

Recent trends in Pd-catalyzed remote functionalization of carbonyl compounds

Ivan Franzoni and Clément Mazet*

Cite this: *Org. Biomol. Chem.*, 2014, **12**, 233

Received 14th October 2013,
Accepted 5th November 2013

DOI: 10.1039/c3ob42050g

www.rsc.org/obc

Recent advances in the palladium-catalyzed remote functionalization of carbonyl derivatives are highlighted in this review. The structure of the article is based on the three strategies that have emerged in recent years as the most viable tactics to construct C(sp³)-C, C(sp³)-N or C(sp³)-O bonds at a distance of at least three carbon atoms from the carbonyl functionality. The specific aspects of each of these approaches are discussed in detail (reaction conditions, substrate scope and limitations). As the reading progresses, their complementarity should also appear clearly.

1. Introduction

The ability of certain enzymes to interact with a functional group and to selectively modify a remote C-H or C-C bond has been a continuous source of inspiration for the synthetic chemist. The introduction of the concept of *remote functionalization* in chemical synthesis can be traced back to the 1970s with the emergence of biomimetic approaches.¹ Since then, numerous effective synthetic methods have been developed to selectively activate and modify a distal position from a functional handle.² Even more remarkably, cases of remote stereochemical control have also been reported, in particular with

the recent explosion of organocatalytic methods.^{3,4} As many of these achievements have already been reviewed, for the sake of clarity and brevity, they will not be discussed again in this perspective. Instead, emphasis will be specifically placed on the most recent and important tactics to remotely functionalize carbonyl derivatives (or surrogates thereof) using Pd catalysis and that generate C(sp³) centers. The structure of this article is articulated around the three strategies that have been dominating over the last two decades: (i) the remote functionalization of carbonyls by direct C-H activation of C(sp³)-H bonds; (ii) the direct arylation of extended enolates (or enolate equivalents); (iii) the Heck arylation of isomerizable double bonds remote from a carbonyl or an alcohol.

Furthermore, as the notion of 'remote functionalization' might be somewhat subjective, a distance of at least 3 carbon atoms between the carbonyl group and the position being

Organic Chemistry Department, University of Geneva, 30 quai Ernest Ansermet, Geneva 1204-CH, Switzerland. E-mail: clement.mazet@unige.ch; Fax: (+41)22 379 32 15; Tel: (+41)22 379 62 88



Ivan Franzoni

Ivan Franzoni was born in Edolo (Italy) in 1986. He studied chemistry at the University of Milano (Italy) and received his M.Sc. in Chemistry in October 2011. He subsequently enrolled as a graduate student in the group of Prof. Clément Mazet in the Organic Chemistry Department of the University of Geneva. His research project focuses on the development of novel catalytic methods for the vicinal and distal stereoselective arylation of aldehydes.



Clément Mazet

Clément Mazet received his PhD from the University of Strasbourg under the supervision of Prof. L. H. Gade (2002). After postdoctoral stays with Prof. A. Pfaltz (University of Basel - 2003-2005) and Prof. E. N. Jacobsen (Harvard University - 2006-2007) he joined the University of Geneva to establish his independent research program. His interests include mechanistic and synthetic chemistry with particular emphasis on asymmetric catalysis. He recently received the Zasshikai Lectureship Award from the University of Tokyo (2012) and the Werner Prize from the Swiss Chemical Society (2013).



modified (γ position or further) was used as the threshold for our selection.

2. C–H activation of C(sp³)–H bonds

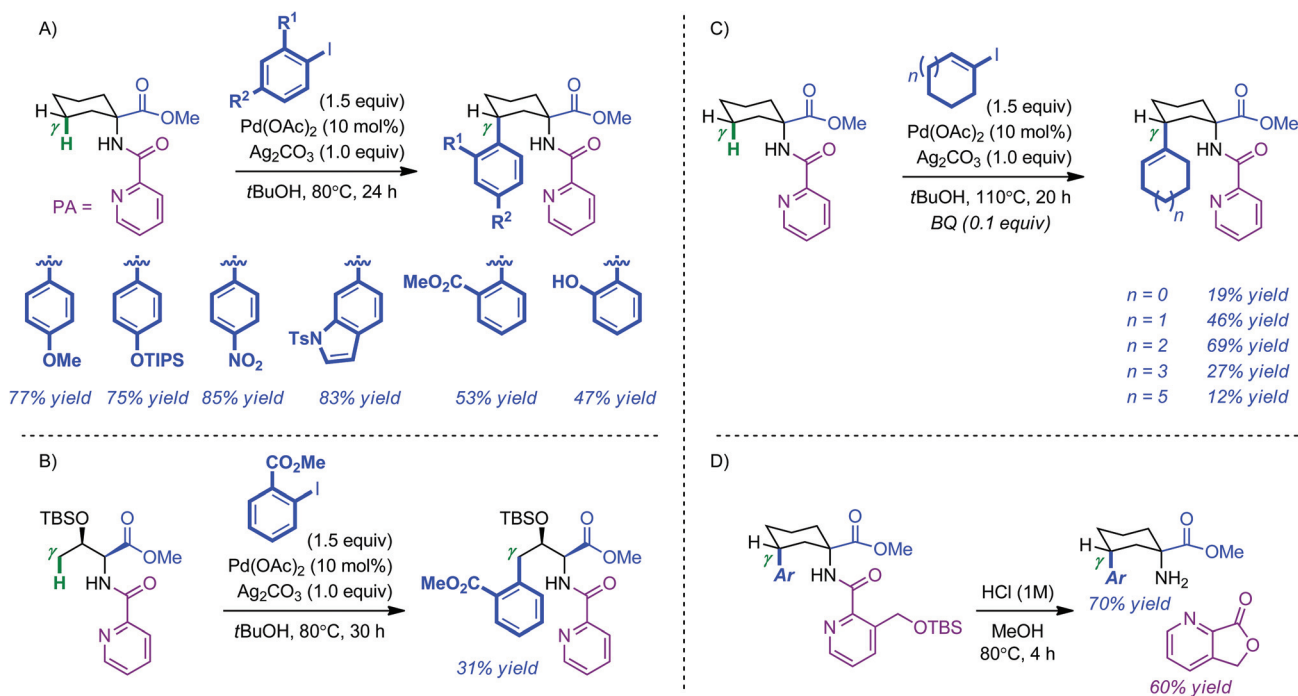
An ever increasing number of methods based on the regioselective functionalization of C(sp²)–H bonds in heteroarenes and arenes have been reported in recent years.⁵ In comparison, less progress has been made for the selective modification of C(sp³)–H bonds, in particular for remote aliphatic positions.⁶ A common aspect of these approaches is the use of directing groups to facilitate C–H activation. Nonetheless, from a practical point of view, this imposes that, beyond its efficiency, the prerequisite auxiliary must be easily installed and equally easily removed.

Daugulis and co-workers first demonstrated the ability of picolinamide (PA) to facilitate the Pd-catalyzed direct C–H arylation of remote aliphatic positions.⁷ However, this was achieved at the cost of relatively demanding reaction conditions (5 mol% Pd(OAc)₂, 4 equiv. aryl iodide, 1 equiv. AgOAc, 130–150 °C). Inspired by this precedent, the Chen group has elaborated milder and more general methods for the remote functionalization of γ C(sp³)–H bonds (5 mol% Pd(OAc)₂, 1.5 equiv. aryl iodide, 1.0 equiv. Ag₂CO₃, 80 °C).⁸ Electron-rich, electron-neutral, electron-poor as well as *ortho*-substituted iodoarenes were coupled with cyclic PA-modified amino esters in moderate to excellent yields (47–85%) (Scheme 1A). In contrast, linear substrates such as the threonine methyl ester shown in Scheme 1(B) were usually obtained in reduced yields.

Both an increase in temperature and the use of benzoquinone (BQ, 10 mol%) allowed γ -alkenylation to occur selectively on cyclic systems; albeit in more modest yields (12–69%). The removal of the PA directing group was found to be challenging. The use of an *ortho* methyloxysilyl analogue proved to be a viable alternative as it could be cleaved under acidic conditions (Scheme 1D). Preliminary mechanistic investigations were supportive of a C–H activation/arylation sequence.

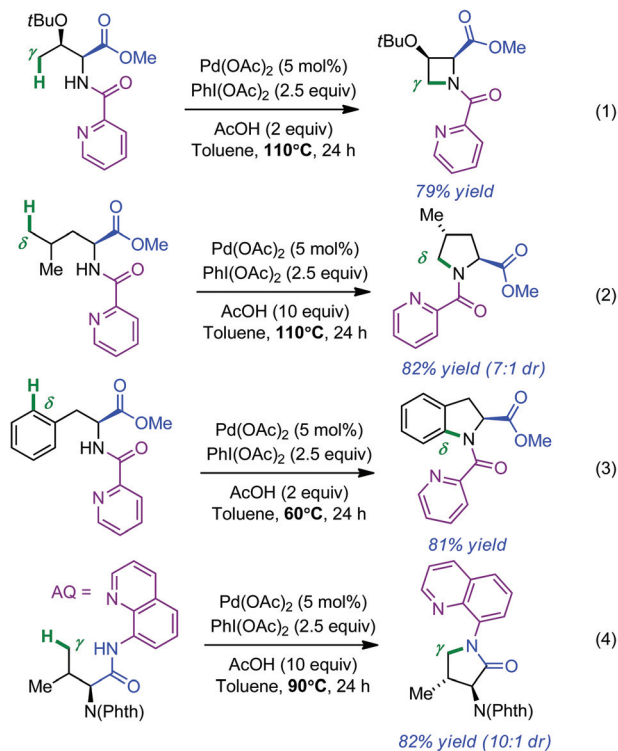
Having laid the fundamental basis in the remote functionalization of C(sp³)–H bonds, the same group next developed a set of Pd-catalyzed directed intramolecular amination reactions leading to azetidines, pyrrolidines, indolines and pyrrolidones by simply changing the oxidant (PhI(OAc)₂, 2.5 equiv.).^{9,10} Representative examples for each compound are shown in Scheme 2 (respectively in eqn (1)–(4)). Yields and diastereoselectivities were usually high when 2.0 equivalents of AcOH were used. Of note, access to pyrrolidinones required the use of 8-aminoquinoline (AQ) as the directing group. Additional studies showed that 8-methyl-5-methoxyquinoline (MQ) not only provided enhanced performances in the cyclization reactions but was also best suited in view of post-reaction ablation (3 equiv. CAN in CH₃CN–H₂O at room temperature).¹¹

The observation that acetoxylation took place preferentially over intramolecular amination when AcOH was used as a solvent prompted the authors to develop an intermolecular Pd-catalyzed PA-directed remote alkoxylation.¹² The isolated yields were usually high and the reaction featured a broad substrate scope by simply changing the nature of the alcohol. Typical products obtained by this method are shown in Scheme 3.

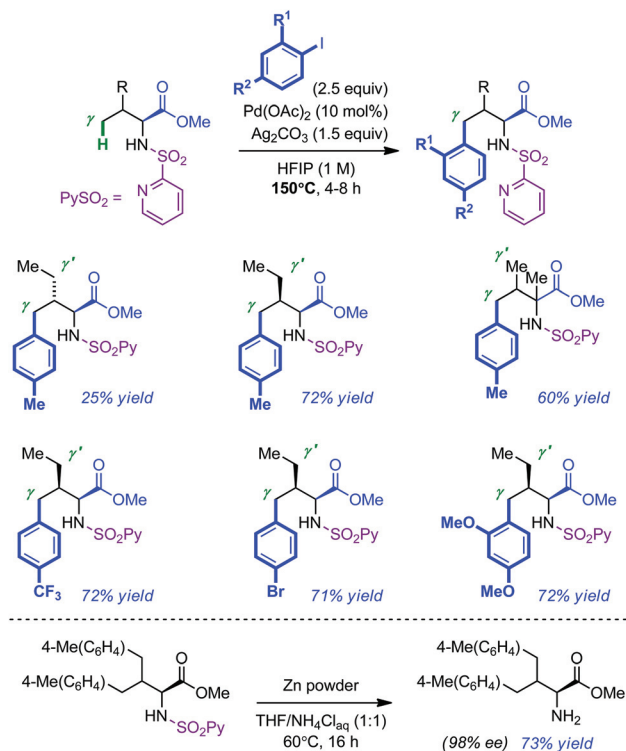


Scheme 1 Pd-catalyzed picolinamide-directed C(sp³)–H activation of the γ position of modified-amino esters. (A) Arylation of cyclic systems. (B) Arylation of linear systems. (C) Vinylation of cyclic systems. (D) Directing-group removal and recovery.

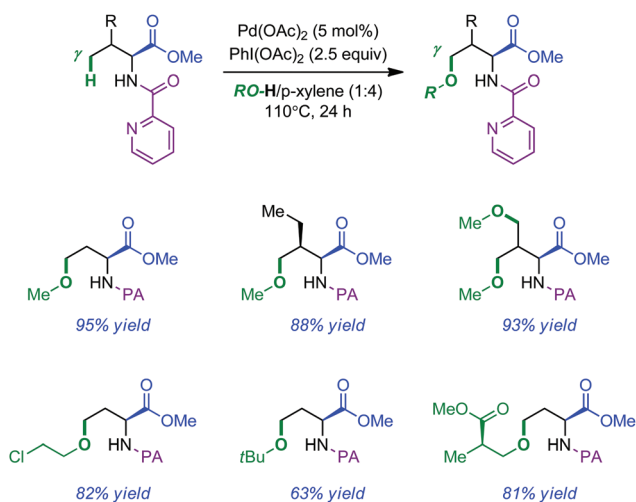




Scheme 2 Pd-catalyzed directed azetidines (eqn (1)), pyrrolidines (eqn (2)), indolines (eqn (3)) and pyrrolidinones (eqn (4)) formation.



Scheme 4 Pd-catalyzed PySO₂-directed intermolecular arylation of amino esters.



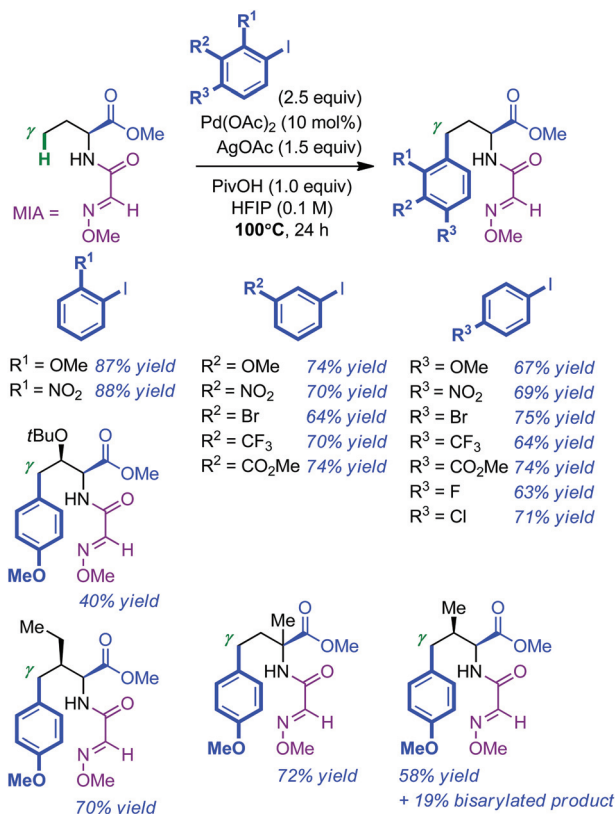
Scheme 3 Intermolecular alkoxylation of PA-modified amino esters.

The Carretero group has recently introduced 2-pyridylsulfonyl (PySO₂) as an easily installable and removable directing group in a variety of Pd-catalyzed C(sp²)-H bond activation protocols.¹³ As a further extension of this work, they recently demonstrated that PySO₂ was also efficient in the remote activation of C(sp³)-H bonds of amino ester derivatives.¹⁴ Their optimized conditions for the γ -arylation of L-valine methyl ester required elevated temperature (150 °C) and the use of 10 mol% of Pd(OAc)₂, 2.5 equivalents of iodo arene,

1.5 equivalent of oxidant (Ag₂CO₃) and 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP), a privileged solvent in C-H activation processes.¹⁵ Although the desired mono-arylated product could be isolated in appreciable yield, its formation was often plagued by the presence of the bis-arylated compound. Consequently, the scope of the reaction was mostly centred on PySO₂-modified amino esters with sterically distinct γ and γ' positions (Scheme 4). The corresponding mono-arylated products were usually isolated in good yields (60–72%). Further investigations established that electron-rich and electron-poor iodo arenes were equally tolerated (average yield of 72% with alloisoleucine). Removal of the 2-pyridylsulfonyl auxiliary was achieved by treatment of the arylated product with Zn in THF-NH₄Cl_{aq} for 16 h at 60 °C without any erosion of the enantiopurity (73% yield).

A related study was reported almost concomitantly by Fan and Ma.¹⁶ Their approach for the Pd-catalyzed γ -arylation of amino esters consisted in demonstrating the superior utility of 2-methoxyiminoacetyl (MIA) as a directing group for C(sp³)-H bond activation. Whereas Pd(OAc)₂ and HFIP proved to be the catalyst and the solvent of choice respectively, a combination of AgOAc and PivOH as additives was found to be optimal (100 °C, 24 h). Under these conditions, a wide number of iodoarenes (2.5 equiv.) were efficiently introduced in the γ position of MIA-modified homoalanine (see Scheme 5 for representative examples). Substituted amino esters were usually arylated at the less hindered γ position. Consistent with the observation of the Carretero group, the L-valine





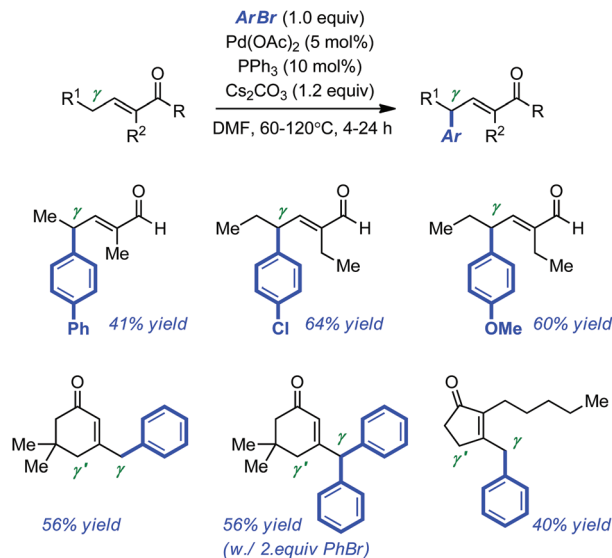
Scheme 5 Pd-catalyzed MIA-directed intermolecular arylation of amino esters.

derivative resulted in a mixture of mono and bis-arylated products (58% and 19% respectively). The MIA directing group was found to be easily removed by simple hydrolysis of a variety of arylation products with 1 M KOH in dioxane at room temperature (the corresponding amino acids were usually directly re-protected using (Boc)₂O: 5 examples, 65–82% yield; no erosion of the enantiopurity).

3. Arylations of extended enolates

The Pd-catalyzed arylation of enolates with aryl halides or aryl halide equivalents has progressed at an impressive pace over the last decade,¹⁷ culminating with the development of enantioselective α -arylations of amides,¹⁸ esters,¹⁹ ketones²⁰ and even of the more sensitive aldehydes.²¹ Yet, the corresponding vinylogous coupling of α,β -unsaturated carbonyl derivatives has not reached the same level of refinement.²² A first explanation for this discrepancy is certainly, the regioselectivity issues associated with the development of such coupling methods (α vs. β vs. γ arylation).

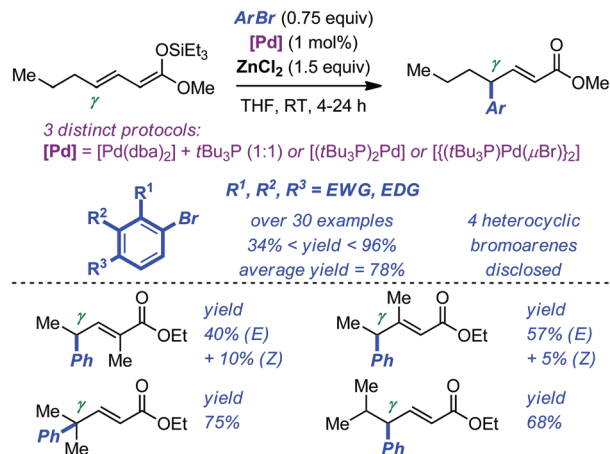
Miura and co-workers first disclosed that the Pd-catalyzed arylation of enals and enones can occur with perfect γ regioselectivity under conditions similar to those developed for the α -arylation of ketones or aldehydes.²³ With a combination of PPh₃ and Pd(OAc)₂ (5 mol% and 10 mol% respectively),



Scheme 6 Pd-catalyzed γ -arylation of α,β -unsaturated aldehydes and ketones.

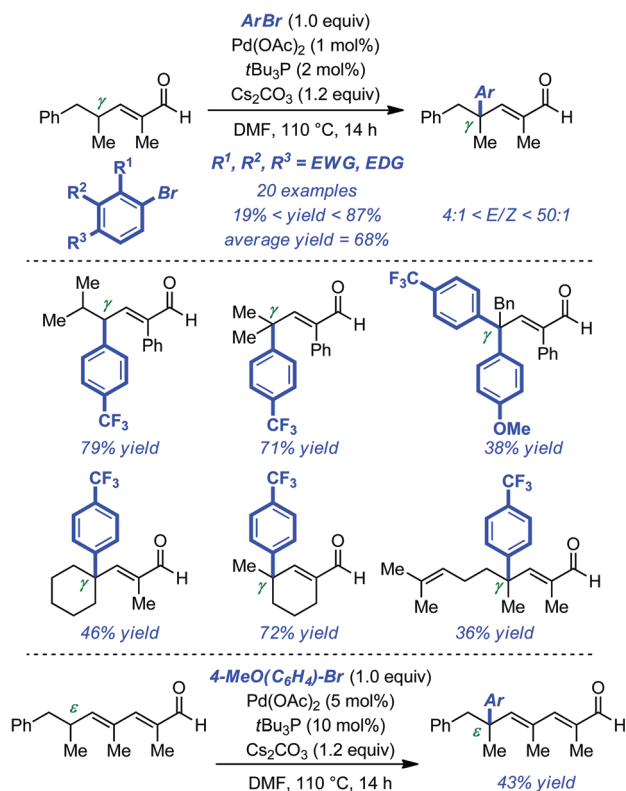
temperatures ranging from 60 to 120 °C, DMF and Cs₂CO₃ as the solvent and the base of choice, they were able to couple electron-rich and electron-neutral bromoarenes with linear aldehydes and cyclic ketones. A representative selection of the products formed is displayed in Scheme 6. Most of them were obtained in moderate isolated yields (40–64% yield). The authors noticed that the presence of an α -blocking substituent in the linear aldehydes was required to favor perfect γ -selective arylation.

Huang and Hartwig have developed an efficient Pd-catalyzed γ -arylation and γ -vinylation of α,β -unsaturated esters starting from the corresponding preformed silyl ketene acetals.²⁴ The reaction proceeds at room temperature with low loadings of palladium (typically 1 mol%). Optimization of the catalytic system led to the identification of three different sets of reaction conditions (see Scheme 7 for details). However, the



Scheme 7 Pd-catalyzed γ -arylation of silyl ketene acetals.





Scheme 8 Pd-catalyzed γ -arylation of α,β -unsaturated γ -branched aldehydes and ϵ -arylation of $\alpha,\beta,\gamma,\delta$ -unsaturated ϵ -branched aldehydes.

beneficial use of a triethylsilyl group (TES) on the ketene acetals, of ZnCl_2 as the additive and of $t\text{Bu}_3\text{P}$ as the ligand, was found to be a common feature of these arylations. Impressively, more than 30 different bromoarenes were evaluated to delineate the scope of the electrophilic component of the reaction. Noticeably, the average yield is approaching 80% and functional groups such as amine, ether, ketone, ester, cyano, indole, pyridine, quinoline, and thiophene were all tolerated. Introduction of a substituent on either the α or the β position of the silyl ketene acetals resulted in the formation of stereoisomeric mixtures of the corresponding arylated products. Although the exact role of ZnCl_2 in these reactions is not fully understood, preliminary investigations suggested that it did not only serve as a source of chlorine but also facilitated labilization of the halide on palladium.

As part of their program aimed at the enantioselective arylation of aldehydes, Mazet and co-workers recently developed an efficient Pd-catalyzed γ -arylation of γ -branched α,β -unsaturated aldehydes.²⁵ The precedent from Miura with linear substrates favoured rapid optimization of the reaction conditions and led to the identification of a robust protocol that allowed quaternary centers to be forged exclusively at the γ position without any detectable traces of the α - or β -arylation products ($\text{Pd}(\text{OAc})_2$ (1 mol%), $t\text{Bu}_3\text{P}$ (2 mol%), Cs_2CO_3 (1.2 equiv.), DMF, 110 °C).²⁶ The reaction proved very general with respect to the electrophile as electron-rich, electron-poor and sterically

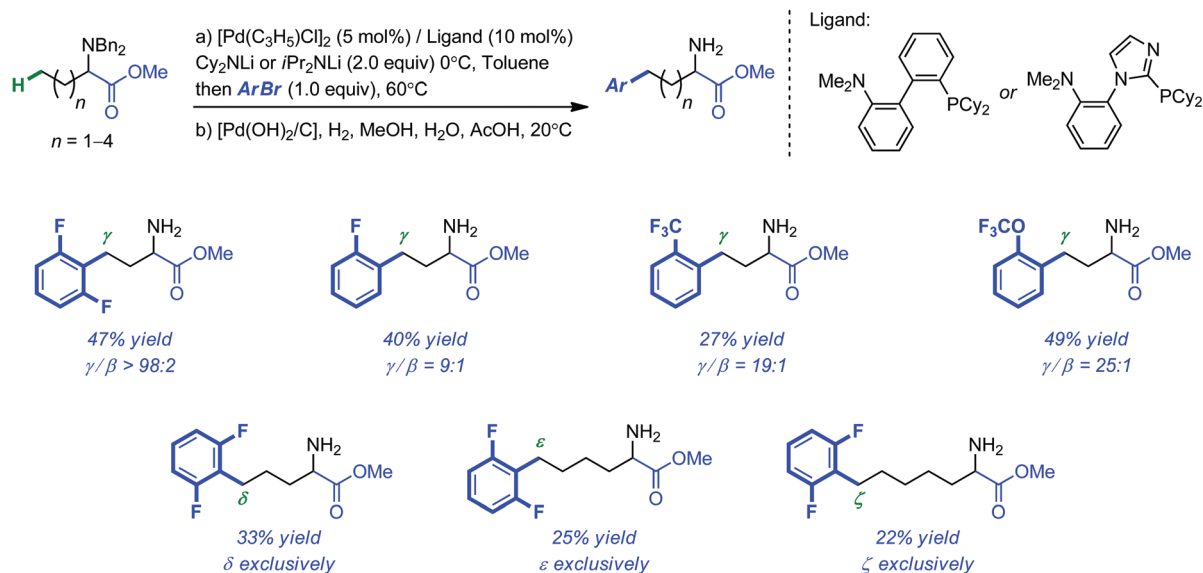
demanding aryl bromides were coupled in usually practical yields (19–87%, average = 68% yield). A slight erosion of the geometrical purity of the substrates was observed upon coupling, although the *E/Z* ratio was still relatively high (typically >10:1). The orthogonal reactivity of the α,β -unsaturated aldehydes obtained was demonstrated through a series of standard derivatizations and – more remarkably – without any optimization of the initial protocol the ϵ arylation of $\alpha,\beta,\gamma,\delta$ -unsaturated was achieved and the corresponding product isolated in a promising 43% yield (Scheme 8).

The Pd-catalyzed intermolecular migrative β -arylation of ester lithium enolates was recently reported by Baudoin and co-workers as a viable alternative to directed C–H activation strategies for the functionalization of carbonyl compounds.²⁷ This original approach relies on the rearrangement of congested palladium enolates to palladium homoenolates. Detailed theoretical mechanistic investigations have lent support on the β -H elimination, rotation and re-insertion sequence initially proposed.²⁸ Although the primary system, focused on carboxylic esters, was limited to aryl halides with an electron-withdrawing substituent in the *ortho* position, a much more general β -arylation was subsequently reported for benzyl-protected alanine esters.²⁹ During these investigations, the authors were also able to extend the arylations to the terminal position of amino esters with longer alkyl chains. Even if, in these cases again, only aryl halides with at least one fluorinated withdrawing *ortho* substituent were tolerated and the yields of the products decreased as the alkyl chain length increased, it is remarkable that arylation occurred *exclusively* at the terminal position of the carbon chain (as far as 6 carbon atoms away from the carbonyl functionality, see Scheme 9).

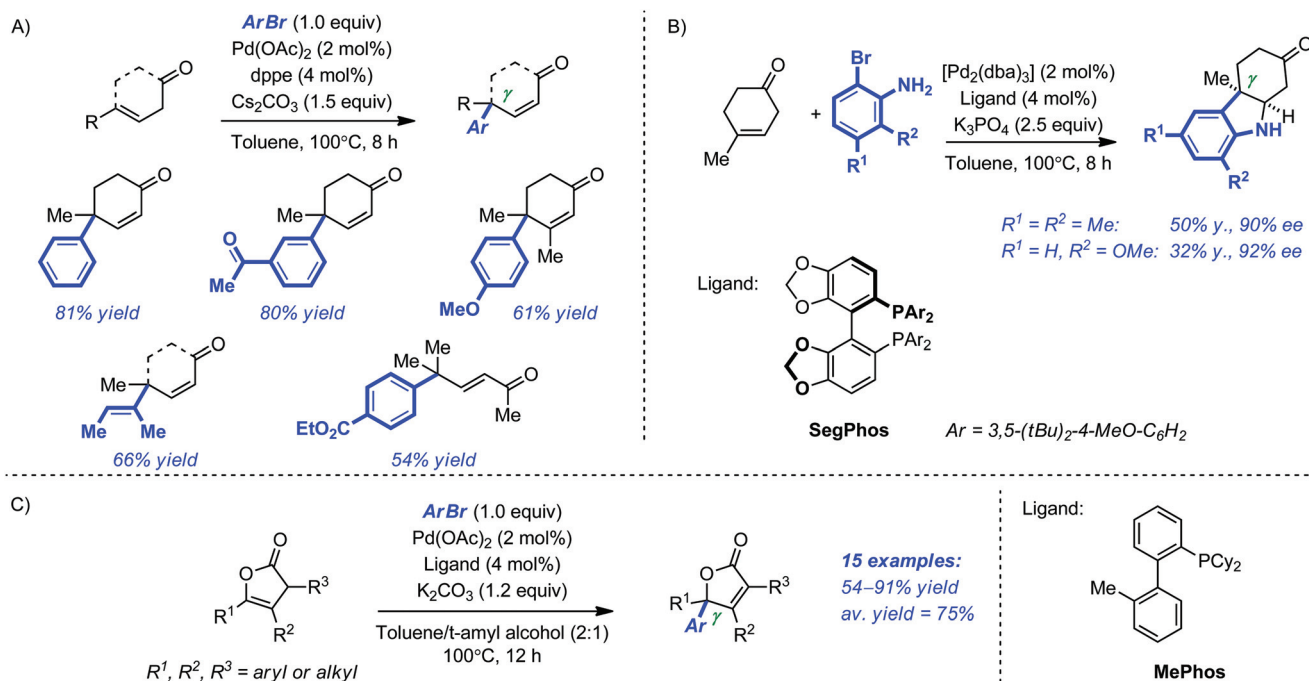
4. Heck-type

Buchwald and Hyde recently reported the Pd-catalyzed γ -arylation of γ -substituted β,γ -unsaturated linear and cyclic ketones.³⁰ Although this reaction could have well been categorized as a direct arylation of dienolates, we *arbitrarily* decided to discuss it in the ‘Heck-type’ section. Our decision stemmed essentially from the striking difference observed between the γ -arylation of α,β -unsaturated and β,γ -unsaturated ketones. In the optimized system, a ligand as simple as dppe (dppe = diphenylphosphinoethane; 4 mol%) in combination with $\text{Pd}(\text{OAc})_2$ (2 mol%) Cs_2CO_3 (1.5 equiv.) proved general for a variety of vinyl and aryl bromides (toluene, 100 °C, 8 h) and the products possessing a congested γ quaternary center were systematically obtained in good to excellent yield (50–81% yield; Scheme 10A). Subsequently, the use of *ortho*-bromoaniline derivatives allowed the efficient one-pot preparation of several ketoindolines by a γ -arylation/conjugate addition sequence. Further refinement of the system led to the identification of DTBM-Segphos as a highly selective ligand for the asymmetric variant of this transformation. Although the yields of the product were modest, the enantioselectivity levels were very high (Scheme 10B). Later, the same authors reported the





Scheme 9 Selective and remote arylation of linear amino esters using *ortho*-substituted electron-deficient aryl bromides.



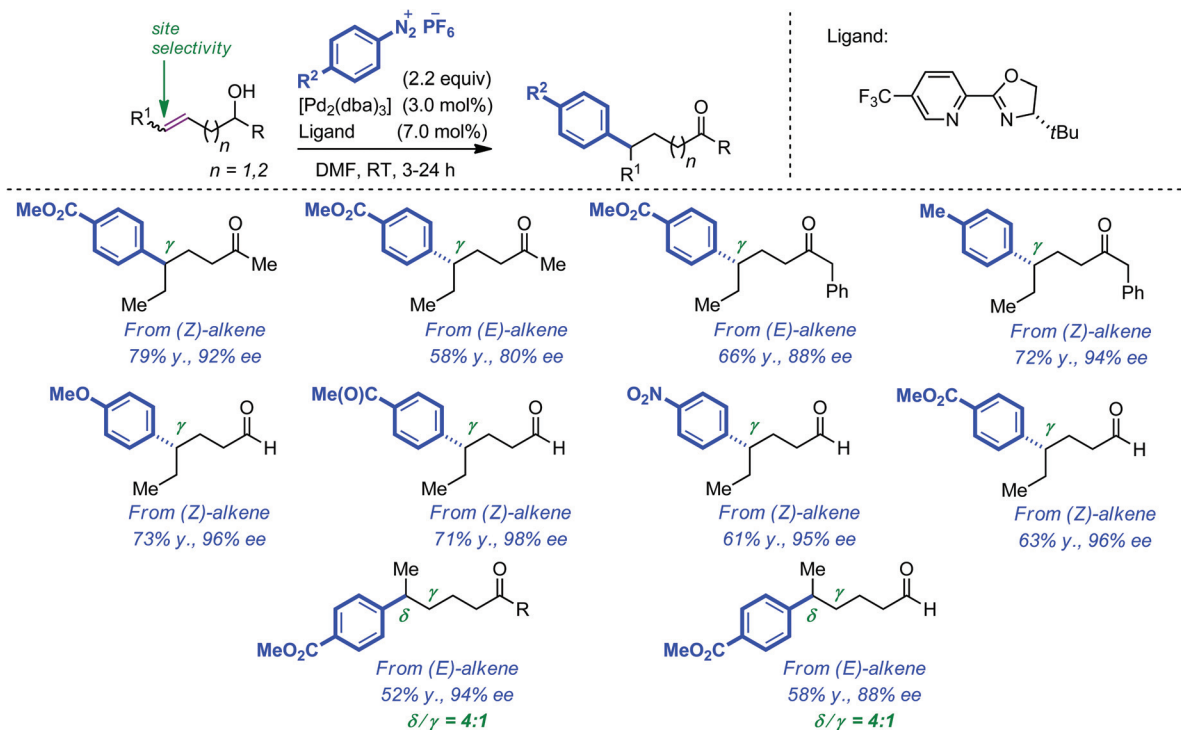
Scheme 10 γ -Arylation of γ -substituted cyclic ketones (A) and Angelica lactone derivatives (C). Enantioselective synthesis of ketoindolines by a γ -arylation conjugate/addition sequence (B).

regioselective γ -arylation of Angelica lactone and some analogues using a related procedure (2 mol% $[\text{Pd}_2(\text{dba})_3]$, 4 mol% Me-Phos, 1.2 equiv. K_2CO_3 , toluene-*t*-amyl alcohol (2 : 1), 100 °C, 12 h), thus providing a straightforward access to 5,5-disubstituted butenolides – a ubiquitous structural motif in the synthesis of biologically active compounds (Scheme 10C).^{31,32}

Distinct from the above-described approach, Sigman and co-workers have recently reported a Pd-catalyzed enantioselective redox-relay Heck reaction of acyclic alkenyl 1° and 2°

alcohols where aryl addition is followed by olefin migration through the carbon chain and eventually oxidation of the alcohol.³³ The authors capitalized on previous works which described similar outcome when allylic alcohols were employed.³⁴ Key to the success of this approach was the use of an electrophilic catalyst which favoured migration (*i.e.* chain walking) over dissociation from the C=C bond. This was achieved by combining aryldiazonium salts with a chiral pyridine-oxazoline ligand. The overall results are spectacular as chiral centers were installed with almost perfect stereocontrol





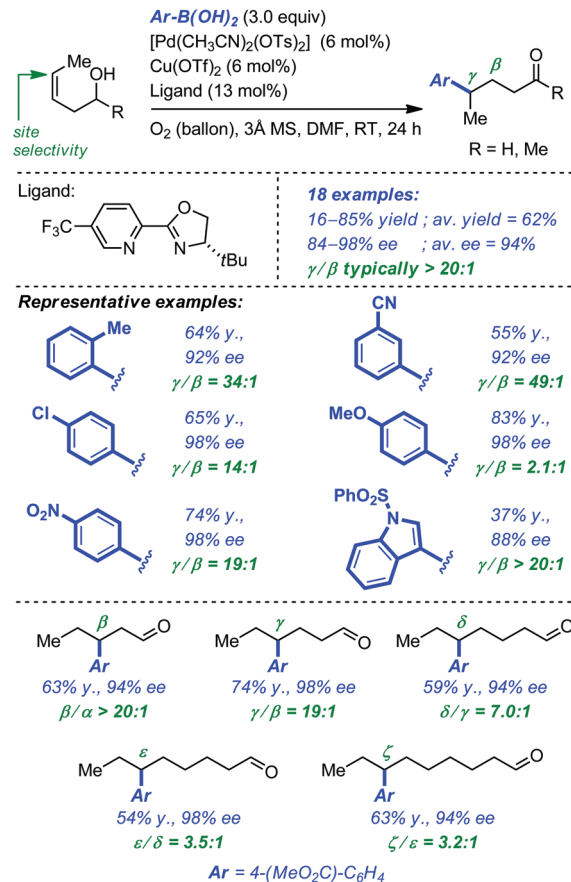
Scheme 11 Pd-catalyzed enantioselective redox-relay Heck reaction of alkenyl primary and secondary alcohols.

not only at the β but also at the γ and δ positions of the resulting carbonyl group. In the latter cases, site selectivity remains to be improved. Of additional note, higher yields were obtained when starting from (*Z*) rather than (*E*) alkenes. A selection of the aldehydes and ketones that were obtained by this method is represented in Scheme 11.

To circumvent some of the limitations of their initial system, Sigman and co-workers have next developed an enantioselective redox-relay Heck arylation of linear alkenyl alcohols using boronic acids (*i.e.* oxidative Heck).³⁵ In addition to the coupling partners, the optimized protocol required the combination of [Pd(CH₃CN)₂(OTf)₂] (6 mol%), [Cu(OTf)₂] (6 mol%), the chiral ligand (13 mol%), O₂ and 3 Å molecular sieves in DMF. The reaction for homoallylic alcohols turned out to be very general as electron-rich or electron-deficient aryl boronic acids were equally tolerated, producing the corresponding γ -arylated aldehydes and ketones in high yield, excellent enantioselectivity and much improved regioselectivity. Extension to alkenols with longer carbon chains proved also feasible, with the more distant C(sp²) being systematically preferred for aryl insertion. While the yields and enantioselectivities remained very high, the regioselectivity of arene insertion was found to decrease as the length of the alkyl chain increased (Scheme 12).

5. Conclusions

The ability to selectively functionalize carbonyl derivatives at more and more distant positions from the C=O bond using



Scheme 12 Pd-catalyzed oxidative redox-relay Heck reaction of alkenyl alcohols.



Pd catalysis has attracted increased interest in recent years. Progress in the directing group remote functionalization strategy has been mainly focused on elaborating more efficient auxiliaries that can not only facilitate the C(sp³)-H bond functionalization but can also be installed and removed easily. In addition to the C-C bond forming process, intra and intermolecular C-N and C-O bond forming reactions leading to ubiquitous heterocyclic structures have also been developed. As most of the approaches reported permit functionalization at the γ position, it would be interesting to develop novel directing groups to functionalize efficiently more remote C-H bonds. The vinylogous coupling of enolates now allows to introduce indifferently electron-rich, electron-poor or even congested aryl halides with high levels of efficiency at the γ position of α,β -unsaturated carbonyl derivatives. Generalization of these arylations to more remote positions is certainly an important objective. Furthermore, additional studies are expected to be focused on the development of the asymmetric versions of these reactions. The formidable achievements in the enantioselective redox-relay Heck and oxidative Heck reactions of alkenyl alcohols developed by Sigman and co-workers have paved the way for further studies. The scope of these reactions certainly needs to be extended. For instance generating quaternary stereocenters at remote positions from the carbonyl group produced upon isomerization of the C=C bond would constitute an important achievement.

Acknowledgements

The authors thank the University of Geneva and the Swiss National Foundation (Project PP00P2_133482) for financial support.

Notes and references

- (a) R. Breslow, *Chem. Soc. Rev.*, 1972, **1**, 553; (b) R. Breslow, *Acc. Chem. Res.*, 1980, **13**, 170.
- (a) H. Schwarz, *Acc. Chem. Res.*, 1989, **22**, 282; (b) H. Sailes and A. Whiting, *J. Chem. Soc., Perkin Trans. 1*, 2000, 1785; (c) J. S. Johnson and D. A. Evans, *Acc. Chem. Res.*, 2000, **33**, 325; (d) K. Mikami, M. Shimizu, H.-C. Zhang and B. E. Maryanoff, *Tetrahedron*, 2001, **57**, 2917.
- (a) J. Clayden and N. Vassiliou, *Org. Biomol. Chem.*, 2006, **4**, 2667; (b) J. Clayden, *Chem. Soc. Rev.*, 2009, **38**, 817.
- For recent reviews on organocatalytic methods: (a) D. W. C. MacMillan, *Nature*, 2008, **455**, 304; (b) S. Bertelsen and K. A. Jørgensen, *Chem. Soc. Rev.*, 2009, **38**, 2178; (c) K. L. Jensen, G. Dickmeiss, H. Jiang, L. Albrecht and K. A. Jørgensen, *Acc. Chem. Res.*, 2012, **45**, 248; (d) E. Arceo and P. Melchiorre, *Angew. Chem., Int. Ed.*, 2012, **51**, 5290; (e) J.-L. Li, T.-Y. Liu and Y.-C. Chen, *Acc. Chem. Res.*, 2012, **45**, 1491; (f) I. Kumar, P. Ramaraju and N. A. Mir, *Org. Biomol. Chem.*, 2013, **11**, 709; (g) H. Jiang, L. Albrecht and K. A. Jørgensen, *Chem. Sci.*, 2013, **4**, 2287.
- For general reviews on C-H functionalization see: (a) K. Godula and D. Sames, *Science*, 2006, **312**, 67; (b) X. Chen, K. M. Engle, D.-H. Wang and J.-Q. Yu, *Angew. Chem., Int. Ed.*, 2009, **48**, 5094; (c) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147; (d) L. McMurray, F. O'Hara and M. J. Gaunt, *Chem. Soc. Rev.*, 2011, **40**, 1885; (e) J. Wencel-Delord, T. Dröge, F. Liu and F. Glorius, *Chem. Soc. Rev.*, 2011, **40**, 4740; for a recent special issue, see: (f) *Acc. Chem. Res.*, 2012, **6**, 777-958. For a review on C(sp²)-H functionalization, see: (g) D. Alberico, M. E. Scott and M. Lautens, *Chem. Rev.*, 2007, **107**, 174; for reviews on C(sp³)-H functionalization, see: (h) R. Jazzar, J. Hitce, A. Renaudat, J. Sofack-Kreutzer and O. Baudoin, *Chem.-Eur. J.*, 2010, **16**, 265; (i) O. Baudoin, *Chem. Soc. Rev.*, 2011, **40**, 4902.
- The Yu group has recently reported a fascinating Pd-catalyzed meta functionalization of aromatic substrates using a quite elaborated directing group: D. Leow, G. Li, T.-S. Mei and J.-Q. Yu, *Nature*, 2012, **486**, 518. Upon unmasking a carbonyl group is potentially revealed. As this is a formal C(sp²)-H bond activation, this paper will not be discussed here. For a recent highlight, see: T. Truong and O. Daugulis, *Angew. Chem., Int. Ed.*, 2012, **51**, 11677.
- V. G. Zaitsev, D. Shabashov and O. Daugulis, *J. Am. Chem. Soc.*, 2005, **127**, 13154.
- G. He and G. Chen, *Angew. Chem., Int. Ed.*, 2011, **50**, 5192.
- G. He, Y. Zhao, S. Zhang, C. Lu and G. Chen, *J. Am. Chem. Soc.*, 2012, **134**, 3.
- G. He, S.-Y. Zhang, W. A. Nack, Q. Li and G. Chen, *Angew. Chem., Int. Ed.*, 2013, **52**, 11124.
- For other examples of Pd-catalyzed AQ-directed C-H functionalization see: (a) B. V. S. Reddy, L. R. Reddy and E. J. Corey, *Org. Lett.*, 2006, **8**, 3391; (b) Y. Feng and G. Chen, *Angew. Chem., Int. Ed.*, 2010, **49**, 958; (c) Y. Feng, Y. Wang, B. Landgraf, S. Liu and G. Chen, *Org. Lett.*, 2010, **12**, 3414; (d) Y. Ano, M. Tobisu and N. Chatani, *J. Am. Chem. Soc.*, 2011, **133**, 12984; (e) W. R. Gutekunst, R. Gianatassio and P. S. Baran, *Angew. Chem., Int. Ed.*, 2012, **51**, 7507.
- S.-H. Yu, G. He, Y. Zhao, K. Wright, W. A. Nack and G. Chen, *J. Am. Chem. Soc.*, 2012, **134**, 7313.
- (a) A. García-Rubia, R. Gómez Arrayás and J. C. Carretero, *Angew. Chem., Int. Ed.*, 2009, **48**, 6511; (b) A. García-Rubia, B. Urones, R. Gómez Arrayás and J. C. Carretero, *Chem.-Eur. J.*, 2010, **16**, 9676; (c) A. García-Rubia, B. Urones, R. Gómez Arrayás and J. C. Carretero, *Angew. Chem., Int. Ed.*, 2011, **50**, 10927.
- C. N. Rodríguez, J. A. Romero-Revilla, M. A. Fernández-Ibáñez and J. C. Carretero, *Chem. Sci.*, 2013, **4**, 175.
- M. Ochiai, K. Miyamoto, T. Kaneaki, S. Hayashi and W. Nakanishi, *Science*, 2011, **332**, 448.
- M. Fan and D. Ma, *Angew. Chem., Int. Ed.*, 2013, **52**, 12152.
- For recent reviews on enolates α -arylation (a) D. A. Culkin and J. F. Hartwig, *Acc. Chem. Res.*, 2003, **36**, 234;



- (b) A. C. B. Bortoloso, *Synlett*, 2009, 320; (c) C. C. C. Johansson and T. J. Colacot, *Angew. Chem., Int. Ed.*, 2010, **49**, 676; (d) F. Bellina and R. Rossi, *Chem. Rev.*, 2010, **110**, 1082; (e) C. Mazet, *Synlett*, 2012, 1999.
- 18 For recent examples of enantioselective α -arylations of amides: (a) S. Lee and J. F. Hartwig, *J. Org. Chem.*, 2001, **66**, 3402; (b) T. Aaro, K. Kondo and T. Aoyama, *Tetrahedron Lett.*, 2006, **47**, 1417; (c) E. P. Kündig, T. M. Seidel, Y.-X. Jia and G. Bernardinelli, *Angew. Chem., Int. Ed.*, 2007, **46**, 8484; (d) X. Luan, R. Mariz, C. Robert, M. Gatti, S. Blumentritt, A. Linden and R. Dorta, *Org. Lett.*, 2008, **10**, 5569; (e) S. Würtz, C. Lohre, R. Fröhlich, K. Bergander and F. Glorius, *J. Am. Chem. Soc.*, 2009, **131**, 8344; (f) X. Luan, L. Wu, E. Drinkel, R. Mariz, M. Gatti and R. Dorta, *Org. Lett.*, 2010, **12**, 1912; (g) Y.-X. Jia, D. Katayev, G. Bernardinelli, T. M. Seidel and E. P. Kündig, *Chem.-Eur. J.*, 2010, **16**, 6300; (h) L. Liu, N. Ishida, S. Ashida and M. Murakami, *Org. Lett.*, 2011, **13**, 1666.
- 19 For recent examples on the asymmetric α -arylation of esters: (a) R. Galarini, A. Musco, R. Pontellini and R. Santi, *J. Mol. Catal.*, 1992, **72**, L11; (b) D. J. Spielvogel and S. L. Buchwald, *J. Am. Chem. Soc.*, 2002, **124**, 3500; (c) D. J. Spielvogel, W. M. Davis and S. L. Buchwald, *J. Am. Chem. Soc.*, 2002, **121**, 3833; (d) X. Liu and J. F. Hartwig, *J. Am. Chem. Soc.*, 2004, **126**, 5182.
- 20 For recent examples of the enantioselective α -arylation of ketones: (a) J. M. Fox, X. Huang, A. Chieffi and S. L. Buchwald, *J. Am. Chem. Soc.*, 2000, **122**, 1360; (b) T. Hamada, A. Chieffi, J. Åhman and S. L. Buchwald, *J. Am. Chem. Soc.*, 2002, **124**, 1261; (c) G. Chen, F. Y. Kwong, H. O. Chan, W.-Y. Yu and A. S. C. Chan, *Chem. Commun.*, 2006, 1413; (d) X. Liao, Z. Weng and J. F. Hartwig, *J. Am. Chem. Soc.*, 2008, **130**, 195; (e) S. Ge and J. F. Hartwig, *J. Am. Chem. Soc.*, 2011, **133**, 1633.
- 21 For the enantioselective α -arylations of aldehydes: (a) J. Garcia-Fortanet and S. L. Buchwald, *Angew. Chem., Int. Ed.*, 2008, **47**, 8108; (b) P. Nareddy, L. Mantilli, L. Guénée and C. Mazet, *Angew. Chem., Int. Ed.*, 2012, **51**, 3826; (c) N. Humbert, E. Larionov, L. Mantilli, P. Nareddy, C. Besnard, L. Guénée and C. Mazet, *Chem.-Eur. J.*, 2013, DOI: 10.1002/chem.201303146.
- 22 For reviews on reactions based on the vinylogous analogy: (a) R. C. Fuson, *Chem. Rev.*, 1935, **16**, 1; (b) G. Casiraghi, F. Zanardi, G. Appendino and G. Rassu, *Chem. Rev.*, 2000, **100**, 1929; (c) S. F. Martin, *Acc. Chem. Res.*, 2002, **35**, 895; (d) S. E. Denmark, J. R. Heemstra Jr. and G. L. Beutner, *Angew. Chem., Int. Ed.*, 2005, **44**, 4682.
- 23 (a) Y. Terao, T. Satoh, M. Miura and M. Nomura, *Tetrahedron Lett.*, 1998, **39**, 6203; (b) Y. Terao, T. Satoh, M. Miura and M. Nomura, *Bull. Chem. Soc. Jpn.*, 1999, **72**, 2345.
- 24 D. S. Huang and J. F. Hartwig, *Angew. Chem., Int. Ed.*, 2010, **49**, 5757.
- 25 I. Franzoni, L. Guénée and C. Mazet, *Chem. Sci.*, 2013, **4**, 2619.
- 26 (a) *Quaternary Stereocentres: Challenges and Solutions for Organic Synthesis*, ed. J. Christoffers and A. Baro, Wiley-VCH, Weinheim, 2005; (b) K. Fuji, *Chem. Rev.*, 1993, **93**, 2037; (c) A. B. Dounay and L. E. Overman, *Chem. Rev.*, 2003, **103**, 2945; (d) M. Bella and T. Gasperi, *Synlett*, 2009, 1583; (e) I. Marek and G. Sklute, *Chem. Commun.*, 2007, 1683; (f) C. Hawner and A. Alexakis, *Chem. Commun.*, 2010, **46**, 7295.
- 27 (a) A. Renaudat, L. Jean-Gérard, R. Jazzar, C. E. Kefalidis, E. Clot and O. Baudoin, *Angew. Chem., Int. Ed.*, 2010, **49**, 7261. For a seminal contribution, see: (b) M. Jørgensen, S. Lee, X. Liu, J. P. Wolkowski and J. F. Hartwig, *J. Am. Chem. Soc.*, 2002, **124**, 12557.
- 28 P. Larini, C. E. Kefalidis, R. Jazzar, A. Renaudat, E. Clot and O. Baudoin, *Chem.-Eur. J.*, 2012, **18**, 1932.
- 29 S. Aspin, A.-S. Goutierre, P. Larini, R. Jazzar and O. Baudoin, *Angew. Chem., Int. Ed.*, 2012, **124**, 10966.
- 30 A. M. Hyde and S. L. Buchwald, *Angew. Chem., Int. Ed.*, 2008, **47**, 177.
- 31 A. M. Hyde and S. L. Buchwald, *Org. Lett.*, 2009, **11**, 2663.
- 32 (a) B. Figadère, *Acc. Chem. Res.*, 1995, **28**, 359; (b) M. F. Braña, M. L. García, B. López, B. de Pascual-Teresa, A. Ramos, J. M. Pozuelo and M. T. Domínguez, *Org. Biomol. Chem.*, 2004, **2**, 1864; (c) J. Dogné, C. T. Supuran and D. J. Pratico, *J. Med. Chem.*, 2005, **48**, 2251.
- 33 E. W. Werner, T.-S. Mei, A. J. Burckle and M. S. Sigman, *Science*, 2012, **338**, 1455.
- 34 (a) R. C. Larock, W.-Y. Leung and S. Stolz-Dunn, *Tetrahedron Lett.*, 1989, **30**, 6629; (b) K. Yonehara, K. Mori, T. Hashizume, K.-G. Chung, K. Ohe and S. Uemura, *J. Organomet. Chem.*, 2000, **603**, 40; (c) S. Bouquillon, B. Ganchev, B. Estrine, F. Hémin and J. Muzart, *J. Organomet. Chem.*, 2001, **634**, 153; (d) M. Quadir, R. E. Priestley, T. W. D. F. Rising, T. Gelbrich, S. J. Coles, M. B. Hursthouse, P. W. Shedrake, N. Whittall and K. K. (Mimi) Hii, *Tetrahedron Lett.*, 2003, **44**, 3675; (e) F. Berthiol, H. Doucet and M. Santelli, *Tetrahedron*, 2006, **62**, 4372.
- 35 T.-S. Mei, E. W. Werner, A. J. Burckle and M. S. Sigman, *J. Am. Chem. Soc.*, 2013, **135**, 6830.

