Natural products as an inspiration in the diversity-oriented synthesis of bioactive compound libraries

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The purpose of diversity-oriented synthesis is to drive the discovery of small molecules with previously unknown biological functions. Natural products necessarily populate biologically relevant chemical space, since they bind both their biosynthetic enzymes and their target macromolecules. Natural product families are, therefore, libraries of pre-validated, functionally diverse structures in which individual compounds selectively modulate unrelated macromolecular targets. This review describes examples of diversity-oriented syntheses which have, to some extent, been inspired by the structures of natural products. Particular emphasis is placed on innovations that allow the synthesis of compound libraries that, like natural products, are skeletally diverse. Mimicking the broad structural features of natural products may allow the discovery of compounds that modulate the functions of macromolecules for which ligands are not known. The ability of innovations in diversity-oriented synthesis to deliver such compounds is critically assessed.

1 Introduction

The interrogation of fundamental biological mechanisms is an immense challenge for the next decade and beyond. The functions of all proteins identified from the genome must be determined, the *in vivo* relevance of post-translation states and interaction targets recognised, and the effects on cellular physiology established. The field of chemical genetics is based on the premise that small molecules may be used to perturb the functions of specific biological macromolecules, thus revealing insights into their roles in complex biological mechanisms.[1] Chemical genetics, and its

impact on our understanding of fundamental biological mechanisms, has been comprehensively reviewed.[1a]

The use of small molecule probes in chemical genetic studies confers a number of advantages over conventional genetic manipulations which may be used to vary the structure, and hence the function, of encoded protein(s). First, the effects of small molecules are usually rapid, reversible and conditional. Second, by varying the concentration of the small molecule tool(s), the activity of a target macromolecule may be tuned in a refined way. Third, the chemical genetic approach can sometimes allow the many functions of individual proteins to be teased apart.[2] Finally, by using more than one small molecule effector in combination, the interplay between proteins may be studied.[3] The approach is particularly useful for investigating tightly coordinated, dynamic biological mechanisms, especially when classical genetics may render the organism inviable.[4]

1.1 Diversity-oriented synthesis

A key challenge in chemical genetics is the design and synthesis of libraries which span large tracts of biologically relevant chemical space.[5] This challenge has spawned a new field in synthetic chemistry – diversity-oriented synthesis[6] (DOS). The synthetic challenges of DOS are very different to target-oriented synthesis, where tailored methods may be developed to solve the synthetic problems raised. DOS methods must be sufficiently robust to allow diverse compounds to be prepared simultaneously, deliberately and combinatorially. In general, the assembly of diverse small molecule libraries must be accomplished in up to *ca.* five highly reliable synthetic steps, leaving little or no scope for protecting group chemistry.

Chemical libraries prepared using diversity-oriented synthetic approaches can often be used to identify new lead molecules.[1] In contrast, combinatorial chemistry is used to prepare much more

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focussed libraries which target narrower regions of chemical space. Indeed, such libraries are widely used in the pharmaceutical and agrochemical industries to optimise the structure, and hence the activity, of a particular ligand.

The diversity of chemical libraries may derive from the high substitutional†, stereochemical and skeletal diversity of its members. Combinatorial chemistry is focussed on the synthesis of chemical libraries with only high substitutional diversity. In general, combinatorial syntheses exploit a rather limited palette of chemical transformations to vary the substitution of library

† Substitutional diversity has also been described as 'appendage diversity'.

members: for example, amide and sulfonamide formation, the Mitsunobu reaction, and Pd-catalysed processes. The broad combinatorial chemistry approach can, however, be extended to the synthesis of libraries based on much more complex, natural product-like scaffolds.

Combinatorial variation of the configuration of library members is rather more challenging. However, a broad range of reliable asymmetric transformations has been developed over the past 25 years. In many cases, the high enantioselectivity of these transformations can dominate over substrate-based diastereoselectivity. Thus, by harnessing the power of modern asymmetric chemistry, highly stereochemically diverse chemical libraries may be prepared.
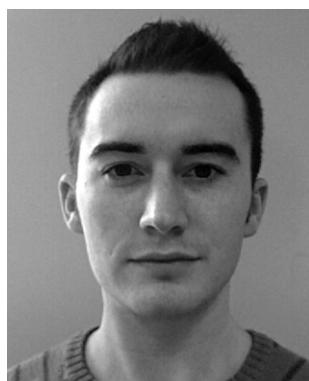
Chris Cordier

Chris Cordier graduated in 2003 from the University of Leeds with a first class BSc (Hons) degree in Chemistry and Pharmacology. His PhD, under the supervision of Prof. Adam Nelson and Dr Stuart Warriner, concerned the synthesis of diverse compounds with natural product-like structural motifs. He has been awarded a Herchel Smith postdoctoral research fellowship in organic chemistry at the University of Cambridge.

Sarah Murrison

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Daniel Morton

Daniel Morton obtained his MChem from the University of East Anglia in 2001, where he remained for his PhD (supervised by Dr Rob Stockman and Prof. Rob Field) which concerned the synthesis of chiral aziridines. In 2005, he joined the Nelson group, where he has worked on the total synthesis of hemibrevetoxin B and the diversity-oriented synthesis of natural product-like molecules.

Adam Nelson

Adam Nelson was appointed at the University of Leeds in 1998, and is currently Professor of Chemical Biology and Deputy Director of the Astbury Centre. Prof. Nelson was awarded the RSC Meldola medal (in 2001), the Pfizer academic award (in 2002), an EPSRC advanced research fellowship (2004–2009) and the AstraZeneca award in organic chemistry (in 2005).

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Preparing libraries with high skeletal diversity is an altogether more challenging proposition. A range of new synthetic innovations has been required to vary systematically the underlying scaffolds of ligands. In this review, we will place particular emphasis on the approaches which have been used to vary molecular scaffolds systematically. We have made no attempt to discuss every aspect of each synthetic strategy. In general, we have indicated in the starting materials the scope for substitutional diversity (through the use of generic substituents R^n). The ability of each approach to deliver alternative scaffolds is indicated (with a yield) for specific combinations of generic substituents. In most syntheses, however, each molecular scaffold was diversely substituted, and, thus, represented by more than one compound; the reader is directed to the primary literature for a full description of the synthetic chemistry undertaken.

1.2 Natural products as an inspiration in the design of compound libraries

Natural products necessarily reside in biologically relevant chemical space, since they bind both their biosynthetic enzymes and their target macromolecules. Natural product families are libraries of pre-validated, functionally diverse structures: these families are "privileged"[7] since individual compounds selectively modulate unrelated targets. Indeed, some natural products, such as trichostatin,[8] wortmannin[9] and brefeldin A,[10] have been directly exploited as valuable tools in chemical genetic studies. It is, therefore, no surprise that natural products have provided an inspiration for ligand library design.

In this review, examples of diversity-oriented chemistry are described which have, to some extent, been inspired by the structures of natural products. The purpose of DOS, however, is to enable the discovery of small molecules with previously unknown biological functions. Natural products do not adhere to principles that have been formulated to guide the structures of drug molecules.[11] Indeed, the only generally accepted principle uniting the structures of protein-binding small molecules is that conformational restriction is advisable. Natural products do, of course, adhere to this principle through the presence of a range of structural features: dense substitution, unsaturated functionality, and ring systems.

Although the inspiration which may be provided by natural products is useful, it is essential that it does not inform DOS chemistry to the extent that only compounds with the same biological functions as known natural products models emerge. We have deliberately omitted, however, studies in which the inherent biological activity of a natural product was optimised through the synthesis of derivatives.[12] In some cases, the scaffold (or scaffolds) represented within a DOS library closely resemble those found in natural products. Inspiration from natural product scaffolds is valuable in DOS, as ligands based on specific natural product scaffolds have been shown to bind selectively to different proteins with similar folds.[5b] In other examples of DOS chemistry, the relationship to natural products is much looser, with perhaps just some of the broad structural features of a natural product family being retained. Given that natural products necessarily have the broad structural features needed to recognise macromolecules, this level of inspiration is appropriate: indeed, it can allow the
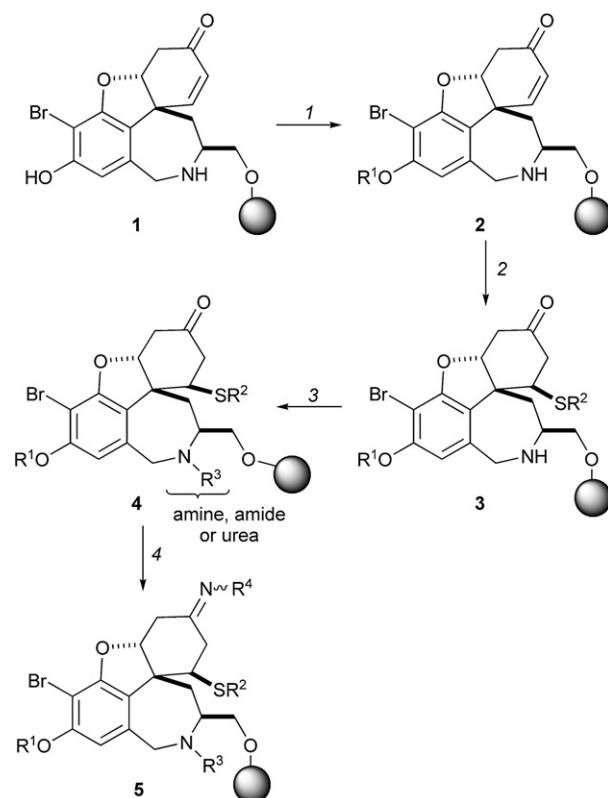

Scheme 1 Shair's synthesis of galanthamine-like molecules with high substitutional diversity. *Reagents and conditions*: (*1*) R^1OH, PPh_3, DIAD, THF, 0 °C (×2); (*2*) R^2SH, 2,6-lutidine, nBuLi, THF, 0→40 °C; (*3*) (i) R^3CHO, AcOH, MeOH–THF; (ii) $NaBH_3CN$, MeOH or R^3COCl, 2,6-lutidine, CH_2Cl_2 or R^3NCO, CH_2Cl_2; (*4*) R^4NH_2, AcOH, MeOH–CH_2Cl_2.

identification of ligands with functions that natural products have not yet been found to possess.

## 2 Strategies for diversity-oriented synthesis

2.1 Synthesis of compound libraries with high substitutional diversity

The combinatorial chemistry approach may be extended to the diversification of complex natural product scaffolds. The scaffold, **1**, of the alkaloid galanthamine was prepared in five steps from a solid-supported tyrosine derivative.[13] A sequence of reactions was then used to vary four of the substituents (R^1 to R^4) in the final compounds (Scheme 1). Functionalisation of the phenol using a Mitsunobu reaction (→ **2**) was followed by conjugate addition
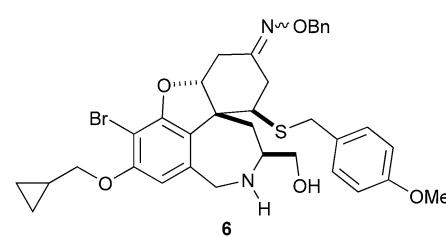
of a thiol to the α,β-unsaturated ketone (→ **3**), and functionali-sation of the secondary amine (→ **4**). Finally, the ketones **4** were converted into imines or oxime ethers **5**. Following release from solid support, the successful formation of 2527 of 2946 possible compounds was confirmed by mass spectrometry. A cell-based phenotypic assay was used to identify secramine **6** as a potent inhibitor of protein trafficking from the Golgi apparatus to the plasma membrane. The discovery of secramine underlines the validity of exploiting natural product scaffolds in the discovery of small molecules with novel functions: secramine's structure was inspired by galanthamine, but it has a novel biological function.

A similar approach was used to prepare a library of conformationally restricted 1,3-dioxanes **10** with high substitu-tional diversity (Scheme 2).[14] A split-pool approach was used to diversify three substituents in the final compounds **10**. The supported epoxides **7** were opened with either a thiol or an amine nucleophile to yield 1,3-diols which were converted into Fmoc-substituted 1,3-dioxanes. Some of the nucleophiles used in the first step were hydroxy-substituted, leading to the formation of mixed acetals in the second step: these acyclic acetals were, therefore, hydrolysed by treatment of the resins with PPTS in THF–MeOH. Following removal of the Fmoc


Scheme 3 Schreiber's synthesis of dihydropyrans with high substitu-tional and stereochemical diversity. *Reagents and conditions*: (*1*) **15** or 20 mol% **16** (or enantiomer), THF, rt; (*2*) (i) Pd(PPh₃)₄, thiosalicylic acid, THF, rt; (ii) R²R³NH, PyBOP, *i*Pr₂NEt, CH₂Cl₂–DMF, rt; (iii) HF·pyridine, THF, rt, then Me₃SiOMe.

group (→ **9**), the resulting free amines were converted into amides, ureas, thioureas or sulfonamides (→ **10**), and the final compounds cleaved from the beads. Using material cleaved from a single bead, and with knowledge of the substituent added into the final step, it was possible to use mass spectrometry to identify possible combinations of substituents added in the first two diversification steps.

Scheme 2 Schreiber's split-pool synthesis of 1,3-dioxanes with high substitutional diversity. *Reagents and conditions*: (*1*) R²SH or R¹R²NH; (*2*) (i) ArCH(OMe)₂, Me₃SiCl, HCl, dioxane; (ii) PPTS, THF–MeOH; (iii) piperidine; (iv) Me₃SiCl; (*3*) amide, urea, thiourea or sulfonamide formation.
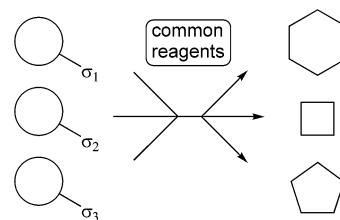
Scheme 4 The folding pathway approach to generating skeletal diversity; the skeletal information is pre-encoded into the substrate through σ-elements which programme folding to yield distinct molecular skeletons under common reaction conditions.
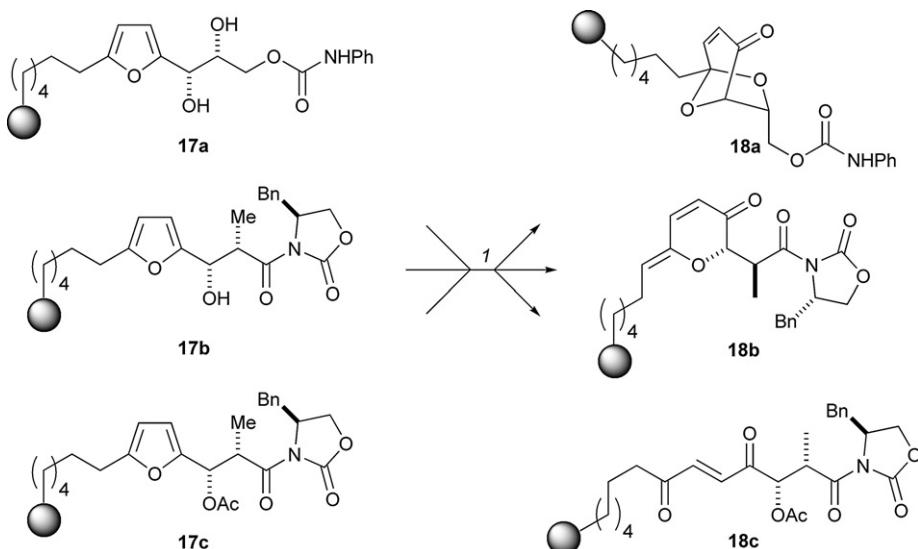
Scheme 5 Schreiber's approach to pre-encoding skeletal information using the Achmatowicz reaction. *Reagents and conditions*: (*1*) *N*-bromosuccini-mide, NaHCO$_3$, NaOAc, THF–H$_2$O (4 : 1), rt, 1 h; PPTS, CH$_2$Cl$_2$, 40–45 °C, 20 h.

Using this general synthetic approach, a 3780-member library of small molecules was used to create a small molecule microarray.[15] The microarray was then challenged with a fluo-rescently labelled derivative of Ure2 p, the central repressor of genes involved in nitrogen metabolism. The approach enabled the identification of the first small molecule ligand for Ure2 p: uretupamine, **11**. Uretupamine was found to modulate only a subset of the functions of Ure2 p, and thus its effects were more specific than those resulting from deletion of the *URE2* gene.

An asymmetric hetero-Diels–Alder reaction was used to prepare a library of dihydropyrancarboxamides which resemble
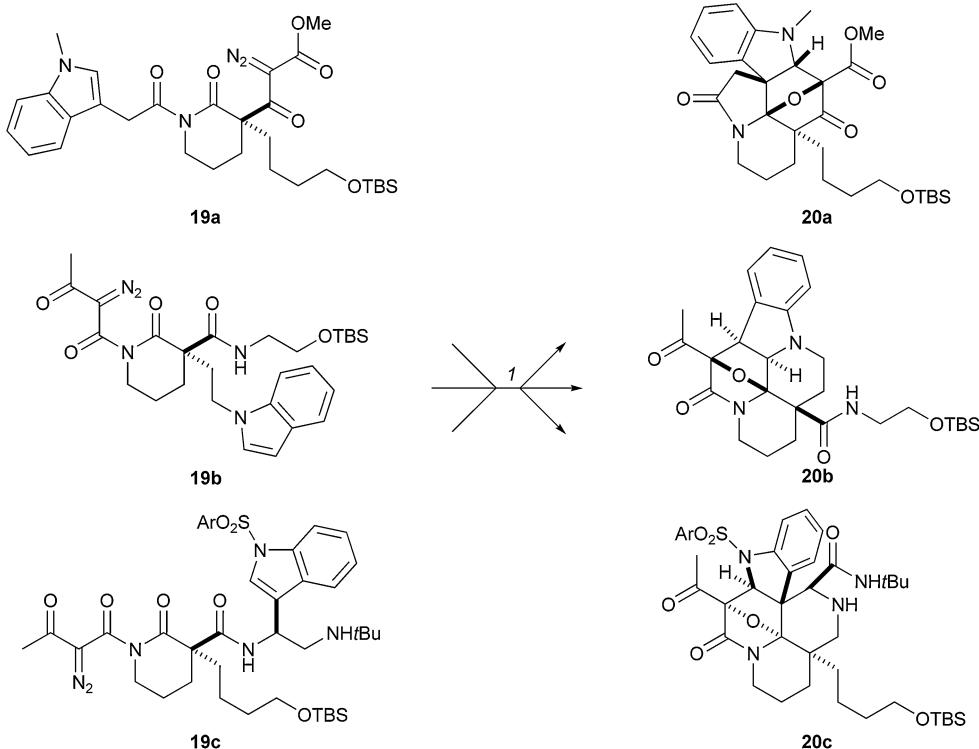
Scheme 6 Schreiber's approach to pre-encoding skeletal information in rhodium-catalysed tandem cyclisation–cycloaddition cascade. *Reagents and conditions*: (*1*) Rh$_2$(O$_2$CC$_7$H$_{15}$)$_4$, benzene, 50 °C; **20a**: 74%; **20b**: 73%; **20c**: 57%.

21a

21b

21c

22a

22b

22c
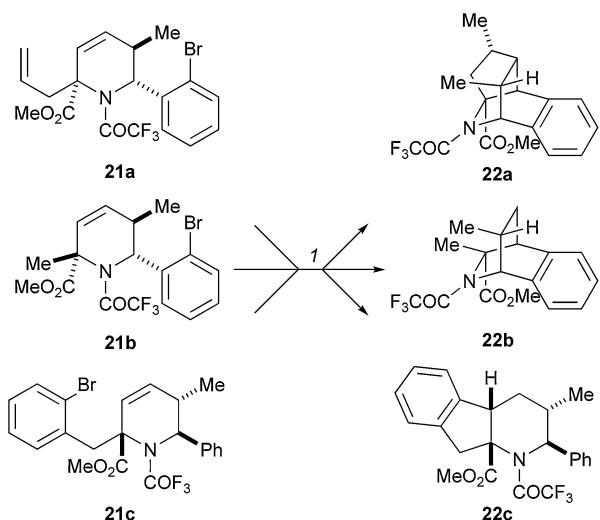
Scheme 7 Panek and Porco's folding pathway using functionalised tetrahydropyridines. *Reagents and conditions*: (*1*) AIBN, Bu₃SnH, 80 °C, 4 h, benzene, **22a**: 65%; **22b**: 89%; **22c**: 81%.

sialic acid mimetics (structures **14** and their enantiomers) (Scheme 3).[15] The library of 4320 structures was encoded using a binary encoding protocol which involved the attachment of tags to individual macrobeads. The asymmetric step (**12** → **13**) served a number of purposes: it generated the molecular scaffolds; it introduced some stereochemical diversity; and it incorporated the variable substituent R¹. Deprotection of the ester, amide formation, and release from solid support gave the final compounds.

A small molecule microarray was created by immobilisation of three small molecule libraries (including the libraries of the 1,3-dioxanes **10** and the dihydropyrancarboxamides **14**).[16] The microarray was probed with a Hap3 p-GST fusion protein, and hits identified using a fluorescently-labelled anti-GST antibody. The approach enabled the discovery of haptamide A **14b**, a ligand which targeted a subunit of the transcription factor, Hap3p. Haptamide B, an analogue of haptamide A, was subsequently discovered using a more traditional ligand optimisation approach.

23a

23b

23c

24a

24b

24c

25a

25b

25c
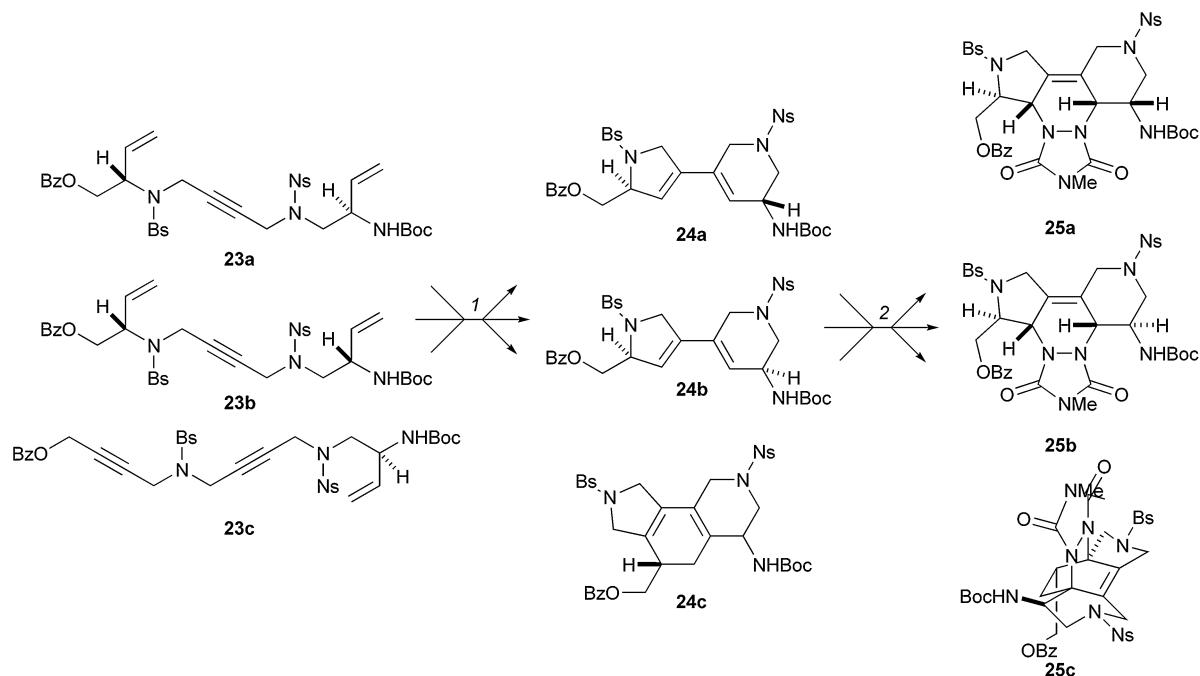
Scheme 8 Oligomer-based approach to skeletal diversity employing metathesis cascades and Diels–Alder reactions as folding processes. *Reagents and conditions*: (*1*) Grubbs' first-generation catalyst (5 mol%), benzene, ethylene, reflux, 12–24 h; **24a**: d.r. >20 : 1, 57%; **24b**: d.r. >15 : 1, 74%; **24c**: d.r. >20 : 1, 56%; (*2*) 4-methyl-1,2,4-triazoline-3,5-dione, CH₂Cl₂, 0 °C to rt.; **25a**: 93%; **25b**: 90%; **25c**: 46%. Ns = 2-nitrobenzenesulfonyl, Bs = 4-bromo-benzenesulfonyl.

26

27

28

29

30

31
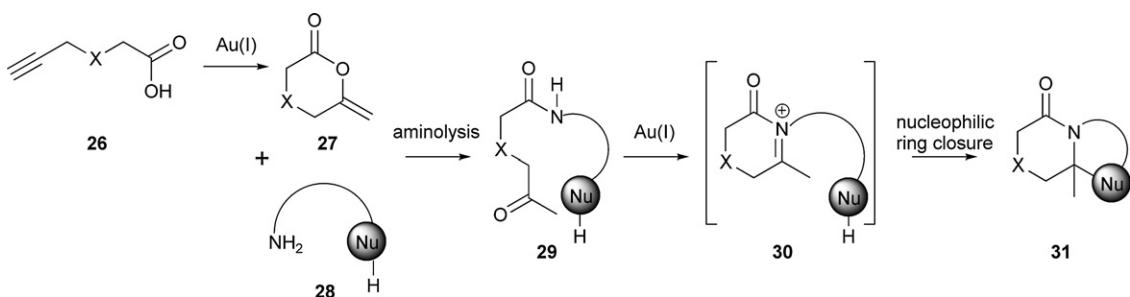
Scheme 9 Outline of Dixon's approach for the synthesis of skeletally diverse, bicyclic heterocycles. X = O or CH₂.

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14b

[R^1=3-thiophenyl, R^2=H
R^3=o-hydroxyphenyl, X=(CH$_2$)$_3$]
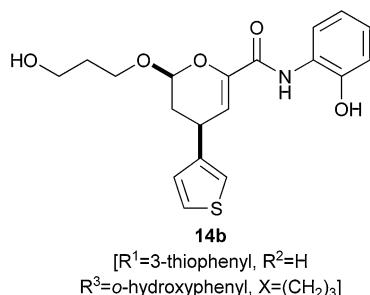
In the three diversity-oriented syntheses described in Schemes 1–3, a library of substitutionally diverse small molecules was prepared. The inspiration provided by natural products, however, varied hugely. Each of the compounds **5** had the same skeleton as the alkaloid galanthamine, whereas the relationship between natural products and the other compounds was much looser. Schreiber has also described the preparation of a substitutionally diverse library of biaryl-containing medium-ring compounds (which have some similar structural features to biaryl-containing natural products such as vanomycin and pterocaryanin C).[17]

In the three examples described in detail (Schemes 1–3), a high-throughput screen enabled the identification of a small molecule with a novel function. A forward chemical genetic (phenotypic) screen was used to identify secramine (**6**) whereas reverse chemical genetic screens were used to identify small molecule partners (**11** and **14b**) for the proteins Ure2 p and Hap3p.‡ Substitutionally diverse libraries are, therefore, valuable tools which may be used to discover novel small molecules for interrogating biological mechanisms.

2.2 Use of 'folding pathways' to introduce skeletal diversity

The 'folding pathway' approach uses a common set of reaction conditions to transform a range of substrates into products with alternative molecular skeletons (Scheme 4). The substrates are encoded to 'fold' into the alternative scaffolds through strategically appended groups, known as 'σ-elements'. This strategy leads skeletal diversification under substrate control.

Schreiber has developed a folding pathway which exploits the Achmatowicz reaction (Scheme 5).[18] The fate of oxidation of the furan substrates **17** depends on the functionalisation of groups elsewhere in the molecule (the σ-elements). Hence, oxidation of the furan **17c**, which does not bear any free hydroxy groups, simply generated the ene-dione **18c**. With suitably positioned nucleophilic groups in the starting material however, a *cis*-enedione intermediate could be intercepted. Hence, upon oxidation, the furyl alcohol **17b** folded, and eliminated water, to yield the alkylidene-pyran-3-one **18b**. In contrast, with two free hydroxyl groups, **17a** folded to yield the bicyclic ketal **18a**. The combination of solid-phase chemistry

and common reaction conditions made the strategy amenable to split-pool synthesis, which significantly increased the efficiency of library generation.

Schreiber used substrate-based control to create alternative indole alkaloid-like skeletons (Scheme 6).[19] An α-diazo-β-keto-carbonyl group and an indole ring were strategically placed at different positions on the scaffolds of the starting materials (**19**). The densely functionalised starting materials were assembled either using an Ugi four-component coupling reaction or by
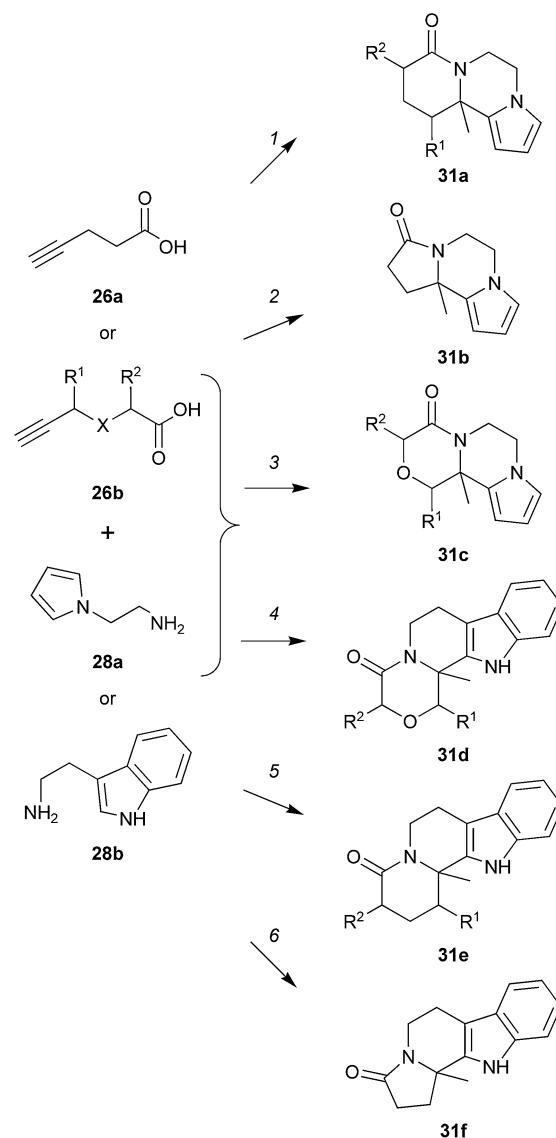
Scheme 10 Dixon's Au(I)-catalysed cascade. *Reagents and conditions*: (*1*) AuPPh$_3$Cl/AgOTf (1%), **26b**, **28a**, toluene, rt, 3 h, then reflux, 2 d, d.r. 2 : 1, 81% [R^1 = H, R^2 = CH$_2$Ph, X = CH$_2$]; (*2*) AuPPh$_3$Cl/AgOTf (1%), **26a**, **28a**, toluene, rt, 3 h then reflux, 1 d, 91%; (*3*) AuPPh$_3$Cl/AgOTf (1%), **26b**, **28a**, toluene, 75 °C, then reflux 1 d, d.r. 3 : 1, 64% [R^1 = H, R^2 = Me, X = O]; (*4*) AuPPh$_3$Cl/AgOTf (1%), **26b**, **28b**, xylene, 75 °C, 3 h, then 125 °C, 1 d, d.r. 1 : 5 : 1, 97%, [R^1 = n-pentyl, R^2 = H, X = O]; (*5*) AuPPh$_3$Cl/AgOTf (1%), **26b**, **28b**, toluene, 75 °C, 3 h, then reflux, 2 d, d.r. 1 : 1 : 5, 82%, [R^1 = H, R^2 = CH$_2$Ph, X = CH$_2$]; (*6*) AuPPh$_3$Cl/AgOTf (1%), **26a**, **28b**, toluene, 75 C, 3 h, reflux, 1 d, 87%.

‡ The forward chemical genetic approach involves the discovery of chemical probes through their phenotypic effect. In contrast, the reverse chemical genetic approach involves the discovery of chemical probes which modulate a predetermined macromolecular target.
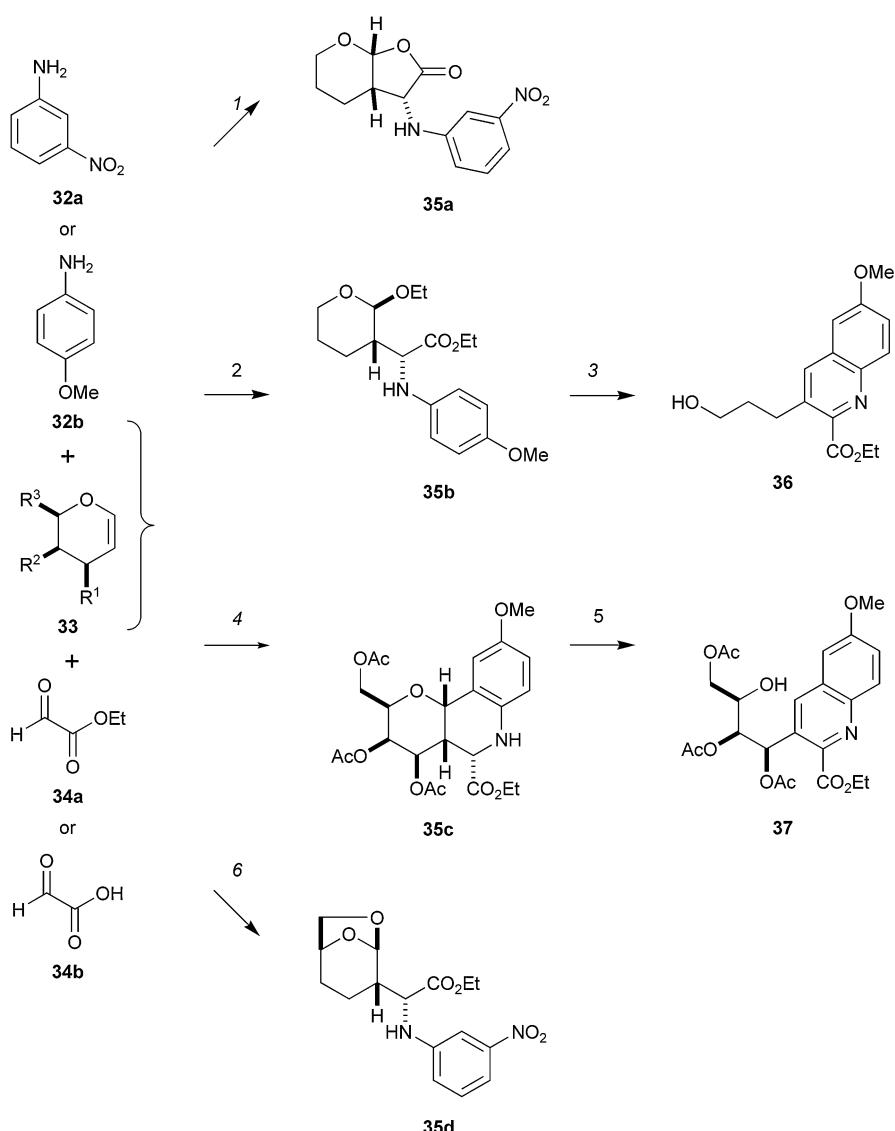
Scheme 11 Latvilla's multicomponent reaction. *Reagents and conditions*: (*1*) Sc(OTf)$_3$ (20%), **32a**, **33**, **34b**, 4 Å MS, MeCN, Ar, 2 d, rt, d.r. 70 : 30, 83%, [R^1,R^2,R^3 = H]; (*2*) Sc(OTf)$_3$ (20%), **32b**, **33**, **34a**, EtOH, d.r. 2 : 3 : 1, 82%, [R^1,R^2,R^3=H]; (*3*) TFA (O$_2$), MeCN–H$_2$O, 71%; (*4*) Sc(OTf)$_3$ (20%), **32b**, **33**, **34a**, single stereoisomer, [R^1,R^2 =OAc; R^3 =(*R*)-CH$_2$OAc]; (*5*) CAN, MeCN, 58%; (*6*) Sc(OTf)$_3$ (20%), **32a**, **33**, **34a**, d.r. 2 : 5 : 1, 46%, [R^1,R^2 = H, R^3= (*R*)-CH$_2$OH].

alkylation of a common scaffold; subsequent conversion to a diazo compound yielded the cyclisation precursors. The substrates, **19**, were treated with a catalytic amount of rhodium(II) octanoate dimer in benzene (80 °C). Presumably, generation of a carbonyl ylid was followed by intramolecular 1,3-dipolar cycloaddition with the indole ring to give the alternative polycyclic skeletons, **20**.

Panek and Porco used the cyclisation chemistry of radical intermediates to generate skeletal diversity (Scheme 7).[20] The skeletons of the products **22** were pre-encoded in the substrates **21** by the location of the radical-initiating sites and the unsaturated groups. The tetrahydropyridine substrates **21** were prepared by stereoselective alkylation with a range of alkyl, alkenyl and alkynyl halides. The folding processes were triggered by treatment of the tetrahydropyridines **21** with

tributyltin hydride and a substoichiometric amount of AIBN at 80 °C. Abstraction of a strategically-positioned bromine atom and cyclisation, yielded a range of distinct polycyclic alkaloid-like frameworks. Particularly high levels of skeletal complexity could be created when tandem radical cyclisation processes were possible (*e.g.* to yield **22a**).

Schreiber has also developed an oligomer-based 'folding' strategy (Scheme 8).[21] In this approach, metathesis substrates were assembled iteratively from monomer starting materials: Fukuyama–Mitsunobu reactions were used to prepare the substrates **23**. The substrates **23** were folded by treatment with Grubbs' first-generation catalyst to yield 1,3-dienes **24**. The 1,3-diene products **24** were substrates for a second diversification step: Diels–Alder reaction with a triazoline-3,5-dione dienophile gave the polycyclic products **25**.
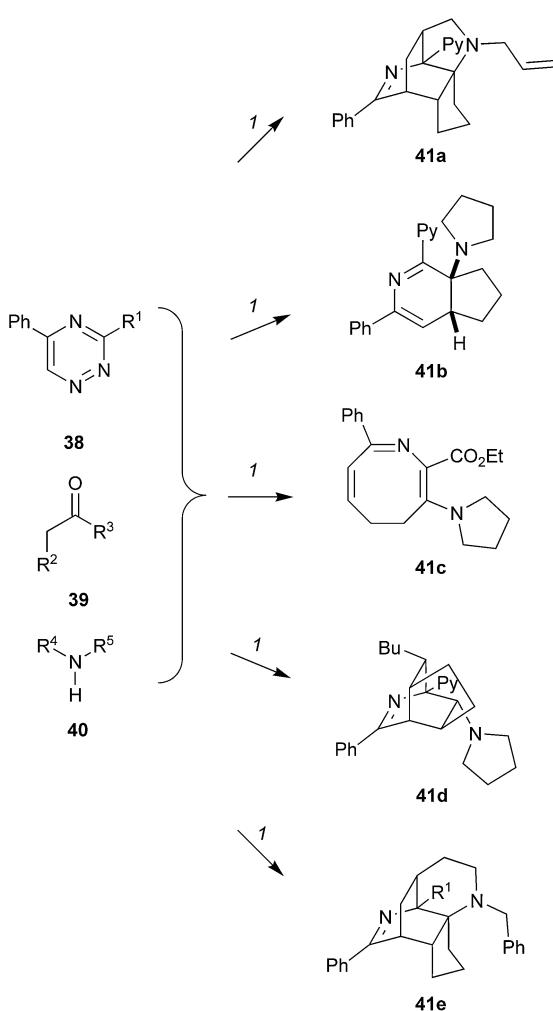
Scheme 12 Taylor's cascade for the synthesis of polycyclic compounds. *Reagents and conditions*: (*1*) 4 Å MS, CHCl$_3$, Δ; **41a**: 96% [R^1 = Py, R^2,R^3 = –(CH$_2$)$_3$–, R^4,R^5 = allyl]; **41b**: 83% [R^1 = Py, R^2,R^3 = –(CH$_2$)$_3$–, R^4,R^5 = –(CH$_2$)$_4$–]; **41c**: 56% [R^1= CO$_2$Et, R^2,R^3 = –(CH$_2$)$_2$–, R^4,R^5= –(CH$_2$)$_4$–]; **41d**: 67% [R^1 = Py, R^2 = *trans*-3-octane, R^3 = H, R^4, R^5 = –(CH$_2$)$_4$–]; **41e**: 39% [R^1 = CO$_2$Et, R^2,R^3 = –(CH$_2$)$_3$–, R^4 = Bn, R^5 = but-3-enyl]. Py = 2-pyridyl.

2.3 Assembly of alternative scaffolds using multi-component reactions

Innovative multi-component reactions have been developed to allow variation of the product scaffolds.§ In essence, alternative cyclisation precursors are assembled *and* 'folded' under common reaction conditions. The approach combines the power of multicomponent reactions with that of folding pathways.

Dixon has used a one-pot gold(I)-catalysed cascade to prepare skeletally-diverse alkaloid-like small molecules (Scheme 9).[22] Initially, Au(I) catalysed the cyclisation of an alkynyl carboxylic acid **26** to give a cyclic enol ether **27**. Attack of an amine nucleophile **28** on the cyclic enol ether

§ In the context of this review, a multi-component reaction is simply considered to involve the assembly of two or more components.
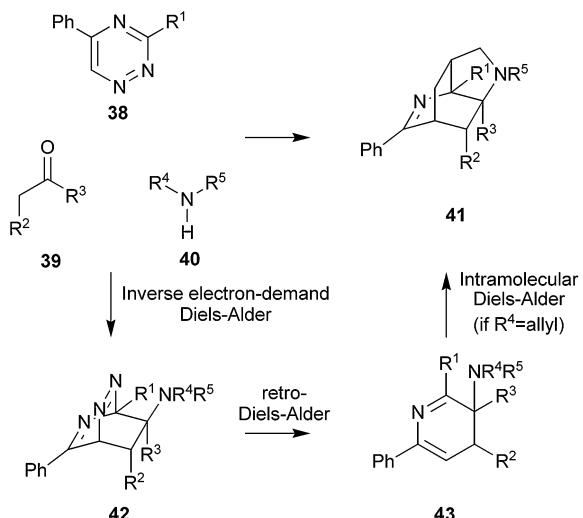
Scheme 13 Mechanism of Taylor's reaction cascade.

subsequently generated a ketoamide **29**. In other words, the combination of alkynyl carboxylic acid **26** and amine **28** dictated the structure of the cyclisation precursor **29**. Under the same reaction conditions, the cyclisation precursor was converted into an *N*-acyl iminium ion **30**, which was then trapped by a tethered nucleophile (→ **31**).

The scope of Dixon's methodology is summarized in Scheme 10. Using combinations of the alkynyl carboxylic acids **26a–b**, and the pyrrole- and indole-tethered amines **28a–b**, a remarkable range of alternative alkaloid-like scaffolds were prepared.

A Lewis acid-catalysed, multicomponent reaction involving an aniline **32**, a cyclic enol ether **33** and a glyoxaldehyde **34**, has been developed (Scheme 11).[23] The cyclic enol ether **33** attacked an iminium ion derived from the aniline **32** and the aldehyde **34**. The outcome of the reaction depended on the fate of the resulting oxocarbenium ion. Intramolecular capture can occur through participation of a pendant carboxylic acid (→ γ-lactone **35a**), aryl ring (→ tetrahydroquinoline **35c**), or alcohol (→ bicyclic acetal **35d**). Alternatively, capture by the solvent (ethanol) yielded a THP acetal (**35b**). The scaffolds **35b–c** were further skeletally diversified by oxidation to the quinolines **36** and **37**.

Taylor and Raw have devised a multicomponent reaction which can yield skeletally diverse products (Scheme 12).[24] The mechanism of formation of some of the products is shown in Scheme 13. Condensation of a carbonyl compound **39** with an amine **40** yields an enamine which can participate in an inverse electron-demand Diels–Alder with a triazine **38** (→ **42**). Expulsion of molecular nitrogen, *via* a retro-Diels–Alder reaction, yields a 2-aza-diene **43**. The outcome of the reaction depends on the fate of the 2-aza-diene **43**. In the absence of a pendant alkene, the final product is usually a 2-aza-diene (*e.g.* **41b**, Scheme 12), although aromatisation to yield a highly substituted pyridine is also possible.[24b,c] If cyclobutanone is used as the carbonyl compound, electrocyclic ring-opening can occur to yield an eight-membered heterocycle (*e.g.* **41c**). However, with a suitably placed pendant alkene, an intramolecular Diels–Alder reaction is possible to generate yet more complex skeletons
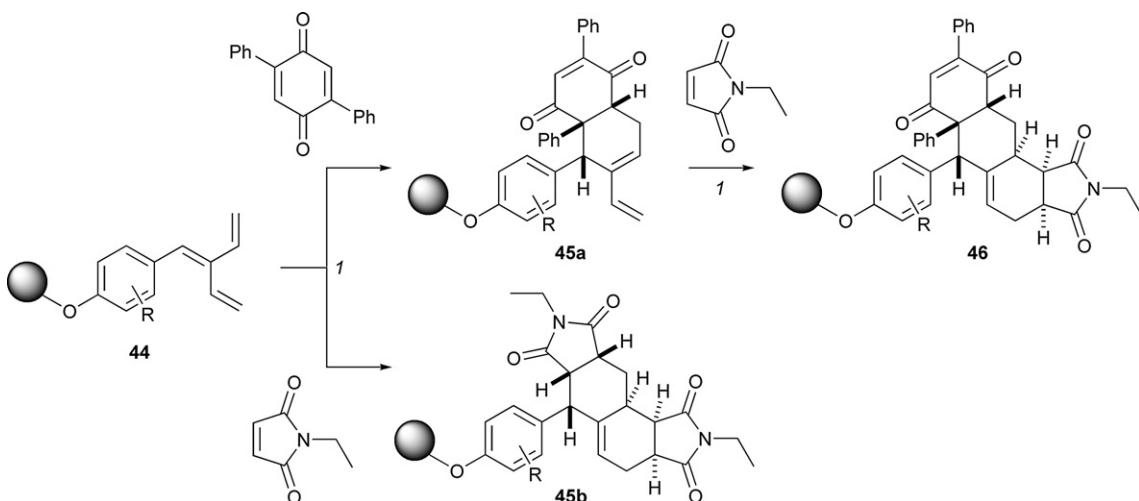
Scheme 14 Summary of Schreiber's consecutive use of Diels–Alder reactions to construct 29 400 discrete polycyclic compounds. *Reagents and conditions*: (*1*) toluene.

(*e.g.* **41a**, **41d** or **41e**). The skeletons **41** are diverse, and broadly resemble those of alkaloids.

## 2.4 Exploitation of combinations of complexity-generating reactions

Using complexity-generating reactions in combination can lead to rapid increases in molecular complexity. Indeed, earlier in this review, a four-component Ugi reaction was used to prepare starting materials for Rh-catalysed dipolar cycloaddition reactions (Scheme 6).[19] Furthermore, the use of metathesis cascade folding processes in combination with Diels–Alder reactions has also been described (Scheme 8).[20] Both of these approaches were discussed in the context of folding pathways, but the use of two complexity-generating reactions in combination allowed some highly complex, polycyclic alkaloid-like molecules to be prepared. Here, we describe further examples of diversity-oriented syntheses in which complexity-generating reactions were used in combination.

The Diels–Alder reactions of the solid-supported trienes **44** yielded skeletally diverse products (Scheme 14).[25] The issue of regioselectivity is raised in the Diels–Alder reactions of the cross-conjugated trienes **44**. With tri- and tetra-substituted dienophiles, one of the two available dienes reacted to yield a diene product (*e.g.* **45a**), which could then undergo a subsequent Diels–Alder reaction with a second dienophile (→ **46**). Remarkably, however, with a disubstituted dienophile, such as *N*-ethyl maleimide, the other available diene reacted initially, and after a second Diels–Alder reaction, an alternative skeleton (*e.g.* **45b**) was obtained. The approach was used to prepare 29 400 polycyclic compounds with ten distinct molecular skeletons. Some of the compounds were immobilised on a small molecule microarray, which was then challenged with Cy5-conjugated calmodulin. The approach allowed novel sub-micromolar ligands for calmodulin to be discovered.

Schreiber used the concept of 'skeletal transformation' to yield 4275 products with three underlying skeletons (Scheme 15).[26] Each step of the synthetic route was adapted and optimised for use on macrobeads. To start with, the steroid-based epoxide **47** was opened, and then functionalised, to yield 171 distinct
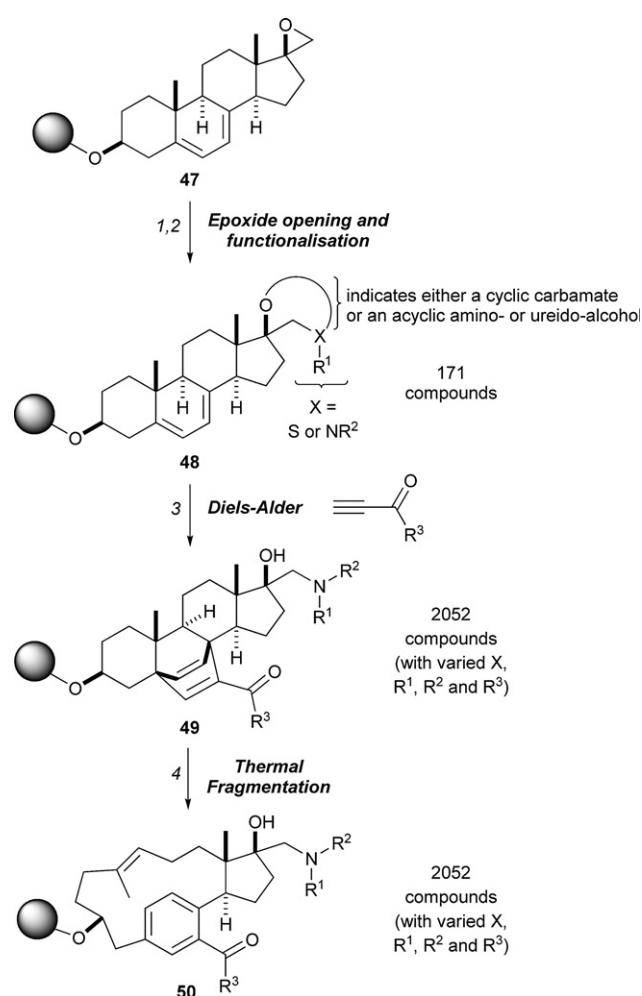
Scheme 15 An overview of the skeletal transformations applied by Schreiber in the preparation of >4000 skeletally diverse small molecules. *Reagents and conditions*: (*1*) amine, LiClO$_4$ (or thiol); (*2*) alkylation, acylation or oxazolidone formation [for X = NR^2]; (*3*) R^{3}COC≡CH, Et$_2$AlCl; (*4*) heat.
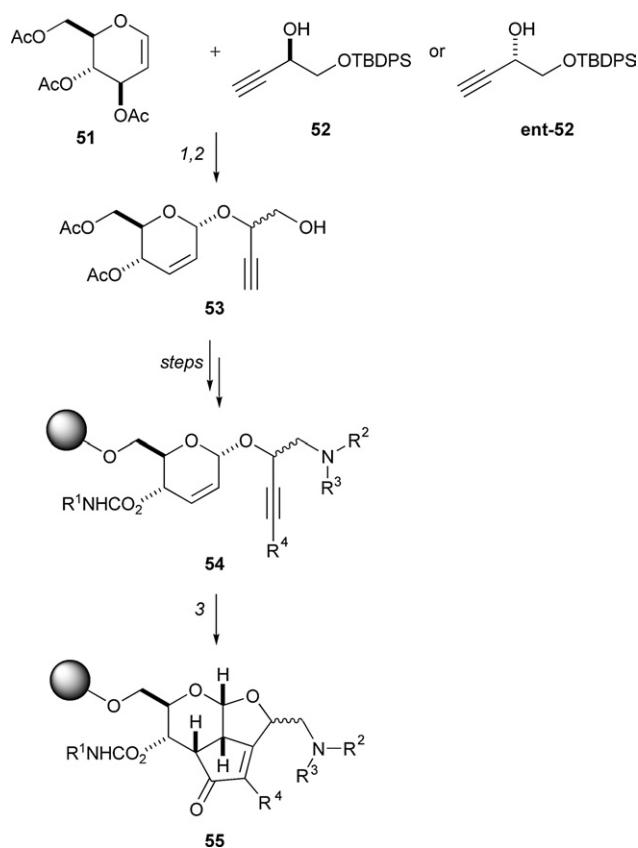
Scheme 16 Summary of Schreiber's pairwise use of Ferrier and Pauson–Khand reactions for the synthesis of polycyclic skeletons. *Reagents and conditions*: (*1*) cat. BF$_3$·OEt$_2$, CH$_2$Cl$_2$, −78 to −20 °C, 92%; (*2*) TBAF, AcOH, THF; purification, 72%; (*3*) Co$_2$(CO)$_8$, CH$_2$Cl$_2$; NMO, CH$_2$Cl$_2$–CH$_3$CN.

compounds with the scaffold **48**. Methodology developed by Winterfeld was then used to sequentially transform the scaffold **48** into two alternative scaffolds, **49** and **50**. First, a Diels–Alder reaction was used to introduce the variable substituent, R^3, and to switch the scaffold (**48** → **49**); 2052 compounds with the scaffold **49** were prepared. Then, a retro-Diels–Alder reaction was used to fragment each of the compounds **49** to yield 2052 *para*-cyclophanes **50**.

Schreiber used the Ferrier reaction in combination with the Pauson–Khand reaction to prepare a library of polycyclic molecules (Scheme 16).[27] Initially, a Ferrier reaction was used to condense a glycal (*e.g.* **51**) with a propargylic alcohol (*e.g.* **52**); the resulting enyne **53** was immobilised on a solid support and then further functionalised (→ **54**). Intramolecular Pauson–Khand reactions were used to prepare fused cyclopentenones **55** to which nucleophiles could be added in a conjugate sense (not illustrated). In a pilot study, 2500 tricyclic compounds were prepared.

Schreiber used three consecutive complexity-generating transformations in combination to prepare compounds of general structure **63** (Scheme 17):[28] an Ugi reaction, a Diels–Alder reaction and a ring-closing metathesis. Thus, the Ugi reaction of the solid-supported amine **56**, an isocyanide (*e.g.* **59**), furfural **57** and a fumaric acid derivative (*e.g.* **58**) yielded an initial adduct **60**. However, the intramolecular Diels–Alder reaction of the furan and a dienophile – the fumaric acid derivative – was spontaneous, and the adduct **61** was obtained directly. Bis-allylation (→ **62**), and ring-closing metathesis generated the final scaffold **63**.

In a later adaptation of this approach, it was shown that the relative configuration of a metathesis precursor controlled the skeleton of the metathesis product (Scheme 18).[29] Hence, the triene **64a** underwent ring-closing metathesis to give the bridged
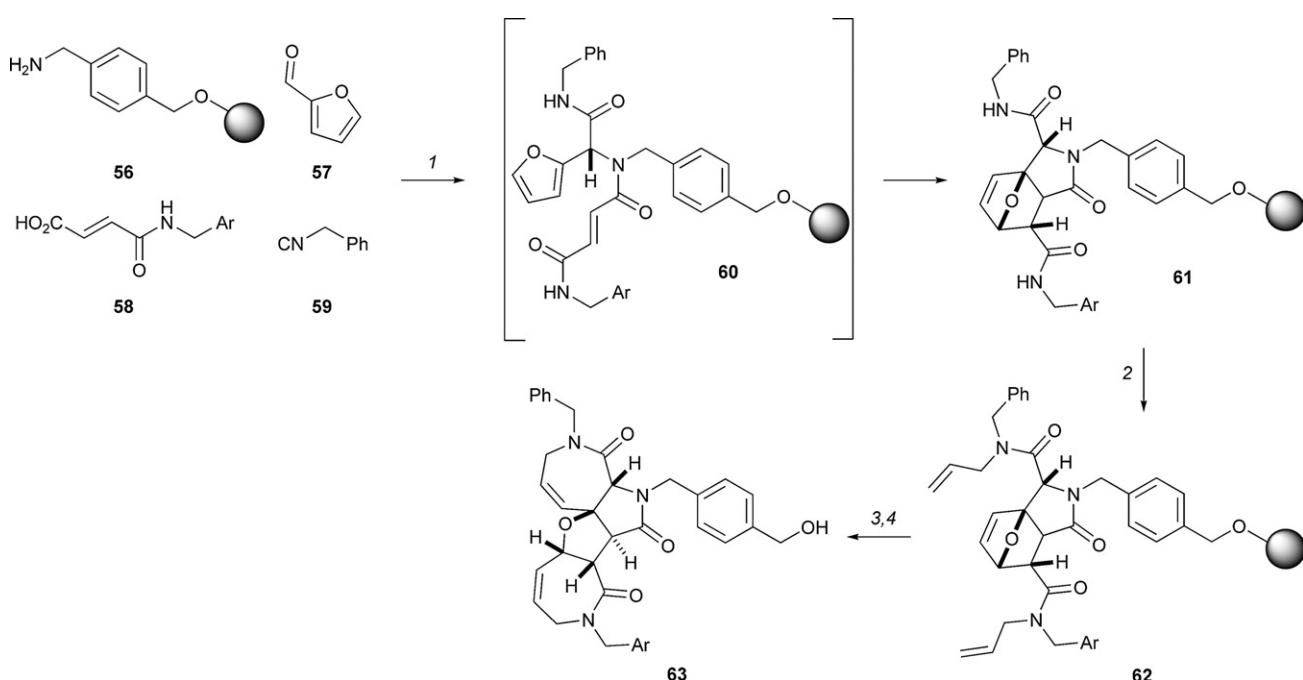
Scheme 17 Schreiber's exploitation of the Ugi and Diels–Alder reactions and metathesis chemistry in combination. *Reagents and conditions*: (*1*) MeOH–THF, (2 : 1), 60 h; (*2*) KHMDS, allyl bromide; (*3*) Grubbs' 2nd-gen. cat., CH$_2$Cl$_2$, 40 °C, 60 h; (*4*) HF·pyridine, CH$_2$Cl$_2$, 2 h. Ar = *m*-bromophenyl.
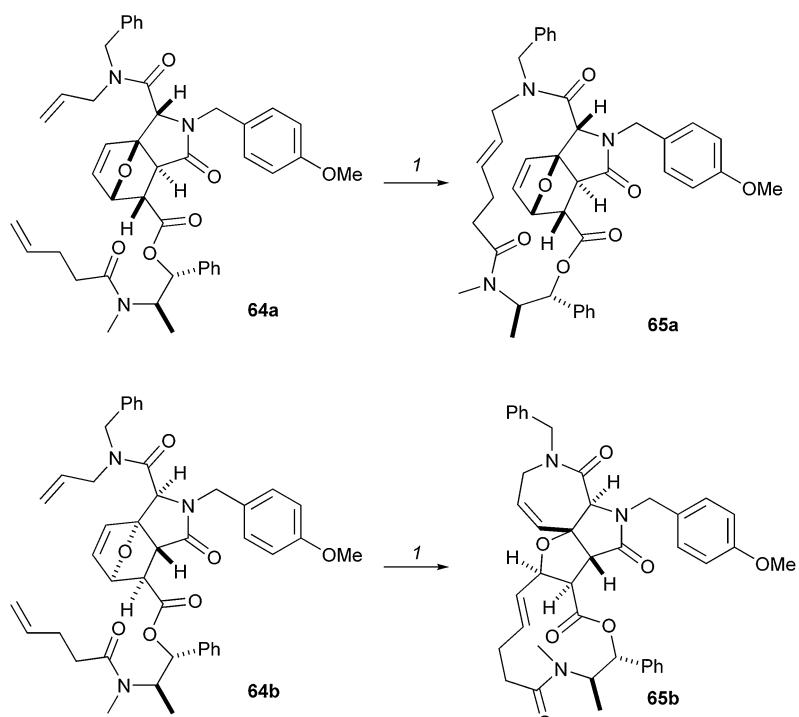
Scheme 18 Stereochemical control of skeletal diversity. *Reagents and conditions*: (*1*) 10 mol% Grubbs' 2nd-gen. cat., CH$_2$Cl$_2$, [**65a**: 87%; **65b**: 65%].
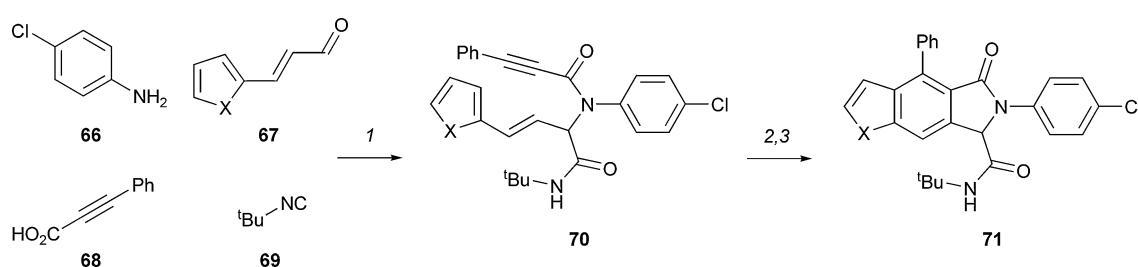
Scheme 19 Summary of Yang's approach to the diversity-oriented synthesis of benzofurans and indoles. *Reagents and conditions*: (*1*) MeOH, 25 °C, 12 h; (*2*) xylene, 140 °C; (*3*) CH$_2$Cl$_2$–benzene (1 : 25), DDQ, (X = O, 87% over 3 steps; X = NMe, 61% over 3 steps).
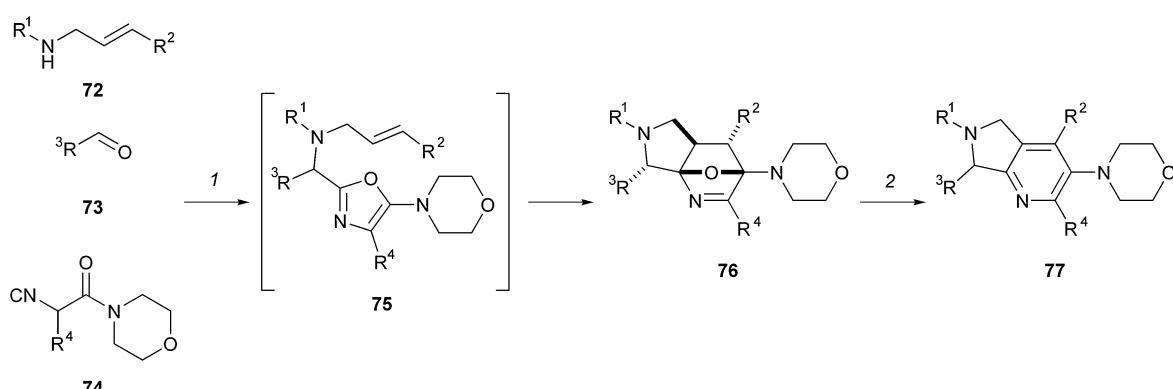
Scheme 20 Summary of Zhu's approach to the diversity-oriented synthesis of polyheterocycles and pyrrolopyridines. *Reagents and conditions*: (*1*) MeOH, rt, [R^1 = Bn, R^2 = CO$_2$Et, R^3 = C$_6$H$_{13}$, R^4 = Bn, 92%]; (*2*) MeOH, TFA, −78 °C, [R^1 = Bn, R^2 = CO$_2$Et, R^3 = C$_6$H$_{13}$, R^4 = Bn, 85%].

macrocycle **65a** in 87% yield. However, a diastereomeric triene **64b** underwent a cascade of metathesis reactions to give the fused product **65b** in 65% yield. This powerful example of a match–mismatch effect was exploited in the synthesis of a library of compounds based on the two very distinct product skeletons.

Fathi and Yang used a diversity-oriented synthetic approach to prepare a library of benzofurans, benzothiophenes and indoles (Scheme 19).[30] A four-component Ugi reaction was used to prepare amides (*e.g.* **70**) which were substrates for intramolecular Diels–Alder reactions. After aromatisation, benzofurans, benzothiophenes or indoles were obtained. The approach was used to prepare 28 compounds with six distinct polycyclic frameworks.

Zhu had previously used a conceptually similar approach to prepare oxa-bridged polyheterocycles **76** and pyrrolopyridines **77** (Scheme 20).[31] Hence, after formation of an iminium ion (from **72** and **73**), and nucleophilic addition of an isocyanide **74**, the usual Ugi pathway was intercepted: the resulting nitrilium ion was attacked intramolecularly by the amide to yield oxazoles **75**. The oxazoles **75** spontaneously underwent intramolecular hetero-Diels–Alder reactions to yield bridged products **76**. Acid-catalysed ring-opening induced pyrrolopyridine formation (→ **77**). In essence, the aromatisation step is a powerful skeletal

transformation which converts one library of compounds (**76**) into another (**77**).

The examples described in this section are a testament to the power of exploiting combinations of complexity-generating reactions. The approach is particularly powerful when a skeletal transformation is used to convert one library of compounds into another library based on an entirely different scaffold (see Schemes 15 and 20).

2.5 Use of branching pathways to yield skeletal diversity

The 'branching pathway' strategy involves the conversion of common precursors into a range of distinct molecular scaffolds (Scheme 21). Ideally, it should be possible to design a range of flexible precursors which participate in complementary skeletal transformations. The development of a branching pathway may require considerable optimisation of the individual skeletal transforming steps. It is important, however, to ensure that the ethos of diversity-oriented synthesis is retained: *i.e.* that a library is prepared in a deliberate and simultaneous fashion. In some cases, the implementation of an intricate branching pathway, which exploits a very limited range of precursors, may cause the parallel nature of diversity-oriented synthesis to be lost.

DNA-templated synthesis is emerging as a valuable tool in the implementation of branching pathways.[32] By using a DNA-tagged substrate, a reaction sequence mediated by (DNA-tagged) reactants may be encoded. Remarkably, it is possible for a number of different reaction sequences to be undertaken in the same reaction vessel. Active compounds may be directly selected from the mixture of products, and identified by PCR-amplification of the DNA tag.

The branching pathway strategy has, however, largely been demonstrated using more conventional synthetic approaches. Schreiber has exploited the chemistry of boronic esters to prepare a polyketide-like library (Scheme 22).[33] Trans-esterification of
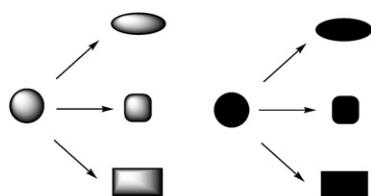

Scheme 21 The 'branching pathway' approach to skeletal diversity. Common precursors (circles) are converted, under different conditions, into distinct molecular skeletons.
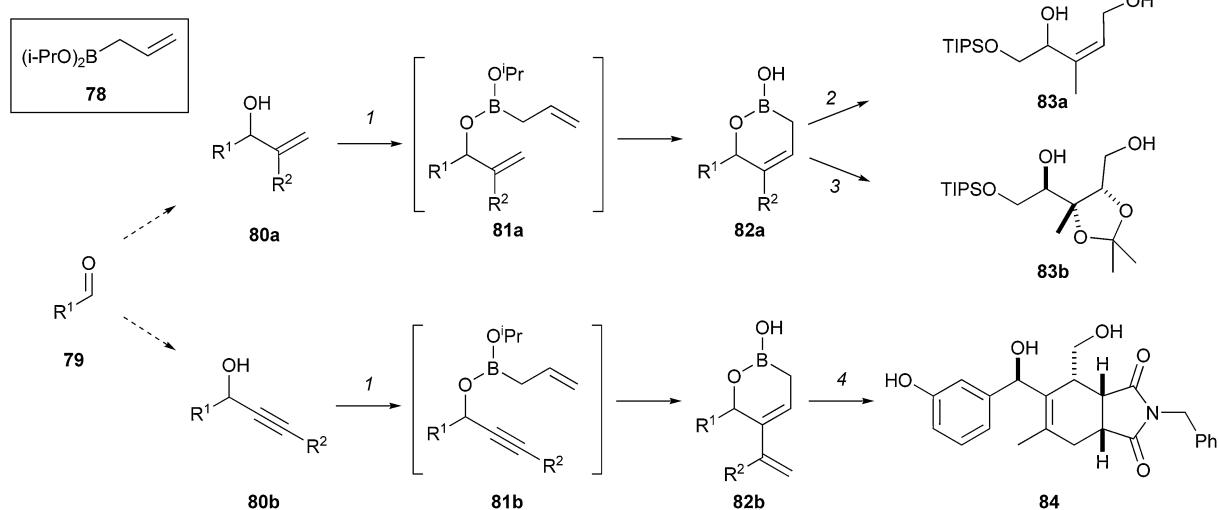
Scheme 22 Schreiber's branching pathway using boronic esters. *Reagents and conditions*: (*1*) **78** (3 eq.), Grubbs' 2nd-gen. cat. (20 mol%), CH$_2$Cl$_2$, reflux, then H$_2$O; yield range: 78–92%; (*2*) H$_2$O$_2$, NaOH, THF, rt, 56% (over 2 steps) [R^1 = CH$_2$OTIPS; R^2 = Me]; (*3*) OsO$_4$, NMO, acetone, pH 7 buffer; then 2,2-dimethoxypropane, CH$_2$Cl$_2$, PPTs; then H$_2$O$_2$, NaOH, THF, 66% (over 2 steps) [R^1 = CH$_2$OTIPS; R^2 = Me]; (*4*) *N*-benzylmaleimide (8 eq.), PhMe, 80 °C; then H$_2$O$_2$, NaOH, THF; carried out on solid support [typical yields for the Diels–Alder reaction of related substrates in solution: 81–91%].

Scheme 24 Schreiber's branching pathway exploiting complementary cyclisation reactions. *Reagents and conditions*: (*1*) 10 mol% Pd(PPh₃)₂(OAc)₂, benzene, 80 °C, [**88a**: single diastereomer, 81%; **90d**: single diastereomer, 70%] (*2*) 10 mol% Hoveyda–Grubbs' 2nd-gen. cat., CH₂Cl₂, reflux, [**88b**: *trans:cis* 85 : 15, 89%; **90a**: *trans:cis* 85 : 15, 87%; **91** : *trans:cis* 75 : 25, 90%]; (*3*) 10 mol% CpRu(MeCN)₃PF₆, acetone, rt, [**88c**: single diastereomer, 85%; **90c**: single diastereomer, 91%]; (*4*) 10 mol% NaAuCl₄, MeOH, rt, single diastereomer, 80%; (*5*) NaH, toluene, rt, 88%; (*6*) Co₂(CO)₈, Et₃NO, NH₄Cl, benzene, rt, [**88f**: d.r. >90 : 10, 85%; **90b**: single diastereomer, 85%]; (*7*) *m*-CPBA, THF, −78 → 0 °C, single diastereomer, 87%; (*8*) 4-methyl-1,2,4-triazoline-3,5-dione, CH₂Cl₂, rt [**89**: single diastereomer from *trans* diene, 72%; **92**: single diastereomer, 65%; **93**: 80% (combined yield from mixture of dienes)].
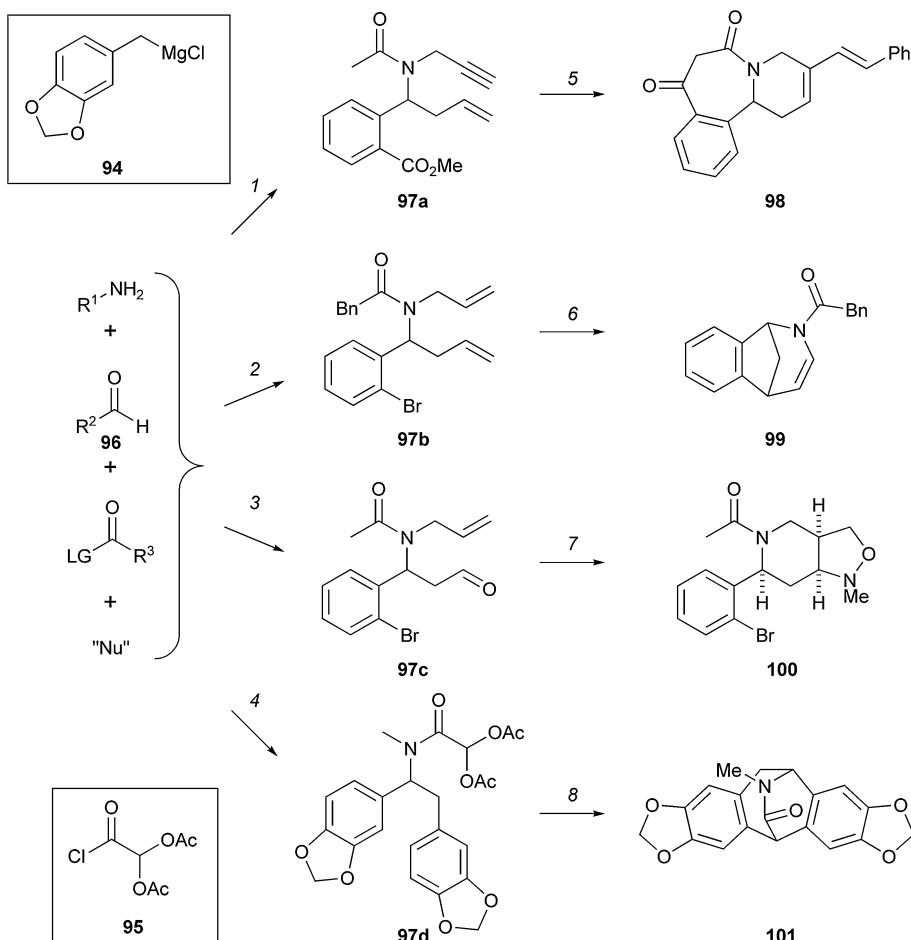
Scheme 25 Martin's branching pathway for heterocycle synthesis following a multicomponent reaction. *Reagents and conditions*: (*1*) propargyl amine, AcCl, allylZnBr, CH₂Cl₂, 63%; (*2*) allyl amine, BnCOCl, allylZnBr 4 Å MS, THF, 91%; (*3*) allylN(TMS)₂, 10% TMSOTf, CH₂Cl₂ then CH₂CHOTBS, AcCl, 78%; (*4*) MeNH₂, **95**, THF, Δ, then **94**, −78 °C, 61%; (*5*) Hoveyda–Grubbs' 2nd-gen. cat., CH₂Cl₂, styrene, then NaHMDS, 55%; (*6*) Grubbs' 2nd-gen. cat., CH₂Cl₂, then Pd[P(*t*Bu)₃]₂, DIPEA, Bu₄NCl, MeCN, MW, 120 °C, 65%; (*7*) MeNHOH·HCl, Et₃N, PhMe, Δ, 87%; (*8*) conc. HCl–MeOH (2 : 1), 71%.

the 1,3-oxygenation and the diverse stereochemistry of polyketide natural products. In addition, Tan and Ley have prepared libraries of natural-product-like acetals.[39,40]

Spring[41] has harnessed the diverse reactivity of diazo compounds in diversity-oriented synthesis (Scheme 27). Thus, the fluorous-tagged diazoacetate **110** was reacted with a reactive dipolarophile, DMAD, to give **111b**, and, under rhodium catalysis, to yield the cyclic compounds **111a** and **111c** and the β-keto esters **111d–e**. The cyclopropane **111a** is in dynamic equilibrium (by electrocyclic ring-opening) with a cycloheptatriene. The cyclopropane was, therefore, reacted directly with DMAD to yield the Diels–Alder adduct **112a**. Alternatively, methylamine was condensed with the cycloheptatriene isomer to give the alkaloid-like bicyclic compound **112b**. In addition, the β-keto esters **111d-e** were converted into heterocyclic compounds such as **113a** and **113b**. Spring's branching pathway is extremely powerful because the rich reactivity of diazo compounds allowed skeletally diverse products to be prepared by simply varying the reacting partner used. In total, a library of 223 small molecules

was prepared in which 30 distinct molecular skeletons were represented.

'Branching pathways' are probably the most powerful strategy devised to date for diversity-oriented synthesis. Provided that the substrates are designed carefully, many molecular skeletons can emerge through the use of a limited range of powerful transformations in combination. The key to the approach is the identification of starting materials, such as the alkynyl-substituted vinyl cyclopropanes **87** or the diazoacetates **110**, which are able to participate in a wide range of complexity-generating reactions.

3   Summary and outlook

In this review, we have described some of the remarkable innovations which have been devised to address the challenges posed by diversity-oriented synthesis. We have concentrated on the diversity-oriented synthesis of compound libraries which
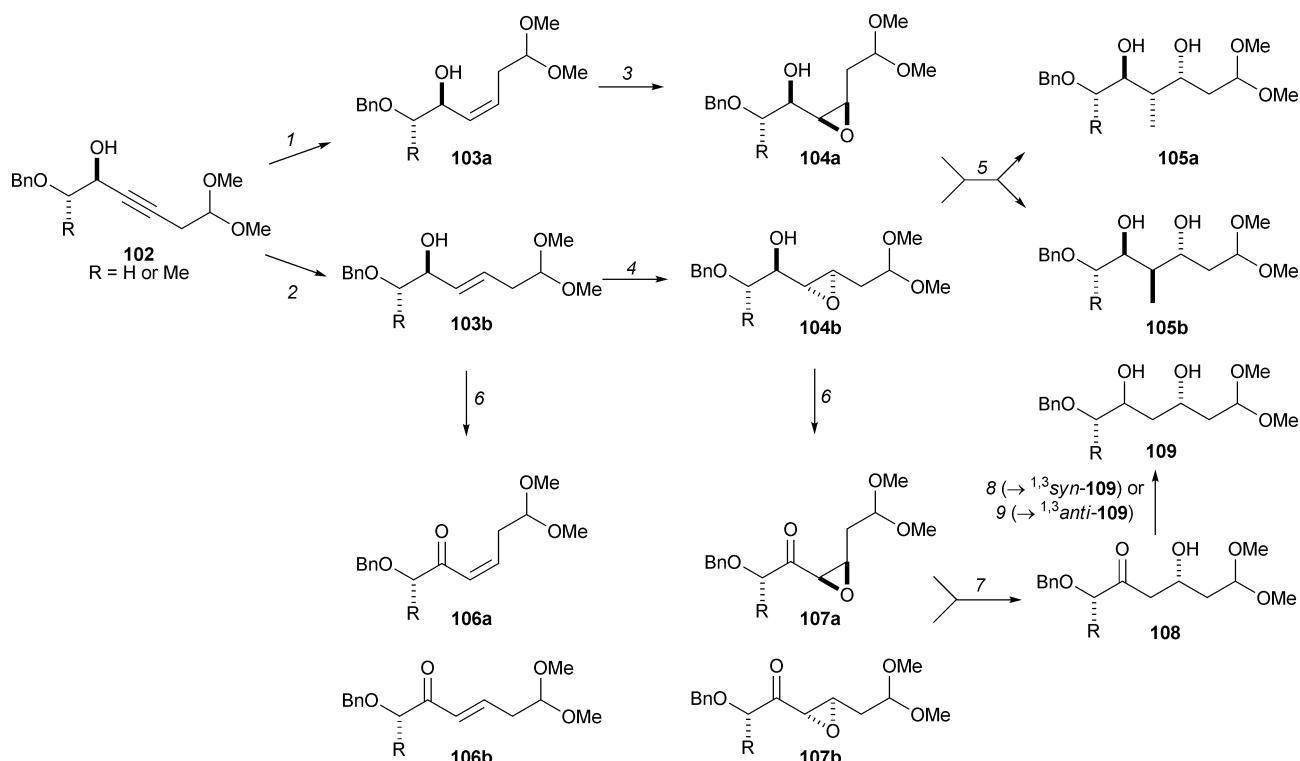
Scheme 26 Tan's branching pathways which yield stereochemically and substitutionally diverse polyketide-like molecules. *Reagents and conditions*: (*1*) Cp$_2$TiCl$_2$, *i*BuMgBr, Et$_2$O, 0 °C → rt, 81–86%, >88 : 12 *Z*:*E*; (*2*) Red-Al, Et$_2$O, 0 °C → rt, 99%, >98 : 2 *E*:*Z*; (*3*) *m*-CPBA, NaHCO$_3$, CH$_2$Cl$_2$, 0 °C → rt (R = H) or −4 °C (R = Me), 91–99%, d.r. >90 : 10; (*4*) diethyl tartrate (matched enantiomer), Ti(O*i*Pr)$_4$, *t*BuOOH, 4 Å sieves, CH$_2$Cl$_2$, −20 °C, (R = H) or −20 °C, (R = Me), 85–93%, d.r. 94 : 6; (*5*) (i) *n*BuLi, Me$_3$Al, 1,2-dichloroethane, 0 °C → rt; (ii) NaIO$_4$, MeCN–H$_2$O, 0 °C → rt, 68–69% (over 2 steps); (*6*) Dess–Martin periodinane, CH$_2$Cl$_2$, 0 °C → rt, 85–100%; (*7*) SmI$_2$, THF, −90 → −78 °C, 65–98%; (*8*) NaBH(OAc)$_3$, AcOH, MeCN, −20 °C, 87–95%, d.r. >93 : 7; (*9*) Et$_2$BOMe, NaBH$_4$.

were – in some sense – inspired by the structures of natural products.

In diversity-oriented synthesis, the most challenging issue is the synthesis of skeletally diverse molecules. A range of 'folding pathways' have been developed in which molecules with distinct scaffolds are prepared under common reaction conditions: this approach has allowed the synthesis of libraries in which a handful of scaffolds are represented. Multicomponent reactions are extremely valuable in diversity-oriented synthesis: such reactions are particularly valuable when they can be manipulated to yield alternative scaffolds, or when they can be used in combination. In diversity-oriented synthesis, there is still a heavy reliance on the Ugi reaction to prepare substrates for other complexity-generating processes. There is, thus, still an urgent need for new multicomponent reactions,[42] particularly those that yield molecules with alternative scaffolds. The 'branching pathway' strategy can allow libraries to be prepared in which tens of scaffolds are represented. A major challenge associated with developing branching pathways is to retain the ethos of diversity-oriented synthesis: that is, that diverse compound libraries are prepared using a limited repertoire of highly optimised reactions.

The success of using natural products to inspire diversity-oriented synthesis can ultimately only be gauged by the discovery of new biologically active molecules: not close derivatives of natural products with related functions, but molecules, which through populating productive regions of biologically-relevant chemical space, have novel biological functions. In this regard, the strategy of appending natural product scaffolds with diverse substituents has performed remarkably well, yielding chemical probes such as secramine (**6**), uretupamine (**11**), and haptamide B (**14b**). It is not yet possible to assess the success of the more recent innovations in diversity-oriented synthesis: in a few years' time, with the benefit of hindsight, it will be possible to assess critically the ability of these new approaches to deliver new small molecule tools for use in chemical genetic studies. It is not the structural similarity of small molecules to natural products that is ultimately important: it is discovery of valuable small molecule tools with biological functions which known natural products do not possess.

4 Acknowledgements
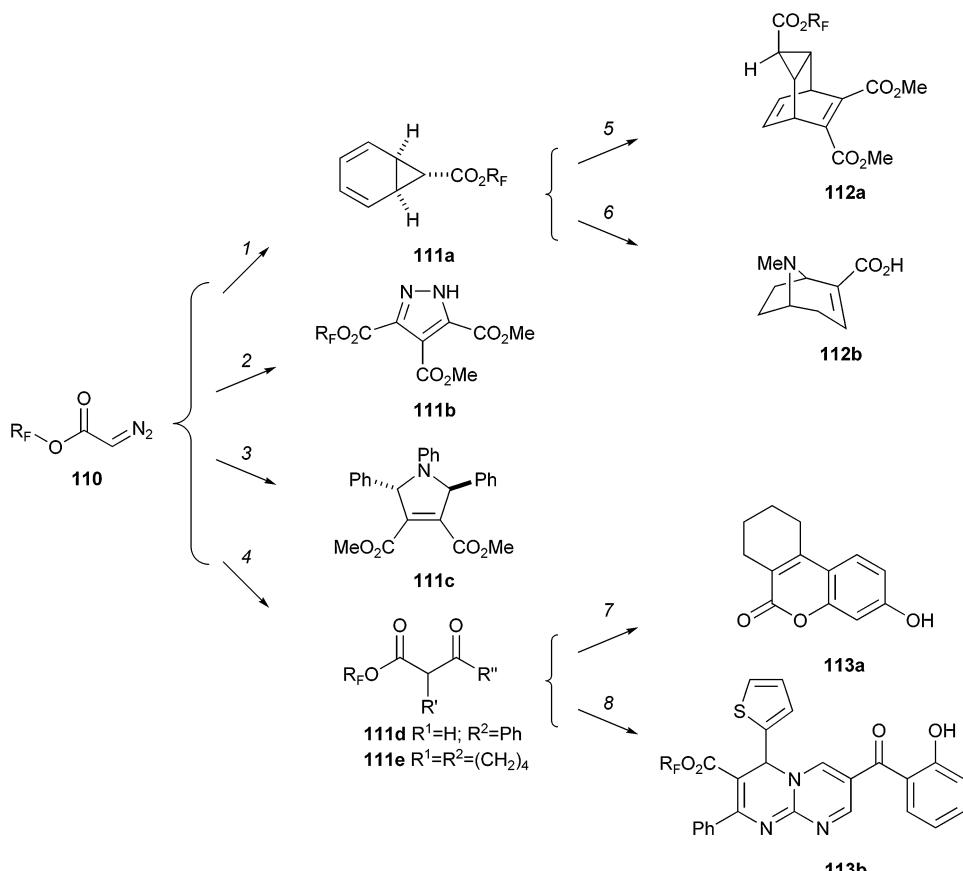
Scheme 27 Spring's branching pathways strategy using the fluorous-tagged diazoacetate **110**. *Reagents and conditions*: (*1*) C₆H₆, Rh₂(O₂CCF₃)₄, 70%; (*2*) dimethyl acetylenedicarboxylate [DMAD], 84% (88%); (*3*) PhCHO, PhNH₂, then DMAD, Rh₂(OAc)₄, d.r. 95 : 5, 51% (80%); (*4*) LDA, R¹COR², then Rh₂(OAc)₄; **111d**: 49% (90%); **111e**: 68% (97%); (*5*) DMAD, 59%; (*6*) MeNH₂, NaOH, then MeOH, H₂SO₄, 35%; (*7*) resorcinol, H₂SO₄, 74% (95%); (*8*) thiophene-2-carboxaldehyde, guanidine carbonate, then 3-formylchromone, 43% (98%). R_F = C₆F₁₃CH₂CH₂–. Numbers in parentheses are purities determined by HPLC, LC-MS or ¹H NMR spectroscopy.

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