# An enantioselective synthesis of the $\mathrm{C}_{24}-\mathrm{C}_{40}$ fragment of (-)-pulvomycin $\dagger$ 

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The $\mathrm{C}_{24}-\mathrm{C}_{40}$ fragment of (-)-pulvomycin was prepared in enantiomerically pure form using a concise synthesis method (15 linear steps from D-fucose, $6.8 \%$ overall yield) featuring a diastereoselective addition to an aldehyde, a $\beta$-selective glycosylation and a Stille crosscoupling as the key steps.

The antibiotic pulvomycin was first isolated in 1957 from a Streptomyces species but due to the limited analytical data no structure was assigned to the compound. ${ }^{1}$ In 1963, Akita et al. isolated a natural product from Streptomyces albosporeus var. labilomyceticus, ${ }^{2}$ which they called labilomycin and which was later shown to be identical to pulvomycin. ${ }^{3}$ Extensive analytical work by Smith et al. revealed the constitution of the natural product (Fig. 1) as well as the absolute and relative configuration at most stereogenic centers except for $\mathrm{C}_{32}$ and $\mathrm{C}_{33} .{ }^{4}$ The assignment was confirmed and the complete configuration was eventually proven by a crystal structure ( $1.4 \AA$ resolution) of pulvomycin with the bacterial elongation factor Tu (EF-Tu). ${ }^{5}$ It is well established that pulvomycin is a potent inhibitor of EF-Tu and it therefore represents a promising lead compound for the development of new antibiotics. ${ }^{6}$

While synthetic reports on pulvomycin are scarce, the biosynthesis of the pulvomycin aglycone has been elucidated by labeling experiments. ${ }^{7}$ Our own interest in pulvomycin was triggered by our previous studies on the synthesis ${ }^{8}$ and antibiotic activity ${ }^{9}$ of thiazole peptides, such as the GE factors and the amythiamicins. It has been shown that the EF-Tu binding site of pulvomycin is in close proximity to the binding site of thiazole peptides. ${ }^{10}$ The synthesis of pulvomycin and pulvomycin analogues might consequently help to further investigate the many facets of EF-Tu activity. ${ }^{11}$ Apart from its biological activity, pulvomycin presents itself as a formidable synthetic challenge due to its complex and labile structure. In this communication

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Fig. 1 Structure and compound numbering of (-)-pulvomycin.


Scheme 1 Retrosynthetic disconnection of the title compound 1 leading to D -fucose (2) as an appropriate carbohydrate substrate.
we disclose the enantioselective synthesis of a suitably protected $\mathrm{C}_{24}-\mathrm{C}_{40}$ fragment 1 (Scheme 1) of pulvomycin.

Retrosynthetically, it was envisioned that ketone 1 (TBDPS = tert-butyldiphenylsilyl) could be derived from commercially available d-fucose (2), which shows the correct configuration at the stereogenic centers $\left(\mathrm{C}_{36}-\mathrm{C}_{39}\right)$ of the pyranose ring. In order to establish the desired $\beta$-configuration at the glycosidic center an appropriate neighbouring group, e.g. an acetate, was required (at carbon atom $\left.\mathrm{C}_{36}\right)^{12}$ and the methyl ether linkage was to be introduced after glycosylation. There was precedence for the differentiation of the two equatorial hydroxy groups at $\mathrm{C}_{36}$ and $\mathrm{C}_{37}$ of d-fucose. ${ }^{13}$

Regarding the $\mathrm{C}_{24}-\mathrm{C}_{34}$ fragment, it seemed best to assemble the triene ${ }^{14}$ after the glycosylation step by an appropriate crosscoupling reaction, e.g. between $\mathrm{C}_{29}$ and $\mathrm{C}_{30}$. The stereogenic center at $\mathrm{C}_{33}$ appeared to be accessible from the chiral pool, e.g. from lactic acid, while the adjacent stereogenic center was to be introduced by a diastereoselective reaction.

The acetylation of d -fucose (2) (Scheme 2) proceeded quantitatively delivering the tetraacetate as an $\alpha / \beta$-mixture $(\alpha / \beta=95 / 5)$



Scheme 2 Synthesis of the protected glycosyl donor 5 from D-fucose (2). DMAP $=4$-( $N, N$-dimethylamino)pyridine, py = pyridine, im = imidazole.
of anomers. ${ }^{13}$ Conversion to the required thioacetal 3 proceeded best in our hands with ethanethiol and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2},{ }^{15}$ which delivered depending on the reaction conditions and on the reaction scale variable amounts of separable $\alpha / \beta$-isomers (see the ESI $\dagger$ for further details).

Since the relative configuration at the anomeric center was irrelevant for the desired glycosylation reaction, the $\alpha / \beta$-mixture of 3 was taken into the four-step procedure previously described for the selective preparation of alcohol $\beta-\mathbf{4}^{13}$ and it furnished the desired product 4 as an $\alpha / \beta$-mixture $(\alpha / \beta \cong 50 / 50)$ in a total yield of $60 \%$ over six steps from o-fucose (2). Conversion of the equatorial alcohol 4 to silyl ether 5 required elevated temperature ( $60{ }^{\circ} \mathrm{C}$ ) and a prolonged reaction time ( 3 d ).

As mentioned above, it was planned to introduce the stereogenic center at $\mathrm{C}_{32}$ by a diastereoselective reaction induced by the adjacent stereogenic center at the carbon atom $\mathrm{C}_{33}$. Surprisingly, the reduction of a ( $S$ )-lactate-derived, para-methoxybenzyl (PMB) protected alkynyl ketone ${ }^{16}$ produced the desired alcohol 7 either in low yields or with insufficient diastereoselectivity (see the ESI $\dagger$ for further details). As an alternative approach, $(S)$-lactate-derived aldehyde $\mathbf{6}^{17}$ was alkynylated with TMSacetylene under chelation control ${ }^{18}$ yielding alcohol 7 and its epimer epi-7 in $81 \%$ yield and in a diastereomeric ratio (d.r.) of 87/13 (Scheme 3). The diastereomerically pure product 7 was isolated in $65 \%$ yield.

Protection of the secondary alcohol proceeded smoothly at ambient temperature and the PMB group was cleaved oxidatively with 2,3-dicloro-5,6-dicyanobenzoquinone (DDQ) ${ }^{19}$ to deliver alcohol 8. The enantiomeric excess (ee) of alcohol 8 was established by chiral HPLC analysis and comparison with a racemic sample (see the ESI $\dagger$ for further details). Gratifyingly, the glycosylation reaction, when performed with $N$-iodosuccinimide (NIS) and trifluoromethanesulfonic acid (HOTf) as activating agents, ${ }^{20}$ delivered a single diastereomerically pure product 9 , which was shown to have the desired $\beta$-configuration. ${ }^{21}$ Reductive removal of the acetyl groups with diisobutylaluminium hydride (DIBAL-H) ${ }^{22}$ produced 1,3-diol 10, which was converted into the respective dimethylether 11 upon treatment with an excess (10 equiv.) of Meerwein salt and proton sponge [1,8-bis(dimethylamino)naphthalene]. ${ }^{23}$ Less electrophilic methylating reagents (MeOTf, MeI ) in combination with appropriate bases failed to react or led to substrate decomposition. Selective desilylation of the alkyne was achieved with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU). ${ }^{24}$

With alkyne 12 in hand, various approaches to potential cross-coupling substrates were pursued. It was found that the


Scheme 3 Assembly of the $\mathrm{C}_{28}-\mathrm{C}_{40}$ fragment via glycosylation of enantiomerially pure ( $\geq 98 \%$ ee) alcohol 8 with glycosyl donor 5 .

Pd-catalyzed hydrostannylation with $\mathrm{Bu}_{3} \mathrm{SnH}^{25}$ can be successfully performed with alkyne 12 delivering stannane 13 in $44 \%$ yield (Scheme 4). Iodide 14 was obtained from stannane 13 upon treatment with iodine in dichloromethane ( $85 \%$ yield). ${ }^{26}$ The alkyne hept-3-en-1-yne-5-ol ${ }^{27}$ seemed to be the most suitable precursor for iodide 15 and stannane 16. The compound was available from bis-1,4-(trimethylsilyl)buta-1,3-diyne in four steps and an overall yield of $53 \%$ (see the ESI $\dagger$ for further details). Stannylation of hept-3-en-1-yne-5-ol with $\mathrm{Bu}_{3} \mathrm{SnH}$ was readily achieved employing the Cu-based protocol of Betzer et al. ${ }^{28}$ to deliver stannane 16 in $\mathbf{7 9 \%}$ yield. As for $\mathbf{1 4}$, iodide 15 was generated by iodo-de-stannylation employing iodine in dichloromethane ( $79 \%$ yield).

While attempted Stille cross-coupling reactions ${ }^{29}$ of stannane 13 and iodide 15 failed, the desired C-C bond formation proceeded smoothly, when performed with the carbohydrate building block as the electrophile. Iodide 14 and stannane 16 underwent a clean cross-coupling employing $\operatorname{Pd}(\mathrm{MeCN})_{2} \mathrm{Cl}_{2}$


Scheme 4 Stille cross-coupling of building blocks 13 and 15 as key step for the assembly of the title compound.
( $10 \mathrm{~mol} \%$ ) as the catalyst. ${ }^{30}$ Alcohol 17 was obtained in $87 \%$ yield and was immediately further oxidized to the desired ketone by treatment with an excess ( 30 equiv.) of $\mathrm{MnO}_{2}$. Despite a pronounced long wavelength absorption $\left(\lambda_{\max }=308 \mathrm{~nm}\right.$, $\varepsilon=28035 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ in MeCN), trienone 1 appears to be more stable than alcohol $17\left(\lambda_{\text {max }}=271 \mathrm{~nm}, \varepsilon=39350 \mathrm{M}^{-1} \mathrm{~cm}^{-1}\right.$ in MeCN; shoulder at $\lambda_{\text {max }}=282 \mathrm{~nm}, \varepsilon=31180 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$ ) and could be stored for one week at $-25{ }^{\circ} \mathrm{C}$ in the dark.

In summary, the enantiomerically pure western fragment 1 of ( - )-pulvomycin was synthesized in 15 linear steps. The fragment comprises the carbohydrate part (labilose, $\mathrm{C}_{35}-\mathrm{C}_{40}$ ) of the natural product and one of its three triene components $\left(\mathrm{C}_{24}-\mathrm{C}_{34}\right)$. Should an aldol-type reaction of fragment 1 with a suitable Eastern fragment not be successful, stannane 13 and iodide $\mathbf{1 4}$ offer suitable options to connect the protected glycoside fragment to the rest of the molecule.

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