CrossMark
click for updates

Cite this: *RSC Adv.*, 2016, 6, 37994

DOI: 10.1039/c6ra90036d

www.rsc.org/advances

Correction: Calcite precipitation from by-product red gypsum in aqueous carbonation process

Omeid Rahmani,^{*ab} Mark Tyrer^c and Radzuan Junin^a

Correction for 'Calcite precipitation from by-product red gypsum in aqueous carbonation process' by Omeid Rahmani et al., *RSC Adv.*, 2014, 4, 45548–45557.

The authors regret that the reproduction of images from their previous work was not acknowledged in their article. Permission for the reproduction of images from O. Rahmani, R. Junin, M. Tyrer, R. Mohsin, *Energy Fuels*, 2014, 28, 5953–5958 (cited as ref. 11 in the original article) had been obtained from the publisher, but was not declared alongside the relevant figures.

The captions for Fig. 2 and 6 have therefore been revised to the following:

"Fig. 2 X-ray diffraction of bulk red gypsum sample. Reproduced with permission from ref. 11. Copyright 2014, American Chemical Society."

"Fig. 6 The effect of procedure variables (a) reaction temperature, (b) stirring rate, (c) liquid to solid ratio, and (d) particle size on conversion of Ca to CaCO₃ in mineral carbonation process. Reproduced with permission from ref. 11. Copyright 2014, American Chemical Society."

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

^aDepartment of Petroleum Engineering, Faculty of Petroleum and Renewable Energy Engineering, Universiti Teknologi Malaysia (UTM), Johor 81310, Malaysia. E-mail: omeid.rahmani@iau-mahabad.ac.ir; omeidrahmani@gmail.com; Tel: +98 9144422009

^bDepartment of Petroleum Engineering, Mahabad Branch, Islamic Azad University, Mahabad 59135–433, Iran

^cMineral Industry Research Organisation, Wellington House, Starley Way, Birmingham International Park, Solihull, Birmingham, B37 7HB, UK