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Bioactive and functional properties of sour cherry juice (*Prunus cerasus*)

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Sour cherry juice (*Prunus cerasus*) is consumed as a nutritional supplement claiming health effects. The aim of the study was to evaluate the different properties of sour cherry juice in terms of antioxidant activity and inhibition of target enzymes in the central nervous system and diabetes. The content of polyphenols and anthocyanins was quantified. Different experiments were carried out to determine the radical scavenging properties of the juice. The activity of sour cherry juice was also tested in physiological relevant enzymes of the central nervous system (acetylcholinesterase, monoamine oxidase A, tyrosinase) and others involved in type 2 diabetes (α -glucosidase, dipeptidyl peptidase-4). Sour cherry juice showed significant antioxidant effects but the activity of the lyophilized juice was not superior to compounds such as ascorbic, gallic or chlorogenic acid. Furthermore, sour cherry juice and one of its main polyphenols known as chlorogenic acid were also able to inhibit monoamine oxidase A and tyrosinase as well as enzymes involved in diabetes. This is the first time that sour cherry juice is reported to inhibit monoamine oxidase A, α -glucosidase and dipeptidyl peptidase-4 in a dose dependent manner, which may be of interest for human health and the prevention of certain diseases.

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1. Introduction

Cherry belongs to the Rosaceae family, and specifically to the genus *Prunus*. The most common types of *Prunus* are *Prunus cerasus* and *Prunus avium*, the first one is known as sour cherry and the other is called sweet cherry. Both are considered nutrient dense food with a relatively low caloric content and a significant amount of important nutrients and bioactive food components.¹ Several studies have confirmed that eating a diet rich in fruit is related to a reduced risk of oxidative stress, cardiovascular disease, cancer, neurodegenerative disorders and diabetes.^{2–6} This may be due to dietary polyphenols, which are formed by at least one aromatic ring with one or more hydroxyl groups attached.⁷

Some of the most common dietary polyphenols present in fruits and berries are anthocyanidins, which generate several anthocyanins. These anthocyanins are responsible for the red colour of fruits and the potential antioxidant activity. Although cherry is botanically classified as a stone fruit (drupe) due to the pit in the centre, it has the appearance of a berry. Several

studies in animal models and in human subjects have demonstrated that dietary polyphenols are bioavailable and exert a protective role against oxidative stress and free radical damage.⁷ Antioxidants have the ability to scavenge or to neutralize free radicals, or are necessary to enable other molecules to perform such a function.⁸

There is strong evidence demonstrating that several ROS-mediated pathways may be involved in the neurodegenerative diseases, like Alzheimer's disease (AD) and Parkinson's disease (PD). It has been described that the accumulation of iron ions in the brain leads to higher ROS generation, involvement of mitochondrial pathways and to a decrease of endogenous antioxidants levels. Thus, natural antioxidants may prevent neurodegenerative disorders.⁹

Although mechanisms remain unclear, a body of evidence links type-2 diabetes with dementia and neurodegenerative diseases.¹⁰ One therapeutic approach to treat diabetes is to retard the absorption of glucose *via* inhibition of enzymes, such as α -glucosidase, in the digestive organs. It has been confirmed that α -glucosidase activity *in vitro* can be inhibited by berry extracts, *i.e.* blueberry, blackcurrant, strawberry, and raspberry rich in polyphenols.¹¹ In recent years, there has also been an increasing interest in the ability of dietary factors to treat diabetes *via* modulating GLP-1 levels. GLP-1 is secreted from enteroendocrine L cells, which are present in the lower small intestine and large intestine, and stimulates insulin secretion in a blood glucose concentration dependent manner.

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GLP-1 is inactivated by dipeptidyl peptidase-4 (DPP-4), a circulating catabolic enzyme, resulting in a rather short half-life of about two minutes in the blood. There are reports that non-nutrient dietary factors such as polyphenols can affect GLP-1 levels.¹²

The aim of this study is to evaluate the bioactive properties of sour cherry juice in terms of antioxidant potential as well as activity in pharmacological targets of neurological diseases and diabetes. Antioxidant and protective effects of the juice have been studied in cellular and cell free systems. Potential inhibition of enzymes with relevant biological properties such as acetylcholinesterase, monoamine oxidase-A, tyrosinase, α -glucosidase and dipeptidyl peptidase 4 has also been carried out.

2. Materials and methods

2.1. Reagents and chemicals

Chemical reagents were acquired through Sigma-Aldrich, Cayman Chemical, Cymit química and Panreac (Spain). Sour cherry juice (*Prunus cerasus*) from Rabenhorst® was kindly supplied by Natur Import. According to the manufacturer, the juice is 100% organic, additives free and was obtained by pressing and pasteurization. The juice is bottled into amber bottles (expiration date: 20/01/2017).

2.2. Sour cherry juice lyophilization

330 mL of Rabenhorst® sour cherry juice were lyophilized using a Genesis VirTis 25 EL lyophilizer (Wizard 2.0 control system) over 7 days. The liquid sample was frozen at $-80\text{ }^{\circ}\text{C}$ for 2 h while the lyophilizer was freezing at $-80\text{ }^{\circ}\text{C}$. Afterwards, the temperature was modified to $-30\text{ }^{\circ}\text{C}$ for a couple of hours and for 96 h to $-60\text{ }^{\circ}\text{C}$. Next transition was to $-40\text{ }^{\circ}\text{C}$ again (4 h) and 24 h to $-60\text{ }^{\circ}\text{C}$. Finally, temperature reached until $-15\text{ }^{\circ}\text{C}$ (7 h) and dried 22 h at $20\text{ }^{\circ}\text{C}$. The last 2 h temperature was fixed at $40\text{ }^{\circ}\text{C}$. A sticky red residue was obtained and kept at $-20\text{ }^{\circ}\text{C}$ in a freezer until experiments were done.

2.3. Phytochemical analyses of lyophilized sour cherry juice

2.3.1. Total polyphenols quantification. The Folin-Ciocalteu assay was used to quantify total phenolic compounds in sour cherry juice. 9 μL of the sample was mixed with 201 μL of the Folin-Ciocalteu reagent. The sample was incubated 5 min at room temperature and preserved for the light; 90 μL of Na_2CO_3 (10%) was added to the mixture and incubated in the dark at room temperature for 40 min. Absorbance was measured at 752 nm. The standard curve was performed with different concentrations of gallic acid: 1–0.0078 mg mL^{-1} . The total polyphenol content is expressed as mg GAE (Gallic Acid Equivalents) per mg lyophilized sour cherry juice.¹³

2.3.2. HPLC-DAD analysis and anthocyanins quantification. Phytochemical screening of sour cherry juice was performed by HPLC using an Agilent 1260 Infinity LC (column

Eclipse plus C18 4.6 \times 100 mm, 5 μm) equipped with a photodiode array detector. A two-phase gradient system of trifluoroacetic acid/water (0.5/99.5, v/v) as mobile phase A and trifluoroacetic acid/acetonitrile/water (0.5/50/49.5, v/v) as mobile phase B was used.¹⁴ The gradient started at 92% of mobile phase A and 8% of phase B, reaching 18% mobile phase B at 1.2 min, 32% at 14 min, 60% of mobile phase B at 28 min and 100% at 34 min, at isocratic elution until 38.8 min. The gradient reached the initial conditions at 39.2 min and was maintained at isocratic elution for 0.8 min. Elution was carried out at a flow rate of 1 mL min^{-1} . The injection volume was 10 μL and the concentration of the sample (lyophilized juice) was 100 mg mL^{-1} . Ascorbic acid, gallic acid, ellagic acid, chlorogenic acid, catechin, as well as anthocyanosides were used as standards in order to detect and compare peaks in sour cherry juice. Standards were dissolved in methanol and calibration curves were calculated. Lyophilized juice was dissolved in ultra-pure water and filtered with nylon filters of 0.22 μm in order to remove any solid residue. For the detection of compounds, the chromatograms were recorded at 260, 280, 320, 360 and 520 nm. Polyphenols were identified according to retention times of standard pure compounds, elution order and comparing with a bibliographic revision of main phenolic compounds. Total anthocyanins were also quantified by HPLC at 520 nm using cyanidin 3-*o*-rutinoside (keracyanin chloride) as the standard for a calibration curve.

2.4. Cytotoxicity screening in HeLa cells

HeLa cells were used to perform a cell viability test (MTT assay).¹⁵ HeLa cells were grown in DMEM supplemented with 10% fetal bovine serum and 1% penicillin–streptomycin–glutamine. Cultures were incubated in the presence of 5% CO_2 at $37\text{ }^{\circ}\text{C}$ under a 100% relative humidified atmosphere. Cells were seeded in 96-well microplates at a density of 7×10^3 cells per well and grown for 24 h at $37\text{ }^{\circ}\text{C}$. Cells were then treated with various concentrations of lyophilized sour cherry juice (0.001–1 mg mL^{-1}) for 72 h and a MTT solution was added and incubated for 3 h at $37\text{ }^{\circ}\text{C}$. Cell survival was measured as the reduction of MTT into formazan at 550 nm in a microplate reader. Experiments were performed twice.

2.5. Antioxidant activity assays

2.5.1. Protective effects of sour cherry juice against hydrogen peroxide induced toxicity in *Artemia salina*. Dried cysts of *Artemia salina* were hatched in seawater with aeration for a whole week. Lyophilized cherry juice was dissolved in seawater and transferred to a 6-well plate at different concentrations (250, 500 and 1000 $\mu\text{g mL}^{-1}$) in 5 mL seawater with 10 nauplii in each well. Control wells were filled with 5 mL seawater and 10 nauplii also. After 24 h incubation at room temperature, survival viability was calculated.

As the viability of *Artemia salina* nauplii was not affected by different concentrations of sour cherry juice, the experiment was performed adding hydrogen peroxide to the wells at a concentration of 0.4 g L^{-1} . Two different control wells without sour cherry juice were also set, one with hydrogen peroxide



and another with seawater. The viability of the nauplii study was measured every 24 h for 72 h.¹²

2.5.2. Superoxide radical scavenging activity. Sour cherry juice was tested in the xanthine/xanthine oxidase assay in order to measure the capacity to scavenge superoxide radicals.¹⁶ 90 μM xanthine, 16 mM Na_2CO_3 , and 22.8 μM NBT were dissolved in phosphate buffer pH = 7 to reach a volume of 240 μL . Then, 30 μL of sample and 30 μL of xanthine oxidase (168 U L^{-1}) were added to start the reaction. The mixture was incubated for 2 min at 37 $^\circ\text{C}$. Absorbance was measured at 560 nm and the activity of cherry juice was determined by checking the transformation of NBT to the blue chromogen dye by the superoxide radical (O_2^-). Decreased absorbance of the reaction mixture indicated increased superoxide anion scavenging activity. Gallic acid was used as a reference compound.

2.5.3. Antiradical activity against 2,2-diphenyl-1-picrylhydrazyl (DPPH) radicals. This assay is based on the measurement of the scavenging capacity of antioxidants.¹⁷ The odd electron of the nitrogen atom in DPPH is reduced by receiving a hydrogen atom from antioxidants to the corresponding hydrazine. 150 μL of a DPPH methanolic solution were added to 150 μL of different sour cherry juice concentrations dissolved in MeOH. Absorbance was measured at 517 nm after 30 min of reaction in a microplate reader. Controls contained a DPPH solution and the solvent. Radical scavenging capacity was calculated by the formula: $\text{RSC (\%)} = \frac{[\text{Abs}_{\text{control}} - \text{Abs}_{\text{sample}}]}{\text{Abs}_{\text{control}}} \times 100$. Ascorbic, gallic and chlorogenic acids were also measured to compare with sour cherry juice antioxidant activity.

2.6. Bioassays regarding CNS enzymes

2.6.1. Acetylcholinesterase (AChE) inhibition. The Ellman's method was selected to perform the experiment using a 96-microplate reader.¹⁸ Each well contained 25 μL of 15 mM ATCI in Millipore water, 125 μL of 3 mM DTNB in buffer C (50 mM Tris-HCl, pH = 8, 0.1 M NaCl, 0.02 M $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$), 50 μL buffer B (50 mM Tris-HCl, pH = 8, 0.1% bovine serum), 25 μL juice in buffer A (50 mM Tris-HCl, pH = 8). Finally, 25 μL of the enzyme (0.22 U L^{-1}) was added to start the reaction. Absorbance was read 13 times every 13 s at 405 nm. Galantamine was used as the reference substance.

2.6.2. Monoamine oxidase A (MAO-A) inhibition. The activity was measured in a 96-well microplate using a described procedure.¹⁹ Each well contained 50 μL of sour cherry juice in MilliQ water, 50 μL chromogenic solution (0.8 mM vanillic acid, 417 mM 4-aminoantipyrine and 4 U mL^{-1} horseradish peroxidase in potassium phosphate buffer pH = 7.6), 100 μL of 3 mM tyramine and 50 μL of 8 U mL^{-1} MAO-A. Control wells contained 50 μL of solvent instead of sour cherry juice. The absorbance was read at 490 nm every 5 min for 30 min. Clorgyline was used as the reference substance.

2.6.3. Tyrosinase (TYR) inhibition. The assay was conducted in 96-well microplates using a microplate reader to measure absorbance at 475 nm.²⁰ 10 μL of cherry juice in MilliQ water, 40 μL of *L*-DOPA, 80 μL phosphate buffer, pH =

6.8 and 40 μL of tyrosinase were mixed in each well. Controls contained 50 μL of solvent instead of sour cherry juice. α -Kojic acid was used as the reference substance.

2.7. Bioassays regarding enzymes involved in type 2 diabetes

2.7.1. Inhibition of α -glucosidase (α -GLU). The capacity of sour cherry juice to inhibit α -glucosidase was measured in a 96-well microplate reader at 405 nm.²¹ Each well contained 50 μL sample and 100 μL enzyme. After 10 min, 50 μL pNPG were added and incubated at 37 $^\circ\text{C}$ for 20 min. Control wells contained 50 μL of solvent (Millipore water) instead of lyophilized juice. Acarbose was used as the reference compound.

2.7.2. Inhibition of dipeptidyl peptidase-4 (DPP-4). The capacity of sour cherry juice and its reference compounds to inhibit the enzyme DPP-4 was measured using the fluorogenic substrate Gly-Pro-Aminomethylcoumarin (AMC) with a commercial kit (Cayman, item no. 700210). The authors followed the kit instructions. Sour cherry juice was tested at four different concentrations (5, 1, 0.1 and 0.01 mg mL^{-1} in the reaction mixture) and sitagliptin as a reference inhibitor of the enzyme. The percentages of inhibition of cherry juice and other compounds were determined with the following formula: $\% \text{ Inhibition} = \frac{[\text{Initial Activity} - \text{Inhibitor}]}{\text{Initial Activity}} \times 100$.

2.8. Statistical analysis

Results were expressed as the mean \pm standard error of experiments performed in triplicate. GraphPad Prism v.5 was required to perform data analyses, nonlinear regressions and statistics.

3. Results

3.1. Phytochemical analysis of lyophilized sour cherry juice

The polyphenol content was measured by the Folin-Ciocalteu method and expressed as gallic acid equivalents (GAE). Sour cherry juice contained $9.835 \pm 1.092 \mu\text{g GAE mg}^{-1}$ of lyophilized juice. Five compounds (ascorbic acid, gallic acid, chlorogenic acid, neochlorogenic acid and cyanidin 3-glucosyl-rutinoside) out of different monitored polyphenolic compounds were detected, quantified and confirmed comparing retention times and UV spectra with standards (Table 1 and Fig. 1). Total anthocyanins were quantified by HPLC at 520 nm using cyanidin-3-*o*-rutinoside (keracyanin chloride) as the external standard following the literature. Anthocyanins were found to be $0.194 \pm 0.004 \mu\text{g cyanidin-3-}o\text{-rutinoside equivalents per mg}$. However, more anthocyanins different from cyanidin 3-glucosyl-rutinoside might be responsible for the red colour as other minor peaks can be seen at 520 nm (Fig. 1B).

3.2. Cytotoxicity screening in HeLa cells

The lyophilized juice showed very mild antiproliferative effects in HeLa cells. Significant differences were detected at concentrations over 0.125 mg mL^{-1} , which indicates that this cell line seems to be partially sensitive to cherry components. Cell viability was approximately 60% at the highest tested concentration (1 mg mL^{-1}), which means that the juice



Table 1 Quantification of individual polyphenols in sour cherry juice (*Prunus cerasus*) by HPLC

| Compound | $\mu\text{g mg}^{-1}$ of lyophilized juice (S.E.) |
|----------------------------------|---|
| Ascorbic acid | 0.011 (0.003) |
| Gallic acid | 0.047 (0.001) |
| Ellagic acid | — |
| Catechin | — |
| Chlorogenic acid | 0.593 (0.001) |
| Neochlorogenic acid | 1.580 (0.012) |
| Cyanidin 3-glucosyl-rutinoside | 0.081 (0.000) |
| Total polyphenols (Folin method) | 9.835 ± 1.092 |
| Total anthocyanins (HPLC method) | 0.194 (0.003) |

—: Not detected; S.E. = standard error.

is not considered cytotoxic in this type of cervical cancer cells (Fig. 2).

3.3. Antioxidant activity assays

Fig. 3 indicates that sour cherry juice increased the survival of *Artemia salina* nauplii compared to 0.4 g L^{-1} hydrogen peroxide

at 24, 48 and 72 h. Different concentrations of lyophilized juice enhanced survival of nauplii exposed to hydrogen peroxide, reaching more than 90% at 24 h. At 48 h, survival of nauplii was around 30–50% and finally, at 72 h between 10–20% of the nauplii survived compared to 0% survival of nauplii exposed to hydrogen peroxide. Significant differences were only obtained at doses of 1000 and 500 $\mu\text{g mL}^{-1}$ at 24 h and 72 h.

Fig. 4 shows the antioxidant effect of the juice compared to a reference standard such as gallic acid on superoxide radicals generated by the xanthine/xanthine oxidase system. IC_{50} values in this case were $54 \mu\text{g mL}^{-1}$ for sour cherry juice and $0.044 \mu\text{g mL}^{-1}$ for gallic acid.

The DPPH radical scavenging effects of sour cherry juice are shown in Fig. 5. The antiradical activity of the juice was compared to gallic, ascorbic and chlorogenic acid. IC_{50} values were also calculated by nonlinear regression. IC_{50} values were $236 \mu\text{g mL}^{-1}$ for cherry juice, $10 \mu\text{g mL}^{-1}$ for chlorogenic acid, $3 \mu\text{g mL}^{-1}$ for ascorbic acid and $1 \mu\text{g mL}^{-1}$ for gallic acid.

All these data indicate that the antioxidant and potential protective effects of sour cherry juice may be due to radical

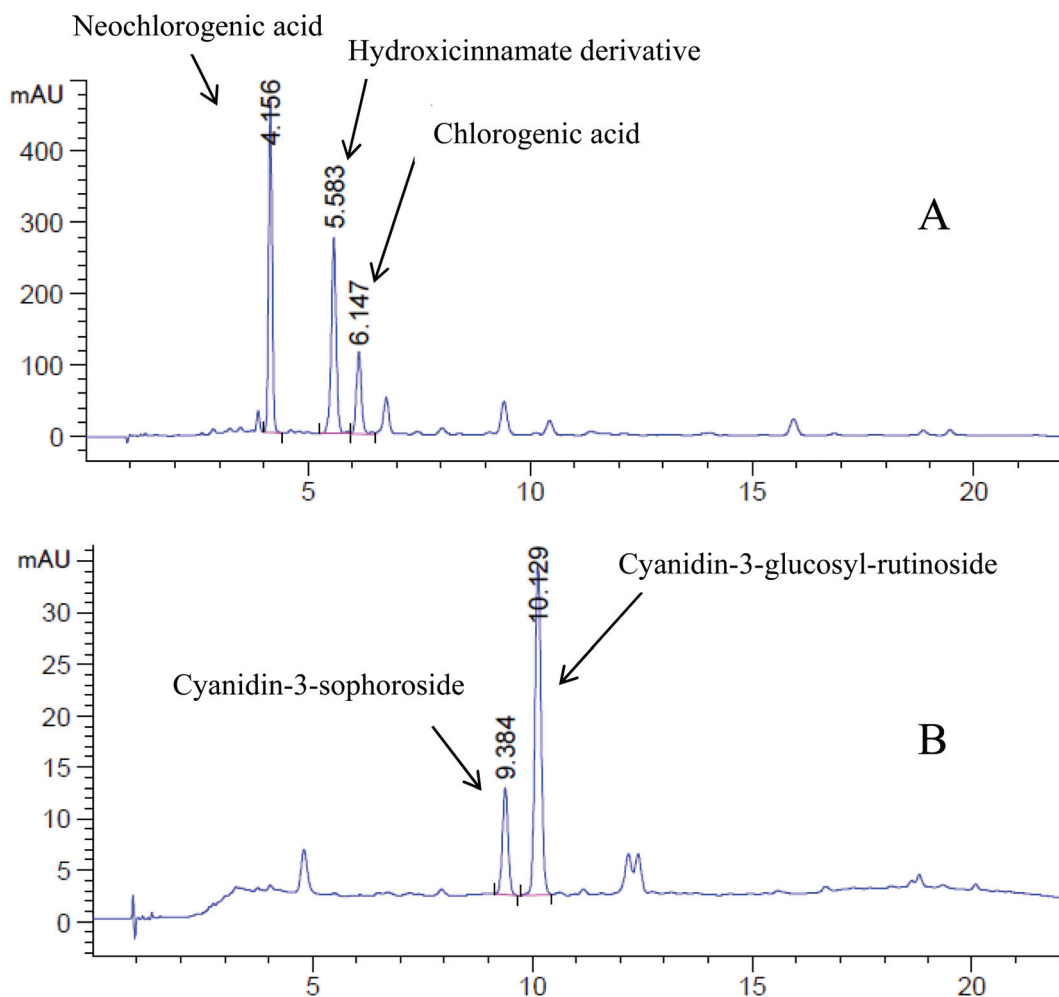


Fig. 1 HPLC chromatograms of lyophilized sour cherry juice at 320 nm (A) and 520 nm (B).



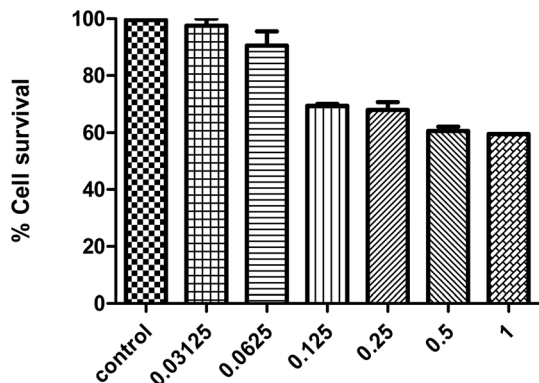


Fig. 2 Viability of HeLa cells exposed to different concentration (mg mL⁻¹) of sour cherry juice for 72 hours in the MTT assay. IC₅₀ was not calculated as percentages of viability were more than 50%.

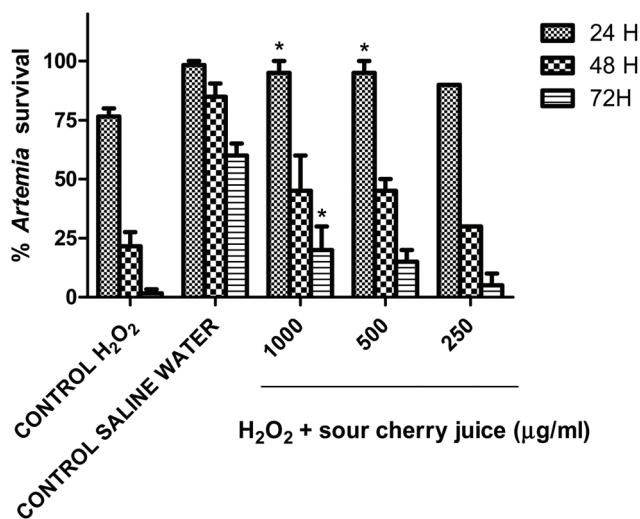


Fig. 3 Effects of sour cherry juice in *Artemia salina* nauplii exposed to hydrogen peroxide (0.4 g L⁻¹). *Significant differences ($p < 0.05$) were detected between nauplii exposed to hydrogen peroxide (control H₂O₂) and nauplii exposed to hydrogen peroxide + 1000 µg mL⁻¹ cherry juice at 24 h and 72 h. A lower dose of cherry juice (500 µg mL⁻¹) also showed a protective effect at 72 h. Differences were calculated using Student *t* tests.

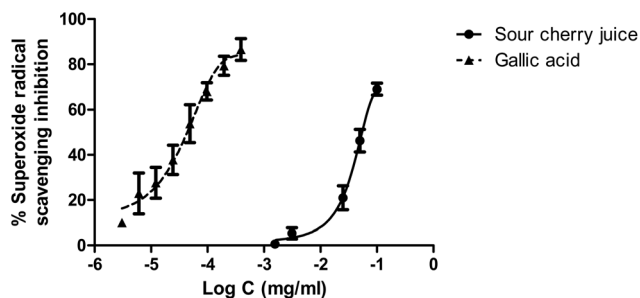


Fig. 4 Antioxidant activity of sour cherry juice and gallic acid against superoxide radicals generated by the xanthine/xanthine oxidase system.

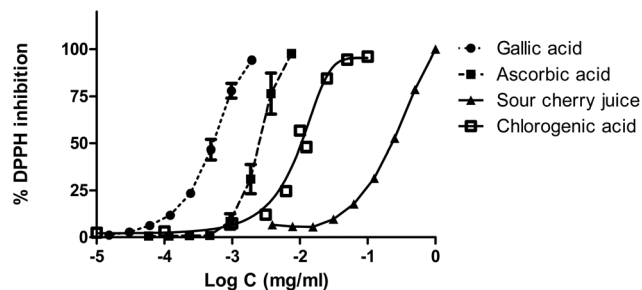


Fig. 5 Antiradical activity of sour cherry juice, ascorbic acid, gallic acid and chlorogenic acid against DPPH.

scavenging properties as it has been demonstrated through the DPPH and superoxide radical assays. In addition, the presence of polyphenols such as chlorogenic acid and anthocyanins in the juice, which was confirmed in the phytochemical analyses, seems to be crucial for the antioxidant and antiradical activities.

3.4. Bioassays regarding CNS enzymes

Sour cherry juice did not show activity in the AChE assay; however, it showed a clear dose dependent MAO-A inhibitory activity. These inhibitions are shown in Fig. 6. IC₅₀ values were calculated by nonlinear regression (0.02 µg mL⁻¹ for the selective inhibitor clorgyline, 246.19 µg mL⁻¹ for sour cherry juice). TYR inhibitory activity was not as clear as for MAO-A. Sour cherry juice produced a very mild inhibition (28% at 1 mg mL⁻¹) like Fig. 7 shows. In the MAO-A and TYR bioassays, chlorogenic acid showed a higher inhibition than the lyophilized juice.

3.5. Bioassays regarding type 2 diabetes enzymes

Sour cherry juice exhibited *in vitro* an inhibition of α-GLU, but this activity was moderate compared to chlorogenic acid and acarbose, which is a reference inhibitor of this enzyme. As shown in Fig. 8, IC₅₀ values were calculated by nonlinear regression (2783 µg mL⁻¹ for sour cherry juice, 996 µg mL⁻¹ for chlorogenic acid and 380 µg mL⁻¹ for acarbose).

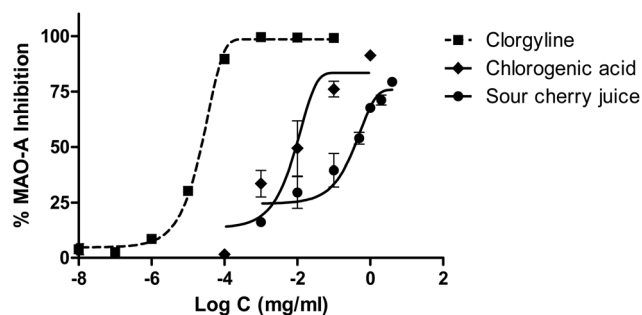


Fig. 6 MAO-A inhibition performed by sour cherry juice, chlorogenic acid and clorgyline.



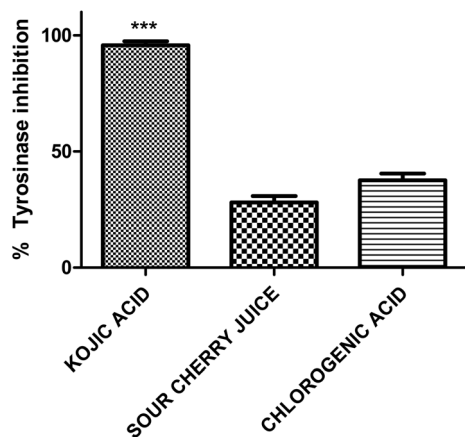


Fig. 7 Tyrosinase inhibition performed by sour cherry juice (1 mg mL^{-1}), chlorogenic acid (1 mg mL^{-1}) and kojic acid (0.2 mg mL^{-1}) as standard. *** $p < 0.0001$ versus sour cherry juice and chlorogenic acid (One way ANOVA with *post-hoc* Tukey test).

Sitagliptin, an antidiabetic drug, showed a clear dose dependent DPP-4 inhibition. In addition, the effects of the juice and chlorogenic acid in this enzyme are shown in Fig. 9. IC_{50} values were calculated by nonlinear regression ($0.1 \mu\text{g mL}^{-1}$ for sitagliptin and $1003.41 \mu\text{g mL}^{-1}$ for sour cherry juice) (Fig. 9).

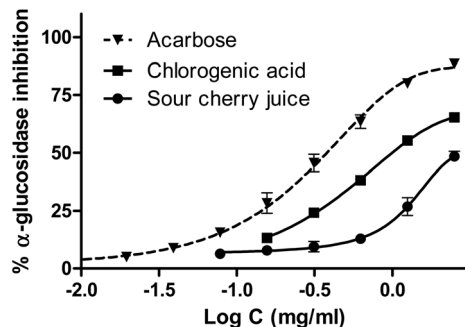


Fig. 8 α -Glucosidase inhibition performed by sour cherry juice, chlorogenic acid and acarbose as the standard.

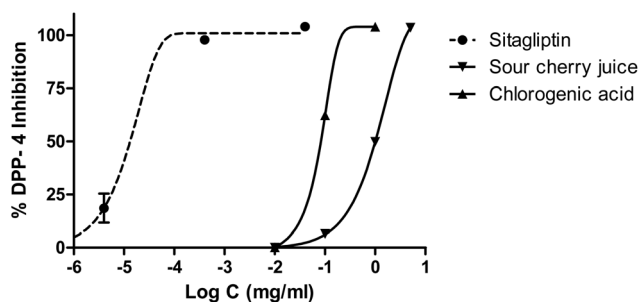


Fig. 9 Dipeptidyl peptidase-4 inhibition performed by sour cherry juice, chlorogenic acid and sitagliptin as the standard.

4. Discussion

Sour cherry juice is a good source of phytochemicals, specifically polyphenols and anthocyanins, which are the polyphenols responsible for the red skin and flesh colour.²²

The concentration of total phenolics (TP) was $9.835 \pm 1.092 \mu\text{g GAE per mg}$ of lyophilized sour cherry juice (approx. $100 \text{ mg per } 100 \text{ g}$), which is a lower concentration compared to other juices such as pomegranate¹³ but can still be of significant importance to produce health benefits. HPLC-DAD analysis showed a peak of chlorogenic acid and previous studies reveal that hydroxycinnamates such as caffeoylquinic acids are the main polyphenols in sweet and sour cherries.²² Anthocyanins were also quantified by HPLC-DAD, obtaining approximately $20 \text{ mg per } 100 \text{ g}$. Other authors such as Wojdyło *et al.* compared 33 types of sour cherry in terms of the polyphenol content and antioxidant activity;²³ our results are within the range of anthocyanins calculated for different cherry cultivars ($7.56\text{--}94.20 \text{ mg per } 100 \text{ g}$) although the authors studied the fruit content instead of juices as in our case.

Sour cherry juice is used in sports medicine to prevent muscle damage as some studies have shown that this product is able to prevent these symptoms through anti-inflammatory and antioxidant properties.^{24–29} Our results confirm the antioxidant potential of sour cherry juice. The protective effects against toxicity induced by hydrogen peroxide were measured using living organisms (*Artemia salina*). This experiment was performed by the authors using lyophilized sour cherry juice as a co-treatment with hydrogen peroxide and the study demonstrated significant differences of 1000 and $500 \mu\text{g mL}^{-1}$ at 24 h and $1000 \mu\text{g mL}^{-1}$ at 72 h . The juice was also able to scavenge both DPPH and superoxide radicals. It can be deduced that the antioxidant activity is mainly provided by polyphenols such as chlorogenic acid and anthocyanins. Other studies quantified *Prunus cerasus* antioxidant capacity using trolox as the standard, which makes it difficult to compare with our results.³⁰

In addition, the study of the activity of sour cherry juice on enzymes was divided into two main groups, one related to the central nervous system and the other to glucose metabolism. We found for the first time that sour cherry juice was able to inhibit MAO-A and TYR. MAO-A is involved in the deamination of catecholamines and serotonin and certain polyphenols such as anthocyanins are involved in this inhibition, which can drive an antidepressant and anxiolytic effect.³¹ Tyrosinase is a copper-containing enzyme essential for tyrosine-melanin pigmentation and the role of toxic quinones in dopamine-induced neuronal damage catalyzed by TYR has been cleared in a number of studies.³² According to our data, this juice may have potential as a neuroprotective agent *via* MAO-A or TYR inhibition; in fact, a recent interventional human study showed that the consumption of anthocyanin-rich cherry juice for 12 weeks improves memory and cognition in older adults with mild-to-moderate dementia.³³

Finally, the inhibition of enzymes involved in glucose metabolism and type 2 diabetes was studied; this is the first



time that sour cherry juice is reported to inhibit α -glucosidase and DPP-4 in a dose dependent manner. The anthocyanin content in fruits is also related with α -glucosidase inhibition.³⁴ According to our results, chlorogenic acid is also responsible for the activity. Polyphenols have also shown to facilitate the insulin response and attenuate secretion of glucose-dependent insulinotropic polypeptide and GLP-1. The DPP-4 enzyme also regulates glycaemia and its inhibitors such as sitagliptin represent some of the new treatments for type 2 diabetes. Taking in consideration that type 2 diabetes is linked to neurodegenerative diseases due to the production of superoxide radicals, sour cherry (*Prunus cerasus*) juice might be an interesting antioxidant nutritional tool to prevent these disorders.

Conflicts of interest

The authors declare that they do not have any conflicts of interest.

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