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# A competitive immunoassay system for microfluidic paper-based analytical detection of small size molecules†

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The development of a competitive immunoassay system for colorimetric detection on microfluidic paper-based analytical devices (μPADs) is reported. The μPADs were fabricated via photolithography to define hydrophilic flow channels and consisted of three main elements: the control and test zones, where target detection was performed, the sample introduction zone, and the competitive capture zone located between the sample introduction zone and the test zone. The chromogenic substrate 3,3',5,5'-tetramethylbenzidine (TMB) was deposited at the control and test zones. μPAD surface modification was performed at the capture zone first via chitosan activation, then the BSA-conjugated target compound was immobilized. The sample solution consisting of the target compound, the peroxidase-conjugated antibody, and the hydrogen peroxide oxidizing agent was introduced into the device and competition occurred at the capture zone, allowing only the target-bound peroxidase-conjugated antibody to travel past the capture zone and into the test zone via capillary action. The developed competitive immunoassay system was successfully demonstrated on the  $\mu PAD$  detection of biotin as a model compound with a detection limit of 0.10 µg mL<sup>-1</sup>. The applicability of the proposed immunoassay system for point-of-need testing was further demonstrated using aflatoxin B<sub>1</sub>, a highly toxic foodborne substance, with a detection limit of 1.31 ng mL $^{-1}$ . The  $\mu$ PAD with the competitive immunoassay format showed promising results for practical applications in point-of-need testing involving small molecular weight targets in food and water safety and quality monitoring, environmental analysis, and clinical diagnostics.

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## Introduction

Studies involving the development of microfluidic paper-based analytical devices (µPADs) have become an area of growing interest specifically in the fields of clinical diagnostics, environmental monitoring, and food and nutrition safety.

Since the pioneering work of Whitesides' group in 2007 involving multiplexed diagnostic detection using a photoresistpatterned paper device, 1 µPADs have emerged as a tool for rapid and easy-to-perform detection that requires small amounts of reagents with little to no external supporting equipment or power, making the µPADs suitable for point-of-need detection that may not even require trained personnel to perform the measurement.<sup>2-5</sup> Using such μPADs for specific target detection, several methods have been incorporated including colorimetric,6-10 electrochemical,11-15 fluorescence,16-18 chemiluminescence, 19-22 and electrochemiluminescence<sup>23</sup> methods. Among these, colorimetric methods offer the simplest means to display detection results without the use of additional read-out devices since a simple production of color, or a change in color intensity as a result of the presence or absence of a target compound is easily obtained. Such colorimetric methods include immunoassays, 24-26 which involve the recognition and binding of antibodies to specific molecules in what might be a complex mixture of molecules, providing high specificity for target detection. In most applications, proteins and other large biomolecules including bacteria are the targets for

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detection.<sup>27–30</sup> In such cases, a sandwich immunoassay is highly applicable for detection, as in the works of Wang et al. 31 and Ge et al., 32 wherein, sandwich ELISA detection systems on μPAD have been demonstrated for the clinical diagnosis of several tumor markers including α-fetoprotein, cancer antigen, and carcinoembryonic antigen (CEA). In food and water safety and quality monitoring, however, with the exception of foodborne and waterborne microorganisms, the targets usually involve small molecular weight compounds.33-37 With small sized targets, a competitive immunoassay format is more applicable.

In our previous report, 38 we demonstrated a simple colorimetric assay system of horseradish peroxidase (HRP) on µPADs. The color production reaction was based on the oxidation of the chromogen, 3,3',5,5'-tetramethylbenzidine (TMB), by hydrogen peroxide, H<sub>2</sub>O<sub>2</sub>, in the presence of the HRP enzyme. In the present work, a similar assay system has been applied. Here, however, an antigen-antibody interaction is involved, the antigen being of a small molecular weight molecule; hence, a competitive immunoassay system using uPADs fabricated via photolithography is designed as the platform. Liu et al.39 demonstrated a laminated paper-based analytical device for the chemiluminescence immunoassay of cotinine (FW 176.2), a second-hand smoke biomarker. Their competitive immunoassay platform consisted of a detection zone, wherein antibodies were immobilized and the sample solution containing HRP-labelled cotinine and the target cotinine was introduced, and a reagent pad with luminol reagent was introduced for the μPAD chemiluminescence detection via the HRP-catalyzed oxidation of luminol. The reported format, however, has the disadvantage of requiring an additional washing step after introducing the sample solution to the µPAD, which might possibly cause some loss of the target compound before detection and quantification. Hence, a further reconfiguration of the competitive immunoassay system on the µPAD is necessary.

Here, we report the development of a competitive immunoassay system for the µPAD detection of small sized target molecules. The µPADs consist of control and test zones, where the TMB chromogen is deposited, a competitive capture zone, where the capture reagent is immobilized, and a sample zone, where the sample solution is introduced into the device. We successfully demonstrated good performance of the competitive immunoassay system using biotin as a model compound and aflatoxin B<sub>1</sub> as a target compound for practical application of the system on µPADs. To the best of our knowledge, this is the first demonstration of such a type of competitive immunoassay system on microfluidic paper-based devices.

# Experimental

## Reagents

All reagents were of analytical grade. 35.7 mM of 3,3',5,5'-tetramethylbenzidine chromogen reagent (Dojindo Laboratories) was prepared with acetonitrile. 1× blocker bovine serum albumin in phosphate-buffered saline (BSA-PBS) (Thermo Fisher Scientific Inc.) was diluted with 1× phosphate-buffered

saline Tween® 20 (PBST), pH 7.5 (Thermo Fisher Scientific Inc.), which was used as the washing solution during immunoassay. For the biotin immunoassay, 200 μg mL<sup>-1</sup> biotin stock solution (Sigma-Aldrich, Inc.) and 10 mg mL<sup>-1</sup> biotin-BSA conjugate (Sigma-Aldrich, Inc.) were prepared by dissolving the solids separately in PBST. For the aflatoxin B<sub>1</sub> (AFB<sub>1</sub>) immunoassay, 1 mg AFB<sub>1</sub> powder (Sigma-Aldrich, Inc.) was dissolved in 1 mL acetonitrile, and 1 mg mL<sup>-1</sup> AFB<sub>1</sub>-BSA conjugate (Sigma-Aldrich, Inc.) was dissolved in PBST.

#### Fabrication of µPADs

The µPADs were fabricated by photolithography as described in our previous work,38 with a slight modification of the first reported method.<sup>1</sup> The fabrication procedure includes the following steps: (1) soaking of Whatman<sup>TM</sup> filter paper grade 41 (GE Healthcare Life Sciences) or Ahlstrom grade 319 paper (Ahlstrom Corporation) in an SU-8 2010 photoresist (Microchem) for about 30 s; (2) spinning for 5 s at 500 rpm, then for 30 s at 2000 rpm to remove excess photoresist using a spin coater (Mikasa MS-A100); (3) prebaking for 5 min at 95 °C; (4) aligning under a photomask (designed using AutoCAD 2015 (Autodesk, Inc.), and then ordered from Unno Giken Co., Ltd for printing with a resolution of 12 700 dpi) using a mask aligner (Mikasa M-1S) after cooling to room temperature for 30 s and before being exposed to UV radiation for 18 s; (5) post-baking for another 5 min at 95 °C; and (6) developing (SU-8 developer, Microchem) for 6 min, then washing 3 times with 2-propanol. The fabricated µPADs were dried with high-pressured air and then stored in a sealed plastic bag until use. The μPADs fabricated using the Whatman<sup>TM</sup> filter paper grade 41 were labelled as FP41 and those fabricated with Ahlstrom grade 319 paper were labelled as A319.

#### Preparation of µPADs for competitive immunoassay

Before the immunoassay, one side of the fabricated µPADs was bonded to acrylic double adhesive tape without removing the rayon of the other side of the tape to limit the reagents to the hydrophilic regions and to avoid leakage (Fig. 1A). The capture zone was then chitosan-activated as described previously<sup>28</sup> before immobilizing the capture reagents. In brief, 0.25 mg mL<sup>-1</sup> of chitosan (Wako Pure Chemical Industries Ltd) solution was prepared by dissolving the flakes in hot (80-90 °C) aqueous solution of 0.05 M HCl, and then adjusting the pH to 3.5-5.0 with sodium hydroxide solution after cooling to room temperature. 0.6 µL of the chitosan solution was added to the capture zone and allowed to dry for at least 5 min. Then, chitosan was activated by adding 0.6 µL of 2.5% glutaraldehyde in PBST and incubated for 1 h at room temperature. After incubation, the capture zone with glutaraldehyde-activated chitosan was washed three times with PBST, and then sequentially wiped by simply pressing a cellulose absorbent sheet on top of the µPADs with the washing solution before allowing it to dry for at least 5 min. For the biotin immunoassay, a total of 6.0 μL of 10 mg mL<sup>-1</sup> of the biotin-BSA conjugate was added 10 times (0.6 μL volume each time) onto the capture zone for immobilization. For the AFB<sub>1</sub> immunoassay, a total of 10 μL of

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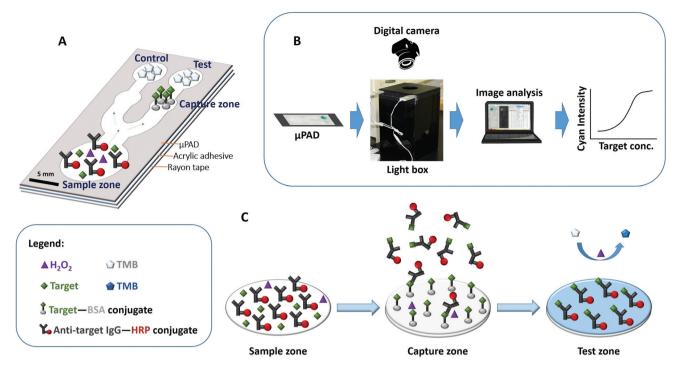


Fig. 1 (A) Schematic illustration of the competitive immunoassay system on μPAD. (B) Process of data acquisition using the digital camera and data quantification using ImageJ software. (C) Schematic illustration of the competitive immunoassay on μPAD.

1.0 mg mL<sup>-1</sup> capture AFB<sub>1</sub>-BSA was added 10 times (1 μL volume each time) onto the capture zone for immobilization. After incubation for 20 min, the capture zone was similarly washed three times with PBST and then sequentially wiped. Then, 1.4 µL of 35.7 mM TMB solution (50 nmol TMB) was added onto the test and control zones of the µPADs and they were allowed to air-dry for at least 2 min before being blocked with 100 µL of BSA-PBS solution for 20 min. After blocking, the μPADs were washed three times with PBST washing solution (75 µL each time), sequentially wiped with the cellulose absorbent sheet, and then allowed to air-dry at room temperature. In real-world point-of-need applications, reagent immobilization on the respective µPAD zones should be completed during device manufacture. An acceptable shelf life of antibodies and reagents on a paper device has been proven possible in commercially available pregnancy and drug test strips. Then, the only step necessary in the field should be the application of the sample in the liquid form to the sample introduction zone, allowing the reconstitution of other immunoassay reagents. 40

Competitive immunoassay of biotin on  $\mu PADs$ . For the biotin immunoassay, the biotin standard solutions were composed of increasing concentrations (0–500 ng mL<sup>-1</sup>) of the standard, 1:500 dilution (1.6  $\mu g$  mL<sup>-1</sup>) of anti-biotin IgG-peroxidase (Sigma-Aldrich, Inc.), and 0.001% (v/v) hydrogen peroxide solution. 50  $\mu L$  each of the biotin standard solutions were introduced separately on the  $\mu PADs$  for colorimetric detection. The images of the  $\mu PADs$  were captured using a digital camera (EOS Kiss X6i Canon), and then analyzed using ImageJ software.

Competitive immunoassay of aflatoxin  $B_1$  on  $\mu PADs$ . For the AFB<sub>1</sub> immunoassay, the standard solutions were composed of increasing concentrations of the AFB<sub>1</sub> standard, 1:25 000 dilution (0.163 pmol or 0.68  $\mu g$  mL<sup>-1</sup>) of anti-AFB<sub>1</sub> IgG-peroxidase, and 0.001% (v/v) hydrogen peroxide solution. 50  $\mu$ L each of the AFB<sub>1</sub> standard solutions were introduced separately on the  $\mu$ PADs for colorimetric detection. The images were captured using the digital camera and analyzed using ImageJ software.

#### Image analysis for colorimetric measurements

The blue color intensities produced during the competitive immunoassay were captured using the digital camera equipped with a standard 18-55 mm objective lens. For biotin measurements, the photos were taken using the ISO AUTO Close-up Mode with an exposure time of 1/100 s, an aperture of F/5, an ISO of 100, a focal distance of 42 mm, and color representation in standard RGB (sRGB). The µPADs were placed in a light box made of acrylic and painted black, with two LED lights positioned parallel to each other on top of the box (Fig. 1B). The light box had a cover with a hole that exactly fits the camera lens, where the camera was positioned above the μPAD during image acquisition. For the AFB<sub>1</sub> measurements, the photos were taken using the ISO AUTO Close-up Mode with an exposure time of 1/60 s, an aperture of F/4.5, an ISO of 1250, a focal distance of 36 mm, and color representation in sRGB. The camera was hand-held and was positioned above the µPAD during image acquisition.

The RAW file formats of the images were processed with Digital Photo Professional software (Canon) and stored as a 16-bit color TIF file (see Fig. S1 in the ESI†). The RGB file was then split to obtain the CMYK (cyan–magenta–yellow-key) profiles using ImageJ software. The cyan profile of each image was used and a circular region of interest (ROI) around each of the test and control zones of the  $\mu$ PAD was drawn for quantitative determinations. The histogram of each of these ROIs provided

a mean cyan intensity value that was used for quantitative

determinations and construction of calibration plots.

## Results and discussion

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#### Competitive immunoassay on µPAD

The µPAD detection system presented in this work demonstrates a competitive type of immunoassay. The µPAD has a sample introduction zone located on one end of it, and a control zone and a test zone that branch out from the sample introduction zone which are located on the other end. A capture zone is located between the test zone and the sample introduction zone where the competitive immunoassay takes place. Upon mixing of the components of the sample solutions, the peroxidase-conjugated antibody binds to the free antigen in the solution. As the sample solution is introduced into the sample zone, all the components flow via capillary action into the capture zone. Here, unbound peroxidase-conjugated antibodies are captured and they bind to the immobilized BSA-conjugated antigen. The previously antigen-bound peroxidase-conjugated antibodies however flow past the capture zone and reach the TMB-immobilized test zone as illustrated in Fig. 1C. A blue-colored TMB diimine product is then formed at the test zone for quantitative measurement. Hence, competition occurs at the capture zone. Prior to investigation, suitable proportion of the target antigen, the capture reagent, and the peroxidase-conjugated antibody was determined. The number of moles of the capture reagent is kept higher than the peroxidase-conjugated antibody so that even in the absence of any target antigen, which is at blank concentration, we assume that all the peroxidase-conjugated antibody will be captured at the capture zone, thereby hindering the formation of any blue-colored product at the test zone. Then, with increasing antigen concentration in the sample solution, more of the antigen-bound peroxidase-conjugated antibody component flows past the capture zone to the test zone which results in an increase in the intensity of the blue colored product. At the control zone however, all antigen-bound and free peroxidase-conjugated antibodies flow into the TMBimmobilized control zone, hence producing a constant blue color intensity. Quantification is then performed by measuring the relative intensity,  $I_R$ , which is computed by simply dividing the test intensity,  $I_t$ , by the control intensity,  $I_c$ , as in the following equation:

$$I_{\mathrm{R}} = \frac{I_{\mathrm{t}}}{I_{\mathrm{c}}}.$$

#### Data evaluation

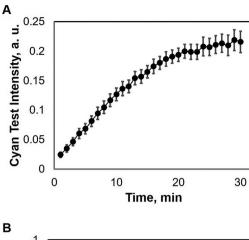
The competitive immunoassay system demonstrated a colorimetric detection of the target on a  $\mu$ PAD using the common enzymatically catalyzed TMB-H<sub>2</sub>O<sub>2</sub> system. Upon enzymatic oxidation of TMB by hydrogen peroxide in the presence of peroxidase, a blue TMB diimine product is produced with water as a by-product. The blue intensity of the TMB diimine product depends on the amount of peroxidase that catalyzes the reaction of TMB and H<sub>2</sub>O<sub>2</sub> as demonstrated in our previous work.<sup>38</sup> The intensities were recorded using the digital camera, and then image-processed using Digital Photo Professional software and analyzed using ImageJ software.

During image processing, the RGB images were first split to CMYK (cyan–magenta–yellow-key) profiles to determine which profile would best provide quantitative results (see Fig. S1†). Based on the different profiles, we found that the cyan profile provided the best quantitation for target measurement. Therefore, the TMB– $H_2O_2$  target detection at different concentrations, producing a varying immunoassay system was quantitatively measured using the cyan profiles of each  $\mu$ PAD.

#### Applications to specific target detection

Biotin immunoassay. The proposed competitive immunoassay system on µPADs was tested using biotin as the model compound. To determine the time necessary for the reaction to produce the maximum relative intensity, we plotted the development profile of the biotin immunoassay. As shown in Fig. 2A, the cyan test intensity increases with time. However, the relative intensity is almost constant as expected (Fig. 2B) since the rate of the enzymatic reaction proceeds simultaneously at the control and test zones. Hence, to allow enough time for color development, we drew the calibration plot for biotin measurement using the color intensities obtained after an incubation time of 20 min. Fig. 3 shows the calibration plot for the biotin measurements after analyzing the cyan intensities produced during the biotin µPAD immunoassay. The limit of detection (LOD), determined experimentally as the lowest biotin concentration that gives a cyan intensity equal to the sum of the cyan intensity of the blank and five times its standard deviation, was 0.10 μg mL<sup>-1</sup> biotin.

**AFB**<sub>1</sub> **immunoassay.** To demonstrate the versatility of the proposed competitive immunoassay system for detecting target compounds on μPADs for practical applications, we detected a highly toxic foodborne substance in the form of aflatoxin  $B_1$  (AFB<sub>1</sub>) along with the comparison of AFB<sub>1</sub> immunoassay results using two different paper substrates – Whatman<sup>TM</sup> filter paper grade 41 (FP41) as in our previous report,<sup>38</sup> and Ahlstrom filter paper grade 319 (A319), a commonly used absorbent pad in chromatographic immunoassays. Although the previous demonstrations of HRP-catalyzed TMB– $H_2O_2$  assay on different cellulose-based μPADs showed a constant tint of blue color in the oxidized TMB product, TMB oxidation using the A319 μPAD during the AFB<sub>1</sub> immunoassay revealed a bluish yellow-green color of the product as shown in



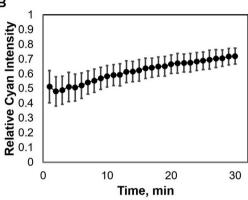


Fig. 2 Color development profile showing (A) the cyan test intensity and (B) the relative cyan intensity of biotin immunoassay on a FP41  $\mu\text{PAD}$  (n = 5). [Conditions: 500 ng mL $^{-1}$  biotin with 1:500 dilution of antibiotin IgG-peroxidase conjugate and 0.001% (v/v)  $\text{H}_2\text{O}_2$ ; BSA-PBS blocking; PBST washing; 50 nmol TMB.]

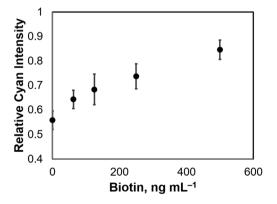


Fig. 3 Biotin immunoassay on a FP41  $\mu$ PAD. The error bars represent 5 replicate measurements. [Conditions: biotin standard solutions with 1:500 dilution of anti-biotin IgG-peroxidase conjugate and 0.001% (v/v)  $H_2O_2$ : BSA-PBS blocking; PBST washing; 50 nmol TMB.]

Fig. 4A. Since the same competitive immunoassay system for the identical target compound was being evaluated using the two μPADs, the measured relative cyan intensities were similar

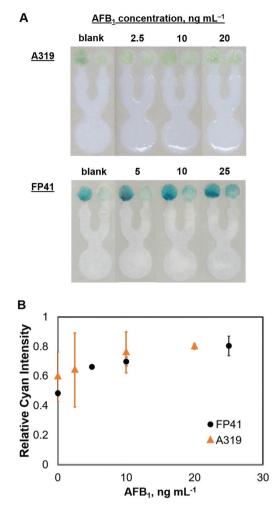


Fig. 4 (A) Images of the A319 and FP41  $\mu$ PADs. (B) Comparative results of the competitive immunoassay on the  $\mu$ PADs. All measurements were performed in duplicate. [Conditions: AFB<sub>1</sub> standard solutions with 1:25 000 dilution of anti-AFB<sub>1</sub> IgG-peroxidase conjugate and 0.001% (v/v) H<sub>2</sub>O<sub>2</sub>; BSA-PBS blocking; PBST washing; 50 nmol TMB.]

(Fig. 4B). However, with the different color tint of the product being measured with the A319  $\mu PAD$ , the standard deviations represented by the error bars for the A319  $\mu PAD$  measurements were significantly higher than the results using the FP41. Moreover, visual analysis was more reliably performed using the FP41  $\mu PAD$  with its more intense blue colored product after the immunoassay. Hence, from the quantitative measurements of AFB1 on the FP41  $\mu PAD$ , we obtained a detection limit of 1.31 ng mL $^{-1}$  at the highest test intensities produced for incubations of 6 to 9 min, which is about 3.8 times lower than the usual limit for AFB1 in food.  $^{41}$ 

## Conclusions

We have demonstrated a competitive immunoassay system on a microfluidic paper-based device. The  $\mu$ PAD consisted of three main elements: (1) test and control zones; (2) a sample introduction zone; and (3) a capture zone. Competition happens in the capture zone, where antigen-free peroxidase-conjugated antibodies are captured before reaching the test zone, allowing only the target-bound peroxidase-conjugated antibodies to travel into the test zone. The proposed competitive immunoassay system was demonstrated on  $\mu PADs$  with biotin as a model compound and with aflatoxin  $B_1$  for analytical testing as a practical application for food monitoring, with limits of 0.10  $\mu g$  mL $^{-1}$  for biotin and 1.31 ng mL $^{-1}$  for AFB $_1$ , respectively. Although the same competitive immunoassay system for both targets was used, the difference in sensitivity was highly attributed to the different affinity of the antibodies used for biotin and AFB $_1$  during the immunoassay. This simple competitive immunoassay system introduces the fundamental principle of competi

tive ELISA on microfluidic paper-based devices, and with this

study, verified its promising applications for a broad range of

analytical testing, not only in food monitoring, but also in

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environmental and clinical applications.

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# References

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- 1 A. W. Martinez, S. T. Phillips, M. J. Butte and G. M. Whitesides, *Angew. Chem., Int. Ed.*, 2007, 46, 1318–1320.
- 2 L. S. A. Busa, S. Mohammadi, M. Maeki, A. Ishida, H. Tani and M. Tokeshi, *Micromachines*, 2016, 7, 86.
- 3 J. A. Adkins and C. S. Henry, *Anal. Chim. Acta*, 2015, **891**, 247–254.
- 4 X. Li, J. Tian, T. Nguyen and W. Shen, *Anal. Chem.*, 2008, **80**, 9131–9134.
- 5 S. Mohammadi, M. Maeki, R. M. Mohamadi, A. Ishida, H. Tani and M. Tokeshi, *Analyst*, 2015, **140**, 6493–6499.
- 6 J. C. Jokerst, J. A. Adkins, B. Bisha, M. M. Mentele, L. D. Goodridge and C. S. Henry, *Anal. Chem.*, 2012, 84, 2900–2907.
- 7 C. Sicard, C. Glen, B. Aubie, D. Wallace, S. Jahanshahi-Anbuhi, K. Pennings, G. T. Daigger, R. Pelton, J. D. Brennan and C. D. M. Filipe, *Water Res.*, 2015, 70, 360–369.
- 8 T. M. G. Cardoso, P. T. Garcia and W. K. T. Coltro, *Anal. Methods*, 2015, 7, 7311–7317.
- 9 S. M. Z. Hossain and J. D. Brennan, *Anal. Chem.*, 2011, 83, 8772–8778.
- 10 N. M. Myers, E. N. Kernisan and M. Lieberman, *Anal. Chem.*, 2015, **87**, 3764–3770.
- 11 J. A. Adkins and C. S. Henry, *Anal. Chim. Acta*, 2015, **891**, 247–254.
- 12 J. Shi, F. Tang, H. Xing, H. Zheng, B. Lianhua and W. Wei, *J. Braz. Chem. Soc.*, 2012, **23**, 1124–1130.
- 13 M. Cuartero, G. A. Crespo and E. Bakker, *Anal. Chem.*, 2015, **87**, 1981–1990.

- 14 S. Ge, W. Liu, L. Ge, M. Yan, J. Yan, J. Huang and J. Yu, Biosens. Bioelectron., 2013, 49, 111-117.
- 15 Z. Nie, F. Deiss, X. Liu, O. Akbulut and G. M. Whitesides, *Lab Chip*, 2010, **10**, 3163–3169.
- 16 P. Zuo, X. Li, D. C. Dominguez and B.-C. Ye, *Lab Chip*, 2013, 13, 3921–3928.
- 17 Y. Su, S. Ma, K. Jiang and X. Han, *Chin. J. Chem.*, 2015, 33, 446–450.
- 18 Y. Zhang, P. Zuo and B.-C. Ye, Biosens. Bioelectron., 2015, 68, 14-19.
- 19 S. Wang, L. Ge, L. Li, M. Yan, S. Ge and J. Yu, *Biosens. Bioelectron.*, 2013, 50, 262–268.
- 20 S.-Q. Jin, S.-M. Guo, P. Zuo and B.-C. Ye, *Biosens. Bioelectron.*, 2014, **63**, 379–383.
- 21 W. Liu, J. Kou, H. Xing and B. Li, *Biosens. Bioelectron.*, 2014, 52, 76–81.
- 22 W. Liu, Y. Guo, J. Luo, J. Kou, H. Zheng, B. Li and Z. Zhang, *Spectrochim. Acta, Part A*, 2015, **141**, 51–57.
- 23 V. Mani, K. Kadimisetty, S. Malla, A. A. Joshi and J. F. Rusling, *Environ. Sci. Technol.*, 2013, 47, 1937–1944.
- 24 Y. Kasahara and Y. Ashihara, *Clin. Chim. Acta*, 1997, **267**, 87–102.
- 25 G. A. Posthuma-Trumpie, J. Korf and A. van Amerongen, *Anal. Bioanal. Chem.*, 2009, **393**, 569–582.
- 26 S. Rattle, O. Hofmann, C. P. Price and L. J. Kricka, *The Immunoassay Handbook*, Elsevier, 2013.
- 27 S. Wang, L. Ge, X. Song, M. Yan, S. Ge, J. Yu and F. Zeng, Analyst, 2012, 137, 3821–3827.
- 28 D. Zang, L. Ge, M. Yan, X. Song and J. Yu, Chem. Commun., 2012, 48, 4683–4685.
- 29 L. Zhang, L. Li, C. Ma, S. Ge, M. Yan and C. Bian, *Sens. Actuators*, *B*, 2015, 221, 799–806.
- 30 S. Ma, Y. Tang, J. Liu and J. Wu, Talanta, 2014, 120, 135-140.
- 31 S. Wang, L. Ge, X. Song, J. Yu, S. Ge, J. Huang and F. Zeng, *Biosens. Bioelectron.*, 2012, 31, 212–218.
- 32 L. Ge, S. Wang, X. Song, S. Ge and J. Yu, *Lab Chip*, 2012, 12, 3150–3158.
- 33 S. Wang, L. Ge, L. Li, M. Yan, S. Ge and J. Yu, *Biosens. Bioelectron.*, 2013, **50**, 262–268.
- 34 J. Shi, F. Tang, H. Xing, H. Zheng, B. Lianhua and W. Wei, *J. Braz. Chem. Soc.*, 2012, **23**, 1124–1130.
- 35 T. M. G. Cardoso, P. T. Garcia and W. K. T. Coltro, *Anal. Methods*, 2015, 7, 7311–7317.
- 36 M. Cuartero, G. A. Crespo and E. Bakker, *Anal. Chem.*, 2015, **87**, 1981–1990.
- 37 N. M. Myers, E. N. Kernisan and M. Lieberman, *Anal. Chem.*, 2015, **87**, 3764–3770.
- 38 L. S. A. Busa, M. Maeki, A. Ishida, H. Tani and M. Tokeshi, *Sens. Actuators, B*, 2016, **236**, 433–441.
- 39 W. Liu, C. L. Cassano, X. Xu and Z. H. Fan, *Anal. Chem.*, 2013, **85**, 10270–10276.
- 40 G. E. Fridley, H. Q. Le, E. Fu and P. Yager, *Lab Chip*, 2012, **12**, 4321–4327.
- 41 W.-B. Shim, K. Kim, J. A. Ofori, Y.-C. Chung and D.-H. Chung, *J. Food Prot.*, 2012, 75, 1991–1999.