

Cite this: *Chem. Sci.*, 2020, **11**, 6830

All publication charges for this article have been paid for by the Royal Society of Chemistry

Received 18th May 2020
Accepted 10th June 2020

DOI: 10.1039/d0sc02816a

rsc.li/chemical-science

Palladium-catalyzed dearomative 1,4-difunctionalization of naphthalenes†

Ping Yang, Chao Zheng, Yu-Han Nie and Shu-Li You *

A highly diastereoselective dearomatization of naphthalenes *via* a Pd-catalyzed 1,4-difunctionalization reaction is described. In the presence of a commercially available palladium precursor and ligand, intramolecular dearomative Heck-type insertion provides π -allylpalladium intermediates which are readily captured by a series of nucleophiles in excellent yields (up to 99%). This reaction features mild conditions, broad substrate scope, and useful transformations of the products.

Introduction

The difunctionalization of alkenes is widely recognized as a powerful approach to generate significant molecular complexity from simple chemical feedstock.¹ In particular, Pd-catalyzed difunctionalization of 1,3-dienes, which introduces two functional groups across the conjugated C=C double bonds, has witnessed significant progress in the past decade.² Mechanistically, Pd(II) complexes, usually generated from the oxidative addition of Pd(0) precursors, have been employed to achieve these reactions through Heck insertion³ to form a π -allylpalladium species. The subsequent regioselective nucleophilic attacks afford 1,2- or 1,4-addition-like products (Scheme 1a).^{2d} However, the scope of these transformations was in general limited to structurally simple dienes⁴ and furans.⁵

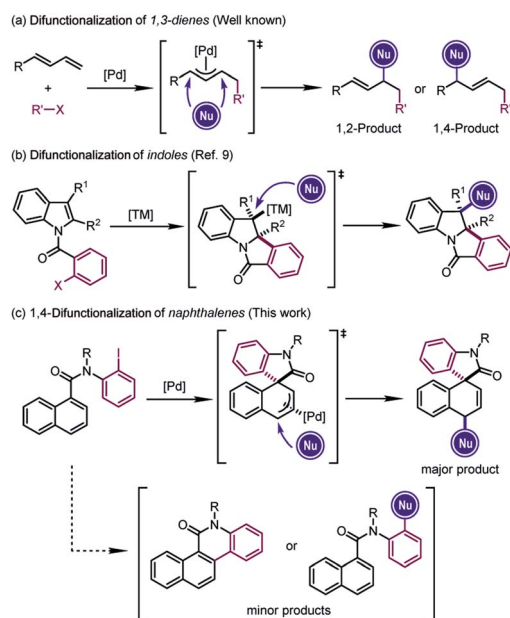
In line with our continuous interest in catalytic dearomatization reactions,⁶ we envisioned that the formal “conjugated diene” structure of the phenyl ring might serve as the equivalent of 1,3-dienes. However, the dearomatization of electronically unbiased aromatic compounds such as naphthalenes and benzenes remained challenging,⁷ due to the generally higher aromatic stabilization energies of plain arenes compared with their heteroaromatic counterparts (36 kcal mol^{−1} for benzene and 22 kcal mol^{−1} for pyrrole).^{6r} Therefore, the translation of Pd-catalyzed difunctionalization of 1,3-dienes to aromatic systems would open a new window for the dearomatization of non-activated arenes.

Inspired by the recent developments in Pd-catalyzed dearomative Heck reactions⁸ that are terminated by the nucleophilic attack on allylpalladium intermediates (Scheme 1b),⁹ we realized

that the π -allylpalladium intermediates formed from the Heck-type insertion into the naphthalene ring might also be captured using external nucleophiles, furnishing the dearomative 1,4-difunctionalization of naphthalenes (Scheme 1c).¹⁰ The successful execution of this reaction design relied on the judicious selection of the catalytic system which could overcome the thermodynamic disadvantage of the dearomatization process and at the same time avoid the competitive C–H activation or direct cross-coupling reactions. Herein we report our results from this study.

Results and discussion

We began our investigation by studying the 1,4-difunctionalization of *N*-(2-iodophenyl)-*N*-methyl-1-naphthamide (**1a**) with



Scheme 1 Difunctionalization of 1,3-dienes, indoles and naphthalenes.

State Key Laboratory of Organometallic Chemistry, Center for Excellence in Molecular Synthesis, Shanghai Institute of Organic Chemistry, University of Chinese Academy of Sciences, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China. E-mail: slyou@sioac.ac.cn


† Electronic supplementary information (ESI) available. CCDC 1982544–1982545. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d0sc02816a

dimethyl malonate (**2a**) (Table 1). Firstly, we tested different ligands in the presence of PdCl₂ (10 mol%), NaH (2.0 equiv.), and Ag₃PO₄ (1.0 equiv.) in DMA at 100 °C. When BINAP (**L1**), BINAP(O) (**L2**), and Feringa phosphoramidite (*rac*-**L3**) were used as the ligand respectively, the dearomatized product **3a** was obtained as a single diastereomeric isomer in good yields (67–90%) (entries 1–3), whose structure and relative configuration were confirmed by X-ray crystallographic analysis. The relative configuration of **3a** revealed that the *in situ* formed π -allylpalladium intermediate was attacked by the nucleophile *via* an outer sphere mechanism. On the other hand, the reaction with the PHOX ligand (**L4**) was sluggish (entry 4). Subsequently, different solvents were examined by using *rac*-**L3** as the ligand (entries 5–8). DMA was found to be the optimal one among those tested. In particular, the desired reaction was prohibited significantly when *t*BuOH was employed (entry 6). Next, the effects of various bases were examined (entries 9–12). This revealed that **3a** could be obtained in moderate yields by using K₂CO₃ or K₃PO₄, while the target product was not observed when 1,2,2,6,6-pentamethylpiperidine (PMP) or *t*BuOK was

employed. Surprisingly, the judicious choice of silver salts was quite critical to this reaction. Among those tested, Ag₃PO₄ provided the optimal reaction outcomes. Other commonly used silver salts including AgOTf, AgNTf₂, and AgBF₄ were ineffective (entries 13–15). Finally, the optimal yield of **3a** (95%) was obtained by using a pre-prepared sodium salt of **2a** with lower catalyst loading (5 mol%) (entry 16).

With the optimal conditions in hand, we surveyed the generality of this novel 1,4-difunctionalization by allowing various naphthalene derivatives **1** to react with sodium salts of dialkyl malonates **2** (Table 2). When the methyl group on the nitrogen tether of **1** was changed to isopropyl or benzyl groups, the desired products **3b** and **3c** were obtained in good yields (82–86%). The substrates bearing an electron-donating group (Me and OMe) or halide (F, Cl, and Br) at the *para* position of the phenyl ring led to **3d–3h** in 75–98% yields. The good tolerance with halides would offer a handle on subsequent transformations. In contrast, when an electron-withdrawing group (CF₃, CO₂Me, and CN) was incorporated into the aryl iodide moiety, the corresponding products (**3i**, **3j**, **3l**) were formed in moderate yields (40–51%). Notably, an *ortho* substituent on the phenyl ring was well tolerated, affording the dearomatized product **3m** in 80% yield. Naphthalene derivatives bearing a substituent at the C6 position furnished the products **3n–3p** in reasonable yields (39–82%). It is worth noting that **3a** could be obtained in 75% yield by using *N*-(2-bromophenyl)-*N*-methyl-1-

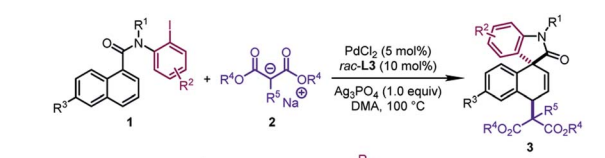
Table 1 Optimization of the reaction conditions^a



Entry	Ligand	Base	Solvent	1a ^b (%)	3a ^b (%)
1 ^c	L1	NaH	DMA	—	79
2 ^c	L2	NaH	DMA	15	67
3	<i>rac</i> - L3 ^d	NaH	DMA	—	90 (90 ^e)
4 ^c	L4	NaH	DMA	44	43
5	<i>rac</i> - L3 ^d	NaH	Toluene	69	22
6	<i>rac</i> - L3 ^d	NaH	<i>t</i> BuOH	Quant.	Trace
7	<i>rac</i> - L3 ^d	NaH	DCE	63	36
8	<i>rac</i> - L3 ^d	NaH	Dioxane	52	51
9	<i>rac</i> - L3 ^d	K ₂ CO ₃	DMA	43	46
10	<i>rac</i> - L3 ^d	K ₃ PO ₄	DMA	51	29
11	<i>rac</i> - L3 ^d	PMP ^f	DMA	Quant.	Trace
12	<i>rac</i> - L3 ^d	<i>t</i> BuOK	DMA	66	Trace
13 ^g	<i>rac</i> - L3 ^d	NaH	DMA	66	27
14 ^h	<i>rac</i> - L3 ^d	NaH	DMA	84	8
15 ⁱ	<i>rac</i> - L3 ^d	NaH	DMA	77	5
16 ^j	<i>rac</i> - L3 ^d	NaH	DMA	—	97 (95 ^e)

^a Reaction conditions: **1a** (0.2 mmol), **2a** (0.4 mmol), PdCl₂ (0.02 mmol), ligand (0.04 mmol), base (0.4 mmol), and Ag₃PO₄ (0.2 mmol) in solvent (1.0 mL) at 100 °C. ^b Yield determined by ¹H NMR using CH₂Br₂ (0.2 mmol) as an internal standard. ^c Ligand (0.02 mmol). ^d *rac*-**L3**: (*R*_a,*R*_b,*R*_c + *S*_a,*S*_b,*S*_c):(*S*_a,*R*_b,*R*_c + *R*_a,*S*_b,*S*_c) = 1 : 4. ^e Isolated yield. ^f PMP: 1,2,2,6,6-pentamethylpiperidine. ^g AgOTf as the silver salt. ^h AgNTf₂ as the silver salt. ⁱ Sodium salt of **2a** (pre-prepared by mixing NaH (0.4 mmol) and **2a** (0.4 mmol) in DMA at room temperature for 0.5 h), PdCl₂ (0.01 mmol), and *rac*-**L3** (0.02 mmol).

Table 2 Substrate scope of naphthalenes and malonic diesters^a



Product	Yield (%)
3a	95% yield
3b	82% yield
3c	86% yield
3d (R = Me)	93% yield
3e (R = OMe)	98% yield
3f (R = F)	83% yield
3g (R = Cl)	83% yield
3h (R = Br)	75% yield ^b
3i (R = CF ₃)	40% yield
3j (R = CO ₂ Me)	51% yield
3k	94% yield
3l	50% yield
3m	80% yield
3n	78% yield
3o	39% yield
3p	82% yield
3q	87% yield
3r	81% yield
3s	28% yield ^d

^a **1** (0.2 mmol), **2** (0.4 mmol, pre-prepared from malonic ester with NaH), PdCl₂ (0.01 mmol), *rac*-**L3** (0.02 mmol), and Ag₃PO₄ (0.2 mmol) in DMA (1.0 mL) at 100 °C. ^b Pre-synthesized PdCl₂-BINAP complex (0.01 mmol) was used. ^c *N*-(2-bromophenyl)-*N*-methyl-1-naphthamide was used at 120 °C. ^d PdCl₂ (0.02 mmol) and *rac*-**L3** (0.04 mmol) at 120 °C.



naphthamide as the substrate and BINAP as the ligand. In addition, when diethyl malonate or dibenzyl malonate was utilized as the nucleophile, the desired products **3q** and **3r** were obtained with good yields (81–87%). The dearomatized product **3s** bearing two all-carbon quaternary stereocenters could be afforded in 28% yield. The low yield probably resulted from the unfavorable steric hindrance in the second step.

Next, different 1,3-diketones were explored under slightly modified conditions (Table 3). When acetylacetone and 3,5-heptanedione were employed as the coupling partners, the corresponding products **4a** and **4b** were delivered in good yields (67–69%). Notably, cyclic β -diketones were also tolerated, leading to **4c** and **4d** in moderate yields (42–55%).

The reaction design was successfully applied to the sodium salts of β -ketoesters (Table 4). The reactions of diverse alkyl

substituted β -ketoesters furnished the desired products in good to excellent yields (**5a–5f**, 56–99%). Aryl β -ketoesters also participated smoothly in the reaction, regardless of the electronic properties of the aryl group, delivering **5g–5i** in good yields (59–76%). Of particular note is that 2-thienyl and 2-furyl β -ketoesters could be converted to **5j** and **5k** in moderate to good yields (45–89%).

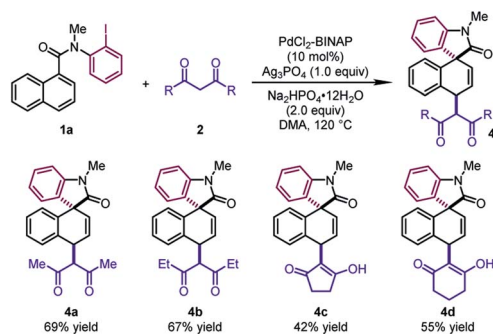
In addition to β -ketoesters, the esters bearing an electron-withdrawing group (NO_2 and SO_2Ph) at the α -position were proved to be viable participants, leading to **5l** and **5m** in moderate yields (24–51%) (Scheme 2). Although obtained as a mixture of a pair of diastereoisomers with poor dr values, the products could undergo decarboxylation to deliver pure dearomatized compounds (*vide infra*).

To further demonstrate the generality of this reaction, we focused on developing the 1,4-difunctionalization with nitrogen-based nucleophiles (Table 5). Various benzyl amines were investigated as the nucleophiles. The corresponding products **6a** and **6b** could be obtained in excellent yields (97–99%). Moreover, furfurylamine and 2-thiophenemethylamine were also compatible, leading to dearomatized products **6c** and **6d** in good yields (76–98%). Cyclic secondary amines, such as pyrrolidine, piperidine and morpholine, reacted as the coupling partners to form the tertiary amines **6f–6h** in excellent yields (98–99%). Diethylamine and cyclohexylamine also participated in the reaction smoothly, leading to **6e** and **6i** in excellent yields (88–95%). It is noteworthy that various anilines regardless of electronic properties were also viable reaction partners. The desired products **6k–6m** could be delivered in high yields (80–99%). The structure of **6k** was determined by X-ray crystallographic analysis. Notably, the reactions of tryptamine and allylamine led to **6j** and **6n** in low yields. When (*R*)-1-phenylethylamine was used, **6o** was obtained in 96% yield with a 1 : 1 dr.

Preliminary investigations on the enantioselective variants of the dearomative 1,4-difunctionalization reactions of naphthalene derivatives with dimethyl malonate or aniline were performed. Unfortunately, the utilization of a series of chiral ligands did not afford satisfactory asymmetric induction (see the ESI† for details).

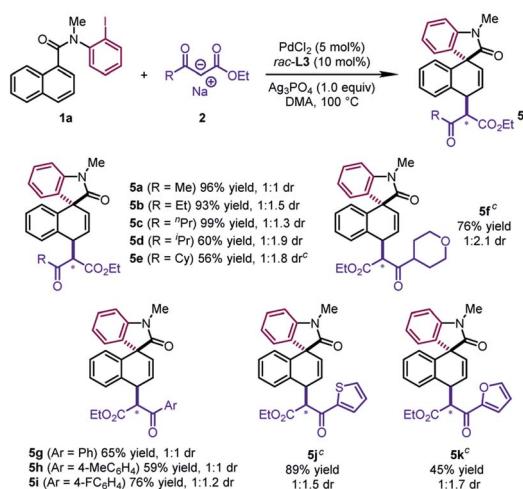
To test the practicality of this method, a gram-scale dearomative 1,4-difunctionalization reaction of **1a** (4.0 mmol) and aniline with a lower catalyst loading (2.5 mol%) was carried out (Scheme 3). The desired product **6k** could be afforded in 96% yield (1.3 g). Some synthetic transformations of the dearomatized products have been examined. The mixture of two diastereoisomers of **5g** (1 : 1 dr) could be decarboxylated to form **7** in 80% yield. The newly formed C–N bond of **6k** could be cleaved

Table 3 Substrate scope of 1,3-diketones^a

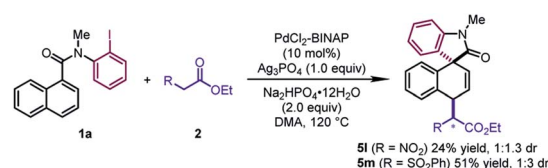


^a **1a** (0.2 mmol), **2** (0.4 mmol), pre-synthesized $\text{PdCl}_2\text{-BINAP}$ complex (0.02 mmol), Ag_3PO_4 (0.2 mmol), and $\text{Na}_2\text{HPO}_4\cdot 12\text{H}_2\text{O}$ (0.4 mmol) in DMA (1.0 mL) at 120 $^\circ\text{C}$.

Table 4 Substrate scope of β -ketoesters^{a,b}



^a **1a** (0.2 mmol), **2** (0.4 mmol), pre-prepared from β -ketoester with NaH), PdCl_2 (0.01 mmol), *rac*-L3 (0.02 mmol), and Ag_3PO_4 (0.2 mmol) in DMA (1.0 mL) at 100 $^\circ\text{C}$. ^b The diastereomeric selectivity originates from the reversal of the relative configuration at the position denoted with an asterisk. ^c PdCl_2 (0.02 mmol), *rac*-L3 (0.04 mmol).



Scheme 2 Substrate scope of esters.

monohydrate, furnishing the corresponding isoxazole **9** and pyrazole **10** in good yields, respectively.



In summary, we have developed a Pd-catalyzed dearomative 1,4-difunctionalization of naphthalene derivatives by mimicking the reactivity of simple conjugated dienes in more challenging electronically unbiased aromatic systems. Diverse nucleophiles were found to be compatible with the reaction conditions. Various functionalized spirooxindoles could be obtained efficiently in good to excellent yields (up to 99%) with exclusive diastereoselectivity. Further exploration of the application of this methodology is currently underway in our laboratory.

Conflicts of interest

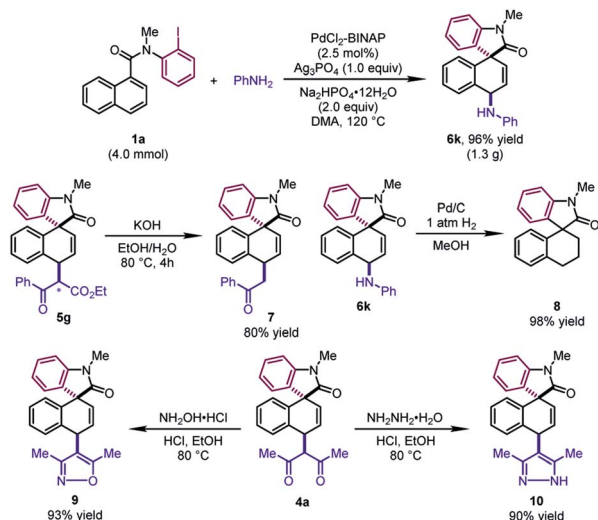
There are no conflicts to declare.

Acknowledgements

We thank the MOST (2016YFA0202900), NSFC (21821002 and 91856201) and Chinese Academy of Sciences (XDB20000000 and QYZDY-SSW-SLH012) for generous financial support. S.-L. Y. acknowledges the support from the Tencent Foundation through the XPLOER PRIZE.

Notes and references

^a **1a** (0.2 mmol), **2** (0.4 mmol), pre-synthesized PdCl₂-BINAP complex (0.01 mmol), Ag₃PO₄ (0.2 mmol), and Na₂HPO₄·12H₂O (0.4 mmol) in DMA (1.0 mL) at 120 °C.



Scheme 3 Gram-scale reaction and transformations of the products.

by the hydrogenolysis reaction with Pd/C, leading to **8** in excellent yield. In addition, the β -diketone compound **4a** could be condensed with hydroxylamine hydrochloride and hydrazine

- 1 For selected reviews: (a) K. H. Jensen and M. S. Sigman, *Org. Biomol. Chem.*, 2008, **6**, 4083–4088; (b) R. I. McDonald, G. Liu and S. S. Stahl, *Chem. Rev.*, 2011, **111**, 2981–3019; (c) D. M. Schultz and J. P. Wolfe, *Synthesis*, 2012, **44**, 351–361; (d) M. S. Sigman and E. W. Werner, *Acc. Chem. Res.*, 2012, **45**, 874–884; (e) G. Yin, X. Mu and G. Liu, *Acc. Chem. Res.*, 2016, **49**, 2413–2423.
- 2 For selected reviews: (a) J. W. Han and T. Hayashi, *Tetrahedron: Asymmetry*, 2010, **21**, 2193–2197; (b) Z. Wu and W. Zhang, *Chin. J. Org. Chem.*, 2017, **37**, 2250–2262; (c) Y. Xiong, Y. Sun and G. Zhang, *Tetrahedron Lett.*, 2018, **59**, 347–355; (d) X. Wu and L.-Z. Gong, *Synthesis*, 2019, **51**, 122–134; (e) G. Li, X. Huo, X. Jiang and W. Zhang, *Chem. Soc. Rev.*, 2020, **49**, 2060–2118.
- 3 For selected reviews: (a) R. F. Heck, *Acc. Chem. Res.*, 1979, **12**, 146–151; (b) I. P. Beletskaya and A. V. Cheprakov, *Chem. Rev.*, 2000, **100**, 3009–3066; (c) A. B. Dounay and L. E. Overman, *Chem. Rev.*, 2003, **103**, 2945–2964; (d) M. Shibasaki, E. M. Vogl and T. Ohshima, *Adv. Synth. Catal.*, 2004, **346**, 1533–1552; (e) D. Mc Cartney and P. J. Guiry, *Chem. Soc. Rev.*, 2011, **40**, 5122–5150; (f) H. Li, C. Ding, B. Xu and X. Hou, *Acta Chim. Sin.*, 2014, **72**, 765–770; (g) A. Biffis, P. Centomo, A. Del Zotto and M. Zecca, *Chem. Rev.*, 2018, **118**, 2249–2295; (h) Y. Ping, Y. Li, J. Zhu and W. Kong, *Angew. Chem., Int. Ed.*, 2019, **58**, 1562–1573.
- 4 For selected examples: (a) K. Kagechika and M. Shibasaki, *J. Org. Chem.*, 1991, **56**, 4093–4094; (b) K. Kagechika, T. Ohshima and M. Shibasaki, *Tetrahedron*, 1993, **49**, 1773–

- 1782; (c) C. S. Nylund, J. M. Klopp and S. M. Weinreb, *Tetrahedron Lett.*, 1994, **35**, 4287–4290; (d) T. Ohshima, K. Kagechika, M. Adachi, M. Sodeoka and M. Shibasaki, *J. Am. Chem. Soc.*, 1996, **118**, 7108–7116; (e) D. Flubacher and G. Helmchen, *Tetrahedron Lett.*, 1999, **40**, 3867–3868; (f) X. Wu, H.-C. Lin, M.-L. Li, L.-L. Li, Z.-Y. Han and L.-Z. Gong, *J. Am. Chem. Soc.*, 2015, **137**, 13476–13479; (g) Y. Liu, Y. Xie, H. Wang and H. Huang, *J. Am. Chem. Soc.*, 2016, **138**, 4314–4317; (h) Z.-L. Tao, A. Adili, H.-C. Shen, Z.-Y. Han and L.-Z. Gong, *Angew. Chem., Int. Ed.*, 2016, **55**, 4322–4326; (i) L. Luo, H. Zheng, J. Liu, H. Wang, Y. Wang and X. Luan, *Org. Lett.*, 2016, **18**, 2082–2085; (j) L. Xu, X. Zhang, M. S. McCamant, M. S. Sigman, Y.-D. Wu and O. Wiest, *J. Org. Chem.*, 2016, **81**, 7604–7611; (k) N. J. Adamson, E. Hull and S. J. Malcolmson, *J. Am. Chem. Soc.*, 2017, **139**, 7180–7183; (l) H.-C. Shen, P.-S. Wang, Z.-L. Tao, Z.-Y. Han and L.-Z. Gong, *Adv. Synth. Catal.*, 2017, **359**, 2383–2389; (m) N. J. Adamson, K. C. E. Wilbur and S. J. Malcolmson, *J. Am. Chem. Soc.*, 2018, **140**, 2761–2764; (n) S. Park and S. J. Malcolmson, *ACS Catal.*, 2018, **8**, 8468–8476; (o) X. Wu, S.-S. Chen, L. Zhang, H.-J. Wang and L.-Z. Gong, *Chem. Commun.*, 2018, **54**, 9595–9598; (p) T. Zhang, H.-C. Shen, J.-C. Xu, T. Fan, Z.-Y. Han and L.-Z. Gong, *Org. Lett.*, 2019, **21**, 2048–2051; (q) Y. Zhang, H.-C. Shen, Y.-Y. Li, Y.-S. Huang, Z.-Y. Han and X. Wu, *Chem. Commun.*, 2019, **55**, 3769–3772; (r) D. Zhu, Z. Jiao, Y. R. Chi, T. P. Gonçalves, K.-W. Huang and J. S. Zhou, *Angew. Chem., Int. Ed.*, 2020, **59**, 5341–5345.
- 5 (a) J. Li, H. Peng, F. Wang, X. Wang, H. Jiang and B. Yin, *Org. Lett.*, 2016, **18**, 3226–3229; (b) J. Liu, X. Xu, J. Li, B. Liu, H. Jiang and B. Yin, *Chem. Commun.*, 2016, **52**, 9550–9553; (c) Z. Wang, W. Luo, L. Lu and B. Yin, *J. Org. Chem.*, 2018, **83**, 10080–10088.
- 6 For recent reviews of dearomative reactions: (a) A. R. Pape, K. P. Kaliappan and E. P. Kundig, *Chem. Rev.*, 2000, **100**, 2917–2940; (b) S. Quideau, L. Pouységu and D. Deffieux, *Synlett*, 2008, 467–495; (c) L. Pouységu, D. Deffieux and S. Quideau, *Tetrahedron*, 2010, **66**, 2235–2261; (d) L. Pouységu, T. Sylla, T. Garnier, L. B. Rojas, J. Charris, D. Deffieux and S. Quideau, *Tetrahedron*, 2010, **66**, 5908–5917; (e) S. P. Roche and J. A. Porco Jr, *Angew. Chem., Int. Ed.*, 2011, **50**, 4068–4093; (f) D.-S. Wang, Q.-A. Chen, S.-M. Lu and Y.-G. Zhou, *Chem. Rev.*, 2012, **112**, 2557–2590; (g) C.-X. Zhuo, W. Zhang and S.-L. You, *Angew. Chem., Int. Ed.*, 2012, **51**, 12662–12686; (h) C.-X. Zhuo, C. Zheng and S.-L. You, *Acc. Chem. Res.*, 2014, **47**, 2558–2573; (i) Q. Ding, X. Zhou and R. Fan, *Org. Biomol. Chem.*, 2014, **12**, 4807–4815; (j) S. P. Roche, J.-J. Youte Tendoung and B. Tréguier, *Tetrahedron*, 2015, **71**, 3549–3591; (k) W.-T. Wu, L. Zhang and S.-L. You, *Chem. Soc. Rev.*, 2016, **45**, 1570–1580; (l) C. Zheng and S.-L. You, *Chem*, 2016, **1**, 830–857; (m) M. J. James, P. O'Brien, R. J. K. Taylor and W. P. Unsworth, *Chem.-Eur. J.*, 2016, **22**, 2856–2881; (n) W. Sun, G. Li, L. Hong and R. Wang, *Org. Biomol. Chem.*, 2016, **14**, 2164–2176; (o) J.-B. Chen and Y.-X. Jia, *Org. Biomol. Chem.*, 2017, **15**, 3550–3567; (p) W.-T. Wu, L. Zhang and S.-L. You, *Acta Chim. Sin.*, 2017, **75**, 419–438; (q) J. Bariwal, L. G. Voskressensky and E. V. Van der Eycken, *Chem. Soc. Rev.*, 2018, **47**, 3831–3848; (r) W. C. Wertjes, E. H. Southgate and D. Sarlah, *Chem. Soc. Rev.*, 2018, **47**, 7996–8017; (s) J. An and M. Bandini, *Chimia*, 2018, **72**, 610–613; (t) Y.-Z. Cheng, X. Zhang and S.-L. You, *Sci. Bull.*, 2018, **63**, 809–811; (u) G. Huang and B. Yin, *Adv. Synth. Catal.*, 2019, **361**, 405–425; for a book: (v) S.-L. You, *Asymmetric Dearomatization Reactions*, Wiley-VCH, Weinheim, 2016.
- 7 For selected recent examples: (a) M. Bao, H. Nakamura and Y. Yamamoto, *J. Am. Chem. Soc.*, 2001, **123**, 759–760; (b) B. Peng, X. Feng, X. Zhang, L. Ji and M. Bao, *Tetrahedron*, 2010, **66**, 6013–6018; (c) B. Peng, S. Zhang, X. Yu, X. Feng and M. Bao, *Org. Lett.*, 2011, **13**, 5402–5405; (d) B. M. Trost, V. Ehmke, B. M. O'Keefe and D. A. Bringley, *J. Am. Chem. Soc.*, 2014, **136**, 8213–8216; (e) E. H. Southgate, J. Pospech, J. Fu, D. R. Holycross and D. Sarlah, *Nat. Chem.*, 2016, **8**, 922–928; (f) M. Okumura, S. M. Nakamata Huynh, J. Pospech and D. Sarlah, *Angew. Chem., Int. Ed.*, 2016, **55**, 15910–15914; (g) S. Zhang, J. Cai, Y. Yamamoto and M. Bao, *J. Org. Chem.*, 2017, **82**, 5974–5980; (h) S. Zhang, A. Ullah, Y. Yamamoto and M. Bao, *Adv. Synth. Catal.*, 2017, **359**, 2723–2728; (i) M. Okumura, A. S. Shved and D. Sarlah, *J. Am. Chem. Soc.*, 2017, **139**, 17787–17790; (j) L. W. Hernandez, U. Klöckner, J. Pospech, L. Hauss and D. Sarlah, *J. Am. Chem. Soc.*, 2018, **140**, 4503–4507; (k) M. Komatsuda, K. Muto and J. Yamaguchi, *Org. Lett.*, 2018, **20**, 4354–4357; (l) Z.-P. Yang, R. Jiang, Q.-F. Wu, L. Huang, C. Zheng and S.-L. You, *Angew. Chem., Int. Ed.*, 2018, **57**, 16190–16193; (m) W. C. Wertjes, M. Okumura and D. Sarlah, *J. Am. Chem. Soc.*, 2019, **141**, 163–167; (n) M. Komatsuda, H. Kato, K. Muto and J. Yamaguchi, *ACS Catal.*, 2019, **9**, 8991–8995; (o) C. Tang, M. Okumura, H. Deng and D. Sarlah, *Angew. Chem., Int. Ed.*, 2019, **58**, 15762–15766; (p) R.-Q. Xu, P. Yang, C. Zheng and S.-L. You, *Chin. J. Chem.*, 2020, **38**, 683–689.
- 8 For a recent review: (a) N. Zeidan and M. Lautens, *Synthesis*, 2019, **51**, 4137–4146; for selected recent examples: (b) P. Yang and S.-L. You, *Org. Lett.*, 2018, **20**, 7684–7688; (c) P. Yang, R.-Q. Xu, C. Zheng and S.-L. You, *Chin. J. Chem.*, 2020, **38**, 235–241.
- 9 (a) C. Shen, R.-R. Liu, R.-J. Fan, Y.-L. Li, T.-F. Xu, J.-R. Gao and Y.-X. Jia, *J. Am. Chem. Soc.*, 2015, **137**, 4936–4939; (b) D. A. Petrone, A. Yen, N. Zeidan and M. Lautens, *Org. Lett.*, 2015, **17**, 4838–4841; (c) S. Chen, X.-X. Wu, J. Wang, X.-H. Hao, Y. Xia, Y. Shen, H. Jing and Y.-M. Liang, *Org. Lett.*, 2016, **18**, 4016–4019; (d) D. A. Petrone, M. Kondo, N. Zeidan and M. Lautens, *Chem.-Eur. J.*, 2016, **22**, 5684–5691; (e) R.-R. Liu, T.-F. Xu, Y.-G. Wang, B. Xiang, J.-R. Gao and Y.-X. Jia, *Chem. Commun.*, 2016, **52**, 13664–13667; (f) R.-R. Liu, Y. Xu, R.-X. Liang, B. Xiang, H.-J. Xie, J.-R. Gao and Y.-X. Jia, *Org. Biomol. Chem.*, 2017, **15**, 2711–2715; (g) R.-R. Liu, Y.-G. Wang, Y.-L. Li, B.-B. Huang, R.-X. Liang and Y.-X. Jia, *Angew. Chem., Int. Ed.*, 2017, **56**, 7475–7478; (h) N. Zeidan, T. Beisel, R. Ross and M. Lautens, *Org. Lett.*, 2018, **20**, 7332–7335; (i) R.-X. Liang, R.-Z. Yang, R.-R. Liu and Y.-X. Jia, *Org. Chem. Front.*, 2018, **5**, 1840–1843; (j)



- A. D. Marchese, F. Lind, Á. E. Mahon, H. Yoon and M. Lautens, *Angew. Chem., Int. Ed.*, 2019, **58**, 5095–5099; (k) C. Shen, N. Zeidan, Q. Wu, C. B. J. Breuers, R.-R. Liu, Y.-X. Jia and M. Lautens, *Chem. Sci.*, 2019, **10**, 3118–3122; (l) R.-X. Liang, K. Wang, Q. Wu, W.-J. Sheng and Y.-X. Jia, *Organometallics*, 2019, **38**, 3927–3930.
- 10 (a) Z. Zuo, H. Wang, Y. Diao, Y. Ge, J. Liu and X. Luan, *ACS Catal.*, 2018, **8**, 11029–11034; (b) X. Liao, D. Wang, Y. Huang, Y. Yang and J. You, *Org. Lett.*, 2019, **21**, 1152–1155.

