


Cite this: *RSC Adv.*, 2022, 12, 998

The interlayer coupling modulation of a g-C₃N₄/WTe₂ heterostructure for solar cell applications

Peng Lin,^a Nengshen Xu,^a Xiaolin Tan,^a Xuhui Yang,^{ID} *^b Rui Xiong,^a Cuilian Wen,^{ID} *^a Bo Wu,^a Qilang Lin^{*a} and Baisheng Sa^{ID} *^a

Constructing van der Waals (vdW) heterostructures has been proved to be an excellent strategy to design or modulate the physical and chemical properties of 2D materials. Here, we investigated the electronic structures and solar cell performances of the g-C₃N₄/WTe₂ heterostructure via first-principles calculations. It is highlighted that the g-C₃N₄/WTe₂ heterostructure presents a type-II band edge alignment with a band gap of 1.24 eV and a corresponding visible light absorption coefficient of $\sim 10^6$ cm⁻¹ scale. Interestingly, the band gap of the g-C₃N₄/WTe₂ heterostructure could increase to 1.44 eV by enlarging the vdW gap to harvest more visible light energy. It is worth noting that the decreased band alignment difference resulting from tuning the vdW gap, leads to a promotion of the power conversion efficiency up to 17.68%. This work may provide theoretical insights into g-C₃N₄/WTe₂ heterostructure-based next-generation solar cells, as well as a guide for tuning properties of vdW heterostructures.

Received 16th November 2021

Accepted 21st December 2021

DOI: 10.1039/d1ra08397j

rsc.li/rsc-advances

Introduction

From graphene, two-dimensional (2D) materials open a new gate to the material society and provide us with unprecedented insight to understanding and exploring materials.^{1,2} Generally speaking, 2D materials could show distinguished physical and chemical properties due to their giant specific surface areas.³ For example, as the first discovered two-dimensional material, graphene has been demonstrated to be an outstanding candidate in tremendous applications such as Li-ion batteries, supercapacitors, and beyond.⁴⁻⁶ So far, the applications of various typical 2D materials have been investigated, involved in MXene, graphene-based materials, transition metal oxides, and so on.⁷⁻¹⁰ Besides, 2D materials present high performance not only in energy storages but also in catalysts, thermoelectric devices, electronic devices, and optoelectronic devices.¹¹⁻¹⁴ Especially, many 2D semiconducting materials show dramatic light harvesting properties, inspiring global researchers to explore their applications in solar cells.¹⁵ Currently, the 2D transition metal dichalcogenides materials (TMDs) have been a research hotspot.^{16,17} TMDs are a class of materials with the formula MX₂, where M is a transition metal element, and X presents for S, Se, and Te. These materials form layered structures with the X-M-X stacking configuration, where the

chalcogens in two hexagonal planes are separated by a plane of transition metal atoms.¹⁸ The bulk TMDs have various properties ranging from insulators, semiconductors, semi-metals, and metals; meanwhile, their corresponding monolayers or few layers essentially preserve these properties.¹⁹ Multitudinous researches illustrated that TMDs could be a class of excellent materials in applications of photovoltaics and solar cells.²⁰ On the other hand, the g-C₃N₄ and its isomers have been widely explored aiming at solar energy converting because of their high surface activities and easily modulated surface chemistry by means of surface engineerings.^{21,22} Monolayer g-C₃N₄ presents a suitable band gap leading to its favorable absorption properties in the visible light spectrum.^{23,24} However, the high recombination rate of electrons and holes in these individual 2D materials limits their performance in photocatalysts and solar cells.^{25,26} Hence, promoting the efficiency of carrier separations in 2D materials is of great interest and importance.^{27,28}

Constructing van der Waals (vdW) heterostructures with different types of 2D materials stacking in a vertical direction has been proved an accessible approach to tune the properties and performance of 2D materials,²⁹⁻³² which have been proved to be one of the most efficient categories to enhance the performance of TMDs and g-C₃N₄. It is noted that heterostructure solar cells, considered as next-generation solar cell technology, have attracted great attention because of their fascinating properties in solar cell application.^{33,34} For example, compared to single-layer structures, the optical properties under visible-light irradiation of Blue_P/TMDs vdW heterostructures are significantly improved combined, which achieves higher efficiency in solar energy conversions.³⁵ Similarly, the g-

^aKey Laboratory of Eco-materials Advanced Technology, College of Materials Science and Engineering, Fuzhou University, Fuzhou, 350108, P. R. China. E-mail: linqilang@fzu.edu.cn; bssa@fzu.edu.cn

^bCollege of Environmental Science and Engineering, Fujian Key Laboratory of Pollution Control & Resource Reuse, Fujian Normal University, Fuzhou 350007 Fujian, P. R. China. E-mail: xhyang@fjnu.edu.cn



C_3N_4 based heterostructures have tunable electric properties, stronger optical properties as well as higher catalytic activity.^{36,37} Especially, $\text{g-C}_3\text{N}_4/\text{WTe}_2$ vdW heterostructure has been proved to be a potential electrocatalyst for hydrogen evolution reaction.³⁸ At the same time, challenges and opportunities for exploring advanced $\text{g-C}_3\text{N}_4$ based heterostructure are still ongoing.

In this work, we investigated the interlayer interactions, electronic structures, and optical properties of an artificial $\text{g-C}_3\text{N}_4/\text{WTe}_2$ vdW heterostructure. It is worth noting that vertical strains can modify the band gap and further result in a better light harvest with a light absorption coefficient up to $\sim 10^6 \text{ cm}^{-1}$ in the process. The decreased band alignment difference caused by the increased vdW gap gives rise to the promotion of power conversion efficiency are unraveled. Our findings provide significant guidance to design and modulate the performance of 2D materials applied in next-generation optoelectronic devices.

Computational methods

In our work, we adopted the ALKEMIE platform³⁹ together with the Vienna *ab initio* simulation package (VASP) based on density functional theory (DFT) to perform the first-principles calculations.⁴⁰ The projection-augmented wave (PAW) exchange and correlation effects potential was used in the term of generalized gradient approximation (GGA) Perdew–Burke–Ernzerhof (PBE).^{41–43} We introduced the DFT-D3 method⁴⁴ to correct the vdW interactions. A vacuum space of 20 Å along the z-direction was built to avoid periodic interactions. Energy cutoff of 500 eV was set, and $8 \times 8 \times 1$ Γ -centered k -mesh was used for Brillouin zone (BZ) integrations. To overcome the underestimation of the band gap by the standard semilocal DFT functionals, we introduced the Heyd–Scuseria–Ernzerhof (HSE06) function⁴⁵ for the electronic structure calculations. The relaxation convergence for electrons and ions were 1×10^{-6} eV and 1×10^{-5} eV, respectively. To obtain accurate dielectric functions comparable to the experimental results, time-dependent Hartree–Fock calculation (TDHF) was introduced to calculate the response functions by including the excitonic effects based on the HSE06 wavefunctions.

Results and discussion

Geometry and electronic structure

Firstly, we analyzed the geometry and electronic structures of monolayer $\text{g-C}_3\text{N}_4$ and WTe_2 . As shown in Fig. 1(a) and (b), $\text{g-C}_3\text{N}_4$ consists of N and C atoms in a staggered fashion similar to graphene with the optimized constant lattice of 6.95 Å, while monolayer WTe_2 shows 2H phase with the optimized constant lattice of 3.52 Å, which agree well with previous works.^{46,47} We, therefore, built a $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure by stacking a $2 \times 2 \times 1$ supercell of WTe_2 upon the unit cell of $\text{g-C}_3\text{N}_4$ together with a lattice constant mismatch of 1.3%. Furthermore, we considered 6 possible stacking configurations by shifting $\text{g-C}_3\text{N}_4$ in a certain direction to explore the energetically favorable

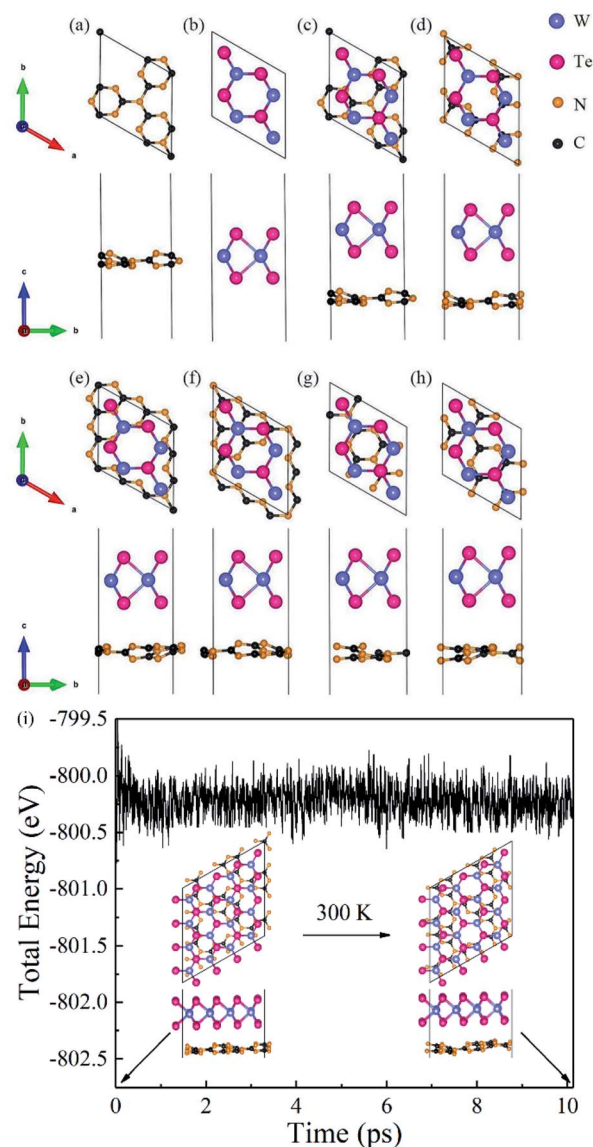


Fig. 1 Top and side views for the structure of (a) $\text{g-C}_3\text{N}_4$ monolayer, (b) WTe_2 monolayer, and (c–h) the six different stacking configurations of heterostructure. (i) The total energy changes and snapshots from AIMD calculations at 0 and 10 ps of $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure.

structure of the heterostructure, as illustrated in Fig. 1(c–h). Herein, the formation energy E_{form} was defined as

$$E_{\text{form}} = E_{\text{heterostructure}}^{\text{total}} - E_{\text{g-C}_3\text{N}_4}^{\text{free}} - E_{\text{WTe}_2}^{\text{free}} \quad (1)$$

where $E_{\text{heterostructure}}^{\text{total}}$, $E_{\text{g-C}_3\text{N}_4}^{\text{free}}$ and $E_{\text{WTe}_2}^{\text{free}}$ are the total energy of the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure, freestanding $\text{g-C}_3\text{N}_4$ and WTe_2 monolayer, respectively. On the other hand, the vdW binding energy E_b was defined as⁴⁸

$$E_b = -\frac{E_{\text{heterostructure}}^{\text{total}} - E_{\text{g-C}_3\text{N}_4 + \text{WTe}_2}}{A} \quad (2)$$

where A is the interface area of a heterostructure unit cell, $E_{\text{g-C}_3\text{N}_4 + \text{WTe}_2}$ is the sum of the total energies of the mutually independent $\text{g-C}_3\text{N}_4$ and WTe_2 monolayers fixed in the



Table 1 The calculated lattice constants a , the vdW gap d_{layer} , formation energy E_{form} and binding energy E_{b} of g-C₃N₄/WTe₂ heterostructure with possible stacking configurations

Configurations	I	II	III	IV	V	VI
a (Å)	6.993	7.098	6.989	6.988	6.993	6.986
d_{layer} (Å)	3.076	3.653	3.312	3.111	3.177	3.325
E_{f} (eV)	−0.652	−0.117	−0.645	−0.640	−0.604	−0.634
E_{b} (meV Å ^{−2})	15.81	15.33	15.79	14.35	13.46	15.07

corresponding heterostructure lattice, respectively. The optimized lattice constant a , the calculated vdW gap d_{layer} , formation energy E_{form} and binding energy E_{b} are listed in Table 1. It is interesting to note that the values of E_{form} for all these 6 configurations are negative, indicating these heterostructures are energetic favorable. In addition, the calculated E_{b} between the g-C₃N₄ and WTe₂ monolayers is around 15 meV Å^{−2}, which is close to the typical vdW binding energy.^{49,50} Therefore, the g-C₃N₄/WTe₂ heterostructure can be defined as a vdW heterostructure. We chose configuration-I as the object to study in the subsequent work since stacking configuration-I exhibits the most favorable E_{form} and smallest d_{layer} .

To prove the thermodynamically stability, Born–Oppenheimer *ab initio* molecular dynamics (AIMD) simulations were adopted for the proposed g-C₃N₄/WTe₂ heterostructure at 300 K for 10 ps. A 2×2 supercell has been constructed for the AIMD calculations. Fig. 1(i) displays the energy evolution and structure snapshots after 300 K annealing for 10 ps of the g-C₃N₄/WTe₂ heterostructures. It is noted that the structure snapshots suggest that atoms just move near their equilibrium location during the simulations, and there is no structural reconstruction at 300 K. At the same time, the changes of the total energy are very small during the simulations from Fig. 1(i), indicating that the proposed g-C₃N₄/WTe₂ vdW heterostructure is thermodynamically stability at 300 K.

Fig. 2(a) shows the band structures of freestanding g-C₃N₄ and WTe₂ monolayers using HSE06 calculations. To compare clearly, the vacuum level was set to 0 eV as a baseline. It can be found that g-C₃N₄ has an indirect band gap of 3.21 eV, where CBM and VBM locate at the K (1/3, 1/3, 0) and Γ (0, 0, 0) point, respectively. Meanwhile, the WTe₂ shows the direct gap feature with the band gap of 1.60 eV, where both CBM and VBM locate at the K (1/3, 1/3, 0) point. These results agree well with the previously published studies.^{46,47} On the other hand, the projected band structure and partial density of states of g-C₃N₄/WTe₂ heterostructure is plotted in Fig. 2(b), in which the projected weight of g-C₃N₄ and WTe₂ are distinguished by size and color. The pink and blue balls represent the contributions from g-C₃N₄ and WTe₂, respectively. For g-C₃N₄/WTe₂ heterostructure, both CBM and VBM locate at the K (1/3, 1/3, 0) point, showing the direct band gap feature, with the calculated HSE06 band gap of 1.24 eV. Interestingly, the g-C₃N₄/WTe₂ heterostructure shows the band structure feature of a type-II heterostructure,⁵¹ where CBM is contributed by the g-C₃N₄ layer and VBM is occupied by the WTe₂ layer. The band alignment diagrams for isolated g-C₃N₄, WTe₂ monolayer, and

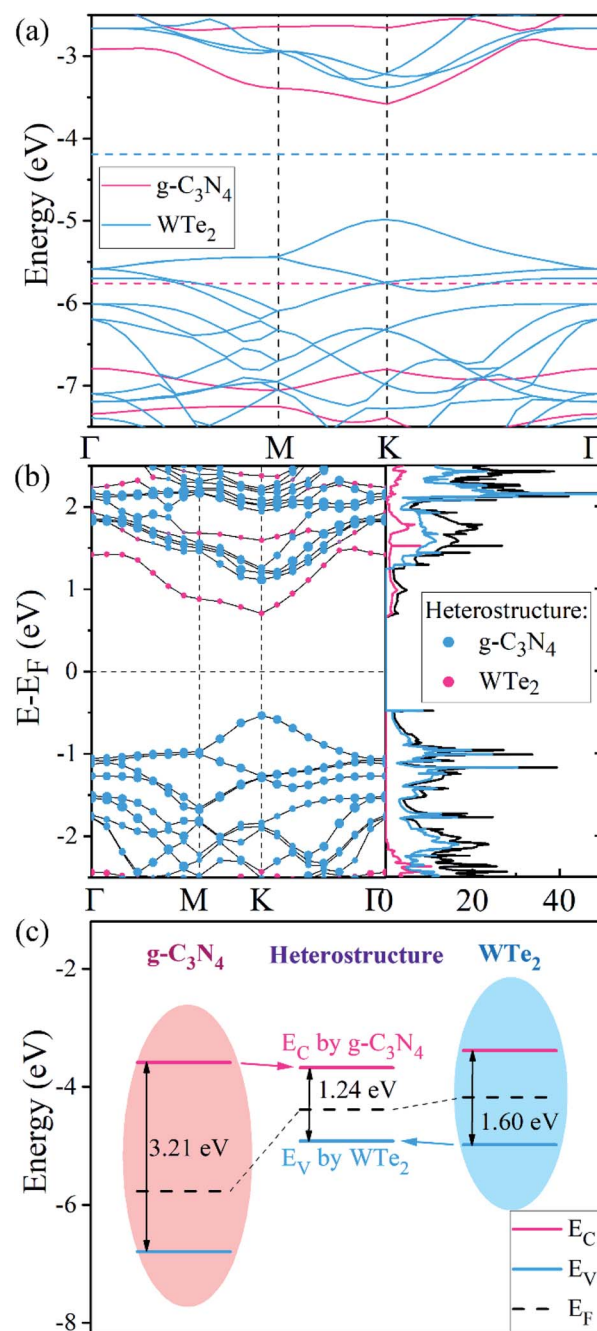


Fig. 2 (a) The HSE06 band structure of free-standing g-C₃N₄ and WTe₂ monolayer, respectively. (b) The projected HSE06 band structure and partial density of states of g-C₃N₄/WTe₂ heterostructure. The size of pink and blue balls present the contributions from g-C₃N₄ and WTe₂, respectively. (c) The band alignment diagrams for isolated g-C₃N₄, WTe₂ monolayer and heterostructure interface.

heterostructure interface are illustrated in Fig. 2(c). Obviously, the work function of the g-C₃N₄/WTe₂ heterostructure lies between the g-C₃N₄ and WTe₂ monolayers. When g-C₃N₄ and WTe₂ come into contact, the electrons flow from WTe₂ to g-C₃N₄ due to the lower work function of WTe₂ and *vice versa* for the holes. As a result of the increased transfer of electrons, the Fermi level shifts and finally reaches the same energy level. The

differences between the band structure of g-C₃N₄/WTe₂ heterostructure and corresponding monolayers indicate that the vdW interactions play an essential role in the electronic structures.

To understand the vdW interlayer interaction between the different parts of the heterostructure, we further investigated g-C₃N₄/WTe₂ heterostructure with different interlayer distance d_{layer} of the vdW gap. As shown in Fig. 3, both E_{form} and $-E_{\text{b}}$ follow the Lenard-Jones type relation as a function of d_{layer} ,⁵² and a lower value of $-E_{\text{b}}$ correspond to a stronger binding. Clearly, g-C₃N₄/WTe₂ heterostructure with the equilibrium d_{layer} holds the most negative E_{form} and $-E_{\text{b}}$. As the d_{layer} decreases, both E_{form} and $-E_{\text{b}}$ increase dramatically. As the d_{layer} increases, E_{form} and $-E_{\text{b}}$ gradually increases towards zero. Herein, E_{form} and $-E_{\text{b}}$ remain negative among an extensive range of d_{layer} , indicating the possibility to tune the interlayer interaction by varying d_{layer} . As mentioned before, there is the transfer of electrons within the vdW gap, which affects the electronic structure of the g-C₃N₄/WTe₂ heterostructure. Thereby, we calculated the planar-averaged charge density differences of g-C₃N₄/WTe₂ heterostructure with different d_{layer} , as shown in Fig. 4. Here, the plane-averaged electron density difference $\Delta\rho$ was calculated by

$$\Delta\rho = \rho_{\text{g-C}_3\text{N}_4/\text{WTe}_2} - \rho_{\text{g-C}_3\text{N}_4} - \rho_{\text{WTe}_2} \quad (3)$$

where $\rho_{\text{g-C}_3\text{N}_4/\text{WTe}_2}$ is the charge density of the heterostructure, $\rho_{\text{g-C}_3\text{N}_4}$ and ρ_{WTe_2} are charge densities of the g-C₃N₄ and WTe₂ parts in the heterostructure, respectively. The positive and negative values denote charge accumulation and depletion in the combined system comparing with the two isolated monolayers, respectively. Fig. 4 clear presents the charge redistribution in the vdW gap of g-C₃N₄/WTe₂ heterostructure: the charge depletion around the g-C₃N₄ part and the charge accumulation around the WTe₂ region, indicating the charge transfer from g-C₃N₄ to WTe₂. As the d_{layer} decreases, the stronger interlayer interaction results in the more obvious charge transfer. Oppositely, the charge transfer weakens when d_{layer} increases. The similar shape of $\Delta\rho$ for g-

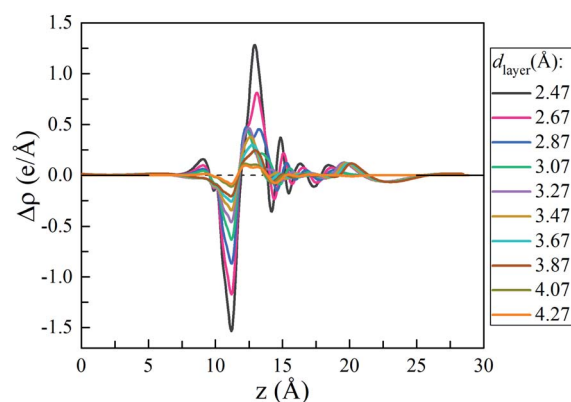


Fig. 4 Plane-averaged charge density difference as a function of vdW gap.

C₃N₄/WTe₂ heterostructure with different d_{layer} indicates the excellent stability of the heterostructure from the electronic structure point of view. This phenomenon suggests a possible method to tune the band structure of the g-C₃N₄/WTe₂ heterostructure by modifying the interlayer interaction.

To further explore the influence of the vdW interactions on the electronic structures of the g-C₃N₄/WTe₂ heterostructure, we plotted the band gap, band alignment, and work function of g-

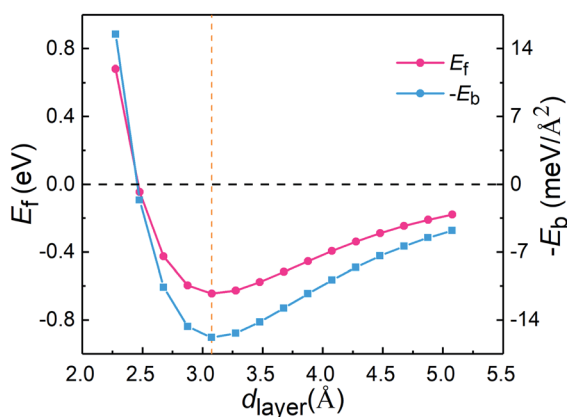


Fig. 3 Formation energy E_{form} and binding energy $-E_{\text{b}}$ as a function of interlayer distance d_{layer} of the vdW gap.

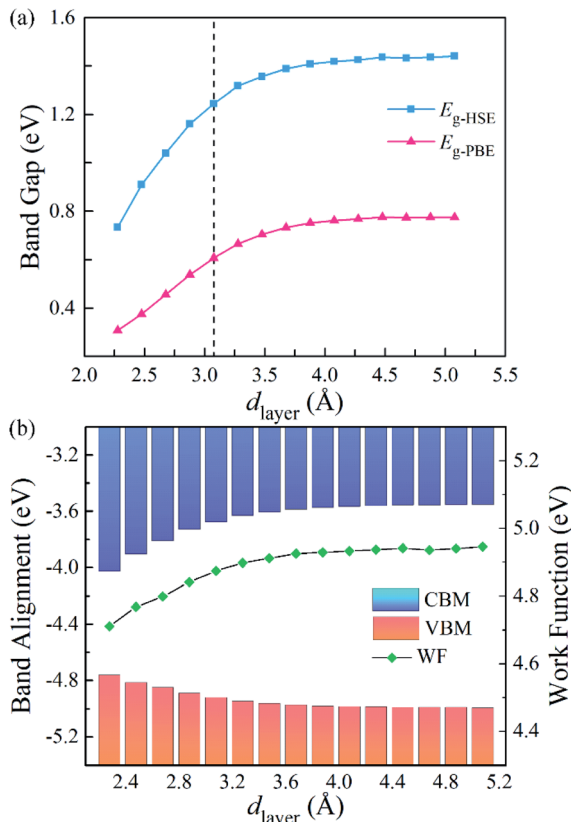


Fig. 5 (a) Band gap, (b) band edge alignments and work function of g-C₃N₄/WTe₂ heterostructure as a function of vdW gap.



$\text{C}_3\text{N}_4/\text{WTe}_2$ heterostructure with different d_{layer} in Fig. 5(a). The PBE and HSE06 results show similar trends that the band gap decreases continuously as d_{layer} decreases. Oppositely, as d_{layer} increases, the band gap increases towards a balance value of 1.44 eV (HSE06). In addition, since the band alignment and work function are crucial in semiconductor heterostructure-based functional device designs, we plotted the band alignment and work function of the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure corresponding to the vacuum level, as shown in Fig. 5(b). Correspondingly, the band alignment and work function show similar trends of band gap with different d_{layer} . As the d_{layer} decreases, CBM shifts downward continuously, and VBM shifts upward continuously, which reduces the band gap. On the contrary, as d_{layer} increases, CBM and VBM shift oppositely and towards convergent.

To explore the solar light-harvesting ability of the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure, we calculated the optical absorption coefficients with a series of d_{layer} . As presented in Fig. 6, there are three absorption peaks in the visible light region for the equilibrium vdW gap $d_{\text{layer}} = 3.07 \text{ \AA}$. The first absorption peak locates at $\sim 1.9 \text{ eV}$, and the main peak covers the light energy region of 2.25–2.6 eV with an ultra-high light absorption coefficient up to $1.22 \times 10^6 \text{ cm}^{-1}$. And the third absorption peak locating at $\sim 2.8 \text{ eV}$ presents the absorption coefficient of about $\sim 1 \times 10^6 \text{ cm}^{-1}$. It is worth noting that the light-harvesting ability in the entire visible solar spectrum is elevated when the d_{layer} increases. Interestingly, the absorption peaks shift weakly towards the lower energy region as the d_{layer} rises, and the absorption coefficient increases the maximum value up to $1.34 \times 10^6 \text{ cm}^{-1}$ when $d_{\text{layer}} = 3.47 \text{ \AA}$. Due to the direct band gap feature being beneficial for separating photo-excited electron-hole pairs and strong light absorption, the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure could be a promising material for efficient photovoltaic solar cells and optoelectronic devices.

Furthermore, we estimated the power conversion efficiency (PCE) by the method proposed by Scharber *et al.*,⁵³ which is widely used in efficiency estimation. The upper limited PCE of the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure is described by^{54,55}

$$\eta = \frac{0.65 \left(E_{\text{g}}^{\text{opt,d}} - \Delta E_{\text{c}} - 0.3 \right) \int_{E_{\text{g}}^{\text{opt,d}}}^{\infty} \frac{J_{\text{ph}}(\hbar\omega)}{\hbar\omega} d(\hbar\omega)}{\int_0^{\infty} \frac{J_{\text{ph}}(\hbar\omega)}{\hbar\omega} d(\hbar\omega)} \quad (4)$$

where 0.65 is the fill factor (β_{FF}), $P(\hbar\omega)$ is the AM1.5 solar energy flux at the photon energy $\hbar\omega$, E_{g} and ΔE_{c} are the band gaps of the donor and conduction band offset between donor and acceptor respectively. The $(E_{\text{g}}^{\text{opt,d}} - \Delta E_{\text{c}} - 0.3)$ term is an estimation of the open-circuit voltage (V_{oc}). The integral term in the numerator is the short-circuit current density (J_{sc}) assuming external quantum efficiency to be 100%, while the energy integral from 0 to infinity in the denominator is the power of incident solar radiation. Fig. 7(a) illustrates the donor band gap $\text{Gap}_{\text{donor}}$ and conduction band offset ΔE_{c} , which are critical to the maximum PCE, as well as simulated PCE of heterostructures with different d_{layer} . Interestingly, due to a suitable band gap of about 1.4 eV with a $\text{Gap}_{\text{donor}}$ of about 1.65 eV, the $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure shows an excellent solar spectrum absorption. Furthermore, the $\text{Gap}_{\text{donor}}$ hardly changes, but the ΔE_{c} decreases about 70% in the process of compression and stretching. The reduced band offset differences in the stretching process lead to a higher PCE. Dramatically, the PCE improves considerably with a maximum value of 17.68% for the

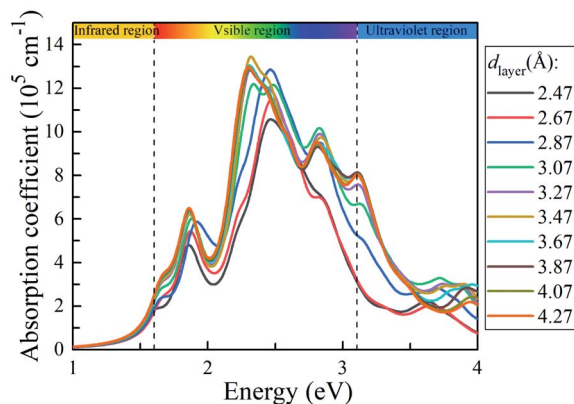


Fig. 6 Absorption coefficient of $\text{g-C}_3\text{N}_4/\text{WTe}_2$ with different vdW gap.

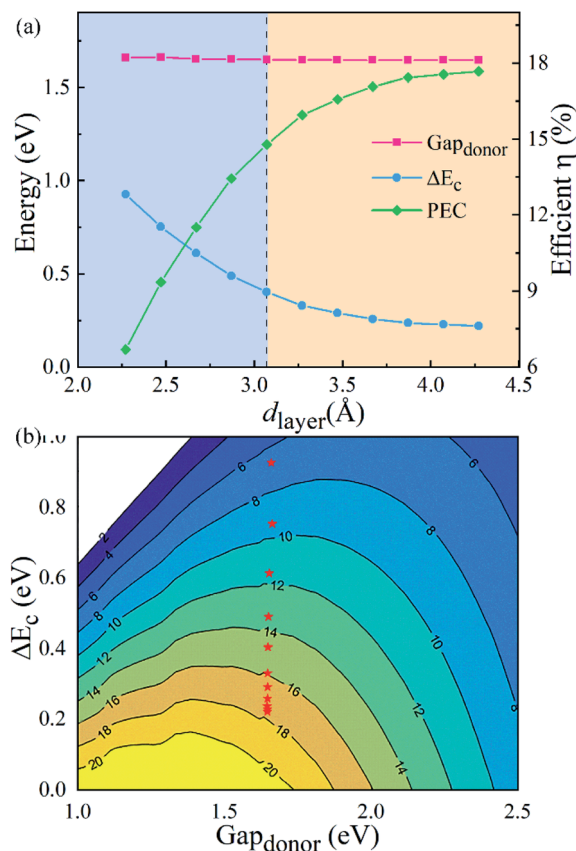


Fig. 7 (a) The donor band gap $\text{Gap}_{\text{donor}}$, conduction band offset ΔE_{c} and PCE of $\text{g-C}_3\text{N}_4/\text{WTe}_2$ heterostructure as a function of vdW gap. (b) Contour plots for PCE as a function of the donor band gap and conduction band offset ΔE_{c} .

g-C₃N₄/WTe₂ heterostructure. Fig. 7(b) depicts PCE variation with the Gap_{donor} and ΔE_c. Therefore, we concluded that the g-C₃N₄/WTe₂ heterostructure could show a better performance in solar cell applications by modifying the vdW gap.

Conclusion

To conclude, based on the first-principle calculations, we have constructed the g-C₃N₄/WTe₂ heterostructure and systematically analyzed the corresponding electronic band structure, optical properties with different *d*_{layer}. As the *d*_{layer} increases, the band gap rises from 1.24 to 1.44 eV when the interlayer interactions become weaker, which brings an augmented light harvest in the visible range. Significantly, the maximum optical absorption coefficient can reach ~10⁶ cm⁻¹ level. Furthermore, the larger band gap and smaller band alignment difference make it better for light absorption and energy conversion. Finally, we found that the PCE of g-C₃N₄/WTe₂ heterostructure has been promoted obviously during vdW gap tuning. The optimized PCE can reach up to 17.68%. Our results show that the g-C₃N₄/WTe₂ heterostructure is favorable in solar cell applications. Here, we gave a tasteful way to realize the better performances of heterostructures, which is vital in the future study of vdW heterostructures.

Conflicts of interest

The authors declare no competing financial interest.

Acknowledgements

This work was supported by the National Natural Science Foundation of China (No. 21973012 and 51872049), the Natural Science Foundation of Fujian Province (Grant No. 2021J06011, 2020J01351, and 2020J01474), the Fujian Provincial Department of Science & Technology (2021H6011), and the "Qishan Scholar" Scientific Research Project of Fuzhou University.

References

- 1 K. S. Novoselov, A. K. Geim, S. V. Morozov, D. Jiang, Y. Zhang, S. V. Dubonos, I. V. Grigorieva and A. A. Firsov, *Science*, 2004, **306**, 666–669.
- 2 C. Tan, X. Cao, X. J. Wu, Q. He, J. Yang, X. Zhang, J. Chen, W. Zhao, S. Han, G. H. Nam, M. Sindoro and H. Zhang, *Chem. Rev.*, 2017, **117**, 6225–6331.
- 3 C. Chang, W. Chen, Y. Chen, Y. Chen, Y. Chen, F. Ding, C. Fan, H. Jin Fan, Z. Fan, C. Gong, Y. Gong, Q. He, X. Hong, S. Hu, W. Hu, W. Huang, Y. Huang, W. Ji, D. Li, L.-J. Li, Q. Li, L. Lin, C. Ling, M. Liu, N. Liu, Z. Liu, K. Ping Loh, J. Ma, F. Miao, H. Peng, M. Shao, L. Song, S. Su, S. Sun, C. Tan, Z. Tang, D. Wang, H. Wang, J. Wang, X. Wang, X. Wang, A. T. S. Wee, Z. Wei, Y. Wu, Z.-S. Wu, J. Xiong, Q. Xiong, W. Xu, P. Yin, H. Zeng, Z. Zeng, T. Zhai, H. Zhang, H. Zhang, Q. Zhang, T. Zhang, X. Zhang, L.-D. Zhao, M. Zhao, W. Zhao, Y. Zhao, K.-G. Zhou, X. Zhou, Y. Zhou, H. Zhu, H. Zhang and Z. Liu, *Acta Phys.-Chim. Sin.*, 2021, **37**, 2108017.
- 4 X. Li and L. Zhi, *Chem. Soc. Rev.*, 2018, **47**, 3189–3216.
- 5 I. V. Lightcap and P. V. Kamat, *Acc. Chem. Res.*, 2012, **46**, 2235–2243.
- 6 Y. T. Du, X. Kan, F. Yang, L. Y. Gan and U. Schwingenschlogl, *ACS Appl. Mater. Interfaces*, 2018, **10**, 32867–32873.
- 7 B. Anasori, M. R. Lukatskaya and Y. Gogotsi, *Nat. Rev. Mater.*, 2017, **2**, 16098.
- 8 R. Li, Y. Cheng and W. Huang, *Small*, 2018, **14**, 1802091.
- 9 Y. Wang, Y. Xu, M. Hu, H. Ling and X. Zhu, *Nanophotonics*, 2020, **9**, 1601–1620.
- 10 C. Liu, H. Chen, S. Wang, Q. Liu, Y. G. Jiang, D. W. Zhang, M. Liu and P. Zhou, *Nat. Nanotechnol.*, 2020, **15**, 545–557.
- 11 J. Chen, S. Cai, R. Xiong, B. Sa, C. Wen, B. Wu and Z. Sun, *Phys. Chem. Chem. Phys.*, 2020, **22**, 7039–7047.
- 12 Y. Zhu, L. Gong, D. Zhang, X. Wang, J. Zhang, L. Zhang, L. Dai and Z. Xia, *Nano Energy*, 2019, **63**, 103819.
- 13 G. P. Gao, A. P. O'Mullane and A. J. Du, *ACS Catal.*, 2017, **7**, 494–500.
- 14 Q. Peng, Z. Wang, B. Sa, B. Wu and Z. Sun, *ACS Appl. Mater. Interfaces*, 2016, **8**, 13449–13457.
- 15 S. J. Kang, D. H. Lee, J. Kim, A. Capasso, H. S. Kang, J. W. Park, C. H. Lee and G. H. Lee, *2D Materials*, 2020, **7**, 022003.
- 16 L. Ju, M. Bie, X. Tang, J. Shang and L. Kou, *ACS Appl. Mater. Interfaces*, 2020, **12**, 29335–29343.
- 17 S. Arra, R. Babar and M. Kabir, *Phys. Rev. Mater.*, 2019, **3**, 095402.
- 18 X. Huang, Z. Zeng and H. Zhang, *Chem. Soc. Rev.*, 2013, **42**, 1934–1946.
- 19 S. Manzeli, D. Ovchinnikov, D. Pasquier, O. V. Yazyev and A. Kis, *Nat. Rev. Mater.*, 2017, **2**, 17033.
- 20 C. K. Sumesh, *Sol. Energy Mater. Sol. Cells*, 2019, **192**, 16–23.
- 21 J. Fu, J. Yu, C. Jiang and B. Cheng, *Adv. Energy Mater.*, 2018, **8**, 1701503.
- 22 G. Mamba and A. K. Mishra, *Appl. Catal., B*, 2016, **198**, 347–377.
- 23 S. Cao and J. Yu, *J. Phys. Chem. Lett.*, 2014, **5**, 2101–2107.
- 24 J. Wen, J. Xie, X. Chen and X. Li, *Appl. Surf. Sci.*, 2017, **391**, 72–123.
- 25 M. D. Hernandez-Alonso, F. Fresno, S. Suarez and J. M. Coronado, *Energy Environ. Sci.*, 2009, **2**, 1231–1257.
- 26 Y. H. Li, M. L. Gu, X. M. Zhang, J. J. Fan, K. L. Lv, S. A. C. Carabineiro and F. Dong, *Mater. Today*, 2020, **41**, 270–303.
- 27 P. Niu, L. L. Zhang, G. Liu and H. M. Cheng, *Adv. Funct. Mater.*, 2012, **22**, 4763–4770.
- 28 X. H. Niu, Y. H. Li, Y. H. Zhang, Q. J. Zheng, J. Zhao and J. L. Wang, *J. Mater. Chem. C*, 2019, **7**, 1864–1870.
- 29 A. K. Geim and I. V. Grigorieva, *Nature*, 2013, **499**, 419–425.
- 30 Q. Wang, J. Li, Y. Liang, B. Wang and Y. Nie, *J. Mater. Chem. A*, 2019, **7**, 10684–10695.
- 31 X. H. Yang, B. S. Sa, P. Lin, C. Xu, Q. Zhu, H. B. Zhan and Z. M. Sun, *J. Phys. Chem. C*, 2020, **124**, 23699–23706.
- 32 S. Y. Xie, H. Jin, J. W. Li and Y. D. Wei, *ACS Appl. Electron. Mater.*, 2021, **3**, 898–904.



- 33 M. M. Furchi, A. Pospischil, F. Libisch, J. Burgdorfer and T. Mueller, *Nano Lett.*, 2014, **14**, 4785–4791.
- 34 C. Li, Q. Cao, F. Wang, Y. Xiao, Y. Li, J. J. Delaunay and H. Zhu, *Chem. Soc. Rev.*, 2018, **47**, 4981–5037.
- 35 Q. Peng, Z. Wang, B. Sa, B. Wu and Z. Sun, *Sci. Rep.*, 2016, **6**, 31994.
- 36 L. Xu, J. Zeng, Q. Li, T. Chen, K.-W. Luo, X. Luo, B. Peng, Z. Ma, L.-L. Wang, X. Zhu, S. Huang, D. Liu, S. X. Xiong and C. Shuai, *Appl. Surf. Sci.*, 2020, **530**, 147181.
- 37 B. Li, H. Song, F. Han and L. Wei, *Appl. Catal., B*, 2020, **269**, 118845.
- 38 J. L. Silva, B. Brena and C. M. Araujo, *J. Phys. Chem. C*, 2020, **124**, 8726–8735.
- 39 G. J. Wang, L. Y. Peng, K. Q. Li, L. G. Zhu, J. Zhou, N. H. Miao and Z. M. Sun, *Comput. Mater. Sci.*, 2021, **186**, 110064.
- 40 J. Hafner, *J. Comput. Chem.*, 2008, **29**, 2044–2078.
- 41 J. P. Perdew and Y. Wang, *Phys. Rev. B*, 1992, **45**, 13244–13249.
- 42 G. Kresse and J. Hafner, *Phys. Rev. B*, 1993, **48**, 13115–13118.
- 43 J. P. Perdew, K. Burke and Y. Wang, *Phys. Rev. B*, 1996, **54**, 16533–16539.
- 44 S. Grimme, J. Antony, S. Ehrlich and H. Krieg, *J. Chem. Phys.*, 2010, **132**, 154104.
- 45 J. Paier, M. Marsman, K. Hummer, G. Kresse, I. C. Gerber and J. G. Angyan, *J. Chem. Phys.*, 2006, **124**, 154709.
- 46 H. H. Huang, X. Fan, D. J. Singh, H. Chen, Q. Jiang and W. T. Zheng, *Phys. Chem. Chem. Phys.*, 2016, **18**, 4086–4094.
- 47 Q. Gao, S. Hu, Y. Du and Z. Hu, *J. Mater. Chem. A*, 2017, **5**, 4827–4834.
- 48 Z. Ma, R. Li, R. Xiong, Y. Zhang, C. Xu, C. Wen and B. Sa, *Materials*, 2021, **14**, 3768.
- 49 T. Bjorkman, A. Gulans, A. V. Krasheninnikov and R. M. Nieminen, *Phys. Rev. Lett.*, 2012, **108**, 235502.
- 50 J. Chen, X. He, B. Sa, J. Zhou, C. Xu, C. Wen and Z. Sun, *Nanoscale*, 2019, **11**, 6431–6444.
- 51 R. Xiong, R. Hu, Y. Zhang, X. Yang, P. Lin, C. Wen, B. Sa and Z. Sun, *Phys. Chem. Chem. Phys.*, 2021, **23**, 20163–20173.
- 52 B. Sa, N. Miao, J. Zhou, Z. Sun and R. Ahuja, *Phys. Chem. Chem. Phys.*, 2010, **12**, 1585–1588.
- 53 M. C. Scharber, D. Mühlbacher, M. Koppe, P. Denk, C. Waldauf, A. J. Heeger and C. J. Brabec, *Adv. Mater.*, 2006, **18**, 789–794.
- 54 M. Bernardi, M. Palummo and J. C. Grossman, *ACS Nano*, 2012, **6**, 10082–10089.
- 55 J. Dai and X. C. Zeng, *J. Phys. Chem. Lett.*, 2014, **5**, 1289–1293.

