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Synthesis of polysubstituted arenes through organocatalytic benzannulation

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Polysubstituted arenes serve as ubiquitous structural cores of aromatic compounds with significant applications in chemistry, biological science, and materials science. Among all the synthetic approaches toward these highly functionalized arenes, organocatalytic benzannulation represents one of the most efficient and versatile transformations in the assembly of structurally diverse arene architectures under mild conditions with exceptional chemo-, regio- or stereoselectivities. Thus, the development of new benzannulation reactions through organocatalysis has attracted much attention in the past ten years. This review systemically presents recent advances in the organocatalytic benzannulation strategies, categorized as follows: (1) Brønsted acid-catalysis, (2) secondary amine catalysis, (3) primary amine catalysis, (4) tertiary amine catalysis, (5) tertiary phosphine catalysis, and (6) N-heterocyclic carbene catalysis. Each part is further classified into several types according to the number of carbon atoms contributed by different synthons participating in the cyclization reaction. The reaction mechanisms involved in different benzannulation strategies were highlighted.

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1 Introduction

Polysubstituted arenes represent a core structural framework that ubiquitously occurs in various natural products, pharmaceuticals, and agrochemicals. They are also universal synthetic building blocks in preparing most of today's industrial chemicals and functional materials.¹ The functions of these molecules are directly associated with the substituents on the benzene core.² Therefore, flexible substitution patterns are highly desirable for the functionalized benzenes, and the synthesis of various multi-substituted benzenes is a long-standing focus and has drawn much attention in the organic chemistry community.³ Fig. 1 shows several selected molecules containing a highly functionalized benzene.

Synthetic methodologies for the construction of polysubstituted benzenes mainly rely on the modification of existing six-membered carbocyclic compounds or arene substrates, such as catalytic dehydrogenation of carbocyclic rings,⁴ electrophilic and nucleophilic aromatic substitution reactions,⁵ as well as transition-metal-catalyzed cross-coupling reactions⁶ or direct C–H functionalization of arenes⁷ (Scheme 1). Despite providing impressive progresses for benzene substitutions, some of these approaches may suffer from expensive metal catalysts, harsh reaction conditions, multistep synthetic

sequences to introduce directing or functional groups into the parent benzene rings, and, in particular, the limited substitution patterns. Besides, different functional groups on the aromatic cores may hugely affect both the selectivity and reactivity of these reactions.

Benzannulation is a method to assemble the arene rings through C–C bonds formation between two or more appropriate building blocks. It is a valid alternative strategy for the synthesis of diverse polyfunctionalized arenes, because these approaches

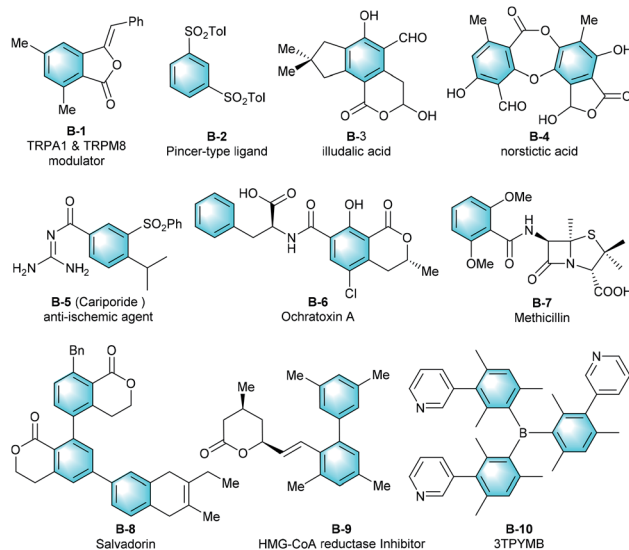


Fig. 1 Examples of multi-substituted benzenes.

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Scheme 1 Strategies for the synthesis of polysubstituted arenes.

could reduce the number of steps and usually exhibit better regioselectivities and flexible substitution patterns. These reactions have been summarized in several reviews.⁸ In 2008, Wessig and Müller reviewed the dehydro-Diels–Alder reactions, which mainly provided benzannulation products.^{8a} In the same year, Kotha and co-workers comprehensively discussed different types of benzannulation reactions for the creation of benzene rings, including Diels–Alder reaction, ring-closing metathesis (RCM), cycloaddition reactions and other transition-metal-promoted processes.^{8b} In 2017, Dichtel and co-workers summarized the transition-metal-catalyzed benzannulation reactions of alkynes.^{8c} Later, Lee *et al.* reviewed the various base-mediated benzannulations under transition-metal-free conditions, such as Cs_2CO_3 , *t*-BuOK, and DBU-catalyzed and -mediated reactions.^{8m}

In recent ten years, from the standpoint of green and sustainable chemistry, the ongoing research on the construction of the functionalized benzene compounds through organocatalytic benzannulation strategies has emerged as a hot topic in organic synthesis. Taking advantage of various organocatalysts, different activation modes, and versatile annulation strategies, the scope of benzannulation approaches were remarkably broadened. Comparing with the conventional benzannulation methods, these approaches feature numerous beneficial attributes, including low economic cost, mild reaction conditions, good functional group compatibility, flexible substitution patterns, high regio- and stereoselectivity (Scheme 1). Organocatalytic benzannulation process can also avoid the problems of removing metal residues or directing groups. On the other hand, organocatalytic benzannulation is a useful tool for the construction of various biaryl scaffolds, which are essential structural motifs found in natural products, biologically active compounds, and functional materials.⁹ Comparing with the transition-metal-catalyzed cross-coupling or C–H arylation strategies,¹⁰ which form the aryl–aryl bonds, these tactics enables the facile access to biaryls from a wide range of acyclic starting materials with more flexible substitution patterns and excellent tolerance of functional groups.

Despite the great advance of organocatalytic benzannulation strategies over the last decade, no comprehensive review has been devoted to this topic.¹¹ In this review, we systemically present recent advances in the organocatalytic benzannulation strategies categorized as follows: (1) Brønsted acid-catalysis, (2) secondary amine catalysis, (3) primary amine catalysis, (4)

tertiary amine catalysis, (5) tertiary phosphine catalysis, and (6) N-heterocyclic carbene catalysis. To discuss the reactions in a more logical way, each part is further classified into several types according to the number of carbon atoms contributed by different synthons participating in the cyclization reaction, such as [4 + 2], [3 + 3], [3 + 2 + 1], [2 + 2 + 2] annulation, *etc.* (Scheme 1). The reaction mechanisms involved in different benzannulation strategies were highlighted and enantioselective versions were emphasized. Organocatalytic strategies for the synthesis of heteroaromatics and the aromatization using quinones as the cyclic precursors would not be mentioned in this review. The literature has been surveyed mainly from 2010 to April of 2020. We believe that this review will be instructive for researchers in different fields and will inspire further reaction design and developments.

2 Brønsted acid-catalyzed benzannulation

Brønsted acids have been widely used as efficient and eco-friendly organocatalysts. Over recent decades, significant progress has been achieved in the Brønsted acid-catalyzed annulation reactions due to their versatile potential in the activation of carbonyl groups, hydroxyl groups, imines, and amines. In particular, the Brønsted acid-catalyzed benzannulation has provided a powerful approach to the synthetically useful polysubstituted arenes. In this section, we will systemically discuss the application of the Brønsted acid-catalysis strategy in benzannulation reactions.

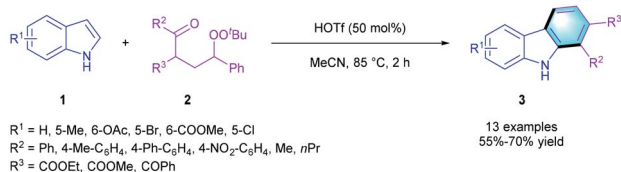
2.1 [4 + 2] benzannulation

Because of the extensive application of [4 + 2] annulation in the *de novo* assembly of new arenes, these methods have drawn much attention. Considering the privileged biological activities of highly functionalized aromatic compounds like polysubstituted carbazoles and indolocarbazoles, tremendous efforts have been made to construct these useful building blocks from simple starting materials.¹² In 2014, the Li group found that trifluoromethanesulfonic acid (HOTf) could catalyze the cyclization between indole **1** and γ -carbonyl *tert*-butylperoxide **2** (Scheme 2).¹³ The corresponding polysubstituted carbazole derivatives **3** were obtained in 55–70% yields. Moreover, this transformation of indole-to-carbazole was successfully applied to the total synthesis of natural carbazole alkaloid olivacine (**5**).

In 2015, Gu *et al.* reported another convenient method to form polysubstituted carbazoles from 2-alkoxy-2,3-dihydrofuran **6** and indoles **7** (Scheme 3).¹⁴ In the presence of a catalytic amount of TsOH (5 mol%), carbazole derivatives **8** were delivered with moderate to excellent yields (58–99%) in MeCN. The authors further applied the strategy to the synthesis of other benzoarenes. In the presence of a catalytic amount of TfOH, the reaction of 2-alkoxy-2,3-dihydrofuran **6** with 2-methylfuran **9**, thiophene **10** and 1,2-dimethoxybenzene **11** could provide corresponding products **12–14** in 46–66% yields.



Review



Application to total synthesis of natural carbazole alkaloid



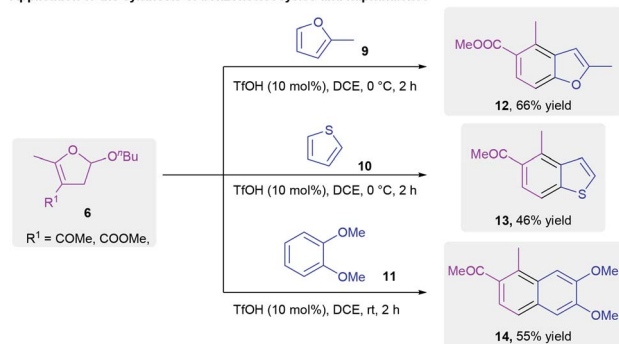
Scheme 2 HOTf-catalyzed [4 + 2] benzannulation to synthesize carbazoles.

Later in 2018, the Maji group employed Brønsted acid catalysis in constructing polyfunctionalized carbazoles and its derivatives through a protecting-group-free benzannulation of 2-alkenylindoles **15** with 1,3-dicarbonyls **16/18** (Scheme 4).¹⁵ Notably, this method using cheap diphenylphosphoric acid as the catalyst and eco-friendly air as an ideal oxidant. Substrates with various substituents were well tolerated in the protocol, and all the reactions smoothly provided the target aromatic molecules **17/19** regardless of their electronic properties. The total synthesis of three carbazole alkaloids further demonstrated the practical applicability of the novel strategy. Besides, the same group successfully expanded the [4 + 2] benzannulation reaction of **15** with aldehydes for the synthesis of other multifunctionalized aromatic compounds.¹⁶

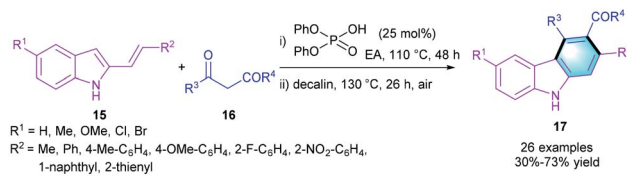
In 2018, Chang's group reported an inverse-electron-demand Diels-Alder reaction between 1,2-diazines **20** and ynamides **21**.¹⁷ Optimizing the reaction conditions showed that the reaction proceeded well under the catalysis of TiF_2O in DCE at 100 °C. The catalytic protocol could tolerate various functional groups,



Application to the synthesis of benzoheterocycles and naphthalenes



Scheme 3 Acid-catalyzed [4 + 2] benzannulation to synthesize substituted carbazoles.



Total synthesis of three medically important alkaloids



Scheme 4 Phosphoric acid-catalyzed [4 + 2] benzannulation to synthesize carbazoles and indolocarbazoles.

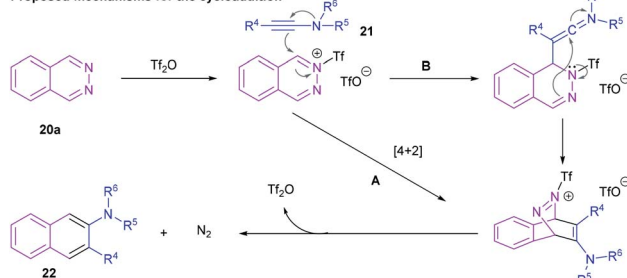
both electron-rich and electron-deficient substituents on 1,2-diazines are compatible in the [4 + 2] benzannulation, giving a series of 2-aminonaphthalenes or 2-aminoanthracenes **22** in moderate to excellent yields (56–95%). As shown in Scheme 5, a possible mechanism of the reaction was proposed.

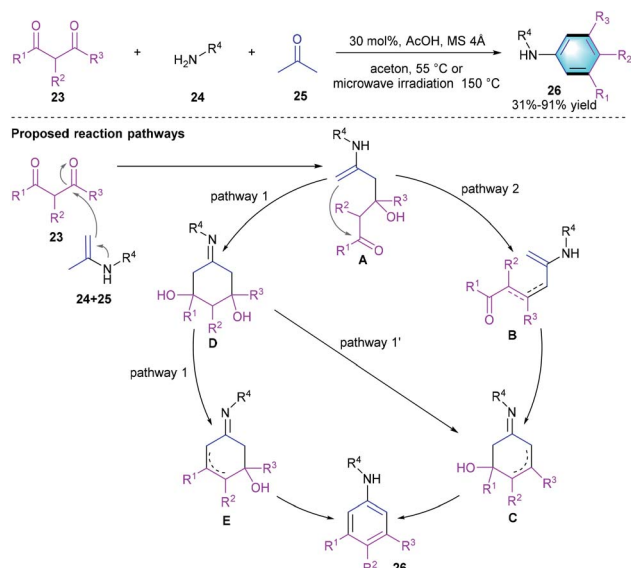
2.2 [3 + 3] benzannulation

Comparing with the acid-catalyzed [4 + 2] benzannulations, [3 + 3] benzannulation strategy has been much less reported in the literatures. In 2019, Rubin *et al.* developed a facile [3 + 3] cyclization to access meta-substituted anilines from a three-component acyclic precursor, 1,3-diketones **23**, amines **24**, and acetone **25** (Scheme 6).¹⁸ The optimum reaction condition is using 30 mol% of acetic acid in acetone under conventional heating (55 °C) or microwave irradiation (150 °C). The compatibility of this procedure was well-investigated, and



Proposed mechanisms for the cycloaddition

Scheme 5 TiF_2O -catalyzed [4 + 2] benzannulation to synthesize 2-aminonaphthalenes and 2-aminoanthracenes.



Scheme 6 AcOH-catalyzed [3 + 3] benzannulation to synthesize *meta*-substituted anilines.

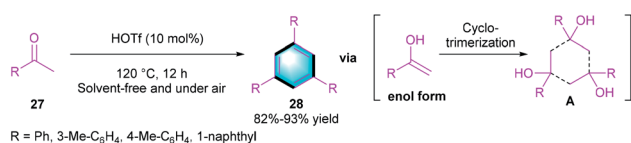
various substrates participate in the reaction successfully to provide the *meta*-substituted anilines **26** with acceptable-to-excellent yields (31–91%). A subsequent gram-scale reaction afforded the aniline product in favorable yield after crystallization. The proposed mechanism was shown in Scheme 6, the key step of this [4 + 2] benzannulation is the cyclocondensation/aromatization of the *in situ* formed imines of acetone 1,3-diketones.

2.3 [2 + 2 + 2] benzannulation

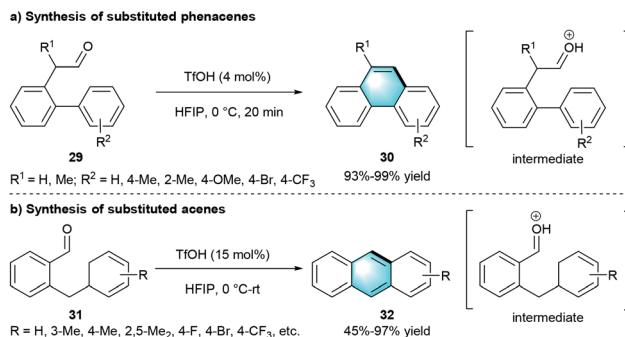
Taking advantage of the unique catalytic activity of HOTf, Yi and co-workers realized a condensation reaction for assembling privileged benzene using ketones **27** as simple substrates (Scheme 7).¹⁹ The cyclo-trimerization of ketones in the presence of HOTf (10 mol%) at 120 °C, furnishes a series of tri-substituted arenes **28** in good to high yields (82–93%).

2.4 Intramolecular benzannulation

Following the concept of atom economy, the group of Ichikawa disclosed an intramolecular benzannulation of aromatic aldehydes to construct polycyclic aromatic hydrocarbons (Scheme 8).²⁰ Using a catalytic amount of TfOH as catalyst and 1,1,1,3,3,3-hexafluoropropan-2-ol (HFIP) as the solvent enabled access to functionalized phenacenes **30** with high yields (Scheme 8a, 93–99%). Furthermore, the protocol was suitable



Scheme 7 HOTf-catalyzed [2 + 2 + 2] benzannulation to synthesize tri-substituted arenes.



Scheme 8 TfOH-catalyzed intramolecular benzannulation to synthesize phenacenes and acenes.

for the cyclization precursor 2-benzylbenzaldehydes **31**, resulting in moderate-to-excellent yields (45–97%) of the corresponding acenes **32** through dehydrative benzannulation (Scheme 8b). The author further proved that HFIP was important for the cyclizations because of its high ionizing power.

Meanwhile, an intramolecular 6 π -electrocyclization–aromatization process of ynedienamine **33** to multi-substituted benzene was reported by Huang's group in 2017 (Scheme 9).²¹ The investigation of the reaction conditions showed that *p*-nitrobenzoic acid (*p*NBA) with NaOAc in chlorobenzene provided product **35a** with good yield (85%). Various acyclic or cyclic ynedienamines **33** and five-/six-/seven-/eight-membered cyclic amines **34** endured the cyclization under the optimized conditions, achieving 44–85% yields for the functionalized arenes **35** bearing two distinct meta-amino groups.

In addition, Jiang and co-workers disclosed that yne-allenone had the potential for constructing 1-naphthols *via* cleavage/rearrangement of carbon–carbon triple bonds in the presence of equivalent NBS and NCS.²² Encouraged by this work, the same group developed an AcOH-catalyzed intramolecular benzannulation of readily available yne-allenone esters **36**, allowing the construction of α -naphthol derivatives **38** in high to excellent yields (85–97%).²³ The reaction featured broad substrate scope, functional group tolerance, as well as an atom- and step-economical manner (Scheme 10). The authors also proposed a possible mechanism for the reaction. Initially, vinyl cation intermediates **B** was generated from **36** in the presence of AcOH. Then, [2 + 2] cycloaddition of **B** occurred to form intermediate **D**. Subsequent ring opening and



Scheme 9 *p*NBA-catalyzed intramolecular benzannulation to synthesize substituted benzenes.





Scheme 10 AcOH-catalyzed intramolecular benzannulation to synthesize α -naphthol derivatives.

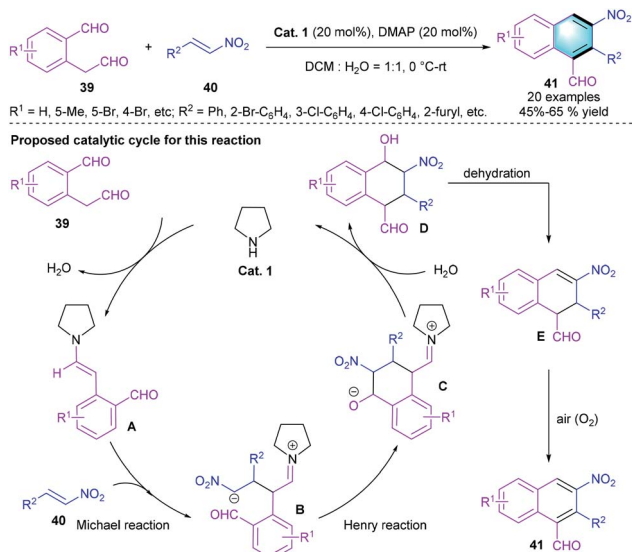
deprotonation/isomerization generated **G**, which reacted with aromatic amine **37** and delivered the final products **38**. In the same year, the group of Jiang presented another novel alternative methodology for the synthesis of substituted α -naphthols *via* NHC-catalyzed C–C triple bond cleavage of yne-allenone under mild conditions.²⁴

3 Secondary amine-catalyzed benzannulation

Secondary amines are considered as the dominant organic amine catalysts. At present, secondary amines, such as Jørgensen–Hayashi catalyst (diphenylprolinol silyl ether), MacMillan's imidazolinone catalyst, and proline, have been commonly applied in the assembling of complex cyclic and polycyclic skeletons *via* enamine or iminium activation and showed extraordinary ability in stereocontrol.²⁵ In addition, some simple achiral secondary amines (*e.g.*, pyrrolidine and piperidine) have been used in the highly efficient cyclization reactions.²⁶ Taking advantage of these powerful organo-catalysts, a series of benzannulation reactions have been developed.

3.1 [4 + 2] benzannulation

In 2010, Xu and co-workers reported a [4 + 2] cascade reaction between dialdehydes **39** and nitroalkenes **40** *via* a Michael–Henry–dehydration–aromatization sequence.²⁷ Using pyrrolidine (**Cat. 1**) as the catalyst and DMAP as the additive in DCM and water ($V_{\text{DCM}}/V_{\text{H}_2\text{O}} = 1/1$) afforded benzene derivatives **41** in 45–65% yields, which bears three adjacent substituents. Notably, nitroalkene with an alkyl group, such as *n*-propyl, exhibits poor reactivity, and no final product is observed



Scheme 11 Pyrrolidine catalyzed [4 + 2] benzannulation to synthesize polysubstituted naphthalene derivatives.

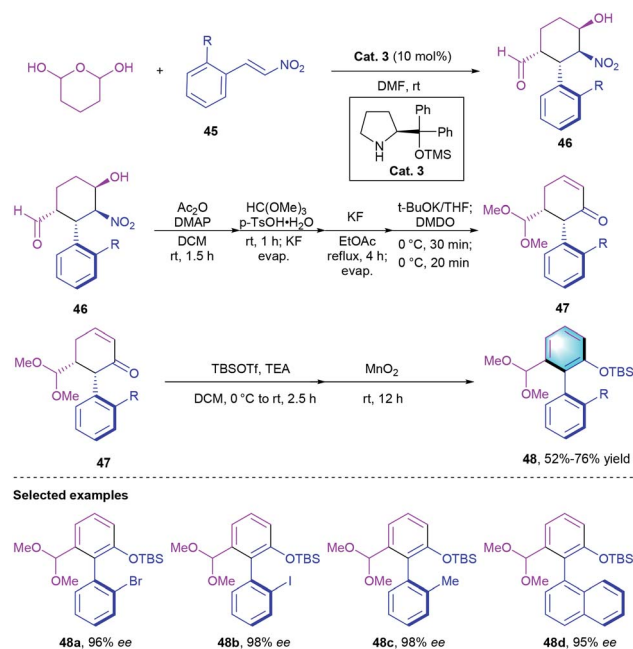
probably because of its instability. Scheme 11 shows the plausible pathway for the reaction. The catalytic cycle begins with enamine activation of dialdehyde **39**, the Michael reaction between enamine intermediate **A** and nitroolefin **40** formed the zwitterion **B**, which further underwent the intramolecular Henry reaction. Subsequent hydrolysis afforded the unstable intermediate **D**, followed by dehydration and aromatization to furnish the corresponding product.

Secondary amine-catalyzed [4 + 2] benzannulation is also considered as an important method to prepare polycyclic



Scheme 12 *L*-Proline-catalyzed [4 + 2] benzannulation to synthesize polycyclic arenes.





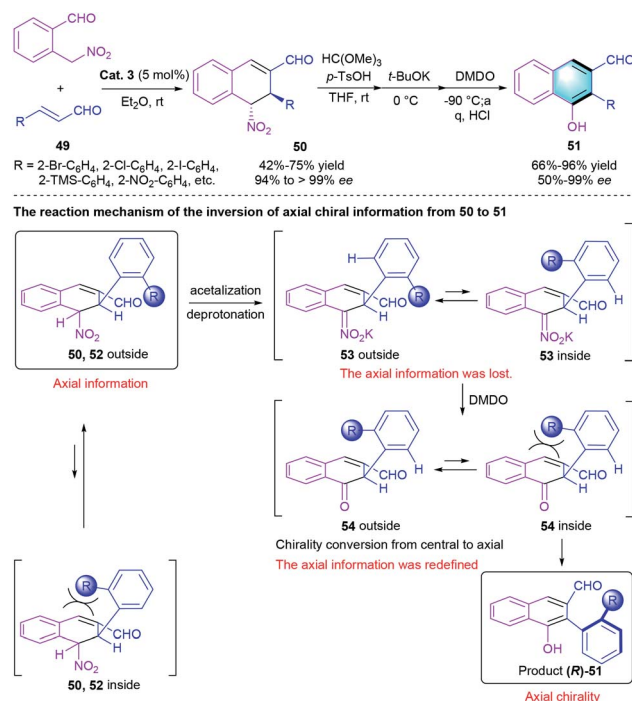
Scheme 13 Secondary amine-catalyzed [4 + 2] benzannulation to synthesize axially chiral nitrocyclohexanecarbaldehyde.

aromatic hydrocarbons. For example, in 2015, Lee's group developed a one-pot reaction for the synthesis of anthraquinones through a [4 + 2] annulation between 1,4-naphthoquinones **42** and α,β -unsaturated aldehydes **43** (Scheme 12).²⁸ In the condition screening, using *L*-proline as the catalyst, benzoic acid as the additive provided the best catalytic efficiency. The tandem benzannulation tolerated a series of substrates with substituents of different electronic properties, affording product **44** in moderate to excellent yields (47–93%). Notably, the reaction of **42a** or **42d** with citral (**43e**) under the standard conditions gave a mixture of regioisomers, **44h** and **44h'**. A plausible mechanism was proposed to explain the formation of these two regioisomers (Scheme 12c).

Axially chiral molecules can be found in many natural products and are widely applied in chiral ligands, material chemistry, and drug discovery. Thus, the enantioselective synthesis of axially chiral compounds is of great significance.

Recently, the *de novo* construction of new arenes *via* benzannulation becomes efficient access to synthesize axially chiral scaffolds.²⁹ In this context, Hayashi's group designed a chiral secondary amine-catalyzed [4 + 2] cyclization of tetrahydro-2H-pyran-2,6-diol with nitroalkenes **45** through a domino Michael–Henry reaction. Chiral nitrocyclohexanecarbaldehyde **46** bearing four contiguous stereogenic centers were obtained. Removing the central chirality of cyclohexane moiety, the corresponding enantioenriched biaryl atropisomers **48** could be acquired with up to 98% ee (Scheme 13).

In 2020, the same group described a similar reaction catalyzed by chiral diphenylprolinol silyl ether (Scheme 14).³⁰ In this work, Michael addition between 2-(nitromethyl)-benzaldehyde and α,β -unsaturated aldehyde **49**, followed by intramolecular aldol condensation and aromatization, would generate the final



Scheme 14 Diphenylprolinol silyl ether-catalyzed [4 + 2] benzannulation to synthesize axially chiral biaryls.

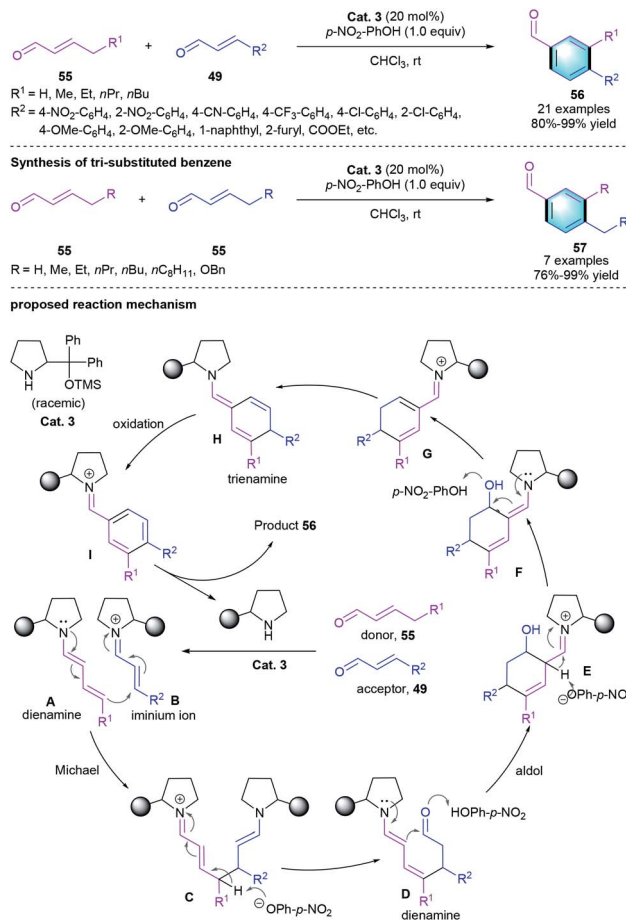
atropisomeric biaryl product **51** with axial chirality in moderate-to-excellent yields (66–96%) with up to 99% ee. Notably, a sufficiently bulky nitro group led to a moderate ee value (50%), whereas the less bulky substituent only afforded racemic biaryls probably because of the decrease of rotational barrier around the axis. The author also proposed a possible reaction mechanism to explain the inversion of axial chirality information from **50** to **51**.

3.2 [3 + 3] benzannulation

Besides [4 + 2] benzannulation, secondary amine-catalyzed [3 + 3] benzannulation strategy was also successfully applied in the construction of polysubstituted benzenes. For example, in 2012, Wang *et al.* developed a [3 + 3] benzannulation from easily accessible enals using diphenylprolinol silyl ether **Cat. 3** and additive (Scheme 15).³¹ Two kinds of α,β -unsaturated aldehydes **55/49** were used as substrates in the reaction, which exclusively produced the corresponding polysubstituted aromatic aldehydes **56** in good yields (80–99%) with high chemo- and regioselectivities. Moreover, the self-dimerized homo-coupling processes smoothly with aliphatic **55** under the standard conditions to give the desired products **57**. The author also proposed a possible mechanism for this Michael-aldol aromatization process. First, iminium ion **B** was generated from a certain enal and the catalyst. Then intermediate **B** underwent the Michael addition with dienamine **A**, which was derived from another enal and the secondary amine. Subsequently, the intermediate Michael adduct was converted into the new dienamine **D** in the presence of *p*-nitrophenoxide. An intramolecular aldol reaction formed enamine **E**, followed by the



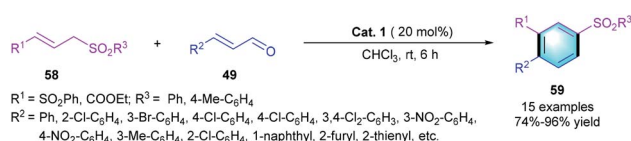
Review



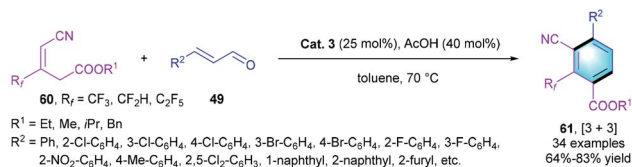
Scheme 15 Diphenylprolinol silyl ether-catalyzed [3 + 3] benzannulation to synthesize tri-substituted benzenes.

formation of electron-rich trienamine **H**. Finally, oxidation of the trienamine intermediate would afford arene **I**. Hydrolysis released the catalyst and furnished the tri-substituted benzene product **56**.

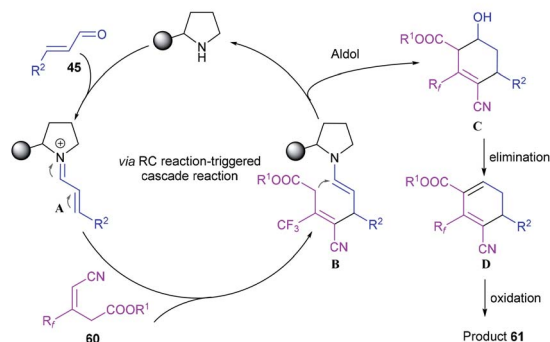
Based on the previous work, Jiang and co-workers reported a secondary amine-catalyzed [3 + 3] cascade reaction.³² Different α,β -unsaturated aldehydes **49** and allylic sulfone-containing 1,3-bisnucleophiles **58** were smoothly converted into multi-substituted benzenes **59** in the presence of pyrrolidine (20 mol%) at room temperature (Scheme 16). This reaction involves iminium–enamine–iminium activation, and the variation of substituents on substrate **49** or **58** was well tolerated, providing the desired product in high-to-excellent yields (74–96%) with good regioselectivities. However, using *trans*-



Scheme 16 Pyrrolidine catalyzed [3 + 3] benzannulation to synthesize tri-substituted benzenes.



Plausible mechanism for the [3 + 3] benzannulation

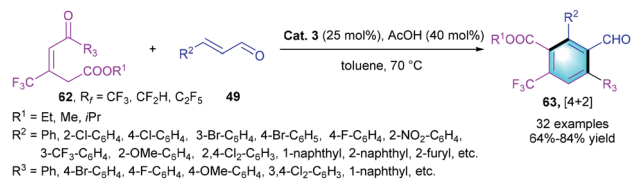


Scheme 17 Diphenylprolinol silyl ether catalyzed [3 + 3] benzannulation to synthesize tetra-substituted benzenes.

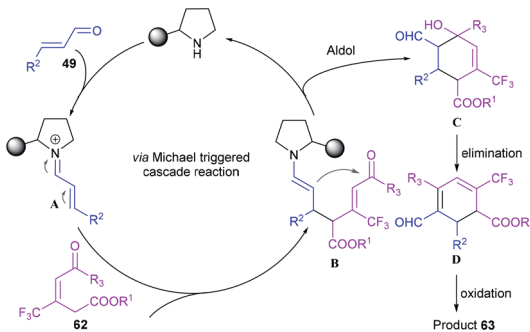
crotonaldehyde as the substrate, only a complex mixture was detected, probably because of the coexistence of dienamine activation compared with aromatic enals.

Given our experience to assemble pharmacologically important biologically interesting frameworks and the positive pharmacokinetic effects brought by CF₃ group-containing

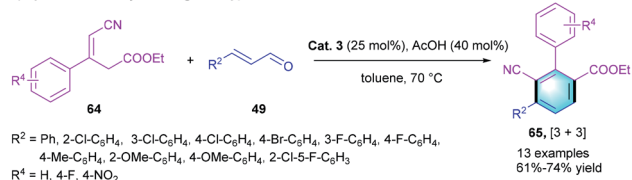
a) Substrate scope of Michael-initiated [4 + 2] aromatization



b) Plausible mechanism for the [4+2] benzannulation



c) Synthesis of biaryls through RC-type reaction



Scheme 18 Diphenylprolinol silyl ether catalyzed benzannulation to synthesize penta-substituted benzenes and biaryls.

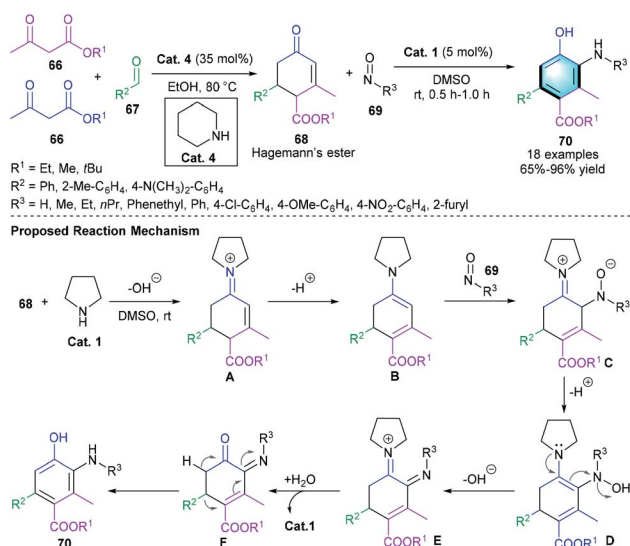


compounds,³³ our group recently reported a novel protocol for the construction of trifluoromethylated multi-functionalized benzenes through a secondary amine-catalyzed [3 + 3] benzannulation.³⁴ The tri-substituted alkene with cyano group **60** was used as a nucleophile.³⁵ α,α -Diphenylprolinol OTMS ether **Cat. 3** was selected to facilitate the generation of iminium species **A** with the assistance of AcOH. The subsequent 1,4-addition reaction between **60** and **A** afforded enamine intermediate **B**, followed by an intramolecular aldol condensation to produce cyclic intermediate **C**. Simultaneous elimination and oxidation of **C** resulted in the [3 + 3] benzannulation product **61**, which bears three adjacent substituents (Scheme 17). The investigation of substrate scope showed that the electron-deficient and electron-rich substituents could be well tolerated, producing the corresponding CF₃-substituted benzenes in moderate-to-good yields (64–83%). A gram-scale reaction proceeded smoothly under the optimal condition, which verified the practical utility of this novel strategy.

Moreover, the regioselectivity of the reaction was controlled by the functional groups of the tri-substituted alkenes. When a benzoyl group was introduced into the alkene substrate instead of the cyano group, a Michael-initialized [4 + 2] aromatization occurred, leading to the formation of a penta-substituted trifluoromethyl benzenes **63** (Scheme 18a). Diverse α,β -unsaturated aldehydes **49** and alkenes **62** with different substituents could participate in the reaction and furnished the target products in good yields (64–84%), which proved the generality of this aromatization. Notably, our strategy also provided a straightforward method for the synthesis of biaryls (Scheme 18c). The reaction between substrate **64** and unsaturated aldehydes **49** produced the desired aryl-substituted benzenes **65** in 61–74% yields.

3.3 [3 + 2 + 1] benzannulation

In 2006, Ramachary *et al.* presented a secondary amine-catalyzed multi-component benzannulation process for the



Scheme 19 Pyrrolidine catalyzed [3 + 2 + 1] benzannulation to synthesize penta-substituted benzenes.



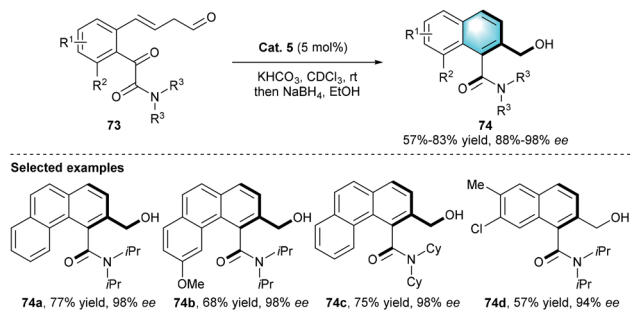
Scheme 20 Proline derivatives catalyzed intramolecular benzannulation to synthesize axially chiral biaryls.

construction of highly functionalized arenes.³⁶ The secondary amine, piperidine or pyrrolidine, could smoothly catalyze the Knoevenagel–Michael–aldol condensation–decarboxylation cascade reaction of alkyl acetoacetates **66**, aldehydes **67**, and nitrosoarenes **69**, giving the final substituted benzenes in good yields with excellent chemo- and regioselectivities (Scheme 19). In the reaction, Hagemann's ester **68** was firstly obtained *via* a [3 + 2 + 1] annulation of **66** and **67** in the presence of piperidine. The postulated reaction pathway of the following steps is illustrated in Scheme 20. The iminium activation of Hagemann's ester **68** formed intermediate **A**, which was converted into dienamine **B**. The reaction of **B** with nitrosoarenes **69** furnished intermediate **D** and released the hydroxide ion to give the imine product **E**. Spontaneous hydrolysis and isoaromatization of **E** generated the penta-substituted products **70**.

3.4 Intramolecular benzannulation

Apart from the above-mentioned strategies, secondary amine-catalyzed intramolecular cyclization is also a powerful route to achieve diverse arenes or biaryls. In 2014, Sparr and co-workers presented an efficient method for the preparation of optically active biaryl atropisomer through an intramolecular benzannulation.³⁷ Considering the excellent reactivity of β,γ -unsaturated aldehydes *via* dienamine catalysis, the exploration was started with ketoaldehyde **71**, which could be obtained from a precursor through double oxidation (Scheme 20). The enantiocontrol of the process was determined by the asymmetric catalytic aldol condensation through enamine activation and the following central-to-axial chirality transfer. The evaluation of several privileged secondary amines indicated that proline derivatives **Cat. 5** was the optimal catalyst. Using CDCl₃ as the reaction medium in the absence of additive at room temperature proved to be the optimal reaction conditions. In the further generality and limitation investigation, various substituents with different electronic properties participated in the reaction successfully, providing the corresponding axially chiral products **72** with excellent enantioselectivities (up to 98%). Based on the literatures, this study was the first report to assemble axially chiral biaryls by aromatic annulation using secondary amine catalyst.





Scheme 21 Proline derivatives-catalyzed intramolecular benzannulation to synthesize axially chiral aromatic amides.

In a follow-up study, the Sparr group developed another stereoselective arene-forming aldol condensation using the chiral secondary amine catalysts mentioned above (Scheme 21).³⁸ Under certain reaction conditions, axially chiral aromatic amides **74** were produced with high stereoselectivities.

Compared with their previous work, this reaction proceeded faster and could complete within minutes because of the enhanced reactivity of glyoxylic amide substrates.

4 Primary amine-catalyzed benzannulation

The development of primary amine catalysis has provided a possible solution for the challenging scenarios of catalytic benzannulation.³⁹ Many elegant publications on primary amine catalysis to access various complex pharmaceutically and structurally important cyclic scaffolds have been reported.⁴⁰ However, primary amine-catalyzed benzannulation is less reported. Up to now, the only example was reported by Sparr and co-workers in 2016.⁴¹ In the protocol, the oxidation of binaphthalene derivative **75** led to the corresponding ketoaldehyde, which then underwent a stereoselective intramolecular arene-forming aldol condensation catalyzed by natural primary amine catalyst (Scheme 22). The strategy provided an efficient access to 1,2-naphthalene oligomers **76** with two chiral axes in good yield with excellent enantioselectivity (90% ee).

5 Tertiary amine-catalyzed benzannulation

Tertiary amine catalysts, which could serve as Lewis or Brønsted base, have been widely applied in facilitating the annulation reactions, thereby synthesizing numerous important



Scheme 22 Primary amine-catalyzed intramolecular benzannulation to synthesize axially chiral 1,2-naphthalene oligomers.

carbocyclic and heterocyclic compounds. In this part, tertiary amine-catalyzed benzannulation for the construction of highly functionalized arenes was discussed.

5.1 [4 + 2] benzannulation

In 2013, the group of Samanta developed a one-pot [4 + 2] cyclization for the preparation of carbazole derivatives using methyl 2-(3-formyl-indolin-2-yl)acetates **77** and nitroolefins **40** as substrates (Scheme 23a).⁴² This Michael–Henry–aromatization cascade reaction proceeded smoothly in the presence of DABCO, and the optimal reaction efficiency was observed at 70 °C in water. The variation of substitution R^1 did not affect the reaction much. **77** with both electron-donating and electron-withdrawing substituents gave the corresponding products **78** in high-to-ideal yields (82–97%). A gram-scale reaction was carried out under standard conditions without loss of yield and efficiency. Moreover, the practicability was proved by using the strategy to synthesize new fused carbazole **79** (Scheme 23b). Later, the same group further reported a one-pot two-step approach to construct highly functionalized carbazole scaffolds. The process was initiated with the DABCO-catalyzed Michael–Henry reaction between **77** and substituted nitro-diene **80**. Subsequent elimination occurred under the treatment of 2 N HCl, forming target compound **81** (Scheme 23c).⁴³ This work achieved a flexible and direct synthetic route to multi-substituted carbazole derivatives.

As a versatile synthon to access cyclic and polycyclic compounds, α -substituted allenoates and relevant [4 + 2] reactions have attracted considerable attention. For example, in 2016, Huang's group disclosed an efficient method to attain polysubstituted benzenes **84** via the annulation of dienic sulfones **82** and α -substituted allenoates **83** (Scheme 24).⁴⁴ The reaction proceeded successfully with DMAP as the catalyst and CH_3COONa as the additive in CHCl_3 at 60 °C. Various substituents on **82** or **83** were well tolerated, providing product **84** in



Scheme 23 DABCO-catalyzed [4 + 2] benzannulation to synthesize carbazole derivatives.



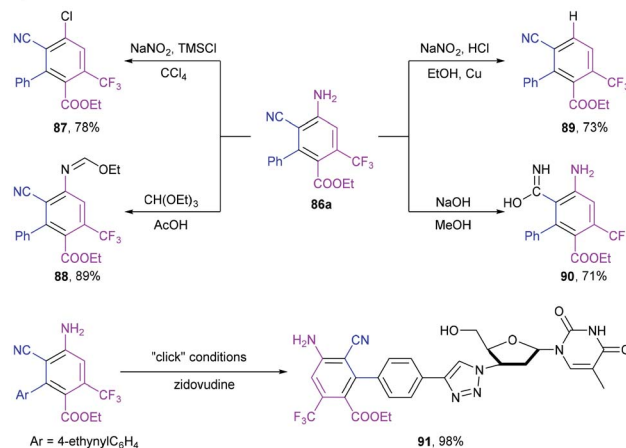


Scheme 24 DMAP-catalyzed [4 + 2] benzannulation to synthesize penta-substituted benzenes.

acceptable-to-good yields (29–72%) with high regioselectivity. However, low yields were observed when the benzene core bore an electron-deficient (NO₂) or an electron-rich group (OMe).



Synthetic transformations of 3a



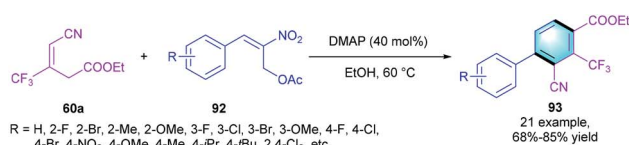
Scheme 25 DABCO-catalyzed [4 + 2] benzannulation to synthesize biaryls.

Moreover, some control experiments were conducted to reveal the mechanism. Isolated intermediates **D** and **F** could be converted into **84** efficiently in the presence of CH₃COONa, proving the reaction pathway in Scheme 24.

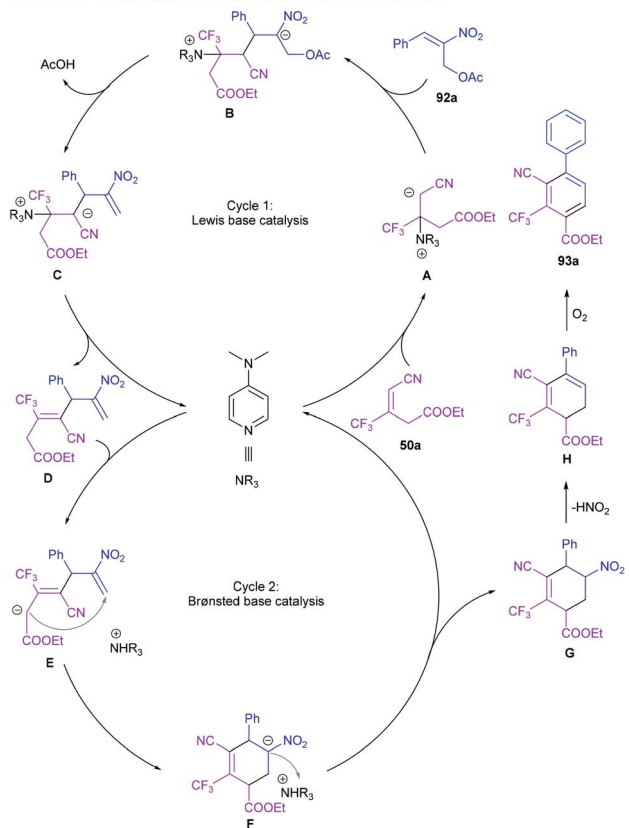
More recently, our group reported a highly chemoselective benzannulation cascade reaction to assemble CF₃-functionalized biaryls (Scheme 25).⁴⁵ The Michael addition-initiated [4 + 2] annulation between electron-deficient alkene **85** and fluoro-substituted **60** occurred uneventfully using DABCO as the catalyst under mild conditions. Diverse R² substituted on the phenyl ring at different positions were compatible in the reaction, providing polysubstituted biaryls **86** in good-to-high yields (71–87%). A gram-scale reaction proceeded smoothly under the optimal conditions with comparable reaction efficiency. Poly-substituted biaryl **86a** could be smoothly converted to diverse derivatives through functional group transformations, which illustrated the synthetic utility of this reaction.

5.2 [3 + 3] benzannulation

Our group had demonstrated that the trifluoromethylated alkenes **60** could act as useful C4 synthons in the [4 + 2]



Two possible catalytic cycle of Rauhut–Currier-initiated benzannulation



Scheme 26 DMAP-catalyzed [3 + 3] benzannulation to synthesize biaryls.





Scheme 27 DABCO-catalyzed [3 + 3] benzannulation to synthesize fluorine/fluorenone derivatives.

benzannulation with enals and electron-deficient alkenes. In the following work, an appropriate dielectrophilic synthon was applied to probe whether CF_3 -alkenes **60** could serve as the dinucleophilic C3 synthons. Consequently, a novel biaryl was obtained through the Rauhut–Currier reaction-initiated [3 + 3] aromatization using 2-nitroallylic acetate **92a** as the C3 counterpart. The screening of tertiary amine catalysts indicated that the nucleophilic DMAP has the best efficiency. A broad spectrum of tetrasubstituted trifluoromethylbenzene **93** was obtained regardless of the electronic properties of R. As shown in the proposed mechanism, DMAP served as a Lewis base in the first cycle and as a Brønsted base in cycle 2 (Scheme 26).

Recently, Yuan's group reported a [3 + 3] benzannulation of 1-indanylidene malononitrile with MBH carbonate to deliver polyfunctionalized fluorene **96** or fluorenone **97** (Scheme 27).⁴⁶ Using DABCO in the presence of ionic liquid, a series of functional fluorenes **96** was produced in moderate-to-high yields (52–86%). However, the electronic properties of substituents R^1 and R^2 showed a remarkable influence on the yields, wherein substrates with the electron-deficient groups providing higher yields than those with the electron-rich groups. Substituents at the *ortho*-position of **95** led to low yields because of their bulky steric hindrance. Interestingly, when oxygen was introduced to

this reaction, a variety of fluorenone derivatives **97** could be obtained in 55–60% yields.

5.3 [2 + 2 + 2] benzannulation

A three-component [2 + 2 + 2] annulation of ethyl propiolate **98** and 1,3-dicarbonyl compounds **99** was reported by Xue and co-workers, enabling the access to highly substituted benzenes **100** (Scheme 28a).⁴⁷ The favorable yields were obtained in the presence of DMAP as the catalyst. Afterward, the same group developed a similar reaction of **99** with dimethyl acetylenedicarboxylate **101** using pyridine as the catalyst with the assistance of potassium *tert*-butoxide (Scheme 28b). The fully substituted benzenes **103** were smoothly formed with good-to-excellent yields (72–98%).⁴⁸

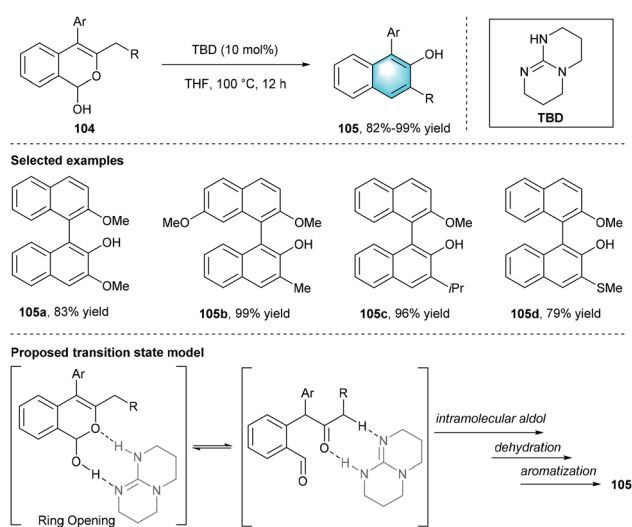
5.4 Intramolecular benzannulation

In 2012, Terada's group developed a direct intramolecular method to synthesize polycyclic substituted arenes through the Ferrier-type rearrangement of lactols **104**.⁴⁹ The strong organic base, 1,5,7-triazabicyclo[4.4.0]dec-5-ene (TBD), was identified to be the optimal catalyst. Further evaluation of the solvents indicated that THF was important for the reaction. Under the standard conditions, the intramolecular multi-step transformation provided unsymmetrically substituted biaryls **105** with high yields (up to 98%). The plausible transition state was depicted in Scheme 29.

In 2019, Yan *et al.* reported the first diastereo- and enantioselective synthesis of axially and helically chiral arenes through tertiary amine-catalyzed intramolecular benzannulation.⁵⁰ After careful investigation of the reaction parameters, the optimal conditions were proven to be using 10 mol% bisquinine squaramide as the catalyst in DCE at 40 °C. The axially and helically chiral polycyclic arenes **107** could be obtained in acceptable-to-high yields (43–97%) with excellent diastereoselectivities and enantioselectivities (99% ee, up to >20 : 1 dr).

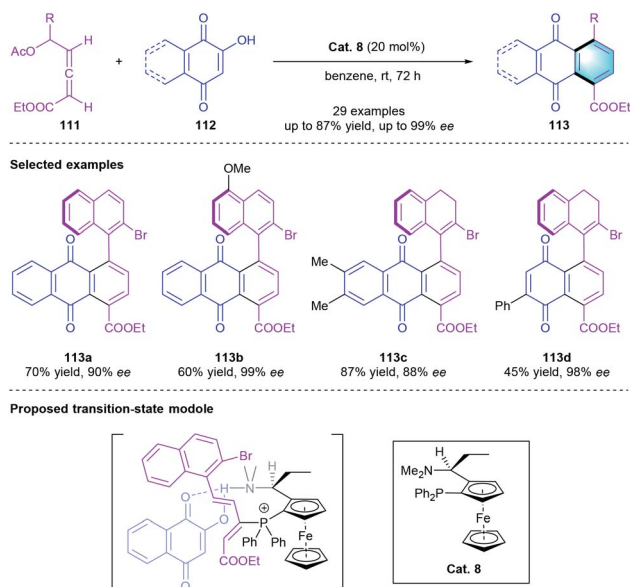


Scheme 28 DMAP-catalyzed [2 + 2 + 2] benzannulation to synthesize highly substituted benzenes.



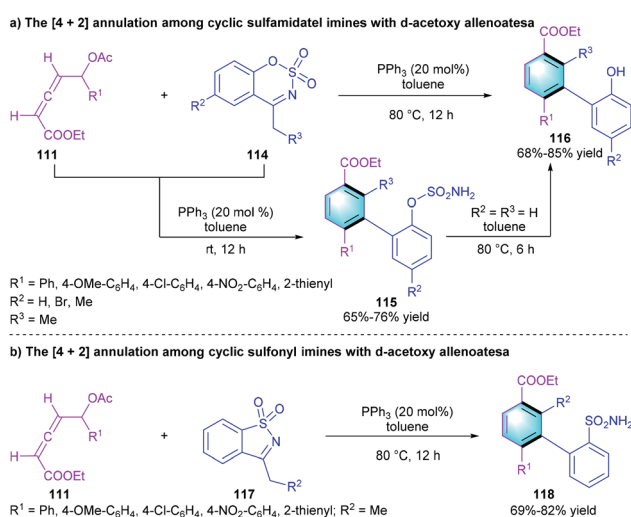
Scheme 29 TBD-catalyzed intramolecular benzannulation to synthesize substituted biaryls.





Scheme 32 Bifunctional ferrocenylphosphine-catalyzed atroposelective [4 + 2] benzannulation to synthesize aryl-naphthaquinones.

Given the extensive application of phosphine catalysis in the reaction of allenates, in 2020, Swamy and co-workers uncovered a [4 + 2] Michael-vinyllogous Mannich-aromatization cascade reaction to synthesize the polysubstituted arenes.⁵³ In this protocol, using Ph_3P as the catalyst, a series of functionalized phenyl sulfamates **115** was attained from **111** and cyclic sulfamidate imines **114** in 68–85% yields. Notably, if the reaction temperature was raised to 80 °C, the S–O bond would be cleaved to afford product **116** (Scheme 33a). Moreover, the [4 + 2] benzannulation could be extended to cyclic sulfonyl imine **117**, generating various substituted arenes **118** with a sulfonamide group in 69–82% yield (Scheme 33b).



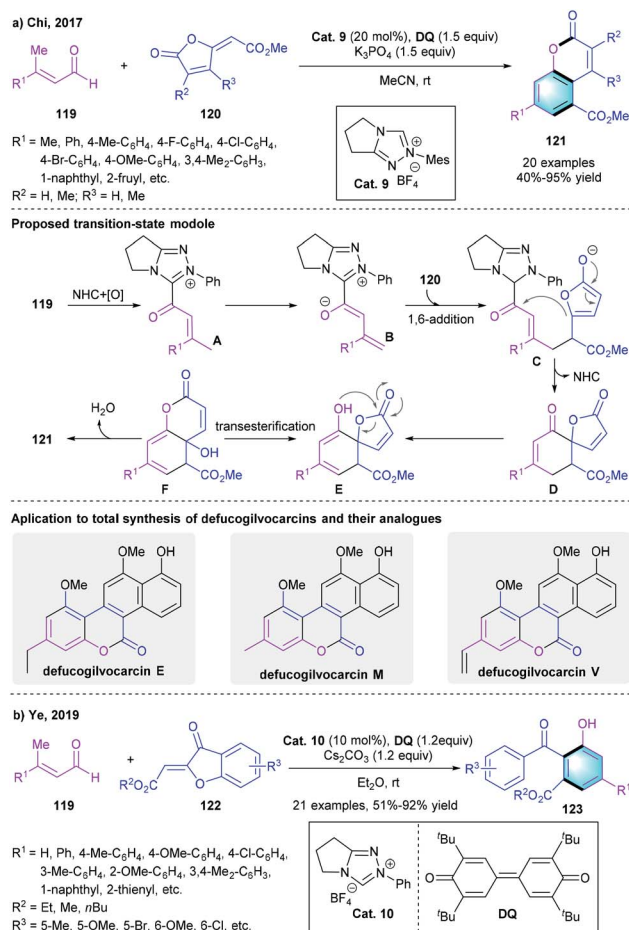
Scheme 33 Ph_3P -catalyzed [4 + 2] benzannulation to synthesize highly functionalized arenes.

7 N-Heterocyclic carbene catalyzed-benzannulation

Since the discovery of stable nucleophilic carbenes by the pioneering work of the Bertrand and Arduengo groups,⁵⁴ NHCs have emerged as exceptional organocatalysts and ligands because of their excellent σ -donor characteristics and moderate π -acidity, leading to the formation of useful metal complexes and active reaction intermediates.⁵⁵ The ability of NHCs to serve as active organocatalyst enabled access to unique reaction modes, which were extensively explored and applied in catalytic C–C bond formation and cyclization reactions for the synthesis of challenging molecules.⁵⁶ The following part discussed the achiral or chiral NHC-catalyzed benzannulations as powerful tools in the construction of multi-substituted arenes.

7.1 [4 + 2] benzannulation

Among the NHC-catalyzed cyclization reactions, the [4 + 2] benzannulation has been extensively studied and has emerged as a useful tool to access new aromatic cores. Early in 2011, Lupton's group disclosed a groundbreaking work of the NHC-catalyzed all-carbon [4 + 2] cycloaddition to synthesize 1,3-



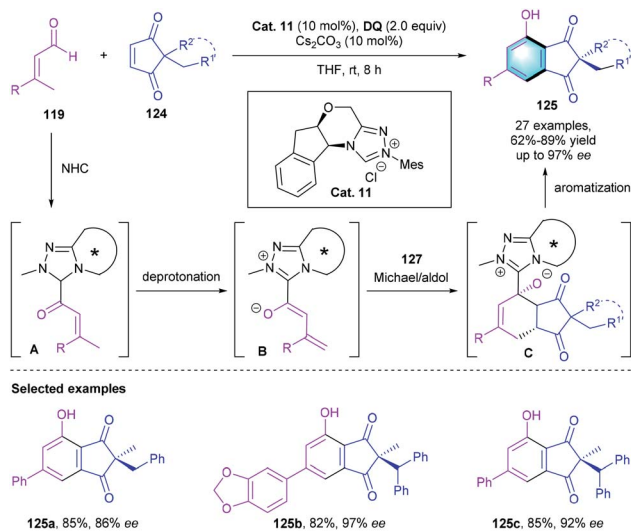
Scheme 34 NHC-catalyzed formal [4 + 2] benzannulation to synthesize coumarins analogs and dihydroxybenzophenones.



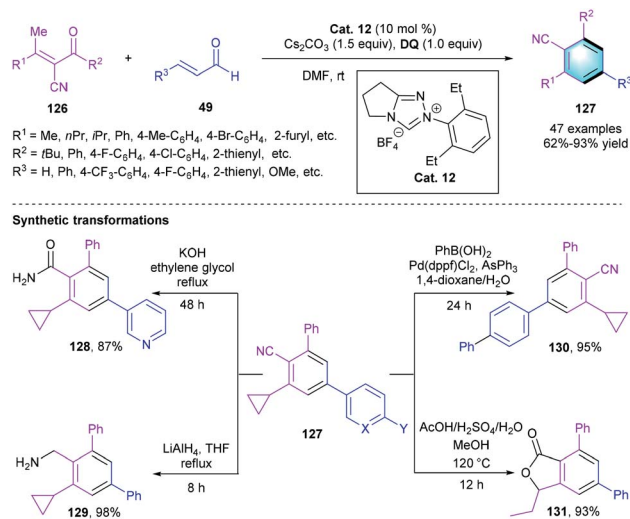
cyclohexadienes with complete diastereocontrol.⁵⁷ In the next nine years, various useful NHC-catalyzed [4 + 2] strategies have been developed for constructing highly functionalized arene molecules.

α,β -Unsaturated aldehydes could be used to afford azolium dienolate species under the catalysis of NHCs after oxidation. Based on this useful activation model, in 2017, Chi and co-workers developed a formal [4 + 2] process for the rapid construction of coumarins analogs (Scheme 34a).⁵⁸ The [4 + 2] reaction of enals **119** with furanones **120** proceeded smoothly in the presence of **Cat. 9** and **DQ**. In the proposed mechanism, the oxidation of Breslow intermediate would generate intermediate **A**. Deprotonation of **A** formed intermediate **B**, which underwent 1,6-addition to afford Intermediate **C** further underwent an intramolecular vinylogous Claisen-type condensation to generate intermediate **D**. Spontaneous enolization, transesterification and dehydration would afford the final multi-substituted coumarins **121**. This [4 + 2] benzannulation process was found compatible with different substituted enals or α/β -substituted furanones, producing the corresponding arenes in 40–95% yield. A gram-scale reaction also proceeded efficiently under the standard conditions. Moreover, the author successfully applied the method in the preparation of bioactive compound defucogilvocarcins **E**, **M**, and **V**. In 2019, the group of Ye extended the [4 + 2] benzannulation strategy to the reaction of β -methyleneals **119** with auronones **122** for the synthesis of dihydroxybenzophenones (Scheme 34b).⁵⁹

Further exploring the application of NHC catalysis in [4 + 2] benzannulation, the group of Chi described a chiral NHC-catalyzed desymmetrization of easily accessible diketones **124** with enals **119** (Scheme 35).⁶⁰ In the presence of the NHC catalyst **Cat. 11**, the reaction proceeded efficiently, affording the corresponding multi-substituted phenols in 62–89% yields, with up to 97% ee. The reaction was supposed to start with the NHC-initiated formation of intermediate **A**, followed by γ -CH deprotonation, Michael reaction, aldol annulation, and



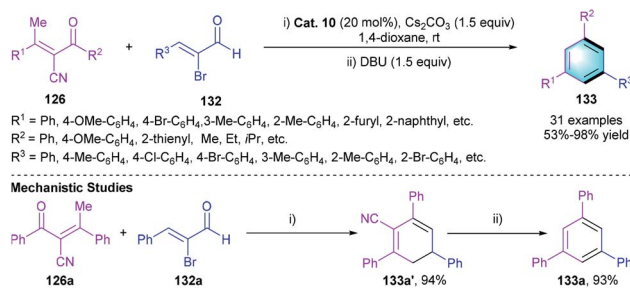
Scheme 35 NHC-catalyzed [4 + 2] benzannulation to synthesize multisubstituted phenols.



Scheme 36 NHC-catalyzed [4 + 2] benzannulation to synthesize cyano-bearing functional benzenes.

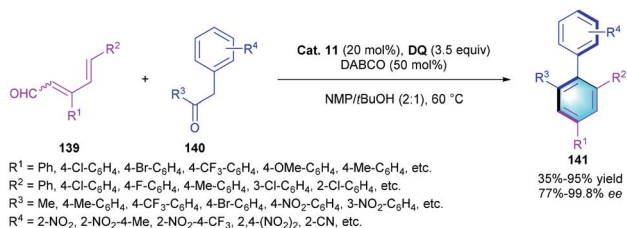
oxidative aromatization to produce the final arenes **125** with an all-carbon quaternary chiral center. Notably, using *E*- or *Z*-enals did not affect the reaction outcomes, leading to the same yields and enantioselectivities. Afterward, a similar asymmetric desymmetrization reaction for constructing functionalized indandione derivatives was reported by Wang's group.⁶¹

Besides the nucleophilic activation of the γ -position, the electrophilic activation of β -position of α,β -unsaturated aldehyde was also applied in NHC-catalyzed [4 + 2] benzannulations. In 2016, Wang's group presented an NHC-catalyzed [4 + 2] annulation between α -cyano- β -methyleneones **126** and α,β -unsaturated aldehydes **49** for assembling benzonitrile unit (Scheme 36).⁶² The optimization of reaction conditions indicated that NHC precursors and solvents had a significant effect on the reactivity. Enhanced efficiency and yield were observed when using achiral **Cat. 12** as the catalyst and tetratertbutyldiphenoquinone as the oxidant in the presence of Cs_2CO_3 in DMF. The substrate scope of enones **126** and enals **49** were carefully evaluated, giving a series of cyano-substituted benzene **127** with up to 93% yield regardless of the electronic properties of the substituents. Transformations of products **127** were carried out to demonstrate the applicability of the method.

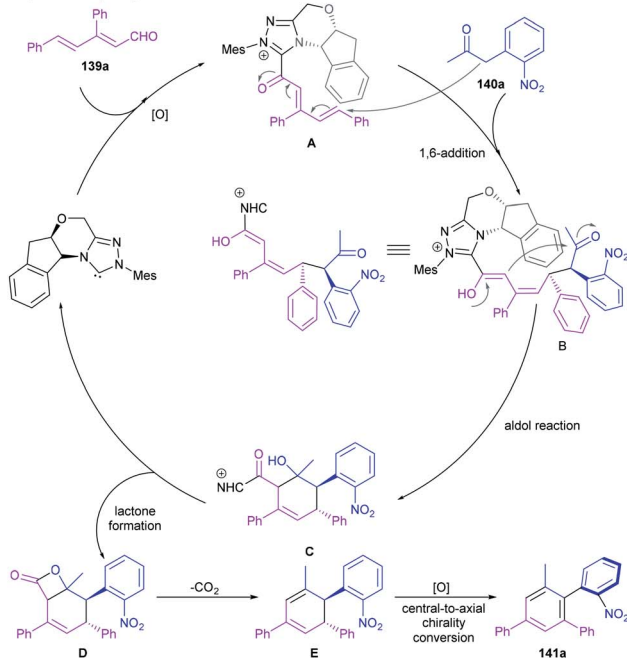


Scheme 37 NHC-catalyzed [4 + 2] benzannulation to synthesize tri-substituted benzenes.





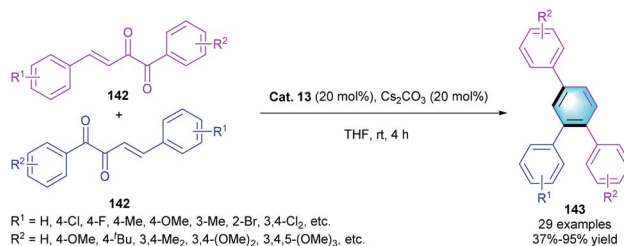
Proposed catalytic cycle



Scheme 40 NHC-catalyzed [4 + 2] benzannulation to synthesize axially chiral biaryl.

(Scheme 40). As shown in the catalytic cycle, the reaction was initiated with the formation of acyl azolium intermediate **A** through NHC activation and oxidation, followed by the addition of **140a** to generate adduct **B**. Then intermediate **B** underwent intramolecular aldol condensation to form the cyclic adduct **C**. Lactonization ended the catalytic cycle and regenerated the NHC catalyst. Thermodynamic controlled decarboxylation and aromatization led to the central-to-axial chirality transfer, providing the axially chiral arene **141**. Notably, this study significantly expanded the application of NHC-catalyzed benzannulation in the assembly of enantioenriched atropisomeric biaryls.

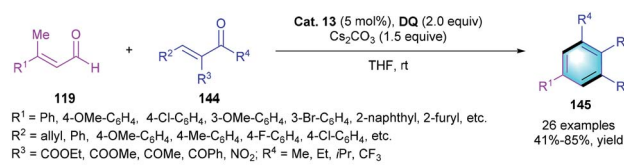
Comparing with the aldehyde, NHC catalysts were much less used to activate ketone derivatives. In 2017, Fang and co-workers developed a direct and efficient method for the synthesis of 1,2,4-triarylbenzenes through NHC-catalyzed umpolung [4 + 2] benzannulation of β,γ -unsaturated diketones **142** (Scheme 41).⁶⁸ The reaction proceeded smoothly to deliver the corresponding benzene derivatives **143** by using **Cat. 13** as the catalyst. Electron-donating or electron-withdrawing substituents at *ortho*-/*meta*-/*para*-positions on the aryl rings were all compatible in the reaction, resulting in the final products with favorable yields (up to 98%).



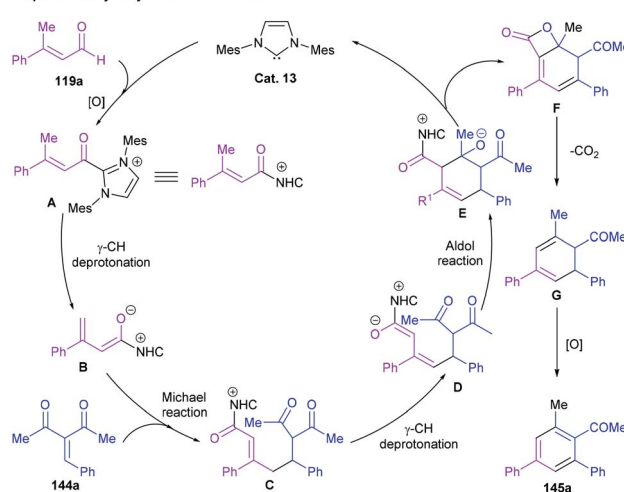
Scheme 41 NHC-catalyzed [4 + 2] benzannulation to synthesize triaryl substituted benzenes.

7.2 [3 + 3] benzannulation

As mentioned above, α,β -unsaturated aldehydes could be used to afford azolium dienolate species in the presence of NHC catalyst. Based on this useful activation model, Chi and co-workers developed an NHC-catalyzed [3 + 3] cyclization of enals with enones (Scheme 42).⁶⁹ In the reaction, β -methyl enal **119** was used as the precursor to form azolium dienolates, then reacted with enones **144**, producing corresponding polyfunctionalized benzenes **145** in medium-to-good yields (41–85%). Different substituents at the β -position (R^1) of enals **119** and the α/β -position (R^2/R^3) of enones **144** were well tolerated. Notably, using *Z*- or *E*-isomer of the substrates barely affected the reaction outcomes, which simplified the preparation of the substrates. A possible mechanism for this reaction was proposed. The catalytic cycle started with the generation of **A** through oxidation. Following γ -deprotonation formed azolium dienolate intermediate **B**. Subsequent nucleophilic addition of **B** to **144a** resulted in intermediate **C**, which further underwent

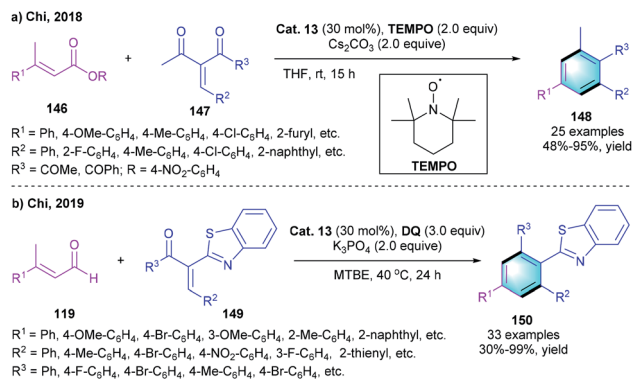


Proposed catalytic cycle for this reaction



Scheme 42 NHC-catalyzed [3 + 3] benzannulation to synthesize polyfunctionalized benzenes.

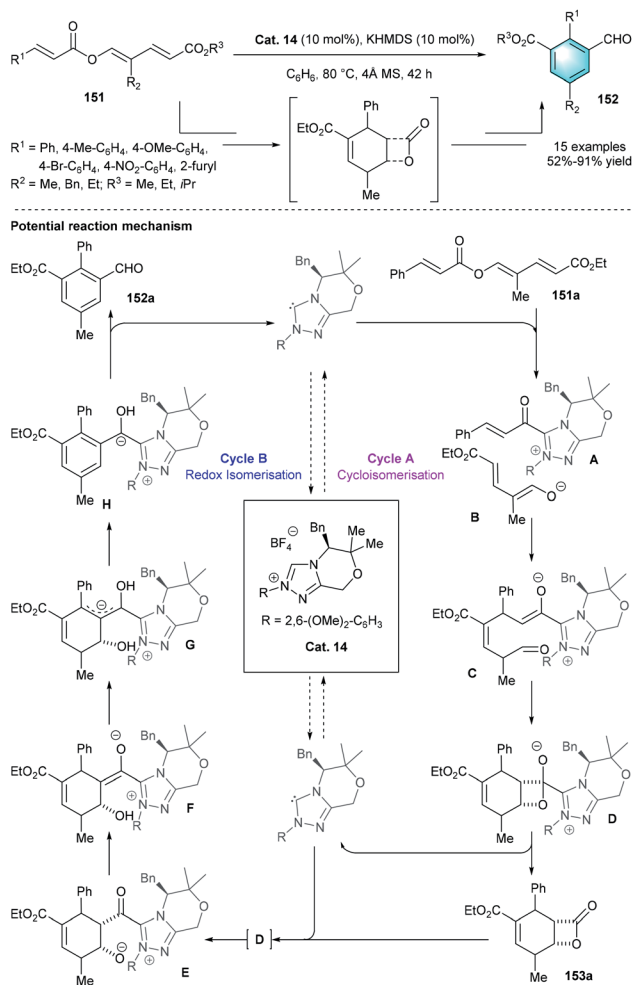




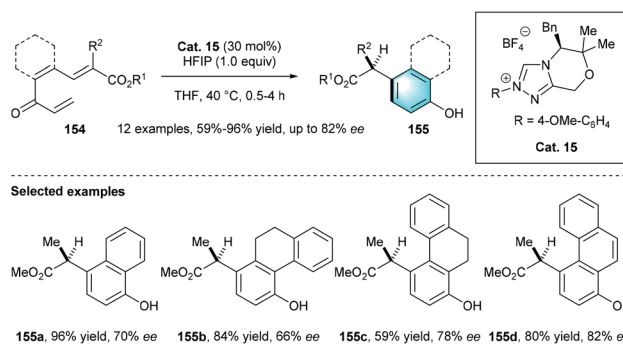
Scheme 43 NHC-catalyzed [3 + 3] benzannulation to synthesize polyfunctionalized benzenes and 2-phenylbenzothiazoles.

an intramolecular aldol reaction to generate intermediate **E**. Esterification completed the catalytic cycle and released bicyclic adduct **F**. Spontaneous decarboxylation and oxidation of **F** generated the tetra-substituted benzene products.

In recent years, esters were found to be applicable substrates in NHC catalysis.⁷⁰ In 2018, the chi group reported a similar [3 +



Scheme 44 NHC-catalyzed intramolecular benzannulation to synthesize tri-substituted benzaldehydes.



Scheme 45 NHC-catalyzed intramolecular benzannulation to synthesize multi-functionalized phenols.

3] benzannulation *via* the direct activation of the ester substrates (Scheme 43a).⁷¹ The reaction of α,β -unsaturated carboxylic ester **146** with enones **147** successfully afforded the tetra-substituted benzenes **148** with up to 95% yield under the catalysis of **Cat. 13** using TEMPO as the oxidant. However, β -dialkyl ester substrates were not compatible with this process, which might be caused by the insufficient acidity of the C-H bond on the β -alkyl group. Compared with previous work using aldehyde as the substrates, this process lowers the economic cost because of the use of readily available ester substrates and inexpensive TEMPO oxidant. Later in 2019, Chi and co-workers presented another [3 + 3] benzannulation to provide substituted 2-phenylbenzothiazoles **150** in acceptable-to-perfect yields (Scheme 43b).⁷²

7.3 Intramolecular benzannulation

Unlike the above-mentioned conversions, in 2015, the group of Lupton uncovered an elegant NHC-catalyzed redox isomerization process for the synthesis of functionalized benzaldehydes from trienyl esters (Scheme 44).⁷³ By using **Cat.14** and KHMDS in benzene at 80 °C, the desired redox-isomerised aldehyde **152** was synthesized in moderate-to-excellent yields (52–91%). The author proposed a possible mechanism, including two linked catalytic cycles. First, acyl azolium **A** and dienolate **B** were originated from the fragmentation of **151a**. Then a vinylogous Michael addition between **A** and **B** generated intermediate **C**. Following β -lactone formation and release of NHC catalyst would produce bicyclic adduct **153a**. The addition of **Cat. 14** to **153a** resulted in acyl azolium intermediate **E**, which ultimately led to the generation of tri-substituted benzaldehyde **152**. In 2016, by using a similar NHC pre-catalyst **Cat. 15**, the same group disclosed another intramolecular benzannulation process, realizing the first enantioselective synthesis of highly functionalized phenol product **155** *via* the umpolung of α, β -unsaturated ketones **154** (Scheme 45).⁷⁴

8 Summary and outlook

This review systemically summarized recent advances in the organocatalytic benzannulation strategies, including Brønsted acid-catalyzed benzannulation, secondary amine-catalyzed



benzannulation, primary amine-catalyzed benzannulation, tertiary amine-catalyzed benzannulation, tertiary phosphine-catalyzed benzannulation, and NHC-catalyzed benzannulation. As presented in the review, recent ten years have witnessed the impressive progress on the organocatalytic benzannulation, which has become an important tool for the construction of structurally complex and densely functionalized arene systems. This review revealed that the organocatalytic benzannulation is a quite different strategy from the conventional benzannulation methods. The main advantage is that versatile catalysts and diversified activation modes make abundant cyclization reactions possible, thereby allowing the construction of poly-substituted arenes with flexible substitution patterns that other methods cannot provide. In different catalytic systems, various types of substrates can be used in the cyclization reaction. As classified in the review, according to the number of carbon atoms contributed by different synthons participating in the cyclization reaction, benzannulation mainly includes [4 + 2], [3 + 3], [3 + 2 + 1], [2 + 2 + 2], and intramolecular annulations. In addition, low economic cost, mild reaction conditions, good functional group compatibility, and excellent chemo-/regioselectivity enable this strategy to be applied in more scenarios, such as in the multistep synthesis of natural products, biologically active compounds, and functional molecules. Moreover, the intrinsic structural features of organocatalyst have enabled the efficient *de novo* construction of chiral arenes.

Based on these impressive advances, new challenges are likely to be addressed in future research of this field. For example, although several examples of asymmetric catalysis have been developed, the methodological development to construct a diverse range of aromatic ring systems containing different chiral elements including central chirality, axial chirality, and helical chirality is still highly desired. So far, aminocatalysis is the main strategy applied in asymmetric organocatalytic benzannulation to achieve the construction of chiral arenes. In future research, the application of chiral tertiary amine catalysis, chiral tertiary phosphine catalysis, chiral Brønsted acid catalysis, and chiral NHC catalysis in asymmetric benzannulation are worth exploring. In addition, the application of dual catalysis of organocatalysis with metal catalysis or photocatalysis in benzannulation may break through the limitations of conventional activation modes, and provide a useful solution for the construction of new functionalized aromatic ring systems. We hope that this review will promote future studies in this field and prompt the application of organocatalytic benzannulation strategy in the construction of densely functionalized arenes.

Conflicts of interest

The authors confirm that this article content has no conflicts of interest.

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Notes and references

- (a) D. A. Horton, G. T. Bourne and M. L. Smythe, *Chem. Rev.*, 2003, **103**, 893; (b) A. Bouisseau, A. Roland, F. Reillon, R. Schneider and F. Cavelier, *Org. Lett.*, 2013, **15**, 3888; (c) M. Li, Y. W. Liu, R. H. Duan, X. F. Wei, Y. P. Yi, Y. Wang and C.-F. Chen, *Angew. Chem., Int. Ed.*, 2017, **56**, 8818; (d) C. L. Zhuang, W. Zhang, C. Q. Sheng, W. N. Zhang, C. G. Xing and Z. Y. Miao, *Chem. Rev.*, 2017, **117**, 7762–7810; (e) T. K. Allred, F. Manoni and P. G. Harran, *Chem. Rev.*, 2017, **117**, 11994; (f) E. K. Davison and J. Sperry, *J. Nat. Prod.*, 2017, **80**, 3060; (g) K. Kitamura, Y. Ando, T. Matsumoto and K. Suzuki, *Chem. Rev.*, 2018, **118**, 1495; (h) M. M. Unterlass, Hot Water Generates, *Angew. Chem., Int. Ed.*, 2018, **57**, 2292.
- (a) J. A. Bull, M. G. Hutchings and P. Quayle, *Angew. Chem., Int. Ed.*, 2007, **46**, 1869; (b) B. Crone, S. F. Kirsch and K. D. Umland, *Angew. Chem., Int. Ed.*, 2010, **49**, 4661; (c) V. Snieckus, *Beilstein J. Org. Chem.*, 2011, **7**, 1215; (d) Y. Izawa, D. Pun and S. S. Stahl, *Science*, 2011, **333**, 209.
- For selected reviews, see: (a) T. B. Poulsen and K. A. Jørgensen, *Chem. Rev.*, 2008, **108**, 2903; (b) W. A. L. van Otterlo and C. B. de Koning, *Chem. Rev.*, 2009, **109**, 3743; (c) S. L. You, Q. Cai and M. Zeng, *Chem. Soc. Rev.*, 2009, **38**, 2190; (d) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147; (e) J. Wencel-Delord, T. Dröge, F. Liu and F. Glorius, *Chem. Soc. Rev.*, 2011, **40**, 4740; (f) K. M. Engle, T.-S. Mei, M. Wasa and J.-Q. Yu, *Acc. Chem. Res.*, 2012, **45**, 788.
- For selected examples of catalytic dehydrogenation of carbocyclic rings, see: (a) T. Diao and S. S. Stahl, *J. Am. Chem. Soc.*, 2011, **133**, 14566; (b) A. V. Iosub and S. S. Stahl, *J. Am. Chem. Soc.*, 2015, **137**, 3454; (c) A. V. Iosub and S. S. Stahl, *ACS Catal.*, 2016, **6**, 8201; (d) Y. F. Liang, X. Li, X. Wang, M. Zou, C. Tang, Y. Liang, S. Song and N. Jiao, *J. Am. Chem. Soc.*, 2016, **138**, 12271.
- For selected examples of electrophilic and nucleophilic aromatic substitution reactions, see: (a) G. Illuminati and H. Gilman, *J. Am. Chem. Soc.*, 1952, **74**, 2896; (b) T. Nakagome, A. Kobayash and A. Misaki, *Chem. Pharm. Bull.*, 1966, **14**, 1074; (c) W. J. Coates and A. McKillop, *J. Org. Chem.*, 1990, **55**, 5418; (d) J. D. Kirkham, R. J. Butlin and J. P. Harrity, *Angew. Chem., Int. Ed.*, 2012, **51**, 6402; (e) C. Liu, M. Zhang, Z. Zhang, S. B. Zhang, S. Yang, A. Zhang, L. Yin, S. Swarts, S. Vidyasagar, L. Zhang and P. Okunieff, *Bioorg. Med. Chem.*, 2016, **24**, 4263; (f) S. Sivanathan, F. Körber and J. Scherkenbeck, *Bioorg. Med. Chem.*, 2016, **24**, 873.
- For selected examples of transition-metal-catalyzed cross-coupling reactions, see: (a) S. M. Bonesi, M. Fagnoni and A. Albini, *Angew. Chem., Int. Ed.*, 2008, **47**, 10022; (b) D.-G. Yu, M. Yu, B.-T. Guan, B.-J. Li, Y. Zheng, Z.-H. Wu



- and Z.-J. Shi, *Org. Lett.*, 2009, **11**, 3374; (c) J. L. Gustafson, D. Lim, K. T. Barrett and S. J. Miller, *Angew. Chem., Int. Ed.*, 2011, **50**, 5125; (d) A. Rahimi, J. C. Namyslo, M. H. H. Drafz, J. Halm, E. Huebner, M. Nieger, N. Rautzenberg and A. Schmidt, *J. Org. Chem.*, 2011, **76**, 7316; (e) C. L. Ciana, R. J. Phipps, J. R. Brandt, F. M. Meyer and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2011, **50**, 458; (f) F. Zhao, Y. F. Zhang, J. Wen, D. G. Yu, J. B. Wei, Z. Xi and Z. J. Shi, *Org. Lett.*, 2013, **15**, 3230; (g) M. Xie, S. Wang, J. Wang, K. Fang, C. Liu, C. Zha and J. Jia, *J. Org. Chem.*, 2016, **81**, 3329.
- 7 For selected examples of direct C-H functionalization of arenes, see: (a) Y. W. Zheng, B. Chen, P. Ye, K. Feng, W. Wang, Q. Y. Meng, L. Z. Wu and C. H. Tung, *J. Am. Chem. Soc.*, 2016, **138**, 10080; (b) S. Bag, R. Jayarajan, U. Dutta, R. Chowdhury, R. Mondal and D. Maiti, *Angew. Chem., Int. Ed.*, 2017, **56**, 12538; (c) B. Ma, Z. Chu, B. Huang, Z. Liu, L. Liu and J. Zhang, *Angew. Chem., Int. Ed.*, 2017, **56**, 2749; (d) H. W. Wang, Y. Lu, B. Zhang, J. He, H. J. Xu, Y. S. Kang, W. Y. Sun and J. Q. Yu, *Angew. Chem., Int. Ed.*, 2017, **56**, 7449; (e) P. Wang, P. Verma, G. Xia, J. Shi, J. X. Qiao, S. Tao, P. T. W. Cheng, M. A. Poss, M. E. Farmer, K. S. Yeung and J. Q. Yu, *Nature*, 2017, **551**, 489; (f) M. Z. Lu, X. R. Chen, H. Xu, H. X. Dai and J. Q. Yu, *Chem. Sci.*, 2018, **9**, 1311.
- 8 For selected reviews of benzannulation reactions in the recent years, (a) P. Wessig and G. Müller, *Chem. Rev.*, 2008, **108**, 2051; (b) S. Kotha, S. Misra and S. Halder, *Tetrahedron*, 2008, **64**, 10775; (c) G. Dominguez and J. Perez-Castells, *Chem. Soc. Rev.*, 2011, **40**, 3430; (d) K. Tanaka and Y. Shibata, *Synthesis*, 2012, **44**, 323; (e) R. Fernandes and S. Mulay, *Synthesis*, 2014, **46**, 1836; (f) C. Holden and M. F. Greaney, *Angew. Chem., Int. Ed.*, 2014, **53**, 5746; (g) A. L. Siva Kumari, A. Siva Reddy and K. C. K. Swamy, *Org. Biomol. Chem.*, 2016, **14**, 6651; (h) P. Loxq, E. Manoury, R. Poli, E. Deydier and A. Labande, *Coord. Chem. Rev.*, 2016, **308**, 131; (i) S. J. Hein, D. Lehnerr, H. Arslan, F. J. Uribe-Romo and W. R. Dichtel, *Acc. Chem. Res.*, 2017, **50**, 2776; (j) W. Song, S. A. Blaszczyk, J. Liu, S. Wang and W. Tang, *Org. Biomol. Chem.*, 2017, **15**, 7490; (k) L. Chen, Z. Liu and S. Zhu, *Org. Biomol. Chem.*, 2018, **16**, 8884; (l) A. Link and C. Sparr, *Chem. Soc. Rev.*, 2018, **47**, 3804; (m) T. N. Poudel, R. J. I. Tamargo, H. Cai and Y. R. Lee, *Asian J. Org. Chem.*, 2018, **7**, 985; (n) A. D. Senese and W. A. Chalifoux, *Molecules*, 2018, **24**, 118; (o) X. Zhang and S. Ma, *Isr. J. Chem.*, 2018, **58**, 608; (p) T. Bera, K. Pandey and R. Ali, *ChemistrySelect*, 2020, **5**, 5239; (q) Q. Jia, Y. Li and Y. Lin, *Chin. J. Org. Chem.*, 2020, **40**, 1502; (r) K. M. Magiera, V. Aryal and W. A. Chalifoux, *Org. Biomol. Chem.*, 2020, **18**, 2372; (s) T. Mies, T.-K. Ma and A. G. M. Barrett, *Russ. Chem. Rev.*, 2020, **89**, 917.
- 9 (a) T. P. Yoon and E. N. Jacobsen, Privileged Chiral Catalysts, *Science*, 2003, **299**, 1691; (b) M. C. Kozlowski, B. J. Morgan and E. C. Linton, *Chem. Soc. Rev.*, 2009, **38**, 3193; (c) G. Bringmann, T. Gulder, T. A. M. Gulder and M. Breuning, *Chem. Rev.*, 2011, **111**, 563.
- 10 For selected reviews see: (a) G. P. McGlacken and L. M. Bateman, *Chem. Soc. Rev.*, 2009, **38**, 2447; (b) C. M. So and F. Y. Kwong, *Chem. Soc. Rev.*, 2011, **40**, 4963; (c) N. Kuhl, M. N. Hopkinson, J. Wencel-Delord and F. Glorius, *Angew. Chem., Int. Ed.*, 2012, **51**, 10236; (d) L. T. Ball, G. C. Lloyd-Jones and C. A. Russell, *Science*, 2012, **337**, 1644; (e) P. Loxq, E. Manoury, R. Poli, E. Deydier and A. Labande, *Coord. Chem. Rev.*, 2016, **308**, 131; (f) P. Y. Choy, S. M. Wong, A. Kapdi and F. Y. Kwong, *Org. Chem. Front.*, 2018, **5**, 288; (g) F.-X. Felpin and S. Sengupta, *Chem. Soc. Rev.*, 2019, **48**, 1150; (h) Y. F. Zhang and Z. J. Shi, *Acc. Chem. Res.*, 2019, **52**, 161.
- 11 Very recently, after we finished the preparation of the manuscript, we found that a similar review on this topic occurred, which did not include Brønsted acid-catalyzed benzannulation. See: L. Wu, B. Yu and E.-Q. Li, *Adv. Synth. Catal.*, 2020, **362**, 4010.
- 12 (a) W. R. Chao, D. Yean, K. Amin, C. Green and L. Jong, *J. Med. Chem.*, 2007, **50**, 3412; (b) T. Janosik, N. Wahlstrom and J. Bergman, *Tetrahedron*, 2008, **64**, 9159; (c) J. Roy, A. K. Jana and D. Mal, *Tetrahedron*, 2012, **68**, 6099; (d) A. W. Schmidt, K. R. Reddy and H. J. Knölker, *Chem. Rev.*, 2012, **112**, 3193; (e) T. Janosik, A. Rannug, U. Rannug, N. Wahlstrom, J. Slätt and J. Bergman, *Chem. Rev.*, 2018, **118**, 9058; (f) J. N. Wang, H. J. Zhang, J. Q. Li, W. J. Ding and Z. J. Ma, *J. Nat. Prod.*, 2018, **81**, 949.
- 13 X. Zheng, L. Lv, S. Lu, W. Wang and Z. Li, *Org. Lett.*, 2014, **16**, 5156.
- 14 C. Liu, W. Huang, M. Wang, B. Pan and Y. Gu, *Adv. Synth. Catal.*, 2016, **358**, 2260.
- 15 S. Saha, A. Banerjee and M. S. Maji, *Org. Lett.*, 2018, **20**, 6920.
- 16 A. Banerjee, A. Guin, S. Saha, A. Mondal and M. S. Maji, *Org. Biomol. Chem.*, 2019, **17**, 1822.
- 17 J. Xue, E. Gao, X. N. Wang and J. Chang, *Org. Lett.*, 2018, **20**, 6055.
- 18 A. R. Galeev, M. V. Dmitriev, I. G. Mokrushin, I. V. Mashevskaya, A. N. Maslivets and M. Rubin, *Org. Biomol. Chem.*, 2019, **17**, 10030.
- 19 X. Zhang, Z. Wang, K. Xu, Y. Feng, W. Zhao, X. Xu, Y. Yan and W. Yi, *Green Chem.*, 2016, **18**, 2313.
- 20 T. Fujita, I. Takahashi, M. Hayashi, J. Wang, K. Fuchibe and J. Ichikawa, *Eur. J. Org. Chem.*, 2017, **2017**, 262.
- 21 X. Li, H. Yu and Y. Huang, *Adv. Synth. Catal.*, 2017, **359**, 1379.
- 22 J. Y. Wang, P. Zhou, G. Li, W. J. Hao, S. J. Tu and B. Jiang, *Org. Lett.*, 2017, **19**, 6682.
- 23 H.-K. Sha, F. Liu, J. Lu, Z.-Q. Liu, W.-J. Hao, J.-L. Tang, S.-J. Tu and B. Jiang, *Green Chem.*, 2018, **20**, 3476.
- 24 P. Zhou, J. Y. Wang, T. S. Zhang, G. Li, W. J. Hao, S. J. Tu and B. Jiang, *Chem. Commun.*, 2018, **54**, 164.
- 25 (a) S. P. Brown, N. C. Goodwin, W. C. David and D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2003, **125**, 1192; (b) J. W. Yang, M. T. H. Fonseca and B. List, *Angew. Chem., Int. Ed.*, 2004, **43**, 6660; (c) M. T. H. Fonseca and B. List, *Angew. Chem., Int. Ed.*, 2004, **43**, 3958; (d) A. J. Cobb, D. M. Shaw, D. A. Longbottom, J. B. Gold and S. V. Ley, *Org. Biomol. Chem.*, 2005, **3**, 84; (e) S. Brandau, A. Landa, J. Franzén, M. Marigo and K. A. Jørgensen, *Angew. Chem.*,



- Int. Ed.*, 2006, **45**, 4305; (f) K. R. Knudsen, C. E. T. Mitchell and S. V. Ley, *Chem. Commun.*, 2006, **1**, 66.
- 26 (a) V. K. Aggarwal, C. Lopin and F. Sandrinelli, *J. Am. Chem. Soc.*, 2003, **125**, 7596; (b) W. Zhuang and A. Jørgensen, *Angew. Chem., Int. Ed.*, 2004, **43**, 4476; (c) A. Armstrong, *Angew. Chem., Int. Ed.*, 2004, **43**, 1460; (d) D. B. Ramachary and C. F. Barbas, *J. Org. Chem.*, 2004, **69**, 5838; (e) N. S. Chowdari and C. F. Barbas, *Org. Lett.*, 2005, **7**, 807; (f) A. McNally, B. Evans and M. J. Gaunt, *Angew. Chem., Int. Ed.*, 2006, **45**, 2116.
- 27 S. G. Li, X. Q. Hu, Z. X. Jia and P. F. Xu, *Tetrahedron*, 2010, **66**, 8557.
- 28 K. B. Somai Magar, L. Xia and Y. R. Lee, *Chem. Commun.*, 2015, **51**, 8592.
- 29 Y. Hayashi, A. Takikawa, S. Koshino and K. Ishida, *Chem.–Eur. J.*, 2019, **25**, 10319.
- 30 S. Koshino, A. Takikawa, K. Ishida, T. Taniguchi, K. Monde, E. Kwon, S. Umemiya and Y. Hayashi, *Chem.–Eur. J.*, 2020, **26**, 4524.
- 31 X. Song, X. Zhang, S. Zhang, H. Li and W. Wang, *Chem.–Eur. J.*, 2012, **18**, 9770.
- 32 L. Jiang, H. Li, J.-F. Zhou, M.-W. Yuan, H.-L. Li, Y.-M. Chuan and M.-L. Yuan, *Synth. Commun.*, 2018, **48**, 336.
- 33 (a) F. M. D. Ismail, *J. Fluorine Chem.*, 2002, **118**, 27; (b) D. O'Hagan, *Chem. Soc. Rev.*, 2008, **37**, 308; (c) S. Purser, P. R. Moore, S. Swallow and V. Gouverneur, *Chem. Soc. Rev.*, 2008, **37**, 320; (d) T. Furuya, A. S. Kamlet and T. Ritter, *Nature*, 2011, **473**, 470; (e) T. Liang, C. N. Neumann and T. Ritter, *Angew. Chem., Int. Ed.*, 2013, **52**, 8214; (f) J. Wang, M. Sánchez-Roselló, J. Luis Aceña, C. del Pozo, A. E. Sorochinsky, S. Fustero, V. A. Soloshonok and H. Liu, *Chem. Rev.*, 2014, **114**, 2432; (g) C. Ni, M. Hu and J. Hu, *Chem. Rev.*, 2015, **115**, 765; (h) X. Yang, T. Wu, R. J. Phipps and F. D. Toste, *Chem. Rev.*, 2015, **115**, 826; (i) T. Ahrens, J. Kohlmann, M. Ahrens and T. Braun, *Chem. Rev.*, 2015, **115**, 931; (j) C. Ni and J. Hu, *Chem. Soc. Rev.*, 2016, **45**, 5441; (k) Q. Zhao, C. Peng, H. Huang, S.-J. Liu, Y.-J. Zhong, W. Huang, G. He and B. Han, *Chem. Commun.*, 2018, **54**, 8359; (l) X. H. He, Y. L. Ji, C. Peng and B. Han, *Adv. Synth. Catal.*, 2019, **361**, 1923; (m) X. Zhang, X. Li, J.-L. Li, Q.-W. Wang, W.-L. Zou, Y.-Q. Liu, Z.-Q. Jia, F. Peng and B. Han, *Chem. Sci.*, 2020, **11**, 2888; (n) J.-L. Li, Y. Q. Liu, W. L. Zou, R. Zeng, X. Zhang, Y. Liu, B. Han, Y. He, H. J. Leng and Q. Z. Li, *Angew. Chem., Int. Ed.*, 2020, **59**, 1863.
- 34 F. Peng, Q. Zhao, W. Huang, S.-J. Liu, Y.-J. Zhong, Q. Mao, N. Zhang, G. He and B. Han, *Green Chem.*, 2019, **21**, 6179.
- 35 X. H. He, L. Yang, Y. L. Ji, Q. Zhao, M. C. Yang, W. Huang, C. Peng and B. Han, *Chem.–Eur. J.*, 2018, **24**, 1947.
- 36 D. B. Ramachary, K. Ramakumar and V. V. Narayana, *J. Org. Chem.*, 2007, **72**, 1458.
- 37 A. Link and C. Sparr, *Angew. Chem., Int. Ed.*, 2014, **53**, 5458.
- 38 V. C. Faseke and C. Sparr, *Angew. Chem., Int. Ed.*, 2016, **55**, 7261.
- 39 L. Zhang, N. Fu and S. Luo, *Acc. Chem. Res.*, 2015, **48**, 986.
- 40 (a) Y.-C. Chen, *Synlett*, 2008, 1919; (b) F. Peng and Z. Shao, *J. Mol. Catal. A: Chem.*, 2008, **285**, 1; (c) L.-W. Xu, J. Luo and Y. Lu, *Chem. Commun.*, 2009, 1807; (d) P. Melchiorre, *Angew. Chem., Int. Ed.*, 2012, **51**, 9748.
- 41 D. Lotter, M. Neuburger, M. Rickhaus, D. Haussinger and C. Sparr, *Angew. Chem., Int. Ed.*, 2016, **55**, 2920–2923.
- 42 P. K. Jaiswal, S. Biswas, S. Singh and S. Samanta, *Org. Biomol. Chem.*, 2013, **11**, 8410.
- 43 S. Biswas, A. Dagar, A. Srivastava and S. Samanta, *Eur. J. Org. Chem.*, 2015, **2015**, 4493.
- 44 M. Chang, C. Wu, J. Zheng and Y. Huang, *Chem.–Asian J.*, 2016, **11**, 1512.
- 45 Y.-L. Ji, X.-H. He, G. Li, Y.-Y. Ai, H.-P. Li, C. Peng and B. Han, *Org. Chem. Front.*, 2020, **7**, 563.
- 46 Y. S. Xie, R. F. Huang, R. Li, C. B. Zhang, J. Y. Fu, L. L. Zhao and J. F. Yuan, *Chem. Commun.*, 2020, **56**, 1948.
- 47 S. Xue, Q.-F. Zhou, F. Yang and Q.-X. Guo, *Synlett*, 2007, 2073.
- 48 S. Xue, B. Hu, L.-G. Meng, Y.-L. Liu and M. Liang, *Synlett*, 2009, **24**, 4137.
- 49 M. Terada and K. Dan, *Chem. Commun.*, 2012, **48**, 5781.
- 50 S. Jia, S. Li, Y. Liu, W. Qin and H. Yan, *Angew. Chem., Int. Ed.*, 2019, **58**, 18496.
- 51 J. Zheng, Y. Huang and Z. M. Li, *Org. Lett.*, 2013, **15**, 5064.
- 52 X. Chen, D. Gao, D. Wang, T. Xu, W. Liu, P. Tian and X. Tong, *Angew. Chem., Int. Ed.*, 2019, **58**, 15334.
- 53 S. K. Arupula, A. A. Qureshi and K. C. K. Swamy, *J. Org. Chem.*, 2020, **85**, 4130.
- 54 (a) A. Igau, H. Grutzmacher, A. Baceiredo and G. Bertrand, *J. Am. Chem. Soc.*, 1988, **110**, 6463; (b) A. Igau, A. Baceiredo, G. Trinquier and G. Bertrand, *Angew. Chem., Int. Ed.*, 1989, **28**, 621; (c) A. J. Arduengo, R. L. Harlow and M. Kline, A Stable Crystalline Carbene, *J. Am. Chem. Soc.*, 1991, **113**, 361.
- 55 T. Dröge and F. Glorius, *Angew. Chem., Int. Ed.*, 2010, **49**, 6940.
- 56 (a) D. Enders, O. Niemeier and A. Henseler, *Chem. Rev.*, 2007, **107**, 5606; (b) A. Grossmann and D. Enders, *Angew. Chem., Int. Ed.*, 2012, **51**, 314; (c) X. Bugaut and F. Glorius, *Chem. Soc. Rev.*, 2012, **41**, 3511; (d) S. J. Ryan, L. Candish and D. W. Lupton, *Chem. Soc. Rev.*, 2013, **42**, 4906; (e) K. P. Jang, G. E. Hutson, R. C. Johnston, E. O. McCusker, P. H. Y. Cheong and K. A. Scheidt, *J. Am. Chem. Soc.*, 2014, **136**, 76; (f) M. N. Hopkinson, C. Richter, M. Schedler and F. Glorius, *Nature*, 2014, **510**, 485; (g) Q. Zhao, B. Han, B. Wang, H.-J. Leng, C. Peng and W. Huang, *RSC Adv.*, 2015, **5**, 26972; (h) X. Y. Chen, Q. Liu, P. Chauhan and D. Enders, *Angew. Chem., Int. Ed.*, 2018, **57**, 3862.
- 57 S. J. Ryan, L. Candish and D. W. Lupton, *J. Am. Chem. Soc.*, 2011, **133**, 4694.
- 58 X. Huang, T. Zhu, Z. Huang, Y. Zhang, Z. Jin, G. Zononi and Y. R. Chi, *Org. Lett.*, 2017, **19**, 6188.
- 59 K. Q. Chen, Z. Luo, Z. H. Gao and S. Ye, *Chem.–Eur. J.*, 2019, **25**, 3253.
- 60 T. Zhu, Y. Liu, M. Smetankova, S. Zhuo, C. Mou, H. Chai, Z. Jin and Y. R. Chi, *Angew. Chem., Int. Ed.*, 2019, **58**, 15778.
- 61 J. M. Hu, J. Q. Zhang, B. B. Sun, J. B. Chen, J. Q. Yu, X. P. Yang, H. P. Lv, Z. Wang and X. W. Wang, *Org. Lett.*, 2019, **21**, 8582.
- 62 Q. Jia and J. Wang, *Org. Lett.*, 2016, **18**, 2212.



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- 63 C.-L. Zhang, Z.-H. Gao, Z.-Q. Liang and S. Ye, *Adv. Synth. Catal.*, 2016, **358**, 2862.
- 64 C. L. Zhang and S. Ye, *Org. Lett.*, 2016, **18**, 6408.
- 65 D. Liu, Y. Gao, J. Huang, Z. Fu and W. Huang, *J. Org. Chem.*, 2018, **83**, 14210.
- 66 T. Zhu, C. Mou, B. Li, M. Smetankova, B. A. Song and Y. R. Chi, *J. Am. Chem. Soc.*, 2015, **137**, 5658.
- 67 K. Xu, W. Li, S. Zhu and T. Zhu, *Angew. Chem., Int. Ed.*, 2019, **58**, 17625.
- 68 J. Liu, D. K. Das, G. Zhang, S. Yang, H. Zhang and X. Fang, *Org. Lett.*, 2018, **20**, 64.
- 69 T. Zhu, P. Zheng, C. Mou, S. Yang, B. A. Song and Y. R. Chi, *Nat. Commun.*, 2014, **5**, 5027.
- 70 P. Chauhan and D. Enders, *Angew. Chem., Int. Ed.*, 2014, **53**, 1485.
- 71 J. Wu, C. Mou and Y. R. Chi, *Chin. J. Chem.*, 2018, **36**, 333.
- 72 Z. Ni, C. Mou, X. Zhu, P. Qi, S. Yang, Y. R. Chi and Z. Jin, *Eur. J. Org. Chem.*, 2020, **2020**, 492.
- 73 L. Candish, A. Levens and D. W. Lupton, *Chem. Sci.*, 2015, **6**, 2366.
- 74 Y. Nakano and D. W. Lupton, *Angew. Chem., Int. Ed.*, 2016, **55**, 3135.

