

Biosynthesis of natural products containing β -amino acids

Cite this: *Nat. Prod. Rep.*, 2014, **31**, 1056

Fumitaka Kudo,^a Akimasa Miyanaga^a and Tadashi Eguchi^{*b}

Covering: up to January, 2014

We focus here on β -amino acids as components of complex natural products because the presence of β -amino acids produces structural diversity in natural products and provides characteristic architectures beyond those of ordinary α -L-amino acids, thus generating significant and unique biological functions in nature. In this review, we first survey the known bioactive β -amino acid-containing natural products including nonribosomal peptides, macrolactam polyketides, and nucleoside- β -amino acid hybrids. Next, the biosynthetic enzymes that form β -amino acids from α -amino acids and the *de novo* synthesis of β -amino acids are summarized. Then, the mechanisms of β -amino acid incorporation into natural products are reviewed. Because it is anticipated that the rational swapping of the β -amino acid moieties with various side chains and stereochemistries by biosynthetic engineering should lead to the creation of novel architectures and bioactive compounds, the accumulation of knowledge regarding β -amino acid-containing natural product biosynthetic machinery could have a significant impact in this field. In addition, genome mining of characteristic β -amino acid biosynthetic genes and unique β -amino acid incorporation machinery could lead to the discovery of new β -amino acid-containing natural products.

Received 20th January 2014

DOI: 10.1039/c4np00007b

www.rsc.org/npr

- 1 Introduction
- 2 Structures
 - 2.1 Nonribosomal peptides and hybrids with polyketides
 - 2.2 Macrolactam polyketides (β -amino acid starter polyketides)
 - 2.3 Polyketide- β -amino acid hybrids
 - 2.4 Terpenoid- β -amino acid hybrids
 - 2.5 Sugar- β -amino acid hybrids
- 3 Formation of β -amino acids in nature
 - 3.1 Origin of β -alanine and β -aminoisobutyrate
 - 3.2 Aminomutase-type reactions (MIO-dependent and radical SAM types)
 - 3.3 Glutamate mutase-type reactions (C-C bond rearrangements)
 - 3.4 Decarboxylation of α -aspartate, 3-methylaspartate, and β -glutamate
 - 3.5 Michael addition of ammonia or amino groups onto α,β -unsaturated carboxylates
 - 3.6 Aminotransferase-type reactions (transamination of β -ketocarboxylates)

- 4 Installation of β -amino acid into natural products
 - 4.1 Acyl transfer reactions of β -amino acids to a hydroxyl or amino group (O- and N-acylation)
 - 4.2 Macrolactam polyketides with β -amino acid starters
 - 4.3 NRPS and NRPS-PKS hybrids through enzymatic adenylation of β -amino acids (specific recognition of β -amino acids by the A domain)
 - 4.4 Amide formation of β -amino acid β -amino groups with carboxylate substrates
 - 4.5 NRPS-PKS hybrids from transamination of β -keto polyketide chains
 - 4.6 Conjugate addition of amino acids in the NRPS assembly line
- 5 Future perspectives
- 6 Acknowledgements
- 7 References

1 Introduction

β -Amino acids are rare substances by comparison with proteinogenic α -L-amino acids in nature. However, β -amino acids have been frequently found as important components of bioactive natural products, such as the anticancer agent taxol (paclitaxel)¹ produced by the western yew *Taxus brevifolia*, anticancer agent bleomycin² by *Streptomyces*, and cytotoxic microcystin³ by cyanobacteria. Incorporation of β -amino acids

^aDepartment of Chemistry, Tokyo Institute of Technology, 2-12-1, O-okayama, Meguro-ku, Tokyo 152-8551, Japan. E-mail: fkudo@chem.titech.ac.jp; Fax: +81 (0)3 5734 2607; Tel: +81 (0)3 5734 2607

^bDepartment of Chemistry and Materials Science, Tokyo Institute of Technology, 2-12-1, O-okayama, Meguro-ku, Tokyo 152-8551, Japan. E-mail: eguchi@cms.titech.ac.jp; Fax: +81 (0)3 5734 2631; Tel: +81 (0)3 5734 2631



instead of α -L-amino acids into natural products generates structurally distinctive substances with similar molecular polarity. In addition, β -amino acid-containing peptides have greater stability than regular peptides, because usual protease-type hydrolases recognize peptides consisting of proteinogenic α -L-amino acids. In most cases, the carboxylic acid moieties of β -amino acids are condensed with the amino group of other amino acids to form an amide bond, a hydroxyl group to form an ester bond, or an α -carbon atom of an enolate derived from malonyl-ACP or its derivative to produce a C–C bond (Fig. 1). Additionally, in many cases, β -amino groups of β -amino acids are masked with carboxylates in amide bonds. Thus, the functionalities of both the carboxylic acid and β -amino groups are important to the structural design of natural products.

Biosynthetic pathways for β -amino acids have been extensively studied. Spiteller has elegantly reviewed the occurrence of β -amino acids and their biosynthetic pathways in *Amino Acids, Peptides and Proteins in Organic Chemistry* in 2009,⁴ and also

summarized most of the known β -amino acid-containing natural products in *Enantioselective Synthesis of β -Amino Acids* in 2005,⁵ and in the chapter “ β -Amino Acids in Nature” in *Highlights in Bioorganic Chemistry* in 2004.⁶ In addition, Walsh and Khosla have summarized the biosynthesis of the major classes of nonproteinogenic amino acids including some β -amino acids in nonribosomal peptides and hybrids of nonribosomal peptide–polyketides in 2013.⁷ Here, as it is important to introduce key bioactive natural products containing β -amino acids including recently identified compounds, we first classify these compounds based on the natural product core structures, such as nonribosomal peptides, nonribosomal peptide–polyketide hybrids, terpenoids, and sugars. Then, the biosynthetic pathways for the β -amino acids are reviewed based on reaction types including rearrangements of the α -amino groups of α -amino acids, conjugate additions of ammonia/amino acids to α,β -unsaturated fatty acids, and transamination of β -ketocarboxylates. Finally, the mechanisms of β -amino acid incorporation into complex natural products are reviewed. Information regarding genome sequences containing unknown biosynthetic gene clusters has accumulated rapidly through sequencing of various microorganisms and plants including some from uncultured and environmental sources. Thus, this review will help in the identification of new β -amino acid-containing natural product biosynthetic machinery via a genome mining approach.

2 Structures

2.1 Nonribosomal peptides and hybrids with polyketides

β -Amino acids are most frequently found in cyclic non-ribosomal peptides and their polyketide hybrids, such as microcystin-LR,⁸ iturin A,⁹ puwainaphycin A,¹⁰ guineamide A,¹¹ dolastatin D,¹² chondramide C,¹³ jaspamide,¹⁴ cryptophycin 1,¹⁵ keramamide H,¹⁶ orbiculamide A,¹⁷ cyclotheonamide,¹⁸ capreomycin 1A,¹⁹ viomycin,²⁰ and in their linear forms, such as



Tadashi Eguchi received his B.Sc. from Yokohama City University in 1980, M.Sc. in 1982 and Ph.D. in 1990 from Tokyo Institute of Technology. After a stay in the Pharmaceutical Company, he joined Iwaki Meisei University as an assistant professor in 1987. He moved to the Tokyo Institute of Technology as an assistant professor in 1990, and was appointed to associate professor in 1995 and full professor in

2005. His research group is currently working in general research areas with a focus on both the elucidation of biosynthetic pathways of biologically active natural products and the mechanisms of novel enzymatic reactions.



Fumitaka Kudo received his B.Sc. in 1994, M.Sc. in 1996, and Ph.D. in 1999 from the Tokyo Institute of Technology under the direction of Katusumi Kakinuma. He was a post-doctoral fellow at Brown University working with David E. Cane from 1999 to 2001 and at Johns Hopkins University working with Craig A. Townsend from 2001 to 2003. In 2003, he joined the Department of

Chemistry at the Tokyo Institute of Technology as an assistant professor and was promoted to associate professor in 2010. His research focuses on bioorganic chemistry in microbial natural product biosynthesis, including aminoglycoside-aminocyclitol antibiotics and polyketide antibiotics.



Akimasa Miyanaga received his B.Sc. in 2001, M.Sc. in 2003, and Ph.D. in 2006 from The University of Tokyo under the direction of Hirofumi Shoun. He was a postdoctoral fellow at The University of Tokyo working with Sueharu Horinouchi (2006–2009) and at the University of California at San Diego working with Bradley S. Moore (2009–2011). He was appointed to assistant professor at the Tokyo

University of Science in 2011 and moved to the Tokyo Institute of Technology as an assistant professor in 2012. His research concerns biochemical and structural studies of natural product biosynthetic enzymes.



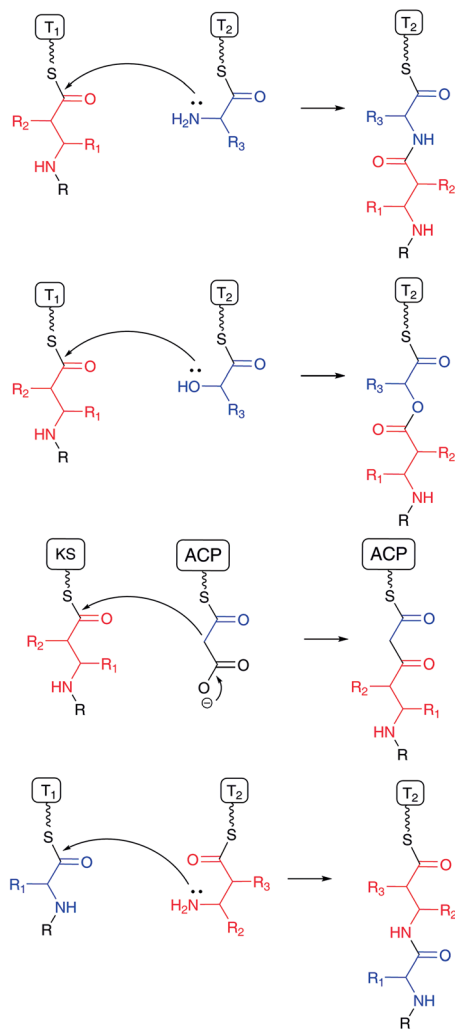


Fig. 1 Condensation mechanism of β -amino acid between carboxylates and other nucleophiles (upper three schemes), and β -amino group and carboxylate (bottom). β -Amino acids and counter substrates in red and blue, respectively.

bestatin,²¹ resormycin,²² TAN-1057 A,²³ edeine A,²⁴ andrimid,²⁵ sperabilline,²⁶ zwittermicin,²⁷ and bleomycin (Fig. 2).² Most of these have been isolated as the metabolites of various microorganisms including Cyanobacteria, *Bacillus*, Actinomycetales, Myxobacteria, and *Enterobacter*. Although some of them have been isolated from marine sources, such as sponges, symbiotic microorganisms are believed to be their true producers. Both the carboxylate and β -amino groups of β -amino acids in many cyclic nonribosomal peptides such as guineamide A are involved in the cyclic structures. Thus, it is unclear whether biosynthesis starts from β -amino acid or the other amino acids only from the chemical structures. The β -amino acid side chains of capreomycin 1A and viomycin are known to be attached after core cyclic peptide formation. Several linear-type peptide-polyketide hybrids, such as bestatin, resormycin, TAN-1057 A, and edeine, possess β -amino acids at the N-terminus. These nonribosomal peptides and their polyketide hybrids possess a wide variety of bioactivities, including anticancer,

antibacterial, and toxic effects. Thus, these compounds offer great potential as drug candidates that bind target biomolecules selectively.

2.2 Macrolactam polyketides (β -amino acid starter polyketides)

A wide variety of macrolactam compounds with β -amino acids as their polyketide starter units have been identified, mostly from Actinomycetales. Identified starter units include a β -aminoisobutyrate unit in vicenistatin²⁸ and piceamycin,²⁹ a 3-amino medium chain fatty acid unit in cremimycin,³⁰ BE-14106,³¹ aureoverticillactam,³² ML-449,³³ and heronamide,^{34,35} a β -aminobutyrate unit in incednine,³⁶ salinilactam,³⁷ and silvalactam,³⁸ a β -phenylalanine unit in hitachimycin,³⁹ and a β -alanine unit in fluvirucin (Fig. 3).⁴⁰ As some of these compounds are glycosylated and possess potent toxic activity against mammalian cells, these macrolactam scaffolds can be used to produce great pharmacological activity in potential drugs. The β -amino groups of β -amino acids are necessary to form these macrocyclic structures and thus the installation of β -amino acids into biosynthetic machinery is critical to synthesizing characteristic macromolecules. Except for cremimycin-type macrolactams with 3-amino medium chain fatty acids, β -amino acid moieties are derived from proteinogenic α -amino acids, such as L-glutamate, L-aspartate, and L-phenylalanine. The carboxylate of the β -amino acid starter unit of macrolactams is condensed with an enolate α -carbon on an ACP-tethered acyl intermediate by a Claisen-type condensation (Fig. 1). The resulting β -ketoacyl intermediate is then reduced to a hydroxyl group, enoyl group, or fully reduced methylene by the domains of PKS. Feeding experiments and enzymatic analyses have illustrated that this biosynthetic machinery contributes greatly to the chemistry of β -amino acid-containing natural products.

2.3 Polyketide- β -amino acid hybrids

The enediyne chromophore of C-1027⁴¹ (Fig. 4) contains a β -tyrosine derivative, (*S*)-3-chloro-4,5-dihydroxy- β -phenylalanine, which is directly connected to the labile enediyne structure. It is also known that a related enediyne antibiotic, maduropeptin,⁴² contains a β -tyrosine-derived unit at a similarly oriented position. In the C-1027 chromophore, the β -amino group of the β -amino acid moiety stands as a free amino group. The *p*-hydroxyl and carboxylate groups are linked with the enediyne core structure. Nakinadines⁴³ isolated from sponge *Amphimedon* sp. contain 2-phenyl-3-aminopropanoic acid whose β -amino group is alkylated with pyridine-containing alkyl chain.

2.4 Terpenoid- β -amino acid hybrids

Taxol¹ (Fig. 5) isolated from western yew stem bark is one of the most famous natural products because of its importance to anticancer therapies. In addition, due to its structural complexity, many synthetic challenges have been reported.⁴⁴ Taxol and related compounds contain a β -phenylalanine derivative at the C-13 hydroxyl group of its characteristic diterpene skeleton. In taxine A, a different position in the



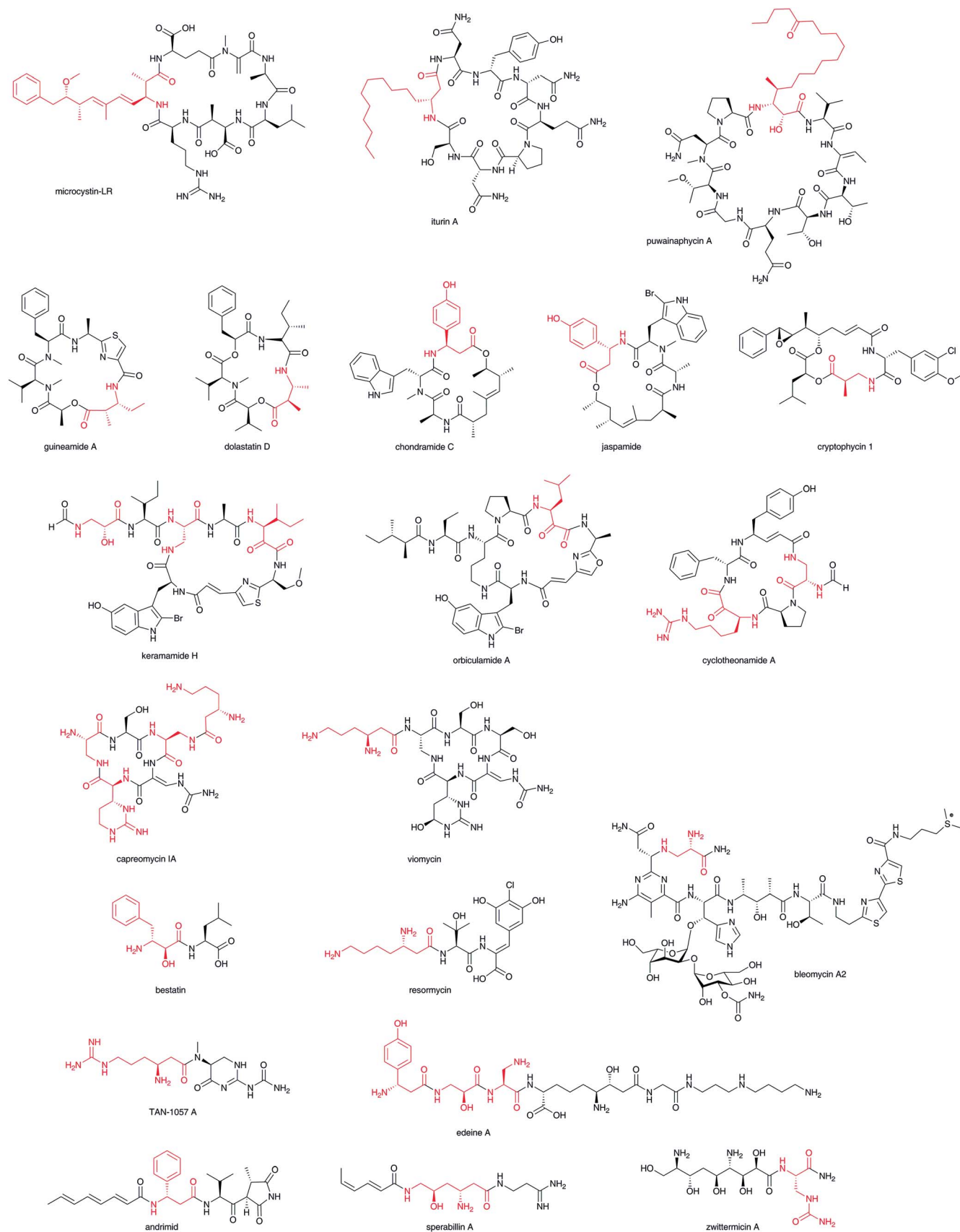


Fig. 2 Structures of β -amino acid-containing nonribosomal peptides and nonribosomal peptide–polyketide hybrids. β -Amino acid moiety in red.



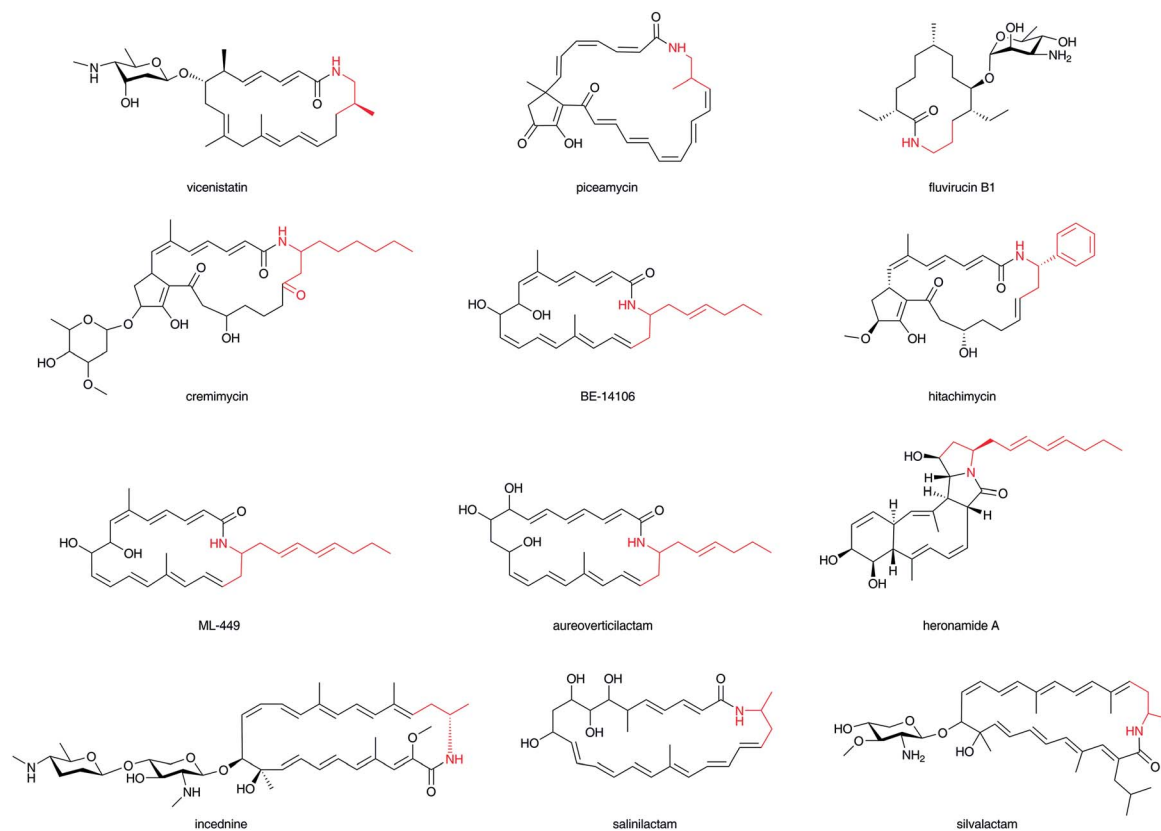


Fig. 3 Structures of macrolactam antibiotics. β -Amino acid moiety in red.

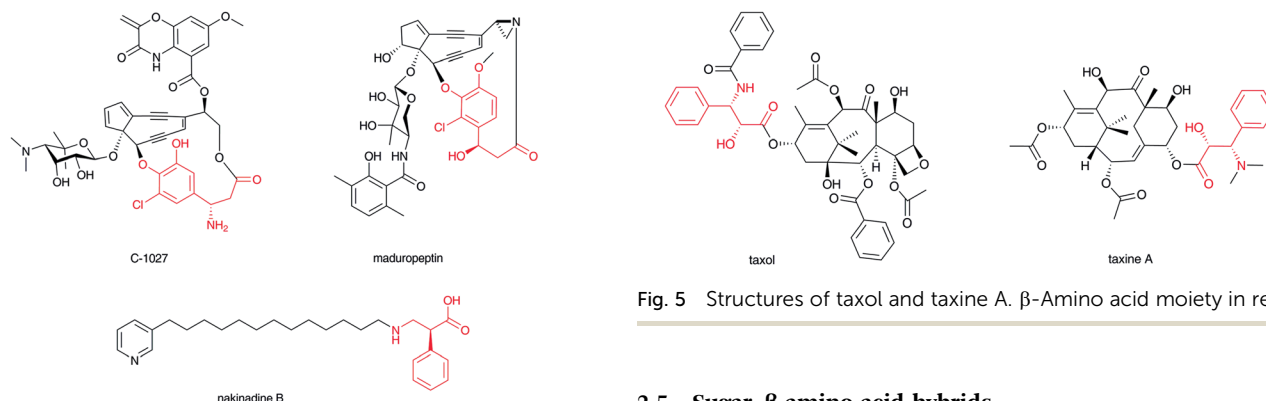


Fig. 5 Structures of taxol and taxine A. β -Amino acid moiety in red.

Fig. 4 Structures of C-1027 chromophore, maduropeptin, and nakinadine B. β -Amino acid moiety in red.

diterpene skeleton is aminoacylated with the carboxylate of the β -phenylalanine derivative.⁴⁵ The α -position of the β -phenylalanine is usually hydroxylated to form phenylisoserine and the β -amino group acylated or methylated to mask the primary amino group. To date, not many terpenoid- β -amino acid hybrid compounds have been isolated. Although terpenoid indole alkaloids are the most abundant hybrids among the family of related natural products, a β -amino acid-derived complex terpenoid alkaloid has not been recognized thus far.

2.5 Sugar- β -amino acid hybrids

The aminosugar moiety of the nucleoside antibiotic blasticidin S is decorated with a β -arginine carboxylate,⁴⁶ and the aminosugar of streptothricin with a β -lysine or poly- β -lysine carboxylate.⁴⁷ In both blasticidin S and streptothricin, the β -amino group of their β -amino acids is not acylated; a free amino group is a characteristic feature of water-soluble, basic antibiotics, as in lysinomycin⁴⁸ and myomycin,⁴⁹ which belong to this β -lysine-containing glycoside family (Fig. 6). In the pacidamycin⁵⁰ and mureidomycin-type⁵¹ of nucleoside-peptide hybrid natural product, characteristic aminosugar moieties of 3'-deoxyuridine nucleoside are connected to a carboxylate of the 2,3-diaminobutyrate moiety of peptide backbones. Both amino groups of the 2,3-diaminobutyrate moiety in this type of natural product are aminoacylated with other α -amino acids to produce



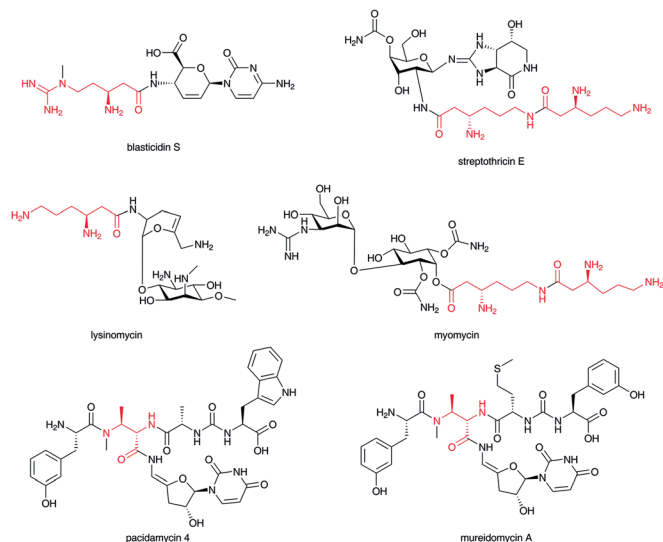


Fig. 6 Structures of sugar-β-amino acid hybrid compounds. β-Amino acid moiety in red.

polypeptide backbones. Thus, 2,3-diaminobutyrate appears to function as a crucial scaffold in this type of natural product.

3 Formation of β-amino acids in nature

It is required to construct β-amino acids before their installation into natural products. In fact, there are various enzyme-mediated mechanisms to construct β-amino acids in nature. Most of β-amino acids are derived from L-amino acids *via* intramolecular rearrangement of the α-amino group to the β-position, C–C-bond rearrangement of glutamate, decarboxylation, and Michael addition to dehydroalanine. On the other hand, the transamination of a β-keto acyl intermediate and Michael addition to an α,β-unsaturated acyl intermediate in the polyketide pathway are also involved. In this section, the detail of the formation mechanism of β-amino acids is described.

3.1 Origin of β-alanine and β-aminoisobutyrate

β-Alanine is a component of vitamin B₅ (pantothenate), which is a precursor for coenzyme A (CoA). In bacteria, β-alanine is formed from α-L-aspartate by aspartate decarboxylase (ADC),⁵² a small member of the pyruvoyl-dependent enzymes on which extensive mechanistic studies have been performed.⁵³ It is known that, in mammals, degradation of uracil and thymine leads to β-alanine and (R)-β-aminoisobutyric acid, respectively.^{4,54,55} Additionally, degradation of L-valine to (S)-β-aminoisobutyric acid in mammals is recognized.^{4,55} In yeasts, spermine is known to be a source of β-alanine *via* 3-amino-propanal.⁵⁶ Thus, β-alanine and β-aminoisobutyric acid occur widely in various organisms. The ADC homolog KirD is involved in kirromycin biosynthesis to supply β-alanine (Fig. 7).⁵⁷ This example suggests that homologous genes for ADCs are required

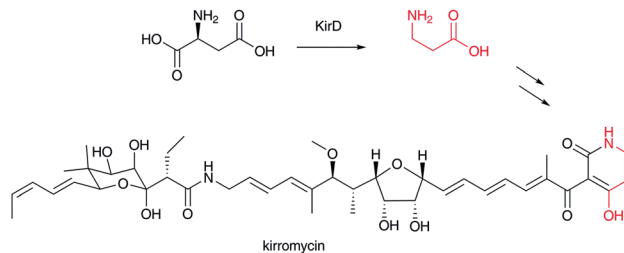


Fig. 7 Biosynthesis of kirromycin. β-Alanine moiety in red.

for secondary metabolism, although β-alanine and β-aminoisobutyrate would generally occur from primary metabolism.

3.2 Aminomutase-type reactions (MIO-dependent and radical SAM types)

Among the various routes of formation for different β-amino acids, the rearrangement of the α-amino group of proteinogenic α-L-amino acid to the β-position on the same molecule is the simplest transformation, although the reaction mechanisms are generally complex (Fig. 8). Among these, a family of phenylalanine 2,3-aminomutases (PAM) and tyrosine 2,3-aminomutases (TAM) use 4-methylideneimidazol-5-one (MIO) as a cofactor in the rearrangement reaction (Fig. 8-I).⁵⁸ The MIO moiety is formed autocatalytically from an Ala-Ser-Gly (Thr-Ser-Gly in *PaPAM*) motif at the enzyme active site. The MIO motif is also present in the ammonia lyase family, which catalyzes a half-reaction in the aminomutase reaction by releasing free ammonia.⁵⁸ The crystal structure of SgcC4 (*SgTAM*) with α,α-difluoro-β-tyrosine revealed that the MIO-amino complex was trapped as a lyase reaction intermediate.⁵⁹ Thus, the first step of this reaction is a nucleophilic attack of a tyrosine α-amino group on electrophilic MIO to generate an alkylated amino complex. Subsequent stereospecific deprotonation at C-3 of L-tyrosine and amino group elimination affords an α,β-unsaturated intermediate. The MIO amino group then prompts nucleophilic attack at the β-position of *p*-hydroxycinnamic acid to form a new C–N bond, and then, the α-position is protonated to produce a β-tyrosine–MIO complex. MIO is regenerated from the complex for the next catalytic cycle. This amino–MIO adduct reaction mechanism has been supported by another crystal structure of *PaPAM* with a phenylpropanoid-amino–MIO adduct at the active site.⁶⁰

As (S)- and (R)-β-phenylalanines (*R* is selected by *TcPAM* in taxol biosynthesis in *Taxus cuspidata*⁶¹ and *S* selected by *PaPAM* in andrimid biosynthesis in *Pantoea agglomerans*⁶²) and (S)- and (R)-β-tyrosines (*R* is selected by *CmdF* in chondramides biosynthesis in *Chondromyces crocatus*⁶³ and *S* selected by *SgTAM* in enediyne antibiotic C-1027 biosynthesis in *Streptomyces globisporus*⁶⁴) exist in nature, the stereochemistries at their β-positions should be determined according to face selectivity of the nucleophilic attack of the amino–MIO adduct on cinnamic acid and *p*-hydroxycinnamic acid by aminomutases. In the enediyne antibiotic kedarcidin biosynthesis, a rare (R)-2-aza-β-tyrosine is known to be constructed from 2-aza-



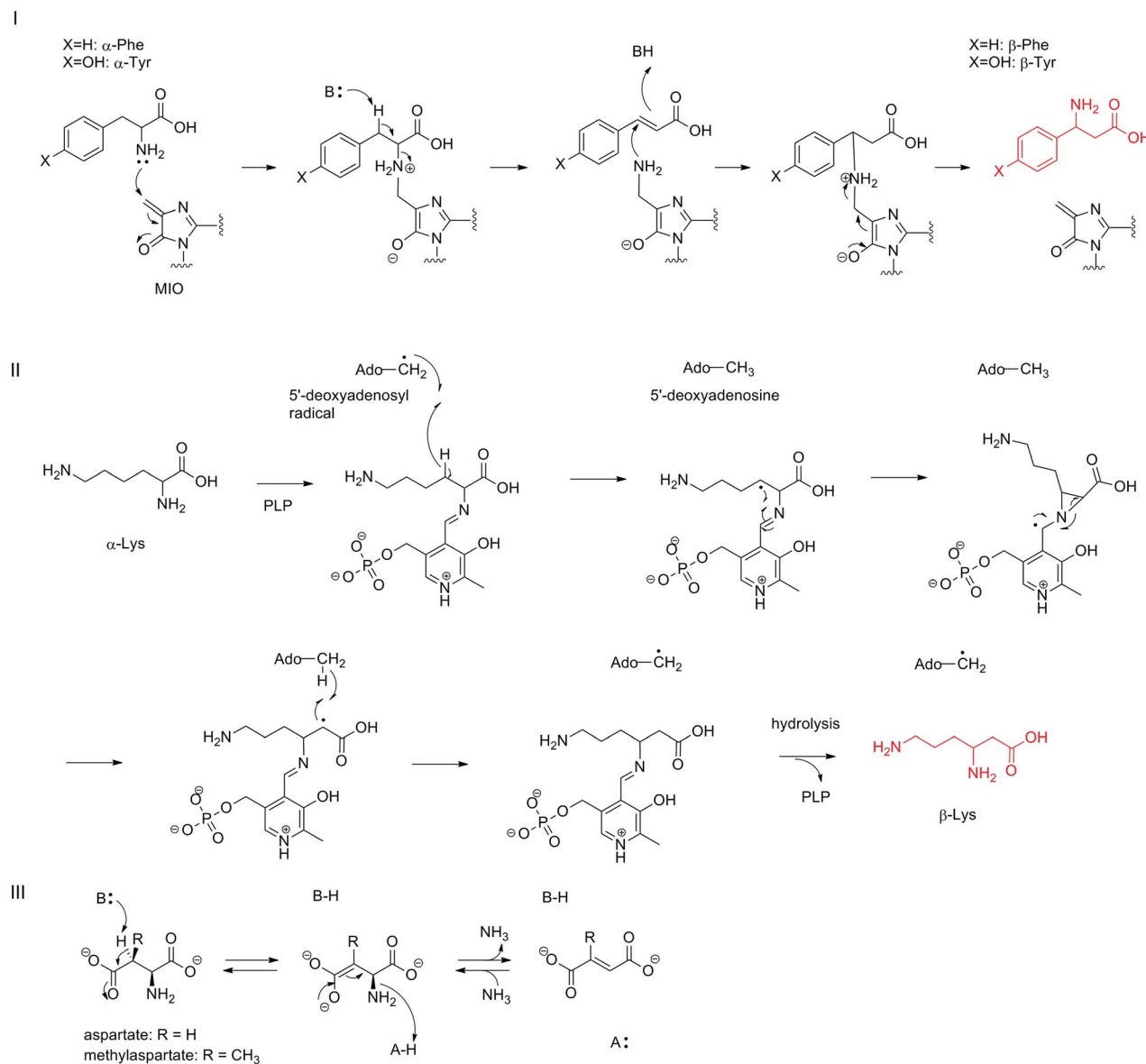


Fig. 8 Aminomutase-type reaction mechanisms: (I) MIO-dependent reaction mechanism; (II) radical SAM-type reaction mechanism; (III) the reaction mechanism of α -L-aspartate lyase.

L-tyrosine by KedY4, which belongs to this family of MIO-dependent aminomutases.⁶⁵

Another family of aminomutases, including lysine 2,3-aminomutase (LAM),⁶⁶ glutamate 2,3-aminomutase (GAM),⁶⁷ and arginine 2,3-aminomutase (AAM),⁶⁸ contains a [4Fe-4S] cluster as a radical SAM (*S*-adenosyl-L-methionine) motif and PLP (pyridoxal 5'-phosphate) binding motif (Fig. 8-II). In this reaction, an α -amino acid-PLP aldimine complex is proposed as a reaction intermediate. A hydrogen atom at the β -position of the intermediate is abstracted by a 5'-deoxyadenosyl radical generated from SAM by a reductive cleavage to yield a radical intermediate and 5'-deoxyadenosine. The radical intermediate next converts to the three-membered aziridine intermediate and the α -amino group rearranges to the β -position. The generated radical at the α -position then abstracts a hydrogen atom from 5'-deoxyadenosine to regenerate a 5'-deoxyadenosyl radical for

the next catalytic cycle. Finally, hydrolysis of β -lysyl aldimine completes the reaction. Thus far, LAM and GAM have been characterized *in vitro*.⁶⁹ β -Arginine in blasticidin S is proposed to be formed by the same enzymatic reaction because the sequentially similar gene *blsG* is encoded in the biosynthetic gene cluster.⁶⁸ β -Lysine 5,6-aminomutase is also known to catalyze an amino group rearrangement at C-6 to C-5 by a similar reaction mechanism.⁷⁰ In this reaction, adenosylcobalamin is a cofactor instead of SAM for generating a 5'-deoxyadenosyl radical, which causes a similar radical reaction to rearrange the 6-amino group to the C-5 position forming 3,5-diaminohexanoic acid. In higher plants, L-leucine 2,3-aminomutase is known to catalyze an α -amino group 2,3-rearrangement in a cobalamin-dependent manner to yield (*R*)- β -leucine, but the detailed reaction mechanism at the molecular level has not been investigated.⁷¹



It should also be noted that α -L-aspartate lyase catalyzes the removal of an α -amino group *via* an acid-base catalytic mechanism to generate fumarate (Fig. 8-III).⁷² This enzyme requires no cofactor and also catalyzes the reverse reaction to synthesize aspartate from fumarate and ammonia. One of this enzyme-type, methylaspartate ammonia lyase, has been engineered to stereoselectively produce various β -amino acids from mesaconate.⁷³

3.3 Glutamate mutase-type reactions (C-C bond rearrangements)

(2*S*,3*S*)-3-Methylaspartate (3-MeAsp) is formed by the action of adenosylcobalamin-dependent glutamate mutase (Fig. 9).⁷⁴ This reaction is related to methylmalonyl-CoA formation from succinyl-CoA by methylmalonyl-CoA mutase.^{75,76} In this reaction, adenosylcobalamin is used as a cofactor in a similar manner to the β -lysine 5,6-aminomutase reaction mentioned above. The generated 5'-deoxyadenosyl radical abstracts a hydrogen atom at the γ -position of L-glutamate to generate a radical intermediate, which causes a homolytic cleavage of the C-2-C-3 bond to form a glycy radical and acrylate as intermediates. Re-linkage between the C-2 carbon of acrylate and the glycy radical affords the primary radical intermediate of (2*S*,3*S*)-3-methylaspartate, which then abstracts a hydrogen atom from 5'-deoxyadenosine to regenerate the 5'-deoxyadenosyl radical for the next catalytic cycle. This is only one example of the generation of a β -amino acid using a C-C bond rearrangement.

3.4 Decarboxylation of α -aspartate, 3-methylaspartate, and β -glutamate

Decarboxylation of α -L-aspartate and 3-methylaspartate is one type of transformation for creating structurally different types of β -amino acids, producing β -alanine and β -aminoisobutyrate, respectively (Fig. 10). (2*S*,3*R*)-3-Methylaspartate is decarboxylated to (*R*)-3-aminoisobutyrate by the pyruvoyl-dependent enzyme CrpG in cryptophycin biosynthesis.⁷⁷ On the other hand, in vicenistatin biosynthesis, ACP-tethered (2*S*,3*S*)-3-methylaspartate is recognized by the PLP-dependent decarboxylase VinO to produce 3-aminoisobutyryl-ACP,⁷⁸ although the

precise stereochemistry has yet to be elucidated. In the fluvirucin B1 biosynthetic gene cluster, a *vinO* homologous gene *fluI* is encoded, indicating that FluI appears to be involved in ACP-tethered aspartate decarboxylation to yield a β -alanine starter unit.⁷⁹ It is noted that this family of decarboxylases is PLP-dependent not pyruvoyl-dependent as is the aforementioned ADC. In addition, aminoacyl-ACPs appear to be substrates. Thus, the free forms of β -alanine and 3-aminoisobutyric acid are not released from this biosynthetic machinery. In fact, the free form of 3-aminoisobutyric acid is not incorporated into vicenistatin.⁸⁰

In incednine biosynthesis, β -glutamate derived from α -L-glutamate by the action of glutamate 2,3-aminomutase is decarboxylated to (*S*)- β -aminobutyrate by the PLP-dependent β -glutamate decarboxylase IdnL3 (Fig. 10).⁸¹ As only one stereoisomer is generated in this reaction, IdnL3 should recognize the prochirality of β -glutamate to produce (*S*)- β -aminobutyrate. This enzyme specifically recognizes β -glutamate but not α -L-glutamate or other proteinogenic α -L-amino acids. As β -aminobutyrate is efficiently incorporated into incednine, this transformation should be unique in the incednine-producing strain. A set of glutamate 2,3-aminomutase and β -glutamate decarboxylase genes is encoded in the salinilactam³⁷ and micromonolactam⁸² biosynthetic gene clusters, suggesting that the same scenario provides β -aminobutyrate as the starter unit in these polyketide pathways.

3.5 Michael addition of ammonia or amino groups onto α,β -unsaturated carboxylates

In nature, an amino group is introduced at the β -position of a fatty acyl moiety *via* an aza-Michael addition of an ammonia molecule or an α -amino group of α -L-amino acids to the β -position of α,β -unsaturated carboxylates. Recently, β -aminononanoate formation in cremimycin biosynthesis has been found to use this type of transformation.⁸³ Here, a non-2-enoyl-ACP thioester that appears to be constructed by PKSs is used as a Michael acceptor (Fig. 11-I). A thioesterase family enzyme, CmiS1, that is similar to the fatty acyl-CoA thioesterase FcoT in *Mycobacterium tuberculosis*,⁸⁴ catalyzes a Michael addition of glycine to the non-2-enoyl-ACP thioester to yield 3-carboxymethylaminononanoate ACP thioester. This intermediate is

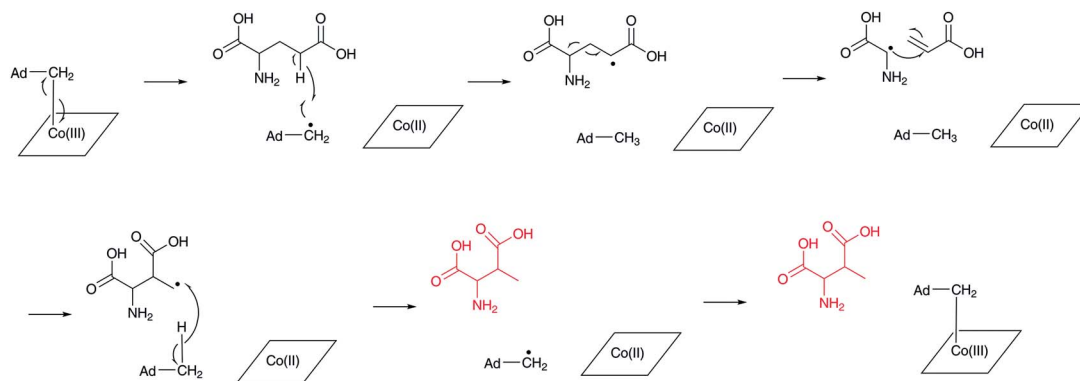


Fig. 9 Reaction mechanism of adenosylcobalamin-dependent glutamate mutase.



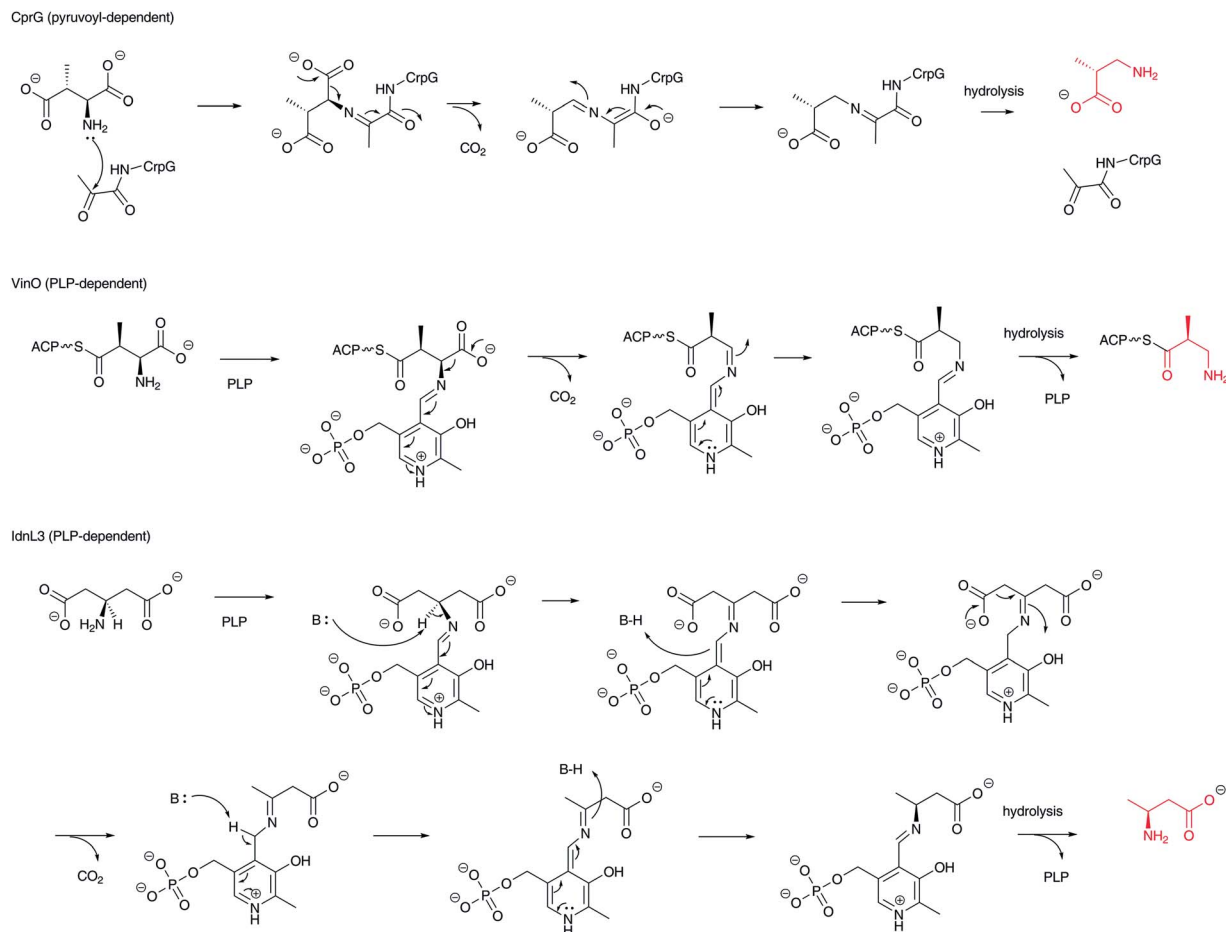


Fig. 10 Reaction mechanism of pyruvoyl-dependent and PLP-dependent decarboxylases to produce β -amino acids, such as β -aminoisobutyrate and β -aminobutyrate.

subsequently hydrolyzed to the corresponding free carboxylic acid. The resulting 3-carboxymethylaminononanoic acid is then oxidized by a glycine oxidase-type of FAD-dependent enzyme, CmiS2, to generate an imine intermediate, which is simultaneously hydrolyzed to yield 3-aminononanoate and glyoxylate. Homologous genes for *cmiS1* and *cmiS2* are conserved in related macrolactam antibiotic biosynthetic gene clusters, such as BE-14106⁸⁵ and ML-449.³³ Therefore, the same scenario is proposed for the structurally similar β -amino fatty acid biosynthesis. A relevant oxidative hydrolysis of *N*-(1-amino-1-carboxyl-2-ethyl)-glutamic acid to afford L-2,3-diaminopropionic acid and α -ketoglutarate has been reported recently.⁸⁶

2,3-Diamino acids, such as 2,3-diaminopropionic and 2,3-diaminobutanoic acids, are also believed to be constructed through a Michael addition of ammonia or an amino acid α -amino group to dehydroalanine and dehydrobutyryne, respectively. In bleomycin biosynthesis, a formed dehydroalanine on NRPS becomes a Michael acceptor and the α -amino group of α -L-asparagine on the next NRPS module attacks the C-3 position of the dehydroalanine moiety in the condensation domain to yield a formal 2,3-diaminopropanoic acid moiety (Fig. 11-II).⁸⁷ Similar scenarios for α,β -diaminopropionate formation in viomycin and capreomycin have been proposed (Fig. 11-III).⁸⁸ In

viomycin biosynthesis, the PLP-dependent enzyme VioB recognizes α -L-serine or *O*-acetylated serine to form dehydroalanine as the Michael acceptor. An associated enzyme, VioK, a putative ornithine cyclodeaminase, provides ammonia as a nucleophile to construct 2,3-diaminopropionate at the VioB active site, although the incorporation mechanism of ammonia from VioK to VioB remains unclear. The lysinoalanine formation in cinnamycin biosynthesis is a relevant mechanism to construct the 2,3-diaminopropionic acid moiety.⁸⁹ In this case, the nucleophile is the ϵ -amino group of the lysine residue in cyclic peptide. In the biosynthesis of the pacidamycin family of nucleoside antibiotics, a PLP-dependent enzyme appears to be involved in the replacement of a threonine hydroxyl group, using ammonia or an amino acid amino group as a nucleophile (Fig. 11-IV), although the precise functions of biosynthetic enzymes are unclear.^{90–92}

Similarly, the capreomycin moiety of viomycin is constructed through a dehydration catalyzed by the PLP-dependent enzyme VioD, followed by intramolecular conjugate addition from the terminal guanidino group to complete the installation of the amino group at the β -position of the amino acid.^{93–95} A similar reaction mechanism can be applied in streptolidine formation in streptothricin biosynthesis, as homologous genes



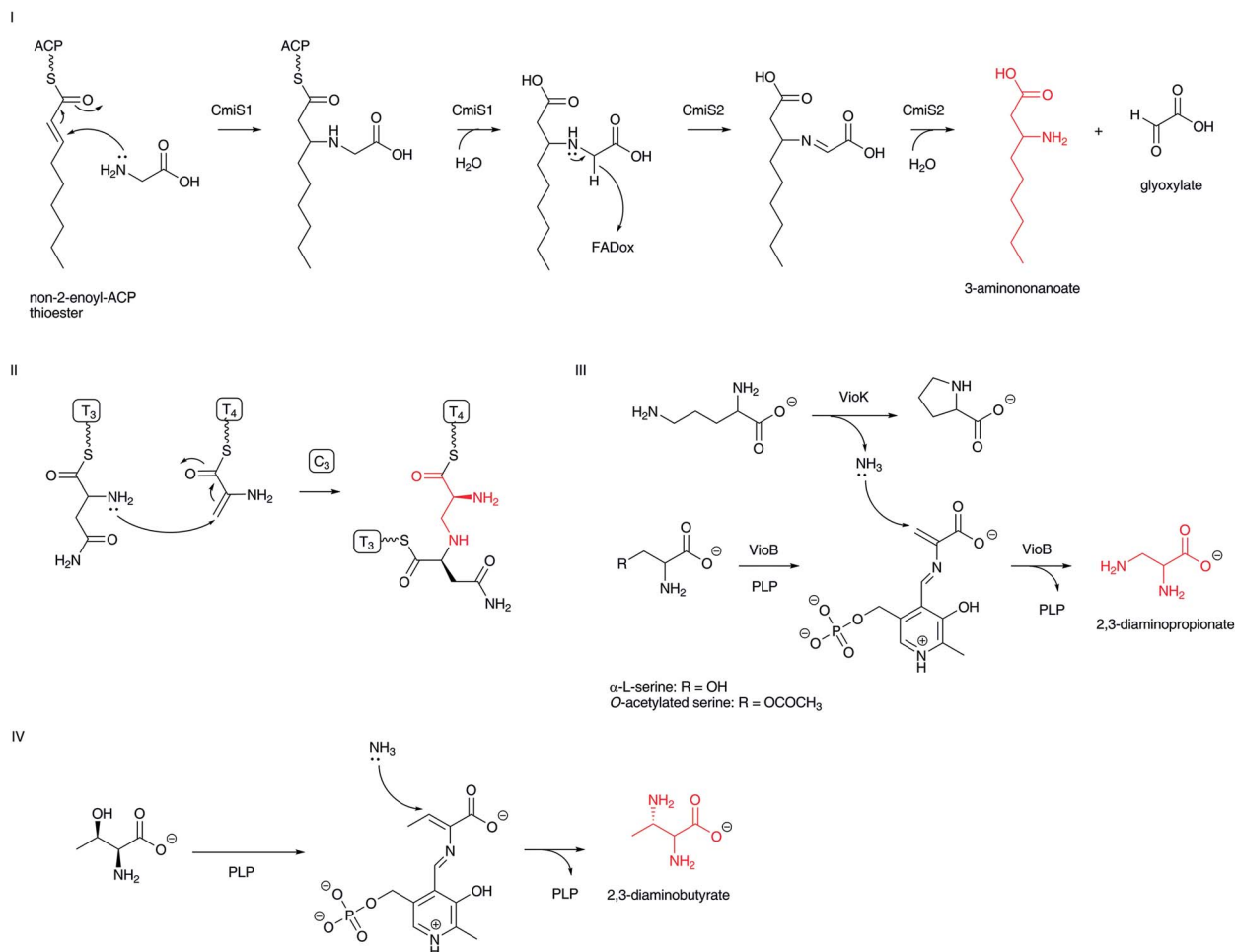


Fig. 11 Mechanism of Michael addition of ammonia or amino acid amino group to α,β -unsaturated carboxylates: (I) formation of β -amino-nonanoate in cremimycin biosynthesis; (II) formation of 2,3-diaminopropionate in bleomycin biosynthesis; (III) formation of 2,3-diaminopropionate in viomycin biosynthesis; (IV) formation of 2,3-diaminobutanoate in the biosynthesis of the pacidamycin family of nucleoside antibiotics.

are conserved in the streptothricin biosynthetic gene cluster,^{96,97} although incorporation studies have suggested that 4-hydroxycapreomycin is a true biosynthetic intermediate but not capreomycin.^{98,99}

3.6 Aminotransferase-type reactions (transamination of β -ketocarboxylates)

Transamination of α -keto acids, such as α -ketoglutarate and pyruvate, by PLP-dependent aminotransferases is well-known for installing an amino group to generate an α -L-amino acid, yielding α -L-glutamate and α -L-alanine, respectively.^{100,101} Using the same PLP chemistry, it is also possible to introduce an amino group at the β -carbonyl group of elongated β -keto fatty acids to form β -amino acids. In the biosynthesis of microcystin-type¹⁰² cyclic peptides, including nodularin,¹⁰³ iturin A,¹⁰⁴ mycosubtilin,¹⁰⁵ bacillomycin D,¹⁰⁶ nostophycin,¹⁰⁷ and microginin,¹⁰⁸ a characteristic aminotransferase domain in a PKS module catalyzes transamination to add an amino group at the β -position of polyketides to yield β -aminoacyl-ACP intermediates (Fig. 12). The aminotransferase domain shows similarity to

typical PLP-dependent aminotransferases, which use α -amino acids, such as L-glutamate and L-glutamine, as the amino donor. The free forms of β -amino fatty acids cannot be released from this type of biosynthetic assembly line. Instead, the formed β -amino fatty acyl-ACPs are condensed with another aminoacyl-ACP at a condensation domain for the next cycle of peptide elongation. Incorporation experiments of the pahayokolide producer *Lyngbya* sp. have clearly revealed that the β -carbonyl group derived from acetic acid is transaminated.^{109,110}

4 Installation of β -amino acid into natural products

Mechanisms for incorporating β -amino acids into natural products are categorized into two types: (1) installation of free form β -amino acids and (2) *de novo* synthesis of β -amino acid moieties during the biosynthetic process. The first mechanism contains: (i) simple acylation with the carboxylic acid of β -amino acids, (ii) a modified β -amino acids transfer mechanism, (iii) an adenylation-mediated transfer mechanism, and (iv)



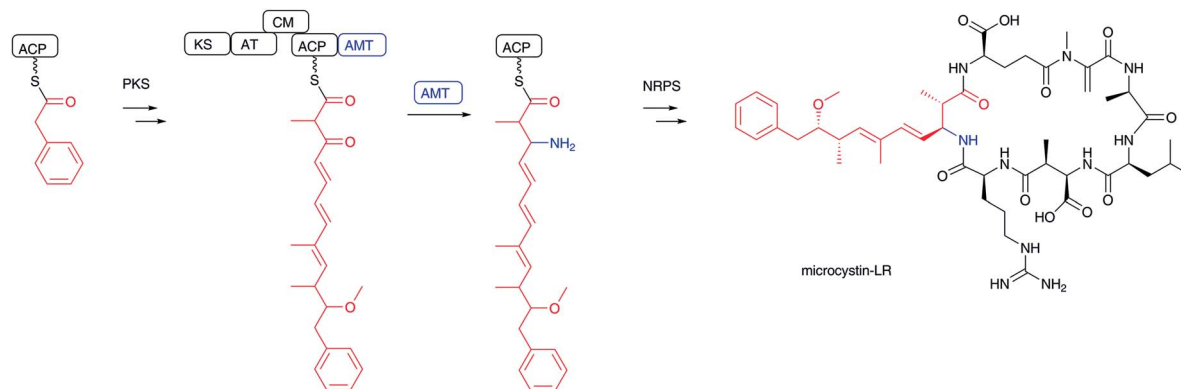


Fig. 12 Biosynthesis of microcystin-LR. β -Ketocarboxylate is transaminated by PLP-dependent aminotransferase and then incorporated into microcystin-LR.

amide bond formation with the β -amino group of β -amino acids. The second mechanism is related to the two β -amino acid formations (Michael addition and transamination) described in the previous section. The first mechanism is a quite unique system for installing β -amino acids because the biosynthetic system must discriminate between structurally related α -amino acids and β -amino acids. In fact, unique enzymes and biosynthetic machinery are involved.

4.1 Acyl transfer reactions of β -amino acids to a hydroxyl or amino group (*O*- and *N*-acylation)

Acylation of hydroxyl and amino groups using the carboxyl groups of β -amino acids is the simplest method for introducing β -amino acids into complex natural products (Fig. 13). In general, β -amino acids are initially activated as CoA thioesters by a unique β -aminoacyl-CoA ligase. Then, the β -aminoacyl moieties are transferred to the target acceptors by β -aminoacyl-CoA transferases. The combination of β -aminoacyl-CoA ligase and β -aminoacyl-CoA transferase distinguishes between α - and β -amino acids and selectively constructs a β -aminoacyl ester or amide bond. In general, this transformation occurs in the late stage of biosynthesis, adding relatively polar amino acids onto hydrophobic core structures.

In taxol biosynthesis (Fig. 13-I), a unique (*R*)- β -phenylalanyl-CoA transferase attaches (*R*)- β -phenylalanine on the hydroxyl group at C-13 of baccatin III.¹¹¹ This enzyme selectively recognizes (*R*)- β -phenylalanyl-CoA over phenylisoserine-CoA. Thus, this β -aminoacyl-CoA transferase discriminates between functional groups as well as the stereochemistry of the aminoacyl moiety of CoA substrates. Although a presumed specific (*R*)- β -phenylalanyl-CoA ligase remains to be identified from the taxol-producing yew, a family of β -aminoacyl-CoA ligases that selectively recognizes (*R*)- β -phenylalanine should be involved prior to the β -phenylalanyl-CoA transfer reaction.

In the enediyne antibiotic C-1027 biosynthesis (Fig. 13-II),¹¹² a free-standing condensation enzyme, SgcC5, catalyzes β -tyrosyl-PCP thioester transfer to the hydroxyl group of 1-phenyl-1,2-ethanediol, aminophenylethanol, and 2-phenylglycinol as enediyne core mimics.^{113,114} The β -tyrosyl-PCP thioester is generated from β -tyrosine and a stand-alone PCP, SgcC2, through

activation by a unique adenylation enzyme, SgcC1.^{115,116} The generated β -tyrosyl-PCP is modified by the chlorination enzyme SgcC3 and oxidized by SgcC to yield a (*S*)-3-chloro-4,5-dihydroxy- β -phenylalanine moiety on PCP.^{117,118} A similar scenario for the construction of a (*S*)-3-(2-chloro-3-hydroxy-4-methoxyphenyl)-3-hydroxypropionate moiety in an enediyne antibiotic maduropeptin has been proposed, although the β -tyrosine β -amino group is removed by an aminotransferase.¹¹⁹ The formal reaction scheme is the same as for taxol biosynthesis, although the enzyme types are completely different from each other.

In streptothricin biosynthesis (Fig. 13-III), a stand-alone adenylation enzyme, ORF 5, catalyzes adenylation of β -lysine and transfers it to the ORF 18 PCP-C didomain.⁹⁶ Then, the C domain of ORF 18 catalyzes a β -lysine transfer to streptothrisamine. Interestingly, another stand-alone adenylation enzyme, ORF 19, catalyzes β -lysine oligomerization and transfers it to ORF 18. Poly- β -lysine is also transferred to the core structure of the natural product.⁹⁶ A similar scenario is proposed for installation of the β -lysine moiety of viomycin. In viomycin biosynthesis, the module VioO, containing adenylation and PCP domains, synthesizes β -lysyl-PCP, which is then transferred to the cyclic peptide core structure by the stand-alone condensation enzyme VioM.⁸⁸ In tuberactinomycin and capreomycin biosyntheses,¹²⁰ homologous genes are encoded that are responsible for β -lysine installation. Overall, the combination of β -amino acyl-CoA ligase/adenylation and β -aminoacyltransferase/condensation enzymes plays a key role for installing unique β -amino acids into matured complex natural products in the late stages of biosynthesis.

4.2 Macrolactam polyketides with β -amino acid starters

Among the β -amino acid-containing macrocyclic lactams mentioned in Section 2.2, the biosynthetic gene cluster for vicenistatin was first determined in 2004¹²¹ and the functions of the biosynthetic enzymes extensively analyzed (Fig. 14).⁷⁸ Glutamate mutase S subunit VinH and E subunit VinI are presumed to be responsible for the formation of (2*S*,3*S*)-3-MeAsp, which was incorporated into vicenistatin in feeding experiments.⁸⁰ (2*S*,3*S*)-3-MeAsp is then recognized by an ATP-dependent ligase, VinN, and transferred onto a stand-alone



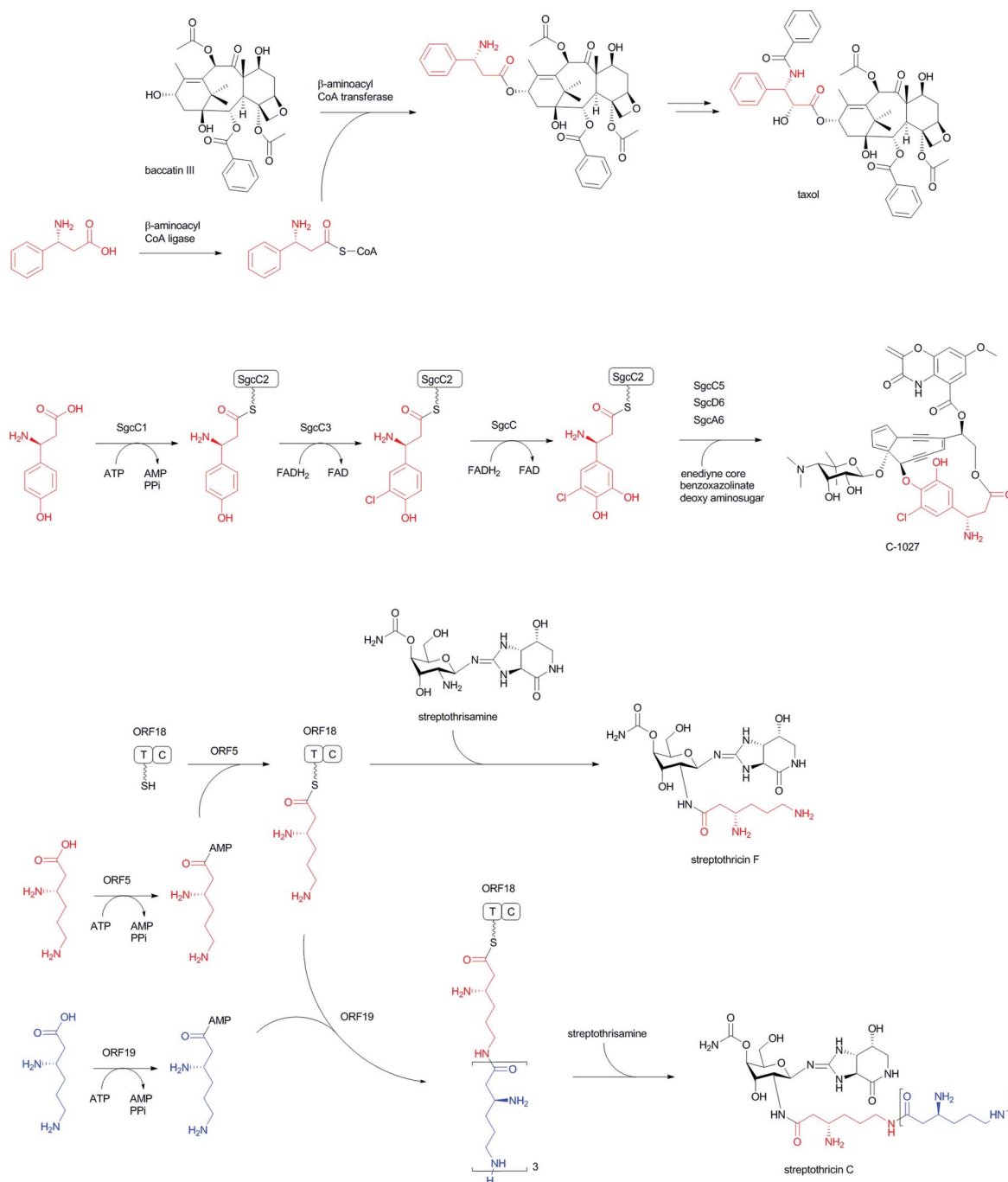
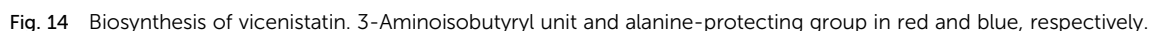


Fig. 13 Acyl transfer reactions of β -amino acids to hydroxyl or amino group acceptors: (I) installation of β -phenylalanine into taxol; (II) installation of (*S*)-3-chloro-4,5-dihydroxy- β -phenylalanine into C-1027; (III) installation of poly- β -lysine into streptothricin. β -Lysine (in red) activated by ORF 5 is initially loaded on the T (PCP) domain of ORF 18. β -Lysine molecules (in blue) activated by ORF 19 are then used as extender units for elongation reactions of β -lysine oligopeptide.

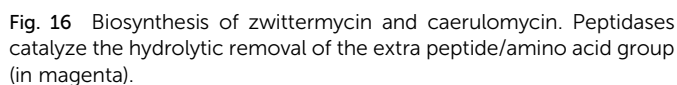
ACP, VinL, to produce (2*S*,3*S*)-3-MeAsp-ACP.⁷⁸ VinN recognizes 3-MeAsp selectively over α -L-amino and 3-aminoisobutyric acids, which were not incorporated into vicanistatin in feeding experiments. Decarboxylation of the 3-MeAsp moiety of the formed 3-MeAsp-ACP occurs by the action of the PLP-dependent decarboxylase VinO, which thus appears to recognize not only the 3-MeAsp moiety but also ACP as a substrate. The formed 3-aminoisobutyryl-ACP is then further aminoacylated with α -L-

alanine by another ATP-dependent ligase, VinM, to yield dipeptidyl-ACP. The acyltransferase VinK then selectively recognizes the dipeptidyl-ACP over 3-aminoisobutyryl-ACP and 3-MeAsp-ACP and transfers the dipeptide moiety to the ACP domain at the PKS loading module. Presumably, the alanyl attachment is due to the protection of the amino group of 3-MeAsp-ACP to prevent formation of a thermodynamically stable six-membered lactam during the polyketide elongation at the





Related amide bond hydrolysis mechanisms in late-stage biosynthesis have also been recently recognized.¹²² In xenocoumacin,¹²³ zwittermicin,¹²⁴ and colibactin^{125,126} biosyntheses, N-terminal signal peptides are trimmed by peptidase enzymes to yield matured bioactive natural products. In caerulomycin biosynthesis, a peptidase enzyme removes the C-terminal



amino acid.^{127,128} Therefore, this type of peptidase has a key role in controlling the final structures of natural products (Fig. 16).

4.3 NRPS and NRPS-PKS hybrids through enzymatic adenylation of β -amino acids (specific recognition of β -amino acids by the A domain)

Some nonribosomal peptides and their polyketide hybrids contain a β -amino acid as an extender unit, including β -alanine in destruxin,¹²⁹ (*R*)-3-aminoisobutyrate/ β -alanine in cryptophycins,¹³⁰ β -tyrosine in chondramide,¹³¹ β -phenylalanine in andrimid,^{132,133} and 3-amino-2-methylbutanoic acid in dolastatin D (Fig. 2). In viomycin biosynthesis,⁸⁸ two β -amino acids, 2,3-diaminopropionic acid and capreomycinidine, are incorporated through the NRPS system. In edeine, three β -amino acids, β -tyrosine, isoserine, and 2,3-diaminopropionic acid, are linked to produce a β -tripeptide moiety.

In most cases, free β -amino acids are activated by an adenylation enzyme and transferred to the PCP/ACP domain according to NRPS chemistry (Fig. 17). The β -amino group of β -amino acyl-ACP serves as a nucleophile to form an amide bond with an elongated amino acid/peptide or polyketide on the former ACP domain. Activated carboxylate, as a thioester, is detached from the ACP domain by nucleophilic attack of an amino acid amino group or hydroxyl acid hydroxyl group on the next module. Sometimes, the thioester bond is cleaved by a hydroxyl group on a parent polyketide chain backbone or an enolate of a β -ketocarbonyl moiety of a polyketide chain. In andrimid biosynthesis, the transglutaminase-type enzyme AdmF, instead of a condensation domain, catalyzes the condensation between β -phenylalanyl-ACP and octatrienoyl-ACP.¹³³

The mechanism for the activation of select β -amino acids and their transfer to ACP is the same as in the aforementioned macrolactam biosynthesis. In the NRPS/NRPS-PKS hybrid system, the β -amino group of β -aminoacyl-ACP subsequently becomes a nucleophile to form a new amide bond with elongated peptidyl-PCP or acyl-ACP. Conversely, in macrolactam biosynthesis, a β -amino acid β -amino group is masked with a small amino acid, such as alanine and glycine, and the protected β -amino group is regenerated before macrocyclization. In either case, selective β -amino acid activation by pathway-specific adenylation enzymes is important for installing unique β -amino acids into natural products.

In streptolydigin biosynthesis, (2*S*,3*S*)-3-MeAsp is used as a component of a β -amino acid-polyketide hybrid molecule.^{134,135} In this case, the adenylation enzyme SlgN1 activates the 3-MeAsp α -amino group.¹³⁶

4.4 Amide formation of β -amino acid β -amino groups with carboxylate substrates

In dapdiamide biosynthesis by *Pantoea agglomerans*, 2,3-diaminopropionate (DAP) and fumarate are first condensed to produce fumaroyl-DAP (Fig. 18). This amide formation is catalyzed by an ATP-dependent amide ligase, DdaG, which activates the fumarate carboxylic acid as an adenylate intermediate for subsequent nucleophilic substitution with the DAP β -amino group, forming an amide bond.¹³⁷ In this case, DAP is not

activated as in the aforementioned adenylation mechanism. The carboxylate of the fumaroyl moiety in fumaroyl-DAP is amidated to fumaramoyl-DAP, which is then condensed by another ATP-dependent amide ligase, DdaF, with valine, isoleucine, and leucine to yield dapdiamide A, B, and C, respectively.

4.5 NRPS-PKS hybrids from transamination of β -keto polyketide chains

As described above, during PKS-mediated polyketide chain elongation, an amino group is introduced at the β -keto position of an elongated polyketide chain. Unique aminotransferase domains have been identified in the biosynthetic gene clusters for microcystin,¹⁰² nodularin,^{103,138} mycosubtilin,¹⁰⁵ iturin,¹³⁹ bacillomycin D,¹⁰⁶ and nostophycin.¹⁴⁰ Here, a β -amino acid-adenylation enzyme is not required and a condensation domain in the subsequent module might recognize the β -amino acid moiety to form a new amide bond. These PKS module-generated β -amino acids should not be released from the biosynthetic assembly line as free amino acids. The PKS domain organization located upstream of a β -amino group introduction domain has a large influence on the structure of β -amino acid moieties.

4.6 Conjugate addition of amino acids in the NRPS assembly line

A β -amino acid β -amino group is also introduced during chain elongation of nonribosomal peptides, as in the biosynthesis of bleomycin (Fig. 11-II).¹⁴¹ The double bonds of dehydroalanine and dehydrobutyrine on ACP/PCP serve as Michael acceptors, to which ammonia or an amino acid amino group on another ACP bring about nucleophilic attack. This reaction thus introduces an amino group at the C-3 position of the dehydroalanine and dehydrobutyrine moieties. In this case, these β -amino acids are not released from the biosynthetic assembly line as free amino acids. Although it remains unclear how this type of aza-Michael addition is catalyzed, it is presumed that the condensation domain adjusts the position of the Michael acceptor toward the nucleophilic ammonia or amino acid. A dehydration enzyme for generating a 2,3-double bond and an associated enzyme for providing ammonia might play key roles in introducing an amino group at the C-3 position of the β -aminoacyl moiety.

5 Future perspectives

Selective installation of β -amino acids into natural biosynthetic assembly lines by pathway engineering should be a powerful tool for creating novel bioactive substances. The synthetic preparation of β -amino acids, extensively developed by many researchers, can be one way to provide various β -amino acids for natural product assembly lines. For example, chemical reductive amination of β -ketoesters, synthesized from various lengths of carboxylic acids by coupling with Meldrum's acid, should afford various lengths of β -amino fatty acids. Indeed, characteristic β -amino acid-forming enzymes, such as PAM and GAM, can be used. The recognition mechanisms of β -amino acid-



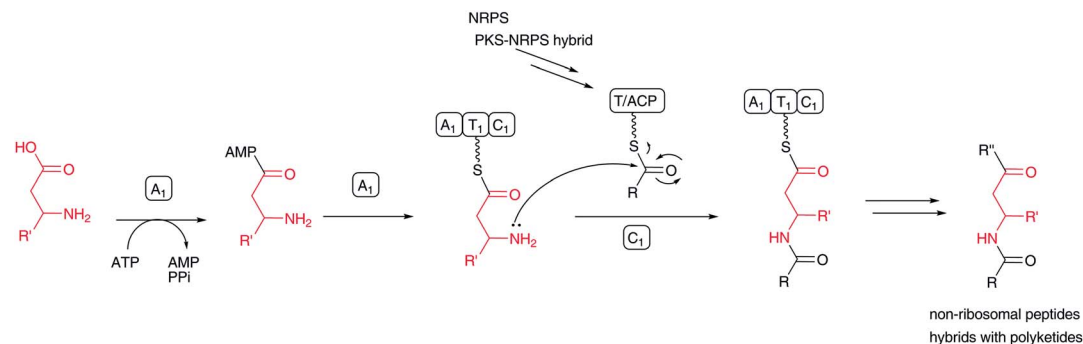


Fig. 17 Biosynthesis of β -amino acid-containing nonribosomal peptides and nonribosomal peptide-polyketide hybrids.

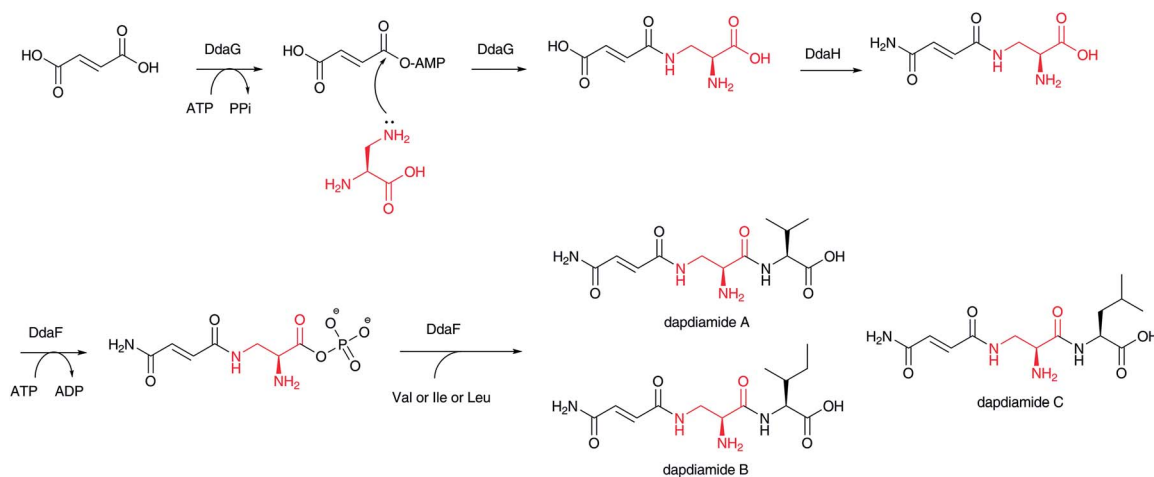


Fig. 18 Biosynthesis of dapdiamides.

activating enzymes are important to determine selective incorporation for hybridizing unique β -amino acids and the core structures of natural products. It is still unclear how β -amino acid-adenylation enzymes selectively recognize substrates. Crystal structural analysis of such enzymes should help to understand the recognition mechanism. Furthermore, β -amino acid-transfer enzymes, such as CoA transferase and the condensation domain, have certain selectivity in distinguishing from proteogenic α -L-amino acids. The specific combination of acceptor molecules is also important. Protein engineering of the β -amino acid-adenylation enzyme and CoA ligase could lead to alterations in substrate specificity to introduce β -amino acids instead of α -amino acids.

A genome mining approach is indeed an effective way to identify new natural products. The unique enzymes for β -amino acid-forming and activation enzymes described in this review should be excellent clues for detecting biosynthetic machinery for new or different types of β -amino acid-containing natural products. As β -amino acid-containing natural products historically have had great potential as bioactive molecules, further investigation using our current knowledge of β -amino acid-containing natural product biosyntheses should lead to the discovery of novel and useful bioactive substances for sustaining human health.

6 Acknowledgements

This work was supported in part by Grants-in-Aid for Scientific Research on Innovative Areas from the Ministry of Education, Culture, Sports, Science, and Technology (MEXT), Nagase Science and Technology Foundation, and Takeda Science Foundation.

7 References

- 1 M. C. Wani, H. L. Taylor, M. E. Wall, P. Coggon and A. T. McPhail, *J. Am. Chem. Soc.*, 1971, **93**, 2325–2327.
- 2 H. Umezawa, K. Maeda, T. Takeuchi and Y. Okami, *J. Antibiot.*, 1966, **19**, 200–209.
- 3 D. P. Botes, A. A. Tuinman, P. L. Wessels, C. C. Viljoen, H. Kruger, D. H. Williams, S. Santikarn, R. J. Smith and S. J. Hammond, *J. Chem. Soc., Perkin Trans. 1*, 1984, 2311–2318.
- 4 P. Spiteller, *Amino Acids, Peptides and Proteins in Organic Chemistry*, 2009, vol. 1, pp. 119–161.
- 5 P. Spiteller and F. Von Nussbaum, *Enantioselective Synthesis of β -Amino Acids*, 2nd edn, 2005, pp. 19–91.
- 6 F. von Nussbaum and P. Spiteller, in *Highlights in Bioorganic Chemistry*, 2004, pp. 63–89.



- 7 C. T. Walsh, R. V. O'Brien and C. Khosla, *Angew. Chem., Int. Ed.*, 2013, **52**, 7098–7124.
- 8 D. P. Botes, P. L. Wessels, H. Kruger, M. T. C. Runnegar, S. Santikarn, R. J. Smith, J. C. J. Barna and D. H. Williams, *J. Chem. Soc., Perkin Trans. 1*, 1985, 2747–2748.
- 9 F. Besson, F. Peypoux, G. Michel and L. Delcambe, *J. Antibiot.*, 1976, **29**, 1043–1049.
- 10 J. M. Gregson, J. L. Chen, G. M. L. Patterson and R. E. Moore, *Tetrahedron*, 1992, **48**, 3727–3734.
- 11 L. T. Tan, N. Sitachitta and W. H. Gerwick, *J. Nat. Prod.*, 2003, **66**, 764–771.
- 12 H. Sone, T. Nemoto, H. Ishiwata, M. Ojika and K. Yamada, *Tetrahedron Lett.*, 1993, **34**, 8449–8452.
- 13 B. Kunze, R. Jansen, F. Sasse, G. Hofle and H. Reichenbach, *J. Antibiot.*, 1995, **48**, 1262–1266.
- 14 T. M. Zabriskie, J. A. Klocke, C. M. Ireland, A. H. Marcus, T. F. Molinski, D. J. Faulkner, C. Xu and J. Clardy, *J. Am. Chem. Soc.*, 1986, **108**, 3123–3124.
- 15 T. Golakoti, J. Ogino, C. E. Heltzel, T. Le Husebo, C. M. Jensen, L. K. Larsen, G. M. L. Patterson, R. E. Moore, S. L. Mooberry, *et al.*, *J. Am. Chem. Soc.*, 1995, **117**, 12030–12049.
- 16 J. Kobayashi, F. Itagaki, H. Shigemori, T. Takao and Y. Shimonishi, *Tetrahedron*, 1995, **51**, 2525–2532.
- 17 N. Fusetani, T. Sugawara, S. Matsunaga and H. Hirota, *J. Am. Chem. Soc.*, 1991, **113**, 7811–7812.
- 18 N. Fusetani, S. Matsunaga, H. Matsumoto and Y. Takebayashi, *J. Am. Chem. Soc.*, 1990, **112**, 7053–7054.
- 19 T. Shiba, S. Nomoto and T. Wakamiya, *Experientia*, 1976, **32**, 1109–1111.
- 20 A. C. Finlay, G. L. Hobby, F. Hoschstein, T. M. Lees, T. F. Lenert, J. A. Means, S. Y. P'An, P. P. Regna, J. B. Routien, B. A. Sobin, K. B. Tate and J. H. Kane, *Am. Rev. Tuberc.*, 1951, **63**, 1–3.
- 21 H. Suda, T. Takita, T. Aoyagi and H. Umezawa, *J. Antibiot.*, 1976, **29**, 100–101.
- 22 M. Igarashi, H. Nakamura, H. Naganawa and T. Takeuchi, *J. Antibiot.*, 1997, **50**, 1026–1031.
- 23 Y. Funabashi, S. Tsubotani, K. Koyama, N. Katayama and S. Harada, *Tetrahedron*, 1993, **49**, 13–28.
- 24 T. P. Hettinger and L. C. Craig, *Biochemistry*, 1970, **9**, 1224–1232.
- 25 A. Fredenhagen, S. Y. Tamura, P. T. M. Kenny, H. Komura, Y. Naya, K. Nakanishi, K. Nishiyama, M. Sugiura and H. Kita, *J. Am. Chem. Soc.*, 1987, **109**, 4409–4411.
- 26 N. Katayama, Y. Nozaki, S. Tsubotani, M. Kondo, S. Harada and H. Ono, *J. Antibiot.*, 1992, **45**, 10–19.
- 27 H. He, L. A. Silo-Suh, J. Handelsman and J. Clardy, *Tetrahedron Lett.*, 1994, **35**, 2499–2502.
- 28 K. Shindo, M. Kamishohara, A. Odagawa, M. Matsuoaka and H. Kawai, *J. Antibiot.*, 1993, **46**, 1076–1081.
- 29 D. Schulz, J. Nachtigall, J. Riedlinger, K. Schneider, K. Poralla, J. F. Imhoff, W. Beil, G. Nicholson, H. P. Fiedler and R. D. Sussmuth, *J. Antibiot.*, 2009, **62**, 513–518.
- 30 M. Igarashi, T. Tsuchida, N. Kinoshita, M. Kamijima, R. Sawa, T. Sawa, H. Naganawa, M. Hamada, T. Takeuchi, K. Yamazaki and M. Ishizuka, *J. Antibiot.*, 1998, **51**, 123–129.
- 31 K. Kojiri, S. Nakajima, H. Suzuki, H. Kondo and H. Suda, *J. Antibiot.*, 1992, **45**, 868–874.
- 32 S. S. Mitchell, B. Nicholson, S. Teisan, K. S. Lam and B. C. Potts, *J. Nat. Prod.*, 2004, **67**, 1400–1402.
- 33 H. Jorgensen, K. F. Degnes, A. Dikiy, E. Fjaervik, G. Klinkenberg and S. B. Zotchev, *Appl. Environ. Microbiol.*, 2010, **76**, 283–293.
- 34 R. Raju, A. M. Piggott, M. M. Conte and R. J. Capon, *Org. Biomol. Chem.*, 2010, **8**, 4682–4689.
- 35 R. Sugiyama, S. Nishimura and H. Kakeya, *Tetrahedron Lett.*, 2013, **54**, 1531–1533.
- 36 Y. Futamura, R. Sawa, Y. Umezawa, M. Igarashi, H. Nakamura, K. Hasegawa, M. Yamasaki, E. Tashiro, Y. Takahashi, Y. Akamatsu and M. Imoto, *J. Am. Chem. Soc.*, 2008, **130**, 1822–1823.
- 37 D. W. Udway, L. Zeigler, R. N. Asolkar, V. Singan, A. Lapidus, W. Fenical, P. R. Jensen and B. S. Moore, *Proc. Natl. Acad. Sci. U. S. A.*, 2007, **104**, 10376–10381.
- 38 D. Schulz, J. Nachtigall, U. Geisen, H. Kalthoff, J. F. Imhoff, H. P. Fiedler and R. D. Sussmuth, *J. Antibiot.*, 2012, **65**, 369–372.
- 39 S. Omura, A. Nakagawa, K. Shibata and H. Sano, *Tetrahedron Lett.*, 1982, **23**, 4713–4716.
- 40 N. Naruse, T. Tsuno, Y. Sawada, M. Konishi and T. Oki, *J. Antibiot.*, 1991, **44**, 741–755.
- 41 K. Yoshida, Y. Minami, R. Azuma, M. Saeki and T. Otani, *Tetrahedron Lett.*, 1993, **34**, 2637–2640.
- 42 D. R. Schroeder, K. L. Colson, S. E. Klohr, N. Zein, D. R. Langley, M. S. Lee, J. A. Matson and T. W. Doyle, *J. Am. Chem. Soc.*, 1994, **116**, 9351–9352.
- 43 T. Nishi, T. Kubota, J. Fromont, T. Sasaki and J. Kobayashi, *Tetrahedron*, 2008, **64**, 3127–3132.
- 44 K. C. Nicolaou, C. R. H. Hale and C. Nilewski, *Chem. Rec.*, 2012, **12**, 407–441.
- 45 E. Graf, A. Kirfel, G. J. Wolff and E. Breitmaier, *Liebigs Ann. Chem.*, 1982, 376–381.
- 46 S. Takeuchi, K. Hirayama, K. Ueda, H. Sakai and H. Yonehara, *J. Antibiot.*, 1958, **11**, 1–5.
- 47 S. Kusumoto, Y. Kambayashi, S. Imaoka, K. Shima and T. Shiba, *J. Antibiot.*, 1982, **35**, 925–927.
- 48 P. Kurath, W. Rosenbrook, Jr., D. A. Dunnigan, J. B. McAlpine, R. S. Egan, R. S. Stanaszek, M. Cirovic, S. L. Mueller and W. H. Washburn, *J. Antibiot.*, 1984, **37**, 1130–1143.
- 49 J. C. French, Q. R. Bartz and H. W. Dion, *J. Antibiot.*, 1973, **26**, 272–283.
- 50 R. H. Chen, A. M. Buko, D. N. Whittern and J. B. McAlpine, *J. Antibiot.*, 1989, **42**, 512–520.
- 51 F. Isono, M. Inukai, S. Takahashi, T. Haneishi, T. Kinoshita and H. Kuwano, *J. Antibiot.*, 1989, **42**, 667–673.
- 52 U. Genschel, *Mol. Biol. Evol.*, 2004, **21**, 1242–1251.
- 53 M. E. Webb, A. G. Smith and C. Abell, *Nat. Prod. Rep.*, 2004, **21**, 695–721.
- 54 K. D. Schnackerz and D. Dobritzsch, *Biochim. Biophys. Acta, Proteins Proteomics*, 2008, **1784**, 431–444.
- 55 A. B. van Kuilenburg, A. E. Stroomeer, H. van Lenthe, N. G. Abeling and A. H. van Gennip, *Biochem. J.*, 2004, **379**, 119–124.



- 56 Q. Huang, Q. Liu and Q. Hao, *J. Mol. Biol.*, 2005, **348**, 951–959.
- 57 K. J. Laible, T. Hartner, H. P. Fiedler, W. Wohlleben and T. Weber, *J. Antibiot.*, 2009, **62**, 465–468.
- 58 N. J. Turner, *Curr. Opin. Chem. Biol.*, 2011, **15**, 234–240.
- 59 C. V. Christianson, T. J. Montavon, G. M. Festin, H. A. Cooke, B. Shen and S. D. Bruner, *J. Am. Chem. Soc.*, 2007, **129**, 15744–15745.
- 60 S. Strom, U. Wanninayake, N. D. Ratnayake, K. D. Walker and J. H. Geiger, *Angew. Chem., Int. Ed.*, 2012, **51**, 2898–2902.
- 61 K. D. Walker, K. Klettke, T. Akiyama and R. Croteau, *J. Biol. Chem.*, 2004, **279**, 53947–53954.
- 62 N. D. Ratnayake, U. Wanninayake, J. H. Geiger and K. D. Walker, *J. Am. Chem. Soc.*, 2011, **133**, 8531–8533.
- 63 S. Rachid, D. Krug, K. J. Weissman and R. Müller, *J. Biol. Chem.*, 2007, **282**, 21810–21817.
- 64 S. D. Christenson, W. Liu, M. D. Toney and B. Shen, *J. Am. Chem. Soc.*, 2003, **125**, 6062–6063.
- 65 S. X. Huang, J. R. Lohman, T. Huang and B. Shen, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 8069–8074.
- 66 T. P. Chirpich, V. Zappia, R. N. Costilow and H. A. Barker, *J. Biol. Chem.*, 1970, **245**, 1778–1789.
- 67 F. J. Ruzicka and P. A. Frey, *Biochim. Biophys. Acta, Proteins Proteomics*, 2007, **1774**, 286–296.
- 68 M. C. Cone, X. Yin, L. L. Grochowski, M. R. Parker and T. M. Zabriskie, *ChemBioChem*, 2003, **4**, 821–828.
- 69 P. A. Frey and G. H. Reed, *Biochim. Biophys. Acta, Proteins Proteomics*, 2011, **1814**, 1548–1557.
- 70 F. Berkovitch, E. Behshad, K. H. Tang, E. A. Enns, P. A. Frey and C. L. Drennan, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 15870–15875.
- 71 J. M. Poston, *Science*, 1977, **195**, 301–302.
- 72 M. de Villiers, V. Puthan Veetil, H. Raj, J. de Villiers and G. J. Poelarends, *ACS Chem. Biol.*, 2012, **7**, 1618–1628.
- 73 H. Raj, W. Szymanski, J. de Villiers, H. J. Rozeboom, V. P. Veetil, C. R. Reis, M. de Villiers, F. J. Dekker, S. de Wildeman, W. J. Quax, A. M. Thunnissen, B. L. Feringa, D. B. Janssen and G. J. Poelarends, *Nat. Chem.*, 2012, **4**, 478–484.
- 74 K. Gruber and C. Kratky, *Curr. Opin. Chem. Biol.*, 2002, **6**, 598–603.
- 75 R. Banerjee, *Chem. Rev.*, 2003, **103**, 2083–2094.
- 76 G. H. Reed, *Curr. Opin. Chem. Biol.*, 2004, **8**, 477–483.
- 77 Z. Q. Beck, D. A. Burr and D. H. Sherman, *ChemBioChem*, 2007, **8**, 1373–1375.
- 78 Y. Shinohara, F. Kudo and T. Eguchi, *J. Am. Chem. Soc.*, 2011, **133**, 18134–18137.
- 79 T. Y. Lin, L. S. Borketey, G. Prasad, S. A. Waters and N. A. Schnarr, *ACS Synth. Biol.*, 2013, **2**, 635–642.
- 80 H. Nishida, T. Eguchi and K. Kakinuma, *Tetrahedron*, 2001, **57**, 8237–8242.
- 81 M. Takaishi, F. Kudo and T. Eguchi, *J. Antibiot.*, 2013, **66**, 691–699.
- 82 E. J. Skellam, A. K. Stewart, W. K. Strangman and J. L. Wright, *J. Antibiot.*, 2013, **66**, 431–441.
- 83 K. Amagai, R. Takaku, F. Kudo and T. Eguchi, *ChemBioChem*, 2013, **14**, 1998–2006.
- 84 F. Wang, R. Langley, G. Gulten, L. Wang and J. C. Sacchettini, *Chem. Biol.*, 2007, **14**, 543–551.
- 85 H. Jorgensen, K. F. Degnes, H. Sletta, E. Fjaervik, A. Dikiy, L. Herfindal, P. Bruheim, G. Klinkenberg, H. Bredholt, G. Nygard, S. O. Doskeland, T. E. Ellingsen and S. B. Zotchev, *Chem. Biol.*, 2009, **16**, 1109–1121.
- 86 M. J. Kobylarz, J. C. Grigg, S. J. Takayama, D. K. Rai, D. E. Heinrichs and M. E. Murphy, *Chem. Biol.*, 2014, **21**, 379–388.
- 87 L. Du, C. Sanchez, M. Chen, D. J. Edwards and B. Shen, *Chem. Biol.*, 2000, **7**, 623–642.
- 88 M. G. Thomas, Y. A. Chan and S. G. Ozanick, *Antimicrob. Agents Chemother.*, 2003, **47**, 2823–2830.
- 89 A. Okesli, L. E. Cooper, E. J. Fogle and W. A. van der Donk, *J. Am. Chem. Soc.*, 2011, **133**, 13753–13760.
- 90 W. H. Lam, K. Rychli and T. D. Bugg, *Org. Biomol. Chem.*, 2008, **6**, 1912–1917.
- 91 E. J. Rackham, S. Gruschow, A. E. Ragab, S. Dickens and R. J. Goss, *ChemBioChem*, 2010, **11**, 1700–1709.
- 92 W. Zhang, B. Ostash and C. T. Walsh, *Proc. Natl. Acad. Sci. U. S. A.*, 2010, **107**, 16828–16833.
- 93 X. Yin, K. L. McPhail, K. J. Kim and T. M. Zabriskie, *ChemBioChem*, 2004, **5**, 1278–1281.
- 94 J. Ju, S. G. Ozanick, B. Shen and M. G. Thomas, *ChemBioChem*, 2004, **5**, 1281–1285.
- 95 X. Yin and T. M. Zabriskie, *ChemBioChem*, 2004, **5**, 1274–1277.
- 96 C. Maruyama, J. Toyoda, Y. Kato, M. Izumikawa, M. Takagi, K. Shin-ya, H. Katano, T. Utagawa and Y. Hamano, *Nat. Chem. Biol.*, 2012, **8**, 791–797.
- 97 J. Li, Z. Guo, W. Huang, X. Meng, G. Ai, G. Tang and Y. Chen, *Sci. China, Ser. C: Life Sci.*, 2013, **56**, 619–627.
- 98 M. D. Jackson, S. J. Gould and T. M. Zabriskie, *J. Org. Chem.*, 2002, **67**, 2934–2941.
- 99 C. Y. Chang, S. Y. Lyu, Y. C. Liu, N. S. Hsu, C. C. Wu, C. F. Tang, K. H. Lin, J. Y. Ho, C. J. Wu, M. D. Tsai and T. L. Li, *Angew. Chem., Int. Ed.*, 2014, **53**, 1943–1948.
- 100 M. D. Toney, *Biochim. Biophys. Acta, Proteins Proteomics*, 2011, **1814**, 1405–1406.
- 101 M. D. Toney, *Biochim. Biophys. Acta, Proteins Proteomics*, 2011, **1814**, 1407–1418.
- 102 D. Tillett, E. Dittmann, M. Erhard, H. von Dohren, T. Borner and B. A. Neilan, *Chem. Biol.*, 2000, **7**, 753–764.
- 103 M. C. Moffitt and B. A. Neilan, *Appl. Environ. Microbiol.*, 2004, **70**, 6353–6362.
- 104 K. Tsuge, T. Ano and M. Shoda, *Arch. Microbiol.*, 1996, **165**, 243–251.
- 105 E. H. Duitman, L. W. Hamoen, M. Rembold, G. Venema, H. Seitz, W. Saenger, F. Bernhard, R. Reinhardt, M. Schmidt, C. Ullrich, T. Stein, F. Leenders and J. Vater, *Proc. Natl. Acad. Sci. U. S. A.*, 1999, **96**, 13294–13299.
- 106 A. L. Moyne, T. E. Cleveland and S. Tuzun, *FEMS Microbiol. Lett.*, 2004, **234**, 43–49.
- 107 D. P. Fewer, J. Osterholm, L. Rouhiainen, J. Jokela, M. Wahlsten and K. Sivonen, *Appl. Environ. Microbiol.*, 2011, **77**, 8034–8040.



- 108 T. B. Rounge, T. Rohrlack, A. J. Nederbragt, T. Kristensen and K. S. Jakobsen, *BMC Genomics*, 2009, **10**, 396.
- 109 L. Liu, D. W. Bearden, J. C. Rodriguez and K. S. Rein, *Tetrahedron Lett.*, 2012, **53**, 6758–6760.
- 110 L. Liu, D. W. Bearden and K. S. Rein, *J. Nat. Prod.*, 2011, **74**, 1535–1538.
- 111 K. Walker, S. Fujisaki, R. Long and R. Croteau, *Proc. Natl. Acad. Sci. U. S. A.*, 2002, **99**, 12715–12720.
- 112 W. Liu, S. D. Christenson, S. Standage and B. Shen, *Science*, 2002, **297**, 1170–1173.
- 113 S. Lin, S. G. Van Lanen and B. Shen, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 4183–4188.
- 114 S. Lin, T. Huang, G. P. Horsman, S. X. Huang, X. Guo and B. Shen, *Org. Lett.*, 2012, **14**, 2300–2303.
- 115 S. G. Van Lanen, P. C. Dorrestein, S. D. Christenson, W. Liu, J. Ju, N. L. Kelleher and B. Shen, *J. Am. Chem. Soc.*, 2005, **127**, 11594–11595.
- 116 S. G. Van Lanen, S. Lin, P. C. Dorrestein, N. L. Kelleher and B. Shen, *J. Biol. Chem.*, 2006, **281**, 29633–29640.
- 117 S. Lin, S. G. Van Lanen and B. Shen, *J. Am. Chem. Soc.*, 2007, **129**, 12432–12438.
- 118 S. Lin, S. G. Van Lanen and B. Shen, *J. Am. Chem. Soc.*, 2008, **130**, 6616–6623.
- 119 S. G. Van Lanen, T. J. Oh, W. Liu, E. Wendt-Pienkowski and B. Shen, *J. Am. Chem. Soc.*, 2007, **129**, 13082–13094.
- 120 E. A. Felnagle, M. R. Rondon, A. D. Berti, H. A. Crosby and M. G. Thomas, *Appl. Environ. Microbiol.*, 2007, **73**, 4162–4170.
- 121 Y. Ogasawara, K. Katayama, A. Minami, M. Otsuka, T. Eguchi and K. Kakinuma, *Chem. Biol.*, 2004, **11**, 79–86.
- 122 D. Reimer and H. B. Bode, *Nat. Prod. Rep.*, 2014, **31**, 154–159.
- 123 D. Reimer, K. M. Pos, M. Thines, P. Grun and H. B. Bode, *Nat. Chem. Biol.*, 2011, **7**, 888–890.
- 124 B. M. Kevany, D. A. Rasko and M. G. Thomas, *Appl. Environ. Microbiol.*, 2009, **75**, 1144–1155.
- 125 C. A. Brotherton and E. P. Balskus, *J. Am. Chem. Soc.*, 2013, **135**, 3359–3362.
- 126 X. Bian, J. Fu, A. Plaza, J. Herrmann, D. Pistorius, A. F. Stewart, Y. Zhang and R. Müller, *ChemBioChem*, 2013, **14**, 1194–1197.
- 127 Y. Zhu, P. Fu, Q. Lin, G. Zhang, H. Zhang, S. Li, J. Ju, W. Zhu and C. Zhang, *Org. Lett.*, 2012, **14**, 2666–2669.
- 128 I. Garcia, N. M. Vior, A. F. Brana, J. Gonzalez-Sabin, J. Rohr, F. Moris, C. Mendez and J. A. Salas, *Chem. Biol.*, 2012, **19**, 399–413.
- 129 B. Wang, Q. Kang, Y. Lu, L. Bai and C. Wang, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 1287–1292.
- 130 N. A. Magarvey, Z. Q. Beck, T. Golakoti, Y. Ding, U. Huber, T. K. Hemscheidt, D. Abelson, R. E. Moore and D. H. Sherman, *ACS Chem. Biol.*, 2006, **1**, 766–779.
- 131 S. Rachid, D. Krug, B. Kunze, I. Kochems, M. Scharfe, T. M. Zabriskie, H. Blocker and R. Müller, *Chem. Biol.*, 2006, **13**, 667–681.
- 132 M. Jin, M. A. Fischbach and J. Clardy, *J. Am. Chem. Soc.*, 2006, **128**, 10660–10661.
- 133 P. D. Fortin, C. T. Walsh and N. A. Magarvey, *Nature*, 2007, **448**, 824–827.
- 134 C. Gomez, D. H. Horna, C. Olano, M. Palomino-Schatzlein, A. Pineda-Lucena, R. J. Carbajo, A. F. Brana, C. Mendez and J. A. Salas, *J. Bacteriol.*, 2011, **193**, 4214–4223.
- 135 C. Olano, C. Gomez, M. Perez, M. Palomino, A. Pineda-Lucena, R. J. Carbajo, A. F. Brana, C. Mendez and J. A. Salas, *Chem. Biol.*, 2009, **16**, 1031–1044.
- 136 D. A. Herbst, B. Boll, G. Zocher, T. Stehle and L. Heide, *J. Biol. Chem.*, 2013, **288**, 1991–2003.
- 137 M. A. Hollenhorst, J. Clardy and C. T. Walsh, *Biochemistry*, 2009, **48**, 10467–10472.
- 138 M. M. Gehringer, L. Adler, A. A. Roberts, M. C. Moffitt, T. K. Mihali, T. J. Mills, C. Fieker and B. A. Neilan, *ISME J.*, 2012, **6**, 1834–1847.
- 139 K. Tsuge, T. Akiyama and M. Shoda, *J. Bacteriol.*, 2001, **183**, 6265–6273.
- 140 D. P. Fewer, J. Osterholm, L. Rouhiainen, J. Jokela, M. Wahlsten and K. Sivonen, *Appl. Environ. Microbiol.*, 2011, **77**, 8034–8040.
- 141 U. Galm, E. Wendt-Pienkowski, L. Wang, S. X. Huang, C. Unsin, M. Tao, J. M. Coughlin and B. Shen, *J. Nat. Prod.*, 2011, **74**, 526–536.

