1 Introduction

Helium nanodroplets (HeN) represent a well-established matrix for the spectroscopic investigation of isolated, cold (0.4 K) atoms, molecules, and clusters1 and offer a unique method for cold chemistry on a single molecule level. The spatial separation of surface located and solvated dopants in HeN provides a promising approach to control and monitor chemical reactions, especially if the location is state specific and can thus be changed by photoexcitation. In this article we demonstrate a complete cycle of photoinduced molecular dissociation, spatial separation of the fragments to a fully solvated and a stable, surface bound state and finally, photoinduced geminate recombination of the fragments.

Increasing interest in exploring chemical reactions in HeN originates from the ability to stabilize and investigate radical reaction intermediates2–7 or confine dissociation fragments to the volume of the droplet.8,9 High cooling rates permit efficient relaxation of excited molecules,1,8 the formation and stabilization of metastable clusters,10,11 or aggregation of weak van der Waals bound complexes (see, e.g., ref. 12). Shallow energy barriers along reaction pathways can lead to trapping in local reactive potential-energy surface minima4 which can be overcome by photo-activation to trigger chemical reactions.13,14 The location of a dopant is dictated by the interaction with helium (the pair potential) and can be estimated by the dimensionless Ancilotto parameter.19 While the majority of atoms and molecules in their ground state are located inside the droplet, an electronic transition from the ground state to an excited state can initiate migration from inside to the droplet surface. This has been observed, for example, for Ag8,16–18 Cu,19,20 Cr,21–23 CF3,24,25 and NO.26 Usually, surface migration is followed by desorption from the droplet, only for NO* there was indication that the excited molecule remains in a surface-bound state.26 For droplets doped with both surface located and solvated species the helium matrix can form a barrier that prevents molecule formation or clustering, even in the presence of long range van der Waals interaction.27 On the other side, van der Waals forces between solvated noble gas atoms and surface-located alkali line earth metal atoms can be exploited to overcome the separating character of HeN.28,29 Transition from a surface location to solvation was achieved for the HCN–Sr complex (and vice versa for HCN–Ca) by vibrational excitation,30 and in general, surface-located species migrate inside the droplet upon ionization.31–34

Here we demonstrate that photoexcitation of solvated Cr2 molecules results in dissociation to one atom in a solvated state (a7S3) and a second atom in a surface bound state (a5S2). Both fragments are sufficiently cooled to prevent ejection from the droplet. A resonant three photon ionization scheme is applied to the surface atom to verify its location and, at the same time, trigger solvation and recombination with its original partner.
The stable interior location of the other fragment is proven with a two laser experiment, where both lasers are scanned individually.

2 Experimental

Chromium clusters (Cr$_2$–Cr$_4$) are formed inside helium nanodroplets by the pickup of single Cr atoms. The setup follows a Helium Nanodroplet Isolation (HENDI) apparatus, described in detail in ref. 35. In brief, He$_N$ are formed by the supersonic expansion of high purity $^4$He gas through a cooled nozzle (diameter $d = 5 \mu m$, stagnation pressure $p_0 = 50$ bar). Droplet sizes follow a log-normal distribution with a distribution maximum of $\bar{N} = 6300$ He atoms (mean droplet size $\bar{N} = 16000$) for a nozzle temperature of 14 K. The collimated He$_N$ beam is crossed at right angles by an effusive beam of Cr atoms from an electron bombardment Cr oven$^{36}$ beneath the He$_N$ beam. This setup ensures that no free atoms reach the detection region. To characterize the pick-up conditions, the number of Cr$_{1,2,\ldots}$ ions formed by electron bombardment ionization is monitored as function of the oven heating power. A quadrupole mass spectrometer (QMS, Balzers QMG 422) is attached at the end of the measurement chamber to facilitate mass selective ion detection. This allows an assignment of the photoion signal to a specific Cr$_n$ cluster size, by comparing the photoion yield to the electron bombardment Cr$_{1,2,\ldots}$ yield as a function of heating power.

Excitation spectra of the Cr$_n$–He$_N$ ($n = 2–4$) complexes are recorded with resonance-enhanced multiphoton ionization (REMPI) spectroscopy. Laser pulses are obtained from excimer (XeCl, Radiant Dyes RD-EXC-200, 308 nm, ~20 ns pulse duration) pumped dye lasers. For one-color REMPI, a pulsed dye laser beam (Lamda Physik, FL3002, dyes: Coumarin 2, Coumarin 102) intersects the doped He$_N$ beam at right angles in the extraction region of the QMS. The mass window of the QMS is set to integrate the signal of the Cr$_n^+$ isotopes and Cr$_n^+$–He around the central cluster mass to increase the signal. To further increase the signal quality, two-color REMPI is used in addition. Therefore, a fraction of the 308 nm pump laser light is overlapped with the dye laser.

A pump–probe scheme is applied to observe the formation of ground state atoms after dissociating Cr$_2$. Two dye laser beams (Radiant Dyes, DL-midi, dye: Coumarin 2; Lamda Physik, FL3002, dye: RDC360 neu) intersect the doped He$_N$ beam, where the first one dissociates the molecules, while the second ionizes the produced ground state Cr atoms state selectively where the first one dissociates the molecules, while the second

![Fig. 1](image)

**Fig. 1.** Cr$_2$ and Cr$_2^+$ potential energy curves$^{37,38}$ (left) and Cr atom energy levels$^{39}$ (right). A one-color ionization scheme is indicated by a combination of solid arrows (laser excitation) and dashed arrows (relaxation). Upon Cr$_2$ excitation to the A$^1\Sigma^+$ state inside He$_N$, the molecule undergoes relaxation to predissociating states resulting in a S$_2$ Cr and a S$_2$ Cr atom states. While the first remains solvated inside the droplet, the latter migrates to the surface where it is ionized by a resonant three photon scheme. This triggers solvation and recombination to Cr$_2^+$ (top of the figure). The ionization path to probe the ground state atom with a second dye laser is indicated by dash-dotted arrows. A suggested ionization path that does not include Cr$_2$ dissociation is indicated by gray arrows.

3 Results

One-color REMPI excitation spectra of Cr$_2$ via the A$^1\Sigma^+$ $\rightarrow$ X$^1\Sigma^+$ transition for Cr$^+$ and Cr$^+$ detection are shown in Fig. 2(a). The broad band, stretching from 21 500 to 22 750 cm$^{-1}$, is attributed to the Cr$_2$ A$^1\Sigma^+_u^+$ $\rightarrow$ X$^1\Sigma^+_g^+$ transition and the three pronounced
peaks can be assigned to Cr atom \( y^7P_j \leftarrow a^5S_2 \) transitions (indicated by triangles\(^{39}\)). Because the \( A^1\Sigma_u^+ \leftarrow X^1\Sigma_g^+ \) molecular band overlaps with the atomic transition energies, these peaks indicate the photoinduced dissociation with one of the fragments being in the \( a^5S_2 \) state, as will be discussed in detail below. The whole excitation-ionization process from \( a^5S_2 \) is indicated with solid, upwards pointing arrows in Fig. 1. \( a^5S_2 \) is excited to a predissociating state that leads to the formation of a ground state \( a^5S_3 \) and a metastable state \( a^5S_2 \) atom.\(^{41}\) The latter is excited via the \( y^5P_{1,2,3} \leftarrow a^5S_2 \) transition and ionized with either two [one-color REMPI] or one [two-color REMPI] photon. This ionization path selectively addresses \( a^5S_2 \) metastable Cr atoms and will be called \( a^5S \) REMPI from now on. For further insight into the dynamics after dissociation, we present detailed scans across the \( y^5P_j \leftarrow a^5S_2 \) transition for the detection of \( Cr^+ \) to \( Cr^+ \) in Fig. 3. Although one-color and two-color REMPI give equal results, the latter provides a better signal to noise ratio and was thus chosen. One 308 nm photon is used for ionization of excited \( y^5P_3 \) state atoms instead of two dye laser photons (cf. Fig. 1). The spectral signature is similar for detection of \( Cr_{1,2,3,4}^+ \), where the appearance of \( Cr_{3,4}^+ \) is connected to larger \( Cr_n (n = 3, 4, \ldots) \) clusters, as will be discussed in Section 4.3.

While the production of the metastable \( a^5S_2 \) atom as one of the dissociation products is verified with \( a^5S \) REMPI (see Fig. 2(a) and 3), the verification of the other dissociation product – the \( a^5S_3 \) ground state atom – requires a second laser. It ionizes the \( a^5S_3 \) atom with a well known Cr one-color REMPI scheme\(^{31}\) that includes the \( y^7P_{2,3,4} \leftarrow a^5S_3 \) ground state transition (Fig. 1, dash-dotted arrows) and is called \( a^5S \) REMPI from now on. The two-laser scheme opens the possibility to scan both laser wavelengths individually. First, the dissociation process is studied (Fig. 2(b)) by scanning the dissociation laser across the \( Cr_2 \) excitation band and photoionizing the produced ground state atoms with \( a^5S \) REMPI (includes the \( y^5P_{1,2,3} \leftarrow a^5S_2 \) transition). Narrow dips and peaks are artifacts due to experimental instabilities.

Fig. 2 (a) One-color REMPI spectra in the range of the \( Cr_2 \) \( A^1\Sigma_u^+ \leftarrow X^1\Sigma_g^+ \) transition. Photoions are detected at the mass of \( Cr^+ \) (black line) and \( Cr^+ \) (blue line, multiplied by 2). The bare Cr atom \( y^5P_{1,2,3} \leftarrow a^5S_2 \) transition energies are indicated with triangles.\(^{39}\) Asterisks mark the wavelengths at which the photoion signal is compared to the \( Cr_{1,2,\ldots}^+ \) ion yield obtained with electron impact ionization in dependence on the pick-up oven temperature. (b) Excitation spectrum in the range of the \( Cr_2 \) \( A^1\Sigma_u^+ \leftarrow X^1\Sigma_g^+ \) transition, recorded with a pump–probe laser scheme. The dissociating laser is scanned across the \( Cr_2 \) band, while the second laser ionizes the produced ground state atoms with \( a^5S \) REMPI (includes the \( y^5P_{1,2,3} \leftarrow a^5S_2 \) transition). Narrow dips and peaks are artifacts due to experimental instabilities.

Fig. 3 High resolution REMPI scan of the \( y^7P_j \leftarrow a^5S_2 \) transition within the \( Cr_2 \) \( A^1\Sigma_u^+ \leftarrow X^1\Sigma_g^+ \) band (Fig. 2) for detection of \( Cr_{1,2,3,4}^+ \) ions (two-color ionization scheme with the XeCl laser at 308 nm as second wavelength). The bare Cr atom \( y^7P_j \leftarrow a^5S_2 \) transition energy is indicated with a vertical line.\(^{39}\)
4 Discussion

4.1 Predissociation

Photoinduced dissociation in He$_n$ has been observed for various species.$^{13,25,28,42}$ Predissociation of Cr$_2$ upon excitation to $^1$S$_u$* has been observed in gas phase$^{37,41,43}$ and in matrix.$^{40}$ It was concluded, that the $^1$S$_u$* state is crossed by at least one perturbing state and one predissociating state, where possible assignments were discussed by Andersson.$^{37}$ Riley et al.$^{41}$ propose the nonradiative transition from the $^1$S$_u$* to another bound state. The latter is rapidly dissociated by a state correlating to the separate atom limit states Cr ($^3$S$_i$) and Cr (a$^3$S$_2$, a$^3$D).$^{41}$ From the observation of the $^5$P$_{1,2,3}$ $\rightarrow$ a$^5$S$_2$ and $^5$P$_{2,3,4}$ $\rightarrow$ a$^7$S$_3$ atomic transitions, we clearly identify the population of the a$^5$S$_2$ and a$^7$S$_3$ atom states and assign the following dissociative process:

$$\text{Cr}_2 \left( X^1 \Sigma_g^+ \right) \xrightarrow{h\nu} \text{Cr}_2 \left( A^1 \Sigma_u^+ \right) \rightarrow \text{Cr} \left( a^5S_2 \right) + \text{Cr} \left( a^7S_3 \right) \ (1)$$

4.2 Separation to a surface located a$^5$S$_2$ and a solvated a$^7$S$_3$ atom

The fate after dissociation is dictated by the interaction of the Cr atoms in the respective state (a$^5$S$_2$, a$^7$S$_3$) with He atoms.$^{44,45}$ In principle, the atoms can remain solvated inside the He$_n$, stay bound on the surface, or detach from the He$_n$ with a certain probability to take a few He atoms along. Since the droplet acts as a heat bath, kinetic energy can be dissipated to prevent ejection.$^{8,9}$ An excess energy of about 0.4 eV (3200 cm$^{-1}$), given by the difference of excitation energy and the dissociation limit (cf. Fig. 1), can be compensated by the evaporation of 640 He atoms.

So the location of the dopants after dissociation is governed by the electronic states of the atoms. Experimentally, the location of the two fragments is obtained from the line-shapes of the a$^5$S REMPI ($via$ $^5$P$_{2,3,4}$ $\rightarrow$ a$^5$S$_2$) and the a$^7$S REMPI ($via$ $^5$P$_{1,2,3}$ $\rightarrow$ a$^7$S$_3$) spectra. As observed for single Cr atoms in earlier experiments$^{21-23}$ and as predicted by theoretical studies,$^{44,45}$ the ground state (a$^5$S$_2$) atom remains solvated in the droplet. The metastable state (a$^7$S$_3$) atom, in contrast, favors a surface location, which is suggested by DFT calculations (ref. 45) and is supported by the experimental observations, as follows.

We start with the a$^5$S$_2$ atom, because the in-droplet A$^1$S$_u^+$ molecule band overlaps in energy with the atom $^5$P$_{1,2,3}$ $\rightarrow$ a$^5$S$_2$ transition, leading to excited Cr$^*$ ($^5$P$^o$) atoms which are photoionized by absorption of two photons (Fig. 1), so that Cr$^+$ ions are detected upon detachment from the droplet. In the Cr$^+$ signal in Fig. 2(a), all three peaks of the $^5$P$_{1,2,3}$ $\rightarrow$ a$^5$S$_2$ transition are observable. Considering the detailed Cr$^+$ scan (Fig. 3) of the most intense $^5$P$_3$ $\rightarrow$ a$^5$S$_2$ peak, the onset of the main feature is shifted approximately 5 cm$^{-1}$ to the blue relative to the bare atom transition energy. A maximum is reached at 22 240 cm$^{-1}$ followed by an almost steady signal decrease, giving a total peak width of about 50 cm$^{-1}$. These spectral characteristics are comparable to transitions of heavier alkali metal atoms that reside in a surface dimple.$^{27,35,46}$ which is a first indication for the surface location of the a$^5$S$_2$ atom (others will follow). At the bare atom transition energy (indicated with a vertical line in the detailed scan in Fig. 3), the absence of sharp atomic lines proves that hardly any atoms are ejected from the droplet upon dissociation, encouraging the interpretation in terms of the effective kinetic energy dissipation upon predissociation.

We consider it important to mention that the Cr $^5$P$_{1,2,3}$ $\rightarrow$ a$^5$S$_2$ excitation observed here on the He$_n$ surface is of completely different nature than the $^5$P$^o$ $\rightarrow$ a$^5$S$_2$ excitation of bare Cr atoms and small Cr-He$_n$ ($n = 1, 2, \ldots$) exciplexes observed in our previous studies.$^{23}$ This becomes obvious by comparing the
corresponding line shapes. In the two-color REMPI scheme in ref. 23, 308 nm excitation and droplet mediated relaxation results in bare a5S2 Cr atoms and Cr–He exciplexes, both being ejected from the HeN. The corresponding excitation spectrum is composed of a sharp atom line accompanied by a wing to the blue (due to Cr–He+). Here, in contrast, we find hardly any signal at the bare atom line position (Fig. 3) and observe the onset of the peak 5 cm−1 to the blue. Although the XeCl laser was present for recording the spectra in Fig. 3, a bare atom transition peak is hardly detectable. Systematic studies showed an increased bare atom signal for higher XeCl laser fluence, which starts causing the ejection process.

We now turn to the location and environment of the other dissociation fragment – the aS ground state atom – based on the two dye laser experiment. The first laser excites the dissociating A1Σu+ ↔ X1Σg+ Cr2 transition and the second laser the aS REMPI. Fig. 4(a) shows the aS REMPI spectrum, recorded from the dissociation fragment. The fact that aS atoms are detected by the differential scheme proves the proposed predissociation mechanism. The spectrum is compared to the reference spectrum obtained from single atom doped HeN (Fig. 4(b)).21 The spectral similarity of the broad structure, which represents the in-droplet broadened y5P2,3,4 → a5S1 transition, indicates a well defined separation after dissociation of the ground state atom from the a5S2 atom on the surface. If both dissociation fragments resided inside the HeN, one would expect a spectral line shift like it was observed for Mg atoms (Δν/ν ≈ 2%).47 Hence, the spectral similarity of a5S3 atoms from dissociated Cr2 and original a7S3 atoms is a further indication of the interior location of the ground state fragment and the surface location of the a5S2 atom fragment.

To show the stable position of both fragments after dissociation, the probe laser was time-delayed (Δτ ≈ 20 ns) to avoid a temporal overlap between the dissociation pump laser and the ionizing probe laser pulse. The pump–probe delay has no significant influence on the spectrum. Hence, no indication for a recombination or spatial approach between both neutral fragments within the delay time is found. We thus conclude that the Cr a5S2 atom is stable on the HeN surface, where it is available for further excitation.

It is important to point out that despite the attractive interaction between the center-located ground state and the surface-located metastable state atoms, molecule formation does not occur. This indicates an energy barrier between the two atoms as a result of the confining potential energy curve of the solvated atom and the attractive Cr2 (a5S1–a5S2) potential energy curve, in a similar way as it was recently calculated for the Rb–Xe van der Waals system in a He506 droplet.27

### 4.3 Cr2,3,4+ from Cr2 and small Crn cluster

The characteristic atomic y5P1,2,3 → a5S2 transition peaks are also present in the REMPI spectra employing Cr2+ detection (see Fig. 2(a) and 3), with exactly the same position and shape as for Cr+ detection. This is very surprising, because the Cr atom y5P1,2,3 → a5S2 transition obviously has to be a step in the ionization process leading from Cr2 to Cr2+.

The observation becomes plausible when we build on the results from before, namely that upon Cr2 dissociation the a5S2 atom is residing on the surface while the a5S3 atom remains solvated inside the droplet. After resonant ionization of the surface atom with aS REMPI (see Fig. 1), the Cr+ ion has a fair chance to be pulled inside the HeN32,34,45 where it recombines with its original partner to form Cr2+.

To verify Cr2 doped HeN as the parent species of the Cr2+ ions, the Cr2+ ion yield was recorded while increasing the Cr oven temperature, and hence, the probability distribution for the formation of different Crn cluster sizes.26 The Cr2+ ion signal shows a clear dependence on the pick-up oven temperature. For reference, the number of Cr2,3,4+ ions obtained with electron impact ionization of the doped HeN, is also recorded in dependence on the pick-up oven temperature. Keeping possible fragmentation in mind, the comparison allows an assignment of the Cr2+ y5P3 → a5S2 signal (Fig. 2(a), blue line, 22 257 cm−1) to Cr2 doped HeN.

We mention that in addition to the Cr + Cr+ → Cr2+ recombination, another recombination path is possible, although unlikely. According to DFT calculations,45 Cr+ y5P+ atoms have a stable position on the surface and inside the HeN, with a small energy barrier in between. Hence, the Cr+ (y5P+) atom might move back inside the droplet to form an excited Cr2 molecule with the Cr (a5S1) ground state atom. As above, evaporation of a certain fraction of the droplets and subsequent two photon ionization would lead to Cr2+ detection. Irrespective of the recombination path, the recombination process is restricted to the volume of the droplet. The Cr2+ detection after aS REMPI proves the location of both dissociation fragments to be on the droplet.

Now we consider HeN doped with Cr3,4 clusters. Fig. 3 shows that at the Cr atom y5P3 → a5S2 transition also Cr2+ and Cr4+ ion clusters are detected. This means, HeN doped with Cr3 and Cr4 (probably also Cr5, –) clusters can undergo a similar excitation process as Cr2+, including the characteristic aS REMPI of a surface located atom and subsequent recombination with the solvated fragments. Because of the strong Cr2 bonding nature, Cr3 is composed of a dimer plus a loosely bonded atom48,49 and Cr4 of two dimers with strong intradimer but weak interdimer bonding.50 This dimerization effect controls the cluster growth up to Cr31, yielding similarities between the ground state photoabsorption spectra of dimers and small chromium clusters.50–52 The detection of Cr2+ and Cr3+ therefore suggests that the Cr2 dissociation process and surface migration of one atom is not disturbed by the presence of a further Cr atom or dimer in the droplet.
4.4 Cr₂ excitation spectrum

Now we turn to the Cr₂ $A'$ $\Sigma_{u}^+$ $\leftrightarrow X'$ $\Sigma_{g}^+$ excitation band, in particular to the broad band stretching from 21,500 to 22,750 cm$^{-1}$ that is present in the Cr$^+$ signal of the one-color REMPI scheme, shown as black curve in Fig. 2(a). We will now show that the signal of the broad band can be assigned to Cr₂ molecules and arises from direct multiphoton ionization and subsequent dissociation to Cr and Cr$^+$, an ionization path that does not include Cr₂ predissociation. Outside the spectral regions of the $y^{2}P_{1,2,3} \leftrightarrow a^{2}S$ transitions, i.e., without resonance enhancement, the ionization probability of both Cr atoms resulting from dissociation is estimated to be lower than a resonant ionization via vibronic levels of the Cr₂ molecule before dissociation (see Fig. 1). Hence, we account resonant multiphoton ionization of Cr₂ in combination with efficient dissociation to Cr and Cr$^+$ to be responsible for the observed signal.

It is instructive to compare the band obtained with one-color REMPI (Fig. 2(a)) to Fig. 2(b), which shows the Cr$^+$ signal of the two-laser $a'S$ REMPI scheme where the dissociating laser is scanned. It is important to recall the following two points: (1) the $a'S$ REMPI is obtained from resonant ionization of the center located ground state atoms that are produced via Cr₂ predissociation. (2) The $a'S$ REMPI is exclusively sensitive to Cr₂ molecules and not to larger Crₙ clusters, as discussed in Section 4.2. Therefore, Fig. 2(b) represents the Cr₂ excitation spectrum, which shows good overall agreement with the one-color REMPI band in Fig. 2(a). The extra features in the one-color REMPI spectrum in Fig. 2(a) with respect to the $a'S$ REMPI in Fig. 2(b) indicate the presence of resonances in this ionization path. This interpretation is supported by monitoring the one-color REMPI Cr$^+$ yield at the spectral positions marked with asterisks in Fig. 2(a) while increasing the evaporation oven temperature and comparing this signal to electron bombardment ion yield, as above. It is found that at all three spectral positions the Cr$^+$ ion signal can be assigned predominantly to Cr₂ and not to larger Crₙ cluster.

The predissociation process, which is observed by the formation of ground state $a'S$ and metastable state $a'S$ atoms, has been reported for the $A'S_{u}^+ \leftrightarrow X'S_{g}^+$ excitation in literature.$^{17,40,41,43}$ We note that the Cr₂ might additionally be formed in higher multiplicities as it was observed for Ag₂ in Hg N$^{53}$ However, currently we cannot identify other multiplicities. For the strongly bound Cr₂ singlet ground state (binding energy = 1.42 eV (11 450 cm$^{-1}$))$^{38}$, the majority of droplets are expected to survive the formation for a size distribution maximum of $N$ = 6300 and an energy of 5 cm$^{-1}$ dissipated per evaporated He atom. In gas phase, the $A'S_{u}^+ \leftrightarrow X'S_{g}^+$ Cr₂ transition is observed at an energy of 21 751 cm$^{-1}$ (0-0 band, 2.7 eV)$^{41,43,54}$ Cr₂ excitation spectra in other solid rare gas matrices yield usually shifted and broadened spectral features, but cannot provide a consistent picture about the magnitude of the shift.$^{40,51,52,55,56}$

5 Conclusions

In conclusion, we have investigated the electronic $A'S_{u}^+ \leftrightarrow X'S_{g}^+$ excitation of Cr₂ molecules located inside HeN with resonant multiphoton ionization spectroscopy, applying one- and two-laser schemes. The ionization spectra show, in addition to the droplet broadened molecular excitation structure, three pronounced peaks which we allocate to the $y^{2}P_{1,2,3} \leftrightarrow a^{2}S$ Cr atomic transitions. We conclude that the Cr₂ molecule dissociates upon excitation$^{41}$ into a solvated, ground state ($a'S$) atom and a surface-located, metastable ($a'S$) atom. The latter is ionized by resonance-enhanced multiphoton ionization. The conclusion is based on (a) the $y^{2}P_{1,2,3} \leftrightarrow a^{2}S$ transition line shape (50 cm$^{-1}$ width and a 5 cm$^{-1}$ blueshift of the onset with respect to the bare atom lines), (b) a two laser experiment with REMPI detection of the produced $a'S$ atoms, and (c) theoreetical simulations of the $a'S$ atom location.$^{45}$ Surprisingly, we detect these three $y^{2}P_{1,2,3} \leftrightarrow a^{2}S$ atom peaks not only at the Cr$^+$ ion mass, but also at Cr₂,3,4 masses. Cr₂ detection demonstrates that upon photoionization, a fraction of the surface atoms migrates back inside the droplet to recombine with their ground state counterpart. The Cr₃,4$^+$ detection indicates that this pairwise dissociation and surface migration process is not disturbed by the presence of additional Cr atoms, a fact which we attribute to the strong dimerization effect in small Cr clusters.$^{30}$

The elucidated mechanisms open a possibility for photoinduced chemistry mediated by HeN. Chemical reactions in HeN doped with both surface located (e.g., alkali-metal or alkaline earth metal atoms) and solvated species could be triggered by photoexcitation of the latter to a surface located state. The bond formation can then be followed, e.g., with time-resolved femtosecond spectroscopy.

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