Cite this: *RSC Adv.*, 2015, 5, 93513Received 5th September 2015
Accepted 29th September 2015

DOI: 10.1039/c5ra18118f

www.rsc.org/advances

The design and investigation of the self-assembly of dimers with two nematic phases†

Z. Ahmed, C. Welch and G. H. Mehl*

A series of non-symmetric dimers were synthesised containing either cyanobiphenyl or difluoroterphenyl moieties on one side and a range of long, short, bent, polar or apolar mesogens on the other side of the molecules. The dielectric anisotropy of the mesogens was varied systematically. The systems were characterised by differential scanning calorimetry (DSC), optical polarizing microscopy (OPM) and detailed X-ray diffraction (XRD) studies, both in the nematic and the N_x phase. The results are compared and structure–property relationships are discussed. A model for the assembly in the N_x phase is developed discussing N_{tb} structures, coaxial helices, swiss roll structures and chiral domain formation.

Introduction

Since the report of an additional nematic phase, initially termed N_x , in cyanobiphenyl based dimers, research into this additional nematic phase has increased rapidly.^{1,2} Other materials with nematic–nematic transitions have been investigated and sets of new materials have been reported.^{2–4} This research has gained additional impetus as it was recognized from the beginning that in the low temperature nematic phase chiral domains are formed. Moreover very fast switching in these materials in the low microsecond regime with a linear electronic effect open up the potential for very fast displays, obviating the need for colour filters and being very energy efficient.^{2,5} Surprisingly, the structure of this lower nematic phase is still being actively discussed.^{2–16} Studies in thin films are consistent with a model where extremely short pitch helices (8–11 nm) are formed, with the pitch being not longer than two to three molecular lengths. These helices are almost two orders of magnitude shorter than those formed by cholesteric liquid crystals, with the mesogens twisting and bending, hence the term N_{tb} . This view is supported too by freeze fracture TEM studies,^{7–9} though AFM results point towards the possibility of surface crystallisation.⁶ Solid state NMR studies of samples in the bulk, used initially to support the simple N_{tb} model¹⁰ cannot, according to more recent results,¹¹ be employed for this without invoking very unusual molecular features, such as very fast migration along helix axes. The optical defect textures of the N_x phase are distinctly different to that of a typical nematic and are more reminiscent of those of smectic phases.¹² The crucial experiment to distinguish between an N_x and a layered phase

are XRD studies, where the absence of intensities associated with layering confirms the nematic character. Theoretical research quickly associated this low temperature nematic phase with potential splay bend and twist bend structures¹³ and this was developed in more detail theoretically as well as in simulation studies.¹⁴ It is noted that a number of additional different varying models are discussed theoretically, ranging from double wall assemblies¹⁵ to hexatic structures.¹⁶ Other complex modes of assembly such as chiral domain structures,¹¹ which are conceptually close to the random domain phase discussed for bent core molecules,¹⁹ have also been considered. The formation of bent conformers has linked this low temperature nematic phase to research on bent-core systems which have an extremely rich phase morphology.

We note that for bent core materials phase structures with C_2 and tetrahedral symmetries and also polar structures have been proposed.^{17–19} In the absence of full consistency of experimental results with one theoretical model the term N_x is retained in this report. Here we report the synthesis and investigation of the liquid crystal properties by OPM, DSC and XRD studies of a series non-symmetric dimeric systems and propose, directed by the experimental data, an assembly mode. The purpose of this study is to understand structure–property correlations of dimeric materials and the structure of the N_x phase with the view to open this class of materials up to technological applications. Bearing in mind the importance of dipoles for any LC device, these were systematically varied in the molecules.

Synthesis

A summary of the synthesis of the materials is shown in Scheme 1. The materials are based on a nonyl spacer linking two mesogenic groups and were designed to exhibit a low temperature N_x mesophase. This was to facilitate ease of investigation and the systematic exploration of the modulation of molecular

Department of Chemistry, University of Hull, HU6 7RX, UK. E-mail: g.h.mehl@hull.ac.uk

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c5ra18118f

Scheme 1 Synthetic scheme leading to the target materials.

structure on liquid crystal phase formation. For mesogenic groups the classical cyanobiphenyl group and the difluoroterphenyl system reported before are used. The molecule based on the mixed cyanobiphenyl/difluoroterphenyl system has been reported before²⁰ and the properties are listed here for clarity. Whilst either the cyanobiphenyl or the difluoroterphenyl group are kept constant on one side of the dimer, the length of the mesogenic group on the other was varied systematically. This was done in such a manner that the mesogenic group consisting of two aromatic rings contain either lateral fluorine or hydrogen atoms and the termini are formed by either pentyl or butyloxy groups, selected for similarity in their overall length. In order to explore the effects of bend in the aromatic mesogenic unit a nitrile functionalized thiophene unit was included in the series. The asymmetric final compounds **1**, **2**, **3** and **4** were prepared by two successive Suzuki coupling reactions. The first reaction was carried out with 4-cyanophenyl boronic acid with a large excess (4×) of the spacer **I1**. The spacer **I1** was synthesized as described previously.⁷ The mono substituted compound was the major product with the conditions mentioned in the procedure, and a 10 times dilution was also tried but no significant difference was observed. The yield was 65% and the unreacted spacer was easily isolated by chromatography and reused. The second reaction was carried out with the appropriate boronic acid in excess this time to achieve complete conversion. The yields were found to be between 42 and 71% with the lowest yielding reaction containing cyanothiophene boronic acid. This is possibly due to the deboration of the boronic acid. Strongly electron withdrawing groups are known to promote deboration in basic conditions.²¹ The asymmetric final compounds **5**, **6**, **7** and **8** were also prepared by two successive Suzuki coupling reactions. The first reaction was carried out with difluorobiphenylpentyl boronic acid with a large excess of the spacer **I1** (4×). The yield was 65% and again the unreacted spacer was isolated by chromatography and reused. The second reaction was carried out with the appropriate boronic acid in excess this time to obtain complete conversion. The yields were found to be between 44 and 77% with the lowest yielding reaction again containing cyanothiophene boronic acid. For the details of the synthesis and the chemical characterisation see the ESI.†

Results and discussion

All of the compounds show nematic phase behaviour. A summary of the results, based on the OPM and DSC investigations is listed in Table 1. For the materials bearing a cyanobiphenyl unit on one side, containing thus in total four aromatic rings, a clear trend is detectable. The nonlinear mesogen **1**, closest in structure to the cyanobiphenyl analogue CB-C9-CB shows the highest LC phase stability, becoming isotropic at 101.1 °C a reduction of 23.1 °C when compared to CB-C9-CB.²² N_x phase formation is destabilized too, resulting in a monotropic phase at 89.5 °C, compared to an enantiotropic N_x phase at 107.3 °C for CB-C9-CB. This is attributed to the reduced packing efficiency of this non-linear molecule when compared to the parent system. For the linear mesogens the trend is as anticipated, and compound **3** without the lateral fluoro groups has the highest transition temperatures of the series, forming a nematic phase at 81.9 °C. Introduction of two lateral fluoro groups in **2** decreases nematic stability to 43.8 °C and the N_x phase becomes monotropic at 38.3 °C, about two degrees below the melting point of 40.6 °C, although it remains stable for several days at room temperature. Introduction of an ether function enhances phase stability, with the material **4** becoming isotropic at 62.7 °C, though the N_x phase is only monotropic at 54.9 °C, 1.4 °C below the melting point at 56.3 °C. The overall trend seen is that a reduction of molecular symmetry lowers the transition temperatures and the melting points with a concomitant reduction in the stability of the low temperature nematic phase. The transition enthalpies range from 0.46 kJ mol⁻¹ to 1.31 kJ mol⁻¹ and are in line with typical nematic materials, but noticeable is the much higher value of 2.10 kJ mol⁻¹ for CB-C9-CB. For the materials bearing a terphenyl unit at one end of the molecule the phase behaviour is considerably more complex. All these materials are non-symmetric with regards to the length of the aromatic groups, with the systems each bearing five aromatic rings in total. For molecule **5** with a thiophene 5-cyano group at the end a clearing temperature of 114.7 °C from the nematic is observed with only a monotropic N_x phase detectable at 83.9 °C. For the linear molecules the differences are quite considerable and these values are substantially lower than those for the already reported molecules DTC5C9 and DTC9-CB. Isotropisation temperatures of 144.1 °C and 165.3 °C and N_x phase formation at 109.8 °C and 120.3 °C for DTC5C9 and DTC9-CB have been reported.^{7,20} Compound **6** with two lateral fluoro groups on the short mesogen melts at 47.4 °C, and shows a highly ordered smectic phase up to 62.2 °C, where the material transforms to a SmA phase which becomes nematic at 71.6 °C before the material turns isotropic at 81.3 °C. Removal of the fluoro groups on one side in compound **7**, enhanced the phase stability and lead to an N_x phase ranging from 78.2 °C to 86.4 °C with the material turning isotropic at 105.5 °C. Introducing a terminal butyloxy group changes the phase behaviour considerably, as seen for compound **8**. A monotropic highly ordered LC phase is formed on cooling at 61.7 °C, with the melting point at 69.5 °C, when an N_x phase is formed which is stable up to 82.8 °C. This is

Table 1 Structures of the materials and transition temperatures

No.	R	R'	Transition temperature [°C] ΔH [kJ mol ⁻¹]
CB-C9-CB ^a			Cr-85.2-N _x -107.3-N-124.2-iso 0.86 2.10
1			Cr-(N _x 89.5) 90.4-N-101.1-iso 0.83 0.10
2			Cr-(N _x 38.3) 40.6-N-43.8-iso 1.31 0.078
3			Cr-81.9-N-85.8-iso 0.37
4			Cr-(N _x 54.9) 56.3-N-62.7-iso 0.46 0.15
5			Cr-(N _x 83.9) 89.7-N-114.7-iso 0.078 0.52
6			Cr-47.4-SmX-62.2-SmA-71.6-N-81.3-iso 3.98 0.98
7			Cr-78.2-N _x -86.4-N-105.5-iso 0.37 0.99
8			Cr-(SmX 61.7) 69.5-N _x -82.8-N-103.4-iso 1.01 0.10 0.38

Transition temperatures and enthalpy values of the compounds were taken from DSC scans (10 °C min⁻¹). Monotropic phases are shown in parentheses; ^a has been reported previously,^{1,22} and is presented here for comparison.

followed by a nematic phase before the material turns isotropic at 103.4 °C. It is noted that the molar transition enthalpies for the lower temperature nematic phase for this set of molecules, ranging from 0.078 kJ mol⁻¹ for 5 to 0.37 kJ mol⁻¹ for 7 are higher than for the previous set of materials, however being still in the range of typical nematics.²³ It is interesting that the intermediate **I3** (see ESI†) shows nematic phase behaviour up to 85 °C. A low temperature phase which was initially identified as N_x based on optical studies, is verified to be a SmA phase based on XRD investigations, for details see the ESI.† In the process of changing the molecular structures, the dielectric anisotropies have been varied, lateral and terminal dipoles were introduced. The effect on the transitions are as expected. It was found that lateral difluoro groups in the aromatic core reduce the transition temperatures, while the introduction of terminal cyano groups and the introduction of ether linkages to the terminal chains enhances transition temperatures. The introduction of a bend in the mesogen, such as in compounds **1** and **5**, does not

stabilize N_x phase formation, however as monotropic phase behaviour was recorded close to the melting points, it can be assumed that small variations of the architecture could lead to enantiotropic N_x phase formation. There is no indication that the position of dipoles or their symmetry plays a crucial role for N_x phase formation, but they are important for dialling in the temperature range of N_x phase formation.

Polarising optical microscopy

The optical defect textures of the N_x phase are distinctly different to those of a typical nematic material. Though dark homeotropic areas can be observed, more typical are the observations of “tile” or “rope” like features often formed when cooling from the nematic. These defect textures can easily be mistaken for those of a smectic phase; indeed smectic focal conic like features are also found for such materials using three photon correlation spectroscopy.¹² Fig. 1 shows the change of

Fig. 1 Micrographs of compound 7 at 101 °C (a) and 85 °C (b) under crossed polarizers and on untreated slides.

texture for compound 7 going from a nematic (Fig. 1a) to the rope like and tile like features in the N_x phase at 85 °C (Fig. 1b).

Fig. 2a shows the plated texture close to the nematic- N_x transition whereas Fig. 2b shows the texture 2 °C lower at 33 °C. The texture observed can be described as polygonal with almost fish scale like resemblance. It is important to note that this texture change does not register as a thermal event in calorimetric studies. As both types of textures are kinetically stable it is likely the plated texture retains some characteristics of the preceding conventional nematic phase. If that were to be the case the natural texture of the N_x phase would be the polygonal fish scale texture on untreated slides. This texture is shown below in Fig. 3a. A characteristic observation for the nematic phase is spontaneous homeotropic alignment under shearing. This phenomenon has been attributed to the small elastic constants of the nematic phase.²⁴ The same is observed for the N_x phase and is shown in the shearing experiments in Fig. 3b. The dependence of the periodicity of the stripes on the thickness of the sample can be observed for the micrographs shown in Fig. 4 and is in line with earlier observations.

X-ray diffraction investigations

The XRD investigation of samples in capillaries in the presence of a magnet field of ~ 0.5 T perpendicular to the XRD beam were performed on heating and cooling. A number of general trends could be observed. For the wide angle reflections the difference between the nematic and N_x phase is very small, the values for lateral distances of the molecules vary between 4.5 and 4.8 Å.

In the N_x phase the wide angle maxima become sharper, suggesting a more ordered lateral register, resulting in a more defined positional ordering. Common for all materials is an excellent macroscopic ordering parallel to the external magnetic field in the nematic (N) phase, evidenced by equatorial wide angle maxima. At the transition to the N_x phase this ordering is

Fig. 2 Example of textures observed on untreated slides of compound 2 in the N_x phase at 33 °C (a) and 31 °C (b).

Fig. 3 Polygonal fish scale texture of N_x (a) and upon shearing (b) on untreated slides of compound 2 at 33 °C under crossed polarizers. Magnification $\times 100$, scale bar 100 μm .

reduced. The maxima of the wide angle arcs are less well centred on the equator and more widely distributed, indicative of either loss of macroscopic order or some form of loss of long range orientational ordering. This is conceivable in a process where two mesogens linked at a fixed angle compete for orientation in an external field. Alternatively, this could be explained by an increased mosaicity of the sample, or in other words domains are formed and the order within the domains can increase. This picture would tie in with the sharpening of the wide angle intensities; the observed macroscopic ordering by the external magnetic field is however lower than in the nematic phase. For the cyanobiphenyl containing systems the small angle intensities are very small, with intensity values considerably lower than those of 5CB, examined under similar conditions for which data was collected to be used as a reference (see Table 2). We note that the data collected here for 5CB is in line with results reported earlier.²⁵ Fig. 5 shows diffractograms of 4 in the nematic phase at 58 °C and at 52 °C in the N_x phase, with a vertical magnetic field. The loss of macroscopic ordering going to the N_x phase is easily visually detectable, with the wide angle arcs less centred on the equator and more spread out. The very low intensity for the weak small angle reflections, hardly detectable for both the N and N_x phase is noticeable and more clearly depicted in a pseudo 1D plot, shown in Fig. 6 where the intensity of collected data against the diffraction angle 2θ is plotted.

Fig. 4 Micrographs of CBC9CB in planar aligned cells in the N_x phase of: (a) 88 °C, 2 μm (b) 103 °C, 6 μm (c) 103 °C, 10 μm (d) 104 °C, 25 μm thickness under crossed polarisers. Indentations represent 10 μm ; magnification $\times 500$; crossed polarizers.

Table 2 Summary of the X-ray data; Comp. = compound

Comp.	Phase	$T/^{\circ}\text{C}$	d -spacing SAXS/ $\text{\AA}$				Longitudinal correlation length (ξ)/ $\text{\AA}$	d -spacing WAXS/ $\text{\AA}$	$I_{\text{SAX}}/I_{\text{WAX}}$
1	N	95.0					—	4.7	—
	N _x	85.0					—	4.6	—
2	N	40.0	17.5				12	4.6	0.69
	N _x	35.0	15.9				12	4.6	0.65
3	N	84.0	18.8				12	4.8	0.72
4	N	56.0	17.6				13	4.7	0.86
	N _x	42.0	17.4				12	4.6	0.80
5	N	90.0	21.3				14	4.8	1.19
	N _x	80.0	20.8				13	4.8	1.02
6	N	70.0	18.4				—	4.6	0.90
	SmA	50.0	47.4	22.9			1176	4.6	19.70
	SmX	30.0	44.2	28.4	21.4	16.8	1166	4.5	18.13
7	N	99.0	19.2				10	4.8	0.87
	N _x	83.0	19.2				11	4.7	0.82
8	N	90.0	17.5				14	4.7	1.04
	N _x	70.0	17.6				16	4.6	0.95
	SmX	60.0	18.7				—	4.5	—
5CB	N	22.0	23.6	13.0			20	4.5	0.92

For the materials containing a difluoroterphenyl group in the mesogenic unit the overall behaviour is very similar, but the stronger intensity of the small angle reflections is noticeable. The results for the investigated materials are summarized in Table 2. The maxima for the wide angle reflections shift for most of the investigated systems by 0.1 $\text{\AA}$ when going from the N to the N_x phase, typically from 4.7 to 4.6 $\text{\AA}$, although there are exceptions. An example is compound 5, where there is no change and an overall lateral distance of 4.8 $\text{\AA}$ is measured. Though not unusual, these distances are slightly larger than for most nematics, where values of 4.5–4.6 $\text{\AA}$ are common. These results might simply be due to packing inefficiencies of the odd membered hydrocarbon structures, connected to the aromatic rings *via* a fairly rigid methylene group. The apparent pseudo d -spacings derived from the weak small angle intensities are all quite similar in value, ranging between 15.9 $\text{\AA}$ and 20.8 $\text{\AA}$. This is larger than a single aromatic mesogenic group, but slightly smaller than that of half of the full molecules. This is indicative of a local interdigitated structure of the SmA_d type. The exception is molecule 1, where the small angle intensities were so weak (see ESI† for diffractogram) that it was not possible to define a pseudo d -spacing for the nematic phase. Looking at the temperature dependence of the small angle data it is noticeable that there is no clear trend observable for the investigated set of materials. For some of the materials on cooling to the N_x phase from the nematic, a contraction in the value of the pseudo d -spacings is noted such as for material 2, where a contraction from 17.5 $\text{\AA}$ at 40 $^{\circ}\text{C}$ to 15.9 $\text{\AA}$ at 35 $^{\circ}\text{C}$ was recorded. In compound 4 the contraction was only from 17.6 $\text{\AA}$ to 17.4 $\text{\AA}$ going from 56 $^{\circ}\text{C}$ to 42 $^{\circ}\text{C}$ and for 5 the contraction was from 21.3 $\text{\AA}$ to 20.8 $\text{\AA}$ going from 90 $^{\circ}\text{C}$ to 80 $^{\circ}\text{C}$. For material 7 no detectable change in the apparent layer spacing of 19.2 $\text{\AA}$ could be detected. To quantify the low intensities in the small angle region further a simple ratio between the individual small angle and wide angle reflections is calculated, termed $I_{\text{SAX}}/I_{\text{WAX}}$, using

the integrated data assembled for the pseudo 1D plots. This indicates for the cyanobiphenyl based systems a value below 1, and for most systems a value just below or above unity is found. Notable is that the cyanobiphenyl based systems tend to have values lower than 1. On the whole all these values are similar to those of which can be detected for 5CB, the archetypal nematic material, for which data was collected in the nematic phase at 22.0 $^{\circ}\text{C}$ under the same conditions as for the dimeric materials, confirming the essentially low ordered character of the N_x phase. Using the small angle intensities, it is possible to estimate the longitudinal correlation lengths. The calculated correlation lengths are a quantity used to indicate the length scale over which particle–particle correlations are lost and has been adopted to describe short range order. Correlation lengths were calculated using $\xi = c/\Delta q$ where c is the function used to describe the scattered intensity and Δq is the full width at half maximum (FWHM).²⁶ For compound 1 and the nematic phase of compound 6, the scattering intensity is described by a non-regular line shape and an accurate measurement of FWHM is not possible. The values for the correlation lengths are on the order of one mesogenic length, between 10 and 15 $\text{\AA}$ (see Table 2), and broadly similar to those found for 5CB^{25,27,28} indicating that the materials are only oriented really over a very short range. The small changes in values for the correlation lengths, going from the nematic to the N_x phase by only 1–3 $\text{\AA}$, if at all, are qualitatively indicative that no substantive increase in long range order takes place and the ordering is very short range indeed, even if one considers, as has been done previously²⁶ that measured correlation lengths might underestimate true ordering by a factor of three or four. We note here that when a material shows a SmA phase such as observed for compound 6 at 50 $^{\circ}\text{C}$, typical correlation lengths of 1176 $\text{\AA}$, correlating to ~ 30 molecular lengths are detectable. A higher ordered low temperature phase of 6 termed SmX needs to be investigated further. Considering that the molecules with polar end groups

Fig. 5 XRD diffraction data for compound **4** at 58 °C (left) and 52 °C (right), the magnetic field is vertical. The loss of macroscopic ordering going to the N_x phase is easily visually detectable, the wide angle arcs are less centred on the equator, but more spread out.

such as materials **1–4** exist in associated and non-associated species, similar to **5CB**^{25–29} and some asymmetric dimers, changes in the mode of assembly in the LC phase can be investigated by XRD. Noticeable too is that none of the materials show cybotactic clustering of the SmC type in the nematic phase or in the N_x phase. This is a feature that would be anticipated in tilted molecules and is a feature common in bent core systems.^{30–32} Moreover, there is no indication of a diffraction pattern indicative of short pitch helical structures. An “X-shaped pattern” would be anticipated with the first meridional reflection indicative of the subunit axial translation of the molecules and off-meridional intensities relating to the pitch of the helix. Patterns of this type have been observed for many natural and synthetic helical materials, with DNA being the most notable.³³ We note too, that when tight pitch chiral assemblies with some layered ordering are formed, as in the B4 phase or in LC nanofilaments, they are clearly detectable by XRD experiments as well as in TGB or smectic blue phases.^{34–39} Given that the materials investigated here are low ordered one would have anticipated at least intensities in the form of a “straight line” in the 2D data of the small angle intensities of the oriented samples, providing a simple helix was present. This was not found, but crescent like small angle intensities were detected with the maxima of the small angle intensities being at an angle of 90° to the maxima of the wide angle data. Occasionally on the transition from the nematic to the N_x phase a small misalignment to the external magnetic field can be detected by ~5–6°, which suggests that even in capillaries with a diameter of 0.8 mm surface effects can play a role. The results require consideration of the possible assembly structure of the N_x phase. Based on XRD data we arrive at a picture indicating that on cooling from the nematic phase, the lateral packing tends to be closer, however there is no or very little increase in long range orientational ordering and macroscopic ordering decreases. This is consistent with results from calorimetric studies, which show that the transition enthalpies are very low, and the specific heat decreases when adding a low molar mass compound, such as **5CB**, however N_x phase formation still occurs.²² Data from OPM experiments and shearing experiments indicate an increased viscosity on shearing going to the N_x phase and textures which point to a highly ordered smectic like phase. The available data constrains model formation for

the structure of the N_x phase considerably. There is no positive experimental evidence based on the XRD data gained in this study to positively support a nematic phase formed from helices with constant pitch and radii, and to the best knowledge of the authors, no XRD experimental evidence has so far been reported in the literature on other dimers either. Short range ordering and closer lateral packing than that in the nematic phase is supported by experimental evidence. NMR data for samples in the bulk in other studies shows clearly the formation of chiral structures of opposite handedness and formation of domains.^{9–11} On the other hand, thin film studies are clearly consistent with heliconical ordering, though recent studies suggest that splay components could play a role too.^{40,41} Closer inspection of the TEM reports show that the periodicities observed are not fully fixed in the 8–10 nm range, but shorter and longer pitches are found too, as well as domain structures.⁹ This is quite different to cholesterics where, at a given temperature, only one defined pitch is observable. This is due to the different origin of chirality. For the dimers in the N_x phase the dynamic interconversion of twisted conformations, chiral for non-symmetric molecules, is slowed down. This occurs to such an extent, that when compared to the N phase, they persist sufficiently long, so that neighbouring molecules in the close vicinity arrange accordingly. A structural model needs to incorporate these findings and in the following an effort is made to reconcile the apparently contradictory findings using different experimental techniques. Fig. 7a shows a schematic of the N_{tb} model which is supported by thin film studies, with a uniform helical pitch of a few dimers in length. Fig. 7b shows a variation of this structure, in the centre the tight pitch N_{tb} structure is shown, which is surrounded, idealized in a coaxial manner, with assemblies of varying pitch formed by molecules packing onto the tightly folded N_{tb} like structure. The formation of such an assembly mode could explain the varying helix lengths observed by TEM experiments, as well the very small correlation lengths for the materials in XRD experiments. Moreover, for biomaterials such as cellulose this varying pitch mode with spiralling helicity explains the lack of the typical helical diffraction pattern for fibrous structures.^{42,43} In this model the uniform ordering in the direction of the z-axis (helix

Fig. 6 Plot of 2θ against intensity for compound **4** in both nematic phases, recorded at 58 °C and 52 °C.

Fig. 7 Possible arrangement of molecules. (a) Schematic side view of the N_{tb} model, below top view; (b) side view coaxial helices, below side view; (c) spiralling helicoid; below top view; (d) side view: spiralling helicoid with finite extension in of the z -axis, below top view.

axis) has been reduced, however additional ordering perpendicular to the main axis due to the surrounding enveloping spiralling helices is required. Incidentally this concept bears on first sight similarity with the double twist model for high chirality cholesterics.⁴⁴ Fig. 7c represents a schematic showing how this level of ordering can be relaxed. The coaxial structure is reduced, the strict layering is broken and the helical pitch changes, shown in an idealized form spiralling out from the centre to the periphery in a type of swiss roll structure, a feature observed in some nanorods.⁴⁵ Nematic splay components which have been discussed in terms of the N_{tb} phase could possibly be accommodated by such a model.^{40,41} Fig. 7d shows a further level of disordering as the length of the helix is now finite and the material is low ordered resulting in a domain structure, or to use polymer nomenclature in analogy, the persistence length of ordering in the z -axis is not large. Surprisingly, this depiction is close to the picture often used intuitively for chiral structures, such as gastropods or chiral swirls.^{46–48} The difference between Fig. 7c and d is that the main axis is of finite length and pitch and radii taper, but they could interconvert, simply by the alignment of domains. Considering the very low correlation lengths found by XRD studies in the dimeric materials, structures such Fig. 7a and b seem to be less likely than Fig. 7c or d, which additionally could easily explain the formation of domains observed by NMR studies and by the XRD data. External fields could favour the structure shown in Fig. 7b or a, the currently discussed N_{tb} structure. As the driving force for this assembly is just the self-sorting of enantiomorphic conformers^{48,49} in the confinement of thin electro-optical cells with alignment layers and strong fields the formation of helices could be rationalized. This might, under the right conditions be close to the structure formed in single crystals⁵⁰ shown for

CB-C7-CB to be a helical structure in the 8–10 nm range. This easy deformation of the domains by external fields would explain the staggeringly diverse range of defect textures formed by materials in the N_x phase, where cell thickness, alignment layers and external fields play a crucial role. The easy change from a ground state could possibly explain too, the diverse AFM and TEM data. Such chameleon like assembly behaviour has not been discussed much for nematics, with the exception of a model for nematic biaxiality,⁵¹ however it is more common for dimeric and oligomeric smectic systems.^{52,53} The concept discussed here is akin to those developed to account for the formation of cybotactic clustering in essentially nematic materials.⁵⁴ There, on decreasing temperature smectic fluctuations occur and this results in the dynamic formation of species of essentially nanometric size. Cybotactic ordering has been observed for a wide range of materials ranging from bent core systems to nematic dimers, oligomers, polymers and dendrimers and often persists over the full temperature range of the nematic phase. In these systems analysis is eased, as dynamic local layer formation can be interrogated by diffraction experiments, due to modulation of the electron density in the domains. In the case of the N_x phase the fluctuations are not smectic like, they are chiral, due to the self-sorting of chiral conformers into dynamically assembling and disassembling conformations and domains and the ordering is still nematic; hence the analysis is more difficult. Finally, whether tightly packed chiral domains assemble in hierarchical superstructures, which may be responsible for the smectic like OPM textures as has been discussed for short pitch smectics such as the SmQ phase, blue phases or TGB phases needs to be explored and concomitant technological exploitation following the understanding of the phase structure is to be expected.^{37–39,44} Hence there is clearly a need to correlate the multitude of theoretical efforts with experimental data and to explore whether there are a number of nematic sub-phases and how they connect to low temperature highly ordered LC structures, such as the smectic (Sm) and bent-core (B) phases.

Conclusions

For the investigated systems the variation of the molecular symmetry by using mesogens of different length and shape and of different dielectric anisotropy, by introducing strong terminal or lateral dipoles or both was investigated. As a general trend it was found that the reduction of molecular symmetry reduces melting points and transition temperatures up to almost room temperature range. Shorter mesogens tend to have lower transition temperatures. The position of dipoles whether positioned terminally or laterally is not critical for the formation of the N_x phase as such, but they are of course critical for technological applications. Detailed XRD investigations show that all of the materials are low ordered in the nematic phase, similar to 5CB and that the orientational ordering, as defined by correlation lengths increases, if at all only very slightly in the N_x phase, though macroscopic loss of order and closer lateral packing was observed for most systems. Based on the available data, parameters for model formation are discussed, bearing in

mind the need to reconcile the results from complimentary experimental methods. Here the merits of the N_{tb} structure, coaxial helices, swiss roll structures and chiral domain formation and their interconversion are discussed.

Acknowledgements

The authors thank the colleagues in the EU project BIND (contract: 216025) for discussions and acknowledge the EPSRC projects EP/J004480/1 and EP/M015726/1 for funding and the EPSRC NMSF in Swansea is acknowledged for mass spectrometry data.

Notes and references

- 1 V. P. Panov, M. Nagaraj, J. K. Vij, Yu. P. Panarin, A. Kohlmeier, M. G. Tamba, R. A. Lewis and G. H. Mehl, *Phys. Rev. Lett.*, 2010, **105**, 167801.
- 2 V. P. Panov, R. Balachandran, M. Nagaraj, J. K. Vij, M. G. Tamba, A. Kohlmeier and G. H. Mehl, *Appl. Phys. Lett.*, 2011, **99**, 261903.
- 3 P. A. Henderson and C. T. Imrie, *Liq. Cryst.*, 2011, **38**, 1407.
- 4 R. J. Mandle, E. J. Davis, C. C. A. Voll, C. T. Archbold, S. J. Cowling and J. W. Goodby, *J. Mater. Chem. C*, 2014, **2**, 556.
- 5 R. Balachandran, V. P. Panov, M. G. Tamba, G. H. Mehl, J. K. Song, Y. Panarin and J. K. Vij, *J. Mater. Chem. C*, 2014, **2**, 8179–8186.
- 6 E. Gorecka, M. Salamonczyk, A. Zep, D. Pocięcha, C. Welch, Z. Ahmed and G. H. Mehl, *Liq. Cryst.*, 2015, **42**, 1.
- 7 V. Borshch, Y. K. Kim, J. Xiang, M. Gao, A. Jákli, V. P. Panov, J. K. Vij, C. T. Imrie, M. G. Tamba, G. H. Mehl and O. D. Lavrentovich, *Nat. Commun.*, 2013, **4**, 2635.
- 8 D. Chen, J. H. Porada, J. B. Hooper, A. Klitnick, Y. Shen, M. R. Tuchband, E. Korblova, D. Bedrov, D. M. Walba, M. A. Glaser, J. E. MacLennan and N. A. Clark, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 15931.
- 9 M. Gao, Y. K. Kim, C. Zhang, V. Borshch, S. Zhou, A. Kohlmeier, G. H. Mehl, W. Weissflog, D. Studer, B. Zuber, H. Gnägi and F. Lin, *Microsc. Res. Tech.*, 2014, **77**, 754.
- 10 L. Beguin, J. W. Emsley, M. Lelli, A. Lesage, G. R. Luckhurst, B. A. Timimi and H. Zimmermann, *J. Phys. Chem. B*, 2012, **116**, 7940.
- 11 A. Hoffmann, A. G. Vanakaras, A. Kohlmeier, G. H. Mehl and D. J. Photinos, *Soft Matter*, 2015, **11**, 850.
- 12 V. P. Panov, M. C. M. Varney, I. I. Smalyukh, J. K. Vij, M. G. Tamba and G. H. Mehl, *Mol. Cryst. Liq. Cryst.*, 2015, **611**, 180–185.
- 13 R. Meyer, in *Molecular Fluids*, ed. R. Balian and G. Weill, Gordon and Breach Science Publishers, 1976.
- 14 R. Memmer, *Liq. Cryst.*, 2002, **29**, 483.
- 15 E. G. Virga, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2014, **89**, 052502.
- 16 R. D. Kamien, *J. Phys. II*, 1996, **6**, 461.
- 17 D. Chen, M. Nakata, R. Shao, M. R. Tuchband, M. Shuai, U. Baumeister, W. Weissflog, D. M. Walba, M. A. Glaser, J. E. MacLennan and N. A. Clark, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 2014, **89**, 022506.
- 18 H. R. Brand and H. Pleiner, *Eur. Phys. J. E*, 2010, **31**, 37.
- 19 M. Koike, C. C. Yen, L. Yuquing, H. Tsuchiya, M. Tokita, S. Kawuchi, H. Takezoe and J. Watanabe, *Macromolecules*, 2007, **40**, 2524.
- 20 M. G. Tamba, A. Kohlmeier and G. H. Mehl, in *Proc. European Liquid Crystal Conference*, ECLC, Greece, 2013.
- 21 D. M. Perrin, Z. Liu and J. Lozada, *J. Org. Chem.*, 2014, **79**, 5365.
- 22 C. S. P. Tripathi, P. Losada-Perez, J. Leys, A. Kohlmeier, M. G. Tamba, G. H. Mehl and C. Glorieux, *Phys. Rev. E*, 2011, **84**, 041707.
- 23 O. Patrick and P. Pieranski, *Nematic and Cholesteric Liquid Crystals*, Taylor & Francis, Boca Raton, 2005.
- 24 I. Dierking, *Textures of Liquid Crystals*, Wiley-VCH, Weinheim, 2003.
- 25 V. Castelletto, A. M. Squires, I. W. Hamley, J. Stasiak and G. D. Moggridge, *Liq. Cryst.*, 2009, **36**, 435.
- 26 O. Francescangeli, M. Laus and G. Galli, *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.*, 1997, **55**, 481.
- 27 S. Barnard, *Investigation of Polar Nematics using X-Ray Diffraction and Dielectric Techniques and X-Ray Diffraction Studies of Disc-Shaped Mesogens*, PhD Thesis, University of Bristol, UK, 1991.
- 28 A. J. Leadbetter, R. M. Richardson and C. N. Colling, *J. Phys., Colloq.*, 1975, **36**, 1.
- 29 F. Vita, K. Spennacci, G. Panzarasa, I. F. Placentino, S. Marino, N. Scaramuzza, G. Portale, E. Di Cola, C. Ferrero, S. I. Torgova, G. Galli, M. Laus and O. Francescangeli, *ACS Macro Lett.*, 2014, **3**, 91–95.
- 30 O. Francescangeli and E. T. Samulski, *Soft Matter*, 2010, **6**, 2413.
- 31 V. Faye, H. Nguyen and P. Barois, *J. Phys. II*, 1997, **7**, 1245–1260.
- 32 C. Keith, A. Lehmann, U. Baumeister, M. Prehm and C. Tschierske, *Soft Matter*, 2010, **6**, 1704.
- 33 D. Sherwood, *Crystals, X-Rays and Proteins*, Wiley, New York, 1976.
- 34 E. Gorecka, N. Vaupotic and D. Pocięcha, *Handbook of Liquid Crystals*, ed. J. W. Goodby, P. J. Collings, T. Kato, C. Tschierske, H. T. Gleeson and P. Raynes, Wiley-VCH, 2nd edn, 2014, vol. 4, pp. 681–713.
- 35 E. Bialecka-Flojanczyk, I. Sledzinka, E. Gorecka and J. Przedmojski, *Liq. Cryst.*, 2008, **35**, 401.
- 36 L. E. Hough, H. T. Jung, D. Krüerke, M. S. Heberling, M. Nakata, C. D. Jones, D. Chen, D. R. Link, J. Zasadzinski, G. Heppke, J. P. Rabe, W. Stocker, E. Korblova, D. M. Walba, M. A. Glaser and N. A. Clark, *Science*, 2009, **325**, 456.
- 37 M. Levelut, E. Hailoin, D. Bennemann, G. Heppke and D. Lotsch, *J. Phys. II*, 1997, **7**, 981.
- 38 P. J. Collings, *Handbook of Liquid Crystals*, ed. J. W. Goodby, P. J. Collings, T. Kato, C. Tschierske, H. T. Gleeson and P. Raynes, VCH, 2nd edn, 2014, vol. 4, pp. 445–464.
- 39 B. Pansu, M. Li and H. Nguyen, *J. Phys. II*, 1997, **7**, 751.

- 40 C.-J. Yun, M. R. Vengatesan, J. K. Vij and J.-K. Song, *Appl. Phys. Lett.*, 2015, **106**, 173102.
- 41 C. Meyer, G. R. Luckhurst and I. Dozov, *J. Mater. Chem. C*, 2015, **3**, 318.
- 42 A. Reiterer, H. F. Jakob, S. E. Stanzl-Tschegg and P. Frantzl, *Wood Sci. Technol.*, 1998, **32**, 335.
- 43 A. Boettiger, B. Ermentrout and G. Oster, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 6837.
- 44 P. G. de Gennes and J. Prost, *The Physics of Liquid Crystals*, Clarendon Press, 2nd edn, 1993, p. 323.
- 45 S. Iijima, *Nature*, 1991, **354**, 56.
- 46 I. Dierking, *Symmetry*, 2014, **6**, 444.
- 47 D. W. Thompson, *On Growth and Form*, University Press, Cambridge, 1992.
- 48 M. M. Safont-Sempere, G. Fernandez and F. Wurthner, *Chem. Rev.*, 2011, **111**, 5784.
- 49 C. Dressel, T. Reppe, M. Prehm, M. Brautzsch and C. Tschierske, *Nat. Chem.*, 2014, **6**, 971.
- 50 K. Hori, M. Iimuro, A. Nakao and H. Toriumi, *J. Mol. Struct.*, 2004, **699**, 23.
- 51 A. G. Vanakaras and D. J. Photinos, *J. Chem. Phys.*, 2008, **128**, 154512.
- 52 J. P. F. Lagerwall, F. Giesselmann, M. D. Wand and D. M. Walba, *Chem. Mater.*, 2004, **16**, 3606.
- 53 H. Sasaki, Y. Takanishi, J. Yamamoto and A. Yoshizawa, *J. Phys. Chem. B*, 2015, **119**, 4531.
- 54 C. Keith, G. Dantlgraber, R. A. Reddy, U. Baumeister, M. Prehm, H. Hahn, H. Lang and C. Tschierske, *J. Mater. Chem.*, 2007, **17**, 3796.