

Palladium-catalyzed cyclization of bromoenynamides to tricyclic azacycles: synthesis of trikentrin-like frameworks†

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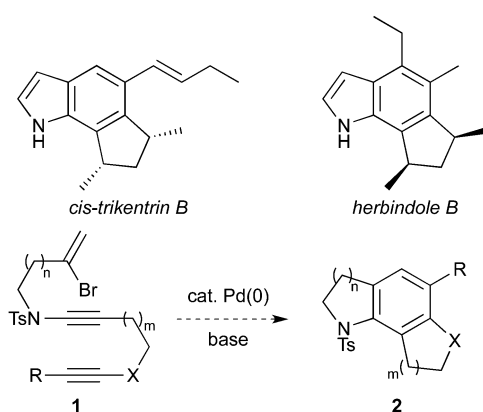
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Palladium-catalyzed cascade cyclization of bromoenynamides equipped with an additional alkyne or ynamide substituent affords azatricyclic products. Using 5- to 7-membered ring tethers, this chemistry offers a regiospecific route to highly-functionalized azacycles. Elaboration to the trikentrin B skeleton is achieved from the arylsilane cyclization products.

The trikentrin and herbindole families of natural products, isolated from the marine sponges *Trikentrion flabelliforme*^{1a} and *Axinella* sp.,^{1b} display a range of bioactivities including antimicrobial, anti-feedant and cytotoxic properties. The heavily-substituted tricyclic indole systems which feature in these compounds (e.g. *cis*-trikentrin B and herbindole B, Scheme 1) represent a particular synthetic challenge that has inspired a number of elegant solutions.¹

In recent work, we have developed a number of routes to azabicycles from ynamides, including *via* ynamide carbopalladation.² We have also reported a strategy to access the tricyclic 7,6,5-CDE ring cores of rubrifloridilactones A and B, which contain penta- and tetrasubstituted arenes respectively, through palladium-catalyzed cascade cyclization of bromoenediynes.³ We noted that the combination of these two methodologies could provide access to the tricyclic indole core of the trikentrins, *via* cyclization of a bromoenynamide equipped with a remote alkyne (**1** → **2**, Scheme 1). The construction of fused ring arenes in this manner⁴ has rarely been employed in synthetic endeavours,³ and in the context of ynamides offers a useful alternative to the elegant cyclotrimerization methodology pioneered by Witulski,⁵ which has recently been applied to the herbindole system.^{1c} This sequenced carbopalladation strategy offers advantages over intramolecular cyclotrimerization, where long tethers restrict the formation of larger rings, presumably due to competing intermolecular reactions.⁶ Here we describe the development of this novel ynamide chemistry,⁷ and its application to a number of azatricyclic systems, including aza- and benzazepine trikentrin analogues. Elaboration to the bis-desmethyl-trikentrin B framework is also described.

We first set about the synthesis of a series of bromoenynamide alkynes suitable for cyclization, through the preparation of appropriate sulfonamide and bromoalkyne precursors (Scheme 2). The bromoalkenyl sulfonamides **3a** and **3b** were prepared from alkynes **4a** and **4b** *via* bromoboration/protodeborylation (with simultaneous Boc deprotection), whilst bromoalkynes **5a** and **5b** were synthesized from 1,6-heptadiyne by monosilylation then bromination. Both trimethylsilyl and benzyldimethylsilyl groups were installed, which we anticipated would enable various strategies for the attachment of trikentrin-like sidechains following cyclization. These building blocks were coupled using Hsung's copper-catalyzed methodology for ynamide formation,⁸ which provided ynamides **1a–c** in moderate to good yield, albeit accompanied by a degree of desilylation in the case of TMS-substituted diyne **5a**. To further extend the cyclization methodology, we targeted a substrate featuring two ynamides, which would lead to an aza-trikentrin (pyrroloindoline) framework. The bis-ynamide **1d** was readily prepared in four steps from **4a** by bromination/carbamate deprotection (to afford the sulfonamide **6**), followed by



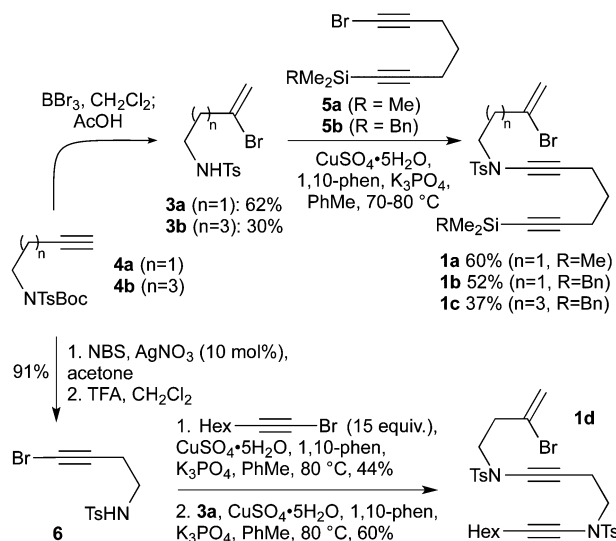
Scheme 1 *cis*-Trikentrin B, herbindole B, and the general bromoenynamide cyclization strategy.

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Scheme 2 Preparation of ynamide and bis-ynamide cyclization substrates.

sequential Hsung couplings – firstly with 1-bromo-oct-1-yne (used in excess to minimize intermolecular homocoupling of **6**), then with sulfonamide **3a**.

With a selection of substrates in hand, the cascade cyclizations were investigated (Table 1). Using our previously reported conditions (10 mol% $\text{Pd}(\text{PPh}_3)_4$, Et_3N , 0.017 M in MeCN , 80°C),³ we were pleased to obtain the 5,6,5-tricyclic trikentrin frameworks **2a** and **2b** in excellent yields (90%, entries 1 and 4). The catalyst loading could be lowered to 5 mol% with a slight reduction in yield (entries 2 and 5); however, by increasing the concentration (to 0.16 M), catalytic efficiency was restored, with **2a** isolated in 88% yield (entry 3).

Cyclization to the challenging 7,6,5-tricyclic analogue of the trikentrin framework was next attempted. At higher catalyst loading and dilution, the desired tricycle **2c** was obtained as a component of a complex mixture (entry 6). However, by performing this reaction at higher concentration, **2c** was formed as the sole product in excellent yield (entry 7), a result that highlights the advantages of the sequenced carbopalladation strategy. Finally, diynamide **1d** was subjected to the range of reaction conditions (entries 8–10). To our delight, tricycle **2d**, which represents the first example of such a pyrroloindoline framework, was isolated in high yield when reacted at the higher concentration (70%, entry 10).

With efficient access to azatricycles established, we aimed to demonstrate the utility of the methodology by preparing a natural product analogue – bis-desmethyl-trikentrin B **13** (see Scheme 3) – from the 5,6,5-indolines **2a** or **2b**. This required installation of the requisite butenyl sidechain, and conversion of the protected indoline to the free indole. For the former of these tasks, we recognised the synthetic value of the silane present in **2a/b**, which enables a variety of sidechain attachment strategies. We first addressed Hiyama cross-coupling of **2b**, which offers a direct route to the butenyl substituent and is an attractive alternative to other coupling methods (e.g. Stille, Suzuki) due to the low toxicity of silicon and its stability to multistep synthesis.⁹ To our knowledge, no Hiyama couplings between *aryl*-benzyl dimethylsilanes and *alkenyl* halides have been reported, with only the reverse process being described (i.e. the coupling of *alkenyl*/benzyl dimethylsilanes with *aryl* halides).¹⁰

Table 1 Bromoenynamide–alkyne cascade cyclizations^a

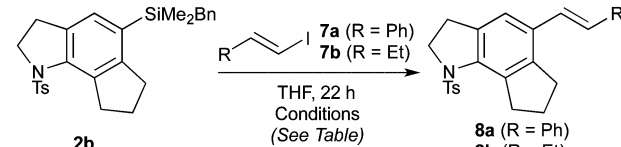
Entry	Substrate	Catalyst loading (mol%)	Product	Yield ^b (%)
1	1a	10		90
2		5		78
3		5 ^c		88
4	1b	10		90
5		5		73
6	1c	10		— ^d
7		10 ^c		78
8	1d	10		49
9		5		52
10		5 ^c		70

^a Reaction concentration 0.017 M unless indicated otherwise. ^b Isolated yield. ^c Reaction concentration 0.16 M. ^d Complex mixture.

Standard conditions for the coupling of alkenyl benzyl dimethylsilanes (TBAF, $\text{Pd}_2\text{dba}_3 \cdot \text{CHCl}_3$ or $\text{Pd}(\text{dba})_2$),¹⁰ using either β -styrenyl iodide **7a** or butenyl iodide **7b** as the halide partner, afforded no cross-coupling product (Table 2, entries 1 and 2). As benzylsilanes are ‘safety-catch’ silanols, and indeed are hydrolysed to the latter on treatment with TBAF, alternative conditions for the coupling of alkenylsilanols¹¹ were also investigated, without success (entry 3). In all of these trials, mixtures of silanol, disiloxane, and desilylated arene were recovered,¹² suggesting that the aryl silanol revealed on unmasking of the benzylsilane was resistant to transmetalation. The addition of Ag_2O has been reported by Hiyama to accelerate transmetalation,¹³ and we were delighted to find that the coupling of styrenyl iodide **7a** under these conditions smoothly afforded the styrenyl trikentrin framework **8a** (68%). Disappointingly, only desilylated arene was returned on attempted coupling with butenyl iodide **7b**, which for this study presented an insurmountable limitation.

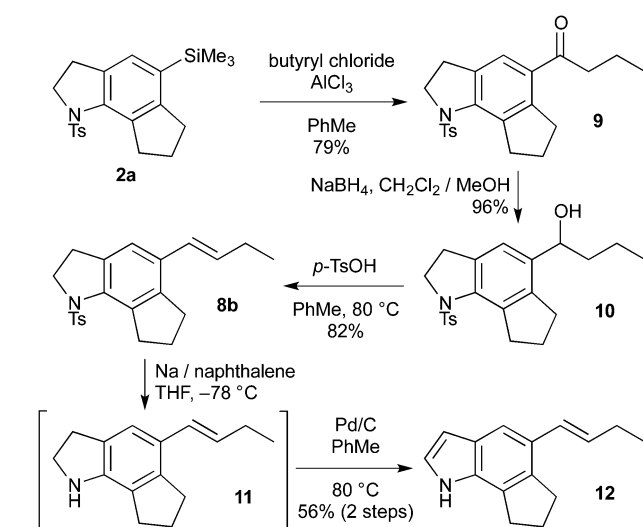
A more classical route to install the butenyl sidechain was thus developed (Scheme 3).¹⁴ Aryltrimethylsilane **2a** was subjected to a Friedel–Crafts acylation, which proceeded with exclusive *ipso*-selectivity to give ketone **9** (79%). This ketone then underwent a high-yielding reduction–dehydration sequence to deliver the targeted butenyl sidechain (**8b**). Completion of the synthesis now required indoline detosylation and oxidation to reveal the indole moiety. However, all attempts to oxidise **8b** to the corresponding sulfonyl indole were unsuccessful, leading mainly to degradation.¹⁵ Inverting this sequence of events resolved this issue; although Mg/MeOH /sonication (which is usually effective for such detosylations)^{2a,16} effected partial deprotection



Table 2 Hiyama cross-coupling of arylsilane **2b**


Entry	Alkenyl iodide	[Pd] cat. (mol%)	TBAF (equiv.)	Temp (°C)	Yield ^a (%)
1	7a	Pd ₂ dba ₃ ·CHCl ₃ (2.5) or Pd(dba) ₃ (5)	2.2	20 → 50	— ^b
2	7b	Pd ₂ dba ₃ ·CHCl ₃ (2.5)	2.2	20 → 50	— ^b
3	7a	(AllylPdCl) ₂ (2.5)	2.2	20 → 50	— ^b
4	7a	Pd(PPh ₃) ₄ (5), Ag ₂ O ^c	1.1	20	68
5	7b	Pd(PPh ₃) ₄ (5), Ag ₂ O ^c	1.1	20	— ^d

^a Isolated yield. ^b A mixture of silanol, disiloxane, and desilylated arene was recovered. ^c 1.1 equiv. Ag₂O. ^d Desilylated **2b** was isolated (67%).



Scheme 3 Synthesis of bis-desmethyl-trikentrin B.

(<25%), treatment of **8b** with sodium naphthalenide gave the deprotected indoline **11** with high efficiency. Somewhat surprisingly, **11** underwent rapid aerobic decomposition, presumably due to the indoline-enhanced reactivity of the electron-rich styrene,¹⁷ and isolation of the pure indoline proved difficult. However, we were pleased to find that direct dehydrogenation of the crude indoline using Pd/C in degassed toluene completed the synthesis, giving bis-desmethyl-trikentrin **12** in good yield over the two steps.

In conclusion, we have developed a facile method for the preparation of azatricycles from bromoalkenyl ynamides. The reaction enables formation of five- to seven-membered rings, and offers an attractive alternative to cyclotrimerization strategies. The utility of this chemistry is demonstrated by installation of the trikentrin B alkenyl sidechain in a further four steps using Friedel-Crafts *ipso*-substitution of the arylsilane cyclization products. As an alternative, we report the first example of an alkenyl iodide/arylbenzylsilane Hiyama cross-coupling, which affords a styrenyl-trikentrin analogue.

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- Oxidation using MnO₂, Mn(OAc)₃, DDQ, or AIBN/NBS led to complete degradation of material; Co(salen)/O₂ led to no reaction.
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