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Proof-of-principle direct double cyclisation of a linear C₁₅-precursor to a dibrominated bicyclic medium-ring ether relevant to Laurencia species†

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Bicyclic dibrominated C₁₅ medium-ring ether hexahydrolaureoxanyne was produced directly from an acyclic model C₁₅-epoxide when treated with NBS with water as the solvent.

Since the original isolation of Laurencin (1a) in 1965, marine red algae of Laurencia species have provided a wide variety of C₁₅-acetogenic halogenated diastereo- and constitutional isomeric monocyclic (C₁₅H₂₁BrO₂) and bicyclic (C₁₅H₂₀Br₂O₂) medium-ring ethers that are oxygenated at both C-6 and C-7 (Fig. 1).2 Both the monocyclic and bicyclic metabolites have received considerable synthetic attention, with numerous necessarily different strategies used to forge the 7-, 8-, or 9-membered medium-ring, control the cis or trans α, α' -ether stereochemistry, install the requisite halogen(s), and - in the case of the bicyclic ethers - to fashion the second ring.³⁻⁵ Various recent studies have also been directed at the further understanding of their biogenesis, 6 where the early pioneering work of Murai⁷ demonstrated enzymatic bromoetherifications of straight-chain co-isolated unsaturated C_{15} -diols – laurediols (3E,6R,7R)-7a and (3Z,6S,7S)-7b⁸ – to monocyclic medium-ring ethers deacetyl laurencin 1b and prelaureatin 2 respectively, albeit in very low yields (Scheme 1, top). We have recently advanced an alternative biogenesis for the monocyclic (C₁₅H₂₁BrO₂) medium-ring ethers from *Laurencia* species from (6S,7R)-epoxide 8 via an intramolecular bromonium ion assisted epoxide ring-opening (IBIAERO) reaction with water functioning as the external nucleophile (Scheme 1, bottom, $8 \rightarrow B \rightarrow O/O' \rightarrow 1b/2$), and experimentally corroborated this with a model epoxide for the concurrent formation of 7-, 8- and 9-ring ethers corresponding to the halogenated medium-ring ethers of known metabolites from Laurencia species. 10,11 The bicyclic metabolites are generally considered to originate by further bromoetherification of the residual unsaturation of the monocyclic compounds - the Z-configured

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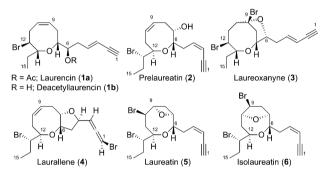


Fig. 1 Representative monocyclic and bicyclic halogenated medium-ring ethers of formulae $C_{15}H_{21}BrO_2$ (**1b**, **2**) and $C_{15}H_{20}Br_2O_2$ (**3–6**) from Laurencia species that are oxygenated at C-6 and C-7. Laurencin 1a is related as the acetate of 1b.

medium-ring alkene or the pendant enyne - using the free alcohol of the original monocyclic compound located either at C-6 or C-7 as the nucleophile (Scheme 1, top).7 Several laboratory demonstrations of these later transformations have been successful, either as enzymatic-mediated bromoetherifications of naturally occurring monocycles, 12 or as part of the synthetic strategy in a total synthesis of the bicyclic natural products. 13 Interestingly, although bromocyclisation events had been postulated for both monocycle and bicycle formation, prior to our 2012 report¹⁰ and Snyder's recent elegant work, 6b,c a non-enzymatic bromonium-ion induced cyclisation process to directly form medium-ring ether cores relevant to Laurencia species had not been reported. Moreover, to the best of our knowledge, there has been no report of a C₁₅-dibrominated bicyclic medium-ring ether relevant to Laurencia species being formed directly from a linear unsaturated C₁₅-precursor by two successive bromination events in the same pot. Herein we report on a successful strategy to effect such a transformation.

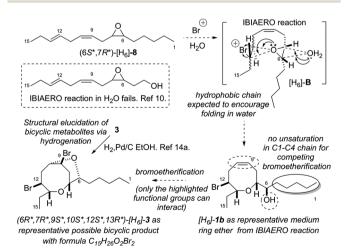
To investigate the proof-of-principle demonstration of a direct double cyclisation of a C15 unsaturated linear precursor to a bicyclic medium-ring ether relevant to Laurencia species we targeted hexahydroepoxide $(6S^*,7R^*)$ - $[H_6]$ -8, with the aim that this

[†] Electronic supplementary information (ESI) available: Experimental procedures, characterising data and 1H and 13C NMR spectra for all compounds; a comparison of ¹H NMR data for (\pm) -[H₆]-3 with the literature data. See DOI: 10.1039/c4cc06402j

Communication ChemComm

Scheme 1 Irie–Murai biogenesis of monocyclic medium-ring ethers from laurediols 7a and 7b (top); alternative biogenesis of deacetyllaurencin 1b and prelaureatin 2 via IBIAERO reaction with water functioning as the external nucleophile (bottom). The other six possible monocyclic ethers of formulae $C_{15}H_{21}BrO_2$ are not shown.

would undergo an initial IBIAERO reaction via [H₆]-**B** where water functions as both the solvent and the nucleophile (Scheme 2). The use of water in this manner thus guarantees a free hydroxyl group for any subsequent bromoetherification reaction (e.g., [H₆]-**1b** \rightarrow [H₆]-**3**, Scheme 2) with a second equivalent of an electrophilic bromine source. While we had previously demonstrated successful IBIAERO reactions in water with NBS as the electrophilic bromine source, ¹¹ the attempted IBIAERO reaction of a model epoxide as a truncated C₁₂ alcohol (inset, Scheme 2) under the same conditions had failed. ¹⁰‡ We considered that hexahydroepoxide [H₆]-**8** offered distinct benefits compared to this earlier model and also to epoxide **8** for the proposed experiment: (i) the hydrophilic hexahydro chain may encourage



Scheme 2 Proposed proof-of-principle direct cyclisation of $(6S^*,7R^*)$ - $[H_6]$ -8 to bicyclic medium ring ethers via IBIAERO reaction and subsequent bromoetherification of the remaining unsaturation.

$$\begin{array}{c} OH \\ \hline & PrMgCl, \, CuCl, \, THF-\\ Et_2O, \, 70\,^{\circ}C, \, 5h, \, 81\% \\ \hline \\ 9\, X = OH \\ 10\, X = Br \end{array} \begin{array}{c} PPh_3Br_2, \\ py, \, CH_2Cl_2, \\ r.t., \, 3h, \, 68\% \\ \hline \\ 12\, X = Br \end{array} \begin{array}{c} 11\, X = OH \\ 12\, X = Br \end{array} \begin{array}{c} PPh_3, \, CBr_4, \\ CH_2Cl_2, \, -15\,^{\circ}C, \, > r.t., \, 2h \\ \hline \\ Hept-1-yne, \, K_2CO_3, \, Nal, \, Cul, \, Acetone, \, 70\,^{\circ}C, \, 20\, h \\ 70\,^{\circ}C, \, 20\, h \\ \hline \\ 14\, \end{array} \begin{array}{c} 5\% \, Pd\text{-BasO}_4, \, H_2, \\ quinoline, \, MeOH, \, cyclohexene \\ r.t., \, 0.75\, h, \, 94\% \\ \hline \\ 14\, \end{array} \begin{array}{c} S:15 = 0.41:0.34 \\ 68\% \, (90\% \, brsm) \\ acetone, \, r.t. \, 16\, h \\ \hline \\ (6S^*, 7R^*) - [H_6] - 8. \end{array} \begin{array}{c} 15\, \end{array}$$

folding of the substrate in water thus inherently facilitating the IBIAREO reaction; (ii) post-IBIAERO reaction, the only region of unsaturation will be located in the medium ring and – compared with the hypothetical use of the putative biosynthetic precursor itself, epoxide 8 – there can be no complicating bromoetherifications to form bromoallene adducts by cyclisation onto any C_1 – C_4 enyne moiety; (iii) hexahydrobicyclic compounds of formulae $C_{15}H_{26}O_2Br_2$ are known in the literature as a consequence of the structural elucidation of the naturally occurring compounds *via* hydrogenation, ¹⁴ providing data for identification of bicyclic products.

Accordingly, epoxide $(6S^*,7R^*)$ -[H₆]-8 was synthesised from bromide **12**, itself prepared from (E)-2-penten-1-ol (9) via a known sequence^{10,15} with minor modifications. Subsequent coppermediated coupling¹⁶ with hept-1-yne gave novel enediyne **13** (Scheme 3).† Chemoselective and stereoselective hydrogenation¹⁷ afforded (E,Z,Z)-doubly skipped triene **14**. Epoxidation of triene **14** with DMDO¹⁸ was found to be entirely selective for the Z-olefins,¹⁹ giving a mixture of mono epoxides $(6S^*,7R^*)$ -[H₆]-8 and **15** which could be separated by chromatography. \P

With epoxide $(6S^*,7R^*)$ - $[H_6]$ -8 in hand, it was treated with two equivalents of NBS - a water stable reagent - under high dilution conditions in water (Scheme 4).** Here, various dibromination adducts, bromohydrin regioisomers, and dibromotetrahydrofurans are expected to be formed by competing processes. 10 In the event, as expected, a complex mixture was obtained that was subjected to extensive chromatography, where 'non-polar' components could be separated away from 'polar' components.†† Much to our delight, by further chromatography of the non-polar components, hexahydrolaureoxanyne $[(\pm)-[H_6]-3]^{12a}$ was isolated as a bicyclic medium-ring ether with ¹H NMR data identical to that previously reported,†‡‡ along with dibromoepoxides 16. Thus the desired proof-of-principle has been achieved. This also constitutes the first synthetic route to the laureoxanyne bicyclic medium-ring ether scaffold, and the isolated yield of (\pm) - $[H_6]$ -3 (2.5%) from $(6S^*,7R^*)$ - $[H_6]$ -8 compares well with the reported enzymatic conversion of deacetyl laurencin 1b (obtained from natural laurencin 1a) into 3 (3%).12a

ChemComm Communication

Scheme 4 Proof-of-principle direct double cyclisation of (6S*,7R*)-[H₆]-8 into (±)-[H₆]-3 via IBIAERO reaction and subsequent bromoetherification of the remaining unsaturation (cf., Scheme 2).

In conclusion, we have demonstrated the proof-of-principle direct cyclisation of a linear unsaturated C15-precursor into a C₁₅-dibrominated bicyclic medium-ring ether relevant to Laurencia species – where hexahydrolaureoxanyne (\pm)-[H₆]-3 has an identical bicyclic medium ring ether framework to laureoxanyne 3 - by two successive bromination events in the same pot. These studies are also consistent with epoxide (6S,7R)-8 acting as the biogenetic precursor¹⁰ for bromocyclisation to bicyclic medium-ring ethers of Laurencia species via IBIAERO reactions followed by subsequent bromoetherification events.

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Notes and references

‡ We speculate that the truncated C12 epoxide suffers from an intramolecular hydrogen bond from the alcohol functional group reducing its nucleophilicity.

- § 25% of a bis-epoxide was also observed.
- \P Attempted epoxidation of 14 with mCPBA was unselective for the Z-olefins.
- || 1H-13C and 1H-1H NMR correlation spectroscopy were used to distinguish between epoxides $(6S^*,7R^*)$ - $[H_6]$ -8 and 15.†
- In an experiment with 1 equivalent of NBS in water, (\pm) -[H₆]-3 was isolated in 1.8% yield after extensive chromatography.
- †† The 'polar' components were expected to contain regioisomeric bromohydrins and dibromohydrins by reference to our earlier work (ref. 10) and were not further characterised.
- ‡‡ The medium-ring bicyclic structure of [H₆]-3 is also supported by a characteristic NOESY cross-peak between H₇ and H₉ as previously reported (as an nOe) for 3 (ref. 12a).†
- 1 (a) T. Irie, M. Suzuki and T. Masamune, Tetrahedron Lett., 1965, 16, 1091-1099; (b) T. Irie, M. Suzuki and T. Masumune, Tetrahedron, 1968, 24, 4193-4205.
- 2 For comprehensive reviews see: (a) B.-G. Wang, J. B. Gloer, N.-Y. Ji and J.-C. Zhao, Chem. Rev., 2013, 113, 3632-3685; (b) J. W. Blunt, B. R. Copp, R. A. Keyzers, M. H. G. Munro and M. R. Prinsep, Nat. Prod. Rep., 2014, 31, 160-258 and earlier reviews in this series.
- 3 For a comprehensive review of the synthesis of medium-ring ethers from Laurencia sp., see: K. Fujiwara, Top. Heterocycl. Chem., 2006, 5, 97-148. See also ref. 2a.
- 4 For recent leading syntheses of C_{15} Laurencia metabolites see: (a) G. Kim, T.-i. Sohn, D. Kim and R. S. Paton, Angew. Chem., 2014, 126, 276-280; (b) C. Recsei, B. Chan and C. S. P. McErlean, J. Org. Chem., 2014, 79, 880-887; (c) J. Rodríguez-López, N. Ortega, V. S. Martín and T. Martín, Chem. Commun., 2014, 50, 3685-3688; (d) M. T. Holmes and R. Britton, Chem. - Eur. J., 2013, 19, 12649-12652; (e) D. J. Shepherd, P. A. Broadwith, B. S. Dyson, R. S. Paton and J. W. Burton, Chem. - Eur. J., 2013, 19, 12644-12648; (f) B. S. Dyson, J. W. Burton, T.-i. Sohn, B. Kim, H. Bae and D. Kim, J. Am. Chem. Soc., 2012, 134, 11781-11790; (g) M. J. Kim, T.-i. Sohn, D. Kim and

- R. S. Paton, J. Am. Chem. Soc., 2012, 134, 20178-20188 and references cited therein.
- 5 For two recent accounts of research in the arena see: (a) T. Martín, J. I. Padrón and V. S. Martín, Synlett, 2014, 12-32; (b) D. Kim, Synlett, 2014, 33-57,
- 6 For recent representative examples see: (a) S. Keshipeddy, I. Martínez, B. F. Castillo II, M. D. Morton and A. R. Howell, J. Org. Chem., 2012, 77, 7883-7890; (b) S. A. Snyder, A. P. Brucks, D. S. Treitler and I. Moga, J. Am. Chem. Soc., 2012, 134, 17714-17721; (c) S. A. Snyder, D. S. Treitler, A. P. Brucks and W. Sattler, J. Am. Chem. Soc., 2011, 133, 15898-15901.
- 7 For a review see: A. Murai, in *Comprehensive Natural Products Chemistry*, ed. D. H. R. Barton, O. Meth-Cohn and K. Nakinishi, Elsevier, Oxford, 1999, vol. 1, pp. 303-324 and references cited therein.
- 8 E. Kurosawa, A. Fukuzawa and T. Irie, Tetrahedron Lett., 1972, 13, 2121-2124.
- 9 (a) Lactoperoxidase (LPO) mediated cyclisation of 7a into 1b (characterized as 1a after acetylation; 0.73% yield): A. Fukuzawa, M. Aye and A. Murai, Chem. Lett., 1990, 1579–1580(b) LPO mediated cyclisation of 7b into 2 (3%): A. Fukuzawa, Y. Takasugi, A. Murai, M. Nakamura and M. Tamura, Tetrahedron Lett., 1992, 33, 2017–2018(c) Bromoperoxidase (BPO) mediated cyclisation of 7a into 1b (0.015%) and 7b into 2 ('trace' amount): A. Fukuzawa, M. Aye, Y. Takasugi, M. Nakamura, M. Tamura and A. Murai, Chem. Lett., 1994, 2307-2310.
- K. J. Bonney and D. C. Braddock, J. Org. Chem., 2012, 77, 9574-9584 and references cited therein.
- 11 For an IBIAERO reaction with capture of the oxonium ion with an added external nucleophile, see: K. J. Bonney, D. C. Braddock, A. J. P. White and M. Yaqoob, J. Org. Chem., 2011, 76, 97-104 and references cited therein.
- 12 (a) BPO mediated cyclisation of **1b** into **3** (3%): A. Fukuzawa, M. Aye, M. Nakamura, M. Tamura and A. Murai, Tetrahedron Lett., 1990, 31, 4895-4898. See ref. 9b for LPO mediated conversion of 1b into laureatin 5 (0.3%). See ref. 9c for BPO mediated conversion of 1b into laureoxanyne 3 (0.8%), and [1-2H]-2 into [1-2H]-5 (laureatin) (0.07%) and $[1-^2H]$ -6 (isolaureatin) (0.05%)(b) For a chemical conversion of [1-2H]-2 into [1-2H]-4 (12%) see: J. Ishihara, Y. Shimada, N. Kanoh, Y. Takasugi, A. Fukuzawa and A. Murai, Tetrahedron, 1997, 53, 8371-8382.
- (a) For the formation of the bromoallene of laurallene 4 from (E)-prelaureatin (24%) see: M. T. Crimmins and E. A. Tabet, J. Am. Chem. Soc., 2000, 122, 5473-5476; for the formation of the tetrahydrofuran ring of (–)-isoprelaurefucin from a pre-existing oxepene (92%) see: (b) H. Lee, H. Kim, T. Yoon, B. Kim, S. Kim, H.-D. Kim and D. Kim, J. Org. Chem., 2005, 70, 8723-8729; the actual chemical conversion of the prelaureatin skeleton into laureatin and/or isolaureatin bicyclics has proved challenging: (c) H. Kim, H. Lee, D. Lee, S. Kim and D. Kim, J. Am. Chem. Soc., 2007, 129, 2269-2274; (d) M. Sugimoto, T. Suzuki, H. Hagiwara and T. Hoshi, Tetrahedron Lett., 2007, 48, 1109-1112.
- 14 (a) Hexahydrolaureoxanyne ($[H_6]$ -3): ref. 12a; (b) hexahydrolaureatin ([H₆]-5): T. Irie, M. Izawa and E. Kurosawa, Tetrahedron Lett., 1968, 2091–2096(c) hexahydroisolaureatin ($[H_6]$ -6): T. Irie, M. Izawa and E. Kurosawa, Tetrahedron Lett., 1968, 2735-2738; (d) hexahydroisoprelaurefucin: M. Suzuki, K. Kurata, T. Suzuki and E. Kurosawa, Bull. Chem. Soc. Jpn., 1986, 59, 2953-2955.
- 15 (a) O. Loreau, A. Maret, D. Poullain, J. M. Chardigny, J. L. Sébédio, B. Beaufrère and J. P. Noël, Chem. Phys. Lipids, 2000, 106, 65-78; see also: (b) W. G. Young, L. Richards and J. Azorlosa, J. Am. Chem. Soc., 1939, 61, 3070-3074; (c) F. P. Cossío, I. Ganboa and C. Palomo, Tetrahedron Lett., 1985, 26, 3041-3044; (d) L. M. Smith, R. G. Smith, T. M. Loehr, G. D. Daves Jr., G. E. Daterman and R. H. Wohleb, J. Org. Chem., 1978, 43, 2361-2366; (e) B. Añorbe, V. S. Martin, J. M. Palazón and J. M. Trujillo, Tetrahedron Lett., 1986, 27, 4991-4994.
- 16 N. P. Villalva-Servin, A. Laurent and A. G. Fallis, Can. J. Chem., 2004, 82, 227-239.
- 17 C. Oger, L. Balas, T. Durand and J.-M. Galano, Chem. Rev., 2013, 113, 1313-1350.
- 18 R. Murray and P. Singh, Org. Synth., 1997, 74, 91.
- 19 DMDO epoxidations of cis/trans-dialkylalkene pairs have been reported to have a ca. 10-fold greater reactivity for the former: A. L. Baumstark and P. C. Vasquez, J. Org. Chem., 1988, 53, 3437-3439.