



Cite this: *Org. Biomol. Chem.*, 2015, **13**, 10705

Synthesis of aza and carbocyclic β -carbolines for the treatment of alcohol abuse. Regiospecific solution to the problem of 3,6-disubstituted β - and aza- β -carboline specificity†

V. V. N. Phani Babu Tiruveedhula,^a Kashi Reddy Methuku,^a Jeffrey R. Deschamps^b and James M. Cook^{*a}

A novel two step protocol was developed to gain regiospecific access to 3-substituted β - and aza- β -carbolines, 3-PBC (**1**), 3-ISOPBC (**2**), β CCt (**3**), 6-aza-3-PBC (**4**) and 6-aza-3-ISOPBC (**5**). These β -carbolines (**1–3**) are potential clinical agents to reduce alcohol self-administration, especially 3-ISOPBC·HCl (**2·HCl**) which appears to be a potent anti-alcohol agent active against binge drinking in a rat model of maternally deprived (MD) rats. The method consists of two consecutive palladium-catalyzed reactions: a Buchwald–Hartwig amination followed by an intramolecular Heck-type cyclization in high yield.

Received 29th July 2015,
Accepted 3rd September 2015

DOI: 10.1039/c5ob01572c

www.rsc.org/obc

Introduction

β -Carbolines, aza- β -carbolines and their derivatives are important targets in synthetic chemistry.¹ In addition, they are found in a large number of natural products, many of which demonstrate novel biological activity, especially in regard to the reduction of alcohol self-administration [binge drinking (BD)]. This is proposed to be due to the activity at the benzodiazepine site of the GABA_A receptor.² Surprisingly, BD kills six people a day, most of which are men, and approximately 88 000 people die from alcohol related issues annually making it the third leading preventable cause of death in the United States.³ In 2006, this alcohol misuse cost the US government approximately \$223.5 billion dollars.³ BD (Blood-alcohol level ≥ 0.08 g% in a 2 hour period) is one form of excessive drinking and because of it, alcohol addiction and dependence remain a significant public health concern.⁴ Maternal separation and early life events can cause profound neurochemical and behavioral alterations in childhood that persist into adulthood, enhance the risk to develop alcohol use disorders and excessive drinking.^{5–7} Consequently, the development of clinically safe and cost effective therapeutic agents to reduce alcohol

addiction and dependence remain essential for the future treatment of alcoholism.^{8,9}

One influence on alcohol abuse is known to be mediated by GABA_A receptors, the major inhibitory chloride ion gated channels with γ -aminobutyric acid (GABA) as the endogenous ligand in the central nervous system. It plays a vital role in several neuronal disorders including anxiety, epilepsy, insomnia, depression, bipolar disorder, schizophrenia, as well as mild cognitive impairments and Alzheimer's disease.^{10–15} The pentameric structure of the GABA_A receptor is made up of 2 α , 2 β and 1 γ subunits, with a higher distribution of the α 1-subunit in the mesolimbic system of the ventral pallidum (VP) possibly playing an important role in regulating alcohol abuse.^{16–20} However, the precise neuromechanisms of regulating alcohol-seeking behavior remain unknown. In addition to the ventral pallidum, there is now compelling evidence that the GABA_A receptors within the striatopallidal and extended amygdala system are involved in the 'acute' reinforcing actions of alcohol.^{21–23}

To evaluate the role of the α 1 receptor in regulating alcohol reinforcement, the orally active β -carbolines 3-propoxy- β -carboline hydrochloride **1·HCl** (3-PBC·HCl) and β -carboline-3-carboxylate-*tert*-butyl ester **3** (β CCt), the mixed benzodiazepine (BDZ) agonist-antagonists with binding selectivity at the α 1 Bz/GABA_A receptor were developed (see Fig. 1).^{18,24,25} Behavioral studies in several species (*e.g.*, rats, mice, primates) show that these ligands were BDZ antagonists, at the α 1 Bz/GABA_A subtype exhibiting competitive binding-site interactions with BDZ agonists over a broad range of doses.^{18,24,26} In studies which involved the α 1 subtype, they were shown to selectively reduce alcohol-motivated behaviors and more importantly,

^aDepartment of Chemistry and Biochemistry, University of Wisconsin-Milwaukee, Milwaukee, WI 53201, USA. E-mail: capncook@uwm.edu

^bCenter for Biomolecular Science and Engineering, Naval Research Laboratory, Code 6930, Washington, D. C. 20375, USA

† Electronic supplementary information (ESI) available: Copies of spectra and crystallographic information files in CIF format. CCDC 1040831–1040833 and 1044936. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c5ob01572c



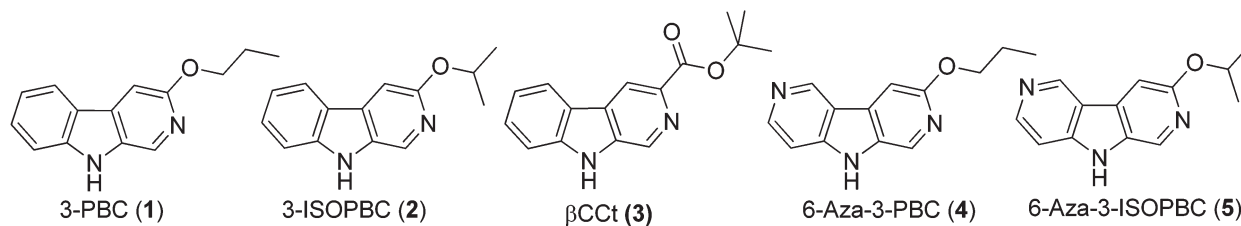


Fig. 1 Structures of 3-PBC (1), 3-ISOPBC (2), βCct (3), 6-aza-3-PBC (4) and 6-aza-3-ISOPBC (5).

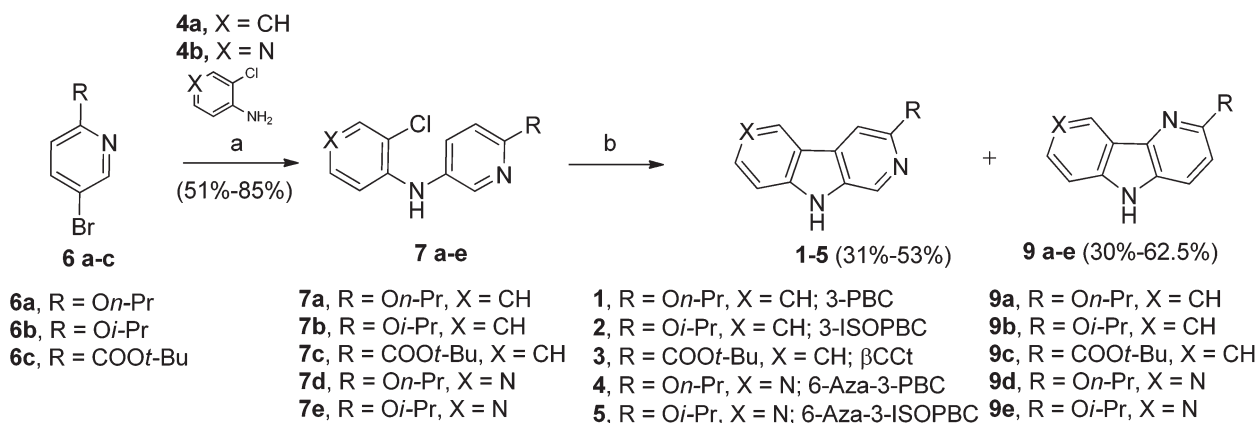
3-PBC-HCl significantly reduced alcohol self-administration and reduced craving in baboons.²⁶ β-Carbolines **1-HCl** and **3** displayed mixed weak agonist-antagonist profiles *in vivo* in alcohol preferring (P) and high alcohol drinking (HAD) rats.^{18,26–28} Therefore, in addition to their use to study the molecular basis of alcohol reinforcement, α1 Bz β-carboline ligands which display mixed pharmacological antagonist-agonist activity in alcohol P and HAD rats may be capable of reducing alcohol intake while eliminating or greatly reducing the anxiety associated with habitual alcohol, abstinence or detoxification.^{18,28–30} Consequently, these types of ligands may be ideal clinical agents for the treatment of alcohol dependent individuals.

Results and discussion

Previously, the β-carbolines **1** and **3** have been synthesized from DL-tryptophan. The overall yield of **1** (*via* 6 steps) as reported previously was 8%, while the combined yield of **3** (5 steps) was 35%. A few key steps occurred in low yields which was something of which we sought to improve on^{31–34} in a continued effort to find more potent subtype selective ligands for GABA_A receptors. This interest resulted in a short and concise synthesis of **1** and **3**. In 2011, a palladium catalyzed two-step protocol for the synthesis of **1**, and **3** as well as analogs of **1** was reported.³⁵ In the search for a more potent subtype selec-

tive ligand for the GABA_A receptor, with the knowledge that many 3-substituted β-carbolines and more water soluble aza-β-carbolines might exhibit greater subtype selectivity at α1β_{2/3}γ2 BZR/GABAergic receptors,^{31–33,36–38} the ligands 3-ISOPBC (**2**), 6-aza-3-PBC (**4**), and 6-aza-3-ISOPBC (**5**) were designed (see Fig. 1) and synthesized using a two-step protocol (Scheme 1).

As shown in Scheme 1, bromopyridines **6a–c**^{39,40} were reacted with anilines **4a–b** in toluene at 100–140 °C in the presence of 5 mol% Pd(OAc)₂ and 7.5 mol% X-Phos to obtain the corresponding diarylamines **7a–e** in moderate to good yields. Unfortunately, the intramolecular Heck cyclization [Pd(OAc)₂, (*t*-Bu)₃P·HBF₄, K₂CO₃, DMA, 120 °C] of **7a–e** afforded both the β-carbolines **1–5** (individually) and their regioisomeric δ-carbolines **9a–e**, respectively. Carbolines **2**, **3**, **9a**, and **9d** were subjected to X-ray crystallographic analysis (see Fig. 2, Scheme 4, and the ESI[†]) to confirm the regiochemistry. Although this protocol permitted synthesis of β-carbolines on gram scale for *in vivo* studies, occasionally the first step in the Buckwald–Hartwig coupling failed to give complete conversion into the carboline. This complicated purification for the diarylamine was difficult to purify *via* column chromatography because the diarylamine and one of the starting anilines had almost identical *R_f* values. Furthermore, in the case of the water soluble aza-β-carboline the yields (51%) were very poor and importantly, since the second step was not regioselective, this required careful purification to remove the unwanted δ-carboline present in 30 to 62.5% yield (Scheme 1). Interest-



Scheme 1 Synthesis of substituted carboline analogues. Reagents and conditions: (a) Pd(OAc)₂, X-Phos, Cs₂CO₃, toluene, 100–140 °C, 15–24 h (b) Pd(OAc)₂, (*t*-Bu)₃P·HBF₄, K₂CO₃, DMA, 120 °C, 16 h.



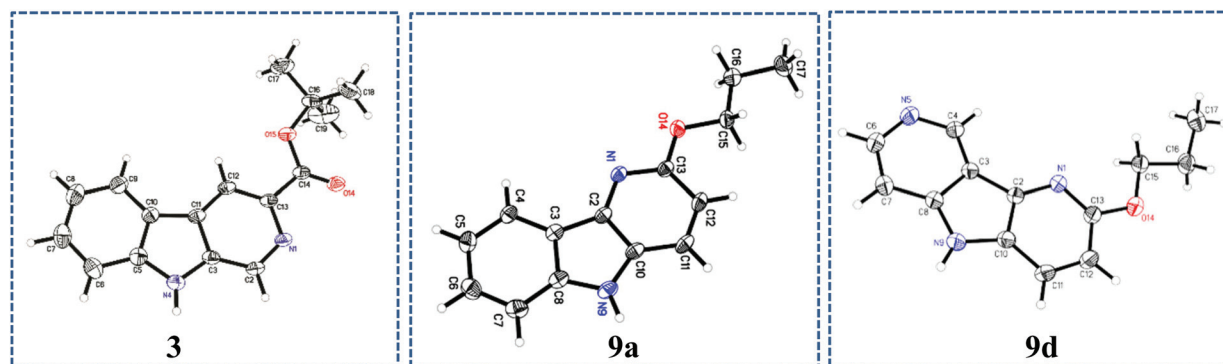
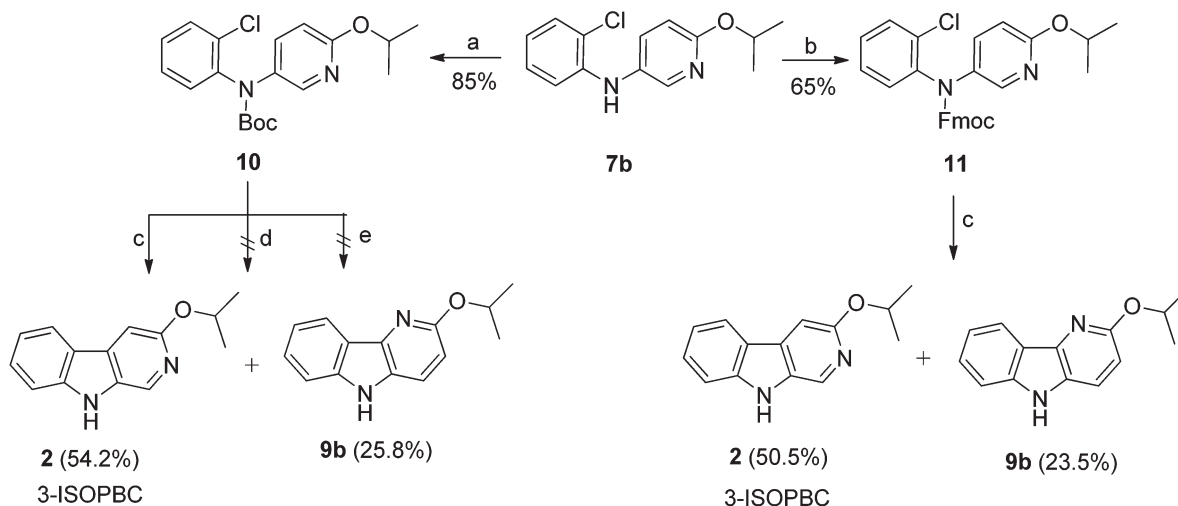


Fig. 2 ORTEP view of the crystal structure of substituted carbolines **3**, **9a**, and **9d** (displacement ellipsoids are at the 50% level (β -carboline numbering not followed).

ingly, the *in vivo* results (unpublished) for 3-isopropoxy- β -carboline hydrochloride **2·HCl** (3-ISOPBC·HCl) carried out in maternally deprived rats for binge drinking decreased dramatically this self-administration compared to **1·HCl** without affecting the overall activity of the rats (*i.e.* no sedation). This important finding led to the interest in a regiospecific synthesis of 3-ISOPBC (**2**) on large scale.

The revised synthetic strategy for the regiospecific synthesis of **2** began with the protection of the intermediate amine **7b** (N_a -H) with bulkier groups such as *tert*-butyloxycarbonyl (Boc) **10** or a fluorenylmethyloxy group (Fmoc) **11**, which might block the formation of the Pd^{II} π -complex that is required to obtain the undesired regioisomeric δ -carboline. The Boc protected amine **10** was easily accessible by treating the amine **7b** with di-*tert*-butyl dicarbonate (Boc)₂O and 4-(dimethylamino)pyridine (DMAP) in good yield (85%). The Fmoc protected amine **11** was synthesized under solvent free conditions by reaction of the amine **7b** and Fmoc-Cl by microwave

irradiation at 80 °C in moderate yield (65%, Scheme 2).⁴¹ Once protected, diarylamines **10** and **11** were subjected to a palladium catalyzed Heck-type cyclization using similar conditions to those from above. Unfortunately, both reactions afforded the deprotected regioisomers 3-ISOPBC (**2**) and δ -isomer **9b** in approximately the same 2 : 1 ratio, as compared to cyclization with the previously unprotected diarylamine **7b** (see Scheme 1 above). It was felt that deprotection of the carbamate occurred once the indole ring had formed (Scheme 2) which provided the better indole leaving group. To test the thermal stability of the carbamate starting materials, diarylamines **10** and **11** were heated at 120 °C in DMA; they were stable to these conditions. In addition, the cyclization with PdCl₂(PPh₃)₂ as a palladium source was also attempted using standard Heck-type reaction conditions with a milder base (NaOAc), but this failed to give the cyclized product. We also explored the reaction by varying the water content using NaOAc·3H₂O as a base; however, there was no cyclization (Scheme 2).



Scheme 2 Synthesis of the carbamate protected analogs from intermediate **7b**. Reagents and conditions: (a) (Boc)₂O, DMAP, THF, rt, 24 h; (b) Fmoc-Cl, 80 °C, microwave, 1 h; (c) Pd(OAc)₂, (t-Bu)₃P·HBF₄, K₂CO₃, DMA, 120 °C, 16 h; (d) PdCl₂(PPh₃)₂, NaOAc·3H₂O, DMA, 120 °C, 14 h; (e) PdCl₂(PPh₃)₂, NaOAc, DMA, 120 °C, 14 h.



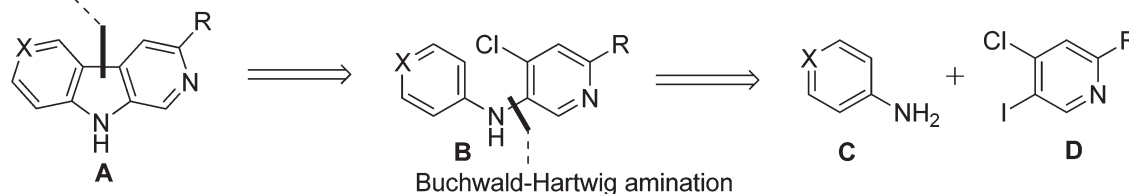
The second approach rested on the important switch of the chlorine atom from the benzene ring to the pyridine ring in amine **7b**. Retrosynthetically, it was envisioned that the core structure of 3,6-disubstituted β -carboline **A** could be obtained from diarylamine **B** via an intramolecular Heck cyclization and it was anticipated that diarylamine **B** could arise from a substituted aniline **C** and a substituted pyridine derivative **D** via a Buchwald–Hartwig amination (Scheme 3).

At this point it was decided to explore the regioselective palladium catalyzed Buchwald–Hartwig coupling between aniline and pyridine **14**⁴² for the synthesis of diarylamine **16** (Table 1). With the previous history in mind,³⁵ the initial attempt was made with 5 mol% Pd(OAc)₂, 7.5 mol% X-Phos and Cs₂CO₃ (1.5 equiv.) in toluene at 110 °C which gave only 18% of the diarylamine **16** with a large excess of unreacted starting material even after heating for 24 hours (Table 1, entry 1). However, the catalyst based on the combination of Pd₂(dba)₃, Xantphos and Pd(OAc)₂, Xantphos with Cs₂CO₃ in toluene and dioxane gave the desired product diarylamine **16** in up to 62% yield (Table 1, entries 2–3). The ligand Xantphos has been shown to be efficient in cross coupling reactions of C–N bond formation because of a wider bite angle,⁴³ which facilitates the reductive elimination. In addition, the excess base may also

play a role in the improvement of the yield.⁴³ In recent years rapid synthesis with microwave technology has attracted a considerable amount of attention for C–N bond formation.^{44–46} All three previous cyclizations were attempted with microwave irradiation (for 1 hour) in order to decrease the duration of the reaction time, as well as increase the selectivity under similar reaction conditions. However, the results were the same except that in the Xantphos-based ligand systems the cyclizations were completed in 1 hour. During continuation of the study of this selective amination, recent reports from Buchwald and co-workers⁴⁷ demonstrated air- and moisture-stable palladacyclic precatalysts, when employed with aryl iodides and heteroaryl iodides were attractive substrates in Pd-catalyzed C–N cross-coupling reactions. This process works by preventing formation of the stable bridging iodide dimers and also using a solvent system in which iodide salts were insoluble. These complexes easily undergo deprotonation and reductive elimination to generate LPd(0) along with relatively inert indoline (for generation of **1**) or carbazole (for generation of **2** and **3**). These conditions also permit the successful coupling of aryl iodides with amines at ambient temperature.^{47–50}

The first attempt in this modification was to use the Buchwald 3rd generation palladacycle precatalyst (BrettPhos Pd G3)

Intramolecular Heck Cyclization



Scheme 3 Retrosynthetic analysis of 3,6-disubstituted β -carbolines.

Table 1 Optimization of conditions for regioselective synthesis of intermediate **16** from **14**^a

Entry	Pd source	Ligand	Base (equiv.)	Solvent	Temp (time)	Yield ^b (%)
1	Pd(OAc) ₂	X-Phos	Cs ₂ CO ₃ (1.5)	Toluene	110 °C (24 h)	18 ^c
2	Pd ₂ (dba) ₃	Xantphos	Cs ₂ CO ₃ (2)	Dioxane	110 °C (6 h)	51
3	Pd(OAc) ₂	Xantphos	Cs ₂ CO ₃ (4)	Toluene	110 °C (6 h)	62
4	BrettPhos Pd G3	BrettPhos	Cs ₂ CO ₃ (1.5)	Toluene	110 °C (14 h)	45
5	BrettPhos Pd G3	BrettPhos	Cs ₂ CO ₃ (3)	Toluene	110 °C (5 h)	66
6	BrettPhos Pd G3	BrettPhos	NaOt-Bu (1.5)	Toluene	110 °C (5 h)	52
7	BrettPhos Pd G3	BrettPhos	Cs ₂ CO ₃ (5)	Toluene	110 °C (5 h)	0 ^e
8	Pd ₂ (dba) ₃	Xantphos	Cs ₂ CO ₃ (5)	Toluene	110 °C (3 h)	74
9	Pd(OAc) ₂	<i>rac</i> -BINAP	Cs ₂ CO ₃ (5)	Toluene	110 °C (5 h)	80
10	Pd(OAc) ₂	<i>rac</i> -BINAP	K ₂ CO ₃ (5)	Toluene	110 °C (24 h)	22
11	Pd(OAc)₂	<i>rac</i> -BINAP	Cs₂CO₃ (5)	Toluene	110 °C (5 h)	92^d

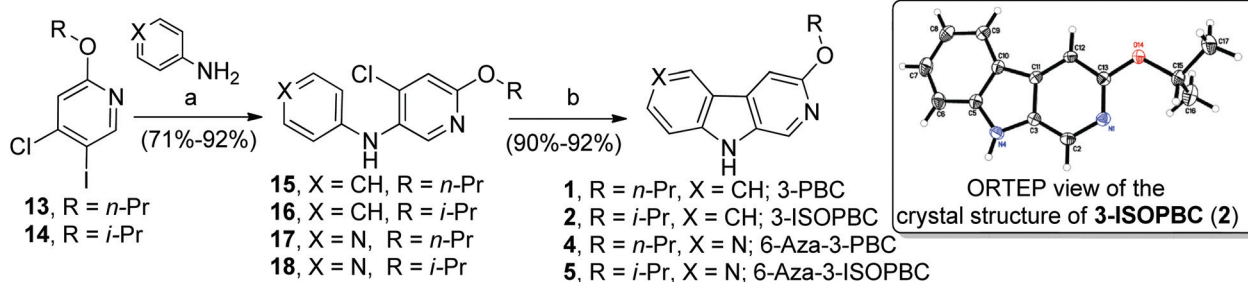
^a **14** (0.1 mmol), aniline (0.12 mmol), Pd (3 mol%), ligand (3 mol%), base, and solvent (1 mL). ^b Isolated yields. ^c Pd (5 mol%), ligand (7.5 mol%). ^d Aniline (0.1 mmol). ^e 90% of diaminated product [6-isopropoxy-*N*³,*N*⁴-diphenylpyridine-3,4-diamine] was observed.



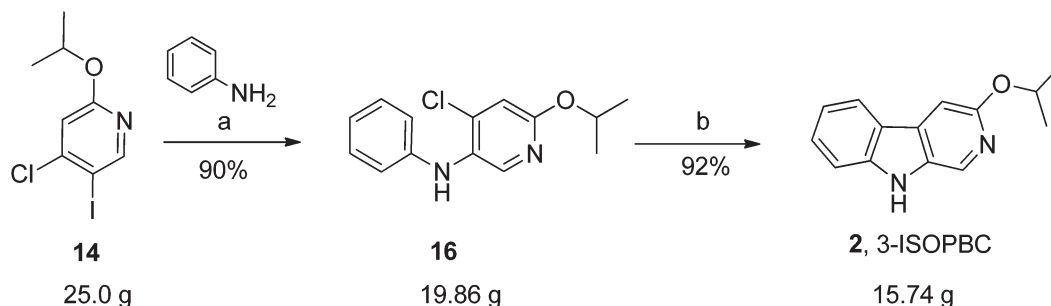
with the BrettPhos ligand in the presence of Cs_2CO_3 or NaOt-Bu in toluene at room temperature. This failed to give the desired product and there was no consumption of starting material. Following this attempt, the temperature was raised to reflux, with the addition of 3 equivalents of Cs_2CO_3 and the reaction went to completion within 5 hours. However, it only gave the desired amine **16** in 66% yield (Table 1, entry 5). When the same experiment was performed using only 1.5 equiv. of Cs_2CO_3 the process took a longer time to go to completion with an isolated yield of 45% of the desired amine **16**. This was accompanied by the diaminated product [6-isopropoxy- N^3,N^4 -diphenylpyridine-3,4-diamine] in ~18% yield (Table 1, entry 4). Unfortunately, when the stronger base NaOt-Bu was employed comparable results to the above reaction (Table 1, entry 4) were obtained accompanied by more decomposed material [TLC(silica gel; Table 1, entry 6)]. The use of excess base (Cs_2CO_3) gave only the unwanted diaminated product in 90% yield (Table 1, entry 7). It was found the $\text{Pd}(\text{OAc})_2$, *rac*-BINAP and K_2CO_3 combination, unfortunately, did not lead to full conversion even after heating for 24 hours (Table 1, entry 10). Interestingly, the catalyst system $\text{Pd}_2(\text{dba})_3$ and Xantphos with a large excess of base [Cs_2CO_3 (5 equiv.)] gave 74% yield of **16**, whereas the catalyst system $\text{Pd}(\text{OAc})_2$, *rac*-BINAP under similar reaction conditions yielded 80% (Table 1, entry 8 and 9) of the desired amine **16**. Remarkably, these data indicated a large excess of mild base was essential to obtain good yields, as well as selectivity. Furthermore, a rate-limiting interphase deprotonation of the Pd(II)-amine complex inter-

mediate has occurred in the catalytic cycle.^{51–53} Encouraged by these promising results, efforts turned toward lowering the aniline loading from 1.2 equivalents to 1 equivalent for regioselectivity. In doing so we achieved selective amination of pyridine **14** with aniline. Interestingly, neither a 4- nor 4,5-diaminated pyridine product was obtained. Using this catalyst-base combination in refluxing toluene, the desired cross-coupling proceeded smoothly to provide the desired anilino-pyridine **16** in excellent yield (92%, Table 1, entry 11). Interestingly, the same reaction conditions gave good yields in the case of the more polar starting 4-amino pyridine (Scheme 4); however, the temperature was necessarily increased to 140 °C to increase the solubility of the starting material, 4-amino pyridine. In contrast, when a polar solvent such as DMA was employed, the result was either inferior yields and/or deiodination of pyridine **16**, as mentioned above.

Once the diarylamines **15–18** were in hand in good to excellent yields, the previously applied Heck-type conditions [$\text{Pd}(\text{OAc})_2$, (*t*-Bu)₃P-HBF₄, K_2CO_3 , DMA, 120 °C] were employed for cyclization. Gratifyingly, this catalyst system gave excellent yields of 91–92% and 90–92% for β -carboline **1–2** and aza- β -carboline **4–5**, respectively (Scheme 4). The switch of the chlorine position from the benzene ring to the pyridine ring worked regiospecifically and completely eliminated the corresponding unwanted δ regioisomer. This completely eliminated the difficult chromatography required to separate β and δ carbolines. The 3-ISOPBC **2** has now been prepared on 15–25 gram scale for studies *in vivo* (Scheme 5) and it is very



Scheme 4 Regiospecific synthesis of β -carboline (**1–2**) and aza- β -carboline (**4–5**). Reagents and conditions: (a) $\text{Pd}(\text{OAc})_2$, *rac*-BINAP, Cs_2CO_3 , toluene, 110–140 °C, 5–6 h (b) $\text{Pd}(\text{OAc})_2$, (*t*-Bu)₃P-HBF₄, K_2CO_3 , DMA, 120 °C, 16 h.



Scheme 5 Large-scale regiospecific synthesis of β -carboline 3-ISOPBC (**2**). Reagents and conditions: (a) $\text{Pd}(\text{OAc})_2$, *rac*-BINAP, Cs_2CO_3 , toluene, 110 °C, 15 h (b) $\text{Pd}(\text{OAc})_2$, (*t*-Bu)₃P-HBF₄, K_2CO_3 , DMA, 120 °C, 16 h.



easy to scale up to 50–100 gram level. Finally, the overall yield increased from 43% to 84% compared to the previous syntheses.^{33,35}

Conclusions

In conclusion, a novel two-step regioselective route to the four anti-alcohol agents of biological interest, 3-PBC (**1**), 3-ISOPBC (**2**), 6-aza-3-PBC (**4**) and 6-aza-3-ISOPBC (**5**), has been developed. The process provided improved yields when compared to the earlier reported syntheses.^{33,35} This two-step protocol consists of the combination of a regioselective Buchwald–Hartwig amination and an intramolecular Heck-type cyclization. The first step, regioselective arylamination, was achieved by using a Pd-BINAP catalytic system in combination with a large excess of Cs₂CO₃, while the latter intramolecular Heck-type cyclization went smoothly with Pd(OAc)₂ in combination with the air-stable monodentate ligand (*t*-Bu)₃·HBF₄ and K₂CO₃. These conditions permit the presence of base sensitive functional groups in the substrates. Regioselective synthesis of β- and aza-β-carbolines was achieved by simply changing the chlorine position from the benzene ring to the pyridine derivatives. Importantly, these reactions are capable of scale-up to multigram quantities and were performed on 25 gram scale level for *in vivo* biology. We observed similar results except in the case of the Buchwald–Hartwig amination step, where it required an increase of the catalyst loading from 3 to 6 mol% whenever the starting material was not consumed. This new process reported here provides the material necessary to study alcohol self-administration and reduction thereof in MD rats and in primates. This regioselective two-step synthetic protocol increased the overall yield from 43% to 84% in the case of β-carbolines **1–2** and from 16% to 66% for aza-β-carbolines **4–5** respectively, and negated the need for a difficult chromatographic step.

Experimental

General considerations

All reactions were carried out in oven-dried, round-bottom flasks or in resealable screw-cap test tubes or heavy-wall pressure vessels under an argon atmosphere. The solvents were anhydrous unless otherwise stated. Stainless steel syringes were used to transfer air-sensitive liquids. Organic solvents were purified when necessary by standard methods or purchased from commercial suppliers. Anhydrous solvents of toluene, dioxane and *N,N*-dimethylacetamide (DMA) were subjected to the freeze-thaw method to render them oxygen free to execute the Buchwald–Hartwig coupling and intramolecular Heck reactions. All chemicals purchased from commercial suppliers were employed as is, unless stated otherwise in regard to purification. Silica gel (230–400 mesh) for flash chromatography was utilized to purify the analogues. The ¹H and ¹³C NMR data were obtained on an NMR spectrometer (300 MHz/500 MHz) instrument with chemical shifts in δ (ppm) reported

relative to TMS. The HRMS were obtained on a LCMS-IT-TOF mass spectrometer by Dr Mark Wang.

General procedure for the Buchwald–Hartwig coupling reaction between substituted anilines and substituted pyridines: representative procedure for the synthesis of *N*-(2-chlorophenyl)-6-propoxy-pyridin-3-amine (**7a**)

A heavy-wall pressure tube was equipped with a magnetic stir bar and fitted with a rubber septum. It was then charged with 5-bromo-2-propoxy-pyridine **6a** (1.3 g, 6 mmol), Pd(OAc)₂ (67.4 mg, 0.3 mmol), X-Phos (214 mg, 0.45 mmol) and Cs₂CO₃ (2.34 g, 7.2 mmol). The vessel was evacuated and backfilled with argon (this process was repeated a total of 3 times). The 2-chloroaniline **4a** (0.8 g, 6.3 mmol) and freeze-thawed toluene (20 mL) was injected into the tube with a degassed syringe under a positive pressure of argon. The rubber septum was replaced with a screw-cap by quickly removing the rubber septum under the flow of argon and the sealed tube was introduced into a pre-heated oil bath at 110 °C. The reaction mixture was maintained at this temperature for 15 h. At the end of this time period, the pressure tube was allowed to cool to rt. The reaction mixture was filtered through a short pad of celite, and the pad was washed with ethyl acetate (until no more product could be obtained; ≈100 mL; TLC, silica gel). The combined organic fractions were washed with water (100 mL), brine (100 mL), dried (Na₂SO₄) and concentrated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 20 : 1 hexanes/ethyl acetate) to afford **7a** (0.64 g, 81%) as a pale yellow oil: ¹H NMR (300 MHz, CDCl₃) δ 8.04 (d, *J* = 2.6 Hz, 1H), 7.47 (dd, *J* = 8.8, 2.8 Hz, 1H), 7.33 (dd, *J* = 7.9, 1.4 Hz, 1H), 7.12–7.02 (m, 1H), 6.84 (dd, *J* = 8.2, 1.3 Hz, 1H), 6.74 (dd, *J* = 11.5, 5.1 Hz, 2H), 5.88 (br, 1H), 4.24 (t, *J* = 6.7 Hz, 2H), 1.90–1.72 (m, 2H), 1.04 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 161.2, 142.1, 141.9, 135.3, 131.0, 129.7, 127.6, 120.1, 119.5, 113.5, 111.4, 67.9, 22.4, 10.6; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₆ClN₂O: 263.0951, found: 263.0958.

N-(2-Chlorophenyl)-6-isopropoxy-pyridin-3-amine (**7b**)

Following the general procedure, 5-bromo-2-isopropoxy-pyridine **6b** (0.44 g, 2.0 mmol) with 2-chloroaniline **4a** (0.268 g, 2.1 mmol), Pd(OAc)₂ (22.4 mg, 0.1 mmol), X-Phos (71.4 mg, 0.15 mmol), and Cs₂CO₃ (0.78 g, 2.4 mmol) were heated to 110 °C in toluene. After flash chromatography (silica gel, 20 : 1 hexane/ethyl acetate), the process afforded **7b** (0.215 g, 82%) as a pale yellow oil: ¹H NMR (300 MHz, CDCl₃) δ 8.06 (d, *J* = 2.7 Hz, 1H), 7.47 (dd, *J* = 8.7, 2.8 Hz, 1H), 7.34 (dd, *J* = 7.9, 1.3 Hz, 1H), 7.13–7.04 (m, 1H), 6.87 (dd, *J* = 8.2, 1.2 Hz, 1H), 6.81–6.67 (m, 2H), 5.90 (br, 1H), 5.36–5.24 (m, 1H), 1.39 (d, *J* = 6.2 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 160.6, 142.3, 142.0, 135.3, 130.7, 129.6, 127.6, 120.1, 119.5, 113.5, 111.9, 68.2, 22.1; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₆ClN₂O: 263.0951, found: 263.0935.

tert-Butyl 5-[(2-chlorophenyl)amino]picolinate (**7c**)

Following the general procedure, *tert*-butyl 5-bromopicolinate **6c** (5 g, 19.4 mmol) with 2-chloroaniline **4a** (2.6 g, 20.3 mmol),



Pd(OAc)₂ (0.22 g, 0.97 mmol), X-Phos (0.69 g, 1.45 mmol), and Cs₂CO₃ (7.59 g, 23.3 mmol) was heated to 110 °C in toluene. After flash chromatography (silica gel, 5:1 hexanes/ethyl acetate), this process afforded **7c** (5.02 g, 85%) as an off-white solid; mp 148–149 °C: ¹H NMR (300 MHz, CDCl₃) δ 8.54 (d, *J* = 2.7 Hz, 1H), 8.00 (d, *J* = 8.6 Hz, 1H), 7.45 (dd, *J* = 8.1, 1.7 Hz, 2H), 7.39 (dd, *J* = 8.1, 1.2 Hz, 1H), 7.27–7.20 (m, 1H), 7.02 (td, *J* = 7.9, 1.4 Hz, 1H), 6.45 (br, 1H), 1.65 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 163.8, 141.5, 141.3, 139.3, 137.3, 130.3, 127.7, 125.9, 124.6, 123.7, 122.6, 118.8, 81.9, 28.2; HRMS (ESI-TOF) (*m/z*): [M + Na]⁺ calcd for C₁₆H₁₇ClN₂O₂Na: 327.0876, found: 327.0857.

N-(3-Chloropyridin-4-yl)-6-propoxy-pyridin-3-amine (**7d**)

Following the general procedure for 24 h at 140 °C, 5-bromo-2-propoxy-pyridine **6a** (13.45 g, 62.50 mmol) was heated with 4-amino-3-chloropyridine **4b** (8.0 g, 62.5 mmol), Pd(OAc)₂ (697 mg, 3.1 mmol), X-Phos (1.46 g, 3.1 mmol) and Cs₂CO₃ (40.6 g, 125 mmol) in refluxing toluene to yield the crude diaza material **7d**. After flash chromatography (silica gel, 1:1 ethyl acetate/hexane), this afforded the pure diaza material **7d** (8.29 g, 51%) as a white solid; mp 71.6–72.6 °C: ¹H NMR (300 MHz, CDCl₃): δ 8.35 (s, 1H), 8.10–8.07 (m, 2H), 7.48 (dd, *J* = 6.0, 3.0 Hz, 1H), 6.80 (d, *J* = 6.0 Hz, 1H), 6.60 (d, *J* = 6.0 Hz, 1H), δ 6.45 (br, 1H), 4.25 (t, *J* = 6.9 Hz, 2H), 1.87–1.75 (m, 2H), 1.03 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 162.3, 148.8, 148.3, 148.2, 144.0, 136.56, 128.2, 117.1, 111.7, 106.9, 68.0, 22.3, 10.5; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₅ClN₃O 264.0904, found 264.0893.

N-(3-Chloropyridin-4-yl)-6-isopropoxy-pyridin-3-amine (**7e**)

Following the general procedure for 24 h at 140 °C, 5-bromo-2-isopropoxy-pyridine **6b** (8.09 g, 37.20 mmol) was heated with 4-amino-3-chloropyridine **4b** (4.74 g, 37.20 mmol), Pd(OAc)₂ (419 mg, 1.87 mmol), X-Phos (608 mg, 1.87 mmol), Cs₂CO₃ (15.25 g, 46.80 mmol) in refluxing toluene to afford a crude solid which was purified by flash chromatography (silica gel, 1:1 ethyl acetate/hexane) to furnish a white solid **7e** (5.20 g, 52.4%); mp 76–78 °C: ¹H NMR (300 MHz, CDCl₃): δ 8.35 (s, 1H), 8.10–8.06 (m, 2H), 7.46 (dd, *J* = 6.0, 3.0 Hz, 1H), 6.74 (d, *J* = 6.0 Hz, 1H), 6.61 (d, *J* = 6.0 Hz, 1H), 6.42 (br s, 1H), 5.35–5.23 (m, 1H), 1.36 (d, *J* = 6.0 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃): δ 161.7, 148.7, 148.3, 144.0, 136.5, 127.9, 117.0, 112.2, 106.9, 68.5, 22.0; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₅ClN₃O 264.0904, found 264.0909.

General procedure for the intramolecular Heck cyclization: representative procedure for the synthesis of 3-propoxy-9*H*-pyrido[3,4-*b*]indole (3-PBC, **1**) and 2-propoxy-5*H*-pyrido[3,2-*b*]indole (**9a**)

A heavy-wall pressure tube was equipped with a magnetic stir bar and fitted with a rubber septum and loaded with *N*-(2-chlorophenyl)-6-propoxy-pyridin-3-amine **7a** (526 mg, 2.0 mmol), Pd(OAc)₂ (44.8 mg, 0.2 mmol), (*t*-Bu)₃P-HBF₄ (116 mg, 0.4 mmol) and K₂CO₃ (552 mg, 4.0 mmol). The vessel was evacuated and backfilled with argon (this process was

repeated a total of 3 times) and degassed DMA (8 mL) was injected into the tube with a degassed syringe under a positive pressure of argon. The rubber septum was replaced with a screw-cap by quickly removing the rubber septum under the flow of argon and the sealed tube was introduced into a pre-heated oil bath at 120 °C. The reaction mixture was maintained at this temperature for 16 h. At the end of this period, the reaction mixture was allowed to cool to rt. The dark brown mixture which resulted was then passed through a short pad of celite. The celite pad was further washed with ethyl acetate (150 mL) until no more product (TLC; silica gel) was detected in the eluent. The combined filtrate was washed with water (100 mL × 3), brine (100 mL), dried (Na₂SO₄) and concentrated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 5:1 hexanes/ethyl acetate) to afford 3-PBC (**1**) (235 mg, 52%) as an off white solid. mp 120.5–121.5 °C (lit.³⁵ mp 119.3–120.5 °C): **1**, ¹H NMR (300 MHz, CDCl₃) δ 8.66 (br, 1H), 8.42 (s, 1H), 8.05 (d, *J* = 7.9 Hz, 1H), 7.50 (t, *J* = 7.6 Hz, 1H), 7.45–7.38 (m, 1H), 7.35 (s, 1H), 7.21 (t, *J* = 7.4 Hz, 1H), 4.28 (t, *J* = 6.7 Hz, 2H), 1.94–1.78 (m, 2H), 1.06 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.9, 142.4, 133.8, 132.7, 128.9, 128.7, 122.0, 121.4, 119.5, 111.5, 99.1, 68.6, 22.7, 10.6; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₅N₂O: 227.1184, found: 227.1174. A hydrochloride salt of **1** was prepared by the reported method³¹ to obtain 3-PBC-HCl (**1·HCl**): yellow solid; mp 194.5–195.5 °C (lit.³¹ 194.0–195.0 °C). The spectral data for this **1·HCl** were in excellent agreement with the reported values (mp, ¹H NMR).³¹

9a (145 mg, 32%) as a white solid; mp 125–126 °C: ¹H NMR (300 MHz, CDCl₃) δ 8.28 (t, *J* = 8.8 Hz, 1H), 8.20 (br, 1H), 7.60 (d, *J* = 8.7 Hz, 1H), 7.51–7.34 (m, 2H), 7.27 (t, *J* = 7.3 Hz, 1H), 6.83 (d, *J* = 8.7 Hz, 1H), 4.46 (t, *J* = 6.7 Hz, 2H), 1.99–1.80 (m, 2H), 1.10 (t, *J* = 7.4 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.5, 140.2, 138.2, 128.4, 126.8, 122.3, 121.6, 120.6, 119.7, 111.3, 108.6, 67.9, 22.6, 10.7; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₅N₂O: 227.1184, found: 227.1180.

3-Isopropoxy-9*H*-pyrido[3,4-*b*]indole (3-ISOPBC, **2**) and 3-isopropoxy-5*H*-pyrido[3,2-*b*]indole (**9b**)

Following the general procedure for the intramolecular Heck cyclization, **7b** (526 mg, 2.0 mmol) was heated with Pd(OAc)₂ (45 mg, 0.2 mmol), (*t*-Bu)₃P-HBF₄ (116 mg, 0.4 mmol) and K₂CO₃ (552 mg, 4.0 mmol) in DMA at 120 °C to afford a mixture of regioisomers **2** and **9b**. After flash chromatography (silica gel, 5:1 hexanes/ethyl acetate), this procedure yielded pure 3-ISOPBC (**2**) and the byproduct **9b**.

2 (239.5 mg, 53%): off-white solid; mp 134–136 °C: ¹H NMR (300 MHz, CDCl₃) δ 8.41 (s, 1H), 8.19 (br, 1H), 8.04 (d, *J* = 7.8 Hz, 1H), 7.50 (t, *J* = 7.6 Hz, 1H), 7.40 (d, *J* = 8.1 Hz, 1H), 7.34 (s, 1H), 7.21 (t, *J* = 7.4 Hz, 1H), 5.35–5.23 (m, 1H), 1.40 (d, *J* = 6.1 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 157.4, 142.1, 133.7, 132.5, 128.9, 128.8, 122.0, 121.6, 119.5, 111.3, 100.5, 68.6, 22.3; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₅N₂O: 227.1184, found: 227.1184. A hydrochloride salt of **2** was prepared by the reported method³⁷ to obtain 3-ISOPBC-HCl (**2·HCl**): light greenish yellow solid; mp



169–171 °C (lit.³⁷ 168–172 °C). The data for this compound matched in all respects (¹H NMR, mp) with that reported in the literature.³⁷

9b (163.1 mg, 36%): light brown solid; mp 110.4–111.5 °C: ¹H NMR (300 MHz, CDCl₃) δ 8.27 (d, *J* = 7.8 Hz, 1H), 7.99 (br, 1H), 7.67 (d, *J* = 8.7 Hz, 1H), 7.49–7.45 (m, 2H), 7.30–7.25 (m, 1H), 6.79 (d, *J* = 8.7 Hz, 1H), 5.60–5.48 (m, 1H), 1.45 (d, *J* = 6.1 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 158.9, 140.1, 138.4, 128.1, 126.7, 122.6, 121.3, 120.5, 119.7, 111.1, 109.4, 67.9, 22.2; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₄H₁₅N₂O: 227.1184, found: 227.1185.

tert-Butyl 9*H*-pyrido[3,4-*b*]indole-3-carboxylate (**βCct**; **3**) and *tert*-butyl 5*H*-pyrido[3,2-*b*]indole-3-carboxylate (**9c**)

Following the general procedure for the intramolecular Heck cyclization, **7c** (2 g, 16.4 mmol), was heated with Pd(OAc)₂ (147 mg, 0.656 mmol), (*t*-Bu)₃P·HBF₄ (380 mg, 0.4 mmol) and K₂CO₃ (1.8 g, 13.12 mmol) in DMA at 120 °C to afford crude **3** and **9c**. After flash chromatography (silica gel, 1:1 hexanes/ethyl acetate), this afforded pure **βCct** (**3**) and **9c**.

3 (885 mg, 50%), white solid; mp 302.5–304.5 °C (lit.³³ 301–303 °C): ¹H NMR (300 MHz, CDCl₃) δ 10.35 (br, 1H), 9.23 (s, 1H), 8.86 (s, 1H), 8.25 (d, *J* = 7.9 Hz, 1H), 7.80 (d, *J* = 8.3 Hz, 1H), 7.66–7.61 (m, 1H), 7.38 (t, *J* = 7.5 Hz, 1H), 1.75 (s, 9H); ¹³C NMR (75 MHz, CD₃COCD₃) δ 164.9, 141.2, 139.2, 137.7, 133.4, 128.6, 128.1, 121.8, 121.5, 120.3, 116.9, 112.2, 80.1, 27.6; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₆H₁₇N₂O₂: 269.1290, found: 269.1286. The spectral data are in excellent agreement with the published values.³³

9c (531 mg, 30%), fluffy white solid; mp 216.0–218.2 °C: ¹H NMR (300 MHz, CDCl₃) δ 9.46 (br, 1H), 8.38 (d, *J* = 7.8 Hz, 1H), 8.18 (d, *J* = 8.5 Hz, 1H), 7.82 (d, *J* = 8.5 Hz, 1H), 7.53–7.49 (m, 2H), 7.25–7.23 (m, 1H), 1.67 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 164.9, 142.4, 141.4, 141.1, 134.7, 128.6, 122.0, 121.9, 121.0, 120.8, 117.4, 111.5, 81.9, 28.2; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₆H₁₇N₂O₂: 269.1290, found: 269.1289.

8-Propoxy-5*H*-pyrrolo[2,3-*c*:4,5-*c'*]dipyridine (6-aza-3-PBC, **4**) and 2-propoxy-5*H*-pyrrolo[3,2-*b*:4,5-*c'*]dipyridine (**9d**)

Following the general procedure for the intramolecular Heck cyclization, the diaza compound **7d** (3.0 g, 11.30 mmol) was heated with Pd(OAc)₂ (255.0 mg, 1.13 mmol), (*t*-Bu)₃P·HBF₄ (657.0 mg, 2.26 mmol) and K₂CO₃ (3.2 g, 22.60 mmol) in DMA at 120 °C to afford crude **4** and **9d**. After flash chromatography (silica gel, 1:24 methanol/dichloromethane) this process afforded the pure regioisomers 6-aza-3-PBC (**4**) and **9d** as white solids.

4 (820 mg, 31.8%): mp 166–168 °C: ¹H NMR (300 MHz, CDCl₃) δ 12.13 (br, 1H), 9.51 (s, 1H), 8.57 (br, 2H), 7.68 (s, 1H), 7.61 (d, *J* = 5.7 Hz, 1H), 4.26 (t, *J* = 6.0 Hz, 2H), 1.83–1.71 (m, 2H), 1.01 (t, *J* = 6.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 158.4, 147.0, 144.5, 143.4, 133.0, 131.5, 130.7, 118.2, 108.1, 100.4, 68.0, 22.5, 10.9; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₄N₃O: 228.1137, found: 228.1144.

9d (1.62 g, 62.5%): mp 192–194 °C: ¹H NMR (300 MHz, CDCl₃) δ 11.70 (s, 1H), 9.25 (s, 1H), 8.43 (d, *J* = 6.0 Hz, 1H),

7.94 (d, *J* = 9.0 Hz, 1H), 7.51 (d, *J* = 6.0 Hz, 1H), 6.93 (d, *J* = 9.0 Hz, 1H), 4.36 (t, *J* = 6.0 Hz, 2H), 1.84–1.77 (m, 2H), 1.03 (t, *J* = 6.0 Hz, 3H); ¹³C NMR (75 MHz, (CD₃)₂SO): δ 159.7, 145.2, 143.8, 142.8, 136.0, 128.8, 123.5, 118.5, 110.2, 107.6, 67.4, 22.4, 11.0; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₄N₃O: 228.1137, found: 228.1140.

8-Isopropoxy-5*H*-pyrrolo[2,3-*c*:4,5-*c'*]dipyridine (6-aza-3-ISOPBC, **5**) and 2-isopropoxy-5*H*-pyrrolo[3,2-*b*:4,5-*c'*]dipyridine (**9e**)

Following the general procedure for the intramolecular Heck cyclization, pyridine **7e** (3.0 g, 11.30 mmol) was heated with Pd(OAc)₂ (255.0 mg, 1.13 mmol), (*t*-Bu)₃P·HBF₄ (657.0 mg, 2.26 mmol) and K₂CO₃ (3.2 g, 22.60 mmol) in DMA at 120 °C to afford crude **5** and **9e**. After flash chromatography (silica gel, 1:24 methanol/dichloromethane) this afforded the regioisomeric 6-aza-3-ISOPBC (**5**) and **9e** as white solids.

5 (800 mg, 31.0%): mp 180.2–183.2 °C: ¹H NMR (300 MHz, CDCl₃) δ 11.66 (s, 1H), 9.37 (s, 1H), 8.51 (s, 1H), 8.48 (d, *J* = 6.0 Hz, 1H), 7.56 (s, 1H), 7.46 (d, *J* = 6.0 Hz, 1H), 5.32–5.20 (m, 1H), 1.32 (d, *J* = 6.0 Hz, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 157.4, 147.3, 146.3, 145.4, 132.7, 131.6, 130.1, 118.2, 107.4, 100.0, 68.0, 22.6; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₄N₃O: 228.1137, found: 228.1150.

9e (1.6 g, 62.3%): mp 207.4–208.6 °C: ¹H NMR (500 MHz, CDCl₃) δ 11.85 (s, 1H), 9.28 (s, 1H), 8.44 (d, *J* = 3.0 Hz, 1H), 7.94 (d, *J* = 6.0 Hz, 1H), 7.55 (d, *J* = 3.0 Hz, 1H), 6.89 (d, *J* = 6.0 Hz, 1H), 5.49–5.41 (m, 1H), 1.36 (d, *J* = 3.0 Hz, 6H); ¹³C NMR (125 MHz, CDCl₃) δ 159.2, 144.2, 144.0, 142.1, 136.0, 128.9, 123.7, 111.1, 107.8, 67.8, 22.4; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₃H₁₄N₃O: 228.1137, found: 228.1140.

tert-Butyl (2-chlorophenyl)(6-isopropoxy-pyridin-3-yl)-carbamate (**10**)

To the amine **7b** (275 mg, 1.05 mmol) in THF (6 mL) was added the di-*tert*-butyl dicarbonate (320 mg, 1.46 mmol) and 4-(dimethylamino)pyridine (DMAP) (51.1 mg, 0.42 mmol) and this mixture was stirred at rt for 24 h. The organic solvent was removed under reduced pressure and the crude product which resulted was purified by flash column chromatography (silica gel, 1:9 ethylacetate/hexane) to give the pure BOC protected amine **10** (323 mg, 85%).

¹H NMR (300 MHz, CDCl₃) δ 8.03 (d, *J* = 2.6 Hz, 1H), 7.60 (s, 1H), 7.44 (dd, *J* = 8.1, 5.9 Hz, 1H), 7.32–7.20 (m, 3H), 6.62 (d, *J* = 8.9 Hz, 1H), 5.30–5.16 (m, 1H), 1.43 (s, 9H), 1.31 (d, *J* = 6.2 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 160.9, 153.3, 143.9, 139.8, 136.7, 133.3, 132.2, 130.4, 130.3, 128.6, 127.7, 111.1, 81.6, 68.2, 28.1; HRMS (ESI-TOF) (*m/z*): [M + H]⁺ calcd for C₁₉H₂₄ClN₂O₃: 363.1475, found: 363.1469.

(9*H*-Fluoren-9-yl)methyl (2-chlorophenyl)(6-isopropoxy-pyridin-3-yl)carbamate (**11**)

The microwave tube was loaded with amine **7b** (300 mg, 1.14 mmol) and Fmoc chloride (325 mg, 1.25 mmol). The tube was sealed and placed into a microwave apparatus (with a



power of 100 W) at 80 °C for 1 h with stirring. At the end of this period, the reaction was directly purified, without quenching, by flash column chromatography (silica gel, 1 : 4 ethylacetate/hexane) to give pure Fmoc protected pyridine **11** (360 mg, 65%).

$^1\text{H NMR}$ (300 MHz, CDCl_3) δ 8.08 (d, $J = 2.7$ Hz, 1H), 7.70 (d, $J = 7.6$ Hz, 3H), 7.50 (d, $J = 3.8$ Hz, 1H), 7.42–7.28 (m, 5H), 7.20–7.06 (m, 4H), 6.64 (d, $J = 8.5$ Hz, 1H), 5.33–5.15 (m, 1H), 4.49–4.41 (m, 2H), 4.16–4.09 (m, 1H), 1.34 (d, $J = 6.1$ Hz, 6H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 154.3, 143.6, 141.3, 139.2, 139.1, 131.6, 130.60, 130.5, 129.2, 127.9, 127.7, 126.9, 125.0, 119.9, 111.4, 68.4, 68.2, 46.9, 22.1; HRMS (ESI-TOF) (m/z): $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{29}\text{H}_{25}\text{ClN}_2\text{O}_3\text{Na}$: 507.1451, found: 507.1448.

4-Chloro-6-isopropoxy-*N*-phenylpyridin-3-amine (16)

A heavy-wall pressure tube was equipped with a magnetic stir bar and fitted with a rubber septum that had been charged with 4-chloro-5-iodo-2-isopropoxy-pyridine **14** (75 mg, 0.252 mmol), aniline (27.6 μL , 0.256 mmol) and Cs_2CO_3 (410 mg, 1.26 mmol). The vessel was evacuated and backfilled with argon (this process was repeated a total of 3 times) and degassed toluene (1 mL) was injected into the tube with a degassed syringe under a positive pressure of argon. In another round bottom flask fitted with a rubber septum, $\text{Pd}(\text{OAc})_2$ (1.7 mg, 0.0076 mmol) and *rac*-BINAP (4.7 mg, 0.0076 mmol) was charged. This flask was evacuated and backfilled with argon (this process was repeated a total of 3 times) and then degassed toluene (0.5 mL) was added under a positive pressure of argon. This mixture was stirred for 10 min and then the mixture which resulted was added to the above pressure tube. The rubber septum was replaced with a screw-cap by quickly removing the rubber septum under the flow of argon and the sealed tube was introduced into a pre-heated oil bath at 110 °C. The reaction mixture was maintained at this temperature for 5 h. At the end of this time period the pressure tube was allowed to cool to rt. The reaction mixture was filtered through a short pad of celite, and the pad was washed with ethyl acetate (until no more product could be obtained; ≈ 50 mL). The combined organic eluents were washed with water (50 mL), brine (50 mL), dried (Na_2SO_4) and concentrated under reduced pressure. The crude product was purified by flash column chromatography (silica gel, 20 : 1 hexanes/ethyl acetate) to afford only **16** (61 mg, 92%) as a pale yellow oil.

$^1\text{H NMR}$ (300 MHz, CDCl_3) δ 8.18 (s, 1H), 7.28 (t, $J = 7.9$ Hz, 2H), 6.98–6.92 (m, 3H), 6.83 (s, 1H), 5.52 (s, 1H), 5.29–5.17 (m, 1H), 1.36 (d, $J = 6.2$ Hz, 6H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 159.2, 143.5, 139.1, 138.2, 130.4, 129.5, 121.1, 116.8, 111.9, 68.8, 22.1; HRMS (ESI-TOF) (m/z): $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{16}\text{ClN}_2\text{O}$: 263.0951, found: 263.0958.

4-Chloro-6-propoxy-*N*-phenylpyridin-3-amine (15)

Following the above general procedure for 5 h at 110 °C, 4-chloro-5-iodo-2-propoxy-pyridine **13** (75 mg, 0.252 mmol), aniline (27.6 μL , 0.256 mmol), $\text{Pd}(\text{OAc})_2$ (1.7 mg, 0.0076 mmol), *rac*-BINAP (4.7 mg, 0.0076 mmol) and Cs_2CO_3 (410 mg, 1.26 mmol) were heated in toluene at reflux to afford

a crude liquid which was purified by flash chromatography (silica gel, 20 : 1 hexanes/ethyl acetate) to furnish a pale yellow oil **15** (60.33 mg, 91%).

$^1\text{H NMR}$ (300 MHz, CDCl_3) δ 8.20 (s, 1H), 7.28 (t, $J = 7.9$ Hz, 2H), 6.98–6.93 (m, 3H), 6.89 (s, 1H), 5.57 (s, 1H), 4.26 (t, $J = 6.6$ Hz, 2H), 1.89–1.77 (m, 2H), 1.06 (t, $J = 7.2$ Hz, 2H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3) δ 159.9, 143.6, 139.3, 138.1, 130.6, 129.5, 121.1, 116.7, 111.4, 68.1, 22.4, 10.5; HRMS (ESI-TOF) (m/z): $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{16}\text{ClN}_2\text{O}$: 263.0951, found: 263.0946.

4-Chloro-6-propoxy-*N*-(pyridin-4-yl)pyridine-3-amine (17)

Following the above general procedure for 6 h at 140 °C, 4-chloro-5-iodo-2-propoxy-pyridine **13** (214 mg, 0.72 mmol), 4-aminopyridine (68.8 mg, 0.73 mmol), $\text{Pd}(\text{OAc})_2$ (4.8 mg, 0.0216 mmol) and *rac*-BINAP (13.4 mg, 0.0216 mmol) as well as Cs_2CO_3 (1.17 g, 3.6 mmol) were heated in toluene at reflux to afford a crude solid which was purified by flash chromatography (silica gel, ethyl acetate) to furnish a white solid **17** (137 mg, 72%); mp 119–120 °C, $^1\text{H NMR}$ (300 MHz, CDCl_3): δ 8.28 (d, $J = 4.8$ Hz, 2H), 8.19 (s, 1H), 6.92 (s, 1H), 6.65 (d, $J = 5.4$ Hz, 2H), 6.18 (br, 1H), 4.27 (t, $J = 6.6$ Hz, 2H), 1.88–1.76 (m, 2H), 1.04 (t, $J = 7.5$ Hz, 2H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3): δ 162.3, 151.7, 150.0, 144.6, 142.0, 126.9, 111.9, 108.9, 68.4, 22.3, 10.5; HRMS (ESI-TOF) (m/z): $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{13}\text{H}_{15}\text{ClN}_3\text{O}$ 264.0904, found 264.0898.

4-Chloro-6-isopropoxy-*N*-(pyridin-4-yl)pyridine-3-amine (18)

Following the above general procedure for 6 h at 140 °C, 4-chloro-5-iodo-2-isopropoxy-pyridine **13** (214 mg, 0.72 mmol), 4-aminopyridine (68.8 mg, 0.73 mmol), $\text{Pd}(\text{OAc})_2$ (4.8 mg, 0.0216 mmol) and *rac*-BINAP (13.4 mg, 0.0216 mmol) as well as Cs_2CO_3 (1.17 g, 3.6 mmol) were heated in toluene at reflux to afford a crude solid which was purified by flash chromatography (silica gel, ethyl acetate) to furnish a white solid **18** (135 mg, 71%); $^1\text{H NMR}$ (300 MHz, CDCl_3): δ 8.24 (d, $J = 4.2$ Hz, 2H), 8.18 (s, 1H), 6.87 (s, 1H), 6.74 (d, $J = 5.7$ Hz, 2H), 5.36–5.23 (m, 1H), 1.37 (d, $J = 6.3$ Hz, 6H); $^{13}\text{C NMR}$ (75 MHz, CDCl_3): δ 161.9, 152.4, 148.9, 144.8, 142.1, 126.4, 112.4, 108.9, 69.3, 22.0; HRMS (ESI-TOF) (m/z): $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{13}\text{H}_{15}\text{ClN}_3\text{O}$ 264.0904, found 264.0910.

3-Propoxy-9*H*-pyrido[3,4-*b*]indole (3-PBC, 1)

Following the general procedure for the Heck cyclization for 16 h at 120 °C, 4-chloro-6-propoxy-*N*-phenylpyridin-3-amine **15** (526 mg, 2.0 mmol), $\text{Pd}(\text{OAc})_2$ (44.8 mg, 0.2 mmol), (*t*-Bu) $_3\text{P}$ -HBF $_4$ (116 mg, 0.4 mmol) and K_2CO_3 (552 mg, 4.0 mmol) were heated to give a solid which was purified by a wash column (silica gel, 5 : 1 hexanes/ethyl acetate) to yield 3-PBC **1** (416.80 mg, 92%).

3-Isopropoxy-9*H*-pyrido[3,4-*b*]indole (3-ISOPBC, 2)

Following the general procedure for the Heck cyclization for 16 h at 120 °C, 4-chloro-6-isopropoxy-*N*-phenylpyridin-3-amine **16** (526 mg, 2.0 mmol), $\text{Pd}(\text{OAc})_2$ (44.8 mg, 0.2 mmol), (*t*-Bu) $_3\text{P}$ -HBF $_4$ (116 mg, 0.4 mmol) and K_2CO_3 (552 mg,



4.0 mmol) were heated to give a solid which was purified by a wash column (silica gel, 5 : 1 hexanes/ethyl acetate) to yield 3-ISOPBC 2 (412.30 mg, 91%).

8-Propoxy-5H-pyrrolo[2,3-c:4,5-c']dipyridine (6-aza-3-PBC, 4)

Following the general procedure for the Heck cyclization for 16 h at 120 °C, 4-chloro-6-propoxy-*N*-(pyridin-4-yl)pyridine-3-amine **17** (125 mg, 0.475 mmol), Pd(OAc)₂ (10.7 mg, 0.047 mmol), (*t*-Bu)₃P·HBF₄ (27.6 mg, 0.095 mmol) and K₂CO₃ (131.3 mg, 0.95 mmol) were heated to give a solid which was purified by a wash column (silica gel, 1 : 24 methanol/dichloromethane) to yield 6-aza-3-PBC **4** (97.15 mg, 90%).

8-Isopropoxy-5H-pyrrolo[2,3-c:4,5-c']dipyridine (6-aza-3-ISOPBC, 5)

Following the general procedure for the Heck cyclization for 16 h at 120 °C, 4-chloro-6-isopropoxy-*N*-(pyridin-4-yl)pyridine-3-amine **18** (125 mg, 0.475 mmol), Pd(OAc)₂ (10.7 mg, 0.047 mmol), (*t*-Bu)₃P·HBF₄ (27.6 mg, 0.095 mmol) and K₂CO₃ (131.3 mg, 0.95 mmol) were heated to give a solid which was purified by a wash column (silica gel, 1 : 24 methanol/dichloromethane) to yield 6-aza-3-ISOPBC **5** (99.31 mg, 92%).

Large-scale synthesis of 3-ISOPBC (2)

Step 1: Synthesis of 4-chloro-6-isopropoxy-*N*-phenylpyridin-3-amine (16). 4-Chloro-5-iodo-2-isopropoxy-pyridine **14** (25 g, 84.03 mmol), aniline (7.65 mL, 84.03 mmol), Pd(OAc)₂ (0.57 g, 2.52 mmol) and *rac*-BINAP (1.57 g, 2.52 mmol) as well as Cs₂CO₃ (136.84 g, 420 mmol) were added to a three-neck flask with a reflux condenser. The flask was evacuated and back-filled with argon. Degassed toluene (300 mL) was added *via* a cannula, and the flask was introduced into a preheated oil bath at 110 °C. After 15 h at 110 °C the reaction mixture was cooled to rt and filtered through a short pad of celite, and the pad was washed with ethyl acetate. The combined organic eluents were washed with water and brine, dried (Na₂SO₄), and concentrated under reduced pressure. The crude product was purified by flash chromatography (silica gel, 20 : 1 hexanes/ethyl acetate) to afford only **16** (19.86 g, 90%) as a pale yellow oil.

Step 2: Synthesis of 3-isopropoxy-9H-pyrido[3,4-*b*]indole (2). A heavy-wall pressure tube was equipped with a magnetic stir bar and fitted with a rubber septum loaded with 4-chloro-6-isopropoxy-*N*-phenylpyridin-3-amine **16** (19.86 g, 75.58 mmol), Pd(OAc)₂ (1.70 g, 7.558 mmol), (*t*-Bu)₃P·HBF₄ (4.39 g, 15.12 mmol) and K₂CO₃ (20.89 g, 151.16 mmol). The vessel was evacuated and backfilled with argon (this process was repeated a total of 3 times) and degassed DMA (200 mL) was added to this vial *via* a cannula. The rubber septum was replaced with a screw-cap by quickly removing the rubber septum under the flow of argon and the sealed tube was introduced into a pre-heated oil bath at 120 °C. The reaction mixture was maintained at this temperature for 16 h. At the end of this period, the reaction mixture was allowed to cool to rt. The dark brown mixture which resulted was then passed through a short pad of celite. The celite pad was further

washed with ethyl acetate until no product (TLC; silica gel) was detected in the eluent. The combined filtrate was washed with water, brine, dried (Na₂SO₄) and concentrated under reduced pressure. The solid product was purified by a wash column (silica gel, 5 : 1 hexanes/ethyl acetate) to afford 3-ISOPBC (**2**) (15.74 g, 92%) as an off white solid.

Acknowledgements

We thank the NIMH (1R01MH096463-01A1), NIAAA (7R01AA-016179) and NINDS (5R01NS076517-03) for generous financial support. The X-ray crystallographic work was supported by NIDA through Interagency Agreement #ADA 12003 with the Naval Research Laboratory (NRL). We also wish to acknowledge the Bradley-Herzfeld Foundation and The Milwaukee Institute for Drug Discovery for support.

References

- R. Cao, W. Peng, Z. Wang and A. Xu, *Curr. Med. Chem.*, 2007, **14**, 479–500.
- P. Venault and G. Chapouthier, *Sci. World J.*, 2007, **7**, 204–223.
- M. Stahre, J. Roeber, D. Kanny, R. D. Brewer and X. Zhang, *Prev. Chronic Dis.*, 2014, **11**, E109.
- A. R. Yang, J. Liu, H. S. Yi, K. T. Warnock, M. Wang, H. L. June Jr., A. C. Puche, A. Elnabawi, W. Sieghart, L. Aurelian and H. L. June Sr., *Front. Neurosci.*, 2011, **5**, 123.
- M. C. Moffett, A. Vicentic, M. Kozel, P. Plotsky, D. D. Francis and M. J. Kuhar, *Biochem. Pharmacol.*, 2007, **73**, 321–330.
- I. Nylander and E. Roman, *Psychopharmacology*, 2013, **229**, 555–569.
- J. N. Jaworski, D. D. Francis, C. L. Brommer, E. T. Morgan and M. J. Kuhar, *Psychopharmacology*, 2005, **181**, 8–15.
- B. A. Johnson and N. Ait-Daoud, *Psychopharmacology*, 2000, **149**, 327–344.
- H. R. Kranzler, *Alcohol Alcohol.*, 2000, **35**, 537–547.
- M. Davies, *J. Psychiatry Neurosci.*, 2003, **28**, 263–274.
- G. Kalsi, C. A. Prescott, K. S. Kendler and B. P. Riley, *Trends Genet.*, 2009, **25**, 49–55.
- G. F. Koob, *Biochem. Pharmacol.*, 2004, **68**, 1515–1525.
- W. Sieghart and M. Ernst, *Curr. Med. Chem.: Cent. Nerv. Syst. Agents*, 2005, **5**, 217–242.
- M. M. Silveri, J. T. Sneider, D. J. Crowley, M. J. Covell, D. Acharya, I. M. Rosso and J. E. Jensen, *Biol. Psychiatry*, 2013, **74**, 296–304.
- S. Kumar, P. Porcu, D. F. Werner, D. B. Matthews, J. L. Diaz-Granados, R. S. Helfand and A. L. Morrow, *Psychopharmacology*, 2009, **205**, 529–564.
- L. Churchill, A. Bourdelais, M. C. Austin, S. J. Lolait, L. C. Mahan, A. M. O'Carroll and P. W. Kalivas, *Synapse*, 1991, **8**, 75–85.
- G. E. Duncan, G. R. Breese, H. E. Criswell, T. J. McCown, J. S. Herbert, L. L. Devaud and A. L. Morrow, *Neuroscience*, 1995, **64**, 1113–1128.



- 18 S. C. Harvey, K. L. Foster, P. F. McKay, M. R. Carroll, R. Seyoum, J. E. Woods 2nd, C. Grey, C. M. Jones, S. McCane, R. Cummings, D. Mason, C. Ma, J. M. Cook and H. L. June, *J. Neurosci.*, 2002, **22**, 3765–3775.
- 19 W. Wisden, D. J. Laurie, H. Monyer and P. H. Seeburg, *J. Neurosci.*, 1992, **12**, 1040–1062.
- 20 J. Liu, A. R. Yang, T. Kelly, A. Puche, C. Esoga, H. L. June Jr., A. Elnabawi, I. Merchenthaler, W. Sieghart, H. L. June Sr. and L. Aurelian, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 4465–4470.
- 21 G. F. Koob and M. Le Moal, *Nat. Neurosci.*, 2005, **8**, 1442–1444.
- 22 G. F. Koob, A. J. Roberts, G. Schulteis, L. H. Parsons, C. J. Heyser, P. Hyytia, E. Merlo-Pich and F. Weiss, *Alcohol. Clin. Exp. Res.*, 1998, **22**, 3–9.
- 23 W. J. McBride and T. K. Li, *Crit. Rev. Neurobiol.*, 1998, **12**, 339–369.
- 24 H. L. June, K. L. Foster, P. F. McKay, R. Seyoum, J. E. Woods II, S. C. Harvey, W. J. A. Eiler, II, C. Grey, M. R. Carroll, S. McCane, C. M. Jones, W. Yin, D. Mason, R. Cummings, M. Garcia, C. Ma, P. V. V. S. Sarma, J. M. Cook and P. Skolnick, *Neuropsychopharmacology*, 2003, **28**, 2124–2137.
- 25 E. D. Cox, T. J. Hagen, R. M. Mckernan and J. M. Cook, *Med. Chem. Res.*, 1995, **5**, 710.
- 26 E. K. Sawyer, C. Moran, M. H. Sirbu, M. Szafir, M. Van Linn, O. Namjoshi, V. V. N. P. B. Tiruveedhula, J. M. Cook and D. M. Platt, *Alcohol. Clin. Exp. Res.*, 2014, **38**, 1108–1117.
- 27 D. S. O'Tousa, K. T. Warnock, L. M. Matson, O. A. Namjoshi, M. V. Linn, V. V. Tiruveedhula, M. E. Halcomb, J. Cook, N. J. Grahame and H. L. June, *Addict. Biol.*, 2015, **20**, 236–247.
- 28 J. K. Rowlett, R. D. Spealman, S. Lelas, J. M. Cook and W. Yin, *Psychopharmacology*, 2003, **165**, 209–215.
- 29 J. Kovacevic, T. Timic, V. V. Tiruveedhula, B. Batinic, O. A. Namjoshi, M. Milic, S. Joksimovic, J. M. Cook and M. M. Savic, *Brain Res. Bull.*, 2014, **104**, 1–6.
- 30 J. Divljakovic, M. Milic, O. A. Namjoshi, V. V. Tiruveedhula, T. Timic, J. M. Cook and M. M. Savic, *Brain Res. Bull.*, 2013, **91**, 1–7.
- 31 M. S. Allen, T. J. Hagen, M. L. Trudell, P. W. Coddington, P. Skolnick and J. M. Cook, *J. Med. Chem.*, 1988, **31**, 1854–1861.
- 32 T. J. Hagen, F. Guzman, C. Schultz, J. M. Cook, P. Skolnick and H. E. Shannon, *Heterocycles*, 1986, **24**, 2845–2855.
- 33 W. Yin, S. Majumder, T. Clayton, S. Petrou, M. L. VanLinn, O. A. Namjoshi, C. Ma, B. A. Cromer, B. L. Roth, D. M. Platt and J. M. Cook, *Bioorg. Med. Chem.*, 2010, **18**, 7548–7564.
- 34 W. Yin, P. V. V. S. Sarma, J. Ma, D. Han, J. L. Chen and J. M. Cook, *Tetrahedron Lett.*, 2005, **46**, 6363–6368.
- 35 O. A. Namjoshi, A. Gryboski, G. O. Fonseca, M. L. Van Linn, Z.-j. Wang, J. R. Deschamps and J. M. Cook, *J. Org. Chem.*, 2011, **76**, 4721–4727.
- 36 M. S. Allen, A. J. LaLoggia, L. J. Dorn, M. J. Martin, G. Costantino, T. J. Hagen, K. F. Koehler, P. Skolnick and J. M. Cook, *J. Med. Chem.*, 1992, **35**, 4001–4010.
- 37 M. S. Allen, Y. C. Tan, M. L. Trudell, K. Narayanan, L. R. Schindler, M. J. Martin, C. Schultz, T. J. Hagen, K. F. Koehler, P. W. Coddington, P. Skolnick and J. M. Cook, *J. Med. Chem.*, 1990, **33**, 2343–2357.
- 38 J. M. Cook, M. L. Van Linn and W. Yin, *U. S. Patent*, 8268854, 2012.
- 39 J. M. Bailey, G. Bruton, A. Huxley, P. H. Milner and B. S. Orlek, *PCT Int. Patent*, 2005014571, 2005.
- 40 J.-R. Jansen, M. Fuesslein, W. Hallenbach, O. Ort, C. Arnold, E.-M. Franken, O. Malsam, U. Reckmann, E. Sanwald and U. Goergens, *PCT Int. Patent*, 2009068194, 2009.
- 41 M. Godoi, G. V. Botteselle, J. Rafique, M. S. T. Rocha, J. M. Pena and A. L. Braga, *Asian J. Org. Chem.*, 2013, **2**, 746–749.
- 42 T. Heffron, B. Safina, S. Staben, D. P. Sutherlin, B. Wei, R. Elliott, R. Heald, E. M. Seward, E. Gancia and B. Waskowycs, *US Patent*, 20120245144, 2012.
- 43 M.-N. Birkholz, Z. Freixa and P. W. N. M. van Leeuwen, *Chem. Soc. Rev.*, 2009, **38**, 1099–1118.
- 44 K. T. J. Loones, B. U. W. Maes, G. Rombouts, S. Hostyn and G. Diels, *Tetrahedron*, 2005, **61**, 10338–10348.
- 45 S. Hostyn, B. U. W. Maes, G. Van Baelen, A. Gulevskaya, C. Meyers and K. Smits, *Tetrahedron*, 2006, **62**, 4676–4684.
- 46 C. O. Kappe, *Angew. Chem., Int. Ed.*, 2004, **43**, 6250–6284.
- 47 N. C. Bruno and S. L. Buchwald, *Org. Lett.*, 2013, **15**, 2876–2879.
- 48 B. P. Fors, N. R. Davis and S. L. Buchwald, *J. Am. Chem. Soc.*, 2009, **131**, 5766–5768.
- 49 B. P. Fors, D. A. Watson, M. R. Biscoe and S. L. Buchwald, *J. Am. Chem. Soc.*, 2008, **130**, 13552–13554.
- 50 J. P. Wolfe and S. L. Buchwald, *J. Org. Chem.*, 1997, **62**, 6066–6068.
- 51 B. U. W. Maes, K. T. J. Loones, T. H. M. Jonckers, G. L. F. Lemièrre, R. A. Dommissie and A. Haemers, *Synlett*, 2002, 1995–1998.
- 52 C. Meyers, B. U. W. Maes, K. T. J. Loones, G. Bal, G. L. F. Lemièrre and R. A. Dommissie, *J. Org. Chem.*, 2004, **69**, 6010–6017.
- 53 Y. Sunesson, E. Limé, S. O. Nilsson Lill, R. E. Meadows and P.-O. Norrby, *J. Org. Chem.*, 2014, **79**, 11961–11969.

