimidyl dipoles arranged in a 2D pattern), the electrostatic energy in the topmost rings is shifted relative to  $E_F$  (as schematically shown in Fig. 5, a plot of the calculated plane-averaged potentials can be found in ref. 42). This shift has been confirmed by high-resolution XPS experiments.<sup>42</sup> As the occupied frontier states are largely delocalized over the SAM, a shift in the electrostatic energy induces a shift in the SAM eigenstates relative to  $E_F$  (see Fig. 5).

The frontier orbitals are largely delocalized over the molecular backbone, associated with highly transmissive channels in the transport experiments. Nevertheless, one can see that in the TP1-down (TP1-up) case the HOMO-derived PDOS has a larger weight on the ring far from (close to) the Au substrate, which is the behavior expected for such a situation,<sup>63</sup> as can be understood, for example, from the analogy of SAM-states and

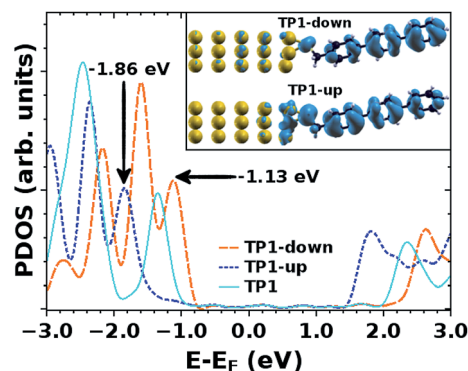


Fig. 4 Density of states of TP1, TP1-up and TP1-down projected (PDOS) onto the molecular region as calculated with the HSE functional. The energy scale is given relative to the Fermi-energy,  $E_F$ ; the inset depicts the charge density associated with the highest occupied peaks in the PDOS (derived from the molecular HOMO) of TP1-down (top) and TP1-up (bottom). These are calculated per system in a  $\pm 0.1$  eV interval centered at the energy indicated by an arrow (isodensity value:  $0.01 \text{ e}^- \text{ \AA}^{-3}$ ).





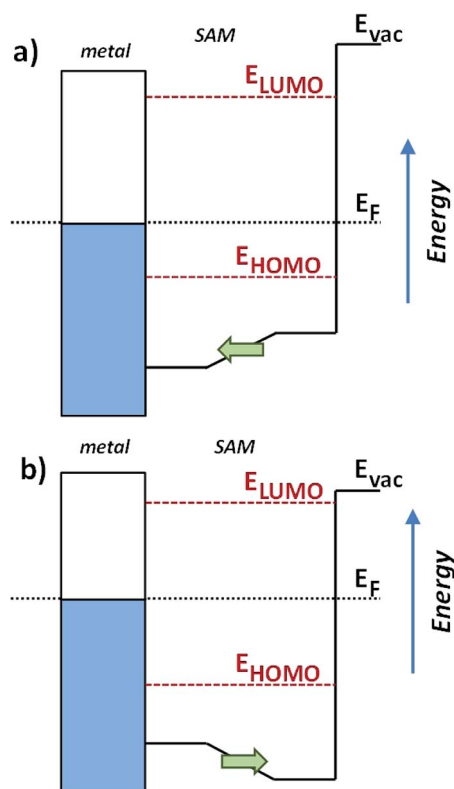


Fig. 5 Schematics of the electrostatic energy distribution and the resulting energy-level alignment in TP1-down (a) and TP1-up (b) SAMs on a Au electrode. The right (upper) parts of the potential well are shifted up, respectively down in energy as a consequence of the pyrimidyl dipoles arranged in a 2D plane. The SAM eigenstates (partially) follow that shift.

electron- and hole-states in quantum-well states in the presence of a potential gradient.<sup>56</sup> This difference in the spatial distribution of PDOS densities might also be responsible for the qualitative differences in the shapes of the  $J/V$  curves for SAMs of TP1-up and TP1-down (Fig. 2).

The question remains as to what exactly happens at  $V_T$ , *e.g.*, if the tail of density of states comes into resonance with  $E_F$ . A calculation of the PDOS for SAMs bound to a metal surface made by plotting the peaks in Fig. 4 produces good correlation of  $V_T$  versus peak values of HOMO levels (Fig. 6). The slopes of linear fits for both  $V_T^+$  and  $V_T^-$  are almost equal (0.56 and 0.55 respectively) and in good agreement with the experimentally determined slope of 0.55 reported by Beebe *et al.*<sup>21</sup> Regardless of the exact physical meaning of the magnitude of  $V_T$ , from the trend it is clearly possible to “feel” energy level alignment in these SAMs. Moreover, the agreement in the slopes suggests that shifting the vacuum level by embedding dipoles in a SAM is physically similar to changing the identity of the electrodes, while the effects of dipoles placed at the physisorbed interface are more convoluted.<sup>46</sup>

### Trends in transition voltages

Just as the absolute value of  $J$  for an isolated member of a series of molecules (from which one cannot make a  $J$  vs.  $d$  plot to

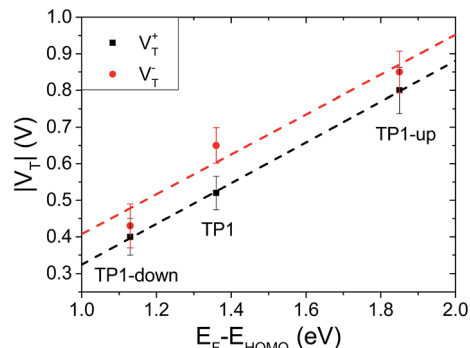


Fig. 6 Plot of  $V_T^+$  (black squares, fitting with the slope of 0.56 and  $R^2 = 1$ ) and  $V_T^-$  (red circles, slope of 0.55 and  $R^2 = 0.87$ ) vs.  $E_F - E_{HOMO}$  from the calculated density of states. Error bars are 95% CI.

extract  $\beta$ ) is significantly less useful than  $\beta$ , the absolute value of  $V_T$  carries complex, inseparable information and is less useful than a trend that relates a shift in  $V_T$  to a controllable variable. The trend presented in Fig. 3 shows that a shift in  $V_T$  is correlated to a change in  $\Phi$  (hence dipole moment) revealing a molecular fingerprint in the transport properties. For any series of molecules of equal length  $\beta$  is obviously not applicable, thus trends in  $V_T$  might serve as empirical evidence that transport is dominated by tunneling through molecules (Fig. S3†). The ability to make this distinction is both important and non-trivial. For example, one can observe quantum interference effects as a length-independent decrease in  $J$  with varying conjugation patterns,<sup>14</sup> a lack of measurable current in *meta* substituted stilbene thiols<sup>64</sup> or negative curvature in  $\log \left| \frac{dI}{dV} \right|$ ,<sup>16</sup> but these interpretations all rely on the underlying

assumption that  $I$  and  $J$  are dominated by transport through molecules. Likewise, applying theoretical models to explain the interference effects relies on the same assumptions. This problem is particularly evident when experimental observations that disagree with theory are based on a somewhat ambiguous interpretations of data (*i.e.*, bi-modal distributions of conductance).<sup>65</sup> The series of molecules in this paper is not expected to exhibit any unusual transport properties, but despite the lack of a distance-dependence the  $J/V$  data presented in Fig. 2 are unambiguously dominated by transport through molecules. And we have shown that embedded dipoles have a measurable influence on the energetics within molecular tunneling junctions comprising TP1, TP1-down, and TP1-up, but do not have a significant influence on the magnitude of tunneling charge-transport.

## 3 Conclusions

We examined tunneling junctions comprising SAMs of three molecules of nearly identical length, packing density, tilt angle, torsional angle and gas-phase HOMO energies.<sup>42</sup> The only difference is the inclusion of a central pyrimidine ring, which introduces a dipole moment, the direction of which is synthetically controllable by adjusting the orientation of the



ring. The resulting dipole moments are embedded in the SAM as opposed to being introduced as a head (tail) group in contact with the top (bottom) electrode. Thus, we can eliminate both electrode interfaces, tunneling distance, packing, tilt, torsional angles, and gas-phase HOMO energies as variables and compare the tunneling transport properties.

We find that, outside of a slight difference in  $J$  at +1 V, the  $J/V$  curves are indistinguishable and this slight difference may be the result of the dipole moments affecting the distribution of HOMO-derived PDOS on or off of the Au electrode. The transition voltages, however, differ systemically and follow the same trend as the experimentally-determined vacuum level shift induced by the direction and magnitude of the embedded dipoles. The trends in Fig. 3 and 6 capture the critical aspect of investigating systematic behavior in  $V_T$ . The former relates an external experimental observable,  $\Phi$ , to an internal experimental observable,  $V_T$ . The latter relates this internal observable to the details of the level alignment that takes place when molecules are chemisorbed to a metal, which can in turn be related to experimentally observable energy positions of frontier electronic states.<sup>24</sup> Thus, the ability to manipulate  $V_T$  systematically through synthetic modifications away from the electrode interfaces simultaneously provides evidence that the charge transport is dominated by molecules and provides quantitative information about their electronic states. This physical interpretation of  $V_T$  is not new, but the isolation of the internal electrostatic profile of a molecule as a variable that affects  $V_T$  is an important step forward in the fundamental understanding of tunneling transport through molecular junctions and, ultimately, control over functionality.

This result demonstrates that (i)  $V_T$  can be manipulated synthetically in a predictable manner, (ii) changes to  $V_T$  can be ascribed to an intrinsic property of the molecules inside the tunneling junction, (iii) the energy level alignment can be adjusted using embedded dipoles without altering any other characteristic of a SAM. And, while the length dependence of conductance can be described by  $\beta$ ,  $V_T$  carries information about energy levels; trends in  $V_T$  can separate some of these influences. The inclusion of embedded dipoles (or specifically pyrimidine rings) instills a “molecular fingerprint” to tunneling transport that is separate from the magnitude of  $I$  or  $J$ . This observation is in agreement with studies showing that polar groups (and embedded dipoles in saturated molecules) have no influence on  $\beta$ .<sup>9</sup> While the lone pairs of a pyrimidyl moiety can interfere with edge-to- $\pi$  interactions, in this particular case all three SAMs pack nearly identically.<sup>42</sup> Thus, this effect is sufficiently weak that it is overcome by the flanking phenyl rings, suggesting that the use of pyrimidine moieties specifically to create a dipole moment is generalizable. We suggest that, irrespective of the precise physical interpretation of transition voltages, trends in  $V_T$ —specifically  $V_T$  versus  $\Delta\Phi$ —are particularly useful for unsaturated molecules in which molecular length is synthetically or experimentally inaccessible or in cases where  $\beta$  is not sensitive to synthetic alterations.

## Acknowledgements

RCC, DF, and AK acknowledge the European Research Council for the ERC Starting Grant 335473 (MOLECSYNCON). TW and MZ thank the Max-IV staff, and A. Preobrajenski in particular, for the technical support during the experiments at the synchrotron. Part of this work was supported financially by the Deutsche Forschungsgemeinschaft (grants ZH 63/17-1 and ZH 63/14-2). DAE and EZ acknowledge financial support by the Austrian Science Fund (FWF): project I937-N19 within the ERAChemistry framework and project P24666-N20. The computational studies presented have been mainly performed using the clusters of the division for high-performance computing at the Graz University of Technology.

## References

- 1 L. A. Bumm, *ACS Nano*, 2008, **2**, 403–407.
- 2 S. L. Bernasek, *Angew. Chem., Int. Ed.*, 2012, **51**, 9737–9738.
- 3 L. Jiang, C. S. S. Sangeetha, A. Wan, A. Vilan and C. A. Nijhuis, *J. Phys. Chem. C*, 2015, **119**, 960–969.
- 4 B. Gotsmann, H. Riel and E. Lörtscher, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 205408.
- 5 E. A. Weiss, R. C. Chiechi, G. K. Kaufman, J. K. Kriebel, Z. Li, M. Duati, M. A. Rampi and G. M. Whitesides, *J. Am. Chem. Soc.*, 2007, **129**, 4336–4349.
- 6 F. C. Simeone, H. J. Yoon, M. M. Thuo, J. R. Barber, B. Smith and G. M. Whitesides, *J. Am. Chem. Soc.*, 2013, **135**, 18131–18144.
- 7 C. A. Nijhuis, W. F. Reus, J. R. Barber, M. D. Dickey and G. M. Whitesides, *Nano Lett.*, 2010, **10**, 3611–3619.
- 8 H. J. Yoon, N. D. Shapiro, K. M. Park, M. M. Thuo, S. Soh and G. M. Whitesides, *Angew. Chem., Int. Ed.*, 2012, **51**, 4658–4661.
- 9 H. J. Yoon, C. M. Bowers, M. Baghbanzadeh and G. M. Whitesides, *J. Am. Chem. Soc.*, 2014, **136**, 16–19.
- 10 F. Mirjani, J. M. Thijssen, G. M. Whitesides and M. A. Ratner, *ACS Nano*, 2014, **8**, 12428–12436.
- 11 K. C. Liao, H. J. Yoon, C. M. Bowers, F. C. Simeone and G. M. Whitesides, *Angew. Chem., Int. Ed.*, 2014, **53**, 3889–3893.
- 12 D. Fracasso, S. Kumar, P. Rudolf and R. C. Chiechi, *RSC Adv.*, 2014, **4**, 56026–56030.
- 13 M. M. Thuo, W. F. Reus, F. C. Simeone, C. Kim, M. D. Schulz, H. J. Yoon and G. M. Whitesides, *J. Am. Chem. Soc.*, 2012, **134**, 10876–10884.
- 14 D. Fracasso, H. Valkenier, J. C. Hummelen, G. C. Solomon and R. C. Chiechi, *J. Am. Chem. Soc.*, 2011, **133**, 9556–9563.
- 15 N. Gorczak, N. Renaud, S. Tarkuç, A. J. Houtepen, R. Eelkema, L. D. A. Siebbeles and F. C. Grozema, *Chem. Sci.*, 2015, **6**, 4196–4206.
- 16 C. M. Guédon, H. Valkenier, T. Markussen, K. S. Thygesen, J. C. Hummelen and S. J. van der Molen, *Nat. Nanotechnol.*, 2012, **7**, 305–309.
- 17 C. R. Arroyo, S. Tarkuc, R. Frisenda, J. S. Seldenthuis, C. H. M. Woerde, R. Eelkema, F. C. Grozema and



- H. S. J. van der Zant, *Angew. Chem., Int. Ed.*, 2013, **52**, 3152–3155.
- 18 H. Valkenier, C. M. Guédon, T. Markussen, K. S. Thygesen, S. J. van der Molen and J. C. Hummelen, *Phys. Chem. Chem. Phys.*, 2014, **16**, 653–662.
- 19 Q. Lu, K. Liu, H. Zhang, Z. Du, X. Wang and F. Wang, *ACS Nano*, 2009, **3**, 3861–3868.
- 20 J. M. Beebe, B. Kim, J. W. Gadzuk, C. D. Frisbie and J. G. Kushmerick, *Phys. Rev. Lett.*, 2006, **97**, 026801.
- 21 J. M. Beebe, B. Kim, C. D. Frisbie and J. G. Kushmerick, *ACS Nano*, 2008, **2**, 827–832.
- 22 N. Bennett, G. Xu, L. J. Esdaile, H. L. Anderson, J. E. Macdonald and M. Elliott, *Small*, 2010, **6**, 2604–2611.
- 23 S. Guo, J. Hihath, I. Díez-Pérez and N. Tao, *J. Am. Chem. Soc.*, 2011, **133**, 19189–19197.
- 24 B. Kim, S. H. Choi, X.-Y. Zhu and C. D. Frisbie, *J. Am. Chem. Soc.*, 2011, **133**, 19864–19877.
- 25 M. C. Lennartz, N. Atodiresei, V. Caciuc and S. Karthäuser, *J. Phys. Chem. C*, 2011, **115**, 15025–15030.
- 26 G. Ricœur, S. Lenfant, D. Guérin and D. Vuillaume, *J. Phys. Chem. C*, 2012, **116**, 20722–20730.
- 27 X. Lefèvre, F. Moggia, O. Segut, Y.-P. Lin, Y. Ksari, G. Delafosse, K. Smaali, D. Guérin, V. Derycke, D. Vuillaume, S. Lenfant, L. Patrone and B. Jousset, *J. Phys. Chem. C*, 2015, **119**, 5703–5713.
- 28 S. Guo, G. Zhou and N. Tao, *Nano Lett.*, 2013, **13**, 4326–4332.
- 29 I. Bâldea, *J. Am. Chem. Soc.*, 2012, **134**, 7958–7962.
- 30 M. Aradai and M. Tsukada, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **81**, 235114.
- 31 E. H. Huisman, C. M. Guédon, B. J. van Wees and S. J. van der Molen, *Nano Lett.*, 2009, **9**, 3909–3913.
- 32 J. Chen, T. Markussen and K. S. Thygesen, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2010, **82**, 121412.
- 33 I. Bâldea, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **85**, 035442.
- 34 A. Tan, J. Balachandran, B. D. Dunietz, S.-Y. Jang, V. Gavini and P. Reddy, *Appl. Phys. Lett.*, 2012, **101**, 243107.
- 35 G. Wang, Y. Kim, S.-I. Na, Y. H. Kahng, J. Ku, S. Park, Y. H. Jang, D.-Y. Kim and T. Lee, *J. Phys. Chem. C*, 2011, **115**, 17979–17985.
- 36 A. Vilan, D. Cahen and E. Kraisler, *ACS Nano*, 2013, **7**, 695–706.
- 37 K. Soththewes, C. Hellenthal, A. Kumar and H. J. W. Zandvliet, *RSC Adv.*, 2014, **4**, 32438–32442.
- 38 R. C. Chiechi, E. A. Weiss, M. D. Dickey and G. M. Whitesides, *Angew. Chem., Int. Ed.*, 2008, **120**, 148–150.
- 39 W. F. Reus, M. M. Thuo, N. D. Shapiro, C. A. Nijhuis and G. M. Whitesides, *ACS Nano*, 2012, **6**, 4806–4822.
- 40 C. A. Nijhuis, W. F. Reus, J. R. Barber and G. M. Whitesides, *J. Phys. Chem. C*, 2012, **116**, 14139–14150.
- 41 E. A. Weiss, G. K. Kaufman, J. K. Kriebel, Z. Li, R. Schalek and G. M. Whitesides, *Langmuir*, 2007, **23**, 9686–9694.
- 42 T. Abu-Husein, S. Schuster, D. A. Egger, M. Kind, T. Santowski, A. Wiesner, R. Chiechi, E. Zojer, A. Terfort and M. Zharnikov, *Adv. Funct. Mater.*, 2015, **25**, 3943–3957.
- 43 J. B. Neaton, M. S. Hybertsen and S. G. Louie, *Phys. Rev. Lett.*, 2006, **97**, 216405.
- 44 S. Y. Quek, L. Venkataraman, H. J. Choi, S. G. Louie, M. S. Hybertsen and J. B. Neaton, *Nano Lett.*, 2007, **7**, 3477–3482.
- 45 D. M. Alloway, M. Hofmann, D. L. Smith, N. E. Gruhn, A. L. Graham, R. Colorado, V. H. Wysocki, T. R. Lee, P. A. Lee and N. R. Armstrong, *J. Phys. Chem. B*, 2003, **107**, 11690–11699.
- 46 D. Fracasso, M. I. Muglali, M. Rohwerder, A. Terfort and R. C. Chiechi, *J. Phys. Chem. C*, 2013, **117**, 11367–11376.
- 47 H. J. Yoon, K. C. Liao, M. R. Lockett, S. W. Kwok, M. Baghbanzadeh and G. M. Whitesides, *J. Am. Chem. Soc.*, 2014, **136**, 17155–17162.
- 48 G. M. Morales, P. Jiang, S. Yuan, Y. Lee, A. Sanchez, W. You and L. Yu, *J. Am. Chem. Soc.*, 2005, **127**, 10456–10457.
- 49 Y. Lee, B. Carsten and L. Yu, *Langmuir*, 2009, **25**, 1495–1499.
- 50 I. Díez-Pérez, J. Hihath, Y. Lee, L. Yu, L. Adamska, M. A. Kozhushner, I. I. Oleynik and N. Tao, *Nat. Chem.*, 2009, **1**, 635–641.
- 51 G. Zhang, M. A. Ratner and M. G. Reuter, *J. Phys. Chem. C*, 2015, **119**, 6254–6260.
- 52 O. M. Cabarcos, A. Shaporenko, T. Weidner, S. Uppili, L. S. Dake, M. Zharnikov and D. L. Allara, *J. Phys. Chem. C*, 2008, **112**, 10842–10854.
- 53 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2003, **118**, 8207–8215.
- 54 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2006, **124**, 219906.
- 55 G. Kresse and J. Furthmüller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, **54**, 11169–11186.
- 56 F. Rissner, D. A. Egger, A. Natan, T. Körzdörfer, S. Kümmel, L. Kronik and E. Zojer, *J. Am. Chem. Soc.*, 2011, **133**, 18634–18645.
- 57 N. Dori, M. Menon, L. Kilian, M. Sokolowski, L. Kronik and E. Umbach, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2006, **73**, 195208.
- 58 A. M. Track, F. Rissner, G. Heimel, L. Romaner, D. Käfer, A. Bashir, G. M. Rangger, O. T. Hofmann, T. Bučko, G. Witte and E. Zojer, *J. Phys. Chem. C*, 2010, **114**, 2677–2684.
- 59 A. Biller, I. Tamblyn, J. B. Neaton and L. Kronik, *J. Chem. Phys.*, 2011, **135**, 164706.
- 60 D. A. Egger, Z.-F. Liu, J. B. Neaton and L. Kronik, *Nano Lett.*, 2015, **15**, 2448–2455.
- 61 A. Natan, L. Kronik, H. Haick and R. Tung, *Adv. Mater.*, 2007, **19**, 4103–4117.
- 62 G. Heimel, F. Rissner and E. Zojer, *Adv. Mater.*, 2010, **22**, 2494–2513.
- 63 B. Kretz, D. A. Egger and E. Zojer, *Adv. Sci.*, 2015, **2**, 1400016.
- 64 S. V. Aradhya, J. S. Meisner, M. Krikorian, S. Ahn, R. Parameswaran, M. L. Steigerwald, C. Nuckolls and L. Venkataraman, *Nano Lett.*, 2012, **12**, 1643–1647.
- 65 J. Xia, B. Capozzi, S. Wei, M. Strange, A. Batra, J. R. Moreno, R. J. Amir, E. Amir, G. C. Solomon, L. Venkataraman and L. M. Campos, *Nano Lett.*, 2014, **14**, 2941–2945.

