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CORRECTION

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Correction: Theoretical study and design of multifunctional phosphorescent platinum(II) complexes containing triarylboron moieties for efficient OLED emitters

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Correction for 'Theoretical study and design of multifunctional phosphorescent platinum(II) complexes containing triarylboron moieties for efficient OLED emitters' by Yong Wu *et al.*, *Phys. Chem. Chem. Phys.*, 2015, DOI: 10.1039/c4cp04919e.

The authors regret that some values of the transition dipole moments, $\mu(S_n)$, in Table 6 of the article are incorrect. The revised version of Table 6 is as shown below. The corrections of $\mu(S_n)$ have no effect on the other calculations, discussions and conclusions.

Table 6 Transition dipole moments $\mu(S_n)$ [Debye] for S_0–S_n transitions, singlet–triplet splitting energies $\Delta E(S_n–T_1)$ [eV] and the SOC matrix elements $\langle T_1|H_{SOC}|S_n\rangle$ [cm^{-1}] of **1–6** at their respective T_1 optimized geometries obtained from SOC-TD-B3LYP + COSMO calculations in CH_2Cl_2 solution

	1			**2**			**3**		
S_n	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$
S_1	3.58	0.423	36.6	6.68	0.419	2.28	6.03	0.383	17.6
S_2	2.37	0.875	349	1.42	0.990	106	1.82	0.682	144
S_3	1.32	0.919	145	1.83	1.275	86.5	2.09	0.995	127
S_4	1.00	1.041	215	0.53	1.289	76.2	1.23	1.051	24.0
S_5	0.66	1.070	739	0.78	1.402	76.4	1.04	1.194	373
S_6	1.76	1.128	132	0.44	1.439	276	1.90	1.262	40.3
	4			**5**			**6**		
S_n	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$	$\mu(S_n)$	$\Delta E(S_n–T_1)$	$\langle T_1\|H_{SOC}\|S_n\rangle$
S_1	2.87	0.539	37.3	3.64	0.457	28.4	2.01	0.348	143
S_2	4.76	0.742	33.4	1.67	0.690	297	3.04	0.519	273
S_3	0.19	0.843	45.0	1.68	0.897	82.8	0.67	0.832	952
S_4	3.15	1.051	25.3	1.73	1.008	97.9	0.59	0.972	168
S_5	0.72	1.438	136	1.63	1.088	340	1.52	0.998	33.0
S_6	1.66	1.564	43.2	0.82	1.104	650	2.38	1.096	129

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

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