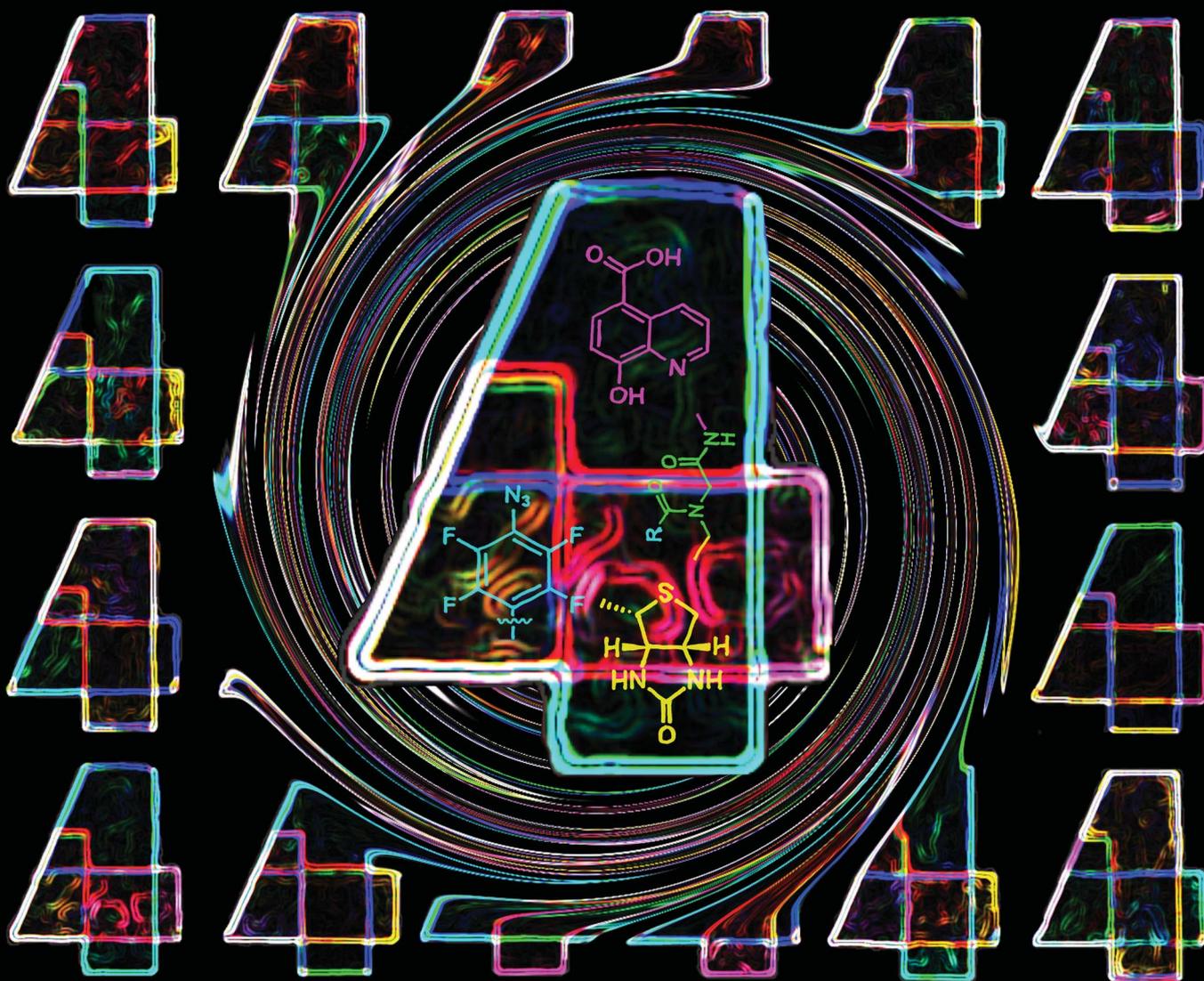


Chemical Science

www.rsc.org/chemicalscience

Volume 4 | Number 11 | November 2013 | Pages 4081–4324



ISSN 2041-6520

RSC Publishing

EDGE ARTICLE

Christopher J. Schofield *et al.*

The Ugi four-component reaction enables expedient synthesis and comparison of photoaffinity probes



2041-6520 (2013) 4:11;1-J

The Ugi four-component reaction enables expedient synthesis and comparison of photoaffinity probes†

Cite this: *Chem. Sci.*, 2013, **4**, 4115

Jacob T. Bush,^a Louise J. Walport,^a Joanna F. McGouran,^b Ivanhoe K. H. Leung,^a Georgina Berridge,^a Sander S. van Berkel,^a Amit Basak,^c Benedikt M. Kessler^b and Christopher J. Schofield^{*a}

Photoaffinity probes are increasingly being used for the study of biological interactions; however, the lack of structure–activity relationship studies has hindered their rational application. We describe the use of the Ugi four-component reaction (U-4CR) for the expedient and versatile assembly of photoaffinity scaffolds that can be linked to small molecule probes. The rates, yields and sites of crosslinking of five commonly used photoreactive groups comprising diazirines, aryl azides and a benzophenone, were compared using a human 2-oxoglutarate oxygenase as a model protein. The results reveal significant differences in the behavior of the probes and suggest that empirically guided optimization of probes for specific tasks is desirable. In the absence of such optimization it may be advisable to use a set of crosslinking probes/conditions; the U-4CR provides a convenient method for obtaining such a set.

Received 19th June 2013

Accepted 17th July 2013

DOI: 10.1039/c3sc51708j

www.rsc.org/chemicalscience

Introduction

Photoaffinity probes (or capture compounds) show promise as useful tools for enabling the selective identification and quantification of proteins, as well as for mapping molecular interactions (for reviews see ref. 1–4). They can be used to profile multiple members of protein families simultaneously, offering potential for probing the biochemical origins of a phenotype.^{5,6} They can also be used to investigate small molecule–protein interactions providing insight into mechanisms of action and selectivity.^{7–12} Photoaffinity probes typically consist of three functional moieties: (i) a selectivity function, such as an enzyme inhibitor, (ii) a photoreactive crosslinking group, such as an aryl azide or diazirine, and (iii) a detection handle, such as biotin or a fluorescent dye.⁷ In a typical protocol the probe is incubated with cell lysate/cells, and binds to the target protein(s) (or other biopolymers) *via* its selectivity function (Fig. 1). The photoreactive group is activated by irradiation which enables reaction leading to covalent attachment to the target enzyme. The handle is then used to isolate and/or identify captured proteins, *e.g.* by mass spectrometry or antibody-based detection.

A challenge in the design of photoaffinity probes is the attachment of a detection handle and photoreactive group to

the selectivity function in orientations that do not compromise target (protein) binding and enable sufficiently efficient crosslinking to the target(s) upon photoirradiation. Further, despite differences in the crosslinking properties and commercial/synthetic accessibility of popular photoreactive groups, there have been few studies to compare their efficiency.^{13,14} Thus, to

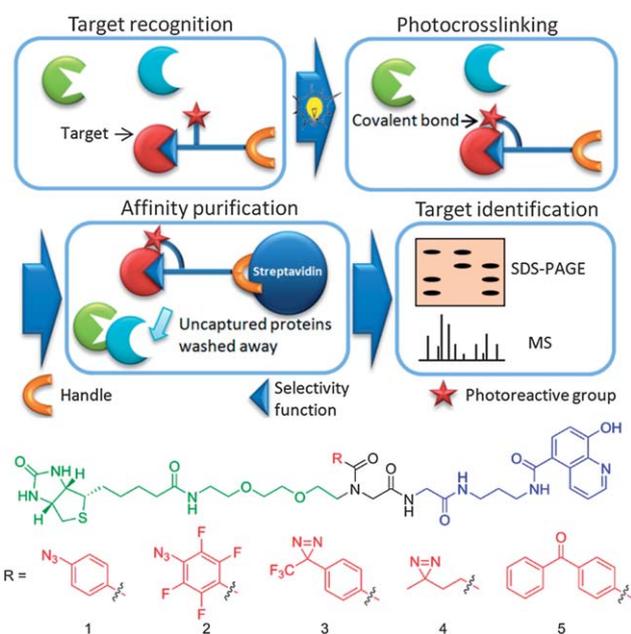


Fig. 1 Above: schematic representation of the application of photoaffinity probes. Below: potential photoaffinity probes for 2-oxoglutarate oxygenases incorporating 5 different photoreactive groups.

^aDepartment of Chemistry, University of Oxford, Oxford OX1 3TA, UK. E-mail: christopher.schofield@chem.ox.ac.uk

^bTarget Discovery Institute, University of Oxford, Roosevelt Drive, Oxford OX3 7FZ, UK

^cDepartment of Chemistry, Indian Institute of Technology, Kharagpur 721302, India

† Electronic supplementary information (ESI) available: Details for probe synthesis, production of PHD2, K_D determination by NMR, photocrosslinking experiments, protein digestion, ESI-MS, MALDI and LC-MS/MS analysis. See DOI: 10.1039/c3sc51708j



maximize the likelihood of specific and efficient crosslinking, the use of a set of probes is desirable, with variation of the attachment point on the selectivity function and the nature of the photoreactive group. Although various tripodal templates and photoactivatable groups have been used in probe construction, their synthesis typically requires multiple steps, which has limited structure–activity relationship studies.

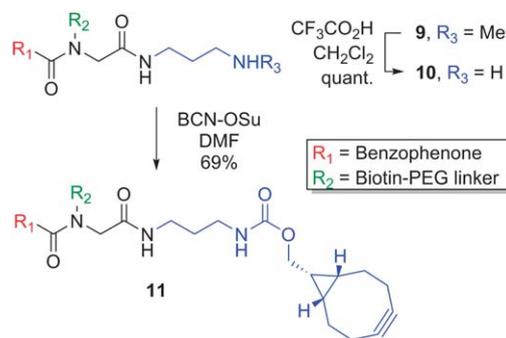
Here we report the use of the Ugi four-component reaction (U-4CR) for the straightforward synthesis of photoaffinity scaffolds, enabling the efficient conjugation of a range of photoreactive groups, purification/visualisation handles and functionalities for inhibitor attachment in a single step. The efficacy of 5 different photoaffinity probes 1–5, prepared by the U-4CR, was investigated using a 2-oxoglutarate (2-OG) oxygenase inhibitor (IOX1) as the selectivity function (Fig. 1).¹⁵ Substantially different reaction rates, yields and crosslinking sites were observed in studies on the human hypoxia sensing enzyme prolyl hydroxylase domain-containing protein 2 (PHD2/EGLN1).¹⁶

Results and discussion

The U-4CR,¹⁷ which couples a carboxylic acid, an amine, an aldehyde and an isonitrile in one pot to provide a peptoid structure, has been extensively used for the synthesis of inhibitor libraries and there are examples of its use for the conjugation of photoreactive groups or dyes with small molecules/inhibitors.^{18–21} We envisioned that the U-4CR would be well-suited for the synthesis of photoaffinity scaffolds since commonly used photoreactive groups can be obtained as carboxylic acids, purification handles are available as amines, and isonitriles can be used to provide an inhibitor attachment point. Initially, we tested the applicability of the U-4CR by reacting *p*-azidobenzoic acid (A), isonitrile methyl ester (I), paraformaldehyde and amine modified biotin (F), under

microwave irradiation (100 °C, 20 min), producing the desired photoaffinity scaffold 6a in good yield (62%). The scope of the reaction was then explored by using other commonly applied photoreactive groups, tetrafluoroaryl azide B, aryl trifluoromethyl diazirine C, benzophenone D and alkyl diazirine E, pleasingly affording the corresponding photoaffinity scaffolds 6b–e in good yields (Table 1).

We also investigated the incorporation of alternative detection handles in place of biotin, such as dyes and ‘clickable’ moieties. Dyes allow direct in-gel visualisation of captured proteins, whereas alkynes allow conjugation of a desired tag (e.g. biotin or dye) after photoreaction *via* a [3 + 2] cycloaddition reaction with an azide.^{2,22} Thus, amine functionalised rhodamine H was reacted with *p*-azidobenzoic acid (A), isonitrile methyl ester (I), and paraformaldehyde to provide the desired photoaffinity scaffold 7 (56%). The use of propargylamine to incorporate an alkyne into the scaffold gave low yields of the desired product (<30%), likely due to the tendency of propargyl amides to undergo rearrangement;²³ however, the use of 3-butynylamine resulted in formation of alkyne functionalised photoaffinity scaffold 8, (73%).



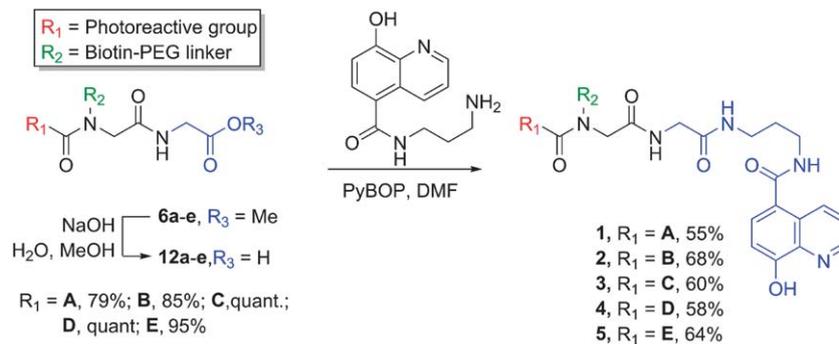
Scheme 1

Table 1 Application of the Ugi four-component reaction for the synthesis of photoaffinity scaffolds

Product	R ₁	R ₂	R ₃	Yield (%)
6a	A	F	I	62
6b	B	F	I	59
6c	C	F	I	66
6d	D	F	I	63
6e	E	F	I	62
7	A	H	I	56
8	A	G	I	73
9	D	F	J	59

R₁: A - E (red)
 R₂: F - H (green)
 R₃: I and J (blue)





Scheme 2

Finally, the scope for variation of the inhibitor attachment point was investigated. To this end, Boc-protected diamine-derived isonitrile **J** was used instead of an ester, which after deprotection provides a complementary handle for attachment of a selectivity function. The U-4CR provided **9** (53%), which was Boc-deprotected to give the desired amine **10**. A “clickable” photoaffinity scaffold (**11**) suitable for strained cycloaddition reactions with azide functionalized inhibitors was obtained by reaction of **10** with a bicyclononyne-OSu ester (BCN-OSu) (Scheme 1).²⁴

The U-4CR was subsequently used to synthesize photoaffinity probes containing various photoreactive groups for

comparison of their crosslinking efficiency. We have previously reported that 5-carboxy-8-hydroxyquinoline (**IOX1**) can be utilized as a probe for 2-OG dependent oxygenases including PHD2 (EGLN1).^{9,15,25} We used the PHD2 catalytic domain (residues 181–426, hereafter PHD2) to investigate the rates, yields and sites of photocrosslinking obtained with **IOX1** functionalised photoaffinity probes bearing commonly used photoreactive groups A–E. The biotinylated scaffolds **6a–e** were hydrolysed prior to coupling with **IOX1**, to provide photoaffinity probes 1–5 (Scheme 2). Dissociation constants (K_D) were determined by NMR spectroscopy using a solvent water relaxation technique and were found to be similar to the reported value for **IOX1** (K_D 60 μM).²⁶ Of the five probes, **4** was found to bind to PHD2–Mn^{II} with the highest affinity (K_D 9 μM). Probes **1**, **2** and **5** bound with medium strength (K_D 19, 28 and 25 μM respectively), and **3** bound more weakly (K_D 126 μM) (Fig. S1†).

Crosslinking rates and yields were determined using 5 μM PHD2, 10 μM Mn^{II} (substituting Fe^{II}), with probes 1–5 at twice the determined K_D values. Irradiation was carried out using a CaproBox (Caprotec, Berlin) at two different wavelengths, 310 nm, reported to be suited for the activation of aryl azides, and 350 nm, more suited for benzophenones.¹ Probes 1–5 were incubated with PHD2 and Mn^{II} at room temperature (20 min) to allow equilibration prior to irradiation (4 °C). The extent of crosslinking was determined by electrospray ionization (ESI) mass spectrometry (MS) and plotted against time (Fig. 2 (310 nm), Fig. S2† (350 nm)). Control experiments demonstrated that crosslinking was competitively inhibited in the presence of 200 μM **IOX1** (Table S11†). Additionally, irradiation of the unfunctionalised scaffolds **6a–e** did not lead to modification (Table S12†), thus supporting the proposal of selective reaction and highlighting the potential use of the probes in inhibitor binding assays.

The best results in terms of initial rate and yield of crosslinking were achieved by irradiating aryl azide **2** at 310 nm, which yielded 55% crosslinked enzyme after just 10 min. Using this wavelength, aryl azide **1** and alkyl diazirine **4** also gave good overall crosslinking yields (30% and 28% respectively), although the alkyl diazirine was slower to react. Aryl azides **1** and **2** gave lower levels of crosslinking at 350 nm (13% and 11% respectively), consistent with literature reports on other proteins.¹ At this wavelength, species crosslinked with the aryl azide **2** appeared to degrade with increased irradiation time,

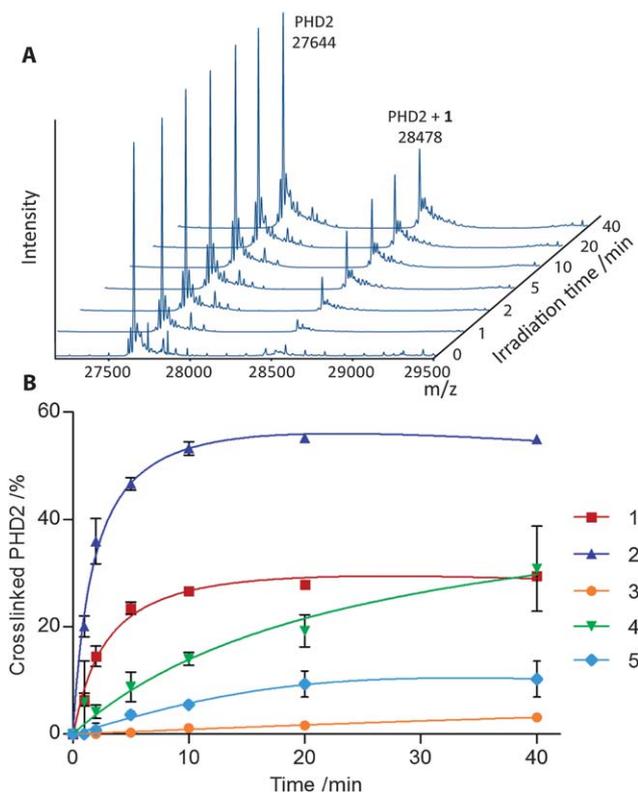


Fig. 2 (A) Deconvoluted ESI MS of PHD2 after irradiation (310 nm) in the presence of Mn^{II} and **1**. (B) Percentage of crosslinked PHD2 upon irradiation (310 nm) in the presence of probes **1–5**, as determined by ESI MS. Data represent mean intensities of repeats ($n = 3$), error bars show 1 standard deviation.



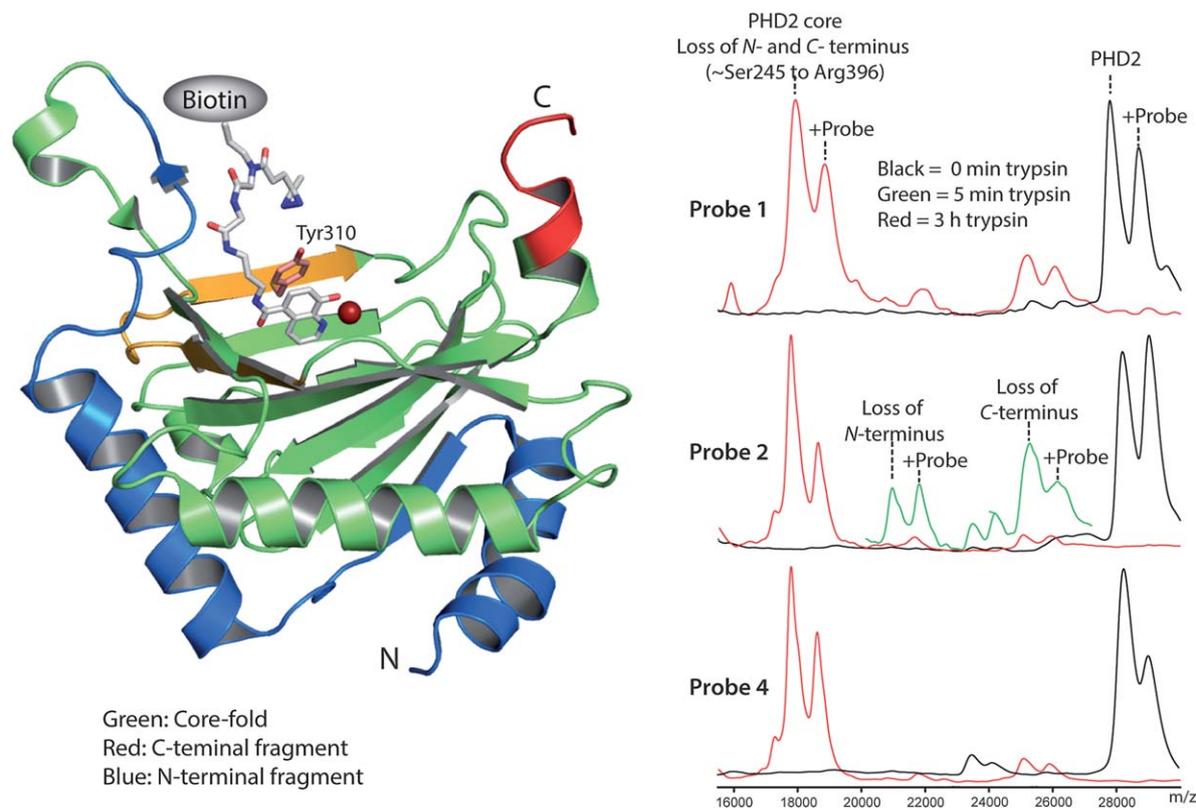


Fig. 3 Limited trypsinolysis of PHD2 identifies the region of crosslinking. Above: view derived from a crystal structure of PHD2 (PDB ID: 3HQJ)¹⁶ docked with diazirine **4** (grey) indicating the sites of photocrosslinking. The 'core-fold' (green), N-terminal (blue) and beginning of the C-terminal (red) region as defined by partial trypsinolysis are shown. The trypsin-derived peptide and tyrosine-residue identified as crosslinking to **4** by LCMS/MS in yellow and pink. Below: MALDI mass spectra derived from partial trypsinolysis of PHD2 crosslinked to probes **1**, **2** and **4**. Black line: before trypsin digest. Red line: 3 h trypsin digest. Green line: 5 min trypsin digest.

possibly due to further photoreactions. Interestingly, the commonly used aryl trifluoromethyl diazirine and benzophenone groups, probes **3** and **5**, resulted in relatively low levels of crosslinking at both wavelengths (3–12%). In the case of **3** this appeared to be due, at least in part, to competitive quenching of the carbene with water, as determined by MS (Fig. S3†).

The rate of activation of the probes at 310 nm was inferred from the rate of decay of the parent peak in the mass spectrum and found to increase in the order $4 < 1 < 3 < 2$ (Fig. S3†). The rate of activation of **5** could not be determined in this manner since both intramolecular reactions and reaction with water can afford products with the same mass as the parent compound, including *via* hemi-acetal/ketone equilibrium. Probe molecules that did not crosslink to enzyme were found to give a variety of products, depending on the nature of the photoreactive group. Irradiation of **3** predominantly formed a species with a mass corresponding to reaction with water, while **1** and **4** gave products with masses corresponding to intramolecular insertion of the nitrene/carbene. Azide **2** formed primarily a species with a mass that suggested reduction of the nitrene to the amine. These differences may arise from the relative reactivity of the activated species towards water/buffer *versus* intramolecular reactions, or from conformational constraints.

The crosslinking site(s) of probes **1**, **2** and **4** on PHD2 were investigated by partial trypsinolysis followed by matrix-assisted laser desorption/ionization (MALDI) MS. We have found that

trypsinolysis of the PHD2 catalytic domain results in rapid cleavage of N- (~7 kDa) and C- (~3 kDa) terminal PHD2 fragments to give a 'core-fold' (~Ser245–Arg396), which is digested more slowly (Fig. 3).²⁷ **1**, **2** and **4** (50 μ M) were irradiated (12 min, 310 nm, 4 $^{\circ}$ C) in the presence of PHD2 (20 μ M) and Mn^{II} (20 μ M), after which reaction mixtures were incubated with trypsin. Aliquots were removed at intervals, quenched with acid and analyzed (Fig. 3). The MALDI mass spectra of the undigested samples (black lines, Fig. 3) exhibited 2 peaks at ~28 kDa corresponding to unmodified and probe conjugated PHD2 respectively. After three hours of trypsin digestion the 2 peaks shifted to ~18 kDa (red lines, Fig. 3), corresponding to complete loss of N- and C-terminal fragments to give the modified and unmodified 'core-fold'. The ratio of the 2 peaks remained unchanged within the limit of detection for **1** and **4**, indicating crosslinking predominantly to the core-fold region. In the case of **2**, the ratio of crosslinking to the core-fold was lower than on the undigested enzyme, suggesting that crosslinking for **2** was divided over the core and the N- or C-terminal fragments. The MALDI mass spectrum for probe **2** at $t = 5$ min (green line, Fig. 3) exhibited 2 additional pairs of peaks at ~25 kDa and ~21 kDa corresponding to loss of the C- or N-terminal fragment, respectively. The decrease in the ratio of modified to unmodified PHD2 upon loss of the C-terminus, but not upon loss of the N-terminus, indicates that **2** crosslinks to both the core-fold and the C-terminus.



Finally, we performed MS/MS fragmentation analyses of extended trypsin digestions to further investigate the sites of photocrosslinking. Probes **1**, **2** and **4** were crosslinked to PHD2 under the conditions described for the MALDI studies followed by overnight trypsinolysis and MS/MS. No masses corresponding to trypsinolysed peptides plus probe were observed for **1** and **2**, possibly either due to poor stability of the crosslinked peptides under our MS conditions, or due to dilution of the crosslinking over a number of peptides. However, with compound **4**, crosslinked peptides were detected which consistently contained the peptide fragment ${}_{298}\text{AMVA-CYPGNGTGYVR}_{312}$ (Table S13[†]). MS/MS analysis afforded peaks corresponding to probe conjugated peptide for γ fragments from y_3 to y_{10} and peaks for unmodified fragments y_1 and y_2 (Fig. S4[†]), indicating crosslinking to tyrosine 310. Tyrosine residues have recently been shown to react with diazirine generated carbenes *via* insertion into both O–H and *ortho*-C–H bonds of the phenol ring.²⁸ Manual docking of the selectivity function of **1**–**5** with the active site of PHD2 (Fig. S5[†]), indicates that tyrosine 310, located at the edge of the PHD2 active site, is well situated for crosslinking (Fig. 3 and S5[†]). Collectively, the MS studies provide insights into the selectivity of photocrosslinking; notably they demonstrate that the use of different photoreactive groups can substantially alter the efficiency and site(s) of crosslinking.

Conclusions

In conclusion, the U-4CR provides a useful route for the synthesis of photoaffinity probes, enabling the expedient assembly of photoaffinity scaffolds with a variety of photoreactive groups, detection handles and inhibitor attachment points. The U-4CR was used for the synthesis of a set of photoaffinity probes for 2-OG oxygenases, containing different photoreactive groups. The photocrosslinking rates, yields and sites were investigated using PHD2 as a model system, revealing substantial differences between the probes. Aryl azides **1** and **2** and alkyl diazirine **4** were found to crosslink efficiently, while aryl trifluoromethyl diazirine **3** and benzophenone **5** gave low crosslinking yields. The relatively poor results observed for **3** and **5** contrast with other studies where these photoreactive groups have been reported to enable good crosslinking,^{14,15} suggesting that the optimum photoreactive group varies depending on the 'intrinsic' photochemical properties of the probe and the nature of its interaction with the target protein. The proximity and orientation of the photoreactive group to an appropriately functionalized group on the target, and the rate of reaction with enzyme *versus* quenching reactions, are likely to be important and context dependent factors in determining the crosslinking yield. The available evidence suggests that in most cases experimentally guided tailoring of the crosslinking reagent/protocol to particular targets is desirable.

Acknowledgements

JTB was supported by the EPSRC and Pfizer Ltd. LJW was supported by the BBSRC. BMK and JFM were supported by the

Biomedical Research Centre (NIHR), Oxford, UK. IKHL is supported by the British Heart Foundation (BHF). SSVB was supported by CRUK. We thank David Pryde and Rachel Grimley for helpful discussions.

References

- 1 P. P. Geurink, L. M. Prely, G. A. van der Marel, R. Bischoff and H. S. Overkleeft, *Top. Curr. Chem.*, 2012, **324**, 85–113.
- 2 D. J. Lapinsky, *Bioorg. Med. Chem.*, 2012, **20**, 6237–6247.
- 3 L. Dubinsky, B. P. Krom and M. M. Meijler, *Bioorg. Med. Chem.*, 2012, **20**, 554–570.
- 4 T. Lenz, J. J. Fischer and M. Dreger, *J. Proteomics*, 2011, **75**, 100–115.
- 5 D. K. Nomura, J. Z. Long, S. Niessen, H. S. Hoover, S. W. Ng and B. F. Cravatt, *Cell*, 2010, **140**, 49–61.
- 6 P. Ranjitkar, B. G. Perera, D. L. Swaney, S. B. Hari, E. T. Larson, R. Krishnamurthy, E. A. Merritt, J. Villen and D. J. Maly, *J. Am. Chem. Soc.*, 2012, **134**, 19017–19025.
- 7 H. Koster, D. P. Little, P. Luan, R. Muller, S. M. Siddiqi, S. Marappan and P. Yip, *Assay Drug Dev. Technol.*, 2007, **5**, 381–390.
- 8 D. Rotili, M. Altun, R. B. Hamed, C. Loenarz, A. Thalhammer, R. J. Hopkinson, Y. M. Tian, P. J. Ratcliffe, A. Mai, B. M. Kessler and C. J. Schofield, *Chem. Commun.*, 2011, **47**, 1488–1490.
- 9 D. Rotili, M. Altun, A. Kawamura, A. Wolf, R. Fischer, I. K. Leung, M. M. Mackeen, Y. M. Tian, P. J. Ratcliffe, A. Mai, B. M. Kessler and C. J. Schofield, *Chem. Biol.*, 2011, **18**, 642–654.
- 10 K. Liu, H. B. Shi, H. G. Xiao, A. G. L. Chong, X. Z. Bi, Y. T. Chang, K. S. W. Tan, R. Y. Yada and S. Q. Yao, *Angew. Chem., Int. Ed.*, 2009, **48**, 8293–8297.
- 11 C. M. Salisbury and B. F. Cravatt, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 1171–1176.
- 12 T. Kinoshita, A. Cano-Delgado, H. Seto, S. Hiranuma, S. Fujioka, S. Yoshida and J. Chory, *Nature*, 2005, **433**, 167–171.
- 13 J. J. Tate, J. Persinger and B. Bartholomew, *Nucleic Acids Res.*, 1998, **26**, 1421–1426.
- 14 C. Dalhoff, M. Huben, T. Lenz, P. Poot, E. Nordhoff, H. Koster and E. Weinhold, *ChemBioChem*, 2010, **11**, 256–265.
- 15 O. N. F. King, X. S. Li, M. Sakurai, A. Kawamura, N. R. Rose, S. S. Ng, A. M. Quinn, G. Rai, B. T. Mott, P. Beswick, R. J. Klose, U. Oppermann, A. Jadhav, T. D. Heightman, D. J. Maloney, C. J. Schofield and A. Simeonov, *PLoS One*, 2010, **5**, e15535.
- 16 R. Chowdhury, M. A. McDonough, J. Mecinovic, C. Loenarz, E. Flashman, K. S. Hewitson, C. Domene and C. J. Schofield, *Structure*, 2009, **17**, 981–989.
- 17 A. Domling and I. I. Ugi, *Angew. Chem., Int. Ed.*, 2000, **39**, 3168–3210.
- 18 J. Ge, X. Cheng, L. P. Tan and S. Q. Yao, *Chem. Commun.*, 2012, **48**, 4453–4455.
- 19 S. Brauch, M. Henze, B. Osswald, K. Naumann, L. A. Wessjohann, S. S. van Berkel and B. Westermann, *Org. Biomol. Chem.*, 2012, **10**, 958–965.



- 20 A. Domling, *Chem. Rev.*, 2006, **106**, 17–89.
- 21 M. Henze, O. Kreye, S. Brauch, C. Nitsche, K. Naumann, L. A. Wessjohann and B. Westermann, *Synthesis*, 2010, 2997–3003.
- 22 P. Sarathi Addy, B. Saha, N. D. Pradeep Singh, A. K. Das, J. T. Bush, C. Lejeune, C. J. Schofield and A. Basak, *Chem. Commun.*, 2013, **49**, 1930–1932.
- 23 J. Huang, H. Xiong, R. P. Hsung, C. Rameshkumar, J. A. Mulder and T. P. Grebe, *Org. Lett.*, 2002, **4**, 2417–2420.
- 24 J. Dommerholt, S. Schmidt, R. Temming, L. J. Hendriks, F. P. Rutjes, J. C. van Hest, D. J. Lefeber, P. Friedl and F. L. van Delft, *Angew. Chem., Int. Ed.*, 2010, **49**, 9422–9425.
- 25 R. J. Hopkinson, A. Tumber, C. Yapp, R. Chowdhury, W. Aik, K. H. Che, X. S. Li, J. B. L. Kristensen, O. N. F. King, K. K. Y. Mun Chiang Chan, H. Choi, L. J. Walport, C. C. Thinnis, J. T. Bush, C. Lejeune, A. M. Rydzik, N. R. Rose, E. A. Bagg, M. A. McDonough, T. J. Krojer, W. W. Yue, S. S. Ng, L. Olsen, P. E. Brennan, U. Oppermann, S. Müller, R. J. Klose, P. J. Ratcliffe, C. J. Schofield and A. Kawamura, *Chem. Sci.*, 2013, **4**, 3110–3117.
- 26 I. K. Leung, M. Demetriades, A. P. Hardy, C. Lejeune, T. J. Smart, A. Szollossi, A. Kawamura, C. J. Schofield and T. D. Claridge, *J. Med. Chem.*, 2013, **56**, 547–555.
- 27 C. J. Stubbs, C. Loenarz, J. Mecinovic, K. K. Yeoh, N. Hindley, B. M. Lienard, F. Sobott, C. J. Schofield and E. Flashman, *J. Med. Chem.*, 2009, **52**, 2799–2805.
- 28 B. Raimier and T. Lindel, *Chem.–Eur. J.*, 2013, 1521–3765.

