

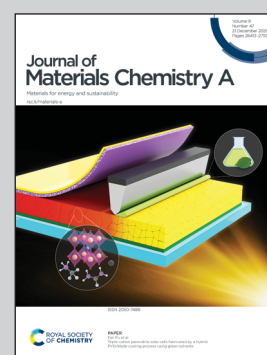


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A synopsis of progressive transition in precursor inks development for metal halide perovskites-based photovoltaic technology

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Perovskite solar cell (PSC) technology has received considerable attention due to the rapid escalation of their solar-to-electrical energy conversion, which has recently surpassed 25% for lab-sized solar cells. Other benefits such as their fabrication through solution processing enable new opportunities for scaling up and rapid production. These features may play a key role in realizing quick installations worldwide, helping to meet the global energy production and consumption demand with a realistic energy pay-back time. This report provides an overview of the progress in developing liquid precursor inks for producing a variety of organic–inorganic halide perovskite-based light absorbing layers. In recent years, a variety of configurations for PSC technology have been reported, where intelligent inks of perovskite precursors have been formulated to facilitate novel designs with impressive solar-to-electrical energy conversions and promising stability. This report highlights the evolution of these novel perovskite precursor ink formulations, and discusses the emerging trends in developing efficient, scalable, and robust PSC technology. Moreover, the classification, advantages, and limitations of various types of perovskite precursor ink are addressed. Specifically, single- and multi-cation-based ink formulations are discussed in relation to their impact on producing efficient solar cells, which provides an overview of the recent progress in the development of this emerging and low-cost solar cell technology. Overall, this synopsis provides the current state of the art in designing novel perovskite precursor inks to be used in producing high performance, efficient, scalable, and stable configurations of perovskite solar cell technology.

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1. Introduction

The growing population and resulting human activity continue to increase energy consumption, imposing new challenges for energy supply and production. Solutions must meet sustainable production criteria, especially under the current climate change scenarios.^{1,2} These stringent conditions have become a source of intense motivation for researchers worldwide to develop clean and alternative energy production schemes focused on reducing greenhouse gas effects and environmental pollution.^{3,4}

Among the various renewable energy sources, photovoltaics has been considered as a potential source for providing clean, silent, and affordable energy with high solar-to-electrical energy conversion.⁵⁻⁷ Despite the increase in installations of dominant Si-based photovoltaic systems, these have various limitations such as saturated performance for bulk outdoor electricity production, limited aesthetics for modern architecture-based building integrated photovoltaic (BIPV) applications, and non-eco-friendly and high energy consumption-based production methods. Moreover, they display limited performance under low light intensity conditions; efficient conversion of electrical energy under simulated light is needed to energize maintenance-free internet of things (IoT) devices and portable electronics.⁸⁻¹¹

In this regard, third generation-based photovoltaic (PV) technologies such as dye-sensitized solar cells (DSSCs), organic solar cells (OSCs) and perovskite solar cells (PSCs) offer numerous advantages over Si-based PV systems.⁹⁻¹¹ For example, they provide flexibility in designing solar cells and

modules in a variety of sizes, their fundamental materials are abundantly available, and they are fabricated with non-vacuum-based scalable production methods such as established slot-die coating, screen printing or inkjet printing-based material deposition technologies.^{12,13} Such unique features make these next generation-based PV technologies a potential source for energizing modern electronics and utilizing their integration in the next generation of BIPV products. They also provide the possibility to produce bulk electricity with much higher solar-to-electrical energy conversion efficiencies compared to stand-alone Si-based PV systems, for example when smartly used in tandem configurations or installed outdoors under natural climatic conditions.¹⁴⁻¹⁶

Among these next generation-based PV technologies, PSCs have received considerable attention due to the rapid escalation of their solar-to-electrical energy conversion, which has recently surpassed 25%.¹⁷ In addition to these reported improvements in conversion efficiencies, other benefits such as their fabrication through solution processing enable new opportunities for scaling up and rapid production.¹⁸ Importantly, this may play a key role in realizing quick installations worldwide, thus providing a realistic way of meeting the global energy production and consumption demand.^{19,20}

This report provides an overview of the progress in developing liquid precursor inks for producing a variety of organic-inorganic halide perovskite-based light absorbing layers in emerging PSC technology. Many configurations for perovskite solar cell technology have been reported in recent years. Specifically, intelligent inks of perovskite precursors have been



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formulated to facilitate novel designs with impressive solar-to-electrical energy conversions and promising stability.^{21–24} This report highlights the evolution of these novel perovskite precursor ink formulations and discusses the emerging trends towards developing efficient and robust PSC technology. In addition, the classification, advantages, and limitations of various types of perovskite precursor ink are addressed. Single- and multi-cation-based ink formulations are discussed in light of their impact on producing efficient solar cells. Progress in developing lead (Pb)-free precursor inks is also reviewed, as well as the scalable precursors that have been identified for producing large-area solar modules in a variety of configurations with popular scalable fabrication methods. Overall, this review provides the most up-to-date advancements being made in designing novel perovskite precursor inks for producing high performance, efficient, scalable, and stable configurations of perovskite solar cell technology.

2. Role of precursor inks in establishing the current state of the art for conversion efficiencies of PSC technology

The progressive advances in PSC solar-to-electrical conversion efficiencies (Fig. 1) may be attributed to many factors, for example, the known knowledge of design principles for producing solid state DSSCs (ssDSSCs),^{25,26} optimizing the active layer thicknesses for improving the diffusion lengths of carriers,^{27–29} or even proposing intelligent strategies for minimizing both the radiative and non-radiative losses observed in device architectures.^{30–35}

The advances in designing novel perovskite precursor inks^{36–39} have played a key role in addressing the aforementioned key issues, thus leading to extraordinary improvements in solar-to-electrical energy conversions in recent years (Fig. 1).

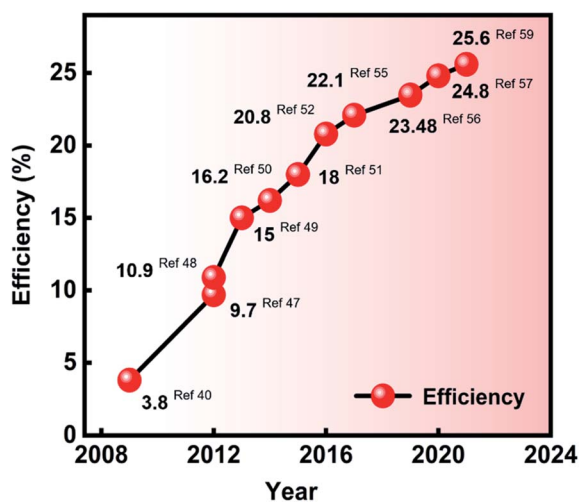


Fig. 1 Solar-to-electrical conversion efficiency evolution of lab-sized perovskite solar cells.

The striking discovery of potential applications of perovskite-based light harvesters in photovoltaics was preliminarily reported by formulating two types of precursor ink.⁴⁰ Briefly, $\text{CH}_3\text{NH}_3\text{Br}$ and PbBr_2 were dissolved in *N,N*-dimethylformamide (DMF) solvent, while $\text{CH}_3\text{NH}_3\text{I}$ and PbI_2 were mixed in γ -butyrolactone (GBL) as a high viscosity solvent, cast on a TiO_2 nanocrystalline film *via* the spin coating method (also known as the single-step deposition method).^{41–43} The solar cells produced in a liquid junction DSSC assembly fashion resulted in 3.81% solar-to-electrical energy conversion when tested under full sunlight illumination.⁴⁰

Following these findings, several additional reports have assessed other novel precursor inks including lead-based compounds. For example, lead iodide (PbI_2) or lead chloride (PbCl_2) have been combined with other organic–inorganic salts such as methylammonium iodide ($\text{CH}_3\text{NH}_3\text{I}$) or cesium iodide (CsI) and a variety of solvents to further improve the solar-to-electrical conversion efficiencies of the fabricated solar cells.^{44–46}

Among these investigations, the study by Kim *et al.* mimics the solid-state dye-sensitized solar cell (ssDSSC) device design, in which the TiO_2 -based electron transport layer (ETL) was first infiltrated with MAI and PbI_2 containing a perovskite precursor ink with a GBL solvent kept overnight while stirring at 60 °C.⁴⁷ The solid-state cell architecture was further processed *via* casting a hole transporting layer (HTL), *i.e.* Spiro-OMeTAD, followed by thermal evaporation of gold to produce the final contact layer. Interestingly, the fabricated solar cells not only exhibited far higher conversion efficiencies (>9%) compared to previous reports, but they also exhibited striking photovoltaic performance stability during periodic measurements when tested without encapsulation for a period of 500 hours in air combined with room temperature (RT)-based ecological conditions.

Building on the outcomes of the aforementioned research, >10% conversion efficiencies were promptly demonstrated for lab-sized PSC devices, also by incorporating a mixed cation-based perovskite precursor solution ink that was used to achieve the perovskite light absorbing layer through a single-step spin coating method.⁴⁸

Unlike the single-step processing-based precursor inks, the pioneering work reported by Burschka *et al.* used a two-step processing scheme. In this approach, the perovskite-based light absorbing layer in the PSC device structure was achieved by fabricating two individual solutions.⁴⁹ The first precursor ink was obtained by dissolving PbI_2 in DMF and was kept at 70 °C before the infiltration in step 1 through spin coating. Next, glass electrodes containing the PbI_2 -infiltrated electron transport layer (ETL, TiO_2) were dipped into a solution of MAI containing 2-propanol solvent for 20 seconds to obtain the desired perovskite crystal-based light harvesting layer. Striking solar-to-electrical conversion efficiencies reached 15% due to achieving a controlled perovskite crystal morphology in response to the previously reported single-step mixed halide perovskite inks.⁴⁹

Since the description of these initial techniques, progressive enhancements in conversion efficiencies have continued. The solar-to-electrical energy conversion milestone of 25% was



Table 1 List of the numerous halide perovskite precursor inks used in the gradual enhancements in the solar-to-electrical conversion efficiencies of lab-sized PSCs in recent years

Perovskite precursor ink recipe	Method of deposition	Cell structure	Active area (cm ²)	η (%)	Ref.
MAPbI ₃ -8 wt% of MAI and PbI ₂ in GBL MAPbBr ₃ -20 wt% of MABr and PbBr ₂ in DMF	Spin-coating	DSSC liquid cell fashion Glass/FTO/TiO ₂ /PVK/thermo-plastic separator film/Pt-coated FTO-glass	0.238	3.81	40
0.395 g synthesized MAI and 1.157 g PbI ₂ mixed in GBL (2 mL) at 60 °C overnight with stirring	Spin-coating	All-solid-state PSC: glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.207	9.7	47
MAI and PbCl ₂ (3 : 1 molar ratio) dissolved in anhydrous DMF → MAPbI ₂ Cl (20 wt%)	Spin-coating	Glass/FTO/c-TiO ₂ /m-Al ₂ O ₃ /PVK/Spiro-OMeTAD/Au	0.09	10.9	48
PbI ₂ in DMF (462 mg mL ⁻¹) (stirring at 70 °C) (spin coating)	Two-step deposition ^a	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.285	15	49
MAI in IPA (10 mg mL ⁻¹) (dipping) 0.8 M MAPb(I _{1-x} Br _x) ₃ (x = 0.1-0.15) stirred in a mixture of GBL and DMSO (7 : 3, v/v) at 60 °C for 12 h	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ -PVK nanocomposite layer/PVK upper layer/PTAA/Au	0.094	16.2	50
H ₂ O (0.5-4 wt% vs. DMF) added to PbI ₂ /DMF (0.8 M) MAI in IPA (50 mg mL ⁻¹)	Two-step deposition ^b	Glass/ITO/PEDOT:PSS/PVK/PC ₇₁ BM/Ca/Al	0.1	18	51
FAI, PbI ₂ , MABr, and PbBr ₂ in mixed solvent DMF and DMSO (4 : 1, v/v) <i>R</i> _{PbI₂/FAI} (molar ratio) varies from 0.85 to 1.54 <i>R</i> _{PbI₂/PbBr₂} fixed at 5.67 <i>R</i> _{MABr/PbBr₂} fixed at 1 : 1 Best: <i>R</i> _{PbI₂=FAI} = 1 : 0.5, corresponding to a 3% weight excess of PbI ₂ in the perovskite	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.16	20.8	52
(FAI) _{0.81} (PbI ₂) _{0.85} (MAPbBr ₃) _{0.15} precursor solution prepared in a glovebox from 1.35 M Pb ²⁺ (PbI ₂ and PbBr ₂) in a mixed solvent of DMF, NMP and DMSO (where the molar ratio of DMF/DMSO is 5 : 1, and the molar ratio of Pb ²⁺ /[(DMSO) _{0.8} (NMP) _{0.2}] is 1 : 1)	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.1051	21.6	53
RbI, pre-dissolved as a 1.5 M stock solution in DMF : DMSO 4 : 1 (v/v), added to the Cs/FA/MA triple cation perovskite to achieve the desired quadruple composition	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.16	21.6 (stabilized)	54
PbI ₂ (1.30 M) containing 2.5 mol% PbBr ₂ dissolved in a 1 mL mixture of DMF and DMSO (8 : 2, v/v) at 80 °C → Spin coating to form a PbI ₂ (PbBr ₂)-DMSO film	Two-step deposition ^b	Glass/FTO/c-TiO ₂ /m-TiO ₂ -PVK nanocomposite layer/PVK upper layer/PTAA/Au	0.096	22.1	55
In the second step, a solution of FAI (80 mg) and MABr (10 mg), containing iodide ions (I ⁻) (0-5 mM) dissolved in 1 mL IPA was spin coated	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/Spiro-OMeTAD/Au	0.0804	23.48	56
Mixing 1.139 mg mL ⁻¹ FAPbI ₃ (0.8 M solution of PbI ₂ and FAI in GBL) in a mixture of DMF and DMSO in a 4 : 1 ratio, v/v. MAcl added in the range of 0-50 mol%	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/HTM/Au	0.0819	24.8 and 22.31 (a large area of 1 cm ² on the active area)	57
1.550 mg mL ⁻¹ FAPbI ₃ and 61 mg MAcl dissolved in a mixture of DMF and DMSO (4 : 1, v/v)	Spin-coating	HTMs: two fluorinated isomeric analogs of Spiro-OMeTAD (Spiro-mF and Spiro-oF)	0.0937	25.2	58
Perovskite solution was prepared by mixing 1.53 M PbI ₂ , 1.4 M FAI, 0.5 M MAcl (as an additive), and 0.0122 M - 0.153 M MAPbBr ₃ in DMF : DMSO = 8 : 1 Bulk perovskite layer is passivated with a 2D perovskite	Spin-coating	Glass/FTO/SnO ₂ /PVK/Spiro-OMeTAD/Au	0.0937	25.2	58



Table 1 (Contd.)

Perovskite precursor ink recipe	Method of deposition	Cell structure	Active area (cm ²)	η (%)	Ref.
For the 2D perovskite passivation, 10 mM of alkylammonium bromide was deposited at 3000 rpm for 30 s The reference perovskite precursor solution was prepared by mixing 1.139 mg FAPbI ₃ and 35 mol% MACI in a mixture of DMF and DMSO (4 : 1). For the formate-doped FAPbI ₃ perovskite film, extra FAHCOO was added to the reference solution in the range of 1–4 mol%	Spin-coating	Glass/FTO/c-TiO ₂ /m-TiO ₂ /PVK/octylammonium iodide/Spiro-OmeTAD/Au	0.0804	25.6	59

^a Two-step deposition: step 1 = spin coating, step 2 = dipping. ^b Two-step deposition: step 1 = spin coating, step 2 = spin coating. PVK = perovskite.

recently surpassed for lab-sized PSCs¹⁷ due to the tremendous efforts being made in developing individual components of this solution-processed PV technology. Designing and developing novel precursor inks for producing high quality perovskite-based light absorbing active layers has remained the focus of reaching high performance and striking stability under several simulated and natural climatic conditions. Fig. 1 summarizes the solar-to-electrical energy conversion evolution. Table 1 provides a list of various perovskite precursor inks used to contribute towards enhancing the conversion efficiencies of lab-sized PSCs in recent years.

3. Emerging research trends in perovskite precursor ink development

There are a variety of precursor inks that have been developed by incorporating various novel materials to produce perovskite-based light absorbing active layers. These precursor inks (Fig. 2) can be categorized as classical single-cation based,^{60–62} multi-cation based,^{63–65} additive based,^{53,66–68} or lead-free.^{69–71}

Traditional PSC devices have been initially reported with a classical ABX₃-based crystalline perovskite active layer. Here, A generally represents methyl-ammonium CH₃NH₃⁺, formamidinium NH₂CH = NH₂⁺ or cesium (Cs)-based monovalent organic or inorganic cations.^{40,46} B typically represents Pb²⁺, Sn²⁺ or even Ge²⁺-based divalent metal cations, and X₃ shares I⁻, Cl⁻, Br⁻ or F⁻ as monovalent anions during the crystal formation of the traditional perovskite-based light absorbing layer.^{72,73}

Due to the inherited characteristics such as stable crystal formations,^{74–78} good chemical compatibility with other active layers, high efficiency and robust stability under initially simulated environmental conditions,^{79,80} single cation-based perovskite precursor inks have been the focus of developing the first generation of lab-sized PSC devices.^{81–83} Interestingly, these first-generation precursor inks exhibited strong processing compatibility when incorporated utilizing both the single-step and two-step fabrication schemes to produce lab-sized PSC devices.^{84–88} Nevertheless, single-step processable precursor inks have remained the preferred choice for scaling up various device designs of PSC technology.^{89,90}

Recently, trends of multi-cation-based ink formulations have also emerged, not only for suppressing non-radiative losses but also for achieving high performance and better stability. This has mainly been achieved through large grain boundary-based crystal formulations, along with physical and chemical stabilities of perovskites as light harvesters in PSCs.^{63,91–94}

Bi *et al.* demonstrated a unique multi-cation-based halide precursor ink by incorporating multi-cations such as formamidinium iodide (FAI) and methylammonium bromide (MABr) salts with lead bromide (PbBr₂) and PbI₂. These were dissolved in a mixed solvent system containing dimethyl formamide (DMF) and dimethyl sulfoxide (DMSO) solvents. This recipe of perovskite precursor ink resulted in PSC devices with >20% conversion efficiency.⁵²

In addition to multi-cations, other novel additives have also been introduced in precursor inks to overcome/achieve some of the key bottlenecks and milestones related to technological

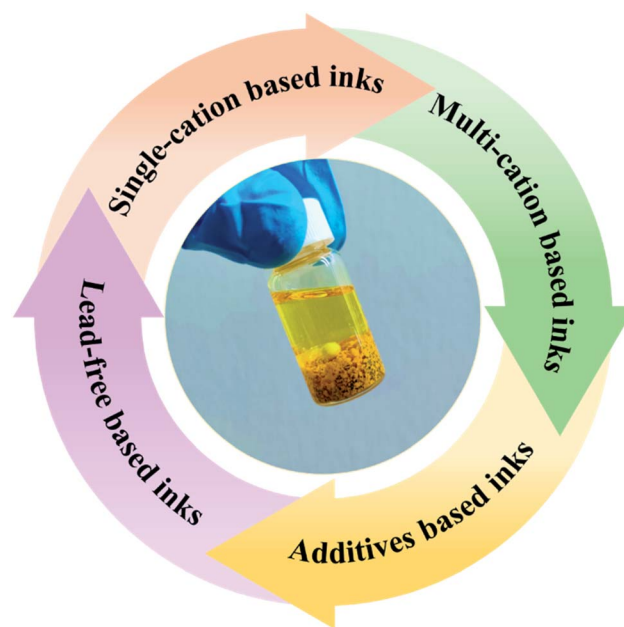


Fig. 2 General classification trends of perovskite precursor inks that have emerged in recent years.



challenges and long-term performance stabilities necessary for the successful commercialization of PSC technology.^{95–99} In this regard, one ground-breaking research study in terms of designing a novel perovskite precursor ink was reported by Mei and co-workers. The striking mesoscopic carbon-based printable perovskite solar cell (CPSC) device design featured the introduction of an additive, *i.e.* 5-ammonium valeric acid iodide (5-AVAI), in the solution recipe of the traditional perovskite precursor ink.²¹ The additive (5-AVAI) induces control over the crystallization process during the infiltration of the formulated perovskite ink in a much thicker and more porous structure compared to the traditional PSC device configurations.^{100–102} As a result, the formulated ink contributed to surpassing the first >1000 hours stability test of these carbon-based printable perovskite solar cells when exposed to ambient air under full sunlight intensity conditions.²¹ The promising characteristics of this device design have yielded beneficial features, including successful scaling up and promising conversion efficiencies demonstrated in PSCs ranging from lab-sized to various module sizes, along with robust stability when deployed under various natural and simulated climatic conditions.^{103–105} Interestingly, 5-AVAI containing a novel precursor ink formulation has also been incorporated with scalable material deposition methods such as inkjet printing or slot-die coating-based technologies.^{106–108}

In contrast to CPSCs, Bi and co-workers used polymer poly(methyl methacrylate) (PMMA) as the templating agent-based additive in the multi-cation-based precursor ink. They demonstrated perovskite films of high electronic quality for producing very high conversion efficiency (>21%) in one of the traditional mesoporous n–i–p PSC devices.⁵³ With the impressive solar-to-electrical energy conversion, the photovoltaic parameters were shown to have an influence upon adjusting the concentration of PMMA, where the champion efficiencies (21.6%) were recorded with a controlled nucleation and crystal growth based light absorbing layer when a specific concentration (0.6 mg mL⁻¹) was used in the fabricated perovskite precursor ink.⁵³

Moreover, the concept of introducing ionic liquids (ILs) in perovskite precursor inks as one of the other promising additive components has also received attention for achieving high performance and robust stability when tested in several configurations.^{99,109–113} For example, Bai and co-workers introduced 1-butyl-3-methylimidazolium tetrafluoroborate (BMIMBF₄) ionic liquid in the multi-cation perovskite precursor ink. They observed striking efficiencies of 20% at the steady state when fabricated as p–i–n configuration-based PSC devices containing this IL-based perovskite light absorbing layer.¹⁰⁹ The inclusion of BMIMBF₄ IL as an additive shows promise for influencing the crystallinity and reducing the defects in the incorporated perovskite-based light absorbing layer in the fabricated PSCs. This also contributed to the stability results when exposed to full-spectrum sunlight in air (RH 40–50%) at 60–65 °C, or full-spectrum sunlight at 70–75 °C based on the environmental testing conditions.¹⁰⁹

Importantly, incorporating ILs in perovskite precursor inks not only offers control over crystallization dynamics and improved charge transport in the perovskite light absorbing

layers, but it also provides a possibility for using them as green solvents for realizing the steps towards making PSCs an eco-friendly technology.^{114–117} For example, Chao and co-workers demonstrated a novel precursor ink utilizing an eco-friendly IL, *i.e.* methylammonium acetate (MAAc), as one of the possible green solvents for producing an efficient and pinhole-free perovskite light absorbing layer. This method revealed promising environmental stability for a period of more than 3000 hours under high humidity (60–80%) conditions.¹¹⁵ Motivated by such promising findings, both n–i–p and p–i–n configuration-based PSCs were fabricated by incorporating this novel precursor ink that showed promising solar-to-electrical energy conversions (~16–20%). Notably, 20.05% is the champion conversion efficiency marked for the n–i–p based planar PSC device design under full sunlight illumination. More impressively, the non-sealed fabricated PSCs also exhibited striking long-term photovoltaic performance stability by retaining 93% of the initial device efficiency under air for more than 1000 hours. A minor deviation (<20%) in the initially achieved efficiency was observed after 700 hours of testing under light stress under the conditions maintained in the glovebox.¹¹⁵

Despite the numerous striking features acknowledged for this emerging low-cost PV technology, one of the key concerns raised relates to incorporating hazardous and toxic compounds such as lead (Pb)-based compounds, which have thus far remained an essential element in producing high performance PSCs.^{118–120} Use of Pb in a PSC, however, restricts its deployments in many portable and consumer electronics-based applications since it may leak from such devices and cause bodily harm.¹²¹ Similarly, deploying PSCs outdoors for bulk electricity generation also requires robust encapsulation procedures due to the solubility of Pb as a heavy metal in water. This may generate a negative environmental impact as indicated in several life cycle assessment reports for PSC technology.^{122–124} Hence, efforts are being made to either reduce or completely replace Pb-based compounds from the next generation of PSC devices.^{125,126}

In this regard, the emerging trends for demonstrating lead-free precursor formulations include tin (Sn)-^{127–129} or germanium (Ge)-based halide perovskite precursor inks.^{130,131} Similarly, trivalent antimony (Sb³⁺)-^{132–134} or bismuth (Bi)-^{135–137} based perovskite precursor formulations have also been explored as possible alternatives for replacing hazardous and toxic lead-based halide perovskite precursor inks.^{125,126,138}

Among these formulations, Sn-based formulations have thus far contributed to a gradual increase in the solar-to-electrical conversion efficiencies,^{128,129,139–141} which has recently exceeded above 13% when tested in inverted p–i–n device design-based PSCs under full sun illumination.¹²⁷ Nevertheless, the long-term photovoltaic performance stability was not demonstrated, hence further investigations are needed for producing robust lead-free PSCs in the near future. Table 2 summarizes versatile ink formulations that have been incorporated in many device designs in recent years to produce high performance perovskite solar cell devices.



Table 2 Classifications of perovskite precursor inks highlighting the champion photovoltaic performance with some of the best recipes used in various PSC configurations

Single cation	Perovskite precursor inks	Deposition method	Perovskite	η (%)	Stability	Ref.
	1.6 M MAPbI ₃ (MAI : PbI ₂ molar ratio 1 : 1) in GBL, stirred overnight at 60 °C	Solution space-limited inverse-temperature crystal growth method	MAPbI ₃ single crystal	21.09	—	142
	663 mg of the synthesized FAPbI ₃ powder and 71 μ L of DMSO (molar ratio 1 : 1) in 568 μ L of DMF	Spin coating	FAPbI ₃	21.07 (certified)	80% of the original is retained after 30 days (in the dark, RH 20%) and 92% of the original is retained after 100 min (1 sun, RH 30%, without encapsulation)	143
	The 0.6 M CsPbI ₃ perovskite precursor was prepared by dissolving stoichiometric CsI, PbI ₂ , and xDMAI in a 1 : 1 : x molar ratio in DMF (x = 0.5, 0.7, 1.0 and 1.5)	Spin coating	Phenyltrimethylammonium chloride (PTACl) passivated CsPbI ₃	19.03	90% of the original is retained after 500 h continuous white light LED illumination (100 mW cm ⁻²) in a N ₂ glove without encapsulation	144
Multi cations	Mixing FAI (1 M), PbI ₂ (1.1 M), MABr (0.2 M) and PbBr ₂ (0.2 M) in anhydrous DMF : DMSO 4 : 1 (v/v). Then CsI, pre-dissolved as a 1.5 M stock solution in DMSO, was added to the mixed perovskite precursor to achieve the desired triple cation composition	Spin coating	Cs _{0.05} (MA _{0.47} FA _{0.83}) _{0.95} -Pb(I _{0.83} Br _{0.17}) ₃	21.1	The device efficiency drops from 20% to 18% where it stays relatively stable for at least 250 h (in a N ₂ atmosphere and room temperature under continuous illumination at MPP)	145
	889 mg mL ⁻¹ of FAPbI ₃ , 33 mg mL ⁻¹ of MAPbBr ₃ and 33 mg mL ⁻¹ of MACl in DMF/DMSO (8 : 1, v/v) mixed solvent	Spin coating	(FAPb ₃) _{0.95} (MAPbBr ₃) _{0.05}	22.7 (certified)	95% of the initial efficiency retained after 1370 h under 1-sun illumination at room temperature (with encapsulation)	146
	1.53 M PbI ₂ , 1.4 M FAI, 0.11 M MAPbBr ₃ , 0.5 M MACl in DMF : DMSO = 8 : 1 volume ratio	Spin coating	(FAPbI ₃) _{0.92} (MAPbBr ₃) _{0.08}	23.4	The device maintained 85% of its efficiency after 500 h under full solar illumination (AM 1.5G, 100 mW cm ⁻²) without a UV-filter at MPP tracking (with encapsulation) in ambient air	147
	0.02 M GAI, 0.09 M CsI, 0.2 M MABr, 1.1 M FAI, 1.2 M PbI ₂ , and 0.3 M PbBr ₂ in anhydrous DMF : DMSO (4 : 1, v/v) mixed solvent	Spin coating	GA _{0.015} Cs _{0.046} MA _{0.152} FA _{0.787} Pb(I _{0.815} Br _{0.185}) ₃	20.96	—	148

Table 2 (Contd.)

Perovskite precursor inks	Deposition method	Perovskite	η (%)	Stability	Ref.
<p>Additives</p> <p>1.35 M Pb^{2+} (PbI_2 and PbBr_2) and 0.6 mg mL^{-1} BrPh-ThR in the mixed solvent of DMF and DMSO (volume ratio 4 : 1)</p>	Spin coating	$(\text{FAI})_{0.81}(\text{PbI}_2)_{0.85}(\text{MABr})_{0.15}(\text{PbBr}_2)_{0.15}$; (Lewis base BrPh-ThR and Lewis acid bis-PCBM (<i>N</i> -(4-bromophenyl)thiourea))	21.7	The unsealed devices remained at 93% of the initial efficiency value in ambient air (10–20% relative humidity) after 3600 h at 20–25 °C, and dropped by 10% after 1500 h under continuous operation at 1-sun illumination and 55 °C in nitrogen with MPPT	149
<p>Two-step solution process:</p> <p>Mixing 1.3 M PbI_2 in DMF/DMSO (9 : 1, v/v) with 5% CsI</p> <p>Note: for the NaX-containing perovskite film, NaX was added</p> <p>Mixing 0.12 M MAI, 0.05 M MABr, 0.07 M MACl, and 0.23 M FAI in IPA</p>	Spin coating	$(\text{Cs}_{0.05}\text{FA}_{0.54}\text{MA}_{0.41})\text{-Pb}(\text{I}_{0.98}\text{Br}_{0.02})_3$; (NaF)	21.46 (21.43% certified)	The unencapsulated device retained over 90% of its initial PCE after 1000 h under MPP tracking and continuous light irradiation with a white LED lamp, 100 mW cm^{-2} in a nitrogen atmosphere	150
<p>1.139 mg mL^{-1} FAPbI₃ in a mixture of DMF and DMSO (4 : 1, v/v). MACl was added in the range 0–50 mol%</p>	Spin coating	MACl additive (40 mol%) was applied in FAPbI ₃	24.02 (certified 23.48%)	The stability at RT showed 90% retention of the device's initial performance over 1200 h. The 40 °C thermal stability test showed 90% retention of the device's initial performances over 300 h	56
<p>573.0 mg PbI_2, 197.5 mg MAI, and 13.70 mg 5-AVAI in 1.0 mL GBL</p>	Slot-die coating	5-AVAI in MAPbI ₃	12.87	—	106
<p>Lead free</p> <p>Mixing the GeI₂ doped $\text{FA}_{0.98}\text{EDA}_{0.01}\text{SnI}_3$ and GeI₂ doped $\text{EA}_{0.98}\text{EDA}_{0.01}\text{SnI}_3$ precursor solution at a molar ratio of $x = 0, 0.05, 0.1, 0.2$</p> <p>First dissolving 129 mg FAI, 39.7 mg MAI, 372.5 mg SnI_2, and 15.7 mg SnF_2 in 875 μL DMF : 125 μL DMSO. Meanwhile, 55.8 mg GeI₂ was dissolved in 200 μL DMF. Different concentrations of GeI₂ (5 mol% and 10 mol%) were subsequently added into the bulk precursor</p> <p>Dissolving AgI, BiI₃, and $[\text{Bi}(\text{S}_2\text{CAR})_3]$ at the desired ratios in a mixed solvent system (DMSO : DMF : HI, 3 : 1, v/v), to which HI (0.12 vol.) was slowly added dropwise</p>	Spin coating	$(\text{FA}_{0.9}\text{EA}_{0.1})_{0.98}\text{EDA}_{0.01}\text{SnI}_3$. 5 mol% germanium doped $\text{FA}_{0.75}\text{MA}_{0.25}\text{Sn}_{1-x}\text{Ge}_x\text{I}_3$	13.24 7.9	— 90% of the unencapsulated device's initial efficiency retained over 600 min under constant illumination under ambient conditions	127 151
<p>Dissolving AgI, BiI₃, and $[\text{Bi}(\text{S}_2\text{CAR})_3]$ at the desired ratios in a mixed solvent system (DMSO : DMF : HI, 3 : 1, v/v), to which HI (0.12 vol.) was slowly added dropwise</p>	Spin coating	Ag ₃ BiI _{5.92} Sb _{0.04}	5.44	When stored in air at room temperature ($\approx 22\text{--}24$ °C) at RH of 40% under diffuse light, the device showed a promising stability and retained more than 90% of the initial PCE after 45 days	152



4. Research and development trends for advanced scalable precursor inks

Motivated by the successes achieved from solution processed lab-sized PSCs with traditional perovskite precursor inks, efforts have also been made towards scaling up PSC technology on a variety of substrates by designing novel precursor inks with key characteristics as shown in Fig. 3.^{153–158} In this regard, spin coating has remained the foremost choice for perovskite precursor ink deposition, as well as for the initial scaling up demonstrations of PSC technology (Table 3).^{159–161}

Among these initial pioneering efforts, Matteocci *et al.* demonstrated the first n-i-p structure-based solid-state monolithic module¹⁶² with organometal halide perovskite $\text{CH}_3\text{NH}_3\text{-PbI}_{3-x}\text{Cl}_x$ ink, produced by dissolving methylammonium iodide $\text{CH}_3\text{NH}_3\text{I}$ and lead chloride PbCl_2 in a *N,N*-dimethylformamide (DMF)-based solvent. The ink was spin-coated at 2000 rpm for 40 s over the nanocrystalline TiO_2 -based electron transport layer (ETL) in air and successively heated at 120 °C for 45 minutes to obtain the final crystalline structure. The fabricated modules exhibited similar (~5.1%) conversion efficiencies when tested with two types of hole transporting layer (Spiro-OMeTAD and poly(3-hexylthiophene-2,5-diyl) (P_3HT) polymer) and thermally evaporated gold (Au)-based contact layers over an active area of 16.8 cm².

As a further advancement, Seo *et al.* reported an inverted (p-i-n) structure-based module design with a perovskite ink composed of $\text{CH}_3\text{NH}_3\text{I}$ and PbI_2 , which were stirred in a mixture of dimethyl sulfoxide (DMSO) and γ -butyrolactone (GBL)-based solvents at 60 °C for 12 h.¹⁶³ The ink was deposited onto a PEDOT:PSS/ITO substrate by a consecutive two-step spin-coating process at 1000 and 4000 rpm for 20 and 60 s, respectively. The toluene in the final spin-stage was dripped onto the substrate during spin coating. The perovskite precursor-coated

substrate was then dried on a hot plate at 100 °C. The fabricated module with the phenyl-C61-butyric acid methyl ester (PCBM)-based ETL and Al coated contact layer showed an impressive conversion efficiency (8.7%) over a far larger active area (60 cm²) when tested under full sunlight illumination.

In addition to these earlier reports of precursor ink development with the spin coating-based material deposition method, advanced single- and two-step processable precursor inks have also been developed and tested with more established scalable approaches (as depicted in Fig. 4), such as blade coating and slot-die coating to fabricate large area perovskite-based solar modules.^{164–167} For example, Razza *et al.* used a two-step approach where the pre-heated PbI_2 precursor ink was first coated on ETL layers using a blade coating method followed by rapid evaporation of its solvent (DMF) through an airflow.¹⁶⁴ Next, the perovskite crystalline layer was achieved by directly dipping the substrates containing PbI_2 -coated ETL layers into the MAI solution with different dipping times. The perovskite modules fabricated with this approach exhibited 4.3% conversion efficiency under full sunlight illumination over an active area of 100 cm².

Deng *et al.* introduced the novel concept of adding a surfactant to the perovskite precursor ink and demonstrated impressive control over the crystallization morphology of the large-area perovskite-based light absorbing layer when deposited with a blade coating scheme.¹⁶⁸ The fabricated modules showed very high (~15%) conversion efficiency when tested under full sunlight illumination with a p-i-n device design. Similarly, Giacomo *et al.* demonstrated slot-die coating of a precursor solution for a perovskite layer, which was prepared under room temperature conditions in an inert N_2 environment.¹⁶⁹ The solution was prepared by dissolving 1 mM of lead compounds and 3 mM of $\text{CH}_3\text{NH}_3\text{I}$ in 1 mL of DMF solvent. The solution was stirred for at least 10 minutes before final use. Slot-die coating of this perovskite precursor ink was performed inside a glovebox, which allowed deposition of a wide range of thicknesses. The wet layer thickness of the perovskite was in the range of 3–4 μm to get a dry layer thickness of 350 nm. With these optimizations, a 12.5 × 13.5 cm² module with power conversion efficiencies above 10% over an active area of approximately 152 cm² was achieved in the planar n-i-p configuration.¹⁶⁹

Recently, slot-die coating of a novel perovskite precursor ink containing MAI, PbI_2 and 5-ammonium valeric acid (5-AVAI) in GBL (tested with the drop-casting¹⁷⁰ and inkjet printing methods^{107,108}) has also been demonstrated.¹⁰⁶ The fabricated triple mesoscopic printable perovskite solar modules showed an impressive conversion efficiency of 12.87%, which was attained over an active area of 60.08 cm² under full sunlight illumination.¹⁰⁶

Similar scalable demonstrations of organic-inorganic lead halide perovskite-based novel inks with the techniques discussed above have also been utilized to scale up PSC technology on flexible substrates. Flexible modules in various device designs have been developed and reported both by research labs and commercial players in recent years.^{171–174} Therefore, there is promising evidence of successful fabrication of the novel organic-inorganic lead halide-based precursor ink formulations for achieving module production, where several device

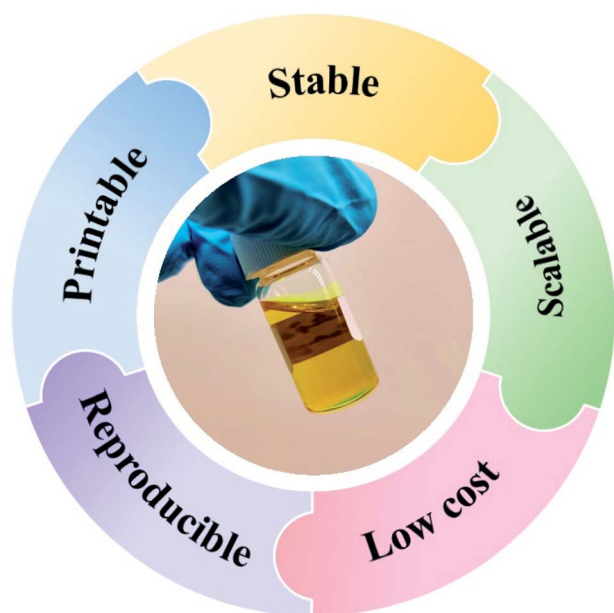


Fig. 3 General characteristics of an efficient perovskite precursor ink.



Table 3 Summary of the scalable device configurations developed in recent years with novel perovskite precursor inks^a

Ink recipe (PVK)	Method	Device structure	Active area (cm ²)	Module area (cm ²)	η (%)	Ref.
MAI and PbCl ₂ (molar ratio 3 : 1) in DMF	Spin coating	Glass/FTO/TiO ₂ /PVK/HTM/Au	16.8	5 × 5	5.1 ^b	162
MAI : PbI ₂ (molar ratio 1 : 1) in GBL : DMSO (3 : 7, v/v)	Spin coating	Glass/ITO/PEDOT:PSS/PVK/PCBM/LiF/Al	60	10 × 10	8.7 ^c	163
PbI ₂ in DMF (285 mg mL ⁻¹) MAI solution in IPA (10 mg mL ⁻¹)	Spin coating and dipping	Glass/FTO/c-TiO ₂ /graphane + m-TiO ₂ /Go-Li/PVK/Spiro/Au	50.6	10 × 10	12.6 ^b	175
PbI ₂ : MAI (1 : 1 molar ratio) in GBL 5-AVA-I added to the perovskite solutions in a molar ratio of 1 : 20 (AVA-I to MAI)	Drop casting	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/PVK	70	10 × 10	10.75 ^b	176
1.2 M MAI and 1.2 M PbI ₂ dissolved in GBL For 2D perovskite: 1.2 M AVAI and 1.2 M PbI ₂ dissolved in GBL	Drop casting	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/PVK	47.6	10 × 10	11.16 ^b	177
Mixing 0.07 g 5-AVAI, 4.61 g PbI ₂ and 1.59 g MAI in 10.5 mL GBL	Drop casting	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/PVK	49	10 × 10	10.4 ^b	170
PbI ₂ and MAI (1 : 1 molar ratio) in GBL prepared by adding 5-AVAI to obtain a 3% molar ratio between 5-AVAI and MAI	Drop casting	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/PVK	198	435.6 (aperture)	6.3 ^b (stabilized) (after 2 months since fabrication)	178
Blade coating of PbI ₂ in DMF (330 mg mL ⁻¹) Dipping into MAI solution (in IPA) (10 mg mL ⁻¹)	Blade coating	Glass/FTO/c-TiO ₂ /mp-TiO ₂ /PVK/P ₃ HT/Au	100	16 × 11	4.3 ^b	164
30 wt% and 45 wt% equimolar ratio of MAI and PbI ₂ precursors with different amounts of MAI additive in a mixed solvent (NMP/DMF 9 : 8, v/v)	Blade coating	Glass/FTO/c-TiO ₂ /PVK/Spiro-OMeTAD/Au	11.09	12.6	13.3 ^b (stabilized)	179
0.8 M MAPbI ₃ in DMF (496 mg mL ⁻¹) and 0.25 M LP surfactant in DMF (0.2 mg mL ⁻¹)	Blade coating	Glass/ITO/PTAA/PVK/C ₆₀ /BCP/Cu	57.2	13 × 8.5	14.6 ^d (stabilized)	180
276.7 mg PbI ₂ in 1 mL DMF or mixed solvent (DMF : additives) (0.6 M) The solution mixture of FAI : MABr : MAI (60 mg : 6 mg : 6 mg in 1 mL IPA) was bladed onto the PbI ₂ films	Blade coating	Glass/FTO/SnO ₂ /PVK/Spiro-OMeTAD/Au	16.54% for 5 × 5 cm ² perovskite solar modules, and 13.32% for 10 × 10 cm ² modules			167
The solid MAPbI ₃ crystals changed to liquid <i>via</i> MA gas-assisted solid-liquid transition. The viscous liquid was diluted with ACN in a weight ratio of viscous liquid to ACN = 60 : 40	D-bar coating	Glass/FTO/SnO ₂ /PVK/Spiro-OMeTAD/Au	10 × 10	10 × 12	Many PCEs ^b were reported (~16-18%)	181
PbI ₂ in DMF (0.7 M, 322 mg mL ⁻¹) MAI (10 mg mL ⁻¹) in IPA	Slot-die coating	PET/ITO/ZnO/PVK/P ₃ HT/Ag	5 × 8	10 × 10	—	165
40% MAPbI ₃ in DMF	Slot-die coating	Glass/FTO/ZnO/PVK/home-made carbon paste	17.6	5 × 5	10.6 ^b	182



Table 3 (Contd.)

Ink recipe (PVK)	Method	Device structure	Active area (cm ²)	Module area (cm ²)	η (%)	Ref.
1 mM of lead compounds PbCl ₂ and Pb(CH ₃ COO)·3H ₂ O at a molar ratio of 1 : 4 and 3 mM of MAI per 1 mL of DMF	Slot-die coating	Glass/ITO/c-TiO ₂ /PVK/Spiro-OMeTAD/Au	151.88	12.5 × 13.5	11.1 ^b	183
573.0 mg PbI ₂ , 197.5 mg MAI, and 13.70 mg 5-AVAI in 1.0 mL GBL	Slot-die coating	Glass/FTO/m-TiO ₂ /ZrO ₂ /carbon/PVK	60.08	80.55 (aperture area)	12.87 ^b	106
0.8 M MAPbI _{3-x} Cl _x in DMF/GBL mixed solvent	Spray coating	Glass/FTO/TiO ₂ /PVK/PTAA/Au	40	10 × 10	15.5 ^b	184
0.15 M Cs _{0.17} FA _{0.83} PbI ₃ in DMF : DMSO (1 : 2, v/v)	Spray coating	Glass/ITO/NiO/PVK/C ₆₀ /BCP/Au	5.9	—	15.5 ^b	185
0.53 g of PbI ₂ , 0.19 g of MAI and 0.0176 g of 5-AVAI in 1 mL of GBL	Ink-jet printing	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/printed PVK	—	10 × 10 (18 cells)	—	107
1.2 M PbI ₂ , 1.2 M MAI and 0.06 M 5-AVAI in 1 mL of GBL	Ink-jet printing	Glass/FTO/TiO ₂ /ZrO ₂ /carbon/printed PVK	0.64	1.5	9.1 ^b	108
<i>l</i> - α -Phosphatidylcholine and methylammonium hypophosphite were added into \approx 1.45 M MAPbI ₃ /2-ME solution at a concentration of \approx 0.3 mg mL ⁻¹ and \approx 0.15 vol%, respectively	Blade coating	Willow glass/ITO/PTAA/PVK/C ₆₀ /BCP/Cu	Geometric fill factor 92%	42.9 (aperture area)	15.86 ^d	186
1.26 mM PbI ₂ , 1.26 mM FAI, 0.06 mM MAPbBr ₃ and 0.5 mM MAI in DMF/DMSO (8 : 1 v/v) mixed solvent	Spin coating	PEN/ITO/np-SnO ₂ and porous-ZSO (Zn ₂ SnO ₄)/PVK/Spiro-OMeTAD/Au	PCE of 15.5%, 12.9% and 11.8% on an aperture area ^d of 100 cm ² , 225 cm ² and 400 cm ² , respectively			187
1.3 M organic cation (0.85 FAI and 0.15 MABr) and 1.4 M mixture of metal lead salts (0.85 PbI ₂ and 0.15 PbBr ₂) in a solvent mixture of DMF/DMSO (4 : 1, v/v)	Spin coating	PET/ITO/SnO ₂ /PVK/Spiro-OMeTAD/Au	16.07	30	15.22 ^b	188
460 mg mL ⁻¹ PbI ₂ in DMF 40 mg mL ⁻¹ MAI in 2-propanal	Spin coating	Cu foil/CuI/PVK/ZnO/Ag	0.8	—	—	189

^a PVK = perovskite. ^b PCE based on active area. ^c Not mentioned. ^d PCE based on aperture area.

designs can be potentially achieved on both rigid and flexible-based substrates according to the targeted applications. Table 3 provides a summary of the recently reported scalable device configurations of perovskite solar modules designed using scalable perovskite precursor ink formulations on rigid glass and flexible substrates.

5. Forthcoming directions in precursor inks, technological research, and development of next-generation PSC technology

The progressive research and development of PSCs over the past decade provide future directions and guidelines to overcome

the remaining key challenges necessary for reaching commercial breakthroughs imagined for this efficient PV technology as envisioned in Fig. 5.

Among them, the toxicity and environmental impact of traditional lead-based precursor inks pose a challenge for the safe deployment of PSCs. In this regard, not only should the safety protocols but also the effective recycling procedures be chosen for the safe removal of hazardous and toxic materials such as PbI₂ with high yield from the installed panels of PSC technology after their end of life. Interestingly, notable progress has been reported in many scientific reports where effective recycling of PbI₂, methylammonium iodide (MAI) and expensive contact layers has been successfully demonstrated for the lab-sized PSCs.^{190–192} These proof-of-concept based striking demonstrations provide pathways in designing effective



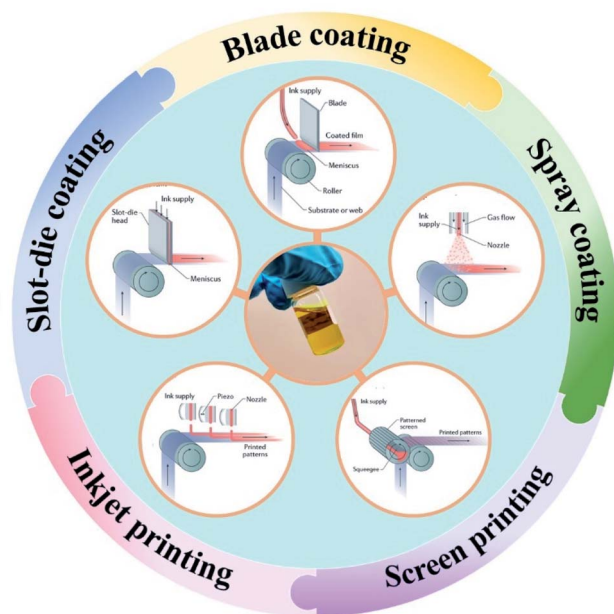


Fig. 4 Some of the popular fabrication schemes opted to produce large area perovskite solar cells and modules. Reproduced from ref. 158 with permission.

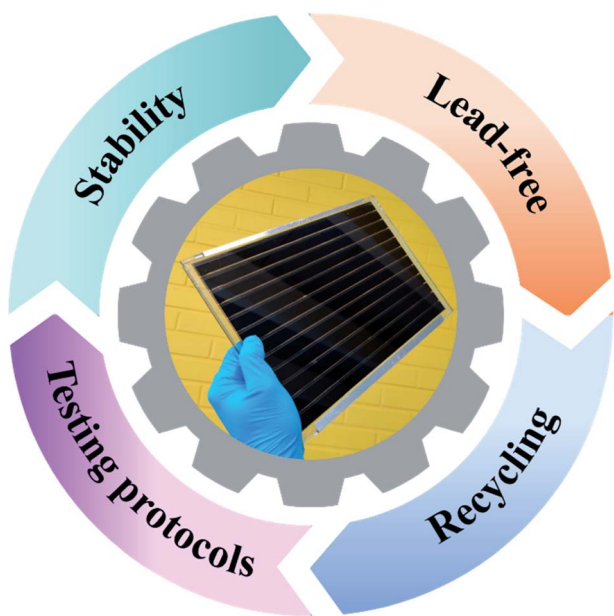


Fig. 5 Summary of the key futuristic directives envisioned for the emerging perovskite solar cell technology.

recycling protocols for large area perovskite solar modules, which may significantly suppress the environmental concerns, which have been heavily debated during the emergence of this low-cost PV technology.^{193–195} Moreover, the abundance, availability, cost and sustainability of potential lead compound substitutes and their respective precursor inks must be further investigated to justify their bulk-scale usage to produce the so-called large area eco-friendly perovskite-based PV panels.¹⁹⁶

On the other hand, despite the growing number of reports on achieving stable photovoltaic performance under various environmental conditions while incorporating advanced perovskite precursor inks,^{103,104,197} realistic assessments of PSC device stability based on recently suggested stability testing protocols¹⁹⁷ have yet to be established. Thus far, there are several reports of either stable performances or degradation of initial efficiencies in various device designs of PSCs containing novel precursor inks, using the suggested stability testing protocols for PSC technology. This again raises concerns due to the uncertainty in deploying this emerging PV technology in many suitable applications.

In addition, new consensus statements need to be developed to address advanced stability testing protocols from the perspective of precisely controlled indoor ecological conditions and some severely cold outdoor climatic conditions, which have not been specifically covered in the current stability testing protocols.⁸ The possible upcoming deployments of PSC technology in batteryless and maintenance-free IoT devices^{198–200} inside modern buildings with precisely controlled ecological conditions or stand-alone communication gadgets in severely cold outdoor climatic conditions call for new testing standards. Notably, degradation rates could vary under such environmental conditions, but these are not known or have rarely been reported to date.

Interesting research questions may provide future research and development directives towards the forthcoming generation of PSC devices over the next decade. Considering the progress that has already been made, it is expected that future technological developments would resolve key research problems of achieving eco-friendly and robust PSC technology with certified integration in vast applications.

6. Summary and conclusions

The development of progressive precursor inks reviewed in this work provides evidence for their contributions towards achieving striking milestones such as high-performance or versatile configuration designs and highlights unique integration possibilities of emerging PSC technology. Moreover, the intelligent precursor inks designed with novel materials reported in the recent past have offered pathways for scalability over both rigid glass and flexible polymer foil-based substrates. As a result, activities related to commercializing PSC technology have greatly accelerated as its integration in smart sensors, charging stations and BIPV based applications continues to be demonstrated.²⁰¹

Nevertheless, additional materials screening is needed for the development of the next generation of novel precursor inks of the perovskite light absorbing layer in order to meet the growing demand, especially from the construction industry where there is keen interest in aesthetically appealing solar panels for smart buildings and nature-friendly ecological architecture.^{202–209} This demand may potentially be met by creating precise and flexible patterning that allows individual PSCs to be assembled into aesthetic structures based on solar modules for energy generation.



The current methods such as slot-die coating, spraying or blade coating techniques offer limited flexibility, and have mostly been used to produce serially connected pattern-based modules in traditional n-i-p or p-i-n based device configurations.^{188,210,211} Notably, alternative schemes such as drop on demand inkjet printing may provide more freedom of design, since advanced precursor inks of both perovskite-based materials and other active materials can be fabricated to create individual PSCs. These PSCs can be designed with aesthetically attractive arbitrary patterns, which may be intelligently connected to create electricity at targeted watt scales.

Surprisingly, except for the few demonstrations of printing individual active layers^{212–215} (including the perovskite-based light absorbing layer^{107,216}), inkjet printing has been rarely reported in fabricating complete device designs of PSC technology, despite knowing its potential for fabricating large area-based active layers at high speed. In this regard, one of the most critical limitations realized in producing active layers through inkjet printing is when the nozzles choke the cartridges, which severely limits drop volumes (1–10 picolitres). This in turn makes it difficult to print nanoparticle-based solution processable inks with the necessary precision and resolution. Moreover, the delicate cartridges frequently exhibit chemical incompatibility and often react with the fabricated inks that are composed of novel materials and solvents, thus making it difficult to achieve the desired patterns with ease and accurate precision.^{217–220}

These key challenges call for the development of advanced precursor inks to be used with high chemical compatibility, not only to achieve precisely patterned perovskite-based light absorbing layers but also for other active layers to produce fully printed PSC technology with pattern flexibility. One of the potential precursor inks to achieve a perovskite light absorbing layer with high precision has been demonstrated utilizing an inkjet printing technique to produce carbon-based printable perovskite solar cells.¹⁰⁷ The precursor ink showed great compatibility during inkjet printing when fabricated with a novel additive (5-ammonium valeric acid iodide; 5-AVAI), which significantly slows the perovskite crystallization formation and allows the fabricated ink to pass through the inkjet nozzles without reacting with them. Thus far, this stable device configuration^{103,105} of PSCs has been demonstrated with schemes such as inkjet printing and screen-printing methods. There is immense potential to achieve further module fabrication with customized designs to generate appealing patterns based on individual cells as discussed earlier in this section.²²¹

Hence, a transformation from the first generation of serially connected carbon-based printable perovskite solar modules^{106,178} to aesthetically appealing solar panels can be anticipated, which could provide guidelines to produce the next generation of traditional n-i-p and p-i-n configuration-based PSCs, especially when considering evolved design principles. This could assuredly enable realistic outcomes such as the rapid deployment of PSCs worldwide and the resulting positive influence on climate change expected to arise from this promising photovoltaic technology.

Author contributions

C. M. T. K contributed with manuscript planning and communication, writing Sections 1, 2 and 4, drafting tables and figures, screening, and furnishing the manuscript text and references. L. A and M. O contributed with writing and drafting Sections 3 and 5. S. G. H. supervised the research work and contributed with funding acquisition, outline drafting and writing sections, reviewing, and editing the overall manuscript.

Conflicts of interest

The authors have no conflicts to declare.

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