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Nanozyme-enhanced ferroptosis for cancer treatment

 Yue Ming,  ^{†a} Mingshu Huang,  ^{†a} Yisheng Huang,  ^a Danqing Liu, ^{bc}
 Min Sun,  ^{*bcd} Bo Jia  ^{*a} and Jianzhong Du  ^{*bcd}

Ferroptosis is a programmed, iron-dependent, oxidative cell death that was discovered recently. It is usually accompanied by iron accumulation and lipid peroxidation during the cell death process. Ferroptosis-inducing factors affect glutathione (GSH) peroxidase directly or indirectly, leading to a decrease in antioxidant capacity and accumulation of lipid reactive oxygen species (ROS). Ferroptosis has garnered much interest in the field of cancer treatment. However, the therapeutic efficacy through the ferroptosis pathway by directly increasing the levels of iron ions at cancer lesion is not ideal due to the inefficient enrichment of iron ions at the lesion site, the uncontrolled Fenton reaction and a single apoptotic pathway. Nanozymes are nanomaterials that can catalyse enzyme substrates into products following enzyme kinetics under physiological conditions. Nanozymes offer advantages such as enhanced stability, simplified preparation, and cost-effectiveness compared to natural enzymes. Notably, nanozymes can serve as self-activated cascade reagents, elevating the therapeutic efficacy of cancer through the ferroptosis pathway by effectively generating reactive ROS and depleting GSH. Furthermore, nanozymes can induce ferroptosis and synergize with other approaches such as photothermal therapy (PTT), photodynamic therapy (PDT), and immunotherapy. Presented in this review are the definition, structure, classification, and features of nanozymes, the fundamental mechanisms of ferroptosis in cancer cells, and the combined strategies employed to combat cancer by leveraging nanozymes to induce or enhance ferroptosis.

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1. Introduction

Cancer refers to a group of diseases characterized by the uncontrolled growth and spread of abnormal cells. The Global Cancer Observatory estimated nearly 1.93 billion new cancer cases and approximately 10 million cancer-related deaths worldwide in 2020.¹ The most common types of cancer are breast cancer, lung cancer, colorectal cancer, prostate cancer, and stomach cancer.² These statistics indicated the cruciality to deeply understand cancer and develop suitable diagnosis and treatment options. Traditional methods for cancer treatment

include surgery, radiation therapy, and chemotherapy, which have limitations such as high recurrence rates, poor postoperative quality of life, and lack of target specificity.¹ Given the current limitations in precision cancer treatment, there is an urgent need for new anticancer approaches.³

Cancer treatment based on ferroptosis can target cancer cells, including those that are resistant to drugs. The iron sensitivity of different types of cancer cells provides the basis for this therapy. By controlling the iron metabolism and lipid peroxidation,⁴ or inhibiting glutathione (GSH) synthesis or glutathione peroxidase 4 (GPX4) activity, this therapy can selectively trigger the iron dependency of cancer cells without adversely affecting healthy cells.⁵ Prospects for iron-based inducers are expanding, including erastin, GPX4 inhibitors, and iron chelators.⁶ Ferroptosis, distinct from traditional forms of cell death such as apoptosis, necrosis, and autophagy, is a regulated form of cell death driven by iron-dependent lipid peroxidation.⁷ Its essence lies in the depletion of GSH and the decrease in the activity of GPX4, leading to the inability of lipid peroxides (LPO) to be metabolized through the GPX4-catalysed GSH reductase reaction. Subsequently, the oxidation of lipid precursors by ferrous iron produces reactive oxygen species (ROS), thereby triggering ferroptosis. This can serve as a novel anticancer method.

^a Department of Oral Surgery, Stomatological Hospital, School of Stomatology, Southern Medical University, Guangzhou, China. E-mail: dentist-jia@163.com

^b Department of Gynaecology and Obstetrics, Shanghai Key Laboratory of Anesthesiology and Brain Functional Modulation, Clinical Research Center for Anesthesiology and Perioperative Medicine, Translational Research Institute of Brain and Brain-Like Intelligence, Shanghai Fourth People's Hospital, School of Medicine, Tongji University, Shanghai 200434, China. E-mail: 1911422@tongji.edu.cn, jzdu@tongji.edu.cn

^c Department of Orthopedics, Shanghai Tenth People's Hospital, School of Medicine, Tongji University, Shanghai 200072, China

^d Department of Polymeric Materials, School of Materials Science and Engineering, Tongji University, 4800 Caoan Road, Shanghai, 201804, China

[†] These authors contributed equally to this work.



Currently, cancer treatment methods based on iron-induced cell death are actively explored. Non-targeted strategies based on nanoparticle delivery to transport iron, peroxides, and other toxic substances have been proposed to kill cancer cells.⁸ The existence of various enzymes that regulate ferroptosis makes the development of targeted therapy possible.⁹ Nanozymes are nanomaterials with enzyme-like characteristics, possess good stability, low cost, and adjustable catalytic activity. Recently, their role in inducing ferroptosis in cancer treatment has attracted much attention.¹⁰ For example, nanozymes were designed to encapsulate or carry anticancer drugs or molecules that influence the ferroptosis pathway, and then delivered to specific cancer cells.¹¹ This further enhanced the selectivity of ferroptosis in cancer cells and minimized the damage to healthy cells.

The process of ferroptosis relies on lipid peroxidation, where the Fenton reaction is a key pathway. The Fe^{2+} ions react with H_2O_2 to generate hydroxyl radicals, leading to further lipid peroxidation and cell death.¹² Nanozymes can effectively catalyse the Fenton reaction or mimic the functions of natural enzymes such as peroxidase (POD), superoxide dismutase (SOD), catalase (CAT), and oxidase (OXD). They regulate the balance between the generation and elimination of ROS, thereby controlling the level of lipid peroxidation to influence the process of ferroptosis (Fig. 1).¹³

More and more preclinical studies have shown that many cancer cells have developed resistance to traditional therapies such as chemotherapy¹⁴ and radiation.¹⁵ However, these conventional mechanisms of drug resistance are often unable to counteract ferroptosis. The employment of nanozymes as a mean to trigger ferroptosis provides an alternative strategy to eradicate previously challenging-to-treat cancer cells. Three main pathways to reverse drug resistance include the canonical GPX4-regulated pathway, iron metabolism pathway, and lipid metabolism pathway. Nanozymes can be designed to achieve precise targeting of specific cells or tissues for precision medicine in cancer treatment. Additionally, nanozymes can be combined

with other anticancer drugs or molecules that affect the pathways of ferroptosis, serving as drug delivery systems for cancer treatment. Compared to natural enzymes,¹⁶ nanozymes have lower production costs, and their excellent stability and adjustability contribute to their potential benefits and feasibility in future cancer treatments. Thus, nanozyme-triggered ferroptosis to induce cancer cell death shows great potential as an innovative anticancer method.

2. Design and synthesis of nanozymes

2.1 Definition of nanozymes

Nanozymes first appeared in the literature in 2004, gained development in 2007 with Yan's discovery of peroxidase-like activity in magnetic iron oxide (Fe_3O_4) nanoparticles.¹⁷ Nanozymes refer in a broad sense to nanoparticles possessing catalytic functions similar to natural enzymes.¹⁸ However, enzymes and nanozymes exhibit significant mechanistic differences. For example, peroxidases confine activated H_2O_2 to the active center, while iron oxide is likely to exert more ROS-related toxicity (Fig. 2).¹⁹ Enzymatic reactions are highly substrate-selective, whereas nanozymes can exhibit broader substrate specificity than their natural counterparts.²⁰

Nanozymes are catalytic nanomaterials, but their functionality extends beyond catalysis. In addition to their catalytic activity, nanozymes can possess other properties and functionalities that make them versatile and valuable in various applications, including target recognition,²¹ specific binding,²² controlled release and so on.²³

2.2 Classification of nanozymes

Existing reports on nanozymes are mainly focused on simulating the activity of oxidation–reduction enzymes,²⁴ including SOD activity that generates H_2O_2 and oxygen (O_2) from a superoxide anion (O_2^-), CAT activity that decomposes H_2O_2 to produce H_2O and O_2 , POD activity that catalyses the redox of H_2O_2 substrates, the oxidoreductase activity that directly oxidizes substrates in the presence of O_2 , some reductase activities, *etc.*²⁵ Among them, SOD, CAT, and POD, as natural antioxidants in the body, play a vital role in clearing cellular oxygen-free radicals and maintaining the normal oxidative-reduction level of the organism. Additionally, anaerobic glycolysis, also known as the “Warburg effect”,²⁶ is one of the inherent markers of cancer metabolism, characterized by high levels of glucose intake and an increased rate of glucose conversion to lactate. Tumour tissues exhibit an acidic micro-environment and are highly sensitive to fluctuations in glucose content. Inspired by this characteristic, glucose oxidase (GOD) has been employed to serve as the nutritional source of cancer cells in tumour starvation treatments.²⁷ In this section, we summarize the mechanisms and kinetics of typical nanozymes with unique catalytic activities. We analyse the catalytic mechanisms of different types of nanozymes (Table 1).

POD-like activity. Fe_3O_4 was the first nanozyme which catalysed the reaction of peroxides, converting H_2O_2 to O_2 and



Fig. 1 An overview of synergistic mechanisms of nanozymes and ferroptosis. The diagram demonstrates that nanozymes mimic the catalytic activity of diverse natural enzymes (including structural modeling of CAT/POD/SOD-like enzymes), thus stimulating the Fenton reaction. This stimulation leads to the production of reactive oxygen species, GSH and GPX4 depletion, and lipid peroxidation, ultimately inducing ferroptosis to exert anticancer effects.



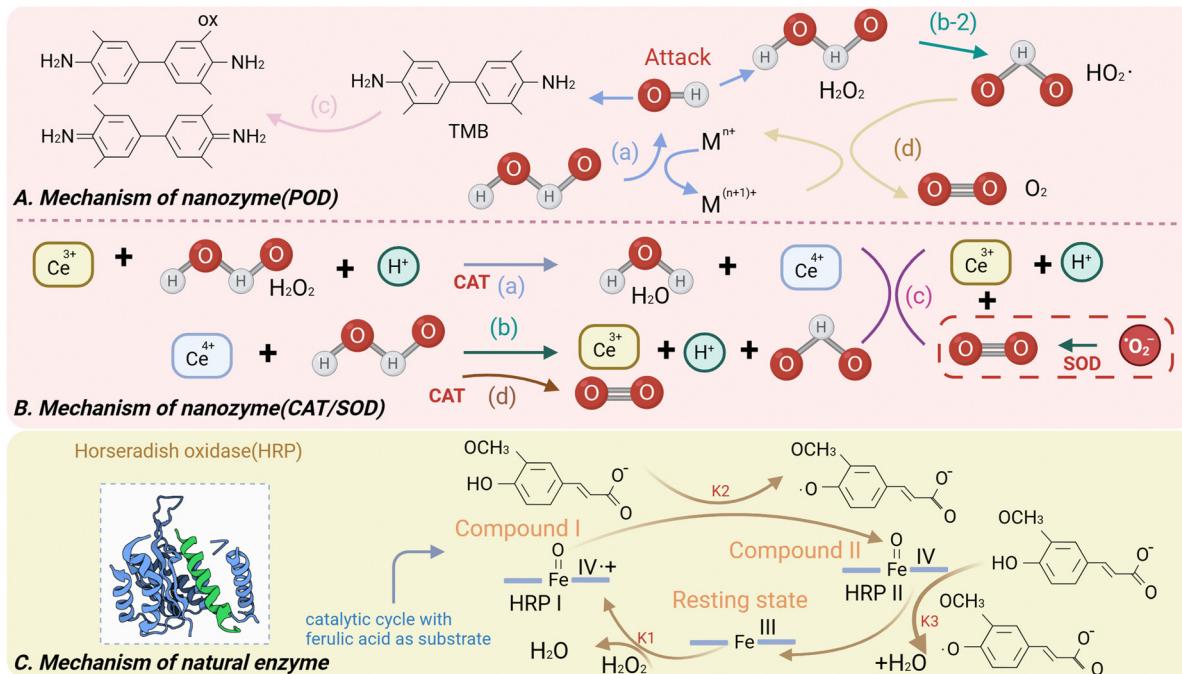


Fig. 2 The catalytic mechanism of (A) POD activity by nanozymes with Fe_3O_4 as the model; (B) CAT/SOD activity by nanozymes with cerium oxide (CeO_2) as the model; and (C) Natural enzyme with horseradish peroxidase (HRP) as the model.

Table 1 Classification of nanozymes

Nanozymes	Catalytic activities ^a	Ref.
Iron oxide	POD	41
Gold	CAT and POD	42
Platinum	POD, OXD, and SOD	42
CeO_2	OXD and SOD	43
Copper	OXD	44
Manganese	OXD, CAT, and POD	45
Carbon	POD	46

^a SOD: superoxide dismutase; CAT: catalase; OXD: oxidase; POD: peroxidase.

hydroxyl radicals.²⁸ The experimental data suggest that the catalysis mechanism of nano Fe_3O_4 may follow the “ping-pong reaction mechanism”. As shown in Fig. 2A, Fe_3O_4 can combine with the first substrate H_2O_2 to produce the intermediate $\cdot\text{OH}$. Then, the generated $\cdot\text{OH}$ captures an H^+ from a hydrogen donor such as 3,3',5,5'-tetramethylbenzidine. Subsequently, through combined electron spin resonance measurements and free radical inhibition experiments,²⁹ Tang *et al.* proposed a possible catalytic mechanism for peroxidase analogues based on Fe_3O_4 nanozymes, and monitored the production of the intermediate $\cdot\text{OH}$ in the catalytic reaction using electron spin resonance.³⁰ The study showed that Fe_3O_4 NPs can also produce the intermediate $\cdot\text{OH}$, indicating that they have similar properties to peroxides.³¹

CAT-like and SOD-like activities. Cerium dioxide (CeO_2) NPs, due to their unique chemical and physical properties, are among the most widely studied nanomaterials. They exhibit enzyme-like activity, especially for CAT and SOD.^{32,33} CeO_2 NPs are primarily associated with the oxidation state of their surface, which simultaneously includes Ce^{3+} and Ce^{4+} sites, granting them

the SOD and CAT activities. Therefore, by manipulating the redox state of cerium, CeO_2 NPs can simulate the dual antioxidant enzymes of SOD and CAT (Fig. 2B). Engaging in redox cycling reactions,³⁴ they regenerate the initial oxidation state and regulate ROS levels. A lower $\text{Ce}^{3+}/\text{Ce}^{4+}$ ratio favours the simulation of CAT-like activity in CeO_2 NPs, reducing Ce^{4+} to Ce^{3+} while catalysing the decomposition of H_2O_2 into O_2 . After generating hydrogen peroxide, CeO_2 NPs can utilize their SOD-like activity to simulate CAT-like activity. Following adsorption onto the CeO_2 surface, H_2O_2 preferentially dissociates, releasing two protons and O_2 , and simultaneously reducing Ce^{4+} to Ce^{3+} . Then, another molecule of H_2O_2 is absorbed and combines with the newly released protons to form two molecules of water, during which Ce^{3+} is re-oxidized to Ce^{4+} .³⁵

GOD-like activities. Compared to natural GOD, nanozymes mimicking GOD offer higher catalytic stability, easier modifications, and lower production costs for tumour treatment. Currently, the use of GOD-mimicking nanozymes in cancer therapy primarily includes substances like AuNPs,³⁶ and manganese dioxide (MnO_2)³⁷ and carbon nitride (C_3N_4) nanomaterials. Other materials also include metal-organic frameworks (MOF), covalent-organic frameworks (COF),³⁸ mesoporous silica or hollow black TiO_2 nanoparticles.³⁹ Much like their natural GOD counterpart, nanozymes, such as AuNPs, catalyse glucose oxidation by dehydrogenating glucose and reducing O_2 through a two-electron mechanism, resulting in the production of gluconate and H_2O_2 .⁴⁰

2.3 Strategies for improving the catalytic activity and specificity of nanozymes

So far, few nanozymes have matched the effectiveness of their natural enzyme counterparts. Usually, an enzymatic reaction



follows a three-step process where a substrate attaches to the enzyme's active site, undergoes transformation into a product and then is released – making way for the next substrate. This mechanism holds true for nanozymes as well, but the reaction happens on the nanomaterial's surface. The substrate binds *via* adsorption on the nanozyme.⁴⁷ The key characteristics affecting the catalytic activity of nanozymes include: (1) molecular structure: nanozymes possess a unique and often complex molecular structure, which generally contains active sites or centers that facilitate catalytic activity;³⁵ (2) electron properties: the ways in which electrons are arranged and behave in nanozymes significantly contribute to their catalytic activities;⁴⁸ and (3) large surface area: the nanoscale size increased surface area allows for more active sites.⁴⁹ By optimizing each stage of the catalytic cycle, it is possible to narrow the performance gap between nanozymes and natural enzymes.⁵⁰

Size and shape optimization. By adjusting the size and shape of nanozymes, one can control the enzymatic activity, stability, and specificity.³⁵ For example, Fan *et al.* proposed that the GOD-like catalytic activity of AuNPs is size-dependent. By comparing the reaction rates of AuNPs with different particle sizes (13, 20, 30, and 50 nm) under the same conditions, it was found that the catalytic performance of AuNPs decreased with increasing particle size. In addition, shape modification (*e.g.*, nanoflowers) of the nanozymes can endow them with a high density of spikes, good optical properties, and larger surface-to-volume ratios, which can enhance the catalytic activity, adsorption, and loading capacity of the nanozymes.⁵¹

Surface modification. By modifying the surface of nanozymes with specific functional groups or targeting ligands, their catalytic activity and specificity for cancer cells can be increased. For example, the same surface charge of the modified materials can affect the catalytic activity of the nanozymes similar to POD. The positively charged PLL (poly-lysine) and PEI (polyethyleneimine)-modified Fe₃O₄ nano-enzymes demonstrated higher catalytic activity when anionic 2,2'-azinobis-(3-ethylbenzothiazoline-6-sulphonate) was used as the substrate.⁵²

Environment-responsive design. Nanozymes are able to modulate the tumour physiological environment, including pH, O₂ levels, and immunosuppressive tumour cells, reprogramming the immunosuppressive tumour microenvironment (TME).⁴⁵ Ideally, nanozymes should also induce immunogenic cell death (ICD) of tumour cells, releasing tumour-associated antigens, which can be trapped, processed, and presented by antigen-presenting cells (APCs), thereby increasing the response rate and therapeutic efficacy of the immune system and immunotherapeutic drugs.⁵³

Composition tweaking. CeO₂ nanozymes exhibit excellent CAT-like activity and O₂ production capacity, while AuNPs exhibit more GOD-like activity and detection ability.⁵⁴ Changing the material compositions of the nanozymes could help in modulating and increasing the catalytic activity.

Hybrid enzyme design. Designing hybrid nanozymes that combine the properties of different enzymes could offer enhanced catalytic activity or specificity.⁵⁵ Sun *et al.* prepared OAm-PEG modified Pt₄₈Pd₅₂-Fe₃O₄ nanocomposites with a dumbbell-like

morphology. The Pt₄₈Pd₅₂-Fe₃O₄/TMB kit was used for the detection of H₂O₂ in biological solutions with a detection limit as low as 2 μM, and was successfully used for the quantitative monitoring of extracellular H₂O₂ produced by neutrophils. Compared with single Pt, Pd and Fe₃O₄ NPs, the obtained Pt₄₈Pd₅₂-Fe₃O₄ hybrid nanomaterials have higher peroxidase-like activity compared to single Pt, Pd and Fe₃O₄ NPs.⁵⁶

By implementing these strategies, the catalytic activity and specificity of nanozymes can be tuned to achieve the desired therapeutic outcomes in cancer treatment.

3. Mechanisms of nanozyme-induced ferroptosis

Ferroptosis is a prevalent form of cell death that occurs as a result of lipid peroxidation, which is initiated by hydroxyl radicals generated in the iron-catalysed Fenton reaction. Lipid peroxidation involves the oxidative degradation of lipids within the cell membrane, with a particular impact on polyunsaturated fatty acids (PUFAs). The resulting LPO species critically disturb the stability of the cell membrane's structure and functionality, ultimately leading to cellular damage and demise.¹¹

In the cellular environment, an excessive quantity of iron is initially produced, and the entry of Fe³⁺ into the cell occurs *via* the transferrin receptor 1 (TFR1) located on the cell membrane. Fe²⁺ was reduced within the cell, creating an unstable iron pool. Peroxidation of PUFA-containing membrane lipids is facilitated by both an unstable iron pool, which promotes the Fenton reaction and subsequent lipid peroxidation, and enzymes that utilize iron as a cofactor, such as arachidonic acid lipoxygenase (ALOX), which triggers the formation of lipid hydroperoxides, serving as substrates for the Fenton reaction.⁵⁷

Accumulating evidence regarding the mechanism of LPO and GPX4 enzymology indicates that LPO is triggered by alkoxy radicals generated through the action of ferrous iron on lipid hydroperoxide derivatives (LOOH), which are unavoidable byproducts of aerobic metabolism. These LOOH molecules are continuously reduced to LOH by GPX4 in the presence of reduced GSH. However, when the activity of GPX4 is limited, such as in the case of GPX4 inhibition or GSH depletion, ferrous iron triggers LPO by catalysing the breakdown of LOOH, ultimately leading to iron-mediated cell death.⁵⁸ The significance of LPO in ferroptosis is further supported by biological studies. For instance, the presence of LPO products, such as hydroxynonenal and malondialdehyde, has been detected during ferroptosis. Although the precise quantitative evidence for LPO in ferroptosis is still limited, the observed link between these two phenomena is compelling. We emphasize here the concept of thresholds, which are surpassed when the essential constraints of LPO are fulfilled, the activity of GPX4 is insufficient, and ferroptosis develops. LPO has been identified as a reliable indicator of ferroptosis.⁵⁹

The regulation of ferroptosis is primarily controlled by the GSH/GPX4 pathway. GSH, a crucial intracellular antioxidant,



plays a vital role in impeding iron metabolism by effectively eliminating LPO through the catalytic activity of GPX4. GPX4 is selenium-containing protein abundantly present in mammalian cells, serving as an enzyme that neutralizes LPO and providing protection. It efficiently converts the substrate GSH to GSSG while reducing intracellular toxic LPO to non-toxic alcohols (L-OH).⁶¹ Additionally, GPX4 effectively converts free H₂O₂ to water, safeguarding the structural integrity and functionality of cell membranes against peroxides, interference and damage.

Commonly used inducers of ferroptosis, such as erastin or Ras selective lethal small molecule 3 (RSL3, Fig. 3), can inhibit the activity of GPX4, thereby diminishing the antioxidant capacity of cells and triggering lipid reactive oxygen species (ROS), ultimately leading to ferroptosis.⁶² Furthermore, the mevalonate (MVA) pathway influences GPX4 synthesis by regulating the maturation of selenocysteine tRNA, which in turn controls the onset of ferroptosis.

Cysteine is closely associated with the glutamate transporter (system X_c⁻, an essential antioxidant system widely distributed in the phospholipid bilayer), another important mechanism for inducing ferroptosis. System X_c⁻ is a heterodimer composed of two subunits, solute carrier family 7 member 11 (SLC7A11) and SLC3A2. It facilitates the exchange of cystine and glutamate in and out of the cell in a 1 : 1 ratio.⁶³ Cystine, an oxidized form of cysteine, is transported into the cell *via* system X_c⁻ and subsequently reduced to cysteine. Cysteine plays a crucial role in the synthesis of GSH, an intracellular antioxidant.⁵⁸ Erastin can inhibit cystine transport causing cells, which leads to the depletion of cysteine (the fundamental component of intracellular GSH) indirectly inducing ferroptosis.⁵⁹

Recent studies have also shown that immune cells in the tumour microenvironment determine the occurrence of ferroptosis in tumours. Arachidonic acid and interferon- γ (IFN- γ) released by CD8⁺ T cells have been identified as potential natural triggers of ferroptosis. Cytotoxic T cells (CTL, *i.e.*, CD8⁺ T cells) downregulate the expression of SLC7A11 and inhibit the uptake of cystine in cancer cells by secreting IFN- γ . Cystine participates in the synthesis of GSH, the main process which involves system X_c⁻ taking extracellular cystine into cells and exporting glutamate outside the cells.⁶⁴ After entering the



Fig. 3 Schematic illustration of a composite nanosheet designed for tumour targeting, with the aim of sensitizing tumour ferroptosis through the disruption of the GPX4/GSH and FSP1/CoQ10H2 pathways.⁶⁰ Reproduced with permission from ACS Publications, copyright 2022.

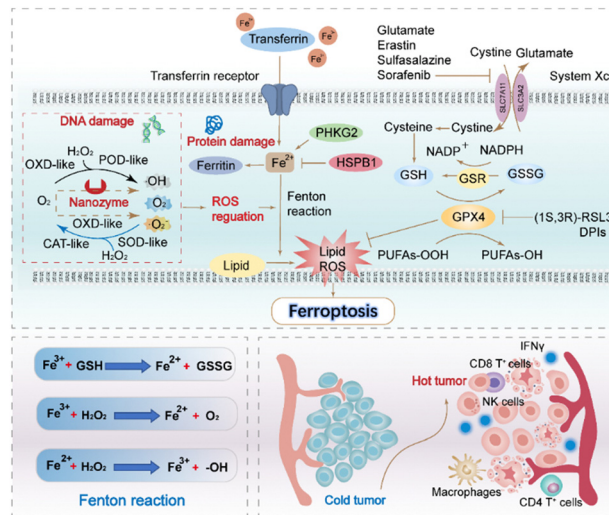


Fig. 4 The classical GPX4 mechanism of ferroptosis and the synergistic role of nanozymes are illustrated in a schematic chart depicting the canonical ferroptosis control axis. This axis involves several key steps, including the uptake of cystine *via* the cystine-glutamate antiporter, known as system X_c⁻. Cystine is then reduced to cysteine in a GSH-dependent manner.⁵⁷ The biosynthesis of GSH and the GPX4-mediated reduction of phospholipid hydroperoxides (PL-OOH) to corresponding alcohols (P-OH) are also part of this axis. The recycling of oxidized glutathione (GSSG) is achieved through glutathione-disulfide reductase (GSR), utilizing electrons provided by NADPH/H⁺. The figure also illustrates the regulatory role of nanozymes in the process of ferroptosis, particularly their ability to facilitate Fenton reactions through various classes of enzyme catalytic activities. Additionally, the chart describes the immune effects induced by ferroptosis, such as the activation of various immune cells and the conversion of “cold” tumours to “hot” tumours, among other immune-related phenomena.⁶⁵

cell, cystine is rapidly reduced to cysteine, which subsequently contributes to the formation of GSH. Several studies have confirmed a positive correlation between the occurrence of ferroptosis and immune-related pathways. The findings indicate that higher expression levels of HLA, as well as increased infiltration of CD8⁺ T cells and tumour infiltrating lymphocyte, are associated with ferroptosis (Fig. 4).⁶⁶

4. Nanozymes play an intrinsic catalytic role in solid tumours

In the treatment of solid tumours such as lung and breast cancers, nanozyme-triggered ferroptosis has been utilized as a new strategy. The reasons for this include: (1) high demand for iron by cancer cells: cancer cells usually require many irons to support their rapid growth and division, which makes them more dependent on iron ions and more susceptible to ferroptosis. Nanozymes can catalyse these iron ions, triggering a more intense ferroptosis effect that kills solid tumour cells.⁶⁷ (2) Integral compatibility: While causing ferroptosis in cancer cells, the impact on regular healthy cells can be controlled to a certain extent, especially with some highly selective and targeted nanozymes. (3) Resistance to treatment: Some cancers



have evolved resistance to therapeutic treatments by way of resistance to conventional programmed cell death (e.g., autophagy).⁶⁸ Since ferroptosis is an iron-dependent programmed necrosis, it may ignore these resistance mechanisms and still effectively kill cancer cells.⁶⁹ For some cancer cells that have become resistant to conventional chemotherapy or radiotherapy, nanozyme-triggered ferroptosis may be an effective alternative or adjuvant therapeutic strategy. The cancers currently being treated using nanozyme-triggered ferroptosis are described below.

An *et al.* prepared a $\text{MnO}_2@\text{HMCu}_{2-x}\text{S}$ nanocomposite (HMCMS).⁷⁰ Mn^{2+} release based on GSH response can generate ROS via the Fenton reaction, which enhances intracellular oxidative stress and leads to LPO accumulation (Fig. 5). With the introduction of photothermal therapy, the GSH-consuming capacity of HMCMS could also be further enhanced to induce iron phagocytosis for efficient tumour ablation.

However, the inefficient conversion of Fe^{3+} to Fe^{2+} during the Fenton reaction of ferroptosis in tumour environments often leads to the blockage of the whole reaction process and the reduction of catalytic efficiency (Table 2). To improve the catalytic efficiency of the Fenton reaction in ferroptosis, Jiang *et al.* developed a hybrid semiconducting nanozyme with high photothermal conversion efficiency for photoacoustic imaging-guided second near-infrared photothermal ferrotherapy.⁷⁸ Such semiconducting polymer NPs composed of highly π -conjugated backbones can act as iron chelators. Their chelating ability is derived from the high binding affinity for ferrous ions of the main chain sulfur and nitrogen atoms, enhancing the Fenton reaction to enhance apoptosis and ferroptosis in A549 cells *in vitro* and *in vivo*. To address the hypoxia-limited nature in tumour therapy, some peroxidase-like nanozymes with intrinsic catalytic ability have been widely studied, such as biodegradable boron oxynitride,⁷⁹ CeO_2 ,⁸⁰ TiO_2 ,⁷⁹ and FeS_2 .⁸¹

Furthermore, nanozymes exhibit the capacity to modulate iron-related proteins and genes within cellular systems,

consequently influencing iron metabolism, a pivotal process underlying the occurrence of ferroptosis. Wang *et al.* highlighted an interesting finding regarding the conserved sequence HEXXH in the fifth structural domain of the Solute Carrier Family 39.⁸² In this case, the conserved sequence was replaced by EEXXH, which imparts the function of transporting divalent metal ions such as Mn^{2+} , Fe^{2+} , or Zn^{2+} . This transporter is involved in the cellular ferroptosis process by facilitating the transport of Fe^{2+} . Nanozymes have the potential to increase iron accumulation in cells by increasing iron uptake or limiting its export, thereby increasing cellular sensitivity to ferroptosis. Overall, nanozymes in cancer treatment showed that in addition to the transport ability of traditional nanomaterials, they played a catalytic role in the body's own biochemical reactions, which successfully improved the catalytic efficiency of the Fenton reaction in ferroptosis and overcame the deficiencies against hypoxia in chemotherapy, PTT, and PDT, resulting in a significant increase in the efficiency of lung cancer treatment. This feature can also be seen in ferroptosis-based therapy for other types of tumours (Table 3).

5. Synergistic therapies involving nanozymes in cancer treatment

Nanozymes have a broader range of applications in cancer therapy compared to natural enzymes. They possess inherent biological enzyme catalytic properties and can be further enhanced by combining them with natural enzymes or modifying them with enzyme-like active groups on nanomaterials, resulting in improved targeting, stability, and durability. Moreover, nanozymes offer high controllability, convertible enzyme activities, and multiple enzyme activities, leading to enhanced effectiveness in targeted therapy, immunotherapy, radiotherapy, and other cancer treatment modalities (Fig. 6). The following are some combination strategies.

5.1 Chemotherapy

Conventional chemotherapy is associated with limitations such as inadequate targeting and the development of tumour resistance. Therefore, targeted therapy which can precisely combat cancer has been a promising research direction.⁹⁸ Nanozymes can be engineered to incorporate specific ligands or antibodies that selectively target molecular markers on cancer cells, thereby improving their specificity and minimizing off-target effects on healthy cells.⁹⁹

Dong *et al.* developed a Cu–Ag alloy nanozyme loaded with banoxadone hydrochloride (AQ4N) (Fig. 7).¹⁰⁰ The Cu–Ag nanozyme acted as a catalyst, facilitating the reduction of oxygen to cytotoxic superoxide (O_2^-), which could intensify tumour hypoxia and activate AQ4N, a highly selective chemotherapeutic drug for cancer therapy. Additionally, the production of hydroxyl radicals (OH^\bullet) induced by the Cu–Ag nanozyme triggered ferroptosis in tumour cells by promoting LPO accumulation and inhibiting GPX4. More importantly, the experiment did not reveal any off-target toxicity. This study effectively combines



Fig. 5 Sequential synthesis process of HMCMS nanocomposites, depicted in the out ring. The central portion of the figure demonstrates a schematic representation of the mechanism involving photothermal (PT) effects and autophagy enhancement, leading to the promotion of iron mutagenesis facilitated by the HMCMS nanocomposites.⁷⁰ Reproduced with permission from ACS Publications, copyright 2019.



Table 2 Factors influencing the Fenton reaction

Factors	Characteristics	Ref.
pH	The optimum pH is usually between 3.0 and 4.0	71
Concentration of reactants	Concentrations of iron ions or hydroxide ions shouldn't be too low or too high	72
Temperature	Temperature (not too high) is positively correlated with the reaction rate	73
Concentration of H ₂ O ₂	Higher H ₂ O ₂ concentration promotes hydroxyl radical formation	74
Light	Light stimulation affects free radical production and activity	75
Electronic field	Electric field inducing cavities for ROS regeneration	76
X-ray	Nano-radiosensitizers can enhance energy absorption and deposition for escalating ROS generation	77

Table 3 Nanozymes for treating various types of tumours^a

Tumour type	Components	Characteristics	Ref.
Lymphoma	Fe ₂ O ₃ /Au nanozymes	Catalysing the production of gluconic acid and H ₂ O ₂ to induce ferroptosis	83
	Au/Cu-TCPP(Fe) nanosheets	Inhibiting GPX4- and FSP1-mediated anti-ferroptosis mechanisms; reducing GSH synthesis and GPX4 expression	84
	Pt-MIL-101 (Fe)	Oxidizing NADPH and subsequent cascade catalytic with nanozymes; generating hydroxyl radicals and blocking GSH regeneration	85
	PtN ₄ C-SAzyme	Self-sufficiency of O ₂ through multi enzyme-like catalytic activity	86
	Manganese-based nanozymes	Depleting GSH and inhibiting GPX4 activity due to LPO accumulation; activating the cGAS-STING pathway	87
	MOF(Fe)	GPX4 depletion and elevated LPO; enhancing SDMD cleavage and elevating the expression of IL-1 β and LDH	88
	FeCo/Fe-Co DAzyme	Activating IFN- γ and targeting arachidonic acid metabolism	89
Glioma	Au/CeO ₂ nanoparticles	Inducing ferroptosis through GSH inhibition and ROS generation; exerting efficient blood-brain barrier permeability and glioblastoma localization	90
	Magnetic nanoparticles	Combining triple actions of dihydroorotic acid dehydrogenase catabolism; GPX4-ferroptosis defense axis with Fe ₃ O ₄ nanoparticle-mediated Fe ²⁺ release	91
Lung cancer	Ru nanozymes	Generating NO to co-activate macrophage M1 polarization	92
	CuCP Lipo nanoparticles	Eliminating GSH from the tumour mass and producing a large amount of LPO; inhibiting tumour antioxidant response through modulation of the tumour microenvironment	93
Colon	MoS ₂ nanoparticles	Self-supply of H ₂ O ₂ ; integrating charge-enhanced enzyme activity; dysregulating redox homeostasis via GSH depletion	94
Liver cancer	Carbon quantum dots	Promoting cancer cell ferroptosis by perturbing the GPX4-catalysed lipid repair system; activating systemic antitumor immune response	95
	Cu-Hemin nanosheets	Up-regulating HMOX1 expression and down-regulating GPX4 expression	96
Pancreatic cancer	NAC-RuO ₂ nanozymes	Depleting GSH and simultaneous generating ONOO; down-regulating GSH reductase to avoid GSH regeneration	97

^a TCPP(Fe): tetra(4-carboxyphenyl)porphyrin chloride(Fe(III)); MIL: materials of institut lavoisier; SAzyme: single-atom nanozyme; DAzyme: dual-metal atom nanozyme; Lipo: liposome; NAC: *N*-acetyl-L-cysteine.

principles from starvation, chemotherapy, and ferroptosis, thereby enhancing the specificity of chemotherapy treatment.

Shen *et al.* conducted a modification of transferrin on a MOF to enhance the targeting of specific sites, thereby promoting iron-induced cell death.¹⁰¹ This modification resulted in an increased anticancer effect through the augmentation of cysteinyl asparagine-1-mediated GSDMD cleavage, as well as the elevation of IL-1 β levels and the release of lactate dehydrogenase (LDH).¹⁰¹

In the context of ferroptosis-mediated reversal of chemotherapy resistance, three main pathways have been identified: the canonical GPX4-regulated pathway, iron metabolism pathway, and lipid metabolism pathway. Fu *et al.* provided evidence that inhibiting Nrf2/Keap1/xCT signalling to induce ferroptosis effectively sensitized cisplatin-resistant gastric cancer cells to cisplatin treatment.⁹⁹ Similarly, PARP inhibition promotes ferroptosis by inhibiting SLC7A11-mediated GSH synthesis, and synergistically sensitizes xenografts to the PARP,

which reverse intrinsic resistance. In addition, CD8⁺ T cells play a role in inducing cellular focal death by activating caspases, which subsequently cleave GSDMB. The secretion of IFN- γ by CD8⁺ T cells down-regulates SLC7A11, disrupting the cellular antioxidant system and leading to the accumulation of lipid ROS, thereby inducing ferroptosis.⁶⁷ Clinically, the induction of ferroptosis has shown promise as a therapeutic approach to effectively inhibit the development of drug resistance, including resistance to drugs such as erlotinib and trametinib. These findings hold promise for the development of novel therapeutics by inducing ferroptosis for overcoming drug resistance in cancer. After sophisticated design based on the structure-activity relationship, nanozymes can exert powerful antitumor effects with ferroptosis as well as chemotherapeutic drugs, not only catalytic effects, but also nanozymes are potentially promising as drug carriers.

Targeted drugs are developed to target oncogenes, which identify characteristic sites on tumour cells determined by





Fig. 6 The figure visually represents the synergistic antitumor effects achieved by integrating nanozymes with diverse therapeutic strategies. Additionally, the figure highlights the advantages and categorization of nanozymes in the context of tumour treatment.¹⁰²

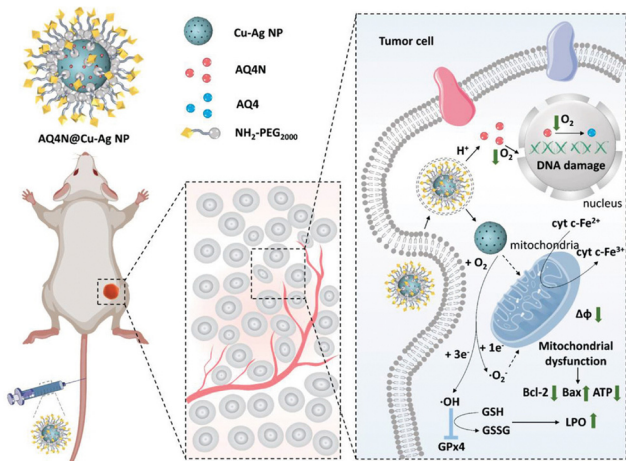


Fig. 7 Utilization of cytochrome c oxidase-like nanozymes, comprising copper–silver alloy nanoparticles (Cu–Ag NPs), as vehicles for delivering chemotherapeutic drugs (AQ4N). The objective is to employ a targeted strategy that combines concurrent starvation therapy, iron-induced cell death, and chemotherapy, with the aim of enhancing the efficacy of anticancer treatment.¹⁰⁰ Reproduced with permission from John Wiley and Sons, copyright 2022.

genes specific to the tumour cells. The study of different oncogene profiles can help to target tumours more precisely. Ferroptosis genes play important roles in different cancers. Ferroptosis genes are expressed at different levels in different cancers with different expression models and prognosis, including ACSL4, GPX4, SLC7A11, RAS, TP53, *etc.*

In addition to the classical GPX4 pathway, in 2019 scientists discovered an enzyme catalytic system independent of the classical GPX4 signalling pathway, a pathway based on the ferroptosis suppressor FSP1.¹⁰³ And the iron sinking mechanism similar to water–oil separation was developed to avoid



Fig. 8 Ferroptosis pathway independent of GPX4 pathways. (A) Graphical abstract depicting icFSP1-induced FSP1 condensate formation, lipid peroxidation and ferroptosis;¹⁰⁴ (B) FSP1/ubiquinone (CoQ10) system has been recently identified to completely protect against ferroptosis induced by pharmacological inhibition or genetic deletion of GPX4. FSP1 prevents lipid peroxidation and associated ferroptosis *via* reduction of ubiquinol/ α -tocopherol on the level of lipid radicals unlike GPX4/GSH;¹¹¹ (C) alternate ferroptosis-suppressive mechanisms include squalene- and di-/tetrahydrobiopterin (BH₂/BH₄)-mediated inhibition of lipid peroxidation.¹¹²

FSP1 inhibitors from exhibiting off-target effects *in vivo*. By triggering FSP1 phase separation, subcellular repositioning of FSP1 can be achieved, ultimately leading to the presence of distinct FSP1 agglutinates in tumour tissues, and significantly inhibited tumour growth.¹⁰⁴

SLC7A11, as a major acting component of system X_c⁻, is highly expressed in a variety of solid malignant tumours, such as breast cancer, pancreatic cancer, and glioma, among others. Erastin, a classical ferroptosis promoter, can target SLC7A11, induce ferroptosis and reverse colorectal cancer drug resistance by activating the GPX4 regulatory pathway. Knockdown of SLC7A11 increases ROS levels and decreases GSH levels to promote apoptosis (Fig. 8).¹⁰⁵

5.2 Immunotherapy

Tumour immunotherapy aims to enhance the body's immune system to directly or indirectly eliminate and control cancer by mobilizing immune cells capable of recognizing tumours. Tumour cells often possess low immunogenicity as a means to evade immune cell recognition.¹⁰⁶ The interaction between ferroptosis and immune cells plays a crucial role, as ferroptosis can impede the immune system's ability to combat pathogenic bacteria by reducing the populations of T cells and B cells. Nanozymes have emerged as effective modulators of the TME, capable of enhancing immunotherapy through diverse mechanisms such as macrophage polarization, antigen presentation, and activation of cellular pathways.¹⁰⁷

Zeng *et al.* proved that polyethylene glycol (PEG)-denatured Cu-doped polypyrrole nanozyme (CuP) effectively reversed the immunosuppressive TME by overcoming tumour hypoxia and polarizing macrophages from a tumour-promoting M2 phenotype to an antitumor M1 phenotype.¹⁰⁸ Due to the high temperature enhancement, the catalytic activity can be further enhanced with the addition of a 1064 nm laser, which helps to promote macrophage polarization. The modification of PEG



can provide a hydration layer that hinders the formation of protein corona avoiding phagocytosis of the nanozyme.¹⁰⁹ Meanwhile, Zhao *et al.* designed a tumour cell membrane camouflaged multi-enzyme activity mimicking manganese oxide nanozymes (CM@Mn) to provide a biomimetic surface,¹¹⁰ further prolonging the *in vivo* circulation of the nanoparticles. Through their inherent POD-like and OXD activities, they generate $\bullet\text{OH}$ and O_2^- to kill tumour cells and trigger immunogenic cell death. In addition, most nanozyme-based immunotherapies are combined with the anti-PD-1 monoclonal antibody ($\alpha\text{PD-1}$) for precise molecular targeting, which transforms the tumour microenvironment from “cold” to “hot” and generates long-term immune-memory effects to further enhance antitumor activity. The tumour microenvironment will be transformed from “cold” to “hot” and a long-term immune memory effect will be generated to further enhance the antitumor effect.

Interferon gamma secreted by CD8^+ T cells down-regulates the levels of SLC3A2 and SLC7A11 in tumour cells, thereby decreasing cystine uptake and ultimately leading to the onset of ferroptosis; knockdown of the GPX4 gene in B cells triggers cellular ferroptosis by inducing lipid peroxidation.¹¹³ Liu *et al.* developed a nanoplatform for the co-expression of six enzymes: Lipoyxygenase (LOX) and phospholipase A2 (PLA2)-co-loaded FeCo/Fe-Co bimetallic atom nanozymes (FeCo/Fe-Co DAZyme/PL), which not only induced initial immunogenic tumour desferrioxidation through their own multi-enzyme mimetic activity, but also up-regulated arachidonic acid (AA) expression, which, together with CD8^+ T cell-derived IFN- γ , synergistically induces ACSL4-mediated immunogenic tumour desferrioxidation. In this process, FeCo/Fe-Co DAZyme/PL could induce lipid peroxidation by efficiently generating ROS and depleting GSH and GPX4 at the tumour site. This study promotes irreversible cascading immunogenic desferrioxidation through multiple ROS storms, GSH/GPX4 depletion, LOX catalysis, and IFN- γ -mediated ACSL4 activation to achieve a synergistic effect of immunogenic death and ferroptosis, which further enhances the antitumor effect of immunotherapy.¹¹⁴

The immune response is commonly used as an adjunct in tumour treatment. For instance, photothermal therapy (PTT) frequently elicits an immune response. PTT induces immunogenic cell death in tumour cells and enhances the body's immune response by improving antigen presentation of APC cells.¹¹⁵ Carbon nanozymes can activate the tumour immune microenvironment by recruiting a large number of tumour-infiltrating immune cells, including T cells, NK cells, and macrophages, thereby converting “cold” tumours into “hot” tumours and activating systemic antitumor immune responses. Yao *et al.* discovered that carbon quantum dot nanozymes (ChA CQDs) derived from coffee exhibited notable GSH scavenging activity and followed Michaelis-Menten kinetics. These nanozymes exhibited the ability to convert GSH to GSSG, thereby inducing ferroptosis, and effectively suppressed tumour growth in mice implanted with HepG2 cells. Other nanozymes, such as Fe_3O_4 , can enhance antigen presentation by dendritic cells;¹¹⁶ Fe_2O_3 can modulate the polarization of tumour-associated macrophages;¹¹⁷ and manganese ion-based



Fig. 9 Illustration of the synthesis process and therapeutic mechanism of TME-responsive Mn(III)-SS NEs, aiming to enhance ferroptosis-induced ICD and immune-driven ferroptosis.¹²¹ Reproduced with permission from John Wiley and Sons, copyright 2023.

nanozymes activate the cGAS-STING pathway to increase tumour-infiltrating CD8^+ T cells.⁸⁷ Shen *et al.* designed a nano-metal-organic framework (referred to as mFe (SS)/DG) that incorporated glucose oxidase (GOx) and adriamycin. Their study demonstrated that both adriamycin and the MOF could induce immunogenic cell death. Fe^{3+} and organic ligands containing disulfide bonds within this structure down-regulate GPX4, leading to ferroptosis. Ferroptosis, in turn, amplifies the immune response by releasing “find me” signals such as lipid mediators and high mobility group box 1 (HMGB1), resembling the mechanism observed in vaccination.¹¹⁸

It can be seen that the cell death mechanism induced by immunogenic death can synergistically enhance the efficacy of antibody-based tumour treatments when combined with ferroptosis (Fig. 9). Currently, ICD is induced not only by intracellular pathogens and chemotherapeutic agents, but also by a variety of physical therapies, including PTT, photodynamic therapy (PDT), extracorporeal photochemotherapy (ECP), and various forms of ionizing radiation, which further suggests that the mainstream platform for the next-generation antitumor therapy will be the synergistic interaction between a variety of therapies in order to effectively increase the efficacy of anti-tumor therapy.^{119,120} Currently, the study of nanozymes is still in its infancy, and their unique roles such as immunogenicity and enzyme-like activity need to be further investigated.

5.3 Photodynamic therapy

Photodynamic therapy (PDT) is an emerging antitumor therapy that incorporates photosensitizers, oxygen molecules, and excitation light to generate ROS to kill cancer cells.¹²² However, there are still some problems that need to be solved. For example, the hypoxic microenvironment in solid tumours can reduce the therapeutic effect of photodynamic therapy; some cancer cells tend to be resistant to apoptosis and necrotic cell death; not all photosensitizers have the capacity to trigger immunogenic cell death pathways.¹²³ Therefore, a new therapy is needed to improve the efficiency of PDT. Fortunately, PDT



can not only induce ferroptosis alone, but also in combination with ferroptosis inducers and metal ions.¹²⁴ The nanozymes can produce ROS under light irradiation, thus co-regulating the intracellular redox homeostasis leading to cell death, and combining them with PDT can enhance the anticancer effect.

In the realm of nanozymes, MOF-based nanozymes stand out due to their structural similarity to natural enzymes, such as binding and stabilization of the transition state, involvement of functional groups for proton transfer, charge stabilization, metal coordination, and supplying nucleophiles or electrophiles. For example, Ye *et al.* integrated Pt NPs and the NIR photosensitizer CyI into iron-based MOFs. The MOFs containing photosensitizers can convert oxygen from tumour tissues into monoclinic oxygen.¹²⁵ The MOFs not only act as delivery carriers for photosensitizers to improve water solubility, but also act as inducers of ferroptosis, which converts O₂ from tumour tissues into monoclinic O₂.

To provide a location for the multi-enzyme cascade reaction and at the same time avoid the rapid removal of nanoparticles from the blood stream *via* the mononuclear phagocyte system, pre-coated protein crowns on the surface of nanoparticles were suggested as they can effectively shield plasma proteins from uptake. By using mesoporous silica nanoparticles (MSNs) with high specific surface area and large pore size as substrates, ultrasmall AuNPs with GOx-like activity were grown *in situ* based on a reduction reaction, and then the photosensitizers cerium chloride (Ce6) and FTn-Ru (Ru nanozymes with cat-like activity) were co-loaded on the surface of the MSN-Au *via* a simple mixing method by amidation reactions or within the pores.¹²⁶ The results showed that human ferritin heavy chain nanocage (HTn)-based protein crown-forming and cascade nanozymes consumed glucose and H₂O₂ in the TME, sustained O₂ production, and enhanced starvation therapy and PDT-targeted tumour treatment protocols. This study achieved an enzymatic cascade reaction of two nanozymes on a single nanoplatform, demonstrating that genetically engineered human ferritin heavy chain nanocages can actively bind tumour cells. By constructing Cu-tetrakis(4-carboxyphenyl)porphyrin-based chloroporphyrin (Fe(III)) (Cu-TCPP(Fe)) MOF nanosystems for the efficient doping of AuNPs and RSL3, Kobayashi and Choyke *et al.* found that it was possible to use these nanocatalytic activities to inhibit both the GPX4/GSH and FSP1/CoQ10H2 pathways.⁸⁸ In conjunction with the GPX3 inactivation function of RSL4, the nanosystems have the ability to universally inhibit three anti-ferroptosis pathways in tumour cells.

It has been found that photosensitizers such as the dihydroporphyrin (Ce6), which is one of the three elements of photodynamic therapy, can itself potentiate the ferroptosis-activating effect of hemin through both classical and non-classical modes.¹²⁷ The photosensitizer Aloe emodin induces cellular ferroptosis by specifically inhibiting GSH S-transferase P1 (GSTP1) production.¹²⁸ By binding to various cell membranes, such as erythrocyte membranes and macrophage membranes, it is possible to specifically target tumours, evade immune clearance or activate T-cells, while achieving the effect of ferroptosis activation/photodynamic/immunotherapy interactions.⁶⁰

In addition to multiple synergistic therapeutic effects, iron-based compounds often used as nanozymes can be used as magnetic resonance imaging (MRI) contrast agents. Wang *et al.* doped gadolinium (Gd) into Prussian blue nanozymes (PBzyme), thereby modulating the density of states of Fe in PBzyme, which lowered the Gibbs free energy of the catalytic cycle of catalysis of CAT, resulting in the enhancement of its CAT-like activity (4-fold higher than that of PBzyme). Other metal ions such as Mn²⁺ and Cu²⁺ also have excellent performance in enhanced T₁-weighted MRI and can improve the relaxation efficiency to realize the integration of nanozymes into diagnostic and therapeutic treatments, which will be a novel platform for the next-generation tumour therapeutics.

5.4 Photothermal therapy

Photothermal therapy (PTT) is an emerging non-invasive and highly targeted modality for cancer treatment. It harnesses the potential of photothermal agents (PTAs) characterized by their high photothermal conversion efficiency.^{129,130} However, PTT often involves reaching temperatures of 50 °C or higher, resulting in unintended damage to both tumour and normal tissues.¹³¹ This excessive heat generation triggers the body's defense mechanism, leading to the production of heat shock proteins (HSPs) that can decrease the sensitivity of tumour cells. The limited solubility, serum instability, and cytotoxicity associated with a traditional HSP inhibitor, including garcinia cambogia, 17-AAG, tretinoin, and siRNA, have hindered their effective utilization in PTT.¹³² How to inhibit HSPs at relatively mild temperatures to improve the efficiency of PTT is an important issue to be addressed in the future.

Metal nanoparticles can absorb visible light by localized surface plasmon resonance (LSPR) and be used for photocatalytic reactions.^{133,134} Pd nanozyme has strong light absorption properties and exhibits excellent photothermal conversion performance in the near infrared region (NIR-II) (1000–1400 nm).¹³⁵ In 2021, Chang *et al.* prepared single-atom Pd nanozymes based on Pd SAzyme with peroxidase (POD) and glutathione oxidase-like activities and excellent photothermal conversion properties for the enhancement of ferroptosis-based cryogenic PTT antitumor therapies.¹³⁶ Unlike the conventional iron-dependent ferroptosis approach, in this study the single-atom Pd was dispersed in the metal centres of nanozymes to maximize the atomic utilization of the active centres and the density of the features.¹³⁷ Thus, the GPX4 pathway is directly modulated to directly resist LPO in biofilms to trigger the ferroptosis-like cell death process. LPO can spontaneously form aldehyde degradation products that cross-link the primary amines of proteins, thereby disrupting the protein structure and function. The study showed a significant reduction in the GPX4 expression, loss of mitochondrial membrane potential and damage under 1064 nm laser irradiation. Compared with the control group, HSP70 was significantly down-regulated in the Pd SAzyme-treated tumour group, and tumour growth was significantly inhibited. The maximum temperature in this study was only 42.68 °C, which shows that Pd SAzyme can promote low temperature PTT with a significant therapeutic effect by depleting HSPs through a large amount of



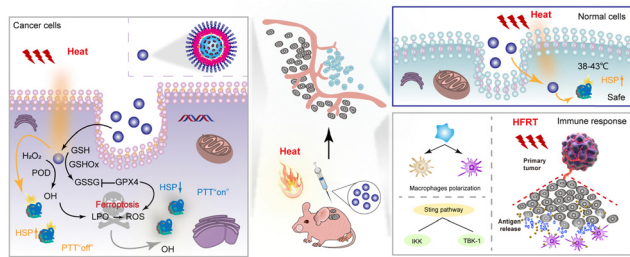


Fig. 10 Nanozymes synergizing with ferroptosis in photothermal therapy, inhibiting HSPs while also triggering an immune response, including macrophage polarization,¹³⁸ antigen presentation,¹³⁹ and activation of the STING pathway.¹⁴⁰

LPO and ROS produced in the process of ferroptosis at a low temperature. In addition, under NIR irradiation, the activity of nanozymes can be increased by 3.5-fold to effectively digest collagen in the tumour extracellular matrix (ECM), thus enhancing the accumulation of nanoparticles in the tumour, and thus improving PTT (Fig. 10).^{141,142}

Copper single-atom nanozymes (Cu SAzyme) have been employed for the inhibition of heat shock proteins (HSPs).¹⁴³ In addition to its intrinsic enzyme-like activity, the incorporation of LIK066 (a sodium-glucose co-transporter protein inhibitor) into the nanozyme system can disrupt existing HSPs through multiple mechanisms. It inhibits the energy required for HSP synthesis and induces disruption of ROS-active centres, leading to the inhibition of HSP function. The study was conducted through a two-pronged strategy of sodium-dependent glucose-transporting protein (SGLT) inhibitors and ROS centres, thereby inducing photothermal therapy.

Nanozyme-based photothermal therapy demonstrates the potential for enhanced antitumor efficacy when synergistically integrated with a diverse array of complementary anticancer modalities (Fig. 11).¹⁴⁴ To achieve more precise tumour targeting, a nanozyme $\text{Fe}_3\text{O}_4@(\text{Cu}_{1.77}\text{Se})$ has been designed to release



Fig. 11 Using a derivative strategy for fabrication of $\text{Fe}_2\text{O}_3@(\text{Cu}_{1.77}\text{Se})$ hybrid nanozyme that holds peroxidase and glucose oxidase mimicking properties, which realized highly efficient treatment of triple negative breast cancer through starvation/PTT/CDT/ferroptosis concerted effects.¹⁴⁴ Reproduced with permission from Elsevier, copyright 2023.

ultra-abundant, ultra-small Fe_3O_4 and $\text{Cu}_{1.77}\text{Se}$ nanoparticles in the presence of matrix metalloproteinases, which are over-expressed in many types of cancer.¹⁴⁵ The released nanoparticles facilitate the elevation of hydroxyl radicals through the Fenton reaction and depletion of GSH. Due to the unique magnetic properties and high biocompatibility of Fe_3O_4 nanoparticles, $\text{Cu}_{1.77}\text{Se}$ has a better photothermal conversion efficiency in the NIR region II. The study demonstrated a significant enhancement of the NMR T2 signal and increased efficiency. This synergistic effect enabled precisely targeted antitumor therapy through NIR photothermal ferroptosis, guided by tumour-enhanced MRI.¹⁴⁶

5.5 Radiotherapy

The hypoxic environment in tumour tissues plays a crucial role in the development of treatment tolerance and reduced efficacy of radiation therapy. There is a strong association between radiotherapy and ferroptosis.¹⁴⁷ Radiotherapy induces multiple changes associated with ferroptosis, including promotion of lipid peroxidation, induction of the expression of key ferroptosis-related proteins such as acyl coenzyme A synthetase long-chain family member 4 (ACSL4), GPX, and the ferroptosis marker gene PTGS2, as well as the reduction of cystine uptake through inhibition of SLC7A11. Additionally, tumour cells treated with radiotherapy exhibit characteristic morphological features of ferroptosis, such as mitochondrial contraction and increased membrane density.¹⁴⁸ Moreover, the use of ferroptosis inhibitors can partially restore cell survival following radiotherapy, demonstrating a stronger recovery compared to other regulatory cell death (RCD) inhibitors.¹⁴⁹ Notably, Isozaki *et al.* uncovered that mutations or deletions in the KEAP1 in lung cancer cells upregulate FSP1 expression through NRF2, resulting in resistance to both ferroptosis and radiotherapy. This finding led to the development of the therapeutic strategy for KEAP1-inactivated lung cancer, involving the targeting of the CoQ-FSP1 axis-mediated ferroptosis defense to sensitize the cancer cells to radiotherapy.¹⁵⁰ More and more studies are focusing on the promotion of the Fenton effect and enhancement of catalytic activity through the action of external fields, such as temperature gradients, microwaves, and X-rays.¹⁵¹ X-rays can be better used to treat deep-seated tumours as opposed to NIR light. Similar to the traditional photothermal reaction in which UV irradiation reduces Fe^{3+} ions to Fe^{2+} , high-energy radiation can also promote Fenton reactions.

Nanozyme activity can be further enhanced by the introduction of X-rays, which promote the frequency of $\text{Fe}^{\text{II}}/\text{Fe}^{\text{III}}$ conversion and sustain the catalytic activity. For example, it has been suggested that the coordination of Fe with N on carbon nanosheets allows for the incorporation of Fe atoms into the carbon nanosheet structure, with N atoms serving as coordinating ligands.¹⁵² This coordination creates high atomic utilization efficiency and catalytic activity. The obtained FeN_4 -SAzyme can initiate a self-cascading enzyme reaction at the site, converting intracellular H_2O_2 into $\cdot\text{OH}$ radicals and consuming glutathione (GSH), thereby promoting ferroptosis. Moreover, the loading of glucose oxidase (GOD) enables the generation of



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