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Recent advances in the synthesis of indolizines and their π -expanded analogues

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Indolizine (pyrrolo[1,2-*a*]pyridine) is one of the five isomers of indole and it serves as a precursor for widespread indolizidine alkaloids. The straightforward synthesis of indolizines based on classical methodologies such as Scholtz or Chichibabin reactions has overshadowed numerous new strategies that have been revealed especially within the last ten years. The desire to achieve substitution patterns which were hard to build sparked the discovery of completely new pathways, e.g. transition metal-catalyzed reactions and approaches based on oxidative coupling. In this review, selected strategies toward indolizines published since 2005 are briefly summarized, commented upon, compared, and illustrated. The literature discussed here involves reactions based on either pyridine or pyrrole scaffolds, as well as selected methodologies leading to π -expanded indolizines.

1. Introduction

Nitrogen-containing heterocycles play a key role in both medicinal and materials chemistry. Numerous drugs, dyes, and high-performance materials are inherently heterocyclic.¹ The methods for their synthesis are continuously improving. One of the most important *N*-fused heterocyclic skeletons is indolizine (1) (Fig. 1), which is composed of two condensed rings: a

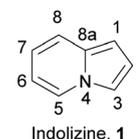


Fig. 1 Structure and numbering system of indolizine skeleton.

pyrrole-type five-membered ring and a pyridine-type six-membered ring.

The partially or more often wholly hydrogenated derivatives of indolizine can be frequently found as backbones in many bioactive natural products such as swainsonine **2**^{2a} or

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cryptowoline 3 (Fig. 2).^{2b} Moreover, their synthetic derivatives have been found to exhibit a variety of biological activities, including as a potential inhibitor of vascular endothelial growth factor (VEGF) and for neuropilin-1 (NRP1) synergy (4),^{3a} as agents with antibacterial, antiviral, anti-inflammatory properties,^{3b-d} as well as phosphodiesterase V inhibitors (5),^{3e} histamine H3 receptor antagonists,^{3f} or as selective chemical probes for the bromodomains BAZ2A and BAZ2B.^{3g} Indolizine

derivatives have also been investigated for their suitability as dyes for dye sensitized solar cells (DSSC) (6)^{4a} or organic light emitting devices (OLEDs),^{4b,c} owing to their appreciable photo-physical properties.

The first report on indolizine comes from the 19th century.⁵ Since that time, numerous studies on their synthesis, properties, and applications have been published, which have been partially collected in earlier reviews.⁶ In this contribution, we have highlighted the most important methodologies leading to the formation of the indolizine core developed over the past ten years. The number of valuable contributions has compelled us to be selective. Our choices featured in this review were based on a combination of the novelty and utility of these novel approaches. Synthetic strategies leading to indolizines possessing additional exocyclic conjugated C–C double or triple bonds, as well as functionalization of the indolizine core⁷ have been omitted here for the sake of brevity.

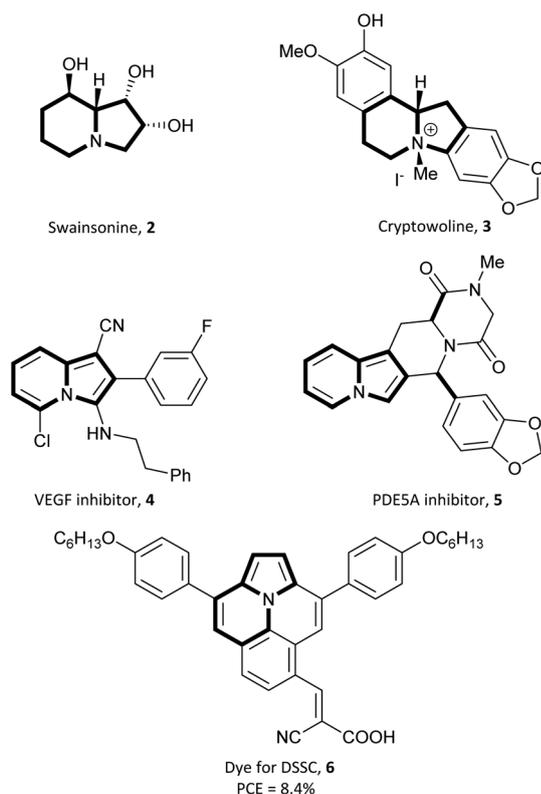


Fig. 2 Examples of useful indolizine derivatives.



Daniel T. Gryko

Daniel T. Gryko obtained his PhD from the Institute of Organic Chemistry of the Polish Academy of Sciences in 1997 under the supervision of J. Jurczak. After post-doctoral studies with J. Lindsey at the North Carolina State University (1998–2000), he began his independent career in Poland. He became Full Professor in 2008. His current research interests are focused on the synthesis of corroles, π -expanded porphyrinoids

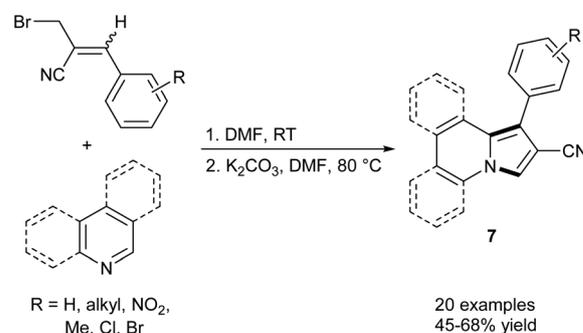
and other functional dyes as well as on two-photon absorption, artificial photosynthesis, excited-state intramolecular proton transfer and fluorescence imaging.

2. Methods based on pyridine derivatives

Starting the indolizine synthesis from pyridine derivatives has the fundamental advantage that pyridine derivatives are easily available *via* numerous methods and >2000 of them are commercially available. Due to their electron-deficient nature, they are typically stable under ambient conditions. Although the pyridine ring is stable due to its aromaticity, it readily undergoes various transformations of its core as long as the final product is also aromatic.

2.1 From pyridines with a free C2 position

Basavaiah *et al.* synthesized a series of indolizines and their benzofused analogues (7) starting from pyridine, quinoline, or isoquinoline and Baylis–Hillman bromides (*i.e.*, allyl bromides which can be prepared directly from products of the Baylis–Hillman reaction) *via* a one pot sequence, which includes salt formation and subsequent 1,5-cyclization of the nitrogen ylides (Scheme 1).⁸ The described reaction is a convenient way to access 1-aryl-2-cyanoindolizines and their π -expanded derivatives under mild conditions despite moderate yields.

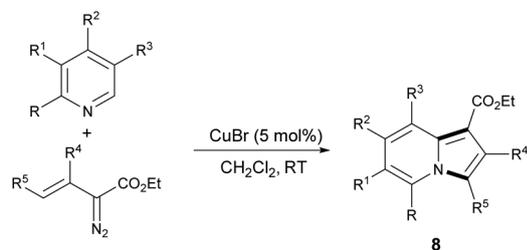


Scheme 1



Attempting to use eco-friendly copper as a less expensive alternative to rhodium-based catalysts, Barluenga, Tomás and co-workers developed a direct annulation of pyridine derivatives with diazoacetates possessing carbon-carbon double bonds, forming indolizine derivatives **8** (Scheme 2).⁹ The broad scope of substrates and mild reaction conditions caused this method to be very useful in the synthesis of highly functionalized indolizines. Furthermore, fused pyridines such as quinoline, isoquinoline, and phenanthridine also reacted easily with ethyl isopropenyldiazoacetate under these conditions to give π -expanded analogues of indolizines in 65, 80, and 30% yield, respectively. Additionally, the selectivity of this method, depending on the type of substituent located at position 3 of pyridine, was investigated. It appeared, that in the case of 3-methylpyridine, the C₃=C₄ bond is preferred for the coordination with the metal centre because of the electron donating influence of the methyl group, while for 3-nitropyridine the C₄=C₅ bond is much more preferable for the coordination owing to the strong electron-withdrawing effect of the NO₂ group, thus forming more stable complex intermediates **I** (Scheme 2, bottom).

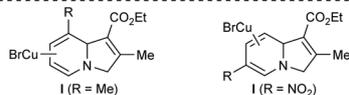
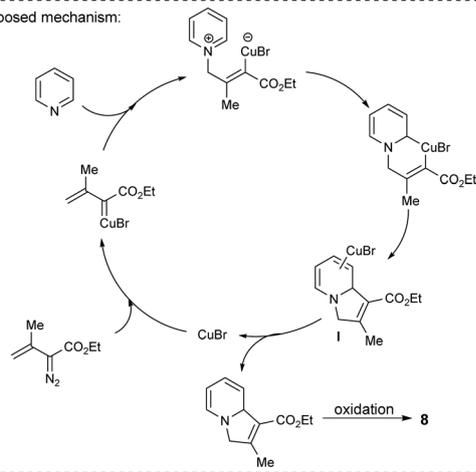
Recently, Wang's group proposed an elegant route leading to highly-substituted 1-cyanoindolizines (**9**) and



R = H, Me; R¹ = H, Me, Cl
 R² = H, Me, Ph, CH=CH₂, Cl,
 OTs, COMe, CO₂Me, CHO, CN
 R³ = H, Me, Cl; R⁴ = H, alkyl
 R⁵ = H, alkyl

8
 19 examples
 25–94% yield

Proposed mechanism:



Scheme 2

3-cyanoindolizines (**10**) starting from esters of 1-cyanocyclopropane-1-carboxylic acid (Scheme 3).¹⁰ This iodine-catalyzed reaction also proceeded well with quinoline and isoquinoline to give π -expanded indolizines **11** and **12**, respectively, in good yields. The application of pyridine as a solvent led to heterocycle **10**, suggesting that different reaction mechanisms apply in the case of either toluene or pyridine. During the course of the reaction, 1-cyanocyclopropane-1-carboxylic acid is converted under the influence of a base and iodine into a carboanion, which can exist in two mesomeric forms – **A** and **A'**. The stability of these two species strongly depends on both solvent polarity and steric hindrances presented therein. The α -carboanion **A** with two strongly electron-withdrawing groups (cyano and ester) is stable in a weak polar solvent such as toluene and due to large steric hindrances is solvated with difficulty by pyridine. Consequently, the 1,3-dipolar cycloaddition reaction of **A** and the pyridine derivative in toluene gave **9**. In contrast, the benzyl anion **A'**, which is a stable resonance in pyridine,^{10b} reacted easily in pyridine to give **10**. The position of CN, aryl and aroyl substituents at C1, C2 and C3 in compounds **9** and **10** was confirmed by X-ray analyses. Surprisingly, esters of 1-cyanocyclopropane-1-carboxylic acid showed no reactivity toward Lewis acids such as aluminum(III) chloride, zinc(II) chloride, iron(III) chloride or boron trifluoride.

2.2 From pyridines bearing formyl or keto functionalities at position 2

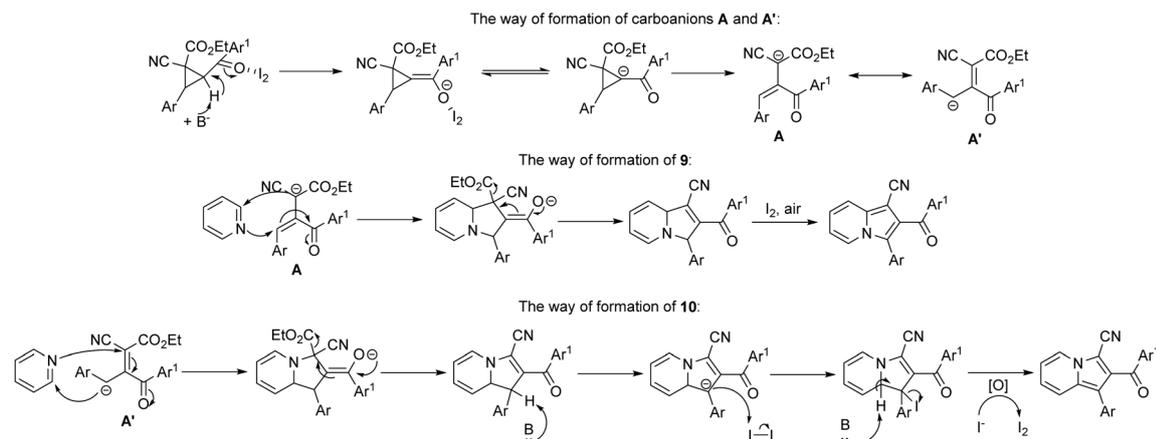
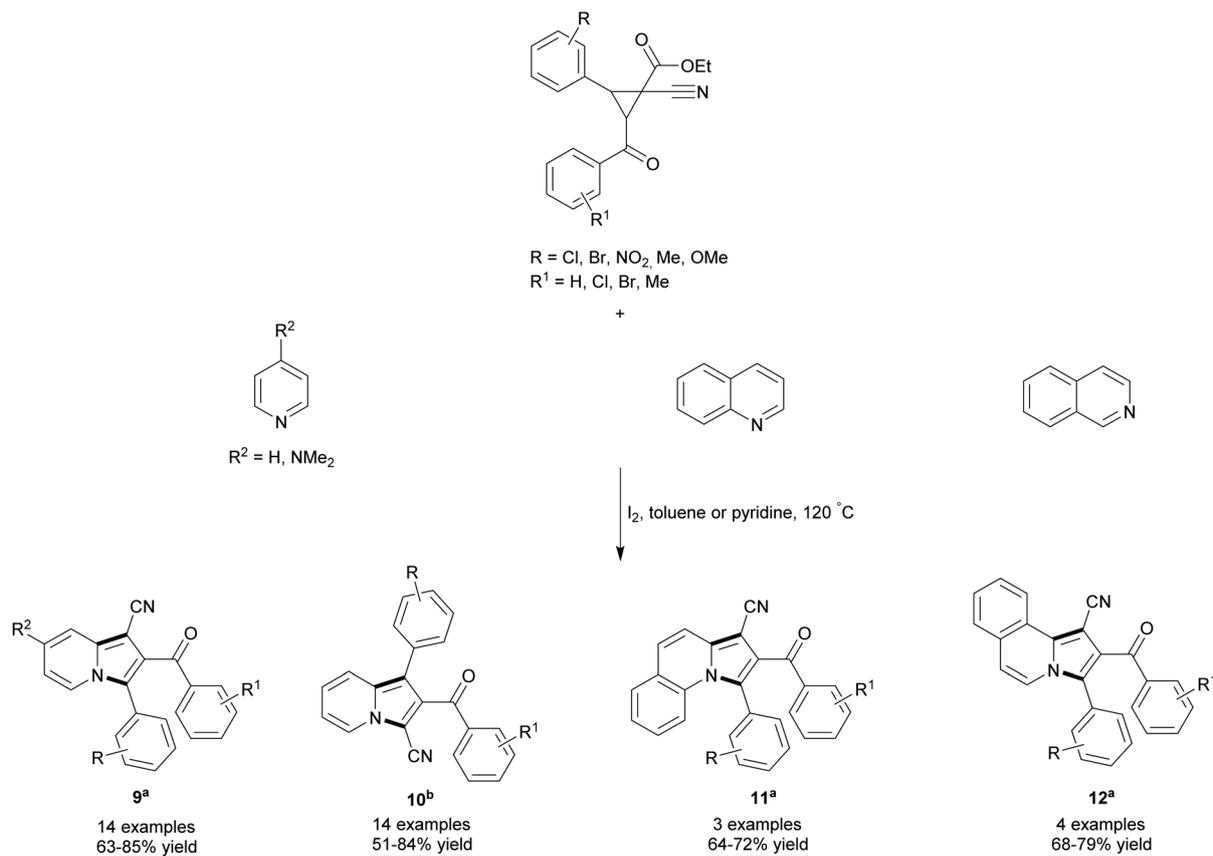
Although the sequence of the reaction leading to the desired product starting from unsubstituted pyridine has the advantage of simplicity, only very few such processes are available as far as indolizine synthesis is concerned. The overwhelming majority of strategies require mono-substituted pyridines as starting materials.

Rao and Basavaiah reported the first example of an electrophile-induced Baylis–Hillman reaction between pyridine-2-carboxaldehyde and activated alkenes such as alkyl vinyl ketones or cyclic enones under the influence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) (Scheme 4).¹¹ The authors assumed that this reaction would proceed according to the most generally accepted Baylis–Hillman reaction mechanism (Scheme 4, bottom).¹²

Later, the same group extended application of this method employing 2-alkanoyl- or 2-aroilpyridines and vinyl ketones in refluxing acetonitrile to produce **14a–b** in moderate yields (Scheme 5).¹³ Unfortunately, the subsequent Michael addition of **14a** and the corresponding vinyl ketone resulted in a decrease in the overall yield. The reaction of isoquinolin-1-yl phenyl ketone and cyclohex-2-enone gave the corresponding π -expanded indolizine in 45% yield.

Liu and Yan were the first to report the Au-catalyzed reaction between aldehydes, secondary amines, and terminal alkynes leading to 1-aminoindolizines in a simple and high-atom economical route (Scheme 6).¹⁴ This reaction proceeds either under solvent-free conditions or in water. Importantly, use of aromatic amines in this transformation resulted in relatively low yields of the corresponding products. A series of





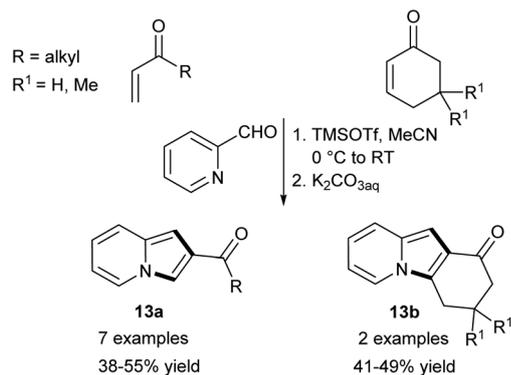
Scheme 3 ^aToluene was used as a solvent. ^bPyridine was used as a solvent.

N-benzyl amino acid derivatives was used as the amine partner according to this protocol, giving **15** in good yields without loss of enantiomeric purity. It is noteworthy that while unsubstituted indolizine is not stable under ambient conditions, its derivatives **15** bearing an amino group attached to a five-membered ring are very unstable both in solution and in the solid state and they must be handled with care. This multicomponent reaction was a breakthrough that prompted other research groups to seek less expensive catalysts. Eventually it was found that the same type of reaction can be realized under the influence of AgBF_4 ,¹⁵ $\text{Fe}(\text{acac})_3/\text{TBAOH}$,¹⁶ or copper nano-

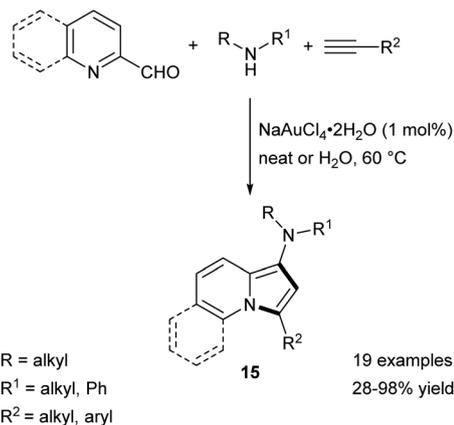
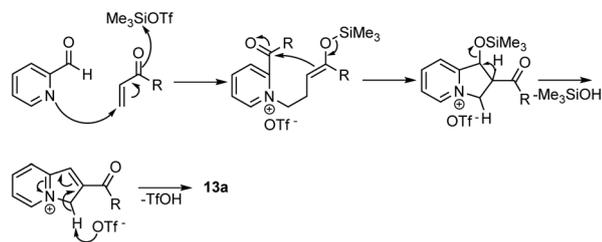
particles supported on activated carbon (CNPs/C)¹⁷ systems. Recently, fully recyclable, green, and highly efficient conditions (CuI/CSP in ethylene glycol) have been proposed for this process with high atom economy (94%), and a small environmental (*E*) factor (0.06).¹⁸

The series of compounds **16** containing the indolizino[8,7-*b*]indole's skeleton, occurring in many alkaloids, was obtained similarly by Batra's group *via* a coupling/cycloisomerization sequence of *N*-substituted 1-formyl-9*H*- β -carbolines, secondary amines, and terminal alkynes (Scheme 7).¹⁹ The method was equally efficient for both aromatic and aliphatic alkynes,

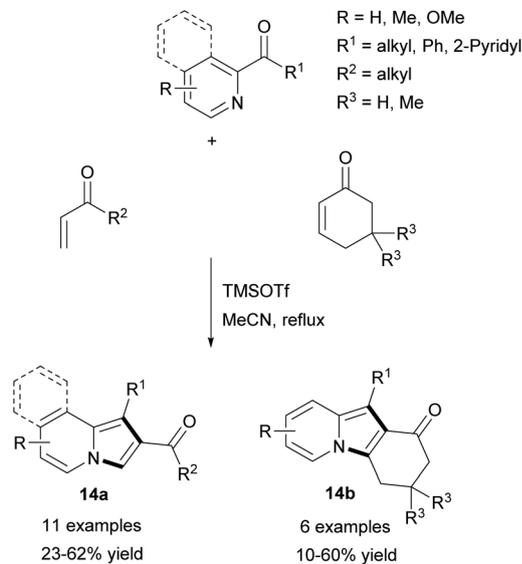
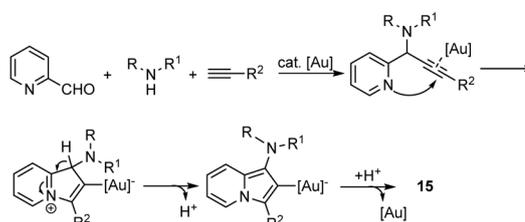




Proposed mechanism:



Proposed mechanism:

**Scheme 5**

however the authors obtained no product in the case of trimethylsilylacetylene. Moreover, while aliphatic amines gave the corresponding products in good yields, diphenylamine showed no reactivity in this reaction.

2.3 From 2-halo- or 2-triflylpyridines and alkynes

The ability to introduce an amino group at any aromatic hydrocarbon or heterocycle opens a plethora of possibilities for further functionalization *via* electrophilic aromatic substitution. An efficient one-pot route leading to 3-aminoindolizines and its expanded analogues **17** has been developed *via* a Pd/Cu-catalyzed sequential Sonogashira cross-coupling/cycloisomerization process (Scheme 8).²⁰ Interestingly, a few other derivatives (**18** and **19a-c**) containing two indolizine moieties within a molecule were also synthesized according to the above protocol in satisfactory yields, demonstrating its high synthetic potential. The use of flow technology in this process allowed linear scale-up from milligram- to gram-scale in the synthesis of a broad range of 3-aminoindolizines.²¹

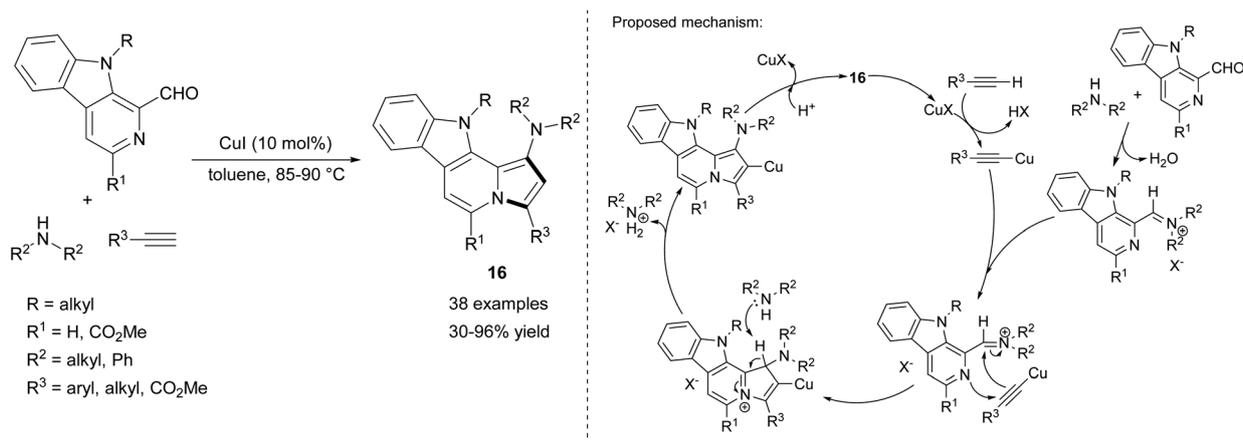
Kim *et al.* reported the direct formation of indolizines **20** through a tandem one-pot Pd-catalyzed cross-coupling reaction using organoindium reagents formed *in situ*, followed by subsequent cycloisomerization of the allenylpyridine intermediate (Scheme 9).²² The role of NaI has not yet been established, but the authors suspect it facilitates the formation of the organoindium reagent through the substitution of bromine. The presence of a strong electron-withdrawing group in the substrate had a negative influence on the reaction outcome.

2.4 From 2-alkylpyridines

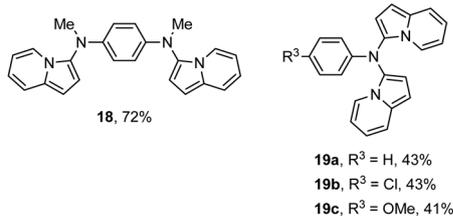
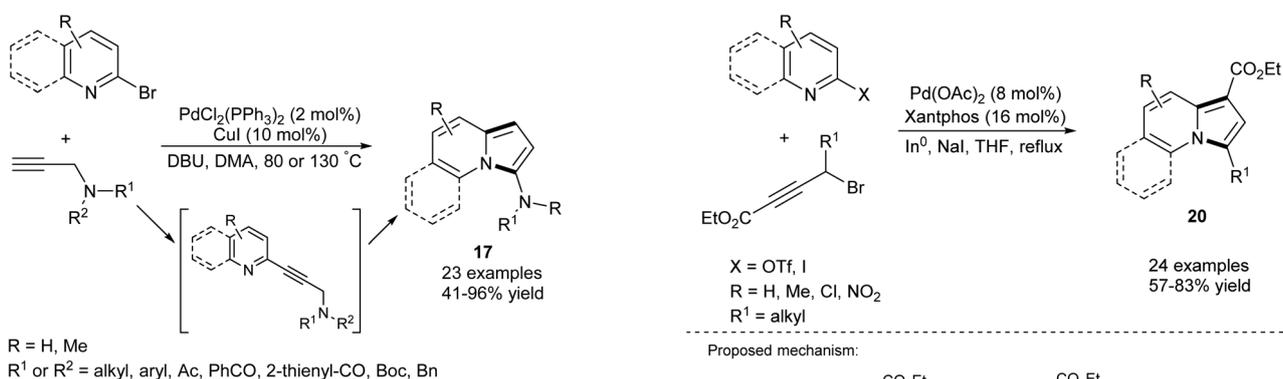
2-Alkylpyridines (especially α -picolines) have long been exploited as valuable substrates to access the indolizine skeleton (*e.g.*, Scholtz or Chichibabin methods for indolizine synthesis). In stark contrast to these older methodologies, the new approaches toward indolizines starting from 2-alkylpyridines rely on oxidation of the initially formed intermediate. A survey of the more recent advances within this area is presented in the following section.

Zhang and co-workers developed a protocol for the annulation of substituted 2-alkylpyridines and cinnamic acids in the

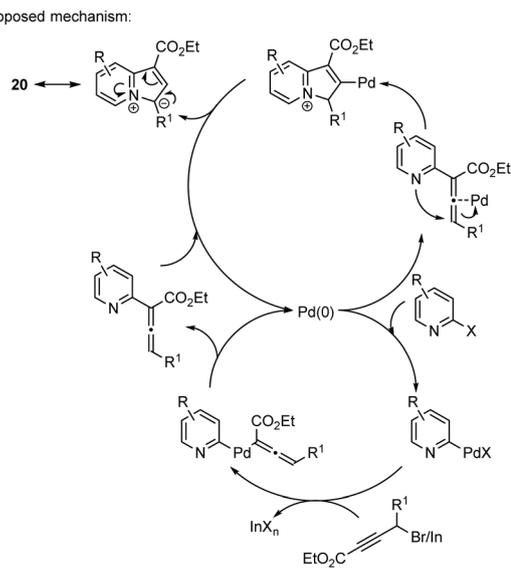




Scheme 7



Scheme 8



Scheme 9

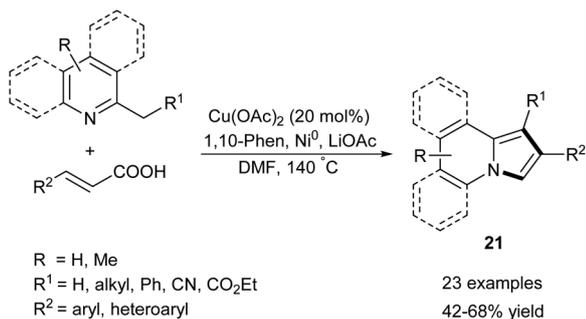
presence of copper(II) acetate, powdered nickel, 1,10-phenanthroline and lithium acetate (Scheme 10).²³ Although the method gave good results for the broad scope of substrates, 2-alkylindolizines cannot be obtained according to this method. It is worth noting that this approach leads to the same pattern of substituents as the Chichibabin reaction.

2-Arylated indolizines (**22**) can also be synthesized by the reaction between 2-alkylpyridines and Morita-Baylis-Hillman acetates (MBHAs) only in the presence of triethylamine (Scheme 11).²⁴ 2-Pyridylacetonitrile reacted well also as did ethyl 2-pyridylacetate, although the yields were found to be slightly lower. Notably, this strategy bears a resemblance to the pyrrole synthesis from imines and nitroolefins, which also proceeds with extrusion of HNO₂.²⁵

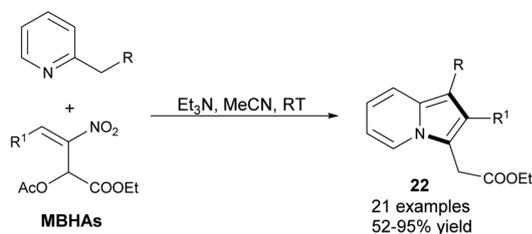
The synthesis of π -expanded indolizines starting from 2-(quinolin-2-yl)acetates and nitroolefins in the presence of

cerium(III) salt has been reported (Scheme 12).²⁶ The method was almost equally efficient when either CoCl₂·6H₂O or ZrCl₄ was used as a catalyst. The authors proposed that the reaction was initiated by the Michael addition to give an adduct, which after cyclization and aromatization with loss of a water molecule and a nitroxyl group (HNO) afforded the final product.

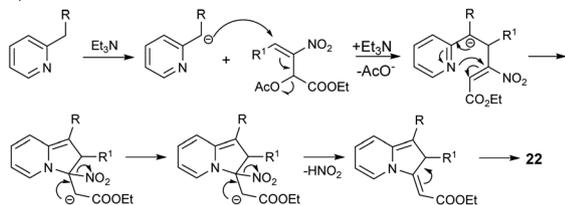




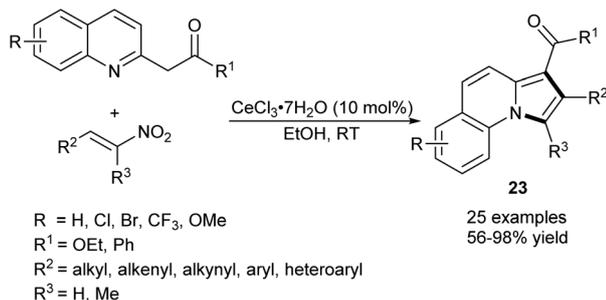
Scheme 10



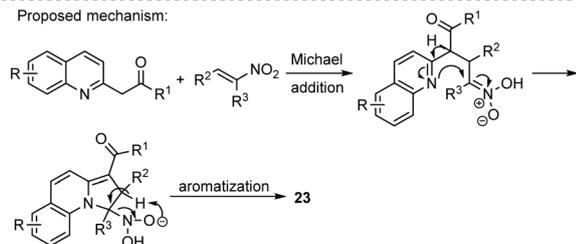
Proposed mechanism:



Scheme 11



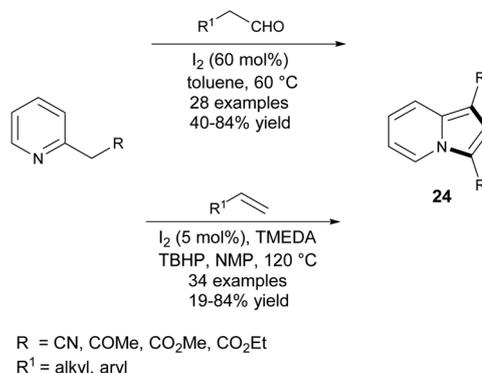
Proposed mechanism:



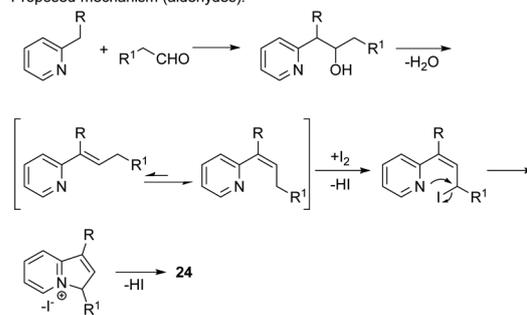
Scheme 12

Lei's group proved that the reaction between 2-(pyridin-2-yl) acetates and nitroolefins can also proceed smoothly without any catalyst.²⁷ Heating the substrates only in DMSO under a nitrogen atmosphere at 100 °C gave various indolizine derivatives in moderate to good yield. However, when 10 mol% copper(II) acetate was used, the rate of the reaction was 2.5-fold faster at the same temperature.

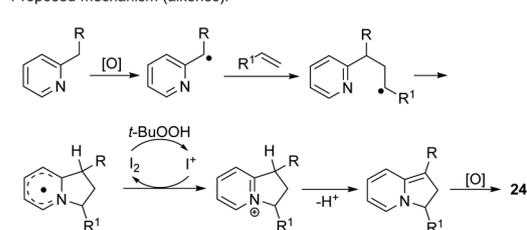
The iodine-mediated, direct oxidative cyclization between 2-alkylpyridines and enolizable aldehydes was reported by Yan's group (Scheme 13).²⁸ Except for *n*-butanal, all tested aldehydes exhibited good reactivity, giving 1,3-disubstituted indolizines **24** in moderate to good yields. The complementary methodology exploiting either alkyl-alkenes or styrenes instead of aldehydes was developed by the same group one year later (Scheme 13).²⁹ Generally, styrenes react with activated 2-alkylpyridines much easier than alkenes, providing better yields of **24**. In two cases, the authors obtained only traces of the expected products (unsubstituted styrene and cyclohexene). An analogous protocol utilizing 2-alkylpyridine derivatives and styrenes was proposed by Tang *et al.*³⁰ This iodine-catalyzed



Proposed mechanism (aldehydes):



Proposed mechanism (alkenes):



Scheme 13



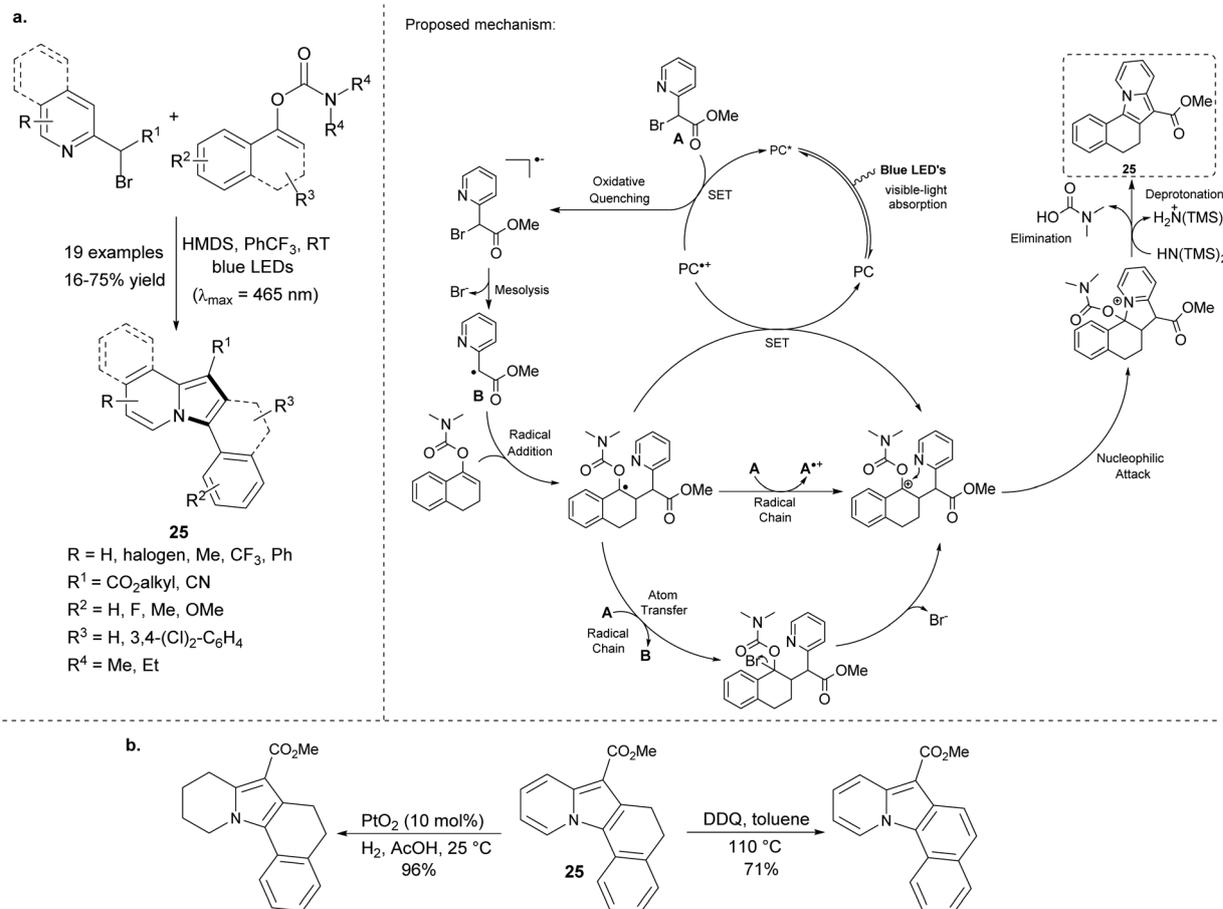
radical oxidative annulation afforded 3-arylidolizines in yields of 25–59% (9 examples) with good regioselectivity. Aliphatic alkenes were not evaluated in this report.

Very recently, Sahoo *et al.* reported the visible-light-mediated preparation of polycyclic indolizines without use of organometallic photosensitizers (Scheme 14a).³¹ The reaction between 2-bromomethyl-pyridine derivatives and enol carbamates gave tetra- and pentacyclic products **25** in moderate to good overall yields. The lowest yield was obtained for a pyridine substrate bearing the CN group in the side chain. The attempts to obtain 2,3-dialkylindolizine were not successful either under the standard conditions or in the presence of the photosensitizer – [Ir(ppy)₂(dtbbpy)](PF₆). Moreover, **25** could be easily oxidized or partially reduced in excellent yields, showing the great synthetic potential of this method (Scheme 14b).

Wang *et al.* proved that propargyl alcohols are also good substrates in the preparation of 2-alkylindolizines (Scheme 15).³² This samarium(III) triflate-catalyzed reaction proceeded under solvent free conditions. Depending on the substituent R², one of the regioisomers **26** or **27** could be obtained over the reaction course.

The silver-mediated cyclization between 2-alkylpyridines and terminal alkynes was investigated by Agrawal's group (Scheme 16).³³ The presence of an ester group in the side chain of pyridine provided generally higher yields of **28**. The authors have stated that the electronic character of substituents of the alkynes had no influence on the reaction outcome. Interestingly, due to the required substantial excess and cost of Ag₂CO₃, the possibility of the salt's regeneration was investigated. It appeared that the recovered salt possessed analogous activity to the commercially available salt.

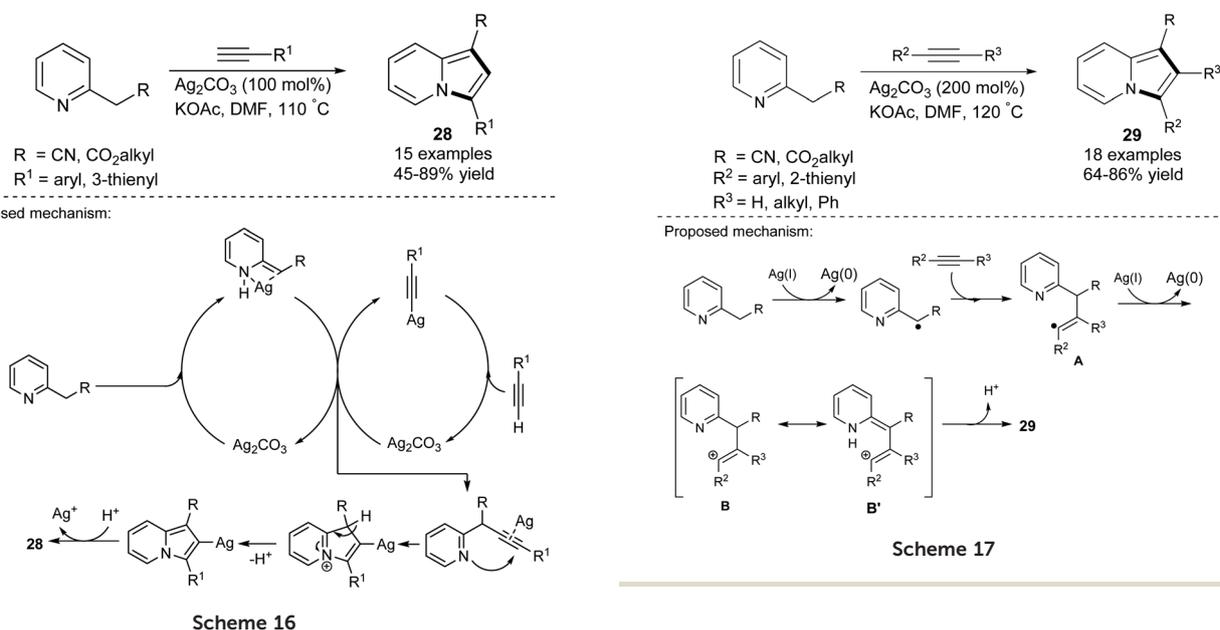
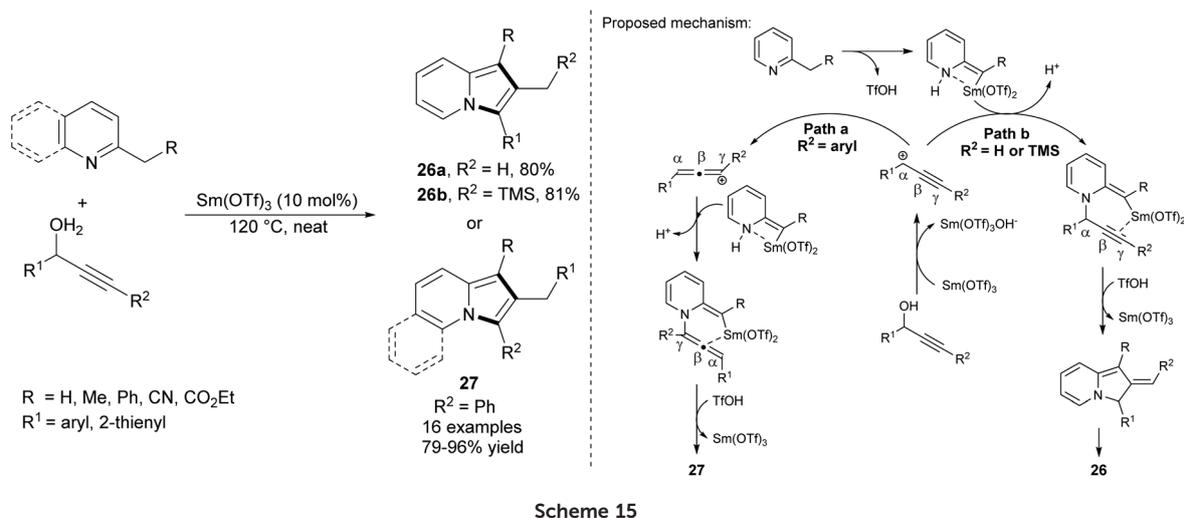
Tan *et al.* proved that disubstituted alkynes also reacted smoothly with 2-alkylpyridines in the presence of silver(I) carbonate to give 1,2,3-trisubstituted indolizines **29** with excellent regioselectivity in moderate to good yields (Scheme 17).³⁴ In contrast to the previous work,³³ higher yields were obtained when electron-rich alkynes were used as a starting material. In the case of both 1-octyne and 4-octyne, the method gave only traces of the corresponding products. The excellent regioselectivity of this method is probably due to the presence of either the aryl or heteroaryl moiety as R², which enhanced the stability of radical **A** as well as carbocations **B** and **B'** (Scheme 17, bottom).



PC – photocatalyst

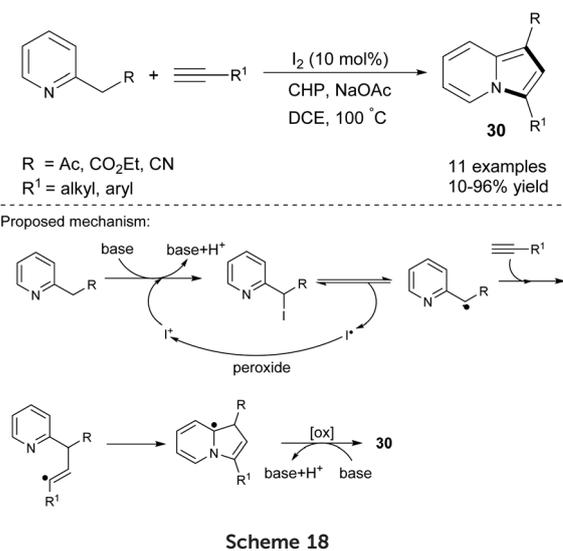
Scheme 14

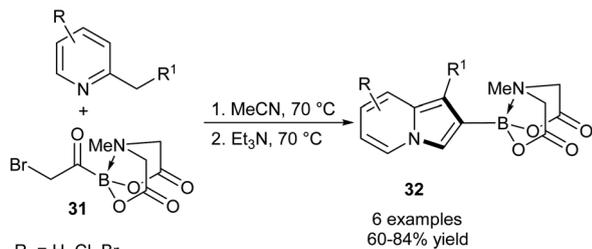




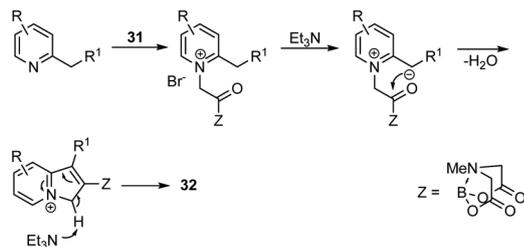
One major drawback of the above reactions is the significant amount of Ag₂CO₃ that was required. Tang *et al.* solved this problem partially by investigating TM-free catalysts in the reaction of 2-alkylpyridines and terminal alkynes. These substrates in the presence of molecular iodine, cumyl hydroperoxide (CHP), and sodium acetate gave indolizines **30** in low to excellent yields (Scheme 18).³⁵ Electron-donating and electron-withdrawing substituents on the aromatic ring in the alkyne were tolerated, while aliphatic alkynes showed poor reactivity under these conditions.

A breakthrough approach toward the synthesis of 2-substituted indolizines was made by Yudin's group (Scheme 19).³⁶ According to their protocol, MIDA-protected bromomethyl acylboronates **31** reacted smoothly with 2-alkylpyridines giving 2-borylated indolizines **32** in good to excellent yields. The

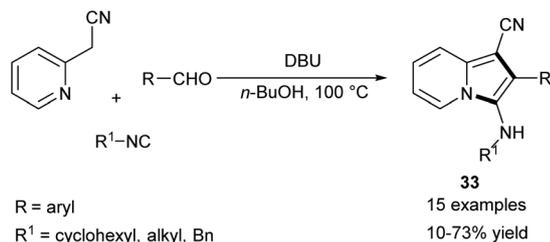




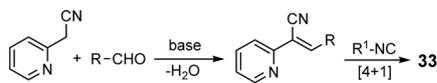
Proposed mechanism:



Scheme 19



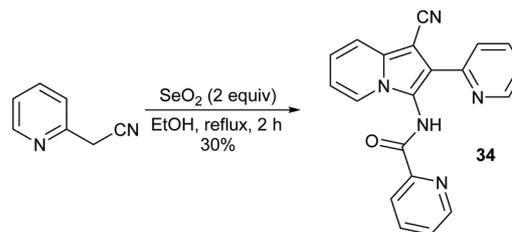
Proposed mechanism:



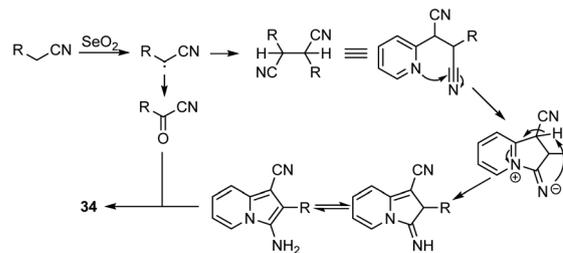
Scheme 20

latter compounds may serve as useful substrates in Suzuki coupling, which makes compounds **32** very promising intermediates in the synthesis of a broad range of 2-functionalized indolizines in a simple manner.

In 2006, Bienaymé and co-workers published the highly efficient, three-component synthesis of diversely substituted indolizines (**33**) by the condensation of 2-(pyridin-2-yl)acetonitrile with aldehydes and isonitriles (Scheme 20).³⁷ An advantage of this methodology is that it gives rise to indolizines possessing a readily functionalizable primary amino group available after removal of a benzyl group. The significant disadvantage of this strategy laid in the use of isonitriles – compounds possessing limited commercial availability and an unpleasant smell. Two years later, Okamoto and Niyomura successfully synthesized indolizine **34** starting from 2-(pyridin-2-yl)acetonitrile and selenium dioxide (Scheme 21).³⁸ Interestingly, the authors found that **34** easily formed the NiL_2 complex by simply heating with $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ in ethanol.



Proposed mechanism:

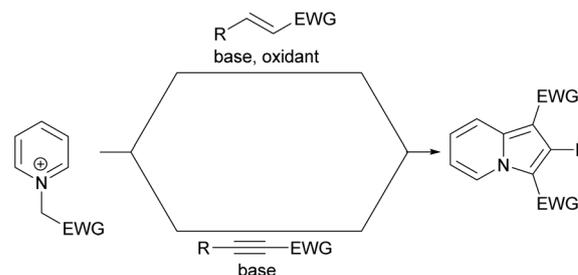


Scheme 21

2.5 From pyridinium salts and dipolarophiles (via 1,3-dipolar cycloaddition)

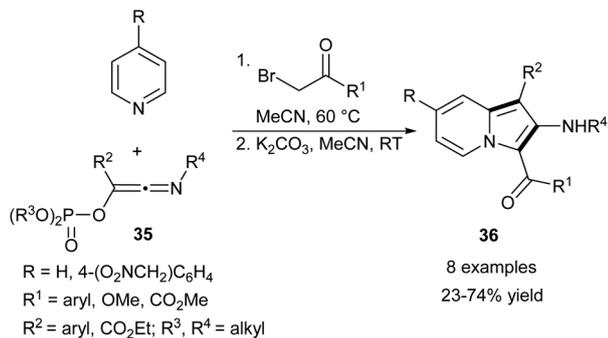
The 1,3-dipolar cycloaddition reaction between pyridinium salts and dipolarophiles is a widely used method for the synthesis of indolizines with acceptor moieties at positions C1 and/or C3 (Scheme 22). Both activated alkenes and alkynes have been used in this approach, with the fundamental difference being that in the first case the addition of a very selective oxidant is required. The first example of successful realization of this type of strategy with alkenes was presented by Hu and co-workers, who used $\text{Cr}(\text{VI})$ -salts as oxidants.³⁹ Taking into consideration the overall number of papers devoted to this topic, in this section we will focus only on the most important reports, which have appeared during the last decade.

In 2010, Coffinier *et al.* succeeded in the synthesis of 2-aminoindolizines (**36**) using phosphorylated hydroxyketenimines (**35**) (Scheme 23).⁴⁰ These reagents were prepared by the Nef-Perkov cascade, exploiting isocyanides as a starting material and they showed good stability in air and on silica. Interestingly, the authors did not detect any side product resulting from a cycloaddition to the carbon–nitrogen moiety of the ketenimine.

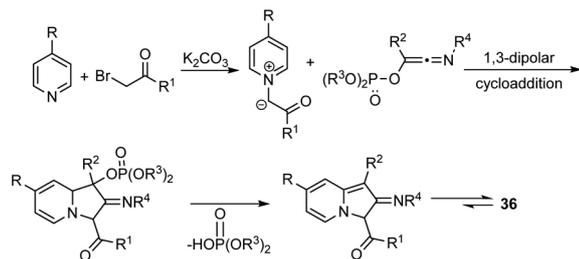


Scheme 22





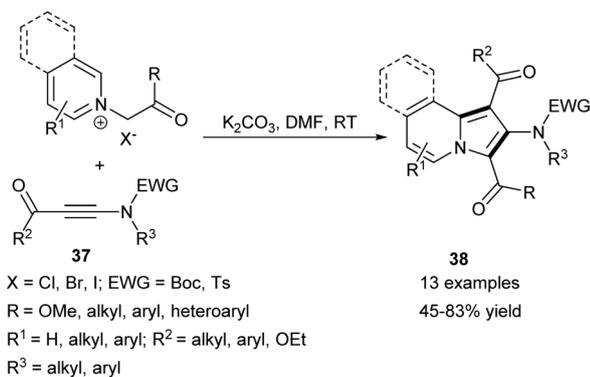
Proposed mechanism:



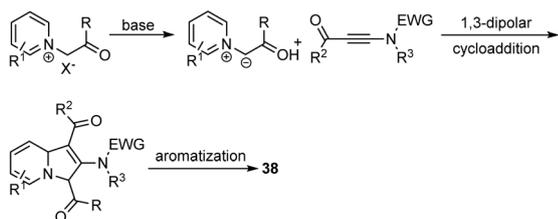
Scheme 23

2-Aminoindolizines **38** can also be successfully obtained in the reaction of either pyridinium or isoquinolinium salts and electron-deficient ynamides **37** (Scheme 24).⁴¹ The synthetic potential of this method was confirmed in the synthesis of 1,3-dialkyl-2-aminoindolizines or a molecule possessing the indoquinolizidine scaffold.

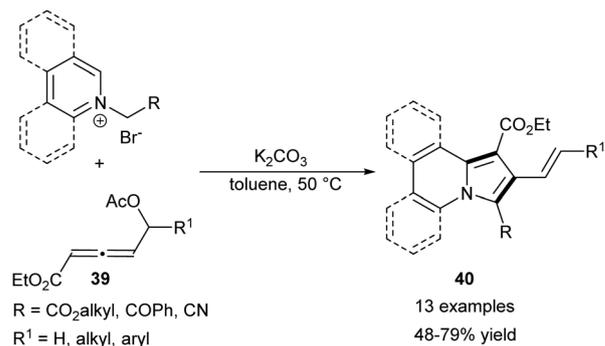
Another interesting finding by Chen *et al.* was that 4-acetoxyallenoates (**39**) are capable of undergoing 1,3-dipolar



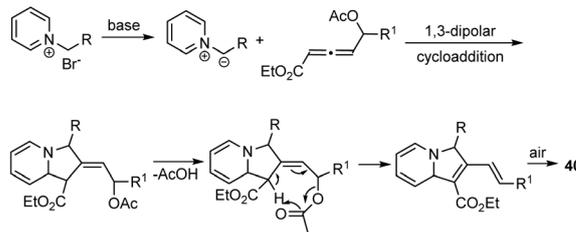
Proposed mechanism:



Scheme 24



Proposed mechanism:



Scheme 25

cycloaddition reactions with various pyridinium salts (Scheme 25).⁴² According to this protocol, a set of 2-styrylindolizines (**40**) as well as their π -expanded analogues was synthesized in good yields with an excellent tolerance of substituents. It is worth pointing out that the oxidation step occurs in the presence of the most environmentally preferred oxidant – air.

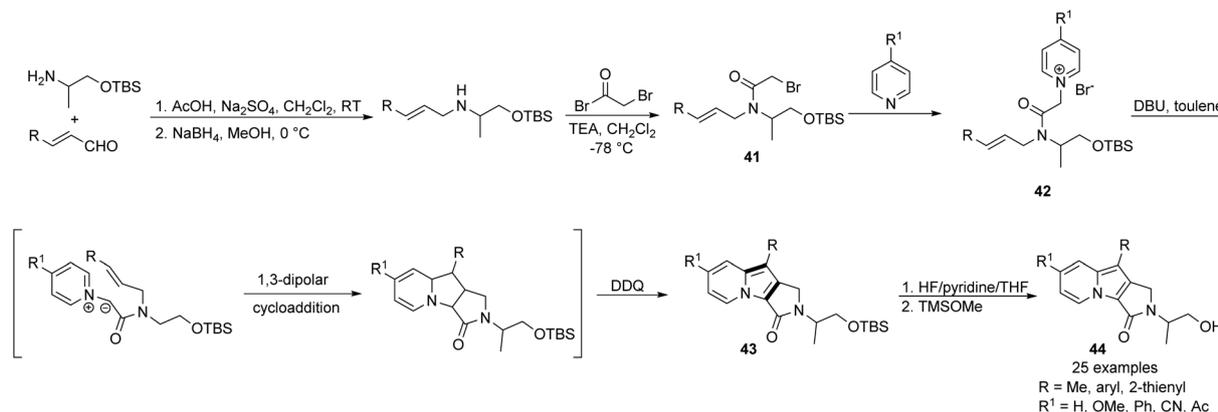
Park and co-workers designed an intramolecular version of that strategy with the aim of obtaining indolizine derivatives possessing additional saturated rings (Scheme 26).⁴³ Pyridinium salts **42**, which were prepared from the pyridine derivatives and the corresponding bromides **41**, underwent an *in situ* cycloaddition/oxidation sequence to give **43**.

Subsequent deprotection led to rather complex and highly fluorescent 1,2-dihydropyrrolo[3,4- β]indolizin-3-ones (**44**) in a relatively simple manner. This method was limited only by the synthesis of building blocks **41**, which could be completed in two steps from readily available chemicals. Advantages of compounds **44** such as pH-dependent fluorescence properties and drug-like lipophilicity for membrane permeability resulted in successful application as bioprobes in HeLa cells. Particular studies on the structure, photophysical property relationship, and potential applications of this intriguing skeleton, termed Seoul-Fluors, have appeared in other articles.⁴⁴

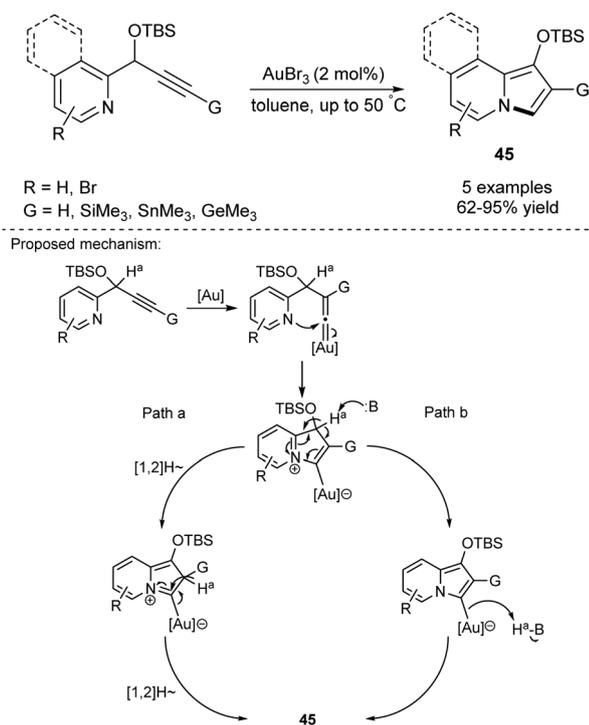
2.6 From 2-pyridine derivatives via cycloisomerization process

In 2006, Gevorgyan and Seregin started a new chapter in the synthesis of indolizines. They developed an Au-catalyzed method for introduction of 2-silyl-, 2-stannyl-, and 2-germyl groups into the indolizine skeleton (Scheme 27).⁴⁵ The reaction proceeded by an alkyne-vinylidene isomerization step fol-





Scheme 26



Scheme 27

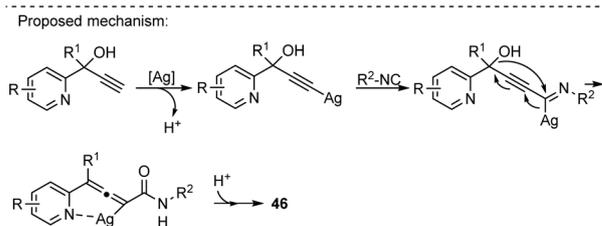
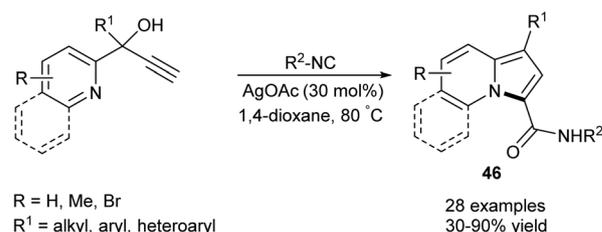
lowed by the 1,2-migration of the corresponding group G to give indolizines **45** in good to excellent yields. Silyl- and stannyl-based compounds were useful reagents in either Hiyama or Stille coupling, thus this method possessed enormous potential in the synthesis of C-2-functionalized indolizines. The detailed mechanism of this process has also been investigated.⁴⁶

This initial report encouraged both Gevorgyan and co-workers as well as other researchers to further exploit this chemistry. Later it was shown that this type of reaction can be also catalyzed by Pt(II),⁴⁷ Cu(I),⁴⁸ or Ag(I)⁴⁹ species and the scope of the reaction was significantly extended. Notably, Sarpong and Narayan proved that cycloisomerization of

propargyl esters may be accomplished by simply heating in pure water.⁵⁰ Other solvents such as methanol, ethanol, or 1,1,1-trifluoroethanol also promoted the formation of product in high yields, probably due to the beneficial effect of hydrogen bonding.

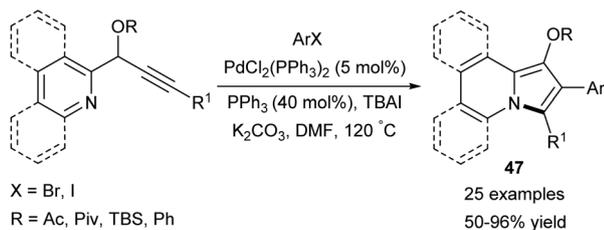
Later, Bi and co-workers studied the Ag-catalyzed cyclization of 2-pyridyl alkynyl carbinols in the presence of isocyanides leading to 1-alkyl- or 1-aryl-3-carboxamidindolizines (**46**) (Scheme 28).⁵¹ This reaction occurred smoothly in most cases, although propargyl alcohol bearing the *i*-Pr group cyclized less efficiently, giving the corresponding indolizine in 30% yield.

Based on their earlier report,⁴⁵ Gevorgyan and co-workers proposed an efficient, one-pot route to 1,2,3-substituted indolizines **47**, where an aryl moiety is introduced into position C2 (Scheme 29).⁵² This cycloisomerization/arylation cascade process exhibited tolerance against a wide range of functional groups present in the coupling partner (ArX) such as halo or nitro, trifluoromethyl, or ester groups. Notably, propargyl phenyl ethers were used as a substrate for the first time to give indolizines in high yields.

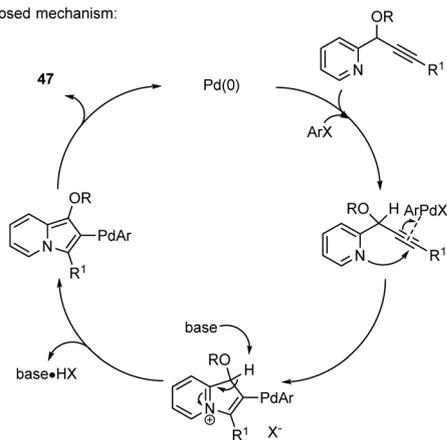


Scheme 28



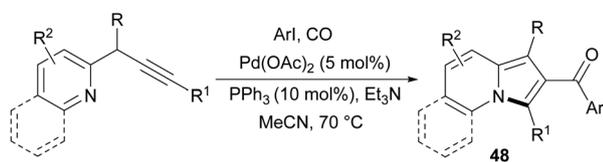


Proposed mechanism:

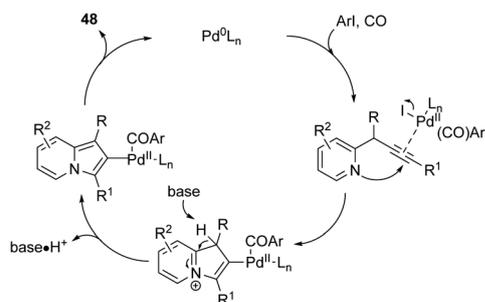


Scheme 29

The series 2-aryloxyindolizines (**48**) can be obtained in a similar manner *via* the palladium-catalyzed carbonylative cyclization/arylation cascade utilizing carbon monoxide as a source of the carbonyl group (Scheme 30).⁵³ This method fully tolerated electron-rich, neutral, or electron-deficient aryl iodides and various pyridine derivatives, although using a 3-methylpyridine derivative led to low yield (35%).



Proposed mechanism:



Scheme 30

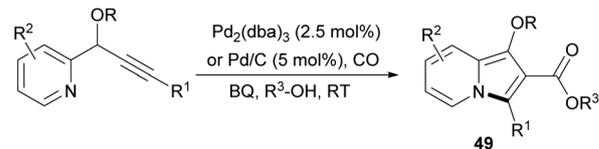
Very recently, Alper and Xu employed a similar carbonylative methodology leading to 2-alkoxycarbonylindolizines (**49**) instead of 2-aryloxyindolizines (Scheme 31).⁵⁴ This reaction, catalyzed by either $\text{Pd}_2(\text{dba})_3$ or Pd/C , proceeds easily at room temperature in the presence of carbon monoxide and benzoquinone as an oxidant. $\text{Pd}_2(\text{dba})_3$ usually gave slightly higher yields in comparison with Pd/C , although palladium on carbon could be easily separated and recovered after the reaction. Relatively low yields were obtained for indolizines bearing the *n*-Bu group at position C3 or when *i*-PrOH was used as an alcohol substrate.

Chernyak and Gevorgyan proposed a palladium-catalyzed carbopalladation/cyclization cascade leading to polycyclic fused heterocycles **50** based on the indolizine skeleton as an intramolecular version of a previously described⁵² method (Scheme 32).⁵⁵ The generality of the process was established by employing substituted propargyl esters, ethers, or silyl ethers to obtain molecules containing the indolizine skeleton fused with coumarin, benzofuran, or the novel silacyclic scaffold, respectively.

Alternatively, indolizines can be synthesized by cycloisomerization of 2-alkynylpyridines. Sarpong and Hardin utilized substrates of this type in the synthesis of 2,3- or 1,3-disubstituted indolizines **51** and **52**, respectively (Scheme 33).⁵⁶ The ratio of **51/52** (determined by comparison of ¹H NMR spectra) was highly dependent on the electronic effect of both R and R¹. The initial 5-*exo-dig* cyclization leading to **A** and then to **51** was more favored in the case of the electron-donating character of both R and R¹, otherwise **B** is formed efficiently *via* 6-*endo-dig* cyclization with a substantial change in **51/52** from >20 : 1 to 2 : 1. The lowest yields were obtained for indolizines bearing the *tert*-butyl or 2-furyl moiety at position C3.

The organocopper-mediated cross coupling/cycloisomerization sequence at low temperatures, leading to **53** was proposed afterward (Scheme 34).⁵⁷ During this process, the organocopper reagent acted as both a catalyst and a nucleophile. It is noteworthy that this methodology allowed the preparation of 1-alkyl or 1-aryloxyindolizines, not accessible by previously described cycloisomerization-based methods. Moreover, the quinoline or isoquinoline derivative could be also used as a substrate leading to π -expanded indolizines.

Very recently, the aryl group was introduced at position C1 *via* a palladium-catalyzed cascade (Scheme 35).⁵⁸ At the initial

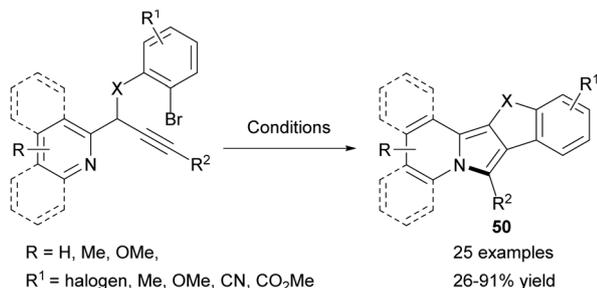


$\text{R} = \text{alkanoyl}, \text{aroyl}, \text{alkenoyl}, \text{TBS}$
 $\text{R}^1 = \text{alkyl}, \text{Ph}, \text{cyclohexenyl}$
 $\text{R}^2 = \text{H}, \text{Br}$
 $\text{R}^3 = \text{alkyl}$

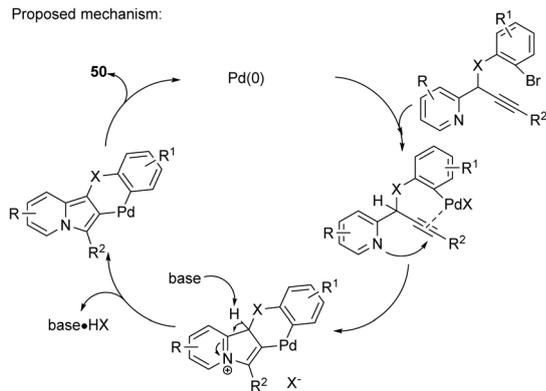
18 examples
24-83% yield

Scheme 31

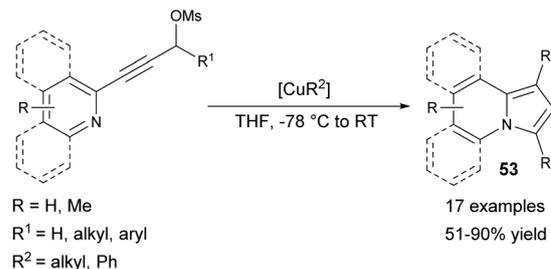




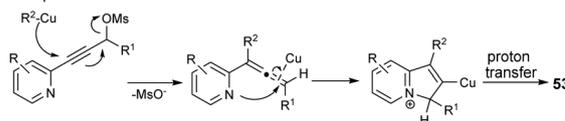
Proposed mechanism:



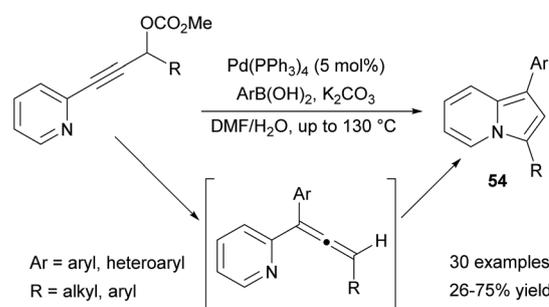
Scheme 32 Reagents and conditions: X = OC(O): Pd(OAc)₂ (5 mol%), K₂CO₃, DMA, 120–130 °C; X = O: Pd(OAc)₂ (5 mol%), TBACl, K₂CO₃, DMF, 120 °C; X = OSi(*i*-Pr)₂O: Pd(OAc)₂ (5 mol%), LiCl, K₂CO₃, *p*-xylene, 135 °C.



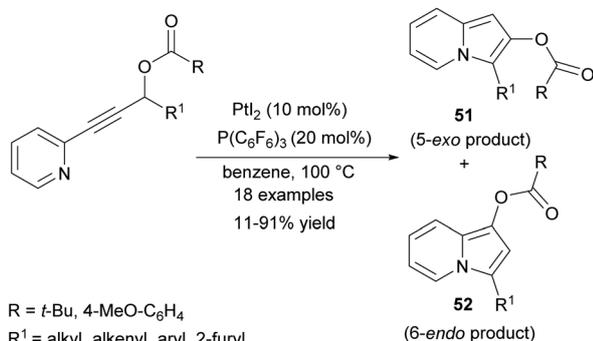
Proposed mechanism:



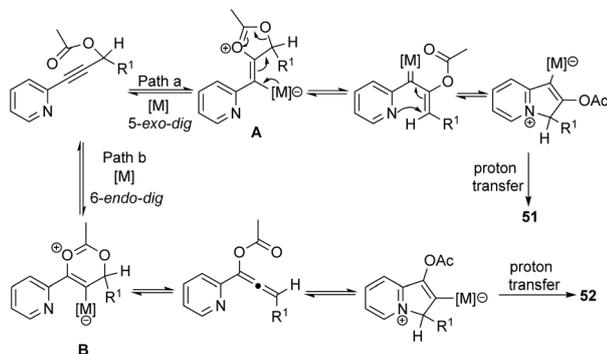
Scheme 34



Scheme 35



Proposed mechanism:



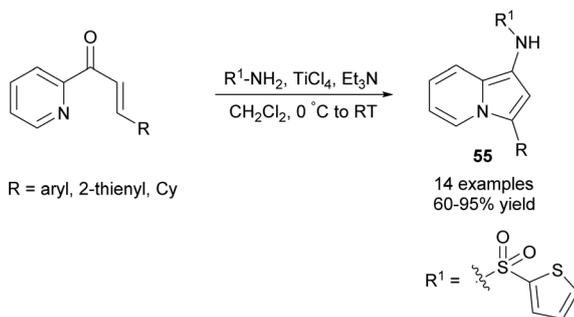
Scheme 33

step, Suzuki coupling of 3-(2-pyridyl)-propargyl carbonates with the corresponding boronic acids gave an allenyl pyridine intermediate, which subsequently underwent spontaneous cycloisomerization to **54**. The above methodology was efficient for a wide range of boronic acids, although 2-furyl boronic acid proved to be less reactive. Taking into account the structure of the starting material, this method was less efficient for pyridines bearing 4-CF₃-C₆H₄, 4-MeO-C₆H₄, or 2,4,6-(MeO)₃-C₆H₄ groups.

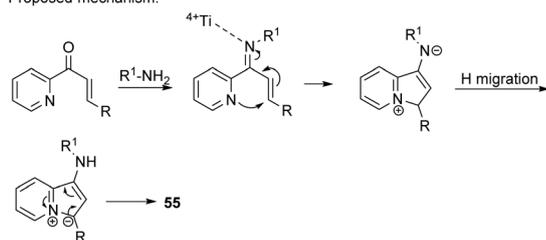
Further research has proven that not only 2-alkynylpyridines and 2-propargylpyridines but also α,β -unsaturated ketones derived from 2-acetylpyridines can serve as substrates for the synthesis of indolizines. In particular, Shi and co-workers demonstrated that 1-aminoindolizines **55** could be easily obtained *via* the titanium(IV)-mediated reaction of 2-enoylpyridines with 2-thienylsulfonamide (Scheme 36).⁵⁹ Both electron-rich and electron-deficient substituents were suitable for this process with the exception of the cinnamyl group which gave no desired indolizine.

Another strategy to utilize 2-enoylpyridines was revealed by Liu and colleagues. These authors discovered that such α,β -unsaturated ketones can be effectively transformed into indolizines **56** under the influence of symmetrical acid anhydrides (Scheme 37).⁶⁰ This method proceeded *via* TM-free C–H bond activation and afforded indolizines in moderate to excellent yields.

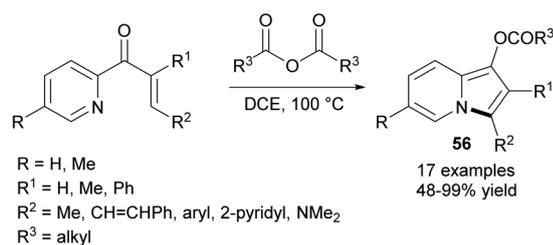




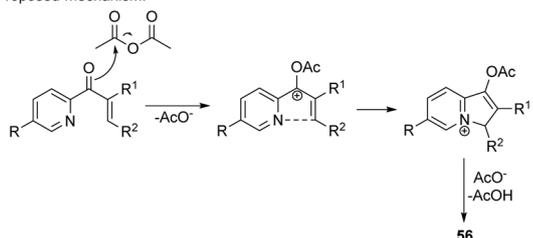
Proposed mechanism:



Scheme 36



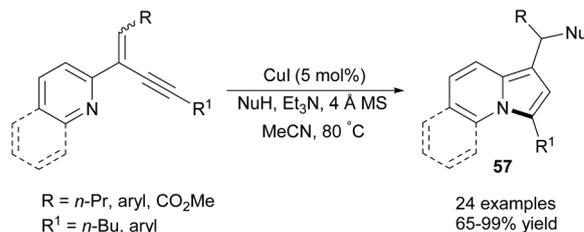
Proposed mechanism:



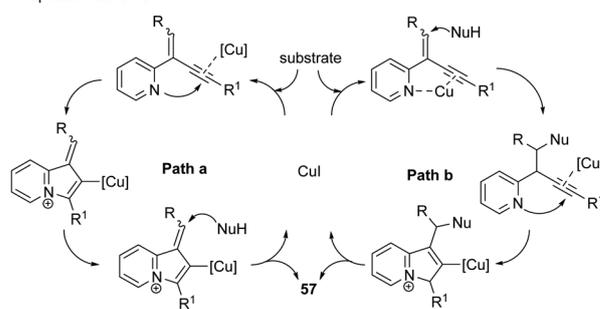
Scheme 37

2-(2-Enynyl)pyridines are also effective substrates toward various indolizine molecules. Jia's group recently proved that this type of cyclization is catalyzed by copper(I) iodide and proceeded readily in the presence of a nucleophile (NuH) such as 1,3-dicarbonyl compounds, indoles, amides, alcohols, and even water and can achieve excellent yields (Scheme 38).⁶¹ In their next paper, they applied PdCl₂(MeCN)₂ as a catalyst and allyl chlorides as an electrophile and obtained 2-allylindolizines (58) in good yields (Scheme 39).⁶² Here, the same set of nucleophiles was used as in the initial study.

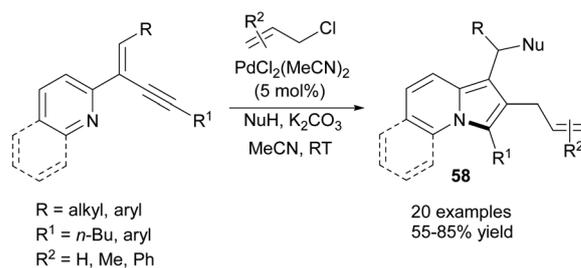
This same research group also developed Au-catalyzed cycloisomerization of 2-(1-alkynyl-cyclopropyl)pyridines in the presence of nucleophiles (Scheme 40).⁶³ In this case, the IPrAuCl/AgNTf catalytic system proved to be very efficient and



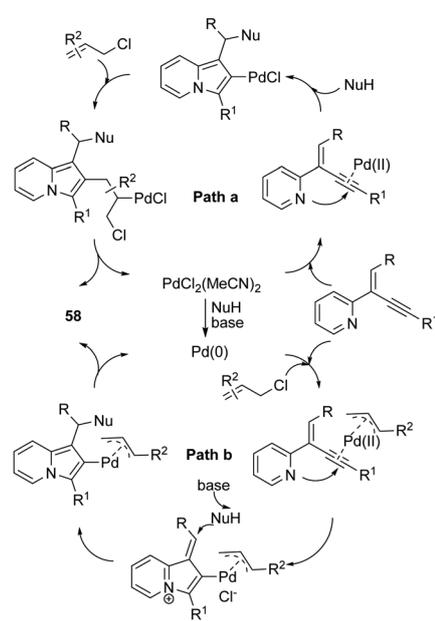
Proposed mechanism:



Scheme 38



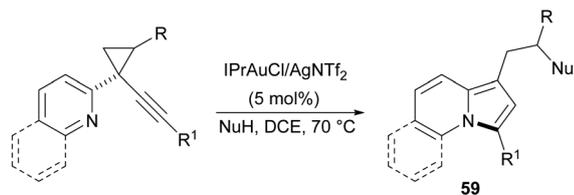
Proposed mechanism:



Scheme 39

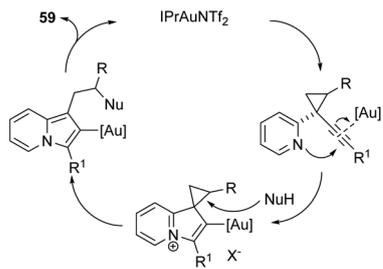
tolerated a variety of functional groups as well as different *N*-, *C*- and *O*-based nucleophiles such as amines, indoles, 1,3-dicarbonyl compounds, alcohols, carboxylic acids, *etc.*





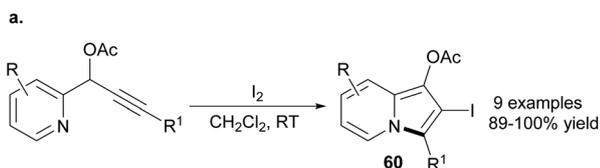
R = aryl
R¹ = *n*-Bu, aryl
27 examples
43-95% yield

Proposed mechanism:



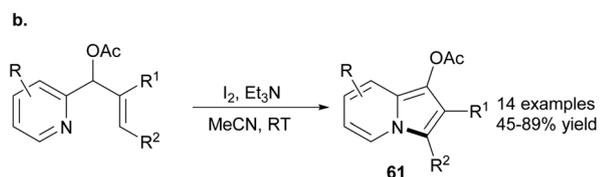
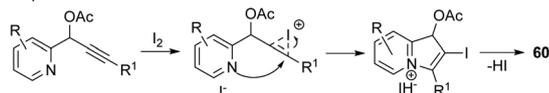
Scheme 40

The cyclization of 2-pyridine derivatives was also achieved in the absence of transition metals. Kim and co-workers developed the iodocyclization of propargylic acetates toward indolizines **60** (Scheme 41a).⁶⁴ The presence of an iodo substituent at position C2 opened new opportunities for the functionalization of **60**. The authors successfully employed **60** as a halogen partner in Suzuki, Heck or Sonogashira couplings in



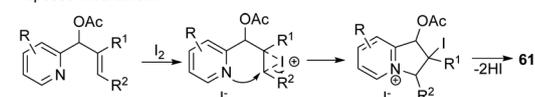
R = H, Me
R¹ = alkyl, cyclohexenyl, aryl, 3-thienyl

Proposed mechanism:



R = H, Me
R¹ = alkyl, Ph
R² = alkyl, alkenyl, aryl, 2-furyl

Proposed mechanism:

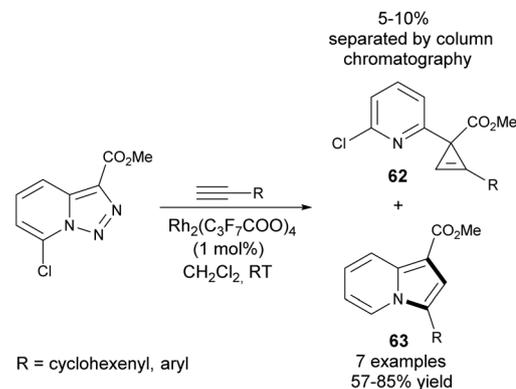


Scheme 41

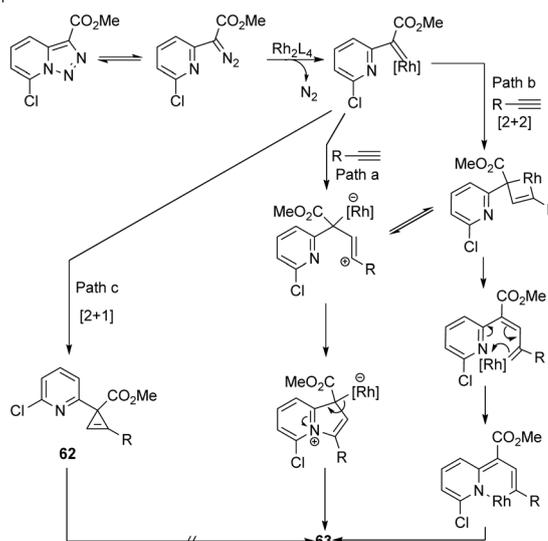
excellent yields. Moreover, the radical reaction of 2-iodoindolizines, *e.g.*, dehalogenation in the presence of the AIBN/*n*-Bu₃SnH system was also possible and provided 1,3-disubstituted indolizines in excellent yields. Additionally, iodocyclization of allylic acetates was also achieved by the same group (Scheme 41b).⁶⁵ In this particular case, the reaction system should be enriched with Et₃N to achieve **61** in satisfactory yields. Later, the series of 3-acylated indolizines⁶⁶ was synthesized by these investigators in a similar manner, providing 1,2,3-trisubstituted indolizines in satisfactory yields. All the aforementioned methodologies tolerated a wide range of functional groups and required mild reaction conditions.

2.7 From pyridotriazoles and alkynes

Gevorgyan and co-workers employed Rh-catalyzed transannulation of activated pyridotriazoles with terminal alkynes for the synthesis of 3-arylindolizines **63** (Scheme 42).⁶⁷ Pyridotriazoles themselves can be obtained either *via* reaction of pyridine-2-yl-acetates with 4-acetamidobenzenesulfonyl azide or from hydrazones *via* oxidative N-N coupling. This reaction proceeded smoothly to produce a mixture of cyclopropene **62** and indolizine **63**, which could be easily separated by column chromatography



Proposed mechanism:

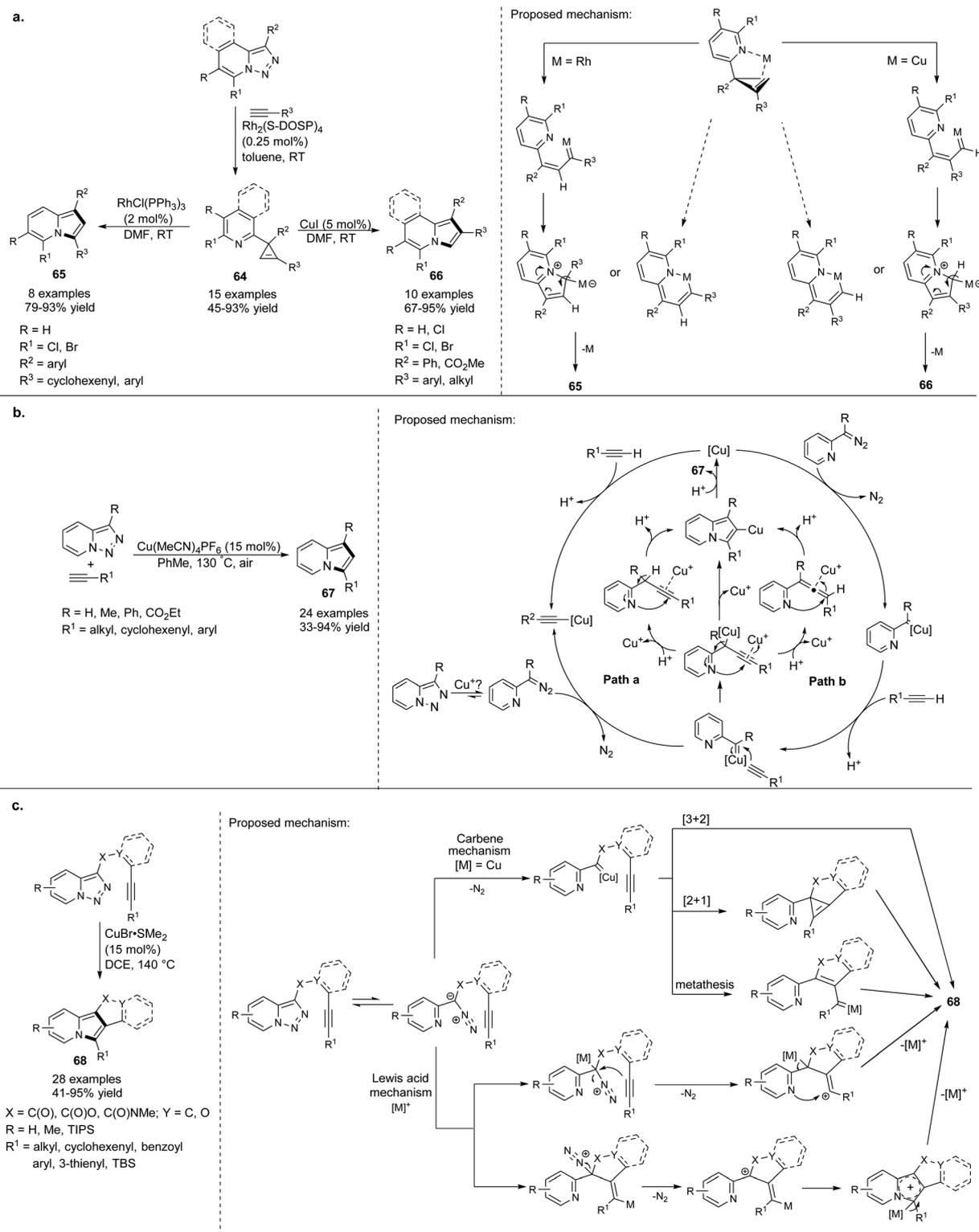


Scheme 42



graphy. Unfortunately, the utilization of aliphatic alkynes resulted in a sluggish or messy reaction. Importantly, it was found that only 7-halopyridotriazoles could indeed serve as convenient substrates in this type of transannulation.

Later, the same group proved that cyclopropenes of type **64** could also be converted into indolizine molecules (Scheme 43a).⁶⁸ They found that depending on the nature of the catalyst, **64** isomerized selectively to 1,3- or 1,2-di-



Scheme 43



substituted indolizines **65** or **66**, respectively. Both methods tolerated electron-rich, electron-deficient, and sterically hindered aryl alkynes. Additionally, aliphatic alkynes were successfully applied in the Cu-catalyzed isomerization of **66** leading to 1,2-disubstituted indolizines **66**. Excellent yields and the presence of halogen in the product provided many opportunities for further transformations, *e.g.*, subjecting these compounds to the coupling reactions.

In the previously described methods^{67,68} 7-halopyridotriazoles were only used as substrates for the Rh-catalyzed transannulation processes. Very recently, Helan *et al.* demonstrated that pyridotriazoles without a halo substituent at position C7 could also be converted into the indolizine skeleton under aerobic conditions (Scheme 43b).⁶⁹ Among copper-based catalysts tested in this transformation, Cu(MeCN)₄PF₆ showed the best catalytic activity. The protocol fully tolerated both aryl and aliphatic alkynes giving **67** in good to excellent yields. A much broader scope of substrates and the lower cost of the catalyst in comparison with previously reported methodologies resulted in a great synthetic potential for this reaction.

Inspired by the aforementioned methodologies in which terminal alkynes were exclusively exploited, Gevorgyan and Shi proposed a simple approach to fused polycyclic indolizines **68** *via* intramolecular transannulation reaction of the pyridotriazole core with an alkyne triple bond catalyzed by CuBr₂·SMe₂ (Scheme 43c).⁷⁰ This reaction was complementary to a previously published method,⁵⁵ and can be catalyzed also by other Lewis acids such as Sc(OTf)₃ or TIPSOTf. Using the appropriately substituted pyridotriazole component, the synthesis of 6-substituted indolizines is possible.

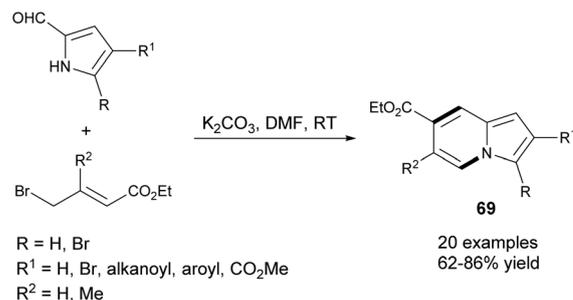
3. Methods based on pyrrole derivatives

In theory, the alternative approach toward assembling the indolizine core starting from pyrrole derivatives and building six-membered rings is simply the mirror image of the previous concept. However, such a synthesis has been a challenge to realize. Numerous difficulties such as significantly fewer commercially available pyrrole derivatives, their lower oxidation potential, and hence their stability are reflected in an overall smaller number of such synthetic strategies.

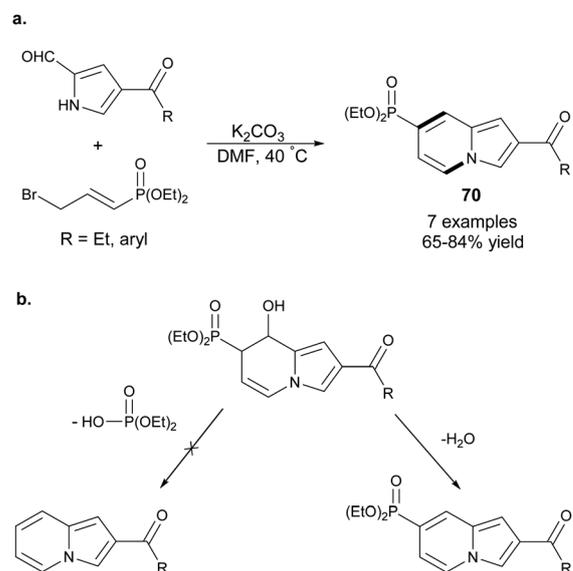
3.1 From N–H free pyrroles

An efficient and highly selective method for the synthesis of **69** starting from 2-formylpyrroles and ethyl γ -bromocrotonates was described by Ge *et al.* (Scheme 44).⁷¹ According to this protocol, it would be possible to obtain 2- or 3-bromoindolizines – useful substrates in a variety of cross-coupling reactions. Subsequently, this method was extended by Yang and co-workers to 2-acetylpyrrole derivatives.⁷²

An analogous type of transformation was used to obtain indolizine-7-phosphonates **70** starting from 2-formylpyrrole derivatives and diethyl 3-bromoprop-1-enylphosphonate (Scheme 45a).⁷³ Mechanistically, this process was similar to



Scheme 44



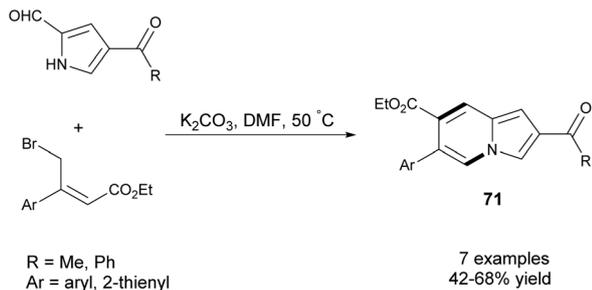
Scheme 45

the above-mentioned methods^{71,72} with the exception of the last stage, in which elimination of a water molecule was more favored due to the conjugation between the indolizine skeleton and the diethyl phosphonate group (Scheme 45b). Moreover, this method was useful in the synthesis of other nitrogen bridgehead heterocycles such as pyrido[1,2-*a*]benzimidazole phosphonates and pyrazolo[1,5-*a*]pyridine phosphonate in moderate to good yields.

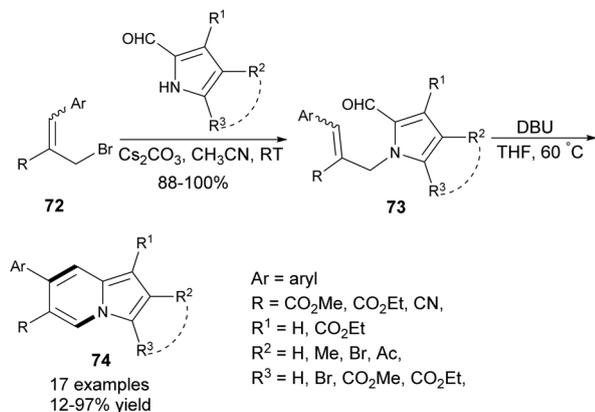
Later, Wang's group utilized β -aryl- γ -bromocrotonates as a synthon to build six-membered rings to give 6-arylindolizines (**71**) in moderate yields (Scheme 46).⁷⁴ In contrast to the protocol used before,⁷¹ higher reaction temperatures (50 °C *vs.* room temperature) and extended reaction times were required due to the electronic effect of the aryl substituent in bromocrotonate.

On the other hand, 7-arylindolizines **74** could be constructed employing the corresponding allyl bromides **72** as a source of the aryl substituent (Scheme 47).⁷⁵ 2-Formylindole or 2-formylpyrrole derivatives smoothly undergo *N*-alkylation with **72**⁷⁶ (pure isomer or *E/Z* mixture) to give **73**, which after isolation were converted into various indolizine derivatives **74**





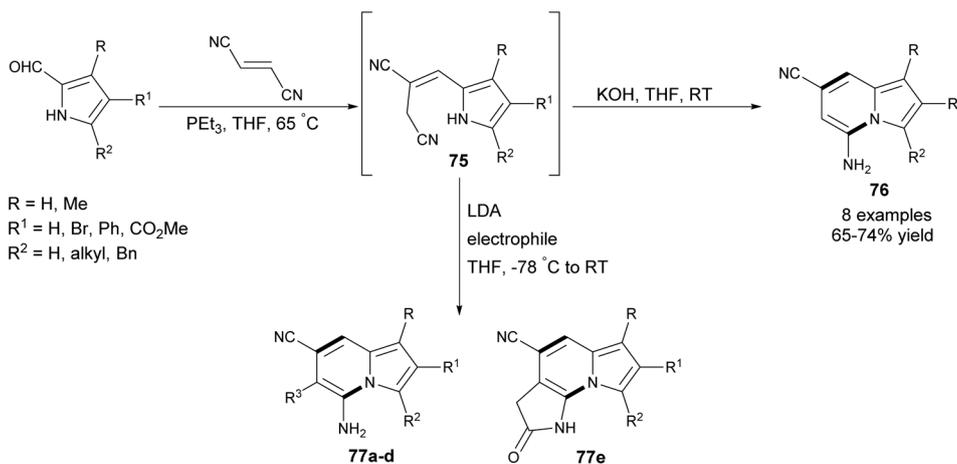
Scheme 46



Scheme 47

in low to moderate yields. Many other heterocyclic skeletons, such as imidazo[1,2-*a*]pyridines and pyrido[1,2-*a*]indoles can be successfully prepared according to this protocol.

Recently, Townsend and co-workers proposed an elegant approach leading to 1,2,3-substituted 5-amino-7-cyanoindolizines **76–77** in good yields *via* a one-pot procedure exploiting 2-formyl pyrrole derivatives and fumaronitrile as substrates (Scheme 48).⁷⁷ In the first step, Wittig olefination took place,



Scheme 48

producing (*E*)-alkene **75** predominately. Next, in the presence of a catalytic amount of KOH, cycloaromatization occurred, giving the corresponding indolizines. Interestingly, there is the possibility of introducing an additional substituent at position C6 by selective alkylation of the allylic position in **75** at low temperature, followed by cyclization induced by a second equivalent of LDA at room temperature (Table 1). This efficient protocol was also applicable to the synthesis of imidazo[1,2-*a*]pyridines and imidazo[1,5-*a*]pyridines in good yields.

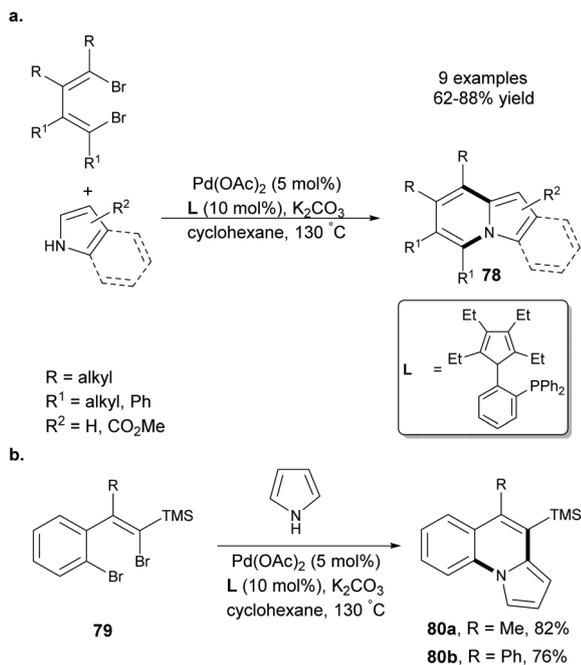
Hao *et al.* demonstrated a facile Pd-catalyzed synthesis of indolizines **78** starting from pyrrole derivatives and 1,4-dibromo-1,3-butadienes (Scheme 49a).⁷⁸ Importantly, in the case of ethyl 3-carboxypyrrole, the mixture of two possible products was obtained in a 1:1 ratio. Additionally, the authors reacted dibromides **79** with pyrrole to obtain π -expanded molecules **80a** and **80b** in 82 and 76% yield, respectively (Scheme 49b). The presence of the trimethylsilyl group (TMS) gave additional possibilities of extending the chromophore, *e.g.*, by Hiyama-type coupling reactions.

An interesting discovery was made recently by Katukojvala and co-workers, who examined the reactivity of 3-substituted indoles toward keto diazoenals in the presence of rhodium-based salts (Scheme 50).⁷⁹ A set of 9,10-disubstituted pyrido[1,2-*a*]indoles **81** was obtained under mild conditions with a wide range of substituents. For example, **81** was selectively hydrogenated (H₂/PtO₂/MeOH) leading to **82** possessing the tetrahydropyrido[1,2-*a*]indole motif, a structure present in many biologically important polycyclic indole alkaloids.

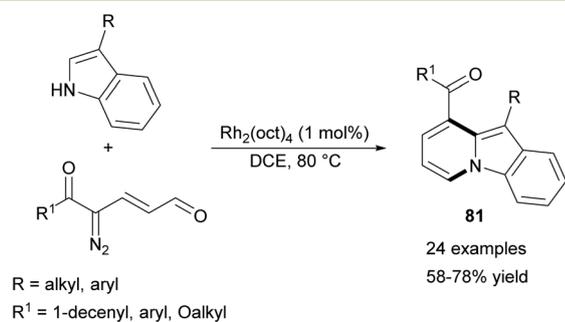
Table 1 Scope of C-6 substituted indolizines depending on an electrophile

Entry	Electrophile	Product	R ³	Yield/%
1	MeI	77a	Me	87
2	H ₂ C=CHCH ₂ Br	77b	-CH ₂ CH=CH ₂	98
3	HC≡CCH ₂ Br	77c	-CH ₂ C≡CH	94
4	PhCH ₂ Br	77d	-CH ₂ Ph	98
5	BrCH ₂ CO ₂ Et	77e		90

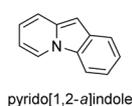
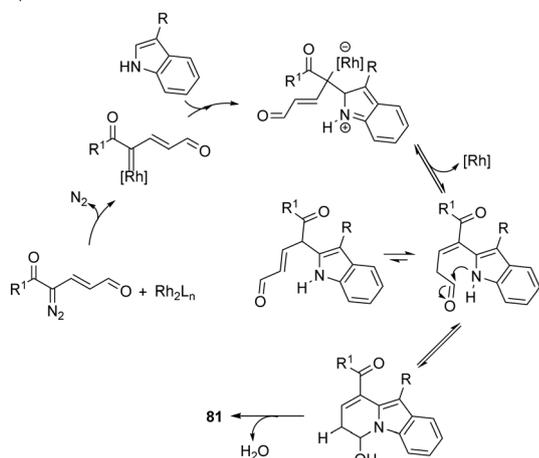




Scheme 49



Proposed mechanism:

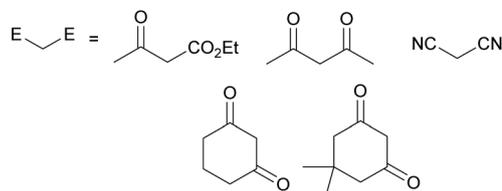
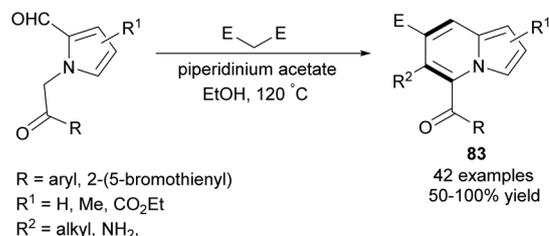


Scheme 50

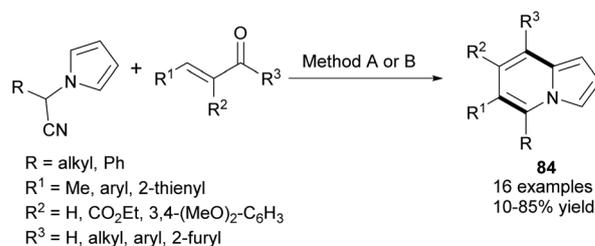
3.2 From N-substituted pyrroles

In 2013, Kim *et al.* developed a new synthetic methodology for the preparation of polysubstituted indolizines **83** (Scheme 51).⁸⁰ This annulation process involves Knoevenagel condensation and intramolecular aldol cyclization and tolerates a broad scope of substrates. A significant advantage of this methodology is the possibility of introducing a variety of substituents into both pyridine-type and pyrrole-type rings. Furthermore, Opatz and Kucukdisli demonstrated a one-pot approach for the synthesis of indolizines **84** fully substituted in the pyridine-type ring (Scheme 52).⁸¹ This method required no expensive reagents or catalysts and afforded products in low to high yields.

At the same time, an easy and efficient route to 6,8-disubstituted indolizines **85** *via* intramolecular cycloaromatization was proposed (Scheme 53).⁸² Once obtained in this way, 8-triflyl indolizines could be further employed in a Pd-catalyzed Suzuki–Miyaura cross-coupling reaction leading to indolizines with various aryl groups at position C8 in 42 to 88% yield (12 examples). In order to maximize diversity, the products of cross-coupling reactions were successfully applied to the Heck coupling, Friedel–Crafts acylation, and Vilsmeier–Haack formylation.

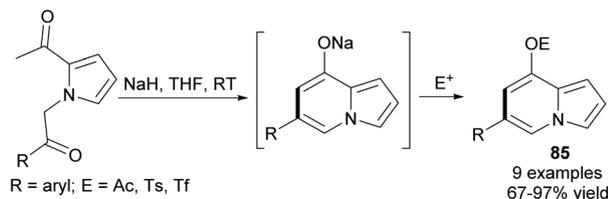


Scheme 51

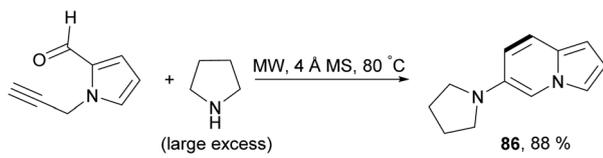


Scheme 52 Reagents and conditions: (A) (i) *t*-BuOK, 0 °C, THF; (ii) AcOH, EtOH, BF₃·OEt₂, RT; (iii) DBU, reflux; (B) (i) *t*-BuOK, DMF, 0 °C; (ii) TfOH, RT; (iii) DBU, 90 °C.

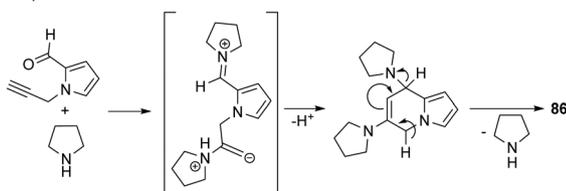




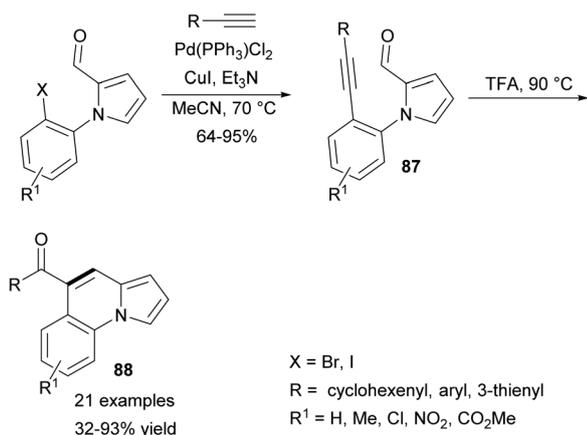
Scheme 53



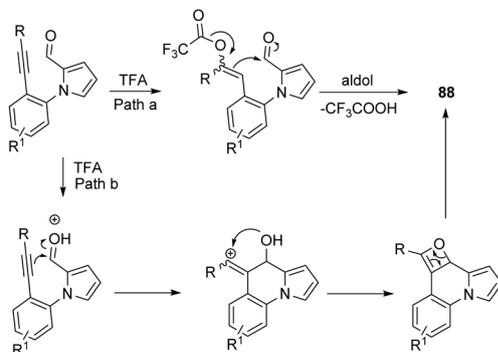
Proposed mechanism:



Scheme 54



Proposed mechanism:



Scheme 55

Microwave-assisted benzannulation of 1-propargyl-2-formylpyrrole in the presence of pyrrolidine gave 6-(pyrrolidin-1-yl)indolizine (**86**) in 88% yield (Scheme 54).⁸³ The reaction proceeded under mild conditions without a metal catalyst and additives. Moreover, the liquid amine acted as a solvent. Undoubtedly, the scope of this process should be examined in detail.

More recently, an intramolecular alkyne-carbonyl metathesis approach toward benzo-fused indolizines **88** was proposed by Kim's group (Scheme 55).⁸⁴ The parent pyrrole derivative **87** was prepared *via* Sonogashira coupling and subsequently transformed into the desired indolizine by heating in TFA. Although a number of TM-based catalysts were tested, none of them proved to be efficient. Substrates bearing alkenyl groups led to the corresponding products in low yields, whereas those possessing alkyl groups were unreactive.

4. Selected examples of methodologies leading to π -expanded indolizines

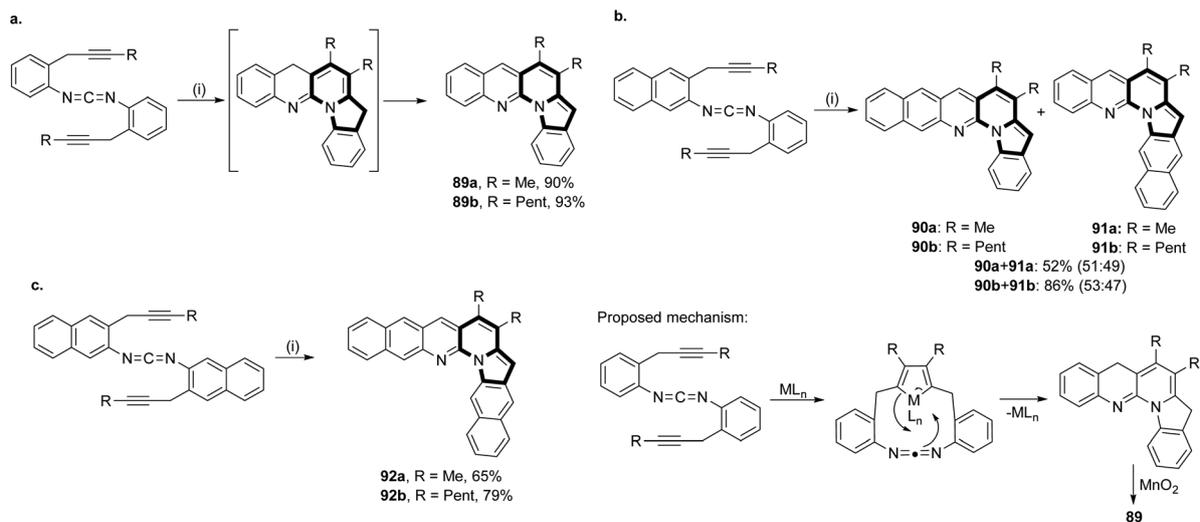
The intramolecular cycloaddition reactions catalyzed by transition metal species are a popular means to construct complex aromatic molecules. Bis(propargylphenyl)carbodiimides can be converted smoothly into L-shaped π -extended compounds with pyrrolo[1,2-*a*][1,8]naphthyridine units **89–92** *via* a [2 + 2 + 2] cycloaddition reaction followed by subsequent oxidation (Scheme 56).⁸⁵ Interestingly, these molecules exhibited sky-blue to golden-orange fluorescence with high quantum yields, even in polar solvents.

In 2015, Kim's group proved that the same type of cyclization used by them previously⁸⁴ could be successfully applied in the synthesis of much more complex derivatives – benzo[*e*]pyrido[1,2-*a*]indoles **93–95** (Scheme 57).⁸⁶ The starting material was obtained in a three step synthesis involving indolizine formation, Vilsmeier–Haack formylation, and Sonogashira coupling. This intramolecular hydroarylation with concomitant extrusion of a formyl group was catalyzed by TFA and the mixture of two products **94–95** was formed. When the reaction was also applied to the substrates bearing the alkyl group at R², the substrates that possessed no formyl group at position C3 gave only **93** as the sole product in the presence of AgOTf. Importantly, applying the substrates bearing an alkyl group such as R² to acidic conditions resulted in a complex mixture, while TM-based conditions gave good yields of products.

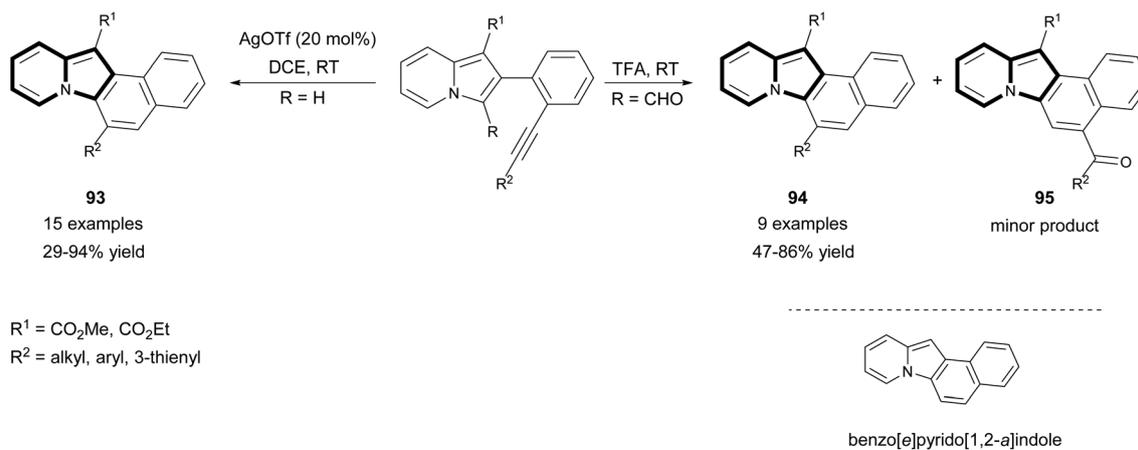
Very recently, the 6-*endo-dig* iodocyclization leading to benzo[*e*]pyrido[1,2-*a*]indoles (**96**) bearing an iodo substituent was developed (Scheme 58).⁸⁷ Some derivatives were applied in Sonogashira or Suzuki coupling, and thus the diversity of products was maximized by introducing two different substituents at positions C5 and C6.

It was recently shown⁸⁸ that the dehydrogenative Heck annelation reaction is a good tool for the synthesis of fused indolizines in a relatively simple manner. Four structurally unique brightly blue or blue-green fluorescent indolizino

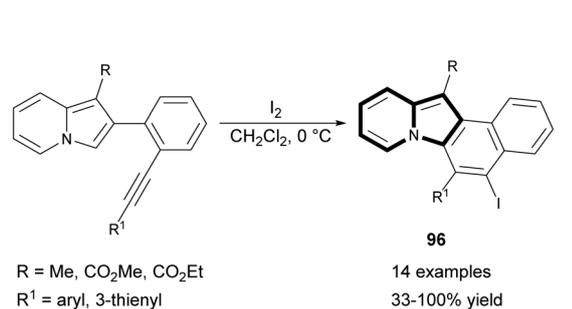




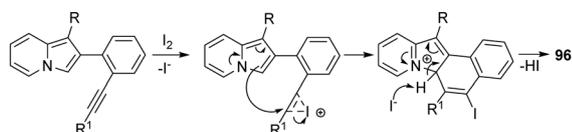
Scheme 56 Conditions: (i) (1) RhCl(PPh₃)₃ (10 mol%), toluene, reflux; (2) MnO₂, RT.



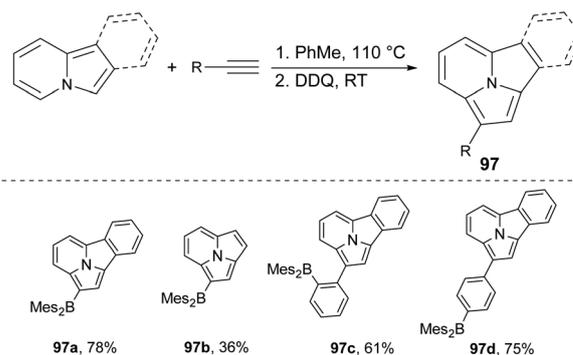
Scheme 57



Proposed mechanism:



Scheme 58



indolizino[3,4,5-ab]isoindole

Scheme 59



[3,4,5-*ab*]isoindoles **97a–d** can be obtained *via* the above mentioned method (Scheme 59).⁸⁹ The X-ray diffraction data of **97a** revealed that there were discrete π -stacked dimers in the crystal lattice, in contrast to previously reported indolizino[3,4,5-*ab*]-isoindole derivatives which showed extended π -stacking interactions.⁹⁰

5. Conclusions

Although the history of indolizine synthesis dates back to the nineteenth century and many methods have been explored to date, there is still a need to develop more versatile and/or eco-friendly methodologies. This review summarizes the greatest achievements in this field over the past ten years and we hope that the advancements presented here, combined with many opportunities in synthetic organic chemistry will be helpful in the development of completely new reactions toward indolizines as well as other heterocyclic motifs.

During the last decade many methods for the synthesis of indolizine molecules have been explored and several prevailing trends can be identified:

(1) while the majority of strategies still have been based on pyridine derivatives as a starting material (with stand-alone contributions from Gevorgyan and co-workers), there have also been a rising number of methodologies employing pyrrole derivatives. Many of these methods are catalyzed by transition metal species, which is undesirable from the viewpoint of medicinal chemistry, but allows, in most cases, the construction of much more diversified molecules.

(2) The unsurpassed simplicity of the Chichibabin synthesis has only one drawback – the substituent in the five-membered ring must always be at position 2. As a consequence, many recent efforts have been concentrated on methods which allow the insertion of substituents at positions 1 and 3. This alteration bears multiple implications, not only for purely academic reasons, but also because these highly reactive positions are responsible for lower stability under ambient conditions. Blocking these sites with substituents will afford more stable indolizines.

(3) One of the recent trends in indolizine synthesis mirrors some more general trends in organic chemistry, namely the attempt to replace a prefunctionalization-based approach with oxidation-based strategies.

(4) In the past, indolizines attracted attention as unsaturated analogs of naturally present indolizidine alkaloids. Interest has shifted in recent years toward fluorescence-related applications. This is perhaps the most profoundly visible in the work of Park's group on the so-called Seoul-Fluor dyes. This, in turn, has influenced the direction of development. The synthesis of indolizines bearing simple alkyl and aryl groups became less important while it also became necessary to introduce substituents possessing a strong electron-donating or electron-withdrawing character. This change has been reflected in the most decisive manner in the emergence of methodologies leading to the formation of indolizines bearing

the amino group (typically tertiary) in positions 1, 2, 3, 5 and 6. Such compounds were hardly available before 2005.

Needless to say, much additional effort is needed to explore the possibilities of some of these molecules. We hope that in addition to organizing knowledge on this topic, this Review will serve as a catalyst to spark further studies. The future target compounds can be extended to more densely substituted indolizines and push-pull indolizines.

Acknowledgements

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