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Correction: Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins

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 Correction for 'Manipulating the nickel shape and catalytic performance: from spheres to chains to urchins' by Lin Chen *et al.*, *CrystEngComm*, 2015, 17, 4343–4348.

Original version: in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0685 g of $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ was dissolved in 30 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$ (500 μL).

Revised version: in the sub-section titled **Synthesis of nickel particles:** In a typical synthesis, 0.0137 g of $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ was dissolved in 6 mL of EG. After that, a certain volume of NaOH aqueous solution (1 M) was added into the above solution under continuous stirring followed by the addition of $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$ (100 μL).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

