



Cite this: *Phys. Chem. Chem. Phys.*,
2016, 18, 30135

Magnetic anisotropy of a Co^{II} single ion magnet with distorted trigonal prismatic coordination: theory and experiment†

Yan Peng,^{‡ab} Tilmann Bodenstein,^{‡bc} Karin Fink,^{*b} Valeriu Mereacre,^a
Christopher E. Anson^a and Annie K. Powell^{*ab}

The single ion magnetic properties of Co(II) are affected by the details of the coordination geometry of the ion. Here we show that a geometry close to trigonal prismatic which arises when the ligand 6,6'-((1Z)-((piperazine-1,4-diylbis(propene-3,1-diyl))bis(azanylylidene))bis(methanylylidene))bis(2-methoxyphenol) coordinates to Co(II) does indeed lead to enhanced single-ion behaviour as has previously been predicted. Synthesis of the compound, structural information, and static as well as dynamic magnetic data are presented along with an analysis using quantum chemical *ab initio* calculations. Though the complex shows a slight deviation from an ideal trigonal prismatic coordination, the zero-field splitting as well as the *g*-tensor are strongly axial with $D = -41 \text{ cm}^{-1}$ and $E < 0.01 \text{ cm}^{-1}$. For the lowest Kramers doublet ($S = 1/2$) $g_{\parallel} = 7.86$ and $g_{\perp} < 0.05$ were found. In contrast, the second Kramers doublet possesses a rhombic *g*-tensor with $g_{\parallel} = 2.75$ and $g_{\perp} = 4.35$. Due to large spin-orbit coupling resulting in very different *g* tensors, it is not possible to simulate the temperature dependence of the magnetic susceptibility with a spin Hamiltonian of the form $H = D(S_z^2 - S(S+1)/3) + E(S_x^2 - S_y^2) + \mu_B \mathbf{g} \cdot \mathbf{S} \cdot \mathbf{B}$ using an effective spin $S = 3/2$. Calculations on model complexes show the influence of the coordinating atoms and the deviation from the ideal trigonal prismatic coordination. As the distortion is reduced towards idealised D_{3h} , the zero field splitting increases and the *g*-tensor of the second Kramers doublet also becomes axial.

Received 10th May 2016,
Accepted 11th October 2016

DOI: 10.1039/c6cp03157a

www.rsc.org/pccp

Introduction

Since the discovery of the first 3d SIM (single-ion magnet), an Fe(II) complex with a trigonal pyramidal coordination geometry in 2010,¹ the family of SIMs based on 3d ions has grown rapidly. The magnetic behaviour of these compounds results from the interplay between ligand-field splitting and spin-orbit interaction that can generate large anisotropy, which is a key ingredient for SMMs (single molecule magnets).^{2–4} With the development of this area, it has become obvious that magnetic anisotropy is the main criterion for engendering slow relaxation of the magnetisation, although other mechanisms may contribute

to the relaxation processes.⁵ In particular, Co(II) complexes are known to show a wide range of zero field splittings. In his review, Boča gives values in the range from -38 cm^{-1} to $+83 \text{ cm}^{-1}$.⁶ Low coordination numbers are promising because they can split the d-orbitals in such a way that a nearly degenerated ground state occurs, facilitating the spin-orbit coupling and thus enhancing the magnetic anisotropy.^{7,8} Currently, the quest for transition-metal-based SIMs is focused on low-coordinate metal species with coordination numbers of two to five, which possess large axial (*D*) and small rhombic zero-field splitting parameters (*E*) resulting in slow magnetic relaxation.^{4,7,8} Only recently have examples of seven- and eight- coordinated Co(II) compounds been reported which show slow relaxation.^{9,10} For 3d ion-based SIMs, the SMM behaviour is usually only visible under a small applied external field that suppresses the fast magnetic tunnelling and very few mononuclear complexes based on 3d ions show slow relaxation of the magnetisation without field.^{11–18} Recently, Ruiz and co-workers predicted the properties of 3d SIMs based on the coordination numbers and electronic structures of paramagnetic centres.² For a Co(II) system, SMM behaviour could occur either *via* the trigonal pyramidal (large positive *D* values) or the trigonal prismatic (large negative *D* values) coordination modes. Such a trigonal prismatic coordination was realised in

^a Institute of Inorganic Chemistry, Karlsruhe Institute of Technology, Engesserstrasse 15, 76131 Karlsruhe, Germany. E-mail: annie.powell@kit.edu

^b Institute of Nanotechnology, Karlsruhe Institute of Technology, P.O. Box 3640, 76021 Karlsruhe, Germany. E-mail: karin.fink@kit.edu

^c qLEAP Center for Theoretical Chemistry, Department of Chemistry, Aarhus University, Langelandsgade 140, 8000 Aarhus C, Denmark

† Electronic supplementary information (ESI) available: Experimental details; crystal data and geometric information for structure 1, additional magnetic data and calculation details. CCDC 1422338. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c6cp03157a

‡ These authors contributed equally.



two homoleptic Co(II) complexes.^{15–17} In the first complex, Co(II) is coordinated by nitrogen (trispyrazoloximate) and derivatives and the complex shows a large negative value for D (-115 cm^{-1}) and a spin reversal barrier of 152 cm^{-1} .¹⁵ The second is coordinated by oxygen via six salicylaldehyde Schiff bases with a barrier of 76 cm^{-1} . In the first coordination sphere, both compounds show only a very small distortion from an ideal trigonal prism. Although both complexes possess a strong axial zero field splitting (E/D is very small), there is a significant difference in the g -tensor. While the anisotropy of the g values obtained from magnetisation and NMR data is small in the first complex ($g_{\parallel} = 2.9$ and $g_{\perp} = 2.2$) it is, according to quantum chemical calculations, strongly axial in the second compound ($g_{\parallel} = 7.57$ and $g_{\perp} = 0.56$) whereas the corresponding magnetisation data was fitted with $S = 3/2$ and an isotropic $g_{\text{iso}} = 3.05$.

Here, we present the synthesis and characterisation of an air stable Co(II) compound with a trigonal prismatic coordination geometry. Although there is deviation from idealised trigonal prismatic coordination arising from the combination of nitrogen and oxygen in the first coordination sphere and the stiffness of the ligand, the compound exhibits slow magnetic relaxation without application of a dc field. The Orbach energy barrier of 56 cm^{-1} is relatively high for a Co(II)-based SIM. The electronic structure and the magnetic properties were analysed in detail by quantum chemical calculations. The effect of the distortion from the ideal trigonal prismatic coordination was investigated by calculating the situation for model complexes where the coordination was changed stepwise from that of the synthesized complex to an ideal trigonal prism with either 6 O or 6 N coordinating atoms.

Experimental

General information

All chemicals were used as received without any further purification and all manipulations were performed under aerobic conditions. Powder X-ray diffraction was carried out on a STOE STADI-P diffractometer, using Cu-K α radiation with $\lambda = 1.5406\text{ \AA}$. Elemental analyses (C H N) were performed using an Elemental Vario EL analyzer.

X-ray crystal structure

Data were measured at 150 K on a Stoe IPDS II diffractometer with graphite-monochromated Mo-K α radiation. The structure was solved by dual-space direct methods (SHELXT) and refined by full-matrix least-squares using SHELX-2014.¹⁹

Magnetic measurements

Magnetic susceptibility data (1.8–300 K) were collected on powdered samples using a SQUID-based sample magnetometer,

Quantum Design model MPMS-XL instrument under a 1000 Oe applied magnetic field. Magnetisation isotherms were collected at 2, 3, and 5 K between 0 and 7 T. Ac susceptibility measurements were carried out under an oscillating ac field of 3 Oe and ac frequencies ranging from 1 to 1500 Hz. Data were corrected for diamagnetism using Pascal constants and a sample holder correction.

Preparation of 6,6'-((1Z)-((piperazine-1,4-diylbis(propene-3,1-diyl))bis(azanylylidene))bis(methanylylidene))bis(2-methoxyphenol) (H₂L)

A mixture of *o*-vanillin (3.04 g, 20 mmol) and 1,4-bis(3-amino-propyl)piperazine (2.0 g, 10 mmol) was refluxed for 4 h in EtOH (50 mL). The resultant yellow precipitate was filtered and washed well with cold EtOH followed by ether and then dried in air. Yield 90% (4.23 g). Anal. calc. (found) % for C₂₆H₃₆N₄O₄: C, 66.64 (66.55); N, 11.96 (11.80); H, 7.74 (7.82).

Preparation of [Co^{II}(L)] (1)

A mixture of H₂L (46.8 mg, 0.1 mmol), Co(NO₃)₂·6H₂O (29 mg, 0.1 mmol) in 2 mL of DMF was stirred for 10 minutes. Then, Et₃N (50 mg, 0.5 mmol) was added to the mixture under stirring. After 1 minute stirring, the solution was filtered and left undisturbed. Red-brown block-shaped crystals of **1** were grown from the filtrate overnight in 50% (37.8 mg, based on Co) yield. Anal. calc. (found)% for C₂₆H₃₄CoN₄O₄: C, 59.42 (59.45); N, 10.66 (10.55); H, 6.52 (6.50).

Quantum chemical methods

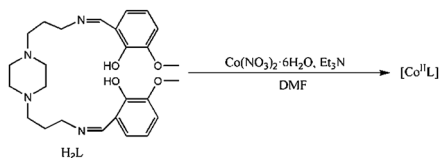
Complete active space self-consistent field (CASSCF) and spin-orbit configuration interaction (SOC) calculations were performed for compound **1**. For Co, O and N, a def2-TZVPP basis set was used while C and H were equipped by a def2-SVP basis set.^{20,21}

All calculations were performed with the Bochum suite of ab initio programs.^{22–26} The active space was spanned by the five 3d orbitals of Co. The CASSCF orbitals were obtained by averaging over the ten $S = 3/2$ states. In the SOC calculations, the scaled-nucleus spin-orbit operator²⁷ ($\xi = 0.61$) was constructed and diagonalised within the active space. Dynamic correlation was considered for the first two quartet states by the averaged coupled pair functional (ACPF)²⁸ variant of the MCCEPA program²⁶ with reference wave-functions based on this active space. The 3d orbitals of Co and the lone pairs of the ligands were included in the correlation treatment. Test calculations showed that an expansion of the active space by addition of a further d-shell or inclusion of occupied ligand orbitals has no influence on the CASSCF reference wave-functions. Since we use a multi-reference configuration interaction type rather than a perturbation approach to treat dynamic correlation it is unlikely that extension of the active space will have a significant influence on the results. For the model complexes, only the core orbitals (1s for O and N, 1s2s2p for Co) were kept frozen. The influence of the correlation energy on the SOC calculations was considered by shifting the diagonal elements of the lowest CASSCF state by $\Delta E_{\text{cor}} = (E(2^4A_1)_{\text{ACPF}} - E(2^4A_1)_{\text{CAS}}) - (E(1^4A_1)_{\text{ACPF}} - E(1^4A_1)_{\text{CAS}})$ thus correcting the first excitation energy. This method is

§ X-Ray crystal structure determination: C₂₆H₃₄CoN₄O₄ (525.50 g mol⁻¹), monoclinic, space group $P2_1/n$, $a = 16.1119(11)$, $b = 7.0151(6)$, $c = 21.1805(14)\text{ \AA}$, $\beta = 96.396(5)^\circ$, $V = 2379.1(3)\text{ \AA}^3$, $Z = 4$, $T = 150\text{ K}$, 16 727 reflections measured, 4819 independent data ($R_{\text{int}} = 0.0322$); 316 parameters, $wR_2 = 0.0867$, $S = 0.989$, R_1 (3898 data with $I > 2\sigma(I)$) = 0.0326. CCDC 1422338.



Phys. Chem. Chem. Phys., 2016, 18, 30135–30143 | 30137



Scheme 1 Synthesis of compound 1.

obtained as dark red crystals from the reaction of H_2L with $Co(NO_3)_2 \cdot 6H_2O$ in DMF in a ratio of 1 : 1 in the presence of Et_3N (Scheme 1). The purity of the complex was confirmed by powder X-ray diffraction (Fig. S1, ESI†) and elemental analysis. The X-ray single-crystal structure analysis reveals that complex **1** crystallises in the monoclinic space group $P2_1/n$. As shown in Fig. 1a, the neutral molecule consists of one deprotonated Schiff-base ligand (L^{2-}) encapsulating one $Co(II)$ ion in a slightly distorted trigonal prismatic coordination environment. The top and bottom planes of the prism are formed by one piperazine nitrogen, and one nitrogen and oxygen of the corresponding Schiff base arms in such a way, that always one oxygen and one nitrogen of the different arms are on top of each other (Fig. 1b). The normals of these planes are tilted with respect to each other by 6° (Fig. 1b) and twisted by 3° (Fig. 1c). We assume that the tilting angle is caused by the piperazine unit. The two N atoms of piperazine (N_1 , N_2) are much closer to each other (2.47 Å) than the N–O pairs (2.81 Å). The Co–N_{piperazine} distances are longer than the Co–N_{imino} distances, and both of these Co–N distances are longer than the Co–O_{phenol} distances (Table 1). The Co(II) atoms are well-separated with an interatomic distance longer than 7.02 Å (Fig. 1d), thus precluding any significant inter-molecular magnetic interactions.

The electronic structure calculations reveal a splitting of the ionic 4F ground state of $Co(II)$ over the range of $10\,180\text{ cm}^{-1}$ as a result of the trigonal prismatic ligand field (see Table 2). The 4A_1 ground state is separated from the next states by about 1924 cm^{-1} (ACPF). Due to spin–orbit interaction, mainly with the second quartet state, the 4A_1 ground state splits into two Kramers doublets, E_1 and E_2 , with a spin–orbit splitting of 82 cm^{-1} (SOC*). The next excited states follow at 2000 cm^{-1}

Table 1 Selected bond lengths (Å) and bond angles ($^\circ$) of compound **1**

Co1–O3	2.0018 (12)	N4–Co1–N3	157.49 (6)
Co1–O1	2.0106 (12)	O3–Co1–N2	136.92 (5)
Co1–N4	2.1547 (14)	O1–Co1–N2	87.16 (5)
Co1–N3	2.1570 (14)	N4–Co1–N2	80.95 (5)
Co1–N2	2.2650 (14)	N3–Co1–N2	118.89 (5)
Co1–N1	2.2880 (14)	O3–Co1–N1	87.55 (5)
O3–Co1–O1	132.22 (6)	O1–Co1–N1	136.20 (5)
O3–Co1–N4	84.85 (5)	N4–Co1–N1	120.19 (5)
O1–Co1–N4	86.01 (5)	N3–Co1–N1	80.02 (5)
O3–Co1–N3	86.39 (5)	N2–Co1–N1	65.80 (5)
O1–Co1–N3	84.61 (5)	Co–Coadj	7.0151 (6)

Table 2 Low energy spectrum of complex **1** (the next states follow at $17\,839\text{ cm}^{-1}$ (CASSCF) and $17\,878\text{ cm}^{-1}$ (SOC*)); the values in parenthesis correspond to the shift of the CASSCF ground state energies by the difference of the ACPF correlation energies for the two lowest quartet states and the SOC* energies obtained with this shift

CASSCF	Energy (cm^{-1})	SOCI	Energy (cm^{-1})
4A_1	0 (–358) ^a	E_1	0
		E_2	101 (82) ^b
4A_1	1567	E_3	1675 (2001)
		E_4	1884 (2189)
4A_1	4667	E_5	4435 (4764)
		E_6	4773 (5102)
4A_1	4915	E_7	5186 (5516)
		E_8	5493 (5821)
4A_1	6340	E_9	6506 (6834)
		E_{10}	6641 (6969)
4A_1	9409	E_{11}	9357 (9686)
		E_{12}	9668 (9998)
4A_1	10 180	E_{13}	10 508 (10 838)
		E_{14}	10 699 (11 028)

^a ACPF energy shift of the ground state (see Experimental section).

^b The values in brackets correspond to the SOC* method calculations. The next states follow at $\sim 18\,000\text{ cm}^{-1}$.

(see Fig. 2). Analysis of the SOC* wave functions shows that the lowest two Kramers doublets consist of 90% contributions of the 4A ground state. Thus, the low-lying energy spectrum, consisting of the two Kramers doublets, can be described by a pseudo-spin of $S = 3/2$ using the spin Hamiltonian $H = D(S_z^2 - S(S+1)/3) + E(S_x^2 - S_y^2)$. The numerical effective Hamiltonian is given in Table 3. The resulting D tensor in cm^{-1} is:

$$D = \begin{pmatrix} 25 & 0 & 0 \\ 0 & 24 & 1 \\ 0 & 1 & -16 \end{pmatrix}$$

Diagonalisation of the D -tensor yields strong axial anisotropy with $D = -41\text{ cm}^{-1}$ and $|E/D| = 0.006$ (for the g -factors, see magnetic properties).

In an ideal trigonal prismatic coordination, the first two quartet states in the CASSCF calculation as well as the two orbitals at 750 cm^{-1} and 1300 cm^{-1} in Fig. 3, should be degenerate. In D_{3h} symmetry, the ground state is an eightfold degenerate $^4E'$. This state is split by first order spin orbit coupling into four Kramers doublets. We expect that the axial zero field splitting is increased if the distortion of the coordination is reduced. We also note that

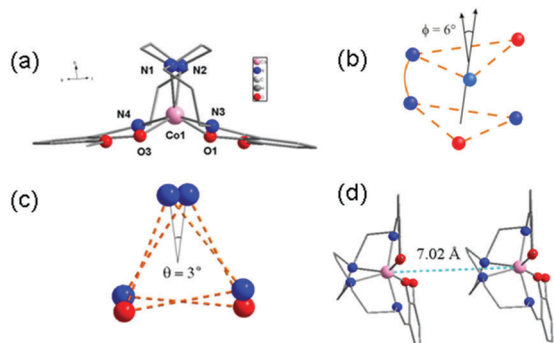


Fig. 1 (a) The molecular structure of **1** (hydrogen atoms have been omitted for clarity). The calculations show that the easy axis points into the plane of the paper. (b and c) Tilt- and twist angles of the coordination polyhedron with respect to ideal prismatic symmetry. (d) The shortest distance of Co(II) ions between neighbouring clusters.



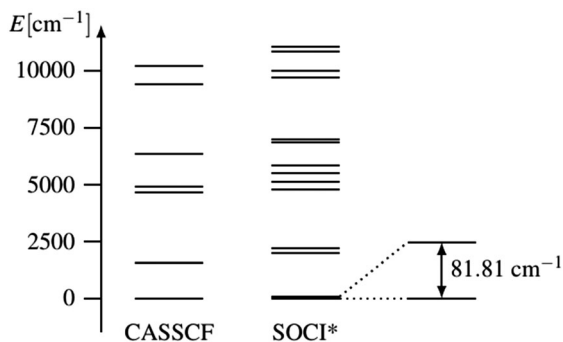


Fig. 2 Lowest electronic states obtained from the *ab initio* calculations. The CASSCF excitation energies are shifted by the ACPF correction (see Table 2).

Table 3 Numerical effective Hamiltonian in the formal spin functions of $S = 3/2$ for the SOCI* method, the energies used in the Hamiltonian are given in cm^{-1}

M_s	$-3/2$	$-1/2$	$1/2$	$3/2$
$-3/2$	$0.06 + 0.00i$	$-0.34 - 2.05i$	$0.48 - 0.04i$	$0.00 + 0.00i$
$-1/2$	$-0.34 + 2.05i$	$81.75 + 0.00i$	$0.00 + 0.00i$	$0.48 - 0.04i$
$1/2$	$0.48 + 0.04i$	$0.00 + 0.00i$	$81.75 + 0.00i$	$0.34 + 2.05i$
$3/2$	$0.00 + 0.00i$	$0.48 + 0.04i$	$0.34 - 2.05i$	$0.06 + 0.00i$

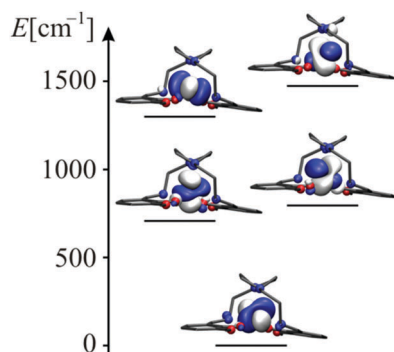


Fig. 3 Natural orbitals of the state-average CASSCF calculation.⁴² On the energy axes, the diagonal elements of the CASSCF Fock-matrix are shown.

in the electronic spectrum of **1** in the UV/Vis range the ligand bands dominate and it is not possible to extract information regarding the d–d transitions (Fig. S4, ESI†).

To gain deeper insights we performed calculations on model complexes where we started from the distorted structure, fixed the O and N atoms of the first coordination sphere and saturated them by H atoms (water and ammonia ligands, respectively). Although modeling nitrogen donors with ammonia and oxygen donors with water is something of a simplification in terms of what the second coordination sphere of the ligand field can provide for further “ultra fine-tuning”, this is also standard practice in theoretical approaches for describing 3d complexes.

Starting from model complex (model a), we went in three steps to the idealised trigonal prismatic coordination (Table 4, see details in ESI†), by first assimilating the angles of the two top and bottom triangles (model b), then making the corresponding

Table 4 Energy differences of the lowest quartet states (ACPF) and the lowest two Kramers doublets (SOCi*), g factors and zero field splitting parameters for the model complexes. In the models, the first coordination sphere is symmetrized in different steps (see text) from the original complex to an ideal trigonal prismatic coordination. In each step, the structure of the previous model is visualized by the grey polyhedron. The asterisk behind the g -values of the second Kramers doublet indicates the direction of the z -axis

Model	ΔE [cm^{-1}]	ΔE [cm^{-1}]	D [cm^{-1}]	E/D	First Kramers doublet		Second Kramers doublet		
	ACPF	SOCI*			g_{\parallel}	g_{\perp}	g_1	g_2	g_3
a	2518	87	−44	0.01	7.9	0.1	4.4	4.3	2.9*
b	1982	124	−62	0.04	8.4	0.3	4.5	4.1	3.0*
c	1591	153	−76	0.05	9.0	0.3	4.5	4.0	3.2*
d	826	233	−129	0.01	9.6	0.1	4.5*	3.4	3.4
e	0	302	−151	0	9.9	0	5.9*	0.0	0.0
f	0	318	−159	0	9.8	0	5.8*	0.0	0.0

planes coplanar (model c) and finally imposing full D_{3h} symmetry to the positions of Co and the first coordination sphere regardless of the kind of atom (model d) and the homoleptic derivatives (models e, and f). In the cases b–f, bond lengths and angles were averaged. For all these model complexes, we calculated the energy of the lowest two quartet states at ACPF level and the lowest two Kramers doublets in SOCI* calculations.

As expected, the two complexes with D_{3h} symmetry (e and f) show the strongest splitting because of first order spin–orbit coupling. The four lowest Kramers doublets which correspond to the $^4E'$ state are located at *ca.* 0 cm^{-1} , 300 cm^{-1} , 700 cm^{-1} , and 1000 cm^{-1} , respectively. With increasing distortion, the energies of the second quartet states are systematically increased. As a result, spin–orbit coupling becomes a second order effect and the splitting of the first quartet state into two Kramers doublets is diminished. We also note that an equally strong reduction of



the zero-field splitting was observed in a study on a Ni(II) complex with large magnetic anisotropy.⁴³

Table S3 and Fig. S5 (ESI†) give the detailed comparison of the results. In particular the zero-field splitting of the first two Kramers doublets is influenced by the details of the calculations. For CASSCF we have a ZFS of 101 cm⁻¹, with two correlated states 82 cm⁻¹ and for seven correlated states 85 cm⁻¹. The latter two values are rather similar and this points to the fact that the underlying physics remains the same for all three methods and although the agreement with the experimental data is not excellent it is in the range of what can be expected from a quantum chemical calculation.

For fields up to ca. 4 T the agreement for the field dependence of the calculated and measured magnetisation is surprisingly good – calculated and measured susceptibilities differ only by ca. 12%. Taking into account that the susceptibilities are obtained as numerical second derivatives and without any fitting parameter the agreement with the experimental values is far better than might be expected and underlines the rather unusual behaviour of the Co(II) ion – but then it always was a “Kobold”.

Magnetic properties

Magnetic measurements were performed on polycrystalline powder samples. As shown in Fig. 4 (left), the χT value at 300 K is 2.46 cm³ K mol⁻¹ and is higher than the expected value of 1.875 cm³ K mol⁻¹ for one isolated Co(II) ion ($S = 3/2$) centre with $g = 2$ but falls within the range 2.1–3.4 cm³ K mol⁻¹ typical for a single non-interacting high spin d⁷ Co(II) ion with a considerable orbital angular momentum contribution. The χT value remains roughly constant in the high temperature range (300–50 K), then decreases abruptly to 1.83 cm³ K mol⁻¹ at 2.0 K due to zero field splitting of the Co(II) ion. The magnetisation was measured up to 7 T dc field at 2, 3, and 5 K. The lack of saturation of the magnetisation at higher field (Fig. 4) also implies the presence of significant magnetic anisotropy. The widely used spin Hamiltonian given in eqn (9) describes the magnetic anisotropy qualitatively:

$$H = D(S_z^2 - S(S+1)/3) + E(S_x^2 - S_y^2) + g\mu_B S \cdot B \quad (9)$$

where μ_B denotes the Bohr magneton and D , E , S and B represent the axial and rhombic zero-field splitting (ZFS) parameters, the spin, and the magnetic field vector, respectively. The magnetisation was fitted using the *ANISOFIT 2.0* program (Fig. 4).⁴⁴ For a spin of $S = 3/2$, the obtained D value was -31 cm⁻¹ with the corresponding E and g being 0.0004 cm⁻¹ and 2.47, respectively.

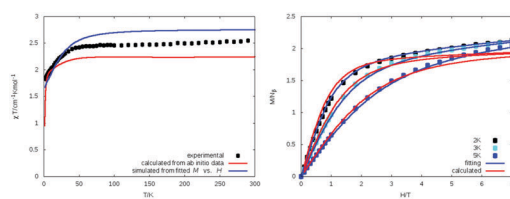


Fig. 4 Experimental, fitted, and calculated plots of χT vs. T (left) and of M vs. H for **1** (right). The χT plot was simulated directly from *ab initio* energies without using a spin Hamiltonian.

The combination of the large negative D value and the very small value of E indicates that **1** has a large uniaxial anisotropy. No reasonable optimisation was obtained when the initial D value was set to a positive value, indicating the correct choice of the negative sign. However, the fitting parameters obtained from the low temperature magnetisation data are not suitable for a simulation of the magnetic susceptibilities (Fig. 4) suggesting a conceptual problem with the use of eqn (9) for describing the magnetic properties of **1**.

In an alternative, and as it turns out, very useful approach the χT characteristics as well as the field dependent magnetisation data (M vs. H) were simulated directly from the SOCI* calculations allowing for the presence of a finite magnetic field and without the use of any effective spin Hamiltonian (Fig. 4). In these calculations, all 3d⁷ states are involved, but only the lowest two Kramers doublets are populated in the Boltzmann distribution. The simulated curves are in a surprisingly good agreement with the experimental data obtained from the magnetisation measurements. This is a further demonstration of the importance of the large magnetic anisotropy of **1** in terms of its affect on the ground and excited state within this compound.⁴⁵

In order to calculate the g -factors and magnetic axes, two different approaches were considered (see Quantum chemical methods for details). In the first approach the Zeeman interaction was included in the SOCI using finite perturbation theory.²⁵ In the second method, the Zeeman splitting is taken into account using first-order degenerate perturbation theory within the ground-state Kramers doublet.³⁰ The first method has two advantages. Firstly, it can be used for any spin multiplicity whereas the second approach is only valid for well-isolated pseudo-spins. Secondly, the response of the wave functions to the magnetic field is included in the calculations and higher order terms can be analysed and this is important when considering the availability of excited states in terms of tunnelling and other relaxation processes. On the other hand, the second method has the advantage that numerical differentiation is avoided and the g -values can be directly obtained from the Zeeman integrals.

Resulting from this, the calculated g -factors using the SOCI wave functions are based on a pseudo-spin of $S = 1/2$ and confirm the high uniaxial anisotropy of **1** with $g_1 = 8.12$, $g_2 = 0.59$, $g_3 = 0.11$, for method 1 and $g_1 = 8.14$, $g_2 = 0.084$ and $g_3 = 0.073$ (Table S3, ESI†) for method 2. The obtained easy axes of these methods differ only by $\sim 4^\circ$ which is in line with the grid resolution of method 1. As expected, the direction of the easy axis is along the C_3 axis of the trigonal prism, *i.e.* perpendicular to the plane of the paper in Fig. 1a. Since the g -factors are very small in the plane perpendicular to the easy axis, the other two axes are not well-defined. The isotropic g -factors amount to 2.94 and 2.74, respectively. The energy differences for the determination of g_2 and g_3 in the finite field approach are rather small, *i.e.* on the limit of the numerical accuracy. Therefore, we assume that the values for g_2 and g_3 obtained by method 2 (directly from the Zeeman integrals) are more reliable in the present case. The principal values of the G -tensor become smaller upon inclusion of dynamic correlation effects (SOCI*) *i.e.* $g_1 = 7.86$, $g_2 = 0.04$, $g_3 = 0.05$. For the second Kramers doublet



we obtained $g_1 = 2.70$, $g_2 = 4.35$ and $g_3 = 4.39$, again with a pseudo-spin of $S = 1/2$. For both Kramers doublets, g_1 is parallel to the easy axis. From these results, it becomes apparent why the description of the magnetic susceptibility by the spin Hamiltonian given in eqn (9) with $S = 3/2$ fails for higher temperatures (Fig. S5, ESI†). Here, the second Kramers doublet becomes populated. In eqn (9), the same g -tensor is used to describe the magnetic behaviour of both Kramers doublets, but simply scaled with different M_s values. Nevertheless, at least in terms of the principal axis the Zeeman energies can be described using $g \approx 8/3$ and $S = 3/2$.

By comparing the g -values of the model complexes (Table 4), the effect of the local coordination symmetry can clearly be seen. As the models become more symmetric, the rhombicity in the D - and G -tensor diminish and eventually in model d these properties become axial. Finally, in the entirely D_{3h} symmetric complexes e and f, the anisotropy of the second Kramers doublet also becomes uniaxial with all anisotropy-axes (D -tensor, g -tensor of the first Kramers doublet and g -tensor of the second Kramers doublet) being collinear. However, because of the first order spin orbit coupling discussed above the Zeeman splittings of the first and second Kramers doublets do not show the 3 : 1 ratio implied by the spin Hamiltonian for $S = 3/2$.

In order to gain further insights and to probe the dynamic magnetic behaviour, ac susceptibility measurements were carried out on **1** in the temperature range 1.8–10 K under zero dc field and 3.0 Oe ac field oscillating at frequencies between 1 and 1500 Hz. A frequency-dependent signal was observed in the χ'' versus T plot below 10 K (Fig. S2, ESI†) suggesting slow relaxation of the magnetisation, generally attributed to a SMM behaviour. However, relaxation barriers cannot be extracted from this data due to the presence of a broad peak at higher frequencies indicating the presence of quantum tunnelling (QTM) effects. Therefore, ac susceptibility measurements were obtained at static dc fields from 300 Oe to 1500 Oe. When different dc fields are applied, the χ'' versus frequency signal at 4.3 K dramatically changes with the magnitude of the applied field (Fig. S3, ESI†). The broad signal observed without dc field shows a clear maximum and diminishes when the dc field is increased. At fields stronger than 600 Oe the maximum in the high frequency range disappears under our measurement conditions. Additionally, a new peak appears at about 10 Hz with the application of the dc field. With increasing strength of the dc field, the peak shifts towards lower frequencies and enhances in magnitude. This behaviour is similar to what was found for previously reported trigonal prismatic mononuclear Co(II) complexes.² The peak shift is negligible between 900 Oe and 1500 Oe, therefore, the variation of the χ'' versus frequency signal with the temperature was studied at 1200 Oe dc applied field. The χ'' versus frequency signal for complex **1** shows only one peak (Fig. 5) which is shifted to higher frequencies when the temperature is increased. This indicates a thermally activated regime and confirms the predicted SMM behaviour.

The field- and temperature-dependence of the measured ac data were analysed using a generalised Debye model^{36–38} (Fig. 6, left and Fig. S3, ESI†). The resulting relaxation times

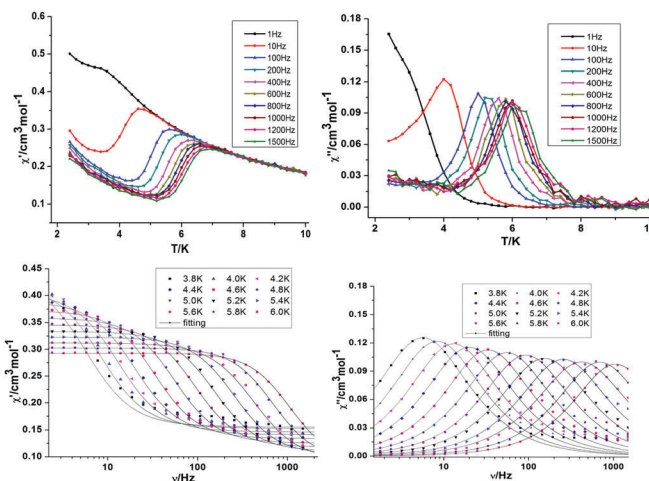


Fig. 5 Plots of χ' (left) and χ'' (right) vs. T (upper) at different frequencies for **1** in 1200 Oe dc field (solid lines are guides for the eye); Plots of χ' (left) and χ'' (right) vs. frequency (lower) at different temperatures in 1200 Oe dc field (solid lines show fitting as Lorentzian function).

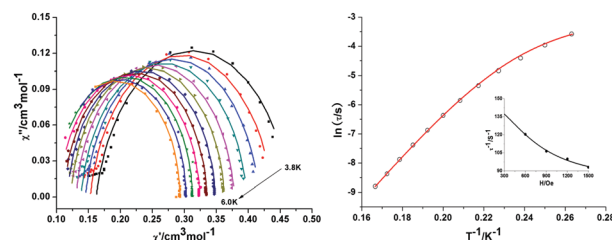


Fig. 6 Cole–Cole plots of data for **1** measured at 1200 Oe between 3.8 K and 6.0 K (lines for fitting) (left); Plots of $\ln(\tau)$ vs. T^{-1} . The red line shows the fit to the data using eqn (10) with $U = 56.65 \text{ cm}^{-1}$ at 1200 Oe dc field. Inset: Field dependence of the magnetic relaxation time, τ , at 4.3 K for a microcrystalline sample of **1** and its approximation by $\tau^{-1} = B_1/(1 + H^2 B_2) + O_H$. Parameters A , B_1 and B_2 have the same meaning as in eqn (10). The parameter O_H accounts for a field-independent offset at 4.3 K (right).

(see Tables S1 and S2, ESI†) were modelled using the expressions for spin–lattice relaxation given in the Experimental section.^{41,46}

Cole–Cole plots of χ' versus χ'' between 3.8 and 6.0 K (Fig. 6, left) have semicircular profiles, indicative of a single relaxation process. The plot was fitted with CC-Fit,⁴⁷ which uses a generalized Debye model.^{36–38} The extracted parameters (see the ESI†) are similar to those extracted from the χ' and χ'' data with $B_1 = 65.60 \text{ s}^{-1}$, and $B_2 = 1.88 \times 10^{-6} \text{ Oe}^{-2}$, respectively, together with a field-independent offset were used and found to be sufficient to model the field-dependence. For the temperature-dependent data, the QTM term itself was used as a constant offset. We thus identify an Orbach process displaying a typical SMM spin-reversal barrier of $U = 56.65 \text{ cm}^{-1}$ together with $\tau_0 = 2.24 \times 10^{-10} \text{ s}^{-1}$, as well as a Raman process with exponent $n = 9$ (typical for a second-order Raman process in an isolated Kramers pair),^{41,46,48} with a prefactor of $C = 1.00 \times 10^{-4} \text{ K}^{-9} \text{ s}^{-1}$. Thus, the full expression for the spin-relaxation containing all terms necessary to describe the observed behaviour of **1** (Fig. 6) is given by eqn (10):

$$\tau^{-1}(H, T) = B_1/(1 + H^2 B_2) + CT^9 + \tau_0^{-1} \exp(-U/kT). \quad (10)$$



The spin reversal barrier of 56.65 cm^{-1} is slightly lower than the calculated energy difference between the lowest Kramers doublets of 82 cm^{-1} . The deviation is typical for such compounds and primarily caused by quantum tunnelling and it is difficult to formulate general rules in terms of optimising SMM behaviour in Co(II) systems as a result of several physical effects influencing the performance of SIMs as reviewed recently by Atanasov *et al.*⁴ Nevertheless, an important message from the analysis we have performed is that the directions of the magnetic axes for the first two Kramers doublets are a key factor. We can show that when, as here, they are rather different and this difference is critically dependent on the distortion from an ideal prismatic coordination, this leads to significant changes in the spin-orbit coupling. That in turn provides a “fine-adjustment” in terms of tuning the parameters relevant to SMM behaviour.

Conclusions

In the mononuclear [Co^{II}(L)] compound the Co(II) ion has a slightly distorted trigonal prismatic coordination geometry. Magnetic studies demonstrate its slow magnetic relaxation even in the absence of an applied dc field and the relaxation barrier of 57 cm^{-1} is found when applying a small dc field. Because of other relaxation mechanisms, most likely QTM,⁴ the measured barrier is smaller than the spin-orbit splitting obtained from *ab initio* calculations. DC magnetic measurements are in good agreement with *ab initio* calculations but could not be modelled reasonably using the single-ion spin Hamiltonian eqn (9) for $S = 3/2$. Although ground state anisotropy is formally considered in this Hamiltonian, it is not sufficient on its own in cases where there are orbitally degenerate states. For the given compound and its models, the calculations show a very different behaviour of the lowest two Kramers doublets with application of an external magnetic field: Whereas the ground state is uniaxial, the first excited state shows a rhombic behaviour, which is not taken into account for in the Zeeman terms in eqn (9). Calculations on the model complexes indicate that axially for both states can be achieved through symmetry constraints within the first coordination sphere. The calculations further indicate that even small deviations from trigonal prismatic coordination reduce spin-orbit coupling and thus deteriorate the SMM behaviour significantly.

In future investigations, we will focus on the effects of structural distortion and coordination number on the magnetic behaviour of mononuclear Co(II) SIMs with a view to finding an optimised building block for the construction of mixed metal polynuclear single molecule magnets incorporating Co(II) ions.

Acknowledgements

We thank the DFG for funding within the collaborative research center SFB/TRR88 “3MET”, projects A1 and A3. T. B. thanks the Danish Research Council (FNU) for funding.

Notes and references

- 1 D. E. Freedman, W. H. Harman, T. D. Harris, G. J. Long, C. J. Chang and J. R. Long, *J. Am. Chem. Soc.*, 2010, **132**, 1224.
- 2 S. Gomez-Coca, E. Cremades, N. Aliaga-Alcalde and E. Ruiz, *J. Am. Chem. Soc.*, 2013, **135**, 7010.
- 3 J. D. Rinehart, M. Fang, W. J. Evans and J. R. Long, *J. Am. Chem. Soc.*, 2011, **133**, 14236.
- 4 M. Atanasov, D. Aravena, E. Suturina, E. Bill, D. Maganas and F. Neese, *Coord. Chem. Rev.*, 2015, **177**, 289.
- 5 W. Lin, T. Bodenstein, V. Mereacre, K. Fink and A. Eichhöfer, *Inorg. Chem.*, 2016, **55**(5), 2091.
- 6 R. Boča, *Coord. Chem. Rev.*, 2004, **248**, 757.
- 7 M. Dey and N. Gogoi, *Angew. Chem., Int. Ed.*, 2013, **52**, 12780.
- 8 P. P. Power, *Chem. Rev.*, 2012, **112**, 3482.
- 9 X.-C. Huang, C. Zhou, D. Shao and X.-Y. Wang, *Inorg. Chem.*, 2014, **53**, 12671.
- 10 L. Chen, J. Wang, J.-M. Wei, W. Wernsdorfer, X.-T. Chen, Y.-Q. Zhang, Y. Song and Z.-L. Xue, *J. Am. Chem. Soc.*, 2014, **136**, 12213.
- 11 T. Jurca, A. Farghal, P.-H. Lin, I. Korobkov, M. Murugesu and D. S. Richeson, *J. Am. Chem. Soc.*, 2011, **133**, 15814.
- 12 W. H. Harman, T. D. Harris, D. E. Freedman, H. Fong, A. Chang, J. D. Rinehart, A. Ozarowski, M. T. Sougrati, F. Grandjean, G. J. Long, J. R. Long and C.-J. Chang, *J. Am. Chem. Soc.*, 2010, **132**, 18115.
- 13 D. Weismann, Y. Sun, Y. Lan, G. Wolmershauser, A. K. Powell and H. Sitzmann, *Chem. – Eur. J.*, 2011, **17**, 4700.
- 14 J. M. Zadrozny, M. Atanasov, A. M. Bryan, C. Y. Lin, B. D. Reinken, P. P. Power, F. Neese and J. R. Long, *Chem. Sci.*, 2013, **4**, 125.
- 15 Y.-Y. Zhu, C. Cui, Y.-Q. Zhang, J.-H. Jia, X. Guo, C. Gao, K. Qian, S.-D. Jiang, B.-W. Wang, Z.-M. Wang and S. Gao, *Chem. Sci.*, 2013, **4**, 1802.
- 16 Y.-Y. Zhu, Y.-Q. Zhang, T.-T. Yin, C. Gao, B.-W. Wang and S. Gao, *Inorg. Chem.*, 2015, **55**, 5475.
- 17 V. V. Novikov, A. A. Pavlov, Y. V. Nelyubina, M. E. Boulon, O. A. Varzatskii, Y. Z. Voloshin and R. E. P. Winpenny, *J. Am. Chem. Soc.*, 2015, **137**, 9792.
- 18 Y. Rechkemmer, F. D. Breitgoff, M. van der Meer, M. Atanasov, M. Hakl, M. Orlita, P. Neugebauer, F. Neese, B. Sarkar and J. van Slageren, *Nat. Commun.*, 2016, **7**, 10467.
- 19 (a) G. M. Sheldrick, *Acta Crystallogr.*, 2015, **A71**, 3; (b) G. M. Sheldrick, *Acta Crystallogr.*, 2015, **C71**, 3.
- 20 F. Weigend and R. Ahlrichs, *Phys. Chem. Chem. Phys.*, 2005, **7**, 2397.
- 21 A. Schäfer, H. Horn and R. Ahlrichs, *J. Chem. Phys.*, 1992, **97**, 2571.
- 22 V. Staemmler, *Theor. Chim. Acta*, 1977, **45**, 89.
- 23 U. Meier and V. Staemmler, *Theor. Chim. Acta*, 1989, **76**, 95.
- 24 J. Wasilewski, *Int. J. Quantum Chem.*, 1989, **36**, 504.
- 25 K. Fink, C. Wang and V. Staemmler, *Inorg. Chem.*, 1999, **38**, 3847.
- 26 K. Fink and V. Staemmler, *Theor. Chim. Acta*, 1993, **87**, 129.
- 27 F. Neese, *J. Chem. Phys.*, 2005, **122**, 034107.
- 28 R. J. Gdanitz and R. Ahlrichs, *Chem. Phys. Lett.*, 1988, **143**, 413.



- 29 *Relativistic electronic structure theory, part 2: Applications*, ed. P. Schwerdtfeger, Elsevier, Amsterdam, 2004.
- 30 H. Bolvin, *ChemPhysChem*, 2006, **7**, 1575.
- 31 R. Maurice, R. Bastardis, C. de Graaf, N. Suaud, T. Mallah and N. Guihéry, *J. Chem. Theory Comput.*, 2009, **5**, 2977.
- 32 TURBOMOLE V 6.6 2014, a development of University of Karlsruhe and Forschungszentrum Karlsruhe GmbH, 1989-2007, TURBOMOLE GmbH since 2007, available from <http://www.turbomole.com>.
- 33 C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1988, **37**, 785.
- 34 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 35 S. Grimme, S. Ehrlich and L. Goerigk, *J. Comput. Chem.*, 2001, **32**, 1456.
- 36 H. B. G. Casimir and F. K. du Pré, *Physica*, 1938, **5**, 507.
- 37 P. Debye, *Polar Molecules*, Chemical Catalogue Company, New York, 1929.
- 38 K. S. Cole and R. H. Cole, *J. Chem. Phys.*, 1941, **9**, 341.
- 39 E. Colacio, J. Ruiz, E. Ruiz, E. Cremades, J. Krzystek, S. Caretta, J. Cano, T. Guidi, W. Wernsdorfer and E. J. K. Brechin, *Angew. Chem., Int. Ed.*, 2013, **52**, 9130.
- 40 E. Carl, S. Demeshko, F. Meyer and D. Stalke, *Chem. – Eur. J.*, 2015, **21**, 1.
- 41 K. N. Shrivastava, *Phys. Status Solidi B*, 1983, **117**, 437.
- 42 R. A. D. Wentworth, *Coord. Chem. Rev.*, 1972, **9**, 171.
- 43 M. Gruden-Pavlović, M. Perić, M. Zlatar and P. García-Fernández, *Chem. Sci.*, 2014, **5**, 1453.
- 44 M. P. Shores, J. J. Sokol and J. R. Long, *J. Am. Chem. Soc.*, 2002, **124**, 2279.
- 45 A. V. Palii, J. M. Clemente-Juan, E. Coronado, S. I. Klokishner, S. M. Ostrovsky and O. S. Reu, *Inorg. Chem.*, 2010, **49**, 8073.
- 46 A. Abragam and B. Bleaney, *Electron Paramagnetic Resonance of Transition Ions*, Clarendon Press, Oxford, 1970.
- 47 N. F. Chilton, CC-Fit, <http://www.nfchilton.com/cc-fit.html>.
- 48 R. Carlin, *Magnetochemistry*, Springer-Verlag, Berlin, 1986.

