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## Correction: Electrochemical reduction of carbon dioxide to formate with a Sn cathode and an $\text{Ir}_x\text{Sn}_y\text{Ru}_z\text{O}_2/\text{Ti}$ anode

Rui Zhang,<sup>a</sup> Weixin Lv,<sup>b</sup> Guanghua Li,<sup>a</sup> Mohammed Adnan Mezaal<sup>a</sup> and Lixu Lei<sup>\*a</sup>Correction for 'Electrochemical reduction of carbon dioxide to formate with a Sn cathode and an  $\text{Ir}_x\text{Sn}_y\text{Ru}_z\text{O}_2/\text{Ti}$  anode' by Rui Zhang *et al.*, *RSC Adv.*, 2015, 5, 68662–68667.

The authors regret that the values presented on the y-axes for Fig. 4 and 10 in the original article are an order of magnitude lower than they ought to be. The correct versions of these figures are shown below.

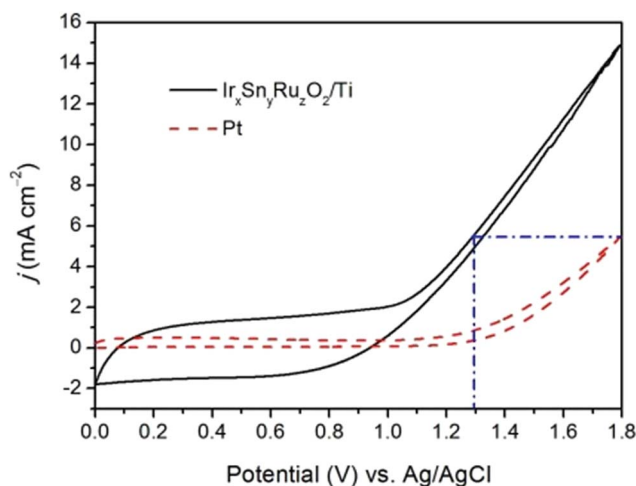


Fig. 4 CV curves of OER on the Pt electrode and the  $\text{Ir}_x\text{Sn}_y\text{Ru}_z\text{O}_2/\text{Ti}$  electrode.

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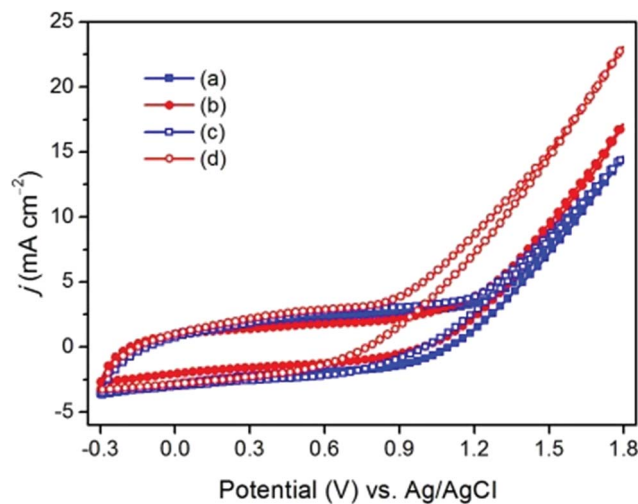


Fig. 10 CV curves on the  $\text{Ir}_x\text{Sn}_y\text{Ru}_z\text{O}_2/\text{Ti}$  electrode with (a) or without (c) Nafion membrane in electrolytes and CV curves on the  $\text{Ir}_x\text{Sn}_y\text{Ru}_z\text{O}_2/\text{Ti}$  electrode with (b) or without (d) Nafion membrane in electrolytes containing  $0.05 \text{ mol L}^{-1} \text{HCOO}^-$ . The measurements were performed in  $0.1 \text{ mol L}^{-1} \text{KHCO}_3$  electrolyte after being bubbled with  $\text{CO}_2$  for 30 min.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

