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A systematic approach to diverse, lead-like scaffolds from α, α -disubstituted amino acids[†]

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A powerful strategy for the efficient lead-oriented synthesis of novel molecular scaffolds is demonstrated. Twenty two scaffolds were prepared from just four α -amino acid-derived building blocks and a toolkit of six connective reactions. Importantly, each individual scaffold has the ability to specifically target lead-like chemical space.

There is an intrinsic link between the molecular properties of initial screening compounds and the likelihood of their successful progression towards drug candidates.^{1,2} Since both molecular weight and lipophilicity tend to increase during the lead optimisation phase,³ controlling the molecular properties of lead molecules is crucial if the final compounds are to remain in drug-like space. However, a recent survey concluded that just 2.6% of 4.9 M commercially-available molecules were lead-like,⁴ even before compound diversity⁵ was considered. It has also been shown that compounds with higher fractions of sp³-hybridised atoms have lower attrition rates than less saturated compounds,² and thus may serve as better leads. Obtaining large numbers of diverse, highly three-dimensional, lead-like small molecules is thus a major challenge in maintaining high-quality screening collections. The concept of lead-oriented synthesis has been proposed⁴ to frame the challenge of delivering diverse molecules with specific molecular properties of direct relevance to the drug discovery and development process.^{4,6,7} Lead-oriented syntheses should, *inter alia*,⁴ focus on the diversity of structures generated, synthetic efficiency, tolerance to polar functionality, and not be susceptible to 'Log P drift', the phenomenon whereby higher failure rates of more polar compounds in a given array synthesis leads to an increase in mean and median $\log P$ for final libraries versus that designed. The significant academic challenge laid down by such demands has

recently begun to prompt the syntheses of specific classes of lead-like molecules. $^{\rm 8}$

Making diverse compound libraries is not necessarily difficult *per se*, but without careful planning comes at the expense of synthetic efficiency and, until recently,^{8a} has lacked a lead-oriented focus. In order to realise efficient lead-oriented synthesis, our approach is to exploit small building blocks that contain multiple chemically-orthogonal functional groups; systematic exploitation of a minimal toolkit of reactions for the pairwise coupling of the functional groups present would rapidly deliver diverse scaffolds that, on decoration of the newly formed or residual functional groups, should target broad regions of lead-like chemical space.

To validate our strategy, we chose readily available⁹ α -allyl α -amino acid derivatives **1** as the substrates, the amine functionality of which would then be armed with a functional group (blue) to yield a trifunctional intermediate **2**, tuning the precursor for cyclisation with either the adjacent ester (green; \rightarrow **3**) or alkene (red; \rightarrow **4**) (Scheme 1). Variation of the resulting scaffold should be



Scheme 1 Overview of the modular synthetic approach. Panel A: attachment of alternative functional groups (blue) to α -allyl α -amino acid building blocks 1 yields intermediates 2 which may cyclise to yield alternative scaffold topologies 3 or 4. Panel B: structures of the specific building blocks used.

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Scheme 2 Synthesis of diverse lead-like molecular scaffolds. Coloured boxes correspond to different cyclisation reactions, new bonds/groups are shown in red. Panel A: synthesis of scaffolds from **1a**. Panel B: extension to alternative scaffolds using building blocks **1b–d**. Standard methods: A: TFA, CH₂Cl₂; B: (i) I₂, THF–H₂O; (ii) NaN₃, DMF; C: (EtO₂C)NCO, CH₂Cl₂; D: (i) I₂, Li[Al(O^tBu)₄], THF, -5° C; (ii) NaN₃, DMF; E: NaOMe, 85:15 MeOH–PhMe, 65 °C; F: *N*-Boc-glycinal, NaBH(OAc)₃, CH₂Cl₂, 4 Å MS; G: (i) TFA, CH₂Cl₂; (ii) Cs₂CO₃, DMF, Δ ; H: 2-Br-C₆H₄CHO, NaBH(OAc)₃, CH₂Cl₂, 4 Å MS; I: 5 mol% Pd(PPh₃)₄, Et₃N, MeCN, μ w, 125 °C; J: allyl bromide, K₂CO₃, DMF; K: 2.5 mol% Grubbs II, TSOH, PhMe, Δ ; L: EDCl, *N*-Boc-glycine, Et₃N, CH₂Cl₂. ^a See ESI[†] for variation from standard methods. ^b Product not purified before subsequent reaction. ^c Yield over 2 steps. ^d 92:8 mixture with the regioisomeric azocane. ^e The regioisomeric azocane **14c** also isolated (31% yield).

possible by exploiting acyclic or cyclic amino acid derivatives (*e.g.* **1a–d**) as substrates; by varying the appended functionality (blue); and by varying the cyclisation reaction.

Firstly, allylation of commercial *N*-Boc-protected amino acid esters furnished the trifunctional building blocks **1**.⁹ Additional intermediates bearing alternative *N*-substituents were prepared to tune the precursors **1** for cyclisation (Scheme 2, Panel A). Removal of the Boc-protecting group was followed by formation of acyl ureas **6** (treatment with ethyl isocyanatoformate) or by alkylation to give protected **1**,2-diamines **9**, *o*-bromobenzylamines **12** and allylamines **15**. We then used these substrates to synthesise cyclic molecular scaffolds, by exploiting the use of a toolkit of three broad reaction types, all targeting medicinally relevant motifs.

1,2-Amino alcohols, 1,2-diamines and their functionalised derivatives are prevalent in many bioactive molecules.¹⁰ We therefore

explored introduction of these motifs by pairwise coupling of the π -nucleophilic allyl group with the nucleophilic *N*-acyl substituents of **1** and **6** (carbamate¹¹ or urea¹² respectively) upon activation with iodine; subsequent treatment with NaN₃ gave cyclic carbamates **5a–5d** and the biologically relevant¹³ ureas **7a–c**.

Pairwise coupling of the electrophilic carboxylate ester with nitrogen-centred nucleophiles led to hydantoins **8a,8c–d** (a motif known for anticonvulsant biological activity, *e.g.* phenytoin) from acyl ureas **6** and lactams **10a–b**/diketopiperazine **11d** from protected amines **9**. Finally, ring closure between the *N*-substituent and the C2-allyl group could be facilitated through transition metal catalysis: Pd-catalysed Heck cyclisations of substrates **12** gave the azepanes **13a–d** and the azocane **14c**. Similarly, Ru-catalysed ring-closing metathesis allowed the synthesis of the indolizidine alkaloid-like tetrahydropyridines **16a** and the congeners **16b–d**. In



Fig. 1 Analysis of molecular properties and shape diversity. A virtual library of 1110 compounds derived from the 22 scaffolds (orange) compared with 90 911 randomly-selected compounds from the ZINC database (blue). Panel A: distribution of molecular properties (black rectangle: guide properties used to delineate lead like space). Panel B: normalised principal moment of inertia plot.

this way we prepared 22 distinct scaffolds from the four building blocks **1a–d** in a total of just 49 synthetic operations‡ using a toolkit consisting of only six reaction methodologies and one *N*-capping event.

To assess the novelty of these scaffolds, substructure searches against the ZINC and CAS databases were performed. A total of 37 variants of the scaffolds (with protecting groups removed and allowing for modifications available through standard functional group interconversions; see ESI[†] for full details) were screened against the ZINC database of 9039756 commercially available compounds. This returned only 28 hits, all of which were derived from just two of the scaffolds (10a/16a). A further analysis of the Murcko assemblies of all scaffolds (with alpha-attachments) against a random 5% of the ZINC database again only returned hits (4) against the same two scaffolds 10a/16a. Thus, it is clear that the scaffolds display a very high degree of novelty relative to commercial collections. Finally, the scaffolds were also assessed for skeletal diversity: using an hierarchical approach^{14,15} it was found that each scaffold comprised a unique molecular framework at the GNB level.§ Thus, the scaffolds are not simple derivatives of each other, but represent a skeletally diverse collection.

The potential of the scaffolds to provide access to lead-like screening compounds was demonstrated by enumeration of a virtual library of 1110 compounds from the deprotected scaffolds and 80 typical medicinal chemistry capping groups (see ESI†). Assessment of the lead-likeness of the virtual library (Fig. 1, Panel A) showed that 66% of compounds survived filtering by molecular size (14 \leq heavy atom count \leq 26; μ = 22.8, σ = 3.57), lipophilicity ($-1 < A \log P < 3$; μ = 0.38, σ = 1.38) and structural filters (see ESI†). By comparison, just 23% of the ~9 M compounds from the ZINC database¹⁵ survived this filtering process, with most compounds lying well outside lead-like chemical space (heavy atoms: μ = 25.9, σ = 5.4; $A \log P$: μ = 1.7, σ = 2.9). Remarkably, using the same set of capping groups, each of the 22 scaffolds allowed significant lead-like chemical space to be targeted. In addition to this *in silico* study, we have shown practically that decoration of exemplar scaffolds is possible (ESI†).

The shape diversity of the virtual library was compared with that of 90 911 randomly-selected compounds from the ZINC database (Fig. 1, Panel B). For each compound, the two normalised principal moments of inertia values were determined for a low energy conformation.¹⁶ The virtual library was significantly more three-dimensional (Fsp³: $\mu = 0.57$) than the commercially-available compounds (Fsp³: $\mu = 0.33$).

In this study we have shown that careful selection of small, polyfunctional substrates facilitates an efficient approach to the synthesis of molecular scaffolds that are novel, diverse, and can specifically target lead-like chemical space. This was achieved using a toolkit of just six cyclisation methodologies, leading to 22 scaffolds in only 49 synthetic operations. In particular, the efficient and diverse coverage of 3D space from such a small set of reagents is notable. This general approach should be applicable to many classes of polyfunctional substrate, enabling the more efficient exploration of lead-like chemical space.

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Notes and references

Defined as a process conducted in a single reaction vessel. § See the scaffold tree in ESI[†] for further detail.

- 1 (a) M. C. Wenlock, R. P. Austin, P. Barton, A. M. Davis and P. D. Leeson, *J. Med. Chem.*, 2003, **46**, 1250; (b) P. D. Leeson and B. Springthorpe, *Nat. Rev. Drug Discovery*, 2007, **6**, 881.
- F. Lovering, J. Bikker and C. Humblet, *J. Med. Chem.*, 2009, **52**, 6752.
 (*a*) T. I. Oprea, A. M. Davis, S. J. Teague and P. D. Leeson, *J. Chem.*
- *Inf. Comput. Sci.*, 2001, **41**, 1308; (*b*) G. M. Keserü and G. M. Makara, *Nat. Rev. Drug Discovery*, 2009, **8**, 203.
- 4 A. Nadin, C. Hattotuwagama and I. Churcher, *Angew. Chem., Int. Ed.*, 2012, **51**, 1114.
- 5 A. H. Lipkus, Q. Yuan, K. A. Lucas, S. A. Funk, W. F. Bartelt, R. J. Schenck and A. J. Trippe, *J. Org. Chem.*, 2008, 73, 4443.
- 6 P. MacLellan and A. Nelson, Chem. Commun., 2013, 49, 2383.
- 7 R. Doveston, S. Marsden and A. Nelson, *Drug Discovery Today*, 2014, **19**, 813.
- 8 (a) R. G. Doveston, P. Tosatti, M. Dow, D. J. Foley, H. Y. Li, A. J. Campbell, D. House, I. Churcher, S. P. Marsden and A. Nelson, Org. Biomol. Chem., 2015, 13, 859; (b) S. V. Ryabukhin, D. M. Panov, D. S. Granat, E. N. Ostapchuk, D. V. Kryvoruchko and O. O. Grygorenko, ACS Comb. Sci., 2014, 16, 146; (c) A. Borisov, V. Voloshchuk, M. Nechayev and O. Grygorenko, Synthesis, 2013, 2413; (d) T. James, I. Simpson, J. A. Grant, V. Sridharan and A. Nelson, Org. Lett., 2013, 15, 6094; (e) T. James, P. MacLellan, G. M. Burslem, I. Simpson, J. A. Grant, S. Warriner, V. Sridharan and A. Nelson, Org. Biomol. Chem., 2014, 12, 2584.
- 9 P. N. Confalone, E. M. Huie, S. S. Ko and G. M. Cole, *J. Org. Chem.*, 1988, 53, 482.

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- 10 (a) T. J. Donohoe, C. K. A. Callens, A. Flores, A. R. Lacy and A. H. Rathi, *Chem. Eur. J.*, 2011, 17, 58; (b) D. Lucet, T. Le Gall and C. Mioskowski, Angew. Chem., Int. Ed., 1998, 37, 2580.
- 11 M. Pattarozzi, C. Zonta, Q. B. Broxterman, B. Kaptein, R. De Zorzi, L. Randaccio, P. Scrimin and G. Licini, Org. Lett., 2007, 9, 2365.
- 12 M. Fujita, O. Kitagawa, T. Suzuki and T. Taguchi, J. Org. Chem., 1997, 62, 7330.
- 13 E. De Clercq, *Biochim. Biophys. Acta*, 2002, **1587**, 258.
 14 A. Schuffenhauer, P. Ertl, S. Roggo, S. Wetzel, M. A. Koch and H. Waldmann, *J. Chem. Inf. Model.*, 2007, **47**, 47.
- 15 J. J. Irwin, T. Sterling, M. M. Mysinger, E. S. Bolstad and R. G. Coleman, J. Chem. Inf. Model., 2012, 52, 1757.
- 16 W. H. Sauer and M. K. Schwarz, J. Chem. Inf. Comput. Sci., 2003, 43, 987.