Phosphorescent columnar hybrid materials containing polyionic inorganic nanoclusters†

S. K. Nayak, a M. Amela-Cortes, a M. M. Neidhardt, b S. Beardsworth, b J. Kirres, b M. Mansueto, b S. Cordier, a S. Laschat* b and Y. Molard*a

The ternary polyionic inorganic compound Cs$_2$Mo$_6$Br$_{14}$ and 18-crown-6 ethers bearing two o-terphenyl units have been combined to design phosphorescent columnar liquid crystalline hybrid materials. The obtained host–guest complexes are very stable even at high temperatures. Depending on their surrounding atmosphere, these hybrids switch reversibly from a high-to-low luminescence state and show a very stable emission intensity up to 140 °C.

Columnar liquid crystals (CLCs) have attracted much interest in materials chemistry since 1977 when Chandrasekhar discovered them during his studies of discotic benzene hexaalkanoates. These materials self-organise spontaneously usually into columns, show self-healing properties and have been shown to be useful in the design of many devices including photovoltaic solar cells, field effect transistors, one dimensional conductors and light emitting diodes. It was also found that non-disk-shaped molecules can form columnar mesophases. Recently, they were also used to organise small amounts of bulky inorganic components such as gold nanoparticles, gold nanorods, or quantum dots over a wide range. Addition of new functionalities to CLCs, e.g., phosphorescence, appears then, to be particularly appealing in terms of application prospects but still remains very challenging in terms of LC phase or emission stability. We present herein the first hybrid CLC containing up to 34.4 wt% of polyionic, bulky, transition metal cluster (TMC) salts. TMCs are aggregates of metal atoms linked by metal–metal bonds. In particular, octahedral TMC based compounds of the general formula A$_n$M$_6$X$_8$ where A = alkali cation, M = Mo or Re, X$^-$ = halogen or chalcogen, X$^\equiv$ = halogen or organic ligands are known for their phosphorescence with evidenced dual emission for Cs$_2$Mo$_6$Br$_{14}$. We recently developed three strategies to integrate these isotropic emitters in smectic or nematic and cholesteric LCs. However, up to now, a TMC containing CLC remains unknown. To design such self-assembling materials, we present herein a supramolecular approach in which the nanocluster alkali counter cations are complexed with 18-crown-6 ether derivatives CE$_n$ as well as the Cs$_2$Mo$_6$Br$_{14}$ cluster.
ternary salt, were prepared according to the reported procedure with conforming analytical data.\textsuperscript{10,12}

This TMC salt was chosen for its high thermal stability, a prerequisite for developing new classes of clusters containing thermotropic hybrid materials. Complexes were obtained by mixing 1 equivalent of Cs\textsubscript{2}Mo\textsubscript{6}Br\textsubscript{14} dissolved in acetone with 4 equivalents of CE\textsubscript{n} dissolved in dichloromethane. The reaction mixture was then stirred at 50 °C for 14 hours. After evaporation of solvents, the residue was taken up by dichloromethane to remove, by filtration through a microfilter, any traces of unreacted TMC, and heated under a high vacuum to yield the respective dicyclohexyl ether complexes \([2\text{CE}_n\cdot Cs]\text{Mo}_6\text{Br}_{14} = \text{CL}_n\). Their purity was verified by elemental analysis (ESI).\textsuperscript{1}H NMR experiments performed on free CE\textsubscript{n} and their respective complexes CL\textsubscript{n} did not reveal many changes due to the Cs\textsuperscript{+} complexation within the crown ether cavity (ESI,† in the dihedral angle of the OCH\textsubscript{2}CH\textsubscript{2}O fragment upon ion complexation to accommodate the bulky Cs\textsuperscript{+} cation and (2) an electric field caused by the presence of the salt. It is noteworthy that the complexation of smaller ions such as K\textsuperscript{+} within the macrocyclic cavity induced the opposite effect.\textsuperscript{13} As previously observed,\textsuperscript{9} it is mandatory to maintain a Cs:CE\textsubscript{n} ratio of 1:2 in order to maintain a full homogeneity of the cluster salt/CE\textsubscript{n} complex.

The LC properties of hybrids were studied by polarized optical microscopy (POM), differential scanning calorimetry (DSC, ESI,† Fig. S4–S6) and small and wide angle X-ray scattering (SAXS, WAXS, ESI,† Fig. S7–S14), the results of which are shown in Table 1. All complexes show anisotropic LC phases extending over a wide temperature range. This behaviour largely contrasts with the native CE\textsubscript{n} precursor, for which LC phases were observed at lower temperatures and over narrower ranges. However, the increase in the mesophase stability as compared to the parent CE\textsubscript{n} precursor is similar to that encountered when soft anions such as iodide or bridging anions such as hexafluorophosphate or thiocyanate are used to counterbalance the positive charge of a potassium cation embedded within the crown cavity.\textsuperscript{11} This unexpected result is of particular interest as host–guest complexes between the small 18-crown-6 cavity and the sterically demanding Cs\textsuperscript{+} cation are usually supposed to be weak. It could eventually be attributed to the polyionic character of the TMC salt and the strong interactions existing between the complexed Cs\textsuperscript{+} and [Mo\textsubscript{6}Br\textsubscript{14}]\textsuperscript{2−} dianion which insure a better stability of the supramolecular assemblies upon heating. Surprisingly, the DSC thermogram of CL\textsubscript{9} reveals, upon heating, the formation of two LC phases, whereas upon cooling, only one first order transition corresponding to the isotropic-to-LC phase transition is observed (ESI,† Fig. S5). Fig. 1 shows the optical fan-shaped focal conic textures observed upon heating for all samples, which are typical indicators of columnar mesophases. Heating CL\textsubscript{9} from 100 °C up to 160 °C did not lead to any noticeable change in the observed textures. Irradiating the samples with UV light illustrates their excellent homogeneity thanks to the well-known red NIR emission of TMC (vide infra). The same experiments performed at 200 °C also reveal a very good homogeneity showing that the interactions between the crown centre and Cs\textsuperscript{+} cations are strong enough to maintain the integrity of the complex even at very high temperatures (ESI,† Fig. S3).

SAXS and WAXS patterns recorded at 127 °C, 119 °C and 131 °C for CL\textsubscript{7}, CL\textsubscript{8}, and CL\textsubscript{9}, respectively, contained sharp and intense reflections in the small angle region. We first analysed the XRD data, to investigate possible lamello-columnar packing, examples of which were independently reported by Donnio, Ziesel and Marcelis for metallomesogens,\textsuperscript{13} donor–acceptor triads,\textsuperscript{14} side-chain LC polymers\textsuperscript{15} or triazines with pendant triphenylenes.\textsuperscript{16} We first checked whether sharp Bragg reflections indicating a smectic order were present for complexes CL\textsubscript{n}. However, neither characteristic layer reflections in the SAXS diffractogram nor the second halo in the wide angle region were visible. Thus, a lamello-columnar phase or a coexisting lamellar and columnar packing could be ruled out.

Hence, the indexation of observed patterns allowed the identification of a rectangular columnar phase with p2mm symmetry.

![Fig. 1](image-url) Polarized Optical Micrographs of (a) CL\textsubscript{9} (T = 150 °C), (b) CL\textsubscript{8} (T = 163 °C) and (c) CL\textsubscript{7} (T = 150 °C) under white light (top) and UV irradiation (bottom).

### Table 1: Thermodynamic and X-ray diffraction data for CL\textsubscript{n} compounds

<table>
<thead>
<tr>
<th>Transition</th>
<th>T/°C</th>
<th>ΔG/ΔH</th>
<th>Mesophase</th>
<th>Lattice spacing/Å</th>
</tr>
</thead>
<tbody>
<tr>
<td>CL\textsubscript{9}</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cr → Col\textsubscript{9}</td>
<td>129.8</td>
<td>7.28\textsuperscript{a}</td>
<td>Col\textsubscript{9}</td>
<td>a = 66.3</td>
</tr>
<tr>
<td>Col\textsubscript{9} → I</td>
<td>150.8 (148.7)\textsuperscript{c}</td>
<td>2.36\textsuperscript{b}</td>
<td>131 °C p2mm</td>
<td>b = 44.4</td>
</tr>
<tr>
<td>CL\textsubscript{9}</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>g → Col\textsubscript{9}</td>
<td>91.6</td>
<td>0.43\textsuperscript{c}</td>
<td>Col\textsubscript{9}</td>
<td>a = 62.5</td>
</tr>
<tr>
<td>Col\textsubscript{9} → Col\textsubscript{9}</td>
<td>146.6</td>
<td>2.54\textsuperscript{b}</td>
<td>119 °C p2mm</td>
<td>b = 43.8</td>
</tr>
<tr>
<td>Col\textsubscript{9} → I</td>
<td>164.4 (162.2)\textsuperscript{c}</td>
<td>2.50\textsuperscript{b}</td>
<td>150 °C Z = 1</td>
<td>h = 4.5</td>
</tr>
<tr>
<td>CL\textsubscript{8}</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>g → Col\textsubscript{8}</td>
<td>96.8 (96.9)</td>
<td>0.141\textsuperscript{a}</td>
<td>Col\textsubscript{8}</td>
<td>a = 63.1</td>
</tr>
<tr>
<td>Col\textsubscript{8} → I</td>
<td>159.2 (147.9)\textsuperscript{c}</td>
<td>2.19\textsuperscript{b}</td>
<td>127 °C p2mm</td>
<td>b = 43.0</td>
</tr>
</tbody>
</table>

\textsuperscript{a} In kJ mol\textsuperscript{−1} K\textsuperscript{−1}. \textsuperscript{b} In kJ mol\textsuperscript{−1}. \textsuperscript{c} Observed on cooling from the isotropic state; Cr: crystalline state; Col\textsubscript{r}: columnar rectangular; Col\textsubscript{h}: columnar hexagonal; I: isotropic state.
for the three complexes (see Tables S1–S4, ESI† for a full indexation). All WAXS patterns contained a broad and large signal around 4.4 Å deriving from the fluid-like behaviour of the molten alkyl chains and from the stacking of the aromatic units within the columns confirming the liquid crystalline nature of all complexes. For CL₉, additional measurements were performed at 150 °C to characterize the unexpected second mesophase detected via DSC upon heating. Three signals with reciprocal d-spacings in the ratio 1:1/√3:1/2 were observed in the SAXS pattern and are in good accordance with the formation of a 2D hexagonal lattice with a p6mm symmetry (Col₉). In this case, the unit cell parameter (a = 35.5 Å) was calculated by looking at the position of the most intense scattering signal assigned to the (10) reflection. A comparison of previously published XRD data for metal-free crown ether CE₁₄ with those for CLₙ (Table 1) revealed that the symmetry of the Colₙ lattice changed upon complexation. While the lattice parameter a remains relatively unaffected, i.e. 66.1 Å (CE₁₄) vs. 66.3 Å (CL₉) despite the difference of 2 CH₂ units in the side chain, the b values almost doubled from 25.1 Å (CE₁₄) to 44.4 Å (CL₉). Halos were in a similar range (4.6 Å vs. 4.0 Å). To rationalize these results, we propose that the nanocluster, because of its bulkiness, cannot be accommodated within the c2mm lattice of CE₁₄, resulting in a p2mm lattice with an elongated b axis, so that the cluster fits in the centre of the unit cell and is surrounded by 4 stacks of crown ethers CE₉ with an intracolumnar distance of 4 Å (Fig. 2). This results in a height of 12 Å for the unit cell and the (1:4) cluster/CE₉ stoichiometry is maintained. In that case the inorganic moieties volumic fraction is around 9 vol%. For the corresponding Col₉ phase of CL₉, the observed a value should allow a hexagonal packing of the discotic crowns CE₉, while each Mo₆Br₁₄²⁻ unit is encapsulated by three crowns and leads to a volumic fraction of 22 vol% for Cs₂Mo₆Br₁₄.

As expected, the emission properties of [Mo₆Br₁₄]²⁻ are retained within the hybrid materials (see ESI† Fig. S15). The photoluminescence spectra (λₑₓ = 380 nm) show a broad and intense band extending from 550 nm up to 950 nm. The influence of the hybrid nanostructures on the inorganic emitter was investigated by temperature-dependent luminescence studies. Samples were heated up to the isotropic state and slowly cooled to reach a nanostructured glassy or crystalline state at 25 °C prior to measurements. The phosphorescence signals were collected at every degree upon heating (see the ESI† Fig. S16 and S17). As previously reported, if the emission intensity variation upon heating is only due to an increase of non-radiative deactivation processes, it should follow an exponential law of the type: I(T) = I₀ exp(−A/T) where A is constant and I₀ the emission intensity at a given temperature. Thus, reporting ln[I(T)/I₀] should give a straight line and this is what is observed for Cs₂Mo₆Br₁₄ (Fig. 3). Surprisingly, increasing the temperature from 25 °C up to 140 °C does not have much effect on the ability of the hybrid materials to emit light as an emission intensity loss of only less than 10% was recorded (the intensity loss at 140 °C is about 77% for Cs₂Mo₆Br₁₄ in its powder form). At higher temperatures, the obtained curves significantly deviate from the straight line with a similar behaviour for CL₇ and CL₉. The separation of the clusters from each other by isolating triangular arrays of crown ethers might be the reason for the increased luminescence intensity of CL₉ in the high temperature Col₉ phase as compared to the low temperature Col₉ phase. Presumably, the crown ethers act as an isolating matrix for luminescent clusters, making these hybrids attractive alternatives with regards to polymers for applications requiring temperatures around 100 °C. Thus, the columnar nanostructures have a strong influence on the emissive ability of hybrids despite the isotropic nature of the inorganic emitter. Let us emphasize that such unprecedented high stability of the emission intensity vs. temperature is particularly rare and has never been reported so far for TMC or TMC-containing hybrid materials.

As illustrated by absolute quantum yield measurements performed in an air or N₂ atmosphere, the emission intensity can also
be modulated by modifying the local atmosphere of the material. In all cases, a reversible and instantaneous switch from 0.05 or 0.07 in air up to 0.20, 0.26 or 0.30 in an N₂ atmosphere was measured for CL₈, CL₉ and CL₁₀, respectively (Fig. S18 and Video S1, ESIF). This phenomenon, which shows that these materials might be used as efficient oxygen sensors, is due to the O₂ permeability of CLₙ and is based on the ability of the excited states of the phosphorescent metal clusters to be quenched by molecular triplet oxygen, thus creating singlet oxygen.²¹ Although we demonstrated recently the same effect when nanoclusters are introduced into gas-permeable polymer matrices,²² this is the first time that such behaviour is reported in a liquid crystalline material.

In conclusion, we show herein, that nanometric luminescent transition metal cluster compounds are easily integrated in a CLC hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach. In these materials, Cs⁺ cluster unit counter cations are complexed within hybrid material by using a supramolecular approach.

References