

REVIEW

View Article Online
View Journal | View Issue



Cite this: *Ind. Chem. Mater.*, 2025, 3, 509

Ionic liquids: a pitocin for next-generation electronic information materials?

Mengyue Li,^{†ab} Bin He,^{†ab} Yangyang Jiang,^c Ruirui Wang,^{ab} Cunliang Gan,^c Fengqi Ji,^c Yao Li^c and Ruixia Liu  ^{*ab}

Electronic information materials (EIMs) are key enablers for building a smart society. As the material carriers of next-generation information technology, the development of EIMs is increasingly constrained by the challenges of manufacturing precision, heterogeneous integration reliability, and circular economy compatibility. As traditional approaches struggle to meet the demands for nanoscale machining, low power consumption, structural flexibility, and environmental compatibility, there is an urgent need for disruptive materials and methodologies. Ionic liquids (ILs), with their unique combination of tunable molecular structures, negligible volatility, broad electrochemical windows, and strong solvation capabilities, offer a promising route to address these bottlenecks. As dynamic reaction media, ILs precisely regulate the nucleation kinetics and interfacial behaviours of zero dimension (0D) quantum dots, one dimension (1D) nanowires, and two dimension (2D) semiconductors through their unique solvation environments, yielding advanced materials with next-generation EIMs. Leveraging hydrogen bonding and ion-exchange interactions, ILs enable selective extraction and recycling of critical electronic chemicals (e.g., rare earth elements, conductive polymers), offering greener alternatives to conventional solvent-based processes. In field-effect transistors and flexible electronics, ILs improve charge transport efficiency, reduce operating voltages, and enhance interfacial stability, while their compatibility with heterogeneous integration addresses reliability challenges in scalable manufacturing. This review systematically examines ILs roles in advancing EIMs and proposes design principles for their targeted application, highlighting their potential to drive sustainable innovation in electronic materials science.

Keywords: Ionic liquids; Electronic information materials; Separation and purification; Electronic devices.

Received 9th June 2025,
Accepted 17th July 2025

DOI: 10.1039/d5im00098j

rscl.li/icm

1 Introduction

The evolution of industrial development has been characterized by a series of revolutionary breakthroughs. Commencing with mechanized production and progressing through the ages of electrification and mass automation, these paradigm shifts have profoundly redefined productivity paradigms, societal frameworks, and the global economic landscape. In the current era, the synergistic integration of digital technologies, intelligent systems, and cyber-physical interfaces is catalyzing a new phase of technological convergence. Transcending

conventional efficiency and automation, the modern industrial paradigm advocates for a symbiotic human-machine coevolution, with heightened emphasis on sustainability, systemic resilience, and corporate social responsibility. This

^a Beijing Key Laboratory of Ionic Liquids Clean Process, CAS Key Laboratory of Green Process and Engineering, Institute of Process Engineering, Innovation Academy for Green Manufacture, CAS, Beijing 100190, P. R. China.
E-mail: rxliu@ipe.ac.cn

^b School of Chemistry and Chemical Engineering, University of Chinese Academy of Sciences, 100049, Beijing, China

^c Sinopec Nanjing Chemical Industrial Co., Ltd., China

[†] These authors contributed equally to this work and should be considered co-first authors.



Mengyue Li

Mengyue Li received her bachelor's degree from Nankai University in 2016. Then she earned her Master's degree in Materials Science and Engineering from North Minzu University in 2019. Now, she works in the Institute of Process Engineering of Chinese Academy of Sciences as research assistant. Her present research focuses on the design of functional ionic liquids and their application in catalysis and electronic chemicals.



paradigm shift represents not merely an intensification of digital transformation, but rather a fundamental reorientation toward innovations that harmonize cutting-edge technological sophistication with human-centric design principles.¹ Within this broader context of technological transformation, the electronic chemicals (ECs) industry is assuming an increasingly pivotal role.^{2,3} As specialized fine chemicals essential to the modern electronics industry, ECs are foundational to the fabrication of integrated circuits, flat-panel displays, photovoltaic devices, and a variety of emerging technologies. Functioning as a critical interface between fundamental scientific inquiry and industrial-scale application (Fig. 1), ECs not only enable but also accelerate material and device innovation.^{4,5}

However, the accelerated advancement of electronic information technologies, propelling by 5G communications, artificial intelligence (AI), cloud computing, the Internet of Things, and smart devices, has imposed increasingly stringent requirements on the properties, performance, and reliability of next-generation electronic materials. Although considerable progress has been achieved, several critical challenges persist in this field. First, the dual imperatives of material innovation and environmental sustainability have become paramount. There is an urgent need to develop novel functional materials that not only exhibit superior performance but also demonstrate eco-compatibility, aiming to replace conventional silicon-based systems. Second, ultra-high chemical purity (*e.g.*, $\geq 99.9999\%$ for semiconductor-grade chemicals) has emerged as an essential criterion, as even trace impurities can significantly degrade device performance and yield. Third, the escalating complexity and integration density of modern electronic devices (*e.g.*, 3D Integrated Circuit, heterogeneous integration packages) necessitate highly tailored chemical formulations with exceptional compatibility, process-specific functionality, and long-term operational stability. These multifaceted demands

have exposed the limitations of traditional material systems, driving the exploration of adaptable, tunable, and multifunctional chemical platforms (*e.g.*, ILs, MOFs, and organic-inorganic hybrids) to meet next-generation technological requirements.

In this context, ILs, a class of thermally stable, non-volatile, and structurally tunable solvents composed entirely of ions, have emerged as promising alternatives. Their inherent properties, including exceptional electrochemical stability,⁶ low volatility,⁷ superb thermal stability,⁸ and high level of customizability,^{9,10} make them particularly attractive for advanced material processing. Beyond bulk properties, ILs can create unique microenvironments *via* specific solute-solvent interactions, such as dipolar interactions, charge delocalization, hydrogen bonding, and Lewis acid/base behaviour, that enable precise control over reaction kinetics and interfacial phenomena.^{11–13} These capabilities have already been widely exploited in other domains of materials science, including catalysis, nanomaterial synthesis, battery electrolytes, and separation science.¹⁴

Despite their success in adjacent fields, the application of ILs in EIMs remains relatively nascent. To date, few systematic studies have addressed how ILs can be strategically leveraged to meet the evolving demands of the ECs industry. This represents both a critical research gap and a significant opportunity. In this review, we aim to bridge this gap by critically examining the physicochemical features of ILs in the context of ECs design and exploring their functional roles in three core areas, EIMs, separation and purification, electronic devices. By analyzing representative case studies and distilling the underlying mechanisms by which ILs exert influence, we propose conceptual frameworks to guide the rational design of IL-assisted ECs. We believe that a deeper understanding of IL-material interactions will facilitate the development of next-generation EIMs with superior performance, precision, and sustainability.



Bin He

Bin He received his MSc in Institute of Process Engineering, Chinese Academy of Sciences in 2019. And then, he earned his PhD degree at Institute of Beijing Technology in 2023. Now, he is a postdoctoral researcher and “Special Research Assistant” at the Institute of Process Engineering. His present research focuses on the development high-powered catalytic materials and design, synthesis and application of functionalized ionic liquids.



Ruixia Liu

Ruixia Liu received her BSc degree from Qufu Normal University in 2005 and PhD degree from the Changchun Institute of Applied Chemistry, Chinese Academy of Sciences in 2010. She works in the Institute of Process Engineering of Chinese Academy of Sciences (2014–now) as the group leader of “Green Catalysis”. She has won the “Introduction of one hundred talents” in 2014 and the “National Science Fund for Excellent Young Scholars” in 2022. Prof. Liu mainly engages in the design of catalysts, and technology for synthesis of oxygen-containing high-value chemicals and electronic chemicals.



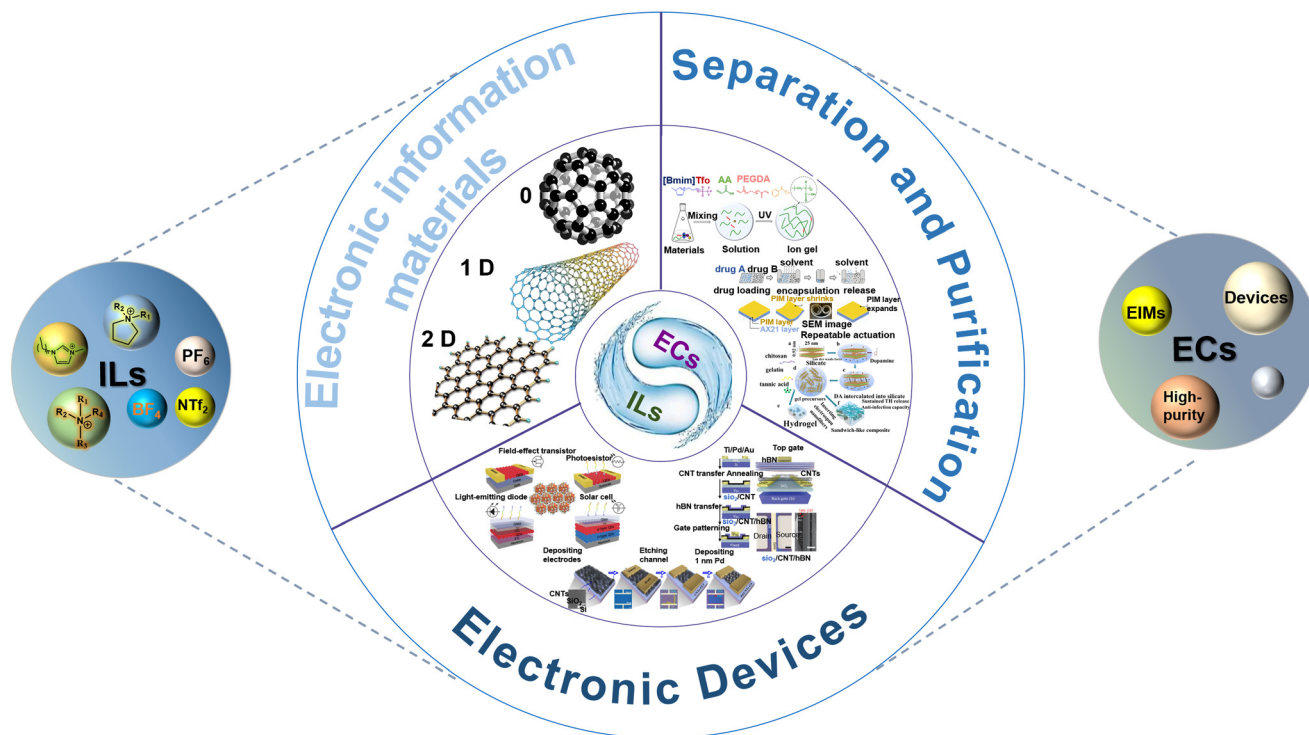


Fig. 1 Dimensional control mechanisms of ILs in ECs.

2 Properties of ILs in EIMs

2.1 Design guidelines for EIMs

Contemporary industrial innovation is undergoing a strategic transition from single technology breakthroughs to a systemic convergence paradigm, which is particularly noticeable in the field of EIMs. In order to break through the high-performance computing demands of von Neumann architectures, material design must go beyond the physical limits of silicon-based systems to address data bottlenecks, high energy consumption and latency challenges with multi-dimensional co-innovation.^{15–17} In order to achieve this objective, it is necessary to enhance the efficiency-orientation of first-generation EIMs, thereby transitioning to a systemic design philosophy that integrates quantum modulation, dynamic interface engineering and sustainable manufacturing.^{18,19} In this section, the design guidelines for EIMs under the technology convergence paradigm are systematically elaborated, based on the requirements of this paradigm shift. These guidelines are derived from three dimensions of electronic structure design, carrier transport optimisation and interfacial physicochemical engineering, based on energy band theory, transport physics and interface science. The central role of ILs as a tool for dynamical regulation is demonstrated.

Energy band theory, a cornerstone of condensed matter physics, reveals the microscopic mechanisms underlying the electrical conductivity, optical properties and magnetic properties of materials. Its core lies in the distribution laws of electrons in the allowed and forbidden bands (*e.g.*, Fermi energy level positions, band gap widths, carrier effective

masses).^{20,21} Dynamic fine-tuning of the energy band structure has become a key challenge in order to achieve precise design of macroscopic physical properties of electronic information materials. The development of ionic liquid gating (ILG) technology, based on the unique electrochemical properties of ILs, is one of the frontiers for achieving “*in situ* editing” of energy bands.²² In 2007, Frisbie group's reported the first double-dielectric layer (EDL) using an ILs as the gate dielectric, with a mobility increase of up to $3 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ for a rubbery conventional SiO_2 gate.²³ The basic principle of ionic liquid gating technology stems from the EDL, *i.e.* the formation of a EDL structure between the ILs and the surface of the material (Fig. 2a). In this structure, the EDL effect formed by the cations and anions of the IL is capable of inducing carrier density changes up to the order of 10^{14} cm^{-2} at the material surface through the construction of a localised electric field modulated by the energy band structure. In contrast to conventional solid oxide gate dielectrics, ILs precisely control the electronic states of materials by dynamically tuning the electric field distribution, especially in two-dimensional materials, where metal–insulator transitions can be dynamically induced or tunable carrier types can be achieved by ionic liquid gating.²⁴

In the development of high-performance functional devices, the energy levels of molecules with quantum interference effects can be tuned to an anti-resonance state by gate electrode potentials gated by ILs.¹⁸ This allows direct observation of sharp destructive quantum interference (DQI) properties due to changes in the relative positions of molecular energy levels. For example, the DQI effect of mesophenyl molecules with



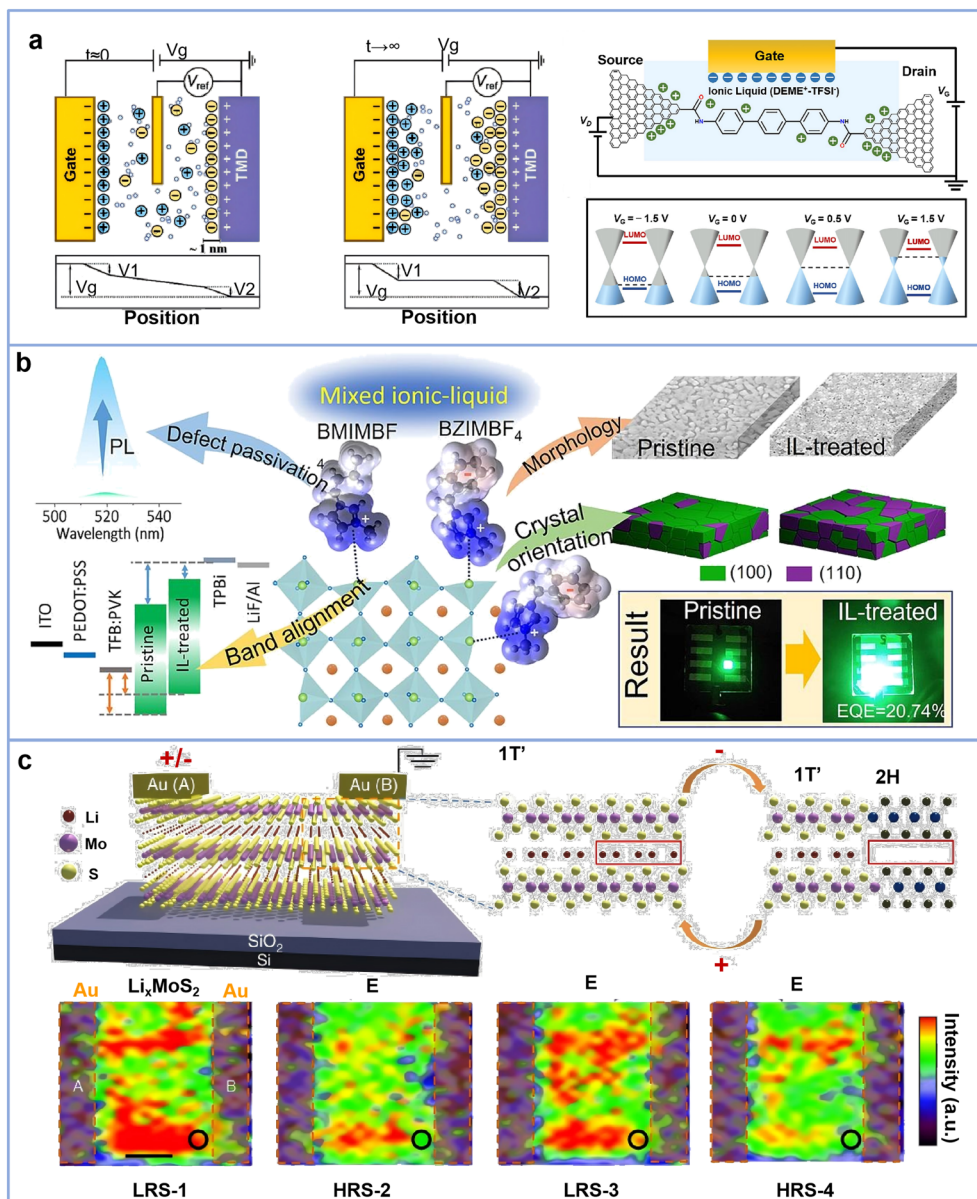


Fig. 2 Design principles for electronic materials under technology integration. (a) ILG based on band engineering;²⁸ (b) green perovskite LED via carrier strategy;²⁹ (c) hetero-structured MoTe_2 for hydrogen evolution reaction enabled by interface design.³⁰ Copyright 2021, MDPI; 2024, Elsevier and 2024, ACS.

dihydrobenzo[*b*]thiophene as an anchoring group (*meta*-BT) can be controlled by the electrode potential of 1-butyl-3-methylimidazolium hexafluorophosphate [Bmim][PF₆].²⁵ At a gate potential of -0.4 V, the molecular conductivity is two orders of magnitude lower than at a gate potential of 0 V, which is mainly due to the DQI effect. Similarly, the junction of 2,4-TP-SAc with DQI can be controlled by the ionic liquid gate.²⁶ The conductance of the molecular junction decreases and then increases in the potential ranges of -0.6 V to -0.4 V (with respect to Ag/AgCl) and -0.4 V to 1.4 V (with respect to Ag/AgCl), resulting in on-off ratios of up to ~ 100 times higher. This demonstrates the feasibility of constructing high-performance single-molecule devices by continuously and efficiently modulating molecules with the DQI effect through ILs. In

addition to individual molecules, the monomolecular layer formed by molecular self-assembly also has the same quantum interference effect as the basic molecules, which is conducive to the construction of efficient and stable quantum tunnelling field effect devices. For example, a vertical molecular tunnelling field effect device with an ILG was created using pseudo-*p*-bis ((4-(acetylthio)phenyl)ethynyl)-*p*-(2,2) cyclohexane (PCP) together with DQI to achieve ~ 330 on-off current ratio.²⁷ In addition to quantum interference effects, the creation of high-performance single-molecule functional devices can be achieved by combining various molecular properties with ILG control. When the structure of the single-molecule junction is fixed, the enhanced molecular electric field response allows for larger switching ratios at lower gate voltages. This is due to the fact



that molecules containing azulene centres have higher dipole moments, resulting in lower leakage currents (currents when the Fermi energy level is between the HOMO and LUMO orbitals). Therefore, this increases the on/off ratio.

In electronic and optoelectronic devices of the next generation, carrier scattering has been shown a significant effect on device performance.³¹ To this end, researchers have proposed various strategies, such as introducing 2D materials (e.g., graphene, h-BN) to construct van der Waals heterostructures, strain modulation, and surface passivation methods to suppress interfacial states and scattering phenomena.^{32,33} In hetero-structured semiconductors, synergistic effects such as interfacial charge transfer, band alignment and spin orbit coupling play a decisive role in the transport performance of devices.³⁴ By accurately designing these interfacial behaviours, not only can carrier complexation be effectively suppressed and scattering be reduced, but also ideal energy band structures and transport channels can be constructed to support the development of cutting-edge applications such as spintronics and valleytronic applications. It is noteworthy that interfacial engineering strategies based on ILs offer more efficient solutions to the aforementioned processes. For instance, in the context of chalcogenide light-emitting diodes (LED), the use of 1-butyl-3-methylimidazolium tetrafluoroborate ([Bmim][BF₄]) and 1-benzyl-3-methylimidazolium tetrafluoroborate ([Bzim][BF₄]) has been demonstrated to facilitate the efficient passivation of surface trap defects through the establishment of hydrogen bonds. This process is concomitant with the modulation of the Fermi energy level of the chalcogenide, thereby ensuring its positioning at the mid-band level. This, in turn, has been shown to result in a substantial enhancement in device performance. The mid-band position has been shown to significantly enhance the performance of the device. The ILs-treated chalcogenide material has been demonstrated to achieve an external quantum efficiency of up to 20.03%, and the prepared Pe LED have exhibited green and efficient luminescence with a brightness of up to 57599.36 cd m⁻² (Fig. 2b).²⁹

The advent of a new generation of EIMs, characterised by heterogeneous integration, has given rise to a series of challenges that have heretofore not been encountered. These challenges are manifest in the interface multidimensional structure of devices.^{30,35} As previously stated, ILs demonstrate distinctive advantages in the modulation of interfacial electronic structure due to their high designability. Recent studies have demonstrated that an IL-system comprising an organic superbase and bis(trifluoromethanesulfonyl)imide (TFSI) can be utilised for the electrochemical doping of carbon nanotubes (CNTs), thereby achieving excellent n-type and p-type doping stability (Fig. 2c). The system has been demonstrated to effectively regulate the Seebeck coefficient and conductivity by adjusting the polarity and amplitude of the applied potential. Furthermore, it has been shown to achieve reversible adjustment of thermoelectric properties in a wide range, and to exhibit a stable operation capability of more than 1500 h at an operating temperature of 100 °C.³⁶

More critically, EDL effect induced by ILs significantly enhances the thermoelectric properties of the doped films. Under n-type (−0.25 V vs. Ag/Ag⁺) and p-type (+1.5 V vs. Ag/Ag⁺) bias, the resulting CNTs films achieved 104 μW m⁻¹ K⁻² and 223 μW m⁻¹ K⁻² power factors, which are close to the highest levels of thermoelectric CNTs in the literature.^{37,38} This suggests that ILs can not only achieve efficient carrier modulation at the interface, but also effectively enhance the performance limit of thermoelectric materials.

As discussed above, the development of next-generation EIMs requires a holistic design strategy encompassing electronic band engineering, carrier transport enhancement, and precise interfacial modulation. ILs have demonstrated unique advantages in each of these domains by enabling non-invasive, reversible, and highly localized tuning of material properties. Their integration into material systems across different spatial dimensions (0D to 2D) reveals significant potential to overcome long-standing limitations in electronic material performance.³⁹ In the following section, we systematically examine how ionic liquids contribute to the performance of electronic information materials across different dimensional regimes, highlighting their role as dynamic enablers of functional enhancement and design flexibility.

2.2 ILs as tunable platforms for EIMs

ILs are generally defined as salts composed entirely of ions, which remain liquid below 100 °C, a criterion that distinguishes them from conventional molten salts.⁴⁰ Their modular structure enables the independent selection and combination of a wide array of organic cations (e.g., imidazolium, pyridinium, phosphonium) and organic/inorganic anions (e.g., [TFSI], [BF₄], [PF₆]), giving rise to a virtually limitless chemical space (up to 10¹⁸ theoretical combinations). This structural diversity is exemplified by aprotic ILs containing large cations (e.g., imidazolium, pyridinium, phosphonium) paired with compact anions, commonly inorganic species ([Br][−], [Cl][−], [PF₆][−], [BF₄][−]) or organic counterparts like TFSI (Fig. 3a).⁴¹ In contrast, protic ILs derive from proton transfer between Brønsted acids and bases, creating proton-donor/acceptor sites that enable hydrogen-bonded networks.^{42,43} This structural flexibility underpins their core advantage that is tunability. Building upon the structural and functional demands outlined for high-performance EIMs, ILs emerge as a versatile class of molecular platforms ideally suited to meet these design challenges. Over the past two decades, ILs have garnered significant academic and industrial attention owing to their exceptional and tunable physicochemical characteristics.

Rather than exhaustively listing IL families, we highlight key structural motifs that underpin their application in electronic materials. Aprotic ILs, particularly 1,3-dialkylimidazolium derivatives with optimized anion pairs (Fig. 3b), dominate electrochemical applications due to their excellent ionic conductivity, low volatility, and high electrochemical stability.⁴⁴ These ILs often organize into nanostructured domains through



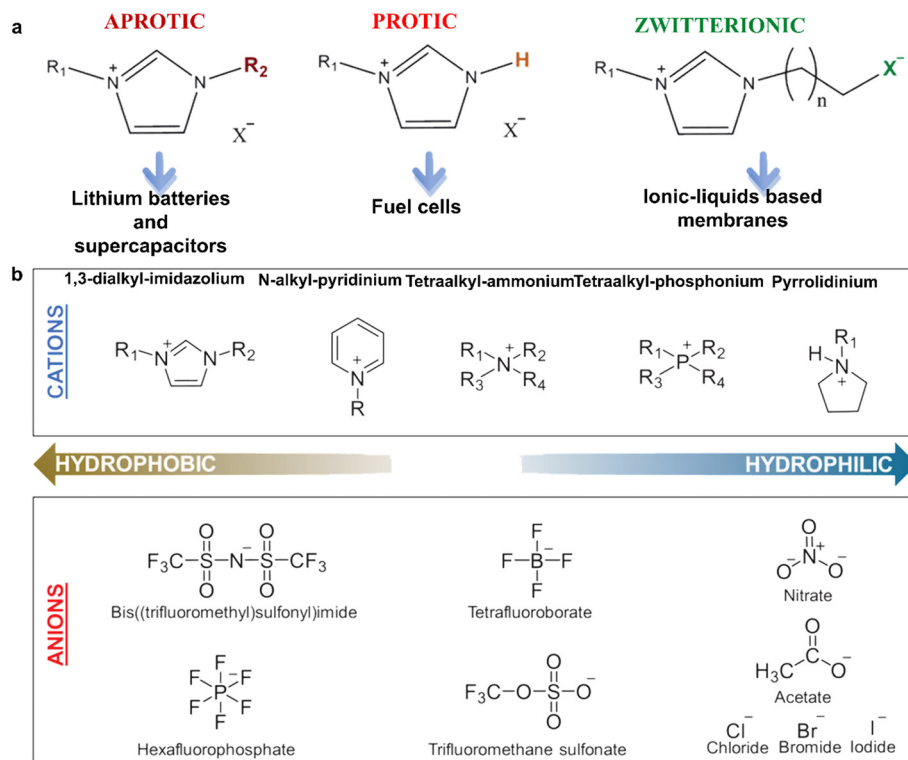


Fig. 3 ILs classification and structure diagram. (a) General classification of ionic liquids with an indication of a representative application area;⁴¹ (b) schematic representation of the most common cations and anions described in the literature. Reproduced with permission.⁴⁴ Copyright 2009, Springer Nature; 2013, RSC.

a balance of coulombic, van der Waals, and hydrogen-bonding interactions, yielding microphase-separated architectures with both polar and nonpolar regions.⁴⁵ This feature facilitates the formation of stable, anisotropic interfaces, which is an essential requirement for emerging electronic devices and nanostructured components. In addition, ILs offer internal charge neutrality while protic ILs provide proton-conductive pathways; both types extend the functionality of ILs into soft electronics and responsive interfaces by modulating complex fluid systems (e.g., nanogels, emulsions).⁴⁶

By tailoring molecular constituents, ILs can be systematically designed to enhance specific performance metrics relevant to EIMs, such as interfacial stability, dielectric response, thermal robustness, or charge carrier mobility. Consequently, they serve not only as solvents or electrolytes but also as active materials capable of modulating electronic behaviour, enabling applications across fields such as energy storage, printed electronics, and sensor technologies. Therefore, the structural diversity and inherent tunability of ILs make them uniquely aligned with the design imperatives for next-generation EIMs. Their role is not ancillary but fundamental, offering a molecular-level toolkit for achieving multi-functional integration, precision control, and high stability in device-relevant environments. The following section will delve deeper into the physicochemical properties that underpin these capabilities.

3 ILs-media for EIMs across dimensional scales

EIMs form the cornerstone of modern electronic and information technologies, underpinning the performance and development of integrated circuits, sensors, display technologies, communication systems, and beyond.⁴⁷ EIMs serve as the functional backbone of electronic hardware and are increasingly expected to meet stringent demands for miniaturization, flexibility, multifunctionality, and energy efficiency in next-generation technologies. However, as the scaling limits of silicon-based materials draw near, marked by quantum tunneling effects and thermodynamic inefficiencies at the nanoscale,^{48–50} conventional materials and processing techniques face diminishing returns. In this context, nanomaterials, classified by their structural dimensionality (0D, 1D, and 2D), have emerged as compelling alternatives. With unique attributes such as quantum confinement, superior charge mobility, and enhanced mechanical compliance, these materials promise to revolutionize electronics, photonics, and energy systems, enabling advances in fields ranging from wearable electronics to quantum computing.^{51,52}

Crucially, ILs have demonstrated exceptional potential in modulating and supporting the synthesis, assembly, and performance optimization of such nanomaterials.⁵³ Their distinctive physicochemical properties, such as high ionic conductivity, tunable polarity, and strong interfacial behaviour,



offer an adaptive medium capable of guiding material growth, morphology, and interfacial interactions with atomic-scale precision. Though many IL-assisted materials have yet to achieve direct commercial deployment in electronic information technologies, their structural and functional traits align strongly with the unmet needs of the field. These include enhanced environmental stability, novel charge transport mechanisms, and tailored interface chemistry, all of which are critical to overcoming current limitations in EIMs development.

This section explores IL-facilitated advances in materials across various dimensions, ranging from 0D to 2D materials, highlighting the ways in which ILs contribute to structural control, property enhancement, and application viability within the EIMs landscape.

3.1 ILs-guided engineering of 0D EIMs

0D nanomaterials, exemplified by quantum dots (QDs), have emerged as foundational building blocks for next-generation electronic and optoelectronic devices due to their quantum size effects, tunable electronic structures, and excellent solution processability.^{54,55} Their size-dependent optical and electronic properties, such as bandgap tunability, high photoluminescence quantum yield, and narrow emission linewidths, have enabled significant advances in displays, photodetectors, and energy-efficient lighting technologies. For instance, CdSe QDs exhibit bandgap modulation from 1.74 eV to 2.36 eV and emission tunability across the visible spectrum, with quantum yields surpassing 90%.^{52,55,56} While significant advances have been made with conventional synthetic approaches, exemplified by the hot-injection technique developed by Bawendi and Murray^{57,58} (awarded the 2023 Nobel Prize in Chemistry), which enable precise control over particle size and uniformity, driving the successful integration of QDs into Samsung's QD display technology.⁵⁹ Despite these advantages, these methods still face limitations including reliance on costly/toxic precursors and scalability challenges.

Against this backdrop, ILs have emerged as versatile media to overcome these constraints. ILs leverage unique properties, electrostatic interactions, solvation power, and steric hindrance, that play pivotal roles in controlling particle size, dispersion, and stability.^{60–62} Moreover, IL-mediated synthesis routes have demonstrated compatibility with green chemistry principles, minimizing the use of volatile organic solvents and enabling recyclable reaction media. These capabilities position ILs not only as enabling tools for scalable, high-quality 0D material production, but also as functional interfaces that can enhance charge transport, energy level alignment, and environmental stability when QDs are incorporated into electronic devices. Therefore, ILs serve as versatile mediators for engineering zero-dimensional information materials with atomic-level precision, offering new synthetic paradigms and interfacial functionalities critical for the advancement of EIMs.

3.1.1 Ionic modulation in nanocrystal growth of 0D EIMs.

Building on the critical role of ILs in 0D nanomaterial

engineering, their multiscale regulation mechanisms during synthesis offer a deeper insight into how atomic-level interactions translate into mesoscopic structural precision and functional enhancement. ILs provide a unique ionic environment wherein both cationic and anionic components contribute to the nucleation, growth, and stabilization of metal and semiconductor nanoparticles (NPs) through distinct but complementary mechanisms.

At the nanoscale, anions in IL are electrostatically stabilised by forming a charged protective shell around the NPs, inhibiting aggregation without the need for additional ligands (Fig. 4a).⁴⁵ The nature of this protective shell is consistent with the description of the EDL in Derjaguin–Landau–Verwey–Overbeek (DLVO) theory.^{63,64} It has been demonstrated that IL ions act as discrete charged entities that counteract van der Waals attraction through space charge repulsion, thus maintaining NPs dispersion stability.^{65,66} Au and Pd NPs are often used in optical contacts or photodetectors for nanoscale devices due to their high electrical conductivity, controlled surface catalytic activity and chemical inertness.⁶⁷ For example, Au NPs synthesized in [Bmim][BF₄] showed monodisperse size distributions, significantly outperforming other ILs, which yielded larger particles under otherwise comparable conditions.⁶⁵ In addition, the type and functionality of the IL anion markedly influence not only particle size but also oxidation resistance and formation kinetics. Hydroxyl-functionalized ILs, such as [C₂OHmim]-based systems paired with [OTf], [TFA], or [Tf₂N] anions, have been shown to produce Pd NPs with smaller average diameters (2.3–4.0 nm) and greater oxidation resistance compared to their non-functionalized analogues.⁶⁸

On a molecular level, ILs form spatially resolved interfacial structures around growing nanocrystals. Electrostatic stabilization is often mediated by layered arrangements, initial anion adsorption followed by loosely ordered cationic shells (Fig. 4b). The thickness of these EDL (typically 2.8–4.0 nm) depends on anion size and symmetry.^{69,70} For instance, the oblique adsorption of asymmetric [Tf₂N] anions on positively charged surfaces creates a dynamic cis-oriented solvation interface, enhancing colloidal dispersion and impeding sintering.⁷¹ The Ir nanoparticle-based TTA conversion system is capable of continuous modulation of the emission spectrum from orange (580 nm) to blue (460 nm).⁷² Upon synthesis of Ir (0) NPs in ILs, SAXS and XPS analyses confirmed the formation of conformal IL layers with precise spatial dimensions (~2–4 nm), which remained associated with the particle surfaces even after isolation.⁷³

Cations of ionic liquids can also coordinate with the surfaces of metal NPs (Fig. 4b). For example, in the case of the interaction between [Bmim] cations and Ag NPs, the cation coordinates with the particle surface in a flat configuration, forming a positively charged protective layer, while [BF₄] anions do not chemically interact with the particle surface (Fig. 4b).⁷⁹ Additionally, the electrostatic layer formed by the interaction of [Emim] cations with the surface of silica has a thickness of 0.5 nm, while the thickness of the



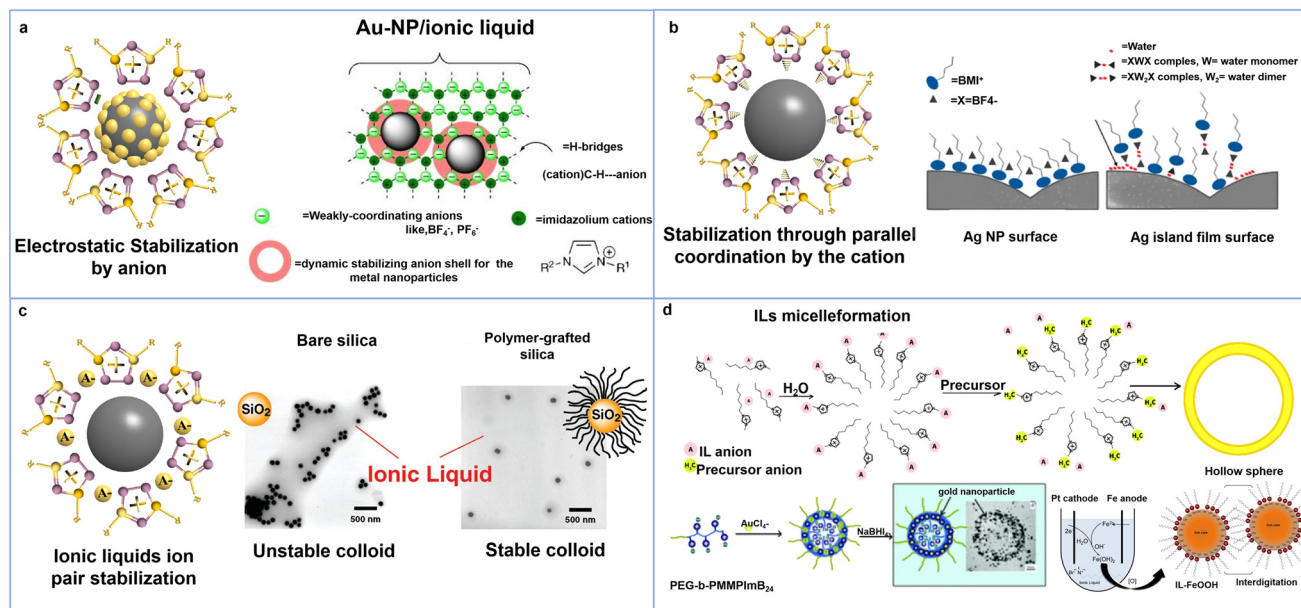


Fig. 4 ILs-guided 0D material construction: (a) anion-based electrostatic stabilization;⁴⁵ (b) cation coordination;⁷⁴ (c) ion-pair stabilization;^{75,76} (d) micelle formation.^{45,77,78} Copyright 2008, ACS; 2014, RSC and 2016, Elsevier.

second solvation layer is 1.15 nm, indicating subtle interactions between the cation and the surface.⁷⁴ Furthermore, in ILs systems, cations and anions act cooperatively as structured ion pairs to form well-ordered interfacial monolayers (Fig. 4c).⁸⁰ This behaviour has been reported at Pt and Ru surfaces, where ion pair orientation depends on surface charge and contributes to enhanced nanoparticle stability (Fig. 4c).^{75,76} Classical DLVO models suggest that the high ionic strength and spatial heterogeneity of ILs favor such ordered interfacial architectures, balancing electrostatic repulsion with van der Waals attractions to achieve thermodynamically stable dispersions.^{63,64}

Taken together, these findings highlight that ILs act not merely as passive media but as active regulators of nanocrystal formation and surface structure through multiscale ionic interactions. Their ability to orchestrate electrostatic, coordinative, and steric stabilization mechanisms positions them as powerful tools in the rational design of 0D EIMs with enhanced precision, stability, and device integration potential.

3.1.2 ILs-driven green nanofabrication of 0D EIMs. In the pursuit of environmentally benign and structurally precise 0D EIMs, ILs offer not only chemical tunability and low volatility but also unique self-assembly behaviors that enable green synthesis strategies. Beyond their intrinsic ionic interactions discussed in section 3.1.1, ILs can participate actively in micro-structured environments, such as micelles and microemulsions, serving as templating agents, phase stabilizers, and growth modulators in the fabrication of 0D nanomaterials.^{45,77}

ILs combined with conventional surfactants can form micellar structures, effectively controlling the size and morphology of nanoparticles (Fig. 4d). The ability of micellar aggregates formed by polymeric ILs to control the size of gold nanoparticles, as well

as the effect of IL concentration, was proposed by Li.⁷⁷ At low metal salt concentrations, vesicular assemblies form with uniformly distributed 4 nm gold NPs localized at the bilayer interfaces (Fig. 4d). In contrast, higher precursor concentrations favor the formation of dense, spherical micelles yielding larger NPs (~10 nm), emphasizing the role of electrostatic repulsion between IL cations in suppressing aggregation within confined environments. This micellar templating effect extends to other systems as well. In aqueous IL-surfactant solutions, *e.g.*, dodecyltrimethylammonium bromide, α -FeOOH NPs exhibited smaller sizes and higher dispersion uniformity compared to control samples lacking ILs. It was proposed that IL headgroups associate with hydroxylated iron oxide surfaces while the hydrophobic tails extend outward, forming reverse-micelle-like structures that encapsulate the growing particles (Fig. 4d).⁷⁸ Additionally, the nature of the solvent modulates micellar structuring and nanoparticle formation.

A more versatile and scalable approach involves IL-based microemulsions, which allow the dispersion of otherwise immiscible phases and enable control over nanostructure dimensions, while also addressing challenges such as precursor solubility and high IL viscosity.⁸¹ In these ternary or quaternary systems, ILs can simultaneously serve as the polar phase, nonpolar medium, and even as surfactant components, creating a unified platform for soft templating.^{82,83} This approach not only eliminates volatile organic solvents but also enhances system conductivity and thermal stability, making it particularly suitable for the fabrication of high-purity 0D materials for electronic applications. In addition, the structural control within IL-based microemulsions arises from the nanodroplet templating effect. Higher IL surfactant concentrations result in smaller and more numerous NPs, while an increase in the



water phase under fixed IL/surfactant ratios leads to larger particle sizes, in line with droplet swelling phenomena.^{83,84}

These studies underscore the emerging paradigm wherein ILs function not only as reaction media but as active supramolecular directors in green synthesis. Through micellar self-assembly and nanodroplet confinement, ILs allow environmentally friendly and compositionally versatile control over 0D nanomaterials, paving the way toward scalable, sustainable, and morphologically tunable platforms for next-generation electronic information technologies.

3.2 ILs in 1D EIMs: harnessing quantum effects

1D nanomaterials are regarded as a robust substitute for silicon-based materials in the domains of nanoelectronics and quantum devices. This is due to the quantum-limited domain effect in the transverse dimension and the intrinsically high carrier mobility, which exhibits excellent electron transport properties and size tunability. 1D materials generally refer to nanostructures with high aspect ratios, such as nanowires, nanotubes, and nanorods.

3.2.1 Quantum confinement effects and fabrication challenges. Quantum confinement effects dominate the

electronic behaviour of 1D EIMs when their characteristic dimensions enter the sub-10 nm domain. This reconfigures the carrier transport mechanism through energy level discretisation. Theoretical studies show that this effect suppresses electron-phonon scattering (increasing the probability of ballistic transport to 85%) and modulates the Fermi energy level depending on its size. This provides a physical basis for 1D materials outperforming conventional silicon-based devices (Fig. 5a).^{85,86} Research shows that specific chiral CNTs with a diameter of 2 nm have a mobility exceeding $10^5 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ and a saturation velocity of $4 \times 10^7 \text{ cm s}^{-1}$ at room temperature. This is two orders of magnitude and four times higher, respectively, than silicon materials.⁸⁷ In contrast, Cu nanowires with a surface step height of less than 0 nm have a lower mobility 5 nm were successfully fabricated through morphological engineering to achieve maximum absorption close to 100 per cent (99.98 per cent) and form ohmic contacts in the deep ultraviolet region.^{88,89}

Despite significant progress being made, the preparation for scaling up and interfacial modulation remain key bottlenecks. In contrast to chemical vapour deposition (CVD), which can produce high-quality CNTs ($I_D/I_G < 0.05$) but is costly and limited in practicality, solution stripping can

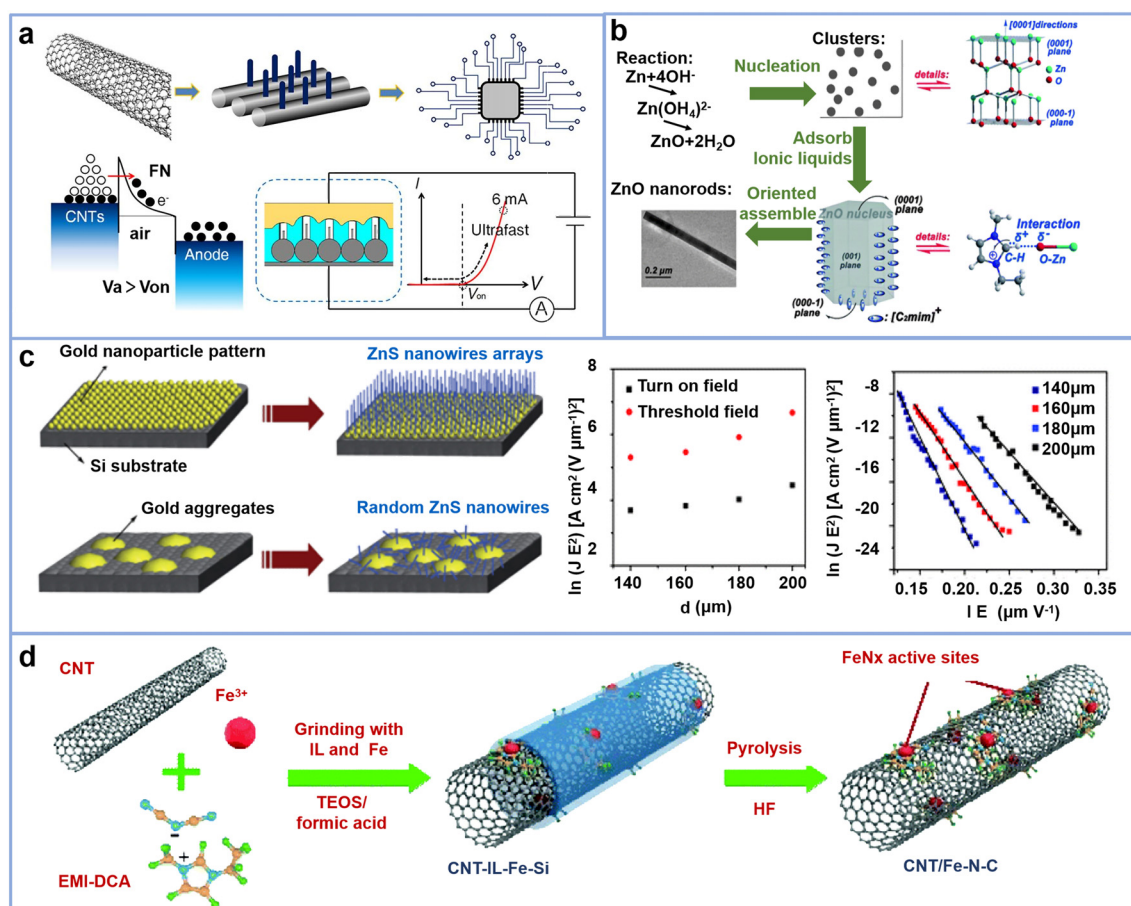


Fig. 5 Representative roles of ILs in 1D EIMs. (a) CNT-based air-channel device;⁸⁵ (b) IL-induced growth of ZnO nanorods;⁹³ (c) IL-guided assembly of ZnS nanowires on gold templates;⁹⁹ (d) IL-assisted synthesis of CNT/Fe-N-C electrocatalysts for ORR.¹⁰¹ Copyright 2023, ACS and 2011, 2021, RSC.



control the diameter deviation within 5% at a tenfold cost reduction by optimizing the solvent system (e.g., DMF/water = 3:1).^{90,91} At the device level, contact resistance can account for up to 70% of the total resistance, which makes it a significant obstacle to improving performance. However, recent interface engineering breakthroughs demonstrate that self-assembled monolayer (SAM) modification can reduce the Schottky barrier by 40% ($\Delta\Phi = 0.12$ eV), while gradient metal stacking (Ag/TiN/Pt) can depress the contact resistance to $8 \times 10^{-8} \Omega \text{ cm}^2$.⁹² These advances reveal that industrial translation of the quantum advantages of 1D materials requires the synergistic optimisation of bottom-up synthetic control and top-down interface design. As discussed in section 2, interfacial modification techniques using ILs provide an innovative solution to the paradox of contact resistance and thermal stability, and are expected to bridge the gap between the properties of quantum confinement and fabricatable device architectures.

3.2.2 ILs as enablers for 1D EIMs' morphological control and functionalization. The unique role of ILs as “intelligent soft templates”, as described in the previous section, also demonstrates excellent capabilities in the precise synthesis, assembly and functionalisation of 1D materials (Fig. 5b).⁹³ 1D carbon nanomaterials occupy a key position in the construction of flexible energy and nanodevices, and ILs intelligent modulation has shown direct results in the modulation of their dispersion and optimisation of their interfacial electronic behaviour. Li prepared metal-free catalysts by introducing imidazolium-based ILs into CNTs through an *in situ* functionalization strategy, which significantly enhanced their performance in electrocatalytic hydrogenolysis reaction (HER).⁹⁴ ILs has also been used as a dispersant in polymer complexes. ILs enable CNTs to form a continuous conductive network in PLA/EVA matrix, which significantly enhances the dielectric and mechanical properties of the composites.⁹⁵ CNTs functionalized based on ILs have also been used to construct highly sensitive electrochemical sensors exhibiting good response properties.⁹⁶

1D semiconductor nanowires, including ZnS and Sb nanowires, are of significant importance as fundamental components in the fabrication of high-density electronic devices and micro/nano-sensing platforms.^{97,98} In related studies, a variety of nanowire systems with fine structures and outstanding functions have been successfully constructed using different ionic liquid environments. ZnS nanowires with diameters of less than 50 nm and lengths of up to 10 μm and centimetre-sized ZnS single-crystal nanowires with aspect ratios of more than 10^5 have been synthesised in the presence of [Bmim][BF₄] (Fig. 5c).⁹⁹ Si nanowires with low deviation etching were realised using [Py1,4][TfO].¹⁰⁰ Furthermore, Si/Tb nanowire structures with homogeneous axial components and room-temperature visible luminescence with millisecond lifetimes were prepared under the template-assisted method (Fig. 5d).¹⁰¹

Boron nitride nanotubes (BNNTs) have been limited in structure functionalisation due to their broad forbidden band and chemical inertness.¹⁰² The introduction of IL provides a novel concept for surface modification.^{103–105}

Hossein investigated the effect of modifications to different organic functional groups (e.g. CH₃CO, 2-methoxy-*N,N*-dimethylethylamine) on the electronic structure of BNNTs. This investigation was based on density functional theory (DFT). The study found that ionic liquid functionalization can significantly enhance the electrical conductivity and chemical softness of BNNTs. This lays the foundation for the practical application of BNNTs in aqueous and non-aqueous media.¹⁰⁶

In summary, IL-assisted 1D material synthesis and functionalization strategy not only promotes the efficient construction of a variety of typical quantum-limited domain structures, but also significantly improves the interfacial properties and integrability of device-level materials. This structural foundation and interfacial guarantee is essential for the large-scale application of 1D EIMs.

3.3 ILs-mediated engineering of 2D EIMs

In the context of EIMs, the dimensionality of the material in question exerts a direct influence on its physicochemical properties and the potential applications thereof. In comparison to 0D EIMs, where carrier localisation is severe and efficient transport is challenging to achieve, and 1D EIMs, where electrical conductivity is limited, 2D EIMs achieve a high degree of synergy between electronic and structural properties. This is due to their atomically hierarchically ordered structure, high carrier mobility and excellent mechanical compliance.^{114,115} Typical 2D EIMs, such as graphene, transition metal dihalides (TMDs) and black phosphorus (BP), are rapidly becoming key material platforms for next-generation nanoelectronic and optoelectronic devices due to their atomic-level thickness, dangling-bond-free surface structure and excellent compatibility with van der Waals heterostructures in FETs (Fig. 6a).¹⁰⁷

Similarly, ILs have demonstrated significant potential in the study of 2D EIMs. Their multifaceted functions, ranging from exfoliation solvents and stabilizing agents to reaction accelerants and structural modulators, enable precise manipulation of 2D nanosheet morphology, thickness, and electronic properties. This section focuses on the primary application directions of ILs in 2D materials.

3.3.1 ILs in liquid-phase exfoliation of traditional EIMs.

Graphene, the most representative 2D material, has attracted significant attention from the research community in the field of developing next-generation electronic devices (e.g. high-frequency transistors, photodetectors) due to its fascinating electrical and optical properties.^{116,117} In order to achieve the potential of the aforementioned device applications, it is imperative that a breakthrough in efficient macro-preparation is realised. The fabrication of graphene can be achieved through a variety of methods, including exfoliation, CVD, and reduction of graphene oxide.^{118,119} CVD is regarded as one of the most promising techniques for the industrial production of graphene (Fig. 6b).⁹¹ It is unfortunate that the preparation of materials by this method frequently necessitates the use of costly catalysts.¹²⁰ However, graphene films grown on solid substrates in the CVD process are



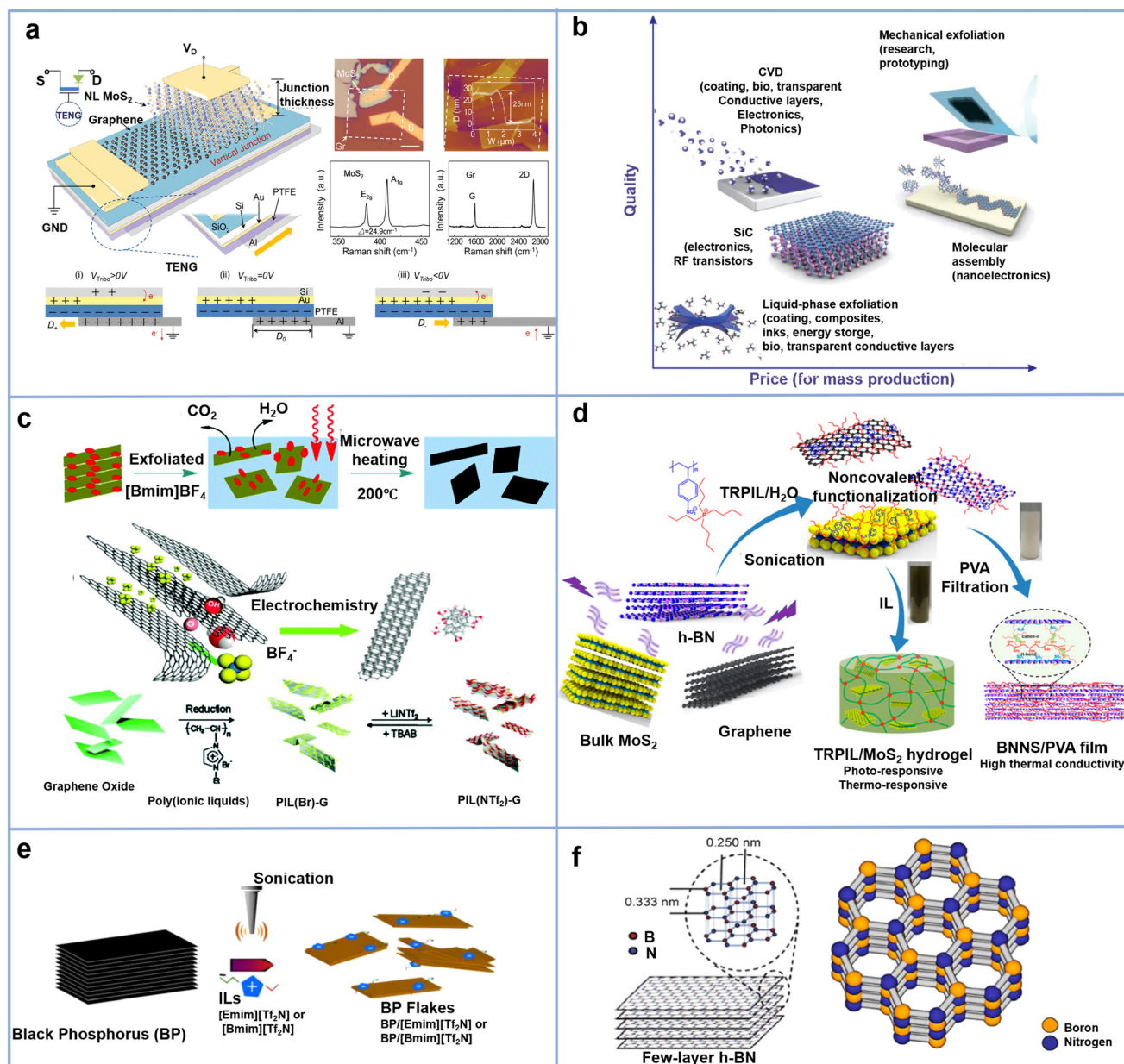


Fig. 6 IL-mediated for 2D EIMs. (a) Morphology and spectral features of vertical field-effect transistors (FETs) based on 2D EIMs, along with a schematic of their integration in sliding-mode triboelectric nanogenerators (TENGs);¹⁰⁷ (b) scalable graphene production methods;⁹¹ (c) microwave-assisted synthesis;^{108,109} radical-induced exfoliation intercalation;¹¹⁰ (d) TRPIL-assisted exfoliation of layered materials;¹¹¹ (e) schematic diagram exfoliation of bulk BP;¹¹² (f) structure and stacking of h-BN nanosheets.¹¹³ Copyright 2024 WILEY; 2018 MDPI; 2015 RSC and soon.

difficult to transfer to other substrates for further applications.¹²¹ Liquid phase exfoliation (LPE) is widely regarded as the most promising technology for industrialisation due to its ambient temperature and pressure operation, adjustable solvents, simple equipment and continuous production advantages.^{122,123} The core mechanism of IL-assisted stripping, as an optimised branch of LPE, relies on strong cation- π interactions between IL cations and graphite layers, significantly improving stripping efficiency and overcoming the limitations of conventional organic solvents.¹²⁴

This technique enabled the preparation of graphene nanosheets (GNS) with concentrations reaching up to 5.33 mg

mL⁻¹.¹²⁵ Subsequent STEM/XPS characterisation confirmed that the products exhibited ultrathin structures (*i.e.* comprising fewer than five layers), low defect rates and high crystalline integrity.^{126–128} These properties are essential for preserving the intrinsic electrical properties of graphene and for constructing high-performance heterojunction devices.¹⁰⁹ The continuous advancement of IL in the field of material optimisation is driven by the utilisation of microwave assistance. The process of microwave assistance involves subjecting graphene oxide (GO) to a treatment at 200 °C for a duration of 40 minutes, resulting in the acquisition of high-conductive, reduced graphene oxide (RGO) with an enhanced electrical conductivity of 4 times. RGO

demonstrates remarkable performance as an electrical capacitor, exhibiting a capacitance of 135 F g^{-1} at 3.5 V (Fig. 6c).¹⁰⁸ The chemical dissolution method employs $[\text{C}_8\text{mim}][\text{PF}_6]$ or water-soluble IL (e.g., $[\text{Bmim}][\text{BF}_4]$) in the $[\text{C}_8\text{mim}][\text{BF}_4]$ system for the preparation of films with a thickness of up to 1 nm . As illustrated in Fig. 6c, the 1 nm thick sensory fossil layer exhibits both efficiency and adaptability in terms of its manufacturing process.^{110,129}

In addition to playing a key role in LPE of graphene, ILs also show unique advantages in the transfer of monolayer graphene. The utilisation of poly(ionic liquid) (PIL) 1-vinyl-3-ethylimidazolium bromide in the formation of $[\text{PIL}][\text{Br}]\text{-G}$ complexes with GO has resulted in the development of a water-soluble system. This system facilitates the reversible phase transfer of reduced RGO between aqueous and organic solvents. Subsequent addition of lithium salt ($[\text{Li}][\text{Tf}_2\text{N}]$) to this system induced an anion exchange between $[\text{Br}]^-$ and $[\text{Tf}_2\text{N}]^-$ to form $[\text{PIL}][\text{NTf}_2]\text{-G}$ complexes that are soluble in organic phases, thus further expanding the application of graphene in the field of electronic information technology (Fig. 6c). In the presence of $[\text{C}_{12}\text{mim}][\text{Br}]$, GO dispersed in the aqueous phase can be transferred to non-polar organic solvents such as *n*-dichlorobenzene. Following high-temperature annealing, the sp^2 properties of RGO are restored, resulting in electrical conductivity levels of up to 823 S cm^{-1} .¹³⁰

3.3.2 ILs in exfoliation and functionalization of emerging 2D EIMs. The maturation of LPE has led to its application being extended from graphene to a wider range of novel 2D EIMs. These include TMDs, BP and BNNTs. It is evident that each of these materials exhibits distinct electronic structural advantages. However, their interlayer bonding strength, surface activity and chemical stability differ significantly, thereby necessitating disparate requirements with regard to the exfoliation mechanism and interface regulation strategy. ILs as an interfacial medium with a highly designable molecular structure, demonstrates unique advantages in regulating the dispersion of the lamellae, the electronic integrity and the environmental adaptability.

MoS_2 , typical materials of TMDs, exhibits excellent flexibility and tunability, making it highly promising for applications in energy storage devices and flexible displays.¹³¹ A simple one-pot sonication approach yielded MoS_2 sheets with intact crystallinity and smooth layered structures, achieving a monolayer yield exceeding 50% .¹³² Further systematic analysis revealed that the exfoliation efficiency of the $[\text{Bpy}][\text{X}]$ series of ILs ($\text{X} = [\text{BF}_4]^-$, $[\text{PF}_6]^-$ and $[\text{Tf}_2\text{N}]^-$) is closely related to their intermolecular aggregation capabilities. The results indicated that among the nine studied IL compounds, the $[\text{Bpy}][\text{X}]$ combinations ($\text{X} = [\text{BF}_4]^-$, $[\text{PF}_6]^-$ and $[\text{Tf}_2\text{N}]^-$) exhibited the strongest aggregation abilities. This finding suggests that intermolecular bonding within the $[\text{Bpy}][\text{X}]$ systems plays a critical role in enhancing the intercalation–exfoliation efficiency of the solvents.¹³³ By optimising the growth and exfoliation interfaces, Gao *et al.* found that bulky cations (e.g. imidazolium) in IL can adsorb synergistically with the MoS_2 surface *via* π -d interactions, significantly enhancing the interlayer

lubricity and intercalation efficiency, leading to the acquisition of high-quality monolayers of MoS_2 with preservation of the in-plane lattice ordering and photoresponsive properties.¹³⁴

Furthermore, thermostable PIL (TRPIL) provides a dynamic peeling strategy (Fig. 6d). At low temperatures, hydrophobic interactions drive peeling through π - π stacking, while at high temperatures, steric hindrance from hydrophilic structural domain expansion inhibits reaggregation. In this manner, dual-responsive TRPIL/ MoS_2 composites for smart hydrogels and high-strength pearl-like films were developed.¹³⁵

BP has found widespread application in near-infrared (NIR) optoelectronic devices, a testament to its tunable bandgap and high mobility.^{136,137} The achievement of high stability liquid-phase stripping of BP is a key prerequisite for its application in the fabrication of optoelectronic devices. China Electronics Technology Group Corporation's patent for a BP-based infrared detector and a prototype BP field effect transistor device developed by STMicroelectronics highlight the importance of BP in this field.¹³⁸ However, its strong oxygen sensitivity severely limits its liquid phase processing window. Zhao prepared highly concentrated, air-stable monolayer to few-layer BP dispersions using ILs such as $[\text{Hoemim}][\text{TfO}]$.¹³⁹ The monolayers obtained from the dispersion of BP nanosheets in ambient air demonstrated stability for up to one month, exhibiting excellent stability. In a similar manner, Lee *et al.* demonstrated that $[\text{Emim}][\text{Tf}_2\text{N}]$ and $[\text{Bmim}][\text{Tf}_2\text{N}]$ not only efficiently removed BP by ultrasonication, but also protected the BP from oxidative degradation by the IL-derived passivation layer. This resulted in scalable processing under ambient conditions. The multilayered BP dispersions obtained have unexpectedly high antioxidant activity as well as chemical and structural integrity (Fig. 6e).¹¹² Known as “white graphene”, BNNS offer excellent thermal conductivity and dielectric performance (Fig. 6f), showing great potential for applications in thermal management, high-frequency electronic devices, and flexible packaging.¹¹³ However, their large-scale exfoliation has been limited. ILs with surface energies comparable to h-BN, such as $[\text{Bmim}][\text{PF}_6]$ and $[\text{Bmim}][\text{Tf}_2\text{N}]$, facilitate effective exfoliation through cation- π and hydrogen bonding interactions.^{140–142} Systematic studies suggest that the anion plays a more decisive role than the cation in determining exfoliation efficiency and final concentration, with the general trend: $[\text{PF}_6]^- > [\text{Tf}_2\text{N}]^- > [\text{BF}_4]^- > [\text{TfO}]^-$, reflecting differences in anion basicity and hydrogen bonding ability.¹⁴¹

In summary, ILs exhibit remarkable versatility and tunability in the synthesis, functionalization, and performance optimization of EIMs spanning 0D, 1D, and 2D systems. For 0D materials such as QDs, ILs enable precise control over particle size, shape, and surface chemistry, thereby enhancing their optical and electronic properties. In 1D systems including CNTs, metal nanowires, and semiconductor nanowires, ILs facilitate selective dispersion, alignment, and doping, contributing to improved electrical conductivity and mechanical integrity. Most notably, in the context of 2D materials, ILs serve as green and highly designable media for efficient LPE and stable dispersion of graphene, TMDs (e.g., MoS_2), BP, and BNNTs. Through noncovalent interactions, thermoresponsive behaviour, and tailored ion-surface



interactions, ILs not only enhance exfoliation yield and material stability but also unlock new routes for multifunctional device integration. These insights collectively underscore the immense potential of ILs as multifunctional agents in the development of next-generation EIMs and devices, offering a powerful platform for green, scalable, and high-performance material engineering across all dimensional hierarchies.

4 ILs in high-purity separation and purification of ECs

In the synthesis of ECs and device fabrication, impurity control and high-purity separation of products are key technical steps to ensure material performance and device stability.^{54,143} In synthesis, purification and process, a variety of metal ion impurities are often introduced or residual, and the main sources of these metal ions include catalyst residues, raw material impurities, equipment corrosion products and process pollution. Among them, platinum group metals (Pt(IV), Pd(II), Rh(III)) and heavy metal ions (e.g., Cr^{3+} , Ni^{2+} , Cu^{2+} , Zn^{2+} , Cd^{2+}) are most representative. The presence of impurities such as metal ions, organic by-products, isomers and solvent residues can trigger electrical property fluctuations and interfacial defects, directly affecting device reliability.^{144,145} Given the demands for high integration, complex process systems, and ultra-trace impurity control, traditional techniques such as distillation, crystallization, extraction, and membrane separation still have limitations in terms of selectivity, energy consumption, and environmental sustainability.¹⁴⁶ Owing to their exceptional molecular designability, tunable polarity, and diverse solvation mechanisms (ion exchange, hydrogen bonding, π - π interaction, hydrophobic association, etc.), ILs have emerged as versatile tools for selective and efficient separation in electronic chemical production.^{147,148}

This section outlines recent advances in IL-enabled separations, focusing on two major categories: (1) metal ions and organic intermediates, and (2) hydrocarbon feedstocks and complex impurities. The corresponding applications, separation mechanisms, and design strategies are discussed, along with industrial demonstrations and emerging trends.

4.1 ILs for metal ion extraction and organic intermediate separation

With regard to the separation of metal ions, ILs have been demonstrated to selectively capture target ions by means of a number of mechanisms, including but not limited to chelation, ion exchange and hydrogen bond complexation. For metal ion extraction, ILs enable high-efficiency separation *via* chelation, ion exchange, and specific solute-IL interactions. For example, imidazolium-based ILs achieve over 90% extraction efficiency for platinum-group metals (Pt(IV), Pd(II), Rh(III)) at elevated temperatures with rapid equilibrium kinetics (Fig. 7a).¹⁴⁹ Cao studied the performance of 1-butyl-3-methylimidazolium hydrogen sulfate ([Bmim][HSO₄]) and sulfonated functionalized imidazolium salts ([PrSO₃Hmim][HSO₄]) in the removal of heavy metals from electroplating sludge. The results showed that the

latter achieved removal efficiencies of 99.8%, 90.1%, and 91.3% for Cr, Ni, and Cu, respectively, significantly higher than non-functionalized ILs (Fig. 7b).¹⁵⁰

The incorporation of chelating functional groups into ILs structures significantly enhances their affinity for metal ions, thereby enabling highly selective and efficient extraction. This design strategy has led to the development of task-specific ILs that can be immobilized within solid matrices, such as polymeric resins, creating hybrid systems that combine the molecular tunability of ILs with the operational ease and recyclability of solid sorbents. Such IL-impregnated resins have shown remarkable performance in the selective recovery of metal ions like Zn^{2+} from complex aqueous environments, underscoring the growing potential of IL-based composite materials in advanced separation technologies.¹⁵⁴ Moreover, ammonium- and phosphonium-based ILs functionalized with tailor-made anions have exhibited strong extraction capabilities for a range of heavy metals, including cadmium, copper, and zinc, demonstrating the versatility of IL chemistry in addressing diverse separation challenges.¹⁵⁵ Importantly, the efficiency of these systems is influenced not only by the chemical nature of the IL but also by operational factors such as contact time, suggesting opportunities for process optimization. These findings collectively affirm the strategic advantage of ILs in the design of next-generation, high-performance separation platforms for critical metal recovery.¹⁵⁶

In the synthesis pathway of bioelectrochemicals, small-molecule organic acids such as lactic, succinic, acetic and nicotinic acids are widely used as functional precursors for the preparation of electronic-grade materials (e.g., polymer dielectrics, photoresist precursors or functional electrode additives). Consequently, the meticulous extraction of these precursors from reaction residues or fermentation broths is of significant value in enhancing yield and purity.¹⁵² Taking nicotinic acid as an example, [C₆mim][ClO₄] exhibits excellent separation performance during its extraction, where the main driving forces include hydrogen bond donation from the IL cation to the carboxyl group of nicotinic acid, along with synergistic effects of π - π stacking and hydrophobic interactions.¹⁵⁷ Fan achieved a recovery rate of 91.7% for nicotinic acid using 0.1 mol L⁻¹ HCl desorption, and further demonstrated the multiple recycling capability of the IL system after drying at 70 °C for 4 h. In the naphthenic acid system, Geng *et al.* employed imidazolium carbonate-based ILs ([Emim]₂CO₃, [Emim]NO₃, [Emim]HSO₄) to achieve efficient extraction under conditions of 40 °C, an IL-to-oil mass ratio of 0.010, a stirring speed of 500 rpm, and a contact time of 1 h, with efficient IL regeneration accomplished *via* hydrochloric acid reflux (Fig. 7c).¹⁵¹

Further studies on process integration indicated that ILs combined with liquid membrane separations, such as spiral module supported liquid membranes (SLM) and emulsion liquid membranes (ELM), exhibit outstanding performance in the separation of lactic acid and succinic acid, as illustrated in the schematic diagram in Fig. 7d. Marták *et al.* utilized a Cyphos IL-104 loaded liquid membrane system supported by PTFE, achieving a high mass transfer coefficient and stable membrane



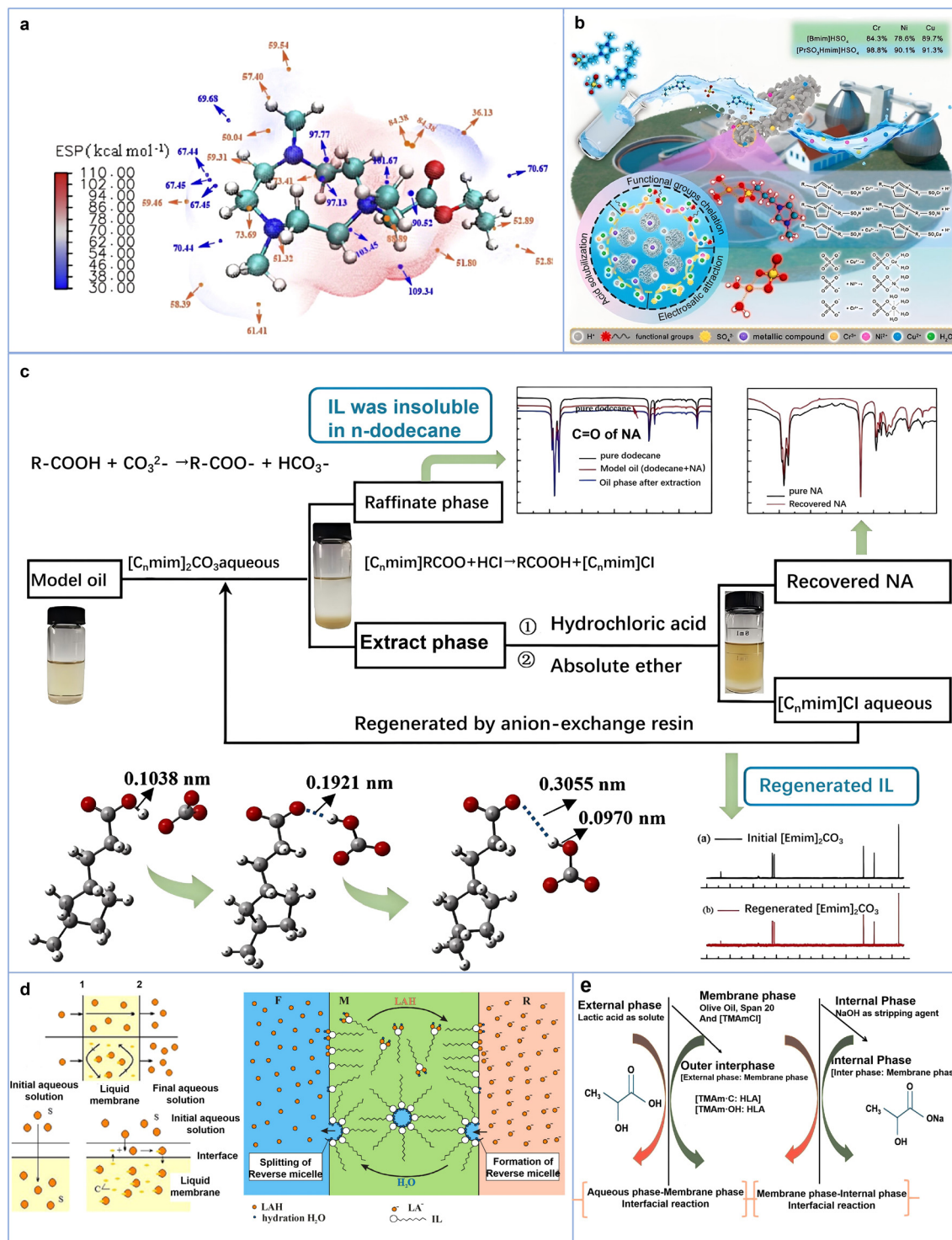


Fig. 7 IL-mediated separation mechanisms: (a) esp map and extraction of Au(III), Pd(II), Pt(IV);¹⁴⁹ (b) removal of Cr, Ni, and Cu;¹⁵⁰ (c) [Emim]-based extraction mechanism;¹⁵¹ (d) IL-SLM separation;¹⁵² (e) reverse micelle-driven transport process.¹⁵³ Copyright 2022, MDPI and 2008, 2021, 2022, ELSEVIER.

operation for 5.3 days during lactic acid recovery.^{158,159} The ILEM system developed by Khan *et al.*, using [Tmam][Ac] as both emulsifier and carrier, achieved a lactic acid separation efficiency of 94.5%, further confirming the process compatibility of ILs in liquid membrane systems (Fig. 7e).¹⁵³ In the recovery of carboxylic acids and bio-fermentation products, ILs achieve high-

selectivity and high-purity extraction through synergistic mechanisms including hydrogen bond complexation, hydrophobic association, and reverse micelle formation. The integration of liquid membrane technology is expected to further enhance the mass transfer efficiency and process sustainability of IL-based separation systems.



4.2 ILs for hydrocarbon separation and complex impurity removal

In the manufacture of high-end electronic chemicals (e.g. electronic-grade solvents, dielectric monomers, photoresist materials, etc.), light hydrocarbon compounds are frequently incorporated into the system as synthetic precursors, diluents, or reaction by-products. Furthermore, some of the hydrocarbons are derived from petrochemical feedstock deep-processing or residues from polymerisation reactions. Typical hydrocarbon impurities include C1–C4 grade straight chain paraffins (e.g. methane, ethane, propane), unsaturated hydrocarbons (e.g. ethylene, propylene, butyne), isomeric alkanes (e.g. isobutane), and alkylation by-products. While these small-molecule hydrocarbons possess inherent value in the pre-fabrication stage of electronic materials, effective separation during subsequent product refinement is imperative to avoid significant interference with the purity regulation of key components, the induction of undesirable side reactions, or degradation of device performance.¹⁶⁰ Conventional cryogenic distillation and extractive distillation are limited in their application to green separation technologies due to their high energy consumption and issues related to organic solvent volatilization. ILs, by virtue of their π -electron-rich cations and tunable anions, can selectively solubilize unsaturated hydrocarbons *via* π - π and van der Waals interactions.^{161,162} In olefin/paraffin separation studies, numerous works have demonstrated that imidazolium-based ILs can effectively enhance ethylene solubility in ethylene/ethane systems, and that separation performance can be further optimized by tuning the cation side-chain length and anion charge density.¹⁶³ Beyond hydrocarbons, ILs are also effective in removing phenolic compounds, dyes, herbicides, and complex metal–organic contaminants from electronic-grade solvents.¹⁶⁴ By designing ILs with specific functional groups, selective extraction of complex pollutants can be achieved, and coupling with technologies such as membrane separation and adsorption can enhance the overall process's flexibility and environmental friendliness.¹⁶⁵

As discussed above, ILs offer a transformative platform for the separation and purification of ECs, addressing key challenges associated with impurity control, molecular selectivity, and process sustainability.¹⁶⁶ Their unique physicochemical properties, including tunable polarity, strong solvation capabilities, and diverse interaction mechanisms such as hydrogen bonding, π - π stacking, ion exchange, and hydrophobic association, enable highly efficient and selective extraction of metal ions, organic acids, and hydrocarbon species. Moreover, the integration of ILs into advanced systems, such as task-specific resins and liquid membranes, further enhances their applicability in complex separation scenarios. These advances not only improve the purity and stability of electronic-grade materials but also align with the growing demand for green and energy-efficient processes. As research continues to optimize IL structures and recovery strategies, IL-based separation technologies are poised to play a critical role in the development of next-generation EIMs and sustainable chemical manufacturing.

5 ILs-enabled electronic devices

Building upon the previous discussions on the regulatory mechanisms of ILs in semiconductor nanomaterials, this section will focus on the value of ILs in electronic devices, particularly their applications in addressing key challenges in the interface engineering of high-performance electronic devices.

5.1 Electric double layer modulation in field-effect transistors

FETs, as the core components in electronic devices, function based on the core mechanism of gate voltage control over carrier distribution, modulating the semiconductor channel's band structure and carrier concentration through the gate electric field, thereby dynamically controlling the formation and cutoff of conductive channels to achieve low-power, high-sensitivity current-switching functionality (Fig. 8a).²² This physical process is primarily achieved through the formation of the EDL at the electrolyte-electrode interface.¹⁶⁷ Effective gate control of the FETs requires meeting four key criteria: (1) the distance between the gate and the molecules should be minimized to ensure that the electric field can directly and effectively act; (2) achieving high electrostatic coupling strength, which can be quantified by the ζ parameter;¹⁸ (3) the dielectric layer must have a high dielectric constant (ϵ_r) to enhance charge accumulation capability;¹⁶⁸ and (4) suppressing the shielding effect of the source-drain electrodes on the gate electric field.^{168–170}

ILs, due to their exceptionally high field-effect coupling capabilities, are particularly well-suited for the ideal gate modulation described above. When a gate voltage is applied, compared to traditional electric double layer interfaces, the cations and anions in ionic liquids can form an ultrathin EDL at the electrode-semiconductor interface under very low voltages.¹⁷⁴ The EDL formed by ILs at the electrode-semiconductor interface can induce carrier concentrations on the order of 10^{14} – 10^{15} cm⁻². Compared to traditional solid-state dielectric layers, the EDL of ILs allows for effective modulation at low gate voltages (<2 V), significantly reducing switching power consumption.^{175–178}

The thickness of the EDL in ILs can be tuned to the sub-nanometer scale through ion size. For example, [Bmpy][FAP], the EDL thickness is only 8 Å, significantly lower than the traditional solid-state gate (about 5 nm), thus generating a higher electric field strength under the same voltage.¹⁷⁹ The conductivity of the *meta*-benzenethiol (*meta*-BT) molecular junction, gated by [Bmim][PF₆], decreases by two orders of magnitude at a -0.4 V gate voltage compared to 0 V (Fig. 8b), confirming the precise tuning capability of ILs over destructive quantum interference (DQI) effects.²⁵ Similarly, the 2,4-TP-SAC molecule achieves a ~100-fold on/off ratio within the range of -0.6 V to -0.4 V (vs. Ag/AgCl) and -0.4 V to 1.4 V (Fig. 8c), verifying the feasibility of continuous IL modulation of the DQI effect to construct high-performance monomolecular devices.²⁶ Additionally, a vertical molecular tunneling field-effect device constructed using pseudo-*para*-



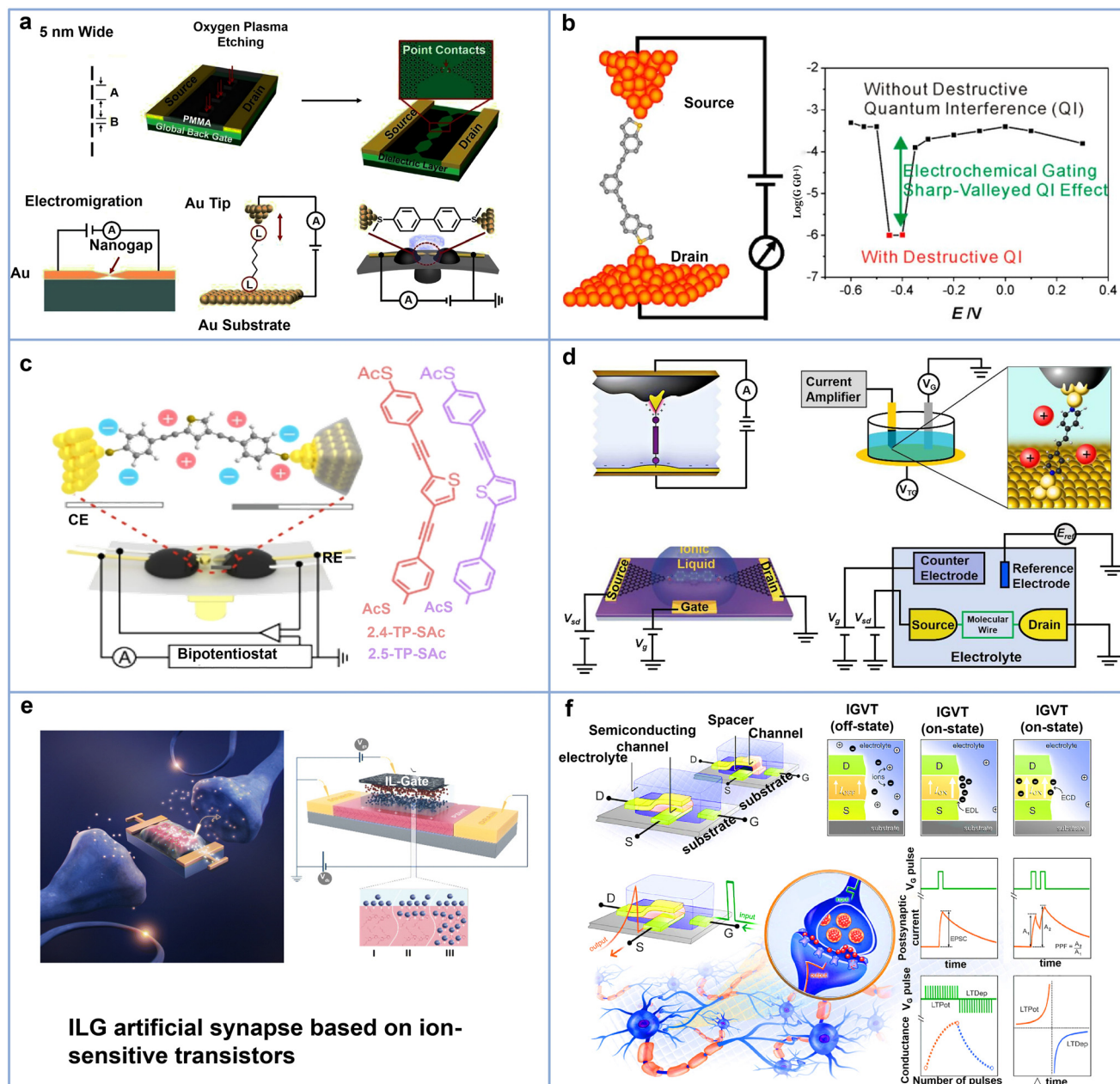


Fig. 8 IL-enabled electronic devices: (a) ILG single-molecule junctions;²² (b) *para/meta*-BT conductance profiles;²⁵ (c) molecular structures of 2,5-/2,4-TP-SAc;²⁶ (d) IL-assisted single-molecule transport characterization;¹⁷¹ (e) ILG artificial synapse based on ion-sensitive transistors;¹⁷² (f) IL-gated neuromorphic array for synaptic modulation.¹⁷³ Copyright 2024, ELSEVIER; 2023, WILEY and 2019, Springer Nature.

diphenylacetylene-*p*-(2,2)-cyclohexane (PCP) achieved a ~ 330 on/off current ratio based on an ILG.¹⁸⁰ The vertical molecular tunneling field-effect devices (such as the PCP structure) achieved a ~ 330 on/off ratio through ILG, further extending the application boundaries of ILs in nanoscale electric field modulation.^{180,181}

In addition, ILs high dielectric constant property greatly improves the efficiency of the interfacial charge response. This is evident in various experimental systems. In dynamic junctions (*e.g.* platinum to electrode systems), the EDL formed after applying a voltage can effectively control carrier transport in single-molecule junctions. In static junctions

(*e.g.* graphene dot electrode systems), the gate-induced EDL can finely regulate the alignment of molecular orbitals with the electrodes Fermi energy levels. In particular, in four-electrode dynamic junction systems with an Ag/AgCl reference electrode, the high-dielectric-constant IL enables the potential applied to the working electrode to modulate its energy levels almost exclusively, achieving extremely high modulation efficiency and significantly reducing the effects of electrode polarisation and solution IR pressure drop (Fig. 8d).¹⁷¹

Recent advances in research, particularly in the field of ion-gated transistors (IGTs) and their derivatives (*e.g.* emerging ion-



gated vertical transistors (IGVTs)), have highlighted the great potential of ILG in optimising the performance of a wide range of low-dimensional semiconductor devices. IGVTs can achieve the long- and short-term enhancement properties required for neuromorphic computation at brain-like processing rates and low power consumption, as well as optimising plasticity by enabling the programmable tuning of the doping mechanism (electrochemical/electrostatic) and time resolved properties through the precise control of gating parameters (Fig. 8e).¹⁷² Notably, the IGVTs architecture demonstrates faster data processing speeds and higher storage densities than conventional devices under low-voltage and low-power operation. It also successfully realises multimodal (vision, touch, taste and hearing) bionic sensory integration, providing a powerful technological route for developing high-performance, multifunctional neuromorphic systems (Fig. 8f).¹⁷³

5.2 ILs for flexible and transparent wearable electronics

Wearable electronic devices are emerging as foundational components of next-generation electronic information systems due to their inherent flexibility, environmental adaptability, and portability.^{182,183} However, they pose demanding material requirements, including optical transparency, mechanical flexibility, and scalable processability, criteria often unmet by conventional substrates.¹⁸⁴ For instance, polyimide (PI), though thermally robust and mechanically strong, suffers from poor visible light transmittance due to its intrinsic chromophores structure. Attempts to improve transparency through chemical modification or nanoparticle doping often compromise flexibility or introduce processing complexity.^{185–188}

ILs offer a transformative solution. When integrated into PI systems, ILs, especially those bearing trifluoromethyl or large anions like [TFSI][−], simultaneously enhance optical transparency, flexibility, and thermal conductivity.¹⁸⁹ IL-modified Kapton film shows an increase in transmittance from 76% to 84.8% at 550 nm, while IL-modified Kapton film shows an increase in transmittance from 76% to 84.8% at 550 nm, while cutoff wavelength (λ_{cutoff}) decreases from 411 nm to 373 nm.¹⁸⁴ Furthermore, ILs combine with the PI side chains through hydrogen bonding, significantly improving the material's flexibility. The L1/CPI sample with 8 wt% ILs achieves a fracture elongation of 108.4% (pure CPI is only 2.9%), and the tensile modulus decreases from 2929.6 MPa to 844.6 MPa, enabling the transition from brittleness to toughness.^{190,191} The molecular structure of ILs plays a decisive role in their performance modulation effects. Smaller cations (such as [Emim]⁺) are more effective than [Bmim]⁺ in reducing molecular stacking density, while larger anions (such as [TFSI][−]) further enhance transparency and flexibility by increasing free volume.^{192,193} In terms of industrial potential, [Emim][TFSI] is preferred due to its low commercial cost. A PI-based graphite film with 5.0 wt% ILs achieves a thermal conductivity of 770 W m^{−1} K^{−1} (1.53 times that of pure PI), and the degree of graphitization increases to 88%.¹⁹⁴ In modification, [Bmim][Cl] significantly enhances

hydrophilicity through thermally induced phase separation, making it suitable for low-cost film processing.

The functionalized applications of ILs are not limited to polyimide modification; their multifunctional integration in flexible electronic devices (FED) also demonstrates unique value. For example, ILs containing hydrophobic FAP anions modify Nafion membranes, enhancing ionic conductivity by more than three times; self-crosslinking PILs, when used in graphene oxide/carbon nanotube composite films, impart the material with ultra-flexibility (bending radius <1 mm) and high conductivity (>1000 S cm^{−1}).^{195,196} It is worth noting that ILs also play a key role in the development of soft ionic conductors.¹⁹⁷ Traditional hydrogels, while exhibiting high stretchability (strain >1000%) and biocompatibility, suffer from poor environmental stability, freezing and cracking at low temperatures, and the introduction of conductive fillers often leads to a decrease in transparency (transmittance <50%).^{198,199} By introducing inorganic salts (e.g., LiCl) or ILs, new ionic conductors can be developed. Yang developed a biomimetic agarose/polyacrylamide hydrogel containing Li⁺, achieving a tensile strength of 0.22 MPa while maintaining 90% transparency.²⁰⁰ The Sun group prepared a [Emim][Cl]-based double-helix structure ion gel *via* photopolymerization, exhibiting fast response (<100 m s^{−1}) and high sensitivity (GF = 2.1).²⁰¹ In particular, the plasticizing effect of ILs can break through the performance limits of traditional materials. The [Bmim][NTf₂]-based ion gel enhances the fracture elongation to 1200% while maintaining 90% transparency by reducing the polymer chain crystallinity, with a triboelectric output density of 12.3 $\mu\text{C m}^{-2}$.²⁰² To address the device interface adhesion issue, Bowen's group prepared ion gels *via* photopolymerization of acrylic acid in [Bmim][TfO], achieving stable adhesion with the substrate through hydrogen bonds/van der Waals forces between the ILs polar groups (trifluoromethyl, sulfonyl) and the substrate, with a peel strength >15 N m^{−1} at 90% transparency.²⁰³ Advanced design strategies such as double-network, double-crosslinked single-network, and supramolecular assembly have further expanded the performance envelope of IL-based ion gels. For instance, [Emim][TFSI]-based DN gels maintain high fracture energy (>5000 J m^{−2}) across a wide temperature range, while supramolecular systems featuring hydrogen-bonded nanoscale channels simultaneously achieve high conductivity and transparency. The solvation and plasticization effects of ILs also enable novel energy dissipation mechanisms and dynamic mechanical reinforcement in wearable sensing platforms.²⁰⁴

In addition, ILs are also applied in reshaping the paradigm of microelectronics patterning. In the field of scanning probe lithography, PIL films exhibit exceptional field-responsive characteristics. The local electric field of an atomic force microscope tip can induce surface morphology reconstruction of PIL under ultra-low voltages (≤ 5 V), reducing energy consumption by 80% compared to traditional insulating polymers. Through bias voltages (1–5 V) and environmental humidity (30–80% RH) synergistic regulation, precise fabrication of feature sizes ranging from 50 to 200 nm is achieved. This is



attributed to the electro-mechanical effect driven by ion migration, which overcomes the dielectric breakdown limit.²⁰⁵ *In situ* photoelectron microscopy based on [Emim][TFSI] shows that ion rearrangement and electrostatic response exhibit a critical thickness effect (>3 molecular layers) and a temperature dependence (activation energy ≈ 0.45 eV), providing a mechanistic basis for designing patterned ionic circuits.²⁰⁶

To address the bottlenecks in functional oxide nanofabrication, RTIL-mediated oxide scanning probe lithography offers a universal solution. Through constructing an IL adsorption layer on substrates such as steel and silicon, the induced voltage is reduced to -2 V, achieving 85 nm resolution for iron oxide patterning, while precisely controlling the $\text{Fe}^{2+}/\text{Fe}^{3+}$ chemical state ratio and sub-2 nm surface roughness. This technology, leveraging the adaptability of ILs to multi-phase solid surfaces, has been extended to 12 metal substrates, promoting the development of corrosion-resistant microelectrodes and neuromorphic arrays.²⁰⁷ These breakthroughs collectively establish ILs as a core medium for cross-scale patterning. From green direct-write processes replacing toxic solvents, decoding interfacial ion transport, to on-demand customization of oxide electronic properties, the ultimate aim is the integration of three-dimensional microelectronic systems on a single substrate: incorporating ion-conductive paths, memristive oxides, and polymer sensor units into a single process, opening new pathways for advanced packaging and wearable electronics.

In summary, ILs exhibit extraordinary potential as multifunctional modulators in the design and fabrication of next-generation flexible and wearable electronic devices and patterning. By virtue of their unique physicochemical properties, such as low volatility, high ionic conductivity, structural tunability, and strong intermolecular interactions, ILs enable simultaneous enhancement of optical transparency, mechanical flexibility, thermal conductivity, and interfacial adhesion in polymer-based systems. Their incorporation into polyimides and ion gels not only overcomes the intrinsic trade-offs between transparency and toughness in conventional materials but also facilitates the development of highly stretchable, transparent, and responsive ionic conductors suitable for wearable sensing, energy harvesting, and soft electronics. Advanced molecular designs, including supramolecular networks and dual crosslinking strategies, further unlock new paradigms for IL-integrated material systems with programmable mechanical and electronic properties. These attributes collectively position ILs as a critical material platform for the scalable engineering of multifunctional, durable, and intelligent wearable technologies.

6 Conclusions and outlooks

EIMs constitute the foundational pillars of modern information technology, playing a strategic role in enabling advancements in integrated circuits, sensors, display technologies, and communication systems. As the digital era accelerates toward ubiquitous connectivity and intelligent systems, the demand for

materials with enhanced electronic, optical, and mechanical performance continues to intensify. While conventional materials such as silicon, gallium nitride, and graphene have driven remarkable progress, limitations in processability, tunability, and multifunctional integration still pose significant constraints. ILs have emerged as promising candidates for next-generation material design, offering unprecedented opportunities for fine-tuned modulation and functional enhancement.

This review systematically delineates the multifaceted roles of ILs in advancing EIMs across several dimensions. First, ILs enable the formation of nanometer-scale electric double layers, allowing for ultra-low-voltage, high-efficiency carrier modulation, thereby enhancing the performance of FETs and related architectures. Second, ILs exhibit superior chemical separation capabilities, aiding the purification of electronic-grade materials by efficiently removing ionic and organic contaminants, which is an essential prerequisite for high-performance device fabrication. Third, ILs not only facilitate performance tuning in traditional semiconductors but also unlock new functionalities in frontier domains such as flexible electronics, quantum devices, and energy-efficient storage systems.

Despite these advances, the application of ILs in EIMs remains in its infancy, and several fundamental and practical challenges must be addressed. Chief among them are: (1) the high synthesis cost, ecological safety concerns, and limited recyclability of many ILs, which restrict their scalability and sustainability; (2) the need for comprehensive studies on long-term stability, compatibility, and interfacial interactions between ILs and diverse EIMs platforms, particularly under harsh operational conditions; (3) industrial bottlenecks such as uniform dispersion, large-scale synthesis, and reproducible doping control, which impede their transition from lab-scale demonstrations to commercial integration.

Looking ahead, the transformative potential of ILs in EIMs lies in the convergence of materials chemistry, device engineering, and sustainable manufacturing. Future research should emphasize the development of low-cost, biodegradable, and task-specific ILs with programmable molecular architectures. Cross-disciplinary approaches combining supramolecular design, machine learning-assisted IL screening, and scalable processing techniques will be critical in driving breakthroughs. Furthermore, integrating ILs with emerging material systems, such as 2D semiconductors, perovskites, and organic-inorganic hybrids, may open new frontiers in reconfigurable electronics, neuromorphic computing, and soft robotics. Ultimately, realizing the full potential of ILs in EIMs will not only enable performance upgrades but also reshape the paradigms of material innovation in the digital age.

Data availability

No primary research results, software or code have been included and no new data were generated or analysed as part of this review.



Author contributions

Mengyue Li: data curation and writing-original draft preparation. He Bin: conceptualization and methodology, writing-review and editing. Yangyang Jiang: data curation. Ruirui Wang: software and writing-review and editing. Cunliang Gan: data curation. Fengqi Ji: data curation. Yao Li: data curation. Ruixia Liu: supervision, project administration, and acquisition, writing-review and editing.

Conflicts of interest

All authors declare that they have no conflicts of interest.

Acknowledgements

This project was supported by the Excellent Young Scholars (22222813), National Natural Science Foundation of China (No. 22408376), the Postdoctoral Fellowship Program of CPSF (GZC20232700).

References

- 1 A. Raja Santhi and P. Muthuswamy, Industry 5.0 or industry 4.0S? Introduction to industry 4.0 and a peek into the prospective industry 5.0 technologies, *Int. J. Interact. Des. Manuf.*, 2023, **17**, 947–979.
- 2 X. Beyrich-Graf and G. Seltensperger, Quality aspects in production of electronic grade chemicals in multipurpose plants, *Chimia*, 2018, **72**, 130–132.
- 3 G. Lopez, D. Keiner, M. Fasihi, T. Koiranen and C. Breyer, From fossil to green chemicals: Sustainable pathways and new carbon feedstocks for the global chemical industry, *Energy Environ. Sci.*, 2023, **16**, 2879–2909.
- 4 C. Lécuyer, Driving semiconductor innovation: moore's law at fairchild and intel, *Enterp. Soc.*, 2020, **23**, 133–163.
- 5 R.-H. Jin, X. Yuan and Q.-S. Li, Research and industrial applications of key technologies of the manufacture of electronic chemicals for chips, *Zhongguo Kexue: Huaxue*, 2023, **53**, 1853–1865.
- 6 Q. Li, J. Li, G. Li, W. Zhao, X. Zhao and T. Mu, The electrochemical stability of ionic liquids and deep eutectic solvents, *Sci. China: Chem.*, 2016, **59**, 571–577.
- 7 S. Ravula, N. E. Larm, M. A. Mottaleb, M. P. Heitz and G. A. Baker, Vapor pressure, mapping of ionic liquids and low volatility fluids using graded isothermal thermogravimetric analysis, *ChemEngineering*, 2019, **3**, 42.
- 8 B. D. Rabideau, K. N. West and J. H. Davis, Making good on a promise: Ionic liquids with genuinely high degrees of thermal stability, *Chem. Commun.*, 2018, **54**, 5019–5031.
- 9 Y. Pei, Y. Zhang, J. Ma, M. Fan, S. Zhang and J. Wang, Ionic liquids for advanced materials, *Mater. Today Nano*, 2022, **17**, 100159.
- 10 B. He, S. Zhong, K. Li, S. Wei, M. Li and R. Liu, Ionic liquids: The emerging “cardiotonic” for photocatalytic materials, *Coord. Chem. Rev.*, 2025, **529**, 216461.
- 11 J. Chen, K. Dong, L. Liu, X. Zhang and S. Zhang, Anti-electrostatic hydrogen bonding between anions of ionic liquids: A density functional theory study, *Phys. Chem. Chem. Phys.*, 2021, **23**, 7426–7433.
- 12 B. He, W. Zhao, W. Li, C. Wei and J. Sun, Ionic liquids microenvironment modulates the interface properties of gC₍₃₎N₍₄₎ for boosting the performance of photodegradation and infected wound-healing therapy, *Small Methods*, 2024, **8**, e2301378.
- 13 Y. Pan, X.-Q. Li, G.-Y. Duan, J. Fang and B.-H. Xu, Microenvironment regulation for high-performance acidic CO₂ electroreduction on poly(ionic liquid)-modified Cu surface, *Appl. Catal., B*, 2025, **361**, 124681.
- 14 K. Bryant, E. Hammond-Pereira and S. Saunders, Ionic liquid aggregation mechanism for nanoparticle synthesis, *J. Phys. Chem. B*, 2021, **125**, 253–263.
- 15 J. Bassaler, J. Mehta, I. Abid, L. Konczewicz, S. Juillaguet, S. Contreras, S. Rennesson, S. Tamariz, M. Nemoz, F. Semond, J. Pernot, F. Medjdoub, Y. Cordier and P. Ferrandis, Al rich AlGaN channel high electron mobility transistors on silicon: A relevant approach for high temperature stability of electron mobility, *Adv. Electron. Mater.*, 2024, **11**, 2400069.
- 16 D. Akinwande, C. Huyghebaert, C. H. Wang, M. I. Serna, S. Goossens, L. J. Li, H. P. Wong and F. H. L. Koppens, Graphene and two dimensional materials for silicon technology, *Nature*, 2019, **573**, 507–518.
- 17 T. K. Nguyen, S. Aberoumand and D. V. Dao, Advances in Si and SiC materials for high performance supercapacitors toward integrated energy storage systems, *Small*, 2021, **17**, e2101775.
- 18 P. Li, L. Zhao, C. Zhao, H. Ju, Q. Gao, W. Si, L. Cheng, J. Hao, M. Li, Y. Chen, C. Jia and X. Guo, Single molecule nano-optoelectronics: Insights from physics, *Rep. Prog. Phys.*, 2022, **85**, 086401.
- 19 C. H. Kim, Nanostructured graphene: An active component in optoelectronic devices, *Nanomaterials*, 2018, **8**, 328.
- 20 Y. Nakayama, S. Kera and N. Ueno, Photoelectron spectroscopy on single crystals of organic semiconductors: Experimental electronic band structure for optoelectronic properties, *J. Mater. Chem. C*, 2020, **8**, 9090–9132.
- 21 I. Piquero-Zulaica, J. Lobo-Checa, Z. M. A. El-Fattah, J. E. Ortega, F. Klappenberger, W. Auwärter and J. V. Barth, Engineering quantum states and electronic landscapes through surface molecular nanoarchitectures, *Rev. Mod. Phys.*, 2022, **94**, 045008.
- 22 L. Zhou, M. Zhang, Y. Huo, L. Bai, S. He, J. Wang, C. Jia and X. Guo, Application of ionic liquids in single-molecule junctions: Recent advances and prospects, *Green Energy Environ.*, 2024, **9**, 1784–1801.
- 23 J. Lee, M. J. Panzer, Y. He, T. P. Lodge and C. D. Frisbie, Ion Gel Gated Polymer Thin-Film Transistors, *J. Am. Chem. Soc.*, 2007, **129**, 4532–4533.
- 24 S. Z. Bisri, C. Piliego, J. Gao and M. A. Loi, Outlook and emerging semiconducting materials for ambipolar transistors, *Adv. Mater.*, 2014, **26**, 1176–1199.



- 25 B. Huang, X. Liu, Y. Yuan, Z. W. Hong, J. F. Zheng, L. Q. Pei, Y. Shao, J. F. Li, X. S. Zhou, J. Z. Chen, S. Jin and B. W. Mao, Controlling and observing sharp-valleyed quantum interference effect in single molecular junctions, *J. Am. Chem. Soc.*, 2018, **140**, 17685–17690.
- 26 A. Daaoub, S. Sangtarash, X. Li, Y. Tang, Q. Zou, H. Sadeghi, S. Liu, X. Huang, Z. Tan, J. Liu, Y. Yang, J. Shi, G. Meszaros, W. Chen, C. Lambert and W. Hong, Anti-resonance features of destructive quantum interference in single-molecule thiophene junctions achieved by electrochemical gating, *Nat. Mater.*, 2019, **18**, 364–369.
- 27 C. Jia, M. Famili, M. Carloti, Y. Liu, P. Wang, I. M. Grace, Z. Feng, Y. Wang, Z. Zhao, M. Ding, X. Xu, C. Wang, S.-J. Lee, Y. Huang, R. C. Chiechi, C. J. Lambert and X. Duan, Quantum interference mediated vertical molecular tunneling transistors, *Sci. Adv.*, 2018, **4**, eaat8237.
- 28 D. Vaquero, V. Clericò, J. Salvador-Sanchez, J. Quereda, E. Diez and A. M. Perez-Munoz, Ionic-Liquid gating in two-dimensional TMDs: The operation principles and spectroscopic capabilities, *Micromachines*, 2021, **12**, 1576.
- 29 P. Sadhukhan, M. S. Kim and J. M. Myoung, Synergetic mixed ionic liquid strategy for comprehensive defect passivation and increased carrier mobility in high-performance green perovskite light-emitting diodes, *Chem. Eng. J.*, 2024, **480**, 148–186.
- 30 G. Hwang, S. Lee, S. Lee, S. Cho and H. Yang, Interface engineering of 2D materials for highly performing electronic and energy devices, *ACS Appl. Electron. Mater.*, 2024, **6**, 4843–4854.
- 31 H. Du, X. Lin, Z. Xu Lin and D. Chu, Electric double-layer transistors: A review of recent progress, *J. Mater. Sci.*, 2015, **50**, 5641–5673.
- 32 T. Uwanno, T. Taniguchi, K. Watanabe and K. Nagashio, Electrically inert h-BN/bilayer graphene interface in all-two-dimensional heterostructure field effect transistors, *ACS Appl. Mater. Interfaces*, 2018, **10**, 28780–28788.
- 33 G. W. King'ori, C. N. M. Ouma, A. K. Mishra, G. O. Amolo and N. W. Makau, Two-dimensional graphene-HfS₂ van der waals heterostructure as electrode material for alkaline batteries, *RSC Adv.*, 2020, **10**, 30127–30138.
- 34 S. Li, J. Zhou, J. Xiong, S. Yang, J. Zhang, W. Fan and J. Li, Two dimensional CuInP₂S₆/h-BN/MoTe₂ van der waals heterostructure phototransistors with double gate control, *J. Mater. Chem. C*, 2025, **13**, 2378–2387.
- 35 L. Ma, Y. Wang and Y. Liu, Van der waals contact for two-dimensional transition metal dichalcogenides, *Chem. Rev.*, 2024, **124**, 2583–2616.
- 36 M. Nishinaka, I. Harada, K. Akaike, Q. Wei, Y. Koshiba, S. Horike and K. Ishida, Electrochemical charge-carrier modulation of carbon nanotubes using ionic liquids derived from organic superbases for stable thermoelectric materials, *Carbon*, 2024, **218**, 118667.
- 37 W. Fan, Y. Zhang, C.-Y. Guo and G. Chen, Toward high thermoelectric performance for polypyrrole composites by dynamic 3-phase interfacial electropolymerization and chemical doping of carbon nanotubes, *Compos. Sci. Technol.*, 2019, **183**, 107794.
- 38 B. Kumanek, G. Stando, P. Stando, K. Matuszek, K. Z. Milowska, M. Krzywiecki, M. Gryglas-Borysiewicz, Z. Ogorzalek, M. C. Payne, D. MacFarlane and D. Janas, Enhancing thermoelectric properties of single-walled carbon nanotubes using halide compounds at room temperature and above, *Sci. Rep.*, 2021, **11**, 8649.
- 39 S. Zhang, Y. Huang, L. Zhang, Y. Liu, Q. Miao, R. Liu, W. Zhao, Y. Diao and K. Dong, The advanced applications of ionic liquids in new energy, electronic information materials, and biotechnologies, *Green Chem.*, 2024, **26**, 9048–9074.
- 40 R. L. Vekariya, A review of ionic liquids: Applications towards catalytic organic transformations, *J. Mol. Liq.*, 2017, **227**, 44–60.
- 41 M. Armand, F. Endres, D. R. MacFarlane, H. Ohno and B. Scrosati, Ionic-liquid materials for the electrochemical challenges of the future, *Nat. Mater.*, 2009, **8**, 621–629.
- 42 Y. Ye, J. Rick and B.-J. Hwang, Ionic liquid polymer electrolytes, *J. Mater. Chem. A*, 2013, **1**, 2719–2743.
- 43 J. M. S. S. Esperança, J. N. C. Lopes, M. Tariq, L. M. N. B. F. Santos, J. W. Magee and L. P. N. Rebelo, Volatility of aprotic ionic liquids-A review, *J. Chem. Eng. Data*, 2010, **55**, 3–12.
- 44 D. M. Correia, L. C. Fernandes, P. M. Martins, C. García Astrain, C. M. Costa, J. Reguera and S. Lanceros Méndez, Ionic liquid polymer composites: A new platform for multifunctional applications, *Adv. Funct. Mater.*, 2020, **30**, 1616–3028.
- 45 J. Luczak, M. Paszkiewicz, A. Krukowska, A. Malankowska and A. Zaleska-Medynska, Ionic liquids for nano- and microstructures preparation. Part 1: Properties and multifunctional role, *Adv. Colloid Interface Sci.*, 2016, **230**, 13–28.
- 46 S. Wang, Q. X. Shi, Y. S. Ye, Y. Xue, Y. Wang, H. Y. Peng, X. L. Xie and Y. W. Mai, Constructing desirable ion-conducting channels within ionic liquid-based composite polymer electrolytes by using polymeric ionic liquid-functionalized 2D mesoporous silica nanoplates, *Nano Energy*, 2017, **33**, 110–123.
- 47 F. Paul, K. N. Manjunatha and S. Paul, Storing electronic information on semi-metal nanoparticles, *Mater. Adv.*, 2022, **3**, 5363–5374.
- 48 L. Meng, N. Xin, C. Hu, H. A. Sabea, M. Zhang, H. Jiang, Y. Ji, C. Jia, Z. Yan, Q. Zhang, L. Gu, X. He, P. Selvanathan, L. Norel, S. Rigaut, H. Guo, S. Meng and X. Guo, Dual-gated single-molecule field-effect transistors beyond Moore's law, *Nat. Commun.*, 2022, **13**, 1410.
- 49 X. Wang, C. Liu, Y. Wei, S. Feng, D. Sun and H. Cheng, Three-dimensional transistors and integration based on low-dimensional materials for the post-Moore's law era, *Mater. Adv.*, 2023, **63**, 170–187.
- 50 X. Xu, T. Guo, H. Kim, M. K. Hota, R. S. Alsaadi, M. Lanza, X. Zhang and H. N. Alshareef, Growth of 2D materials at the wafer scale, *Adv. Mater.*, 2022, **34**, e2108258.
- 51 R. Chen, Y. C. Li, J. M. Cai, K. Cao and H. B. R. Lee, Atomic level deposition to extend Moore's law and beyond, *Int. J. Extreme Manuf.*, 2020, **2**, 022002.



- 52 D. D. Sarma and P. V. Kamat, 2023 Nobel prize in chemistry: A mega recognition for nanosized quantum dots, *ACS Energy Lett.*, 2023, **8**, 5149–5151.
- 53 A. Al Bodour, N. Alomari, G. Perez Duran, S. Rozas, G. Iglesias Silva, S. Aparicio, A. Gutierrez Vega and M. Atilhan, Ionic liquids as multidimensional materials: A review from fundamentals to applications, *Energy Fuels*, 2025, **39**, 12791–12829.
- 54 C. R. Kagan, E. Lifshitz, E. H. Sargent and D. V. Talapin, Building devices from colloidal quantum dots, *Science*, 2016, **353**, 6302.
- 55 A. Mittelstadt, A. Schliwa and P. Klenovsky, Modeling electronic and optical properties of III-V quantum dots—selected recent developments, *Light: Sci. Appl.*, 2022, **11**, 17.
- 56 S. V. Dayneko, P. S. Samokhvalov, D. Lypenko, G. I. Nosova, I. A. Berezin, A. V. Yakimanskii, A. A. Chistyakov and I. Nabiev, A highly efficient white light emitting diode based on a two component polyfluorene/quantum dot composite, *Opt. Spectrosc.*, 2017, **122**, 12–15.
- 57 C. B. Murray, D. J. Norris and M. G. Bawendi, Synthesis and characterization of nearly monodisperse CdE (E = sulfur, selenium, tellurium) semiconductor nanocrystallites, *J. Am. Chem. Soc.*, 1993, **115**, 8706–8715.
- 58 V. I. Klimov, A. A. Mikhailovsky, Xu Su, A. Malko, J. A. Hollingsworth, C. A. Leatherdale, H. J. Eisler and M. G. Bawendi, Optical gain and stimulated emission in nanocrystal quantum dots, *Science*, 2000, **290**, 314–317.
- 59 T. Kim, K. H. Kim, S. Kim, S. M. Choi, H. Jang, H. K. Seo, H. Lee, D. Y. Chung and E. Jang, Efficient and stable blue quantum dot light-emitting diode, *Nature*, 2020, **586**, 385–389.
- 60 F. Endres, Physical chemistry of ionic liquids, *Phys. Chem. Chem. Phys.*, 2010, **12**, 1648–1648.
- 61 F. Bernardi, J. D. Scholten, G. H. Fecher, J. Dupont and J. Morais, Probing the chemical interaction between iridium nanoparticles and ionic liquid by XPS analysis, *Chem. Phys. Lett.*, 2009, **479**, 113–116.
- 62 G. S. F. J. Dupont, A. P. Umpierre, P. F. P. Fichtner and S. R. Teixeira, Transition-metal nanoparticles in imidazolium ionic liquids: Recyclable catalysts for biphasic hydrogenation reactions, *J. Am. Chem. Soc.*, 2002, **124**, 4228–4229.
- 63 V. A. Hernández, An overview of surface forces and the DLVO theory, *ChemTexts*, 2023, **9**, 10.
- 64 J. A. V. Butler, Theory of the stability of lyophobic colloids, *Nature*, 1948, **162**, 315–316.
- 65 E. Redel, M. Walter, R. Thomann, C. Vollmer, L. Hussein, H. Scherer, M. Kruger and C. Janiak, Synthesis, stabilization, functionalization and, DFT calculations of gold nanoparticles in fluorinated phases (PTFE and ionic liquids), *Chemistry*, 2009, **15**, 10047–10059.
- 66 L. L. Lazarus, C. T. Riche, N. Malmstadt and R. L. Brutchey, Effect of ionic liquid impurities on the synthesis of silver nanoparticles, *Langmuir*, 2012, **28**, 15987–15993.
- 67 F. Cao, L. Liu and L. Li, Short-wave infrared photodetector, *Mater. Today*, 2023, **62**, 327–349.
- 68 X. Yuan, N. Yan, S. A. Katsyuba, E. E. Zvereva, Y. Kou and P. J. Dyson, A remarkable anion effect on palladium nanoparticle formation and stabilization in hydroxyl-functionalized ionic liquids, *Phys. Chem. Chem. Phys.*, 2012, **14**, 6026–6033.
- 69 C. W. Scheeren, G. Machado, S. R. Teixeira, J. Morais, J. B. Domingos and J. Dupont, Synthesis and characterization of Pt(0) nanoparticles in imidazolium ionic liquids, *J. Phys. Chem. B*, 2006, **110**, 13011–13020.
- 70 G. S. Fonseca, G. Machado, S. R. Teixeira, G. H. Fecher, J. Morais, M. C. Alves and J. Dupont, Synthesis and characterization of catalytic iridium nanoparticles in imidazolium ionic liquids, *J. Colloid Interface Sci.*, 2006, **301**, 193–204.
- 71 M. Sobota, I. Nikiforidis, W. Hieringer, N. Paape, M. Happel, H. P. Steinruck, A. Gorling, P. Wasserscheid, M. Laurin and J. Libuda, Toward ionic-liquid-based model catalysis: Growth, orientation, conformation, and interaction mechanism of the [Tf₂N][−] anion in [Bmim][Tf₂N] thin films on a well-ordered alumina surface, *Langmuir*, 2010, **26**, 7199–7207.
- 72 C. Zhang, J. Y. Zheng, Y. S. Zhao and J. Yao, Organic core-shell nanostructures: Microemulsion synthesis and upconverted emission, *Chem. Commun.*, 2010, **46**, 4959–4961.
- 73 L. S. Ott, M. L. Cline, M. Deetlefs, K. R. Seddon and R. G. Finke, Nanoclusters in ionic liquids: Evidence for N-heterocyclic carbene formation from Imidazolium based ionic liquids detected by 2H NMR, *J. Am. Chem. Soc.*, 2005, **127**, 5758–5759.
- 74 R. A. G. Warr, Structure in confined room temperature ionic liquids, *J. Phys. Chem. C*, 2007, **111**, 5162–5168.
- 75 K. Ueno, A. Inaba, M. Kondoh and M. Watanabe, Colloidal stability of bare and polymer grafted silica nanoparticles in ionic liquids, *Langmuir*, 2008, **24**, 5253–5259.
- 76 E. Vanecht, K. Binnemans, S. Patskovsky, M. Meunier, J. W. Seo, L. Stappers and J. Fransaer, Stability of sputter-deposited gold nanoparticles in imidazolium ionic liquids, *Phys. Chem. Chem. Phys.*, 2012, **14**, 5662–5671.
- 77 J. Li, J. Liang, W. Wu, S. Zhang, K. Zhang and H. Zhou, AuCl₄-responsive self-assembly of ionic liquid block copolymers for obtaining composite gold nanoparticles and polymeric micelles with controlled morphologies, *New J. Chem.*, 2014, **38**, 2508–2513.
- 78 R. Jusoh, A. A. Jalil, S. Triwahyono, A. Idris, S. Haron, N. Sapawe, N. F. Jaafar and N. W. C. Jusoh, Synthesis of reverse micelle α -FeOOH nanoparticles in ionic liquid as an only electrolyte: Inhibition of electron-hole pair recombination for efficient photoactivity, *Appl. Catal., A*, 2014, **469**, 33–44.
- 79 J. C. Rubim, F. A. Trindade, M. A. Gelesky, R. F. Aroca and J. Dupont, Surface-enhanced vibrational spectroscopy of tetrafluoroborate 1-*n*-butyl-3-methylimidazolium (BMIBF₄) ionic liquid on silver surfaces, *J. Phys. Chem. C*, 2008, **112**, 19670–19675.
- 80 E. E. Zvereva, S. Grimme, S. A. Katsyuba, V. V. Ermolaev, D. A. Arkhipova, N. Yan, V. A. Miluykov, O. G. Sinyashin and A. Aleksandrov, Solvation and stabilization of palladium nanoparticles in phosphonium-based ionic liquids: A



- combined infrared spectroscopic and density functional theory study, *Phys. Chem. Chem. Phys.*, 2014, **16**, 20672–20680.
- 81 S. A. Katsyuba, E. E. Zvereva, N. Yan, X. Yuan, Y. Kou and P. J. Dyson, Rationalization of solvation and stabilization of palladium nanoparticles in imidazolium-based ionic liquids by DFT and vibrational spectroscopy, *ChemPhysChem*, 2012, **13**, 1781–1790.
 - 82 M. Zhao, L. Zheng, X. Bai, N. Li and L. Yu, Fabrication of silica nanoparticles and hollow spheres using ionic liquid microemulsion droplets as templates, *Colloids Surf., A*, 2009, **346**, 229–236.
 - 83 L. G. Wu, J. n. Shen, C. H. Du, T. Wang, Y. Teng and B. V. d. Bruggen, Development of AgCl/poly(MMA-co-AM) hybrid pervaporation membranes containing AgCl nanoparticles through synthesis of ionic liquid microemulsions, *Sep. Purif. Technol.*, 2013, **114**, 117–125.
 - 84 A. Serrà, E. Gómez, J. F. López-Barbera, J. Nogués and E. Vallés, Green electrochemical template synthesis of CoPt nanoparticles with tunable size, composition, and magnetism from microemulsions using an ionic liquid ([Bmim][PF₆]), *ACS Nano*, 2014, **8**, 4630–4639.
 - 85 N. Li, H. Zhang, F. Yan and Y. Luo, Carbon nanotube based high-current nanoscale air channel electronic devices for low power ultrafast electronics, *ACS Appl. Nano Mater.*, 2023, **6**, 18926–18933.
 - 86 Z. Zhang, Y. Ouyang, Y. Cheng, J. Chen, N. Li and G. Zhang, Size dependent phononic thermal transport in low-dimensional nanomaterials, *Phys. Rep.*, 2020, **860**, 1–26.
 - 87 Z. Chen, J. Chen, W. Liao, Y. Zhao, J. Jiang and C. Chen, Progress on a carbon nanotube field-effect transistor integrated circuit: State of the art, challenges, and evolution, *Micromachines*, 2024, **15**, 817.
 - 88 J. Zhou, Z. Liu, X. Liu, P. Pan, X. Zhan and Z. Liu, Silicon-Au nanowire resonators for high-Q multiband near-infrared wave absorption, *Nanotechnology*, 2020, **31**, 375201.
 - 89 Z. Huang, Z. Zhong, H. Wang, S. Lu, J. Wang, G. Liu, T. Wei, J. Yan, J. H. Min, W. L. Jeong, D. S. Lee, X. Cai, F. Xu, X. Chen, D. Cai, J. Wang and J. Kang, Enhanced emission of deep ultraviolet light-emitting diodes through using work function tunable Cu nanowires as the top transparent electrode, *J. Phys. Chem. Lett.*, 2020, **11**, 2559–2569.
 - 90 A. Magrez, J. W. Seo, R. Smajda, M. Mionie and L. Forro, Catalytic CVD synthesis of carbon nanotubes: Towards high yield and low temperature growth, *Materials*, 2010, **3**, 4871–4891.
 - 91 Y. Xu, H. Cao, Y. Xue, B. Li and W. Cai, Liquid-phase exfoliation of graphene: An overview on exfoliation media, techniques, and challenges, *Nanomaterials*, 2018, **8**, 942.
 - 92 M. M. R. Nayini and Z. Ranjbar, Carbon nanotubes: Dispersion challenge and how to overcome it, in *Handbook of Carbon Nanotubes*, Springer, 2022, pp. 1–52.
 - 93 L. Wang, L. X. Chang, L. Q. Wei, S. Z. Xu, M. H. Zeng and S. L. Pan, The effect of 1-N-alkyl chain of ionic liquids [C_n-mim]⁺[Br][−] (n = 2, 4, 6, 8) on the aspect ratio of ZnO nanorods: Syntheses, morphology, forming mechanism, photoluminescence and recyclable photocatalytic activity, *J. Mater. Chem.*, 2011, **21**, 15732.
 - 94 T. Li, Y. Chen, W. Hu, W. Yuan, Q. Zhao, Y. Yao, B. Zhang, C. Qiu and C. M. Li, Ionic liquid in situ functionalized carbon nanotubes as metal-free catalyst for efficient electrocatalytic hydrogen evolution reaction, *Nanoscale*, 2021, **13**, 4444–4450.
 - 95 E. C. Lopes Pereira, M. E. C. F. da Silva, K. Pontes and B. G. Soares, Influence of protonic ionic liquid on the dispersion of carbon nanotube in PLA/EVA blends and blend compatibilization, *Front. Mater.*, 2019, **6**, 00234.
 - 96 A. Cetinkaya, S. I. Kaya, M. Yence, F. Budak and S. A. Ozkan, Ionic liquid-based materials for electrochemical sensor applications in environmental samples, *Trends Environ. Anal. Chem.*, 2023, **37**, e00188.
 - 97 K. Zhang, T. Luo, H. Chen, Z. Lou and G. Shen, Au-nanoparticles-decorated Sb₂S₃ nanowire-based flexible ultraviolet/visible photodetectors, *J. Mater. Chem. C*, 2017, **5**, 3330–3335.
 - 98 T. Sun, J. Huo, Y. Xiao, L. Liu, B. Feng, X. Zhai, W. Wang and G. Zou, Atomic bonding-engineered heterogeneous integration of semiconductor nanowires by femtosecond laser irradiation for a miniaturized photodetector, *Appl. Surf. Sci.*, 2022, **575**, 151709.
 - 99 S. Chen, L. Li, X. Wang, W. Tian, X. Wang, D. M. Tang, Y. Bando and D. Golberg, Dense and vertically-aligned centimetre-long ZnS nanowire arrays: Ionic liquid assisted synthesis and their field emission properties, *Nanoscale*, 2012, **4**, 2658–2662.
 - 100 S. A. Al Kiey, H. K. Farag and S. Z. E. Abedin, Template-assisted electrodeposition of freestanding antimony, tin, and antimony-tin nanowire arrays from an ionic liquid, *J. Solid State Electrochem.*, 2024, **28**, 3285–3295.
 - 101 S. Thomas, J. Mallet, B. K. Bahuleyan and M. Molinari, Growth of homogeneous luminescent silicon-terbium nanowires by one-step electrodeposition in ionic liquids, *Nanomaterials*, 2020, **10**, 2390.
 - 102 H. Roohi, A.-r. Nowroozi, A. Ebrahimi and B. Makiabadi, Effect of CH₃CO functional group on the molecular and electronic properties of BN_{43zz} nanotube: A computational chemistry study, *J. Mol. Struct.: THEOCHEM*, 2010, **952**, 36–45.
 - 103 T. Morishita, H. Okamoto, Y. Katagiri, M. Matsushita and K. Fukumori, A high-yield ionic liquid-promoted synthesis of boron nitride nanosheets by direct exfoliation, *Chem. Commun.*, 2015, **51**, 12068–12071.
 - 104 R. Salehi and H. Roohi, Tuning the physicochemical properties of the single-walled boron nitride nanotube by covalent grafting of triazolium-based [MTZ][X_{1–3}] (X_{1–3} = NTf₂[−], TfO[−] and BF₄[−]) ionic liquids in the gas phase and solvent media: A quantum chemical approach, *J. Mol. Liq.*, 2019, **277**, 726–737.
 - 105 G. Sun and J. Bi, Scalable production of boron nitride nanosheets in ionic liquids by shear-assisted thermal treatment, *Ceram. Int.*, 2021, **47**, 7776–7782.
 - 106 H. Roohi and S. Khyrkah, Green chemical functionalization of single-wall carbon nanotube with methylimidazolium dicyanamide ionic liquid: A first principle computational exploration, *J. Mol. Liq.*, 2015, **211**, 498–505.



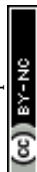
- 107 Y. Wang, X. Lin, G. Gao, J. Yu, Y. Wei, J. Gong, J. Sun, Z. L. Wang and Q. Sun, Tribotronic vertical field-Effect transistor based on van der waals heterostructures, *Adv. Funct. Mater.*, 2024, **34**, 2313210.
- 108 B. Wang, X. Wang, W. Lou and J. Hao, Reduced graphene oxides by microwave-assisted ionothermal treatment, *New J. Chem.*, 2012, **36**, 1684–1690.
- 109 S. Aldroubi, N. Brun, I. Bou Malham and A. Mehdi, When graphene meets ionic liquids: A good match for the design of functional materials, *Nanoscale*, 2021, **13**, 2750–2779.
- 110 J. Lu, J. Yang, J. Wang, A. Lim, S. Wang and K. P. Loh, One-pot synthesis of fluorescent carbon nanoribbons, nanoparticles, and graphene by the exfoliation of graphite in ionic liquids, *ACS Nano*, 2009, **3**, 2367–2375.
- 111 X. Wang and P. Wu, Aqueous phase exfoliation of two-dimensional materials assisted by thermoresponsive polymeric ionic liquid and their applications in stimuli-responsive hydrogels and highly thermally conductive films, *ACS Appl. Mater. Interfaces*, 2018, **10**, 2504–2514.
- 112 M. Lee, A. K. Roy, S. Jo, Y. Choi, A. Chae, B. Kim, S. Y. Park and I. In, Exfoliation of black phosphorus in ionic liquids, *Nanotechnology*, 2017, **28**, 125603.
- 113 C. R. Dean, L. Wang, P. Maher, C. Forsythe, F. Ghahari, Y. Gao, J. Katoch, M. Ishigami, P. Moon, M. Koshino, T. Taniguchi, K. Watanabe, K. L. Shepard, J. Hone and P. Kim, Hofstadter's butterfly and the fractal quantum Hall effect in moire superlattices, *Nature*, 2013, **497**, 598–602.
- 114 A. Khan, M. C. Parameshwara and R. Arya, Defects of quantum dot cellular automata computing devices: An extensive review, evaluation, and future directions, *Microprocess. Microsyst.*, 2023, **101**, 104912.
- 115 Y. Zheng, X. Lin, J. Li, J. Chen, Z. Song, Y. Gao, H. Wang, Z. Ye, H. Qin and X. Peng, Ultra-bright and energy-efficient quantum-dot LEDs by idealizing charge injection, *Appl. Phys.*, 2024, **6**, 09730.
- 116 N. Huo and G. Konstantatos, Recent progress and future prospects of 2D-based photodetectors, *Adv. Mater.*, 2018, **30**, e1801164.
- 117 K. Zhou, Z. Jia, X. Q. Ma, W. Niu, Y. Zhou, N. Huang, G. Ding, Y. Yan, S. T. Han, V. A. L. Roy and Y. Zhou, Manufacturing of graphene based synaptic devices for optoelectronic applications, *Int. J. Extreme Manuf.*, 2023, **5**, 042006.
- 118 X. Wang, H. Feng, Y. Wu and L. Jiao, Controlled synthesis of highly crystalline MoS₂ flakes by chemical vapor deposition, *J. Am. Chem. Soc.*, 2013, **135**, 5304–5307.
- 119 M. Yi and Z. Shen, A review on mechanical exfoliation for the scalable production of graphene, *J. Mater. Chem. A*, 2015, **3**, 11700–11715.
- 120 W. Xing, Y. Chen, X. Wu, X. Xu, P. Ye, T. Zhu, Q. Guo, L. Yang, W. Li and H. Huang, PEDOT: PSS-assisted exfoliation and functionalization of 2D nanosheets for high-performance organic solar cells, *Adv. Funct. Mater.*, 2017, **27**, 1701622.
- 121 A. Reina, X. Jia, J. Ho, D. Nezich, H. Son, V. Bulovic, M. S. Dresselhaus and J. Kong, Large area, few-layer graphene films on arbitrary substrates by chemical vapor deposition, *Nano Lett.*, 2008, **9**, 30–35.
- 122 J. Loomis and B. Panchapakesan, Dimensional dependence of photomechanical response in carbon nanostructure composites: A case for carbon-based mixed-dimensional systems, *Nanotechnology*, 2012, **23**, 215501.
- 123 X. Feng, W. Xing, H. Yang, B. Yuan, L. Song, Y. Hu and K. M. Liew, High-performance poly(ethylene oxide)/molybdenum disulfide nanocomposite films: Reinforcement of properties based on the gradient interface effect, *ACS Appl. Mater. Interfaces*, 2015, **7**, 13164–13173.
- 124 S. Aldroubi, E. Anglaret, I. Bou Malham, P. Hesemann, N. Brun and A. Mehdi, Long-term stable solid concentrated graphene dispersion assisted by a highly aromatic ionic liquid, *J. Colloid Interface Sci.*, 2023, **636**, 668–676.
- 125 D. Nuvoli, L. Valentini, V. Alzari, S. Scognamiglio, S. B. Bon, M. Piccinini, J. Illescas and A. Mariani, High concentration few-layer graphene sheets obtained by liquid phase exfoliation of graphite in ionic liquid, *J. Mater. Chem.*, 2011, **21**, 3428–3431.
- 126 V. V. Chaban, E. E. Fileti and O. V. Prezhdo, Exfoliation of graphene in ionic liquids: Pyridinium versus Pyrrolidinium, *J. Phys. Chem. C*, 2017, **121**, 911–917.
- 127 H. Yang, C. Shan, F. Li, D. Han, Q. Zhang and L. Niu, Covalent functionalization of polydisperse chemically-converted graphene sheets with amine-terminated ionic liquid, *Chem. Commun.*, 2009, 3880–3882.
- 128 E. Vazquez, F. Giacalone and M. Prato, Non-conventional methods and media for the activation and manipulation of carbon nanoforms, *Chem. Soc. Rev.*, 2014, **43**, 58–69.
- 129 T. Kim, H. Chang Kang, T. Thanh Tung, J. Don Lee, H. Kim, W. Seok Yang, H. Gyu Yoon and K. S. Suh, Ionic liquid-assisted microwave reduction of graphite oxide for supercapacitors, *RSC Adv.*, 2012, **2**, 8808–8812.
- 130 W. Yan, Y. Huang, Y. Xu, L. Huang and Y. Chen, Rapid and effective functionalization of graphene oxide by ionic liquid, *J. Nanosci. Nanotechnol.*, 2012, **12**, 2270–2277.
- 131 F. Xie, C. Xu, Y. Song, Q. Liang, J. Ji and S. Wang, 2D-2D heterostructure of ionic liquid-exfoliated MoS₂/MXene as lithium polysulfide barrier for Li-S batteries, *J. Colloid Interface Sci.*, 2023, **636**, 528–536.
- 132 A. Winchester, S. Ghosh, S. Feng, A. L. Elias, T. Mallouk, M. Terrones and S. Talapatra, Electrochemical characterization of liquid phase exfoliated two-dimensional layers of molybdenum disulfide, *ACS Appl. Mater. Interfaces*, 2014, **6**, 2125–2130.
- 133 X. Liu, P. Ding, H. Li, Z. Yuan, Y. Jiao and S. Jameh-Bozorghi, The effect of ionic liquid compounds on the exfoliation of the two-dimensional layer of molybdenum disulfide, *Microporous Mesoporous Mater.*, 2020, **299**, 110127.
- 134 Y. Gao, Z. Liu, D. M. Sun, L. Huang, L. P. Ma, L. C. Yin, T. Ma, Z. Zhang, X. L. Ma, L. M. Peng, H. M. Cheng and W. Ren, Large-area synthesis of high-quality and uniform monolayer WS₂ on reusable Au foils, *Nat. Commun.*, 2015, **6**, 8569.



- 135 Y. Biswas, M. Dule and T. K. Mandal, Poly(ionic liquid)-promoted solvent-borne efficient exfoliation of MoS₂/MoSe₂ nanosheets for dual-responsive dispersion and polymer nanocomposites, *J. Phys. Chem. C*, 2017, **121**, 4747–4759.
- 136 L. Li, Y. Yu, G. J. Ye, Q. Ge, X. Ou, H. Wu, D. Feng, X. H. Chen and Y. Zhang, Black phosphorus field-effect transistors, *Nat. Nanotechnol.*, 2014, **9**, 372–377.
- 137 M. T. Edmonds, A. Tadich, A. Carvalho, A. Ziletti, K. M. O'Donnell, S. P. Koenig, D. F. Coker, B. Ozyilmaz, A. H. Neto and M. S. Fuhrer, Creating a stable oxide at the surface of black phosphorus, *ACS Appl. Mater. Interfaces*, 2015, **7**, 14557–14562.
- 138 Y. Wang, Z. Feng, Y. Lyu, X. Zhou, X. Tan, T. Han and S. Liang, CN110676308A, 2020.
- 139 W. Zhao, Z. Xue, J. Wang, J. Jiang, X. Zhao and T. Mu, Large-scale, highly efficient, and green liquid-Exfoliation of black phosphorus in ionic liquids, *ACS Appl. Mater. Interfaces*, 2015, **7**, 27608–27612.
- 140 Q. Cai, L. H. Li, S. Mateti, A. Bhattacharjee, Y. Fan, S. Huang and Y. I. Chen, Boron nitride nanosheets: Thickness-related properties and applications, *Adv. Funct. Mater.*, 2024, **34**, 2403669.
- 141 J. Restolho, J. L. Mata and B. Saramago, On the interfacial behavior of ionic liquids: Surface tensions and contact angles, *J. Colloid Interface Sci.*, 2009, **340**, 82–86.
- 142 T. Morishita, M. Matsushita, Y. Katagiri and K. Fukumori, Effects of the composition and molecular weight of maleimide polymers on the dispersibility of carbon nanotubes in chloroform, *Carbon*, 2011, **49**, 5185–5195.
- 143 X. Yuan, L. Guo, Q. Lv, D. Zhao, H. Liao, W. Ma, Q. Jiang, G. Xie, Y. Hou and J. Shen, Product quality research in the industrialized preparation of electronic grade polysilicon: A review of impurity sources and control strategies, *Renewable Sustainable Energy Rev.*, 2025, **214**, 115569.
- 144 S. K. Wallace, D. B. Mitzi and A. Walsh, The steady rise of kesterite solar cells, *ACS Energy Lett.*, 2017, **2**, 776–779.
- 145 I. Poli and A. Petrozza, Halide perovskite semiconductors processing: Solvent based or solvent free?, *ACS Energy Lett.*, 2024, **9**, 4596–4597.
- 146 M. R. Begley, D. S. Gianola and T. R. Ray, Bridging functional nanocomposites to robust macroscale devices, *Science*, 2019, **364**, 1250.
- 147 Y. Cao, L. Ge, X. Dong, Q. Yang, Z. Bao, H. Xing and Q. Ren, Separation of hydrophobic compounds differing in a monounsaturated double bond using hydrophilic ionic liquid/water mixtures as extractants, *ACS Sustainable Chem. Eng.*, 2018, **6**, 2379–2385.
- 148 K. K. Thasneema, T. Dipin, M. S. Thayyil, P. K. Sahu, M. Messali, T. Rosalin, K. K. Elyas, P. M. Saharuba, T. Anjitha and T. B. Hadda, Removal of toxic heavy metals, phenolic compounds and textile dyes from industrial waste water using phosphonium based ionic liquids, *J. Mol. Liq.*, 2021, **323**, 114645.
- 149 Y. Wang, S. Chen, R. Liu, L. Zhang, W. Xue and Y. Yang, Toward green and efficient recycling of Au(III), Pd(II) and Pt(IV) from acidic medium using UCST-type ionic liquid, *Sep. Purif. Technol.*, 2022, **298**, 121620.
- 150 C. Cao, X. Xu, G. Wang, Z. Yang, Z. Cheng, S. Zhang, T. Li, Y. Pu, G. Lv, C. Xu, J. Cai, W. Zhou, F. Li, Z. Pu and X. Li, Characterization of ionic liquids removing heavy metals from electroplating sludge: Influencing factors, optimisation strategies and reaction mechanisms, *Chemosphere*, 2023, **324**, 138309.
- 151 F. Geng, R. Zhang, L. Wu, Z. Tang, H. Liu, H. Liu, Z. Liu, C. Xu and X. Meng, High-efficiency separation and extraction of naphthenic acid from high acid oils using imidazolium carbonate ionic liquids, *Chin. J. Chem. Eng.*, 2022, **41**, 252–259.
- 152 A. C. Blaga, A. Tucaliuc and L. Kloetzer, Applications of ionic liquids in carboxylic acids separation, *Membranes*, 2022, **12**, 771.
- 153 H. W. Khan, A. V. B. Reddy, M. A. Bustam, M. Goto and M. Moniruzzaman, Development and optimization of ionic liquid-based emulsion liquid membrane process for efficient recovery of lactic acid from aqueous streams, *Biochem. Eng. J.*, 2021, **176**, 108216.
- 154 C. Xiong, Y. Li, S. Wang and Y. Zhou, Functionalization of nanosilica via guanidinium ionic liquid for the recovery of gold ions from aqueous solutions, *J. Mol. Liq.*, 2018, **256**, 183–190.
- 155 R. Leyma, S. Platzer, F. Jirsa, W. Kandioller, R. Krachler and B. K. Keppler, Novel thiosalicylate-based ionic liquids for heavy metal extractions, *J. Hazard. Mater.*, 2016, **314**, 164–171.
- 156 L. Fischer, T. Falta, G. Koellensperger, A. Stojanovic, D. Kogelnig, M. Galanski, R. Krachler, B. K. Keppler and S. Hann, Ionic liquids for extraction of metals and metal containing compounds from communal and industrial waste water, *Water Res.*, 2011, **45**, 4601–4614.
- 157 Y. Fan, D. Cai, L. Yang, X. Chen and L. Zhang, Extraction behavior of nicotinic acid and nicotinamide in ionic liquids, *Chem. Eng. Res. Des.*, 2019, **146**, 336–343.
- 158 S. Imdad and R. K. Dohare, A critical review on heavy metals removal using ionic liquid membranes from the industrial wastewater, *Chem. Eng. Process.*, 2022, **173**, 108812.
- 159 J. Marták, Š. Schlosser and S. Vlčková, Pertraction of lactic acid through supported liquid membranes containing phosphonium ionic liquid, *J. Membr. Sci.*, 2008, **318**, 298–310.
- 160 S. Zhang, N. Sun, X. He, X. Lu and X. Zhang, Physical properties of ionic liquids: Database and evaluation, *J. Phys. Chem. Ref. Data*, 2006, **35**, 1475–1517.
- 161 S. Luo, T. Han, C. Wang, Y. Sun, H. Zhang, R. Guo and S. Zhang, Hierarchically microporous membranes for highly energy-efficient gas separations, *Ind. Chem. Mater.*, 2023, **1**, 376–387.
- 162 J. Wang, D. Xie, Z. Zhang, Q. Yang, H. Xing, Y. Yang, Q. Ren and Z. Bao, Efficient adsorption separation of acetylene and ethylene via supported ionic liquid on metal-organic framework, *AIChE J.*, 2016, **63**, 2165–2175.



- 163 H. Xing, X. Zhao, Q. Yang, B. Su, Z. Bao, Y. Yang and Q. Ren, Molecular dynamics simulation study on the absorption of ethylene and acetylene in ionic liquids, *Ind. Eng. Chem. Res.*, 2013, **52**, 9308–9316.
- 164 H. Cao, M. Dou, Z. Lyu, Y. Wang, C. M. Pedersen and Y. Qiao, Understanding the interaction mechanism of carbazole/anthracene with N,N-dimethylformamide: NMR study substantiated carbazole separation, *Ind. Chem. Mater.*, 2023, **1**, 240–246.
- 165 P. Isosaari, V. Srivastava and M. Sillanpää, Ionic liquid-based water treatment technologies for organic pollutants: Current status and future prospects of ionic liquid mediated technologies, *Sci. Total Environ.*, 2019, **690**, 604–619.
- 166 H. Yang, N. U. Afsar, Q. Chen, X. Ge, X. Li, L. Ge and T. Xu, Poly(alkyl-biphenyl pyridinium) anion exchange membranes with a hydrophobic side chain for mono-/divalent anion separation, *Ind. Chem. Mater.*, 2023, **1**, 129–139.
- 167 H. Yuan, H. Wang and Y. Cui, Two-dimensional layered chalcogenides: From rational synthesis to property control via orbital occupation and electron filling, *Acc. Chem. Res.*, 2015, **48**, 81–90.
- 168 H. Fu, X. Zhu, P. Li, M. Li, L. Yang, C. Jia and X. Guo, Recent progress in single-molecule transistors: Their designs, mechanisms and applications, *J. Mater. Chem. C*, 2022, **10**, 2375–2389.
- 169 C. Huang, A. V. Rudnev, W. Hong and T. Wandlowski, Break junction under electrochemical gating: Testbed for single-molecule electronics, *Chem. Soc. Rev.*, 2015, **44**, 889–901.
- 170 S. Zhou, K. S. Panse, M. H. Motevaselian, N. R. Aluru and Y. Zhang, Three-dimensional molecular mapping of ionic liquids at electrified interfaces, *ACS Nano*, 2020, **14**, 17515–17523.
- 171 N. Xin, X. Li, C. Jia, Y. Gong, M. Li, S. Wang, G. Zhang, J. Yang and X. Guo, Tuning charge transport in aromatic-ring single-molecule junctions via ionic-liquid gating, *Angew. Chem., Int. Ed.*, 2018, **57**, 14026–14031.
- 172 R. Karimi Azari, T. Lan and C. Santato, On the factors affecting the response time of synaptic ion-gated transistors, *J. Mater. Chem. C*, 2023, **11**, 8293–8299.
- 173 L. Merces, L. M. M. Ferro, A. Nawaz and P. Sonar, Advanced neuromorphic applications enabled by synaptic Ion-gating vertical transistors, *Adv. Sci.*, 2024, **11**, e2305611.
- 174 D. Xiang, X. Wang, C. Jia, T. Lee and X. Guo, Molecular-scale electronics: From concept to function, *Chem. Rev.*, 2016, **116**, 4318–4440.
- 175 C. Leighton, Electrolyte-based ionic control of functional oxides, *Nat. Mater.*, 2019, **18**, 13–18.
- 176 A. M. Goldman, Electrostatic gating of ultrathin films, *Annu. Rev. Mater. Res.*, 2014, **44**, 45–63.
- 177 T. Fujimoto and K. Awaga, Electric-double-layer field-effect transistors with ionic liquids, *Phys. Chem. Chem. Phys.*, 2013, **15**, 8983–9006.
- 178 S. Z. Bisri, S. Shimizu, M. Nakano and Y. Iwasa, Endeavor of iontronics: From fundamentals to applications of ion-controlled electronics, *Adv. Mater.*, 2017, **29**, 1607054.
- 179 M. Mezger, H. Schröder, H. Reichert, S. Schramm, J. S. Okasinski, S. Schöder, V. Honkimäki, M. Deutsch, B. M. Ocko, J. Ralston, M. Rohwerder, M. Stratmann and H. Dosch, Molecular layering of fluorinated ionic liquids at a charged sapphire (0001) surface, *Science*, 2008, **322**, 424–428.
- 180 C. Jia, M. Famili, M. Carlotti, Y. Liu, P. Wang, I. M. Grace, Z. Feng, Y. Wang, Z. Zhao, M. Ding, X. Xu, C. Wang, S.-J. Lee, Y. Huang, R. C. Chiechi, C. J. Lambert and X. Duan, Quantum interference mediated vertical molecular tunneling transistors, *Sci. Adv.*, 2018, **4**, eaat837.
- 181 M. Famili, C. Jia, X. Liu, P. Wang, I. M. Grace, J. Guo, Y. Liu, Z. Feng, Y. Wang, Z. Zhao, S. Decurtins, R. Häner, Y. Huang, S.-X. Liu, C. J. Lambert and X. Duan, Self-assembled molecular-electronic films controlled by room temperature quantum interference, *Chem*, 2019, **5**, 474–484.
- 182 X. Guo, J. Li, F. Wang, J. H. Zhang, J. Zhang, Y. Shi and L. Pan, Application of conductive polymer hydrogels in flexible electronics, *J. Polym. Sci.*, 2022, **60**, 2635–2662.
- 183 J. H. Lee, K. Cho and J. K. Kim, Age of flexible electronics: Emerging trends in soft multifunctional sensors, *Adv. Mater.*, 2024, **36**, e2310505.
- 184 C. Han, J. Li, Z. Ma, H. Xie, G. I. N. Waterhouse, L. Ye and T. Zhang, Black phosphorus quantum dot/g-C₃N₄ composites for enhanced CO₂ photoreduction to CO, *Sci. China Mater.*, 2018, **61**, 1159–1166.
- 185 E. N. Popova, V. E. Yudin, L. A. Myagkova, N. V. Kukarkina, M. Y. Goikhman and V. M. Svetlichnyi, Thermal aging of carbon- and glass-reinforced plastics based on heat-resistant polyimide binders, *Russ. J. Appl. Chem.*, 2009, **82**, 889–893.
- 186 P. Thiruvassagam, Synthesis and characterization of AB-type monomers and polyimides: A review, *Des. Monomers Polym.*, 2013, **16**, 197–221.
- 187 H. Shi, Y. Li and T. Guo, In situ preparation of transparent polyimide nanocomposite with a small load of graphene oxide, *J. Appl. Polym. Sci.*, 2012, **128**, 3163–3169.
- 188 Y. Guan, C. Wang, D. Wang, G. Dang, C. Chen, H. Zhou and X. Zhao, High transparent polyimides containing pyridine and biphenyl units: Synthesis, thermal, mechanical, crystal and optical properties, *Polymer*, 2015, **62**, 1–10.
- 189 J. Lan, Y. Li, B. Yan, C. Yin, R. Ran and L. Y. Shi, Transparent stretchable dual-network ionogel with temperature tolerance for high-performance flexible strain sensors, *ACS Appl. Mater. Interfaces*, 2020, **12**, 37597–37606.
- 190 P. K. Tapaswi, M. C. Choi, Y. S. Jung, H. J. Cho, D. J. Seo and C. S. Ha, Synthesis and characterization of fully aliphatic polyimides from an aliphatic dianhydride with piperazine spacer for enhanced solubility, transparency, and low dielectric constant, *J. Polym. Sci., Part A: Polym. Chem.*, 2014, **52**, 2316–2328.
- 191 J. Abraham, P. Mohammed Arif, L. Kailas, N. Kalarikkal, S. C. George and S. Thomas, Developing highly conducting and mechanically durable styrene butadiene rubber composites with tailored microstructural properties by a green approach using ionic liquid modified MWCNTs, *RSC Adv.*, 2016, **6**, 32493–32504.



- 192 T. Xiao, X. Fan, D. Fan and Q. Li, High thermal conductivity and low absorptivity/ emissivity properties of transparent fluorinated polyimide films, *Polym. Bull.*, 2017, **74**, 4561–4575.
- 193 I. H. Choi, B. Sohn and J. H. Chang, Synthesis and characterization of transparent copolyimide films containing CF₃ groups: Comparison with copolyimide nanocomposites, *Appl. Clay Sci.*, 2010, **48**, 117–126.
- 194 P. Qi, H. Jia, S. Xu, Q. Wang, G. Su, G. Yang, M. Zhang, Y. Qu and F. Pei, Preparation and separation properties of electrospinning modified membrane with ionic liquid terminating polyimide/polyvinyl pyrrolidone@polydopamine, *Membranes*, 2022, **12**, 189.
- 195 C. Schmidt, T. Glück and G. S. Naake, Modification of nafion membranes by impregnation with ionic liquids, *Chem. Eng. Technol.*, 2007, **31**, 13–22.
- 196 J. Han, Z. Dong, L. Hao, J. Gong and Q. Zhao, Poly(ionic liquid)-crosslinked graphene oxide/carbon nanotube membranes as efficient solar steam generators, *Green Energy Environ.*, 2023, **8**, 151–162.
- 197 X. Ye, Z. Geng, X. Rao, Y. Wu, Y. Li, Y. Zhang and P. Wang, Recent advances in transparent materials based on ionic liquids and their applications in sensors, *Adv. Mater. Technol.*, 2024, **9**, 2301292.
- 198 Q. Rong, W. Lei, J. Huang and M. Liu, Low temperature tolerant organohydrogel electrolytes for flexible solid-state supercapacitors, *Adv. Energy Mater.*, 2018, **8**, 1801967.
- 199 X. Pan, Q. Wang, R. Guo, Y. Ni, K. Liu, X. Ouyang, L. Chen, L. Huang, S. Cao and M. Xie, An integrated transparent, UV-filtering organohydrogel sensor via molecular-level ion conductive channels, *J. Mater. Chem. A*, 2019, **7**, 4525.
- 200 C. Qiu, Y. Xu, X. Fan, D. Xu, R. Tandiana, X. Ling, Y. Jiang, C. Liu, L. Yu, W. Chen and C. Su, Highly crystalline k-intercalated polymeric carbon nitride for visible-light photocatalytic alkenes and alkynes deuterations, *Adv. Sci.*, 2019, **6**, 1801403.
- 201 L. Zhang, W. Deng, B. Wu, L. Ye, X. Sun, Z. Wang, K. Gao, H. Wu, C. Duan, F. Huang and Y. Cao, Reduced energy loss in non-fullerene organic solar cells with isomeric donor polymers containing thiazole pi-spacers, *ACS Appl. Mater. Interfaces*, 2020, **12**, 753–762.
- 202 P. Lv, L. Shi, C. Fan, Y. Gao, A. Yang, X. Wang, S. Ding and M. Rong, Hydrophobic ionic liquid gel-based triboelectric nanogenerator: Next generation of ultrastable, flexible, and transparent power sources for sustainable electronics, *ACS Appl. Mater. Interfaces*, 2020, **12**, 15012–15022.
- 203 K. Polak-Krasna, M. Tian, S. Rochat, N. Gathercole, C. Yuan, Z. Hao, M. Pan, A. D. Burrows, T. J. Mays and C. R. Bowen, Solvent sorption-induced actuation of composites based on a polymer of intrinsic microporosity, *ACS Appl. Polym. Mater.*, 2021, **3**, 920–928.
- 204 L. E. Shmukler, M. S. Gruzdev, N. O. Kudryakova, Y. A. Fadeeva, A. M. Kolker and L. P. Safonova, Triethylammonium-based protic ionic liquids with sulfonic acids: Phase behavior and electrochemistry, *J. Mol. Liq.*, 2018, **266**, 139–146.
- 205 V. Bocharova, A. L. Agapov, A. Tselev, L. Collins, R. Kumar, S. Berdzinski, V. Strehmel, A. Kisliuk, I. I. Kravchenko, B. G. Sumpter, A. P. Sokolov, S. V. Kalinin and E. Strelcov, Controlled nanopatterning of a polymerized ionic liquid in a strong electric field, *Adv. Funct. Mater.*, 2014, **25**, 805–811.
- 206 W. Sitaputra, D. Stacchiola, J. F. Wishart, F. Wang and J. T. Sadowski, In situ probing of ion ordering at an electrified ionic liquid/Au interface, *Adv. Mater.*, 2017, **29**, 1606357.
- 207 Z. Li, J. T. Sadowski, A. Dolocan and F. Mangolini, Ionic liquid-mediated scanning probe electro-oxidative lithography as a novel tool for engineering functional oxide micro- and nano-architectures, *Adv. Funct. Mater.*, 2023, **33**, 2306660.

