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Direct C-H amination and C-H chloroamination of 7-deazapurines†

Nazarii Sabat, ab Martin Klečka, ab Lenka Slavětínská, b Blanka Klepetářová and Michal Hocek*ab

Protocols for selective Pd–Cu-catalyzed direct C–H amination or C–H chloroamination of 7-deazapurines with *N*-chloro-*N*-alkyl-arylsulfonamides have been developed leading either to 8-(arylsulfonyl)methylamino-7-deazapurines or to 7-chloro-8-(arylsulfonyl)methylamino-7-deazapurines. The scope and limitations of the methods, as well as synthesis of a small series of 6,8,9-tri- and 6,7,8,9-tetrasubstituted 7-deazapurines and deprotection of the sulfonamide are presented.

Introduction

Modified pyrrolo[2,3-d]pyrimidine (7-deazapurine)¹ bases and nucleosides display a variety of biological effects. Substituted 7-deazapurine bases induct neurogenesis² or exert antitumor³ or antiinflammatory⁴ activity, whereas 6- or 7-aryl-7-deazapurine nucleosides are potent cytostatics. ^{5,6} Therefore, development of regioselective synthesis of 7-deazapurines bearing multiple substituents is a worthwhile goal. Some regioselective cross-couplings of di- or trihalogenated pyrrolo[2,3-d]pyrimidines were recently reported for the synthesis of di- or triaryl derivatives. ^{7,8} We have developed regioselective C–H borylation⁰ and C–H sulfenylation¹⁰ leading to 8-B- or 8-S-substituted 7-deazapurines. Here we report on complementary C–H aminations.

Metal-catalyzed direct C–H aminations are increasingly popular reactions for modifications of arenes and heterocycles. ¹¹⁻¹⁴ One of the most efficient reagents are *N*-chlorosulfonamides under Pd/Cu catalysis. ¹³ Inspired by related C–H aminations of indoles, ¹³ we decided to study the C–H aminations of pyrrolo[2,3-*d*]pyrimidines (7-deazapurines). The only previously reported C–H amination of 7-deazapurine was performed through hypervalent iodanes and proceeded at position 7. ¹⁴

For our initial study, we selected easily accessible 6-phenyl-9-benzyl-7-deazapurine **1a.**9 We started testing its reaction with *N*-chloro-*N*-methyl-tosylamide **2** under literature¹³ conditions in presence of Pd(OAc)₂, Cu(acac)₂, 2,2'-bipyridine (bpy), Na₂CO₃ in dioxane (Scheme 1 and Table 1). The reaction with 2 equiv. of **2** in presence of 2 equiv. of Na₂CO₃ gave the desired 8-tosylamino product **5a** in 13% only (entry 1). The use of a larger excesses of the base (5–7 equiv.) and of reagent **2** (3 equiv.) led to only low increase of yields (18–29%). Only the use of large excess (5 equiv.) of **2** gave product **5a** in acceptable preparative yields of 68%.

In order to have a choice of some more easily cleavable N-protecting groups, 15 we also tested 4-nitrophenylsulfonyl-(p-nosyl, pNs) and 2-nitrophenylsulfonyl (o-nosyl, oNs) chloroamides 3 and 4. The reaction of $\mathbf{1a}$ with pNs reagent 3 (3 equiv.) gave the 8-p-nosylamino product $\mathbf{6a}$ in acceptable 48% yield (entry 6). The reactions of $\mathbf{1a}$ with oNs chloroamide $\mathbf{4}$ (2–3 equiv.) gave very low conversions (see Table S1 in ESI†), whereas the reaction with 5 equiv. of $\mathbf{4}$ gave a mixture of the desired product of 8-amination $\mathbf{7a}$ (28%) accompanied by 7-chloro-

Bn

Scheme 1 C-H aminations of 1a.

9a Bn

Results and discussion

reles University in Prague,
of Sciences of the Czech
n. 2, CZ-16610 Prague 6,

^aDepartment of Organic Chemistry, Faculty of Science, Charles University in Prague, Hlavova 8, CZ-12843 Prague 2, Czech Republic

^bInstitute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, Gilead & IOCB Research Center, Flemingovo nam. 2, CZ-16610 Prague 6, Czech Republic. E-mail: hocek@uochb.cas.cz

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Table 1 Optimization of C-H aminations of 7-deazapurine 1a with N-chloro-N-methyl-arylsulfoneamides $(2-4)^a$

Entry	Ar	2-4 (equiv.)	Na ₂ CO ₃ (equiv.)	Product(s) (yield)
1	4-MePh	2 (2)	2	5a (13%)
2	4-MePh	2 (2)	5	5a (18%)
3	4-MePh	2 (3)	5	5a (25%)
4	4-MePh	2 (3)	7	5a (29%)
5^b	4-MePh	2 (5)	7	5a (68%)
6	4-NO ₂ Ph	3 (3)	7	6a (47%)
7	2-NO ₂ Ph	4 (5)	5	7a(28%) + 8a(33%) + 9a(25%)
8	2-NO ₂ Ph	4 (3)	7	7a (60%)

^a Reagents and conditions: Pd(OAc)₂ (5%), Cu(acac)₂ (10%), bpy (10%), Na₂CO₃, 1,4-dioxane, Ar, rt, 24 h. ^b Reaction time 72 h.

8-amino- 8a and 7-chloro-7-deazapurine 9a as side-products. Apparently, the chloroamide 4 in larger excess can act as an electrophilic chlorination reagent which halogenates the deazapurine at position 7 (similarly as it was shown in indoles¹³). Therefore, we performed a detailed optimization of this reaction using different ratios of reagents, catalysts and additives and different conditions (see Table S1 in ESI†). The optimum protocol for aminations used 3 equiv. of 4 in presence of large excess of Na₂CO₃ (7 equiv.) to give the desired product 7a in 60% yield (Table 1, entry 8).

The detailed optimization also revealed some ratios of reagents and conditions under which the chloroamination proceeds. Also inspired by the related work on indoles, 13 we employed CuCl as copper source, Ag₂CO₃ as base and LiCl as additive (Table S1 in ESI†) to find an optimum protocol leading exclusively to chloroamination, 13,16 employing 4 (3 equiv.) in presence of Pd(OAc)2, CuCl (10 mol%), LiCl (2 equiv.) and Ag_2CO_3 (2 equiv.).

The next step was the study of the scope and limitation of the methods. A series of five 9-benzyl-7-deazapurine derivatives 1a-1e bearing phenyl, methoxy, methyl, chloro or amino group at position 6 was tested in the amination and chloroamination

C-H Amination Β'n Bn 7a-7c C-H Chloroamination C-H Amination i) 8a-8b Bn i) 4, Pd(OAc)₂, Cu(acac)₂, bpy, Na₂CO₃ (7 equiv.), dioxane, rt ii) 4, Pd(OAc)2, CuCl, LiCl, Ag2CO3 (2 equiv.), dioxane, rt

Scheme 2 C-H aminations and chloroaminations of 7-deazapurines.

protocols (Scheme 2 and Table 2). The preparative aminations were performed with choroamide 4 (3 equiv.) in presence of Pd(OAc)₂, Cu(acac)₃, bpy and 7 equiv. of Na₂CO₃. The reactions of 6-phenyl, -methoxy and -methyl deazapurines proceeded give desired 8-(o-nosyl)methylamino-7smoothly deazapurines 7a-7c in acceptable yields of 41-62%. Conversely, analogous reaction of 6-chloro- and 6-aminoderivatives 1d,e led to very complex inseparable mixtures.

Then we tested the chloroamination protocol on the same series of deazapurines 1a-1e. The reactions with 4 (3 equiv.) were performed in presence of Pd(OAc)2, CuCl, LiCl and Ag₂CO₃. The reactions of 6-phenyl and 6-methoxy derivatives 1a,b proceeded well to get desired 7-chloro-8-(oNs)MeNH-7deazapurines 8a,b in acceptable yields of 51 and 42%, whereas the reaction of 6-methyl derivative 1c gave low conversion to inseparable mixture containing products of chlorination and chloroamination. Similarly, the reactions of 6chloro- and 6-aminodeazapurines 1d,e gave complex inseparable mixtures. Finally, 6-phenyl-7-chloro-7-deazapurine 9a was also converted to 7-chloro-8-aminated derivative 8a in 41% yield showing that the chlorine at position 7 is better tolerated (as it is less reactive toward nucleophiles) than the chlorine at position 6.

The last goal in this study was to test a deprotection of the sulfonamides and the stability of the corresponding 8-amino-7deazapurines (2-aminoindoles are prone to tautomerization17 and oxidation18). Any attempt to cleave the Ts- or pNs-groups in

2 Preparative C-H aminations chloroaminations 7-deazapurines

Entry	Starting compd	R	Product (yield)
1	1a	Ph	7a (62%)
2	1b	ОМе	7 b (60%)
3	1c	Me	7 c (41%)
4	1d	Cl	Complex mixture
5	1e	NH_2	Complex mixture
6	1a	Ph	8a (51%)
7	1b	ОМе	8b (42%)
8	1c	Me	Low conversion, complex mixture
9	1d	Cl	Complex mixture
10	1e	NH_2	Complex mixture
11	9a	Ph	8a (41%)

Ph N N N N ONS CH₃CO₃, thiophenol CH₃CN 75% 10a Bn

Scheme 3 Deprotection of 7a

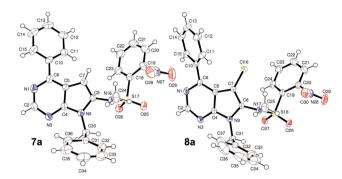


Fig. 1 An ORTEP view of compounds 7a (CCDC 1014820) and 8a (1014817) shown with 50% probability displacement ellipsoids. Structures 5a (1014819) and 7b (1014818) are shown in ESI.†

compounds **5a** or **6a** according to literature¹⁵ either did not work or led to decomposition of the heterocycles. Therefore, major part of this study was performed with *o*Ns-group which is more easily cleavable.¹⁵ The deprotection of compound **7a** was successfully performed using thiophenol and cesium carbonate^{15d} to afford 8-methylamino-7-deazapurine **10a** in 75% yield (Scheme 3). We performed also one-pot C–H amination deprotection sequence to furnish the desired compound **10a** directly in 35% for two steps. The 8-(methylamino)-7-deazapurine **10a** was reasonably stable under neutral conditions but quickly decomposed when exposed to even traces of acid (*e.g.* in chlorinated solvents).

All new compounds were fully characterized by NMR spectroscopy including assignment of all signals. In addition, to confirm the regioselectivity of the reactions, single-crystal X-ray diffraction was performed with compounds **5a**, **7a**, **7b** and **8a**. Fig. 1 shows the crystal structures of compounds **7a** and **8a** (for structures of **5a** and **7b**, see ESI†).

Conclusions

In conclusion, we have developed selective protocols for palladium/copper-catalyzed direct C–H amination and C–H chloroamination of 7-deazapurines with *N*-chloro-*N*-alkylarylsulfoneamides. Reactions proceed under mild conditions regioselectively at position 8 of 7-deazapurines (in analogy to aminations at position 2 of indoles¹³) and are applicable to 6-aryl, -alkyl and -alkoxy 7-deazapurine derivatives. On the other hand, they are not compatible with 6-amino- and 6-chloro derivatives. Apart from the potential for the synthesis of series of 8-(arylsulfonyl)methylamino-7-deazapurines (and their 7-chloro-derivatives), when using *o*Ns sulfonamides, the deprotection is possible to 8-methylamino-7-deazapurines. These

protocols nicely complement the current toolbox of reactions for modifications of these privileged heterocycles and will be used for generation of libraries of compounds for biological activity screening.

Experimental

General procedure for C-H amination of 7-deazapurines

7-Deazapurine **1a–1e** (0.5 mmol), $Pd(OAc)_2$ (0.025 mmol), $Cu(acac)_2$ (0.05 mmol), bpy (0.05 mmol), Na_2CO_3 (3.5 mmol) and chlorosulfonamide (1–1.75 mmol) were placed in a vial which was purged with argon. Then 1,4-dioxane (2 mL) was added and the reaction mixture was then stirred for 24 h at rt, then quenched with H_2O (2 mL), extracted with ethyl acetate (3 × 20 mL) and washed with brine (2 mL). The organic phases were combined and dried over sodium sulphate, filtered, and evaporated under vacuum. The crude product was purified by column chromatography on silica gel, eluting with hexanes/ EtOAc (5 : 1 to 1 : 2) to afford the corresponding product.

N-(7-benzyl-4-phenyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-6-yl)-*N*-methyl-2-nitrobenzenesulfonamide (7a)

1a (285 mg, 1 mmol) and N-chloro-N-methyl-2-nitrobenzenesulfonamide 4 (877 mg, 3.5 mmol) were used as starting compounds to give product 7a (310 mg, 62%) as colourless crystals after chromatography with hexanes/EtOAc 5:1 to 1:1 and crystallization from EtOAc/hexane. M.p. 102–103 °C. ¹H NMR (500.0 MHz, CDCl₃): 2.94 (s, 3H, CH₃N); 5.67 (bs, 2H, CH₂Ph); 6.45 (s, 1H, H-5); 7.22 (m, 2H, H-o-Bn); 7.27 (m, 1H, Hp-Bn); 7.30 (m, 2H, H-m-Bn); 7.50-7.53 (m, 3H, H-m,p-Ph); 7.61 (ddd, 1H, $J_{5,6} = 8.1$, $J_{5,4} = 7.5$, $J_{5,3} = 1.3$, H-5-C₆H₄NO₂); 7.67 (ddd, 1H, $J_{3,4} = 8.0$, $J_{3,5} = 1.3$, $J_{3,6} = 0.5$, H-3-C₆H₄NO₂); 7.76 (ddd, 1H, $J_{6,5} = 8.1$, $J_{6,4} = 1.4$, $J_{6,3} = 0.5$, H-6-C₆H₄NO₂); 7.77 (ddd, 1H, $J_{4,3} = 8.0$, $J_{4,5} = 7.5$, $J_{4,6} = 1.4$, H-4-C₆H₄NO₂); 7.97 (m, 2H, H-o-Ph); 9.10 (s, 1H, H-2). ¹³C NMR (125.7 MHz, CDCl₃): 40.53 (CH₃N); 45.29 (CH₂Ph); 96.40 (CH-5); 113.91 (C-4a); 124.12 (CH-3-C₆H₄NO₂); 127.65 (CH-o-Bn); 127.89 (CH-p-Bn); 128.85 (CH-m-Ph, CH-o-Bn); 128.94 (CH-m-Ph); 130.16 (C-1-C₆H₄NO₂); 130.64 (CH-p-Ph); 131.26 (CH-5-C₆H₄NO₂); 132.23 (CH-6-C₆H₄NO₂); 134.62 (CH-4-C₆H₄NO₂); 136.54 (C-i-Bn); 136.86 (C-i-Ph); 137.04 (C-6); 148.55 (C-2-C₆H₄NO₂); 150.56 (C-7a); 152.40 (CH-2); 157.48 (C-4). IR(KBr): 2821, 1545, 1376, 1368, 1360, 1343, 1318, 1309, 1180, 1163, 924. HRMS (ESI) calculated for C₂₆H₂₂N₅O₄S: 500.1387; found 500.1387.

General procedure for C–H chloroamination of 7-deazapurines

7-Deazapurine **1a–1e** (0.5 mmol), $Pd(OAc)_2$ (0.0125 mmol), CuCl (0.05 mmol), CuCl (0.05 mmol), CuCl (1.0 mmol) and chlorosulfonamide (1.5–1.75 mmol) were placed in a vial which was purged with argon. Then 1,4-dioxane (2 mL) was added and the reaction mixture was stirred for 24 h at room temperature, then quenched with CuCl (2 mL), extracted with ethyl acetate (3 × 20 mL) and washed with brine (2 mL). The organic phases were combined and dried over sodium sulphate, filtered, and evaporated under vacuum. The crude product was purified by

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column chromatography on silica gel, eluting with hexanes/ EtOAc (5:1 to 1:2) to afford the corresponding product.

N-(7-benzyl-5-chloro-4-phenyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-6-yl)-*N*-methyl-2-nitrobenzenesulfonamide (8a)

1a (285 mg, 1 mmol) and 4 (877 mg, 3.5 mmol) were used as starting compounds to give product 8a (273 mg, 51%) as white crystals after chromatography with hexanes/EtOAc 5:1 to 1:1 and crystallization from EtOAc/hexane. M.p. 215-216 °C. ¹H NMR (500 MHz, CDCl₃): 2.91 (s, 3H, CH₃N); 5.42 (d, 1H, J_{gem} = 15.3 Hz, CH_2a-Ph); 6.16 (d, 1H, $J_{gem} = 15.3$ Hz, CH_2b-Ph); 7.26– 7.31 (m, 3H, H-o,p-Bn); 7.33 (m, 2H, H-m-Bn); 7.42-7.50 (m, 3H, H-m,p-Ph); 7.58-7.63 (m, 2H, H-3,5-C₆H₄NO₂); 7.71 (m, 2H, H-o-Ph); 7.73 (ddd, 1H, $J_{4,3} = J_{4,5} = 7.7$ Hz, $J_{4,6} = 1.4$ Hz, H-4- $C_6H_4NO_2$; 7.84 (m, 1H, H-6- $C_6H_4NO_2$); 9.09 (s, 1H, H-2). ¹³C NMR (125.7 MHz, CDCl₃): 38.28 (CH₃-N); 45.68 (CH₂-Ph); 103.08 (C-5); 112.02 (C-4a); 124.00 (CH-3-C₆H₄NO₂); 127.84 (CHm-Ph); 128.09 (CH-p-Bn); 128.11 (CH-o-Bn); 128.95 (CH-m-Bn); 129.86 (CH-p-Ph); 130.20 (CH-o-Ph); 131.45 (CH-5-C₆H₄NO₂); 131.62 (C-1-C₆H₄NO₂); 131.65 (CH-6-C₆H₄NO₂); 131.89 (C-6); 134.49 (CH-4-C₆H₄NO₂); 136.48 (C-i-Bn); 136.62 (C-i-Ph); 148.56 (C-2-C₆H₄NO₂); 148.78 (C-7a); 153.18 (CH-2); 160.20 (C-4). IR(KBr): 3050, 1583, 1545, 1374, 1345, 1165, 826, 558. HRMS (ESI) calculated for C₂₆H₂₁N₅O₄SCl: 534.0998; found: 534.0997.

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