REVIEW

Recent applications of chiral *N-tert*-butanesulfinyl imines, chiral diene ligands and chiral sulfur–olefin ligands in asymmetric synthesis

Han-Qing Dong,†[a] Ming-Hua Xu,*[b] Chen-Guo Feng,[a] Xing-Wen Sun[c] and Guo-Qiang Lin*[a]

This review highlights the recent applications of chiral *N-tert*-butanesulfinyl imines and chiral diene ligands, with bicyclo[3.3.0]octadiene and dicyclopentadiene skeletons, in asymmetric chemical transformations. The chiral sulfinamide–olefin products from allylation of *N-tert*-butanesulfinyl imines can be used as hybrid ligands for transition metal-catalyzed asymmetric reactions. The efforts in the further exploration of chiral sulfur–olefin ligands are also discussed.

[a] CAS Key Laboratory of Synthetic Chemistry of Natural Substances, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China. E-mail: lingq@sioc.ac.cn
[b] State Key Laboratory of Drug Research, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, 555 Zuchongzhi Road, Shanghai 201203, China. E-mail: xumh@simm.ac.cn
[c] Department of Chemistry, Fudan University, 220 Handan Road, Shanghai 200433, China
† Present address: Arvinas Inc., 5 Science Park, New Haven, CT 06511, USA.

1. Introduction

Asymmetric synthesis with both high level of selectivities and yields remains a challenge in spite of significant achievements in the last few decades.[1] Chiral *N-tert*-butanesulfinyl imines have been intensively investigated to generate enantioenriched 1,2-diamines and β-amino alcohols through SmI_2-induced homo-coupling and cross-coupling with nitrones and aldehydes. With our discovery of these useful transformations, we demonstrated their synthetic applicability in the total syntheses of (3*R*,4*S*)-statine, D-*erythro*-sphinganine, (+)-febrifugine, (+)-CP-99 994 and (+)-L-733 060. Additionally, we studied the

Han-Qing Dong

Han-Qing Dong obtained his Ph.D. in organic chemistry from the Shanghai Institute of Organic Chemistry (SIOC), Chinese Academy of Sciences (CAS) in 1997. He was awarded the Alexander von Humboldt (AvH) fellowship and performed research in Prof. Volker Jäger's laboratory at the University of Stuttgart in Germany (1998–1999). Subsequently, he moved to the USA and conducted post-doctoral research in Prof. Andrew S. Kende's group at the University of Rochester (1999–2001). Dr Dong has extensive experience in synthetic organic chemistry and drug discovery. He has worked on multiple oncology targets at OSI Pharmaceuticals/Astellas Pharma (2002–2013), and is currently a lead chemist at Arvinas Inc.

Ming-Hua Xu

Ming-Hua Xu was born in Zhejiang, China. He obtained his Ph.D. from SIOC in 1999. Following postdoctoral research at the University of Virginia and Georgetown University Medical Center, he joined the faculty at SIOC as an associate professor in 2003. He has been a professor at the Shanghai Institute of Materia Medica, Chinese Academy of Sciences since 2005. His research interests mainly involve asymmetric synthesis and drug discovery.

This journal is © the Partner Organisations 2015

Org. Chem. Front., 2015, **2**, 73–89 | 73

Zn/In-mediated allylation of *N-tert*-butanesulfinyl imines to provide a variety of chiral homoallylic amines.[2] In the meantime, we discovered a series of bicyclo[3.3.0]octadiene and dicyclopentadiene-based ligands, and successfully employed them in rhodium-catalyzed asymmetric addition to imines, α,β-unsaturated ketones/esters, nitroalkenes and α,β-unsaturated γ-lactams.[3] In this review, we highlight our recent advancement in the allylation of chiral *N-tert*-butanesulfinyl imines, as well as novel asymmetric reactions using our chiral diene ligands. In addition, by combining *N-tert*-butanesulfinamide and olefin elements into one ligand molecule, an unprecedented type of sulfur–olefin hybrid ligand has been designed. Gratifyingly, the use of chiral sulfinamide-containing allylation products as ligands in rhodium-catalyzed asymmetric reactions proved to be successful (Fig. 1). Furthermore, a variety of sulfur–olefin ligands were prepared and evaluated, and excellent enantioselectivities were achieved in a series of asymmetric arylation reactions of C=C, C=O and C=N bonds.[4]

2. Synthetic applications of chiral *N-tert*-butanesulfinyl imines toward α-chiral amines

Chiral *N-tert*-butanesulfinyl imine **1**, whether in the *R* or *S* form, can be obtained from the corresponding enantiopure *R*
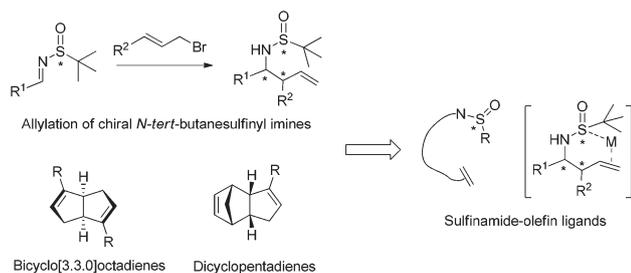

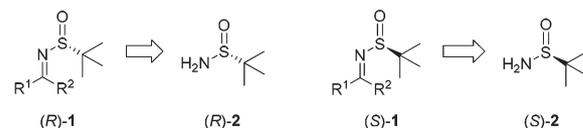**Fig. 1** Our footprints to chiral sulfur−olefin ligands.

Fig. 2 *N-tert*-Butanesulfinyl imines and their synthetic precursors *N-tert*-butanesulfinamides.

or *S N-tert*-butanesulfinamide **2** (Fig. 2). Pioneered by Ellman and co-workers,[5] both enantiomers of imine **1** were applied in the preparation of a wide range of optically active amines. The investigations in this field were rooted in the SmI$_2$-mediated asymmetric synthesis of α,γ-substituted γ-butyrolactones.[6] During the last few years, we found that imine **1** could be subjected to SmI$_2$-mediated pinacol-type coupling reactions, Lewis acid promoted aza-Friedel–Crafts reactions, and Zn/In-mediated allylation reactions, producing a diverse range of new chiral skeletons in organic syntheses.

2.1. SmI$_2$-mediated syntheses of chiral 1,2-diamines and β-amino alcohols

Samarium(II) iodide (SmI$_2$), also known as Kagan's reagent,[7] is a mild one-electron donating agent. Upon treatment with 2 equivalents of SmI$_2$, *N-tert*-butanesulfinyl imines were converted to an anion, which could undergo homo-coupling to provide optically pure C_2-symmetrical 1,2-diamines.[8] The preparation of C_2-unsymmetrical 1,2-diamines requires cross-coupling with different imines. We found that nitrones (iminium ion equivalents) could be applied to achieve our goal. In this case, a nitrone was first reduced by SmI$_2$ to generate an α-aza nucleophilic anion, which subsequently attacked the C=N bond of the *N-tert*-butanesulfinyl imine (Scheme 1).[9] The success of cross-coupling with nitrones encouraged us to explore cross-coupling with aldehydes. Fortunately, such SmI$_2$-mediated cross-coupling reactions could lead to the corresponding chiral β-amino alcohols in excellent diastereoselectivities and enantioselectivities.[10] The synthetic values of SmI$_2$-

Chen-Guo Feng received his MSc (2001) in chemistry from Xiamen University under the supervision of Prof. Pei-Qiang Huang. After he obtained his Ph.D. (2009) from SIOC under the supervision of Prof. Guo-Qiang Lin, he spent two years at the Scripps Research Institute (La Jolla, CA) with Prof. Jin-Quan Yu as a postdoctoral fellow. He is currently an associate professor in SIOC and his research interests include asymmetric catalysis and medicinal chemistry.

Chen-Guo Feng

Xing-Wen Sun was born in Henan province, China. He received his undergraduate education at the Zhengzhou University and graduated in 2002. In the fall of that year, Xing-Wen enrolled in graduate studies at SIOC and joined the research group of Prof. Guo-Qiang Lin. He earned his Ph.D. in organic chemistry in 2007. In July 2007, he began his independent career at Fudan University, where his research program focuses on the total synthesis of natural products and the development of new methods of organic synthesis.

Xing-Wen Sun
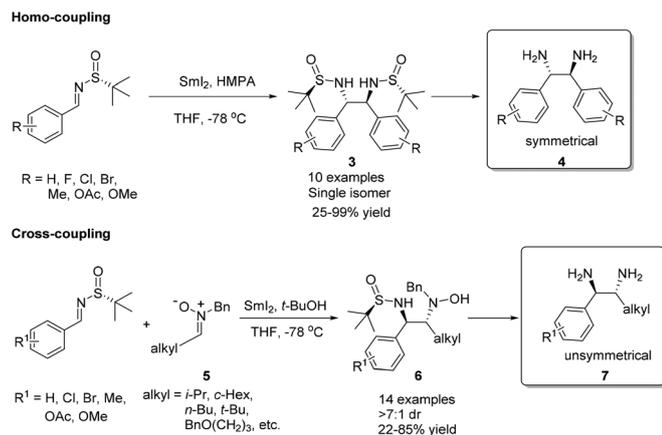
Scheme 1   SmI₂-mediated couplings to provide chiral 1,2-diamines.

Scheme 2   SmI₂-mediated cross-coupling with aldehydes to provide chiral β-amino alcohols.

mediated coupling were demonstrated by the total syntheses of biologically important compounds D-*erythro*-sphinganine, (3*R*,4*S*)-statine and (+)-febrifugine, as well as pharmaceutical agents (+)-CP-99 994 and (+)-L-733 060 (Scheme 2).[11]

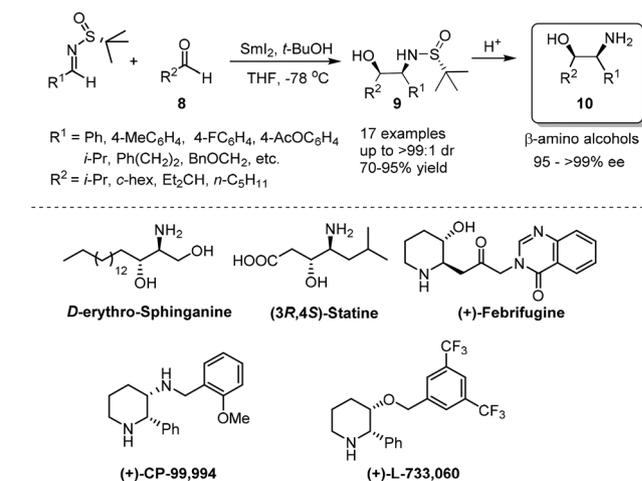## 2.2.   Lewis acid promoted aza-Friedel–Crafts reaction of *N-tert*-butanesulfinyl imines

Most imines possess relatively poor electrophilicity, compared with the corresponding carbonyl compounds.[12] However, *N-tert*-butanesulfinyl imines can provide suitable reactivity toward such nucleophilic addition in the presence of Lewis acids.

Optically active non-proteinogenic amino acids are currently of great interest due to their significant biological activities, and extensive applications in organic synthesis and drug discovery.[13] Asymmetric Friedel–Crafts alkylation of aromatic substrates with *N-tert*-butanesulfinyl glyoxylate imine **12** could be promoted by a proper transition metal-based Lewis acid catalyst, providing a convenient approach to access optically active non-proteinogenic amino acids. Under the optimal reaction conditions, *N-tert*-butanesulfinyl glyoxylate imine **12** could react with a wide range of indoles bearing different substituents, affording α-(3-indolyl)glycines **13** in moderate to

excellent yields and with good to excellent diastereoselectivities (Scheme 3).[14]

Furthermore, a variety of arenes or heteroarenes were also suitable substrates for the reaction with *N-tert*-butanesulfinyl glyoxylate imine **12**. With In(OTf)₃ as the optimal Lewis acid catalyst, the reaction could be accomplished at room temperature, providing a series of enantiomerically enriched α-arylglycines **14** in good yields and with excellent diastereoselectivities (up to 99% de) (Scheme 4). Highly stereoselective double Friedel–Crafts alkylation of 1,3-dimethoxybenzene was investigated, and the reaction proceeded smoothly, with 1 equivalent of In(OTf)₃, to afford a dialkylation product in 43% yield.[15]

9-Aminofluorene and its derivatives are known as important structural motifs in biological compounds.[16] A Lewis-acid-catalyzed asymmetric intramolecular aza-Friedel–Crafts reaction of *N-tert*-butanesulfinyl imine was designed to provide an efficient approach for the preparation of optically active 9-aminofluorenes. Bi(OTf)₃ proved to be the best Lewis acid catalyst in this case, giving the desired products in good yields and with high diastereoselectivities. It is notable that both optically pure diastereomers could be separated by silica gel flash chromatography (Scheme 5).[17]

Guo-Qiang Lin pursued his graduate studies (1964–1968) in SIOC with Prof. Wei-Shan Zhou. He has been a professor in the same institute since 1990. His current research interests include the synthesis of natural products and biologically active compounds, asymmetric catalysis and biotransformation.
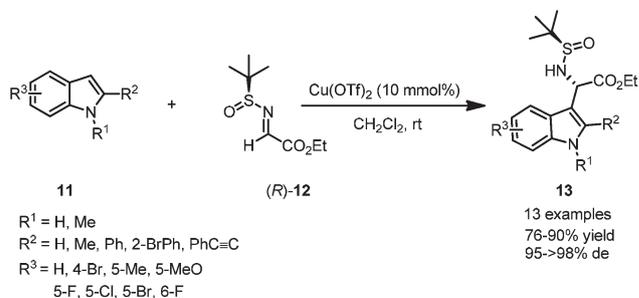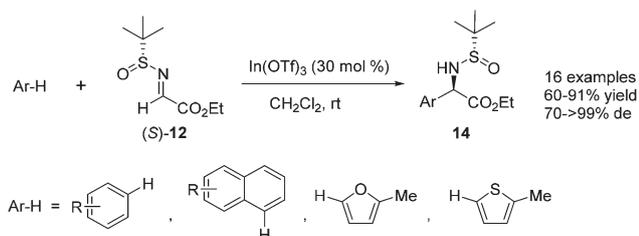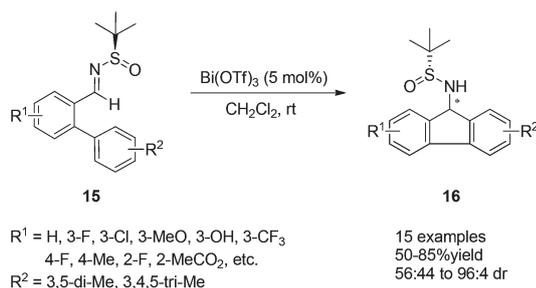
Guo-Qiang Lin

Scheme 3   Lewis acid catalyzed asymmetric aza-Friedel–Crafts reactions of *N-tert*-butanesulfinyl imine to indoles.

Scheme 4 Lewis-acid-catalyzed asymmetric aza–Friedel–Crafts reactions of *N-tert*-butanesulfinyl imine to arenes.

Scheme 5 Lewis acid catalyzed asymmetric intramolecular aza-Friedel–Crafts reactions of *N-tert*-butanesulfinyl imines.

2.3. Zn/In-mediated allylation toward chiral homoallylic amines

Chiral *N-tert*-butanesulfinyl imines could be subjected to the diastereoselective additions of allyl metal reagents. The Ellman group first applied allylmagnesium in addition to chiral *N-tert*-butanesulfinyl imines.[18] Later, the Foubelo group reported the corresponding diastereoselective allylation with allylindium reagents.[19] A six-membered ring transition state with metal cations was proposed to explain the high diastereoselectivities in these two reactions. Our group was also engaged in this kind of allylation, but under the Zn/In-mediated process (Scheme 6). It is worth mentioning that either enantiomeric homoallylic amine could be achieved from a common chiral *N-tert*-butanesulfinyl imine by slightly tuning the reaction conditions. The reaction in HMPA with a small
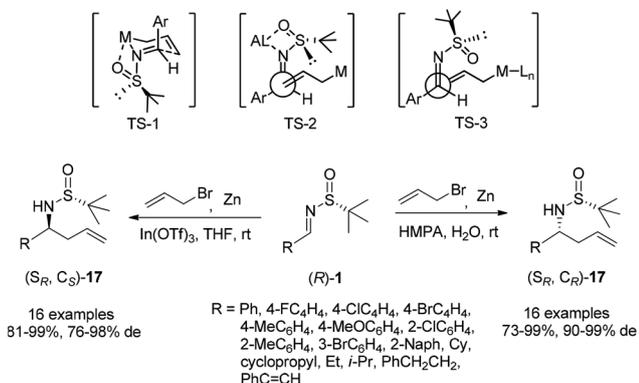
amount of water provided one major diastereomer, while a Lewis acid additive In(OTf)$_3$ in THF led to the preponderant formation of the other diastereomer. The two systems worked well for *N-tert*-butanesulfinyl imines derived from both aliphatic and aromatic aldehydes. Furthermore, the latter was successfully applied in the allylation of ketimines, giving the corresponding quaternary stereocenter-containing chiral homoallylic amines with excellent diastereoselectivities. Two new acyclic transition-states (TS-2 and TS-3), which differ from the previous six-membered chair transition-state model (TS-1), were proposed to explain the stereochemical outcomes (Scheme 6).[20]

The In-mediated allylation could be conducted in aqueous media at room temperature, and saturated aqueous NaBr provided the best result. Both aryl and alkyl aldimines worked well, even in the open air, to afford the corresponding homoallylic amines **17** in high yields and with excellent diastereoselectivities (Scheme 7). The synthetic utility of this method was demonstrated by the convenient preparation of enantio-enriched allylglycine, as well as the synthesis of chiral 3-allyl isoindolinones.[21]

The successful Zn/In-mediated allylation of *N-tert*-butanesulfinyl imines encouraged us to further explore the use of functionalized allylic bromide reagents, in which 3-benzoyloxyallyl bromide (3-bromopropenyl benzoate), 3-phenylallyl bromide (cinnamyl bromide), 3-methyl-2-ethoxycarbonylallylic bromide (ethyl 2-bromomethyl butenoate), as well as heterocyclic allylic bromide (3-bromomethyl-5*H*-furan-2-one) proved to work well, providing diverse chiral allylation products in high yields and with excellent diastereoselectivities.

The reaction of *N-tert*-butanesulfinyl imines with 3-benzoyloxyallyl bromide (3-bromopropenyl benzoate) **22** could generate various β-amino-α-vinyl alcohols.[22] The traditional methods to prepare optically active β-amino alcohols include asymmetric aminohydroxylation of olefins, ring-opening of epoxides and aziridines, nucleophilic addition to α-hydroxyimines or α-aminocarbonyls, α-alkoxyenolate addition to imines, Mannich-type reactions of α-hydroxyketones and imines, as well as direct pinacol-type cross-coupling between carbonyls and imines.[23] This new strategy offers an alternative approach as shown in Fig. 3. In this case, an α-hydroxyallyl carbanion **19** attacks an imine substrate, representing a very straightforward route to β-amino-α-vinyl alcohol derivatives **18**.

The Zn-mediated reaction of 3-benzoyloxyallyl bromide and *N-tert*-butanesulfinyl imines proceeded well in different

Scheme 6 Zn-mediated allylation under two different systems.
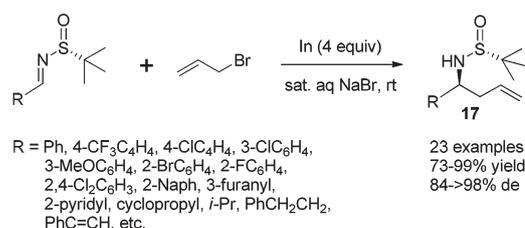
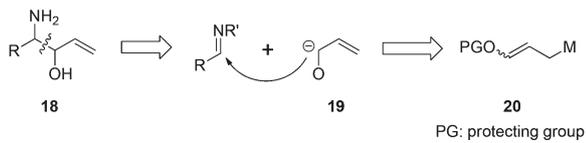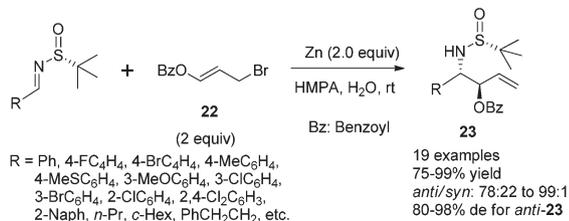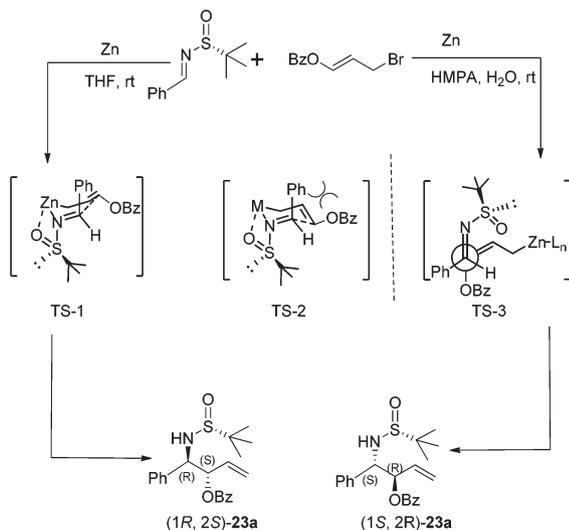**Scheme 7** In-mediated allylation in aqueous media.

Fig. 3 Direct α-hydroxyallylation approach.

solvent systems, and a reversal of stereofacial selectivity was observed again, which has been found in the simple allylic addition as shown in Scheme 8. Two different transition state models (TS-1 and TS-3) were proposed for both reaction pathways (Scheme 9). High reaction yield and stereochemical control can be achieved by using HMPA as the solvent in the presence of a small amount of water (Scheme 9). For aromatic substrates where R were different substituted phenyl or naphthyl groups, the *anti*/*syn* ratio could be achieved up to 99 : 1, and diastereoselectivity up to 98% de. A decreased *anti*/*syn* ratio (78 : 22 to 79 : 21) and diastereoselectivity (80–87% de) were observed when the less sterically hindered *N*-sulfinyl imines, derived from 3-phenylpropanal, butanal and 3-methyl-butanal, were used. Interestingly, the cyclohexyl imine provided the corresponding product with the *anti*/*syn* ratio as 98 : 2 although the diastereoselectivity remained to be moder-
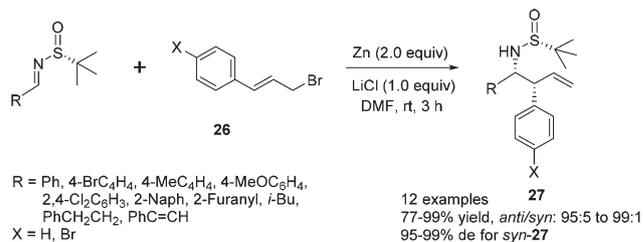
ate (81% de). Notably, we found that the unpleasant solvent HMPA could be replaced by DMF with LiCl as an additive in this reaction.[24] In addition, the use of 1,3-dibromoprop-1-ene (*i.e.*, replacing the benzoyloxy group with bromo in compound **22**) in this Zn-mediated aza-Barbier reaction was found to afford optically active *trans*-2-substituted vinyl aziridines.[25]

The synthetic application of the above benzoylallylation was demonstrated by the efficient asymmetric synthesis of (−)-cytoxazone, a selective modulator of Th2 cytokine secretion isolated from Streptomyces sp.[26] The allylation of *p*-methoxy-benzaldehyde derived (*S*)-imine with 3-benzoyloxyallyl bromide could provide the corresponding *anti*-product **24** in almost quantitative yields, with a 98 : 2 *anti*/*syn* ratio and 92% de. The basic hydrolysis of benzoate and acidic cleavage of the chiral sulfinyl group resulted in β-amino alcohol, which was treated with 1,1′-carbonyldiimidazole (CDI) to give oxazolidinone **25** in 89% overall yield. Ozonolysis of the double bond and its subsequent reduction with sodium borohydride accomplished the total synthesis of (−)-cytoxazone (Scheme 10).

The reaction of *N-tert*-butanesulfinyl imines with 3-arylallyl bromide (cinnamyl bromide) **26** could lead to various β-aryl homoallylic amines **27**.[24] A dramatic lithium chloride effect on the stereocontrol of this reaction was observed after a careful study.[27] Under the optimal conditions, the imines were treated with 2 equivalents of zinc, 2 equivalents of (*E*)-1-bromo-4-(3-bromoprop-1-enyl)benzene and 1 equivalent of LiCl in DMF, which was distilled from activated Al$_2$O$_3$, providing the corresponding homoallylic amines with a *syn*/*anti* ratio up to 99 : 1 and diastereoselectivity up to 99% de (Scheme 11). It is noteworthy that the use of absolute anhydrous DMF, which was

Scheme 8 Zn-mediated benzoyloxyallylation to provide β-amino-α-vinyl alcohols.

Scheme 9 Proposed reaction pathways for Zn-mediated benzoyloxyallylation.
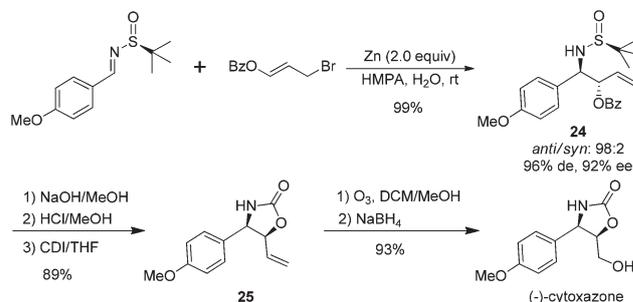
Scheme 10 Total synthesis of (−)-cytoxazone through asymmetric benzoyloxyallylation.

Scheme 11 Zn-mediated cinnamylation to provide β-aryl homoallylic amines.

This journal is © the Partner Organisations 2015

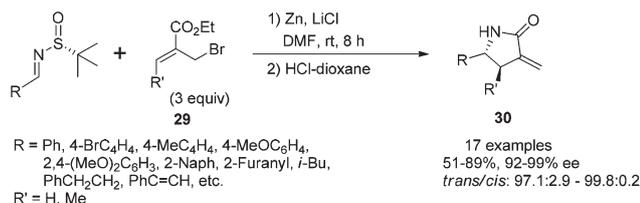*Org. Chem. Front.*, 2015, **2**, 73–89 | **77**

distilled from CaH_2, could cause an obvious decrease in stereo-selectivity (with *syn/anti* ratio of 90 : 10 and 92% de). The cinnamylation of a series of (*R*)-*N-tert*-butanesulfinyl imines was investigated. For aromatic imines, the nature of aryl and heteroaryl R groups did not have a major impact on the *syn/anti* ratios (95 : 5 to 99 : 1), as well as the diastereoselectivities (95–99% de). The reactions with imines derived from aliphatic aldehydes, such as cinnamaldehyde, 3-phenylpropanal and 3-methylbutanal, also proceeded well, giving the corresponding *syn*-products **27** in 95–97% de. Interestingly, the reaction with either (*E*) or (*Z*)-cinnamyl bromide almost exclusively afforded the *syn*-isomer, with *syn/anti* ratios as 97 : 3 and 98 : 2 and diastereoselectivities as 99% and 97% de, respectively.

An attractive application of this cinnamylation method was to use *N*-sulfinyl α-imino ester **12** as an electrophilic substrate (Scheme 12). In this case, the reactions could provide chiral α-amino acid derivatives with a β-vinyl moiety, in higher yields and with better stereoselectivities without the additive LiCl.

The reaction of *N-tert*-butanesulfinyl imines with 3-methyl-2-ethoxycarbonylallylic bromide (ethyl 2-bromomethyl butenoate) **29** could result in highly substituted γ-lactams bearing an α-methylene group (Scheme 13).[28a] Under the optimal conditions, the imines were treated with 3 equivalents of zinc, 3 equivalents of (*Z*)-ethyl 2-(bromomethyl) butenoate and 5 equivalents of LiCl in DMF, followed by the cleavage of the chiral auxiliary with 1 N HCl in dioxane. This one-pot process could produce highly substituted α-methylene-γ-lactams in good yields (51–89%) with excellent *trans/cis* ratios (97.1 : 2.9 to 99.7 : 0.3 when R′ = Me) and enantio-purities (92–99% ee). Various aromatic imines, including aryl and heteroaryl substitutions, afforded the corresponding products with excellent enantioselectivities and diastereoselectivities. The imines derived from aliphatic aldehydes, such as 3-phenylpropanal, 3-methylbutanal, also gave excellent stereoselectivities although only moderate isolated yields were obtained. The syn-

thetic application was demonstrated by the total synthesis of tubulysin V.[28b] As a complementary study for such a one-pot asymmetric synthesis of poly-substituted *trans*-α-methylene-γ-lactams, we achieved the highly efficient construction of *cis*-α-methylene-γ-lactams *via* the Zn-mediated allylation of *N-tert*-butanesulfinyl using 3-(bromomethyl)coumarin as the precursor of an allylic reagent.[28c]
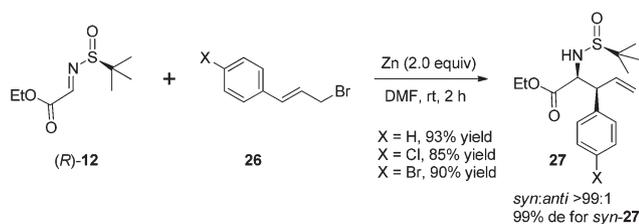
The successful application of *N-tert*-butanesulfinyl imines in the reaction with various linear allylic bromides, including 3-bromopropenyl benzoate, cinnamyl bromide and ethyl 2-bromomethyl butenoate, prompted us to explore the opportunity with heterocyclic allylic bromide. As shown in Scheme 14, 3-bromomethyl-5*H*-furan-2-one **31** could react with *N-tert*-butanesulfinyl imines, providing optically pure α-methylene-γ-butyrolactones bearing chiral amino moieties in the β-position.[29] Under the optimal conditions, the imines were treated with 3 equivalents of zinc and 3 equivalents of 3-bromomethyl-5*H*-furan-2-one in DMF, along with 2 equivalents of water, which likely played an important role to protonate the zinc salt of the allylation adduct. The use of excessive water led to a decrease in the reaction yield, due to the decomposition of the allylic zinc reagent. The reactions produced a mixture of *anti*- and *syn*-isomers, which could be separated by silica gel flash chromatography. For the aryl and alkenyl imines, the reactions afforded the major *anti*-isomers with excellent diastereoselectivities (generally >99% de). However, the diastereoselectivity of the reaction with alkyl imine (derived from 3-phenylpropanal) significantly dropped to 76% de, due to the reduced steric hindrance of the alkyl substituent. Notably, the resulting α,β-unsaturated lactone skeleton from this asymmetric allylation could allow for further rhodium-catalyzed arylation, which introduced a new chiral center in a stereospecific manner (Scheme 15).

In addition to these successful applications of various allylic bromide substrates, we recently conducted a systematic

Scheme 12 Preparation of β-vinyl-containing α-amino acid derivatives.
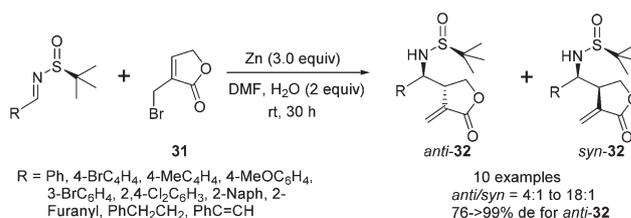
Scheme 14 Zn-mediated allylation of *N-tert*-butanesulfinyl imines using 3-bromomethyl-5*H*-furan-2-one.

Scheme 13 One-pot preparation of highly substituted α-methylene γ-lactams.
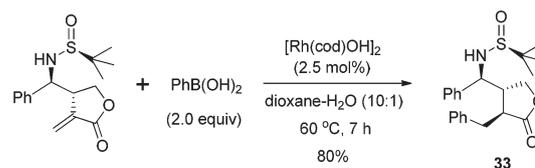
Scheme 15 Rhodium-catalyzed arylation of *anti*-isomer to introduce another chiral center.
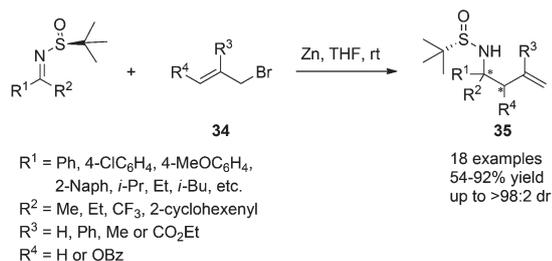
Scheme 16 Zn-mediated allylation of *N-tert*-butanesulfinyl ketimines.
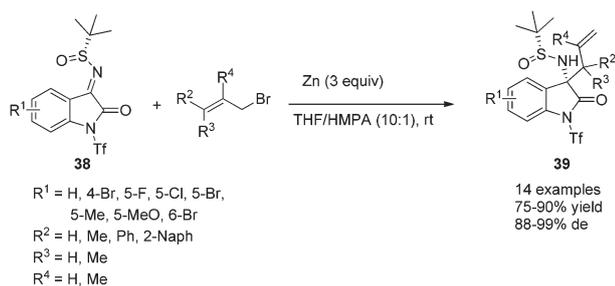
Scheme 17 Zn-mediated allylation of *N-tert*-butanesulfinyl ketimines.

research on the allylation of chiral *N-tert*-butanesulfinyl keti-mines, which were scarcely studied due to their relatively lower reactivity. Although the steric hindrance difference in the two sides of ketimines became less significant, moderate to excel-lent enantioselectivities were still achieved for this Zn-mediated allylation using THF as the solvent (Scheme 16). A six-membered transition-state chair model was proposed for the stereochemical outcome in this case.[30]

The zinc-mediated diastereoselective allylation of isatin-derived *N-tert*-butanesulfinyl ketimines was also achieved at room temperature, providing a highly practical asymmetric approach for the efficient synthesis of chiral tetra-substituted 3-aminooxindoles (Scheme 17).[31]

3. Synthetic applications of chiral diene ligands

Another part of the research focus in our group is to discover new catalytic transformations based on chiral diene ligands. While pursuing the above carbon–carbon bond formation through aza Barbier-type allylation of *N-tert*-butanesulfinyl imines, we further explored the synthetic application of our unique diene ligands highlighted in Fig. 4. The nonbridged bicyclo[3.3.0]octadiene-based ligands were independently introduced by our group[32] and Laschat group[33] in 2007. Both enantiomers in bicyclo[3.3.0]octadiene series could be pre-pared from the commercially available cycloocta-1,5-diene, with lipase-catalyzed enzymatic resolution as a key step. Later, a new family of chiral diend ligand based on the dicyclopenta-diene skeleton was also reported by our group.[34] The synthesis of C_1-symmetric dicyclopentadiene-based ligands applied a
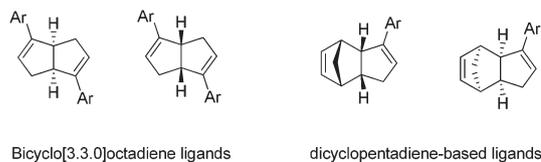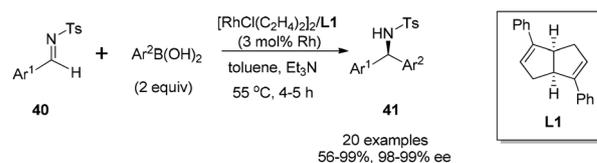


Fig. 4 Our representative chiral diene ligands.

similar synthetic sequence using commercial dicyclopenta-diene as the starting material. In both cases, the kinetic resolu-tion through lipase-catalyzed acetylation could lead to both enantiomers in >98% ee. Recently, the chiral bicyclo[3.3.0] octadiene ligand was successfully incorporated into a polymer chain. Such immobilized chiral diene ligands demonstrated similar catalytic activity and could be recycled for multiple uses.[35]

3.1. Rhodium–diene catalyzed asymmetric addition to imines

Transition metal-catalyzed asymmetric reaction, including the design of efficient chiral ligands, is always an attractive topic in organic synthesis. Since the pioneering work on chiral diene ligands by Hayashi[36] and Carreira,[37] various diene ligands have been designed and evaluated in asymmetric cata-lysis, especially in rhodium- and iridium-catalyzed asymmetric reactions. In many cases, the diene ligands demonstrated better stereoselectivity and catalytic activity than the powerful phosphine ligands.[38] Our diene ligands in both series were successfully utilized in rhodium-catalyzed enantioselective ary-lation of *N*-tosyl imines with aryl boronic acids, and further extended to rhodium-catalyzed conjugate addition of α,β-un-saturated ketones/esters, nitroalkenes, and α,β-unsaturated γ-lactams. In addition to various rhodium-catalyzed reactions, we successfully demonstrated the synthetic utility of chiral pal-ladium–diene complexes in the Suzuki–Miyaura cross-coupling reaction.

Our first synthetic application of C_2-symmetric bicyclo-[3.3.0]octadiene was the rhodium-catalyzed arylation of aryl *N*-tosyl imines with aryl boronic acids.[1c,39] The reaction was carried out with 2 equivalents of aryl boronic acid in the pres-ence of 2 equivalents of triethylamine, 3 mol% of [RhCl-(C₂H₄)₂]₂ and 3.3 mol% of chiral diene at 55 °C for 4–5 h (Scheme 18). A variety of aryl *N*-tosyl imines, regardless of their steric and electronic effects, could afford the corresponding diarylmethyltosylamides **41** with extremely high enantio-selectivities (98–99% ee). In addition to the substituted phenyl

Scheme 18 Catalytic asymmetric arylation of aryl *N*-tosyl imines with arylboronic acids.

This journal is © the Partner Organisations 2015

*Org. Chem. Front.*, 2015, **2**, 73–89 | 79
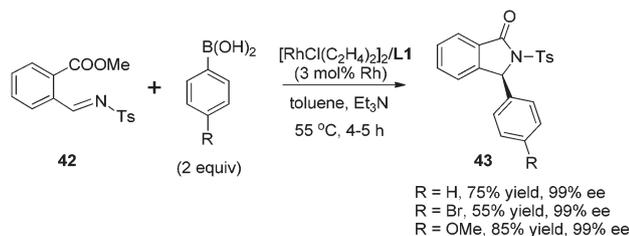
**Scheme 19** Synthesis of chiral 3-aryl-2-tosyl-2,3-dihydroisoindolones.

imines, other aromatic imines including furanyl, thienyl, naphthyl proved to be excellent substrates for this transformation. An attractive application of this asymmetric addition was to use 2-methoxycarbonyl phenyl imine **42** as the reaction substrate, which could conveniently generate chiral phthalimidines (isoindolones) **43** in good yields and excellent enantioselectivities (Scheme 19).[30] Bicyclic heterocyclic imines, for example, *N*-tosyl and *N*-nosyl indolylimines, could also react with aryl boronic acids, providing a new enantioselective approach for the synthesis of α-aryl 2- or 3-indolylmethanamines.[40]

The arylation of aliphatic imines is much more challenging due to the low stability of the imines.[41] Although the arylation of aliphatic *N*-tosyl imines could also yield the corresponding adducts under the above conditions in excellent enantioselectivity (99% ee), the isolated yield was quite low (only 28% yield for imine substrate derived from propanal). Attempts to improve the reaction yield by switching the chiral diene ligand, as well as the additive, turned out to be fruitless. Thus a more reactive catalyst [Rh(OH)(L)]$_2$ and a neutral reaction condition were applied. This replacement significantly improved the arylation of *N*-tosylimine of propanal, providing the desired products in almost quantitative yields (99%), along with excellent enantioselectivity (>99% ee).[42] Under the optimal conditions, various aliphatic *N*-tosyl imines were treated with 2 equivalents of aryl boronic acids, rhodium–diene complex [Rh(OH)(L)]$_2$ (3 mol% Rh) and 4 Å molecular sieves in dioxane, affording the corresponding chiral sulfonamides **45** in high yields (87–99%) and with excellent enantioselectivities (91–>99% ee) (Scheme 20). Similar results were also achieved for *N*-nosylimine substrates.
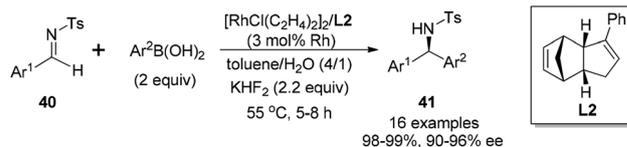
It is noteworthy that the imines with terminal functional groups such as Ph, BnO, and Cl were also suitable substrates, giving the desired adducts in 87–99% yields and >99% ee. The products with terminal Cl substitution could be further
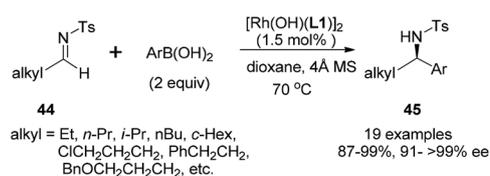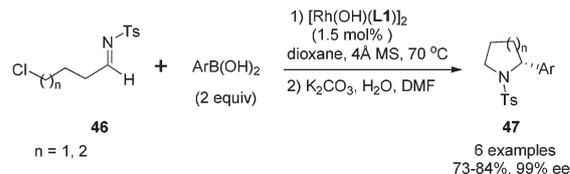
Scheme 21 One-pot synthesis of chiral 2-aryl pyrrolidine and piperidine derivatives.

Scheme 22 Catalytic asymmetric arylation of aryl *N*-tosyl imines with arylboronic acids.
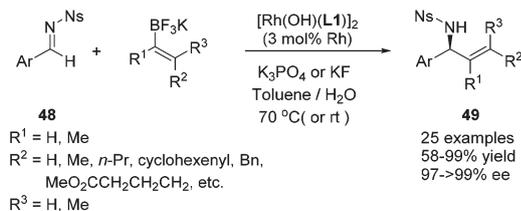
cyclized upon treatment with potassium carbonate to form chiral 2-aryl pyrrolidine and piperidine derivatives **47** in a one-pot fashion (Scheme 21), and the *N*-tosyl group in the cyclized products could be cleaved through the reduction with naphthalene/Li, affording useful optically pure 2-aryl pyrrolidine and piperidine building blocks.

The *C*$_1$-symmetric chiral dicyclopentadiene-based ligands were also applied in the rhodium-catalyzed arylation of aryl *N*-tosyl imines (Scheme 22).[43] In this case, the additive of aqueous KHF$_2$ provided higher yields and better enantioselectivities than the use of Et$_3$N as a base. Compared with the arylation using bicyclo[3.3.0]octadiene ligands, this new series led to the corresponding diarylmethyltosylamides **41** in higher yields (98–99%), but with slightly lower enantioselectivities (90–96% ee).

In addition to these extensive studies on the arylation of imines, we also investigated the rhodium catalyzed alkenylation of arylimines. Due to the relatively lower stability of alkenylboronates and the possible competitive binding between chiral diene and the alkenylation reagent/product, this reaction would be much more challenging. Lam and co-workers reported the alkenylation of stable and relatively reactive cyclic imines.[44] The only single successful example of acyclic imines was provided by Shintani, Hayashi and co-workers during their research on the arylation of imines.[45] Our studies provide a more general approach toward this end. A variety of potassium alkenyltrifluoroborates were successfully added to various *N*-tosyl arylimines, providing the corresponding products in high yields and excellent enantioselectives (Scheme 23). The power of this method was demonstrated by the short formal synthesis of the natural product (−)-aurantioclavine (Scheme 24).[46]

3.2.   Rhodium–diene catalyzed asymmetric conjugate additions

In addition to the rhodium-catalyzed asymmetric arylation to aliphatic and aromatic imines, the chiral diene ligands also

Scheme 20 Catalytic asymmetric arylation of aliphatic *N*-tosyl imines with arylboronic acids.

80 | *Org. Chem. Front.*, 2015, **2**, 73–89

This journal is © the Partner Organisations 2015

Scheme 23 Catalytic asymmetric alkenylation of aryl *N*-nosyl imines with potassium alkenyltrifluoroborates.
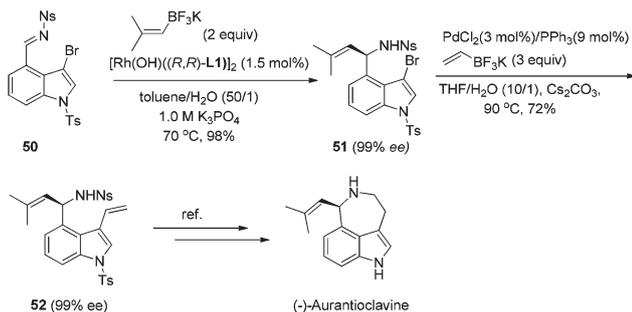
Scheme 24 Formal synthesis of (−)-aurantioclavine.

demonstrated excellent activity and stereoselectivity in the rhodium-catalyzed asymmetric conjugate additions to α,β-unsaturated ketones/esters, nitroalkenes, and α,β-unsaturated γ-lactams.

As shown in Scheme 25, a series of aryl boronic acids could react with 2-cyclopentenone (X = CH₂, n = 1), 2-cyclohexenone (X = CH₂, n = 2), and cyclic α,β-unsaturated esters (X = O, n = 1, 2). The reactions proceeded under very mild conditions with 2 equivalents of boronic acid, 2.5 mol% of $[RhCl(C_2H_4)_2]_2$ and 5.5 mol% of chiral diene ligand at room temperature for 3 h, providing the corresponding chiral adducts **54** in excellent yields and enantioselectivities.[47] In this case, the electron-donating or electron-withdrawing substitution of the phenyl boronic acids did not significantly affect the reaction yields and stereoselectivities. Interestingly, sterically more hindered 1-naphthylboronic acid and 2-tolylboronic acid led to slightly higher enantioselectivities (98% ee) for this conjugate addition.

The above bicyclo[3.3.0]octadiene ligand has two phenyl rings directly connected to two double bonds. It would be interesting to explore other ligands **L3–5** by moving the bulky substitutions from the double bonds to the adjacent carbon atoms. Due to multiple oxygen atom substitutions, such diene ligands are shown to be quite soluble in water (Fig. 5).
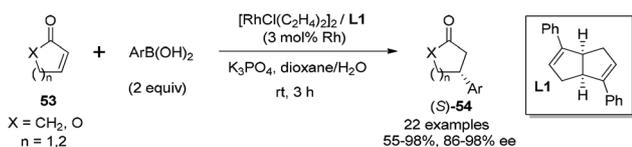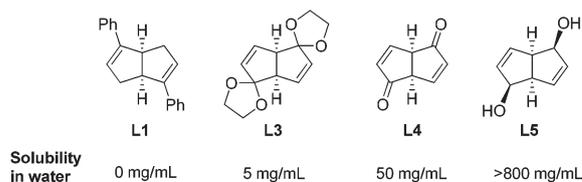
Fig. 5 Solubility of chiral diene ligands in water.

We tried rhodium-catalyzed asymmetric conjugate addition of cyclohexenone and phenyl boronic acid using the above ligands in aqueous media. The reactions were conducted with 2 equivalents of boronic acid, 2.5 mol% of $[RhCl(C_2H_4)_2]_2$, 5.5 mol% of chiral diene ligand at room temperature. To our delight, all three ligands **L3–5** could lead to the corresponding adduct within 0.5 h, and the diketal ligand **L3** was found to be optimal, providing high yield (96%) and excellent enantioselectivity (93% ee).[48] Encouraged by this, we explored such addition reaction with a variety of boronic acids (Scheme 26). Different from the results in Scheme 25, the 2-cyclohexenone (X = CH₂, n = 2) substrate afforded higher enantioselectivities than the 2-cyclopentenone (X = CH₂, n = 1) substrate, providing the corresponding adducts **54** with 93–95% ee and 80–86% ee, respectively. In addition, the sterically more hindered 1-naphthylboronic acid and 2-methoxyphenylboronic acid resulted in significant loss in terms of stereoselectivities (52% ee and 69% ee, respectively). The reactions with cyclic α,β-unsaturated ester (X = O, n = 2) also led to a modest decrease in enantioselectivities (80–82% ee).

In contrast to the ineffectiveness of diphenyl diene ligand **L1** for the linear α,β-unsaturated substrates, this hydrophilic diketal diene ligand **L3** could promote such asymmetric conjugate additions to linear α,β-unsaturated ketones (Scheme 27). This transformation was also very sensitive to the steric factor of boronic acids, but in the opposite direction. Sterically encumbered arylboronic acids, such as 2-chlorophenylboronic, 2-tolylboronic and 1-naphthylboronic acids, proved to be favor-
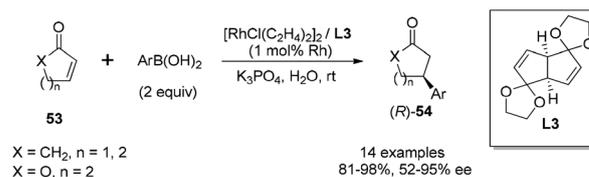
Scheme 26 Catalytic asymmetric conjugate addition with cyclic α,β-unsaturated ketones/esters.

Scheme 25 Catalytic asymmetric conjugate addition with cyclic α,β-unsaturated ketones/esters.
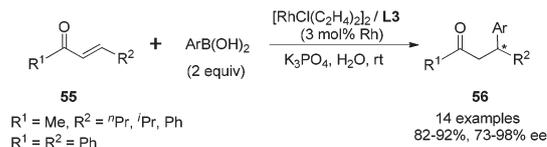
Scheme 27 Catalytic asymmetric conjugate addition with linear α,β-unsaturated ketones.
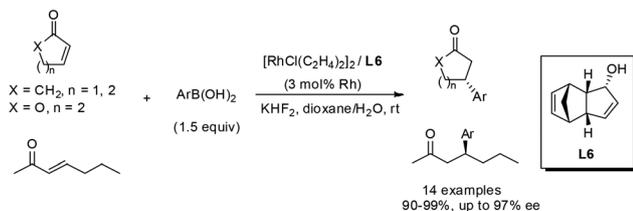
Scheme 28 Catalytic asymmetric conjugate addition with α,β-unsaturated ketones/esters.

able for higher enantioselectivities. To the best of our knowledge, this serves as the first example of an asymmetric reaction employing a chiral metal–diene catalyst in water.
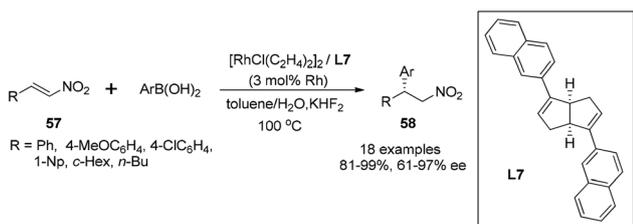
The chiral dicyclopentadiene ligands could also induce excellent enantioselectivities for the rhodium-catalyzed asymmetric conjugate addition.[49] As shown in Scheme 28, both cyclic and linear substrates could undergo this asymmetric addition, affording the corresponding adducts. In general, the 2-cyclohexenone (X = CH$_2$, n = 2) substrate afforded higher enantioselectivities (usually >96% ee) than 2-cyclopentenone and a linear α,β-unsaturated ketone. The electronic properties of the substitution groups at the phenyl ring of arylboronic acids did not affect the yields and enantioselectivities. However, the sterically hindered boronic acids, such as 2-methoxyphenylboronic and 1-naphthylboronic acid, led to a modest decrease in stereoselectivity, giving the corresponding adducts with 87% ee and 85% ee, respectively. As for the linear enone substrates, 2-methylphenylboronic acid afforded much higher enantiomeric excess (87% ee) than phenylboronic acid did (16% ee) although both reactions were accomplished in high yields (90–92%).

The rhodium-catalyzed asymmetric conjugate addition of boronic acids to nitroalkenes was also demonstrated with the chiral bicyclo[3.3.0] diene ligands.[50] In this case, the addition of 3 equivalents of KHF$_2$ proved to be crucial for high yields and enantioselectivities, and the reaction was performed in toluene–water at 100 °C (Scheme 29). As for the aromatic substituted nitroalkenes, the sterically hindered aryl boronic acids, such as 1-naphthylboronic and 2-tolylboronic acid, tended to afford higher enantioselectivities, giving the corresponding adducts **58** with 95% ee and 97% ee, respectively. The electron-withdrawing group at the phenyl ring of boronic acids could enhance the selectivity. In contrast, the electronic properties of the substitution at the phenyl ring of nitroalkene substrates **57** did not significantly affect the reaction selec-

tivities. Interestingly, the aliphatic nitroalkene (R = cyclohexyl) and nitrostyrene (R = Ph) could give similar enantioselectivities (78% ee and 84% ee, respectively) when reacting with 4-anisyl-boronic acid. The reaction with linear 1-nitrohexene (R = n-Bu) led to lower enantioselectivity (61% ee), but the selectivity could get enhanced to 86% ee when 4-methoxyphenylboronic acid was replaced by sterically encumbered 2-tolylboronic acid. It is worth highlighting that our results represent the best enantioselectivity in the asymmetric addition of arylboronic acids to α-unsubstituted nitroalkene substrates.
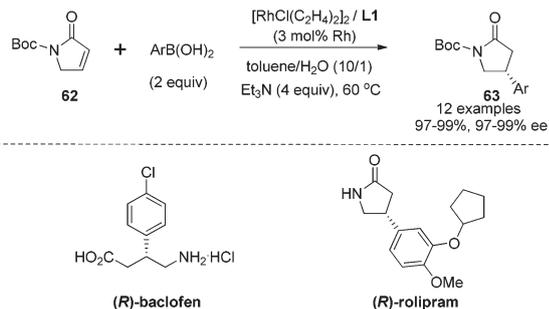
The synthetic utility of this highly selective asymmetric addition was exemplified by the following preparation of isoquinoline derivative (Scheme 30). The corresponding chiral adduct **59** from 4-fluorophenyl boronic acid could be conveniently converted, through the Bischler–Napieralski cyclization, to obtain optically active (S)-4-(4-fluorophenyl)-6,7-dimethoxy-3,4-dihydroquinoline **61** in three steps.

Having established the rhodium-catalyzed asymmetric conjugate addition for α,β-unsaturated ketones/esters and nitroalkenes, we turned our focus to investigate α,β-unsaturated γ-lactam substrates.[51] The studies showed that the N-Boc protection was crucial for this conjugate addition because protection with either benzyl or the 4-methoxyphenyl group led to inseparable alkene by-products, which were generated through Mizoroki–Heck type coupling with aryl boronic acids. As shown in Scheme 31, a series of arylboronic acids successfully reacted with the N-Boc protected α,β-unsaturated γ-lactam **62** under optimal conditions, providing the corresponding adducts **63** in high yields and with excellent enantioselectivities (97–99% ee). The steric properties of arylboronic acids had little effect on the performance of the addition. However, electron-withdrawing groups (4-CF$_3$, 4-Cl, 4-Br) at the phenyl ring of boronic acids led to a significant drop in the reaction yields in spite of excellent enantioselectivities. Fortunately, this issue was solved by switching the additive from Et$_3$N to KHF$_2$, which could dramatically shorten the reaction time and thus improved the yields from 56–78% to >97%. This transformation provided a facile access of chiral γ-lactams, which could be applied in the synthesis of the GABAB receptor agonist (R)-baclofen and the antidepressant (R)-rolipram.

Compared with arylboron reagents, the use of alkenylboron reagents is much less examined for the rhodium-catalyzed asymmetric conjugate addition.[52] Although many groups
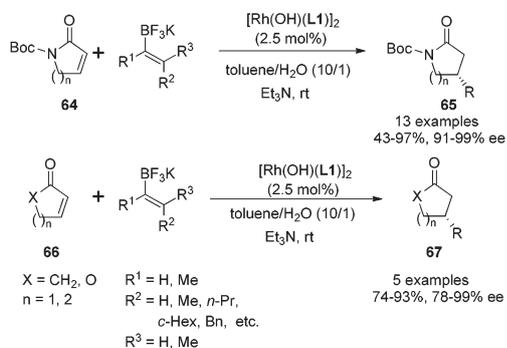

Scheme 29 Catalytic asymmetric conjugate addition with nitroalkenes.
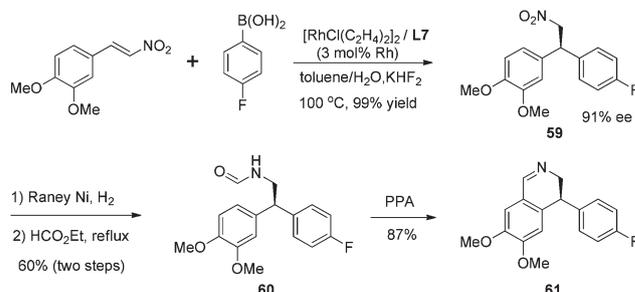
Scheme 30 Synthesis of isoquinoline derivative.

82 | *Org. Chem. Front.*, 2015, **2**, 73–89

This journal is © the Partner Organisations 2015

Scheme 31 Catalytic asymmetric conjugate addition with α,β-unsaturated γ-lactam.

reported the addition of alkenylboron reagents to simple α,β-unsaturated ketones,[53] their addition to α,β-unsaturated amides remains unexplored. Encouraged by the performance of our diene ligands in the conjugate addition to *N*-Boc protected α,β-unsaturated γ-lactam, we turned our focus to extend arylboron reagents to alkenylboron reagents. Due to the low stability of alkenylboronic acids, we chose to apply potassium alkenyltrifluoroborates in the rhodium-catalyzed conjugate addition (Scheme 32).[54] For the γ-lactam substrate **64** (*n* = 1), the reactions with a variety of alkenyltrifluoroborates all afforded excellent enantioselectivities (97–99% ee), but the reaction yields were dependent on both steric and electronic properties of alkenyltrifluoroborates. The reactions with more substituted alkenyltrifluoroborates or electron-rich alkenyltrifluoroborates generally led to higher yields. For the δ-lactam substrate **64** (*n* = 2), the enantioselectivities were slightly lower (91–94% ee) while the yields remained high. In addition to the reactions with α,β-unsaturated lactams, alkenyltrifluoroborates could react with cyclic α,β-unsaturated ketones and esters **66**. In the case of the cyclic α,β-unsaturated esters (lactones), the combination of dioxane and KF turned out to be better than toluene and Et₃N system.

   The synthetic application of this asymmetric transformation was demonstrated by the total synthesis of (−)-α-kainic acid, an alkaloid natural product first isolated in 1953 from the Japanese marine algae *Digenea simplex*.[55] As shown in Scheme 33, the reaction of γ-lactam with potassium 1-methyl-

**Scheme 32** Catalytic asymmetric conjugate addition with alkenyltrifluoroborates.
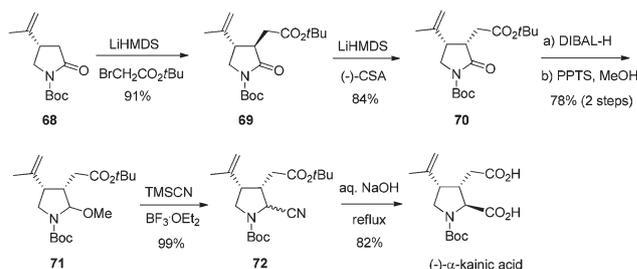
Scheme 33 Synthesis of (−)-α-kainic acid.

vinyltrifluoroborate, by the catalysis of the rhodium-(*R*,*R*)-diene complex, afforded the desired conjugate adduct **68** in 82% yield and with 99% ee. Its subsequent alkylation exclusively gave the *trans*-isomer **69**, which was isomerized to *cis*-isomer **70** through a thermodynamic protonation process. Further manipulations of the amide functional group, through the reaction of an aminal intermediate **71** with TMSCN and a subsequent basic hydrolysis, completed the total synthesis of (−)-α-kainic acid.

3.3. Palladium–diene catalyzed asymmetric Suzuki–Miyaura cross-coupling reactions

Although diene ligands are widely applied in rhodium- and iridium-catalyzed asymmetric reactions, there were very few examples of the use of chiral diene ligands in palladium-catalyzed asymmetric transformations. In 2005, a chiral diene–palladium complex was prepared and tested in enyne cyclization reaction. The desired product was generated, but in a racemic form and with low yield (35%).[56] The application of a palladium–diene complex in asymmetric reactions faces the following two major challenges: first, palladium-catalyzed reactions often require the change of metal valence state in the catalytic cycle, which is different from the rhodium-catalyzed transformations where the metal valence state usually keeps constant. Second, the diene ligands probably possess weaker binding affinity than phosphorus ligands to stabilize palladium in different oxidation states. With significant efforts in further expanding the synthetic application of our diene ligands, we found that bicyclo[3.3.0]octadiene ligands could form air-stable complexes with Pd(II) by treatment with PdCl₂(PhCN)₂ in benzene at room temperature overnight. From the screening of different ligands for Suzuki–Miyaura cross-coupling, the 3,5-dimethylphenyl group substituted bicyclo[3.3.0]octadiene was found to provide the corresponding axially chiral biaryls in the combination of good yields and enantioselectivities (Scheme 34).[57] Under the optimal conditions, a variety of aryl bromides and trifluoromethanesulfonates could react with aryl boronic acids in the presence of Pd–diene complex **75** (5 mol%) and a free ligand (15 mol%) using 2.5 equivalents of Cs₂CO₃ as a base in toluene. It is noteworthy that some reactions, for example, the coupling between 2-bromo-3-methylbenzaldehyde and 2-methylphenyl boronic acid, could occur at 0 °C to provide the corresponding biaryl in 97% yield and with 65% ee. To the
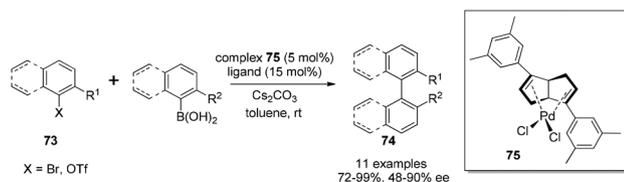
Scheme 34 Catalytic asymmetric Suzuki–Miyaura cross-coupling.

best of our knowledge, this coupling reaction serves as the first successful application of chiral diene ligands in the asymmetric palladium-catalyzed transformation.

X-ray crystallography of two Pd–diene complexes was obtained by coordination with diphenyl- and di(3,5-dimethylphenyl)-substituted dienes. A structural analysis confirms that our *cis*-fused bicyclic diene skeleton creates a wedge shape, providing an excellent chiral environment to coordinate with PdCl$_2$, which is quite similar to the Rh–diene complexes.[41] A detailed comparison of the bond lengths and angles between free dienes and the corresponding Pd-complexes in the X-ray crystallography was conducted through a Dewar–Chatt–Duncanson model. The length of double bond in the complexes (1.37–1.38 Å) is longer than that in the corresponding free ligand (1.32 Å), indicating an sp^3 characteristic after binding with PdCl$_2$. In addition, the dihedral angle of the bicyclic diene unit shrinks (88–91° *versus* 107°) and creates a bite angle (85–86°) after binding with palladium. The dihedral angle in di(3,5-dimethylphenyl)-substituted diene complex (87.6°) is slightly smaller than that in the diphenyl-substituted diene complex (90.8°), indicating that the di(3,5-dimethylphenyl)-substituted diene results in stronger chelating affinity with PdCl$_2$. The mechanism of this palladium–diene catalyzed asymmetric Suzuki–Miyaura cross-coupling was investigated using electrospray ionization mass spectrometry (ESI-MS/MS).[58] In all detected palladium species, the diene ligand was firmly associated with palladium, explaining the high enantioselectivities for this asymmetric transformation.

4. Development of chiral sulfur–olefin ligands for asymmetric reactions

The allylation of chiral *N-tert*-butanesulfinyl imines could generate a variety of enantiopure *N-tert*-butanesulfinyl homoallylic amines bearing different substitution patterns at the *C*-1 and *C*-2 positions. The chiral products from this asymmetric allylation actually could serve as attractive olefin–sulfur hybrid ligands.[59] As shown in Scheme 35, Xu observed that a simple chiral *N-tert*-butanesulfinyl homoallylic amine **L8** led to an excellent conversion for the rhodium-catalyzed asymmetric arylation, affording the desired conjugate adduct in 99% yield and with 82% ee. The modification of either of the functional groups involved in the ligand binding with rhodium, *i.e.*, the reduction of olefin or elimination of sulfur chirality, resulted in detrimental effects on this asymmetric transformation. The former (compound **L9**) did not generate any desired product,
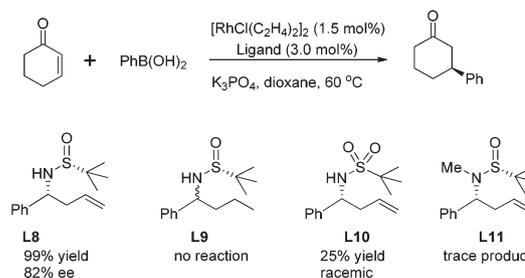
Scheme 35 Proof-of-concept studies on potential *N-tert*-butanesulfinyl homoallylic amine as ligand.

while the latter (compound **L10**) afforded the desired product in a racemic form and with much lower yield (25%). This observation indicates that the sulfinamide and allylic double bond are both involved in the coordination with rhodium during the catalytic process. In addition, the *N*-methyl analog (compound **L11**) resulted in very poor conversion, probably due to the unfavorable coordination geometry. This study represents one of the first examples using chiral sulfur–olefin ligands for transition metal-catalyzed asymmetric transformation.[60]

With the encouraging results, we evaluated a series of *N-tert*-butanesulfinyl homoallylic amines accumulated over the last decade as chiral ligands for this rhodium-catalyzed conjugate addition (Fig. 6). Except the ligand **L19**, which gave quite a poor yield (21%) and low enantioselectivity (−26% ee), other ligands promoted the asymmetric arylation in almost quantitative yields (99%) and with excellent enantioselectivities (82–95% ee). Interestingly, both diastereomers **L8** and **L12**, regardless of the absolute configuration of the carbon chiral center, afforded the addition product displaying the same chirality with similar enantioselectivities (82% and 85% ee, respectively). The benzoates **L13–18**, which were generated by the asymmetric benzoyloxyallylation, all promoted the conjugate addition in high yields (99%) and with excellent enantioselectivities (94–95% ee). It is noteworthy that both the *anti*-isomers **L17** and **L18** led to the same chiral adduct in almost quantitative yields (99%) and with outstanding enantioselectivities (95% and 94% ee, respectively). A separate experiment with a mixture of two *anti*-isomers as ligands also
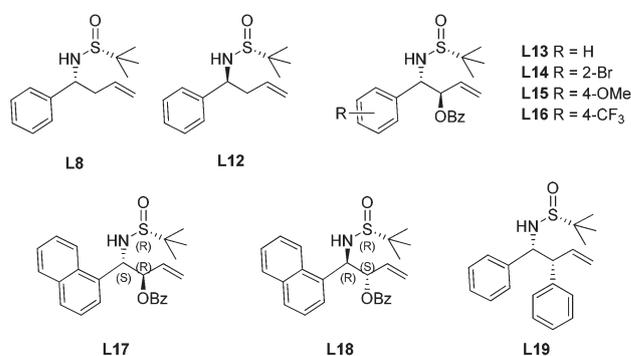

Fig. 6 Sulfinamide–olefin ligands.

demonstrated a highly enantioselective formation of the addition product in 99% yield and with 95% ee. In other words, the enantioselectivity of this conjugate arylation was mainly controlled by the stereochemistry of the chiral *N*-sulfinyl group. This observation provides great advantage for ligands **L17** and **L18** in the rhodium-catalyzed asymmetric arylation as it can simply apply the products from zinc-mediated benzoyloxyallylation of the corresponding chiral *N*-tert-butanesulfinyl imine regardless of its stereoselectivity outcome.

With the simple and readily available chiral ligand **L18**, different substrates for the rhodium-catalyzed asymmetric arylation were examined. A variety of α,β-unsaturated cyclic carbonyl compounds, including 2-cyclohexenone, 2-cyclopentenone, α,β-unsaturated δ-lactone and lactam, could undergo the 1,4-addition, affording the corresponding products in high yields (74–99%) and with excellent enantioselectivities (78–98% ee) (Scheme 36). The steric and electronic properties of aryl boronic acids appeared to have little influence on the reaction. Unfortunately, the reaction of linear enone (*E*)-2-heptenone with phenyl boronic acid gave the desired product in low yield (51%) and stereoselectivity (38% ee). Our continuous work in this field revealed that a simpler ligand **L21** also showed great catalytic activity and enantioselectivity for the conjugate addition to α,β-unsaturated cyclic carbonyl compounds.[61]

With the above significant success in the rhodium-catalyzed asymmetric 1,4-addition using chiral sulfur–olefin ligands, the exploration of this new ligand class and applying them in the rhodium-catalyzed enantioselective 1,2-addition of arylboronic acids to α-ketoesters and α-diketones was expanded.[62] In the initial screening with methyl phenylglyoxylate and *p*-tolylboronic acid as the reaction substrates (Scheme 37), it was found that a simple sulfur–olefin ligand **L21** could lead to 1,2-

addition in high yield (90%) and with excellent enantioselectivity (85% ee). It is noteworthy that the chiral diene ligand (**L1**) and diphosphine ligand (BINAP) afforded the desired adduct with very low stereoselectivities (9% ee and 13% ee, respectively), in spite of excellent chemical yields (90% and 79%, respectively).

The ester group in phenylglyoxylate showed some impact on the 1,2-addition in terms of chemical yield and enantioselectivity. Aromatic groups, such as phenyl, 4-chlorophenyl, 1-naphthyl and 2-naphthyl esters, afforded better enantioselectivities (92–95% ee), albeit with low yields (33–65%), due to basic hydrolysis when 0.5 equivalent of 1.5 M K_3PO_4 was applied at 40 °C. Further optimization of the base and reaction temperature led to the use of 2-naphthyl phenylglyoxylate as the best substrate, and the reaction could be accomplished at room temperature when 0.08 equivalent of 0.1 M KOH was applied, producing the desired tertiary α-hydroxyester in good yield (86%) and with excellent ee (94%).

With the identified optimal ester group and reaction conditions, the scope of different arylglyoxylates **78** and aryl boronic acids was investigated. As shown in Scheme 38, the 1,2-addition reactions consistently generated the corresponding adducts **79** in moderate to good yields (54–89%) and with excellent enantioselectivities (90–95% ee). It is noteworthy that both enantiomers could be obtained with high ee values by simply switching the corresponding Ar^1 and Ar^2 groups of the α-ketoesters and boronic acids.

This method could be extended to heteroaryl α-ketoesters.[60b] A variety of α-ketoesters including 3-indoleglyoxylates, 3-benzofuranglyoxylates and 3-benzothiopheneglyoxylates could react with arylboronic acids, affording corresponding quaternary carbon-containing heteroaromatic α-hydroxy esters in moderate to good yields with high enantiomeric excesses (up to 97%). As for the 1,2-addition to aliphatic α-ketoesters, moderate yields and enantioselectivities were also achieved despite the possible formation of enolate esters under basic reaction conditions.[60c]
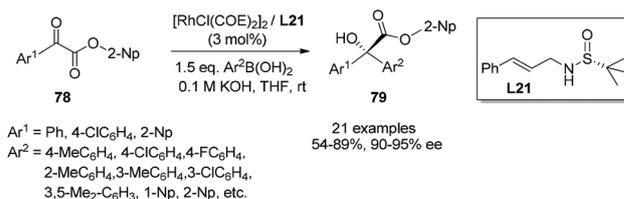
This enantioselective 1,2-addition could also be applied to α-diketone substrates **80**.[60a] As shown in Scheme 39, the enantioselectivities for di-aryl α-diketones were generally quite high (95–99% ee), while the less hindered aliphatic diketone, biacetyl ($R^1 = R^2 = Me$), gave the desired adduct with only 63% ee. For the asymmetric α-diketone ($R^1 = Me$, $R^2 = Ph$), the 1,2-addition predominantly took place at the less hindered carbonyl group (8.5 : 1 regioselectivity), affording the major adduct
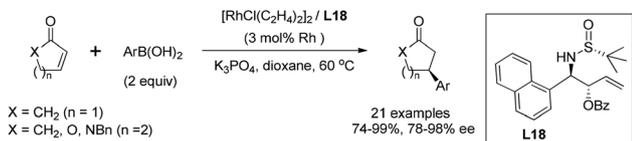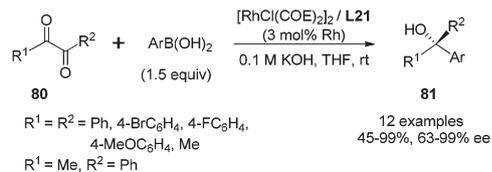

Scheme 36 Asymmetric arylation of α,β-unsaturated cyclic carbonyl compounds catalyzed sulfinamide–olefin ligand.
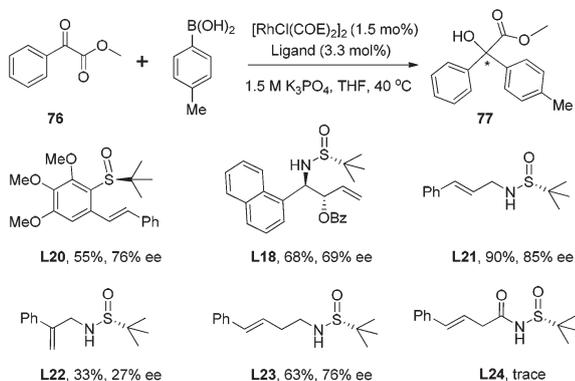
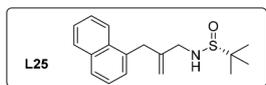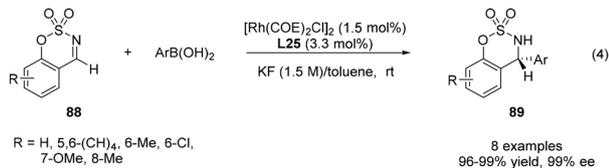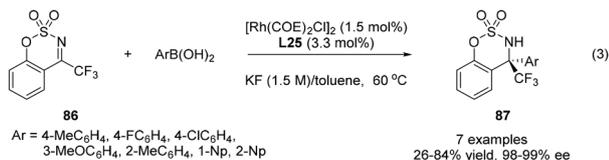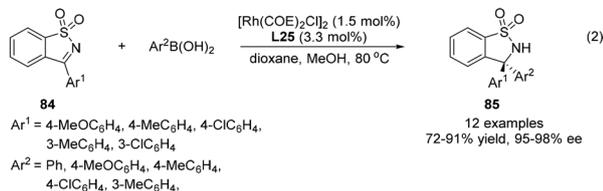**Scheme 37** Ligand screening (COE = cyclooctene, Bz = benzoyl).

Scheme 38 Catalytic asymmetric 1,2-addition of arylboric acids to α-ketoesters.

This journal is © the Partner Organisations 2015

Org. Chem. Front., 2015, **2**, 73–89 | 85

Scheme 39 Catalytic asymmetric 1,2-addition of arylboric acids to α-diketones.

with 80% ee. Several bicyclic heterocycles with quaternary carbon-containing chiral center were successfully obtained through short or one-pot chemical transformations from such 1,2-additions of properly designed α-ketoesters and α-diketones.[63]

Inspired by the success of enantioselective addition of aryl-boron reagents to C=C and C=O double bonds using simple chiral sulfur–olefin ligands, the attention was turned to their applications in the asymmetric 1,2-addition to C=N double bonds to construct pharmacologically attractive chiral amines. Initially, we investigated the rhodium-catalyzed addition of arylboronic acids to cyclic N-sulfonyl α-iminoesters **82**. Simple sulfur–olefins could successfully promote this arylation reaction. Under the optimized catalytic system, the reaction between a wide variety of cyclic N-sulfonyl α-iminoesters and arylboronic acids with diverse steric and electronic properties proceeded well, giving the corresponding addition products **83**

mostly in high yields (up to 96%) and with excellent enantioselectivities (85–99% ee) (Scheme 40).[64] Furthermore, the asymmetric arylation of a serial of other cyclic N-sulfonyl imines (**84**, **86**, **88**) were also subjected to this reaction. Although the reaction yields varied for different substrates, excellent enantioselectivities were observed in all examined cases. Notably, the benzo-fused six-membered cyclic imine **86** yielded particularly interesting CF_3-containing cyclic sulfamidates.

The application of chiral ligand **L25** in the addition of arylboronic acids to 3,4-disubstituted thiadiazoles was also developed, providing direct access to chiral quaternary carbon-containing 1,2,5-thiadiazoline 1,1-dioxide derivatives (Scheme 41). The reactions proceeded under exceptionally mild conditions, giving the corresponding adducts in high yields and with excellent enantioselectivities.[65]

Encouraged by the above results from chiral sulfinamide–olefin ligands, we designed and synthesized a new set of compounds with the sulfinyl group serving as the only chiral center to control the stereoselectivity.[66] In this case, the chiral sulfinyl group and olefin were connected by a benzene ring in a 1,2-fashion (Fig. 7). These new ligands were evaluated in the rhodium-catalyzed 1,4-addition of phenylboronic acid to 2-cyclohexenone.

The ligand with terminal olefin (**L26**, R = H) could promote the addition reaction in high yields (98%), albeit with quite

Scheme 40 Catalytic asymmetric 1,2-addition of arylboric acids to cyclic N-sulfonyl imines.
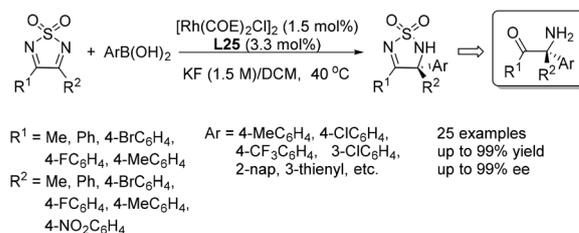
Scheme 41 Asymmetric arylation of 3,4-disubstituted thiadiazoles catalyzed by sulfoxide–olefin ligand.

Fig. 7 Sulfoxide–olefin ligands.

86 | Org. Chem. Front., 2015, **2**, 73–89

This journal is © the Partner Organisations 2015
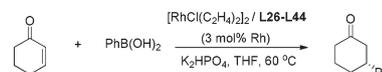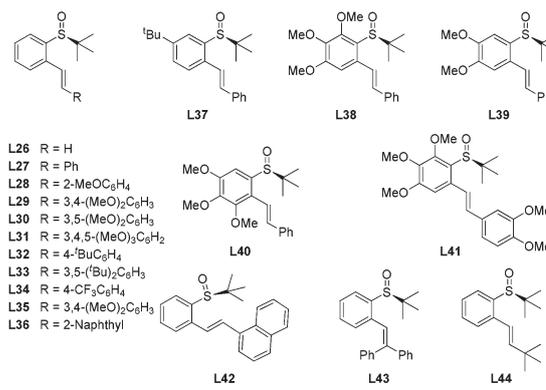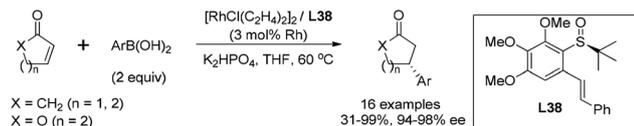
Scheme 42 Asymmetric arylation of α,β-unsaturated cyclic carbonyl compounds catalyzed sulfoxide–olefin ligand.

low enantioselectivity (8% ee). To our delight, most of the ligands (**L27–36**) bearing R substituent displayed dramatic improvement in terms of enantioselectivities (70–95% ee). The bulky terminal group 1-naphthyl (**L42**) led to the desired product in only 4% yield and with 48% ee. The ligands with diphenyl (**L43**) and *t*-butyl groups (**L44**) at the olefin terminal only generated a negligible amount of product, suggesting that the bulky R groups would disfavor the coordination with rhodium. An evaluation of the different substitutions at the central benzene ring revealed that the introduction of multiple methoxy groups was beneficial for the reaction yield and enantioselectivity. The ligand **L38**, with a 2,3,4-trimethoxybenzene ring connecting the sulfinyl and olefin moieties, could afford the desired adduct in 99% yield and with the highest enantioselectivity (95% ee). Further optimization of the base and the reaction solvent for ligand **L38** led to the enantioselectivity of 97% ee along with 99% yield when aqueous K_2HPO_4 and THF were used in this rhodium-catalyzed asymmetric arylation.

With the best ligand **L38** and under the optimal reaction conditions, we examined the substrate scope of this rhodium-catalyzed asymmetric conjugate arylation (Scheme 42). The reactions performed well for α,β-unsaturated cyclic carbonyl compounds, such as 2-cyclohexenone, 2-cyclopentenone, α,β-unsaturated δ-lactone, giving the corresponding adducts with excellent enantioselectivities (94–98% ee), regardless of the electronic and steric properties of arylboronic acids. Unfortunately, the reaction of linear enone (*E*)-2-heptenone with phenyl boronic acid gave the desired product in a low yield (34%) and stereoselectivity (21% ee). The research results revealed that well-designed innovative *S*-stereogenic olefin ligands can be as useful as other classical chiral ligands in asymmetric transformations.

Inspired by the success of the above initial research on chiral sulfur–olefin ligands, further explorations aimed at expanding the applications of these new novel ligands have been carried out by other groups and us, which were described in detail in a recent review.[4]

5. Conclusion and outlook

The development of asymmetric synthesis with a highly enantioselective control continues to be one of the main topics in organic synthesis. Synthetic methods based on chiral auxiliaries provide reliable approaches to access chiral compounds. Using *N-tert*-butansulfinamide as a chiral auxiliary, we successfully developed three different kinds of stereoselective reac-

tions to prepare enantiorich amines. SmI_2-mediated homo-coupling and cross-coupling of *N-tert*-butansulfinyl imines with nitrones and aldehydes provide a simple and efficient way to synthesize enantiorich vicinal diamines and α-amino alcohols. As for the Zn/In-mediated allylation of *N-tert*-butansulfinyl imines, the stereoselectivity reversal can be controlled by switching the reaction conditions. More complicated chiral compounds can be synthesized by using suitable multifunctional allyl bromide reagents. Highly diastereoselective intermolecular or intramolecular aza-Friedel–Crafts reaction have also been realized in the presence of proper Lewis acid catalysts.

In the meantime, highly enantioselective asymmetric transition-metal-catalyzed reactions were also pursued using structurally novel chiral dienes as ligands. Two families of chiral diene ligands bearing bicyclo[3.3.0]octadiene or dicyclopentadiene skeleton were successfully introduced. A variety of highly enantioselective rhodium-catalyzed reactions, including 1,2-addition to imines and 1,4-conjugate additions to several electron-deficient double bonds were performed. The first successful application of chiral diene ligands in palladium-catalyzed cross-coupling reactions was also realized.

Inspired by our success in these two parts, we developed conceptually new chiral sulfur–olefin ligands. These easily accessible new ligands proved to be competent for a variety of rhodium-catalyzed reactions, especially in the addition to specific C=O and C=N double bonds, allowing for highly enantioselective construction of otherwise-difficult-to-access fully-substituted carbon stereocenters.

Despite many successful examples of chiral *N-tert*-butanesulfinamide auxiliary and chiral diene and sulfur–olefin ligands, there are still many challenges we have to face. The application of both chiral auxiliary and ligands requires further expansion to cover more useful and challenging substrates. A more comprehensive mechanistic study on the catalytic role of ligands, as well as the development of new effective ligands, is still underway. Nevertheless, we believe that the power of these asymmetric reactions will be further demonstrated in organic synthesis and drug development.

Acknowledgements

Notes and references

1 (*a*) R. Noyori, *Asymmetric Catalysis in Organic Synthesis*, Wiley, New York, 2nd edn, 1994; (*b*) E. N. Jacobsen,

This journal is © the Partner Organisations 2015

Org. Chem. Front., 2015, **2**, 73–89 | 87

A. Pfaltz and H. Yamamoto, *Comprehensive Asymmetric Catalysis*, Springer, Berlin, 1999, vols. I–III; (*c*) I. Ojima, *Catalytic Asymmetric Synthesis*, John Wiley & Sons, Hoboken/NJ, 3rd edn, 2010.

2 G.-Q. Lin, M.-H. Xu, Y.-W. Zhong and X.-W. Sun, *Acc. Chem. Res.*, 2008, **41**, 831.

3 C.-G. Feng, M.-H. Xu and G.-Q. Lin, *Synlett*, 2011, 1345.

4 Y. Li and M.-H. Xu, *Chem. Commun.*, 2014, **50**, 3771.

5 (*a*) J. A. Ellman, T. D. Owens and T. P. Tang, *Acc. Chem. Res.*, 2002, **35**, 984; (*b*) J. A. Ellman, *Pure Appl. Chem.*, 2003, **75**, 39; (*c*) C. H. Senanayake, D. Krishnamurthy, Z. H. Lu, Z. X. Han and I. Gallou, *Aldrichimica Acta*, 2005, **38**, 93; (*d*) D. Morton and R. A. Stockman, *Tetrahedron*, 2006, **62**, 8869.

6 (*a*) M.-H. Xu, W. Wang and G.-Q. Lin, *Org. Lett.*, 2000, **2**, 2229; (*b*) W. Wang, M.-H. Xu, X.-S. Lei and G.-Q. Lin, *Org. Lett.*, 2000, **2**, 3773; (*c*) M.-H. Xu, W. Wang, L.-J. Xia and G.-Q. Lin, *J. Org. Chem.*, 2001, **66**, 3953.

7 H. B. Kagan, *Tetrahedron*, 2003, **59**, 10351.

8 Y. W. Zhong, K. Izumi, M. H. Xu and G. Q. Lin, *Org. Lett.*, 2004, **6**, 4747.

9 Y.-W. Zhong, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2004, **6**, 3953.

10 Y.-W. Zhong, Y.-Z. Dong, K. Fang, K. Izumi, M.-H. Xu and G.-Q. Lin, *J. Am. Chem. Soc.*, 2005, **127**, 11956.

11 (*a*) Y.-W. Zhong, Y.-Z. Dong, K. Fang, K. Izumi, M.-H. Xu and G.-Q. Lin, *J. Am. Chem. Soc.*, 2005, **127**, 11956; (*b*) R.-H. Liu, K. Fang, B. Wang, M.-H. Xu and G.-Q. Lin, *J. Org. Chem.*, 2008, **73**, 3307; (*c*) R. Wang, K. Fang, B.-F. Sun, M.-H. Xu and G.-Q. Lin, *Synlett*, 2009, 2301.

12 M. Lombardo and C. Trombini, *Chem. Rev.*, 2007, **107**, 3843.

13 (*a*) *Amino Acids, Peptides, and Proteins in Organic Chemistry*, ed. A. B. Hughes, Wiley-VCH, Weinheim, 2009, vol. 1 and 2; (*b*) G. C. Barrett and D. T. Elmore, *Amino Acids and Peptides*, Cambridge University Press, Cambridge, 1998; (*c*) F. P. J. T. Rutjes, L. B. Wolf and H. E. Schoemaker, *J. Chem. Soc., Perkin Trans. 1*, 2000, 4197.

14 D.-M. Ji and M.-H. Xu, *Chem. Commun.*, 2010, **46**, 1550.

15 Y. Li, D.-M. Ji and M.-H. Xu, *Org. Biomol. Chem.*, 2011, **9**, 8452.

16 (*a*) S. Ni, Y. Yuan, J. Huang, X. Mao, M. Lv, J. Zhu, X. Shen, J. Pei, L. Lai, H. Jiang and J. Li, *J. Med. Chem.*, 2009, **52**, 5295; (*b*) M. Sabbah, L. Soulre, S. Reverchon, Y. Queneau and A. Doutheau, *Bioorg. Med. Chem.*, 2011, **19**, 4868; (*c*) X. Zhang, Y. He, S. Liu, Z. Yu, Z.-X. Jiang, Z. Yang, Y. Dong, S. C. Nabinger, L. Wu, A. M. Gunawan, L. Wang, R. J. Chan and Z.-Y. Zhang, *J. Med. Chem.*, 2010, **53**, 2482.

17 Y. Li and M.-H. Xu, *Asian J. Org. Chem.*, 2013, **2**, 50.

18 D. A. Cogan, G. C. Liu and J. Ellman, *Tetrahedron*, 1999, **55**, 8883.

19 F. Foubelo and M. Yus, *Tetrahedron: Asymmetry*, 2004, **15**, 3823.

20 X.-W. Sun, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2006, **8**, 4979.

21 X.-W. Sun, M. Liu, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2008, **10**, 1259.

22 M. Liu, X.-W. Sun, M.-H. Xu and G.-Q. Lin, *Chem. – Eur. J.*, 2009, **15**, 10217.

23 For reviews on the synthesis of β-amino alcohols: (*a*) D. J. Ager, I. Prakash and D. R. Schaad, *Chem. Rev.*, 1996, **96**, 835; (*b*) C. H. Senanayake, *Aldrichimica Acta*, 1998, **31**, 3; (*c*) S. C. Bergmeier, *Tetrahedron*, 2000, **56**, 2561; (*d*) G. Mlostoń, E. Obijalska and H. Heimgartner, *J. Fluorine Chem.*, 2010, **131**, 829.

24 M. Liu, A. Shen, X.-W. Sun, F. Deng, M.-H. Xu and G.-Q. Lin, *Chem. Commun.*, 2010, **46**, 8460.

25 W.-J. Liu, Y.-H. Zhao and X.-W. Sun, *Tetrahedron Lett.*, 2013, **54**, 3586.

26 (*a*) H. Kakeya, M. Morishita, H. Kobinata, M. Osono and H. Osada, *J. Antibiot.*, 1998, **51**, 1126; (*b*) H. Kakeya, M. Morishita, H. Kobinata, T. Morita, K. Kobayashi and H. Osada, *J. Org. Chem.*, 1999, **64**, 1052.

27 For related study, see: L. R. Reddy, B. Hu, M. Prashad and K. Prasad, *Org. Lett.*, 2008, **10**, 3109.

28 (*a*) A. Shen, M. Liu, Z.-S. Jia, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2010, **12**, 5154; (*b*) R. Wang, P. Tian and G.-Q. Lin, *Chin. J. Chem.*, 2013, **31**, 40; (*c*) Y. Tan, X.-D. Yang, W.-J. Liu and X.-W. Sun, *Tetrahedron Lett.*, 2014, **55**, 6105.

29 A. Shen, Z.-T. He, H.-J. Yu, Y. Fukui, P. Tian and G.-Q. Lin, *Synlett*, 2013, 1649.

30 T. Guo, R. Song, B.-H. Yuan, X.-Y. Chen, X.-W. Sun and G.-Q. Lin, *Chem. Commun.*, 2013, **49**, 5402.

31 D. Chen and M.-H. Xu, *Chem. Commun.*, 2013, **49**, 1327.

32 Z.-Q. Wang, C.-G. Feng, M.-H. Xu and G.-Q. Lin, *J. Am. Chem. Soc.*, 2007, **129**, 5336.

33 S. Helbig, S. Sauer, N. Cramer, S. Laschat, A. Baro and W. Frey, *Adv. Synth. Catal.*, 2007, **349**, 2331.

34 C. Shao, H.-J. Yu, N.-Y. Wu, C.-G. Feng and G.-Q. Lin, *Org. Lett.*, 2010, **12**, 3820.

35 H.-Y. Yang and M.-H. Xu, *Chin. J. Chem.*, 2013, **31**, 119.

36 T. Hayashi, K. Ueyama, N. Tokunaga and K. Yoshida, *J. Am. Chem. Soc.*, 2003, **125**, 11508.

37 C. Fischer, C. Defieber, T. Suzuki and E. M. Carreira, *J. Am. Chem. Soc.*, 2004, **126**, 1628.

38 Reviews of chiral olefin ligands: (*a*) C. Defieber, H. Grutzmacher and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2008, **47**, 4482; (*b*) J. B. Johnson and T. Rovis, *Angew. Chem., Int. Ed.*, 2008, **47**, 840; (*c*) R. Shintani and T. Hayashi, *Aldrichimica Acta*, 2009, **42**, 31; (*d*) X. Q. Feng and H. F. Du, *Asian J. Org. Chem.*, 2012, **1**, 204.

39 Reviews on arylation of imines: (*a*) C. S. Marques and A. J. Burke, *ChemCatChem*, 2011, **3**, 635; (*b*) P. Tian, H.-Q. Dong and G.-Q. Lin, *ACS Catal.*, 2012, **2**, 95.

40 H.-Y. Yang and M.-H. Xu, *Chem. Commun.*, 2010, **46**, 9223.

41 M. Trincado and J. A. Ellman, *Angew. Chem., Int. Ed.*, 2008, **47**, 5623.

42 Z. Cui, H.-J. Yu, R.-F. Yang, W.-Y. Gao, C.-G. Feng and G.-Q. Lin, *J. Am. Chem. Soc.*, 2011, **133**, 12394.

43 C. Shao, H.-J. Yu, N.-Y. Wu, C.-G. Feng and G.-Q. Lin, *Org. Lett.*, 2010, **12**, 3820.

44 Y. F. Luo, A. J. Carnell and H. W. Lam, *Angew. Chem., Int. Ed.*, 2012, **51**, 6762.

45 R. Shintani, M. Takeda, Y.-T. Soh, T. Ito and T. Hayashi, *Org. Lett.*, 2011, **13**, 2977.

46 Z. Cui, Y.-J. Chen, W.-Y. Gao, C.-G. Feng and G.-Q. Lin, *Org. Lett.*, 2014, **16**, 1016.

47 C.-G. Feng, Z.-Q. Wang, P. Tian, M.-H. Xu and G.-Q. Lin, *Chem. – Asian J.*, 2008, **3**, 1511.

48 C.-G. Feng, Z.-Q. Wang, C. Shao, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2008, **10**, 4101.

49 C. Shao, H.-J. Yu, C.-G. Feng, R. Wang and G.-Q. Lin, *Tetrahedron Lett.*, 2012, **53**, 2733.

50 Z.-Q. Wang, C.-G. Feng, S.-S. Zhang, M.-H. Xu and G.-Q. Lin, *Angew. Chem., Int. Ed.*, 2010, **49**, 5780.

51 C. Shao, H.-J. Yu, N.-Y. Wu, P. Tian, R. Wang, C.-G. Feng and G.-Q. Lin, *Org. Lett.*, 2011, **13**, 788.

52 For rhodium-catalyzed asymmetric conjugate addition of alkenyl silicon and zirconium reagents, see: (*a*) S. Oi, Y. Honma and Y. Inoue, *Org. Lett.*, 2002, **4**, 667; (*b*) S. Oi, A. Taira, Y. Honma and Y. Inoue, *Org. Lett.*, 2003, **5**, 97; (*c*) Y. Otomaru and T. Hayashi, *Tetrahedron: Asymmetry*, 2004, **15**, 2647; (*d*) Y. Nakao, J. Chen, H. Imanaka, T. Hiyama, Y. Ichikawa, W.-L. Duan, R. Shintani and T. Hayashi, *J. Am. Chem. Soc.*, 2007, **129**, 9137; (*e*) S. Oi, T. Sato and Y. Inoue, *Tetrahedron Lett.*, 2004, **45**, 5051.

53 For selected examples: (*a*) Y. Takaya, M. Ogasawara and T. Hayashi, *Tetrahedron Lett.*, 1998, **39**, 8479; (*b*) A. Duursma, J. G. Boiteau, L. Lefort, J. A. F. Boogers, A. H. M. de Vries, J. G. de Vries, A. J. Minnaard and B. L. Feringa, *J. Org. Chem.*, 2004, **69**, 8045; (*c*) R. T. Stemmler and C. Bolm, *J. Org. Chem.*, 2005, **70**, 9925; (*d*) G. Lalic and E. J. Corey, *Tetrahedron Lett.*, 2008, **49**, 4894; (*e*) K. Okamoto, T. Hayashi and V. H. Rawal, *Org. Lett.*, 2008, **10**, 4387.

54 H.-J. Yu, C. Shao, Z. Cui, C.-G. Feng and G.-Q. Lin, *Chem. – Eur. J.*, 2012, **18**, 13274.

55 S. Murakami, T. Takemoto and Z. Shimizu, *J. Pharm. Soc. Jpn.*, 1953, **73**, 1026.

56 M. A. Grundl, J. J. Kennedy-Smith and D. Trauner, *Organometallics*, 2005, **24**, 2831.

57 S.-S. Zhang, Z.-Q. Wang, M.-H. Xu and G.-Q. Lin, *Org. Lett.*, 2010, **12**, 5546.

58 X.-S. He, S.-S. Zhang, Y.-L. Guo, H.-Y. Wang and G.-Q. Lin, *Organometallics*, 2012, **31**, 2945.

59 S.-S. Jin, H. Wang and M.-H. Xu, *Chem. Commun.*, 2011, **47**, 7230.

60 (*a*) T. Thaler, L.-N. Guo, A. K. Steib, M. Raducan, K. Karaghiosoff, P. Mayer and P. Knochel, *Org. Lett.*, 2011, **13**, 3182; (*b*) G.-H. Chen, J.-Y. Gui, L.-C. Li and J. Liao, *Angew. Chem., Int. Ed.*, 2011, **50**, 7681; (*c*) F. Xue, X.-C. Li and B.-S. Wan, *J. Org. Chem.*, 2011, **76**, 7256; (*d*) X. Q. Feng, Y. Z. Wang, B. B. Wei, J. Yang and H. F. Du, *Org. Lett.*, 2011, **13**, 3300.

61 S.-S. Jin, H. Wang, T.-S. Zhu and M.-H. Xu, *Org. Biomol. Chem.*, 2012, **10**, 1764.

62 (*a*) T.-S. Zhu, S.-S. Jin and M.-H. Xu, *Angew. Chem., Int. Ed.*, 2012, **51**, 780; (*b*) H. Wang, T.-S. Zhu and M.-H. Xu, *Org. Biomol. Chem.*, 2012, **10**, 9158; (*c*) T.-S. Zhu and M.-H. Xu, *Chin. J. Chem.*, 2013, **31**, 321.

63 (*a*) Y. Li, D.-X. Zhu and M.-H. Xu, *Chem. Commun.*, 2013, **49**, 11659; (*b*) T.-S. Zhu, J.-P. Chen and M.-H. Xu, *Chem. – Eur. J.*, 2013, **19**, 865.

64 (*a*) H. Wang, T. Jiang and M.-H. Xu, *J. Am. Chem. Soc.*, 2013, **135**, 971; (*b*) H. Wang and M.-H. Xu, *Synthesis*, 2013, 2125.

65 H. Wang, Y. Li and M.-H. Xu, *Org. Lett.*, 2014, **16**, 3962.

66 W.-Y. Qi, T.-S. Zhu and M.-H. Xu, *Org. Lett.*, 2011, **13**, 410.

This journal is © the Partner Organisations 2015

Org. Chem. Front., 2015, **2**, 73–89 | **89**