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## Correction: Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins

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 Correction for 'Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins' by Lin Chen *et al.*, *CrystEngComm*, 2015, 17, 4343–4348.

**Original version:** in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0685 g of NiCl<sub>2</sub>·6H<sub>2</sub>O was dissolved in 30 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O (500 μL).

**Revised version:** in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0137 g of NiCl<sub>2</sub>·6H<sub>2</sub>O was dissolved in 6 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O (100 μL).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

