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A potent trifluoromethyl ketone histone deacetylase inhibitor exhibits class-dependent mechanism of action

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Histone deacetylase (HDAC) enzymes are validated targets for treatment of certain cancers and have potential as targets for pharmacological intervention in a number of other diseases. Thus, inhibitors of these enzymes have received considerable attention, but these are often evaluated by IC₅₀ value determination, which may vary significantly depending on assay conditions. In this work, we therefore performed detailed kinetic evaluation of inhibitors containing two fundamentally different zinc-binding chemotypes, hydroxamic acid or trifluoromethyl ketone. For the hydroxamic acids, a fast-on-fast-off mechanism was observed, but the trifluoromethyl ketone compound exhibited differential mechanisms depending on the enzyme isoform. The trifluoromethyl ketone compound displayed a fast-on-fast-off mechanism against class-IIa HDACs 4 and 7, but slow-binding mechanisms against class-I and class-IIb enzymes (HDACs 1–3, 6 and 8). Furthermore, different competitive, slow-binding mechanisms were observed for HDACs 1, 2, and 6 vs. HDACs 3 and 8, demonstrating the power of kinetic experiments for characterisation of enzyme inhibitors.

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Introduction

The Zn²⁺-dependent histone deacetylase (HDAC) enzymes catalyse hydrolytic removal of acetyl functionalities from ε-amino groups of lysine residues in a variety of proteins including histones.¹ Posttranslational modification (PTM) plays major roles in regulation of protein function, and histone acetylation affects chromatin packing and recruitment of transcription factors resulting in effects on gene expression.^{2,3} Recent discoveries have underpinned lysine acetylation as a general PTM in proteins,⁴ and a growing list of non-histone proteins are identified as substrates for the HDACs,^{5–8} thereby extending their potential impact on cellular function.^{2,3,9,10} Eleven HDAC isoforms are encoded by the human genome, classified according to sequence similarity:¹¹ class-I (HDACs 1–3 and 8), class-IIa (HDACs 4, 5, 7 and 9), class-IIb (HDACs 6 and 10), and class-IV (HDAC11).

Due to the importance of HDACs in a wide variety of disease states,^{12–15} selective HDAC inhibitors have potential as tool compounds and possibly as novel drugs. Several chemotypes have been described as HDAC inhibitors with the most extensively studied class being inhibitors featuring the strongly zinc-chelating hydroxamic acid moiety,^{16,17} including vorinostat (SAHA, 1) and trichostatin A (2) (Fig. 1).

Due to the strong metal ion-chelating propensity of hydroxamic acids, inhibitors containing this functionality have been suspected to give rise to off-target effects, but recent experiments have indicated that this may be of minor concern.^{18,19} Nevertheless, hydroxamic acids have also been associated with rapid degradation and clearance, and investigation of other zinc-binding functionalities, such as thiols (as in romidepsin (3)),²⁰ trifluoromethyl ketones (4),^{21–24} and *o*-aminoanilides (5)^{25,26} (Fig. 1), may therefore be of interest.

The binding mode of hydroxamic acids, which are generally considered potent class-I, -IIb and -IV inhibitors, but much poorer class-IIa inhibitors,²⁷ is through bidentate coordination to the active site Zn²⁺ ion *via* the carbonyl oxygen and deprotonated nitrogen-bound oxygen atom.²⁸

Trifluoromethyl ketones constitute another strongly zinc-binding functionality. This motif is readily hydrated in aqueous solution²⁹ and presumably binds to the active site Zn²⁺ ion *via* bidentate coordination from the two resulting geminal hydroxy groups.³⁰ Although trifluoromethyl ketones have also been reported to be metabolically labile by reduction to the corresponding alcohol *in vivo*,²¹ they may give rise to interesting tool compounds for investigation of HDAC inhibition *in vitro*. Interestingly, compounds containing this zinc-binding group have been reported to selectively inhibit class-IIa HDAC isoforms;^{22,23,31} however, class-IIa inhibition does not appear to be an inherent capacity of the functional group itself.³² We recently reinvestigated the inhibitory properties of a known trifluoromethyl ketone compound (4)²¹ and

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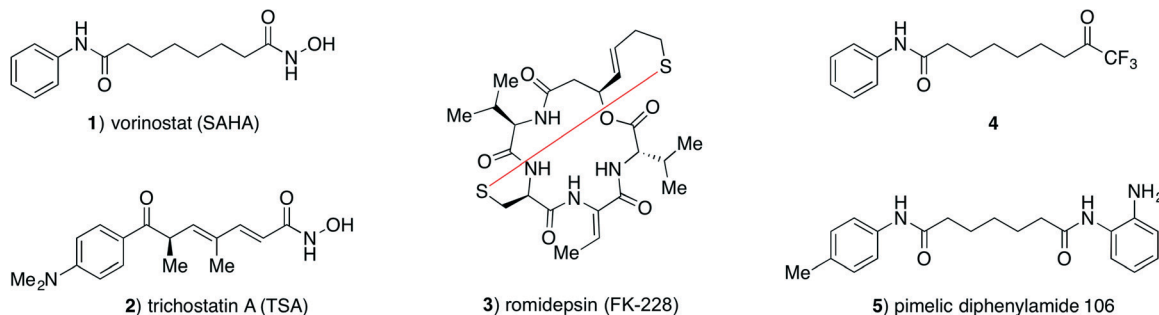


Fig. 1 Structures of selected HDAC inhibitors.

extended the previous studies to characterization of its inhibitory effect against all eleven zinc-dependent HDACs,³² thereby identifying compound 4 as a promising inhibitor of isozymes from all four HDAC classes.³² In our previous study, IC_{50} and K_i values were determined from standard endpoint dose–response assays assuming a fast-on–fast-off binding mechanism. In these experiments, the HDAC enzyme and fluorogenic substrate are incubated with varying concentrations of inhibitor for 1 h, followed by assay development by addition of trypsin to achieve fluorophore release. Thus, a constant rate over the entire incubation period has to be assumed in order to apply the Cheng–Prusoff equation for approximation of the corresponding K_i values.

While endpoint assays are convenient for screening large numbers of inhibitors, more information can be obtained from kinetic inhibition experiments, where progression of enzyme activity is monitored over time. By applying such assays, Gottesfeld and co-workers demonstrated that pimelic diphenylamide 106 (5), an *o*-aminoanilide based class-I HDAC inhibitor, inhibits HDACs 1–3 *via* different mechanisms not revealed by simple IC_{50} determination. They were instead able to show that compound 5 is a slow, tight-binding inhibitor of HDACs 1–3, but acting *via* a different mechanism for HDACs 1 and 2 compared to HDAC3.²⁶ More recently, two groups have also reported on the development of kinetically selective HDAC2 inhibitors based on *o*-aminoanilide moieties, and considering differences in the active sites of HDACs 1–3.^{33,34}

Interestingly, trifluoromethyl ketones have previously been described as slow-binding inhibitors for serine proteases,^{29,35,36} and we therefore found it interesting to perform a more elaborate investigation of the inhibitory mechanism of compound 4 on the different HDAC isoforms. We have previously used continuous rate experiments for investigating inhibitors of the NAD^+ -dependent deacetylase sirtuin 5³⁷ as well as cyclic peptide HDAC inhibitors³⁸ and thought we could modify those protocols for evaluation of compound 4 in continuous assays against a selection of the zinc-dependent HDACs.

Not surprisingly, our investigation showed that the hydroxamic acid-containing inhibitors acted *via* a fast-on–fast-off binding mechanism against all the tested enzymes.

For the trifluoromethyl ketone inhibitor, on the other hand, the mechanism of inhibition proved to vary between enzyme isoforms. Thus, this study provides valuable mechanistic insight for future medicinal chemistry efforts involving HDAC inhibitors containing this zinc-binding group.

Results and discussion

Initially, we investigated the influence of preincubation of HDAC3 and trifluoromethyl ketone 4 on the determined IC_{50} values and found a 17-fold increase in apparent inhibitor potency at 2 h of preincubation time (Fig. 2). Further incubation lead to substantial decrease in enzyme activity, even without addition of inhibitor, thus precluding prolonged preincubation due to enzyme degradation in the assay buffer. This initial observation prompted us to investigate the kinetics of HDAC inhibition in more detail.

Usually, the equilibrium of inhibitor binding to enzyme is assumed to be rapidly obtained, *i.e.*, fast on and off rates for the enzyme–inhibitor equilibrium, which is therefore obtained prior to onset of the experiment. For this fast-on–fast-off mechanism, linear progression curves are observed when monitoring product formation continuously, and the enzyme–inhibitor equilibrium constant (K_i) can be determined by fitting a secondary plot of the relative rates (v_i/v_o) to eqn (1), where v_i and v_o are the observed rates with and without inhibitor, respectively.

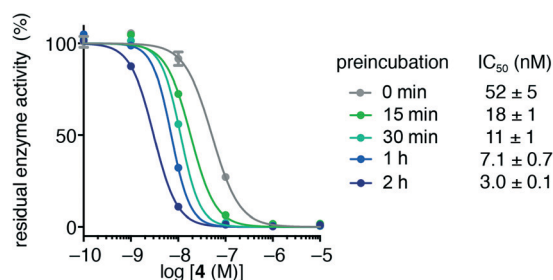


Fig. 2 Residual activity of HDAC3 after preincubation with trifluoromethyl ketone 4. Preincubation times and apparent IC_{50} values are shown.



$$\frac{v_i}{v_o} = \left(\frac{[I]}{K_i \left(1 + \frac{[S]}{K_m} \right)} + 1 \right)^{-1} \quad (1)$$

However, if the enzyme–inhibitor equilibrium is slow and thus not obtained prior to onset of the experiment, non-linear progression curves will be obtained. In such an event, the slow, competitive binding of inhibitor may be divided into two mechanisms as shown in Fig. 3. In mechanism A, the enzyme–inhibitor complex forms without kinetically observable intermediates, whereas in mechanism B, an equilibrium is rapidly established with an initial complex, which is then converted to a second stable complex displaying rate-determining kinetics.

For both types of slow binding kinetics, product formation follows eqn (2), where [P] is the concentration of product at time t . The v_o and v_s are the initial and final steady-state velocities, respectively, and k_{obs} is the apparent first-order rate constant for establishment of the enzyme–inhibitor equilibrium. It should be noted, that for the fast-on-fast-off mechanism, [P] = $v_s t$ is the limiting case of mechanism A when k_{obs} is very large.

$$[P] = v_s t + \frac{v_o - v_s}{k_{obs}} (1 - e^{-k_{obs} t}) \quad (2)$$

For mechanisms A and B, the relationship between k_{obs} and [I] is linear (eqn (3)) and hyperbolic (eqn (4)), respectively. As a result, fitting of these equations to secondary plots of the k_{obs} –[I] relationship allows determination of the binding mechanism, as well as calculation of the rate constants involved and the resulting enzyme–inhibitor equilibrium constants [K_i for mechanism A (eqn (5)) and $K_{i,1}$ and K_i for mechanism B (eqn (6) and (7), respectively)].

Slow binding, mechanism A

$$k_{obs} = k_1 \left(1 + \frac{[S]}{K_m} \right) [I] + k_{-1} \quad (3)$$

$$K_i = \frac{k_{-1}}{k_1} \quad (5)$$

Slow binding, mechanism B

$$k_{obs} = \frac{k_2}{[I] + \frac{k_{-1}}{k_1} \left(1 + \frac{[S]}{K_m} \right)} [I] + k_{-2} \quad (4)$$

$$K_{i,1} = \frac{k_{-1}}{k_1} \quad (6)$$

$$K_i = \frac{k_{-1} k_{-2}}{k_1 (k_2 + k_{-2})} = K_{i,1} \frac{k_{-2}}{(k_2 + k_{-2})} \quad (7)$$

To determine the inhibitory mechanism and associated kinetic values, we initially established assay conditions that

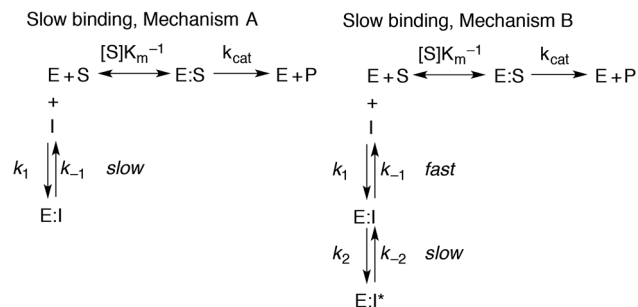


Fig. 3 Mechanism of slow inhibitor binding with the corresponding rate constants.

allowed monitoring of enzyme activity for 60 min. Progression curves were generated by incubating the enzyme–inhibitor combination with an appropriate fluorogenic substrate at concentrations greater than the K_m [Ac-LGK_{ac}-AMC (20 μ M) for HDACs 1–3 and 6 (K_m values 6 μ M, 3 μ M, 6 μ M, and 16 μ M, respectively), Ac-LGK_{tra}-AMC (50 μ M) for HDACs 4 and 7 (K_m values 12 μ M and 26 μ M, respectively), and Ac-RHK_{ac}-AMC (400 μ M) for HDAC8 (K_m value 430 μ M)]. The trypsin concentration was optimised to be high enough to ensure rapid establishment of steady state by cleaving the AMC fluorophore from the deacylated peptide intermediate at a rate that avoids lag time. At the same time, the trypsin concentration should be low enough to ensure a linear progress curve for 60 min without observable reduction of HDAC activity in the experiment without inhibitor added. Previously, Gottesfeld and co-workers reported low stability of HDAC8 under the employed conditions.²⁶ In agreement with this, we also observed rapid loss of activity of HDAC8 when incubated in the standard Tris buffer. However, changing the buffer to either the buffer marketed for HDAC8 (a Tris buffer containing PEG_{8kDa}) or the buffer described by Bradner, Mazitschek and co-workers (a HEPES buffer, data not shown),²⁷ allowed monitoring of HDAC8 activity for 60 min as well.

Having established suitable conditions for all the investigated enzymes, a series of progress curves were generated for HDACs 1–4 and 6–8 against the three investigated inhibitors (1, 2 and 4).

SAHA (1) has previously been demonstrated to be a poor class-IIa inhibitor³² and was not included for HDACs 4 and 7. Instead, TSA was chosen as the hydroxamic acid inhibitor against class-IIa HDACs and HDAC3 was chosen as the representative for TSA inhibition of class-I HDACs. All seven HDACs were included for investigation of compound 4.

Hydroxamate inhibitors such as SAHA have been reported to display fast-on-fast-off kinetics against class-I HDACs,²⁶ and in agreement with these reports, linear progress curves were obtained for inhibition of HDACs 1–3 and 8 as well as the class-IIb enzyme HDAC6 when monitored for 1 h (Fig. 4). Low nanomolar inhibition constants were observed for all investigated enzymes except for HDAC8 (230 \pm 20 nM; Table 1), and all values are in agreement with the literature.²⁷ Similarly, linear progress curves were obtained for TSA inhibition



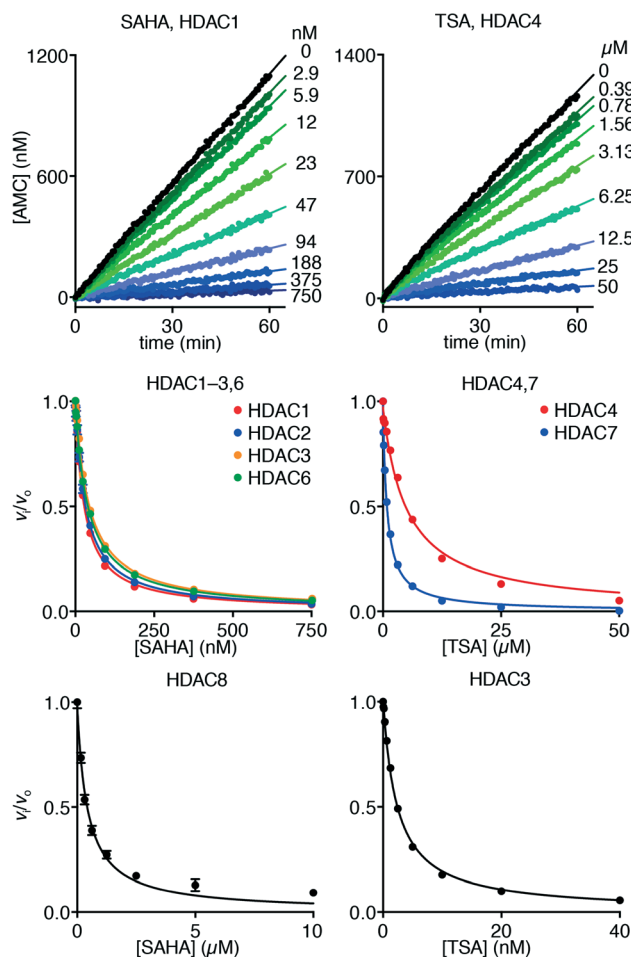


Fig. 4 Progress curves of HDAC1 (LGK_{ac}) and HDAC4 (LGK_{tra}) deacylase activities under SAHA and TSA inhibition, respectively. Relative rates (v_i/v_0), fitted to eqn (1), for HDACs 1–3 and 6 (LGK_{ac}), and HDAC8 (RHK_{ac}K_{ac}) against concentrations of SAHA and HDACs 3, 4 and 7 (LGK_{tra}) against concentrations of TSA.

of all investigated enzymes, including class-IIa enzymes HDAC4 and HDAC7 (Fig. 4). Despite a marked span of potencies of almost 2000-fold (0.54 ± 0.02 nM vs. 980 ± 80 nM for HDACs 3 and 4, respectively), these results indicate that fast-on–fast-off kinetics is general for hydroxamate mediated HDAC inhibition regardless of enzyme class and potency.

Interestingly, trifluoromethyl ketone 4 displayed a differential inhibition mechanism depending on enzyme class. Thus compound 4 was found to be a potent inhibitor of class-IIa enzymes displaying a fast-on–fast-off mechanism with K_i values of 5.1 ± 0.2 nM and 4.7 ± 0.1 nM against HDACs 4 and 7, respectively (Fig. 5).

In contrast, compound 4 displayed slow binding kinetics with non-linear progression curves when incubated with all

the investigated class-I and class-IIb enzymes. The obtained progression curves were fitted to eqn (2) and secondary plots of the calculated k_{obs} values against inhibitor concentration were created (Fig. 5). For HDACs 1, 2 and 6, a linear relationship was observed, consistent with competitive, slow-binding of the inhibitor following mechanism A, and the data were therefore fitted to eqn (3) to determine the kinetic parameters. The overall inhibition constants (K_i) were determined to be 9.8 μ M, 11 μ M and 4.5 μ M, respectively. For HDACs 1 and 6, these are in good agreement with our previous findings using endpoint experiments, while the K_i value against HDAC2 is ~ 30 -fold higher than the previously determined.³² The overall trend that HDACs 4 and 7 are significantly more potently inhibited than HDACs 1, 2 and 6, however, corresponds well with previous findings.

For HDACs 3 and 8, on the other hand, a hyperbolic relationship between k_{obs} and inhibitor concentration was indicated, which is consistent with competitive, slow binding of the inhibitor following mechanism B. Thus, fitting of the data to eqn (4) allowed determination of the accompanying kinetic parameters (Table 2). For HDAC3 the overall inhibition constant (K_i) was estimated to be 11 nM and the overall inhibition constant (K_i) for HDAC8 was estimated to be 33 nM, which is in the range of inhibition constants for class-IIa HDACs 4 and 7. For HDAC3, the K_i value is an order of magnitude lower than previously estimated from endpoint experiments using the Cheng–Prusoff equation,³² which may be explained by a low k_{-2} value indicating a highly stable enzyme–inhibitor complex (E:I*, Fig. 3 mechanism B). This also helps explain the data obtained by preincubation experiments discussed above (Fig. 2), and underlines the power of kinetic evaluation of inhibitors that do not follow standard fast-on–fast-off mechanism of action.

Interestingly, potent inhibition of HDACs 3, 4, 7 and 8 by trifluoromethyl ketone 4 correlates well with the observation that these enzymes all efficiently cleave ϵ -N-trifluoroacetylated lysine substrates.³⁹ Furthermore, the lower affinities for compound 4 against HDACs 1, 2 and 6 also agree with the low detri-fluoroacetylase activities of these enzymes. Taken together, these observations suggest that certain HDACs accommodate the trifluoromethyl group better than others, and that this feature is not in accordance with isoform sub-classification.

Conclusions

In summary, we have evaluated hydroxamate and trifluoromethyl ketone HDAC inhibitors by continuous assays monitoring the progress of substrate conversion over time to enable kinetic insight. We selected enzymes from class-I (HDACs 1–3 and 8), class-IIa (HDACs 4 and 7), and class-IIb (HDAC6), and as

Table 1 K_i values (nM) for HDAC inhibition by SAHA (1) and TSA (2)

	HDAC1	HDAC2	HDAC3	HDAC8	HDAC6	HDAC4	HDAC7
1	6.4 ± 0.2	4.2 ± 0.1	9.9 ± 0.3	230 ± 20	17.7 ± 0.3	n.d.	n.d.
2	n.d.	n.d.	0.54 ± 0.02	200 ± 20	n.d.	980 ± 80	300 ± 10



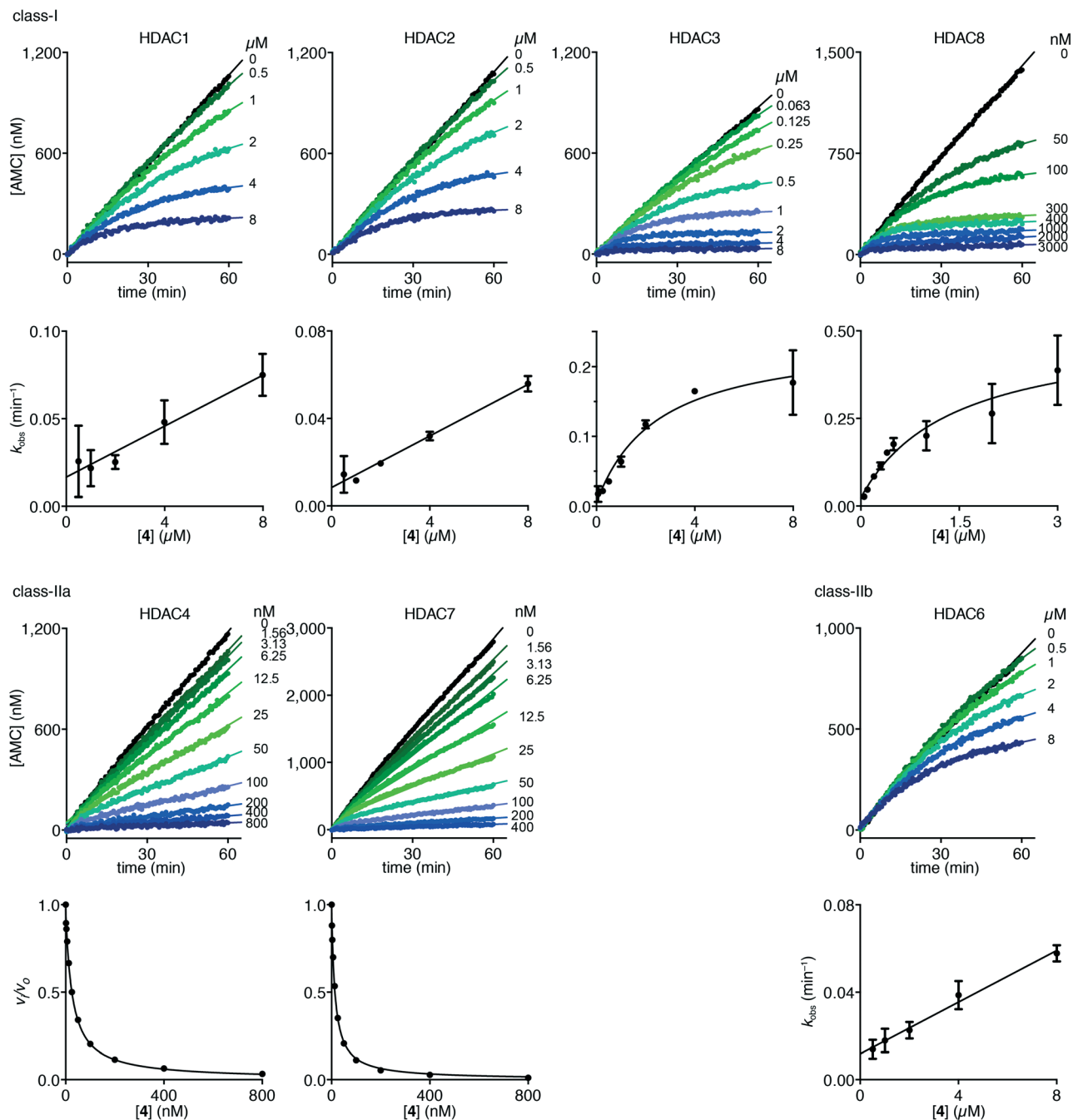


Fig. 5 Progress curves of HDACs 1–3 and 6 (LGK_{ac}), HDAC8 (RHK_{ac}K_{ac}), and HDACs 4 and 7 (LGK_{trf}) deacetylase activities under inhibition by trifluoromethyl ketone 4. Apparent first-order rate constants (k_{obs}) for trifluoromethyl ketone 4 inhibition were fitted to eqn (3) for HDACs 1, 2 and 6 (LGK_{ac}) or eqn (4) for HDAC3 and relative rates (v_i/v_0) were fitted to eqn (1) for HDACs 4 and 7.

Table 2 Kinetic values and inhibition constants for HDAC inhibition by trifluoromethyl ketone 4

	HDAC1	HDAC2	HDAC3	HDAC8	HDAC6	HDAC4	HDAC7
K_i (nM)	$(9.8 \pm 6.7) \times 10^3$	$(1.1 \pm 0.4) \times 10^4$	~ 11	~ 33	$(4.5 \pm 1.6) \times 10^3$	5.1 ± 0.2	4.7 ± 0.1
k_1 ($\text{M}^{-1} \text{min}^{-1}$)	$(1.7 \pm 0.4) \times 10^3$	$(7.7 \pm 0.8) \times 10^2$	—	—	$(2.6 \pm 0.3) \times 10^3$	—	—
k_{-1} (min^{-1})	$(1.7 \pm 0.7) \times 10^{-2}$	$(8 \pm 2) \times 10^{-3}$	—	—	$(1.2 \pm 0.3) \times 10^{-2}$	—	—
$K_{i,1}$ (nM)	—	—	$(5.7 \pm 2.6) \times 10^2$	$(7.3 \pm 5.7) \times 10^3$	—	—	—
k_2 (min^{-1})	—	—	0.24 ± 0.03	0.49 ± 0.13	—	—	—
k_{-2} (min^{-1})	—	—	$(5 \pm 11) \times 10^{-3}$	$(2.3 \pm 3.7) \times 10^{-2}$	—	—	—



suspected based on literature precedent, the hydroxamic acids all exhibited a fast-on-fast-off mechanism of inhibition. However, the SAHA-inspired trifluoromethyl ketone (4) only displayed a fast-on-fast-off mechanism against HDACs 4 and 7, whereas slow-binding mechanisms were observed against class-I and class-IIb enzymes (HDACs 1–3, 6 and 8). Furthermore, a more detailed investigation revealed similar mechanisms for HDACs 1, 2, and 6, which were different from the mechanism exhibited against HDACs 3 and 8. This insight is important for understanding the effects of trifluoromethyl ketone containing inhibitors, and should be taken into account when designing novel chemotypes containing this moiety. Furthermore, it will be interesting to investigate the possible effects of these mechanistic differences as well as kinetics of ketone reduction in cellular environments.

Experimental

Materials

HDAC1, HDAC2, HDAC3-NCoR1, HDAC6 and HDAC8 were purchased from Enzo Life Sciences (Postfach, Switzerland). HDAC4 and HDAC7 were purchased from Millipore (Upstate, Temecula, CA). The HDAC assay buffers used were: Buffer A, Biomol International BML-KI-143 [Tris/Cl (50 mM), NaCl (137 μ M), KCl (2.7 μ M), MgCl₂ (1 μ M), pH 8.0] with BSA (0.5 mg mL⁻¹) added; or Buffer B, Biomol International BML-KI-311 [Tris/Cl (50 mM), NaCl (137 μ M), KCl (2.7 μ M), MgCl₂ (1 μ M), PEG-8 kDa (10% w/v), pH 8.0] with BSA (0.5 mg mL⁻¹) added. Trichostatin A and trypsin (10 000 units per mg, TPCK treated from bovine pancreas) were from Sigma-Aldrich (Steinheim, Germany), and syntheses of SAHA and compound 4 were previously reported.³² All chemicals and solvents were of analytical grade and were used without further purification as obtained from commercial suppliers.

Fluorescence measurements

All microtiter plates were analysed using a Perkin Elmer EnSpire plate reader with excitation at 360 nm and detecting emission at 460 nm. Fluorescence measurements (RFU) were converted to [AMC] concentrations based on an [AMC]-fluorescence standard curve and all data analysis were performed using GraphPad Prism.

Fluorescence-based preincubation inhibition experiments

Hydrolase activity of HDAC3-NCoR1 against substrate Ac-LGK_{ac}-AMC was evaluated under varying concentrations of trifluoromethyl ketone 4 and preincubation times. HDAC3 (1000 pg μ L⁻¹, 10 μ L) was added to a solution of trifluoromethyl ketone 4 (30 μ L total volume, 16.7 μ M–167 pM, 10-fold dilution) at 37 °C (final concentrations 250 pg μ L⁻¹ and 12.5 μ M–125 pM, respectively) with preincubation times of 2 h, 1 h, 30 min, 15 min and 0 min, then a solution of Ac-LGK_{ac}-AMC (10 μ L, 50 μ M) was added to each well and the plate was incubated for additional 30 min at 37 °C (final concentrations HDAC3: 200 pg μ L⁻¹, 4: 10 μ M–100 pM, and Ac-LGK_{ac}-AMC: 10 μ M,

respectively). Then a solution of trypsin (50 μ L, 0.4 mg mL⁻¹) was added and the reaction mixture was developed for 15 min before analysing fluorescence. The resulting data were analysed to obtain residual enzymatic activity relative to control wells.

Fluorescence-based steady-state rate inhibition experiments

Hydrolase activities of HDACs against substrates Ac-LGK_{ac}-AMC, Ac-LGK_{tf}-AMC and Ac-RHK_{ac}K_{ac}-AMC were evaluated under varying concentrations of SAHA (1), TSA (2), and trifluoromethyl ketone 4. The HDAC enzyme was incubated with the relevant substrate, inhibitor, and trypsin in a total volume of 100 μ L of assay buffer (Buffer A, except for HDAC8, where Buffer B was used), using the following concentrations: HDAC1 (2500 pg μ L⁻¹), LGK_{ac} (20 μ M), trypsin (3 ng μ L⁻¹); HDAC2 (4000 pg μ L⁻¹), LGK_{ac} (20 μ M), trypsin (3 ng μ L⁻¹); HDAC3-NCoR1 (200 pg μ L⁻¹), LGK_{ac} (20 μ M), trypsin (3 ng μ L⁻¹); HDAC4 (50 pg μ L⁻¹), LGK_{tf} (50 μ M), trypsin (5 ng μ L⁻¹); HDAC6 (4000 pg μ L⁻¹), LGK_{ac} (20 μ M), trypsin (3 ng μ L⁻¹); HDAC7 (2.5 pg μ L⁻¹), LGK_{tf} (50 μ M), trypsin (5 ng μ L⁻¹); HDAC8 (5 pg μ L⁻¹), LGK_{tf} (200 μ M), trypsin (3 ng μ L⁻¹); HDAC8 (400 pg μ L⁻¹), RHK_{ac}K_{ac} (400 μ M), trypsin (5 ng μ L⁻¹). *In situ* fluorophore release was monitored immediately by fluorescence readings recorded continuously every 10–15 seconds for 60 min at 25 °C. Using GraphPad Prism, background fluorescence was subtracted and the data were fitted to the relevant equations ([P] = $v_s t$, or eqn (2)). Secondary plots were then fitted to the relevant equations (eqn (1), (3) or (4)) to obtain the desired kinetic parameters.

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