

## CORRECTION

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## Correction: Properties and applications of designable and photo/redox dual responsive surfactants with the new head group 2-arylaZO-imidazolium

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The authors wish to amend two statements and a figure in the original article that concern the reference electrode used in the electrochemical studies. The use of a Ag/AgCl reference electrode is reported in the original article. However, the authors actually used a saturated calomel electrode (SCE) as the reference electrode and a glassy carbon electrode (GCE) as the working electrode.

Therefore, in the sub-section entitled *Electrochemical properties*, the reduction and oxidation peaks are centered at  $-0.24$  V and  $-0.17$  V vs. SCE, respectively. Furthermore, in the *Apparatus and procedures* sub-section, the text should be amended to read: 'Electrochemical measurements were performed in a conventional three-electrode system with a glassy carbon electrode (GCE) as the working electrode, a saturated calomel electrode (SCE) as the reference electrode and a Pt wire as the counter electrode.' The change has also been reflected in a modified Fig. 7, in which the x axis label has been altered accordingly.

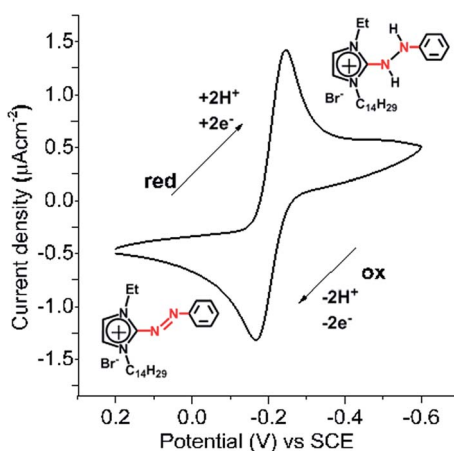


Fig. 7 Cyclic voltammetry of  $0.1 \text{ mg mL}^{-1}$  2-Br in 0.1 M PBS and possible mechanism of redox reactivity.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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