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Base mediated 1,3-dipolar cycloaddition of α -substituted vinyl phosphonates with diazo compounds for synthesis of 3-pyrazolylphosphonates and 5-pyrazolcarboxylates†

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5-Aryl-substituted pyrazol-3-ylphosphonates have been conveniently synthesized by 1,3-dipolar cycloaddition of 1-formamidovinylphosphonates and aryl diazomethanes under $K_2CO_3/MeOH$ conditions at room temperature. These pyrazoles are formed in one pot *via* spontaneous elimination of formamide. Basic conditions prevent competitive formation of cyclopropylphosphonates. 3-Aryl substituted pyrazol-5-carboxylates can be synthesized by the same methodology from 1-arylvinylphosphonates and ethyl diazoacetate, although a stronger base NaH is necessary to ensure the success of the aromatization stage with the elimination of the diethoxylphosphoryl moiety.

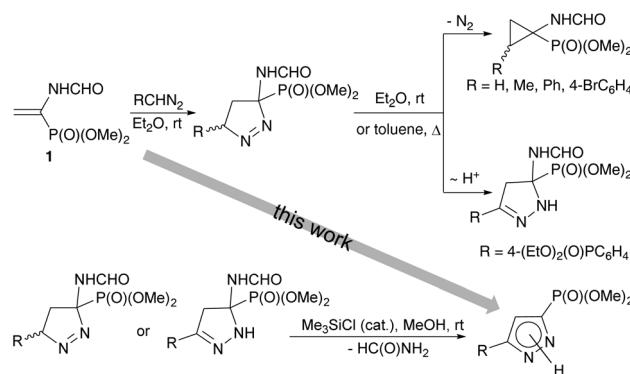
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Introduction

A pyrazole moiety is regarded as a privileged heterocyclic skeleton with multifarious applications. Pyrazoles represent the core structure of numerous naturally occurring molecules (*e.g.*, pyrazol-3(5)-carboxylic acid from the sponge *Tedania anhelans*, (*S*)- β -pyrazolylalanine from *Citrullus vulgaris*, and alkaloid Withasomnine from Indian medicinal plant *Withania somnifera*),¹ pharmaceuticals (*e.g.*, the marketed drugs Celebrex and Viagra), and crop protection agents.² Pyrazoles are also efficient coordinating ligands in catalysis³ and supramolecular chemistry.⁴ Among various functionalized pyrazoles, pyrazolylphosphonates have received considerable attention over the past few decades due to their remarkable bioactivity profiles⁵ and possible applications as organic precursors.⁶ Increasing interest in pyrazolylphosphonates has been reflected by the development of various synthetic methodologies allowing straightforward access to these structural motifs, however, the main focus had been on 4-pyrazolylphosphonates,⁷ while 3(5)-pyrazolylphosphonates⁸ received much less attention. A remarkable progress had been initiated by the pioneering work of Namboothiri *et al.* who proposed to employ the Bestmann–Ohira reagent as a 1,3-dipolar precursor in base-

mediated cycloaddition reactions with conjugated nitroalkenes.⁹ This methodology was further extended on a variety of dipolarophiles and nowadays presents the most developed synthetic route to 3(5)-pyrazolylphosphonates.^{8,10,11} By contrast, an alternative approach utilizing α,β -unsaturated phosphonates as cycloaddition partners has been scarcely reported, is frequently complicated by competitive formation of cyclopropylphosphonates, and thus remains a challenging task.

Recently we have reported an efficient approach to the synthesis of 2-substituted 1-aminocyclopropylphosphonates based on the regioselective 1,3-dipolar cycloaddition reaction of diazo compounds with dimethyl 1-formamidovinylphosphonate (**1**) (Scheme 1).^{12,13} The reaction pathway involves the formation of 1-pyrazoline. The subsequent elimination of a nitro-



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Scheme 1 Summary of our previous results on 1,3-dipolar cycloaddition reaction of vinylphosphonate **1** with diazo compounds.



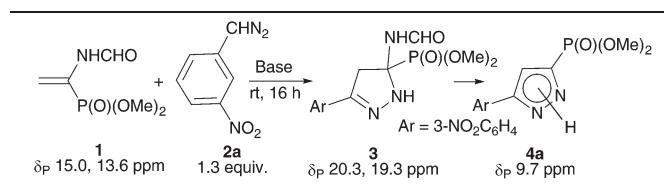
gen molecule affords the cyclopropane ring. However in the case of (4-diethoxyphosphorylphenyl)diazomethane containing a strong electron-acceptor substituent in the ring,¹⁴ the major reaction product is 2-pyrazoline, which is resistant to nitrogen extrusion. It was also shown that the aromatization of 1- and 2-pyrazolines in acidic media is accompanied by the elimination of formamide to give 3-pyrazolylphosphonates.

In the present work, the possibility to give a new turn to the general reaction scheme and to adapt it for the target one-pot synthesis of 3(5)-phosphonylated pyrazoles was demonstrated. All one has to do is provide conditions for the fast isomerization of the initially formed 1-pyrazoline to 2-pyrazoline (thus suppressing the possibility of nitrogen extrusion with cyclopropane formation) and the elimination of formamide. Both steps would be expected to be catalyzed by base.^{15,16} Therefore the preliminary optimization of the reaction conditions included the variation of bases and solvents.

Results and discussion

Reasonably stable and, hence, easy to handle (3-nitrophenyl)diazomethane (**2a**) was chosen as a model diazo compound. The reaction course was monitored using the ³¹P NMR method by the disappearance of signals from two rotamers of vinylphosphonate **1** at δ_p 15.0 and 13.6 ppm (ref. 17) and accumulation of signals from the products. The obtained results presented in Table 1 show that the reaction of **1** with 1.3 equiv. of **2a** in

Table 1 Optimization of the reaction conditions for preparation of pyrazolylphosphonate **4a**^a



Entry	Solvent	Base	Conversion of 1 ^b (%)	Yield ^{b,c} (%)	
				3	4a
1	Et ₂ O	K ₂ CO ₃	96	87 (79)	3
2	CH ₂ Cl ₂	K ₂ CO ₃	98	68	25
3	MeOH	K ₂ CO ₃	98	1	89 (76)
4	MeOH	Cs ₂ CO ₃	99	5	84
5	MeOH	K ₃ PO ₄	99	5	86
6	MeOH	MeONa	99	3	89

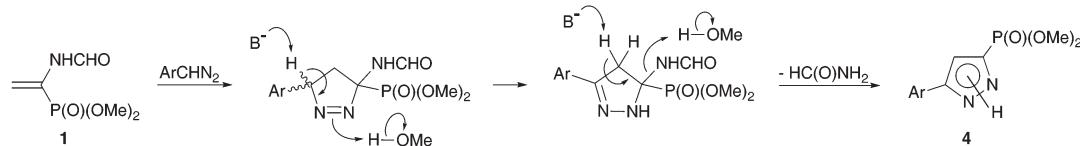
^a Reaction conditions: **1** (0.25 mmol), **2a** (0.325 mmol), base (10 mol%), solvent (0.65 ml), rt, 16 h. ^b Determined by ³¹P NMR analysis of the crude reaction mixture. ^c Isolated yields are given in the parentheses.

ether in the presence of 10 mol% K₂CO₃ completes in 16 h and affords almost exclusively 2-pyrazoline **3** (two rotamers in a ratio of 86:14; δ_p 20.3 and 19.3 ppm), which precipitates (entry 1). The yield of pyrazolylphosphonate **4a** (δ_p 9.7 ppm) was only 3% and increased to 25% when the experiment was repeated in CH₂Cl₂ in which all reaction products are soluble (entry 2). The use of MeOH turned out to be most productive: in this case, the yield of the target pyrazole **4a** was 89% (entry 3). Note that in all experiments the conversion of vinylphosphonate **1** was close to quantitative, which is consistent with the known fact that the rate of 1,3-dipolar cycloaddition is almost insensitive to the polarity of the medium.¹⁸ In all experiments the formation of dimethyl 1-formamido-2-(3-nitrophenyl)cyclopropylphosphonates (δ_p 23.3 and 22.7 ppm for the *cis*-isomer and δ_p 25.7 and 24.9 ppm for the *trans*-isomer) was detected, but their total yield did not exceed 2%.

The variation of bases showed that the replacement of K₂CO₃ by Cs₂CO₃ (entry 4), K₃PO₄ (entry 5), or MeONa (entry 6) exerted almost no effect on the yield of the target product **4a** and, therefore, cheaper potash was used in further experiments.

The absence of a noticeable effect of the nature of the base and a rather strong solvent effect is consistent with the proposed reaction mechanism (Scheme 2), according to which the rate determining stage of 1,3-dipolar cycloaddition leads to 1-pyrazoline formation. The base initiates the fast isomerization of 1-pyrazoline to thermodynamically more stable conjugated 2-pyrazoline. The role of alcohol is most probably the protonation of the leaving group in the next step of formamide elimination rather than increasing the solubility of an inorganic base.¹⁹

Under the optimal conditions found, a large series of aryl-diazomethanes **2a–p** containing both acceptor and donor substituents in the ring were introduced into the reaction with vinylphosphonate **1**, and the corresponding 5-aryl-substituted dimethyl (1*H*-pyrazol-3-yl)phosphonates **4a–p** were obtained in high yields after chromatographic purification (Table 2). Commenting on the obtained results, two points should be mentioned. First, the reaction time, required for quantitative conversion of vinylphosphonate **1**, is noticeably longer in the case of aryl-diazomethanes containing strong electron-acceptor substituents in the ring, especially **2b**.²⁰ The same concerns the sterically hindered aryl-diazomethanes **2e,f** containing *ortho*-substituents. These two factors are combined in (2,4-dinitrophenyl)diazomethane, which turned out to be completely inert under the reaction conditions. The second point is the necessity to use a substantial excess of aryl-diazomethanes with electron-donor substituents in the ring (e.g., 4-EtO (**2p**),



Scheme 2 Proposed pathway for pyrazolylphosphonate **4** formation.

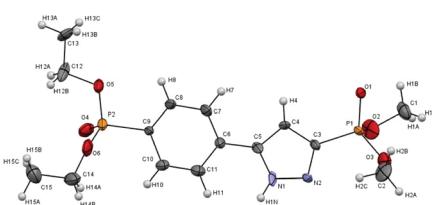
Table 2 Preparation of 5-aryl substituted dimethyl (1*H*-pyrazol-3-yl) phosphonates **4**^a

Entry	2 (equiv.)	Ar	Product	Reaction	Yield of
				time ^b (h)	
1	2a (1.3)	3-NO ₂ C ₆ H ₄	4a	16	76
2	2b (1.3)	4-NO ₂ C ₆ H ₄	4b	30	72
3	2c (1.5)	4-(EtO) ₂ P(O)C ₆ H ₄	4c	24	79
4	2d (1.5)	4-MeO ₂ CC ₆ H ₄	4d	18	78
5	2e (1.4)	2-ClC ₆ H ₄	4e	17	63
6	2f (1.3)	2-BrC ₆ H ₄	4f	16	54
7	2g (1.3)	3-ClC ₆ H ₄	4g	12	78
8	2h (1.3)	4-ClC ₆ H ₄	4h	4	87
9	2i (1.4)	4-BrC ₆ H ₄	4i	3	87
10	2j (3)	4-MeSC ₆ H ₄	4j	3	86
11	2k (3)	3-MeOC ₆ H ₄	4k	3	71
12	2l (1.3)	Ph	4l	3	85 (91) ^d
13	2m (1.5)	4-FC ₆ H ₄	4m	3	67
14	2n (4)	4-i-PrC ₆ H ₄	4n	3	80
15	2o (2)	4-MeOC ₆ H ₄	4o	3	76
16	2p (1.6)	4-EtOC ₆ H ₄	4p	2	63

^a Reaction conditions: **1** (0.25 mmol), **2** (equiv., 0.5 M solution in MeOH), K₂CO₃ (10 mol%), rt. ^b Time to completion, estimated by ³¹P NMR. ^c Isolated yield. ^d The yield in the reaction performed on a 2.5 mmol scale is given in the parentheses.

4-MeO (**2o**), and 4-i-Pr (**2n**)), since these diazo compounds are noticeably decomposed during the reaction to form substituted *cis*- and *trans*-stilbenes and dibenzylidenehydrazines.²¹ Not unexpectedly, the amount of cyclopropanes increases to 9–10% of total yield, since the donor substituents should impede the proton transfer and isomerization of 1-pyrazoline to 2-pyrazoline.

The structure of pyrazolylphosphonate **4c** was unambiguously proved by the X-ray diffraction analysis data (Fig. 1).^{13,22} In the ¹H NMR spectra of pyrazolylphosphonates **4** the doublet of the C(4)H proton of the pyrazole ring at δ_{H} 6.8–7.3 ppm ($^3J_{\text{PH}} = 1.8$ –2.4 Hz) is characteristic. The analysis of the ¹³C NMR spectra is complicated by the fact that the signals of the carbon atoms of the pyrazole ring and the *ipso*-carbon of the aryl fragment broadened, which is explained by prototropic ring tautomerism in the solution.^{11b,c,15,23} The detection of a doublet of the quaternary carbon atom bound to phosphorus and lying at δ_{C} 132.7–138.0 ppm ($^1J_{\text{PC}} = 220$ –229 Hz) was



The synthesis conditions were optimized for the model diethyl 1-phenylvinylphosphonate (**7a**). The reaction course was monitored by ^{31}P and ^1H NMR spectroscopy. On heating at 50 °C a solution of vinylphosphonate **7a** and ethyl diazoacetate (2 equiv.) in THF in the presence of NaH (2.5 equiv.), the reaction took 18 h for completion. After neutralization of the reaction mixture with a solution of HCl in ether followed by treatment with triethylamine, the target pyrazolecarboxylate **9a** was isolated chromatographically in 82% yield (Table 4, entry 1).

The reaction mechanism (Scheme 4), assuming the deprotonation of 1-pyrazoline formed at the first step, followed by the elimination of the diethyl phosphite anion *via* the E1cB mechanism, and repeated deprotonation to form an aromatic structure, explains the necessity to use two equivalents of sodium hydride.

The replacement of NaH by K_2CO_3 results in the termination of the reaction at the step of formation of 2-pyrazoline **8** when using THF (entry 2) or ethanol (entry 3) as a solvent. No formation of target pyrazolecarboxylate **9a** was observed even on increasing the reaction temperature to 80 °C (in dioxane), but the selectivity of the process decreased noticeably because of the formation of by-products (entry 4).

2-Pyrazoline **8** was isolated and characterized. The aromatization of 2-pyrazoline **8** occurs smoothly in the presence of 2 equiv. NaH in THF. An increase of temperature from ambient to 50 °C shortens the reaction time from 24 to 5 h. The aromatization can be carried out both for individual pyrazoline **8** and *in situ*, adding 2 equiv. NaH to the reaction mixture in THF after the reaction of vinylphosphonate **7a** with ethyl

diazoacetate in the presence of K_2CO_3 has ceased. In both cases, pyrazolecarboxylate **9a** was isolated in a yield of 66%.

A series of vinylphosphonates **7b–e** containing 4-chlorophenyl, 4-isobutylphenyl, 4-diphenyl, and 2-naphthyl substituents in the α -position were introduced into the reaction with ethyl diazoacetate to extend the scope of the method. Since the yields of product **9a** were comparable when using both the one-step (method A) and two-step (method B) protocols, we checked both procedures for each substrate of **7b–e**. The obtained results (Table 5) showed that both protocols made

Table 5 Preparation of 3-aryl substituted ethyl 1*H*-pyrazol-5-carboxylates **9**

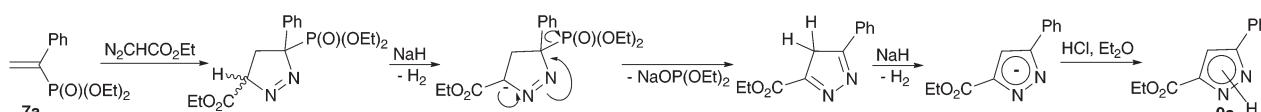
Entry	7	Ar	Product	Isolated yield (%)	
				Method A ^a	Method B ^b
1	7a	Ph	9a	82	66
2	7b	4-ClC ₆ H ₄	9b	57	62
3	7c	4-i-BuC ₆ H ₄	9c	60	56
4	7d	4-PhC ₆ H ₄	9d	76	83
5	7e	2-Naph	9e	74	68

^a Reaction conditions: **7** (0.4 mmol), $\text{N}_2\text{CHCO}_2\text{Et}$ (0.8 mmol), NaH (1.0 mmol), THF (2 ml), 50 °C, 18 h. ^b Reaction conditions: (1) **7** (0.4 mmol), $\text{N}_2\text{CHCO}_2\text{Et}$ (0.6–0.8 mmol), K_2CO_3 (5 mol%), THF (2 ml), 50 °C, 48–55 h (^{31}P NMR monitoring); (2) NaH (0.8 mmol), 50 °C, 5 h.

Table 4 Optimization studies on the reaction of vinylphosphonate **7a** with ethyl diazoacetate^a

Entry	Solvent	Base	T (°C)	Reaction time (h)	Conv. of 7a ^b (%)	Yield (%)	
						8 ^{b,c}	9a
1	THF	NaH (2.5 equiv.)	50	18	97	0	82 ^{d,e}
2	THF	K_2CO_3 (5 mol%)	50	48	97	92(43)	0 ^f
3	EtOH	K_2CO_3 (5 mol%)	60	24	92	86	0 ^f
4	Dioxane	K_2CO_3 (5 mol%)	80	20	92	63	0 ^f

^a Reaction conditions: **7a** (0.4 mmol), $\text{N}_2\text{CHCO}_2\text{Et}$ (0.8 mmol), base, solvent (2 ml). ^b Determined by ^{31}P NMR analysis of the crude reaction mixture. ^c Isolated yield is given in the parentheses. ^d Isolated yield after column chromatography. ^e Formation of tetraethyl (1-phenylethane-1,2-diy)bis(phosphonate) (11%; δ_{P} 27.7 and 29.2 ppm ($^3J_{\text{P-P}} = 82.9$ Hz)) was also detected by NMR analysis of the crude reaction mixture. ^f Determined by ^1H NMR analysis of the crude reaction mixture; doublets due to *ortho*-protons are observed at 7.51, 7.45, and 7.72 ppm for compounds **7a**, **8**, and **9a**, respectively.



Scheme 4 Proposed pathway for pyrazolcarboxylate **9a** formation.

it possible to obtain products **9b–e** in good yields. The use of the longer method B is justified in the case of substrates **7b,d** because it gave better yields of the corresponding products **9b,d**.

Products **9** were characterized by all relevant methods. The spectral characteristics of substrates **9a,b,e** are consistent with those described in the literature.^{25d,l} Note that the published X-ray diffraction data for the 4-methoxyphenyl analog of products **9** show that in the crystalline state pyrazolecarboxylates **9**, unlike pyrazolylphosphonates **4**, exist as an alternative tautomer ethyl 3-aryl-1*H*-pyrazole-5-carboxylate.^{25d}

Conclusions

In conclusion, 1,3-dipolar cycloaddition of aryl diazomethanes with dimethyl 1-formamido vinylphosphonate (**1**) exhibits a condition-controlled product-selectivity. Under K_2CO_3 /MeOH conditions nitrogen extrusion with cyclopropane ring formation is suppressed and the reaction affords 5-aryl substituted dimethyl (1*H*-pyrazol-3-yl)phosphonates **4** in high yields. Mild reaction conditions and broad functional group tolerance make this strategy synthetically useful. In much the same way, 1,3-dipolar cycloaddition of ethyl diazoacetate with diethyl 1-arylvinyolphosphonates **7** under NaH/THF conditions provides regioisomerically pure 3-aryl substituted ethyl 1*H*-pyrazol-5-carboxylates **9**, the reaction being accompanied by the elimination of the diethoxyphosphoryl moiety.

Experimental

Reactions were routinely performed under dry conditions in oven-dried glassware and under an air atmosphere unless otherwise specified. Methanol was distilled from magnesium methoxide, and THF and Et_2O were distilled from sodium benzophenone ketyl prior to use. The starting material vinylphosphonate **1** was prepared according to the known method.^{17,28} Vinylphosphonates **7** were obtained as previously reported.²⁷ Ethyl diazoacetate was purchased from Aldrich and used as received. Aryldiazomethanes **2** were generated from sodium salts of tosylhydrazones of the corresponding aromatic aldehydes by the vacuum pyrolysis method²⁹ (for **2d,o,p**) or by the pyrolysis in the ethylene glycol procedure³⁰ (for **2a–c,e–n**). (Caution! Although we did not experience any problems in handling aryl diazomethanes, full safety precautions should be taken due to their potentially explosive nature.)

NMR data were recorded on Bruker Avance-300, 400 or Agilent 400-MR spectrometers at ambient temperature. ^{13}C and ^{31}P NMR spectra were 1H decoupled. Chemical shifts are reported on the δ -scale in parts per million relative to the solvent ($CDCl_3$: δ_C 77.0; Acetone- d_6 : δ_C 29.8; CD_3OD : δ_C 49.0; $DMSO-d_6$: δ_C 39.5) or the residual solvent peak ($CHCl_3$: δ_H 7.25; Acetone- d_6 : δ_H 2.04; CD_3OD : δ_H 3.30) as internal standards, or to external 85% H_3PO_4 (δ_P 0). The IR spectra were taken on a SPECORD 75 IR instrument using KBr pellets.

High resolution mass spectra (HRMS) were measured on a Bruker maXis spectrometer using electrospray ionization (ESI). Elemental analyses were carried out on an Elementar Vario MICRO Cube analyzer. Melting points were measured with Electrothermal 9100 apparatus and are uncorrected. Column chromatography was carried out using Macherey-Nagel silica gel 60 (0.015–0.04 mm). Preparative thin-layer chromatography (TLS) was performed using 20 × 20 cm pre-coated glass plates SIL G-100 UV254 (Macherey-Nagel) and visualized by UV (254 nm).

Dimethyl [5-(formamido)-3-(3-nitrophenyl)-4,5-dihydro-1*H*-pyrazol-5-yl]phosphonate (3)

A foil-covered 4 ml vial was charged with vinylphosphonate **1** (45 mg, 0.25 mmol), 0.5 M solution of the diazo compound **2a** in Et_2O (0.65 ml, 0.325 mmol), and K_2CO_3 (3.5 mg, 0.025 mmol). The reaction mixture was stirred at rt for 16 h. The resultant precipitate was collected by filtration, washed with Et_2O , acetone, and water and dried in a vacuum desiccator over P_2O_5 to give 2-pyrazoline **3** as a yellowish solid (68 mg) in 79% yield; a 86:14 mixture of two rotamers in $CDCl_3$. ^{31}P NMR (162 MHz, $CDCl_3$): δ 20.3 (major), 19.3 (minor). 1H NMR (400 MHz, $CDCl_3$) (only for the major rotamer): δ 3.53–3.67 m (2H, CH_2), 3.91 (d, $^3J_{H-P}$ = 10.6 Hz, 3H, OCH_3), 3.93 (d, $^3J_{H-P}$ = 10.4 Hz, 3H, OCH_3), 6.36 (br. s, 1H, $NHC=O$), 7.07 (br. s, 1H, NNH), 7.56 (dd, $^3J_{H-H}$ = $^3J_{H-P}$ = 8.0 Hz, 1H, ArH), 8.04 (m, $^3J_{H-H}$ = 8.0 Hz, 1H, ArH), 8.18 (s, 1H, CHO), 8.20 (m, $^3J_{H-H}$ = 8.0 Hz, 1H, ArH), 8.40 (dd, $^4J_{H-H}$ = $^4J_{H-P}$ = 1.7 Hz, 1H, ArH). ^{13}C NMR (101 MHz, $CDCl_3$) (only for the major rotamer): δ 42.2 (d, $^2J_{C-P}$ = 3.1 Hz, CH_2), 54.3 (d, $^2J_{C-P}$ = 7.5 Hz, OCH_3), 55.2 (d, $^2J_{C-P}$ = 6.7 Hz, OCH_3), 76.1 (d, $^1J_{C-P}$ = 197.7 Hz, PC), 121.0 (CH), 123.8 (CH), 129.7 (CH), 131.7 (CH), 133.4 (C_{Ar}), 148.5 (d, J_{C-P} = 14.4 Hz, C=N), 148.7 (CNO₂), 161.1 (d, $^3J_{C-P}$ = 5.2 Hz, C=O). Anal. calcd for $C_{12}H_{15}N_3O_5P\cdot 2H_2O$: C, 38.10; H, 5.06; N, 14.81. Found: C, 38.61; H, 4.55; N, 14.41.

Representative synthesis: dimethyl [5-(3-nitrophenyl)-1*H*-pyrazol-3-yl]phosphonate (4a)

A foil-covered 4 ml vial was charged with vinylphosphonate **1** (45 mg, 0.25 mmol), 0.5 M solution of the diazo compound **2a** in MeOH (0.65 ml, 0.325 mmol), and K_2CO_3 (3.5 mg, 0.025 mmol). The reaction mixture was stirred at rt for 16 h. Volatile components were removed on a rotary evaporator and the residue was purified by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.4) to afford pyrazolylphosphonate **4a** as a colorless solid (56.5 mg) in 76% yield. Mp 185 °C. ^{31}P NMR (162 MHz, $CDCl_3$): δ 9.7. 1H NMR (400 MHz, $CDCl_3$): δ 3.87 (d, $^3J_{H-P}$ = 11.6 Hz, 6H, OCH_3), 7.09 (d, $^3J_{H-P}$ = 2.4 Hz, 1H, C(4)H), 7.61 (dd, $^3J_{H-H}$ = $^3J_{H-P}$ = 8.0 Hz, 1H, ArH), 8.17–8.22 (m, 2H, ArH), 8.67 (dd, $^4J_{H-H}$ = $^4J_{H-P}$ = 1.9 Hz, 1H, ArH), 12.86 (br. s, 1H, NH). ^{13}C NMR (101 MHz, $CDCl_3$): δ 53.6 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 109.1 (d, $^2J_{C-P}$ = 19.1 Hz, C(4)), 120.7 (CH), 122.9 (CH), 129.8 (CH), 131.6 (CH), 132.7 (d, $^1J_{C-P}$ = 223 Hz, CP), 133.6 (C_{Ar}), 148.7 (CNO₂), 149.3 (d, $^3J_{C-P}$ = 14.3 Hz, C(5)). IR (KBr): ν 3400, 1550, 1529, 1344, 1248, 1184, 1063, 1003, 843,



781, 741 cm^{-1} . Anal. calcd for $\text{C}_{11}\text{H}_{12}\text{N}_3\text{O}_5\text{P}$: C, 44.45; H, 4.07; N, 14.14. Found: C, 44.40; H, 4.31; N, 13.84.

Dimethyl [5-(4-nitrophenyl)-1*H*-pyrazol-3-yl]phosphonate (4b)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2b** in MeOH (0.65 ml, 0.325 mmol); the reaction time was 30 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.3) gave pyrazolyl-phosphonate **4b** as a white solid (53.5 mg) in 72% yield. Mp 198 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, acetone- d_6): δ 7.4 (major tautomer), 11.7 (minor tautomer) in 83:17 ratio. ^1H NMR (400 MHz, acetone- d_6): δ 3.80 (d, $^3J_{\text{H-P}} = 11.4$ Hz, 6H, OCH_3), 7.38 (d, $^3J_{\text{H-P}} = 1.9$ Hz, 1H, C(4)H), 8.18 (br. d, $^3J_{\text{H-H}} = 8.8$ Hz, 2H, ArH), 8.31 (br. d, 2H, ArH), 13.59 (br. s, 1H, NH). ^{13}C NMR (101 MHz, acetone- d_6): δ 53.5 (d, $^2J_{\text{C-P}} = 5.6$ Hz, OCH_3), 111.2 (d, $^2J_{\text{C-P}} = 20.1$ Hz, C(4)), 124.9 (2CH), 127.2 (2CH), 139.0 (br., C(5)), 148.3 (CNO₂). IR (KBr): ν 3420, 1518, 1346, 1236, 1184, 1061, 1030, 854, 600 cm^{-1} . Anal. calcd for $\text{C}_{11}\text{H}_{12}\text{N}_3\text{O}_5\text{P}$: C, 44.45; H, 4.07; N, 14.14. Found: C, 44.29; H, 4.12; N, 13.

Dimethyl [5-(4-(diethoxyphosphoryl)phenyl)-1*H*-pyrazol-3-yl]phosphonate (4c)

The reaction was carried out as described for **4a** using 0.5 M solution of diazo compound **2c** in MeOH (0.75 ml, 0.375 mmol); the reaction time was 24 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.17) gave pyrazolyl-phosphonate **4c** as a pale yellow oil (77 mg) in 79% yield. ^{31}P NMR (121 MHz, CD_3OD): δ 11.6, 18.4. ^1H NMR (300 MHz, CD_3OD): δ 1.33 (dt, $^3J_{\text{H-H}} = 7.1$ Hz, $^4J_{\text{H-P}} = 0.5$ Hz, 6H, CCH_3), 3.84 (d, $^3J_{\text{H-P}} = 11.4$ Hz, 6H, OCH_3), 4.13 (m, 4H, OCH_2), 7.24 (d, $^3J_{\text{H-P}} = 1.9$ Hz, 1H, C(4)H), 7.85 (dd, $^3J_{\text{H-P}} = 12.9$ Hz, $^3J_{\text{H-H}} = 8.6$ Hz, 2H, ArH), 7.96 (dd, $^4J_{\text{H-P}} = 4.0$ Hz, $^3J_{\text{H-H}} = 8.6$ Hz, 2H, ArH). ^{13}C NMR (75 MHz, CD_3OD): δ 16.5 (d, $^3J_{\text{C-P}} = 6.2$ Hz, CCH_3), 54.1 (d, $^2J_{\text{C-P}} = 5.9$ Hz, OCH_3), 64.0 (d, $^2J_{\text{C-P}} = 5.9$ Hz, CH_2), 111.0 (d, $^2J_{\text{C-P}} = 22.3$ Hz, C(4)), 127.0 (d, $^3J_{\text{C-P}} = 15.3$ Hz, 2CH), 128.7 (d, $^1J_{\text{C-P}} = 191.5$ Hz, CP), 133.4 (d, $^2J_{\text{C-P}} = 10.3$ Hz, 2CH), 135.5 (d, $^4J_{\text{C-P}} = 3.3$ Hz, C), 138.0 (d, $^1J_{\text{C-P}} = 228.3$ Hz, CP), 147.8 (dd, $^3J_{\text{C-P}} = 13.3$ Hz, $^5J_{\text{C-P}} = 1.0$ Hz, C(5)). IR (KBr): ν 3433, 1480, 1238, 1186, 1130, 1051, 1028, 970 cm^{-1} . HRMS (ESI): m/z [M + H]⁺ calcd for $\text{C}_{15}\text{H}_{24}\text{N}_2\text{O}_6\text{P}_2$: 389.1025. Found 389.1020.

Dimethyl [5-(4-(methoxycarbonyl)phenyl)-1*H*-pyrazol-3-yl]phosphonate (4d)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2d** in MeOH (0.75 ml, 0.375 mmol); the reaction time was 18 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.19) gave pyrazolyl-phosphonate **4d** as a pale yellow solid (61 mg) in 78% yield. Mp 184 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 10.6. ^1H NMR (400 MHz, CDCl_3): δ 3.84 (d, $^3J_{\text{H-P}} = 11.5$ Hz, 6H, POCH_3), 3.92 (s, 3H, COCH_3), 7.06 (d, $^3J_{\text{H-P}} = 2.1$ Hz, 1H, C(4)H), 7.88 (pseudo d, $^3J_{\text{H-H}} = 8.4$ Hz, 2H, ArH), 8.09 (pseudo d, $^3J_{\text{H-H}} = 8.4$ Hz, 2H, ArH), 13.04 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 52.1 (COCH_3), 53.5 (d, $^2J_{\text{C-P}} = 5.5$ Hz, POCH_3), 109.4 (d, $^2J_{\text{C-P}} = 19.8$ Hz, C(4)), 125.6 (2CH), 129.8, 130.2 (2CH),

133.7 (d, $^1J_{\text{C-P}} = 223.4$ Hz, CP), 135.5 (C_{Ar}), 149.5 (br., C(5)), 166.7 (C=O). IR (KBr): ν 3400, 1724, 1457, 1284, 1240, 1188, 1161, 1103, 1057, 1026, 787 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{13}\text{H}_{15}\text{N}_2\text{O}_5\text{PNa}$: 333.0611. Found 333.0602.

Dimethyl [5-(2-chlorophenyl)-1*H*-pyrazol-3-yl]phosphonate (4e)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2e** in MeOH (0.70 ml, 0.35 mmol); the reaction time was 17 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.21) gave pyrazolyl-phosphonate **4e** as a pale yellow solid (45 mg) in 63% yield. Mp 89 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.7. ^1H NMR (400 MHz, CDCl_3): δ 3.82 (d, $^3J_{\text{H-P}} = 11.5$ Hz, 6H, OCH_3), 7.15 (d, $^3J_{\text{H-P}} = 1.8$ Hz, 1H, C(4)H), 7.25–7.31 (m, 2H, ArH), 7.44 (m, 1H, ArH), 7.71 (m, 1H, ArH), 8.94 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.3 (d, $^2J_{\text{C-P}} = 5.5$ Hz, OCH_3), 112.2 (d, $^2J_{\text{C-P}} = 21.5$ Hz, C(4)), 127.2 (CH), 129.2, 129.7 (CH), 130.48 (CH), 130.53 (CH), 131.9 (CCl), 135.4 (d, $^1J_{\text{C-P}} = 226.4$ Hz, CP), 145.6 (d, $^3J_{\text{C-P}} = 14.3$ Hz, C(5)). IR (KBr): ν 3420, 1479, 1244, 1188, 1030, 837, 787, 758 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{11}\text{H}_{12}\text{ClN}_2\text{O}_3\text{PNa}$: 309.0166. Found 309.0166.

Dimethyl [5-(2-bromophenyl)-1*H*-pyrazol-3-yl]phosphonate (4f)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2f** in MeOH (0.65 ml, 0.325 mmol); the reaction time was 16 h. Purification by preparative TLC (EtOAc/hexane: 20/1, R_f 0.21) using EtOAc/hexane as the eluent gave pyrazolylphosphonate **4f** as a white solid (45 mg) in 54% yield. Mp 125 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.3. ^1H NMR (400 MHz, CDCl_3): δ 3.83 (d, $^3J_{\text{H-P}} = 11.5$ Hz, 6H, OCH_3), 7.13 (d, $^3J_{\text{H-P}} = 1.9$ Hz, 1H, C(4)H), 7.20 (ddd, $^3J_{\text{H-H}} = 8.0$ Hz, $^3J_{\text{H-H}} = 7.5$ Hz, $^4J_{\text{H-H}} = 1.6$ Hz, 1H, ArH), 7.35 (ddd, $^3J_{\text{H-H}} = 7.6$ Hz, $^3J_{\text{H-H}} = 7.5$ Hz, $^4J_{\text{H-H}} = 1.2$ Hz, 1H, ArH), 7.63 (dd, $^3J_{\text{H-H}} = 7.6$ Hz, $^4J_{\text{H-H}} = 1.6$ Hz, 1H, ArH), 7.65 (dd, $^3J_{\text{H-H}} = 8.0$ Hz, $^4J_{\text{H-H}} = 1.2$ Hz, 1H, ArH), 12.77 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.4 (d, $^2J_{\text{C-P}} = 5.6$ Hz, OCH_3), 112.5 (d, $^2J_{\text{C-P}} = 21.2$ Hz, C(4)), 121.8 (CBr), 127.6 (CH), 129.9 (CH), 131.1 (CH), 131.7, 133.7 (CH), 134.7 (d, $^1J_{\text{C-P}} = 224$ Hz, CP), 147.3 (br., C(5)). IR (KBr): ν 3471, 1475, 1242, 1180, 1157, 1055, 1030, 987, 958, 860, 845, 835, 766 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{11}\text{H}_{12}\text{BrN}_2\text{O}_3\text{PNa}$: 352.9661. Found 352.9659.

Dimethyl [5-(3-chlorophenyl)-1*H*-pyrazol-3-yl]phosphonate (4g)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2g** in MeOH (0.65 ml, 0.325 mmol); the reaction time was 12 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.29) gave pyrazolyl-phosphonate **4g** as a white solid (56 mg) in 78% yield. Mp 81 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.4. ^1H NMR (400 MHz, CDCl_3): δ 3.81 (d, $^3J_{\text{H-P}} = 11.5$ Hz, 6H, OCH_3), 6.97 (d, $^3J_{\text{H-P}} = 2.0$ Hz, 1H, C(4)H), 7.28 (m, 1H, ArH), 7.31 (m, 1H, ArH), 7.67 (m, 1H, ArH), 7.82 (m, 1H, ArH), 10.26 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.4 (d, $^2J_{\text{C-P}} = 5.5$ Hz, OCH_3), 108.8 (d, $^2J_{\text{C-P}} = 20.2$ Hz, C(4)), 123.9 (CH), 125.9 (CH), 128.4 (CH), 130.1 (CH), 132.8, 134.8 (CCl),



134.1 (d, $^1J_{C-P}$ = 223.4 Hz, CP), 148.6 (d, $^3J_{C-P}$ = 14.4 Hz, C(5)). IR (KBr): ν 3369, 1479, 1246, 1182, 1032, 839, 787, 775 cm^{-1} . Anal. calcd for $\text{C}_{11}\text{H}_{12}\text{N}_2\text{ClO}_3\text{P}$: C, 46.09; H, 4.22; N 9.77. Found: C, 46.07; H, 4.64; N, 9.34.

Dimethyl [5-(4-chlorophenyl)-1*H*-pyrazol-3-yl]phosphonate (4h)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2h** in MeOH (0.65 ml, 0.325 mmol); the reaction time was 4 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.2) gave pyrazolylphosphonate **4h** as a white solid (62 mg) in 87% yield. Mp 143 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.0. ^1H NMR (400 MHz, CDCl_3): δ 3.84 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, OCH_3), 6.98 (d, $^3J_{H-P}$ = 2.1 Hz, 1H, C(4)H), 7.38 (pseudo d, $^3J_{H-H}$ = 8.5 Hz, 2H, ArH), 7.74 (pseudo d, $^3J_{H-H}$ = 8.5 Hz, 2H, ArH), 13.22 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.4 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 108.8 (d, $^2J_{C-P}$ = 20.4 Hz, C(4)), 127.1 (2CH), 129.0 (2CH), 129.6, 134.2 (CCl), 134.1 (d, $^1J_{C-P}$ = 223.3 Hz, CP), 148.8 (d, $^3J_{C-P}$ = 14.4 Hz, C(5)). IR (KBr): ν 3388, 1495, 1246, 1178, 1059, 1018, 833, 771 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{11}\text{H}_{12}\text{ClN}_2\text{O}_3\text{PNa}$: 309.0166. Found 309.0167.

Dimethyl [5-(4-bromophenyl)-1*H*-pyrazol-3-yl]phosphonate (4i)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2i** in MeOH (0.70 ml, 0.35 mmol); the reaction time was 3 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.2) gave pyrazolylphosphonate **4i** as a white solid (72 mg) in 87% yield. Mp 159 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 10.75. ^1H NMR (400 MHz, CDCl_3): δ 3.84 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, OCH_3), 6.98 (d, $^3J_{H-P}$ = 2.1 Hz, 1H, C(4)H), 7.54 (pseudo d, $^3J_{H-H}$ = 8.5 Hz, 2H, ArH), 7.68 (pseudo d, $^3J_{H-H}$ = 8.5 Hz, 2H, ArH), 11.90 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.5 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 108.9 (d, $^2J_{C-P}$ = 19.9 Hz, C(4)), 122.4 (CBr), 127.4 (2CH), 130.2, 132.0 (2CH), 133.6 (d, $^1J_{C-P}$ = 222.6 Hz, CP), 149.4 (d, $^3J_{C-P}$ = 14.8 Hz, C(5)). IR (Nujol): ν 1492, 1385, 1240, 1175, 1080, 1050, 998, 845, 795 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{11}\text{H}_{12}\text{BrN}_2\text{O}_3\text{PNa}$: 352.9661. Found 352.9660.

Dimethyl [5-(4-(methylthio)phenyl)-1*H*-pyrazol-3-yl]phosphonate (4j)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2j** in MeOH (1.5 ml, 0.75 mmol); the reaction time was 3 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.17) gave pyrazolylphosphonate **4j** as a white solid (64 mg) in 86% yield. Mp 138 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.6. ^1H NMR (400 MHz, CDCl_3): δ 2.49 (s, 3H, SCH_3), 3.84 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, OCH_3), 6.98 (d, $^3J_{H-P}$ = 1.9 Hz, 1H, C(4)H), 7.28 (pseudo d, $^3J_{H-H}$ = 8.3 Hz, 2H, ArH), 7.69 (pseudo d, $^3J_{H-H}$ = 8.3 Hz, 2H, ArH), 13.04 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 15.6 (SCH_3), 53.3 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 108.3 (d, $^2J_{C-P}$ = 20.8 Hz, C(4)), 126.1 (2CH), 126.7 (2CH), 127.4, 139.2 (CSMe), 135.0 (d, $^1J_{C-P}$ = 225.6 Hz, CP), 148.8 (d, $^3J_{C-P}$ = 13.2 Hz, C(5)). IR (KBr): ν 3383, 1242, 1493, 1171, 1068, 1038, 985, 841, 779 cm^{-1} . HRMS (ESI):

m/z [M + Na]⁺ calcd for $\text{C}_{12}\text{H}_{15}\text{N}_2\text{O}_3\text{PSNa}$: 321.0433. Found 321.0429.

Dimethyl [5-(3-methoxyphenyl)-1*H*-pyrazol-3-yl]phosphonate (4k)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2k** in MeOH (1.5 ml, 0.75 mmol); the reaction time was 3 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.25) gave pyrazolylphosphonate **4k** as a white solid (50 mg) in 71% yield. Mp 113 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.7. ^1H NMR (400 MHz, CDCl_3): δ 3.83 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, POCH_3), 3.85 (s, 3H, COCH_3), 6.99 (d, $^3J_{H-P}$ = 2.0 Hz, 1H, C(4)H), 6.90 (m, 1H, ArH), 7.30–7.38 (m, 3H, ArH), 12.69 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.3 (d, $^2J_{C-P}$ = 5.5 Hz, POCH_3), 55.3 (OMe), 108.8 (d, $^2J_{C-P}$ = 20.8 Hz, C(4)), 111.1 (CH), 114.4 (CH), 118.3 (CH), 130.0 (CH), 132.0, 135.0 (d, $^1J_{C-P}$ = 222.0 Hz, CP), 149.2 (d, $^3J_{C-P}$ = 14.4 Hz, C(5)), 160.0 (COMe). IR (KBr): ν 3369, 1493, 1248, 1188, 1034, 841, 787 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_{12}\text{H}_{15}\text{N}_2\text{O}_4\text{PNa}$: 305.0662. Found 305.0654.

Dimethyl (5-phenyl-1*H*-pyrazol-3-yl)phosphonate (4l)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2l** in MeOH (0.65 ml, 0.325 mmol); the reaction time was 3 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.24) gave pyrazolylphosphonate **4l** as a white solid (54 mg) in 85% yield. Mp 79 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.85. ^1H NMR (400 MHz, CDCl_3): δ 3.83 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, OCH_3), 7.01 (d, $^3J_{H-P}$ = 2.0 Hz, 1H, C(4)H), 7.33 (t, $^3J_{HH}$ = 7.4 Hz, 1H, ArH), 7.41 (dd, $^3J_{HH}$ = $^3J_{HH}$ = 7.4 Hz, 2H, ArH), 7.78 (d, $^3J_{HH}$ = 7.4 Hz, 2H, ArH), 11.52 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.3 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 108.6 (d, $^2J_{C-P}$ = 21.2 Hz, C(4)), 125.8 (2CH), 128.5 (CH), 128.9 (2CH), 130.5, 135.4 (d, $^1J_{C-P}$ = 225.6 Hz, CP), 148.7 (d, $^3J_{C-P}$ = 13.8 Hz, C(5)). IR (KBr): ν 3406, 1495, 1250, 1180, 1038, 787 cm^{-1} . Anal. calcd for $\text{C}_{11}\text{H}_{13}\text{N}_2\text{O}_3\text{P}$: C, 52.39; H, 5.20; N, 11.11. Found: C, 52.03; H, 5.36; N, 11.20.

This reaction was also repeated using vinylphosphonate **1** (448 mg, 2.5 mmol), 0.5 M solution of the diazo compound **2l** in MeOH (6.5 ml, 3.25 mmol), and K_2CO_3 (35 mg, 0.25 mmol). The product **4l** was isolated by column chromatography in 91% yield (574 mg).

Dimethyl [5-(4-fluorophenyl)-1*H*-pyrazol-3-yl]phosphonate (4m)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2m** in MeOH (0.75 ml, 0.375 mmol); the reaction time was 3 h. Purification by preparative TLC (CH_2Cl_2 /MeOH: 20/1, R_f 0.38) gave pyrazolylphosphonate **4m** as a colorless solid (45 mg) in 67% yield. Mp 117 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.5. ^1H NMR (400 MHz, CDCl_3): δ 3.83 (d, $^3J_{H-P}$ = 11.5 Hz, 6H, OCH_3), 6.96 (d, $^3J_{H-P}$ = 2.0 Hz, 1H, C(4)H), 7.09 (dd, $^3J_{H-H}$ = 8.7 Hz, $^3J_{H-F}$ = 8.7 Hz, 2H, ArH), 7.76 (dd, $^3J_{H-H}$ = 8.7 Hz, $^4J_{H-F}$ = 5.3 Hz, 2H, ArH), 13.16 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.4 (d, $^2J_{C-P}$ = 5.5 Hz, OCH_3), 108.7 (d, $^2J_{C-P}$ = 20.4 Hz, C(4)), 115.8 (d, $^2J_{C-F}$ = 21.8 Hz, 2CH), 127.3 (d, $^4J_{C-F}$ = 2.7 Hz),



127.6 (d, $^3J_{C-F} = 8.2$ Hz, 2CH), 134.1 (d, $^1J_{C-P} = 223.1$ Hz, CP), 149.0 (d, $^3J_{C-P} = 14.2$ Hz, C(5)), 162.8 (d, $^1J_{C-F} = 248.0$ Hz, CF). IR (KBr): ν 3400, 1508, 1246, 1178, 1066, 1034, 837, 785 cm^{-1} . Anal. calcd for $\text{C}_{11}\text{H}_{12}\text{FN}_2\text{O}_3\text{P}$: C, 48.90; H, 4.48; N, 10.37. Found: C, 48.98; H, 4.70; N, 10.07.

Dimethyl [5-(4-isopropylphenyl)-1*H*-pyrazol-3-yl]phosphonate (4n)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2n** in MeOH (2 ml, 1 mmol); the reaction time was 3 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.3) gave pyrazolylphosphonate **4n** as a white solid (59 mg) in 80% yield. Mp 142 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 11.9. ^1H NMR (400 MHz, CDCl_3): δ 1.26 (d, $^3J_{H-H} = 6.9$ Hz, 6H, CCH_3), 2.92 (septet, $^3J_{H-H} = 6.9$ Hz, 1H, $\text{CH}_{i\text{Pr}}$), 3.83 (d, $^3J_{H-P} = 11.4$ Hz, 6H, OCH_3), 6.96 (d, $^3J_{H-P} = 1.9$ Hz, 1H, C(4)H), 7.28 (pseudo d, $^3J_{H-H} = 8.2$ Hz, 2H, ArH), 7.66 (pseudo d, $^3J_{H-H} = 8.2$ Hz, 2H, ArH), 12.97 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 23.8 (CCH_3), 33.9 ($\text{CH}_{i\text{Pr}}$), 53.3 (d, $^2J_{C-P} = 5.6$ Hz, OCH_3), 108.6 (d, $^2J_{C-P} = 21.5$ Hz, C(4)), 125.8 (2CH), 127.0 (2CH), 128.0, 135.7 (d, $^1J_{C-P} = 226.4$ Hz, CP), 148.5 (d, $^3J_{C-P} = 12.9$ Hz, C(5)), 149.4 (CP*t*). IR (KBr): ν 3400, 1502, 1236, 1178, 1043, 787 cm^{-1} . Anal. calcd for $\text{C}_{14}\text{H}_{19}\text{N}_2\text{O}_3\text{P}$: C, 57.14; H, 6.51; N, 9.52. Found: C, 57.35; H, 6.80; N, 9.20.

Dimethyl [5-(4-methoxyphenyl)-1*H*-pyrazol-3-yl]phosphonate (4o)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2o** in MeOH (1 ml, 0.5 mmol); the reaction time was 3 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.24) gave pyrazolylphosphonate **4o** as a white solid (64 mg) in 86% yield. Mp 184 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 12.0. ^1H NMR (400 MHz, CDCl_3): δ 3.80 (s, 3H, COCH_3), 3.82 (d, $^3J_{H-P} = 11.5$ Hz, 6H, POCH_3), 6.91 (d, $^3J_{H-P} = 1.8$ Hz, 1H, C(4)H), 6.92 (pseudo d, $^3J_{H-H} = 8.7$ Hz, 2H, ArH), 7.68 (pseudo d, $^3J_{H-H} = 8.7$ Hz, 2H, ArH), 10.72 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 53.3 (d, $^2J_{C-P} = 5.5$ Hz, POCH_3), 55.3 (COCH_3), 108.0 (d, $^2J_{C-P} = 21.4$ Hz, C(4)), 114.3 (2CH), 123.1, 127.1 (2CH), 135.5 (d, $^1J_{C-P} = 223.8$ Hz, CP), 148.4 (d, $^3J_{C-P} = 11.5$ Hz, C(5)), 159.9 (COMe). IR (KBr): ν 3452, 1618, 1510, 1252, 1230, 1178, 1049, 1028, 841, 827, 785 cm^{-1} . Anal. calcd for $\text{C}_{12}\text{H}_{15}\text{N}_2\text{O}_4\text{P}$: C, 51.07; H, 5.36; N, 9.93. Found: C, 50.81; H, 5.33; N, 9.79.

Dimethyl [5-(4-ethoxyphenyl)-1*H*-pyrazol-3-yl]phosphonate (4p)

The reaction was carried out as described for **4a** using 0.5 M solution of the diazo compound **2p** in MeOH (0.8 ml, 0.4 mmol); the reaction time was 2 h. Purification by preparative TLC ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.26) gave pyrazolylphosphonate **4p** as a yellowish solid (47 mg) in 63% yield. Mp 165 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 12.5. ^1H NMR (400 MHz, CDCl_3): 1.40 (t, $^3J_{H-H} = 7.0$ Hz, 3H, CCH_3), 3.80 (d, $^3J_{H-P} = 11.4$ Hz, 6H, OCH_3), 4.02 (q, $^3J_{H-H} = 7.0$ Hz, 2H, CH_2), 6.88 (d, $^3J_{H-P} = 1.8$ Hz, 1H, C(4)H), 6.90 (pseudo d, $^3J_{H-H} = 8.8$ Hz, 2H, ArH), 7.66 (pseudo d, $^3J_{H-H} = 8.8$ Hz, 2H, ArH), 12.98 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 14.8 (CCH_3), 53.3 (d, $^2J_{C-P} = 5.5$ Hz, OCH_3), 63.5 (CH_2), 107.8 (d, $^2J_{C-P} = 21.6$ Hz, C(4)), 114.8 (2CH), 122.8, 127.1 (2CH), 135.9

(d, $^1J_{C-P} = 229.1$ Hz, CP), 148.2 (br. s, C(5)), 159.3 (COEt). IR (KBr): ν 3369, 1508, 1250, 1178, 1030, 985, 841, 781 cm^{-1} . Anal. calcd for $\text{C}_{13}\text{H}_{17}\text{N}_2\text{O}_4\text{P}$: C, 52.70; H, 5.78; N, 9.46. Found: C, 52.17; H, 5.64; N, 8.94.

Ethyl 5-(dimethoxyphosphoryl)-5-(formamido)-4,5-dihydro-1*H*-pyrazole-3-carboxylate (5)

A foil-covered 4 ml vial was charged with vinylphosphonate **1** (50 mg, 0.28 mmol) and ethyl diazoacetate (60 μl , 0.56 mmol). The reaction mixture was stirred at 80 $^{\circ}\text{C}$ for 4 h. Purification by column chromatography ($\text{CH}_2\text{Cl}_2/\text{MeOH}$: 20/1, R_f 0.32) afforded 2-pyrazoline **5** as a colorless oil (66 mg) in 80% yield; a 84:16 mixture of two rotamers in CDCl_3 . ^{31}P NMR (162 MHz, CDCl_3): δ 19.4 (major), 18.5 (minor). ^1H NMR (400 MHz, CDCl_3) (only for the major rotamer): δ 1.29 (t, $^3J_{H-H} = 7.1$ Hz, 3H, CCH_3), 3.31 (dd, $^3J_{H-P} = 18.2$ Hz, $^2J_{H-H} = 18.2$ Hz, 1H, C(4)H), 3.43 (dd, $^3J_{H-P} = 7.2$ Hz, $^2J_{H-H} = 18.2$ Hz, 1H, C(4)H), 3.80 (d, $^3J_{H-P} = 10.6$ Hz, 3H, OCH_3), 3.83 (d, $^3J_{H-P} = 10.4$ Hz, 3H, OCH_3), 4.25 (q, $^3J_{H-H} = 7.1$ Hz, 2H, OCH_2) 7.77 (br. s, 1H, NHC=O), 7.93 (br. s, 1H, NNH), 8.09 (s, 1H, CHO). ^{13}C NMR (101 MHz, CDCl_3): δ 14.1 (CCH_3 , major), 14.0 (CCH_3 , minor), 41.1 ($^2J_{C-P} = 3.2$ Hz, C(4), major), 41.8 ($^2J_{C-P} = 2.9$ Hz, C(4), minor), 54.0 ($^2J_{C-P} = 7.3$ Hz, OCH_3 , minor), 54.2 ($^2J_{C-P} = 7.3$ Hz, OCH_3 , major), 55.1 ($^2J_{C-P} = 6.6$ Hz, OCH_3 , major), 55.3 ($^2J_{C-P} = 6.3$ Hz, OCH_3 , minor), 61.5 (OCH_2 , major), 61.6 (OCH_2 , minor), 75.1 ($^1J_{C-P} = 198.1$ Hz, CP, minor), 76.2 ($^1J_{C-P} = 199.8$ Hz, CP, major), 141.5 ($^3J_{C-P} = 12.2$ Hz, C(3), minor), 142.3 ($^3J_{C-P} = 11.4$ Hz, C(3), major), 161.7 (CHO, minor), 161.8 ($^3J_{C-P} = 3.1$ Hz, CHO, major), 163.4 (C(O)O, minor), 163.6 (C(O), major). IR (KBr): ν 3192, 1697, 1666, 1460, 1379, 1263, 1213, 1178, 1037, 1022 cm^{-1} . HRMS (ESI): m/z [M + Na]⁺ calcd for $\text{C}_9\text{H}_{16}\text{N}_3\text{O}_6\text{PNa}$: 316.0669. Found 316.0672.

Ethyl 3-(dimethoxyphosphoryl)-1*H*-pyrazole-5-carboxylate (6)

To a solution of 2-pyrazoline **5** (52 mg, 0.18 mmol) in anhydrous methanol (2 mL) two drops of Me_3SiCl were added. After stirring at rt for 4 h under an argon atmosphere, the mixture was evaporated under vacuum 2 mmHg. The residue was suspended in CH_2Cl_2 and filtered. The filtrate was evaporated under vacuum to provide the pure pyrazole **6** as a colorless solid (39 mg) in 89% yield. Mp 109 $^{\circ}\text{C}$. ^{31}P NMR (162 MHz, CDCl_3): δ 9.6. ^1H NMR (400 MHz, CDCl_3): δ 1.36 (t, $^3J_{H-H} = 7.1$ Hz, 3H, CCH_3), 3.79 (d, $^3J_{H-P} = 11.5$ Hz, 6H, 2OCH_3), 4.38 (q, $^3J_{H-H} = 7.1$ Hz, 2H, CH_2), 7.21 (d, $^3J_{H-P} = 11.5$ Hz, 1H, C(4)H). ^{13}C NMR (101 MHz, CDCl_3): δ 14.17 (CCH_3), 53.5 (d, $^2J_{C-P} = 5.6$ Hz, OCH_3), 61.4 (CH_2), 114.6 (d, $^2J_{C-P} = 19.9$ Hz, C(4)), 134.6 ($^1J_{PC} = 224.2$ Hz, CP), 141.5 (d, $^3J_{PC} = 13.6$ Hz, C(5)), 160.8 (CO). IR (KBr): ν 3442, 1734, 1730, 1462, 1257, 1242, 1176, 1047 cm^{-1} . Anal. calcd for $\text{C}_8\text{H}_{13}\text{N}_2\text{O}_5\text{P}$: C, 38.72; H, 5.28; N, 11.29. Found: C, 38.87; H, 5.40; N, 10.99.

Ethyl 5-(diethoxyphosphoryl)-5-phenyl-4,5-dihydro-1*H*-pyrazole-3-carboxylate (8)

To a solution of vinylphosphonate **7a** (96 mg, 0.4 mmol) in THF (2 ml) ethyl diazoacetate (63 μl , 0.6 mmol) and K_2CO_3 (3 mg, 0.02 mmol) were added. After stirring at 50 $^{\circ}\text{C}$ for 48 h,



the mixture was evaporated under vacuum. The viscous residue was solidified by vigorous stirring with hexane. The resultant precipitate was collected by filtration, washed with Et_2O , hexane and dried under vacuum to give 2-pyrazoline **8** as a yellow solid (61 mg) in 43% yield. ^3P NMR (162 MHz, CDCl_3): δ 21.5. ^1H NMR (400 MHz, CDCl_3): δ 1.17 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 1.23 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 1.32 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 3.33 (dd, $^3J_{\text{H}-\text{P}} = 24.2$ Hz, $^2J_{\text{H}-\text{H}} = 17.6$ Hz, 1H, C(4)H), 3.81 (dd, $^3J_{\text{H}-\text{P}} = 26.2$ Hz, $^2J_{\text{H}-\text{H}} = 17.6$ Hz, 1H, C(4)H), 3.86–4.09 M (m, 4H, POCH_2), 4.28 (q, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 2H, COCH_2), 6.98 (br. d, $^3J_{\text{H}-\text{P}} = 4.5$ Hz, 1H, NH), 7.30 (m, 1H, ArH), 7.37 (m, 2H, ArH), 7.45 (m, 2H, ArH). ^{13}C NMR (101 MHz, CDCl_3): δ 14.2 (COCH_2CH_3), 16.27 (POCH_2CH_3), 16.31 (POCH_2CH_3), 41.7 (C(4)), 61.3 (COCH_2), 63.4 (d, $^2J_{\text{C}-\text{P}} = 7.4$ Hz, POCH_2), 64.0 (d, $^2J_{\text{C}-\text{P}} = 6.8$ Hz, POCH_2), 70.5 (d, $^1J_{\text{C}-\text{P}} = 152.6$ Hz, CP), 126.6 (d, $^3J_{\text{C}-\text{P}} = 4.0$ Hz, 2CH), 128.1 (CH), 128.6 (2CH), 138.7, 142.1 (C(3)), 162.0 (C=O). Anal. calcd for $\text{C}_{16}\text{H}_{23}\text{N}_2\text{O}_5\text{P}$: C, 54.23; H, 6.54; N, 7.91. Found: C, 53.99; H, 6.31; N, 7.75.

Representative synthesis: ethyl 3-phenyl-1*H*-pyrazole-5-carboxylate (9a)

Method A. To a solution of vinylphosphonate **7a** (96 mg, 0.4 mmol) and ethyl diazoacetate (84 μl , 0.8 mmol) in THF (2 ml) sodium hydride (60% dispersion in mineral oil, Aldrich; 40 mg, 1 mmol) was added in an argon atmosphere. The reaction mixture was stirred at 50 °C for 18 h, cooled to rt, and neutralized by the slow dropwise addition of the satd. ethereal solution of HCl (300 μl), followed by the addition of NET_3 (130 μl). Volatile components were removed on a rotary evaporator and the residue was purified by column chromatography (EtOAc/hexane: 1/2, R_f 0.8) afforded pyrazolecarboxylate **9a** as a white solid (71 mg) in 82% yield.

Method B. A mixture of vinylphosphonate **7a** (96 mg, 0.4 mmol), ethyl diazoacetate (84 μl , 0.8 mmol), K_2CO_3 (3 mg, 0.8 mmol), and THF (2 ml) was stirred at 50 °C for 48 h and cooled to rt in an argon atmosphere. After the addition of sodium hydride (60% dispersion in mineral oil, Aldrich; 32 mg, 0.8 mmol), the reaction mixture was further stirred at 50 °C for 5 h, cooled, and worked up as described in Method A. Column chromatography afforded pyrazolecarboxylate **9a** (57 mg) in 66% yield.

Mp 185 °C. ^1H NMR (400 MHz, CDCl_3): δ 1.26 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 4.25 (q, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 2H, CH_2), 7.02 (s, 1H, C(4)H), 7.32 (t, $^3J_{\text{H}-\text{H}} = 7.4$ Hz, 1H, ArH), 7.38 (dd, $^3J_{\text{H}-\text{H}} = 3J_{\text{H}-\text{H}} = 7.4$ Hz, 2H, ArH), 7.72 (d, $^3J_{\text{H}-\text{H}} = 7.4$ Hz, 2H, ArH), 11.40 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 14.0 (CH_3), 61.0 (CH_2), 105.1 (C(4)), 125.6 (2CH), 128.5 (CH), 128.8 (2CH), 130.3, 140.1 (br., C(5)), 148.1 (br., C(3)), 160.9 (C=O). IR (KBr): ν 3435, 1726, 1493, 1241, 1194, 1138, 762 cm^{-1} . Anal. calcd for $\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2$: C, 66.65; H, 5.59; N, 12.96. Found: C, 66.56; H, 5.78; N, 12.85.

Ethyl 3-(4-chlorophenyl)-1*H*-pyrazole-5-carboxylate (9b)

Method A. The reaction was carried out as described for **9a** using vinylphosphonate **7b** (110 mg, 0.4 mmol). Purification

by column chromatography (EtOAc/hexane: 1/2, R_f 0.8) gave pyrazolecarboxylate **9b** as a yellow solid (57 mg) in 57% yield.

Method B. The reaction was carried out as described for **9a** using vinylphosphonate **7b** (110 mg, 0.4 mmol). Column chromatography afforded pyrazolecarboxylate **9b** (62 mg) in 62% yield.

Mp 141 °C. ^1H NMR (400 MHz, CDCl_3): δ 1.32 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 4.35 (q, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 2H, CH_2), 7.04 (s, 1H, C(4)H), 7.37 (pseudo d, $^3J_{\text{H}-\text{H}} = 8.5$ Hz, 2H, ArH), 7.69 (pseudo d, $^3J_{\text{H}-\text{H}} = 8.5$ Hz, 2H, ArH), 10.24 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 14.2 (CH_3), 61.4 (CH_2), 105.4 (C(4)), 126.9 (2CH), 129.0 (2CH), 129.5, 134.3 (CCl), 138.4 (br., C(5)), 148.8 (br., C(3)), 160.3 (C=O). IR (KBr): ν 3292, 1697, 1473, 1444, 1304, 1277, 1192, 1093, 1028, 957, 833, 773 cm^{-1} . Anal. calcd for $\text{C}_{12}\text{H}_{11}\text{ClN}_2\text{O}_2$: C, 57.49; H, 4.42; N, 11.17. Found: C, 57.62; H, 4.57; N, 11.01.

Ethyl 3-(4-isobutylphenyl)-1*H*-pyrazole-5-carboxylate (9c)

Method A. The reaction was carried out as described for **9a** using vinylphosphonate **7c** (119 mg, 0.4 mmol). Purification by column chromatography (EtOAc/hexane: 1/2, R_f 0.5) gave pyrazolecarboxylate **9c** as a yellowish solid (65 mg) in 60% yield.

Method B. The reaction was carried out as described for **9a** using vinylphosphonate **7c** (110 mg, 0.4 mmol); the reaction time at the first stage was 53 h. Column chromatography afforded pyrazolecarboxylate **9c** (61 mg) in 56% yield.

Mp 168 °C. ^1H NMR (400 MHz, CDCl_3): δ 0.90 (d, $^3J_{\text{H}-\text{H}} = 6.6$ Hz, 6H, CH_3 iBu), 1.23 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, OCH_2CH_3), 1.86 (m, 1H, CH_3 iBu), 2.47 (d, $^3J_{\text{H}-\text{H}} = 7.2$ Hz, 2H, CH_2 iBu), 4.21 (q, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 2H, OCH_2), 6.98 (s, 1H, C(4)H), 7.15 (pseudo d, $^3J_{\text{H}-\text{H}} = 8.1$ Hz, 2H, ArH), 7.61 (pseudo d, $^3J_{\text{H}-\text{H}} = 8.1$ Hz, 2H, ArH), 11.81 (br. s, 1H, NH). ^{13}C NMR (101 MHz, CDCl_3): δ 14.1 (OCH_2CH_3), 22.3 (CH_3 iBu), 30.2 (CH iBu), 45.2 (CH_2 iBu), 61.0 (OCH_2), 105.0 (C(4)), 125.4 (2CH), 127.8, 129.6 (2CH), 140.2 (br., C(5)), 142.3 (CBuⁱ), 148.4 (br., C(3)), 161.0 (C=O). IR (KBr): ν 3159, 1730, 1464, 1379, 1242, 1130, 985 cm^{-1} . Anal. calcd for $\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}_2$: C, 70.56; H, 7.40; N, 10.29. $\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}_2$. Found: C, 70.34; H, 7.34; N, 10.19.

Ethyl 3-(biphenyl-4-yl)-1*H*-pyrazole-5-carboxylate (9d)

Method A. The reaction was carried out as described for **9a** using vinylphosphonate **7d** (127 mg, 0.4 mmol). Purification by column chromatography (EtOAc/hexane: 1/2, R_f 0.7) gave pyrazolecarboxylate **9d** as a yellowish solid (89 mg) in 76% yield.

Method B. The reaction was carried out as described for **9a** using vinylphosphonate **7d** (127 mg, 0.4 mmol); the reaction time at the first stage was 55 h. Column chromatography afforded pyrazolecarboxylate **9d** (93 mg) in 83% yield.

Mp 192 °C. ^1H NMR (400 MHz, CDCl_3): δ 1.42 (t, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 3H, CH_3), 3.51 (br. s, 1H, NH), 4.42 (q, $^3J_{\text{H}-\text{H}} = 7.1$ Hz, 2H, CH_2), 7.15 (s, 1H, C(4)H), 7.36 (t, $^3J_{\text{H}-\text{H}} = 7.3$ Hz, 1H, ArH), 7.45 (dd, $^3J_{\text{H}-\text{H}} = 7.3$ Hz, $^3J_{\text{H}-\text{H}} = 7.9$ Hz, 2H, ArH), 7.63 (d, $^3J_{\text{H}-\text{H}} = 7.9$ Hz, 2H, ArH), 7.67 (d, $^3J_{\text{H}-\text{H}} = 8.3$ Hz, 2H, ArH), 7.84 (d, $^3J_{\text{H}-\text{H}} = 8.3$ Hz, 2H, ArH). ^{13}C NMR (101 MHz, DMSO-d_6):



δ 14.2 (CH₃), 60.5 (CH₂), 105.3 (C(4)), 125.9 (2CH), 126.6 (2CH), 127.1 (2CH), 127.7 (CH), 129.0 (2CH), 129.3, 139.4, 139.9, 141.6 (br., C(5)), 146.6 (br., C(3)), 160.8 (C=O). IR (KBr): ν 3288, 1697, 1472, 1442, 1267, 1186, 1024, 835, 762 cm⁻¹. Anal. calcd for C₁₆H₂₀N₂O₂: C, 73.95; H, 5.52; N, 9.58. C₁₈H₁₆N₂O₂. Found: C, 74.14; H, 5.86; N, 9.41.

Ethyl 3-(2-naphthyl)-1*H*-pyrazole-5-carboxylate (9e)

Method A. The reaction was carried out as described for 9a using vinylphosphonate 7e (116 mg, 0.4 mmol). Purification by column chromatography (EtOAc/hexane: 1/2, *R*_f 0.75) gave pyrazolecarboxylate 9e as a yellowish solid (79 mg) in 74% yield.

Method B. The reaction was carried out as described for 9a using vinylphosphonate 7e (116 mg, 0.4 mmol); the reaction time at the first stage was 53 h. Column chromatography afforded pyrazolecarboxylate 9e (73 mg) in 68% yield.

Mp 187 °C. ¹H NMR (400 MHz, CDCl₃): δ 1.39 (t, ³J_{H-H} = 7.1 Hz, 3H, CH₃), 4.38 (q, ³J_{H-H} = 7.1 Hz, 2H, CH₂), 7.23 (s, 1H, C(4)H), 7.46–7.51 (m, 2H, ArH), 7.82–7.92 (m, 4H, ArH), 8.23 (s, 1H), 11.70 (br. s, 1H, NH). ¹³C NMR (101 MHz, CDCl₃): δ 14.2 (CH₃), 61.4 (CH₂), 105.8 (C(4)), 123.6 (CH), 124.6 (CH), 126.4 (CH), 126.5 (CH), 127.7 (CH), 128.22 (CH), 128.27, 128.7 (CH), 133.3, 133.4, 138.9 (br., C(5)), 149.8 (br., C(3)), 160.6 (C=O). IR (KBr): ν 3261, 1711, 1469, 1257, 1184, 1144, 1024, 820, 779 cm⁻¹. Anal. calcd for C₁₆H₁₄N₂O₂: C, 72.16; H, 5.30; N, 10.52. C₁₈H₁₆N₂O₂. Found: C, 72.29; H, 5.41; N, 10.29.

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major, δ_p 15.0 ppm

minor, δ_p 13.6 ppm

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