



Cite this: *Org. Biomol. Chem.*, 2016, **14**, 5606

Received 15th February 2016,
Accepted 6th April 2016

DOI: 10.1039/c6ob00365f

www.rsc.org/obc

Protein-specific localization of a rhodamine-based calcium-sensor in living cells†

Marcel Best,^a Isabel Porth,^a Sebastian Hauke,^b Felix Braun,^c Dirk-Peter Herten^c and Richard Wombacher^{*a}

A small synthetic calcium sensor that can be site-specifically coupled to proteins in living cells by utilizing the bio-orthogonal HaloTag labeling strategy is presented. We synthesized an iodo-derivatized BAPTA chelator with a tetramethyl rhodamine fluorophore that allows further modification by Sonogashira cross-coupling. The presented calcium sensitive dye shows a 200-fold increase in fluorescence upon calcium binding. The derivatization with an aliphatic linker bearing a terminal haloalkane-function by Sonogashira cross-coupling allows the localization of the calcium sensor to Halo fusion proteins which we successfully demonstrate in *in vitro* and *in vivo* experiments. The herein reported highly sensitive tetramethyl rhodamine based calcium indicator, which can be selectively localized to proteins, is a powerful tool to determine changes in calcium levels inside living cells with spatiotemporal resolution.

Introduction

Calcium plays a key role as second messenger in living cells and both the spatial and temporal changes of its endogenous concentration $[Ca^{2+}]_i$ have a significant effect on signaling transduction in excitable and non-excitable cells.¹ Different approaches to measure calcium fluctuations in live cell experiments have been developed: the genetically encoded calcium indicators (GECIs)² utilize calcium sensitive fluorescent proteins or fusion proteins with the so-called “cameleons”³ and “camgaroos”⁴ as well known examples. GECIs have the advantage that they can be genetically targeted to any protein of

interest (POI), but they lack fast response to quick changes of $[Ca^{2+}]_i$ in living cells.⁵ Alternatively, small synthetic calcium indicators⁶ as Indo-1/Fura 1-3,⁷ Rhod-1 + 2/Fluo1-3⁸ and Fluo-4⁹ have been developed for calcium imaging in live cells and organisms with the advantage of being much smaller in size compared to fluorescent proteins and therefore less disruptive. Furthermore they can be chemically modified to adapt both the chemical and physical properties to the experimental needs. Over the last decades, numerous small synthetic calcium sensors have been developed and used for calcium measurements in living systems.¹⁰ However, their application is limited to the measurements of $[Ca^{2+}]_i$ in the cytosolic environment as they lack the ability to be specifically localized within the cells. Recently, protein labeling strategies for site-specific localization of small synthetic calcium sensors to a POI utilizing the FAsH-5 and SNAP-tag protein¹¹ labeling strategy have been reported. Whereas the calcium sensor for FAsH-tag applications is based on fluorescein combined with the calcium binding APTRA (2-aminophenol-*N,N,O*-tri-acetat)¹² motive, the SNAP-tag based approach utilizes a BODIPY (boron dipyrromethene)¹³ conjugated to the well-known BAPTA (1,2 bis-(*ortho*-aminophenoxy)ethane-*N,N,N',N'*-tetra-acetate)¹⁴ calcium-ion chelator. The fluorophores used in these studies need to be excited in the blue range of the spectrum, whereas calcium indicators for protein localization, which can be excited at longer wavelengths, have not been reported yet. Here we introduce a new sensitive synthetic calcium indicator based on the fluorophore rhodamine that can be specifically localized to a POI in living cells utilizing the Halo-tag protein labeling strategy (Fig. 1).¹⁵ The new calcium sensor shows a >200-fold increase in fluorescence upon calcium binding and can be excited in the green spectrum ($Ex_{max} = 552$ nm; $Em_{max} = 580$ nm). In a first proof of principle, we demonstrate the specific localization of the calcium sensitive fluorophore to a POI by labeling of a purified Halo fusion protein. For live cell application of the presented strategy, we performed specific labeling of nucleus localized Halo tagged histone protein (H2B) with the calcium sensor. We show that the H2B localized calcium sensor can be used to monitor nuclear calcium

^aRuprecht-Karls-University Heidelberg, Institute for Pharmacy and Molecular Biotechnology, Im Neuenheimer Feld 364, 69120 Heidelberg, Germany.
E-mail: wombacher@uni-heidelberg.de; Fax: +49(0)621 544879;
Tel: +49 (0)621 544879

^bEuropean Molecular Biology Laboratory, Cell Biology and Biophysics Unit, Meyerhofstraße 1, 69117 Heidelberg, Germany

^cRuprecht-Karls-University Heidelberg, Institute for Physical Chemistry, Im Neuenheimer Feld 267, 69120 Heidelberg, Germany

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c6ob00365f



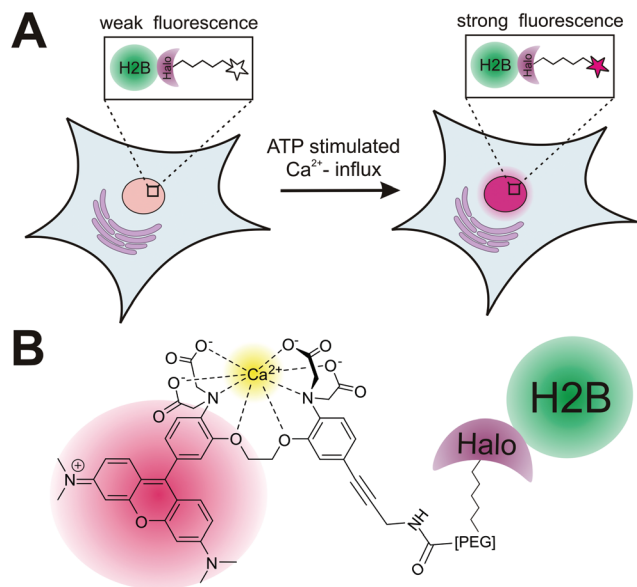
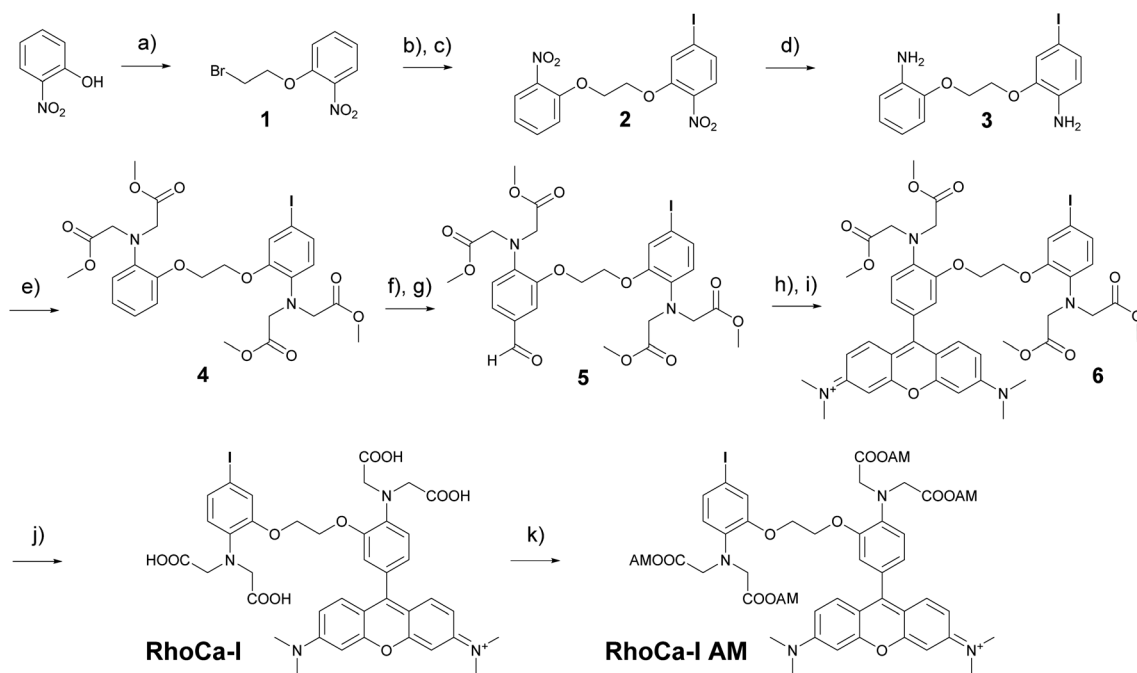


Fig. 1 (A) Measurement of spatiotemporal changes of calcium levels (e.g. in the nucleus) using the RhoCa-Halo probe. Halo-H2B fusion protein, transiently expressed in cells can be specifically labeled with the calcium sensitive probe RhoCa-Halo. ATP stimulated influx of free calcium ions into the cytosol can be visualized by a significant fluorescence increase of the localized probe. (B) Structure of the rhodamine based RhoCa-Halo. The calcium sensor was derivatized at the calcium sensitive BAPTA-site by Sonogashira cross-coupling to introduce a haloalkane function for site-specific labeling by utilizing the HaloTag strategy.

changes in living cells. The herein presented localizable rhodamine derived calcium indicator can be used to measure rapid spatiotemporal changes of $[Ca^{2+}]_i$ in living cells.

Results and discussion

The considerations for the development of the calcium sensor were based on the high photostability and brightness of the fluorophore tetramethyl rhodamine (TAMRA) in aqueous systems¹⁶ and the high selectivity of the BAPTA chelator in the presence of magnesium ions.¹⁴ The commercially available Rhod-2 calcium indicator⁸ which is based on these both chemical components has a K_d -value in the nanomolar range and became a standard tool for calcium measurements in living cells.^{8,17} However, the mitochondrial localization¹⁸ of rhodamine derivatives limits its application. Therefore we sought to develop a calcium indicator with the advantageous photophysical properties of a rhodamine, the excellent calcium selectivity and sensitivity of BAPTA and an additional functionality for further chemical modification. The additional functionality will then be used to attach a haloalkane which serves as a substrate for the HaloTag and enables the covalent attachment to Halo fusion proteins. The highly selective protein localization will then suppress the unwanted mitochondrial localization of the rhodamines. For this reason, we first synthesized a BAPTA core with an iodine modification that allows the selective functionalization by Sonogashira cross-coupling in a final step. Scheme 1 outlines the synthesis



Scheme 1 Synthesis of RhoCa-I and RhoCa-I AM. Reaction conditions: (a) K_2CO_3 , 1,2-dibromoethane, MeCN, reflux; 85%; (b) 5-iodo-2-nitrophenol, K_2CO_3 , DMF, 45 °C; (c) **1**, DMF, 130 °C *o/n*, 84%; (d) Sn/HCl, reflux, quant.; (e) methyl bromoacetate, NaI, proton sponge, MeCN, reflux, 28%; (f) DMF, $POCl_3$, pyridine, 80 °C; (g) H_2O , 0 °C, 27%; (h) 3-(dimethylamino)-phenol, propionic acid, 110 °C; (i) *p*-chloranil, DCM/MeOH, rt, 14%; (j) DCM/MeOH, 1 M NaOH, rt, 67%; (k) bromomethyl acetate, $(iPr)_2NEt$, DCM, rt, 13%.



of **RhoCa-I**, the iodine modified BAPTA calcium chelator coupled to rhodamine. First *ortho*-nitrophenol was alkylated with 1,2-dibromoethane under basic conditions to yield **1**. Further alkylation with 5-iodo-2-nitrophenol and reduction of the nitro functions resulted in the diamine **3**. Alkylation with methyl bromoacetate followed by Vilsmeier–Haack formylation yielded **5** which was then converted to the BAPTA rhodamine **6** by treatment of **5** with 3-(dimethylamino)-phenol under acetic conditions and subsequent oxidation.¹⁹ Finally, the methyl esters of **6** were saponified to obtain the calcium sensitive **RhoCa-I**. To render the newly synthesized calcium indicator cell permeable for its application in living cells, **6** was converted to the respective acetoxymethyl ester (AM-ester) **RhoCa-I AM**.

Calcium titration (in an EGTA-buffered solution with free calcium concentrations ranging from 0 μM to 39 μM , Fig. S1A,† $\lambda_{\text{ex}} = 552 \text{ nm}$) of **RhoCa-I** revealed a 215-fold fluorescence enhancement upon calcium binding and a K_{d} -value of $2.50 \pm 0.17 \mu\text{M}$ (Fig. S1B†). The relative quantum yield²⁰ was determined to be $20.9 \pm 0.6\%$ at $[\text{Ca}^{2+}]_{\text{free}} = 39 \mu\text{M}$ and $0.07 \pm 0.01\%$ at $[\text{Ca}^{2+}]_{\text{free}} = 0 \mu\text{M}$ in reference to Rhodamine B.²¹ As expected, the K_{d} -value of 2.5 μM is significantly higher than the K_{d} -value of Rhod-2 ($\sim 570 \text{ nM}$),¹⁰ which has a methyl group in *para*-position to the alkyl amine groups. We assume that the higher K_{d} -value is due to the iodine substituent as electron withdrawing groups are known to decrease the calcium affinity of the BAPTA chelator.²²

Next, we evaluated the calcium sensitivity of **RhoCa-I** in living cells. Therefore, HeLa cells were incubated with the AM ester of the photo-cleavable *o*-nitrophenyl EGTA (NP-EGTA²³ AM) which has a high affinity for calcium ions ($K_{\text{d}} = 80 \text{ nM}$). One of the chelating moieties can be cleaved off by UV-light radiation, thereby lowering the Ca^{2+} affinity by a factor of more than 10 000. To release calcium ions from cellular internal stores, the HeLa cells were stimulated with 100 μM ATP²⁴ followed by the incubation with 10 μM of the AM ester²⁵ of **RhoCa-I** (**RhoCa-I AM**). As presented in Fig. 2, the calcium sensor shows a significant increase of fluorescence intensity after calcium release from NP-EGTA after UV-radiation. As mentioned before, rhodamine dyes tend to localize to mitochondria what we also observed for the calcium sensor **RhoCa-I** (Fig. 2A). To overcome that limitation and to actively choose the subcellular site for localization, we further synthesized the calcium sensor with a haloalkane functionality for protein labeling utilizing the bioorthogonal HaloTag-strategy (**RhoCa-Halo**, Scheme 2). First we derivatized **6** in a Sonogashira cross coupling reaction with Boc-protected propargylamine. After deprotection of the amine, the haloalkane functionality for protein localization was introduced by peptide bond formation. Finally, the methyl esters were converted to the free acids by saponification to obtain **RhoCa-Halo**.

Next, we evaluated the suitability of the newly synthesized **RhoCa-Halo** for protein labeling by an *in vitro* experiment. **RhoCa-Halo** was incubated with different concentrations of a purified HaloTag fused to green fluorescent protein (GFP-Halo) and the covalent attachment of the probe to the

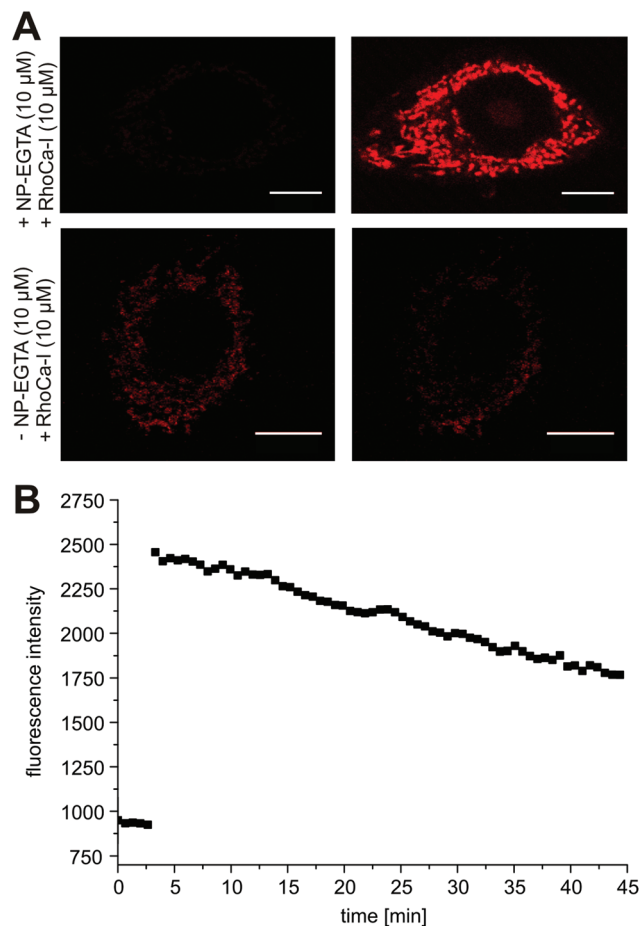


Fig. 2 (A) Live cell imaging of UV-mediated release of Ca^{2+} , $\lambda_{\text{ex}} = 559 \text{ nm}$, detection at 572–672 nm. Upper panel: HeLa cells were incubated with 10 μM NP-EGTA AM and endogenous release of calcium ions from cellular stores was stimulated by ATP. After incubation with 10 μM **RhoCa-I AM** (upper panel, left), NP-EGTA-chelated Ca^{2+} was liberated by UV-radiation (405 nm) leading to a 2.5-fold increase of fluorescence (upper panel, right). Lower panel: Cells were incubated with **RhoCa-I AM** and the sensor shows clear sub-cellular localization (lower panel, left). NP-EGTA AM was omitted in the experiment and no fluorescence increase is monitored upon UV-radiation (lower panel, right). Scale bars represent 10 μm . (B) Corresponding fluorescence intensity trace. UV-mediated uncaging was conducted after 2 min with a short laser pulse.

model protein GFP-Halo was visualized by SDS-PAGE. Fig. 3 shows both the in-gel fluorescence scan of the SDS-gel and the Coomassie stain. From the fluorescence-scan it becomes clear, that GFP-Halo gets efficiently labeled with **RhoCa-Halo**. BSA served as negative control and did not show any labeling by the calcium sensor. As expected, the positive control with a non-calcium sensitive **Halo-rhodamine**, a commercially available substrate for the HaloTag (see ESI† for structure) resulted in labeling of the Halo fusion protein. When fluorescence emission was recorded at $\lambda_{\text{em}} = 580 \text{ nm}$, we still detected fluorescence from GFP. Therefore, we chose to measure the fluorescence at $\lambda_{\text{em}} = 670 \text{ nm}$, a range where GFP does not emit fluorescence light when excited, to selectively record fluorescence from **RhoCa-Halo**. Fig. 3 shows no fluorescence for



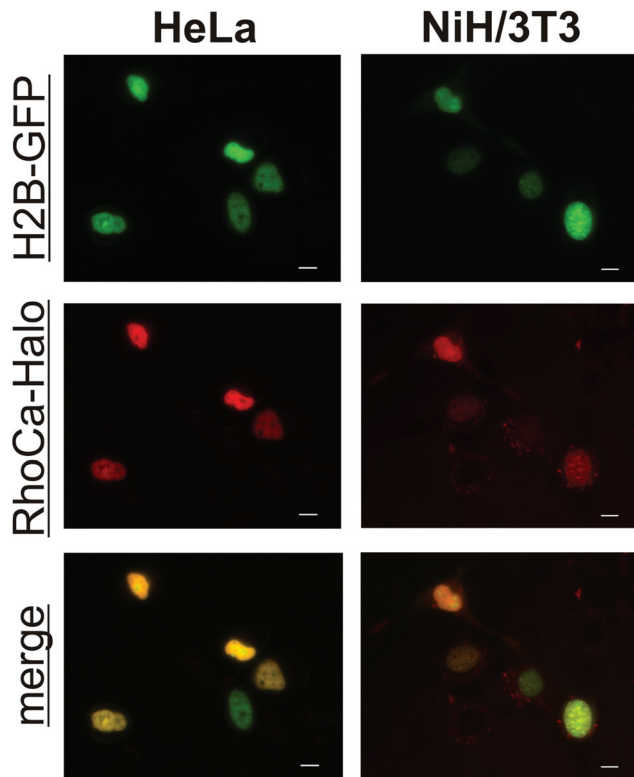


Fig. 4 Epifluorescence microscopic images of H2B-GFP and Halo-H2B co-transfected HeLa and NIH/3T3 cells after incubation with 1 μ M RhoCa-Halo AM. The overlay of the red and the green channel indicates successful labeling of the nucleus in living cells. Scale bars represent 10 μ m. GFP was excited at 472 ± 20 nm (green channel, GFP) and the RhoCa-Halo was excited at 562 ± 20 nm (red channel, RhoCa-Halo).

RhoCa-Halo AM. Non-transfected cells were distinguished from the transfected cells by confocal microscopy. Cytosolic influx of free calcium ions was then stimulated with 100 μ M ATP and a noticeable increase of fluorescence was observed in the nucleus of the transfected and labeled cells directly upon addition of ATP (Fig. 5). The non-transfected cells did not show any detectable increase of fluorescence in the nucleus. This clearly demonstrated the potential of the fluorescent sensor **RhoCa-Halo** as a localizable calcium sensor to monitor spatiotemporal changes of free calcium ions in live cell experiments.

Conclusion

In conclusion, we developed a new localizable rhodamine based calcium sensor which can be applied in live cell imaging. **RhoCa-Halo** is a localizable highly sensitive calcium sensor which exhibits an increase of fluorescence upon calcium-binding, thereby enabling monitoring spatiotemporal changes of calcium concentrations in living cells. We demonstrated its application in *in vitro* and live cell experiments. Moreover, **RhoCa-I** serves as a general building block that can be readily derivatized by Sonogashira cross-coupling to intro-

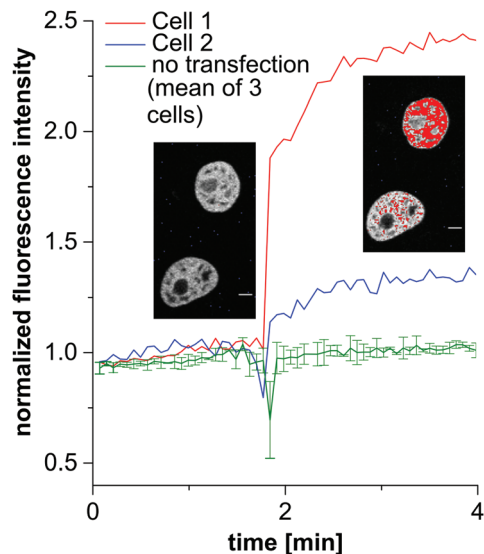


Fig. 5 Confocal microscopic images of Halo-H2B transfected HeLa cells after incubation with 1 μ M RhoCa-Halo AM, $\lambda_{\text{ex}} = 559$ nm, detection at 572–672 nm. Cytosolic influx of free calcium ions was stimulated by addition of ATP. A significant increase of fluorescence intensity in the nucleus was detected immediately after addition of ATP. Shown are the normalized nuclear fluorescence intensity traces of the depicted HeLa cells (blue and red curve). The average normalized fluorescence intensity in the nucleus of three non-transfected cells is shown in green. Scale bars 5 μ m.

duce alternative functionalities to the rhodamine calcium sensor. **RhoCa-Halo** is the first localizable small molecule based calcium sensor with excitation in the green spectrum and represents a valuable addition to the toolbox of calcium imaging. The sensor can be combined with other sensors which are spectrally separated. Furthermore, the application of the HaloTag labeling strategy allows multi-color calcium imaging of different organelles in living cells when combined with the orthogonal SNAP-tag based localizable calcium sensor,¹¹ an Indo-1 derivative with excitation in the near-UV ($\text{Ex}_{\text{max}} = 350$ nm; $\text{Em}_{\text{max}} = 475$ nm). The specific localization of a fluorescent calcium sensor to any POI has great potential to study the spatiotemporal regulation of calcium as a second messenger. Furthermore we would like to point out that the high photo-stability and specific photo-physical properties of the rhodamine moiety makes the newly developed **RhoCa-Halo** probe a potential candidate for application in super-resolution microscopy.²⁹

Acknowledgements

This work was supported by the Deutsche Forschungsgemeinschaft DFG (SPP1623, WO 1888/1-2). We thank the Nikon Imaging Center Heidelberg for access to fluorescence microscopes. We gratefully acknowledge Carsten Schultz (EMBL) for access to confocal microscopes and for providing NP-EGTA AM. We thank the Department of Infectious Diseases



(Medical Biology and Hygiene, group of Konrad Bode) for access to the in-gel fluorescence scanner. We acknowledge the support from the Inorganic Chemical Institute of the University of Heidelberg (Prof. Dr Markus Enders and Beate Termin) for providing access to the NMR facility for spectroscopic measurements. We thank Mathis Baalman, Achim Wieczorek and Michael Heimes for scientific support and Heiko Rudy for technical support.

Notes and references

- D. E. Clapham, *Cell*, 2007, **131**, 1047–1058.
- (a) J. Zhang, R. E. Campbell, A. Y. Ting and R. Y. Tsien, *Nat. Rev. Mol. Cell Biol.*, 2002, **3**, 906–918; (b) M. Mank and O. Griesbeck, *Chem. Rev.*, 2008, **108**, 1550–1564; (c) T. Rose, P. M. Goltstein, R. Portugues and O. Griesbeck, *Front. Mol. Neurosci.*, 2014, **7**, 88.
- A. Miyawaki, J. Llopis, R. Heim, J. M. McCaffery, J. A. Adams, M. Ikura and R. Y. Tsien, *Nature*, 1997, **388**, 882–887.
- O. Griesbeck, G. S. Baird, R. E. Campbell, D. A. Zacharias and R. Y. Tsien, *J. Biol. Chem.*, 2001, **276**, 29188–29194.
- O. Tour, S. R. Adams, R. A. Kerr, R. M. Meijer, T. J. Sejnowski, R. W. Tsien and R. Y. Tsien, *Nat. Chem. Biol.*, 2007, **3**, 423–431.
- M. Oheim, M. van 't Hoff, A. Feltz, A. Zamaleeva, J. M. Mallet and M. Collot, *Biochim. Biophys. Acta*, 2014, **1843**, 2284–2306.
- G. Grynkiewicz, M. Poenie and R. Y. Tsien, *J. Biol. Chem.*, 1985, **260**, 3440–3450.
- A. Minta, J. P. Kao and R. Y. Tsien, *J. Biol. Chem.*, 1989, **264**, 8171–8178.
- K. R. Gee, K. A. Brown, W. N. Chen, J. Bishop-Stewart, D. Gray and I. Johnson, *Cell Calcium*, 2000, **27**, 97–106.
- R. M. Paredes, J. C. Etzler, L. T. Watts, W. Zheng and J. D. Lechleiter, *Methods*, 2008, **46**, 143–151.
- (a) M. Bannwarth, I. R. Correa, M. Sztretye, S. Pouvreau, C. Fellay, A. Aebischer, L. Royer, E. Rois and K. Johnsson, *ACS Chem. Biol.*, 2009, **4**, 179–190; (b) M. Kamiya and K. Johnsson, *Anal. Chem.*, 2010, **82**, 6472–6479.
- B. Raju, E. Murphy, L. A. Levy, R. D. Hall and R. E. London, *Am. J. Physiol.*, 1989, **256**, C540–C548.
- I. J. Arroyo, R. R. Hu, G. Merino, B. Z. Tang and E. Pena-Cabrera, *J. Org. Chem.*, 2009, **74**, 5719–5722.
- R. Y. Tsien, *Biochemistry*, 1980, **19**, 2396–2404.
- G. V. Los, L. P. Encell, M. G. McDougall, D. D. Hartzell, N. Karassina, C. Zimprich, M. G. Wood, R. Learish, R. F. Ohana, M. Urh, D. Simpson, J. Mendez, K. Zimmerman, P. Otto, G. Vidugiris, J. Zhu, A. Darzins, D. H. Klaubert, R. F. Bulleit and K. V. Wood, *ACS Chem. Biol.*, 2008, **3**, 373–382.
- (a) C. Eggeling, J. Widengren, R. Rigler and C. A. M. Seidel, *Anal. Chem.*, 1998, **70**, 2651–2659; (b) B. Hein, K. I. Willig, C. A. Wurm, V. Westphal, S. Jakobs and S. W. Hell, *Biophys. J.*, 2010, **98**, 158–163.
- J. Qiu, Y. W. Tan, A. M. Hagenston, M. A. Martel, N. Kneisel, P. A. Skehel, D. J. A. Wyllie, H. Bading and G. E. Hardingham, *Nat. Commun.*, 2013, **4**.
- (a) G. Hajnoczky, L. D. Robbgaspers, M. B. Seitz and A. P. Thomas, *Cell*, 1995, **82**, 415–424; (b) D. R. Trollinger, W. E. Cascio and J. J. Lemasters, *Biochem. Biophys. Res. Commun.*, 1997, **236**, 738–742.
- B. Agnew, J. Beechem, K. Gee, R. Haugland, T. Steinberg and W. Patton, *PCT/US 2004/036968*, 2004.
- J. N. Demas and G. A. Crosby, *J. Phys. Chem.*, 1971, **75**, 991–1024.
- R. F. Kubin and A. N. Fletcher, *J. Lumin.*, 1982, **27**, 455–462.
- R. Pethig, M. Kuhn, R. Payne, E. Adler, T. H. Chen and L. F. Jaffe, *Cell Calcium*, 1989, **10**, 491–498.
- G. C. Ellis-Davies, *Nat. Methods*, 2007, **4**, 619–628.
- (a) A. Luckhoff and R. Busse, *J. Cell. Physiol.*, 1986, **126**, 414–420; (b) F. A. Gonzalez, E. Rozengurt and L. A. Heppel, *Proc. Natl. Acad. Sci. U. S. A.*, 1989, **86**, 4530–4534; (c) R. Charest, P. F. Blackmore and J. H. Exton, *J. Biol. Chem.*, 1985, **260**, 5789–5794.
- D. G. Lambert, *Calcium signaling protocols*, Springer Science & Business Media, 2006.
- U. Quast, A. M. Labhardt and V. M. Doyle, *Biochem. Biophys. Res. Commun.*, 1984, **123**, 604–611.
- A. T. Harootunian, J. P. Kao, S. Paranjape and R. Y. Tsien, *Science*, 1991, **251**, 75–78.
- M. B. Cannell and C. Soeller, *J. Gen. Physiol.*, 1999, **113**, 373–376.
- (a) G. T. Dempsey, J. C. Vaughan, K. H. Chen, M. Bates and X. Zhuang, *Nat. Methods*, 2011, **8**, 1027–1036; (b) M. Heilemann, S. van de Linde, M. Schüttelz, R. Kasper, B. Seefeldt, A. Mukherjee, P. Tinnefeld and M. Sauer, *Angew. Chem., Int. Ed.*, 2008, **47**, 6172–6176.

