Food & Function

PAPER

The blood pressure effect and related plasma levels of flavan-3-ols in spontaneously hypertensive rats

Mar Quiñones,*†[a] Maria Margalef,[b] Anna Arola-Arnal,[b] Begoña Muguerza,*[b,c] Marta Miguel[d] and Amaya Aleixandre[a]

We studied the short-term antihypertensive effect of flavan-3-ols (−)-epicatechin, (+)-catechin and (−)-catechin, in spontaneously hypertensive rats (SHR). Plasma metabolites and the corresponding plasma antioxidant capacity were determined. All the assayed flavan-3-ols decreased systolic blood pressure (SBP) in SHR. Their antihypertensive effects were less pronounced than that of Captopril (50 mg kg^{-1}) and were not shown in normotensive Wistar-Kyoto rats. 6 mg kg^{-1} (−)-epicatechin caused the maximum decrease in SBP. The maximum effects of the catechin monomers were observed post-administration of 0.5 mg kg^{-1} of flavan-3-ols, (−)-catechin being the least effective among the three assayed compounds. Glucuronide and methyl glucuronide metabolites were obtained in the flavan-3-ol treated SHR, but it was not possible to relate the antihypertensive effect of the assayed flavan-3-ols with a concrete plasma metabolite or with their antioxidant effect. In conclusion, the studied flavan-3-ols could be responsible for the antihypertensive effect of cocoa products.

1. Introduction

In recent years, numerous studies have demonstrated the health benefits of polyphenols, and special attention has been paid to their beneficial effect on hypertension and cardiovascular diseases.[1–3] In particular, flavan-3-ols, also known as flavanols, are the most structurally complex subclass of flavonoids, and have been recognized as antihypertensive agents.[1,4] These compounds are present in different foods such as cocoa, grape, tea, apple and berries.[5] Cocoa, in particular, has a high content in flavan-3-ols.[6,7] Elegant reports have shown the beneficial effect of cocoa on blood pressure.[6,8] Moreover, our research group demonstrated that a polyphenol enriched cocoa powder showed antihypertensive properties,[9] improved endothelial function,[10] and exhibited an antioxidant capacity.[11]

[a]Dpto. Farmacología, Fac. Medicina, U. Complutense, Avda. Complutense s/n, 28040 Madrid, Spain. E-mail: mar.quinones@usc.es; Fax: +34-91-3941463; Tel: +34-91-3941475
[b]Nutrigenomic Group, Department of Biochemistry and Biotechnology, Rovira i Virgili University, Tarragona, Spain. E-mail: begona.muguerza@urv.cat; Fax: +34 977 55 82 32; Tel: +34 977 55 95 66
[c]Centre Tecnològic de Nutrició i Salut (CTNS), TECNIO, CEICS, Reus, Spain
[d]Instituto de Investigación en Ciencias de Alimentación (CIAL, CSIC-UAM, CEI UAM+CSIC), Madrid, Spain
†Present address: Department of Physiology, CIMUS, University of Santiago de Compostela-Instituto de Investigación Sanitaria, Santiago de Compostela, 15782, Spain. CIBER Fisiopatología de la Obesidad y Nutrición (CIBERobn), 15706, Spain.

Cocoa flavan-3-ols consist of a complex mixture of the monomeric (−)-epicatechin and (+)-catechin and the oligomers of these monomeric base units known as procyanidins.[5] It is nevertheless worth nothing that the preservation of flavan-3-ols during cocoa manufacturing is important to exhibit the health effects associated with cocoa consumption, and it has been recently described that roasted cocoa beans and cocoa products additionally contained (−)-catechin. This atypical flavan-3-ol is generally formed during the cocoa manufacturing process by an epimerization which converts (−)-epicatechin to its epimer (−)-catechin. High temperatures during the cocoa bean roasting process, and particularly the alkalization of the cocoa powder, are the main factors inducing the epimerization reaction.[12]

The bioavailability of flavan-3-ols is higher than the bioavailability of other flavonoids, even if it is also relatively poor.[13] Moreover, flavan-3-ols, as other kinds of flavonoids, occur in plasma in more diverse forms than in food.[14,15] In fact, the uptake and metabolism of polyphenols are usually associated with their methylation, sulphation or glucuronidation, afforded by phase-II enzymes.[16] In addition, considerable quantities of ingested flavonoids are degraded by colonic microbiota upon reaching the large intestine, where they yield other smaller molecules that are also absorbed into the body.[17] Therefore, the identification of the specific flavan-3-ol metabolites present in the body is crucial for understanding their biological activities.

Different studies have demonstrated that (−)-epicatechin, (+)-catechin and their oligomers, have important cardiovascular beneficial effects, such as the ability to inhibit LDL oxidation[18] and the capacity to promote endothelium-dependent relaxation.[19] Moreover these compounds can also modulate the production of inflammatory cytokines[20] and can inhibit pro-inflammatory response in *in vitro* systems.[21] Nevertheless, scarce research exists for evaluating the short-term antihypertensive effect of the main cocoa flavan-3-ols. The aim of this study was to characterize the dose dependent short-term antihypertensive effect of (−)-epicatechin, (+)-catechin and the atypical flavan-3-ol (−)-catechin, in spontaneously hypertensive rats (SHR). The corresponding plasma flavan-3-ol compound and its main metabolites in this fluid were quantified. Moreover, since the antioxidant effect of polyphenols enables us to explain many of their health benefits, the corresponding plasma antioxidant effect was also determined.

2. Materials and methods

2.1. Chemicals and reagents

Captopril, (−)-epicatechin, (+)-catechin and (−)-catechin were used for the experiments, and all these compounds were purchased from Sigma Chemical (Fluka/Sigma Aldrich, Madrid, Spain). Captopril was dissolved in water to be administered to the rats. (+)-Catechin, (−)-catechin and (−)-epicatechin were prepared with 10% DMSO in water and sonicated for 45 minutes before administration to the animals.

For the chromatographic analysis, methanol (Scharlab S.L., Barcelona, Spain), acetone (Sigma-Aldrich, Madrid, Spain) and glacial acetic acid (Panreac, Barcelona, Spain) of HPLC analytical grade, were used. Ultrapure water was obtained from a Milli-Q advantage A10 system (Madrid, Spain). The 2000 mg l^{-1} standard stock solutions of (+)-catechin, (−)-epicatechin and pyrocatechol (Sigma Aldrich, Madrid, Spain) in methanol as the internal standard (IS), were stored in dark-glass flasks at −20 °C. A 200 mg l^{-1} stock standard mixture of (+)-catechin and (−)-epicatechin was prepared weekly and stored at −20 °C. The stock standard solution was diluted daily to the desired concentration using an acetone : water : acetic acid (70 : 29.5 : 0.5, v : v : v) solution.

2.2. Experimental procedure in rats

2.2.1. Experiments to evaluate the antihypertensive activity. In this study, we have used thirty 17–22 week-old male SHR, weighing 314 ± 3 g, and ten 17–22 week-old male WKY rats, weighing 337 ± 6 g. All these animals were obtained from Charles River Laboratories España S.A. They were caged in groups of five rats at a temperature of 23 °C with 12 hour light/dark cycles, and they consumed tap water and a standard diet for rats (A04 Panlab, Barcelona, Spain), *ad libitum*, during the experiments.

The assayed flavan-3-ols were orally administered by gastric intubation to the rats and tentative trials were formerly carried out in order to estimate the doses of each monomer that could

be efficient in the SHR. In accordance with these initial trials, we evaluated the effect of four different doses of (−)-epicatechin (1 mg kg^{-1}, 2 mg kg^{-1}, 6 mg kg^{-1} and 12 mg kg^{-1}), four different doses of (+)-catechin (0.25 mg kg^{-1}, 0.5 mg kg^{-1}, 1.5 mg kg^{-1} and 3 mg kg^{-1}) and four different doses of (−)-catechin (0.25 mg kg^{-1}, 0.5 mg kg^{-1}, 1.5 mg kg^{-1} and 3 mg kg^{-1}) in the SHR. Each flavan-3-ol was evaluated by using a minimum of 8 rats and we have always administered increasing doses in the same animal by waiting for at least 4 days between the administrations of two different doses. The most effective dose to lower the arterial blood pressure in the SHR of each flavan-3-ols [6 mg kg^{-1} (−)-epicatechin, 0.5 mg kg^{-1} (+)-catechin and 0.5 mg kg^{-1} (−)-catechin], was also administered to WKY rats. In all cases, 1 ml of the corresponding solution was orally administered by gastric intubation, between 9 and 10 a.m. to the rats. Captopril (50 mg kg^{-1}), a known antihypertensive drug, served as the positive control, and 1 ml 10% DMSO water solution served as the negative control. We measured the SBP of the rats by the tail cuff method[22] before administration and also 2, 4, 6, 8, 24, 48 and 72 hours post-administration. Before the measurement, the rats were maintained at 30 °C for 10 minutes to make the pulsations of the tail artery detectable. The person who measured the arterial blood pressure in the animals did not know either the compound or the dose that had been administered.

2.2.2. Experiments for plasma determination. Fourteen 17–22 week-old male SHR were used for an additional study that was carried out in order to quantify the corresponding flavan-3-ol compound and its main metabolites in plasma, as well as for modification of the plasma antioxidant capacity, after flavan-3-ol administration. For these purposes, the flavan-3-ols were orally administered by gastric intubation, as in the arterial blood pressure trials. Determination was made in the plasma of the rats that had been treated with the most effective dose of each flavan-3-ol, at the moment of maximum SBP decrease [6 hours after 6 mg kg^{-1} (−)-epicatechin; 4 hours after 0.5 mg kg^{-1} (+)-catechin and 6 hours after 0.5 mg kg^{-1} (−)-catechin]. Determination was also made in the plasma of the rats treated with the highest dose of each flavan-3-ol, at the moment of maximum decrease in SBP and also 72 hours post-administration, when SBP had always returned to baseline values [4 and 72 hours after 12 mg kg^{-1} (−)-epicatechin; 6 and 72 hours after 3.0 mg kg^{-1} (+)-catechin; 6 and 72 hours after 3.0 mg kg^{-1} (+)-catechin]. Blood samples were obtained from the saphenous vein by using heparin vials (Starsted, Barcelona, Spain), and the corresponding plasma samples were obtained by centrifugation (2000g, 15 min, 4 °C). They were stored at −80 °C until chromatographic or plasma antioxidant capacity analysis.

2.3. Plasma determination

2.3.1 Plasma flavan-3-ol extraction and quantification. Quantification of flavan-3-ols and their metabolites in plasma was carried out by High-Performance Liquid Chromatography/ Electrospray Ionization coupled with tandem Mass Spectrometry (HPLC-ESI-MS/MS). Plasma samples were first centri-

Table 1 Calibration curve, determination coefficient (R^2) and limits of quantification and detection, for the quantification of flavan-3-ols in the spiked plasma samples by High-Performance Liquid Chromatography/Electrospray Ionization coupled with tandem Mass Spectrometry

Compound	Calibration curve	R^2	LOD (µM)	LOQ (µM)	MDL (µM)	MQL (µM)
(+)-Catechin	$y = 0.0159x$	0.992	0.003	0.009	0.001	0.003
(−)-Epicatechin	$y = 0.0179x$	0.993	0.002	0.008	0.001	0.003

Abbreviations: LOD (limit of detection); LOQ (limit of quantification); MDL (method detection limit); MQL (method quantification limit). The (+)-catechin calibration curve was used to quantify catechin and its metabolites and the (−)-epicatechin calibration curve was used to quantify (−)-epicatechin and its metabolites.

fuged (2000g, 5 min, 4 °C) and the plasma flavan-3-ols and their metabolites were extracted by the off-line µ-solid phase extraction methodology, previously described by Margalef *et al.* in 2014,[23] using 30 µm OASIS HLB µElution Plates (Waters, Barcelona, Spain). Briefly, the micro-cartridges were sequentially conditioned with 250 µl of methanol and 250 µl of 0.2% acetic acid. Plasma (250 µl) was mixed with 300 µl of 4% phosphoric acid and 50 µl of pyrocatechol (2000 ppb), and then loaded onto the plate. The plates were washed with 200 µl of Milli-Q water and 200 µl of 0.2% acetic acid. The retained flavan-3-ols and their metabolites were eluted with 2 × 50 µl of an acetone/Milli-Q water/acetic acid (70 : 29.5 : 0.5, v : v : v) solution. The eluted solutions were then directly injected into the HPLC tandem triple quadrupole mass spectrometer (HPLC-MS/MS) for chromatographic analysis.

The chromatographic analysis was performed using a 1290 Infinity UHPLC coupled with a 6490 QqQ/MS (Agilent Technologies, Palo Alto, CA, USA). The separations were achieved by using a Zorbax SB-Aq (150 mm × 2.1 mm i.d., 3.5 µm of particle size) as a chromatographic column from Agilent Technologies. The mobile phase consisted of 0.2% acetic acid (solvent A) and acetonitrile (solvent B) at a flow rate of 0.4 ml min^{-1}. The elution gradient was 0–10 min, 5–55% B; 10–12 min, 55–80% B; 12–15 min, 80% B isocratic; 15–16 min 80.5% B. A post run of 10 min was applied. The sample volume injected was 2.5 µl. The electrospray ionisation (ESI) conditions were: 150 °C and 14 l min^{-1} of drying gas temperature and flow, respectively, a nebulizer gas pressure of 30 psi, and 3000 V of capillary voltage. MS/MS was operated in a negative mode. MS/MS acquisition was performed in a multiple reaction monitoring (MRM) mode for flavan-3-ols and their metabolites, using the same quantification previously reported by Serra *et al.* (2009).[24] Data acquisition was conducted by using the MassHunter software (Agilent Technologies, Palo Alto, CA, USA).

Spiked blank plasmas with standard compounds at 8 different concentrations were used to obtain calibration curves for quantification. Standard compounds in the samples were quantified by interpolating the analyte/IS peak abundance ratio in these curves. (−)-Epicatechin, (+)-catechin and (−)-catechin metabolites were tentatively quantified by using the standard (−)-epicatechin and (+)-catechin calibration curves respectively. The sensitivity was evaluated by determining the limit of detection (LOD), which is defined as the concentration that corresponds to three times the signal-to-noise ratio, and

the limit of quantification (LOQ), which is defined as the concentration that corresponds to 10 times the signal-to-noise ratio. The method detection and quantification limits (MDL and MQL, respectively) were calculated in the analysis of 250 µl of a sample. Table 1 shows the values that were obtained for each quality parameter.

2.3.2. Plasma antioxidant capacity. To measure the plasma antioxidant capacity we used the oxygen radical absorbance capacity (ORAC) assay, as previously described by Huang *et al.*[25] Briefly, 25 µl of plasma solution and 150 µl of 59.8 nM fluorescein (FL) (Sigma-Aldrich) were added to each well of a 96-well microplate. The fluorescence was measured at λ_{ex} = 485 nm and λ_{em} = 520 nm every min for 90 min in the FLx800 Multi-Detection Microplate Reader (Biotek, Winooski, USA) after the addition by using an injector of 25 µl of 73 mM of the radical generator 2,2′-azobis(2-methylpropionamidine)dihydrochloride (AAPH, Acros Organics, Belgium). As a standard, Trolox (Sigma-Aldrich) solution was used at different concentrations (0, 3.125, 6.25, 12.5, 25, 50, and 100 µM). The final ORAC values were calculated by a regression equation between the Trolox concentration and the net area under the FL decay curve and were expressed as µmol Trolox equivalents (TE)ml^{-1}.

All the above-mentioned experiments were designed and performed in accordance with the European and Spanish legislation on care and use of experimental animals (2010/63/UE; Real Decreto 53/2013), and were approved by the Ethics Committees at Universidad Complutense de Madrid (UCM).

2.4. Statistical analysis

SBP results are expressed as mean values ± standard error of the mean (SEM) for a minimum of 8 rats and the plasma results are expressed as mean values ± SEM for a minimum of 6 rats. They were analyzed by a one- or two-way analysis of variance (ANOVA), using the GraphPad Prism 4 software in the case of SBP values, and using the SPSS software (Version 20.0.0) in the case of plasma values. The differences between the groups were assessed by the Bonferroni test and the differences were always considered to be significant when $p < 0.05$.

3. Results

Before administration of the different products, the SHR showed SBP values of 236 ± 2.5 mmHg; n = 30. The values of

SBP obtained after the oral administration of 1 ml of 10% DMSO solution (used as the negative control), were very similar to those obtained before its administration. On the contrary, the dose of 50 mg kg^{-1} of Captopril (used as the positive control) caused a clear decrease in SBP in the SHR. The maximum decrease in SBP caused by this drug (60.5 ± 2.7 mmHg) was observed 4 hours post-administration, and this variable returned to the baseline 48 hours post-administration. The oral administration of (−)-epicatechin also resulted in a significant decrease of the SBP in the SHR. The maximum effect was attained 6–8 hours post-administration of 6 mg kg^{-1} of this flavan-3-ol. The decreases of SBP at that moment were, respectively, 34.0 ± 4.8 mmHg and 34.3 ± 5.6 mmHg, and SBP remained still lower than the basal SBP value when measurements were made 48 hours post-administration of (−)-epicatechin. Nevertheless, the values of SBP obtained 72 hours post-administration of this flavan-3-ol were similar to those obtained before the administration (Fig. 1A). The administration of (+)-catechin also caused a significant decrease in the SBP of the SHR, and the decrease was maximum 4 hours

post-administration of 0.5 mg kg^{-1} of this flavan-3-ol (29.9 ± 0.8 mmHg). SBP returned to the baseline 48 hours post-administration of this dose of (+)-catechin (Fig. 2A). The administration of (−)-catechin also caused a significant decrease in the SBP of the SHR. Nevertheless, the effect of this flavan-3-ol on this variable was less accentuated than the effect of (−)-epicatechin or (+)-catechin. The more effective dose of (−)-catechin was 0.5 mg kg^{-1}, and the maximum decrease in SBP caused by this dose of this flavan-3-ol was observed 6–8 hours post-administration. At that moment, the decreases in SBP were, respectively, 23.6 ± 5.9 mmHg and 24.5 ± 5.7 mmHg. In addition, we could observe that, 72 hours post-administration of 0.5 mg kg^{-1} of (−)-catechin, the values of SBP were already similar to those obtained before administration (Fig. 3A).

None of the assayed flavan-3-ols modified SBP in the WKY rats. This variable was similar in the WKY rats that were treated with these products and in the WKY rats that were treated with the 10% DMSO solution (Fig. 1B, 2B and 3B).

Fig. 4 shows the different flavan-3-ol signals obtained in the chromatographic analysis of plasma samples. This analysis
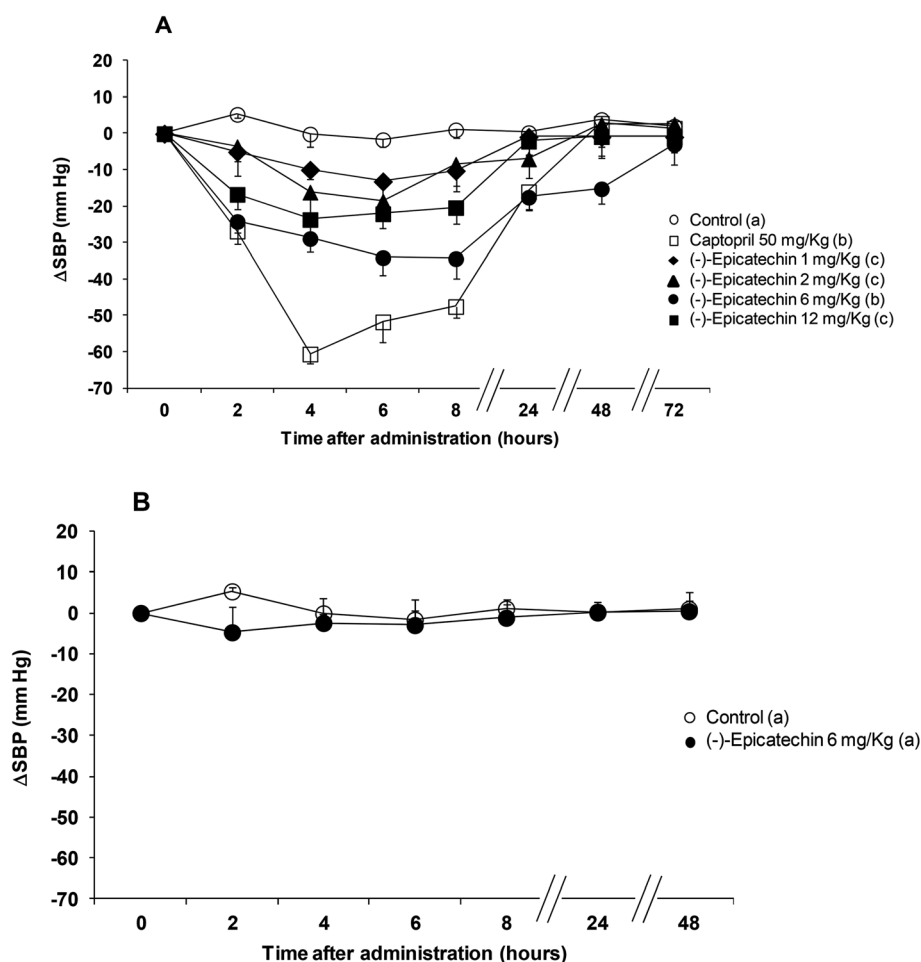

Fig. 1 (A) Decreases in systolic blood pressure (SBP) caused in spontaneously hypertensive rats by the negative control (○), Captopril (50 mg kg^{-1}) (□) or different doses of (−)-epicatechin: 1 mg kg^{-1} (◆), 2 mg kg^{-1} (▲), 6 mg kg^{-1} (●) and 12 mg kg^{-1} (■). (B) Decreases in SBP caused in Wistar-Kyoto rats by the negative control (○) or 6 mg kg^{-1} (−)-epicatechin (●). Data are expressed as mean ± SEM. The experimental groups always have a minimum of 8 animals. Different letters represent the statistical differences ($p < 0.05$). p estimated by two-way ANOVA.
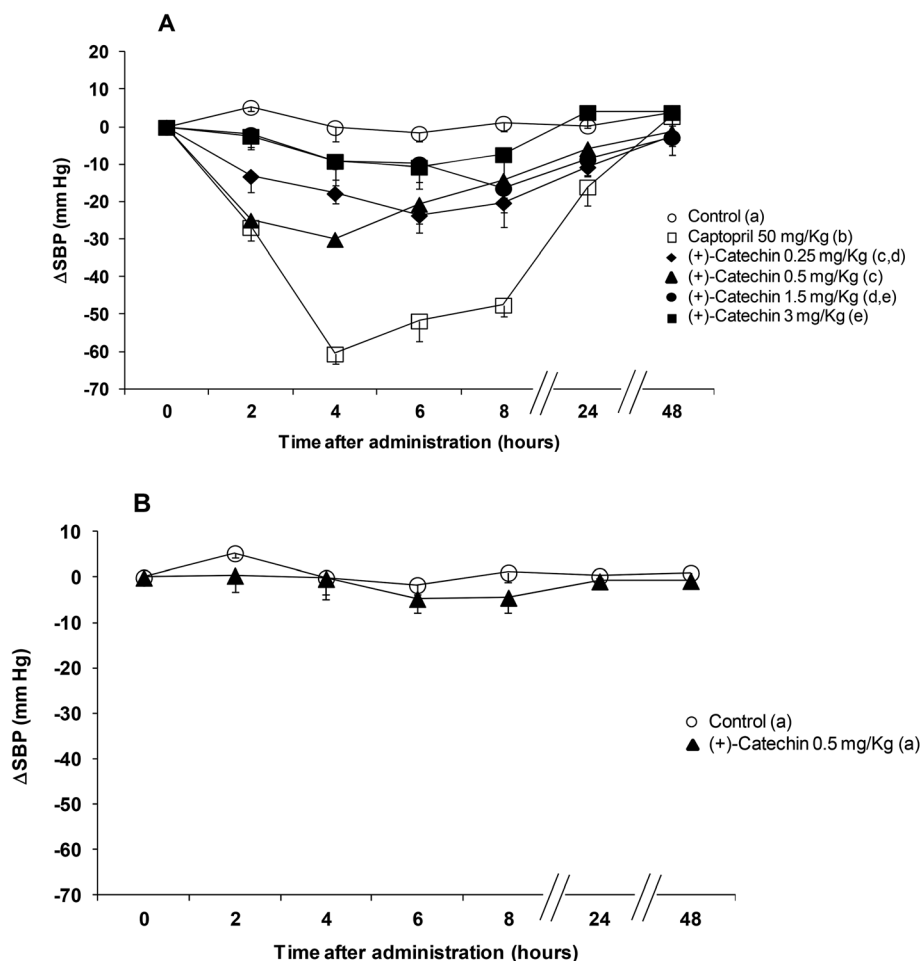
Fig. 2 (A) Decreases in systolic blood pressure (SBP) caused in spontaneously hypertensive rats by the negative control (○), Captopril (50 mg kg^{-1}) (□) or different doses of (+)-catechin: 0.25 mg kg^{-1} (◆), 0.5 mg kg^{-1} (▲), 1.5 mg kg^{-1} (●) and 3 mg kg^{-1} (■). (B) Decreases in SBP caused in Wistar-Kyoto rats by the negative control (○) or 0.5 mg kg^{-1} (+)-catechin (▲). Data are expressed as mean ± SEM. The experimental groups always have a minimum of 8 animals. Different letters represent the statistical differences ($p < 0.05$). p estimated by two-way ANOVA.

revealed only glucuronide metabolites and methyl-glucuronide metabolites, apart from the original flavan-3-ols, as all other potential phase II metabolites, as sulfated and methylated derivatives, were not detected. In detail, the following concrete metabolites were detected: epicatechin glucuronide, methyl-epicatechin glucuronide, catechin glucuronide and methyl-catechin glucuronide. Fig. 5 shows the concentration of the all these flavan-3-ol compounds, that had been detected in the plasma after the different administrations. As this figure shows, scarce concentrations of unconjugated flavan-3-ols appeared in the plasma of the treated rats, regardless of the administered dose and the time elapsed after the treatment. On the contrary, methyl-glucuronide metabolites were always obtained in this biological fluid after different treatments. Significant concentrations of the glucuronide metabolite were appreciated after 6 mg kg^{-1} (−)-epicatechin administration, but these conjugated products did not appear in the plasma when the most effective doses of the other flavan-3-ols were administered (0.5 mg kg^{-1} (+)-catechin or 0.5 mg kg^{-1}

(−)-catechin). However, glucuronide metabolites were obtained when the highest doses of the different flavan-3-ols (12 mg kg^{-1} (−)-epicatechin, 3 mg kg^{-1} (+)-catechin or 3 mg kg^{-1} (−)-catechin) were administered, being the concentration of the glucuronide metabolite particularly high after the administration of 12 mg kg^{-1} (−)-epicatechin. In any case, no correlation could be established between the plasma concentration of glucuronide and/or methyl-glucuronide metabolites, and the antihypertensive effect caused in the animals by (−)-epicatechin or the catechin monomers. 72 hours post-flavan-3-ol administration, neither flavan-3-ol compounds nor flavan-3-ol metabolites could be detected in plasma (data not shown).

As panel A in Fig. 6 shows, no differences were observed in the plasma antioxidant capacity between the rats administered with 10% DMSO solution and the rats administered with more effective antihypertensive doses of the different flavan-3-ols (6 mg kg^{-1} (−)-epicatechin, 0.5 mg kg^{-1} (+)-catechin or 0.5 mg kg^{-1} (−)-catechin). Nevertheless, as panel B in Fig. 6 shows, the administration of the highest doses of the different flavan-
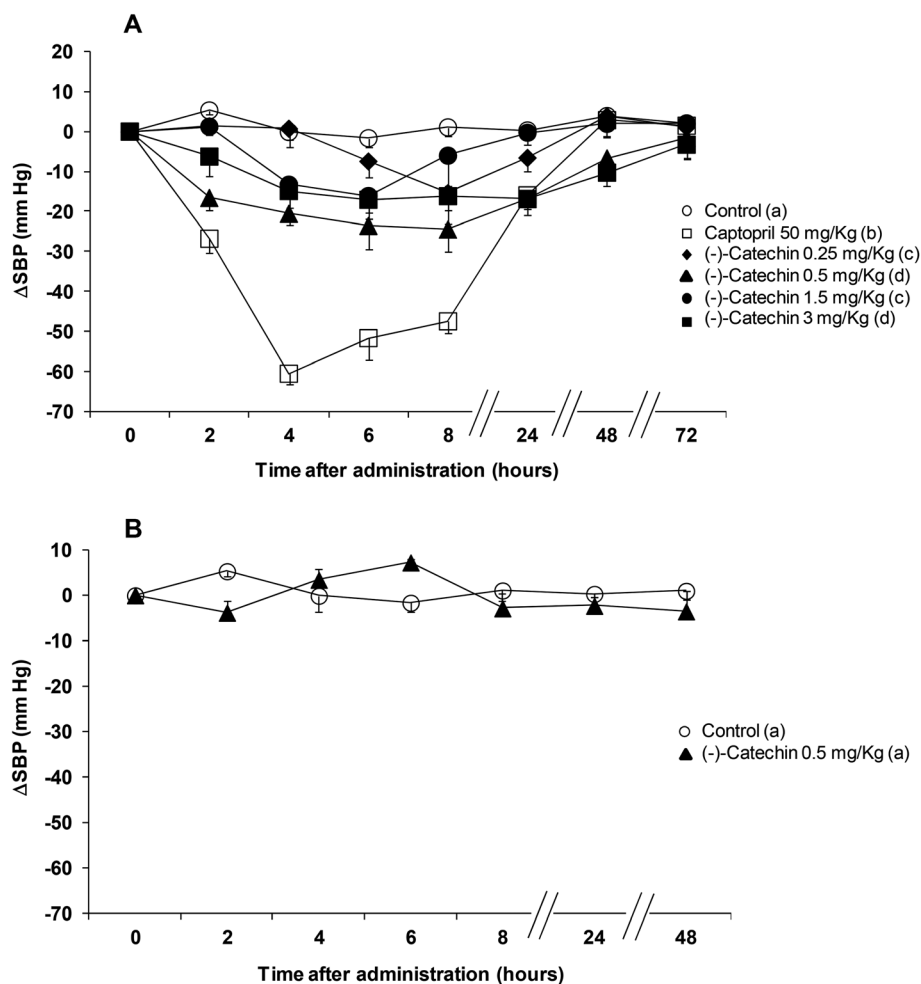
Fig. 3 (A) Decreases in systolic blood pressure (SBP) caused in spontaneously hypertensive rats by the negative control (○), Captopril (50 mg kg^{-1}) (□) or different doses of (−)-catechin: 0.25 mg kg^{-1} (◆), 0.5 mg kg^{-1} (▲), 1.5 mg kg^{-1} (●) and 3 mg kg^{-1} (■). (B) Decreases in SBP caused in Wistar-Kyoto rats by the negative control (○) or 0.5 mg kg^{-1} (−)-catechin (▲). Data are expressed as mean ± SEM. The experimental groups always have a minimum of 8 animals. Different letters represent the statistical differences ($p < 0.05$). p estimated by two-way ANOVA.

3-ols (12 mg kg^{-1} (−)-epicatechin, 3 mg kg^{-1} (+)-catechin or 3 mg kg^{-1} (−)-catechin) caused always an increase in the plasma antioxidant capacity in the rats.

4. Discussion

SHR are frequently used to carry out initial studies with antihypertensive functional food ingredients because these animals represent nowadays the best experimental model for essential hypertension in humans.[26] In the present study, we have demonstrated that the short-term administration of (−)-epicatechin and (+)-catechin, two flavan-3-ols present in cocoa and other different foods, decreased SBP in SHR. (−)-Catechin, an atypical flavan-3-ol produced by changes in the chiral nature of (−)-epicatechin during the cocoa manufacturing process,[12] also decreased the arterial blood pressure in this strain. On the contrary, none of these flavan-3-ols

decreased SBP in the normotensive WKY rats. The arterial blood pressure effects of the assayed flavan-3-ols are therefore specific for the hypertensive condition.

The healthy properties of cocoa seem to be related to its high content in monomeric, dimeric and potentially polymers of flavan-3-ols.[27] (−)-Epicatechin is the most abundant monomer in cocoa seeds and cocoa derived foods. In accordance with the results that we have obtained in SHR, a recent study has demonstrated that the long-term administration of 10 mg kg^{-1} (−)-epicatechin prevented deoxycorticosterone acetate-salt induced hypertension in rats.[28] The high amount of (−)-epicatechin in cocoa seems actually to be important, because, in previous studies an increased plasma level of this flavan-3-ol was accompanied by a dose-dependent increase in the plasma antioxidant capacity,[29,30] and also with a dose-dependent decrease in plasma lipid oxidation[29] and a beneficial effect on the vascular function.[31] Moreover, (−)-epicatechin decreases serum oxidative stress[8] and restores
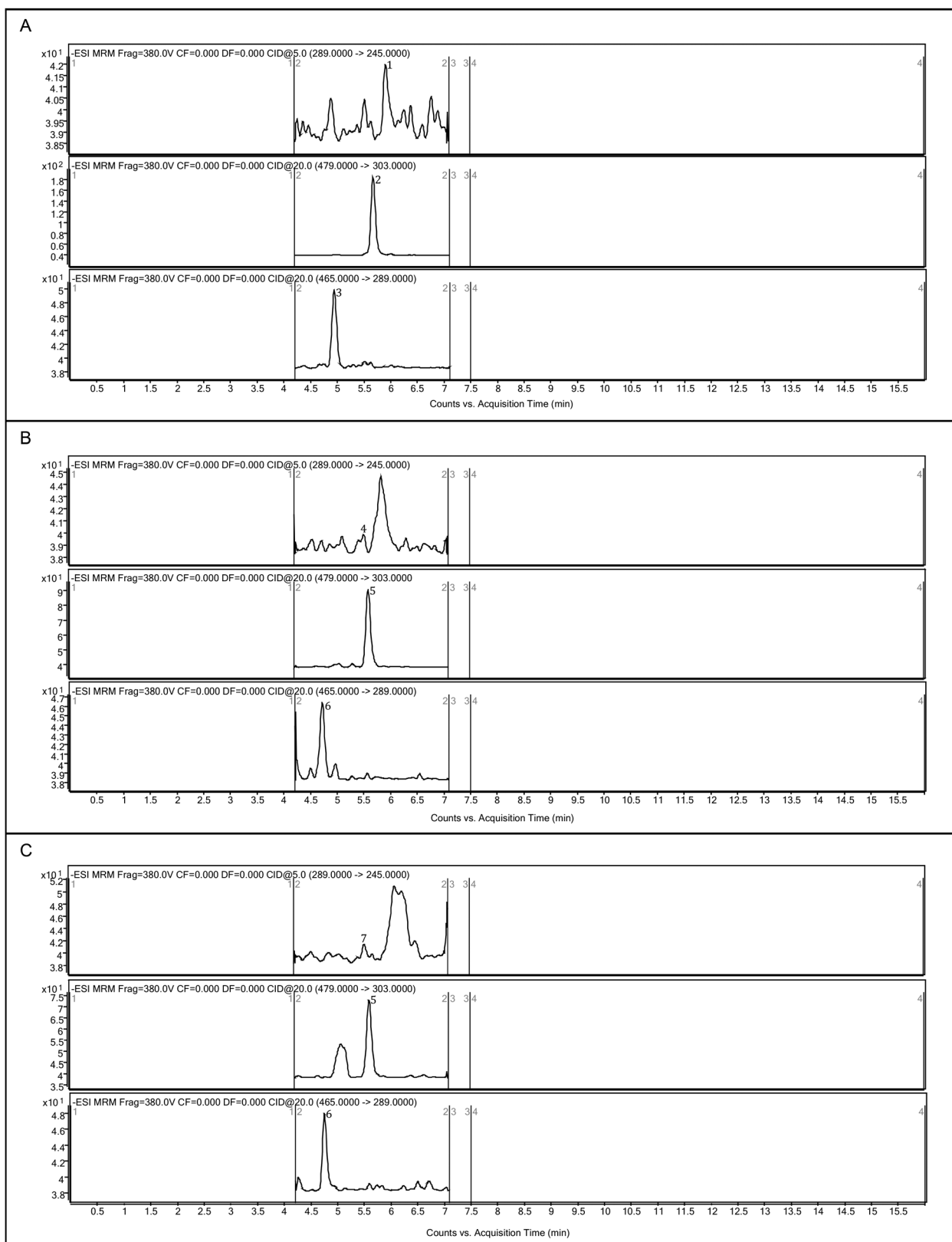
Fig. 4 Multiple reaction monitoring (MRM) signals of the compounds found in the plasma of spontaneously hypertensive rats after oral adminis-tration of different flavan-3-ols: (A) 6 hours after oral administration of 6 mg kg^{-1} (–)-epicatechin. (B) 4 hours after oral administration of 0.5 mg kg^{-1} (+)-catechin. (C) 6 hours after oral administration of 0.5 mg kg^{-1} (–)-catechin. The following compounds were obtained: (–)-epicatechin (1), methyl-epicatechin glucuronide (2), epicatechin glucuronide (3), catechin (4), methyl-catechin glucuronide (5), catechin glucuronide (6) and catechin (7).
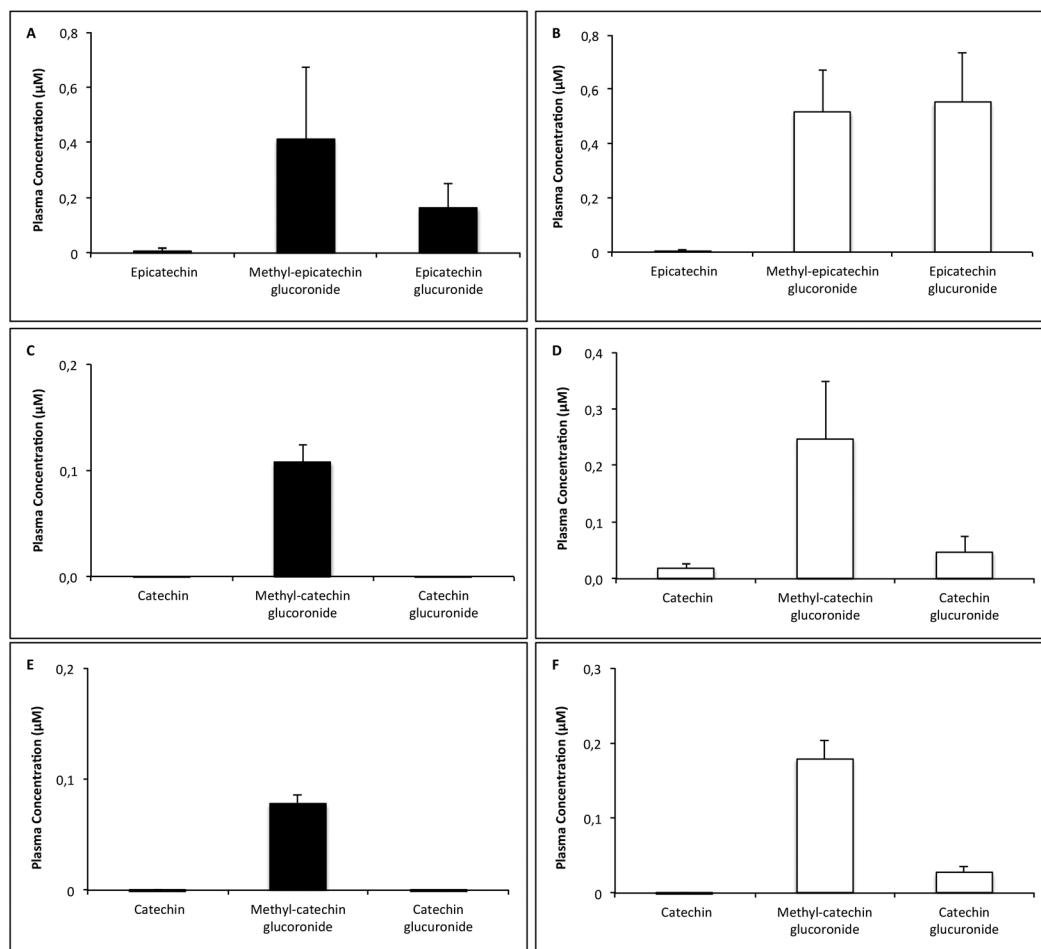
Fig. 5 Concentration of the flavan-3-ol compounds in plasma samples of spontaneously hypertensive rats: 6 hours after oral administration of 6 mg kg^{-1} (−)-epicatechin (panel A); 4 hours after oral administration of 12 mg kg^{-1} (−)-epicatechin (panel B); 4 hours after oral administration of 0.5 mg kg^{-1} (+)-catechin (panel C); 6 hours after oral administration of 3 mg kg^{-1} (+)-catechin (panel D); 6 hours after oral administration of 0.5 mg kg^{-1} (−)-catechin (panel E); 6 hours after oral administration of 3 mg kg^{-1} (−)-catechin (panel F). Data are expressed as mean (μM) ± SEM. The experimental groups always have a minimum of 6 animals.

the NO bioavailability.[32] In addition, a study of our research group has also demonstrated the participation of NO in the antihypertensive effect of a polyphenol-rich cocoa in SHR.[33]

(+)-Catechin predominates in cocoa beans and (−)-catechin in chocolate, and it has been postulated that (+)-catechin is almost 10 times more absorbed than (−)-catechin.[12] In this study, the administration of (+)-catechin, and also the administration of (−)-catechin, caused a significant decrease in the SBP of the SHR. Nevertheless, the effect of (−)-catechin on this variable was less accentuated than the effect of (−)-epicatechin or (+)-catechin. In the last few years, an important role has been attributed to the chiral nature of polyphenols and the effects of chirality on bioavailability. Ottaviani et al. have demonstrated the significance of the stereochemical configuration of flavan-3-ols for their biological activity.[34] These facts may explain why catechin from processed cocoa [mainly (−)-catechin], is not as well absorbed as (+)-catechin.[35] In any case, in this study, the effects of (+)-catechin in SHR, and the effects of (−)-catechin in these animals were not very different,

and the metabolites of both catechin monomers in the plasma of the treated rats were also very similar, indicating a similar oral absorption of these two isomers. The effect seems however to last somewhat more when (−)-catechin was administered and we shall comment on this later.

It is therefore clear that all the flavan-3-ols assayed showed a blood pressure lowering effect in the SHR, and, this study enables us to characterize and to compare their antihypertensive effect in these animals. Among the three used flavan-3-ols, (−)-epicatechin was the most effective one, even if (+)-catechin and (−)-catechin were more potent than (−)-epicatechin to decrease the arterial blood pressure in the SHR. In fact, maximum decreases in SBP were obtained when (−)-epicatechin was administered, but lower doses of (+)-catechin and (−)-catechin were needed for the antihypertensive effect. It is also important to highlight that in this study we failed to demonstrate a dose-dependent antihypertensive effect for the three monomers that we have used. In fact, the maximum effect was attained always with a dose different form the
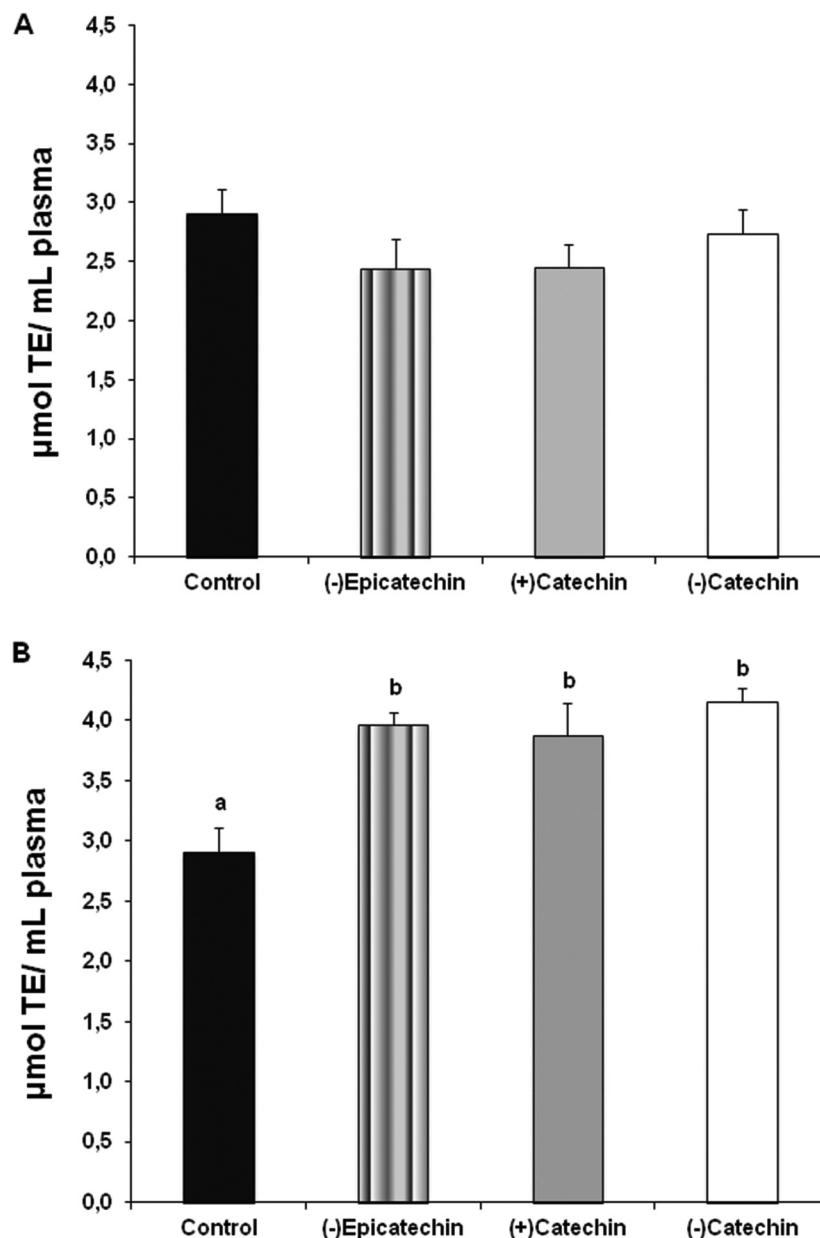
Fig. 6 Oxygen radical absorbance capacity (ORAC) histograms of plasma samples from spontaneously hypertensive rats. Panel A: 4 hours after oral administration of 10% DMSO solution (control group) (■), 6 hours after oral administration of 6 mg kg^{-1} (−)-epicatechin (▥), 4 hours after oral administration of 0.5 mg kg^{-1} (+)-catechin (▨) and 6 hours after oral administration of 0.5 mg kg^{-1} of (−)-catechin (□). Panel B: 4 hours after oral administration of 10% DMSO solution (control group) (■), 4 hours after oral administration of 12 mg kg^{-1} (−)-epicatechin (▥), 6 hours after 3 mg kg^{-1} (+)-catechin (▨) and 6 hours after oral administration of 3 mg kg^{-1} of (−)-catechin (□). Data are expressed as mean (μmol TE ml^{-1}) ± SEM. The experimental groups always have a minimum of 6 animals. Different letters represent the statistical differences ($p < 0.05$). p estimated by two-way ANOVA.

highest one (6 mg kg^{-1} of (−)-epicatechin, 0.5 mg kg^{-1} of (+)-catechin and 0.5 mg kg^{-1} of (−)-catechin). A similar paradox was observed when we had administered a polyphenol-rich cocoa powder in SHR, since the maximum antihypertensive effect in these animals was neither obtained when we administered the highest dose of this cocoa powder.[9] The results obtained by our research group using both the cocoa monomers and the cocoa powder are somewhat difficult to understand. They might be explained by keeping in mind different

studies that demonstrated that a high quantity of polyphenols could exhibit pro-oxidant properties instead of antioxidant properties.[36,37] It is true that, in the present study, an increase in the dose of the assayed flavan-3-ols was related to an increased plasma antioxidant capacity, but the plasma antioxidant capacity cannot totally define the endothelium redox status. An improved vascular oxidative stress in the SHR treated with the most effective doses of the assayed flavan-3-ols, or a pro-oxidant effect on arterial tissue in the rats pre-

viously treated with the highest doses of these compounds, cannot be ruled out. In addition, it is also important to bear in mind that other properties of the assayed flavan-3-ols could explain their blood pressure effects. In this context, activation of the deacetylase sirtuin 1 (SIRT1) and up-regulation of endothelial nitric oxide synthase[38,39] have also been proposed to explain the cardiovascular effects of polyphenols. Moreover, the effects of polyphenols have also been attributed to the induction of antioxidant enzymes in cardiovascular tissues[40,41] and also to the inhibition of the angiotensin converting enzyme.[42]

On seeing the period of time elapsed to recover the baseline values of SBP from administration, we can also assume that the effects of (−)-epicatechin and (−)-catechin (Fig. 1A and 3A) were longer than the antihypertensive effect of (+)-catechin (Fig. 2A). The *in vivo* bioactivity of the flavan-3-ols depends on their process of absorption and metabolism after ingestion, and the reducing properties of resulting metabolites. The highest plasma peak concentrations of flavan-3-ols in humans are obtained 2–3 hours after ingestion of these compounds[29,30] and they are still measurable after 8 hours.[43] Nevertheless, it is important to note that in this study the metabolite profile has been performed for 4 and 6 h in order to evaluate the bioavailable plasma metabolites at the maximum blood pressure decrease time point. In this study, the effect of (−)-epicatechin, (+)-catechin and (−)-catechin could also be appreciated 8 hours post-administration. It should be noted that (−)-epicatechin and (+)-catechin in particular, showed a good bioavailability in humans.[13] Flavan-3-ols are conjugated after ingestion by phase-II enzymes in the small intestine and liver, and it has been demonstrated that some of the beneficial effects of these compounds are due to the metabolized forms.[14] Even if (−)-epicatechin and catechin monomers are usually conjugated to produce sulphate, glucuronide and methyl-glucuronide metabolites,[14,44] in this study, only glucuronide and methyl-glucuronide metabolites were quantified in the plasma from the treated rats. These metabolites, but not sulphate metabolites, could therefore be responsible for the antihypertensive effects of the assayed flavan-3-ols. At first sight, the products enriched in these monomers might represent a good strategy in biomedicine, but it is nevertheless true that no correlation could be established in our study between the plasma concentration of the glucuronide and methyl-glucuronide metabolites, and the antihypertensive effect caused by the (−)-epicatechin or the catechin monomers. Thus, when the most effective doses of these flavan-3-ols were administered, we did not obtain the highest concentration of these metabolites. Moreover, the administration of the most effective doses of the catechin monomers (0.5 mg kg⁻¹ for both catechin monomers) were not accompanied with glucuronide metabolites in plasma, and this suggests that methyl-glucuronide metabolites could actually be responsible for the effect of the catechin monomers. Glucuronide metabolites appeared nevertheless in the plasma obtained from the rats that were treated with the highest dose of these monomers, and also in all the plasma samples from the (−)-epicatechin treated rats, but the concentration of the (−)-epicatechin glu-

curonide metabolite was also lower in the plasma of the rats treated with the most effective antihypertensive dose of (−)-epicatechin (6 mg kg⁻¹) than in the plasma obtained from the rats that had been treated with the highest dose of this flavan-3-ol (12 mg kg⁻¹). It seems in addition that according to our results, the effect of the assayed flavan-3-ols was always elapsed 72 hours after their administration and that at this time no flavan-3-ol metabolites were present in the plasma of the rats.

In conclusion, we have demonstrated the antihypertensive properties of the main cocoa flavan-3-ols in SHR and we have also demonstrated that the effect of the evaluated monomers is specific to the hypertensive conditions. Therefore, the flavan-3-ols (−)-epicatechin and (+)-catechin, and also the atypical epimer (−)-catechin, would be beneficial for controlling arterial blood pressure. They could be responsible for the antihypertensive effect of different cocoa powders and functional foods. The present study also represents a good contribution to clarify the metabolites generated in the SHR when these flavan-3-ols are administered to these animals. Nevertheless, our results neither enable us to relate the antihypertensive effect of the assayed flavan-3-ols with the presence of a concrete flavan-3-ol metabolite in plasma, nor to consider the antioxidant effect of the used flavan-3-ols as their main antihypertensive mechanism. The concentration of flavan-3-ol metabolites in arterial tissues may provide interesting information in the future to elucidate their cardiovascular role, and further research should be interesting to go deep into the mechanisms that could explain the antihypertensive effects of the assayed flavan-3-ols.

Acknowledgements

References

1 M. Galleano, O. Pechanova and C. G. Fraga, *Curr. Pharm. Biotechnol.*, 2010, **11**, 837–848.
2 A. Lynn, H. Hamadeh, W. C. Leung, J. M. Russell and M. E. Barker, *Plant Foods Hum. Nutr.*, 2012, **67**, 309–314.
3 A. S. Mathew, G. M. Capel-Williams, S. E. E. Berry and W. L. Hall, *Plant Foods Hum. Nutr.*, 2012, **67**, 351–357.
4 N. Mihailovic-Stanojevic, A. Belščak-Cvitanović, J. Grujić-Milanović, M. Ivanov, D. Jovović, D. Bugarski and Z. Miloradović, *Plant Foods Hum. Nutr.*, 2013, **68**, 235–240.

5 S. Ellinger, A. Reusch, P. Stehle and H.-P. Helfrich, *Am. J. Clin. Nutr.*, 2012, **95**, 1365–1377.

6 R. Corti, A. J. Flammer, N. K. Hollenberg and T. F. Lüscher, *Circulation*, 2009, **119**, 1433–1441.

7 K. W. Lee, Y. J. Kim, H. J. Lee and C. Y. Lee, *J. Agric. Food Chem.*, 2003, **51**, 7292–7295.

8 H. Schroeter, C. Heiss, J. Balzer, P. Kleinbongard, C. L. Keen, N. K. Hollenberg, H. Sies, C. Kwik-Uribe, H. H. Schmitz and M. Kelm, *Proc. Natl. Acad. Sci. U. S. A.*, 2006, **103**, 1024–1029.

9 E. Cienfuegos-Jovellanos, M. Quiñones, B. Muguerza, L. Moulay, M. Miguel and A. Aleixandre, *J. Agric. Food Chem.*, 2009, **57**, 6156–6162.

10 M. Quiñones, D. Sánchez, B. Muguerza, L. Moulay, S. Laghi, M. Miguel and A. Aleixandre, *Food Chem.*, 2010, **122**, 1013–1019.

11 M. Quiñones, D. Sánchez, B. Muguerza, M. Miguel and A. Aleixandre, *Food Res. Int.*, 2011, **44**, 1203–1208.

12 M. Kofink, M. Papagiannopoulos and R. Galensa, *Molecules*, 2007, **12**, 1274–1288.

13 F. A. Tomas-Barberan, E. Cienfuegos-Jovellanos, A. Marín, B. Muguerza, A. Gil-Izquierdo, B. Cerda, P. Zafrilla, J. Morillas, J. Mulero, A. Ibarra, M. A. Pasamar, D. Ramón and J. C. Espín, *J. Agric. Food Chem.*, 2007, **55**, 3926–3935.

14 L. Guerrero, M. Margalef, Z. Pons, M. Quiñones, L. Arola, A. Arola-Arnal and B. Muguerza, *J. Nutr. Biochem.*, 2013, **24**, 2092–2099.

15 M. Margalef, L. Guerrero, Z. Pons, F. I. Bravo, L. Arola, B. Muguerza and A. Arola-Arnal, *Food Res. Int.*, 2014, **64**, 500–507.

16 M. Monagas, M. Urpi-Sarda, F. Sánchez-Patán, R. Llorach, I. Garrido, C. Gómez-Cordovés, C. Andres-Lacueva and B. Bartolomé, *Food Funct.*, 2010, **1**, 233–253.

17 D. Del Rio, A. Rodriguez-Mateos, J. P. E. Spencer, M. Tognolini, G. Borges and A. Crozier, *Antioxid. Redox Signaling*, 2013, **18**, 1818–1892.

18 A. L. Waterhouse, J. R. Shirley and J. L. Donovan, *Lancet*, 1996, **348**, 834.

19 M. Karim, K. McCormick and C. T. Kappagoda, *J. Nutr.*, 2000, **130**, 2105S–22108.

20 E. Ramiro, A. Franch, C. Castellote, F. Pérez-Cano, J. Permanyer, M. Izquierdo-Pulido and M. Castell, *J. Agric. Food Chem.*, 2005, **53**, 8506–8511.

21 E. O. Cuevas-Rodríguez, V. P. Dia, G. G. Yousef, P. A. García-Saucedo, J. López-Medina, O. Paredes-López, E. Gonzalez de Mejia and M. A. Lila, *J. Agric. Food Chem.*, 2010, **58**, 9542–9548.

22 R. D. Buñag, *J. Appl. Physiol.*, 1973, **34**, 279–282.

23 M. Margalef, Z. Pons, B. Muguerza and A. Arola-Arnal, *J. Agric. Food Chem.*, 2014, **62**, 7698–7706.

24 A. Serra, A. Macià, M. P. Romero, M. J. Salvadó, M. Bustos, J. Fernández-Larrea and M. J. Motilva, *J. Chromatogr., B: Anal. Technol. Biomed. Life Sci.*, 2009, **877**, 1169–1176.

25 D. Huang, B. Ou, M. Hampsch-Woodill, J. A. Flanagan and R. L. Prior, *J. Agric. Food Chem.*, 2002, **50**, 4437–4444.

26 K. Okamoto and K. Aoki, *Jpn. Circ. J.*, 1963, **27**, 282–293.

27 K. A. Cooper, J. L. Donovan, A. L. Waterhouse and G. Williamson, *Br. J. Nutr.*, 2008, **99**, 1–11.

28 M. Gómez-Guzmán, R. Jiménez, M. Sánchez, M. J. Zarzuelo, P. Galindo, A. M. Quintela, R. López-Sepúlveda, M. Romero, J. Tamargo, F. Vargas, F. Pérez-Vizcaíno and J. Duarte, *Free Radical Biol. Med.*, 2012, **52**, 70–79.

29 D. Rein, S. Lotito, R. R. Holt, C. L. Keen, H. H. Schmitz and C. G. Fraga, *J. Nutr.*, 2000, **130**, 2109S–2114S.

30 M. Serafini, R. Bugianesi, G. Maiani, S. Valtuena, S. De Santis and A. Crozier, *Nature*, 2003, **424**, 1013.

31 C. Heiss, A. Dejam, P. Kleinbongard, T. Schewe, H. Sies and M. Kelm, *JAMA*, 2003, **290**, 1030–1031.

32 M. C. Litterio, G. Jaggers, G. Sagdicoglu Celep, A. M. Adamo, M. A. Costa, P. I. Oteiza, C. G. Fraga and M. Galleano, *Free Radical Biol. Med.*, 2012, **53**, 1894–1902.

33 M. Quiñones, B. Muguerza, M. Miguel and A. Aleixandre, *Pharmacol. Res.*, 2011, **64**, 478–481.

34 J. I. Ottaviani, T. Y. Momma, C. Heiss, C. Kwik-Uribe, H. Schroeter and C. L. Keen, *Free Radical Biol. Med.*, 2011, **50**, 237–244.

35 J. L. Donovan, C. Manach, R. M. Faulks and P. A. Kroon, in *Plant Secondary Metabolites: Occurrence, Structure and Role in the Human Diet*, ed. A. Crozier, M. N. Clifford and H. Ashihara, Blackwell Publishing, Oxford, 2006, pp. 303–351.

36 B. S. P. Nicole Cotelle, *Curr. Top. Med. Chem.*, 2001, **1**, 569–590.

37 M. Lahouel, S. Amedah, A. Zellagui, A. Touil, S. Rhouati, F. Benyache, E. Leghouchi and H. Bousseboua, *Therapie*, 2007, **61**, 347–355.

38 I. Mattagajasingh, C.-S. Kim, A. Naqvi, T. Yamamori, T. A. Hoffman, S.-B. Jung, J. DeRicco, K. Kasuno and K. Irani, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 14855–14860.

39 T. Wallerath, *Circulation*, 2002, **106**, 1652–1658.

40 Z. Cao and Y. Li, *Cardiovasc. Toxicol.*, 2004, **4**, 339–354.

41 Y. Li, Z. Cao and H. Zhu, *Pharmacol. Res.*, 2006, **53**, 6–15.

42 L. Guerrero, J. Castillo, M. Quiñones, S. Garcia-Vallvé, L. Arola, G. Pujadas and B. Muguerza, *PLoS One*, 2012, **7**, e49493.

43 M. Richelle, I. Tavazzi, M. Enslen and E. A. Offord, *Eur. J. Clin. Nutr.*, 1999, **53**, 22–26.

44 A. Arola-Arnal, G. Oms-Oliu, A. Crescenti, J. M. Del Bas, M. R. Ras, L. Arola and A. Caimari, *Mol. Nutr. Food Res.*, 2013, **57**, 1741–1752.