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Recent biological applications of heterocyclic hybrids containing s-triazine scaffold

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s-Triazine possesses an auspicious status in the field of drug discovery and development owing to its presence in many naturally occurring compounds as well as commercially available drugs like enasidenib, gedatolisib, bimiralisib, atrazine, indaziflam, and triaziflam. Easy, cost-effective, and efficient access to its derivatives in addition to their splendid biological activities such as anticancer, anti-inflammatory, antiviral, anticonvulsant, anti-tubercular, antidiabetic, antimicrobial, makes it an attractive heterocyclic nucleus in the field of medicinal chemistry. Other than the direct access of its derivatives from simple commercially available starting materials like amidine, the s-triazine derivatives have also been obtained starting from an inexpensive commercially available 2,4,6-trichloro-1,3,5-triazine (TCT) commonly known as cyanuric chloride. Owing to the high reactivity and the possibility of sequential substitution of TCT, a variety of biologically active heterocyclic scaffolds have been installed on this nucleus in order to have more potent compounds. These s-triazine-based heterocyclic hybrids have been reported to show enhanced biological activities in recent years. Therefore, it is important to summarize and highlight recent examples of these hybrids which is imperative to attract the attention of the drug development community.

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1. Introduction

Rapid developments in the field of combinatorial, computational, and experimental chemistry have provided organic chemists access to a library of a large number of bioactive compounds based on heterocycle scaffolds. 1,2 Careful analysis suggests that the compounds having two or more bioactive heterocyclic moieties are superior in their biological applications as compared to the compounds having only one heterocyclic moiety. 3 Therefore, the hybridization of different bioactive heterocyclic moieties has become an effective approach to afford potent compounds and solve resistance and efficacy problems. Resultantly, the heterocyclic hybrids are envisioned as excellent future candidates for the development of drug candidates. 3

Triazine is an important heterocyclic scaffold having a diverse biological profile. Depending on the position of nitrogen, there are three isomeric forms of triazine, namely 1,2,3-triazine (1), 1,2,4-triazine (2), and 1,3,5-triazine (3). Out of these three, 1,3,5-triazine (s-triazine) has a unique place in medicinal chemistry⁴ owing to its symmetric structure and easy access to a variety of its derivatives either directly from simple starting materials (*vide infra*) or indirectly from one of its commercially available derivatives *e.g.*, cyanuric chloride⁵ (2,4,6-trichlorotriazine) (4) (Fig. 1). Having three reactive chloro-

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substituents and the possibility of their sequential substitution at different temperatures has made cyanuric chloride an ideal platform to construct novel drug candidates with optimum physicochemical and biological properties.6 Owing to these reasons, s-triazine is now part of many approved drugs such as altretamine (5) (anti-ovarian cancer), tretamine (6) (antineoplastic), 8,9 azacitidine (7) (antineoplastic agent), 10 almitrine (8) (respiratory stimulant),11 enasidenib (9) (antileukemia),12 gedatolisib (10) (anti-breast cancer), 13,14 bimiralisib (11) (anti-breast cancer)15 melarsoprol (12) (anti-trypanosomiasis or sleeping sickness),16 (Fig. 2a). In addition, this heterocyclic nucleus is also part of many commercially available herbicides. 17 Most importantly, these herbicides are also found in natural sources, especially in plants which help them in proper growth and protection from epiphytes, climbers, and wild wines.18 Atrazine (13), desethylatrazine (14), desisopropylatrazine (15), simazine (16), terbuthylazine (17), propazine (18), cyanazine (19), cyprazine (20), atraton (21), prebane (22), terbutryn (23), prometon (24), prometryn (25), desmetryn (26), ametryn (27) and dimethametryn (28) are the examples of s-triazine based herbicides17,19,20 (Fig. 2b). Likewise, indiaziflam (29)21 and triaziflam (30)22 are US-FDA-approved insecticides that possess s-triazine as a basic structure (Fig. 2c).

Given the tremendous recent importance of heterocyclic derivatives, scientists are continuously working to generate novel libraries of drug candidates by hybridizing *s*-triazine nucleus (*vide supra*) with a wide variety of well-known heterocyclic scaffolds.²³ Therefore, it is imperative to summarize such

Review RSC Advances

Fig. 1 Isomeric forms of triazine (1), (2), and (3), and structure of cyanuric chloride (TCT) (4).

recent studies published in the last five years (2018–2023) to attract not only the attention of the drug development community but also to find the gap in planning novel strategies for the synthesis of pharmacologically more potent novel striazine-based hybrid molecules.

2. Summary of synthetic strategies for s-triazine derivatives

A variety of strategies have been used to synthesize s-triazine derivatives from cheap and readily available starting materials (please see Fig. 3). For instance, from zinc dimethyl imidodicarbonimidic (31) by reacting it with activated acid derivatives in a mixture of solvent DCM and pyridine²⁴ (Fig. 3). Derivatives of s-triazine are also reported to be synthesized from amidines (32) by reacting with phosgene gas,25 with tertiary amines via copper catalysis,26 with primary amines under metal-free conditions,26 with alcohols under highly chemo-selective, base-free conditions via aerobic oxidation,27-30 and with diones.31 Similarly, ethyl acetimidate hydrochloride (33) is reacted with two equivalent monosodium cyanamide (NaNHCN), which on further treatment with hydroxylamine hydrochloride yields derivatives of s-triazine.32 Likewise, cycloaddition reactions of DICY (34) with aryl and aliphatic nitriles to yield substituted striazines are also reported.33 Cyclization of nitriles (35) under Lewis-acid-catalysis yields substituted symmetrical triazines.34 One-pot synthesis of 1,3,5-triazine can also be carried out via a trimerization reaction of nitriles.35 In the same fashion, reactions of metformin (36) with ester³⁶ or other bile acids are reported to synthesize derivatized s-triazines in good yield.³⁷ The metformin (36) on reaction with primary alcohols under graphene oxide catalysis yields substituted triazines.38 Derivatized 1,3,5-triazines can also be synthesized from rearrangement reactions of benzodiazepinedione (37) via hydrolysis or alcoholysis.39 In industries, derivatives of s-triazine can be synthesized starting from hydrogen cyanide (38).40

Apart from the direct access of triazine derivatives (Fig. 3), they are now easily accessible starting from an inexpensive, commercially available cyanuric chloride or trichlorotriazine (TCT) (4). TCT is a useful and reactive source to synthesize the biologically active leads, different multitopic molecules, and molecular hybrids possessing s-triazine as a main ingredient.⁵ Any nucleophilic species can be utilized to substitute chlorine atoms from TCT (4) through S_NAr mechanism at different temperature conditions (Fig. 4).⁴¹

3. Biological activities of *s*-triazine-based heterocyclic hybrids

3.1 Anticancer activity

Cancer is one of the serious health issues globally.⁴² It is the second major cause of death in the USA.^{43,44} Cancer is not just caused by genetic factors but also by other carcinogenic environmental factors.⁴⁵ Some of the recently reported molecules are summarized in Table 1 that have the potential to inhibit the enzymes responsible for cancer and can control angiogenesis.

Conjugates (39) represent the molecular hybrids of s-triazine and genistein.46 These hybrids exhibited remarkable action against the cancer cell-lines MDA-MB-231 (breast), HeLa (cervical), HCT-116 (prostate), and Huh-7 (liver). Structureactivity relationship revealed that the hybrid clubbed with 4methyl piperidine is the most potent and exhibited the highest levels of activity with the least IC50 value in comparison to other substituents for instance N,N-diphenyl amine, N,N-diethyl amine etc. IC₅₀ values of 4-methyl piperidine substituent are $HeLa = 39.13 \pm 0.89$ mM, $HCT-116 = 29.89 \pm 0.87$ mM, and Huh-7 = 44.01 \pm 0.41 mM. The most efficient antiproliferative activity is against MDA-MB-231 cells (IC₅₀ = 23.13 mM), which is even better than the standard drug 5-fluorouracil (IC₅₀ = 78.04 mM) (Table 1). Compound (40) is a bioconjugate incorporating the structural component of s-triazine tethered with tetrazole.47 In vitro, analysis of (40) suggested that exhibited good cytotoxicity against human alveolar basal epithelium adenocarcinoma A549 ($IC_{50} = 41.3 \text{ mmol L}^{-1}$), human ovarian teratocarcinoma PA-1 ($IC_{50} = 10.6 \text{ mmol L}^{-1}$), hepatocarcinoma Huh7 ($IC_{50} = 19.9 \text{ mmol L}^{-1}$), cervical cancer HeLa ($IC_{50} =$ 3.7 mmol L^{-1}), and human embryonic kidney HEK293 (IC₅₀ = 15.8 mmol L⁻¹). After investigation, it was suggested that the possible mechanism of cytotoxicity is through HIF pathway inhibition. This output suggests that these conjugates are remarkable anticancer agents against a wide range of cancerous cell-lines (Table 1).

Sun *et al.*, carried out the synthesis of 28 *s*-triazine-based compounds *via* coupling of thiophene and aryl urea with *s*-triazine (41).⁴⁸ These 2-arylurea-1,3,5-triazine derivatives show prominent inhibition of P13K/mTOR which is an effective anticancer therapy. The cytotoxic activity of these conjugates was evaluated against MCF-7, HeLa, and A549 cancer cell-lines. SAR analysis suggested that cytotoxicity decreases with increasing carbon in the R² chain linked with arylurea moiety and the most efficient cytotoxicity was observed with the

Fig. 2 Commercially available drugs (a), commercially available herbicides (b), and commercially available insecticides (c) that possess s-triazine scaffold

(29)

compound having sulfonamide moiety. For this potent compound, IC₅₀ of cytotoxicity was evaluated to be MCF-7 = $0.03 \pm 0.01~\mu\text{M}, \, \text{HeLa} = 0.27 \pm 0.02~\mu\text{M}, \, \text{A549} = 0.18 \pm 0.01~\mu\text{M}, \,$ while for the enzyme inhibition activity, it is noted to be PI3K =

23.8 nM, and mTOR = 10.9 nM (Table 1). To evaluate the anticancer activities of s-triazine, its three distinct series in conjunction with benzimidazole (42) were prepared and its inhibitor effect against P13K α - β - γ - δ , and three cancer cell-

NH.HCl R-CN Dicyandiamde (DICY) (34)Nitriles Ethyl acetimidate hydrochloride (33)(35)R-CN 2NaNHCN KOH, heat Metformin (36)Amidine (32)NH NaOH. Toluene EEDQ, EtOH (32)s-Triazine DBU, Reflux Phosgene Amidine Hydrolysis or Acoholysis Metformin DCM+Py (2:3) rt, 2h Amidine (32)HCN Benzodiazepinedione (37)Hydrogen Cyanide (38)Amidine (32)

Zinc dimethyl-imidodicarbonimidate

ig. 3 Summary of the various synthetic strategies used for the synthesis of s-triazine derivatives

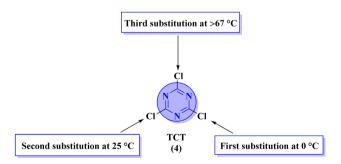


Fig. 4 Showing optimum temperature conditions for sequential substitution of three reactive chloro substituents of TCT (4).

lines, namely, U87-MG, MCF-7, and PC-3 were evaluated.⁴⁹ SAR showed that compounds with $\mathbf{R^1}$ containing 2,4-difluoro substituent and $\mathbf{R^2}$ being H showed maximum inhibition effect this may be due to the fact that these two fluorine atoms can

show more electron-withdrawing effect on the phenyl ring and make it more susceptible to $\pi-\pi$ interactions. Cytotoxicity of these compounds was evaluated to be U87-MG = 0.70 \pm 0.10 mM, MCF-7 = 0.13 \pm 0.02 mM, PC-3 = 0.57 \pm 0.15 mM, PI3K α = 0.32 \pm 0.18 nM, PI3K β = >100 nM, PI3K γ = 16.03 \pm 3.55 nM, and PI3K δ = >100 nM. Moreover, this compound also represented good anticancer activity in, *in vivo* analysis at 20 mg kg day $^{-1}$ in comparison to ZSTK-474 administrated at 40 mg kg day $^{-1}$ (Table 1).

Bruton's tyrosine kinase (BTK) plays a critical and vital role in B-cell development. Its inhibition is proven effective in cancer therapy. A series of 6-amino-s-triazine-based molecular hybrids (43) were synthesized and it's *in vitro* and *in silico* inhibitory action was evaluated.⁵⁰ SAR studies suggested that a compound with R² benzothiophenyl group proved to be a more potent and irreversible inhibitor against BTK than substituted benzene rings. This SAR was further studied

	Most potent compound			
Compounds	Substituents	(Γ_{C_50})	${ m IC}_{50}$ of reference drug	Refs.
HOO (39)	R = N-5-2	MDA-MB-231 = 23.13 \pm 1.29 μ M, HeLa = 39.13 \pm 0.89 μ M, HCT-116 = 29.89 \pm 0.87 μ M, Huh-7 = 44.01 \pm 0.41 μ M	Genistein: MDA-MB-231 = 89.11 ± 6.39 μM, HeLa = >100 μM, HCT-116= >100 μMHuh-7 = >100 μM, 5-FU: MDA-MB-231 = 78.04 ± 8.90 μM, HeLa = 67.64 ± 5.85 μM, HCT-116= 15.42 ± 2.88 μM, Huh-7 = 38.37 ± 1.51 μM	46
N. N	I	A549 (IC ₅₀ 41.3 μ mol L ⁻¹), PA-1 (IC ₅₀ 10.6 μ mol L ⁻¹), Huh7 (IC ₅₀ 19.9 μ mol L ⁻¹), HeEa (IC ₅₀ 3.7 μ mol L ⁻¹), HEK293 (IC ₅₀ 15.8 μ mol L ⁻¹)	I	47
S (41)	$X = CH$ $R^{1} = $	MCF-7 = $0.03 \pm 0.01 \mu M$, Hela = $0.27 \pm 0.02 \mu M$, A549 = $0.18 \pm 0.01 \mu M$, PI3K = $23.8 nM$, mTOR = $10.9 nM$	GDC- 0941: MCF-7 = 0.23 \pm 0.10 μ M, Hela = 0.93 \pm 0.06 μ M, A549 = 0.59 \pm 0.03 μ M, Pl3K = 6.0 nM, mTOR = 525.0 nM	48
R ¹ K K K K K K K K K K K K K K K K K K K	$R^1 = \frac{R}{R}$ $R^2 = \frac{R}{R}$ $\frac{d}{d}$	U87-MG = 0.70 \pm 0.10 mM, MCF-7 = 0.13 \pm 0.02 mM, PC-3 = 0.57 \pm 0.15 mM, PI3K α = 0.32 \pm 0.18 nM, PI3K β = >100 nM, PI3K γ = 16.03 \pm 3.55 nM, PI3K δ = >100 nM	ZSTK-474: U87-MG = 0.71 \pm 0.01 mM, MCF-7 = 0.13 \pm 0.02 mM, PC-3 = 0.58 \pm 0.03 mM, PI3K α = 6.40 \pm 0.12 nM, PI3K β = 23.89 \pm 0.28 nM, PI3K γ = 3.24 \pm 1.84 nM, PI3K δ = 24.76 \pm 3.57 nM	49
R2 H H N H 2 N H 2	R. R. R. C. H. C. C. H. S. C.	BTK = 17.0 ± 1.01 nM, EGFR = >1000 nM, JAK3 = 104.6 ± 0.80 nM	Ibrutinib: BTK = 1.9 \pm 1.71 nM, EGFR = 1.7 \pm 1.22 nM, JAK3 = 47.9 \pm 1.23 nM	50

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	Most potent compound			
Compounds	Substituents	(IC_{50})	${ m IC}_{50}$ of reference drug	Refs.
N NH, (A)	$ \begin{aligned} X &= S \\ m &= 2 \\ R &= H \end{aligned} $	Cytotoxic effect (on mouse T-lymphoma cells): PAR = 11.37 ± 0.33 μM , MDR = 16.73 ± 0.35 μM , antiproliferative effects: (on mouse T-lymphoma cells), PAR = 6.71 ± 0.40 μM , MDR = 5.35 ± 0.52 μM	Doxorubicin Cytotoxic effect (on mouse T-lymphoma cells): PAR = $0.7 \pm 0.56 \mu M$, MDR = $2.14 \pm 0.76 \mu M$, antiproliferative effects: (on mouse T-lymphoma cells), PAR = $0.28 \pm 0.06 \mu M$, MDR = $1.75 \pm 0.38 \mu M$	51
(£)	R = S	PI3K α = 177.41 nM, mTOR = 12.24 nM, MCF-7 = 0.30 \pm 0.06 μ M, Ovcar-2 = 0.25 \pm 0.11 μ M, U87MG = 1.25 \pm 0.37 μ M, A549 = 0.47 \pm 0.09 μ M, NCI-H460 = 2.31 \pm 0.38 μ M, H2228 = 0.22 \pm 0.03 μ M, H1975 = 0.008 \pm 0.002 μ M, HeLa = 1.04 \pm 0.23 μ M, HeLa-MDR = 0.37 \pm 0.13 μ M	GDC-0941: PI3 $K\alpha$ = 6.02 nM, mTOR = 525.03 nM, MCF-7 = 1.20 ± 0.23 μ M, Ovcar-2 = 1.25 ± 0.17 μ M, U87MG = 3.13 ± 0.31 μ M, A549 = 1.03 ± 0.09 μ M, NCF-H460 = 2.68 ± 0.37 μ M, H2228 = 1.07 ± 0.28 μ M, H275 = 2.42 ± 0.74 μ M, HeLa = 3.15 ± 0.17 μ M, HeL-MDR = 2.97 ± 0.31 μ M	22
A STANCE OF THE	R = 0	PC3 = 0.07 \pm 0.0082 μ M, A549 = 0.13 \pm 0.021 μ M, MCF-7 = 0.025 \pm 0.0018 μ M, A2780 = 0.19 \pm 0.073 μ M	Etoposide: PC3 = 2.39 \pm 1.56 μ M, A549 = 3.08 \pm 0.135 μ M, MCF-7 = 2.11 \pm 0.024 μ M, A2780 = 1.38 \pm 0.56 μ M	53
Harman Market Ma	$R = \underset{P}{\overset{\$}{\stackrel{\$}{\sim}}} CH_3$ $R^1 = \underbrace{\begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array}} M$	MDA-MB-231 = 97.3 \pm 9.3 μ M, U-87MG = 10.9 \pm 4.1 μ M, PANC-1 = 26.7 \pm 8.3 μ M, A549 = 12.4 \pm 6.4 μ M, MCF-7 = 8.3 \pm 2.1 μ M, HDFs = 38.9 \pm 8.3 μ M, EGFR-PK = 70.3 \pm 1.34 μ M, PI3K = 6.64 \pm 0.15 μ M mL ⁻¹ , AKT = 37.3 \pm 0.69 μ M mL ⁻¹ , mToR = 69.3 \pm 1.98 μ M mL ⁻¹	Tamoxifen: EGFR-PK inhibition, ${ m IC_{50}}=69.1\pm1.39~{ m nM}$	4.
N H (\$4)	I	HepG-2 = 29.75 ± 0.03 µM, PC-3 = 17.90 ± 0.03 µM, MCF-7 = 4.65 ± 0.07 µM, A-549 = 7.43 ± 0.04 µM, PBMC = 165.23 ± 18.05 µM, EGFR ^{WT} = 0.22 ± 0.05 µM, EGFR ^{T790M} = 0.18 ± 0.11 µM	DOX: HepG-2 = $4.51 \pm 0.26 \mu \text{M}$, PC-3 = $8.11 \pm 0.05 \mu \text{M}$, MCF-7 = $4.17 \pm 0.2 \mu \text{M}$, A-549 = $8.20 \pm 0.08 \mu \text{M}$, PBMC = $250.00 \pm 26.56 \mu \text{M}$, Erlotinib: HepG-2 = $8.19 \pm 0.4 \mu \text{M}$, PC-3 = $8.89 \pm 0.6 \mu \text{M}$, MCF-7 = $4.16 \pm 0.2 \mu \text{M}$, A-549 = $3.76 \pm 0.0 \mu \text{M}$, PBMC = $45.75 \pm 26.56 \mu \text{M}$, EGFR ^{WT} = $0.09 \pm 0.05 \mu \text{M}$, EGFR ^{WT} = $0.09 \pm 0.05 \mu \text{M}$, EGFR ^{WT} = $0.55 \pm 0.10 \mu \text{M}$	55
n _N N N N N N N N N N N N N N N N N N N	$X = \mathbf{Br}$ $\mathbf{N_{II}} = \mathbf{N}$	MCF-7 = 4.53 \pm 0.30 μ M, HCT-116 = 0.50 \pm 0.080 μ M, HepG2 = 3.01 \pm 0.49 μ M, EGFR PK = 61 \pm 0.002 nM	Tamoxifen: MCF-7 = 5.12 \pm 0.36 μ M, HCT-116 = 26.41 \pm 4.11 μ M, HepG2 = 2.45 \pm 0.20 μ M, Erlotinib: EGFR PK = 78.65 \pm 3.54 nM	26

	Most potent compound			
Compounds	Substituents	(IC ₅₀)	${ m IC}_{50}$ of reference drug	Refs.
(59)	$R_1 = H$ $R_2 = OCH_3$ $R_3 = H$ $R_4 = CI$ $X = N$ $Y = O$	Caco-2 = $19.36 \pm 3.71 \mu M$, HT-29 = $19.29 \pm 2.62 \mu M$, HMVEC-d = $10.86 \pm 0.08 \mu M$	1	57
R1 R2 R3 R4	$X = O$ $R = H$ $R_1 = Br$ $R_2 = H$ $R_3 = H$	Cell growth inhibition (GI%) with 10 µM dose, CCRF-CEM = 79.05, SR = 92.07, NCI-H522 = 66.45, HT29 = 93.39, COLO 205 = 76.23, SF-295 = 70.86, LOX IMVI = 94.77, SK-MEL-2 = 97.36, CAKI-1 = 84.64, MDA-MB-468 = 98.91	Enasidenib cell growth inhibition (GI%): CCRF-CEM = Not tested, SR = 31.62, HT29 = 19.95, COLO 205 = 15.84, SF-295 = 19.95, LOX IMVI = 15.84, MDA-MB-468 = 19.95	
R R S S S S S S S S S S S S S S S S S S	R ₃ N ₃ N ₄ N H H H H H H H H H H H H	$\mathrm{pIC}_{50} = 4.29$		29
HN Nu Nu N	Nu ¹ = Morpholine Nu ² = 4-chloroaniline	A549 = 2.40 \pm 0.64 μ M, MCF-7 = 3.28 \pm 0.16 μ M, HDFs = 3.78 \pm 0.55 μ M, EGFR = 34.1 \pm 1.58 nM, CDK-2 = 108.3 \pm 3.12 nM	Doxorubicin (IC ₅₀ μ M mL ⁻¹): A549 = 0.28 \pm 0.11, MCF-7 = 0.06 \pm 0.02, HDFs = 0.42 \pm 0.27, Erlotinib: EGFR = 67.3 \pm 2.54 nM, Roscovitine: CDK-2 = 140 \pm 3.4 nM	09
RI CNH2	54a 54b R \$N(CH ₃);	54a	Staurosporine	61
		MDA-MB-468 = 3.99 \pm 0.21 μ M, CCRF-CM = 4.51 \pm 0.24 μ M, 54b: MDA-MB-468= 1.48 \pm 0.08 μ M, CCRF-CM = 9.83 \pm 0.52 μ M	MDA-MB-468 = 6.07 \pm 0.03 μ M, CCRF-CM = 2.17 \pm 0.11 μ M	

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Table 1 (Contd.)

	Most potent compound			
Compounds	Substituents	(IC_{50})	${ m IC}_{50}$ of reference drug	Refs.
R (55)	R1 R2 R3 NH CH3 NN-N NN-N	Huh-7 0.4 μΜ		62
H N N N N N N N N N N N N N N N N N N N	R = -OH	Kinase inhibition (IC ₅₀ in nM), PI3K $\alpha=3.41\pm0.05$ nM, mTOR = 8.45 ± 1.28 nM, Anticancer activity (IC ₅₀ in μ M), MDA-MB321 = 15.83 ± 1.52 μ M, MGF-7 = 16.32 ± 0.92 μ M, HeIa = 2.21 ± 0.45 μ M, HepG2 = 12.21 ± 0.43 μ M	Gedatolisib kinase inhibition (IC $_{50}$ in nM): PI3K $lpha=6.04\pm0.45$ nM, mTOR $=2.5\pm0.23$ nM	93

RSC Advances Review

through docking analysis, and it was observed that the benzothiophene moiety can bind more properly into the H3 pocket of BTK to show π - π interaction with Tyr551 which improves the inhibitory action even more than that of the standard drug, ibrutinib. Kinase selectivity of this conjugates is BTK = 17.0 \pm 1.01 nM, EGFR = >1000 nM, JAK3 = 104.6 \pm 0.80 nM. These findings suggest that the s-triazine and its molecular hybrids can be a promising candidate for blocking the Bruton's tyrosine kinase enzyme irreversibly (Table 1). In order to treat lymphoma, a molecular hybrid of amino-s-triazine tethered piperazine (44) was synthesized and the main motive of the synthesis was to evaluate the difference of activity between thioand seleno-ether at position X.51 Through all the analyses it was observed that β-naphthyl-thioether derivative proved to be more potent than verapamil by a factor of 11 i.e., $11 \times$ times more potent. The cytotoxicity of this lead on mouse T-lymphoma cells was shown to be (sensitive parental) PAR = 11.37 \pm 0.33 μ M, and (resistant) MDR = $16.73 \pm 0.35 \mu M$ while, the antiproliferative effects was (sensitive parental) PAR $= 6.71 \pm 0.40$ μM and (resistant) MDR = 5.35 \pm 0.52 μM (Table 1).

In their studies, S. Xu et al., reported molecular hybrids of striazine and thiophene (45). The third scaffold which is pendant with this hybrid is morpholine.⁵² The anticancer activity of these conjugates was assessed against a panel of 9 different cancer cell-lines and enzymes PI3K/mTOR. Fortunately, 7 compounds among them proved to be more effective than GDC-0941. The inhibitory action of this conjugate was 302.5× better than the lead compound GDC-0941. With the lowest IC₅₀ value being $0.008 \pm 0.002~\mu\text{M}.$ The kinase inhibitory value was noted to be 177.41 and 12.24 nM for PI3K and mTOR respectively. SAR showed that the presence of the arylurea group significantly improved the activity and the conjugate with substituent R = pyridinyl ring displayed 98.04% inhibition highlighting its anticancer potency. The cytotoxicity of these compounds are MCF-7 = $0.30 \pm 0.06 \,\mu\text{M}$, Ovcar-2 = $0.25 \pm 0.11 \,\mu\text{M}$, U87MG = $1.25 \pm 0.37 \,\mu\text{M}$, A549 = $0.47 \pm 0.09 \,\mu\text{M}$, NCI-H460 = 2.31 ± 0.38 μ M, H2228 = 0.22 \pm 0.03 μ M, H1975 = 0.008 \pm 0.002 μ M, HeLa = 1.04 \pm 0.23 μ M and Hela-MDR = 0.37 \pm 0.13 μ M (Table 1).

Novel conjugates of oxadiazole tethered with s-triazine (46) were reported by V. Balaraju and coworkers.⁵³ These compounds were assessed as antitumor agents using MTT method against a number of anticancer cell-lines for instance, (PC3) prostate cancer, (A549) lung cancer, (MCF-7) breast cancer, and (A2780) ovarian cancer. Their activity was compared with etoposide, a remarkable standard for antitumor activity. The value of IC $_{50}$ was in order of PC3 $=0.07\pm0.0082~\mu\text{M},$ A549 = 0.13 \pm 0.021 $\mu M,$ MCF-7 = 0.025 \pm 0.0018 μM and A2780 = $0.19 \pm 0.073 \mu M$. The SAR studies showed that cytotoxicity is enhanced greatly when substituents are trimethoxy groups on phenyl ring. The reason may lie in the fact that methoxy is a strong electron-donating group and has three strong hydrogen bond acceptor atoms that can bind with the active site of certain enzymes. The data obtained from these studies provide a remarkable insight into the enhanced activity of s-triazine when tethered with oxadiazole (Table 1). Likewise, hybrids of striazine with pyrazole and pyrazole-fused cycloalkanone (47) have been synthesized and evaluated as potential anticancer

agents⁵⁴ against triple-negative breast cancer, glioblastoma, lung cancer, pancreatic cancer, human dermal fibroblast via relevant in vitro analysis. Inhibition of enzymes like P13K, AKT, and mToR was also screened. Intriguingly, it was noted that compounds with morpholine moiety proved to be more potent than standard tamoxifen. The most efficient cytotoxicity values of these compounds are U-87 MG = 10.9 \pm 4.1 μ M, PANC-1 = $26.7 \pm 8.3 \, \mu M$, $A549 = 12.4 \pm 6.4 \, \mu M$, MCF-7 = $8.3 \pm 2.1 \, \mu M$, HDFs = 38.9 \pm 8.3 μ M, EGFR-PK = 70.3 \pm 1.34 nM. While the IC_{50} value for the enzyme inhibitory activity is PI3K = 6.64 \pm 0.15 ng mL^{-1} , AKT = $37.3 \pm 0.69 \text{ ng mL}^{-1}$, and mToR = $69.3 \pm 0.00 \pm 0.00 \pm 0.00 \pm 0.00$ 1.98 ng mL $^{-1}$ (Table 1).

Starting from 2-phenylacetyl isothiocyanate, microwaveassisted synthesis of 1,3,5-triazine tethered thiazole (48) is also reported as a potent anticancer agent.55 This bioconjugate was evaluated against different cancer cell-lines and kinase enzymes epidermal growth factor receptor (EGFR) and results were compared with standard drugs doxorubicin and erlotinib. The notable conjugate was showed the IC_{50} value as follow, HepG-2 = $29.75 \pm 0.03 \, \mu\text{M}$, PC-3 = $17.90 \pm 0.03 \, \mu\text{M}$, MCF-7 = 4.65 ± 0.07 μ M, A-549 = 7.43 \pm 0.04 μ M, PBMC = 165.23 \pm 18.05 μ M, EGFR^{WT} = 0.22 \pm 0.05 μ M, and EGFR^{T790M} = 0.18 \pm 0.11 μ M. Through in silico analysis, it was evaluated that the thiazole ring of conjugate showed good binding with methionine and leucine amino acid which supports the enhanced biological activity of conjugate (Table 1). Herein, hybrids of s-triazine and pyrazole are reported (49).56 These hybrids had variations at 4 positions of the phenyl ring and 5 positions of the triazine ring. The biological activities of these compounds were checked for 3 different cancer cell-lines and kinase enzymes. Via SAR studies and in silico inhibitory effect was also noted it was evaluated that conjugate with morpholine moiety and bromine atom at 4 positions of the amino group was noted as potent in comparison to tamoxifen with cytotoxicity MCF-7 = 4.53 \pm 0.30 μ M, HCT-116 = 0.50 \pm 0.080 μ M, and HepG2 = 3.01 \pm 0.49 μ M and IC₅₀ of enzyme inhibitory activity EGFR PK = 61 ± 0.002 nM (Table 1). In search of novel and potential antiproliferation agents, fused ring derivatives (50) of s-triazine and benzimidazole along with an appended oxazole ring linked via an amide bond were synthesized in high yield.⁵⁷ In vitro analysis against adenocarcinoma cell-lines and endothelial cell-line and IC50 value was quite good for instance, Caco-2 = 19.36 \pm 3.71 μ M, HT-29 = 19.29 \pm 2.62 μ M, and HMVEC-d = $10.86 \pm 0.08 \, \mu M$. Further studies were done to insight into the mechanism of action of these drug candidates and it was investigated that the microtubule skeleton was targeted by these conjugates (Table 1).

A new series of hydrazones of s-triazine tethered to morpholine (51) were synthesized and their antiproliferation activity was evaluated against 60 different cancer cell-lines and percentage growth inhibition was confirmed which provided useful results of these potent antiproliferation agents. Through comprehensive SAR analysis, it was also noted that the activity of conjugates decreases from a change in the hydrazone of aldehyde to the ketone. Likewise, morpholine substituents (X =O) were more potent than other analogs where X = N or X =CH₂. Cell growth inhibition (GI%) with a 10 μM dose can be summed up as, CCRF-CEM = 79.05, SR = 92.07, NCI-H522 =

67

70

Review RSC Advanc

	Most potent compoun	nd		
Compounds	Substituents	MIC/zone of inhibition	MIC/ZOI of ref. drug	Refs
R ¹ N N R ² (57)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	MIC (μM) <i>S. aureus</i> : LATCC 33591 = 3.9, MRSA BRS3 = 0.95, MRSA C1 = 1.95, MRSA C7 = 3.9, MRSA C14 = 3.9, MRSA G1 = 7.8	MIC (μM), AMK (Amikacin) S. aureus: ATCC 33591= 31.3, MRSA BRS3= 15.6, MRSA C1 = 3.9, MRSA C7 = 7.8, MRSA C14 = 7.8, MRSA G1 = 15.6	66
R ¹	R^1 R^2	Zone of inhibition in (mm), Gram positive: <i>S. aureus</i> = 14.84 ± 0.3 , <i>S. faecalis</i> = 14.84 ± 0.4 , <i>B. whetikin</i> = 14.44 ± 0.3 , <i>P. p. Largin</i>	Zone of inhibition in (mm), Norfloxacin Gram positive: <i>S.</i> $aureus = 17.44 \pm 0.3$, <i>S.</i> $faecalis = 17.45 \pm 0.5$, <i>B.</i> $substilis = 16.64 \pm$	

\mathbb{R}^2	
(58)	
R ² NM	

(59)

(MIC)
$$\mu g \text{ mL}^{-1}$$
: *E. coli* = 256, *S. aureus* = 256

 $substilis = 14.4 \pm 0.3$, P. vulgaris =

15.64 \pm 0.8, B. pumilus = 16.64 \pm

0.4, Gram negative: E. coli = 14.04

 \pm 0.44, K. penumoniae = 11.04 \pm

0.43

(MIC)
$$\mu g \text{ mL}^{-1}$$
 Chloramphenicol:
E. coli = 128, S. aureus = 256

0.3, P. vulgaris = 17.64 ± 0.4 , B.

Ciprofloxacin Gram negative: E.

 $coli=17.45\pm0.36$, K. penumoniae

 $pumilus = 18.45 \pm 0.4$,

 $= 17.64 \pm 0.65$

ZOI: **60a**: *E. coli* =
$$30 \pm 0.42$$
 mm, *B. subtillis* = 28 ± 0.22 mm, S. phyogenes = 19 ± 0.56 mm, *P. aregeunosa*, **60b**: *E. coli* = 35 ± 0.12 mm, *B. subtillis* = 18 ± 0.32 mm, *S. phyogenes* = 20 ± 0.56 mm

ZOI: Sulfadiazine: *E. coli* = 26
$$\pm$$
 0.83 mm, *B. subtillis* = $-$, Streptomycin, S. phyogenes = 10 \pm 0.66 mm

$$R = \underbrace{\begin{array}{c} CH_3 \\ CH \end{array}}_{C}$$

MIC (
$$\mu g \ \mu L^{-1}$$
): E. coli = 1.25, S. enteritidis = 1.25, S. aureus = 1.25, B. subtilis = 1.25

MIC (
$$\mu g \mu L^{-1}$$
): Penicillin, *E. coli* = 6, *S. enteritidis* = 6, *S. aureus* = 1.5, *B. subtilis* = 1.5

$$R = \xi C_{14}H_{29}$$

Table 2 (Contd.)

	Most potent compound	d		
Compounds	Substituents	MIC/zone of inhibition	MIC/ZOI of ref. drug	Refs.
Br NH NH NH R-NH (63)	$\mathbf{R} = \mathbf{N}$	MIC ($\mu g \text{ mL}^{-1}$): S. aureus = 6.25, P. aeruginosa = 12.5, B. cereus = 100, K. pneumoniae = 100	MIC ($\mu g \text{ mL}^{-1}$): Ciprofloxacin: <i>S. aureus</i> = 3.125, <i>P. aeruginosa</i> = 3.125, <i>B. cereus</i> = 3.125, <i>K. pneumoniae</i> = 3.125	72
(64b) S Ar (64c) R	$Ar = (64-1) \qquad O$ $C1 \qquad O$	MIC (μ M): 64b -1: DS = 2.49, MDR = 9.91, XDR = 39.72, 64b -2: DS = 2.28, MDR = 18.14, XDR = 36.31, IC ₅₀ (μ M) for InhA: 64b -1: 3.90 \pm 0.09, 64b -2: 2.47 \pm 0.11	MIC (μ M): Isoniazid: DS = 0.88, MDR = Resistant, XDR = resistant, IC ₅₀ (μ M) for InhA: triclosan = 1.22 \pm 0.13	73
HN S N (65)	$\mathbf{R} = -\mathbf{C}\mathbf{F}_3$	MIC (μg mL ⁻¹): Staphylococcus aureus = 25, Bacillus subtilis = 62.5, Clostridium tetani = 100, Salmonella typhi = 100, Escherichia coli = 12.5, Vibrio cholerae = 250	Norfloxacin MIC ($\mu g \ mL^{-1}$): Staphylococcus aureus = 100, Bacillus subtilis = 10, Clostridium tetani = 50, Salmonella typhi = 10, Escherichia coli = 10, Vibrio cholerae = 10	74

66.45, HT29 = 93.39, COLO 205 = 76.23, SF-295 = 70.86, LOX IMVI = 94.77, SK, MEL-2 = 97.36, CAKI-1 = 84.64 and MDA-MB-468 = 98.91. Hence, potent antileukemic agents were reported (Table 1). Thiophene clubbed s-triazine derivatives (52) were synthesized as anticancer agents and their biological activities were evaluated against human lung adenocarcinoma cells (A549). PIC 50 of the series ranged from 4.29 to 6.70, enlightening the range of their potency. In addition to *in vitro* analysis, 3D-QSAR, *in silico* analysis, and molecular docking were also performed in order to gain more output from compound structure–activity relationships and their binding modes with targeted sites. It was finalized from these findings that conjugates with benzimidazole are more potent than other analogs like substituted benzene ring and aminopyrimidine (Table 1).

Bioconjugates of pyrazole clubbed s-triazine (53) with an addition motif of indole ring were designed and its antiproliferation activity was evaluated as a kinase inhibitor. 60 For this purpose, the inhibitory effect was evaluated against two enzymes, EGFR and CDK. The cytotoxicity of conjugates was also evaluated against three different cancer cell-lines and results were obtained to be A549 = 2.40 \pm 0.64 μ M, MCF-7 = 3.28 \pm 0.16 μ M, HDFs = 3.78 \pm 0.55 μ M, EGFR = 34.1 \pm 1.58 nM, and CDK-2 = 108.3 \pm 3.12 nM for the most prominent conjugate. Through SAR analysis, it was noted that conjugate

having the amalgam of indole, triazine, and pyrazole was proven to be the most effective in controlling cancerous cells via apoptosis. The *in silico* analysis was also performed in order to analyze the binding affinities (Table 1). Deregulation or overexpression of different isomeric forms of carbonic anhydrase leads to diseases like glaucoma, epilepsy, and cancer. To target and inhibit the overexpression of this enzyme, sulfonamidebased hybrids of s-triazine were reported.⁶¹ The biological activity of these compounds was evaluated against breast cancer (MDA-MB-468) and leukemia (CCRF-CM) cell-lines. Two hits were observed in the whole set of compounds where compound 54a when $R = N_1N$ -dimethyl and $R^1 = H$ with cytotoxicity observed to be MDA-MB-468 = $3.99 \pm 0.21 \,\mu\text{M}$, CCRF-CM = 4.51 \pm 0.24 μ M and the second one is 54b when R = H and R¹ = F with cytotoxicity value noted to be MDA-MB-468 = 1.48 \pm 0.08 μ M and CCRF-CM = 9.83 \pm 0.52 μ M. The K_i value of 54a and 54b for enzyme hCA-IX is reported to be 190.0 and 38.8 nM respectively. Compound 54b was also able to arrest the G_0 - G_1 and S-phase of the cell cycle (Table 1).

The molecular hybrid of tetrazole and *s*-triazine $(55)^{62}$ was prepared in good yield and their biological activities against cancer cell-lines Huh-7 were checked by MTT analysis which was obtained to be Huh-7 0.4 μ M. The compound possessing a 5-phenyltetrazol-2-ylacetohydrazide moiety showed good

cytotoxic activity. These compounds did not show any cytotoxic activity against normal cells which depicts that these compounds are safe to be used in further clinical trials (Table 1). To the same degree, the molecular hybrids (56) of s-triazine, morpholine, and piperazine were synthesized with good yield. In addition to these three pharmacophores, sulfonamide moiety is also present in these novel molecular hybrids.⁶³ After successful synthesis, these molecular hybrids were evaluated kinase inhibitory action against enzymes i.e., PI3K/mTOR, and four different cancer cell-lines namely, triple-negative breast cancer cells, human breast cancer cells, human cervical cancer cells, and human liver cancer cells and the most efficient potency was found against the human cervical cancer cells (HeLa) with cytotoxicity HeLa $= 2.21 \pm 0.45$ μ M. Evaluating SAR it was noted that a compound with 4-OH substituent was able to arrest the G1 phase of HeLa cell-line due to strong electrondonating effect of the hydroxyl group at position 4. The IC₅₀ of this compound is PI3K $\alpha = 3.41 \pm 0.05$ nM and mTOR = 8.45 \pm 1.28 nM. This compound was able to induce apoptosis in cancer cell-lines eventually leading to a reduction in migration

3.2 Antibacterial activity

and invasion toward normal cells (Table 1).

The prodigious phenomenon shown by bacteria to resist most of the available antibiotics has made a hot issue for scientists.⁶⁴ Multidrug resistance (MDR) is known as a global crisis. Although, as with the passage of time resistance is increasing, the clinical trials for new drugs are decreasing. The development of novel antimicrobial agents with minimal financial investment has always been a viable solution to deal with this situation.⁶⁵ Numerous hybrid analogs of *s*-triazine combined with various other pharmacophores as part of this aim have been developed with potent antimicrobial activities. Some of the latest antibacterial drug candidates have been reported in Table 2.

To cope with developed resistance in bacterial strains, molecular hybrids of s-triazine and pyrrole (57) were synthesized and screened for their biological activities against 27 different kinds of S. aureus.66 The biological screening came out with results indicating the conjugates were potent inhibitors. Through the structure-activity relationship, it was noted that the presence of a single imidazole ring in their structure enhanced the activities. The minium inhibitory concentration of this prominent conjugate with pyrazole ring against S. aureus is outed to be ATCC 33591 = 3.9, MRSA BRS3 = 0.95, MRSA C1 =1.95, MRSA C7 = 3.9, MRSA C14 = 3.9, and MRSA G_1 = 7.8 μ M (Table 2). New molecular hybrids of s-triazine and quinazoline were synthesized starting from anthranilic acid. After complete synthesis and characterization, these molecular hybrids (58)67 were screened for their biological activity again 5 Gram-positive and 2 Gram-negative strains using Saboraud's medium and Nutrient broth method, and ZOI (zone of inhibition) was noted. After a complete screening and activity SAR analysis, it was confirmed that the compound with $R^1 = H$ and $R^2 = H$ is the most potent. Zone of inhibition in (mm) can be summarized as Gram positive strains *S. aureus* = 14.84 ± 0.3 , *S. faecalis* = 14.84

 \pm 0.4, *B. substilis* = 14.4 \pm 0.3, *P. vulgaris* = 15.64 \pm 0.8, and *B. pumilus* = 16.64 \pm 0.4 while for Gram negative, *E. coli* = 14.04 \pm 0.44, and *K. pneumoniae* = 11.04 \pm 0.43 (Table 2).

Herein, fused rings molecular hybrids (59) of pyrimidine and triazine are reported.⁶⁸ Starting from 4-oxopyrimidin-2-ylthioureas and respective aryl aldehydes, these conjugates were synthesized in good yield and their biological activities were evaluated against E. coli and S. aureus, and their minimum inhibition concentration was noted as (MIC) $\mu g \text{ mL}^{-1}$, E. coli =256, and S. aureus = 256 (Table 2). Being a versatile scaffold, striazine is found to be a very potent antibacterial agent. To evaluate this antibacterial activity, molecular hybrids [(60a) and (60b)] of s-triazine and pyrimidine with additional sulfonamide moiety69 were synthesized in good yield and characterized theoretically and experimentally. Through SAR analysis, it was observed that when s-triazine is tethered to the pyrimidine ring its activities are enhanced meanwhile, when the sulfonamide group is attached activity is also increased since it has a synergic effect on the s-triazine ring. Bioassays were performed, zone of inhibition was calculated and following results were obtained for 60a: E. $coli = 30 \pm 0.42$ mm, B. $subtillis = 28 \pm 0.22$ mm, S. pyogenes = 19 ± 0.56 mm, P. aregenosa, and for 60b: E. coli = 35 \pm 0.12 mm, *B. subtillis* = 18 \pm 0.32 mm, and *S. pyogenes* = 20 \pm 0.56 mm (Table 2).

In order to develop new potent antibacterial agents, Asadi and coworkers in 2023, synthesized a series (61) bioconjugates of s-triazine clubbed phthalimide via N-α-amino acids spacer.⁷⁰ These conjugates were synthesized in moderate to good yield and characterized through spectroscopic techniques. The biological activities of these compounds against 4 strains of bacteria (2 Gram-positive and 2 Gram-negative) were evaluated. Through SAR studies, it was concluded that a compound possessing a hydrophobic iso-butyl chain is the most potent and efficient among the series. The MIC values for this compound were noted which are MIC ($\mu g \mu L^{-1}$): E. coli = 1.25, S. enteritidis = 1.25, S. aureus = 1.25, and B. subtilis = 1.25. In silico docking analyses were also performed with DNA gyrase enzyme to check interactions between ligand and DNA gyrase and the presence of imide, amide, and nitrogen atoms within the ring of striazine was highly appreciated for forming hydrogen bonding interactions (Table 2).

Likewise, molecular hybrids of s-triazine and imidazole (62) are also reported as quaternary ammonium salts as potent antibacterial agents.71 Using the sustainable methodology, 2,4dichloro-6-alkylamine-s-triazine was reacted with methyl imidazole to synthesize the targeted products in a very good yield of \sim 90% in lesser time of 1.5 hours involving haloalkanes as the reactant. The chain length of the alkyl group ranged from C10-C18. During bioassay, the compound with a carbon chain of 14 carbon atoms (tetradecane) came out to be the potent one. These compounds were then evaluated against Gram-positive and Gram-negative bacterial cultures and MIC in (mg L⁻¹) was evaluated to be S. aureus = 3.90, E. faecalis = 62.5, E. coli =62.5, and P. aeruginosa = 250 (Table 2). Patel and coworkers reported the synthesis of molecular hybrids of quinazoline (63) and s-triazine and additional semi-carbazide moiety.72 After complete synthesis and characterization through different

RSC Advances Review

 Table 3
 s-Triazine-based molecular hybrids as antifungal and antiviral agent

	Most potent compound			
Compounds	Substituents	MIC/zone of inhibition	MIC/ZOI of ref. drug	Refs.
Ar NH ₂ H NH N-N (66)	66a 66b Ar = NO;	The MIC ($\mu g \ mL^{-1}$) against Candida albicans: $66a = 1.475 \times 10^{-8}, \ 66b = 2.1851 \times 10^{-4}, \ (IC_{50} \ \mu g \ mL^{-1})$ for CYP51 inhibition: $66a = 7.451 \pm 0.404, \ 66b = 7.451 \pm 0.183$	The MIC ($\mu g\ mL^{-1}$) against Candida albicans: fluconazole = 0.857, (IC ₅₀ $\mu g\ mL^{-1}$) for CYP51 inhibition, fluconazole = 1.614 \pm 0.087	77
F F (67)	_	MIC (μ g mL ⁻¹): C. albicans SC5314 = 16.08, C. glabrata = 4, C. tropicalis = 8, C. parapsilosis = 8, C. dubliniensis = 8, S. cerevisiae BY4741 = 4	_	58 and 78
N _≥ C CH ₃ (68)	$\mathbf{X} = -\mathbf{C}\mathbf{H}_2$ -	MIC (mg mL $^{-1}$): <i>C. albicans</i> = 3.191	-	79
HN S HN S (65)	$\mathbf{R} = -\mathbf{C}\mathbf{F}_3$	MIC ($\mu g \ mL^{-1}$): Candida albicans = 250	Griseofulvin MIC ($\mu g \ mL^{-1}$): Candida albicans = 500	74
R N N N N N N N N N N N N N N N N N N N	R = 2-OH	$IC_{50} = 12.1 \; \mu\text{M, } CC_{50} > 400 \mu\text{M}$	Remdesivir: IC $_{50} = 0.07 \pm 0.04$ $\mu\text{M}, \text{ CC}_{50}$ > 94.9 μM	80
Br HN Br NH NH (63)	R = F	IC_{50} (μg mL ⁻¹): HIV-1 (III _B) = 49.13, HIV-2 (ROD) = 49.13, CC_{50} (μg mL ⁻¹): HIV-1 (III _B) = 49.13, HIV-2 (ROD) = 49.13	Azidothymidine IC_{50} (µg mL^{-1}): HIV-1 (III _B) = 0.0049, HIV-2 (ROD) = 0.0061, CC_{50} (µg mL^{-1}), HIV-1 (III _B) = 0.0074, HIV-2 (ROD) = 0.010	72
CHCl ₂ R (70) R ¹	$R = R^1 =$	EC_{50} (μ M): human cytomegalovirus > 1.20, herpes simplex virus 1 > 1.20, Varicellazoster virus > 1.20	EC_{50} (µM) Ganciclovir: human cytomegalovirus = 4.65, acyclovir: herpes simplex virus 1 = 0.78, varicella-zoster virus = 3.28	81

spectroscopic techniques. In SAR analysis, it was noted that the compound possessing 4-methyl cyclohexylamine substituent is the most efficient in comparison to its contrasting scaffold-like aromatic and aliphatic amine. These compounds were evaluated for their biological activities as antibacterial agents against Gram-positive as well as negative strains, and the minimum inhibition concentration (MIC) of the most highlighted compound was noted to be (MIC in μg mL⁻¹) *S. aureus* = 6.25, *P. aeruginosa* = 12.5, *B. cereus* = 100, and *K. pneumoniae* = 100 which are quite comparable with that of standard drug ciprofloxacin (Table 2). Analogs of these compounds *i.e.*, thiosemicarbazides were also synthesized and evaluated in the same way. It was concluded that thiosemicarbazides exhibited

more anti-HIV properties than semicarbazides (Table 3). To combat infectious diseases like tuberculosis, 3 new series of s-triazine pendant thiazolidin-4-one were synthesized in good vield starting from thioureas. This thiazolidine-4-one was then fused with triazine to form thiazolo[3,2-a][1,3,5]triazine. This fused ring was then further clubbed with other heterocycles and sulfonamide moiety to enhance the biological activity.73 These compounds were then evaluated for their activity against Mycobacterium tuberculosis (Mtb) drug-sensitive (DS), extensive drugresistant (XDR), and multi-drug resistant (MDR) strains. In SAR, it was noted compounds with an additional heterocyclic ring furan or an electron-withdrawing group at position 4 of the benzene ring are proven to enhance the biological activity of the compound while any electron-donating group decreases the activities. The MIC in μ M is found to be DS = 2.49, MDR = 9.91, and XDR = 39.72 for 64b-1 while for the 64b-2 DS = 2.28, MDR =18.14, and XDR = 36.3. In addition to tuberculosis, these compounds were also screened for their activity against bronchitis-causing bacteria. Compounds 64b-1 and 64-2 were also screened as inhibiting agents for M. tuberculosis enoyl-acyl carrier protein reductase (InhA) and IC₅₀ (μM) for InhA, 64b-1: 3.90 ± 0.09 and for 64b-2: 2.47 ± 0.11 is calculated (Table 2). In the same manner, two novel series of molecular hybrids (65) of s-triazine with benzothiazole and s-triazine with coumarin were synthesized.74 After complete characterization through FTIR, NMR, and mass spectrometry these novel bioconjugates were further analyzed for their biological activities as a potent medicinal conjugate. From obtained data and in vitro screening it was noted that these series exhibit a remarkable antibacterial activity with MIC in the range of 12.5-62.5 µM and antifungal activity with MIC in the range of 100-200 µM. With the evaluation of SAR analyses, it was noted that hybrid of s-triazine and benzothiazole with electron-withdrawing substituent -CF3 denoted the most promising activities. MIC ($\mu g \ mL^{-1}$) for this compound is calculated to be Staphylococcus aureus = 25, Bacillus subtilis = 62.5, Clostridium tetani = 100, Salmonella typhi = 100, Escherichia coli = 12.5, and *Vibrio cholerae* = 250 (Table 2). These compounds were also evaluated for their anti-fungal activities MIC against C. albicans came to be in the range of 100-200 µM (Table 3).

3.3 Antifungal and antiviral activities

Another major issue in the public health problem is increased due to the ubiquity nature of many fungal species. The developed antifungal drug resistance is the sequel of uncontrolled use of antifungal agents.⁷⁵ Likewise, CoV-2 is a severe acute respiratory syndrome coronavirus type-2 which was discovered in late 2019. This virus is the cause of the pandemic and global crisis.⁷⁶ s-Triazine based drug-candidates are also used as an antiviral agent for this virus. Herein, in Table 3 some notable hybrids of s-triazine are tabulated as potent antifungal and antiviral agents.

Novel compounds (66) of s-triazine conjugated with tetrazole were synthesized from microwave-assisted one-pot synthesis using amino-tetrazole, aromatic/heteroaromatic aldehydes, and cyanamide in acetic acid as a solvent in moderate to good yield.77 The greenly synthesized new conjugates were screened against C. albicans in comparison to fluconazole as an antifungal agent. While evaluating the SAR, it was noted that the compound with the nitro group present at the meta-position appeared to be the most potent among the set, their MIC was noted in ($\mu g \text{ mL}^{-1}$) **66a** = 1.475 × 10⁻⁸ for **66b** = 2.1851 × 10⁻⁴. IC₅₀ in μg mL⁻¹ for enzyme CYP51 inhibition was also calculated $66a = 7.451 \pm 0.404$ and for $66b = 7.451 \pm 0.183$. In addition to this, in vitro analysis against lanosterol 14-α-demethylase (CYP51 protein) braced antifungal analysis (Table 3). Compound (67) Me156 (ref. 78) is an s-triazine-based synthetic compound that was screened for its biological activities against human pathogenic fungus specie namely, C. albicans, several nonpathogenic non-albicans Candida strains, and S. cerevisiae and minimum inhibitory concentration (MIC) was noted as C. albicans SC5314 = 16.08, C. glabrata = 4, C. tropicalis = 8, C. parapsilosis = 8, C. dubliniensis = 8, and S. cerevisiae BY4741 = 4 ($\mu g \text{ mL}^{-1}$). It was concluded that (67) inhibits yeast-hyphae transition at significant MIC (Table 3). Molecular hybrids of striazine (68) with different heterocycles of 6-membered rings were reported by Dabade and his coworkers.⁷⁹ In their research work, the concoction of genetic algorithm multiple-linearregression was used to investigate QSAR and molecular docking of these s-triazine molecular hybrids which braced their motive of potent inhibitors of C. albicans fungal strains. SAR analysis suggested that when the attached 6-memebered heterocyclic ring is piperidine the activity is enhanced remarkably in comparison to other analogs like piperazine. The MIC of this piperidine tethered s-triazine ring is for C. albicans = 3.191 mg mL^{-1} (Table 3).

To cure severe acute respiratory syndromes 3 new series (69) of the *s*-triazine pendant morpholine were synthesized in good yield and its biological activity was screened against SAR-CoV-2. SAR suggested one derivative from the series with a strong electron-donating substituent like 2-OH group came out to be potent with the least inhibitory concentration and less toxicity against Caco-2 cells. *In vitro* ADME analysis and inhibition against helicase enzyme were also studied and $IC_{50} = 12.1 \,\mu\text{M}$ and $IC_{50} = 12.1 \,\mu\text{M$

RSC Advances

Table 4 s-Triazine-based molecular hybrids as antioxidants and anti-Alzheimer agents

	Most potent compound			
Compounds	Substituents	IC_{50}	IC ₅₀ ref. drug	Refs.
O N N N R R R (71)	$R = \frac{\overset{\checkmark}{HN}}{\overset{\dagger}{NH_2}}$	(IC50 μM): BuChE $= 1.0 \pm 0.1$	Donepezil (IC $_{50}$ μ M): BuChE = 2.9 \pm 0.5	85
NH NN NCI	$\mathbf{R} = \left\langle \mathbf{r}^{\mathbf{r}^{\mathbf{r}}} \mathbf{N} \right\rangle$	$IC_{50} = 0.044~\mu M$	Donepezil: IC ₅₀ = 0.052 μ M	86
\mathbb{R}^2 \mathbb{R}^1 \mathbb{R}^1 \mathbb{R}^1	$R^{1} = \bigvee_{N \text{ in } N}$ $R^{2} = \bigvee_{N \text{ in } N}$ NH	IC50 (μ M) hAChE = 10.82 \pm 0.98, hBuChE = 0.046 \pm 0.002	Donepezil: hAChE = 0.027 \pm 0.004, hBuChE = 3.17 \pm 0.06	87
NH ₂ (74)	$R = \sqrt{\frac{1}{2}}$ $n = 1$	$K_{\rm i}=11~{ m nM}$	_	88
HN NH (75)	R = F	$K_{\rm i}=8~{ m nM}$	_	89
R ¹ R ²	$R^{1} = \prod_{\mathbf{F}_{3}\mathbf{C}} \mathbf{R}^{2} = \prod_{\mathbf{F}_{3}\mathbf{C}} \mathbf{R}^{2}$	IC ₅₀ (μ M) EeAChE = 8.49 \pm 0.01, EqBuChE = 24.79 \pm 0.28, DPPH radical scavenger = 16.72 \pm 1.39	IC $_{50}$ (μ M) Galantamine: EeAChE = 5.0 ± 0.295 , EqBuChE = 20.96 ± 0.45 , ascorbic acid: DPPH radical scavenger = 10.74 ± 0.36	90
CI (77)	R 77a = Lys(Boc) 77b = Asp(tBu) 77c = His(Ts)	$\begin{split} & \text{IC}_{50} \text{ (mM) AChE: } 77\text{a} = 0.055 \\ & \pm 0.001, 77\text{b} = 0.065 \pm 0.002, \\ & 77\text{c} = 0.067 \pm 0.003, \text{ BACE1:} \\ & 77\text{a} = 11.09 \pm 2.29, 77\text{b} = \\ & 33.82 \pm 3.91, 77\text{c} = 14.25 \pm \\ & 3.45 \end{split}$	IC ₅₀ (mM) AChE: Tacrine = 0.046 \pm 0.013, donepezil = 0.274 \pm 0.08, BACE1: quercetin = 4.89 \pm 2.31	91
S N R ¹	$R^{2} = HO$ $R^{1} = H_{3}C^{-\frac{2}{5}}$	IC $_{50}$ (µg mL $^{-1}$): DPPH $=25.2\pm3.1$	IC $_{50}$ (µg mL) butylated hydroxytoluene: DPPH $=$ 25.2 \pm 3.1	92

virus. SAR suggested that compounds with simple phenyl rings are more potent than other compounds. EC₅₀ values of these compounds came out to be human cytomegalovirus >1.20 μ M, herpes simplex virus 1 > 1.20 μ M, and Varicella-zoster virus >1.20 μ M (Table 3).

3.4 Anti-alzheimer and antioxidant activity

Alzheimer's disease (AD) is a multifactorial neurodegenerative condition that is defined by loss of memory and other perceptive abilities. S2 Acetylcholinesterase or butyrylcholinesterase can lead to hydrolysis of acetylcholine and ultimately to AD. S3 Likewise, some patients suffering from Alzheimer's are reported to have a high level of free radicals and oxidating agents which promote oxidative stress in the brain. S4 S-Triazine-based molecular hybrids are multi-targeted ligands that can act as an antioxidant as well as anti-acetylcholinesterase agents (Table 4).

Carboline has three fused ring structures with nitrogen as a heteroatom. The molecular hybrids of s-triazine and carboline (71)85 were prepared through a linker approach in good yield and screened in vitro against cholinesterase enzymes (acetyl and butryl). The IC₅₀ value of these compounds ranged between 1.0 and 18.8 μM with the lowest value of 1.0 \pm 0.1 μM which is more efficient than the standard drug Donepezil, $IC_{50} = 2.9 \pm 0.5 \mu M$. Compound with a free amino group $(R = NHNH_2)$ appeared to be the most potent among the series and virtual docking analysis also supported the bio-screening showing this compound binds in the same site with efficient binding where butyrylthiocholine can bind (Table 4). Benzimidazole is a highly biologically active moiety. This potent moiety was clubbed with the magical scaffold s-triazine to synthesize a new series (72).86 A total of 24 compounds were synthesized and characterized via spectroscopic methods. This series was designed as multi-target ligands to treat Alzheimer's disease. Using colorimetric Ellman's method, synthesized compounds were screened for their cholinesterase inhibition activity. It was evaluated through SAR that the presence of cyclohexyl amine makes the inhibitory action even more potent than the standard drug i.e., IC50 for this compound is 0.044 μ M while Donepezil, IC₅₀ = 0.052 μ M. In silico ADMET, docking, and MD stimulation suggested these compounds as efficient anti-Alzheimer drug candidates (Table 4).

Quinoline is a fused bicyclic, nitrogen-containing heterocyclic compound. This heterocyclic compound was linked with striazine to form 16 new s-triazine-quinoline molecular hybrids (73). Using colorimetric Ellman's method, synthesized compounds were screened for their cholinesterase inhibition activity, SAR suggested that when $\mathbf{R^1}$ is pyrrolidine and $\mathbf{R^2}$ the linker between s-triazine and quinoline is piperidin-3-ylmethanamine, the anti-Alzheimer potency is remarkably enhanced than any other amino group present as the linker and the IC50 value was noted to be hAChE = 10.82 \pm 0.98 μ M and hBuChE = 0.046 \pm 0.002 μ M. Docking analysis suggested that this compound can interact with and block the active site of butyrylcholinesterase (Table 4). 5-HT6 receptor is abundant in CNS and maintains the presence of neurotransmitters like

GABA, dopamine, and acetylcholine which are notable in AD. A decrease in the density of this enzyme improves the mental health of Alzheimer's patients. For this purpose, s-triazine-based compounds (74) were synthesized with N-methylpiper-azine scaffold with additional ether linkages. In vitro analysis of these conjugates was performed against the 5-HT $_6$ receptor, and it was noted that when alkyl chains are as short as only one carbon atom and the aromatic ring contains a bulky group like iso-propyl the inhibitory action is enhanced to a prominent level the inhibitor constant (K_i) was calculated to be 11 nM (Table 4).

Likewise, keeping in view the psychiatric function of the 5-HT₇ receptor, novel s-triazine conjugated benzo-pyrrole compounds (75) were synthesized in good yield and their biological activity was screened in vitro against 5-HT₇ receptor and inhibitor constant was noted. From SAR it was evaluated that fluoro-substituent and benzo-pyrrole ring enhances the inhibitory actions of ligand.89 Ki of this compound was evaluated to be 8 nM (Table 4). Reddy and his coworkers reported the molecular hybrids of morpholine and s-triazine (76) to target Alzheimer's disease.90 These molecular hybrids were synthesized in good yield and their inhibitory effect was evaluated against enzymes (acetylcholinesterase and butyrylcholinesterase). Moreover, these compounds were also tested for their antioxidant properties and DPPH free radical scavenger. The most potent compounds in the series exhibited the IC50 for AChE and antioxidants in the range of 7.23 to 10.35 μM and 14.80-27.22 μM respectively. Excellent in vitro and SAR analysis results indicated that the compound with substituted pyridine and R² being simple phenyl ring are far more potent than other similar compounds with IC₅₀ EeAChE = $8.49 \pm 0.01 \mu M$, EqBuChE = 24.79 \pm 0.28 μ M, and DPPH radical scavenger = 16.72 \pm 1.39 μM (Table 4).

A new series containing molecular hybrids of s-triazine and piperazine was synthesized (77).91 The attributes of this series are that it contains nitrogen mustard and dipeptide residue to enhance the biological activities as anti-Alzheimer. These compounds were tested for their biological activities against enzymes related to Alzheimer i.e., acetylcholinesterase and βsecretase. Compounds with amino acids lysine, aspartic acid, and histidine exhibited highly efficient potency in comparison to standard drugs with IC₅₀ (mM) AChE: $77a = 0.055 \pm 0.001$, $77\mathbf{b} = 0.065 \pm 0.002, 77\mathbf{c} = 0.067 \pm 0.003, \text{BACE1: } 77\mathbf{a} = 11.09 \pm 0.003, \text$ 2.29, $77\mathbf{b} = 33.82 \pm 3.91$, and $77\mathbf{c} = 14.25 \pm 3.45$ (Table 4). Compounds (78) are the molecular hybrids of s-triazine and 1,3,5-triazine which are fused with each other giving rise to novel pharmacophores.92 These bioconjugates were synthesized in good yield via the conventional method and their biological activity was examined as DPPH free-radical scavenger, that is to say, antioxidant. After complete examination and evaluation, and through SAR analysis it was confirmed that compounds possessing hydroxyl group in their structures depicted more efficiency as a free radical scavenger in in vitro analysis than other compounds in the same series due to electron-donating effect. IC₅₀ of this compound was noted to be 25.2 \pm 3.1 μg mL⁻¹ and compared with standard antioxidant drugs i.e., BHT $25.2 \pm 3.1 \ \mu g \ mL^{-1}$ (Table 4).

Table 5 s-Triazine-based molecular hybrids as antioxidants and anti-Alzheimer agents

	Most potent compound			
Compounds	Substituents	IC_{50}	IC ₅₀ ref. drug	Refs.
Y NH HOOC NH (79)	Y =	$IC_{50}=9.54~\mu g~mL^{-1}$	_	96
R ¹ NH NH HOOC HO (80)	$R^{2} = \frac{HN}{N} \int_{S^{2}}^{N} ds$ $R^{1} = \frac{N}{N} \int_{S^{2}}^{N} ds$ $R^{2} = \frac{N}{N} \int_{S^{2}}^{N} ds$ $R^{1} = \frac{N}{N} \int_{S^{2}}^{N} ds$	IC ₅₀ (μ M) 80a : 3D7 = 13.25 \pm 0.82, Dd2 = 16.30 \pm 0.93, 80b : 3D7 = 88.74 \pm 2.14, Dd2 = 14.72 \pm 0.82	Chloroquine: $3D7 = 2.18 \pm 0.12$, $Dd2 = 3.70 \pm 0.13$	97
N N R ² (81) OH	R = Cl	Dead rings trophozoites IC_{50} ($\mu g~mL^{-1}$) = 53.85 ± 0.42 , Vero cells: CC50 ($\mu g~mL^{-1}$) = 550.26 , SI index = 10.21	Chloroquine: Dead rings trophozoites: IC_{50} (µg mL ⁻¹) = 0.7 ± 0.004 , Vero cells: CC50 (µg mL ⁻¹) = 560.29 , SI index = 801.28	98
R ² R1 (82)	$R^{2} = N NH$ $R^{1} = F N^{\frac{1}{2}}$	IC_{50} (obs.) = 1.36, $log IC_{50}$ (obs.) = 0.134	_	99

3.5 Antimalarial activity

According to the World Report Malaria-2021 by WHO, in 2020 a major increment in malarial death by 12% was observed estimating 627 000 deaths globally.⁹³ The species which is the main cause of death in Africa and Southeast Asia is *Plasmodium falciparum* (*Pf*). *P. falciparum* is the deadliest plasmodium species that causes malaria in humans.⁹⁴ Plasmodium has developed a great resistance against formerly available drugs and a struggle of decades in going all in vain.⁹⁵ In Table 5 some recently designed antimalarial drug candidates of *s*-triazine are

tabulated which can cope with these resistant plasmodium species and save mankind.

To handle the resistant plasmodium, s-triazine-based novel compounds (79) were designed by Adhikari and coworkers. They designed 120 different compounds, and their biological activity was screened for the Dd2 strain of plasmodium which is a chloroquine-resistant strain and IC_{50} was calculated. SAR suggested that compounds possessing an additional heterocyclic ring of piperidine and a free amino-group on s-triazine have the lowest IC_{50} value i.e., $9.54~\mu g~mL^{-1}$. To support these evaluations, virtual screening like molecular docking against Pf-

Table 6 s-Triazine-based molecular hybrids as anticonvulsant agent

M.	Most potent compound			
Compounds	Substituents	% Protection	IC ₅₀ ref. drug	Refs.
5	R 83a; -F	% Protection MES: $83a = 100$, $83b = 100$, $83c = 100$, $83d = 100$	% Protection MES: valproate = 100, methaqualone = 100	
HZ	83b; -Cl	PTZ: roscovitine = 100 , $83b = 100$, $83c = 100$, $83d = 100$	PTZ: valproate = 100, methaqualone = 100	,
	83c; -CF ₃	Therapeutic index: $83a = 3.35$, $83b = 2.76$, $83c = 5.61$, $83d = 4.41$	Therapeutic index: valproate = 2.59 , methaqualone = 1.40	103
(83)	83d; -NO ₂			
HN (F8)	I	Docking score = -10.79 , Interactions = N-H···O=C (VAL-86), O (ether)···H-O (THR87)	I	104

 Table 7
 s-Triazine-based molecular hybrids as antidiabetic and anti-obesity agent

		Refs.	108	109	110
		IC ₅₀ ref. drug	Acarbose = 118.0 nM, $K_{\rm i} = 132.74 \pm 14.76$ nM	ICso (nM): alogliptin, DPP-4 = 3.56, hERG = >34 μ M	
		$1C_{50}$	$ m IC_{50} = 44.72~nM, extit{K_i} = 41.74 \pm 8.08~nM$	IC ₅₀ (nM): DPP-4 = 2.32, DPP-8 = >25 000, DPP-9 = >25 000, hERG = $505.21 \mu M$	Binding energies (DG): $87 = -9.0$ to 11.2 keal mol ⁻¹ , $88 = -9.2$ keal mol ⁻¹ , $89 = -10.1$ keal mol ⁻¹
s- mazine-based moecdial hybrids as antidiabetic and anti-obesity agent	Most potent compound	Substituents	$R = O_2 N$	$\mathbf{R}=\mathbf{4-F}$	
able / s- mazme-based moecdial myonds		Compounds	H ₂ N S NH ₂ (85)	S NH S NH	MCOOC S NIH (88) MACOOC S NIH (

Review RSC Advances

DHFR-TS protein was performed and interactions were observed (Table 5). Likewise, a new series of 120 compounds of N-(4-aminobenzoyl)-L-glutamic acid pendant s-triazine (80) was successfully synthesized via microwave-assisted synthesis techniques in good yield.97 Before in vitro analysis, these compounds were evaluated through in silico modeling and only 10 compounds out of 120 were selected for further screening. Through SAR analysis, it was observed that this series possessed two main hits, one with piperidine and one with piperazine moiety. These six-membered heterocyclic moieties are the reasons for such splendid antimalarial activities. IC₅₀ of these compounds is obtained as 80a: 3D7 = 13.25 \pm 0.82 μ M, Dd2 = $16.30 \pm 0.93 \,\mu\text{M}$ 81b: $3\text{D7} = 88.74 \pm 2.14 \,\mu\text{M}$, and Dd2 = 14.72 \pm 0.82 μM (Table 5). Gogoi and his coworkers synthesized a novel series of molecular hybrids of pyrazole-tethered striazine (81).98 Completing their wet-lab experiments and characterizations, they moved to virtual analysis of their synthesized molecular hybrids and picked the top-three compounds for in vitro analyses and antimalarial agents. In virtual analyses, it was observed that hydrogen bonding has a major contribution with residual amino acids Asp54, Ile164, Arg122, and Tyr170. In vitro analysis showed that these top-three compounds have IC₅₀ in the range of *P. falciparum* in the range of 53.85 to 100 $\mu g \text{ mL}^{-1}$. SAR suggested that the least IC_{50} 53.85 µg mL⁻¹ value is given by the compound with substituent 4-chloroaniline. This may be due to alkyl-halogen and hydrogen-bond interactions of chlorine and amino group as visualized by molecular docking results (Table 5).

To treat malaria, the major target for medicinal chemists is *Plasmodium falciparum* dihydrofolate reductase (*Pf*-DHFR). Mutation in this enzyme leads to resistance to malaria-causing plasmodium. In this regard, molecular hybrids (82) of *s*-triazine and 4-anilinoquinoline⁹⁹ were synthesized and their biological profiling and SAR suggested that compounds with an addition morpholine ring are highly potent and show remarkable antimalarial activities with $IC_{50} = 1.36 \ \mu g \ mL^{-1}$ and $log \ IC_{50} = 0.134 \ \mu g \ mL^{-1}$ (Table 5).

3.6 Anticonvulsant activity

An abrupt, forceful, erratic movement of the body brought on by the uncontrollable contraction of muscles, particularly linked to brain illnesses like epilepsy is called convulsion. 100,101 Despite the accessibility to advanced anti-epileptic medications, still there are numerous problems for chemists, for instance, cytotoxicity, fatal adverse drug reactions, and patient resistance to well-established drugs. 102 In an attempt to encounter these issues, s-triazine-based conjugates have been represented in Table 6.

To grapple with the health problems due to epilepsy, a new series (83) of s-triazine tethered 1,2,3-triazole was synthesized using click chemistry approach, 103 and their biological activity as anticonvulsant agents was evaluated via maximum electroshock seizure (MES), and pentylenetetrazole (PTZ) induced seizure. In the second phase of analysis, compounds with fluoro (83a), chloro (83b), trifluoro-carbon (83c), and nitro (83d) substituents appeared to be more potent than the standard

drug methaqualone and valproate due to electron-withdrawing effect on the benzene ring. Percentage protection against MES and PTZ was evaluated to be 100% while the therapeutic index for these 4 hits was calculated as 83a = 3.35, 83b = 2.76, 83c =5.61, and 83d = 4.41. Gamma amino butyric acid (GABA) was chosen as the target protein for the molecular docking study of these conjugates because convulsions are usually aroused when there is a low concentration of GABA (Table 6). Through the optimization, multicomponent reactions Knoevenagel condensation (method A) and Bucherer-Bergs (method B), a novel series of compounds (84) with two or three biologically active nuclei of imidazolidine-2,4-dione and thiazolidine-2,4-dione (TZD) were prepared and characterize via FT-IR, ¹HNMR, ¹³CNMR, and elemental analysis. ¹⁰⁴ All the compounds were put through molecular docking experiments for anticonvulsant drug binding (ADB) to the VGCIP i.e., voltage-gated sodium channel inner pore to assess the anticonvulsant activity of these conjugates. According to the outcomes of the in silico molecular interaction conjugate containing the s-triazine scaffold appeared to be the most efficient. The docking score was calculated to be -10.79 kcal mol⁻¹. Major interactions observed are N-H···O=C (VAL-86), O (ether)···H-O (THR87) (Table 6).

3.7 Antidiabetic and anti-obesity activity

Diabetes mellitus is a well-known complex and chronic health issue that greatly adds to high mortality rates by leading to severe health issues, particularly those related to cardiovascular illnesses, renal damage, and paralysis. ¹⁰⁵ Inhibiting the enzyme glucosidase and dipeptidyl peptidase-4, which catalyzes the hydrolysis of starch in the gut, is an efficient treatment method for managing the glucose brought on by type-2 diabetes. ¹⁰⁶ However, the currently FDA-approved medications and inhibitors, including acarbose and voglibose, have a number of unfavorable stomach adverse effects that limit their use. Therefore, an intriguing field of study continues to be the hunt for new and more potent inhibitors with fewer adverse effects and lower costs. ¹⁰⁷ In this regard, *s*-triazine-based molecular hybrids have been synthesized as potent antidiabetic agents (Table 7).

Sulfonamides possess a noble place in the field of medicinal chemistry. This well-potent moiety was clubbed with s-triazine (85) to synthesize molecular hybrids that can serve as multitargeted ligands and cure multiple diseases at the same time.108 Conjugates from this series were evaluated for their inhibitor effect on various enzymes like carbonic anhydrase, acetylcholinesterase, and a-glucosidase and it was concluded form SAR studies that conjugates having an 3-nitro group on linked phenyl ring exhibit more activity than any other analogs due to high electron-withdrawing effect on nitro group. K_i for this conjugate is calculated to be 51.67 \pm 4.76 for hCA-I, 40.35 \pm 5.74 nM for hCA-II, 41.74 \pm 8.08 nM for α -GLY and 335.76 \pm 46.91 nM for AChE (Table 7). Gao et al. reported the molecular hybrids of s-triazine with morpholine and thiazole (86). The bioactive thiazole ring was clubbed with s-triazine via a sulfonamide linker.109 These novel molecular hybrids of sulfonamidethiazole-s-triazine were used as dipeptidyl peptidase-4 (DPP-4)

RSC Advances

Table 8 s-Triazine-based molecular hybrids as anti-inflammatory agent

	Most potent compoun	d		
Compounds	Substituents	% Inhibition	% Inhibition of ref. drug	Refs.
(90)	$R^{2} = \frac{NH}{NH}$ $R^{1} = Cl^{\frac{2}{\xi}}$	% Inhibition at 100 mg kg $^{-1}$, 93% inhibition ($P < 0.001$)	% Inhibition at of indomethacin (10 mg ${\rm kg}^{-1}$), 90% inhibition	118
S N Cl Cl (91)	R = 2,5-difluoro	% Inhibition at 10 μ M, TNF- α = 72%, IL-6 = 79%	% Inhibition of dexamethasone (1 $\mu M)$, TNF- $\alpha=75\%$, IL-6 $=81\%$	119
R O CH ₃ (92)	$R = -OCH_3$	% Change in paw height, 1 h = 0.30 ± 0.02 , 2 h = 0.24 ± 0.01 , 3 h = 0.28 ± 0.01 , % inhibition = 56% , COX1-IC ₅₀ = $6.60~\mu\text{M}$, COX2-IC ₅₀ = $0.080~\mu\text{M}$	_	120

inhibitors to treat diabetes mellitus type II. The in vitro analysis showed that the compound with substituent 4-fluoro has a maximum inhibitory action on DPP-4 with IC₅₀ 2.32 nM which is even lower than that of the standard drug, alogliptin. From in silico analysis, it was observed that this compound also showed prominent interactions with the binding site of DPP-4 with hotspot amino acids Ser630, Asp708, and His740. In vivo analyses of blood glucose lower effect were also evaluated (Table 7).

Obesity is brought on by a confluence of excessive food consumption, inactivity, and hereditary vulnerability. To combat obesity and its consequences Badrey and coworkers synthesized molecular hybrids of the s-triazine pendant with thiazole (87, 88) and thiazine (89).110 After the successful synthesis of these conjugates, these conjugates were evaluated through virtual screening against fat mass and obesityassociated (FTO) proteins to analyze the superimposition of these hybrids with the active site. The results of these molecular docking were given in terms of free binding energies (ΔG), 87 = -9.0 to 11.2 kcal mol⁻¹, 88 = -9.2 kcal mol⁻¹, and 89 = -10.1 kcal mol⁻¹. Moreover, through computer-based studies,

Table 9 s-Triazine-based molecular hybrids as anti-Parkinson agent

	Most potent compo	ound		
Compounds	Substituents	_	IC ₅₀ ref. drug	Refs.
S N-N NH ₂ OH (93)	$\mathbf{R} = \mathbf{4\text{-}NO_2}$	Radioligand binding assay K_i (nM): A ₂ AR = 32.1, A ₁ R = 322.3, Selectivity index = 10.04	_	123
R ^{2-N} NH NC R ¹ (94)	$R^1 = H$ $R^2 = CONH_2$	IC $_{50}$ (µM): SK-3 $\beta=0.17\pm0.02,$ CK-1 $\delta=0.68\pm0.03$	_	124

 Table 10
 s-Triazine-based molecular hybrids as anti-leishmanial agent

	Most potent compound			
Compounds	Substituents	${ m IC}_{50}$	${ m IC}_{50}$ of ref. drug	Refs.
R ¹ H H H R ² (95)	$R^{1} = \begin{cases} x^{2} \\ y^{4} \\ y^{4} \end{cases}$ $R^{2} = \begin{cases} x^{4} \\ y^{4} \\ y^{4} \\ y^{4} \end{cases}$ $n = 2$	IC ₅₀ (μ M): Promastigotes = 6.2 \pm 1.4, Amastigotes = 1.2 \pm 0.5 CC50 (μ M): J774A1 = 145.7 \pm 10.1	Miltefosine IC ₅₀ (μ M): promastigotes = 18.5 \pm 1.1, amastigotes = 2.4 \pm 0.1 CC50 (μ M): J774A1 = 40.5 \pm 1.7, Melarsoprol	126
R ² R ² R ³ Linker R ⁵ R ⁶ R ⁷ R ⁶ R ⁷ R ⁶ R ⁷ R ⁶ R ⁷ R ⁸ R ⁷ R ⁸ R ⁸ R ⁸ R ⁸ R ⁸ R ⁹ R ⁹ R ⁹ R ⁹ R ⁹ R ⁹ R ¹ R ² R ¹ R ² R ³ R ⁴ R ² R ³ R ⁴ R ⁵ R ⁶	$X = NH$ $R^{1} = Me$ $R^{2} = Me$ $R^{3} = Me$ $R^{4} = H$ $R^{5} = H$ $R^{6} = CN$ $R^{7} = H$ $R^{6} = \frac{1}{2}N$ $R^{7} = \frac{1}{2}N$	IC ₅₀ (μM): <i>T. b. bruc.</i> = 0.32, <i>T. b. rhod.</i> = 0.07 CC50: MRC-5 = >64	IC ₅₀ (µM): T. b. bruc. = 0.03, T. b. rhod. = 0.02 CC50: MRC-5 = 7.4	128

RSC Advances Review

ADMET, pharmacokinetic studies, and drug-likeness profiles were also evaluated (Table 7).

3.8 Anti-inflammatory activities

Whenever infectious and pathogenic microbes reside in a specific part of tissues or move around the body *via* the circulatory system they cause inflammation.^{111,112} Other causes of inflammation may include rupturing or damage of tissues, apoptosis or sudden cell death, ischemia, degeneration, or cancer.^{113,114} Both innate as well as adaptive immune systems respond against inflammations.¹¹⁴⁻¹¹⁶ Chronic inflammation is marked by simultaneous sabotage and curing of the damaged and ruptured tissue.¹¹⁷ To encounter the inflammation, molecular hybrids *s*-triazine have been synthesized with prominent biological properties (Table 8).

Molecular hybrids of phenylpyrazole and s-triazine (90) were synthesized using conventional methods and characterized by spectroscopic methods. These nine molecular hybrids were then evaluated for their biological activities as anti-inflammatory agents against the carrageenan-induced paw edema model in male rats. These analyses were compared with a group of rats with indomethacin as standard. SAR analysis suggested that a hybrid with ethyl amine and free chloro group of s-triazine shows a high percentage of inhibition i.e., 93% at 100 mg kg⁻¹ which is greater than the standard drug (Table 8). In the same manner, to cope with inflammation, molecular hybrids (91) of thiazolidine-4-one and striazine were prepared in single step cyclo-condensation reaction of 2-amino-4,6-dichloro-1,3,5-triazine, aromatic aldehydes, and thioglycolic acid.119 These nine molecular hybrids were then evaluated for their biological activities as antiinflammatory agents against the carrageenan-induced paw edema model in male rats. Through the structure-activity relationship, it was evaluated that the 2,5-difluoro substituent present on the phenyl ring of thiazolidine-4-one enhances the inhibition remarkably. Percentage inhibition at 10 µM is evaluated as TNF- $\alpha = 72\%$, and IL-6 = 79% (Table 8). Furochromones have been reported to possess vast applications in medicinal chemistry. Molecular hybrids (92) include the striazine fused with pyrimidine, thiazolidine-4-one, and furochromones. 120 This extensive molecular hybrid was then analyzed for its biological activity as anti-inflammatory affects by evaluating the percentage inhibition of edema and cyclooxygenases (COX1 and COX2). SAR suggested that hybrids possessing strong electron-donating groups like methoxy on furochromones are highly potent in controlling inhibition. The percentage inhibition was calculated after every in-cat's paw which can be summarized as, % change in paw height: 1 h = 0.30 ± 0.02 , $2 \text{ h} = 0.24 \pm 0.01$, $3 \text{ h} = 0.28 \pm 0.01$, % inhibition = 56%. While IC_{50} of enzymes are COX1- $IC_{50} = 6.60 \mu M$, and $COX2-IC_{50} = 0.080 \mu M$ (Table 8).

3.9 Anti-Parkinson activities

Parkinson's disease in one of the neurodegenerative diseases which are caused by the loss of dopamine-secreting neurons.¹²¹ The currently approved drugs are based on the approach as an

alternative to dopamine, but these therapeutic drugs have also been linked to other diseases such as dyskinesia. Enzymes like adenosine A_2A receptor, GSK-3 β , and CK-1 δ are hotspot targets to treat neurodegenerative diseases. To target these enzymes efficient molecular hybrids of s-triazine have been reported which are included in Table 9.

Compounds (93) are molecular hybrids of thiadiazole and striazine which are designed, synthesized, and screened as anti-Parkinson's disease. 123 After completion of synthesis and characterizations, these conjugates were screened as a radioligand binding assay for adenosine receptors A2AR and A1R. After screening it was concluded that conjugates show selective binding toward A₂AR and conjugates with 4-NO₂ substituent show the highest efficacy due to the presence of a strong electron-withdrawing group. The K_i for this compound was evaluated to be $A_2AR = 32.1$ nM, $A_1R = 322.3$ nM, and the selectivity index was 10.04 (Table 9). As mentioned before, GSK- 3β and CK- 1δ are hotspots to cure neurodegenerative disorders. To grapple with these disorders Redenti and coworkers synthesized fused ring molecular hybrids (94) of s-triazine and 1,3,4-triazole and screened these hybrids against GSK-3 \beta and CK-1δ and noted their IC₅₀ values. 124 In vitro analysis confirmed that the compound with amide linkage as R2 is non-cytotoxic, and highest efficacy, and can serve as a hit in drug discovery and development. IC₅₀ for this hit is found to be, SK-3 β = 0.17 \pm 0.02 μ M while CK-1 δ = 0.68 \pm 0.03 μ M (Table 9).

3.10 Anti-leishmanial and anti-sleeping sickness activities (antiprotozoal)

Leishmaniasis is caused by 17 different protozoan species belonging to the genus Leishmania. Many species of Leishmania are resistant to chemotherapy. Moreover, these drugs exhibit high toxicity and other side-effects.125 To cope with this scenario, Barea and co-workers synthesized molecular hybrids (95) of s-triazine and β-carboline alkaloids, and their biological activity was evaluated against promastigote and amastigote forms of Leishmania amazonensis (Table 10). In vitro and SAR analysis showed compound with 4-OMe group at the C1 position of carboline and isopropyl amine linked to s-triazine has the least IC₅₀ value against these forms and low toxicity to J774-A1 cell-lines. IC₅₀ of this compound is Promastigotes = 6.2 ± 1.4 μ M, and for Amastigotes = 1.2 \pm 0.5 μ M, while CC₅₀ is, J774A1 = 145.7 \pm 10.1 μ M. From further findings, it was confirmed that this compound can alter the cell division cycle which ultimately leads to the death of species126 (Table 10).

A parasitic ailment termed human African trypanosomiasis also referred to as sleeping sickness, is spread through the bite of the 'Glossina' bug, also known as the tsetse fly. It generally results in death if untreated. Venkatraj *et al.*, designed, and synthesized dimers (96) of *s*-triazine linked *via* piperazine as a potent antitrypanosomal agent and screened their biological activity through *in vitro* and *in vivo* analyses. It was observed that these dimers show 4–400 times more inhibitory activity as compared to their monomers. IC₅₀ of this dimer is found to be *T. b. bruc.* = 0.32 μ M, *T. b. rhod.* = 0.07 μ M, while CC₅₀ MRC-5 = >64 μ M (Table 10).

4. Conclusion and future perspectives

A library of chemically diverse *s*-triazine-based heterocyclic hybrid molecules has been highlighted as promising drug development candidates. The majority of the summarized hybrid molecules have been reported to show more potent *in vitro* biological activities when compared to the standard drugs. Importantly, the effect of different substituents installed on the heterocyclic hybrid core has also been highlighted. This review having a chemically diverse collection of *s*-triazine-based hybrid molecules may serve as an ideal platform to identify potential leads and therefore can be an important avenue for the development of novel drugs with high efficacy and low toxicity for the treatment of various challenging diseases.

Based on the gilt-edged properties of s-triazine and its presence in a number of already approved commercially available drugs, it can be regarded as the magic nucleus to be incorporated in the design of novel drug candidates. Especially the design of multitarget directed ligands (MTDLs) i.e., the molecules having the capability to act on several targets at a time is a novel approach that is getting increased attention from the scientific community. In this context, the hybrids molecules having multiple heterocyclic moieties including s-triazine may be the ideal MTDLs for future drug design and development.

Data availability

No data was used for the research described in the article.

Author contributions

Muhammad Imran Ali, conceptualization, writing - original draft. Muhammad Moazzam Naseer, conceptualization, supervision, final editing of manuscript.

Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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