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Introduction

PD 404182 (**1a**, 3,4-dihydro-2*H*,6*H*-pyrimido[1,2-*c*][1,3]benzothiazin-6-imine)¹⁻⁴ is an antiviral agent against human hepatitis C virus (HCV),^{5,6} HIV,⁶⁻¹⁰ simian immunodeficiency virus (SIV),⁶ and herpes simplex virus (HSV) (Fig. 1).¹⁰ 3,4-Dihydro-2*H*-benzo[4,5]isothiazolo[2,3-*a*]pyrimidine **2a** with a unique heterocyclic scaffold also exhibits potent anti-HIV activity (EC₅₀ = 0.29 \pm 0.09 μ M) (Fig. 1).¹¹ In our recent studies, we



Fig. 1 Structures of PD 404182 (1a) and benzo[4,5]isothiazolo[2,3-a]-pyrimidine (2a).

Investigations of possible prodrug structures for 2-(2-mercaptophenyl)tetrahydropyrimidines: reductive conversion from anti-HIV agents with pyrimidobenzothiazine and isothiazolopyrimidine scaffolds[†]

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3,4-Dihydro-2*H*,6*H*-pyrimido[1,2-c][1,3]benzothiazin-6-imine (PD 404182) and 3,4-dihydro-2*H*-benzo-[4,5]isothiazolo[2,3-a]pyrimidine are the heterocyclic antiretroviral agents against human immunodeficiency virus type 1 (HIV-1) infection. On the basis of similar structure–activity relationships of anti-HIV activities toward the early-stage of viral infection between these unique scaffolds, the transformations under the bioassay conditions were investigated. The distinctive S–N bond in the isothiazolopyrimidine scaffold was immediately cleaved under reductive conditions in the presence of GSH to generate a thiophenol derivative. A similar rapid conversion of PD 404182 into the same thiophenol derivative was observed, suggesting that pyrimidobenzothiazine and isothiazolopyrimidine scaffolds may work as prodrug forms of the common bioactive thiophenol derivatives.

> revealed that modification of the 9-position of PD 404182 derivatives with hydrophobic aryl groups improved the anti-HIV activity.⁷⁻⁹ For example, the introduction of an aryl (1b, 1c and **1f**) or bromo (**1d**) substituent at the 9-position led to a two- to three-fold increase in the anti-HIV activity compared with compound 1a (Table 1). This improvement was duplicated by the modification of benzo[4,5]isothiazolo[2,3-a]pyrimidine derivatives at the 8-position (2b, 2c, 2d and 2f), which is the equivalent position of the 9-position in pyrimido[1,2-c][1,3]benzothiazin-6-imine. Similarly, introduction of acetamido group at the equivalent positions resulted in a complete loss of the anti-HIV activity of both the PD 404182 derivative (1e) and isothiazolopyrimidine derivative (2e). Furthermore, a complete loss of anti-HIV activity was observed by the introduction of the 8-bromo and 7-bromo groups to the parent compounds (1g and 2g, respectively). These similar structure-activity relationships implied that the same target molecule(s) or mechanism(s) of action could possibly be involved in the anti-HIV activities of PD 404182 and the isothiazolopyrimidine derivatives.

> In addition, heterocyclic compounds **1a** and **2a** both inhibited early-stage HIV infection including virus attachment and membrane fusion to host cells as exemplified in comparison with DS 5000 (adsorption inhibitor)¹² and enfuvirtide (fusion inhibitor).¹³ Although it was suggested that the antiviral effects of compound **1a** are derived from its virucidal effect against viral particles,¹⁴ its antiviral mechanism of action has not yet

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Table 1 Comparison of the structure-activity relationships of PD 404182 derivatives and benzo[4,5]isothiazolo[2,3-a]pyrimidines



 a EC₅₀ values represent the concentration of the compound required to inhibit the HIV-1 infection by 50%. The data were obtained from three independent experiments by the MAGI assay. b EC₅₀ values were obtained by the NCK assay. c EC₅₀ values were obtained by the NCK assay, compound **1a**: 0.30 ± 0.06 μ M; compound **1d**: 0.14 ± 0.02 μ M; compound **1g**: >10 μ M. d Cytotoxicity was observed at 10 μ M.

been sufficiently detailed. In the current study, we investigated the chemical transformations of pyrimido[1,2-c][1,3]benzothiazin-6-imine derivatives **1** and benzo[4,5]isothiazolo[2,3-a]pyrimidine derivatives **2** under the bioassay conditions.

Results and discussion

Synthesis

PD 404182 derivatives 1 and benzo[4,5]isothiazolo[2,3-a]pyrimidine derivatives 2 were synthesized according to the protocols in our previous studies (Scheme 1).^{7,8,11} 2-(2-Mercaptophenyl)tetrahydropyrimidine derivatives 4 were obtained by trifluoroacetic acid (TFA)-mediated ethanolysis of the imino group in pyrimido[1,2-c][1,3]benzothiazin-6-imines 1. The alternative route *via* alkaline hydrolysis of pyrimido[1,2-c][1,3]benzothiazin-6-thione 3 also generated thiophenol derivatives 4. Some thiophenol derivatives 4 could be isolated as stable crystals, such as 2-(2-mercaptophenyl)-, 2-(4-bromo-2-mercaptophenyl)- or 2-(3-bromo-2-mercaptophenyl)-tetrahydropyrimidine (4a, d, g). Thioanisole derivative 7 was obtained by the oxidative amidination of the corresponding benzaldehyde 5.^{15,16} Sulfuric acid-mediated deprotection of *N*-tosylated tetrahydropyrimidine 8, derived from benzaldehyde 6 by the identical protocol,⁷ afforded aniline derivative 9.

Identification of the inactive ingredient by transformation from PD 404182 in aqueous media

Recently, Chamoun-Emanuelli *et al.* reported that the antiviral activity of PD 404182 (**1a**) disappeared during storage under some conditions (for example, in Dulbecco's phosphate-buffered saline (pH 7) at 42 $^{\circ}$ C).¹⁰ During our studies on PD



Scheme 1 Synthesis of benzo[4,5]isothiazolo[2,3-a]pyrimidine and 2-phenyl-1,4,5,6-tetrahydropyrimidine derivatives. Reagents and conditions: (a) TFA, CHCl₃–EtOH or EtOH, rt; (b) NaOH, MeOH–H₂O, reflux; (c) PIDA, CHCl₃ and/or EtOH, rt; (d) 1,3-propanediamine, I₂, K₂CO₃, t-BuOH, 70 °C; (e) conc. H₂SO₄, 100 °C.

404182 derivatives, we also found that the anti-HIV activity from the stock solution of **1a** in DMSO was often attenuated. X-ray crystallography of the predominant component from **1a** revealed that the resulting product was 3,4-dihydro-2*H*,6*H*-



Fig. 2 Transformations of PD 404182 under assay conditions and under reductive conditions in the presence of GSH at 37 °C. EC_{50} values represent the concentration of the compound required to inhibit the HIV-1 infection by 50%. The data were obtained from three independent experiments by the NCK assay.

benzo[*e*]pyrimido[2,1-*b*][1,3]thiazin-6-one **10** (Fig. 2). An apparent rearrangement of the cyclic amidine substructure in **1a** provided pyrimidothiazinone **10** with no anti-HIV activity. In contrast, benzo[4,5]isothiazolo[2,3-*a*]pyrimidine **2a** was stable in acidic and basic solutions, and the assay medium containing fetal calf serum.

Transformation of PD 404182 and benzo[4,5]isothiazolo[2,3-*a*]pyrimidine derivatives under reductive conditions in the presence of GSH

We investigated the conversion of PD 404182 (1a) in the presence of high glutathione (GSH) concentrations that mimic the intracellular environment.¹⁷ Interestingly, the imino group of **1a** decomposed in 10 mM GSH solution at 37 °C, and was converted into thiophenol **4a** within 1 h (Fig. 2 and 3).¹⁸ This thiophenol derivative **4a** exhibited equipotent anti-HIV activity as the parent compound **1a** (EC₅₀ = 0.32 ± 0.06 μ M). The mechanism of conversion from **1a** into thiophenol **4a** could be *via* the release of cyanylated glutathione, which was detected by ESI-MS analysis.¹⁹

Our previous structure-activity relationship study of 1a demonstrated that the 6-imino group and the 7-sulfur atom in 1a were indispensable for the anti-HIV activity (Table 2).⁷ For example, substitution of the imino group in compound 1a with a carbonyl (13) or thiocarbonyl (3a) group resulted in a significant decrease or loss of activity. Neither pyrimido[1,2-c]-[1,3]benzoxazines (14a-c) nor pyrimido[1,2-c]quinazolines (15a-c) exhibited anti-HIV activity (Table 2). In the presence of GSH, the carbonyl (13) and thiocarbonyl (3a) derivatives were slowly converted to compound 4a in 26 and 15% vield, respectively, after 24 h (Table 2, Fig. 3). This is in contrast with the rapid and quantitative conversion from 1a within 1 h, indicating that a more reactive imino group in 1a was a prerequisite for the efficient generation of 4a with potent anti-HIV activity. In the absence of GSH, compound 13 was converted into 4a in only 5% yield for 24 h, and 3a did not produce 4a, indicating that degradation of 13 and 3a was mainly mediated by GSH.²⁰ Compound 14a was unstable in the presence or absence of GSH to give a cyclic carbamate 14b and a phenol derivative 16,^{7,21} which showed no anti-HIV activity. The carbamate 14b and thiocarbamate 14c were also degraded into an inactive phenol 16 in 78 and 71% yield, respectively, in GSH solution over 24 h. The pyrimido-[1,2-c]quinazolines (15a-c) were stable under identical conditions, and thus, did not produce the possible aniline 9. In light of these results, the absent or diminished anti-HIV activity of 3a and 13-15 is likely attributable to decreased conversion to the potent thiophenol 4a or the poor anti-HIV activity of the parent compounds and ring-opened products.



Fig. 3 Transformation of compounds 1a (A), 13 (B) and 3a (C) under reductive conditions in the presence of GSH.^a The concentrations of compounds were calculated by HPLC analysis using previously determined calibration curves.

Table 2 Reactions of PD 404182 derivatives under reductive conditions in the presence of GSH





GSH

<3 min

GSH

1 h

PD 404182 (1a)

Fig. 4 Reactions of potent and inactive pyrimido[1,2-c][1,3]benzothiazin-6-imine and benzo[4,5]isothiazolo[2,3-a]pyrimidine derivatives under reductive conditions in the presence of GSH at 37 °C. EC₅₀ values represent the concentration of the compound required to inhibit the HIV-1 infection by 50%. The data were obtained from three independent experiments by the NCK assay.

^a EC₅₀ values represent the concentration of the compound required to inhibit the HIV-1 infection by 50%. The data were obtained from three independent experiments by the NCK assay. $^b\rm EC_{50}$ values were obtained by the MAGI assay. $^c\rm The$ product yields were calculated by HPLC analysis using previously determined calibration curves. ^d The reactions were also carried out in the absence of GSH: 5% conversion from 13 after 24 h; no reaction from 3a. ^e The accurate product yields of 14b and 16 by GSH-mediated conversion were not calculated because of the chemical instability of the substrate 14a.

We next focused on the isothiazolopyrimidine scaffold of 2a with a characteristic S-N covalent bond, which is contained in several anti-HIV agents against HIV-1 nucleocapsid protein 7 (NCp7).²²⁻²⁴ We assessed the stability of compound 2a under identical GSH-containing conditions (Fig. 4). In 10 mM GSH solution at 37 °C, compound 2a was also immediately converted to a thiophenol derivative 4a with potent anti-HIV activity (<3 min). The two-fold more potent pyrimidobenzothiazine 1d or isothiazolopyrimidine 2d was also converted under identical reductive conditions into the corresponding thiophenol derivative 4d, which has similar anti-HIV activity to those of 1d and 2d.²⁵ The thiophenol derivative 4g, which was obtained by conversion from the inactive pyrimidobenzothiazine 1g or isothiazolopyrimidine 2g under the GSH-mediated conditions, exhibited no anti-HIV activity. Thus, the anti-HIV activities of the PD 404182 derivatives (1a, 1d and 1g) and iso-

thiazolopyrimidine derivatives (2a, 2d and 2g) corresponded, at least in part, to those of the thiophenol derivatives (4a, 4d and 4g). These results suggest that the tricyclic structures of 1a and 2a are the prodrug forms, which could be transformed into 4a under intracellular conditions of high GSH concentrations in host cells. Interestingly, inactive analogues with scaffolds similar to PD 404182 (1a) or compound 2a were not converted into thiophenol 4a (Fig. 5). For example, compound 10,²⁶ thioanisole derivative 7, benzo[4,5]thiazolo[3,2-a]pyrimidine 11 (the structural isomer of 2a),²⁷ and saccharin-like derivative 12²⁸ were completely stable under high concentrations of GSH for 24 hours. Therefore, the pyrimidobenzothiazine scaffold in 1a and the isothiazolopyrimidine scaffold in 2a structurally satisfied two criteria, which would make them good prodrugs of 4a.

Taken together, the structure-activity relationships and antiviral profiles were similar between the PD 404182 derivatives 1 and benzo[4,5]isothiazolo[2,3-a]pyrimidine derivatives 2. Under the reductive conditions corresponding to the intracellular environment, these scaffolds were efficiently converted to the common ring-opened thiophenol derivatives with potent anti-HIV activity. In addition, these two scaffolds inhibited not only HIV-1 infection at an early stage, but also several RNA and DNA viruses as we and others recently reported.5-8,10,11 In light of these findings, it was suggested that PD 404182 and benzo-[4,5]isothiazolo[2,3-a]pyrimidine derivatives may achieve their antiviral activities by acting

13

3a



Fig. 5 Reactions of PD 404182, benzo[4,5]isothiazolo[2,3-a]pyrimidines and the inactive analogues under reductive conditions in the presence of GSH at 37 °C. EC₅₀ values represent the concentration of the compound required to inhibit the HIV-1 infection by 50%. The data were obtained from three independent experiments by the NCK assay.

through an identical intermediate or *via* an interaction with a similar target molecule(s) or mechanism of action in an intracellular compartment. A viral target molecule(s) or mechanism (s) has not yet been determined,¹⁴ but these heterocyclic compounds could conceivably regulate the defense mechanism(s) of the host cells.

Conclusions

We have unveiled the chemical transformations of pyrimido-[1,2-*c*][1,3]benzothiazin-6-imine and benzo[4,5]isothiazolo-[2,3-*a*]pyrimidine scaffolds, which yield compounds with potent antiviral activities. Under the reductive conditions meant to mimic intracellular GSH levels, the covalent S-N bond of the isothiazole substructure in 2a was cleaved to provide the ring-opened 2-(2-mercaptophenyl)tetrahydropyrimidine 4a with potent anti-HIV activity, which was also generated from PD 404182 (1a) under identical conditions. The similar structure-activity relationships and antiviral profiles between compounds 1a and 2a suggested that these could possibly be the prodrug forms for the common bioactive substance 4a that binds with the potential target molecule(s) in host cells.

Experimental section

General

¹H NMR spectra were recorded using a JEOL ECA-500 spectrometer. Chemical shifts are reported in δ (ppm) relative to Me₄Si as an internal standard. ¹³C NMR spectra were referenced to the residual solvent signal. Exact mass (HRMS) spectra were recorded on a JMS-HX/HX 110A mass spectrometer or Shimadzu LC-ESI-IT-TOF-MS equipment. For flash chromatography, Wakogel C-300E (Wako) and CHROMATOREX NH DM1080 (Fuji Silysia) were employed. For analytical HPLC, a Cosmosil $5C_{18}$ -ARII column (4.6 × 250 mm, Nacalai Tesque, Inc.) was employed with a linear gradient of CH₃CN containing 0.1% (v/v) aq. TFA at a flow rate of 1 cm³ min⁻¹, and eluting products were detected by UV at 254 nm. Preparative HPLC was performed using a Cosmosil 5C18-ARII preparative column $(20 \times 250 \text{ mm}, \text{Nacalai Tesque, Inc.})$ with a linear gradient of CH_3CN containing 0.1% (v/v) aq. TFA at a flow rate of 8 cm³ min⁻¹. The purity of the compounds was determined as no less than 95% by combustion analysis or HPLC analysis. Preparation of the compounds 1-3, 4d, 6, 8, 13-15 was already reported from our laboratory.7,8,11,15

2-(2-Mercaptophenyl)-1,4,5,6-tetrahydropyrimidine (4a). Method A: TFA (3.70 cm³, 50.0 mmol) was added to a suspension of PD 404182 (1a) (1.09 g, 5.00 mmol) in CHCl₃ (50.0 cm³) and EtOH (75.0 cm³) dropwise. After being stirred at room temperature for 1 h, the mixture was quenched with Et₃N (7.20 cm³, 50.0 mmol) at 0 °C and was concentrated. The crude product was purified by flash chromatography over NH silica gel with CHCl₃-MeOH (10:0 to 9:1) to give the title compound 4a as yellow crystals (999 mg, quant.); mp 208–210 °C (from *n*-hexane–CHCl₃–MeOH); IR (neat) ν_{max} cm⁻¹: 3185–3092 (SH), 1590 (C=N); $\delta_{\rm H}$ (500 MHz, CD₃OD): 2.05-2.09 (2H, m, CH₂), 3.55 (4H, t, J 5.7, CH₂), 6.87-6.91 (1H, m, Ar), 7.06 (1H, ddd, $J_1 = J_2$ 8.0, J_3 1.7, Ar), 7.28 (1H, dd, J_1 7.7, J_2 1.4, Ar), 7.56 (1H, dd, J_1 8.0, J_2 1.1, Ar); $\delta_{\rm C}$ (125 MHz, DMSOd₆): 18.1, 38.1 (2C), 118.7, 122.8, 126.7, 128.8, 137.8, 158.9, 159.8; HRMS (FAB): m/z calcd for $C_{10}H_{13}N_2S$ [M + H]⁺ 193.0799; found: 193.0801.

Method B: 3,4-dihydro-2*H*,6*H*-pyrimido[1,2-*c*][1,3]benzothiazine-6-thione $3a^{15}$ (1.15 g, 4.90 mmol) was suspended in 0.1 N NaOH in MeOH–H₂O (9:1, 98.0 cm³). After being stirred under reflux for 12 h, the mixture was quenched with 1 N HCl until pH was adjusted to 7. The whole mixture was extracted with CHCl₃–MeOH (95:5), and dried over MgSO₄. After concentration, the resulting solid was recrystallized from Et₂O–CHCl₃–MeOH to give the title compound **4a** as an yellow solid (655 mg, 70%): spectral data were in good agreement with those of the compound that was synthesized using method A.

2-(3-Bromo-2-mercaptophenyl)-1,4,5,6-tetrahydropyrimidine (4g). By using a procedure similar to that described for the preparation of the compound 4a from 1a, compound 1g⁷ (45.0 mg, 0.152 mmol) was converted to the title compound 4g as a colorless solid (29.6 mg, 72%); mp 260–262 °C (from *n*-hexane–CHCl₃–MeOH); IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 2895–2795 (SH), 1626 (C=N); δ_{H} (500 MHz, DMSO-*d*₆): 1.88–1.93 (2H, m, CH₂), 3.44 (4H, t, *J* 5.7, CH₂), 6.58 (1H, dd, $J_1 = J_2$ 7.7, Ar), 7.29 (1H, d, *J* 8.0, Ar), 7.53 (1H, d, *J* 8.0, Ar), 10.85 (2H, br s); δ_{C} (125 MHz, DMSO-*d*₆): 17.8, 38.2 (2C), 117.5, 126.2, 128.1, 132.8, 133.4, 160.1, 160.7; anal. calcd for C₁₀H₁₁BrN₂S: C, 44.29; H, 4.09; N, 10.33. Found: C, 44.12; H, 4.08; N, 10.31.

2-[2-(Methylthio)phenyl]-1,4,5,6-tetrahydropyrimidine (7). 2-(Methylthio)benzaldehyde 5 (90% purity, 0.430 cm³, 3.00 mmol) was reacted with propylenediamine (0.300 cm³, 3.60 mmol), K₂CO₃ (1.24 g, 9.00 mmol) and I₂ (952 mg, 3.75 mmol). The crude compound was purified by flash chromatography over NH silica with CHCl3-MeOH (10:0 to 10:1) followed by recrystallization from n-hexane-CHCl₃ to give the title compound 7 as colorless crystals (294 mg, 48%): mp 110–112 °C; IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1621 (C=N); δ_{H} (500 MHz, CDCl₃): 1.82-1.87 (2H, m, CH₂), 2.43 (3H, s, CH₃), 3.45 (4H, t, J 5.7, CH₂), 4.68 (1H, br s, NH), 7.12 (1H, ddd, J₁ = J_2 7.4, J_3 1.1, Ar), 7.20 (1H, d, J 6.9, Ar), 7.29 (1H, ddd, $J_1 = J_2$ 7.6, J_3 1.5, Ar), 7.36 (dd, J_1 7.4, J_2 1.7, 1H, Ar); $\delta_{\rm C}$ (125 MHz, CDCl₃): 16.0, 20.6, 42.1 (2C), 124.6, 125.5, 128.3, 129.1, 136.7, 137.0, 154.8; HRMS (ESI): m/z calcd for $C_{11}H_{15}N_2S$ [M + H]⁺ 207.0956; found: 207.0954.

2-(2-Aminophenyl)-1,4,5,6-tetrahydropyrimidine (9). A mixture of 2-[2-N-(p-toluenesulfonylamino)phenyl]-1,4,5,6tetrahydropyrimidine 8^7 (494 mg, 1.50 mmol) in conc. H₂SO₄ (10.0 cm³) was stirred at 100 °C for 30 min, the mixture was cooled to 0 °C, and then pH was adjusted to 12-14 with 2 N NaOH. The whole mixture was extracted with CHCl₃, and dried over MgSO₄. After concentration, the resulting solid was recrystallized from *n*-hexane-CHCl₃ to give the title compound 9 as colorless crystals (137 mg, 52%): mp 83-85 °C; IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 2931 (NH₂), 2851 (NH₂), 1620 (C=N); $\delta_{\rm H}$ (500 MHz, CDCl₃): 1.81-1.86 (2H, m, CH₂), 3.48 (4H, t, J 5.7, CH₂), 4.71 (1H, br s, NH), 5.72 (2H, br s, NH₂), 6.61-6.66 (2H, m, Ar), 7.07-7.10 (1H, m, Ar), 7.23-7.26 (1H, m, Ar); $\delta_{\rm C}$ (125 MHz, CDCl₃): 20.7, 42.0 (2C), 116.56, 116.61, 119.3, 126.3, 129.9, 147.1, 155.1; anal. calcd for C10H14N3: C, 68.54; H, 7.48; N, 23.98. Found: C, 68.28; H, 7.57; N, 23.75.

3,4-Dihydro-2H,6H-benzo[e]pyrimido[2,1-b][1,3]thiazin-6one (10). PD 404182 (1a) (20.0 mg, 0.0920 mmol) in DMSO (0.920 cm^3) was dissolved in H₂O (92.0 cm³). After being stirred at 37 °C for 26 h, the mixture was purified by HPLC. The resulting solid was dissolved in EtOAc, and washed with sat. NaHCO₃. The extract was dried over MgSO₄, and concentrated to give the title compound 10 as colorless crystals (5.2 mg, 27%); mp 132–134 °C (from toluene); IR (neat) ν_{max} cm⁻¹: 1662 (C=O), 1608 (C=N); $\delta_{\rm H}$ (500 MHz, CDCl₃): 1.98-2.03 (2H, m, CH₂), 3.57 (2H, t, J 5.7, CH₂), 4.02 (2H, t, J 6.0, CH₂), 7.17 (1H, d, J 8.0, Ar), 7.26-7.29 (1H, m, Ar), 7.47 (1H, ddd, $J_1 = J_2$ 7.7, J_3 1.6, Ar), 8.24 (1H, dd, J_1 8.0, J_2 1.1, Ar); $\delta_{\rm C}$ (125 MHz, CDCl₃): 21.1, 42.3, 46.1, 122.7, 124.0, 126.0, 131.0, 133.2, 134.3, 147.0, 161.7; HRMS (FAB): m/z calcd for $C_{11}H_{11}N_2OS [M + H]^+$ 219.0592; found: 219.0584. Spectral data were in good agreement with those previously reported.²⁶

Crystal structure: C₁₁H₁₀N₂OS, M_w : 218.27, primitive monoclinic, a = 10.7456(6), b = 10.9466(6), c = 16.6473(7) Å, $\beta = 97.422(2)^\circ$, V = 1941.78(17) Å³, space group $P2_1/n$ (no. 14), Z = 8. The data were collected with a Rigaku R-AXIS RAPID diffractometer using graphite monochromated Mo-K α radiation at -93 K. 18 534 Reflections were measured, 4420 unique $(R_{int} = 0.0220)$ which were used in all calculations. The final w*R* was 0.0796. The substance was crystallized from toluene. The CCDC deposition number: 1037500.

3,4-Dihydro-2*H***-benzo[4,5]thiazolo[3,2-***a***]pyrimidine (11). Following the reported procedure,²⁷ 2-chloro-1,3-benzothiazole (0.619 cm³, 5.00 mmol) was converted to the title compound 11**, as a pale yellow solid (407 mg, 2 steps 73%), by the reaction with 3-amino-1-propanol (0.381 cm³, 5.00 mmol). The product (80.8 mg) was purified by HPLC to give the compound **11** as a colorless solid (72.3 mg, TFA salt): IR (neat) ν_{max}/cm^{-1} : 3274–3181 (OH), 1672 (C=O), 1632 (C=N); $\delta_{\rm H}$ (500 MHz, CD₃OD): 2.31–2.36 (2H, m, CH₂), 3.70 (2H, t, *J* 5.7, CH₂), 4.27 (2H, t, *J* 6.0, CH₂), 7.42–7.45 (1H, m, Ar), 7.57–7.61 (2H, m, Ar), 7.84 (1H, d, *J* 8.0, Ar); $\delta_{\rm C}$ (125 MHz, CD₃OD): 19.2, 41.3, 44.0, 113.5, 118.0 (q, *J* 291.5), 123.1, 124.1, 126.7, 129.2, 140.0, 162.4 (q, *J* 36.0), 166.0; HRMS (FAB): *m*/z calcd for C₁₀H₁₁N₂S [M + H]⁺ 191.0643; found: 191.0641.

3,4-Dihydro-2*H*-benzo[4,5]isothiazolo[2,3-*a*]pyrimidine 6,6dioxide (12). 50% H₂O₂ (0.0391 cm³, 0.638 mmol) was added to a suspension of compound 2a (48.5 mg, 0.159 mmol, TFA salt) in TFA (0.610 cm³) dropwise. After being stirred at room temperature for 24 h, the mixture was quenched with Et₃N (1.50 cm^3) , and the whole mixture was extracted with EtOAc. The extract was washed with sat. NaHCO₃, brine, and dried over MgSO₄. After concentration, the residue was purified by flash chromatography over aluminum oxide with n-hexane-EtOAc (3:2 to 1:1) to give the title compound **12** as colorless crystals (25.2 mg, 71%): mp 146-148 °C (from Et₂O-MeOH); IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1670 (C=N); δ_{H} (500 MHz, CDCl₃): 2.01-2.05 (2H, m, CH₂), 3.70 (2H, t, J 5.7, CH₂), 3.79 (2H, t, J 6.0, CH₂), 7.70-7.76 (2H, m, Ar), 7.85-7.88 (1H, m, Ar), 7.98–8.01 (1H, m, Ar); $\delta_{\rm C}$ (125 MHz, CDCl₃): 19.5, 36.5, 44.2, 120.7, 122.8, 129.6, 132.4, 133.6, 134.8, 142.7; anal. calcd for C₁₀H₁₀N₂O₂S: C, 54.04; H, 4.54; N, 12.60. Found: C, 53.89; H, 4.74; N, 12.36.

2-(2-Hydroxyphenyl)-1,4,5,6-tetrahydropyrimidine (16).²¹ A mixture of methyl salicylate (1.29 cm³, 10.0 mmol) and propylenediamine (2.56 cm³, 30.0 mmol) was refluxed for 16 h. The crude product was dissolved in MeOH, and crystallized with Et₂O. The precipitate was filtered, and the unreacted propylenediamine was removed by washing with Et₂O. The resulting solid was purified by flash chromatography over aluminum oxide with $CHCl_3$ -MeOH (10:0 to 10:1) to give the title compound 16 as colorless crystals (497 mg, 28%): mp 259-261 °C (from Et₂O–MeOH); IR (neat) ν_{max} /cm⁻¹: 3212–3050 (OH), 1620 (C=N); $\delta_{\rm H}$ (500 MHz, DMSO- d_6): 1.83–1.88 (2H, m, CH₂), 3.40 (4H, t, J 5.7, CH₂), 6.27-6.30 (1H, m, Ar), 6.46 (1H, d, J 8.6, Ar), 7.04-7.08 (1H, m, Ar), 7.45 (1H, dd, J1 8.0, J2 1.7, Ar), 12.10 (1H, br s, OH); $\delta_{\rm C}$ (125 MHz, CD₃OD): 20.0, 39.3 (2C), 111.2, 114.5, 124.4, 126.3, 135.1, 160.9, 172.5; anal. calcd for C₁₀H₁₂N₂O: C, 68.16; H, 6.86; N, 15.90. Found: C, 68.13; H, 7.03; N, 16.09.

Determination of anti-HIV activity

The anti-HIV activity of a series of compounds against HIV- $1_{\rm IIIB}$ was determined by the NCK assay.²⁹ The target

cells (NCK45- β -Gal; 10⁴ cells per well) were plated in 96-well flat microtiter culture plates. On the following day, the cells were inoculated with HIV-1 (60 NCK U per well, giving 60 blue cells after 48 h of incubation) and cultured in the presence of various concentrations of the test compounds in fresh medium. Forty-eight hours after viral exposure, all the blue cells stained with X-Gal (5-bromo-4-chloro-3-indolyl- β -D-galactopyranoside) were counted in each well. Cytotoxic effects were not observed at 10 μ M except for compound 2g.

The activity of test compounds was determined as the concentration that blocked HIV-1 infection by 50% (50% effective concentration $[EC_{50}]$). EC_{50} was determined by using the following formula:

$$EC_{50} = 10^{[\log(A/B) \times (50-C)/(D-C) + \log(B)]}.$$

where *A*, of the two points on the graph that bracket 50% inhibition, the higher concentration of the test compound; *B*, of the two points on the graph that bracket 50% inhibition, the lower concentration of the test compound; *C*, inhibitory activity (%) at the concentration *B*; *D*, inhibitory activity (%) at the concentration *A*.

Glutathione-mediated transformation of pyrimidobenzothiazine and isothiazolopyrimidine derivatives

The compound (1 mM) was incubated in 10 mM GSH (in a mixture of 50 mM phosphate buffer (pH 7.4) and MeCN [75:25 (v/v) or 1:1 (v/v)]) at 37 °C. The sample was analyzed by HPLC and the peak area was recorded by UV detection at 254 nm. The concentrations of the products were calculated using previously determined calibration curves. Of note, the GSH-mediated transformation from **1a** was slightly accelerated by the addition of MeCN in phosphate buffer.

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