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## Enzymatically degradable EMI shielding materials derived from PCL based nanocomposites

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In this study, two different types of multiwall carbon nanotubes (MWNTs) namely pristine (p-MWNTs) and amine functionalized (a-MWNTs) were melt-mixed with polycaprolactone (PCL) to develop biodegradable electromagnetic interference (EMI) shielding materials. The bulk electrical conductivity of the nanocomposites was assessed using broadband dielectric spectroscopy and the structural properties were evaluated using dynamic mechanical thermal analysis (DMTA). Both the electrical conductivity and the structural properties improved after the addition of MWNTs and were observed to be proportional to the increasing fractions in the nanocomposites. The shielding effectiveness of the nanocomposites was studied using a vector network analyzer (VNA) in a broad range of frequencies, X-band (8 to 12 GHz) and K<sub>u</sub>-band (12 to 18 GHz) on toroidal samples. The shielding effectiveness significantly improved on addition of MWNTs, more in the case of p-MWNTs than in a-MWNTs. For instance, at a given fraction of MWNTs (3 wt%), PCL with p-MWNTs and a-MWNTs showed a shielding effectiveness of −32 dB and −29 dB, respectively. Moreover, it was observed that reflection was the primary mechanism of shielding at lower fractions of MWNTs, while absorption dominated at higher fractions in the composites. As one of the rationales of this work was to develop biodegradable EMI shielding materials to address the challenges concerning electronic waste, the effect of different MWNTs on the biodegradability of PCL composites was assessed through enzymatic degradation. The enzymatic degradation of the samples cut from the hot pressed films by bacterial lipase was investigated. It was noted that a-MWNTs exhibited almost similar degradation rate as the control PCL sample; however, p-MWNTs showed a slower degradation rate. This study demonstrates the potential use of PCL-MWNT composites as flexible, light weight and eco-friendly EMI shielding materials.

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## Introduction

Owing to its versatility, polymeric nanocomposites have gained tremendous commercial importance and are being explored in a wide range of applications ranging from infrastructure, military and automotive to electronic industry. Due to the advancement in electronic industry, electronic-waste (e-waste) is increasing enormously. In order to manage the e-waste, recycling and degradation of electronic components such as control panels, switches, printed circuit boards, instrument cabinets, have become vital. In order to minimize the e-waste, the development of biodegradable materials for various applications is in great demand. Typical biodegradable polymers comprises of ester linkage, which can easily be attacked by enzymes,<sup>1</sup> leading to their degradation. Among the various biodegradable polymers, poly( $\epsilon$ -caprolactone), PCL has emerged as an important candidate<sup>1–7</sup> for polymer based nanocomposites for various applications due to its ease of

processing. However, poor mechanical properties and its insulating nature limit its use in structural and electronic applications. Various strategies, such as blending with another polymer<sup>8–10</sup> or addition of functional nanoparticles,<sup>11–16</sup> have been adopted in the recent past to improve the properties of PCL.

In recent times, there has been a significant increase in a new type of pollution in the form of Electromagnetic interference (EMI) owing to the increasing number of electronics and telecommunication devices. EMI is responsible for the malfunctioning of electronic equipments and can also adversely affect human health. Therefore, it is not surprising that a large volume of research is being dedicated toward developing materials that can reduce EMI.<sup>17–25</sup> In this context, different types of materials, such as metals, intrinsically conducting polymers (ICPs), and conducting polymer composites (CPCs), have been explored to attenuate the EM radiations. However, these materials suffer from different limitations. For instance, ICPs present processing difficulties, and moreover it has to be doped in order to make them conducting, which requires additional preparation steps. In addition, metals are difficult to

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process and are prone to corrosion. Hence, most of the studies have been dedicated to fabricate conducting polymer nanocomposites involving conducting fillers such as carbon black (CB), carbon nanofibers (CNFs), carbon nanotubes (CNTs) and metal nanoparticles. Among the other particles, CNTs<sup>26</sup> have proven to be the most effective on account of their high electrical and thermal conductivity, high aspect ratio and superior mechanical properties. It is well studied and documented that owing to their high aspect ratio, CNTs form conducting networks in polymer matrixes at relatively low fraction. This eventually leads to highly conducting nanocomposites. However, the key challenge associated with CNTs is the difficulty to homogeneously disperse them in a given polymer matrix. A recent review<sup>27</sup> addresses various challenges in dispersing CNTs and also highlights various surface treatments of CNTs.

In light of this, we made an attempt to develop biodegradable nanocomposites for EMI shielding applications in this study. This issue has been addressed here in the context of reducing e-waste that contains large amounts of components made of polymer based composites. Two key properties required for EMI shielding applications are high electrical conductivity and good mechanical strength. In order to improve the electrical and mechanical properties of PCL, two different types of MWNTs were melt mixed, an industrially viable route to produce composites in large scale. The state of dispersion of MWNTs was assessed using a scanning electron microscopy (SEM) and broadband dielectric spectroscopy. The mechanical properties of the nanocomposites were evaluated as a function of temperature using dynamic mechanical thermal analyzer. The shielding effectiveness of the nanocomposites was studied using a vector network analyzer in a broad range of frequencies, X-band and K<sub>u</sub>-band. In order to elucidate its degradation enzymatically, we evaluated the degradation of PCL in the presence of a *Pseudomonas* lipase, which is capable of cleaving ester bonds in PCL. Films were subjected to enzymatic degradation at 37 °C in an enzyme containing pH 7 phosphate buffer solution. The results obtained from various analytical techniques are presented herein.

## Experimental section

### Materials

Poly( $\epsilon$ -caprolactone) (PCL, average molecular weight  $M_n = 45\,000\text{ g mol}^{-1}$ ) was obtained from Sigma-Aldrich in pellet form. Both pristine (p-MWNTs) and amine functionalized multiwall carbon nanotubes (a-MWNTs) were obtained from Nanocyl SA Belgium. The characteristics dimensions of p-MWNTs are 9.5 nm (average diameter) and 1.5  $\mu\text{m}$  (average length) and those of a-MWNTs are 9.5 nm (average diameter) and 900 nm (average length). *Pseudomonas* lipase was obtained from Sigma.

### Preparation of the nanocomposites

Nanocomposites of PCL were processed by melt mixing using a Haake Mini lab II melt compounder. Prior to melt mixing, the p-MWNTs were dispersed in tetrahydrofuran (THF) using a probe

sonicator for 15 min (Hielsher, Ultrasound technology) followed by bath sonication for 45 min. The obtained powder was finally vacuum dried for 48 h at 80 °C. The a-MWNTs were used as received without any pre-treatment. PCL pellets were dried at 40 °C under vacuum for 24 h prior to melt mixing. All the nanocomposites were processed at 80 °C for 20 min at a screw speed of 60 rpm and under nitrogen atmosphere. Various PCL/MWNT composites were prepared by varying the concentration of MWNTs as 1, 2 and 3 wt% of the total batch size.

### Characterization

Dispersion of MWNTs in PCL matrix was analyzed using FEI Sirion XL Fe30 FEG SEM, on cryo-fractured extruded strands. Mechanical properties in the temperature range of  $-80\text{ }^{\circ}\text{C}$  to  $25\text{ }^{\circ}\text{C}$  was studied for various PCL nanocomposites using a Q800 (TA, instruments) dynamic mechanical analyzer in the tension mode at 1 Hz. Specimens of required dimensions ( $25\text{ mm} \times 6.5\text{ mm} \times 1\text{ mm}$ ) were compressed molded at 80 °C. All the experiments were carried out in the linear viscoelastic region, determined *a priori*. Each experiment was repeated three times with a fresh sample, and errors of 2–3% were noted.

Room temperature electrical conductivity of the nanocomposites in the frequency range of 0.1 Hz to 10 MHz was analyzed using Alpha-N Analyser, Novocontrol (Germany). Compression molded discs were used as specimens for these experiments.

Enzymatic degradation of control and PCL nanocomposites was carried out in a  $1\text{ mg ml}^{-1}$  *Pseudomonas* lipase solution in pH 7 phosphate buffered saline (PBS) at 37 °C. The specimens used were 25 mg thin films cut from the compression molded samples. The samples were placed in nylon mesh bags and placed in a 20 ml solution of the enzyme-containing buffer. These tubes were placed in an incubator-shaker (N-Biotek NB-205QF) at 37 °C and 100 rpm. After every 24 h, the samples were removed from the incubator and dried under vacuum to constant weight. The weight losses of the samples were recorded at an interval of 24 h for 10 days. All the measurements were performed in triplicates and the data are presented as mean  $\pm$  S.D.

EMI shielding of the nanocomposites was carried out using an Anritsu MS4642A vector network analyzer (VNA) coupled to a coax (Damaskos M07T) set up. Prior to the measurements, the transmission line and the sample holder were calibrated by a two port (Short-open-load-transmission). Toroidal samples were hot pressed at 80 °C and were used for EMI shielding measurements. The EMI shielding experiments were carried out at room temperature. Total EMI SE was extracted from the scattering parameters ( $S_{11}$ ,  $S_{12}$ ,  $S_{22}$  and  $S_{21}$ ) in X-band (8–12 GHz) and K<sub>u</sub>-band (12–18 GHz) frequency.

## Results and discussion

### State of dispersion of MWNTs in PCL

In order to assess the state of dispersion of p-MWNTs and a-MWNTs in PCL matrix, SEM micrographs of cryo-fractured composites were recorded. Fig. 1 shows the SEM micrographs



of PCL composites with a-MWNTs and p-MWNTs. It is observed that both the MWNTs are relatively well dispersed in the PCL matrix, as can be seen by the white dots denoted by arrows in Fig. 1. As a general observation, the quality of dispersion of a-MWNTs is better than that of p-MWNTs. This can be attributed to specific interactions between the functional groups on a-MWNTs and PCL. It is well understood that beyond the percolation threshold, the conducting particles form a network-like structure and has a significant effect on the bulk electrical conductivity of the composites, and in turn, on the EM shielding. These properties will be discussed in the subsequent sections.

### Dynamic mechanical properties

The dynamic mechanical properties as a function of temperature for PCL nanocomposites were studied under the tension mode. The storage modulus ( $E'$ ) for PCL with a-MWNTs and p-MWNTs are illustrated in Fig. 2(a) and (b), respectively. In all the composites, it is observed that the storage modulus gradually decreased when the temperature increases, a typical characteristic of brittle to ductile transition in polymers. In the case of neat PCL, a drastic decrease in the storage modulus is observed at  $-40^\circ\text{C}$ , which corresponds to the glass transition ( $T_g$ ) temperature. Interestingly, after the addition of MWNTs, the storage modulus, especially in the glassy region, was significantly enhanced, although no significant change is observed in the  $T_g$ . For instance, in the glassy region ( $-70^\circ\text{C}$ ) the neat PCL showed a storage modulus of 3160 MPa, whereas

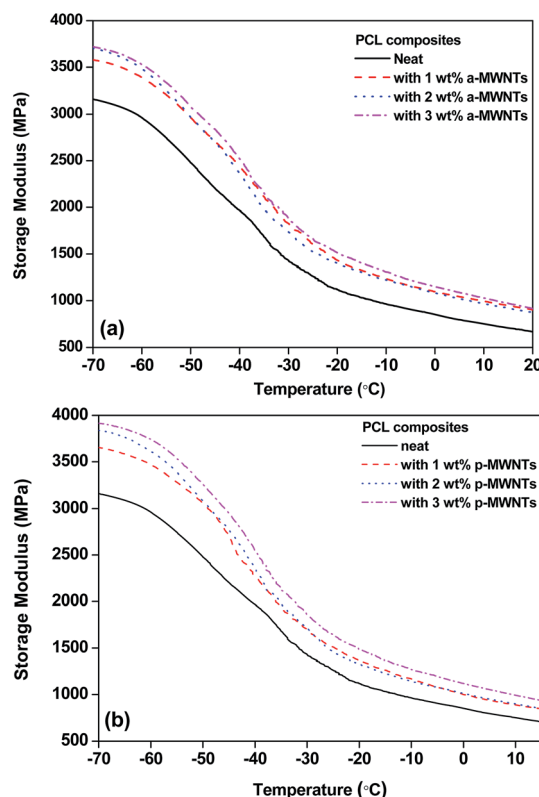


Fig. 2 Storage modulus as a function of temperature for (a) PCL nanocomposites with a-MWNTs along with neat PCL and (b) PCL nanocomposites with p-MWNTs along with neat PCL.

the addition of 1 wt% a-MWNTs and 1 wt% of p-MWNTs exhibited a significant increase in the storage modulus to 3579 MPa and 3657 MPa, respectively. This increase in storage modulus of the nanocomposites is consistent with the increasing fraction of MWNTs. This could be due to the relatively well dispersed MWNTs in the composites leading to improved mechanical properties.

### Bulk electrical conductivity and EMI shielding in the nanocomposites

Fig. 3 illustrates the room temperature AC electrical conductivity ( $\sigma_{AC}$ ) as a function of frequency for various PCL/MWNTs composites. As expected, the neat PCL exhibited a linear increase in AC electrical conductivity as a function of frequency, which is a typical characteristic of an insulator. The electrical conductivity of PCL/MWNT nanocomposites increases along with the concentration of MWNTs in the composites. It is observed that there is a dramatic increase in the bulk electrical conductivity in the case of 1 wt% MWNTs along with a frequency independent plateau in the lower frequencies. This phenomenon clearly suggests that the electrical percolation threshold in PCL nanocomposites lies between 0–1 wt% in the case of both MWNTs. Moreover, this observation is consistent with the type of MWNTs used in this work. For instance, the p-MWNTs are longer by a factor of 2 with respect to a-MWNTs; hence, at any given fraction of MWNTs, the bulk electrical

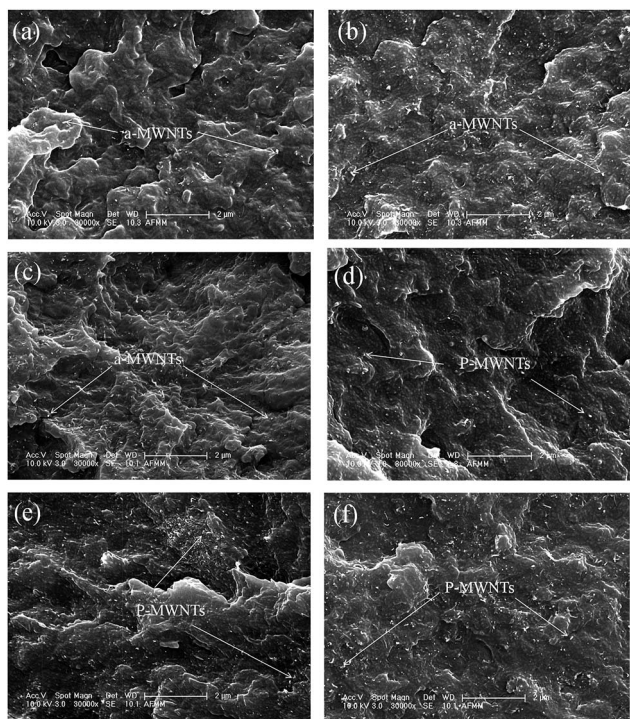


Fig. 1 SEM micrographs of PCL composites with 1, 2 and 3 wt% a-MWNTs respectively (a–c) and 1, 2 and 3 wt% p-MWNTs, respectively, (d–f).





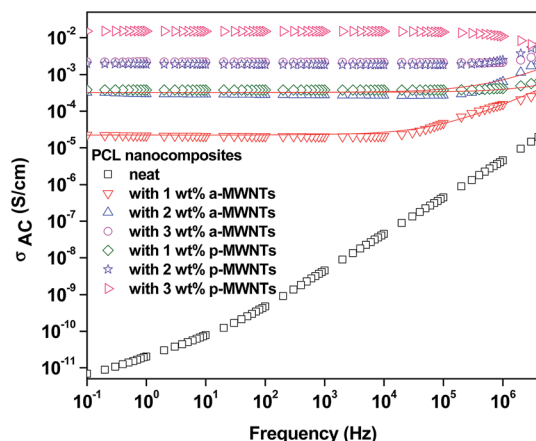


Fig. 3 Alternate current electrical conductivity as a function of frequency for various PCL nanocomposites along with neat PCL. Red colored lines show power law fitting for respective nanocomposites.

conductivity is expected to be higher in PCL/p-MWNTs than in PCL/a-MWNTs composites. Interestingly, it was also observed that the frequency independent plateau extended to higher frequencies with increasing fractions of MWNTs and it became almost independent over the entire measured frequency range. It is reported that the aspect ratio of the conducting nanoparticles has a great impact on the network formation and percolation threshold.<sup>28–31</sup>

We have fitted the AC electrical conductivity scans using a power law to gain more insight into the mechanism of charge transport.<sup>32</sup>

$$\sigma(\omega) = \sigma(0) + \sigma_{ac}(\omega) = \sigma_{dc} + A\omega^s$$

where,  $\omega$  represents angular frequency,  $\sigma_{dc}$  corresponds to direct current (DC) conductivity and 's' is the power, which depends on temperature and frequency and lies between 0–1. 'A' is a constant that depends on temperature. As a general observation, with increasing fractions of MWNTs, the exponent (s) was observed to decrease. Typically for  $s < 0.7$ , it is envisaged that the charge transport is through fractal 3 dimensional network of MWNTs. We could only fit the data for few compositions as the frequency independent plateau extended to the entire measured range of frequencies (see the solid lines through the data in Fig. 3). From the fitting parameters (not shown here), it is observed that at a given fraction of MWNTs (1 wt%), PCL/a-MWNTs represent a 74 : 26 and PCL/p-MWNTs represent 72 : 28 equivalent network of resistance and capacitance. It is evident from the exponent values that hopping dominates the charge transport at lower fractions and becomes tunneling at higher fractions through the fractal network of 3D MWNTs in the composites.

As one of the rationales of the paper is to develop biodegradable EMI shielding materials, the shield effectiveness was measured using a vector network analyzer on toroidal samples. In order to shield electromagnetic radiations, three basic mechanisms have been reported, namely, absorption, reflection and multiple reflections.<sup>21,33,34</sup> EMI shielding materials must

possess high electrical conductivity and electrical/magnetic dipoles. Shielding of EM radiations due to reflection is characteristic of materials with free charge carriers. However, absorption requires the presence of electrical and magnetic dipoles.<sup>35</sup> These dipoles interact with the incident EM radiation and results in further eddy ohmic losses. Absorption losses are strongly dependent on the thickness and often increase with the thickness of the shielding material. Multiple reflections have negative effect on SE and can be neglected for absorption losses above 10 dB.

In this study, the shielding effectiveness was analyzed in broad range of frequencies, X-band (8–12 GHz) and the K<sub>u</sub>-band (12–18 GHz). The X-band frequency is important in the case of commercial, as well as military applications such as transmission line, TV signals and radar transmission systems. Because the total shielding effectiveness is the summation of shielding through absorption ( $SE_A$ ), reflection ( $SE_R$ ) and multiple reflections ( $SE_{MR}$ ), materials exhibiting both reflection and absorption are in great demand. A schematic explaining how the scattering parameters are derived from VNA is shown in Fig. 4a, the mechanism of shielding in composites in Fig. 4b and the VNA attached to a coax set up in Fig. 4c. Total shielding effectiveness can be mathematically expressed as  $SE_T = SE_R + SE_A + SE_{MR}$ . This can be further calculated by knowing the ratio of transmitted to incident power using the following equations:<sup>36–38</sup>

$$SE_T(\text{dB}) = -10 \log \frac{P_T}{P_I} = -20 \log \frac{H_T}{H_I} = -20 \log \frac{E_T}{E_I}$$

Fig. 5a and b depicts the total shielding effectiveness as a function of frequency for various PCL nanocomposites. The total SE was observed to increase with increasing fractions of MWNTs, irrespective of the type of MWNTs. This is consistent

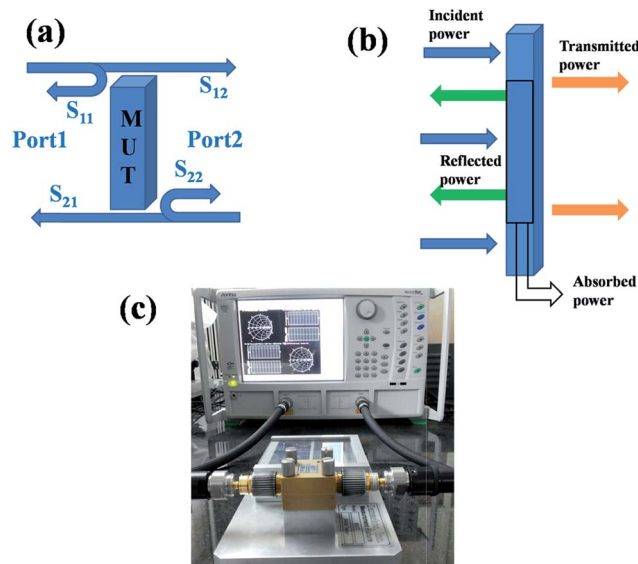


Fig. 4 (a) Schematic representation showing the scattering parameters extracted from two-port VNA, (b) mechanism of shielding and (c) VNA and the coax set up.



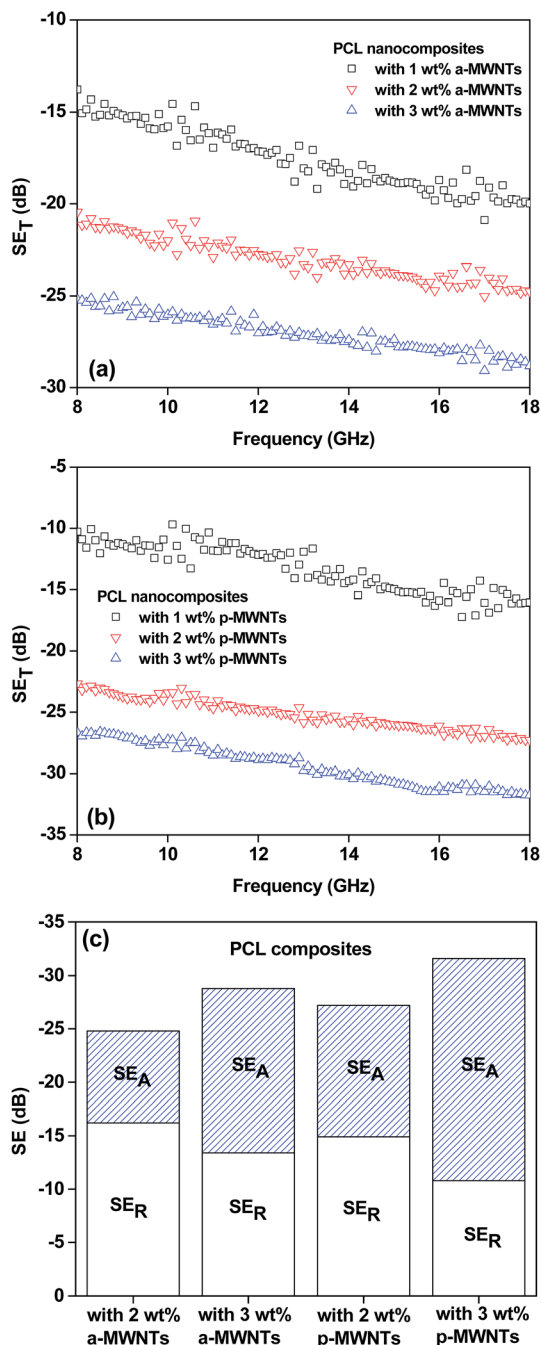


Fig. 5 Total shielding effectiveness as a function of frequency for various PCL nanocomposites with (a) with a-MWNTs, (b) with p-MWNTs and (c) contribution of SE through reflection and absorption to the total SE for PCL composites.

with the observed bulk electrical conductivity of the composites. For instance, the total shielding effectiveness increased from 16 dB for PCL with 1 wt% p-MWNTs to 32 dB for PCL with 3 wt% p-MWNTs at a particular frequency (18 GHz). The relatively higher electrical conductivity and the shielding effectiveness in the nanocomposites suggest an efficient dispersion of MWNTs. Although the a-MWNTs are shorter by a factor of 2 with respect to p-MWNTs, the shielding effectiveness values are only

moderately lower than those of p-MWNTs. It is well understood that the functional groups present on the surface of the MWNTs facilitate an improved dispersion in the matrix.<sup>39</sup> As discussed earlier, the network-like structure of the conducting nanoparticles enhances the attenuation of EM radiations. Interestingly, at higher fractions of MWNTs (>2 wt%), the nanocomposites exhibited an attenuation of EM radiations mostly through absorption.

Fig. 5c depicts the contribution of reflection and absorption to the total EM shielding effectiveness for PCL composites. It is clearly evident that, at 2 wt% of MWNTs loading, the reflection is the dominating mechanism, whereas at 3 wt% MWNTs loading, the absorption is dominating the attenuation process. Similar observation is reported by Liu *et al.*<sup>33</sup> in the case of polyurethane composites with single walled carbon nanotubes (SWNTs), where the attenuation mechanism shifted from reflection to absorption at higher fractions of SWNT loading.

In the case of uniformly conducting shields, the shield thickness has no effect on the reflection mechanism, whereas for multi component systems, such as polymer composites, an increase in shield thickness results in the enhancement of SE through both reflection and absorption. In addition, the shielding mechanism is also dependent on electrical conductivity and connectivity of the conducting nanoparticles, which dictates the spacing between MWNTs in a given polymeric shield. Increasing electrical conductivity and decreasing spacing between MWNTs result in an enhanced attenuation through absorption. A similar observation is also evident in PCL based composites studied in this work, where at lower concentrations of MWNTs, the spacing between MWNTs is very high (less connectivity), and reflection seems to be the dominant mechanism of shielding. In the case of composites with higher concentrations of MWNTs, where spacing between MWNTs is very small (high connectivity), absorption is the dominant mechanism. The shielding mechanism in PP/MWNTs composites was discussed in detail by Al-Saleh *et al.*<sup>19</sup> It was observed that 1 vol% (approx: 2 wt%) MWNTs resulted in 56% attenuation through absorption, which increased to 84% with 5 vol% MWNT in PP. Furthermore, in the case of epoxy composites with single walled carbon nanotubes (SWNTs) Li *et al.*<sup>40</sup> showed that the contribution of reflection is much more than that one of absorption. Similarly, Yang *et al.*<sup>41</sup> reported that reflection is the main mechanism of shielding in polystyrene composites with carbon nanotubes. In this work, PCL composites with MWNTs also showed a similar type of attenuation behavior, as reported in the literature where absorption is dominating at higher fractions of MWNTs. In the case of PCL composites with 2 wt% a-MWNTs and p-MWNTs, the absorption was ca. 34.6% and ca. 45.2%, respectively. However, at 3 wt% loading of a-MWNTs and p-MWNTs SE through absorption was increased to ca. 53.5% and ca. 66%, respectively. The composites with 3 wt% p-MWNTs manifested in highest SE through absorption and can be attributed to enhanced relative permittivity leading to the absorption of electric fields.

In order to assess the effect of the shield thickness on the EM attenuation, specimens with different thicknesses were studied here. Fig. 6a illustrates the total shielding effectiveness as a



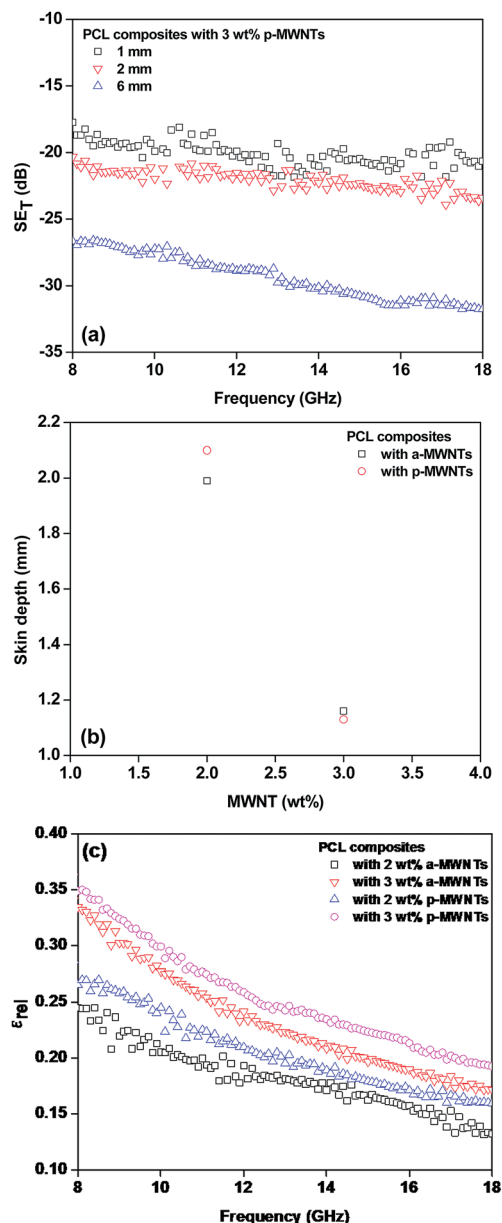


Fig. 6 (a) Total shielding effectiveness as function of frequency at different thicknesses, (b) skin depth as function of wt% of MWNT loading and (c) relative permittivity as function of frequency.

function of shield thickness for various PCL composites with 3 wt% p-MWNTs. It was observed that the total SE increased along with the thickness of the specimen. This enhanced attenuation is mainly due to the increased absorption (not shown here) with increasing thickness of the shield. These observations show that shield thickness and the fraction of conducting filler has prominent effects on the attenuation mechanism. The absorption component increased along with the shield thickness. It is well known that electromagnetic waves of very high frequencies cannot penetrate into an electrically conducting shield. The strength of the radiation decreases exponentially as it penetrates into the shield, and the

critical thickness at which the EM radiation field strength drops to  $1/e$  of the incident field is known as skin depth. The skin depth is dependent on the frequency, permeability and conductivity of shield and can be estimated from the following expression:<sup>42</sup>

$$\delta = \sqrt{\frac{2}{f\mu\sigma}} = -8.68 \times \frac{t}{SE_A}$$

where,  $\delta$ ,  $f$ ,  $\mu$  and  $\sigma$  are the skin depth, frequency, permeability and electrical conductivity, respectively; and  $t$  and  $SE_A$  are the specimen thickness and shielding effectiveness through absorption, respectively. Conductivity of the shield (fraction of conducting filler) has a dominant effect on skin depth, and it decreases with conductivity. In the case of PCL based composite, as conductivity increases with increasing MWNT fraction, the skin depth is expected to decrease. A similar effect was observed here. The PCL composites with 3 wt% p-MWNTs manifested lower skin depth, in contrast to the composites with 1 and 2 wt% MWNTs. Fig. 6b depicts the variation of skin depth as a function of the concentration of MWNTs. It is evident that with increasing fractions of MWNTs, the skin depth decreases. This is mainly due to enhanced electrical conductivity and smaller distance between the MWNTs. It is envisaged that skin depth for polymer composites lie in the range of few mm and decreases with the increasing conductivity of the shield. In the case of polyurethane composites with 1 wt% MWNTs, the skin depth was observed to be 1.4 mm at 12 GHz, which decreases when the fraction of conducting filler<sup>42</sup> increases. Similarly, for polyaniline composites<sup>43</sup> with graphene the skin depth was observed to be in the range of 1 to 3 mm. In this work, toroidal specimens of 6 mm thickness were used, which is well above the expected skin depth.

In order to comprehend the attenuation mechanism, we evaluated the permittivity values of PCL composites, which were estimated through the backward and the forward scattering parameters using the Nicolson and Ross line theory.<sup>44</sup> The reflection coefficient of the incident wave on the specimen and line interface of toroidal specimen with known thickness can be expressed using the following equation:

$$\Gamma = \frac{Z - Z_0}{Z + Z_0} = \frac{\sqrt{\mu_R/\epsilon_R} - 1}{\sqrt{\mu_R/\epsilon_R} + 1}$$

The transmitting coefficient for toroidal specimen across the thickness can be expressed as follows:

$$Z = \exp -j\omega\sqrt{\mu\epsilon}t = \exp \left[ -j \left( \omega/c\sqrt{\mu_R\epsilon_R} t \right) \right]$$

where,  $Z_0$  and  $Z$  are the characteristic impedance of coaxial line and the sample holder. The  $\epsilon_R$  and  $\mu_R$  are the complex permittivity and permeability, respectively.

$V_1$  and  $V_2$  are sum and difference of scattering parameters, respectively

$$V_1 = S_{21} + S_{11}$$

$$V_1 = S_{21} - S_{11}$$



Scattering parameters can be used to estimate the reflection coefficient ( $\Gamma$ ) using the following expression:

$$\Gamma = X \pm \sqrt{X^2 - 1}$$

where,  $X = \frac{1 - V_1 V_2}{V_1 - V_2}$ , and in order to satisfy the condition of  $|\Gamma| \leq 1$ , the appropriate sign was chosen.

From relation 1,

$$\frac{\mu_R}{\epsilon_R} = \left( \frac{1 + \Gamma}{1 - \Gamma} \right)^2 = c_1$$

And from relation 2,

$$\mu_R \epsilon_R = - \left\{ \frac{c}{\omega l} \ln \left( \frac{1}{z} \right) \right\}^2 = c_2$$

Complex permittivity and permeability can be presented as

$$\epsilon_R = \sqrt{\frac{c_2}{c_1}}$$

and

$$\mu_R = \sqrt{c_1 c_2}$$

The relative permittivity obtained through scattering parameters as a function of frequency for PCL composites is illustrated in Fig. 6c. PCL composites with 2 wt% a-MWNTs and 2 wt% p-MWNTs manifested in 0.24 and 0.27, respectively, at 18 GHz frequency, whereas higher loading of MWNTs showed increased relative permittivity. The PCL composites with 3 wt% p-MWNTs manifested the highest permittivity among the composites, 0.35. This particular composite also resulted in the highest attenuation of EM radiation.

As mentioned earlier, different attenuation mechanisms are responsible for EM shielding, namely, reflection, absorption and multiple reflections. Multiple reflections can be ignored for specimens with absorption more than 10 dB, or if the shield thickness is more than the skin depth. In order to have insight of attenuation mechanisms for PCL composites, the shielding effectiveness through absorption and reflection were estimated using  $S$  parameters obtained from vector network analyzer using following relations:

$$SE_A = 10 \times \log_{10}((1 - S_{11}^2)/S_{12}^2)$$

$$SE_R = 10 \times \log_{10}(1/(1 - S_{11}^2))$$

where,  $SE_A$  and  $SE_R$  are shielding effectiveness through absorption and reflection respectively.  $S_{11}$  and  $S_{12}$  are the forward reflection coefficient and the reverse transmission coefficient respectively.

### Enzymatic degradation: effect of MWNTs

The effect of different MWNTs on the biodegradability of PCL composites was assessed through enzymatic degradation. The

enzymatic degradation of samples, cut from the hot pressed films, was investigated at 37 °C in a pH 7 phosphate buffer containing *pseudomonas* lipase. Fig. 7 shows the weight loss as a function of the incubation time for neat PCL and its nanocomposites. From this study, it is clearly evident that PCL has significant degradability in enzymatic environment. In the presence of *pseudomonas* lipase, PCL showed 66% weight loss in 10 days. PCL nanocomposites with p-MWNTs showed a dramatic reduction in the degradation rate. The PCL composites with 3 wt% p-MWNTs resulted in negligible (6 wt%) degradation, even after 10 days of exposure. Paul *et al.*<sup>45</sup> reported that the rate of hydrolytic degradation strongly depends on the surface water-wettability of the nanocomposites. In the case of PCL nanocomposites, Kumar *et al.*<sup>46</sup> showed that the nanocomposites with p-MWNTs are more hydrophobic in nature, in striking contrast to what happens to the neat PCL. Therefore, the suppression of degradation rate of PCL nanocomposites could be attributed to its hydrophobic nature. Interestingly, composites with a-MWNTs manifested nearly the same degradation rate as that of neat PCL. In order to gain more insight into this behavior, we took a closer look at the nature of interactions between PCL and a-MWNTs. The melt mixed composites were dissolved in chloroform to remove the PCL phase and the nanoparticles obtained were filtered and dried. The FTIR spectra of the neat PCL and PCL/a-MWNT composites are shown in Fig. 8a. The characteristic peak at 1734 cm<sup>-1</sup> corresponding to the ester groups of PCL is evident from the spectra. Similarly, the characteristic C-H stretching peaks corresponding to MWNT in the range of 2800 to 2900 cm<sup>-1</sup> is also evident in both p-MWNT and a-MWNTs. In addition, in the PCL/a-MWNT nanocomposites a peak at 1635 cm<sup>-1</sup> that corresponds to ester-amide linkage is observed. The FTIR spectra of the extracted nanoparticles from the composites are shown in Fig. 8b. Interestingly, the presence of ester groups of PCL on the surface of the remaining a-MWNTs clearly indicates the interaction between a-MWNTs and PCL. This was not observed in the case of extracted p-MWNTs. These findings are indicative of covalently-bound PCL on a-MWNTs, which did not dissolve in chloroform. In contrast, due to the lack of strong covalent

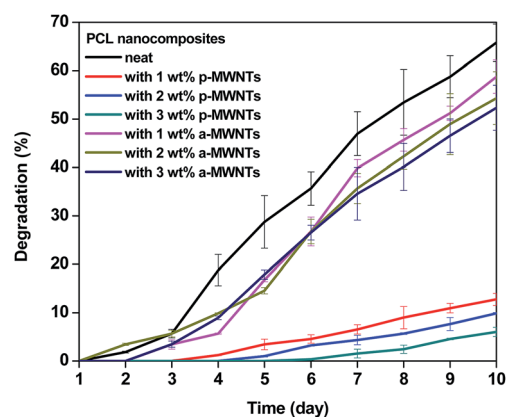


Fig. 7 Temporal degradation profile of PCL nanocomposites at 37 °C in a pH 7 phosphate buffer containing *pseudomonas* lipase.





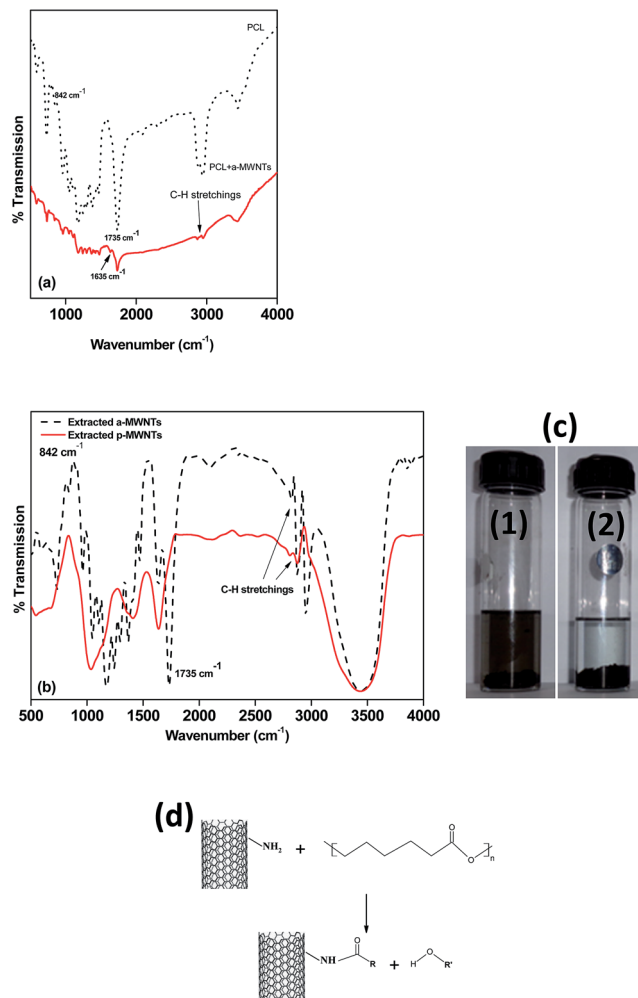


Fig. 8 FTIR spectra of (a) PCL; (b) different MWNTs extracted from the nanocomposites by completely dissolving PCL using chloroform; (c) vials showing the state of dispersion of a-MWNTs (vial 1) and p-MWNTs (vial 2) in the nanocomposites after sonicating for few minutes; (d) possible reaction between a-MWNTs and PCL.

interactions between PCL and p-MWNTs, the PCL matrix was fully dissolved plausibly because the chains were only physically adsorbed. The interaction of a-MWNTs with PCL is further supported by the solution dissolution test, where a fixed amount of the composite samples were dissolved in chloroform and sonicated for few minutes (see Fig. 8c). It is evident that a-MWNTs suspends well (vial 1 in Fig. 8c) in the solution, in striking contrast to p-MWNTs, which quickly settles down (vial 2 in Fig. 8c) after the initial sonication process. The possible reaction mechanism is schematically illustrated in Fig. 8d. This presumably can explain the faster degradation rate in PCL/a-MWNTs. Moreover, as mentioned earlier, the addition of a-MWNTs increases surface hydrophilicity in the nanocomposites, facilitating the faster degradation of PCL, in contrast to p-MWNTs.

The Scanning Electron Microscopy (SEM) images of PCL and its nanocomposites were assessed to analyze the morphological changes due to the degradation process. Fig. 9 depicts the SEM

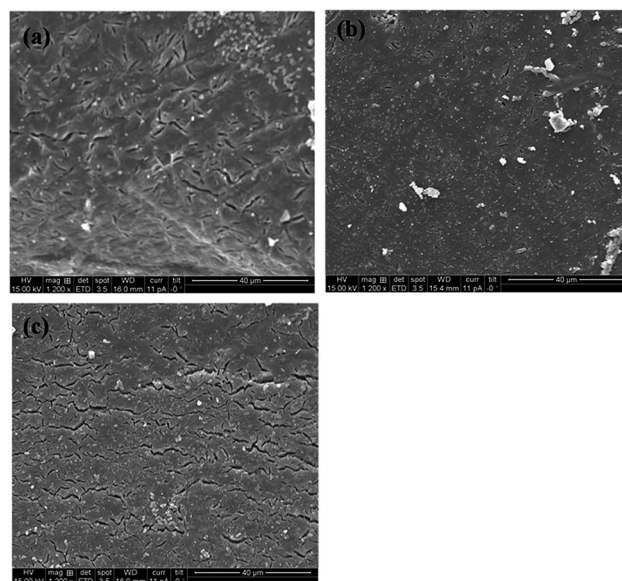


Fig. 9 SEM micrographs of the sample surface after 10 days of enzymatic degradation for (a) neat PCL, (b) PCL with 3 wt% p-MWNTs and (c) PCL with 3 wt% a-MWNTs.

micrographs of PCL and its nanocomposites after the degradation tests at 10 days. It is observed that neat PCL has gone through profound degradation, and this is manifested from the cracks that appeared on the surface (see Fig. 9a). The morphology of the PCL/p-MWNTs films exhibited no significant change (Fig. 9b) and is consistent with the degradation kinetics, as observed in Fig. 7. In the case of PCL with a-MWNTs (Fig. 9c), similar morphological features were recorded as that of neat PCL.

To summarize, we have developed biodegradable nanocomposites based on PCL and MWNTs, which can be explored as potential EMI shielding films. A schematic figure (Fig. 10) further illustrates how these eco-friendly nanocomposites can significantly contribute toward the reduction of e-waste, which is a major concern at the present time.

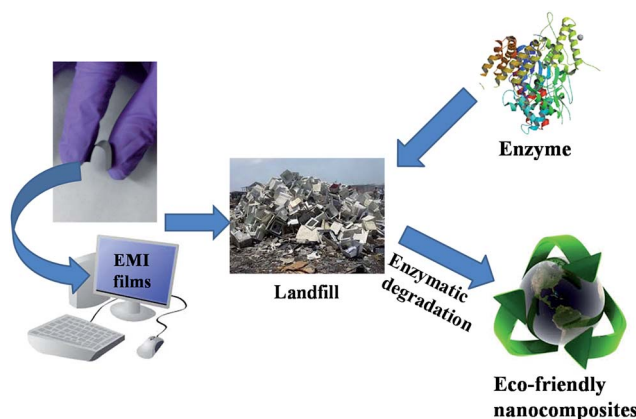


Fig. 10 A schematic illustration of enzymatically degradable PCL/MWNT composites as potential eco friendly EMI shielding materials.





## Conclusion

Bio-degradable nanocomposites with enhanced microwave attenuation ability were designed from PCL and MWNTs. Two different types of MWNTs were incorporated in the PCL matrix by melt mixing process. PCL nanocomposites with 3 wt% p-MWNTs and a-MWNTs manifested SE of  $-32$  and  $-29$  dB, respectively, which shows an approximately 99% attenuation of incident radiation. The enzymatic degradation of the samples, cut from the hot pressed films by bacterial lipase was investigated. It was noted that a-MWNTs exhibited almost the same degradation rate as the control PCL; however, p-MWNTs showed a slower degradation rate. This study opens new avenues in designing biodegradable PCL/MWNT based composites that can be explored as potential EMI shielding materials. These nanocomposites may contribute extensively to control the e-waste generated due to electronic equipments.

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## References

- 1 K. Cho, J. Lee and P. Xing, *J. Appl. Polym. Sci.*, 2002, **83**, 868–879.
- 2 F. Causa, P. A. Netti, L. Ambrosio, G. Ciapetti, N. Baldini, S. Pagani, D. Martini and A. Giunti, *J. Biomed. Mater. Res., Part A*, 2006, **76**, 151–162.
- 3 L. S. Nair and C. T. Laurencin, *Prog. Polym. Sci.*, 2007, **32**, 762–798.
- 4 C. V. Benedict, J. A. Cameron and S. J. Huang, *J. Appl. Polym. Sci.*, 1983, **28**, 335–342.
- 5 W. J. Cook, J. A. Cameron, J. P. Bell and S. J. Huang, *J. Polym. Sci., Polym. Lett. Ed.*, 1981, **19**, 159–165.
- 6 I. Arvanitoyannis, A. Nakayama, N. Kawasaki and N. Yamamoto, *Angew. Makromol. Chem.*, 1994, **222**, 111–123.
- 7 M. A. Woodruff and D. W. Huttmacher, *Prog. Polym. Sci.*, 2010, **35**, 1217–1256.
- 8 C.-S. Wu, *Polym. Degrad. Stab.*, 2003, **80**, 127–134.
- 9 L. Cabedo, J. Luis Feijoo, M. Pilar Villanueva, J. M. Lagarón and E. Giménez, *Macromol. Symp.*, 2006, **233**(1), 191–197.
- 10 M. Todo, S.-D. Park, T. Takayama and K. Arakawa, *Eng. Fract. Mech.*, 2007, **74**, 1872–1883.
- 11 B. Lepoittevin, M. Devalckenaere, N. Pantoustier, M. Alexandre, D. Kubies, C. Calberg, R. Jérôme and P. Dubois, *Polymer*, 2002, **43**, 4017–4023.
- 12 W. Kai, Y. Hirota, L. Hua and Y. Inoue, *J. Appl. Polym. Sci.*, 2008, **107**, 1395–1400.
- 13 A. Arbelaiz, B. Fernández, A. Valea and I. Mondragon, *Carbohydr. Polym.*, 2006, **64**, 224–232.
- 14 B. Chen and K. Sun, *Polym. Test.*, 2005, **24**, 978–982.
- 15 M. C. Azevedo, R. L. Reis, M. B. Claase, D. W. Grijsma and J. Feijen, *J. Mater. Sci.: Mater. Med.*, 2003, **14**, 103–107.
- 16 B. Chen, K. Sun and T. Ren, *Eur. Polym. J.*, 2005, **41**, 453–457.
- 17 S. P. Pawar, K. Pattabhi and S. Bose, *RSC Adv.*, 2014, **4**, 18842–18852.
- 18 H. Kim, K. Kim, C. Y. Lee, J. Joo, S. Cho, H. S. Yoon, D. A. Pejaković, J. W. Yoo and A. J. Epstein, *Appl. Phys. Lett.*, 2004, **84**, 589–591.
- 19 M. H. Al-Saleh and U. Sundararaj, *Carbon*, 2009, **47**, 1738–1746.
- 20 D. D. L. Chung, *Carbon*, 2012, **50**, 3342–3353.
- 21 M. H. Al-Saleh, W. H. Saadeh and U. Sundararaj, *Carbon*, 2013, **60**, 146–156.
- 22 M. Sharma, S. Sharma, J. Abraham, S. Thomas, G. Madras and S. Bose, *Mater. Res. Express*, 2014, **1**, 035003.
- 23 R. Rohini and S. Bose, *ACS Appl. Mater. Interfaces*, 2014, **6**, 11302–11310.
- 24 S. P. Pawar, D. A. Marathe, K. Pattabhi and S. Bose, *J. Mater. Chem. A*, 2015, **3**, 656–669.
- 25 M. Sharma, M. P. Singh, C. Srivastava, G. Madras and S. Bose, *ACS Appl. Mater. Interfaces*, 2014, **6**, 21151–21160.
- 26 O. Breuer and U. Sundararaj, *Polym. Compos.*, 2004, **25**, 630–645.
- 27 X.-L. Xie, Y.-W. Mai and X.-P. Zhou, *Mater. Sci. Eng., R*, 2005, **49**, 89–112.
- 28 J. Li, P. C. Ma, W. S. Chow, C. K. To, B. Z. Tang and J. K. Kim, *Adv. Funct. Mater.*, 2007, **17**, 3207–3215.
- 29 S.-H. Yao, Z.-M. Dang, M.-J. Jiang, H.-P. Xu and J. Bai, *Appl. Phys. Lett.*, 2007, **91**, 212901.
- 30 J. B. Bai and A. Allaoui, *Composites, Part A*, 2003, **34**, 689–694.
- 31 M. R. Ayatollahi, S. Shadlou, M. M. Shokrieh and M. Chitsazadeh, *Polym. Test.*, 2011, **30**, 548–556.
- 32 K. L. Ngai, C. T. White and A. K. Jonscher, *Nature*, 1979, **277**, 185–189.
- 33 Z. Liu, G. Bai, Y. Huang, Y. Ma, F. Du, F. Li, T. Guo and Y. Chen, *Carbon*, 2007, **45**, 821–827.
- 34 K. Lakshmi, H. John, K. T. Mathew, R. Joseph and K. E. George, *Acta Mater.*, 2009, **57**, 371–375.
- 35 D. D. L. Chung, *J. Mater. Eng. Perform.*, 2000, **9**, 350–354.
- 36 B. P. Singh, Prabha, P. Saini, T. Gupta, P. Garg, G. Kumar, I. Pande, S. Pande, R. K. Seth, S. K. Dhawan and R. B. Mathur, *J. Nanopart. Res.*, 2011, **13**, 7065–7074.
- 37 Y.-E. Moon, J. Yun and H.-I. Kim, *J. Ind. Eng. Chem.*, 2013, **19**, 493–497.
- 38 A. P. Singh, M. Mishra, P. Sambyal, B. K. Gupta, B. P. Singh, A. Chandra and S. K. Dhawan, *J. Mater. Chem. A*, 2014, **2**, 3581–3593.
- 39 S. Bose, R. A. Khare and P. Moldenaers, *Polymer*, 2010, **51**, 975–993.
- 40 N. Li, Y. Huang, F. Du, X. He, X. Lin, H. Gao, Y. Ma, F. Li, Y. Chen and P. C. Eklund, *Nano Lett.*, 2006, **6**, 1141–1145.
- 41 Y. Yang, M. C. Gupta, K. L. Dudley and R. W. Lawrence, *Nano Lett.*, 2005, **5**, 2131–2134.
- 42 T. K. Gupta, B. P. Singh, S. R. Dhakate, V. N. Singh and R. B. Mathur, *J. Mater. Chem. A*, 2013, **1**, 9138–9149.
- 43 K. Singh, A. Ohlan, V. H. Pham, R. Balasubramanian, S. Varshney, J. Jang, S. H. Hur, W. M. Choi, M. Kumar,



- S. K. Dhawan, B. S. Kong and J. S. Chung, *Nanoscale*, 2013, **5**, 2411–2420.
- 44 A. M. Nicolson and G. F. Ross, *IEEE Trans. Instrum. Meas.*, 1970, **19**, 377–382.
- 45 M.-A. Paul, C. Delcourt, M. Alexandre, P. Degée, F. Monteverde and P. Dubois, *Polym. Degrad. Stab.*, 2005, **87**, 535–542.
- 46 S. Kumar, S. Bose and K. Chatterjee, *RSC Adv.*, 2014, **4**, 19086–19098.

