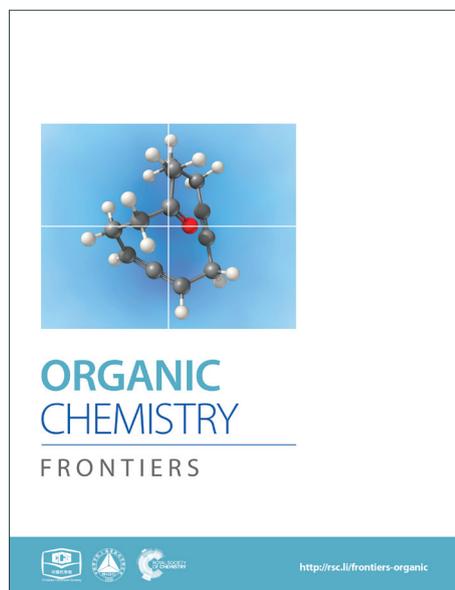
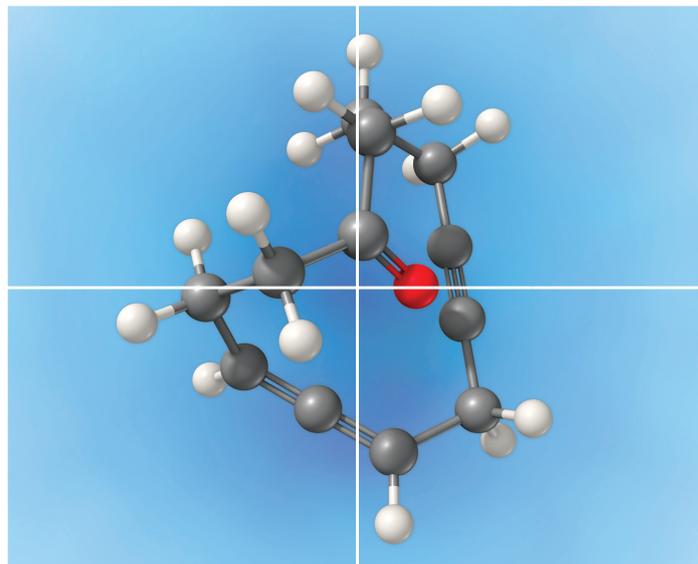


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ARTICLE

One-pot dilithiation-lithium-zinc exchange-Negishi coupling approach to 2,6-di(hetero)aryl substituted dithienothiazines – A novel class of electronically fine-tunable redox systems^{†,‡}

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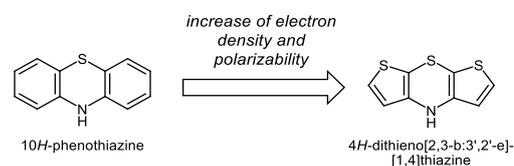
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Catherine Dostert,^a and Thomas J. J. Müller^{*a}

2,6-Di(hetero)aryl and 2-(hetero)aryl substituted dithienothiazines are prepared from *N*-aryl dithienothiazines by a lithiation-lithium-zinc exchange-Negishi cross-coupling sequence with (hetero)aryl iodides in a one-pot fashion in good to excellent yield. These novel extended π -electron systems can be reversibly oxidized and fine-tuned in their electronic properties as supported by cyclic voltammetric, and absorption and emission spectroscopic studies.

Introduction

Over the past years, the interest in electroactive organic molecules as functional π -systems¹ has enormously increased due to important potential technological applications ranging from organic light-emitting diodes² over organic photovoltaic devices³ to organic field-effect transistors.⁴ Main advantages of using organic materials are their low production costs, favorable properties such as flexibility, transparency and light weight, the tunability of their properties, and their good processability.⁵ As a consequence the exploration of novel semiconducting molecular materials has become an ongoing challenge to synthetic organic chemistry. Moreover, small redox-active molecules can be considered as molecular wires⁶ and, therefore, open new alleys to unimolecular electronics.⁷ Recently we presented 4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazines, congeners to the well-established class of phenothiazines, as novel electron-rich organic π -systems (Scheme 1).⁸ As a consequence of their unique electronic properties showing two reversible oxidations with Nernstian behavior at low oxidation potentials dithienothiazines qualify in principle well for the use as conducting materials or as a donor component in donor-acceptor conjugates.



Scheme 1. 4*H*-Dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine – a congener of 10*H*-phenothiazine with increased electron density.

For potential applications, functionalization of the heterocyclic core represents a key step and major challenge. Most interestingly, the thiophene anellation offers an easy entry to thiophene characteristic transformations, such as lithiation in the α -positions with respect to the sulfur atom,⁹ which is also suitable for sequential one-pot processes.¹⁰ This concept was

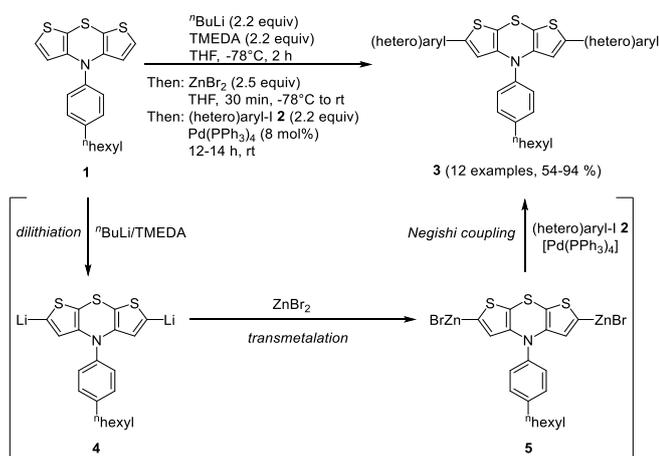
very successfully transposed to the 2,6-difunctionalization of *N*-substituted dithienothiazines via a dilithiation-electrophilic trapping sequence.¹¹ Therefore, a one-pot dilithiation-cross-coupling strategy to 2,6-di(hetero)aryl substituted dithienothiazines lies at hand. Here, we report a consecutive one-pot dilithiation-transmetalation-Negishi coupling sequence to 2,6-di(hetero)aryl substituted dithienothiazines and 2-(hetero)aryl substituted dithienothiazines and the elucidation of their electronic properties by cyclic voltammetry, absorption and emission spectroscopy, and DFT calculations.

Results and discussion

Synthesis and structure

Interestingly, the electrophile addition to 2,6-dilithio dithienothiazine¹¹ opens a rapid access to highly reactive organometallic species setting the stage for a subsequent cross-coupling reaction in a consecutive one-pot fashion.¹⁰ Therefore, we set out to introduce zinc bromide as an electrophile for generating symmetrical bis(organozinc halides), suitable nucleophiles for envisioned Negishi coupling.¹² The twofold lithiation of *N*-[4-(*n*-hexyl)phenyl] 4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (**1**) with a slight excess of *n*-BuLi/TMEDA (*N,N,N',N'*-tetramethylethylenediamine) at -78 °C was ensured by reverse addition¹³ of dithienothiazine **1** to a precooled solution of *n*-BuLi/TMEDA (Scheme 2). Addition of freshly dried zinc bromide to the 2,6-dilithio dithienothiazine **4** furnished the organozinc bromide species **5**. The subsequent addition of (hetero)aryl iodides **2** and a catalytic amount of Pd(PPh₃)₄ gave rise to the twofold Negishi cross-coupling as terminating step of this one-pot process furnishing after workup and flash chromatography on silica gel 2,6-di(hetero)aryl substituted dithienothiazines **3** in moderate to excellent yield (Table 1). The structures of all compounds **3** have been unambiguously assigned by spectroscopy (NMR, IR, MS) and elemental analysis or HRMS. In the NMR spectra, expectedly, characteristic single sets of signals are found for the methine

proton singlets around δ 6.2-6.6 and methine carbon nuclei at δ 115-120 of the central dithienothiazine cores.



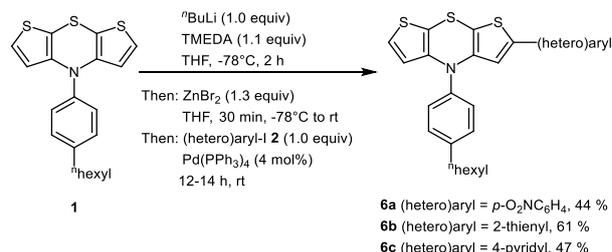
Scheme 2. One-pot synthesis of 2,6-di(hetero)aryl substituted dithienothiazines **3** by dilithiation-transmetalation-Negishi coupling sequence.

Table 1. One-pot synthesis of 2,6-di(hetero)aryl substituted dithienothiazines **3**.

Entry	(Hetero)aryl iodide 2	2,6-Di(hetero)aryl dithienothiazines 3 (yield)
1	(Hetero)aryl = 4-MeOC ₆ H ₄ (2a)	3a (65 %)
2	(Hetero)aryl = 4-MeC ₆ H ₄ (2b)	3b (82 %)
3	(Hetero)aryl = Ph (2c)	3c (87 %)
4	(Hetero)aryl = 4-ClC ₆ H ₄ (2d)	3d (78 %)
5	(Hetero)aryl = 4-MeO ₂ CC ₆ H ₄ (2e)	3e (80 %)
6	(Hetero)aryl = 4-F ₃ CC ₆ H ₄ (2f)	3f (95 %)
7	(Hetero)aryl = 4-NCC ₆ H ₄ (2g)	3g (69 %)
8	(Hetero)aryl = 4-O ₂ NC ₆ H ₄ (2h)	3h (78 %)
9	(Hetero)aryl = 3-O ₂ NC ₆ H ₄ (2i)	3i (77 %)
10	(Hetero)aryl = 2-O ₂ NC ₆ H ₄ (2j)	3j (54 %)
11	(Hetero)aryl = 3-pyridyl (2k)	3k (63 %)
12	(Hetero)aryl = 10- <i>n</i> -hexyl-10 <i>H</i> -phenothiazin-3-yl (2l)	3l (94 %)

The synthesis proceeds smoothly with a variety of aromatic (**2a-j**) and heteroaromatic iodides (**2k** and **2l**) and the substituents can be electron donating (**2a, 2b**), electro-neutral (**2c**), and electron withdrawing substituents (**2d-g**). Moreover, it is remarkable that the sequence can be conducted with nearly equistoichiometric amounts of reagents to give the targeted structures with excellent efficiency.

Likewise we probed the synthesis of 2-(hetero)aryl substituted dithienothiazines **6** in three representative examples by employing equimolar amounts of reagents. With this adapted stoichiometry under dropwise addition of *n*-BuLi to a THF solution of **1** and TMEDA, the intermediacy of a monolithiated specimen furnished after workup and chromatography 2-(hetero)aryl substituted dithienothiazines **6** in moderate to good yields (Scheme 3) and the structures of these monosubstituted derivatives **6** have been unambiguously assigned by spectroscopy (NMR, IR, MS) and elemental analysis or HRMS. The lower symmetry of the structures **6** in comparison to the disubstituted derivatives **3** manifests in the appearance of three distinct methine signals, two as doublets at δ 6.1 and 7.15 with coupling constants $J = 5.5$ Hz and one as singlets between δ 6.2-6.6 in the proton NMR spectra, and three resonances for the methine carbon nuclei of the central dithienothiazine cores between δ 116 and 125 in the ¹³C NMR spectra.



Scheme 3. One-pot synthesis of 2-(hetero)aryl substituted dithienothiazines **6** by monolithiation-transmetalation-Negishi coupling sequence.

The structure of the symmetrically disubstituted dithienothiazines **3** was additionally corroborated by studying the DFT-optimized geometries¹⁴ (B3LYP functional¹⁵ and 6-311G*¹⁶ basis set) for the parent compound **1** and selected derivatives of **3**, where the *n*-hexyl substituent has been truncated to an ethyl group for reducing the computational time (Table 2). The *para*-ethylphenyl substituent on the thiazine core was always oriented in a quasi-axial configuration to ensure global energy minima of the computed structures. The influence of the substituents in positions 2 and 6 on the central 1,4-thiazine folding angle and the thienyl-aryl torsional dihedral angle was studied for electron-releasing (structure **3a**), electroneutral (structure **3c**), electron-withdrawing (structure **3g**), and electron-withdrawing *para*- (structure **3h**), *meta*- (structure **3i**), and *ortho*-substituents (structure **3j**).

Table 2. Selected DFT-calculated dihedral angles of structures **1, 3a, 3c,** and **3g-j**.

Structure ^a	Central 1,4-thiazine folding angle	Thienyl-aryl torsional dihedral angle
1	140.2 °	---
3a	140.3 °	30.8 °
3c	140.7 °	33.4 °
3g	141.1 °	28.2 °
3h	141.1 °	28.0 °
3i	140.6 °	28.1 °
3j	141.1 °	59.8 °

^aIn contrast to the synthesized structures **1** and **3** the *n*-hexyl substituent has been truncated to an ethyl group.

While the central 1,4-thiazine folding angle lies in a narrow margin between 140.2 and 141.1 ° the thienyl-aryl torsional dihedral angle is diminished to 28 ° for remote electron-withdrawing substituents (structures **3g-i**) in contrast to 30 ° (structure **3a**) and 33 ° (structure **3c**) for electron-releasing and electroneutral substituents. For the former a more efficient overlap of the π -systems can be expected in the electronic ground state. Expectedly, an *ortho*-nitrophenyl substituent (structure **3j**) results in torsion of 60 ° from coplanarity.

Electronic properties

The electronic properties of 2,6-di(hetero)aryl substituted dithienothiazines **3** and 2-(hetero)aryl substituted dithienothiazines **6** were experimentally investigated by cyclic voltammetry and by absorption and emission spectroscopy (Table 3), and for elucidating the electronic structure DFT and

TDDFT calculations were performed for selected molecule structures.

The first oxidation $E_{1/2}^{0/+1}$ of 2,6-di(hetero)aryl substituted dithienothiazines **3** and 2-(hetero)aryl substituted dithienothiazines **6** are found between 270 and 500 mV, while the second oxidations $E_{1/2}^{+1/+2}$ from the radical cations to the

dications occur between 1000 and 1250 mV. With exception of compounds **3k** and **6c** all other 2,6-di(hetero)aryl substituted dithienothiazines **3** and 2-(hetero)aryl substituted dithienothiazines **6** show two clearly separated reversible oxidation waves with Nernstian behavior (Figure 1).

Table 3. Selected electronic properties (absorption and emission data, Stokes shift $\Delta\tilde{\nu}$, oxidation potentials, and semiquinone formation constant K_{SEM}) of dithienothiazine **1**, 2,6-di(hetero)aryl and 2-(hetero)aryl substituted dithienothiazines **3** and **6**.

Compound	Absorption maxima $\lambda_{max,abs}$ (ϵ) ^a [nm] (Lcm ⁻¹ mol ⁻¹)	Emission $\lambda_{max,em}$ ^b [nm]	Stokes-shift $\Delta\tilde{\nu}$ [cm ⁻¹]	$E_{1/2}^{0/+1}$ [mV] ^d	$E_{1/2}^{+1/+2}$ [mV] ^d	K_{SEM} ^c
1	240 (22000), 318 (6000)	-	-	360	1230	$55.7 \cdot 10^{13}$
3a	295 (46100), 367sh (6000)	-	-	270	1000	$0.21 \cdot 10^{13}$
3b	294 (48200), 385 (5700)	-	-	320	1080	$1.04 \cdot 10^{13}$
3c	290 (49500), 400 (5300)	-	-	340	1120	$1.26 \cdot 10^{13}$
3d	296 (49550), 409 (6000)	-	-	380	1130	$0.44 \cdot 10^{13}$
3e	309 (49600), 455 (8800)	-	-	420	1160	$0.38 \cdot 10^{13}$
3f	298 (62900), 431 (8500)	-	-	440	1180	$0.44 \cdot 10^{13}$
3g	310 (43600), 457 (7200)	-	-	480	1210	$0.26 \cdot 10^{13}$
3h	254 (31300), 332 (46000), 524 (17900)	-	-	500	1240	$0.32 \cdot 10^{13}$
3i	286 (40100), 440 (4800)	-	-	480	1230	$0.68 \cdot 10^{13}$
3j	264 (30250), 452 (2900)	-	-	440	1250	$4.76 \cdot 10^{13}$
3k	292 (32450), 414 (3900)	434sh, 560 ^f	6300	410	- ^g	-
3l	239sh (27900), 267sh (30300), 291 (36250), 349 (14300), 422sh (8500)	497 ^h	3800	290	840, 1090 ⁱ	$0.07 \cdot 10^{13}$
6a	250 (25550), 328 (22850), 511 (7300)	-	-	430	1210	$1.94 \cdot 10^{13}$
6b	237sh (12400), 300 (12700), 355sh (3000), 420sh (1900)	569 ^k	6200	360	1130	$1.37 \cdot 10^{13}$
6c	238sh (9800), 292 (13300), 367 (1900), 449 (2200)	606 ^l	5800	440	- ^h	-

^aRecorded in CH₂Cl₂ UVASOL at rt. ^bRecorded in CH₂Cl₂ UVASOL at rt with $\lambda_{exc.} = 310.0$ nm. ^c $\Delta\tilde{\nu} = 1/\lambda_{max,abs} - 1/\lambda_{max,em}$ [cm⁻¹].

^dRecorded in CH₂Cl₂, $T = 293$ K, 0.1 M electrolyte [ⁿBu₄N][PF₆], Pt working, Ag/AgCl reference, and Pt counter electrodes.

potentials are corrected against decamethylferrocene as an external standard with $E_0^{0/+1} = -95$ mV. ^e $K_{SEM} = 10^{\frac{E_0^{0/+1} - E_0^{+1/+2}}{0.059}}$

(potentials are inserted in [V] and potential differences are dimensionless). ^f $\lambda_{max,exc} = 380$ nm. ^gThe oxidation is irreversible.

^h $\lambda_{max,exc} = 420$ nm. ⁱ $E_{1/2}^{+1/+3}$. ^j $E_{1/2}^{+3/+4}$. ^k $\lambda_{max,exc} = 425$ nm. ^l $\lambda_{max,exc} = 450$ nm.

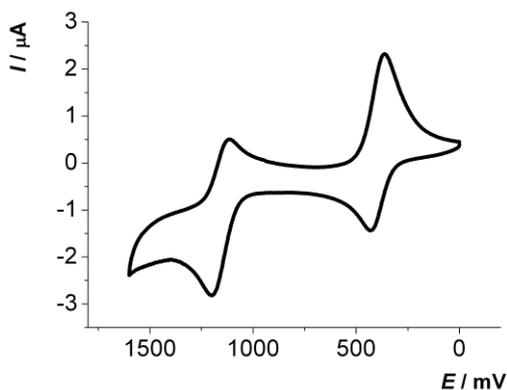


Figure 1. Cyclic voltammogram of compound **3c** (recorded in CH₂Cl₂, $T = 293$ K, 0.1 M electrolyte [ⁿBu₄N][PF₆], scan rate $v = 100$ mV/s, Pt-working, Ag/AgCl reference and Pt counter electrodes).

Interestingly, the oxidation potentials in the series of the 2,6-diaryl substituted dithienothiazines **3a–3i** vary with the electronic nature of the substituent. This strong dependence of the reversible first and second oxidation potentials on the electronic substituent effect is shown in a good linear correlation with the Hammett σ_p or σ_m substitution parameters¹⁷ ($E_0^{0/+1}$, $R^2 = 0.9889$; $E_0^{+1/+2}$, $R^2 = 0.9406$) (Figure 2). Upon

oxidation radical cations **3⁺** and dications **3²⁺** are successively formed and their stabilities are achieved by both inductive and resonance effects. The three nitrosubstituted regioisomers **3h–j** nicely illustrate the influence of remote conjugative and inductive substituent effects on the first oxidation potentials as shown by the *para*-substituted derivative **3h** ($E_{1/2}^{0/+1} = 500$ mV) and the *meta*-substituted compound **3i** ($E_{1/2}^{0/+1} = 480$ mV) whereas for compound **3j** ($E_{1/2}^{0/+1} = 440$ mV) the lack of overlap by torsion out of coplanarity underlines the stereoelectronic effect of the *ortho*-substitution.

The electrochemistry of the 2,6-di(phenothiazine-3-yl) substituted dithienothiazine **3l** is more complicated because phenothiazinyl substituents are additionally redox-active. In this case three distinctly separated, reversible oxidations appear in the cyclic voltammogram. The first and third oxidations at $E_0^{0/+1} = 290$ mV and $E_0^{+3/+4} = 1090$ mV clearly show Nernstian behavior, but the second oxidation at $E_0^{+1/+3} = 840$ mV reveals a large difference $\Delta E_{1/2} = 112$ mV for anodic and cathodic peak potentials with an overall increased intensity of the current density. Therefore, the first and third one-electron oxidations can be assigned as dithienothiazinyl centered oxidation events, whereas the second oxidation rather represents two simultaneously occurring phenothiazinyl centered oxidations as quasi-reversible events.

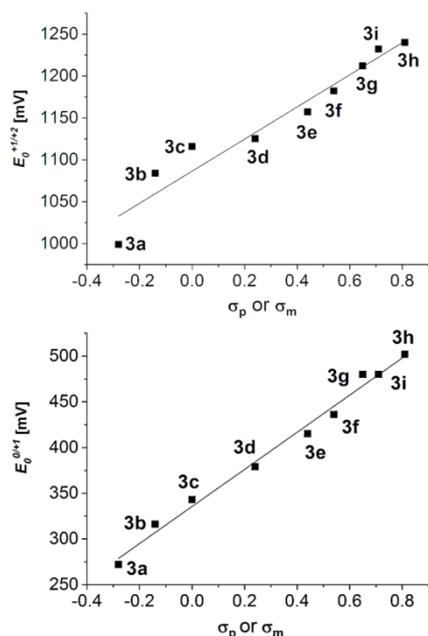


Figure 2. Linear correlation plots of the first oxidation potentials $E_0^{0/+1}$ [mV] (bottom) and second oxidation potentials $E_0^{+1/2}$ [mV] (top) of 2,6-diaryl-substituted dithienothiazines **3a-i** against Hammett σ_p or σ_m parameters ($E_0^{0/+1} = 201 \sigma_{p/m} + 336$ [mV], $r^2 = 0.9889$; $E_0^{+1/2} = 192 \sigma_{p/m} + 1086$ [mV], $r^2 = 0.9406$).

In addition, the differences between first and second oxidations are relatively large and indicate considerable stability of the electrochemically generated radical cations $3^{+\cdot}$. By calculation of the semiquinone formation constants K_{SEM} ¹⁸ for the comproportionation of **3** and 3^{2+} furnishing two moles of $3^{+\cdot}$ the stability can be quantified to lie between $0.07 \cdot 10^{13}$ and $4.76 \cdot 10^{13}$. In comparison to the 2,6-unsubstituted mother compound **1** the K_{SEM} are approximately 1-2 orders of magnitude lower. Likewise, the K_{SEM} of the 2-substituted derivatives **6** are one order of magnitude smaller.

The absorption spectra of all 2,6-di(hetero)aryl substituted dithienothiazines **3** reveal similar absorption characteristics. An intense absorption band appears around 300 nm and the less intensive longest wavelength absorption is found in a range from 367 to 525 nm (Figure 3). In comparison to the 2,6-unsubstituted mother compound **1** ($\lambda_{max,abs} = 318$ nm) the red shift of the longest wavelength absorption bands of the 2,6-di(hetero)aryl substituted derivatives **3** accounts for extended π -electron systems.

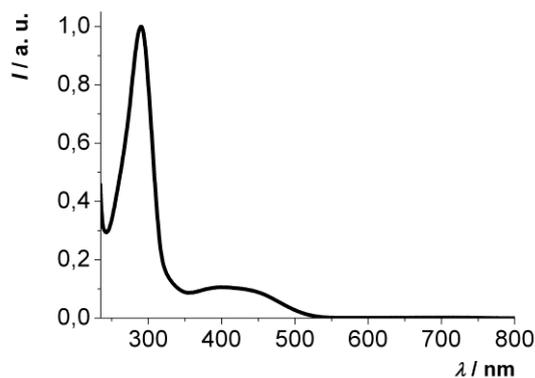


Figure 3. Absorption spectrum of compound **3c** (recorded in CH_2Cl_2 , $T = 293$ K).

The absorption behavior can be plausibly rationalized by a TD-DFT calculation^{14,15,16} of compound **3f**, where the *n*-hexyl substituent has been truncated to an ethyl group for reducing the computational time. The experimental UV/vis spectrum of **3f** is nicely reproduced by the computation (Figure 4). While the longest wavelength absorption maximum at 455 nm consists with a 48 % contribution of the HOMO-LUMO transition the strongest absorption band at around 315 nm constitutes to 43 % from a HOMO-LUMO+2 transition and the shoulder at 350 nm arises to 40 % from HOMO-1-LUMO transition. In addition the Kohn-Sham frontier molecular orbitals of **3f**, i. e. HOMO and LUMO (Figure 5), which are predominantly involved in constituting the longest wavelength absorption band, display considerable coefficient densities in the central *N*-aryl dithienothiazine core in the HOMO and in the 2- and 6-aryl substituents in the LUMO. Thereby this transition is accompanied by a considerable charge transfer character from the central donor moiety to the outer acceptor units.

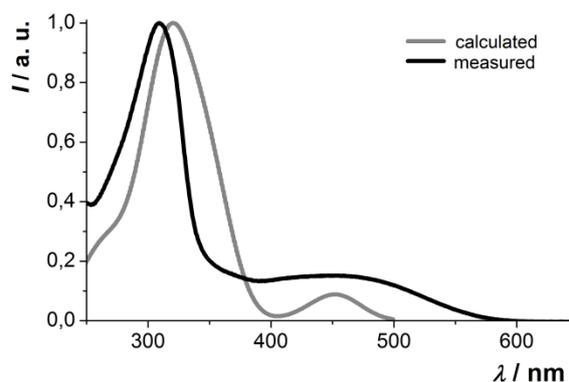


Figure 4. Experimental (black) and calculated (grey) absorption spectra of compound **3f**.

Furthermore the considerable charge transfer character of the longest wavelength absorption band becomes qualitatively apparent upon eyesight by comparing the bathochromic shift of the color of equimolar dichloromethane solutions of compounds **3c-h** with increasing acceptor strength under daylight (Figure 6).

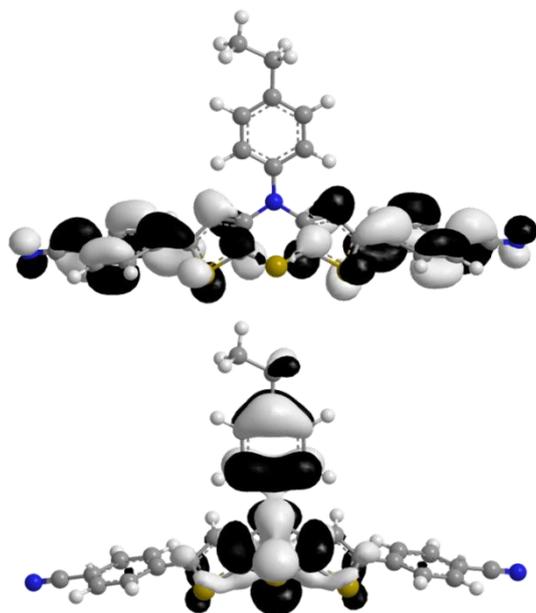


Figure 5. Frontier molecular orbitals (HOMO, bottom, and LUMO, top) of compound **3f** (DFT computations with B3LYP functional and 6-311G* basis set).

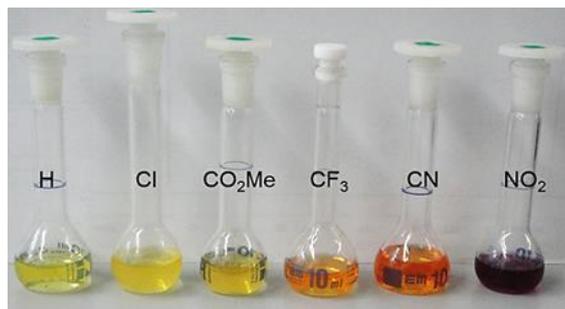


Figure 6. Bathochromic shift of equimolar dichloromethane solutions of compounds **3c-h** with increasing acceptor strength under daylight.

This observation can be additionally quantified by a good linear correlation of longest wavelength absorption maxima of compounds **3a-h** with the Hammett substitution parameter¹⁷ σ_p^- ($R^2 = 0.942$) (Figure 7), whereby σ_p^- indicates the immediate influence of the resonance stabilization of negative charges by mesomeric and inductive substituents.

While the absorption characteristics of 2-(hetero)aryl substituted dithienothiazines **6** indicate three distinct absorption bands, the UV/vis spectrum of 2,6-di(phenothiazine-3-yl) substituted dithienothiazine **3l** shows multiple absorption bands and shoulders indicating the behavior of an extended π -electron system rather than a simple additive behavior of the underlying subchromophores.

Finally, among 2,6-di(hetero)aryl substituted and 2-(hetero)aryl substituted dithienothiazines **3** and **6** the most peculiar electronic feature is reflected that in contrast to many dithienothiazines, which are essentially nonluminescent (see also Table 3),^{8,11} the four derivatives indicate intense luminescence at 560 nm (**3k**), 497 nm (**3l**), 569 nm (**6b**), and 606 nm (**6c**) with broad unstructured emission bands and large Stokes shifts (Figure 8).

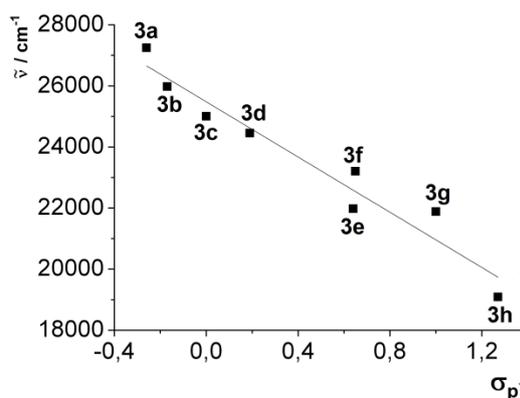


Figure 7. Linear correlation plot of the longest wavelength absorption maxima $\tilde{\nu}_{max,abs}$ [cm^{-1}] of 2,6-diaryl-substituted dithienothiazines **3a-h** against Hammett parameter σ_p^- ($\tilde{\nu} = -4522 \cdot \sigma_p^- + 25480$ [cm^{-1}], $r^2 = 0.9421$).

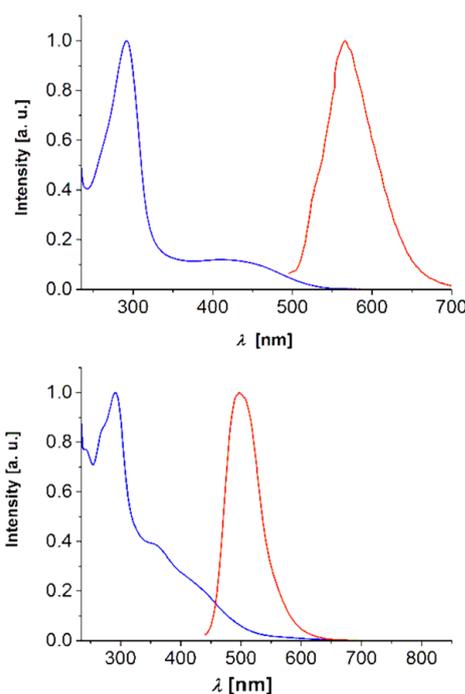


Figure 8. Normalized absorption (blue) and emission (red) spectra of compounds **3k** (top, $\lambda_{max,exc} = 380$ nm) and **3l** (bottom, $\lambda_{max,exc} = 420$ nm) (recorded in CH_2Cl_2 , $T = 293$ K).

Conclusions

The hitherto unknown classes of 2,6-di(hetero)aryl and 2-(hetero)aryl substituted dithienothiazines are readily accessible from *N*-aryl dithienothiazines by lithiation, lithium-zinc exchange, and subsequent Negishi cross-coupling with (hetero)aryl iodides in good to excellent yield and in a one-pot fashion. The title compounds possess an extended π -electron conjugation and display characteristic electronic features. While the electronic ground state properties, investigated by cyclic voltammetry and UV/vis spectroscopy, indicate a strong influence of the remote substituents that can be semiquantitatively treated by excellent Hammett's linear free enthalpy correlations with σ_p (oxidation potentials) and σ_p^-

($\lambda_{max,abs}$), luminescence as an excited state characteristics is only found for four derivatives with characteristic heterocyclic substitution. In principle this emission behavior is unusual for *N*-aryl dithienothiazines, but it establishes a novel class of redox active lumophores. The rapid access of 2,6-di(hetero)aryl dithienothiazines with remote substitution furnishing tunable electronic properties render this class of electron rich reversible multistep redox systems highly intriguing as potential hole conductor materials. Further studies exploiting this expedient synthetic strategy to functional dithienothiazines are currently underway.

Experimental

Synthetic procedures

General Procedure for the Dilithiation-Negishi Coupling Synthesis of 2,6-Di(hetero)aryl Dithienothiazines **3** (GP1).

Dry THF (3 mL) and dry TMEDA (0.17 mL, 1.2 mmol) were placed in a 50 mL Schlenk flask under nitrogen. Then, the solution was cooled to $-78\text{ }^{\circ}\text{C}$ (dry ice/acetone) and 1.6 M *n*-BuLi in hexane (0.72 mL, 1.2 mmol) was added dropwise to the solution. 4-(4-*n*-Hexylphenyl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]-

thiazine (**1**) (186 mg, 0.50 mmol) was added to this solution and the reaction mixture was stirred at $-78\text{ }^{\circ}\text{C}$ for 2 h. In the meantime a solution of freshly dried zinc bromide (300 mg, 1.33 mmol) in dry THF (1.5 mL) was prepared which was added dropwise by syringe to the reaction mixture at $-78\text{ }^{\circ}\text{C}$. After stirring at $-78\text{ }^{\circ}\text{C}$ for 30 min palladium tetrakis(triphenylphosphane) (47 mg, 8 mol%) and 2.2 equivs of the (hetero)aryl iodide **2** (1.1 mmol) were successively added to the reaction mixture (for experimental details see Table 4). The reaction mixture was allowed to come to room temp upon stirring for 12-14 h. Then, a mixture of deionized water (20 mL) and dichloromethane (20 mL) was added, the organic layer was separated and the aqueous phase was extracted with dichloromethane (3 x 20 mL). The combined organic layers were dried with anhydrous magnesium sulfate and the solvents were removed in vacuo. The crude product was purified by flash chromatography on silica gel (eluent containing 2 % of triethylamine) under nitrogen to give the pure products **3**. Further purification could be achieved by fractional crystallization.

Table 4. Experimental details of the dilithiation-Negishi coupling synthesis of 2,6-di(hetero)aryl dithienothiazines **3**.

Entry	(Hetero)aryl iodide 2 [mg] (mmol)	2,6-Di(hetero)aryl dithienothiazines 3 [mg] (%)
1	257 (1.10) of 1-iodo-4-methoxybenzene (2a)	189 (65) of 3a
2	240 (1.10) of 1-iodo-4-methylbenzene (2b)	226 (82) of 3b
3	224 (1.10) of iodobenzene (2c)	227 (87) of 3c
4	262 (1.10) of 1-chloro-4-iodobenzene (2d)	231 (78) of 3d
5	288 (1.10) of methyl 4-iodobenzoate (2e)	257 (80) of 3e
6	299 (1.10) of 4-iodobenzotrifluoride (2f)	313 (95) of 3f
7	252 (1.10) of 4-iodobenzonitrile (2g)	198 (69) of 3g
8	274 (1.10) of 1-iodo-4-nitrobenzene (2h)	241 (78) of 3h
9	274 (1.10) of 1-iodo-3-nitrobenzene (2i)	236 (77) of 3i
10	274 (1.10) of 1-iodo-2-nitrobenzene (2j)	165 (54) of 3j
11	226 (1.10) of 3-iodopyridine (2k)	164 (63) of 3k
12	450 (1.10) of 3-iodo-10- <i>n</i> -hexyl-10 <i>H</i> -phenothiazine (2l)	441 (94) of 3l

4-(4-*n*-Hexylphenyl)-2,6-bis(4-methoxyphenyl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (**3a**)

According to the GP1 and after flash chromatography on silica gel (*n*-hexane/ethyl acetate 20:1 with 2 % of triethylamine) and after crystallization from ethanol/ethyl acetate (1:1) compound **3a** (189 mg, 65 %) was obtained as yellow needles.

R_f (*n*-hexane/ethyl acetate 20:1): 0.19, Mp $170\text{ }^{\circ}\text{C}$. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313\text{ K}$): δ 0.96 (t, $^3J = 7.1\text{ Hz}$, 3 H), 1.37-1.48 (m, 6 H), 1.75 (p, $^3J = 7.4\text{ Hz}$, 2 H), 2.71-2.77 (m, 2 H), 3.79 (s, 6 H), 6.24 (s, 2 H), 6.80-6.85 (m, 4 H), 7.21-7.29 (m, 4 H), 7.33-7.41 (m, 4 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313\text{ K}$): δ 14.9 (CH $_3$), 23.8 (CH $_2$), 29.9 (CH $_2$), 32.3 (CH $_2$), 32.8 (CH $_2$), 36.6 (CH $_2$), 55.7 (CH $_3$), 102.3 (C $_{quat}$), 115.2 (CH), 115.5 (CH), 127.1 (CH), 127.2 (C $_{quat}$), 129.2 (CH), 131.1 (CH), 142.1 (C $_{quat}$), 142.4 (C $_{quat}$), 143.3 (C $_{quat}$), 144.7 (C $_{quat}$), 160.4 (C $_{quat}$). MALDI-TOF MS: m/z 583.2 [M] $^+$. IR (ATR): ν 602 cm $^{-1}$ (m), 615 (s), 627 (m), 633 (m), 644 (s), 698 (s), 723 (s), 760 (m), 795 (s), 802 (s), 812 (s), 829 (m), 947 (m), 999 (w), 1018 (w), 1032 (s), 1057 (w), 1076 (w), 1111 (m), 1153 (w), 1177 (s), 1250 (s), 1296 (m), 1375 (m), 1439 (m), 1463 (m), 1501 (s), 1564 (w), 1568 (w), 1607 (w), 2833 (w), 2853 (w), 2922 (w), 2947 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 295 nm (46100), 367 (6000, sh). Anal. calcd. for C $_{34}$ H $_{33}$ NO $_2$ S $_3$

(583.8): C 69.95, H 5.70, N 2.40, S 16.48; Found: C 69.67, H 5.63, N 2.29, S 16.59.

4-(4-*n*-Hexylphenyl)-2,6-di-*p*-tolyl-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (**3b**)

According to the GP1 and after flash chromatography on silica gel (*n*-pentane with 0.5 % of triethylamine) compound **3b** (134 mg, 78 %) was obtained as an orange oil.

R_f (*n*-hexane): 0.14. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 1:1): δ 0.93 (t, $^3J = 7.0\text{ Hz}$, 3 H), 1.35-1.44 (m, 6 H), 1.70-1.76 (m, 2 H), 2.31 (s, 6 H), 2.72-2.76 (m, 2 H), 6.37 (s, 2 H), 7.11-7.14 (m, 4 H), 7.24-7.27 (m, 4 H), 7.38-7.43 (m, 4 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 1:1): δ 14.7 (CH $_3$), 21.4 (CH $_3$), 23.6 (CH $_2$), 30.0 (CH $_2$), 32.3 (CH $_2$), 32.7 (CH $_2$), 36.5 (CH $_2$), 102.6 (C $_{quat}$), 115.9 (CH), 125.6 (CH), 129.3 (CH), 130.5 (CH), 131.3 (CH), 138.5 (C $_{quat}$), 142.0 (C $_{quat}$), 142.6 (C $_{quat}$), 143.5 (C $_{quat}$), 145.0 (C $_{quat}$). MALDI-TOF MS: m/z 551.2 [M] $^+$. IR (ATR): ν 617 cm $^{-1}$ (w), 638 (w), 667 (w), 710 (w), 760 (m), 804 (s), 945 (w), 999 (w), 1018 (w), 1038 (w), 1057 (w), 1111 (w), 1121 (w), 1182 (w), 1217 (w), 1280 (m), 1310 (w), 1357 (m), 1435 (m), 1504 (s), 1537 (w), 1574 (w), 1611 (w), 2853 (w), 2924 (m), 2953 (w), 3022 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 294 nm (48200), 385 (5700). HRMS (ESI) calcd. for C $_{34}$ H $_{33}$ NS $_3$: 551.17751; Found: 551.17757.

4-(4-Hexylphenyl)-2,6-diphenyl-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3c)

According to the GPI and after flash chromatography on silica gel (*n*-hexane with 2 % of triethylamine) and after crystallization from ethanol/ethyl acetate (20:1) compound **3c** (227 mg, 87 %) was obtained as an orange powder.

R_f (*n*-hexane): 0.09, Mp 114 °C. $^1\text{H NMR}$ (300 MHz, acetone- d_6 /CS $_2$ 1:1): δ 0.96 (t, $^3J = 7.0$ Hz, 3 H), 1.33-1.50 (m, 6 H), 1.76 (p, $^3J = 7.6$ Hz, 2 H), 2.70-2.82 (m, 2 H), 6.36 (s, 2 H), 7.18-7.47 (m, 14 H). $^{13}\text{C NMR}$ (75 MHz, acetone- d_6 /CS $_2$ 1:4): δ 15.1 (CH $_3$), 23.8 (CH $_2$), 29.9 (CH $_2$), 32.3 (CH $_2$), 32.8 (CH $_2$), 36.6 (CH $_2$), 103.1 (C $_{\text{quat}}$), 115.9 (CH), 125.5 (CH), 128.2 (CH), 129.3 (CH), 129.5 (CH), 131.0 (CH), 134.1 (C $_{\text{quat}}$), 141.5 (C $_{\text{quat}}$), 141.9 (C $_{\text{quat}}$), 143.2 (C $_{\text{quat}}$), 144.4 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 523.2 [M] $^+$. IR (ATR): $\tilde{\nu}$ 673 cm $^{-1}$ (w), 685 (s), 712 (w), 748 (s), 810 (w), 822 (m), 889 (w), 945 (w), 1003 (w), 1072 (w), 1098 (w), 1192 (w), 1273 (w), 1368 (w), 1418 (w), 1425 (w), 1451 (m), 1489 (m), 1508 (w), 1528 (w), 1566 (w), 1597 (w), 2359 (w), 2851 (w), 2920 (w), 2053 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 290 nm (49500), 400 (5250). Anal. calcd. for C $_{32}$ H $_{29}$ NS $_3$ (522.8): C 73.38, H 5.58, N 2.67; Found: C 73.19, H 5.55, N 2.74.

2,6-Bis(4-chlorophenyl)-4-(4-hexylphenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3d)

According to the GPI and after flash chromatography on silica gel (*n*-hexane with 2 % of triethylamine) and after crystallization from *n*-hexane compound **3d** (231 mg, 78 %) was obtained as light red crystals.

R_f (*n*-hexane): 0.18, Mp 131 °C. $^1\text{H NMR}$ (500 MHz, CD $_2$ Cl $_2$): δ 0.91 (t, $^3J = 7.0$ Hz, 3 H), 1.30-1.41 (m, 6 H), 1.69 (p, $^3J = 7.5$ Hz, 2 H), 2.67-2.72 (m, 2 H), 6.31 (s, 2 H), 7.24-7.37 (m, 12 H). $^{13}\text{C NMR}$ (125 MHz, CD $_2$ Cl $_2$): δ 14.4 (CH $_3$), 23.2 (CH $_2$), 29.6 (CH $_2$), 31.9 (CH $_2$), 32.2 (CH $_2$), 36.2 (CH $_2$), 103.0 (C $_{\text{quat}}$), 116.2 (CH), 126.6 (CH), 129.0 (CH), 129.5 (CH), 130.9 (CH), 132.7 (C $_{\text{quat}}$), 133.7 (C $_{\text{quat}}$), 140.4 (C $_{\text{quat}}$), 141.2 (C $_{\text{quat}}$), 143.7 (C $_{\text{quat}}$), 144.8 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 591.1 [M] $^+$. IR (ATR): $\tilde{\nu}$ 615 cm $^{-1}$ (w), 627 (w), 644 (w), 669 (w), 694 (w), 714 (w), 731 (w), 766 (w), 806 (s), 814 (s), 820 (m), 943 (w), 1003 (w), 1059 (w), 1094 (m), 1117 (w), 1179 (w), 1192 (w), 1273 (w), 1371 (m), 1398 (w), 1431 (m), 1485 (m), 1508 (w), 1568 (m), 1888 (w), 2853 (w), 2924 (w), 2053 (w), 3030 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 296 nm (49550), 409 (6000). Anal. calcd. for C $_{32}$ H $_{27}$ Cl $_2$ NS $_3$ (591.7): C 64.85, H 4.59, N 2.36; Found: C 64.80, H 4.81, N 2.26.

Dimethyl 4,4'-(4-(4-hexylphenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazin-2,6-diyl)dibenzoate (3e)

According to the GPI and after flash chromatography on silica gel (*n*-hexane/ethyl acetate 10:1 with 2 % of triethylamine) and after crystallization from dichloromethane/*n*-hexane compound **3e** (257 mg, 80 %) was obtained as a red solid.

R_f (*n*-hexane/ethyl acetate 10:1): 0.28, Mp 199 °C. $^1\text{H NMR}$ (300 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 293$ K): δ 0.96 (t, $^3J = 7.0$ Hz, 3 H), 1.36-1.50 (m, 6 H), 1.69 (p, $^3J = 7.6$ Hz, 2 H), 2.74-2.81 (m, 2 H), 3.86 (s, 6 H), 6.51 (s, 2 H), 7.39-7.54 (m, 8 H), 7.89-7.96 (m, 4 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 14.9 (CH $_3$), 23.8 (CH $_2$), 30.2 (CH $_2$), 32.3 (CH $_2$), 32.8 (CH $_2$), 36.6 (CH $_2$), 52.3 (CH $_3$), 98.9 (C $_{\text{quat}}$), 117.4 (CH), 125.3 (C $_{\text{quat}}$), 126.0 (CH), 128.1 (CH), 130.0 (C $_{\text{quat}}$), 131.03 (CH), 131.4 (CH), 138.3 (C $_{\text{quat}}$), 143.9 (C $_{\text{quat}}$), 144.9 (C $_{\text{quat}}$), 146.2 (C $_{\text{quat}}$), 166.2 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 639.1 [M] $^+$. IR (ATR): $\tilde{\nu}$ 665 cm $^{-1}$ (w), 694 (m), 725 (w), 766 (s), 818 (m),

851 (w), 947 (w), 962 (w), 1001 (w), 1016 (w), 1063 (w), 1109 (s), 1186 (m), 1192 (w), 1244 (w), 1273 (s), 1317 (w), 1377 (w), 1406 (w), 1420 (w), 1431 (m), 1500 (w), 1539 (w), 1560 (w), 1572 (w), 1603 (m), 1722 (s), 2332 (w), 2361 (w), 2853 (w), 2924 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 309 nm (49600), 455 (8750). Anal. calcd. for C $_{36}$ H $_{33}$ NO $_4$ S $_3$ (639.9): C 67.58, H 5.20, N 2.19; Found: C 67.51, H 5.01, N 2.16.

4-(4-Hexylphenyl)-2,6-bis(4-(trifluoromethyl)phenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3f)

According to the GPI and after flash chromatography on silica gel (*n*-hexane with 2 % of triethylamine) and after crystallization from *n*-hexane/ethyl acetate compound **3f** (313 mg, 95 %) was obtained as fine coral red crystals.

R_f (*n*-hexane): 0.19, Mp 215 °C. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 0.95 (t, $^3J = 7.0$ Hz, 3 H), 1.37-1.49 (m, 6 H), 1.75 (p, $^3J = 7.6$ Hz, 2 H), 2.74-2.79 (m, 2 H), 6.54 (s, 2 H), 7.44 (q, $^3J = 8.4$ Hz, 4 H), 7.59 (dd, $J = 8.3$ Hz, 29.7 Hz, 8 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 14.7 (CH $_3$), 23.7 (CH $_2$), 30.1 (CH $_2$), 32.3 (CH $_2$), 32.7 (CH $_2$), 36.6 (CH $_2$), 105.4 (C $_{\text{quat}}$), 117.8 (CH), 125.1 (C $_{\text{quat}}$, $^1J_{\text{CF}} = 271.6$ Hz), 126.0 (CH), 126.9 (CH, $^3J_{\text{CF}} = 3.8$ Hz), 129.5 (CH), 129.9 (C $_{\text{quat}}$, $^2J_{\text{CF}} = 32.5$ Hz), 131.5 (CH), 137.9 (C $_{\text{quat}}$), 140.5 (C $_{\text{quat}}$), 141.6 (C $_{\text{quat}}$), 144.0 (C $_{\text{quat}}$), 145.3 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 659.1 [M] $^+$. IR (ATR): $\tilde{\nu}$ 610 cm $^{-1}$ (w), 638 (w), 652 (w), 691 (w), 714 (w), 735 (w), 773 (w), 822 (s), 837 (m), 949 (w), 1003 (w), 1013 (m), 1069 (s), 1109 (s), 1121 (m), 1163 (m), 1194 (w), 1246 (w), 1275 (w), 1290 (w), 1325 (s), 1379 (w), 1406 (w), 1435 (m), 1508 (m), 1566 (w), 1611 (w), 2855 (w), 2901 (w), 2926 (w), 2959 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 298 nm (62850), 431 (8500). Anal. calcd. for C $_{34}$ H $_{27}$ F $_6$ NS $_3$ (659.8): C 61.89, H 4.12, N 2.12; Found: C 61.70, H 4.21, N 2.06.

4,4'-(4-(4-Hexylphenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazin-2,6-diyl)dibenzonitrile (3g)

According to the GPI and after flash chromatography on silica gel (*n*-hexane/ethyl acetate 10:1) compound **3g** (198 mg, 69 %) was obtained as dark red crystals.

R_f (*n*-hexane/ethyl acetate 10:1): 0.11, Mp 175 °C. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 0.96 (t, $^3J = 7.1$ Hz, 3 H), 1.39-1.48 (m, 6 H), 1.75 (p, $^3J = 7.5$ Hz, 2 H), 2.75-2.78 (m, 2 H), 6.53 (s, 2 H), 7.39-7.46 (m, 4 H), 7.53 (d, $J = 8.3$ Hz, 4 H), 7.54-7.58 (d, $J = 8.4$ Hz, 4 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 14.8 (CH $_3$), 23.7 (CH $_2$), 30.1 (CH $_2$), 32.3 (CH $_2$), 32.7 (CH $_2$), 36.6 (CH $_2$), 111.8 (C $_{\text{quat}}$), 116.7 (CH), 118.0 (C $_{\text{quat}}$), 118.8 (C $_{\text{quat}}$), 126.0 (CH), 127.6 (CH), 131.5 (CH), 133.6 (CH), 138.1 (C $_{\text{quat}}$), 140.2 (C $_{\text{quat}}$), 141.8 (C $_{\text{quat}}$), 144.1 (C $_{\text{quat}}$), 146.4 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 573.1 [M] $^+$. IR (ATR): $\tilde{\nu}$ 617 cm $^{-1}$ (w), 632 (w), 719 (w), 760 (w), 773 (w), 814 (s), 835 (m), 950 (w), 1001 (w), 1065 (w), 1109 (w), 1179 (m), 1248 (w), 1277 (m), 1312 (w), 1381 (m), 1404 (s), 1427 (s), 1497 (s), 1507 (m), 1574 (m), 1599 (s), 2220 (m), 2853 (w), 2924 (w), 2953 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 310 nm (43600), 457 (7200). Anal. calcd. for C $_{34}$ H $_{27}$ N $_3$ S $_3$ (573.8): C 71.17, H 4.74, N 7.32; Found: C 70.94, H 4.73, N 7.06.

4-(4-Hexylphenyl)-2,6-bis(4-nitrophenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3h)

According to the GPI and after flash chromatography on silica gel (*n*-hexane/dichloromethane 10:3 with 2 % of triethylamine) and after crystallization from dichloromethane/*n*-hexane compound **3h** (241 mg, 78 %) was obtained as a dark violet powder.

R_f (*n*-hexane/dichloromethane 10:3): 0.21, Mp 192 °C. ^1H NMR (500 MHz, CD_2Cl_2): δ 0.92 (t, $^3J = 7.1$ Hz, 3 H), 1.32-1.43 (m, 6 H), 1.71 (p, $^3J = 7.5$ Hz, 2 H), 2.68-2.76 (m, 2 H), 6.43 (s, 2 H), 7.34 (d, $^3J = 8.3$ Hz, 2 H), 7.40 (d, $^3J = 8.4$ Hz, 2 H), 7.45-7.48 (m, 4 H), 8.10-8.13 (m, 4 H). ^{13}C NMR (125 MHz, CD_2Cl_2): δ 14.4 (CH_3), 23.2 (CH_2), 29.6 (CH_2), 31.9 (CH_2), 32.2 (CH_2), 36.2 (CH_2), 106.0 (C_{quat}), 117.6 (CH), 124.8 (CH), 125.4 (CH), 129.1 (CH), 131.2 (CH), 139.0 (C_{quat}), 139.9 (C_{quat}), 140.7 (C_{quat}), 144.3 (C_{quat}), 145.1 (C_{quat}), 147.0 (C_{quat}). MALDI-TOF MS: m/z 613.1 [$\text{M}]^+$. IR (ATR): $\tilde{\nu}$ 687 cm^{-1} (w), 746 (m), 822 (w), 833 (w), 849 (m), 1069 (w), 1107 (m), 1188 (w), 1215 (w), 1248 (w), 1273 (w), 1329 (s), 1385 (w), 1404 (w), 1427 (m), 1489 (m), 1514 (m), 1572 (w), 1589 (m), 2855 (w), 2926 (w). UV/Vis (CH_2Cl_2): λ_{max} (ϵ) 254 nm (31300), 332 (46000), 524 (17900). Anal. calcd. for $\text{C}_{32}\text{H}_{27}\text{N}_3\text{O}_4\text{S}_3$ (613.8): C 62.62, H 4.43, N 6.85; Found: C 62.43, H 4.44, N 6.57.

4-(4-Hexylphenyl)-2,6-bis(3-nitrophenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3i)

According to the GP1 and after flash chromatography on silica gel (*n*-hexane/dichloromethane 10:3 with 2 % of triethylamine) compound **3i** (236 mg, 77 %) was obtained as a chestnut brown solid.

R_f (*n*-hexane/dichloromethane 10:3): 0.22, Mp 199 °C. ^1H NMR (600 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 0.93 (t, $^3J = 7.1$ Hz, 3 H), 1.37-1.48 (m, 6 H), 1.75 (p, $^3J = 7.6$ Hz, 2 H), 2.74-2.78 (m, 2 H), 6.63 (s, 2 H), 7.45 (s, 4 H), 7.62 (t, $J = 8.0$ Hz, 2 H), 7.80 (ddd, $J = 0.9$ Hz, 1.8 Hz, 7.8 Hz, 2 H), 8.08 (ddd, $J = 0.9$ Hz, 1.8 Hz, 8.2 Hz, 2 H), 8.17 (t, $J = 2.0$ Hz, 2 H). ^{13}C NMR (150 MHz, acetone- d_6 /CS $_2$ 1:1, $T = 313$ K): δ 14.7 (CH_3), 23.6 (CH_2), 30.1 (CH_2), 32.3 (CH_2), 32.7 (CH_2), 36.5 (CH_2), 105.7 (C_{quat}), 118.2 (CH), 120.0 (CH), 122.9 (CH), 129.4 (CH), 131.3 (CH), 131.45 (CH), 131.53 (CH), 136.0 (C_{quat}), 139.7 (C_{quat}), 141.6 (C_{quat}), 144.1 (C_{quat}), 145.4 (C_{quat}), 149.8 (C_{quat}). MALDI-TOF MS: m/z 613.1 [$\text{M}]^+$. IR (ATR): $\tilde{\nu}$ 671 cm^{-1} (m), 694 (w), 712 (w), 727 (m), 732 (m), 762 (w), 797 (m), 835 (w), 860 (w), 897 (w), 997 (w), 1067 (w), 1073 (w), 1101 (w), 1202 (w), 1281 (m), 1306 (w), 1352 (s), 1385 (w), 1427 (m), 1477 (m), 1510 (m), 1524 (s), 1578 (w), 1614 (w), 2335 (w), 2851 (w), 2922 (w), 2951 (w), 3084 (w). UV/Vis (CH_2Cl_2): λ_{max} (ϵ) 286 nm (40100), 440 (4800). Anal. calcd. for $\text{C}_{32}\text{H}_{27}\text{N}_3\text{O}_4\text{S}_3$ (613.8): C 62.62, H 4.43, N 6.85, S 15.67; Found: C 62.40, H 4.51, N 6.72, S 15.44.

4-(4-Hexylphenyl)-2,6-bis(2-nitrophenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3j)

According to the GP1 and after flash chromatography on silica gel (*n*-hexane/dichloromethane 10:3 with 2 % of triethylamine) compound **3j** (165 mg, 54 %) was obtained as a chestnut brown oil.

R_f (*n*-hexane/dichloromethane 10:3): 0.35, ^1H NMR (600 MHz, acetone- d_6 /CS $_2$ 1:1): δ 0.91 (t, $^3J = 7.1$ Hz, 3 H), 1.31-1.42 (m, 6 H), 1.69 (p, $^3J = 7.5$ Hz, 2 H), 2.68-2.72 (m, 2 H), 6.17 (s, 2 H), 7.29-7.32 (m, 2 H), 7.35-7.38 (m, 2 H), 7.52 (dd, $J = 1.3$ Hz, 7.8 Hz, 2 H), 7.57 (td, $J = 1.4$ Hz, 7.6 Hz, 2 H), 7.66 (td, $J = 1.3$ Hz, 7.6 Hz, 2 H), 7.77 (dd, $J = 1.2$ Hz, 8.0 Hz, 2 H). ^{13}C NMR (150 MHz, acetone- d_6 /CS $_2$ 1:1): δ 14.8 (CH_3), 23.6 (CH_2), 30.0 (CH_2), 32.3 (CH_2), 32.7 (CH_2), 36.4 (CH_2), 106.0 (C_{quat}), 120.4 (CH), 124.8 (CH), 127.6 (C_{quat}), 128.7 (CH), 130.2 (CH), 131.2 (CH), 132.3 (CH), 133.1 (CH), 135.6 (C_{quat}), 141.3 (C_{quat}), 143.6 (C_{quat}), 144.8 (C_{quat}), 149.8 (C_{quat}). MALDI-TOF MS: m/z 613.2 [$\text{M}]^+$. IR (ATR): $\tilde{\nu}$ 608 cm^{-1} (w), 652 (m), 681 (w), 696 (w), 710 (s), 721 (m), 748 (s), 775 (s), 824 (m), 831 (w), 837 (w), 853 (m), 949 (w), 988 (w), 999 (w), 1019

(w), 1063 (w), 1113 (w), 1142 (w), 1163 (w), 1188 (w), 1279 (m), 1302 (w), 1358 (s), 1379 (m), 1425 (m), 1481 (m), 1508 (m), 1526 (s), 1566 (w), 1576 (w), 1605 (w), 2855 (w), 2928 (w), 2955 (w). UV/Vis (CH_2Cl_2): λ_{max} (ϵ) 264 nm (30250), 452 (2850). Anal. calcd. for $\text{C}_{32}\text{H}_{27}\text{N}_3\text{O}_4\text{S}_3$ (613.8): C 62.62, H 4.43, N 6.85, S 15.67; Found: C 62.24, H 4.61, N 6.57, S 15.27.

4-(4-Hexylphenyl)-2,6-di(pyridin-3-yl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazine (3k)

According to the GP1 and after flash chromatography on silica gel (*n*-hexane/ethyl acetate 4:1 with 2 % of triethylamine) compound **3k** (164 mg, 63 %) was obtained as an intense orange oil.

R_f (*n*-hexane/ethyl acetate 4:1): 0.15. ^1H NMR (300 MHz, acetone- d_6 /CS $_2$ 1:1): δ 0.95 (t, $^3J = 7.0$ Hz, 3 H), 1.33-1.50 (m, 6 H), 1.74 (p, $^3J = 7.5$ Hz, 2 H), 2.71-2.79 (m, 2 H), 6.45 (s, 2 H), 7.27 (ddd, $J = 0.8$ Hz, 4.8 Hz, 8.0 Hz, 2 H), 7.41 (s, 4 H), 7.67 (ddd, $J = 1.6$ Hz, 2.4 Hz, 8.0 Hz, 2 H), 8.41 (dd, $J = 1.5$ Hz, 4.7 Hz, 2 H), 8.59 (dd, $J = 0.7$ Hz, 2.4 Hz, 2 H). ^{13}C NMR (75 MHz, acetone- d_6 /CS $_2$ 1:1): δ 14.9 (CH_3), 23.7 (CH_2), 30.2 (CH_2), 32.3 (CH_2), 32.7 (CH_2), 36.6 (CH_2), 104.5 (C_{quat}), 117.2 (CH), 124.3 (CH), 129.3 (CH), 130.0 (C_{quat}), 131.3 (CH), 132.2 (CH), 138.6 (C_{quat}), 141.4 (C_{quat}), 143.7 (C_{quat}), 145.1 (C_{quat}), 146.6 (CH), 149.4 (CH). MALDI-TOF MS: m/z 525.2 [$\text{M}]^+$. IR (ATR): $\tilde{\nu}$ 619 cm^{-1} (w), 660 (w), 679 (w), 704 (s), 725 (w), 760 (w), 799 (s), 827 (w), 837 (w), 941 (m), 995 (w), 1022 (w), 1063 (w), 1101 (w), 1180 (w), 1234 (w), 1281 (m), 1339 (w), 1379 (m), 1418 (m), 1429 (s), 1477 (s), 1508 (m), 1539 (w), 1574 (m), 1585 (w), 2853 (w), 2924 (m), 2953 (w). UV/Vis (CH_2Cl_2): λ_{max} (ϵ) 292 nm (32450), 414 (3900). Anal. calcd. for $\text{C}_{30}\text{H}_{27}\text{N}_3\text{S}_3$ (525.7): C 68.53, H 5.18, N 7.99, S 18.30; Found: C 68.13, H 5.26, N 7.75, S 18.51.

3,3'-(4-(4-Hexylphenyl)-4H-dithieno[2,3-b:3',2'-e][1,4]thiazin-2,6-diyl)bis(10-hexyl-10H-phenothiazine) (3l)

According to the GP1 and after flash chromatography on silica gel (*n*-hexane with 2 % of triethylamine) and trituration with *n*-hexane under ultrasound compound **3l** (441 mg, 94 %) was obtained as an orange solid.

R_f (*n*-hexane): 0.32, Mp 95 °C. ^1H NMR (600 MHz, acetone- d_6 /CS $_2$ 1:2): δ 0.90 (t, $^3J = 7.1$ Hz, 6 H), 0.97 (t, $^3J = 7.1$ Hz, 3 H), 1.30-1.44 (m, 18 H), 1.75-1.82 (m, 6 H), 2.73-2.79 (m, 2 H), 3.85-3.89 (m, 4 H), 6.25 (s, 2 H), 6.83 (d, $J = 8.6$ Hz, 2 H), 6.87-6.91 (m, 4 H), 7.04 (dd, $J = 1.5$ Hz, 6.7 Hz, 2 H), 7.06 (d, $J = 2.2$ Hz, 2 H), 7.10-7.15 (m, 4 H), 7.34-7.38 (m, 2 H), 7.39-7.43 (m, 2 H). ^{13}C NMR (150 MHz, acetone- d_6 /CS $_2$ 1:2): δ 14.8 (CH_3), 14.9 (CH_3), 23.7 (CH_2), 23.8 (CH_2), 27.5 (CH_2), 27.7 (CH_2), 30.2 (CH_2), 32.4 (CH_2), 32.5 (CH_2), 32.8 (CH_2), 36.6 (CH_2), 48.1 (CH_2), 102.2 (C_{quat}), 115.5 (CH), 116.4 (2 x CH), 123.4 (CH), 124.1 (CH), 124.78 (C_{quat}), 124.81 (C_{quat}), 126.3 (C_{quat}), 128.0 (CH), 128.2 (CH), 128.8 (CH), 129.3 (CH), 131.2 (CH), 141.4 (C_{quat}), 141.7 (C_{quat}), 143.4 (C_{quat}), 144.8 (C_{quat}), 145.36 (C_{quat}), 145.38 (C_{quat}). MALDI-TOF MS: m/z 933.3 [$\text{M}]^+$. IR (ATR): $\tilde{\nu}$ 613 cm^{-1} (w), 677 (w), 680 (w), 710 (w), 723 (w), 743 (s), 752 (m), 783 (w), 804 (s), 822 (w), 864 (w), 879 (w), 930 (w), 997 (w), 1042 (w), 1059 (w), 1109 (w), 1142 (w), 1167 (w), 1188 (w), 1246 (m), 1285 (w), 1333 (w), 1362 (m), 1373 (m), 1404 (w), 1439 (m), 1466 (s), 1493 (m), 1539 (w), 1572 (w), 1599 (w), 2853 (w), 2868 (w), 2926 (w), 2951 (w). UV/Vis (CH_2Cl_2): λ_{max} (ϵ) 239 nm (sh, 27900), 267 (sh, 30250), 291 (36250), 349 (14300), 422 (sh, 8500). Anal. calcd. for $\text{C}_{56}\text{H}_{59}\text{N}_3\text{S}_5$ (933.4): C 71.98, H 6.36, N 4.50, S 17.16; Found: C 71.83, H 6.22, N 4.42, S 17.31.

General Procedure for the Lithiation-Negishi Coupling Synthesis of 2-(Hetero)aryl Dithienothiazines 6 (GP2). Dry THF (3 mL), dry TMEDA (0.08 mL, 0.6 mmol) and 4-(4-*n*-hexylphenyl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (**1**) (186 mg, 0.50 mmol) were placed in a 50 mL Schlenk flask under nitrogen. Then, the solution was cooled to -78 °C (dry ice/acetone) and 1.6 M *n*-BuLi in hexane (0.32 mL, 0.5 mmol) and *n*-hexane Acrosil® (0.66 mL) were added successively and dropwise to the solution and the reaction mixture was stirred at -78 °C for 2 h. In the meantime a solution of freshly zinc bromide (150 mg, 0.67 mmol) in dry THF (1.0 mL) was prepared which was added dropwise by syringe to the reaction mixture at -78 °C. After stirring at -78 °C for 30 min palladium

tetrakis(triphenylphosphane) (23 mg, 4 mol%) and 1.0 equiv of the (hetero)aryl iodide **2** (0.5 mmol) were successively added to the reaction mixture (for experimental details see Table 5). The reaction mixture was allowed to come to room temp upon stirring for 12-14 h. Then, a mixture of deionized water (20 mL) and dichloromethane (20 mL) was added, the organic layer was separated and the aqueous phase was extracted with dichloromethane (3 x 20 mL). The combined organic layers were dried with anhydrous magnesium sulfate and the solvents were removed in vacuo. The crude product was purified by flash chromatography on silica gel (eluent containing 2 % of triethylamine) to give the pure products **6**. Further purification could be achieved by fractional crystallization.

Table 5. Experimental details of the lithiation-Negishi coupling synthesis of 2-(hetero)aryl dithienothiazines **6**.

Entry	(Hetero)aryl iodide 2 [mg] (mmol)	2-(Hetero)aryl dithienothiazines 6 [mg] (%)
1	125 (0.50) of 1-iodo-4-nitrobenzene (2h)	108 (44) of 6a
2	105 (0.50) of 2-iodothiophene (2m)	138 (61) of 6b
3	103 (0.50) of 4-iodopyridine (2n)	105 (47) of 6c

4-(4-Hexylphenyl)-2-(4-nitrophenyl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (6a**)**

According to the GP2 and after twofold flash chromatography on silica gel (*n*-hexane/dichloromethane with 2 % of triethylamine and *n*-hexane/toluene 10:1 with 1 % of triethylamine) compound **6a** (108 mg, 44 %) was obtained as a dark violet oil.

R_f (*n*-hexane/dichloromethane 10:1): 0.09. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 2:1): δ 0.94 (t, $^3J = 7.1$ Hz, 3 H), 1.36-1.46 (m, 6 H), 1.72 (p, $^3J = 7.5$ Hz, 2 H), 2.72-2.76 (m, 2 H), 6.11 (d, $J = 5.5$ Hz, 1 H), 6.64 (s, 1 H), 7.19 (d, $J = 5.5$ Hz, 1 H), 7.36-7.39 (m, 2 H), 7.41-7.44 (m, 2 H), 7.61-7.65 (m, 2 H), 8.15-8.19 (m, 2 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 2:1): δ 14.7 (CH $_3$), 23.6 (CH $_2$), 30.1 (CH $_2$), 32.4 (CH $_2$), 32.7 (CH $_2$), 36.5 (CH $_2$), 102.7 (C $_{\text{quat}}$), 107.9 (C $_{\text{quat}}$), 118.5 (CH), 120.7 (CH), 125.0 (CH), 125.2 (CH), 125.9 (CH), 129.3 (CH), 131.4 (CH), 139.2 (C $_{\text{quat}}$), 140.3 (C $_{\text{quat}}$), 141.7 (C $_{\text{quat}}$), 143.8 (C $_{\text{quat}}$), 144.4 (C $_{\text{quat}}$), 146.1 (C $_{\text{quat}}$), 147.4 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 492.1 [M] $^+$. IR (ATR): $\tilde{\nu}$ 611 cm $^{-1}$ (m), 629 (m), 687 (s), 696 (s), 714 (m), 748 (s), 799 (s), 818 (s), 847 (s), 948 (w), 999 (m), 1017 (s), 1061 (m), 1096 (s), 1107 (s), 1175 (w), 1203 (w), 1260 (s), 1329 (s), 1380 (m), 1329 (s), 1380 (m), 1402 (m), 1427 (m), 1491 (m), 1506 (s), 1562 (w), 1589 (m), 2853 (w), 2922 (w), 2957 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 250 nm (25550), 328 (22850), 511 (7300). HRMS (ESI) calcd. for C $_{26}$ H $_{24}$ N $_2$ O $_2$ S $_3$: 492.09999, Found: 492.09937.

4-(4-Hexylphenyl)-2-(thiophen-2-yl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazine (6b**)**

According to the GP2 and after flash chromatography on silica gel (*n*-hexane with 2 % of triethylamine) compound **6b** (138 mg, 61 %) was obtained as a pale orange oil.

R_f (*n*-hexane): 0.19. $^1\text{H NMR}$ (600 MHz, acetone- d_6 /CS $_2$ 5:1): δ 0.93 (t, $^3J = 7.0$ Hz, 3 H), 1.36-1.44 (m, 6 H), 1.72 (p, $^3J = 7.6$ Hz, 2 H), 2.71-2.75 (m, 2 H), 6.10 (d, $J = 5.5$ Hz, 1 H), 6.20 (s, 1 H), 6.98 (dd, $J = 3.6$ Hz, 5.1 Hz, 1 H), 7.04 (dd, $J = 1.1$ Hz, 3.6 Hz, 1 H), 7.15 (d, $J = 5.5$ Hz, 1 H), 7.29 (dd, $J = 1.1$ Hz, 5.1 Hz, 1 H), 7.32-7.35 (m, 2 H), 7.39-7.41 (m, 2 H). $^{13}\text{C NMR}$ (150 MHz, acetone- d_6 /CS $_2$ 2:1): δ 14.7 (CH $_3$), 23.6 (CH $_2$), 29.9 (CH $_2$), 32.4 (CH $_2$), 32.7 (CH $_2$), 36.5 (CH $_2$), 102.7 (C $_{\text{quat}}$), 103.2 (C $_{\text{quat}}$), 116.6 (CH), 120.6 (CH), 124.4 (CH), 124.6 (CH), 125.7 (CH), 128.8 (CH), 129.3 (CH), 131.2 (CH), 135.6 (C $_{\text{quat}}$), 137.2

(C $_{\text{quat}}$), 141.8 (C $_{\text{quat}}$), 143.6 (C $_{\text{quat}}$), 144.6 (C $_{\text{quat}}$), 145.0 (C $_{\text{quat}}$). MALDI-TOF MS: m/z 452.9 [M] $^+$. IR (ATR): $\tilde{\nu}$ 627 cm $^{-1}$ (w), 642 (w), 669 (w), 692 (s), 758 (w), 810 (m), 833 (m), 845 (m), 883 (w), 905 (w), 997 (m), 1016 (w), 1043 (w), 1059 (w), 1078 (w), 1098 (w), 1180 (w), 1225 (w), 1279 (m), 1333 (w), 1350 (w), 1377 (m), 1406 (m), 1459 (m), 1508 (s), 1537 (w), 1570 (w), 1611 (w), 2853 (w), 2924 (m), 2953 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 237 nm (sh, 12400), 300 (12700), 355 (sh, 2950), 420 (sh, 1900). HRMS (ESI) calcd. for C $_{24}$ H $_{23}$ NS $_4$: 453.07133, Found: 453.07097.

4-(4-Hexylphenyl)-2-(pyridin-4-yl)-4*H*-dithieno[2,3-*b*:3',2'-*e*][1,4]thiazin (6c**)**

According to the GP2 and after flash chromatography on silica gel (*n*-hexane/ethyl acetate 5:1 with 2 % of triethylamine) compound **6c** (105 mg, 47 %) was obtained as a red oil.

R_f (*n*-hexane/ethyl acetate 5:1): 0.19. $^1\text{H NMR}$ (300 MHz, acetone- d_6 /CS $_2$ 5:1): δ 0.96 (t, $^3J = 7.1$ Hz, 3 H), 1.34-1.47 (m, 6 H), 1.73 (p, $^3J = 7.6$ Hz, 2 H), 2.70-2.77 (m, 2 H), 6.10 (d, $J = 5.5$ Hz, 1 H), 6.60 (s, 1 H), 7.15 (d, $J = 5.5$ Hz, 1 H), 7.24-7.27 (m, 2 H), 7.32-7.37 (m, 2 H), 7.39-7.43 (m, 2 H), 8.43-8.46 (m, 2 H). $^{13}\text{C NMR}$ (75 MHz, acetone- d_6 /CS $_2$ 2:1): δ 14.8 (CH $_3$), 23.7 (CH $_2$), 30.0 (CH $_2$), 32.4 (CH $_2$), 32.7 (CH $_2$), 36.5 (CH $_2$), 102.8 (C $_{\text{quat}}$), 106.8 (C $_{\text{quat}}$), 118.0 (CH), 119.2 (CH), 120.6 (CH), 124.8 (CH), 129.3 (CH), 131.3 (CH), 138.8 (C $_{\text{quat}}$), 140.7 (C $_{\text{quat}}$), 141.7 (C $_{\text{quat}}$), 143.7 (C $_{\text{quat}}$), 144.3 (C $_{\text{quat}}$), 145.7 (C $_{\text{quat}}$), 151.2 (CH). MALDI-TOF MS: m/z 448.2 [M] $^+$. IR (ATR): $\tilde{\nu}$ 627 cm $^{-1}$ (w), 644 (w), 656 (m), 665 (w), 696 (m), 806 (s), 839 (m), 961 (w), 991 (m), 1000 (m), 1016 (w), 1061 (w), 1098 (w), 1117 (w), 1196 (w), 1217 (w), 1279 (m), 1325 (w), 1377 (m), 1404 (s), 1431 (s), 1458 (w), 1491 (s), 1508 (m), 1526 (m), 1545 (w), 1566 (m), 1591 (s), 2853 (w), 2924 (m), 2951 (w), 3028 (w). UV/Vis (CH $_2$ Cl $_2$): λ_{max} (ϵ) 238 nm (sh, 9750), 292 (13250), 367 (1900), 449 (2150). Anal. calcd. for C $_{25}$ H $_{24}$ N $_2$ S $_3$ · 1/3 CH $_3$ CO $_2$ C $_2$ H $_5$ (448.7 + 29.4): C 66.17, H 5.62, N 5.86, S 20.12, Found: C 66.23, H 5.44, N 5.84, S 20.48.

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Notes and references

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‡ Electronic Supplementary Information (ESI) available: ¹H and ¹³C NMR spectra, UV/Vis and fluorescence spectra of compounds **5**, **6**, **7**, and **8**; computed xyz-coordinates of the S₀ state of the pyrazoles **5c**, **6d**, and **8b**, computed UV/Vis spectra of TD-DFT calculated structures of **5c**, **6d**, and **8b**, computed xyz-coordinates of the S₁ state of pyrazole **6d**. See DOI: 10.1039/b000000x/

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