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Virosome engineering of colloidal particles and surfaces: bioinspired fusion to supported lipid layers†

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Immunostimulating reconstituted influenza virosomes (IRIVs) are liposomes with functional viral envelope glycoproteins: influenza virus hemagglutinin (HA) and neuraminidase intercalated in the phospholipid bilayer. Here we address the fusion of IRIVs to artificial supported lipid membranes assembled on polyelectrolyte multilayers on both colloidal particles and planar substrates. The R18 assay is used to prove the IRIV fusion in dependence of pH, temperature and HA concentration. IRIVs display a pH-dependent fusion mechanism, fusing at low pH in analogy to the influenza virus. The pH dependence is confirmed by the Quartz Crystal Microbalance technique. Atomic Force Microscopy imaging shows that at low pH virosomes are integrated in the supported membrane displaying flattened features and a reduced vertical thickness. Virosome fusion offers a new strategy for transferring biological functions on artificial supported membranes with potential applications in targeted delivery and sensing.

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Introduction

Immunostimulating reconstituted influenza virosomes (IRIVs) are virus like nanoparticles similar to influenza virus envelopes but lacking any viral genetic material.¹ IRIVs are essentially spherical, unilamellar vesicles with a mean diameter of less than 200 nm.^{2–4} Their base is a liposome comprised of phosphatidylcholine (PC), phosphatidylethanolamine (PE) and lipids derived from the influenza virus.⁵ In contrast to liposomes, virosomes contain functional viral envelope glycoproteins: influenza virus hemagglutinin (HA) and neuraminidase (NA) intercalated in the phospholipid bilayer. Thus, reconstituted influenza virosomes retain the receptor binding and membrane fusion activity of the influenza virus.¹ Virosomes have demonstrated to be a versatile and efficient carrier system for a variety of antigens, for example proteins, peptides, nucleic acids and carbohydrates.^{3,6}

Virosomes have been mainly developed as prospective adjuvants to potentiate an immune response, since antigens alone are often poor immunogens.^{1,7} While most adjuvants, including IRIVs, induce a humoral immune response, Kammer *et al.* developed a new generation of influenza virosomes (TIRIVs) that induced both cytotoxic T-cell and humoral responses.¹ This new type of IRIV contains TC- or DC-cholesterol as stabilizers, which cause their possible lyophilization. After reconstitution, the immunogenic properties (*e.g.* membrane fusion activity) are maintained.¹ Furthermore, virosomes have potential applications as drug carriers, like liposomes, with the advantage that the viral glycoprotein hemagglutinin promotes the fusion of the virosome with the cell membrane, which can in turn lead to the release of the virosome cargo directly in the cytoplasm of the cell.⁸ An unexplored domain where virosomes can be applied is their use to engineer colloids and surfaces in the way that liposomes have been often applied for the development of biocompatible surfaces, biosensors or drug delivery systems. Since virosomes retain the fusion properties of virus they could also be fused on artificial lipid membranes bringing the functionalities carried on their membranes to the supported lipids. Colloidal particles engineered with virosomes could have interesting applications in drug delivery besides their obvious importance in vaccine development. Virosome engineered particles could find applications as adjuvants, as influenza virus proteins will be present on the surface of the colloidal particles.^{9–11} An additional advantage of virosomes is that other proteins like antibodies or specific ligands can be integrated into the virosome or attached to the surface of the

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virosomes employing a lipid anchor.^{12,13} These biomolecules would be then transferred to the supported membranes.^{14–16} This represents an alternative to the direct functionalization of supported membranes, which is not always possible once the membrane is formed. Also the use of lipid vesicles entailing proteins and other biomolecules does not always result in the formation of a bilayer as they can affect the assembly, rupture and fusion process of liposomes. For the colloidal particles the presence of proteins in liposomes can also lead to major particle aggregation if the biomolecule prevents charge compensation during an electrostatically driven assembly. Therefore, virosomes could act as transfer agents of biological functionalities to supported membranes in a sequential way through a fusion mechanism without jeopardizing membrane formation and reducing particle aggregation in the case of colloids.

In this manuscript we will show that it is possible to fuse influenza virosomes on supported membranes, both assembled on colloidal particles and planar surfaces. We will prove that the fusion of the influenza virosomes is triggered by low pH, as occurring with the influenza virus.¹⁷ Thereby, the fusion of the virosomes on supported membranes can be used as a model for the virosome fusion on cell membranes.

Despite the fact that the fusion of virosomes has been proven in cell cultures,^{12,18,19} the fusion mechanism and the state of virosomes and glycoproteins after fusion have not been studied in detail. Our results can provide a better understanding of the fusion mechanism of virosomes with membranes that is relevant for drug delivery and vaccine applications.

Virosome fusion has been studied *via* flow cytometry, confocal laser scanning microscopy (CLSM), fluorescence spectroscopy, quartz crystal microbalance with dissipation (QCM-D), atomic force microscopy (AFM) and cryo-transmission electron microscopy (Cryo-TEM). By combining these techniques we have been able to address the influence of pH, time, temperature, and virosome concentration on their fusion into lipid bilayer supported membranes.

Supported membranes were fabricated on self-assembled polyelectrolyte multilayers (PEMs), which were chosen as model supports as they can be assembled on both planar and colloidal interfaces. PEMs were chosen as supports since membranes in cells are also supported by a polymer cushion of glycoproteins. The supported membranes were assembled from the mixed vesicles of DOPS and DOPC, under conditions that ensure the formation of a single lipid bilayer.²⁰

Experimental

Materials and methods

Polyallylamine hydrochloride, (PAH, M_w 15 kDa), polystyrene sulfonate sodium salt, (PSS, M_w 70 kDa), and Rhodamine B octadecyl ester perchlorate were obtained from Sigma-Aldrich. PAH (M_w 15 kDa) labeled with Rhodamine B isothiocyanate (Rh) and PAH (M_w 15 kDa) labeled with Cy5 were kindly

offered from Surflay Nanotec GmbH. The phospholipids 1,2-dioleoyl-*sn*-glycero-3-phosphocholine (DOPC, 10 mg mL⁻¹ in chloroform) and 1,2-dioleoyl-*sn*-glycero-3-phospho-L-serine (DOPS, sodium salt, 10 mg mL⁻¹ in chloroform) were purchased from Avanti Polar Lipids, Inc. SiO₂ particles with a diameter of 3 μ m were purchased from Attenbio. The immunostimulating reconstituted influenza virosomes (IRIVs); 7 mg per mL lipid [DOPC : OPPE in a mass ratio 4 : 1] were prepared with 1.3 mg per mL hemagglutinin of the influenza strain A/Sing/6/86 H1N1. Phosphate buffered saline (PBS), sodium chloride (NaCl), citric acid, sodium phosphate dibasic and chloroform anhydrous (>99%) were purchased from Sigma-Aldrich. Acetic acid (99.8%) was obtained from Sigma Aldrich-Fluka. Absolute ethanol (99.9% HPLC) was obtained from Scharlau S.A.

Polyelectrolyte multilayer (PEM) assembly. Polyelectrolyte multilayers based on poly(allylamine hydrochloride) (PAH) and poly(styrene sulfonate) (PSS) were assembled either on silica particles (3 μ m) or on plane surfaces by means of the Layer-by-Layer (LbL) technique. PEMs were formed upon 11 polyelectrolyte layers ((PAH/PSS)_{5,5}) while the outermost layer was always the positively charged PAH. Polyelectrolytes were assembled in acetate buffer containing 0.2 M NaCl (pH 5.6) at a 1 mg per mL concentration. After each layer assembly the surplus polyelectrolyte was removed by washing three times with water. The process was repeated until eleven polyelectrolyte layers were assembled. The step-wise successful coating with positively charged PAH and negatively charged PSS was proved by measuring the zeta potential at each deposition step (Fig. S1†) and monitored by QCM-D (Fig. S2†).

Lipid layer assembly on planar and colloidal supports. For the fusion of influenza virosomes on supported membranes LbL engineered colloids and planar surfaces were initially equipped with a continuous fluid lipid bilayer. The protocol for lipid coating was optimized regarding lipid composition. The lipid layer is assembled from lipid vesicles, which are adsorbing, spreading, and fusing among themselves to form a continuous lipid bilayer on the LbL coated colloids.²⁰

Small unilamellar lipid vesicles (SUVs) composed of mixtures of 1,2-dioleoyl-*sn*-glycero-3-phosphatidylcholine (DOPC) and 1,2-dioleoyl-*sn*-glycero-3-phospho-L-serine (DOPS) were prepared as follows: lipid stock solutions (10 mg mL⁻¹ in chloroform) were mixed together at a molar ratio of 40 : 60 (DOPC : DOPS). The chloroform was evaporated under vacuum (1 mbar, room temperature) for at least 1 h. The lipid film was rehydrated immediately with PBS (10 mM, pH 7.4) and the resulting lipid solution was extruded through a polycarbonate membrane with 50 nm diameter pores to form SUVs.

For the assembly of a lipid bilayer, LbL-coated silica particles and planar surfaces were washed three times with PBS. After addition of SUVs (5 mg mL⁻¹ in PBS) the sample was incubated for 15 min at room temperature under continuous shaking. Surplus SUVs were removed by washing three times with PBS.

SUVs and virosomes were characterized regarding the size and zeta-potential. The size was determined after 1 : 100



dilution (v/v) in buffer (50 mM phosphate, 83 mM NaCl; pH 7.4) by dynamic light scattering (DLS) using a Malvern Zeta-Sizer 3000HS and Nanotracking analysis (NTA) (NanoSight 2.3). The zeta potential was measured in 1 mM Tris (pH 7) with the Malvern ZetaSizer 3000HS.

R18 labeling of virosomes. The fusion of virosomes with SUVs or lipid coated colloids was studied *via* the R18 fusion assay. IRIVs were labeled with octadecyl Rhodamine B (R18) by incubating them with a 1 : 10 dilution in PBS from an ethanolic R18 stock solution (3 mg mL⁻¹ in water containing 20% (v/v) ethanol) at room temperature for 1 h under gentle shaking. Due to the excess of R18 (5 mol% R18 of total lipid) and the resulting dye-dye interaction in the supported lipid membrane, the fluorescence of the probe is self-quenched.^{21,22} Surplus R18 was removed by Sephadex G-50 chromatography. When virosome fusion takes place into an unlabeled lipid bilayer the probe will dilute and the fluorescence increases accordingly.²² The fluorescence increase after fusion was measured by fluorescence spectroscopy and flow cytometry.

R18 assay using flow cytometry. Silica colloids coated with (PAH/PSS)_{5.5} and a DOPC/DOPS bilayer (5 wt%) were incubated with R18 labeled virosomes in citric acid/phosphate buffer (pH 4.5 or pH 7.4). The fluorescence of the colloids was recorded using a CyFlow Space, Sysmex Partec flow cytometer. The mean particle fluorescence intensity was measured at a rate of 50–100 colloids per second in the fluorescence channel 4 (λ_{ex} : 532 nm, λ_{em} : 590 ± 25 nm) at different times: 0, 1, 2, 5, 10, 15, and 20 min. The data were analyzed using flowing software (version 2.5.1; P. Terhu).

R18 assay using fluorescence spectroscopy

Virosome-vesicle fusion. R18 labeled virosomes (0.5 mg per mL lipid) were diluted in citric acid/phosphate buffer (pH 4.5 or pH 7.4). The fluorescence of the R18 probe was monitored immediately after virosome mixing with SUVs for 10 min using a Varian Cary50 spectrofluorimeter (λ_{ex} : 532 nm, λ_{em} : 584 nm). After 1 min unlabeled SUVs (0.5 mg mL⁻¹) were added to the virosome dispersion. Finally Triton X-100 (1% (v/v)) was added to derive the maximum R18 fluorescence value. Due to a small self-fluorescence of Triton X-100, control measurements in the absence of virosomes were performed in buffer and subtracted from the Triton value of the sample.

Virosome fusion with lipid coated colloids. Lipid coated LbL engineered beads (5 wt%) were diluted 1 : 500 in citric acid/phosphate buffer (pH 4.5 or pH 7.4). Control measurements without virosomes were performed. After addition of R18-labeled virosomes (0.9 mg per mL lipid) the sample was incubated at 37 °C in a water bath for different times. The fluorescence of the sample was recorded using a Varian Cary50 Eclipse spectrofluorometer (λ_{ex} : 532 nm, λ_{em} : 584 nm). Triton X-100 (1% (v/v)) was added, mixed and measured to obtain the maximal value of R18 fluorescence.

Confocal fluorescence microscopy. The distribution of the R18 fluorescence on the particle surface was analyzed by confocal fluorescence microscopy (Leica TCS Sp1). Free virosomes were removed by three repeated washing steps. 10 μ L of the sample were diluted in PBS (pH 7.4). Fluorescence was

measured after excitation with a 532 nm laser in the emission range of 570 to 600 nm.

QCM-D measurements. Polyelectrolyte multilayers of PAH and PSS were assembled from 1 mg per mL polyelectrolyte solutions with a concentration of polyelectrolyte in acetate buffer containing 0.2 M NaCl (pH 5.6) inside the E4 QCM-D chamber at 25 °C. A (PAH/PSS)_{5.5} PEM was deposited on top of SiO₂ (50 nm) coated quartz crystals (5 MHz) from Q-Sense. Each polyelectrolyte solution was left for 10 min in the chamber until a stable frequency was reached. Before the injection of the subsequent polyelectrolyte solution the surfaces were rinsed with milli-Q water for 10 min and for another 10 min with acetate buffer containing 0.2 M NaCl. The last deposited layer was PAH (11th layer). Afterwards, PBS (pH 7.4) was added to the chamber and the temperature was increased to 37 °C for the lipid layer deposition. PBS was exchanged with a dispersion of PC : PS (molar ratio 40 : 60) SUVs in PBS (0.1 mg mL⁻¹). The suitable conditions for the formation of a lipid bilayer membrane from a mixture of DOPC and DOPS lipids on top of PAH/PSS PEM have been previously described.^{20,23} When the bilayer was formed and the frequency reached a plateau the membrane was rinsed with PBS to remove surplus vesicles. Subsequently, the chamber was pre-equilibrated with milli-Q water. The QCM-D chamber was filled with CIP buffer (pH 4.5) or PBS (pH 7.4) before injection of the virosomes. Virosomes were added at a concentration of 0.014 μ g mL⁻¹ and left in the chamber for at least 1 h. Finally, the chamber was rinsed with PBS and milli-Q water in order to remove non-fused virosomes from the membrane.

AFM measurements. Atomic force microscopy (AFM) measurements were performed in the liquid state using a Multimode AFM with a Nanoscope V controller (Bruker AXS, Santa Barbara, CA), equipped with a J-scanner. Oxide-sharpened silicon nitride cantilevers with a nominal spring constant of 0.06 N m⁻¹ (Bruker) were used. The QCM-D sensors with the supported membranes and fused virosomes were attached to Teflon-coated metal disks using double-sided tape and placed on the AFM scanner. Images were acquired with a minimal force. Tapping mode images were analyzed using the Gwyddion software (gwyddion.net).

Cryo-TEM imaging. Virosome morphology was characterized by Cryo-TEM with a JEOL JEM-2200FS field emission TEM with a digital camera and an in-column energy filter (Omega filter). Measurements were conducted for IRIVs diluted in PBS at a concentration of 0.35 mg mL⁻¹. The specimen was deposited on Quantifoil grids Holey Carbon Films (shape R2/2). The sample suspended in aqueous solution was then rapidly frozen in liquid ethane and cooled to liquid nitrogen temperature; vitrification was performed on a Vitrobot-FEL.

Results and discussion

Supported membrane assembly

SUVs were prepared as described in the experimental part and characterized regarding the size and zeta potential. The size of



the SUVs was 127.2 ± 2.3 nm and 87 ± 5 nm as measured by DLS and NTA, respectively. The zeta potential of the SUVs was -23.7 ± 7.49 mV in 1 mM Tris (pH 7). The negative zeta potential value is consistent with the presence of DOPS in the vesicles. The SUVs were employed to deposit a lipid bilayer on silica particles covered with a PEM consisting of 11 layers with PAH being the top layer. Cryo-TEM showed that the LbL silica colloids have been successfully coated by a continuous lipid bilayer, which forms the basis for subsequent virosome fusion.²⁰

Fusion of virosomes with PEM supported lipid membranes

Hemagglutinin, the membrane fusion protein of the influenza virus, is activated and triggered by low pH.^{2,3,8,17} Upon exposure to low pH in endosomes, conformational changes of the hemagglutinin protein result in the exposure of the fusion peptide, the hydrophobic N-terminus of the HA2 polypeptide subunit.^{24,25} The hydrophobic peptide is inserted into the hydrocarbon part of the cell membrane.^{8,25,26} At pH 7.4 the virosomes remain attached to the surface of the membrane, since hemagglutinin is not triggered to initiate fusion with the lipid bilayer. Since the pH is important for the influenza virus for the fusion with lipid membranes, the fusion activity of the virosomes was tested on supported lipid membranes at acidic and neutral pH. Furthermore the best conditions for virosome fusion were established by evaluation of the temperature and virosome concentration.

At first, IRIVs and free protein liposomes with the same composition and lipid content of the IRIVs, without hemagglutinin, were characterized in liquid solution regarding size and zeta potential (Fig. 1). The PDI of 0.096 (IRIVs) demonstrates that no aggregation or degradation takes place.

The fusion of the virosomes was studied by means of the fluorescence self-quenching R18 assay. The assay is based on the incorporation of the lipophilic Rhodamin B- octadecylester-perchlorate (R18) into lipid membranes taking advantage of the alkyl chain of the dye conjugate. Virosomes were labeled by adding a R18 solution in ethanol to the virosome suspension at such a concentration that the R18 fluorescence becomes self-quenched as a result of the dye-dye interaction. The R18 quenching was about 50–70% (Fig. S3†).

The fusion of R18-labeled virosomes with SUVs was monitored by fluorescence spectroscopy measurements. Fig. 2 shows that the virosome–liposome fusion depended on the pH value. At pH 4.5 the fluorescence intensity increased after addition of SUVs reaching a plateau after approximately 3 min. At pH 7.4 the fluorescence intensity did not increase after addition of virosomes, which indicates that at neutral pH virosome–liposome fusion did not occur. Control studies with protein free liposomes with otherwise identical lipid composition and concentration to IRIVs showed no increase of fluorescence with time after their addition to the SUVs for both pH 4.5 and pH 7.4.

The fusion of R18-labeled IRIVs with lipid-coated LbL particles was studied by flow cytometry. Fig. 3 shows that fusion was strongly dependent on pH. The mean fluorescence intensity was measured as a function of time and pH. At pH 4.5 the R18 fluorescence intensity increased over 20 min by more than a factor of two, while at pH 7.4 the fluorescence remained approximately constant over this period of time. Further experiments showed that the fusion of virosomes with lipid coated beads was completed within about 30 min at pH 4.5. The fusion of virosomes with the supported lipid layer on the beads proceeded much slower than the virosome–vesicle

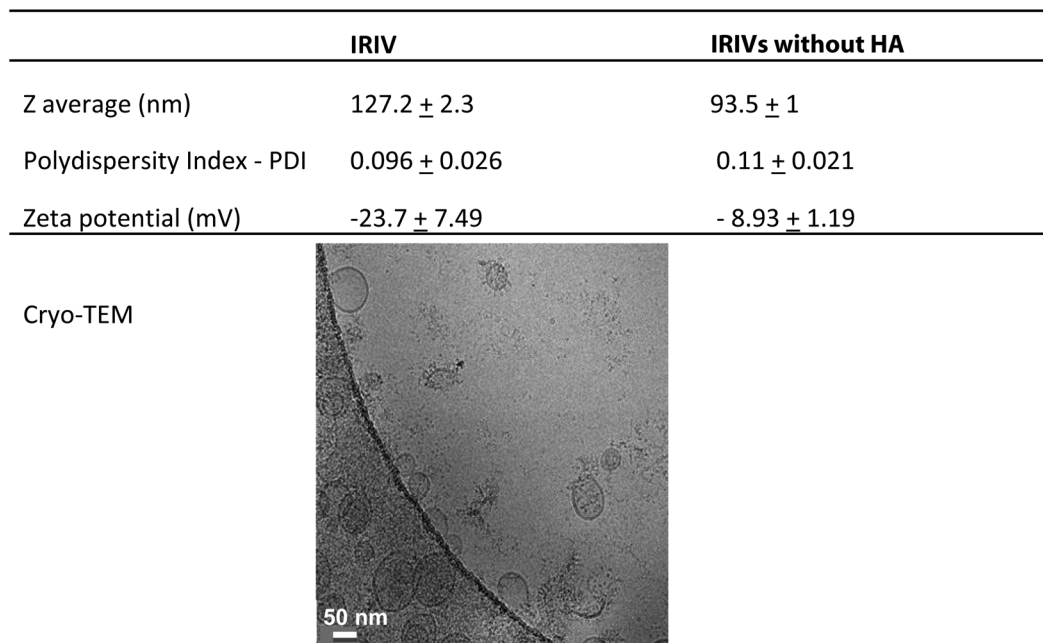


Fig. 1 Particle size, zeta-potential and Cryo-TEM image of IRIVs and protein free equivalent liposomes.



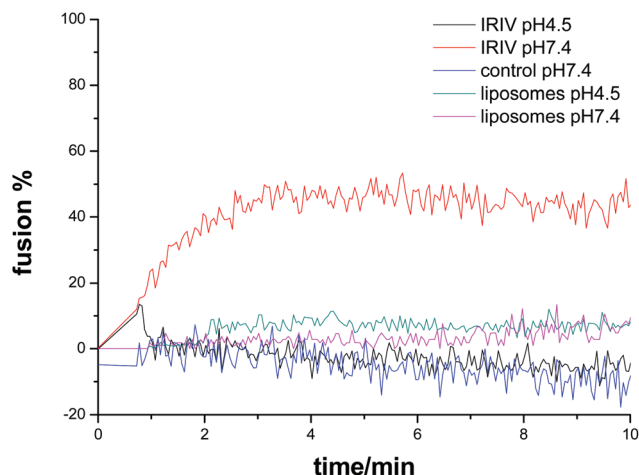


Fig. 2 Kinetic analysis of the virosome fusion with liposomes. SUVs were incubated in citric acid/phosphate buffer with pH 4.5 or pH 7.4. After 1 min R18-labeled IRIVs or R18-labelled free protein liposomes analogous to IRIV (0.5 mg per mL of total lipid) were added. A sample without addition of R18-labeled virosomes was used as a control. The fluorescence intensity (λ_{ex} : 532 nm, λ_{em} : 584 nm) was recorded every 0.05 min. Fusion% was calculated with $(F - F_0)/F_{\text{max}} \times 100$. F is the fluorescence intensity after addition of virosomes or protein free liposomes or PBS (control). F_0 is the fluorescence intensity before addition of virosomes. F_{max} is the maximal fluorescence intensity after lysis with Triton X-100.

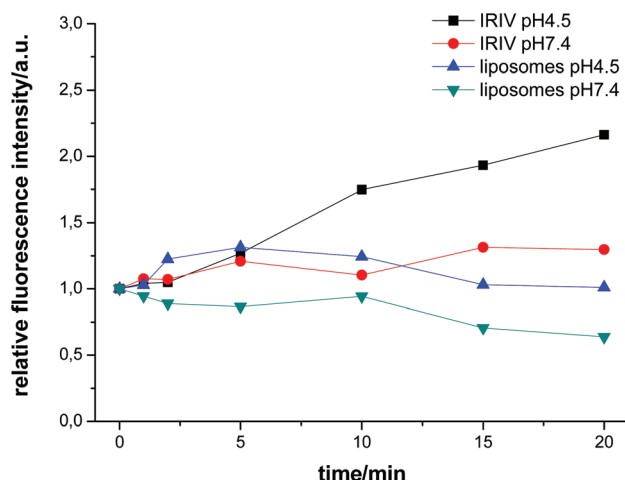


Fig. 3 Kinetic analysis of the virosome fusion with lipid-coated LbL-silica particles. Lipid-coated LbL-colloids were incubated in citric acid/phosphate buffer with pH 4.5 or pH 7.4 before addition of R18-labeled IRIVs or R18-labeled liposomes of IRIV. The mean fluorescence intensity distribution relative to the mean fluorescence intensity of the start value (0 min) is plotted as a function of the time. The fluorescence intensity (λ_{ex} : 532 nm, λ_{em} in FL4) was measured after different times (0, 1, 2, 5, 10, 15, 20 min) by flow cytometry ($n = 200$ –500 events).

fusion. We attribute this difference to the restricted mobility and flexibility of the supported membrane compared with the free membrane of the vesicles. The different lipid surface ratios of liposomes and lipid-coated LbL-colloids may also play

a role. Control measurements with protein free liposomes at both pH values showed also no increase of fluorescence with time, confirming that the presence of HA is responsible for the virosome fusion.

Virosome fusion with lipid-coated LbL-colloids was studied as a function of temperature. Fig. 4 shows that after 10 min the fluorescence intensity at 37 °C was only slightly larger than at 21 °C. The fluorescence intensity increases from 0 min (virosome addition) to 10 min by 2 times for 21 °C and 1.7 times for 37 °C. However, after lysis with Triton X-100 the R18 fluorescence intensity was much larger at 21 °C than at 37 °C. This indicates that a comparatively large amount of virosomes at 21 °C was still adsorbed on the lipid coated particles rather than fused within the lipid bilayer. This finding demonstrates that the temperature increase facilitates fusion as demonstrated by Haywood *et al.* for the fusion of the influenza virus with liposomes.²⁷

Furthermore, the effect of virosome concentration on the membrane fusion was studied. In Fig. 5 two different IRIV concentrations are compared. The virosome concentration, given in the final amount of hemagglutinin per mL, was 0.5 μg and 2 μg HA. At pH 4.5 the fluorescence increases 2 times for 0.5 μg HA and one time for 2 μg HA after 10 min incubation to virosome addition at 0 min. This represents a more efficient fusion for the lower virosome concentration to lipid-coated LbL silica particles. At pH 4.5, after lysis with Triton X-100 the R18-fluorescence intensity increased 14 times for 2 μg hemagglutinin, a significantly larger increase than the observed approximately 4 times increase for 0.5 μg hemagglutinin (Fig. 5a and b). The pronounced difference in the fusion efficiency between 0.5 and 2 μg hemagglutinin suggests that in the latter case a large amount of virosomes remained attached to the supported membrane instead of fusing. It may also be that being the concentration of R18 significantly higher for the

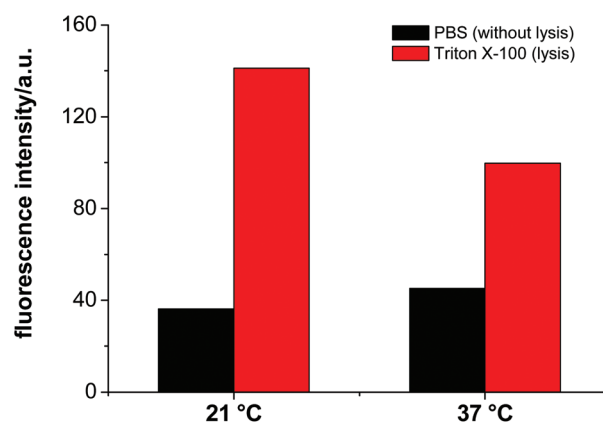


Fig. 4 Fusion of IRIVs with lipid-coated LbL-silica particles in dependence of the temperature measured as the R18 fluorescence intensity. Lipid-coated LbL-silica particles were incubated with R18-labeled IRIVs (0.5 μg hemagglutinin) in citric acid/phosphate buffer with pH 4.5 at room temperature (21 °C) or 37 °C for 10 min. The fluorescence intensity (λ_{ex} : 532 nm, λ_{em} : 584 nm) before and after lysis by addition of Triton X-100 was measured.



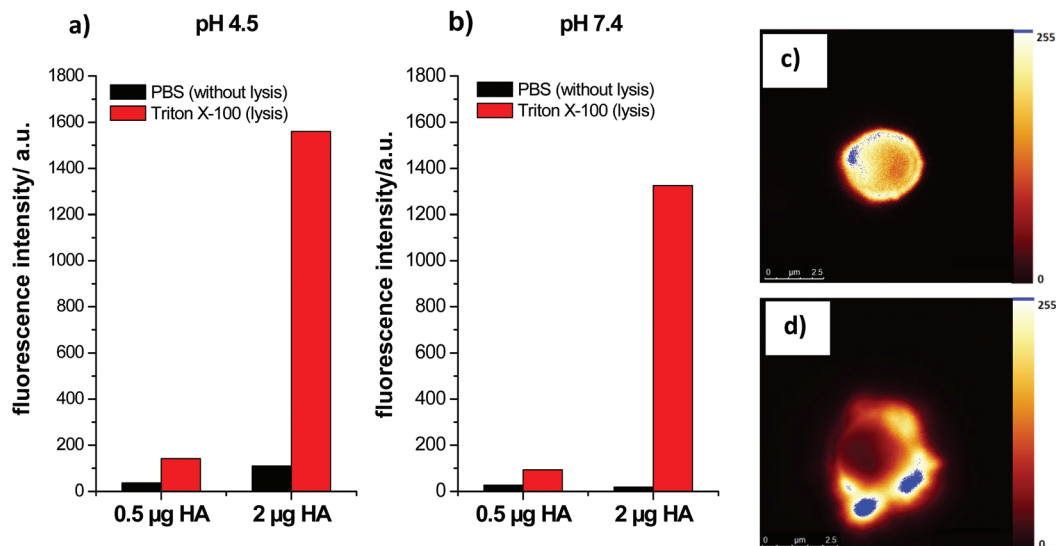


Fig. 5 Fusion of IRIVs with lipid-coated LbL-silica particles in dependence of hemagglutinin concentration. R18 fluorescence intensity (λ_{ex} : 532 nm, λ_{em} : 584 nm) was measured before and after lysis with Triton X-100 for 0.5 and 2 μg hemagglutinin at (a) pH 4.5 and (b) pH 7.4. Confocal images (c) and (d) corresponding to (a) and (b), respectively. Lipid-coated LbL-silica particles were incubated with R18-labeled IRIVs concentrations either 0.5 μg or 2 μg hemagglutinin in citric acid/phosphate buffer with pH 4.5 at 37 $^{\circ}\text{C}$ for 10 min. The color bar indicates the R18-fluorescence intensity from a value of 0 (black) to 255 (blue).

case with the larger concentration of virosomes their fusion does not cause a complete dequenching of the R18. However, the confocal fluorescence microscopy image (Fig. 5d) shows a rather inhomogeneous fluorescence distribution thus confirming the adsorption of virosomes to the supported membrane. It may be that if the lipid membrane is densely covered with virosomes, fusion becomes inhibited as a consequence of topological constraints. In contrast for 0.5 μg hemagglutinin the fluorescence distribution is homogenous indicating the virosome fusion to the supported membrane (Fig. 5c).

Virosome fusion on planar supported membranes

Virosome adsorption and fusion was also studied by means of QCM-D on planar supported membranes. Fig. 6 shows the results from the real time monitoring of fusion of the IRIVs for the two different pH values 4.5 and 7.4. The QCM-D sensor is

sensitive to surface mass; the frequency decrease is related to the increase of the mass deposited on the sensor. As referred to in the Experimental part, a lipid bilayer was deposited on top of the 11 layers of PAH/PSS cushion prior to virosome deposition. Fig. 6a shows the frequency and dissipation changes occurring upon interaction of the IRIVs diluted in CIP buffer with the supported membrane at pH 4.5 as a function of time. The frequency followed a steadily decreasing tendency for more than two hours reaching a total frequency shift of $\Delta f = -99$ Hz. This comparatively large frequency shift demonstrates an ongoing adsorption of IRIVs on the membrane indicating that at this pH the HA protein in the IRIVs is able to recognize the lipid membrane. The strong parallel increase in dissipation confirms adsorption. The increase in dissipation is also indicative of the increase of the surface roughness during the attachment of the virosomes to the membrane leading to

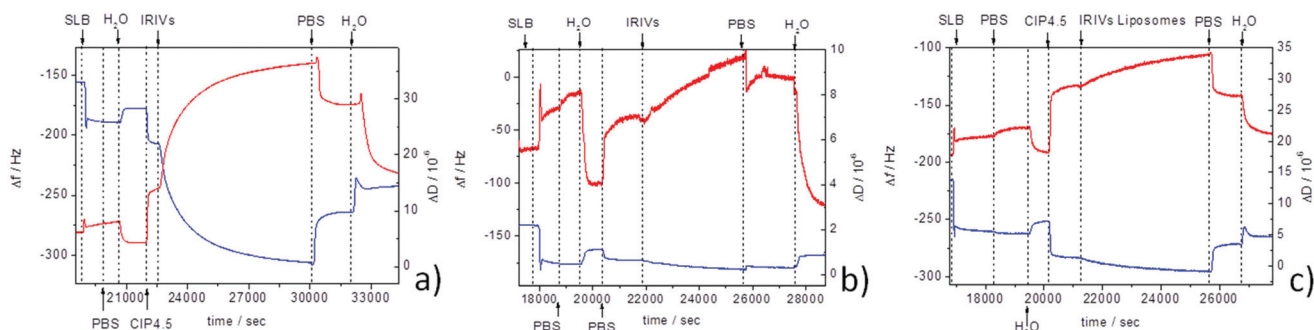


Fig. 6 Frequency (blue) and dissipation (red) curves measured by QCMD for the formation of a phospholipid bilayer deposited on a PEM of 11 layer of PAH/PSS and later exposed to (a) IRIVs in acidic pH (CIP 4.5), (b) IRIVs in neutral pH (PBS 7.4) and (c) protein free liposomes with the same lipid composition and content as IRIVs in acidic pH (CIP 4.5).



an enhanced friction of the soft virosomes. At this stage the QCM-D data do not allow the separation of fusion and adsorption. However, after rinsing with PBS non-fused IRIVs were removed as seen by the decrease in dissipation and the increase in frequency reaching a $\Delta f = -46$ Hz. The frequency increased even further to $\Delta f = -21$ Hz when rinsing with water. So finally the total frequency shift for the fused IRIVs in the supported membrane at acidic pH was $\Delta f = -32$ Hz. We attribute the frequency change after rinsing with PBS and water to the net mass increase caused by virosome membrane fusion. At neutral pH the QCM-D data demonstrate a completely different behavior as shown in Fig. 6b. The frequency changes only slightly upon the deposition of the IRIVs. The small increase in dissipation, note the different scale in Fig. 6a and b, is consistent with the adsorption of a small number of virosomes, which do not subsequently fuse with the lipid membrane. This was followed by the small total change in frequency $\Delta f = -8$ Hz after rinsing.

Liposomes analogous to the IRIV, without HA proteins, did not fuse at pH 4.5 on the supported membranes (Fig. 6c) confirming again that the effect of the pH is only related to the hemagglutinin.

AFM imaging was conducted to have visual proof of the virosome fusion to the lipid membrane. In Fig. 7a the fused IRIVs at pH 4.5 can be recognized as they form features of a varying size of 190–270 nm, which is about twice as much as the diameter of the IRIVs. These features were not present in the supported membrane before the exposure to the virosomes. The observed features have a height of 50–70 nm. These values are consistent with the measured size of the IRIVs of about 115 nm. From the geometrical considerations it is obvious that after fusion of the IRIVs with the lipid

membrane the latter has to be lifted off in the regions of the fusion events, as fusion goes along with a net increase of the total membrane surface. From this it follows that the virosome membranes were embedded in the supported membrane, while the enclosed volume was released during fusion. The fused virosomes appear to be aggregated and randomly distributed on the surface. The control at neutral pH is more revealing. Fig. 7b shows features of a height of ~ 100 nm and lateral dimensions of 150–350 nm, which correspond to non-fused IRIVs displaying a size comparable to the one obtained by DLS and Cryo-TEM measurements for the IRIVs in bulk. The IRIVs at this pH appear as if they remained attached on the supported membrane without fusing. AFM was also conducted for supported membranes exposed to protein free liposomes without HA, at pH 4.5. In this case the lipid surface displays the same features as the supported membranes prior to the exposure to the liposomes but with a few large liposome aggregates on top with a height of 120–130 nm and a width of ~ 500 nm (Fig. 7c). The extended aggregation of the liposomes could be due to the lack of proteins. No hints of fused liposomes could be observed.

The virosome fusion is based on their interaction with the lipid membrane driven by the hemagglutinin (HA) protein present on the surface of the virosomes, which is responsible for the pH dependent fusion. Fig. 8a displays a scheme of the behaviour of the virosomes at neutral pH where they remain on the membrane without fusing, while at low pH they fuse (Fig. 8b and c). In the figure the change in the conformation of the HA with the hydrophobic peptide penetrating in the hydrocarbon part of the cell membrane has been sketched (Fig. 8b).^{8,25,26} In Fig. 8c the possible arrangement of the lipids and proteins after the fusion is shown. It is remarkable

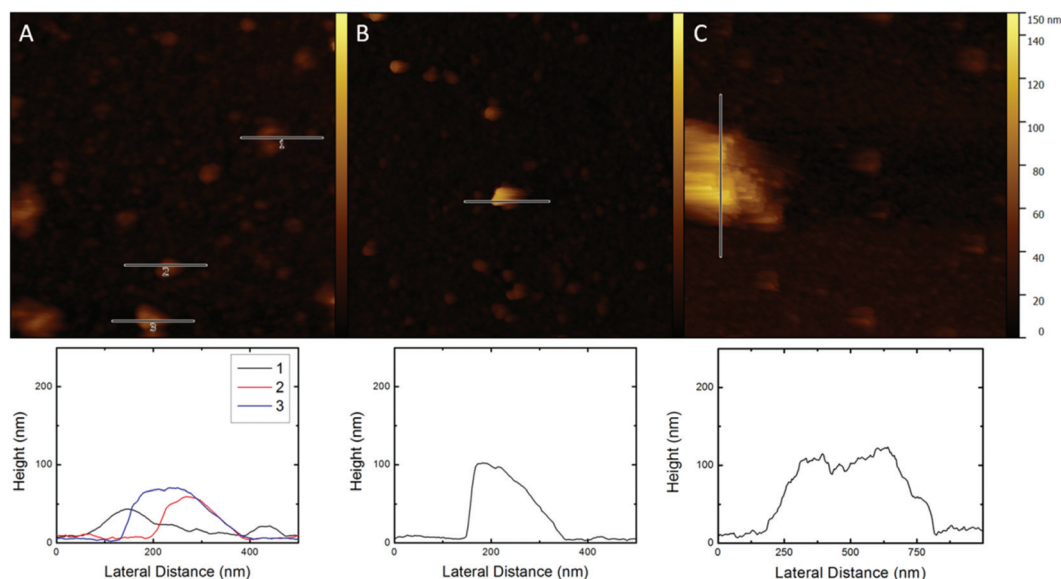


Fig. 7 AFM images ($2 \times 2 \mu\text{m}$) acquired in 1.5 mM NaCl (liquid mode) of (PAH/PSS)_{5.5} PC : PS lipid bilayer membrane after addition of a) IRIVs in pH 4.5 or b) IRIVs in pH 7.4 and c) protein free liposomes in pH 4.5. Bottom panels display roughness profiles taken as cross sections of the above images.



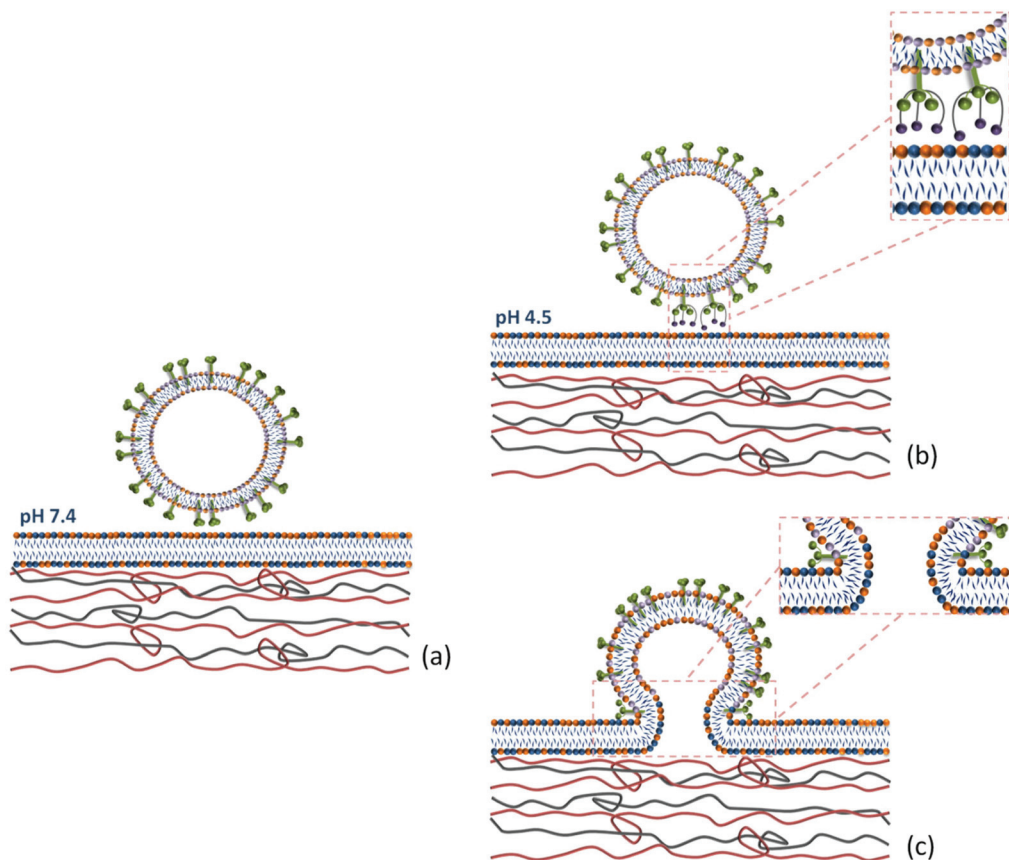


Fig. 8 Schematic illustration of the (a) virosome attached to a supported membrane at neutral pH without fusing; and (b) and (c) fusion of the virosomes at low pH on the supported membrane. The virosome makes contact with the lipid membrane and there is a rearrangement of the hemagglutinin protein at a low pH of 4.5 (b), the zoom out image shows in more detail the rearrangement of the hemagglutinin protein. The virosome is fused to the lipid bilayer membrane (c), the zoom out image shows how the proteins are arranged at the place where the fusion occurs.

that the fusion of virosomes is taking place in the absence of a sialoglycan or any other receptor on the membrane that promotes the interaction of the virosome with the membrane. The fact that our supported membranes are formed by lipids only lacking additional components of cell membranes like the glycocalyx or proteins could explain that the virosomes can fuse directly on the membranes being the lipid layers easily accessible for the virosomes to fuse triggered by the low pH when they get in close vicinity with the supported membrane.

Conclusions

We have shown that influenza virosomes carrying hemagglutinin can fuse on supported membranes on colloids as well as on planar surfaces. The mechanism of fusion is pH dependent as has been confirmed independently by the R18 assay, QCM-D and AFM. Virosomes fuse at low pH 4.5, but not at neutral pH 7.4. AFM imaging proves that the virosomes are integrated in the supported membrane at low pH as they can be recognized in the membrane as flattened features and have a reduced vertical thickness. Virosome fusion on supported membranes offers a novel strategy for the functionalization of

supported membranes that can be used for the design of complex colloidal systems which can have applications in drug delivery or sensing devices.

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