## Dalton Transactions



## CORRECTION

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## Correction: Unexpected optical activity of cerium in $Y_2O_3$ : $Ce^{3+}$ , $Yb^{3+}$ , $Er^{3+}$ up and down-conversion system

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Correction for 'Unexpected optical activity of cerium in  $Y_2O_3$ : $Ce^{3+}$ ,  $Yb^{3+}$ ,  $Er^{3+}$  up and down-conversion system' by Riccardo Marin *et al.*, *Dalton Trans.*, 2013, **42**, 16837–16845.

In the paper, we claim to observe the optical activity of cerium in a triply doped yttrium oxide system displaying both upconverting and downshifting properties. The singly-doped  $Y_2O_3$ : $Ce^{3+}$  system has already been investigated by many groups, with controversial results, as stated in our paper. Some groups suggested an emission centered on cerium ions, while others proposed a defect-related emission, caused by the introduction of defects during the doping process. We propose optical activity of the trivalent cerium ions embedded in yttria.

We observed that a broadband emission feature in the 300-600 nm range was displayed by samples doped with various cerium contents, which was ascribed to the cerium  $5d\rightarrow 4f$  electronic transition. One piece of evidence supporting this interpretation is the shift of this emission upon increasing the cerium content (*i.e.* increasing the cell edge), a behavior already observed in yttrium aluminum garnet (YAG) doped with cerium and increasing gadolinium concentrations. Nonetheless, further analyses of the system demonstrated that this broadband feature cannot be unambiguously assigned to the cerium emission. Actually, this spectral region hosts a plethora of defect-related emissions. In particular, we found that the quartz window used during the analyses has an intrinsic emission in this region. So, the real effect of the increasing cerium content could be distorted somehow

To gain a deeper insight into this issue, we synthesized three more samples with composition  $Y_2O_3$ :Er(2%),  $Y_2O_3$ :Er(2%

As can be noted, when the samples with increasing cerium content are excited at 330 nm, a band centered at 436 nm appears at the expense of the band centered at 413 nm. The latter is ascribed to an intrinsic yttria emission due to defect states; the former originates from features introduced by the cerium ions. The same behavior is observed on excitation at 350 nm. We can observe that the emission of erbium is also enhanced when cerium is present: this corroborates our hypothesis of an energy transfer process taking place between cerium and erbium ions. This mechanism is even more evident when the excitation wavelength is shifted to 379 nm. Direct excitation of erbium is expected in this case. We can observe a decrease in the intensity of the emission arising from this lanthanide ion when the cerium content increases. Concomitantly, the tail of the emission supposed to arise from the  $d\rightarrow f$  electronic transition in  $Ce^{3+}$  increases in intensity. This evidence supports our thesis about the energy transfer mechanism, and makes us confident in the interpretation of the data we presented in the article. Admittedly, the shape of the emission profile of these samples is slightly different from that displayed by the samples analyzed in the paper. This difference could be due to an effect caused by the old quartz window, and it does not compromise to any extent the interpretations presented above and in the article.

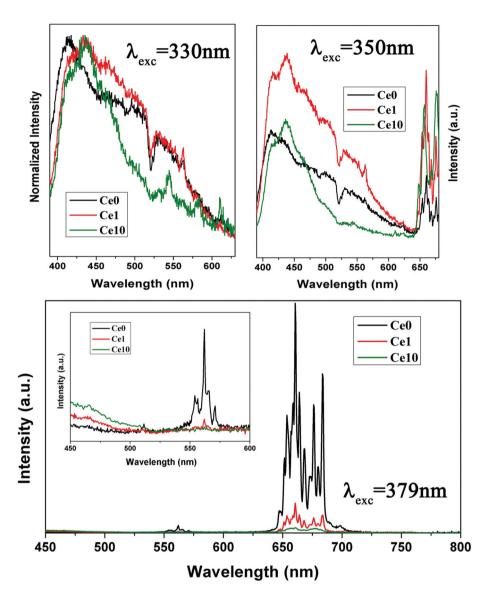
To conclude, although we noted the contamination of the photoluminescence signals of the samples with a signal from the analysis apparatus, we observed that the conclusions drawn for the presented systems are only slightly affected by this

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misinterpretation. The upconverting and downshifting properties of the systems are still confirmed, as is the energy transfer between the erbium and cerium-related states. Nonetheless, we could not definitively exclude the possibility of the broadband emission arising from defect states instead of the cerium ions themselves, the latter being the most plausible interpretation at the time the article was written.



The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.