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## CORRECTION



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## Correction: Relative contributions of quantum and double layer capacitance to the supercapacitor performance of carbon nanotubes in an ionic liquid

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Correction for 'Relative contributions of quantum and double layer capacitance to the supercapacitor performance of carbon nanotubes in an ionic liquid' by Alexander J. Pak, Eunsu Paek and Gyeong S. Hwang, *Phys. Chem. Chem. Phys.*, 2013, **15**, 19741–19747.

The following errata were found in the published article. However, the overall conclusions of the original article remain unaffected.

1. In our analysis, the DOS was mistakenly underestimated by a factor of two. As a result, the *y*-axis labels for the DOS and  $C_Q$  in Fig. 5 should be doubled and are corrected below.



Fig. 5 Comparison of the quantum capacitance ( $C_{O}$ ) between the (6,6) CNT and graphene electrodes.  $E_{F}$  indicates the Fermi level. The inset shows the DOS (in eV<sup>-1</sup> Å<sup>-1</sup>) of the CNT.

- 2. The discussion of Fig. 5 on p. 19745 should be modified to reflect the corrected DOS and  $C_{\rm O}$  as follows:
- (a) "One important feature to note is that the DOS is nearly constant ( $\approx 0.29 \text{ eV}^{-1} \text{ Å}^{-1}$ ) when |E| < 1.0 eV..."
- (b) "The  $C_{\rm Q,CNT}$  exhibits a constant profile with a value around 18.0  $\mu$ F cm<sup>-2</sup> when  $|\phi_{\rm C}| < 1.0$  V"

(c) There is an additional typo which should be corrected as follows: "The approximated DOS in this energy range ( $\approx 0.24 \text{ eV}^{-1} \text{ Å}^{-1}$ ) shows good agreement with DFT calculations..."

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3. In Fig. 6, the  $C_{\rm T}$  and  $\phi_{\rm a}$  are subsequently corrected as shown below.



**Fig. 6** The double layer ( $C_D$ ), quantum ( $C_Q$ ), and total interfacial ( $C_T$ ) capacitance at the listed  $\sigma$  (in  $\mu$ C cm<sup>-2</sup>). The ( $\Psi$ ) indicates the corresponding applied potential ( $\phi_a$ ).

4. The discussion of Fig. 6 on p. 19745 should be corrected as follows:

(a) "When  $\sigma = \pm 0.40 \ \mu\text{C cm}^{-2}$  [Fig. 6(a)], the  $C_{\text{T}}$  for the CNT electrode is predicted to be  $C_{\text{T,CNT}} \approx 3.8-5.7 \ \mu\text{F cm}^{-2}$  which is 2.7-3.2 times larger than the graphene case with  $C_{\text{T,GT}} \approx 1.4-1.8 \ \mu\text{F cm}^{-2}$ ."

(b) "However, when  $\sigma = \pm 6.33 \ \mu\text{C cm}^{-2}$  [Fig. 6(b)],  $C_{T,Gr}$  and  $C_{T,CNT}$  are much more comparable ( $\approx 3.2-4.5 \ \mu\text{F cm}^{-2}$ ) although  $C_{T,CNT}$  is around 28–31% larger than  $C_{T,Gr}$ .

This is primarily due to the large increase (slight decrease) in  $C_{Q,Gr}$  ( $C_{Q,CNT}$ ) such that both  $C_{T,Gr}$  and  $C_{T,CNT}$  are primarily limited by the respective  $C_D$  values."

5. The conclusion on p. 19746 should be corrected as follows:

(a) "... the  $C_Q$  of the (6,6) CNT is nearly constant (18  $\mu$ F cm<sup>-2</sup>)..."

(b) "However, the enhancement of  $C_{\rm T}$  is likely diminished (predicted to be 28–31%)..."

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.