CrystEngComm

CORRECTION

View Article Online
View Journal | View Issue

Cite this: *CrystEngComm*, 2015, **17**, 9240

DOI: 10.1039/c5ce90193f

www.rsc.org/crystengcomm

Correction: Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins

Lin Chen, Minling Fang, Chengzhan Liu, Xianchun Liu* and Shuangxi Xing*

Correction for 'Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins' by Lin Chen *et al.*, *CrystEngComm*, 2015, **17**, 4343–4348.

Original version: in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0685 g of $NiCl_2{\cdot}6H_2O$ was dissolved in 30 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of $N_2H_4{\cdot}H_2O$ (500 μL).

Revised version: in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0137 g of $NiCl_2{\cdot}6H_2O$ was dissolved in 6 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of $N_2H_4{\cdot}H_2O$ (100 μL).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

Faculty of Chemistry, Northeast Normal University, 5268 Renmin Street, Changchun, PR China 130024. E-mail: xingsx737@nenu.edu.cn; Tel: (+86) 85099657