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Photo-induced uncaging of a specific Re(I) organometallic complex in living cells†

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In the last decades, a large number of organometallic complexes have shown promising anti-proliferative activity towards different cancer cell lines. However, these compounds generally had low cellular uptake and low selectivity towards cancer cells over healthy cells. The use of external triggers (e.g. light, ultrasound, temperature, etc.) to modify the cytotoxic effect of a prodrug and the coupling of a targeting vector (e.g. peptides, antibodies, etc.) to a drug were found to be very successful techniques to tackle these drawbacks. Here, we envisioned combining these two methods, namely an external trigger (i.e. light activation) and a targeting vector, in an organometallic compound. More specifically, a Re(I) tricarbonyl *N,N*-bis(quinolinoyl) complex (**Re-NH₂**) was derivatised with a photo-labile protecting group (PLPG) to cage **Re-NH₂** by formation of **Re-PLPG**. For organelle/cellular specificity, **Re-PLPG** was then further coupled to a nuclear localization sequence (NLS) or a bombesin peptide derivative to give **Re-PLPG-NLS** or **Re-PLPG-Bombesin**, respectively. Photolysis experiments in PBS buffer (pH 7.4) demonstrated that **Re-NH₂** was completely photo-released from **Re-PLPG-NLS** and **Re-PLPG-Bombesin** using a very low irradiation dose (1.2 J cm⁻²). To the best of our knowledge, these are the first two examples of the selective photo-release of an intact organometallic compound from a bioconjugate. Of high interest, both derivatives showed toxicity comparable to that of cisplatin towards cervical cancer cells (HeLa) upon light irradiation, although the phototoxic index (PTI) varied greatly with the targeting peptide. The cell death mechanism of **Re-PLPG-NLS** was explored using different techniques, including fluorescence microscopy, ICP-MS, gel electrophoresis, flow cytometry and transmission electron microscopy (TEM). It could be demonstrated that HeLa cells treated with **Re-PLPG-NLS** in the dark and upon irradiation showed severe cell stress (nucleolar segregation, pyknosis and vacuolation). The data obtained from an Annexin V/propidium iodide (PI) assay indicated that, after an early apoptotic stage, the onset induced by **Re-PLPG-NLS** led to cell death, with features ascribable to late apoptosis and necrosis, which were more marked for the treatment involving irradiation.

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Introduction

Photodynamic therapy (PDT) is a medical technique that allows the destruction of target cells by irradiation of a photosensitizer (PS) with light. Offering important advantages, such as temporal and spatial control, PDT is currently used to

treat certain cancers that are readily accessible with lamps or optic fibers (e.g. some types of head and neck, skin, bladder, lung and oesophagus cancers).² Other applications of PDT – treatment of multi-drug resistant bacterial infections in particular – are also being actively explored.^{3–8} However, PDT suffers from a general drawback, namely its reliance on an oxygen-dependent mechanism. In fact, the two main PDT modes of action (type I which involves reactive oxygen species (ROS) production, and type II in which the damage is directly produced by ¹O₂) rely on the conversion of oxygen from its triplet ground state to its singlet excited state by a PS. This can be problematic since tumours, in their inner central part in particular, are often hypoxic.^{9,10} Hence, it would be extremely interesting to develop a light-mediated strategy that does not rely on the presence of oxygen. In this sense, singlet oxygen generation is not the only effect that light can induce. Chemical bonds can also be rearranged and/or cleaved upon light

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irradiation. Indeed, the activity of a compound may be modified by caging some of its functional groups with a photo-removable moiety. The release of the original compound can then be controlled by light irradiation. Such light-activated molecules have proven themselves useful tools in biological/chemical biology/medicinal chemistry studies, such as in the analysis of neurological and enzymatic processes, gene expression, cell signalling and many others.^{11,12} Several drugs and drug candidates have also been photo-caged and efficiently released upon light irradiation.^{13–19} Among them are aspirin,¹³ ibuprofen,¹³ ketoprofen,¹³ carbonic anhydrase II inhibitors,¹⁴ as well as a few anti-cancer agents such as fluorouridine,¹⁵ paclitaxel,¹⁶ tegafur,¹⁷ doxorubicin,¹⁸ and a photo-reactive DNA intercalator.¹⁹ Recently, this photo-caging strategy has also been combined with singlet oxygen generation: an anti-cancer drug candidate, combretastatin A-4, has been coupled to a porphyrin photosensitiser *via* a cleavable singlet oxygen linker.²⁰ Surprisingly, although many studies reporting the use of photo-caged organic molecules have been published, examples of the specific release of intact (organo)metallic compounds using light as a trigger are very scarce.²¹ Surely, some complexes are known to undergo light-induced reactions involving the metal centre itself. This property was used not only to enhance their cytotoxicity,^{22–39} but also to design metal-based photo-cages.^{40–44} As shown by Franz and co-workers using a photo-labile *o*-nitrophenyl moiety, the release of a labile metal complex⁴⁵ or the release of metal ions^{46,47} could also induce cytotoxicity. Of note, the photo-release of metal ions alone to explore neurological response and cell signalling is currently an intensively investigated research area.^{48–52} However, to the best of our knowledge, we were the first to report the photo-release of an intact cytotoxic substitutionally inert Ru(II) complex upon light irradiation.²¹ While most drugs on the market are organic molecules, the platinum(II)-based drug cisplatin and its derivatives oxaliplatin and carboplatin are playing a pivotal role in anti-cancer chemotherapy. In addition to purely metal coordination complexes, a significant number of organometallic compounds have been reported to be strongly toxic towards cancer cells in recent years.^{53–57} Among them, rhenium tricarbonyl complexes may not be the most numerous, but are particularly attractive, as their isostructural ^{99m}Tc analogues permit the use of imaging to observe the compound's behaviour *in vivo*. In this work, we are building on the information gained during a previous study on a Re(I) tricarbonyl bis(quinolinoyl) complex that showed a strong PDT profile¹ to achieve multi-modal activity. We observed that synthetic modifications of the alkyl chain of these Re(I) complexes (see Scheme 1) strongly influenced both their dark and light-mediated cytotoxicities.^{1,58} Hence, photo-caging and peptide conjugation of these compounds could provide temporal and spatial control. Moreover, we envisaged boosting the uptake of our derivative by malignant tissues *via* conjugation of targeting peptides. Herein, we report the synthesis and characterization of photo-activated targeted Re(I) tricarbonyl *N,N*-bis(quinolinoyl) bioconjugates, as well as their photo-products. Their bioactivity towards cervical (HeLa) and prostate (PC-3) cancer cells, as well as towards normal lung

fibroblasts (MRC-5), upon light activation is described. The possible modes of cytotoxic action are also explored in depth.

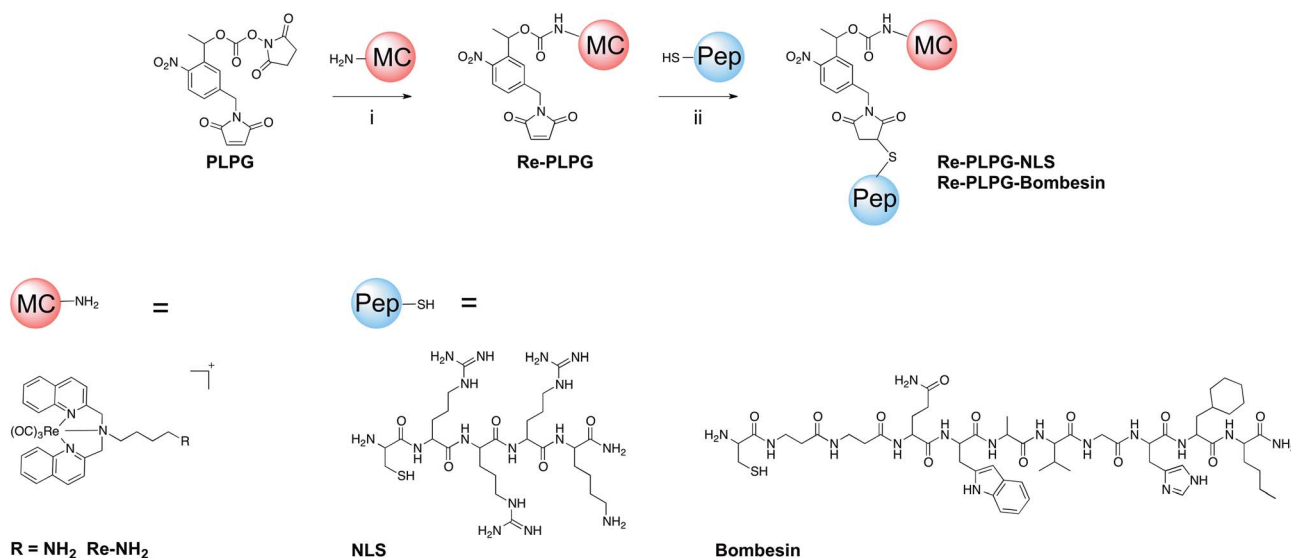
Results and discussion

Synthesis and characterization of the organometallic caged bioconjugates

Re-NH₂⁵⁹ was caged with an *o*-nitrophenyl-based photo-linker, namely the Dmochowski photo-labile protecting group (PLPG).⁶⁰ This type of photo-linker has been systematically studied and applied in numerous biochemical systems.^{12,61–64} Besides, it allows coupling of additional molecules – a property we used to attach targeting peptides to our caged Re(I) derivative. More specifically, the primary amino group of **Re-NH₂** was first reacted with PLPG *via* *N*-hydroxysuccinimide ester coupling to give **Re-PLPG** (Scheme 1).⁶⁰ For this purpose, the original Dmochowski protocol was slightly modified. The phosphate buffer was replaced by organic solvents to solubilise the organometallic compounds, and a weak base was added to catch the protons liberated during the reaction. Although the presence of a base can induce maleimide ring opening,⁶⁰ only a small amount of by-product was observed. The formation of **Re-PLPG** was confirmed by ESI-MS, which showed the expected mass of [M]⁺ at *m/z* 943.2, as well as the isotopic pattern characteristic of rhenium (Fig. S3 in the ESI†). In the ¹H NMR spectrum, a new peak corresponding to the amide proton of the newly formed carbamate bond appeared at approximately 5.7 ppm. As expected, a shift of the signals of the protons in the immediate vicinity of the carbamate was observed. In **Re-PLPG**, the signals of the methyl group of the PLPG moiety and its neighbouring tertiary carbon proton shifted upfield compared to PLPG. Slight (both upfield and downfield) changes in chemical shifts were also observed for the protons of the organometallic part of **Re-PLPG** compared to **Re-NH₂** (Fig. S5–S7†).

The photo-caged complex **Re-PLPG** was then conjugated to two different targeting peptides *via* Michael addition of the cysteine thiol of a peptide to the maleimide bond of the PLPG (Scheme 1). **Re-PLPG** was attached to (1) a nucleus localization signalling (NLS) peptide to target the compound to the nucleus in close proximity to DNA and (2) a bombesin derivative^{65,66} to improve the uptake of the compound by cancer cells over-expressing the gastrin-releasing peptide receptor (GRPR). A cysteine residue was added to each peptide to allow conjugation to the PLPG maleimide. After preparative HPLC purification and lyophilisation, light yellow solids of **Re-PLPG-NLS** and **Re-PLPG-Bombesin** were obtained. The expected masses at *m/z* 415.7 [M + 3H]⁴⁺, 553.9 [M + 2H]³⁺ and 830.3 [M + H]²⁺ (**Re-PLPG-NLS**) and at *m/z* 538.4 [M + 3H]⁴⁺, 717.4 [M + 2H]³⁺ and 1075.3 [M + H]²⁺ (**Re-PLPG-Bombesin**) were observed by ESI-MS (Fig. S9–S16†). The purity of the bioconjugates was unambiguously confirmed by LC-MS, in which two peaks in very close proximity with the same mass pattern were observed, corresponding to the two possible diastereomers formed during the Michael addition. Such behaviour was previously reported for biotin conjugated compounds photo-caged with a similar PLPG.^{18,67} Two small extra *m/z* peaks were sometimes observed





Scheme 1 Synthesis of bioconjugated organometallic photo-caged derivatives **Re-PLPG-NLS** and **Re-PLPG-Bombesin**. (i) TEA, acetonitrile; (ii) H₂O, MeOH.

in the mass spectra during LC-MS. They could be assigned to the photo-uncaged but not yet decarboxylated metal complex, as well as to the fully uncaged complex. This uncaging process clearly happens during the electrospray ionization process due to the harsh ionization conditions used for LC-MS, since the uncaged complex is eluted over a different time frame (Fig. S12 and S16–S18 in the ESI†).

Photo-release of the organometallic complex

In order to verify the photo-uncaging efficiency of our system, the photo-release of **Re-NH₂** from the two prepared bioconjugates, namely **Re-PLPG-NLS** and **Re-PLPG-Bombesin**, was then investigated in phosphate buffer solution (PBS, pH 7.4). To determine the light dose required to photo-release **Re-NH₂** from **Re-PLPG-NLS** and **Re-PLPG-Bombesin** in cells, the two compounds were irradiated using a Rayonet UV reactor (350 nm) that was later used in the cell culture experiments. Both **Re-PLPG-NLS** and **Re-PLPG-Bombesin** completely photo-released **Re-NH₂** using a very low irradiation dose of 1.2 J cm⁻² – four times lower than the dose used to completely photolyze the caged Ru complex previously reported by our group.²¹ During the irradiation, the composition of the solutions was monitored by LC-MS. As shown in Fig. 1 for **Re-PLPG-NLS**, prior to irradiation, only two peaks at 12.6–13.3 min, corresponding to the two diastereomers, were present. As the caged complex was irradiated, new peaks arose in the chromatograms. The first small peaks at 9.8–10 min were assigned to the by-products of the UV decomposition of the PLPG (Fig. S17 and S19†) that have been reported for *o*-nitrobenzyl photo-linkers.¹² The released **Re-NH₂** complex is detected at 14.5 min. In the case of **Re-PLPG-Bombesin**, the order of peak appearance changes, as bombesin is more lipophilic than NLS (Fig. S18†). The released **Re-NH₂** is eluted first, followed by *o*-nitrobenzyl moiety by-products, while the remaining **Re-PLPG-Bombesin** peak is eluted last. Photolysis quantum yields were then measured using a 355 nm laser

setup. As expected, the targeting peptide had no influence, and both **Re-PLPG-NLS** and **Re-PLPG-Bombesin** achieved a good photolysis quantum yield of 10 ± 2% (value obtained using

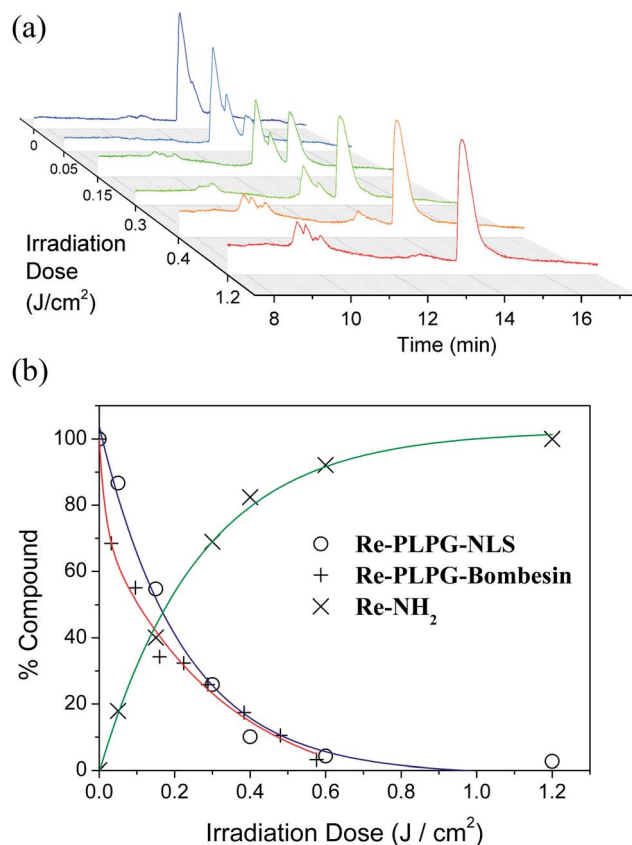


Fig. 1 (a) Steady state photolysis of **Re-PLPG-NLS** irradiated in a UV reactor (lamp max. output at 350 nm, 42 W m⁻², 30 °C); (b) exponential fit of **Re-PLPG-NLS** and **Re-PLPG-Bombesin** photo-decomposition, and of **Re-NH₂** release during photolysis.



azobenzene actinometry using the azobenzene quantum yield of 0.15, Fig. S20 and S21†).⁶⁸

(Photo-)toxicity

The cytotoxicity of the photo-caged bioconjugates **Re-PLPG-NLS** and **Re-PLPG-Bombesin**, as well as of the **Re-NH₂** complex, towards cancerous (HeLa) and non-cancerous (MRC-5) cells was evaluated in the presence and the absence of light. Cisplatin was used as a positive control (although it is not a PDT agent), in order to compare the cytotoxicity of our compounds to that of a successful metal-based drug. **Re-PLPG-Bombesin** was additionally tested on prostate cancer cells (PC-3), as the analogue of bombesin used in this study has been optimised on this cell line, which over-expresses GRPR.^{65,66} HeLa cells also over-express receptors belonging to the bombesin receptor family, namely the bombesin receptor subtype 3 (BSR-3), for which our bombesin derivative should also have some affinity.⁶⁹ As shown in Table 1, 48 h of incubation, photo-caging and peptide conjugation strongly affected both the dark and light toxicity of the compounds. In general, the bioconjugates with NLS peptide (**Re-NLS** and **Re-PLPG-NLS**) showed a marked increase in dark cytotoxicity compared to **Re-NH₂** (up to 13-fold), while the bombesin derivatives displayed only moderate toxicity towards MRC-5 cells and remained inactive up to the highest concentration used in this study (100 μ M) towards HeLa and PC-3. While the presence of the PLPG rendered **Re-PLPG-NLS** almost 2.5 times more toxic towards HeLa cells than **Re-NLS**, it did not affect the cytotoxicity of the bombesin derivative **Re-PLPG-Bombesin** towards HeLa and PC-3, and actually attenuated (up to 2-fold) the effect of the compounds on the non-cancerous cell line MRC-5.

The influence of light on the activity of **Re-PLPG-NLS** and **Re-PLPG-Bombesin** was studied by incubating cells with the compounds for 4 h. This incubation time frame has been shown to allow sufficient uptake of **Re-NH₂** for its phototoxic effect to be observed.¹ After incubation, the medium was replaced with a compound-free fresh medium, the cells were irradiated at 350 nm (UV reactor) with a low light dose of 2.58 J cm⁻², and the viability of the cells was quantified 44 h later. Light irradiation increased the toxicity of almost all tested compounds towards HeLa, PC-3 and MRC-5. The UVA irradiation by itself did not affect cell viability (Fig. S22†). Interestingly, although light significantly increased the cytotoxicity of the **Re-NH₂** complex towards HeLa (10-fold) and MRC-5 cells (2.5-fold), irradiated

PC-3 cells remained unaffected at a **Re-NH₂** concentration of up to 100 μ M. Derivatization of **Re-NH₂** with bombesin allowed much more efficient targeting of this particular cancer cell line (5–7 phototoxic index (PTI)). In addition, the bombesin conjugates **Re-Bombesin** and **Re-PLPG-Bombesin** showed good light-induced increases in cytotoxicity towards the HeLa cell line (10–20 fold). However, no significant difference (within the error range) in phototoxicity between **Re-Bombesin** and **Re-PLPG-Bombesin** – *i.e.* no influence of the PLPG – was observed. These results suggest a predominant singlet oxygen-dependent mode of cytotoxic action that, together with an acceptable PTI towards cancer cell lines and moderate toxicity towards healthy MRC-5 cells, makes these two bombesin derivatives interesting for PDT applications. In fact, some commercially available PSs have comparable PTIs (at a similar light dose of 1.5 J cm⁻²), *e.g.* Photofrin (PTI 3), Hypericin (PTI 27) and ALA-induced PPIX.⁷⁰ Only Foscan and Fospeg achieve PTIs of 268 and 4695 respectively.⁷⁰ Compared with our previously reported caged Ru(II) species that use a similar photo-caging moiety, **Re-Bombesin** and **Re-PLPG-Bombesin** possess better phototoxicity characteristics (higher PTI and higher light toxicity).²¹ However, in the area of photoactive metal compounds acting *via* singlet oxygen production or ligand lability, similar or higher (typically in the 100–200 range) PTIs with both UVA and visible light have already been achieved,^{34,35,44,71–78} although often with higher light doses (up to 100 J cm⁻²) and longer irradiation times.

Upon light irradiation, the NLS derivatives behaved quite differently to the bombesin bioconjugates. **Re-NLS** and **Re-PLPG-NLS** possess lower PTIs (2 or less), and show a more pronounced difference between the PLPG-peptide conjugate (**Re-PLPG-NLS**) and the peptide-only derivative (**Re-NLS**). In this case, the presence of the PLPG confers a certain advantage. **Re-PLPG-NLS** had a stronger (2-fold) anti-proliferative effect on malignant HeLa cells compared with **Re-NLS**. In addition, **Re-PLPG-NLS** affected non-cancerous MRC-5 cells less, compared with **Re-NLS**. To explore the origin of these differences, the effect of **Re-PLPG-NLS** on cells and cellular components was studied in-depth.

Cellular localization

To investigate the biological behaviour of **Re-PLPG-NLS** in more detail, its cellular localisation was assessed. The known luminescence^{1,59,79} of the **Re-NH₂** complex allowed us to visualize the sub-cellular localization of its bioconjugates (Fig. 2). While the

Table 1 Cytotoxicities (IC₅₀) of **Re-NH₂**, **Re-NLS**, **Re-PLPG-NLS**, **Re-Bombesin**, **Re-PLPG-Bombesin** and cisplatin towards human cell lines

Compound	HeLa dark ^a (μ M)	HeLa UV (μ M)	MRC-5 dark ^a (μ M)	MRC-5 UV (μ M)	PC-3 dark ^a (μ M)	PC-3 UV (μ M)
Re-NH₂ ¹	187.1 \pm 17.9	17.3 \pm 2.9	>100	40.3 \pm 5.4	>100	>100
Re-NLS ¹	35.1 \pm 1.8	18.3 \pm 1.4	17.8 \pm 1.8	13.0 \pm 2.5	n.d.	n.d.
Re-PLPG-NLS	14.5 \pm 5.2	9.3 \pm 0.8	36.2 \pm 0.6	20.5 \pm 5.5	n.d.	n.d.
Re-Bombesin ¹	>100	5.3 \pm 1.0	44.1 \pm 9.9	41.6 \pm 15.9	>100	13.6 \pm 1.7
Re-PLPG-Bombesin	>100	9.7 \pm 4.4	72.3 \pm 3.6	23.3 \pm 0.6	>100	19.2 \pm 2.4
Cisplatin	9.2 \pm 0.6	26.8 \pm 1.7	10.5 \pm 2.8	47.8 \pm 1.5	15.7 \pm 3.5	74.8 \pm 14.8

^a IC₅₀ after 48 h treatment and incubation; n.d. = not determined.



complex by itself⁴ and its photo-caged version **Re-PLPG** were observed exclusively in the cytoplasm (Fig. 2B and C), conjugation to an NLS peptide (**Re-PLPG-NLS**) brought the compound inside the nucleus, as designed (Fig. 2D). However, the compound did not show homogeneous distribution inside the nucleus, but accumulated in a particular sub-nuclear domain – the nucleolus. We previously reported this cellular distribution for **Re-NH₂** directly conjugated to NLS peptide.¹ Responsible mainly for ribosomal RNA transcription, processing and assembly, nucleoli also play an important role in cell cycle regulation and stress response.⁸⁰ As up-regulated ribosomal RNA synthesis and the consequent enlargement of nucleoli are characteristic of many cancer cells, these organelles have gained attention as promising drug targets in the field of anti-cancer research.⁸¹

Quantification of Re-PLPG-NLS uptake in whole cells and nucleoli

Due to possible quenching effects inside the cells, fluorescence microscopy is not always reliable when it comes to quantifying sub-cellular accumulation of a compound. Such effects with this class of compounds were recently observed by our group.^{58,82} The cellular distribution of Re complexes can be quantitatively determined by isolating the cellular compartments of interest and measuring the Re content by ICP-MS. According to this method, we incubated HeLa cells with 20 μ M of **Re-NH₂**, **Re-NLS** and **Re-PLPG-NLS** for 2 h, and collected their pellets (whole cell samples) or isolated their nucleoli through differential centrifugation (nucleoli samples), following an established method with opportune modifications (Fig. S23 in the ESI†).⁸³ Both nucleoli and whole cell extracts were then lysed, the protein content was measured, and the obtained

samples were lyophilized and analysed by ICP-MS. The Re content of the whole cell samples followed the order of accumulation **Re-NH₂** \ll **Re-NLS** < **Re-PLPG-NLS**. Interestingly, these data demonstrated very efficient uptake of **Re-PLPG-NLS** by HeLa. Indeed, in just 2 h, the cells internalised almost half of all Re available in solution. Moreover, evaluation of the accumulation of the target Re compound in the nucleoli indicated that approximately 25% of **Re-PLPG-NLS** taken up by the cells was concentrated in these organelles (Fig. S23†). The results obtained from the quantitative determination are in good agreement with the microscopic experiments and emphasize the fast and significant uptake of the Re complex.

Effect of Re-PLPG-NLS on nucleic acids

In order to achieve a better understanding of the possible mode of action of our Re compound, we investigated the possible damage to DNA or RNA induced by **Re-PLPG-NLS** in the dark and upon irradiation. Given the high positive charge of **Re-PLPG-NLS** (5+) at physiological pH and the fact that the NLS sequence is arginine- and lysine-rich, its DNA/RNA aggregating behaviour could be hypothesized. Electrostatic forces indeed play an important role in the interactions of small molecules and proteins with nucleic acids, due to the polyanionic nature of the latter. Arginine and lysine groups in peptides and proteins are heavily involved in H-bonds in DNA/RNA-protein complexes. Most of the small molecule compounds targeting nucleic acids carry multiple positive charges as well.^{84,85} DNA photo-cleavage experiments were performed on a closed circular plasmid DNA (namely pcDNA3, see Fig. 3A). The DNA was incubated with various concentrations of **Re-PLPG-NLS**, and irradiated with UVA for 10 minutes. **Re-PLPG-NLS** cleaved the plasmids upon light irradiation in a concentration-dependent manner, converting the supercoiled DNA form (intact) to a circular relaxed/nicked form (damaged). Linearization of pcDNA3 was observed at a concentration as low as 10 μ M, and complete conversion of the supercoiled to the nicked configuration occurred at a concentration of 50 μ M (Fig. 3A). DNA incubated with the rhenium bioconjugate in the dark did not show a significant effect up to the highest concentration used in this study (50 μ M). It was also observed that the plasmid treated with **Re-PLPG-NLS** displayed a smear band for high concentrations of **Re-PLPG-NLS** in the dark. This effect is due to NLS/DNA interactions, which produced the deformed band, as confirmed by DNA photo-cleavage experiments on NLS alone (see Fig. S24 in the ESI†).

As RNA is one of the major components of nucleoli, we additionally investigated the effect of **Re-PLPG-NLS** on a 638 nucleotide long RNA sequence (D135-L14 ribozyme derived from the *Sc.ai5 γ* group II intron of *S. cerevisiae*).⁸⁶ Denaturing polyacrylamide gel electrophoresis (Fig. 3B) revealed a considerable decrease in intensity and shift of the RNA band due to **Re-PLPG-NLS**/RNA interactions and formation of non-migrating RNA agglomerates (Fig. S26B†) at high concentrations of **Re-PLPG-NLS** (100 μ M). This effect occurred both in the absence and the presence of light, and no difference was observed for UVA-irradiated samples. **Re-NH₂** or **Re-PLPG** alone did not

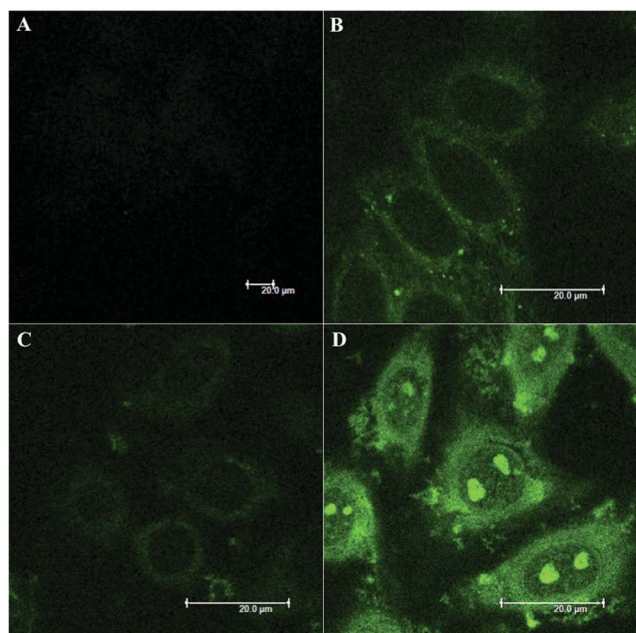


Fig. 2 Fluorescence microscopy of HeLa cells: (A) control; (B) **Re-NH₂**; (C) **Re-PLPG** and (D) **Re-PLPG-NLS** (50 μ M, 2 h).



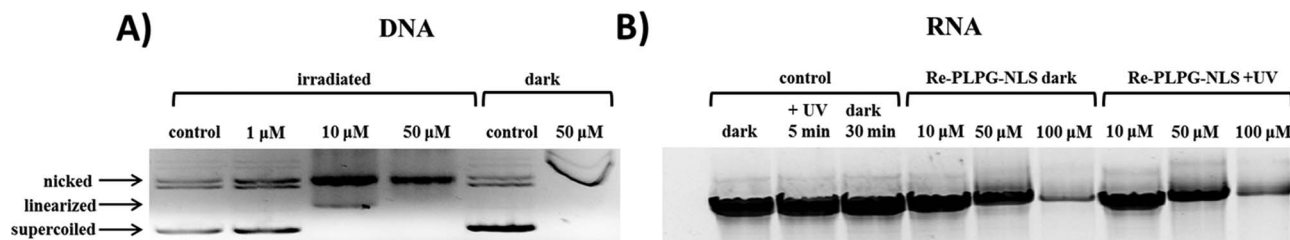


Fig. 3 (A) Electrophoretic resolution of pcDNA3 on 0.8% agarose gel-treated **Re-PLPG-NLS** at increasing concentration; (B) denaturing polyacrylamide gel electrophoresis of D135-L14 ribozyme sequence from *Sc.ai5γ* intron of *S. cerevisiae* treated with **Re-PLPG-NLS** at increasing concentration in the absence and presence of UVA irradiation.

affect the RNA (Fig. S27 and S28†). Given the localization of **Re-PLPG-NLS** in RNA-rich nucleoli, it is possible that the strong dark toxicity of this compound is linked to its effect on RNA.

Morphological features induced by the targeted rhenium complex

The deep cellular onset provoked by the target **Re-PLPG-NLS** is not just related to an isolated biochemical pathway or biological structure, but is also mirrored by macroscopic morphological features. Major alterations can be conveniently monitored by transmission electron microscopy (TEM), which is a particularly powerful technique to distinguish sub-cellular compartments, hence giving precious insights into the overall cell stress response.^{87–90}

Due to its traceability through confocal microscopy, its valuable anti-proliferative effects both in the dark and upon irradiation, and its strong interaction with DNA and RNA, **Re-PLPG-NLS** was chosen for this further biological characterization. TEM images of irradiated and non-irradiated HeLa cells incubated for 2 h with **Re-PLPG-NLS** were recorded. As shown in Fig. 4, the resulting images (selection from a total pool of 40–60 images per sample, see Fig. S29 in the ESI† for other examples) suggested that **Re-PLPG-NLS** induced the onset of severe cellular stress, with features ascribable to late apoptosis and necrosis. The sudden occurrence of membrane blebs, cytoplasm condensation, organelle packaging, extended vacuolation and nuclear pyknosis indicated an initial triggering of apoptosis. However, the extremely rapid uptake and activity of **Re-PLPG-NLS** (confirmed by ICP-MS measurements, see

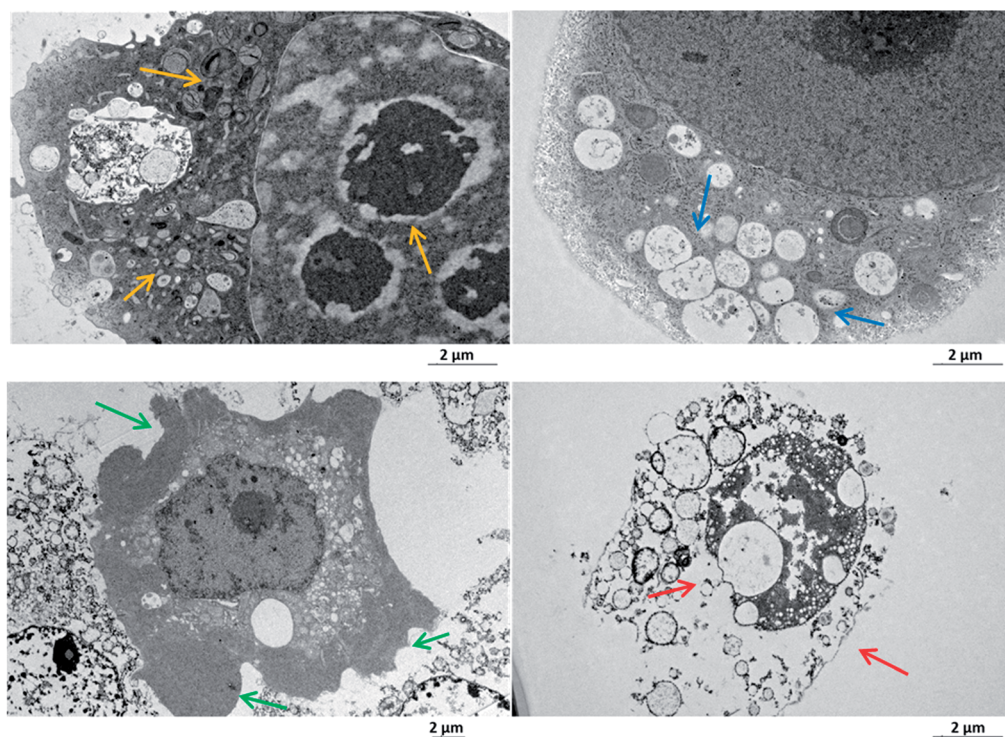


Fig. 4 Transmission electron microscopy study of HeLa cell death morphology upon irradiation treatment at 350 nm with 20 μM **Re-PLPG-NLS** for 2 h; the features detected were organelle packaging, as well as cytoplasm condensation and pyknosis (yellow arrows), extensive vacuolation (blue arrows), shrinkage and membrane blebbing (green arrows) and necrotic profiles (red arrows).

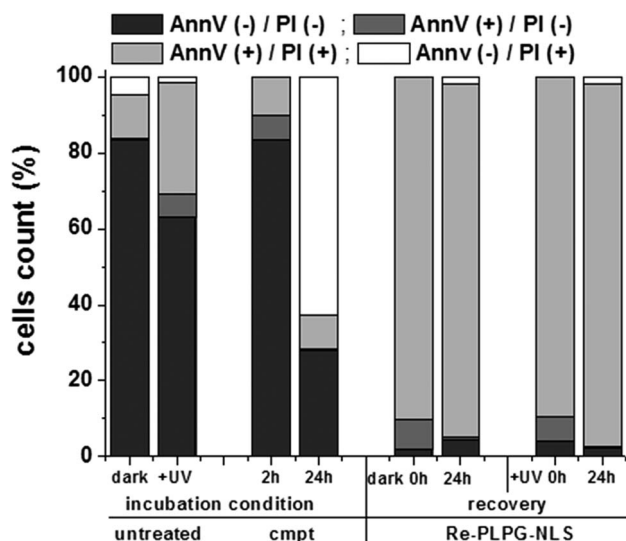


Fig. 5 Induction of apoptosis/necrosis in HeLa cells treated with 20 μ M Re-PLPG-NLS for 2 h in the dark and upon irradiation at 350 nm for 10 minutes at 0 h or 24 h recovery time; cmpt = camptothecin (5 μ M).

“Quantification of **Re-PLPG-NLS** uptake in whole cells and nucleoli” section) caused the balance to shift towards necrosis-related effects, such as loss of membrane integrity and nuclear fragmentation (see Fig. 4).

Moreover, coherent with the insights obtained from the cellular localization studies, major attention was paid to the specific effects triggered in nucleoli by **Re-PLPG-NLS**. Treatment with the rhenium conjugate influenced the nucleoli morphology. In good agreement with established data, the target sub-organelle presented a stress-altered morphology, characterized by indicators such as the formation of nuclear caps around the nucleolar body, separation of the fibrillar centre from the granular centre, unravelling of the fibrillar centre and migration of the whole nucleolar body to the peripheral part of the nucleus (Fig. S30 in the ESI†).^{91,92} All these features represent the main characteristics of nucleolar segregation, which could contribute to the initial apoptotic event, as well as to the overall cytotoxicity of **Re-PLPG-NLS**.

Mechanism of cell death induced by Re-PLPG-NLS

The mobilization of phosphatidylserine from the inner to the outer layer of the cellular membrane promotes phagocytosis and is considered a clear marker of an ongoing early apoptotic event.⁹³ Under severe stress, the loss of membrane integrity enables a large pool of solutes to penetrate the plasma membrane (which is normally impermeable to them), and represents a signal of the occurrence of necrosis. The combination of two staining methods allows evaluation of the cell death mechanism in large cell populations by flow cytometry, using the Annexin V/propidium iodide (PI) assay. The rapid triggering of cell death processes after short incubation treatment of HeLa cells with **Re-PLPG-NLS** investigated by TEM showed a combination of apoptotic and necrotic stimuli. Hence, it was of interest to study the contributions of these two

pathways to the cytotoxicity of **Re-PLPG-NLS**. In agreement with the TEM data, **Re-PLPG-NLS** induced a drastic cellular effect after only 2 h incubation with HeLa cells, with less than 10% of cells still viable (see Fig. 5 and S31 in the ESI†). The majority of the cell population displayed membrane symmetry due to phosphatidylserine migration (Annexin V positive) and membrane permeabilisation, which allowed PI internalization (PI positive). The data support the thesis that **Re-PLPG-NLS** induces severe cellular stress after a short incubation time, to which cells react by activating apoptosis. However, the very dynamic onset of these events resulted in a pronounced late apoptotic profile (see Fig. 5). These effects could be observed for both probes in the dark and upon irradiation, and were irreversible since the cells did not show any recovery over a time frame of 24 h. Camptothecin was used as a positive control.

Conclusions

In this article, we described the efficient caging of a Re(I) tricarbonyl *N,N*-bis(quinolinoyl) complex (**Re-NH₂**) with a photolinker (PLPG) to give **Re-PLPG**. The latter was further conjugated to two different peptides for potential organelle/cellular uptake (*i.e.* bombesin or NLS) to give **Re-PLPG-NLS** or **Re-PLPG-Bombesin**. Photolysis experiments showed that complete release of **Re-NH₂** could be achieved using a very low light dose (1.2 J cm⁻²). To the best of our knowledge, these are the first two examples of the selective release of an organometallic complex from a bioconjugate using light activation. A first biological screening on different cancer and non-tumorigenic cell lines demonstrated the very valuable toxicity of the target complexes, with IC₅₀ values in the range of the established anti-cancer drug cisplatin. It is noteworthy that, while **Re-PLPG-Bombesin** showed a more marked ¹O₂-dependent cytotoxicity at low irradiation energy (2.58 J cm⁻²), with a certain preference for cancer cells over MRC-5, **Re-PLPG-NLS** showed a stronger (2-fold) anti-proliferative effect on malignant HeLa cells compared with **Re-NLS**. In addition, **Re-PLPG-NLS** affected non-cancerous MRC-5 cells less, compared with **Re-NLS**. Due to this interesting behaviour, **Re-PLPG-NLS** was chosen for in-depth biological characterization. Cellular biodistribution studies using ICP-MS indicated an outstanding cellular uptake, close to 50% of all available Re, with very rapid and strong accumulation in the nucleoli. The penetration into sub-cellular compartments (nucleus/nucleoli) allowed interaction between the Re complex and nucleic acids. **Re-PLPG-NLS** was able to photo-cleave DNA and created non-migrating RNA agglomerates. This molecular dysfunction contributed to the induction of late apoptotic and necrotic effects, mirrored by morphological features such as nucleolar segregation, pyknosis and extended vacuolation, which resulted in cell death.

In conclusion, in this article, we present, for the first time, a strategy (organometallic complex – PLPG – targeting vector) to selectively release an organometallic complex upon light irradiation. Current efforts are aimed at increasing the ratio of dark vs. UV light toxicity and producing this photo-cytotoxic action in the PDT window.



Experimental section

Materials

Chemicals and solvents were of reagent grade or better, purchased from commercial suppliers and used without further purification unless otherwise specified. The plasmid pcDNA3 was obtained from Invitrogen.

Instrumentation and methods

^1H and ^{13}C NMR measurements were carried out on Bruker 400 and 500 spectrometers and referenced to residual solvent peaks. UV spectra were recorded on a Cary 50 Scan Varian spectrophotometer. ESI-MS, LC-MS and UPLC-MS were performed using a Bruker Daltonics HCT 6000 mass spectrometer. LC-MS and UPLC-MS spectra were recorded on a Waters AcquityTM system equipped with a PDA detector and an auto-sampler. LC-MS was carried out using an Agilent Zorbax 300 SB-C18 analytical column (3.5 μm particle size, 300 Å pore size, 150 \times 4.6 mm). The LC run (flow rate: 0.5 mL min^{-1}) was performed with a linear gradient of A (double distilled water containing 0.1% v/v formic acid) and B (acetonitrile containing 0.1% v/v formic acid): $t = 0$ min, 5% B; $t = 3$ min, 5% B; $t = 17$ min, 100% B; $t = 20$ min, 100% B; $t = 25$ min, 5% B. UPLC-MS was performed on an Acquity UPLC BEH C18 column (2.1 \times 50 mm, 1.7 μm). The UPLC run (flow rate: 0.6 mL min^{-1}) was carried out with a linear gradient of A (double distilled water containing 0.1% v/v formic acid) and B (acetonitrile containing 0.1% v/v formic acid): $t = 0$ min, 5% B; $t = 0.25$ min, 5% B; $t = 1.5$ min, 100% B; $t = 2.5$ min, 100% B. High resolution ESI-MS spectra were recorded using a Bruker ESQUIRE-LC quadrupole ion trap instrument. Preparative HPLC purification was carried out on a Varian ProStar system and an Agilent Zorbax 300 SB-C18 prep column (5 μm particle size, 300 Å pore size, 150 \times 21.1 mm. Flow rate: 20 mL min^{-1}). The runs were performed with a linear gradient of A (double distilled water containing 0.1% v/v trifluoroacetic acid (TFA)) and B (acetonitrile containing 0.1% v/v TFA, Sigma-Aldrich HPLC grade). Preparative runs: $t = 0$ min, 5% B; $t = 25$ min, 100% B; $t = 30$ min, 100% B; $t = 32$ min, 5% B. Photolysis quantum yields were determined using an Edinburgh LP-920 setup equipped with a Continuum Surelite laser (355 nm). Cell culture irradiation was performed using a Rayonet RPR-200 photochemical reactor with 6 bulbs (14 W each) and maximum intensity output at 350 nm. Samples were irradiated in a fluorescence quartz cuvette (width 1 cm) placed in the centre of the reactor. The light intensity at that spot, measured with an X1₁ optometer (Gigahertz-Optik), was 42 W m^{-2} . The temperature inside the reactor was 30 °C.

Synthesis and characterization

Re-NH₂. The complex was synthesised following a previously published procedure. The analytical data were in agreement with the literature report.⁵⁹

PLPG. 1-5-(1-(5-(*N*-maleimidomethyl)-2-nitrophenyl)ethanol NHS ester). The photo-linker was synthesised following a previously published procedure. The analytical data were in agreement with the literature report.⁶⁰

Re-PLPG. PLPG (30 mg, 0.072 mmol) and **Re-NH₂** (14.5 mg, 0.072 mmol) were dissolved in dry acetonitrile (1 mL). *N,N*-Diisopropylethylamine (DIPEA) (23.5 mg, 31 μL , 0.18 mmol) was then added to the mixture, which was stirred for 2 h at 35 °C. The solvent was then removed under vacuum to give a sticky brown solid, which was then purified by preparative HPLC to afford **Re-PLPG** as a light yellow solid. Yield: 58%. Characterisation data: ^1H NMR (500 MHz, CDCl_3) δ 8.47 (dd, $^3J = 9.0$, 5.5 Hz, 2H); 8.32 (dd, $^3J = 16.5$, 8.5 Hz, 2H); 7.89–7.81 (m, 5H); 7.71 (d, $^3J = 8.0$ Hz, 1H); 7.67–7.63 (m, 4H); 7.29 (dd, $^3J = 8.5$, 1.5 Hz, 1H); 6.73 (s, 2H); 6.17 (q, $^3J = 6.5$ Hz, 1H); 5.68 (m, 1H); 5.37 (d, $^2J = 18$ Hz, 2H); 5.1 (dd, $^2J = 18$, 7.25 Hz, 2H); 4.73 (d, $^3J = 6.0$ Hz, 2H); 4.39 (s, broad, 1H); 3.85 (m, 2H); 3.24–3.21 (m, 2H); 2.07 (m, 2H); 1.64 (m, 2H); 1.57 (d, $^3J = 6.5$ Hz, 3H). ^{13}C NMR (500 MHz, CDCl_3) δ 22.3, 23.0, 27.2, 39.8, 40.9, 68.0, 68.5, 68.6 (2 peaks overlapping), 120.3, 125.2, 127.7, 128.4, 128.5, 128.6, 128.7, 133.2, 134.5, 139.9, 141.6, 141.7, 142.3, 146.8, 147.1, 156.1, 160.4, 160.7, 165.0, 170.3, 194.4, 195.9. ESI-MS m/z 943.2 $[\text{M}]^+$. ESI HRMS calcd for $[\text{C}_{41}\text{H}_{36}\text{N}_6\text{O}_9\text{Re}]^+/z$ 943.20979, found 943.20958.

Peptide synthesis. The peptide synthesis was performed using solid-phase techniques following a previously published procedure.¹

NLS (Cys-Arg-DArg-Arg-Lys(CONH₂)). The peptide was synthesised following a previously published procedure. The analytical data were in agreement with the literature report.¹

Bombesin (Cys- β Ala- β Ala-Gln-Trp-Ala-Val-Gly-His-Cha-Nle-(CONH₂)). The peptide was synthesised following a previously published procedure. The analytical data were in agreement with the literature report.¹

Re-PLPG-NLS. **Re-PLPG** (86.0 mg, 0.0841 mmol) and the NLS peptide (89.5 mg, 0.107 mmol) were pre-dissolved in DMSO (100 μL), and then in a 1 : 2 : 2 mixture of methanol-acetonitrile-water (45 mL). The reaction mixture was stirred overnight at room temperature. MeOH was removed using a rotary evaporator, while water and acetonitrile were removed by lyophilisation. The compound was purified by preparative HPLC to afford a yellow-white solid. Characterisation data: ESI-MS m/z 830.3 $[\text{M} + \text{H}]^{2+}$, 553.9 $[\text{M} + 2\text{H}]^{3+}$, 415.7 $[\text{M} + 3\text{H}]^{4+}$. ESI HRMS calcd for $[\text{C}_{68}\text{H}_{92}\text{N}_{22}\text{O}_{14}\text{ReS}]^+/z$ $[\text{M} + \text{H}]^{2+}$ 830.32599, found 830.32526, calcd for $[\text{M} + 2\text{H}]^{3+}$ 553.88660, found 553.88599, calcd for $[\text{M} + 3\text{H}]^{4+}$ 415.66691, found 415.66631.

Re-PLPG-Bombesin. **Re-PLPG** (86.0 mg, 0.0841 mmol) and bombesin peptide (89.5 mg, 0.107 mmol) were dissolved in a 1 : 1 mixture of acetonitrile-water (45 mL). The reaction mixture was stirred overnight at room temperature. The solvent was removed using a lyophiliser, and the crude compound was purified by HPLC to afford a yellow-white solid. Characterisation data: ESI-MS m/z 1075.3 $[\text{M} + \text{H}]^{2+}$, 717.4 $[\text{M} + 2\text{H}]^{3+}$, 538.4 $[\text{M} + 3\text{H}]^{4+}$. ESI HRMS calcd for $[\text{C}_{68}\text{H}_{92}\text{N}_{22}\text{O}_{14}\text{ReS}]^+/z$ $[\text{M} + \text{H}]^{2+}$ 1075.42557, found 1075.42494, calcd for $[\text{M} + 2\text{H}]^{3+}$ 717.28632, found 717.28571. Anal. found: C, 49.97; H, 4.90; N, 12.01. Calcd for $[\text{C}_{90}\text{H}_{109}\text{F}_9\text{N}_{19}\text{O}_{22}\text{ReS}]$: C, 49.69; H, 4.94; N, 12.38.

Determination of photo-uncaging quantum yields

20 mM DMSO stock solutions of the caged compounds were diluted in phosphate buffer (pH = 7.4) to achieve an optical



density at 355 nm ($OD(\lambda = 355 \text{ nm})$) of approximately 0.2. Aliquots of 100 μL were irradiated at 355 nm using an Edinburgh LP-920 setup equipped with a Continuum Surelite laser. The laser was slightly misaligned to reach a suitable irradiation power. After certain irradiation intervals (number of laser shots), the solutions were transferred to amber HPLC vials with 200 μL inlets. 20 μL of the solutions were injected into an LC-MS system. The percentage of remaining compound was plotted against the number of laser shots and fitted with a single exponential decay curve:

$$C = C_0 \exp(-kx)$$

where C is the remaining concentration of the compound (percentage relative to the initial concentration), C_0 is the initial concentration (percentage), k is the rate constant and x is the irradiation dose.

The initial slope (less than 20% conversion) of the curve (S_{compound}) was determined by taking the tangent to the first (<20% decomposition) points of the exponential fit. It was then used to calculate the photolysis quantum yield by comparing it to the quantum yield of azobenzene photoisomerization. For that purpose, *trans*-azobenzene was dissolved in methanol to obtain an $OD(\lambda = 355 \text{ nm})$ of approximately 0.2. It was then irradiated under the same conditions as the compounds for the measurements. UV/vis absorbance of the azobenzene solution at 355 nm was then monitored. The remaining amount of *trans*-azobenzene was then plotted against the number of laser shots, and the initial slope of the curve ($S_{\text{reference}}$) was then obtained. The photolysis quantum yields were then calculated using:

$$\Phi_{\text{compound}} = \Phi_{\text{reference}} \times \frac{S_{\text{compound}}}{S_{\text{reference}}} \times \frac{OD_{\text{reference}}^{355}}{OD_{\text{compound}}^{355}}$$

where $\Phi_{\text{reference}}$ is the quantum yield of *trans*-*cis* isomerization of azobenzene.⁶⁸

Photolysis in UV reactor

20 mM DMSO stock solutions of the compounds to be irradiated were diluted with 4 mL of PBS buffer (pH 7.4) to obtain absorbance of 0.2 at 350 nm. The solutions were then irradiated in fluorescence quartz cuvettes (width 1 cm). At each irradiation time, the solution compositions were analysed using LC-MS. The relative concentrations of the photolysed compounds were calculated from the areas of the corresponding LC-MS peaks. Concentration % plotted as a function of photolysis time (or irradiation dose) could be fitted with a single exponential (first order kinetics law):

$$C = C_0 \exp(-kx)$$

where C_0 is the initial concentration, k is the rate constant and x is the irradiation dose.

Cell culture

Human cervical carcinoma cells (HeLa) were cultured in DMEM (Gibco) with 5% fetal calf serum (FCS, Gibco), 100 U mL^{-1}

penicillin and 100 $\mu\text{g mL}^{-1}$ streptomycin at 37 °C and 5% CO_2 . Normal lung fibroblasts (MRC-5) were maintained in F-10 medium (Gibco) supplemented with 10% FCS (Gibco), 100 U mL^{-1} penicillin and 100 $\mu\text{g mL}^{-1}$ streptomycin at 37 °C and 5% CO_2 . Prostate carcinoma cells (PC-3) were cultured in Ham's F-12K (Kaighn's) Medium (Gibco) supplemented with 10% FCS (Gibco), 100 U mL^{-1} penicillin and 100 $\mu\text{g mL}^{-1}$ streptomycin at 37 °C and 5% CO_2 .

Cytotoxicity studies

The cytotoxicity of the rhenium complexes and their bioconjugates towards HeLa, MRC-5 and PC-3 cells in the presence and absence of UV irradiation was measured using a fluorometric cell viability assay using resazurin (Promocell GmbH). Cells were seeded in triplicate in 96-well plates at a density of 4×10^3 cells per well in 100 μL medium 24 h prior to treatment. To assess cytotoxicity, cells were treated with increasing concentrations of the compounds for 48 h, whereas, to assess phototoxicity, cells were treated with increasing concentrations of the compounds for 4 h only. After that, the medium was removed and replaced by fresh complete medium prior to 10 min UVA irradiation (2.58 J cm^{-2}). Cells were then returned to the incubator for 48 h. After incubation, the medium was replaced by 100 μL complete medium containing resazurin (0.2 mg mL^{-1} final concentration). After 4 h of incubation at 37 °C (overnight incubation for PC-3 due to slower metabolism of this cell line), fluorescence of the highly red fluorescent resorufin product at 590 nm emission was quantified using a SpectraMax M5 microplate reader with a 540 nm excitation wavelength.

In vitro fluorescence evaluation

Cellular localization of the luminescent rhenium complexes and bioconjugates was assessed using fluorescence microscopy. HeLa cells were grown in 2 mL medium on 18 mm Menzel-glasser coverslips at a density of 1×10^5 cells per mL and incubated with rhenium complexes at 50 μM for 2 h. Cells were fixed in 4% formaldehyde solution and mounted on slides for viewing by confocal microscopy on a CLSM Leica SP5 microscope. The rhenium complexes were excited at 405 nm, and emission above 420 nm was recorded.

DNA photo-cleavage

The DNA photo-cleavage effect provoked by complex **Re-PLP-NLS** was investigated by electrophoresis. Supercoiled pcDNA3 plasmid (0.10 μg) was treated with increasing concentrations of the rhenium compounds in buffer (50 mM Tris-HCl, 18 mM NaCl, pH 7.2), incubated for 30 minutes at 25 °C and irradiated at 350 nm for 10 minutes (Rayomet Chamber Reactor, 2.58 J cm^{-2}). A series of negative controls, in which the plasmid was treated with NLS or **Re-PLP-NLS** in the dark at different incubation temperatures and in the presence of BstXI restriction enzyme (to visualize the linearized form), were used for comparative purposes (see Fig. S18†). After irradiation, sample loading buffer (250 mg xylene cyanol in 33 mL of 150 mM Tris-HCl buffer, pH 7.6, 0.1 M EDTA, 20% Ficoll 400 in 100 mL of H_2O) was added, and the probes were analyzed by



electrophoresis in 0.8% agarose in $1\times$ TBE at 70 V (Biorad Powerpack 1000, BioRad) for 2 h. The gel was pre-stained with $0.5\ \mu\text{g mL}^{-1}$ ethidium bromide, photographed and examined using an AlphaDigiDoc 1000 CCD camera (Buchner Biotec AG) and AlphaImager software v1.3.0.7.

RNA

Preparation of RNA. The plasmid pT7D135-L14, encoding the D135-L14 ribozyme sequence derived from the *Sc.ai5 γ* intron from *S. cerevisiae*, was digested with *Hind*III and subsequently transcribed *in vitro* with home-made T7 RNA polymerase under standard conditions⁸⁶ to give D135-L14 RNA. D135-L14 was purified by denaturing polyacrylamide gel electrophoresis.⁹⁴

Gel electrophoresis studies. D135-L14⁸⁶ RNA ($1\ \mu\text{M}$) was incubated with increasing concentrations of the rhenium compounds in buffer (50 mM Tris-HCl, 18 mM NaCl, pH 7.2) for 30 min at room temperature and irradiated at 350 nm for 5 min (Rayomet Chamber Reactor Complex, $1.29\ \text{J cm}^{-2}$). Control experiments were carried out in the dark. An equal volume of denaturing loading buffer (11.7 M urea, 40 mM Tris buffer, pH 7.5, 0.1% xylene cyanol, 0.1% bromophenol blue, 230 mM sucrose, 0.8 mM EDTA, pH 8.0) was added to each sample, and denaturing polyacrylamide gel electrophoresis was performed using 5% acrylamide in TBE $1\times$ buffer for 3 h at room temperature. RNA bands were detected by UV shadowing. Additionally, RNA bands were stained with ethidium bromide and detected by UV excitation at 260 nm (Gel Visualization System from Vilber).

Determination of the rhenium complex content in biological samples

Sample preparation for ICP-MS

Whole cells. To perform ICP-MS measurements, HeLa cells were seeded two days before treatment at a concentration of 4×10^5 cells per mL in a $75\ \text{cm}^2$ cell culture flask until 80% confluence, and incubated with the target complexes, namely **Re-NH₂**, **Re-NLS** or **Re-PLPG-NLS**, at $20\ \mu\text{M}$ for 2 hours. The medium was removed, and the cells were washed with PBS and trypsinized. After re-suspension in PBS, the pellets were collected by centrifugation at 5500 rpm for 4.5 minutes. The pellets were re-dissolved in 500 μL of PBS, lysed by a freeze-thaw cycle and treated in an ultrasonic bath (Digitana AG) for 20 minutes. The lysates were lyophilized using an Alpha 2-4 LD plus (CHRIST).

Cell nucleoli. Nucleoli of HeLa cells were collected following an established procedure with appropriate modifications.⁸¹ Briefly, HeLa cells were seeded two days before treatment at a concentration of 4×10^5 cells per mL in a $75\ \text{cm}^2$ cell culture flask until 80% confluence. Cells were treated with $20\ \mu\text{M}$ **Re-NH₂**, **Re-NLS** or **Re-PLPG-NLS** for 2 h, and the pellets were collected and washed 3 times with ice cold PBS. The pellets were incubated with lysis buffer (10 mM HEPES, 10 mM KCl, 1.5 mM MgCl_2 , 0.5 mM dithiothreitol, pH 7.9) containing a protease inhibitors cocktail (1/500 v/v, batch number: P8340, Sigma-Aldrich) for 10 minutes, the membranes were disrupted and the

cells were homogenized in a 7 mL Dounce Homogenizator with a tight pestle (type A, 20 strokes). The suspensions were centrifuged at $250g$ at $4\ ^\circ\text{C}$ in a Centrifuge 5810R (Eppendorf) for 5 minutes, and the supernatant was removed. The pellets were re-dissolved in 3 mL of a sucrose solution (0.25 M sucrose, 10 mM MgCl_2) and layered with 3 mL of a second hypertonic sucrose solution (0.35 M sucrose, 0.5 mM MgCl_2). The suspensions were centrifuged at $1450g$ at $4\ ^\circ\text{C}$ for 5 minutes. The supernatant was discarded, and the pellets were re-suspended in 3 mL of the second sucrose solution. The suspensions containing the nuclear fraction were sonicated 5–6 times in 10 s bursts (Bandelin Sonoplus GM70, 50% cycle, intervals of 10 s), layered with a third hypertonic sucrose solution (0.88 M sucrose, 0.5 mM MgCl_2) and centrifuged at $3000g$ at $4\ ^\circ\text{C}$ for 10 minutes. The pellets were re-suspended in the second sucrose solution (0.35 M sucrose, 0.5 mM MgCl_2), centrifuged at $1450g$ at $4\ ^\circ\text{C}$ for 5 minutes to obtain the pure nucleoli extracts (see Fig. S17[†]), and stored at $-80\ ^\circ\text{C}$. All the steps of the isolation procedure were monitored under a phase contrast microscope on Menzela-gläser coverslips (Olympus IX81). The collected nucleoli solutions were lyophilized using an Alpha 2-4 LD plus (CHRIST).

ICP-MS measurements. To measure the Re content, the biological samples were sent to Mikroanalytisches Labor Pascher (Remangen, Germany). The samples were dissolved in a PFA pressure system with nitric and hydrochloric acid at $180\ ^\circ\text{C}$, and Re187 (Ir193 as internal standard) was detected by ICP-MS (Perkin Elmer Elan 5000).

Transmission electron microscopy

HeLa cells were seeded in 6-well plates containing carbon-coated sapphire discs and glass cover slips at a concentration of 100 000 cells per mL and cultured at $37\ ^\circ\text{C}/5\%\ \text{CO}_2$ for one day ($>80\%$ confluence on the discs). The medium was then removed and replaced with medium containing $20\ \mu\text{M}$ **Re-PLG-NLS**. The cells were further incubated for 2 hours and irradiated at 350 nm for 10 minutes ($2.58\ \text{J cm}^{-2}$) or left non-irradiated (control). Subsequently, the cells were processed by chemical fixation and high-pressure freezing, as follows.

Chemical fixation (Fig. S23[†]). Cells were fixed for 20 minutes at room temperature with 2.5% glutaraldehyde and, subsequently, with 1% OsO_4 for 1 hour, both in PBS (pH 7.3). The cells were then incubated with 1% uranyl acetate in water for 1 hour, dehydrated in an ethanol series and embedded into Epon/Araldite.

High-pressure freezing (Fig. 4). Cells on sapphire discs (6 mm diameter) were frozen in an EM HPM100 high-pressure freezing machine (Leica Microsystems, Vienna, Austria) using the 6 mm cartridge system and a middle plate with ridge. Subsequently, the samples were freeze-substituted in an EM AFS2 unit (Leica) for 8 h at $-90\ ^\circ\text{C}$, 6 h at $-60\ ^\circ\text{C}$, 4 h at $-30\ ^\circ\text{C}$ and 1 h at $0\ ^\circ\text{C}$, with periodic temperature transition gradients of $30\ ^\circ\text{C h}^{-1}$. Afterwards, the cells were incubated for 1 hour in 1% uranyl acetate in anhydrous acetone at $4\ ^\circ\text{C}$ prior to embedding in Epon/Araldite.

Ultrathin 50 nm sections of all specimens were contrasted with uranyl acetate and lead citrate, and analysed with a Tecnai



G2 Spirit or CM 100 transmission electron microscope (FEI, Eindhoven, The Netherlands) using an ORIUS 1000 CCD camera (Gatan, Munich, Germany).

Annexin V/PI assay

The induction of apoptosis and necrosis in HeLa cells treated with **Re-PLPG-NLS** was evaluated by means of the well-established Annexin V/PI assay, using flow cytometry (BD Pharmingen, Bioscience) according to the manufacturer's instructions with minor changes (BD Pharmingen, Mat. no. 556419). Briefly, cells were seeded in a Petri dish at a concentration of 2×10^5 cells mL⁻¹ and cultured at 37 °C/6% CO₂ for 24 h. The medium was replaced with fresh medium containing 20 μM **Re-PLPG-NLS**. The cells were incubated for 2 h, and then the medium was replaced with fresh medium and the Petri dish was irradiated at 350 nm for 10 minutes (2.58 J cm⁻²). To observe the direct effects after irradiation, the cells were immediately collected for FACS analysis, whereas, for the recovery experiments, the medium was replaced and the cells were further incubated for 24 h at 37 °C/6% CO₂. The cells were then trypsinized and pelleted, washed twice with ice cold PBS, centrifuged at 250g for 5 minutes and re-suspended in binding buffer (10 mM HEPES, 140 mM NaCl, and 2.5 mM CaCl₂, pH 7.4) at a concentration of 1×10^6 cells per mL. 100 μL of the obtained suspension was transferred into an FACS culture tube (1×10^5 cells), and 5 μL of FITC Annexin V complex solution and 5 μL of a 50 μg mL⁻¹ propidium iodide (PI) solution were added. Samples were incubated for 15 minutes at room temperature (25 °C) in the dark, 400 μL of binding buffer was added, and the probes were analyzed using a CynAn ADP9 flow cytometer with the FITC (for Annexin V-FITC, excitation = 488 nm, emission = 515–545 nm) and PE-Texas Red channels (for PI, excitation = 488 nm, emission = 564–606 nm). The data were analyzed using Summit v4.3 software. Camptothecin, used at 5 μM for 2 or 24 h, and untreated cells in the dark or under irradiation were used as positive controls.

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Notes and references

- 1 A. Leonidova, V. Pierroz, R. Rubbiani, J. Heier, S. Ferrari and G. Gasser, *Dalton Trans.*, 2014, **43**, 4287–4294.

- 2 A. F. Taub, *Dermatol. Clin.*, 2007, **25**, 101–109.
- 3 D. M. A. Vera, M. H. Haynes, A. R. Ball, T. Dai, C. Astrakas, M. J. Kelso, M. R. Hamblin and G. P. Tegos, *Photochem. Photobiol.*, 2012, **88**, 499–511.
- 4 D. Vecchio, T. Dai, L. Huang, L. Fantetti, G. Roncucci and M. R. Hamblin, *J. Biophotonics*, 2013, **6**, 733–742.
- 5 J.-H. Park, M.-Y. Ahn, Y.-C. Kim, S.-A. Kim, Y.-H. Moon, S.-G. Ahn and J.-H. Yoon, *Biol. Pharm. Bull.*, 2012, **35**, 509–514.
- 6 F. Gad, T. Zahra, K. P. Francis, T. Hasan and M. R. Hamblin, *Photochem. Photobiol. Sci.*, 2004, **3**, 451–458.
- 7 X. Ragas, T. Dat, G. P. Tegos, M. Agut, S. Nonell and M. R. Hamblin, *Lasers Surg. Med.*, 2010, **42**, 384–390.
- 8 H. Deppe, T. Mucke, S. Wagenpfeil, M. Kesting and A. Sculean, *Quintessence International*, 2013, **44**, 609–618.
- 9 G. Bozzini, P. Colin, N. Betrouni, C. A. Maurage, X. Leroy, S. Simonin, C. Martin-Schmitt, A. Villers and S. Mordon, *Photodiagn. Photodyn. Ther.*, 2013, **10**, 296–303.
- 10 M. J. Brown, Tumor Hypoxia in Cancer Therapy, in *Methods in Enzymology*, ed. H. Sies and B. Brüne, Academic Press, 2007, vol. 435, pp. 295–321.
- 11 G. Mayer and A. Heckel, *Angew. Chem., Int. Ed.*, 2006, **45**, 4900–4921.
- 12 M. Goeldner and R. Crivens, *Dynamic Studies in Biology*, Wiley-VCH, 2005.
- 13 C. P. McCoy, C. Rooney, C. R. Edwards, D. S. Jones and S. P. Gorman, *J. Am. Chem. Soc.*, 2007, **129**, 9572–9573.
- 14 P. D. Kehayova, C. D. Woodrell, P. J. Dostal, P. P. Chandra and A. Jain, *Photochem. Photobiol. Sci.*, 2002, **1**, 774–779.
- 15 Y. Wei, Y. Yan, D. Pei and B. Gong, *Bioorg. Med. Chem. Lett.*, 1998, **8**, 2419–2422.
- 16 M. Noguchi, M. Skwarczynski, H. Prakash, S. Hirota, T. Kimura, Y. Hayashi and Y. Kiso, *Bioorg. Med. Chem.*, 2008, **16**, 5389–5397.
- 17 W. Lin, D. Peng, B. Wang, L. Long, C. Guo and J. Yuan, *Eur. J. Org. Chem.*, 2008, **2008**, 793–796.
- 18 S. Ibsen, E. Zahavy, W. Wrasdilo, M. Berns, M. Chan and S. Esener, *Pharm. Res.*, 2010, **27**, 1848–1860.
- 19 N. Ueberschaar, H.-M. Dahse, T. Bretschneider and C. Hertweck, *Angew. Chem., Int. Ed.*, 2013, **52**, 6185–6189.
- 20 M. Bio, P. Rajaputra, G. Nkepan, S. G. Awuah, A. M. L. Hossion and Y. You, *J. Med. Chem.*, 2013, **56**, 3936–3942.
- 21 T. Joshi, V. Pierroz, C. Mari, L. Gemperle, S. Ferrari and G. Gasser, *Angew. Chem., Int. Ed.*, 2014, **53**, 2960–2963.
- 22 N. J. Farrer, L. Salassa and P. J. Sadler, *Dalton Trans.*, 2009, 10690–10701.
- 23 D. Maeda, H. Shimakoshi, M. Abe and Y. Hisaeda, *Dalton Trans.*, 2009, 140–145.
- 24 A. F. Westendorf, J. A. Woods, K. Korpis, N. J. Farrer, L. Salassa, K. S. Robinson, V. Appleyard, K. Murray, R. Grünert, A. M. Thompson, P. J. Sadler and P. J. Bednarski, *Mol. Cancer Ther.*, 2012, **11**, 1894–1904.
- 25 C. R. Maldonado, N. Gomez-Blanco, M. Jauregui-Osoro, V. G. Brunton, L. Yate and J. C. Mareque-Rivas, *Chem. Commun.*, 2013, **49**, 3985–3987.
- 26 J. S. Butler and P. J. Sadler, *Curr. Opin. Chem. Biol.*, 2013, **17**, 175–188.



- 27 J. Mlcouskova, J. Stepankova and V. Brabec, *J. Biol. Inorg. Chem.*, 2012, **17**, 891–898.
- 28 Y. Zhao, G. M. Roberts, S. E. Greenough, N. J. Farrer, M. J. Paterson, W. H. Powell, V. G. Stavros and P. J. Sadler, *Angew. Chem., Int. Ed.*, 2012, **51**, 11263–11266.
- 29 M. Roy, T. Bhowmick, R. Santhanagopal, S. Ramakumar and A. R. Chakravarty, *Dalton Trans.*, 2009, 4671–4682.
- 30 J. Talib, D. G. Harman, C. T. Dillon, J. Aldrich-Wright, J. L. Beck and S. F. Ralph, *Dalton Trans.*, 2009, 504–513.
- 31 A. K. Patra, T. Bhowmick, S. Roy, S. Ramakumar and A. R. Chakravarty, *Inorg. Chem.*, 2009, **48**, 2932–2943.
- 32 T. K. Goswami, M. Roy, M. Nethaji and A. R. Chakravarty, *Organometallics*, 2009, **28**, 1992–1994.
- 33 F. Barragan, P. Lopez-Senin, L. Salassa, S. Betanzos-Lara, A. Habtemariam, V. Moreno, P. J. Sadler and V. Marchan, *J. Am. Chem. Soc.*, 2011, **133**, 14098–14108.
- 34 B. S. Howerton, D. K. Heidary and E. C. Glazer, *J. Am. Chem. Soc.*, 2012, **134**, 8324–8327.
- 35 E. Wachter, D. K. Heidary, B. S. Howerton, S. Parkin and E. C. Glazer, *Chem. Commun.*, 2012, **48**, 9649–9651.
- 36 M. Frascioni, Z. Liu, J. Lei, Y. Wu, E. Strekalova, D. Malin, M. W. Ambrogio, X. Chen, Y. Y. Botros, V. L. Cryns, J.-P. Sauvage and J. F. Stoddart, *J. Am. Chem. Soc.*, 2013, **135**, 11603–11613.
- 37 A. Hussain, S. Gadadhar, T. K. Goswami, A. A. Karande and A. R. Chakravarty, *Dalton Trans.*, 2012, **41**, 885–895.
- 38 A. Kastl, A. Wilbuer, A. L. Merkel, L. Feng, P. Di Fazio, M. Ocker and E. Meggers, *Chem. Commun.*, 2012, **48**, 1863–1865.
- 39 Q.-X. Zhou, W.-H. Lei, Y.-J. Hou, Y.-J. Chen, C. Li, B.-W. Zhang and X.-S. Wang, *Dalton Trans.*, 2013, **42**, 2786–2791.
- 40 L. Zayat, C. Calero, P. Albores, L. Baraldo and R. Etchenique, *J. Am. Chem. Soc.*, 2003, **125**, 882–883.
- 41 L. Zayat, M. Salierno and R. Etchenique, *Inorg. Chem.*, 2006, **45**, 1728–1731.
- 42 M. Salierno, C. Farneli and R. Etchenique, *Eur. J. Inorg. Chem.*, 2008, **2008**, 1125–1128.
- 43 L. Zayat, M. G. Noval, J. Campi, C. I. Calero, D. J. Calvo and R. Etchenique, *ChemBioChem*, 2007, **8**, 2035–2038.
- 44 T. Respondek, R. N. Garner, M. K. Herroon, I. Podgorski, C. Turro and J. J. Kodanko, *J. Am. Chem. Soc.*, 2011, **133**, 17164–17167.
- 45 K. L. Ciesinski, L. M. Hyman, D. T. Yang, K. L. Haas, M. G. Dickens, R. J. Holbrook and K. J. Franz, *Eur. J. Inorg. Chem.*, 2010, **2010**, 2224–2228.
- 46 K. L. Ciesinski, K. L. Haas, M. G. Dickens, Y. T. Tesema and K. J. Franz, *J. Am. Chem. Soc.*, 2008, **130**, 12246–12247.
- 47 A. A. Kumbhar, A. T. Franks, R. J. Butcher and K. J. Franz, *Chem. Commun.*, 2013, **49**, 2460–2462.
- 48 G. C. R. Ellis-Davies, *Chem. Rev.*, 2008, **108**, 1603–1613.
- 49 C. Gwizdala, D. P. Kennedy and S. C. Burdette, *Chem. Commun.*, 2009, 6967–6969.
- 50 H. M. D. Bandara, D. P. Kennedy, E. Akin, C. D. Incarvito and S. C. Burdette, *Inorg. Chem.*, 2009, **48**, 8445–8455.
- 51 C. Gwizdala, S. C. Gwizdala and C. Burdette, *Curr. Opin. Chem. Biol.*, 2013, **17**, 137–142.
- 52 H. W. Mbatia and S. C. Burdette, *Biochemistry*, 2012, **51**, 7212–7224.
- 53 G. Gasser, I. Ott and N. Metzler-Nolte, *J. Med. Chem.*, 2011, **54**, 3–25, and references therein.
- 54 C. G. Hartinger, N. Metzler-Nolte and P. J. Dyson, *Organometallics*, 2012, **31**, 5677–5685, and references therein.
- 55 P. C. A. Bruijinx and P. J. Sadler, *Curr. Opin. Chem. Biol.*, 2008, **12**, 197–206, and references therein.
- 56 G. Gasser and N. Metzler-Nolte, *Curr. Opin. Chem. Biol.*, 2012, **16**, 84–91, and references therein.
- 57 G. Jaouen and N. Metzler-Nolte, *Medicinal organometallic chemistry*, Springer-Verlag, 2010.
- 58 I. Kitanovic, S. Can, H. Alborzinia, A. Kitanovic, V. Pierroz, A. Leonidova, A. Pinto, B. Spingler, S. Ferrari, R. Molteni, A. Steffen, N. Metzler-Nolte, S. Wolfl and G. Gasser, *Chem. – Eur. J.*, 2014, **20**, 2496–2507.
- 59 N. Viola-Villegas, A. E. Rabideau, M. Bartholoma, J. Zubieta and R. P. Doyle, *J. Med. Chem.*, 2009, **52**, 5253–5261.
- 60 X. Tang and I. J. Dmochowski, *Nat. Protoc.*, 2007, **1**, 3041–3048.
- 61 C. P. Holmes, *J. Org. Chem.*, 1997, **62**, 2370–2380.
- 62 C.-y. Chang, B. Niblack, B. Walker and H. Bayley, *Chem. Biol.*, 1995, **2**, 391–400.
- 63 L. Niu, R. Wieboldt, D. Ramesh, B. K. Carpenter and G. P. Hess, *Biochemistry*, 1996, **35**, 8136–8142.
- 64 M. J. Brubaker, D. H. Dyer, B. Stoddard and D. E. Koshland, *Biochemistry*, 1996, **35**, 2854–2864.
- 65 E. Garcia Garayoa, D. Rüegg, P. Bläuenstein, M. Zwimpfer, I. U. Khan, V. Maes, A. Blanc, A. G. Beck-Sickinger, D. A. Tourwé and P. A. Schubiger, *Nucl. Med. Biol.*, 2007, **34**, 17–28.
- 66 E. Garcia Garayoa, C. Schweinsberg, V. Maes, D. Rüegg, A. Blanc, P. Bläuenstein, D. A. Tourwé, A. G. Beck-Sickinger and P. A. Schubiger, *Eur. J. Nucl. Med. Mol. Imaging*, 2007, **51**, 42–50.
- 67 J. Olejnik, S. Sonar, E. Krzymańska-Olejnik and K. J. Rothschild, *Proc. Natl. Acad. Sci. U. S. A.*, 1995, **92**, 7590–7594.
- 68 G. Gauglitz and S. J. Hubig, *J. Photochem.*, 1985, **30**, 121–125.
- 69 R. T. Jensen, J. F. Battey, E. R. Spindel and R. V. Benya, *Pharmacol. Rev.*, 2008, **60**, 1–60, and references therein.
- 70 J. Berlanda, T. Kiesslich, V. Engelhardt, B. Krammer and K. Plaetzer, *J. Photochem. Photobiol., B*, 2010, **100**, 173–180.
- 71 F. S. Mackay, J. A. Woods, P. Heringová, J. Kašpárková, A. M. Pizarro, S. A. Moggach, S. Parsons, V. Brabec and P. J. Sadler, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 20743–20748.
- 72 N. J. Farrer, J. A. Woods, L. Salassa, Y. Zhao, K. S. Robinson, G. Clarkson, F. S. Mackay and P. J. Sadler, *Angew. Chem., Int. Ed.*, 2010, **49**, 8905–8908.
- 73 B. Peña, A. David, C. Pavani, M. S. Baptista, J.-P. Pellois, C. Turro and K. R. Dunbar, *Organometallics*, 2014, **33**, 1100–1103.
- 74 R. Lincoln, L. Kohler, S. Monroe, H. Yin, M. Stephenson, R. Zong, A. Chouai, C. Dorsey, R. Hennigar, R. P. Thummel and S. A. McFarland, *J. Am. Chem. Soc.*, 2013, **135**, 17161–17175.



- 75 H. Yin, M. Stephenson, J. Gibson, E. Sampson, G. Shi, T. Sainuddin, S. Monro and S. A. McFarland, *Inorg. Chem.*, 2014, **53**, 4548–4559.
- 76 D. A. Lutterman, P. K. L. Fu and C. Turro, *J. Am. Chem. Soc.*, 2006, **128**, 738–739.
- 77 A. A. Holder, D. F. Zigler, M. T. Tarrago-Trani, B. Storrie and K. J. Brewer, *Inorg. Chem.*, 2007, **46**, 4760–4762.
- 78 M. A. Sgambellone, A. David, R. N. Garner, K. R. Dunbar and C. Turro, *J. Am. Chem. Soc.*, 2013, **135**, 11274–11282.
- 79 G. Gasser, A. Pinto, S. Neumann, A. M. Sosniak, M. Seitz, K. Merz, R. Heumann and N. Metzler-Nolte, *Dalton Trans.*, 2012, **41**, 2304–2313.
- 80 F.-M. Boisvert, J. van Koningsbruggen, A. Navascués and A. Lamond, *Nat. Rev. Mol. Cell Biol.*, 2007, **8**, 574–585, and references therein.
- 81 A. J. Pickard and U. Bierbach, *ChemMedChem*, 2013, **8**, 1441.
- 82 G. Gasser, S. Neumann, I. Ott, M. Seitz, R. Heumann and N. Metzler-Nolte, *Eur. J. Inorg. Chem.*, 2011, **36**, 5471–5478.
- 83 M. Muramatsu, K. Smetana and H. Busch, *Cancer Res.*, 1963, **25**, 693–697.
- 84 T. Hermann, *Angew. Chem., Int. Ed.*, 2000, **39**, 1890–1905.
- 85 S. Phongtongpasuk, S. Paulus, J. Schnabl, R. K. O. Sigel, B. Spingler, M. J. Hannon and E. Freisinger, *Angew. Chem., Int. Ed.*, 2013, **52**, 11513–11516.
- 86 M. Steiner, K. S. Karunatilaka, R. K. O. Sigel and D. Rueda, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 13853–13858.
- 87 M. G. Sun, J. Williams, C. Munoz-Pinedo, G. A. Perkins, J. M. Brown, M. H. Ellisman, D. R. Green and T. G. Frey, *Nat. Cell Biol.*, 2007, **9**, 1057–1065.
- 88 M. R. Gill, D. Cecchini, M. G. Walker, R. S. Mulla, G. Battaglia, W. Smythe and J. A. Thomas, *Chem. Sci.*, 2013, **4**, 4512–4519.
- 89 J. M. Hearn, I. Romero-Canelon, B. Qamar, Z. Liu, I. Hands-Portman and P. J. Sadler, *ACS Chem. Biol.*, 2013, **8**, 1335–1343.
- 90 D. V. Krysko, T. V. Berghe, K. D'Herde and P. Vandenabeele, *Methods*, 2008, **44**, 205–221.
- 91 V. Sirri, S. Urcuqui-Inchima, P. Roussel and D. Hernandez-Verdun, *Histochem. Cell Biol.*, 2008, **129**, 13–31.
- 92 W. J. Welch and J. P. Suhan, *J. Cell Biol.*, 1985, **101**, 1198–1211.
- 93 M. Engeland, L. J. W. Nieland, F. C. S. Ramaekers, B. Schutte and C. P. M. Reutelingsperger, *Cytometry*, 1998, **31**, 1–9.
- 94 S. Gallo, M. Furler and R. K. O. Sigel, *Chimia*, 2005, **59**, 812–816.

