

NMR methodology for complex mixture 'separation'[†]

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Mixture 'separation' by NMR is demonstrated through the development of a pseudo 4D NMR experiment, 3D IPAP INEPT-INADEQUATE-HSQC, designed for the structural elucidation of ^{13}C tagged compounds.

The structural elucidation of compounds contained within complex mixtures is a challenging task. Despite the advances in chromatography, some mixtures cannot be separated. Humic substances (HS), produced by the biodegradation of plant and animal residues,¹ are the best known example of an 'inseparable' mixture consisting of thousands of organic compounds. Ubiquitous in nature, they make up a considerable proportion of the Earth's carbon pool and are key players in many biogeochemical processes.

In order to comprehend the functional roles of HS on a molecular level, their structural composition needs to be deciphered. So far HS have been characterized only on the level of compound classes and presence of individual functional groups.^{2–4} NMR spectroscopy and mass spectrometry are widely regarded as the two most promising analytical techniques for revealing the structure of individual HS molecules.^{5–7} Nevertheless, HS pose a considerable analytical challenge to both techniques, *e.g.*, the signal overlap in NMR spectra of complex mixtures prevents separation of resonances belonging to individual molecules and hence their identification using standard NMR techniques. Increasing the dimensionality of NMR experiments, selective excitation, or DOSY spectroscopy^{8,9} are the three most common approaches attempted to circumvent these problems.

Unfortunately, these quickly fail when the complexity of mixtures increases. Inevitably, for chromatographically inseparable mixtures the "separation" must be done spectroscopically. One way of achieving this is by tagging the molecules with isotopically labelled

nuclei. Once in place, the polarization transfer pathways are directed through these tags, reducing the complexity of spectra significantly. As illustrated here, this approach has a potential to elucidate molecular fragments of compounds contained in complex mixtures.

The bulk of HS is composed of oxygen rich $\text{H}_x\text{C}_y\text{O}_z$ compounds with the M_w range of $\sim 200\text{--}1000\text{ g mol}^{-1}$.¹⁰ Particularly prevalent functionalities decorating aromatic and aliphatic skeletons of HS are the hydroxyl and carboxyl groups. These groups are crucial to the interaction of HS with species such as heavy metals as well as the self-association of HS molecules. The methodology presented here aims at characterizing the aromatic moieties of HS carrying OH and COOH groups and relies on introducing ^{13}C -enriched $-\text{O}^{13}\text{CH}_3$ and $-\text{COO}^{13}\text{CH}_3$ groups into HS compounds. HS have been methylated in the past and the inspection of $-\text{O}^{13}\text{CH}_3$ resonances yielded some rudimentary information about the nature of their COOH and OH groups.¹¹ The novelty of our approach is that it uses labels to spy on their neighbourhood, obtaining the ^1H and ^{13}C chemical shifts and ^1H – ^1H and ^1H – ^{13}C coupling constants of the nuclei in their vicinity.

To achieve this aim we have designed a novel 3D NMR experiments, referred to here as 3D IPAP INEPT-INADEQUATE-HSQC (Fig. 1).

This NMR experiment can be viewed as a 3D extension of a 2D INEPT-INADEQUATE.¹² It is also related to 3D HCCH experiments.^{13,14} 3D IPAP INEPT-INADEQUATE-HSQC exploits polarization transfer pathways shown in Fig. 1, and correlates the double-quantum (DQ) coherencies of long-range coupled carbons (F_1) with corresponding single-quantum ^{13}C chemical shifts (F_2) and the ^1H chemical shifts (F_3). The polarizations transfer is tuned for $^{2,3}\text{J}_{\text{CC}}$ (Fig. S1, ESI[†]) and correlates the chemical shifts of nuclei in $^{13}\text{CH}_3\cdots^{13}\text{CH}_y$ ($y = 0, 1$) fragments. It starts and ends concurrently on methoxy and aromatic protons located next to the methoxy groups. For $\text{CH}_3\cdots\text{CH}$ fragments it provides chemical shifts of all four nuclei and can therefore be regarded as a pseudo 4D experiment; all three chemical shifts are obtained for the $\text{CH}_3\cdots\text{C}_q$ moieties. Since the acquired NMR signal is filtered *via* isotopically enriched $^{13}\text{CH}_3$ groups, the resulting spectra are significantly simplified.

The limited ^1H and ^{13}C chemical shifts range of methoxy groups (Fig. S2, ESI[†]) necessitates the use of high digital resolution in the

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[†] Electronic supplementary information (ESI) available: Materials and methods, details of the optimization of NMR parameters, values of coupling constants, S/N ratios, examples of processing of IPAP spectra and a suppression of signals from aliphatic OMe groups. See DOI: [10.1039/c3cc48907h](https://doi.org/10.1039/c3cc48907h)



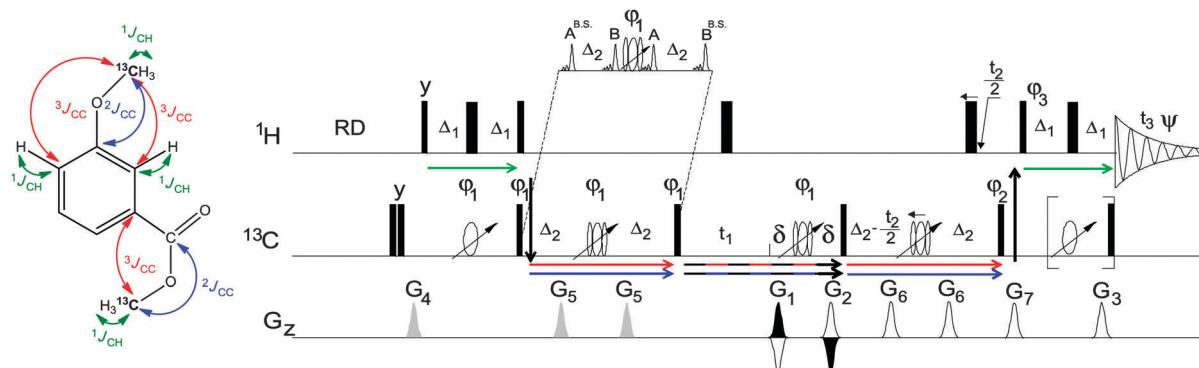


Fig. 1 The pulse sequence of the 3D IPAP INEPT-INADEQUATE-HSQC and its polarisation transfer pathways utilising various coupling constants. The details of the pulse sequence are given in the ESI.[†]

F_2 and F_3 dimensions of the 3D spectra. In order to achieve this in a realistic time, we have (i) optimized ^1H and ^{13}C carrier frequencies and spectral widths in all dimensions, (ii) frequency shifted DQ coherences during t_1 , (iii) aliased signals in the F_2 dimension, (iv) used non-uniform sampling and (v) acquired pure in phase and antiphase (IPAP) 15 ^{13}C -coupled multiplets during t_3 , as detailed in the ESI,[†] Fig. S3–S5.

The proposed method is illustrated on a model mixture of nine benzoic acid derivatives, **1a–9a**, which were fully methylated using $^{13}\text{CH}_3\text{I}$, yielding compounds **1–9** (Scheme 1, average concentration of 1.4 mM).

2D F_2F_3 (HSQC) (Fig. 2a and b) and 2D F_1F_3 (DQ) (Fig. 2c and d) planes of the 3D IPAP INEPT-INADEQUATE-HSQC spectrum of the model mixture illustrate the resolving power of this experiment. The spectrum was analysed as outlined in Fig. 2e–g. Shown here are the 2D DQ (F_1F_3) planes extracted at the ^{13}C shifts of the aromatic and Me carbons (F_2) indicated by red arrows in Fig. 2a and b. The obtained ^{13}C and ^1H chemical shifts allowed unambiguous identification of the fragment highlighted in the inset. The fragment was assigned to the correct molecule by considering the electronic effects of OMe and COOMe groups on the ^{13}C and ^1H chemical shifts of benzene and by analysing the proton–proton and long-range proton–carbon coupling constants determined in F_3 . The inspection of the complete 3D spectrum lead to the identification of the fragments highlighted in Scheme 1 and their assignment to individual molecules.

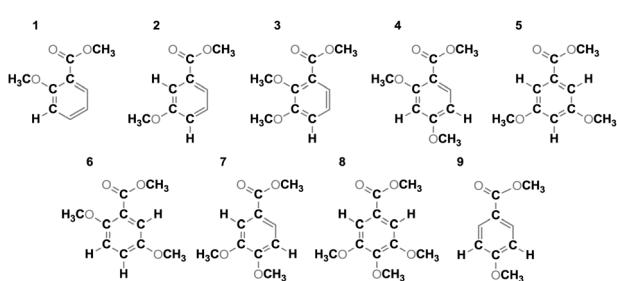
When applied to HS, the signals from methylated aliphatic hydroxyl and carboxyl groups can potentially interfere with the

interpretation of the 3D IPAP INEPT-INADEQUATE-HSQC spectra. These can be eliminated by band-selective inversion of spins resonating within the 65–95 and 15–45 ppm regions of the ^{13}C spectra applied during the initial carbon spin-echo, $2A_2$, (see the inset of Fig. 1). Spins resonating in these regions effectively receive a 360° pulse, thus refocusing their $^{2,3}J_{\text{CC}}$ couplings with the $^{13}\text{CH}_3\text{O}$ carbons and eliminating their signals (see Fig. S6, ESI[†]).

How sensitive is the 3D IPAP INEPT-INADEQUATE-HSQC experiment? Because the DQ coherences are created between the fully labelled $^{13}\text{CH}_3$ carbons and natural abundance ^{13}C spins, the theoretical sensitivity of this experiment is comparable to that of a refocused gradient-selected ^1H , ^{13}C HMBC optimized for $^3J_{\text{CH}}$ couplings. We have determined the S/N ratios in 1D F_3 traces extracted from F_1F_3 planes of the 3D spectrum obtained by the addition of the 3D IP and AP INEPT-INADEQUATE-HSQC spectra. The values were normalized (Fig. S7, ESI[†]) to the average concentration of the methylated compounds (1.4 mM). Due to their singlet character, the S/N is higher for $-\text{O}^{13}\text{CH}_3$ (60–1900:1) than for the aromatic resonances (57–600:1). Somewhat lower values were obtained for $-\text{COO}^{13}\text{CH}_3$ signals (40–240:1). These values clearly reflect the sizes of $^{2,3}J_{\text{CC}}$ couplings (Fig. S1, ESI[†]) involved in the polarisation transfer and can be increased for small coupling constants by optimizing the Δ_2 intervals for smaller J_{CC} couplings (6 Hz at present). This will have to be balanced against faster relaxation in larger molecules. We note that the relaxation effects can be reduced by using $^{13}\text{CD}_2\text{H}$ in place of $^{13}\text{CH}_3$ groups. This modification opens a route toward a design of more efficient 3D INADEQUATE-based experiments thus compensating partially for the loss of protons in $^3\text{CD}_2\text{H}$ groups.

Considering the lowest S/N observed (40:1), a 10-fold reduction of the concentration would still yield spectra of adequate quality. Thus assuming a $140\ \mu\text{M}$ concentration of a $500\ \text{g mol}^{-1} M_w$ compound, $38.5\ \mu\text{g}$ dissolved in $550\ \mu\text{l}$ of CDCl_3 are required per compound. A 19.3 mg strong mixture of compounds containing 500 unique substitution patterns by $-\text{OH}$ and $-\text{COOH}$ groups on aromatic rings will therefore provide 3D spectra with sufficient S/N ratios to be analyzed. As the methylation procedure includes extraction into the organic phase, and thus selects a subset of HS compounds, simplification of these complex mixtures is inherent to this process. This could be further improved by employing partial fractionation of HS prior to methylation.

Obtaining ^1H and ^{13}C chemical shifts of the nuclei in the vicinity OH and COOH by itself does not lead to a structure, and



Scheme 1 Nine methylated compounds. The molecular fragments identified by the analysis of the 3D IPAP INEPT-INADEQUATE-HSQC spectrum based on the chemical shift and coupling constants are highlighted in bold.



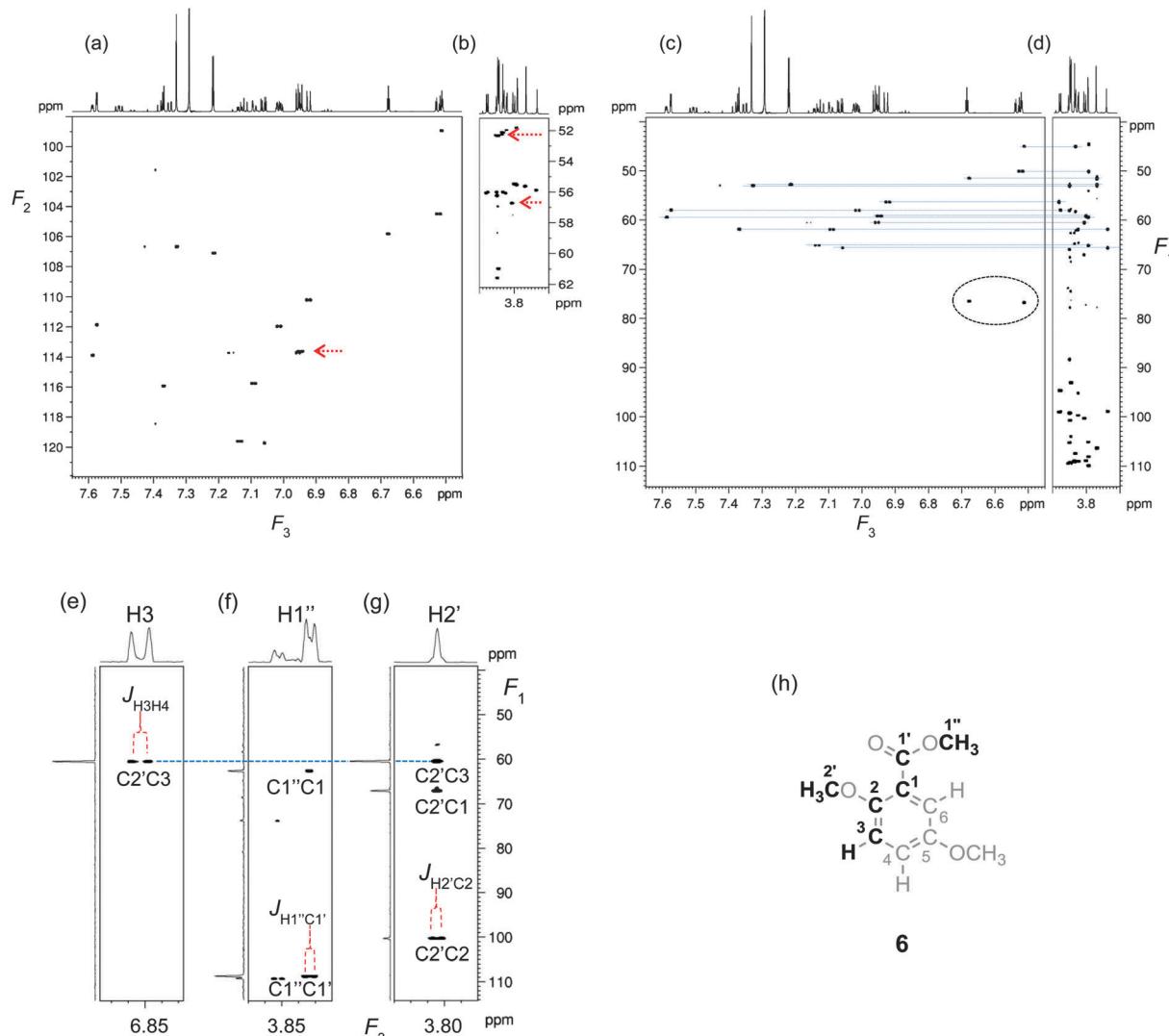


Fig. 2 2D projections of the 3D IPAP INEPT-INADEQUATE-HSQC spectrum obtained by the addition of the inphase and anti-phase 3D spectra. (a, b) F_2F_3 HSQC and (c, d) F_1F_3 DQ projections of the aromatic (a, c) and methyl (b, d) regions of the spectrum. The circled signals belong to the aliased DQ coherences of OMe carbons in compounds **4** and **5**. See Fig. S4 (ESI†) for the explanation of their origin. Blue lines connect DQ coherences of $\text{CH}_3\cdots\text{CH}$ fragments detected simultaneously on the aromatic and methyl protons. The red arrow indicates the F_2 coordinates of planes shown in (e–g). 2D DQ (F_1F_3) planes extracted at the CH and OMe ^{13}C chemical shifts indicated in (a and b) showing (e) aromatic, (f) ester and (g) methoxy proton regions of the spectrum. The blue line indicates a shared DQ frequency. Observed splittings are annotated. (h) The identified molecular fragment.

such data needs to be interpreted by considering the precursors of HS molecules and their chemical shifts. However, it is clear that this methodology supersedes information content of simple ^1H , ^{13}C HSQC spectra of HS.¹⁶ The key is the increased number of nuclei that are identified as belonging to the same molecule. Additional valuable information can be obtained by hydrolysing the $-\text{COO}^{13}\text{CH}_3$ esters, while keeping the $-\text{O}^{13}\text{CH}_3$ groups untouched. The resulting site-specific changes of chemical shifts of aromatic spins can be quantified through recording the 3D IPAP INEPT-INADEQUATE-HSQC of a partially hydrolysed mixture (data not shown). Such information is very valuable for the identification of fragments carrying both OH and COOH groups.

In conclusion, we conjecture that tagging by ^{13}C labelled methyl groups in combination with 3D NMR spectroscopy is a

very promising approach to the analysis of complex mixtures such as HS. To the best of our knowledge, this work is the first example of a sophisticated NMR experiment designed to explore the chemical environment around the tags, rather than just the signals from the tags themselves. This approach is not limited to methylation, other tags, containing NMR active, fully abundant nuclei such as ^{15}N or ^{19}F can also be explored. Finally, many other mixtures can benefit from this approach, including those found in food, beverages, natural products, or biological samples, to name a few.

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