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Two-photon uncageable enzyme inhibitors bearing targeting vectors†

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The activity of two cyclooxygenase-2 enzyme inhibitors, Celecoxib and Lumiracoxib, could be suppressed by coupling to photo-labile protecting groups, so-called photocages. These groups could be further functionalized with a peptide targeting vector for specific cellular delivery. The enzyme inhibition potential of the cyclooxygenase-2 inhibitors could be regained upon two-photon excitation with tissue-transparent near-IR light at 800 nm.

Enzymes are important drug targets, and thus understanding the (indirect) outcome of their inhibition is of fundamental significance.^{1,2} It would therefore be highly desirable to have biochemical tools, which allow for spatio-temporally controlled release of enzyme inhibitors in living cells or organisms. Photo-labile protecting groups (PLPGs), also known as *photocages*, have been shown in different fields of research to enable the release of molecules of interest upon UV-irradiation.^{3–8} Very importantly, two-photon (TP) cages have recently allowed for the release of compounds upon simultaneous TP excitation in the near-IR range.⁹ Consequently, higher spatial precision (sub-cellular) and deeper tissue penetration are possible.¹⁰ A few caged enzyme inhibitors with the potential to be two-photon-uncaged have been reported.^{11–14} However, for those compounds, either no biological TP experiments were performed,¹³ the irradiation times needed were very long (several hours),^{11,12} the enzyme activity upon uncaging was only moderately changed,¹¹ or the uncageable concentrations were physiologically irrelevant.¹⁴ In this article, we fill this gap and report the TP controlled inhibition of cyclooxygenase-2 (COX-2) by the inhibitors Celecoxib and Lumiracoxib.

Based on a TP cage developed by Goeldner and co-workers,¹⁵ which has already been used on HeLa cells,¹⁶ we

developed PLPGs **1** and **2** (see Scheme 1, and ESI† for their synthesis) to cage Celecoxib and Lumiracoxib. Of note, although the main effect of the two compounds is COX-2 inhibition, their indirect effects are quite different. Lumiracoxib is active against certain cancer cells, while Celecoxib is not. The reasons behind this difference are not yet understood. This highlights even further the need for novel biochemical tools as the one developed in this study. Importantly, in view of potential selective drug delivery, we designed the cage structures to allow for the sub-sequent attachment of targeting vectors, as previously performed by our group for a single photon caged Re complex.¹⁷ The carboxylic acid of Lumiracoxib and sulfonamide function of Celecoxib were caged to the alcohol and carboxylic acid moieties of **1** and **2**, respectively, to

Scheme 1 Caging of the COX-2 selective inhibitors Lumiracoxib (top) and Celecoxib (bottom), exemplary attachment of a targeting peptide to caged Celecoxib (bottom right), and the previously reported caged glutamate (top right).¹⁵ Reaction conditions: (a) *N,N'*-dicyclohexylcarbodiimide, DMAP, Lumiracoxib, CH₂Cl₂, 0 °C to rt, 20 h, 80%; (b) 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride, DMAP, Celecoxib, CH₂Cl₂, rt, 2 h, 54%; (c) CuSO₄, sodium ascorbate, IFLLWQR (Pra)RR, THF/H₂O 2 : 1, rt to 60 °C, 30 h. DMAP: 4-dimethylaminopyridine; Pra: (S)-2-amino-4-pentynoic acid.

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give caged Lumiracoxib **3** and caged Celecoxib **4**, respectively. Other cage structures with carboxylic acid attachment groups have been reported before,¹⁸ but to the best of our knowledge, **2** is the first biphenyl nitro cage capable of caging *via* a carboxylic acid function. Consequently, a modified reaction mechanism has to be assumed, possibly in analogy to the uncaging mechanism postulated for 7-nitroindolyl based groups, with subsequent decarboxylation.¹⁹ The biphenyl-core of both cages carries a tetraethylene glycol chain to increase solubility in aqueous media, a critical parameter for the hydrophobic TP cages.²⁰ The terminal azide group can be used to additionally attach any alkyne-containing targeting biomolecules *via* click chemistry. This gives the opportunity to target the biological effect with two orthogonal methods, *i.e.* (1) by directing the caged compound towards the intended tissue or organelle with the targeting vector and (2) by light activation. Importantly, the attachment of the targeting moiety to the caged compounds is by design the last step in the synthesis and hence conveniently adjustable. To exemplify this, a peptide, which was shown to bind to annexin1 and to function as a targeting moiety towards certain cancer cells in mice,²¹ was attached to caged Celecoxib **4**, producing targeted caged Celecoxib **5** (see Scheme 1).

The hydrolytic stability of **3**, **4** and **5** in the dark at room temperature was monitored by HPLC and UV/Vis. After 24 h in solution (acetonitrile/PBS 1 : 1, pH = 7.4), no degradation was observed. The photo-physical properties were characterized and the results are summarized in Table 1. Notably, although photo-degradation of azides to reactive nitrene groups was previously reported,²² such reactions were never observed for **3** or **4**, neither upon UV nor fs-pulsed near-IR irradiation. The single photon uncaging quantum yields were determined upon irradiation with a frequency-tripled Nd-YAG laser at 355 nm with azobenzene as reference, as discussed in recent articles on the accurate determination of uncaging quantum yields.^{23–25} The caged glutamate by Goeldner *et al.* (Scheme 1) was shown to have a quantum yield of 0.1 at 313 nm.¹⁵ For **3**, which links the bio-active compound also *via* an ester to the same cage chromophore, a similar uncaging quantum yield of 0.094 was determined. **4**, which has a sulfonamide as linking group, has a significantly lower quantum yield of 0.013. The quantum yield of **5** is with 0.0047 in the range of **4**, but still

smaller. This difference demonstrates a moderate effect of the targeting moiety on the single-photon uncaging. Low single-photon uncaging rates are in principle desired for two-photon cages since unwanted photolysis under ambient conditions is suppressed. However, in this case, the differences between the compounds are mostly due to the chemical reactions following photo-excitation since the biphenylnitro core is the same for all three compounds. Consequently, also lower TP uncaging action cross-sections ($\delta_a\Phi_u$) can be expected for **4** and **5**, compared to **3**. $\delta_a\Phi_u$ were determined upon irradiation at 800 nm with a fs-pulsed laser with a 5 kHz repetition rate using the known reference 7-hydroxycoumarin-4-ylmethyl acetate.^{26,27} As expected, $\delta_a\Phi_u$ of **4** (0.063 GM) and **5** (0.053 GM) were found to be lower than the one of **3** (0.37 GM). Importantly, the attachment of the targeting peptide was found to not majorly influence the TP uncaging efficiency. As for the single photon uncaging, the value for **3** is in the same range as the one for caged glutamate.¹⁵ Notably, $\delta_a\Phi_u$ of a dye attached to the same biphenylnitro cage core was shown to be more than an order of magnitude greater at wavelengths around 740 nm.¹⁶ Thus, using lower wavelength light would likely also lead to a more efficient release. However, with respect to future applications, the better tissue transparency at 800 nm²⁸ suggests that investigations at this wavelength are more significant.

In a study which investigated TP uncaging at 740 nm, it was stated that $\delta_a\Phi_u$ of at least 0.1 GM are required for biological applications.²⁶ Accordingly, the TP uncaging of **4** and **5** could be insufficient for biological studies. However, for aforementioned reasons, it cannot be taken for granted that this is valid at 800 nm. The minimum value certainly depends on the wavelength. Furthermore, additional laser properties, such as the repetition rate of the laser, play a role. Most TP uncaging studies used lasers with MHz repetition rates.^{27,29} However, 10³ smaller repetition rates (kHz), like in this study, have also been used before.¹⁵ With the same average powers and peak lengths, the peak photon densities differ therefore by the same factor of 10³. Thus, the commonly quoted paradigm that TP uncaging only occurs in the focal point of a laser beam is not generally true. Indeed, we, and others before,¹⁵ found that with a laser beam like the one used in this study it is not necessary to focus the beam since the photon density is sufficient with a collimated beam. Furthermore, undesired light phenomena such as white light generation, which was observed if the laser beam was focused, are avoided with a collimated laser beam. These remarks raise the question if biologically relevant uncaging is still possible with TP uncaging cross-sections below 0.1 GM at 800 nm, and if biological structures can withstand such powerful irradiation. To this end, the inhibition potential of COX-2 by the caged compounds **4** and **5**, whose uncaging efficiencies are clearly below 0.1 GM, was tested with a fluorescent inhibitor screening assay (see Fig. 1). An irradiation time of 15 min was chosen since this time frame was used in biological studies before.^{20,30} Notably, this time is significantly shorter than what was required in other reports on uncaging of drugs.^{11,12,31} To test if the high photon

Table 1 Photophysical uncaging properties of the caged compounds **3**, **4**, **5** and of the literature-known caged glutamate¹⁵

Compound	Φ (λ = 355 nm) ^a	$\delta_a\Phi_u$ (λ = 800 nm) ^b
Caged glutamate	0.1 (λ = 313 nm) ¹⁵	0.45 GM
3	0.094 ± 0.02	0.37 ± 0.04 GM
4	0.013 ± 0.001	0.063 ± 0.008 GM
5	0.0047 ± 0.0004	0.053 ± 0.008 GM

^a Acquired with a Nd-YAG laser, relative to the photoisomerization of azobenzene.^{23–25} ^b Acquired with a fs-pulsed laser with 5 kHz repetition rate relative to 7-hydroxycoumarin-4-ylmethyl acetate.^{26,27} for further details on both methods, see ESI.

Fig. 1 Inhibition constants of Celecoxib, **4**, and **5**, without and upon irradiation with a pulsed laser at 800 nm for 15 min. The values are relative to the initial concentrations of caged compound, or released Celecoxib, as indicated.

densities influence biomolecules, the enzyme was irradiated using these experimental conditions and its activity tested in comparison to a non-irradiated sample. The activity of the enzyme was found to be unaffected by the pulsed laser light. In addition, Celecoxib was irradiated. No decomposition could be detected by HPLC analysis and its inhibitory potential towards COX-2 remained unchanged at around 60–70 nM. Likewise, solutions of the caged compounds **4** and **5** were tested. Before uncaging, no inhibition of COX-2 could be detected within the limits given by the solubility of the caged compounds. After 15 min of irradiation, the amount of free Celecoxib was determined by HPLC. The uncaging progress was around 70% for **4** and 40% for **5**. The IC_{50} values relative to the released Celecoxib are for both **4** and **5** identical to the non-caged Celecoxib. Relative to the caged starting materials, the IC_{50} values are slightly higher due to the incomplete release in the given timeframe. Nevertheless, at least five- and ten-fold increases for **4** and **5**, respectively, were observed.

To assess if the targeting vector afforded the expected selectivity towards specific cancer cells, we prepared a derivative of **5**, which contains an additional fluorescein moiety (**16**, see ESI† for structure and experimental details). By monitoring fluorescence, we could follow the uptake of the bioconjugate into cells. To this end, we selected A549 human lung adenocarcinoma epithelial cells, which express high levels of annexin1, and HEK 293 human embryonic kidney cells, which have low levels of annexin1 expression.^{32,33} Unfortunately, the anticipated preferential uptake of **16** into A549 cells could not be observed (see Fig. S8 and S9†). In a previous study of our group, a luminescent rhenium complex was shown to be taken better up by cells targeted with a peptide attached to a related single photon cage. This suggests that, possibly, biasing caused by the luminescent label itself could be responsible for this observation.³⁴ An effect based on the two-photon cage itself can, however, not be ruled out completely. Nevertheless,

this does not alter the fact that the photocages presented in this article allow for biomolecules to be easily attached.

In summary, we could demonstrate that targetable TP uncaging can be employed to control enzyme activity. The two drugs Celecoxib and Lumiracoxib, which both selectively inhibit COX-2, were deactivated by conjugation to TP cages. A targeting peptide was attached to the cage. Importantly, the peptide did neither significantly alter photorelease nor enzyme inhibition. The uncaging process was shown to allow for efficient control over the inhibition of an enzyme with near-IR light in an *in vitro* assay. The concept holds great promise for the future, not only for chemical biological studies on enzyme function, but potentially also for applications in therapeutic targeted drug delivery. Currently, Photodynamic Therapy (PDT) is the most prominent method which utilizes light in a medicinal context in the treatment of certain skin conditions and cancers.^{35–37} PDT, however, depends on the generation of reactive oxygen species, although many cancerous tissues are hypoxic.^{38,39} Therapeutic TP uncaging, with the ability to activate virtually any drug without depending on additional factors such as oxygen, could therefore further promote non-invasive light based therapies.

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Notes and references

- 1 H. Nam, N. E. Lewis, J. A. Lerman, D.-H. Lee, R. L. Chang, D. Kim and B. O. Palsson, Network Context and Selection in the Evolution to Enzyme Specificity, *Science*, 2012, **337**, 1101–1104.
- 2 P. K. Mazur, N. Reynoird, P. Khatri, P. W. T. C. Jansen, A. W. Wilkinson, S. Liu, O. Barbash, G. S. Van Aller, M. Huddleston, D. Dhanak, P. J. Tummino, R. G. Kruger, B. A. Garcia, A. J. Butte, M. Vermeulen, J. Sage and O. Gozani, SMYD3 links lysine methylation of MAP3K2 to Ras-driven cancer, *Nature*, 2014, **510**, 283–287.
- 3 C. Brieke, F. Rohrbach, A. Gottschalk, G. Mayer and A. Heckel, Light-Controlled Tools, *Angew. Chem., Int. Ed.*, 2012, **51**, 8446–8476.
- 4 H.-M. Lee, D. R. Larson and D. S. Lawrence, Illuminating the Chemistry of Life: Design, Synthesis, and Applications of “Caged” and Related Photoresponsive Compounds, *ACS Chem. Biol.*, 2009, **4**, 409–427.

- 5 G. C. R. Ellis-Davies, Caged compounds: photorelease technology for control of cellular chemistry and physiology, *Nat. Methods*, 2007, **4**, 619–628.
- 6 D. K. Sinha, P. Neveu, N. Gagey, I. Aujard, C. Benbrahim-Bouzi, T. Le Saux, C. Rampon, C. Gauron, B. Goetz, S. Dubruille, M. Baaden, M. Volovitch, D. Bensimon, S. Vriz and L. Jullien, Photocontrol of Protein Activity in Cultured Cells and Zebrafish with One- and Two-Photon Illumination, *ChemBioChem*, 2010, **11**, 653–663.
- 7 A. A. Kumbhar, A. T. Franks, R. J. Butcher and K. J. Franz, Light uncages a copper complex to induce nonapoptotic cell death, *Chem. Commun.*, 2013, **49**, 2460–2462.
- 8 C. Gwizdala and S. C. Burdette, in *Inorganic Chemical Biology: Principles, Techniques and Applications*, ed. G. Gasser, John Wiley & Sons, Ltd, Chichester, 2014, ch. 9, pp. 275–308.
- 9 T. M. Dore, in *Dynamic studies in biology. Phototriggers, photoswitches and caged biomolecules*, ed. M. Goeldner and R. Givens, Wiley-VCH, Weinheim, 2005, ch. 8, pp. 435–459.
- 10 E. B. Brown, J. B. Shear, S. R. Adams, R. Y. Tsien and W. W. Webb, Photolysis of Caged Calcium in Femtoliter Volumes Using Two-Photon Excitation, *Biophys. J.*, 1999, **76**, 489–499.
- 11 H. J. Montgomery, B. Perdicakis, D. Fishlock, G. A. Lajoie, E. Jervis and J. Guy Guillemette, Photo-Control of nitric oxide synthase activity using a caged isoform specific inhibitor, *Bioorg. Med. Chem.*, 2002, **10**, 1919–1927.
- 12 B. Perdicakis, H. J. Montgomery, G. L. Abbott, D. Fishlock, G. A. Lajoie, J. G. Guillemette and E. Jervis, Photocontrol of nitric oxide production in cell culture using a caged isoform selective inhibitor, *Bioorg. Med. Chem.*, 2005, **13**, 47–57.
- 13 M. Goard, G. Aakalu, O. D. Fedoryak, C. Quinonez, J. St. Julien, S. J. Poteet, E. M. Schuman and T. M. Dore, Light-Mediated Inhibition of Protein Synthesis, *Chem. Biol.*, 2005, **12**, 685–693.
- 14 D. Abate-Pella, N. A. Zeliadt, J. D. Ochocki, J. K. Warmka, T. M. Dore, D. A. Blank, E. V. Wattenberg and M. D. Distefano, Photochemical Modulation of Ras-Mediated Signal Transduction Using Caged Farnesyltransferase Inhibitors: Activation by One- and Two-Photon Excitation, *ChemBioChem*, 2012, **13**, 1009–1016.
- 15 S. Gug, S. Charon, A. Specht, K. Alarcon, D. Ogden, B. Zietz, J. Léonard, S. Haacke, F. Bolze, J.-F. Nicoud and M. Goeldner, Photolabile Glutamate Protecting Group with High One- and Two-Photon Uncaging Efficiencies, *ChemBioChem*, 2008, **9**, 1303–1307.
- 16 D. Warther, F. Bolze, J. Léonard, S. Gug, A. Specht, D. Puliti, X.-H. Sun, P. Kessler, Y. Lutz, J.-L. Vonesch, B. Winsor, J.-F. Nicoud and M. Goeldner, Live-Cell One- and Two-Photon Uncaging of a Far-Red Emitting Acridinone Fluorophore, *J. Am. Chem. Soc.*, 2010, **132**, 2585–2590.
- 17 A. Leonidova, V. Pierroz, R. Rubbiani, Y. Lan, A. G. Schmitz, A. Kaech, R. K. O. Sigel, S. Ferrari and G. Gasser, Photo-Induced Uncaging of a Specific Re(I) Organometallic Complex in Living Cells, *Chem. Sci.*, 2014, 4044–4056.
- 18 N. Gagey, M. Emond, P. Neveu, C. Benbrahim, B. Goetz, I. Aujard, J.-B. Baudin and L. Jullien, Alcohol Uncaging with Fluorescence Reporting: Evaluation of o-Acetoxyphenyl Methyloxazolone Precursors, *Org. Lett.*, 2008, **10**, 2341–2344.
- 19 G. Papageorgiou, D. C. Ogden, A. Barth and J. E. T. Corrie, Photorelease of Carboxylic Acids from 1-Acyl-7-nitroindolines in Aqueous Solution: Rapid and Efficient Photorelease of L-Glutamate, *J. Am. Chem. Soc.*, 1999, **121**, 6503–6504.
- 20 S. C. Boca, M. Four, A. Bonne, B. van der Sanden, S. Astilean, P. L. Baldeck and G. Lemercier, An ethylene-glycol decorated ruthenium(ii) complex for two-photon photodynamic therapy, *Chem. Commun.*, 2009, 4590–4592.
- 21 S. Hatakeyama, K. Sugihara, T. K. Shibata, J. Nakayama, T. O. Akama, N. Tamura, S.-M. Wong, A. A. Bobkov, Y. Takano, C. Ohyama, M. Fukuda and M. N. Fukuda, Targeted drug delivery to tumor vasculature by a carbohydrate mimetic peptide, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 19587–19592.
- 22 G. M. Anstead, S. E. Sen and G. D. Prestwich, Evaluation of squalene analogs bearing photoreactive groups as inhibitors of squalene epoxidase and oxidosqualene cyclase, *Bioorg. Chem.*, 1991, **19**, 300–313.
- 23 P. Anstaett, A. Leonidova and G. Gasser, Caged Phosphate and the Slips and Misses in Determination of Quantum Yields for UV-A induced Photouncaging, *ChemPhysChem*, 2015, **16**, 1857–1860.
- 24 J. E. T. Corrie, J. H. Kaplan, B. Forbush, D. C. Ogden and D. R. Trentham, Commentary on “Caged Phosphate and the Slips and Misses in Determination of Quantum Yields for Ultraviolet-A-Induced Photouncaging” by G. Gasser and Co-Workers, *ChemPhysChem*, 2015, **16**, 1861–1862.
- 25 P. Anstaett, A. Leonidova, E. Janett, C. G. Bochet and G. Gasser, Reply to the Commentary by Trentham *et al.* on “Caged Phosphate and the Slips and Misses in Determination of Quantum Yields for Ultraviolet-A-Induced Photouncaging” by Gasser *et al.*, *ChemPhysChem*, 2015, **16**, 1863–1866.
- 26 T. Furuta, S. S.-H. Wang, J. L. Dantzker, T. M. Dore, W. J. Bybee, E. M. Callaway, W. Denk and R. Y. Tsien, Brominated 7-hydroxycoumarin-4-ylmethyls: Photolabile protecting groups with biologically useful cross-sections for two photon photolysis, *Proc. Natl. Acad. Sci. U. S. A.*, 1999, **96**, 1193–1200.
- 27 A. Specht, F. Bolze, J. F. Nicoud and M. Goeldner, in *Chemical Neurobiology*, ed. M. R. Banghart, Humana Press, New York, 2013, ch. 6, vol. 995, pp. 79–87.
- 28 R. R. Anderson and J. A. Parrish, The Optics of Human Skin, *J. Invest. Dermatol.*, 1981, **77**, 13–19.
- 29 L. Donato, A. Mourot, C. M. Davenport, C. Herbivo, D. Warther, J. Léonard, F. Bolze, J.-F. Nicoud, R. H. Kramer, M. Goeldner and A. Specht, Water-Soluble, Donor-Acceptor Biphenyl Derivatives in the 2-(o-Nitrophenyl)propyl Series: Highly Efficient Two-Photon Uncaging of the Neuro-

- transmitter γ -Aminobutyric Acid at $\lambda = 800$ nm, *Angew. Chem., Int. Ed.*, 2012, **51**, 1840–1843.
- 30 H. A. Collins, M. Khurana, E. H. Moriyama, A. Mariampillai, E. Dahlstedt, M. Balaz, M. K. Kuimova, M. Drobizhev, X. D. YangVictor, D. Phillips, A. Rebane, B. C. Wilson and H. L. Anderson, Blood-vessel closure using photosensitizers engineered for two-photon excitation, *Nat. Photonics*, 2008, **2**, 420–424.
 - 31 C. Bao, M. Jin, B. Li, Y. Xu, J. Jin and L. Zhu, Long conjugated 2-nitrobenzyl derivative caged anticancer prodrugs with visible light regulated release: preparation and functionalizations, *Org. Biomol. Chem.*, 2012, **10**, 5238–5244.
 - 32 M. Uhlén, L. Fagerberg, B. M. Hallström, C. Lindskog, P. Oksvold, A. Mardinoglu, Å. Sivertsson, C. Kampf, E. Sjöstedt, A. Asplund, I. Olsson, K. Edlund, E. Lundberg, S. Navani, C. A.-K. Szgyarto, J. Odeberg, D. Djureinovic, J. O. Takanen, S. Hober, T. Alm, P.-H. Edqvist, H. Berling, H. Tegel, J. Mulder, J. Rockberg, P. Nilsson, J. M. Schwenk, M. Hamsten, K. von Feilitzen, M. Forsberg, L. Persson, F. Johansson, M. Zwahlen, G. von Heijne, J. Nielsen and F. Pontén, Tissue-based map of the human proteome, *Science*, 2015, **347**, 1260419.
 - 33 M. Uhlen, P. Oksvold, L. Fagerberg, E. Lundberg, K. Jonasson, M. Forsberg, M. Zwahlen, C. Kampf, K. Wester, S. Hober, H. Wernerus, L. Bjorling and F. Ponten, Towards a knowledge-based Human Protein Atlas, *Nat. Biotechnol.*, 2010, **28**, 1248–1250.
 - 34 C. A. Puckett and J. K. Barton, Fluorescein Redirects a Ruthenium–Octaarginine Conjugate to the Nucleus, *J. Am. Chem. Soc.*, 2009, **131**, 8738–8739.
 - 35 M. Triesscheijn, P. Baas, J. H. M. Schellens and F. A. Stewart, Photodynamic Therapy in Oncology, *Oncologist*, 2006, **11**, 1034–1044.
 - 36 A. Frei, R. Rubbiani, S. Tubafard, O. Blaque, P. Anstaett, A. Felgentraeger, T. Maisch, L. Spiccia and G. Gasser, Synthesis, Characterization and Biological Evaluation of New Ru(II) Polypyridyl Photosensitizers for Photodynamic Therapy, *J. Med. Chem.*, 2014, **57**, 7280–7292.
 - 37 M. R. Hamblin and T. Hasan, Photodynamic therapy: a new antimicrobial approach to infectious disease?, *Photochem. Photobiol. Sci.*, 2004, **3**, 436–450.
 - 38 J. A. Bertout, S. A. Patel and M. C. Simon, The impact of O₂ availability on human cancer, *Nat. Rev. Cancer*, 2008, **8**, 967–975.
 - 39 P. Vaupel and A. Mayer, Hypoxia in cancer: significance and impact on clinical outcome, *Cancer Metastasis Rev.*, 2007, **26**, 225–239.