



Scheme 1 (A) Strategies toward net-reductive decarboxylative-based cross-couplings. (B) Overview of electron donor-acceptor (EDA) photoactivation. (C) Electron donor-acceptor (EDA) complex platform for Ni-mediated alkyl transfer using HE as an organic photoreductant.

based radicals through direct visible-light excitation in Ni-catalyzed cross-couplings remains underdeveloped.

Recently, synthetic methods driven by the photoactivity of electron donor-acceptor (EDA) complexes (Scheme 1B) have gained considerable momentum, including borylation, thioetherification, and sulfonylation.¹⁶ Inspired by this advance, we examined the feasibility of EDA complex photoactivation as an enabling technology in Ni-mediated C(sp³)-C(sp²) cross-couplings (Scheme 1C). Under light irradiation at 390 nm, a commercially available and inexpensive electron donor, Hantzsch ester (HE, diethyl-1,4-dihydro-2,6-dimethyl-3,5-pyridinedicarboxylate), serves as a potent organic photoreductant to deliver diverse radical architectures from carboxylic acid feedstocks¹⁷ for further functionalization in Ni-catalyzed cross-couplings. As part of its dual role, the excited HE modulates the oxidation state of the transition-metal, delivering catalytically active Ni(0) species, thus bypassing the need for exogenous, expensive photocatalysts.

Discussion

Encouraged by the potential synthetic applications of harnessing photoactive EDA complexes toward Ni-mediated bond

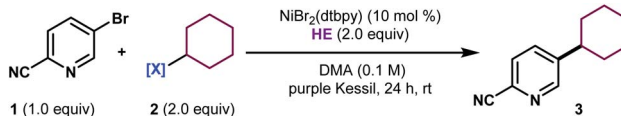
formation, we examined the feasibility of the proposed net-reductive cross-electrophile coupling using 5-bromo-2-cyanopyridine **1** and cyclohexyl-*N*-hydroxyphthalimide-ester **2** as model substrates (Table 1). From the outset of our investigation, it was evident that the solvent plays a key role in this process (entries 1–4), as it heavily affects the molecular assembly and formation of EDA complexes. Once exciplex-based charge transfer occurs, the resulting radical ion pair is stabilized by interaction with solvent dipoles.¹⁸ In this vein, dimethylacetamide (DMA) proved crucial to the success of this photochemical method.

We then studied the influence of the dihydropyridine (DHP) backbone on the efficacy of the cross-coupling (Table 1, entries 5–7). To this end, we subjected four different DHP derivatives to the reaction conditions to gain a deeper understanding of their dual role in EDA complex photoactivation as well as the reduction of Ni species through SET paradigms. It was demonstrated that Hantzsch ester (HE) and cyano substitution at C3 and C5 of the DHP (HE **A**, entry 5) promoted the reaction most efficiently. For experimental simplicity, commercially available and inexpensive Hantzsch ester HE was adopted as the standard photoreductant. The C-4-substituted DHP (HE **B**, entry 6) resulted in diminished reactivity. Not surprisingly, 4,4'-dimethyl HE **C** (entry 7) led to no product formation. This result can be rationalized by the lack of photooxidative aromatization as an intrinsic driving force to suppress a competitive back electron transfer (BET) event from the radical ion pair, restoring the ground-state EDA complex.^{16a,19} Finally, a ligand screen was performed (Table 1, entries 10–12), demonstrating that 4,4'-di-*tert*-butyl-2,2'-bipyridine (dtbpy), 2,2'-bipyridine (bpy), and electron-rich 4,4'-dimethoxy-2,2'-bipyridine (dMeObpy) function as viable ligand frameworks. Of note, modest conversion to **3** was observed using ligand-free nickel(II) bromide trihydrate.

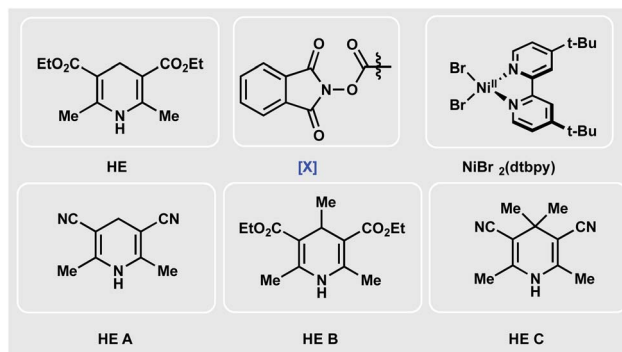
Notably, the developed net-reductive photochemical conditions are user-friendly, employing an air-stable nickel precatalyst and a mild, homogeneous reductant (HE). Deviations from the standard reaction setup are tolerated. For example, modest product formation was observed when the reaction was carried out under air (Table 1, entry 9). Similarly, although superior reactivity was accomplished using purple Kessil irradiation ($\lambda_{\text{max}} = 390 \text{ nm}$), affording a potent photoreductant [$E_{\text{red}}(\text{HE}^*/\text{HE}^+) = -2.28 \text{ V vs. SCE}$],²⁰ comparable results were achieved under blue light (entry 8, $\lambda_{\text{max}} = 456 \text{ nm}$). Control experiments demonstrated that all reaction parameters are key to the formation of C(sp³)-C(sp²) linkages (Table 1, entries 14–16).

With suitable conditions established, we examined the scope of the decarboxylative arylation employing a broad palette of (hetero)aryl bromides (Scheme 2). In general, organic halides substituted with electron-withdrawing groups exhibited excellent reactivity, although electron-neutral and electron-donating groups also afforded the desired products in modest yields. More sterically-encumbered *ortho*-substituted aryl bromides (**5**, **9**) did not hinder the cross-coupling efficacy. Furthermore, substitution at the *meta* position (**4**, **6**) is tolerated. Several sensitive functional groups, including secondary sulfonamides (**8**, **19**, **20**, **31**), ketones (**4**, **11**, **18**), trimethylsilylalkyne (**9**), and



Table 1 Optimization of the reaction conditions^a


Entry	Deviation from std conditions	3/IS ratio ^b	Entry	Deviation from std conditions	3/IS ratio ^b
1	None	1.42 (79%) ^c	9	Under air	0.93
2	DMSO	Traces	10	NiBr ₂ (dMe-phen)	Traces
3	THF	Traces	11	NiBr ₂ (bpy)	1.32
4	MeCN	0.32	12	NiBr ₂ (dMeObpy)	1.37
5	HE A	1.40	13	NiBr ₂ ·3H ₂ O	0.92
6	HE B	0.36	14	No light	0
7	HE C	0	15	No Ni	0
8	Blue Kessil	1.34	16	No HE	Traces



^a Optimization reactions were performed using **1** (0.1 mmol), **2** (0.2 mmol), HE (0.2 mmol), and NiBr₂(dtbpy) (10 mol%) in dry, degassed solvent (1.0 mL, 0.1 M) under purple Kessil irradiation for 24 h at rt. ^b Product to internal standard ratio (P/IS) was calculated using 1,3,5-trimethoxybenzene as internal standard using LC-MS analysis of the crude reaction mixture. ^c Isolated yield of **3** on 0.5 mmol scale.

terminal alkene (**22**) remained intact under the developed photoredox conditions. To this end, substrates **6**, **9** and **22** can be further diversified *via* Kumada-, sila-Sonogashira-couplings, as well as Giese-type additions, respectively.²¹

Notably, several heteroarenes (electron-deficient: **3**, **12**, **13**, **34** and electron-rich: **17**, **18**) were compatible structural motifs. In particular, nitrogen-containing heteroaryl bromides, including quinoline (**13**) and pyridine (**3**, **12**, **34**) scaffolds, reacted in a chemoselective fashion to yield C(sp³)-C(sp²) linkages, despite their propensity to undergo visible light-mediated Minisci C-H alkylation with alkyl-*N*-hydroxyphthalimide-esters.²² This demonstrates a complementary reactivity mode to existing Minisci protocols, delivering linchpins that drive molecular complexity.²³

Next, the aliphatic photocoupling was evaluated with respect to redox-active carboxylate derivatives (Scheme 2). The reaction proceeded smoothly using a diverse array of proteinogenic and non-proteinogenic amino acids (**21**, **23**, **25–30**). Bifunctional reagents, including Boc- (**17**, **23**, **25–26**, **29–30**) and Fmoc-protected (**28**) amines, afforded the arylated products without compromising yields. The scope was further extended to secondary, benzylic (**35**), and stabilized α -oxy (**19**, **24**, **31**, **32**) radical architectures. Remarkably,

primary alkyl systems that lack any radical stabilizing groups displayed exceptional reactivity (**4–8**, **11–18**), providing a clear advantage in terms of scope over previously reported protocols.^{13a} Of note, medically relevant structures including thiophene (**6–8**), piperidine (**23**), and pyrrolidine (**25**) motifs were efficiently incorporated.

To demonstrate the amenability of this cross-coupling for late-stage functionalization, including glycodiversification of drug scaffolds, photoredox-generated glycosyl radicals were successfully harnessed in this dual-catalytic manifold (Scheme 2). The desired C-aryl carbohydrates (**31** and **32**) were obtained in good yields and excellent diastereoselectivity (dr > 20 : 1). The relative configuration of the major diastereomer **32** was elucidated based on X-ray crystallography with the aryl group *cis* with respect to the dimethyl acetal protecting group (Scheme 2). Efficient decarboxylative arylation was observed with pharmaceutically relevant cores, displaying a high density of pendant functional groups, including indomethacin²⁴ and loratadine²⁵ precursors (**33**, **34**). To evaluate the amenability toward bioactive molecules further, carboxylic acid derivatives stemming from dipeptide (**36**) and D-biotin (**37**) were subjected to the reaction conditions. The corresponding cross-coupled products were obtained in moderate





Scheme 2 Scope of the developed C(sp³)-C(sp²) cross-coupling. All values correspond to isolated yields after purification. Reaction conditions as depicted in Table 1, entry 1 (0.5 mmol scale).

to high yields (65–86%). Finally, to demonstrate the versatility of EDA paradigms toward Ni-catalyzed bond-forming processes, the cross-coupling was further extended to

other electron acceptors including redox-active thiols (38a–39a)²⁶ and pyridinium-activated amines (40a–41a)^{9b,16d,27} in the absence of external catalysts.



To gain insight into the mechanism of this photochemical net-reductive cross-coupling, we analysed the reaction components by UV/vis absorption spectroscopy (Fig. 1). In line with seminal reports,¹⁶ although cyclohexyl-*N*-hydroxyphthalimide-ester shows absorption in the visible light region, mixtures of the RAE and HE in DMA at 0.2 M display a significant bathochromic shift (Fig. 1B, brick red and blue lines).

The absorption band (brick red line) stems from the formation of a new molecular aggregation, a colored EDA complex (Fig. 1A), exhibiting a wavelength band tailing to 500 nm. Preliminary studies revealed an association constant of 2.04 M^{-1} of HE with **2**, indicating a plausible EDA complex association event prior to homolytic fragmentation (Fig. 1C). Analysis of this complex using Job's method²⁹ revealed a 1 : 1 stoichiometry of the most absorbing species (Fig. 1D). Notably, concentration is a crucial parameter for effective cross-coupling. A dilute reaction mixture (10^{-4} M) exhibits a blue-shifted absorption band, indicating the inhibition of EDA complex formation (see ESI†). Furthermore, a DMA solution of HE was found to absorb visible light ($\lambda > 400 \text{ nm}$), indicating selective photoexcitation of this species at 390 nm to generate a potent reducing agent (Fig. 1B, purple line). Under the optimized reaction conditions, near full recovery of pyridine was observed (relative to 2.0 equiv. of HE), demonstrating the role of

excited HE as an organic single-electron donor in this system (Scheme 3A).

To probe the formation of alkyl radical intermediates, TEMPO trapping experiments were performed. The corresponding TEMPO-adduct was isolated in 63% yield and confirmed *via* HRMS analysis (Scheme 3B, bottom right). Stoichiometric experiments with Ni complex **46**, synthesized through oxidative addition of 4-bromobenzotrifluoride to Ni(cod)₂/dtbpy, revealed that C(sp³) alkyl transfer does not occur in the absence of HE, thus highlighting the likelihood of EDA complex activation for effective cross-coupling (Scheme 3B, bottom left). As anticipated, Ni complex **46** is catalytically active in the reaction, delivering the desired arylated product in 77% yield (Scheme 3B, bottom left).

Furthermore, negligible conversion of **42** was observed in the presence of stoichiometric amounts of Ni(COD)₂/dtbpy; traces of homocoupling or alkene-side products, if any, were detected in the crude mixture, ruling out the role of Ni as a catalytic reductant toward redox-active esters (Scheme 3B, top right). Finally, it is worth noting that reduction of redox-active esters with photoredox catalysts is reported in the presence of HE as a hydrogen atom transfer (HAT) donor.³⁰ A model reaction in the absence of Ni/dtbpy and aryl halide afforded the hydro-alkylated product in 70% yield with full recovery of pyridine

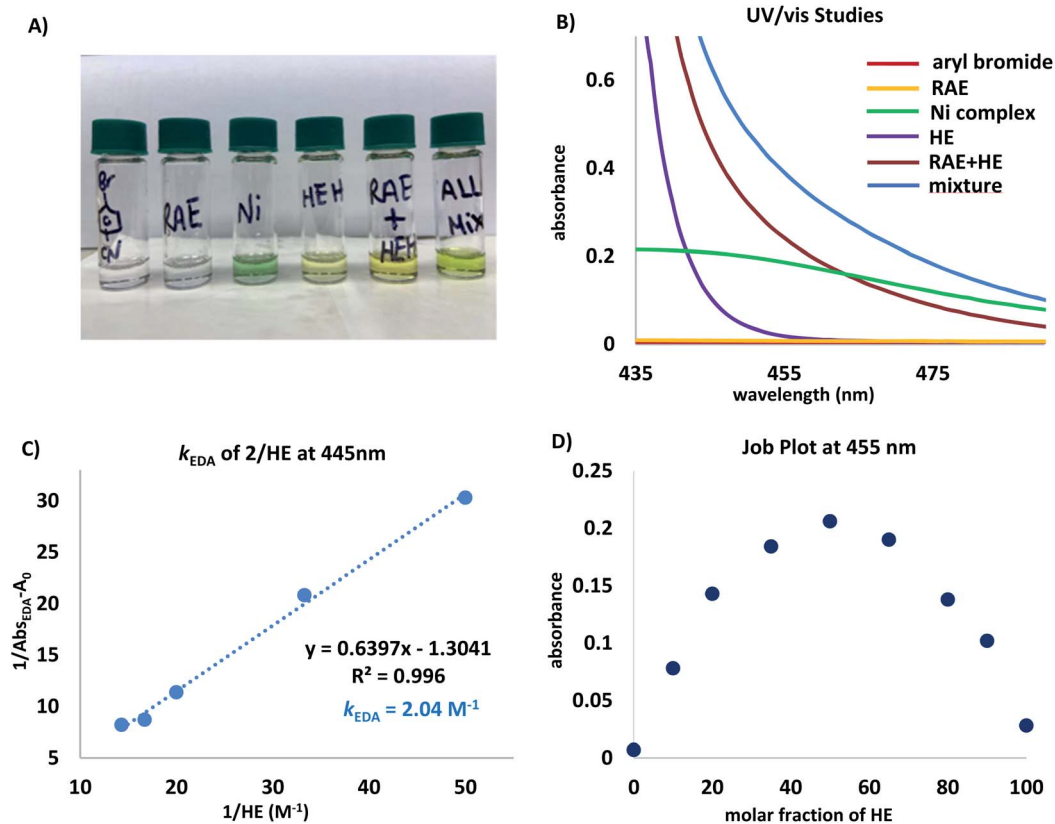
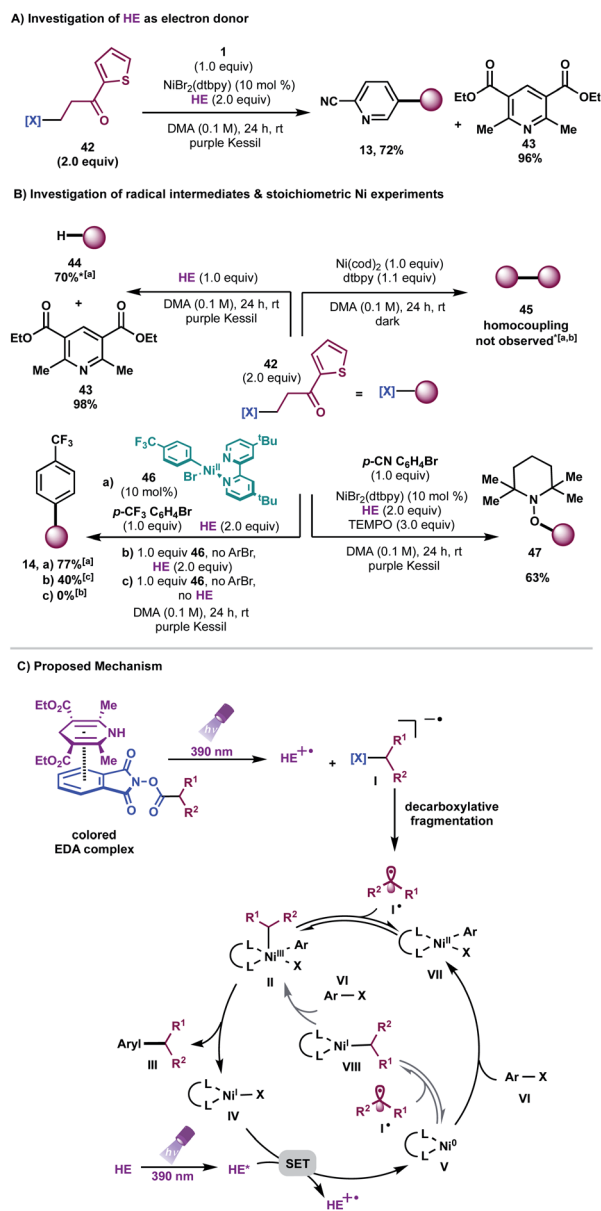


Fig. 1 (A) Visual appearance of reaction components and mixtures thereof. (B) UV/vis absorption spectra measured in DMA (0.1 M) unless otherwise noted. Ni complex = NiBr₂(dtbpy), aryl bromide = 4-bromobenzonitrile, and RAE = cyclohexyl-*N*-hydroxyphthalimide ester. Mixture refers to a DMA solution of all reaction components. (C) Benesi-Hildebrand plot.²⁸ (D) Job plot²⁹ for a mixture of *N*-(cyclohexyl)-hydroxyphthalimide ester (**2**) and HE in DMA (0.2 M).





Scheme 3 (A) Investigation of HE as electron donor. (B) Mechanistic experiments. ^[a]isolated yield on 0.3 mmol scale, ^[b]analysed *via* GC-MS analysis, ^[c]NMR yield, *1.0 equiv. of 42. (C) Proposed mechanism.

(Scheme 3B, top left). These findings demonstrate that the rate of alkyl radical addition to aryl-Ni(II) species **VII** must be faster than hydrogen atom abstraction from HE.

Based on these experiments, we propose a mechanistic scenario involving the intermediacy of an EDA complex between the electron-deficient, aliphatic redox-active esters and the electron-rich HE (Scheme 3C). Photoirradiation at 390 nm triggers an intra-complex SET event, generating a dihydropyridine radical cation and a phthalimide radical anion. The latter species undergoes decarboxylative fragmentation to yield an alkyl radical **I'**. Although the absorptivity of the EDA complex is significantly greater than that of the Hantzsch ester itself, and although the entropic advantage inherent in intra-complex

charge transfer provides an enormous rate enhancement over intermolecular SET, we cannot rule out the intervention of some direct electron transfer from the photoexcited Hantzsch ester to initiate radical formation from the RAE. This high-energy, C(sp³)-hybridized intermediate can suffer two potential fates. Based on previous computational studies,¹¹ one plausible mechanistic pathway involves initial radical combination with a Ni⁰ species, generating a Ni^I intermediate that engages in oxidative addition with the aryl halides to produce high valent Ni^{III} species **II**. Subsequent reductive elimination from this complex yields the desired cross-coupled product and the corresponding L_nNi^I species **IV**. At this juncture, a key SET event from the excited state HE [$E_{\text{red}}(\text{HE}^*/\text{HE}^{+\bullet}) = -2.28 \text{ V vs. SCE}^{20}$] to Ni [$E_{\text{red}}(\text{Ni}^{\text{I}}/\text{Ni}^0) = -1.17 \text{ V vs. SCE in THF}^{10d}$] regenerates the active Ni⁰ catalyst. However, a process that involves initial oxidative addition of the aryl halides to Ni⁰ species **V**, affording aryl-Ni^{II} complex **VII**, cannot be ruled out based on stoichiometric experiments with Ni complex **46** (Scheme 3B). In this scenario, **VII** would engage the radical to generate Ni^{III} complex **II**, which would then be carried on through the catalytic cycle.

Conclusion

In summary, we have demonstrated the implementation of an electron donor-acceptor (EDA) complex platform toward Ni-catalyzed C(sp³)-C(sp²) bond formation, circumventing the need for exogenous photocatalysts, additives, and stoichiometric metal reductants. Under the developed conditions, the generation of heteroatom- and unactivated carbon-based radicals through direct visible-light excitation of diverse electron acceptors is feasible, providing a clear advantage in terms of scope over previously reported photochemical systems.¹³ Upon light irradiation at 390 nm, excited HE functions as a potent organic photoreductant exhibiting a dual role: generation of reactive C(sp³)-hybridized radicals through EDA complex activation as well as restoration of the desired catalytically active Ni(0) species through SET paradigms. This decarboxylative cross-electrophile arylation is amenable to the synthesis of peptidomimetics, drug-like molecules, and the diastereoselective functionalization of carbohydrates. The commercial availability of carboxylic acids and related electron acceptor species facilitates the rapid incorporation of diverse carbon- and heteroatom-based radical architectures with high functional group tolerance. Key mechanistic and spectroscopic studies highlight the necessity for EDA photoactivation for efficient alkyl transfer and help inform the design of improved EDA-based paradigms in Ni-catalyzed cross-couplings.

Author contributions

Shorouk O. Badir conceived the topic. Lisa Marie Kammer, Shorouk O. Badir, and Ren-Ming Hu completed experiments with input from Gary A. Molander. Shorouk O. Badir and Lisa Marie Kammer prepared the manuscript with input from Gary A. Molander.



Conflicts of interest

There are no conflicts to declare.

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